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Suppressing Ag₂Te nanoprecipitates for enhancing thermoelectric efficiency of AgSbTe₂†

 Zichen Gong,^{‡a,b} Kivanc Saglik,^{‡a,c} Jing Wu,^{id a,b} Ady Suwardi^{id a,b} and Jing Cao^{id *a}

Thermoelectrics are a class of materials that provide interconversion between heat and electricity, with desirable traits such as low thermal conductivity and low electrical resistivity. AgSbTe₂ has emerged as one of the leading materials in recent years due to its ultra-low thermal conductivity. However, one major hindrance in undoped AgSbTe₂ is its high electrical resistivity and low Seebeck coefficient due to the presence of Ag₂Te nanoprecipitates. In this work, we leverage on the combination of an off-stoichiometric composition and a non-equilibrium process to simultaneously enhance the properties of AgSbTe₂ and its thermoelectric device performance. Microscopically, the Ag₂Te-deficient starting composition combined with a non-equilibrium thermal process suppresses the Ag₂Te nanoprecipitates in the material. In addition, it is evident from the density functional theory (DFT) electronic structure that Ag₂Te deficiency results in a smaller lattice and higher density-of-states near the Fermi level, which simultaneously lower the electrical resistivity and increase the Seebeck coefficient. As a result, *zT* as high as 1.7 was achieved at 573 K. Additionally, when combined with a high room temperature *zT* of 0.75, a power conversion efficiency of 7.3% was achieved at a ΔT of 290 K. Crucially, the strategy in this work can inspire application in other ABX₂ material systems to achieve improved thermoelectric performances.

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Introduction

The rapidly changing climate, primarily due to the excessive consumption of energy and electricity, has created an urgent need for the development of sustainable materials to meet our renewable energy requirements. Among various renewable energy technologies, thermoelectrics hold a unique position because of their remarkable capacity to convert heat into electricity and *vice versa*. This distinctive capability makes them particularly valuable. One notable application of thermoelectrics lies in their capacity to harness waste heat and convert it into electricity, offering an attractive complement to technologies such as photovoltaics. By capturing and utilizing the energy that would otherwise turn into waste heat, thermoelectrics enhance overall energy conversion efficiency. Moreover, thermoelectrics also excel as solid-state coolers, used for cooling purposes in heat-sensitive devices,

which enhances the reliability and performance of existing technologies such as batteries and modern electronics. Despite their immense potential, the widespread adoption of thermoelectric technologies faces limitations, primarily in terms of their performance and overall cost.^{1–6}

Thermoelectric materials are typically characterized using the figure-of-merit *zT*, where *zT* is calculated as the product of three key parameters: the Seebeck coefficient (*S*), electrical conductivity (σ), and thermal conductivity (κ). To enhance the efficiency of thermoelectric materials and achieve higher *zT* values, various strategies including engineering scattering mechanisms, pursuing band convergence, and implementing nanostructuring techniques are employed.^{7–10} Each of these approaches aims to optimize the interplay between *S*, σ , and κ , ultimately leading to improved thermoelectric performance.^{11–15} Furthermore, researchers have explored a range of strategies to effectively reduce lattice thermal conductivity, including the introduction of various types of defects, the manipulation of lattice anharmonicity, the utilization of complex structures, and the application of lattice strain.^{16–24} In recent years, several standout materials have risen to prominence as top performers in the field of thermoelectrics. Notable among these are compounds based on GeTe, SnSe, Cu₂Se, Mg₃Sb₂, and AgSbTe₂.^{25–37}

In particular, AgSbTe₂ can potentially be used for both near room temperature and medium temperature applications, making it a versatile high performance thermoelectric

^aInstitute of Materials Research and Engineering, Agency for Science, Technology and Research, 138634, Singapore. E-mail: cao_jing@imre.a-star.edu.sg

^bDepartment of Materials Science and Engineering, National University of Singapore, 117575, Singapore

^cSchool of Materials Science and Engineering, Nanyang Technological University, 639798, Singapore

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‡These authors contributed equally to this work.

material.^{38–46} Due to the high intrinsic Seebeck coefficient and low intrinsic thermal conductivity, high thermoelectric performance can be achieved in AgSbTe₂ even in the absence of doping. Despite these advantages, this compound has been historically plagued by several major issues including cation disorder, performance instability, and the presence of n-type Ag₂Te nanoprecipitates resulting from phase decomposition. Recent breakthrough has been accomplished in achieving cation ordered AgSbTe₂ via the process of Cd doping. However, due to the highly toxic nature of Cd, alternatives to this strategy are sought after. In this work, we leverage on off-stoichiometry and non-equilibrium thermal annealing to improve the thermoelectric performance of undoped AgSbTe₂-based compounds. As shown in Fig. 1, samples with different levels of Ag₂Te-deficiency were prepared at different annealing temperatures. As a result, a high *zT* of 1.7 at 573 K was achieved in Ag_{0.91}Sb_{1.09}Te_{2.09} annealed at 773 K. The origin of the performance enhancement can be ascribed to the combination of suppression of Ag₂Te nanoprecipitates and the lowering of lattice parameters in Ag₂Te-deficient samples. This results in simultaneously a higher Seebeck coefficient and lower electrical resistivity, which drastically enhances the power factor and therefore the ultimate *zT* across a wide temperature range. Furthermore, a high power conversion efficiency of 7.3% at ΔT of 290 K was achieved in an Ag_{0.91}Sb_{1.09}Te_{2.09} single-leg with Fe and Cu encapsulation. The off-stoichiometry and non-equilibrium strategy reported in this work can be further applied to other ABX₂ material systems to achieve improved thermoelectric performances.

Methods

Experimental

Stoichiometric batches of Ag (99.99%) powders, Sb (99.99%) shots, and Te (99.99%) chunks from ZhongNuo Advanced



Jing Cao

*Dr Jing Cao is currently a project leader at the Institute of Materials Research and Engineering (IMRE), Agency for Science, Technology and Research (A*STAR), Singapore. She received her Ph.D. degree in Materials Science at the University of Cambridge in 2019. Her research is primarily focused on electronic transport, specifically in thermoelectrics and ferroelectrics, and bulk photovoltaic effects. Besides experiments, she also does first principles calculations to sup-*

plement her work. Outside research, she is active in outreach activities by serving on the scientific committee for Singapore Junior Chemistry Olympiad organized by the SNIC (Singapore National Institute of Chemistry).

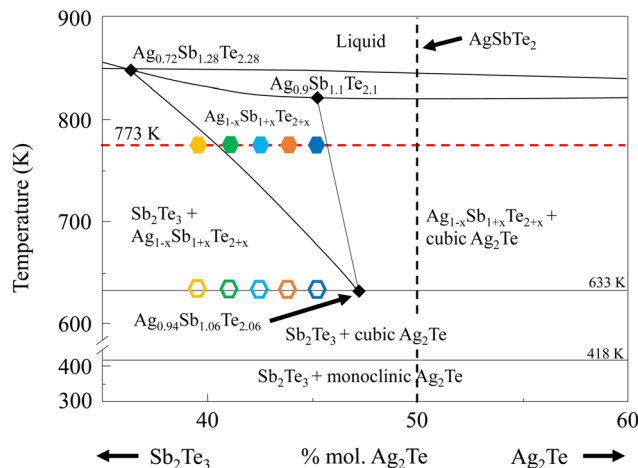


Fig. 1 Phase diagram of AgSbTe₂, plotted as a fractional %mol of Sb₂Te₃ and Ag₂Te. Symbols of distinct colors represent the composition and processing temperature of the respective samples used in this study. Filled hexagon symbols represent samples annealed at 773 K for 24 hours followed by air quenching. Hollow hexagon symbols represent samples without such an annealing process. The figure is replotted from ref. 47.

Material Technology Co. Ltd were precisely weighed to a total weight of 12 grams. Subsequently, these precisely measured mixtures were sealed within quartz tubes under a vacuum environment, maintaining a pressure of 10⁻³ Pa. These sealed tubes were then placed into a rotary furnace and subjected to a gradual heating process, increasing the temperature to 1273 K at a rate of 5 K per minute. Furthermore, the molten samples underwent a homogenization process, achieved by automatic rotation at a constant speed of 2 rpm for a duration of 5 hours. This rigorous and controlled procedure was employed to ensure the uniformity and consistency of the resulting material. To prevent excessive phase decomposition during cooling, the furnace was switched off at 1273 K and allowed to cool down in 3 hours. The obtained ingots were ground into powders, followed by spark plasma sintering (SPS) at 673 K at 50 MPa for 5 minutes. Subsequently, the obtained pellets were cut into various shapes and subjected to thermal processing for subsequent measurements. 5 samples of different compositions (represented as filled hexagon symbols) were annealed at 773 K for 24 hours, followed by air-quenching, whereas the remaining 5 samples of the same composition (represented as hollow hexagon symbols) did not undergo such a process.

Electronic transport properties were measured using a ZEM-3 from ULVAC to determine the Seebeck coefficient and electrical resistivity. For this purpose, the samples were cut into rectangular bars, approximately 2 mm × 2 mm × 12 mm height. The thermal transport properties were measured using a laser LFA 457, NETZSCH, with a laser energy of 10 J. LFA samples are typically thin rectangular samples measuring 6 mm × 6 mm × 1 mm thickness. To calculate thermal conductivity, a heat capacity of 0.217 J g⁻¹ K⁻¹ at room temperature to 0.251 J g⁻¹ K⁻¹ at 673 K was used, based on earlier refer-

ences.³⁸ Single-leg power conversion efficiency measurements were carried out using the commercial system NYDM, a TE device property measurement system from Shanghai Nengyi Semiconductors, Ltd. To ensure impedance matching, a sample with a small cross section of around 1.5 mm × 1.5 mm × 12 mm was prepared. Microstructural analysis was performed using a field-emission scanning electron microscope (FESEM, JEOL JSM7600 and JSM-7800F PRIME) equipped with an energy-dispersive X-ray spectrometer (EDS) detector for elemental analysis. TEM was collected using a Talos F200X. X-ray diffraction (XRD) patterns were collected using a Bruker D8-Advance X-ray powder diffractometer with Cu K α radiation.

Computational

The density functional theory (DFT) calculations were carried out using QUANTUM ESPRESSO,^{48–50} with a generalized gradient approximation (GGA)⁵¹ to the exchange–correlation energy functional as parametrized by Perdew, Burke, and Ernzerhof (PBE),⁵² within the projector augmented wave formalism. During calculation, spin–orbit interaction was not included as it has been reported to have little effect on the band structure in AgSbTe₂.⁵³ We chose the ordered crystal structure of AgSbTe₂: the AF-IIb (*F* $\bar{3}$ *dm*) phase, with the experimental lattice parameter *a* = 12.305 Å, 64 atoms in the conventional cell and 16 atoms in the primitive cell.^{54,55} The cut-off energies for the wave functions and charge density expansion in the plane wave basis sets were truncated at 65 and 260 Ry, respectively. Brillouin zone integration was sampled on a uniform mesh of 4 × 4 × 4 *k*-points. The defect structure Ag_{0.94}Sb_{1.06}Te_{2.06} was built by replacing 2 Ag atoms with 1 Sb

atom and 1 Te atom with the lowest energy. All the structures were fully relaxed before proceeding for band structure and density of states calculations.

Results and discussion

Fig. 2(a) shows the room temperature XRD analysis of all samples. Notably, no Ag₂Te peak can be observed, which is in contrast to typical scans from AgSbTe₂ samples.²⁸ This observation can be ascribed to the phase diagram in Fig. 1 which shows that with increasing annealing temperature above 633 K, Ag₂Te-deficient samples tend to form a stable phase of Ag_{1–x}Sb_{1+x}Te_{2+x}. It is worth noting that a minute amount of the Sb₂Te₃ phase is expected to be present in Ag_{0.79}Sb_{1.21}Te_{2.21}, despite the absence of peaks in the XRD scan. HRTEM (high resolution transmission electron microscopy) scan of the area of interest shown in Fig. S1† confirms the presence of Sb-rich precipitates in the sample. To further analyze the effect of Ag₂Te deficiency on the lattice parameters of the resulting sample, Rietveld refinement was used to derive the lattice volume for all samples, as shown in Fig. 2(b). Evidently, a decreasing trend of lattice parameters can be observed with increasing Ag₂Te-deficiency. This is not surprising considering the larger ionic radii of Ag⁺ (126 pm) than that of Sb³⁺ (76 pm). To further confirm the absence of Ag₂Te nanoprecipitates, SEM EDS scan was conducted on all samples. Fig. 2(c) shows the homogeneous scan of Ag_{0.91}Sb_{1.09}Te_{2.09} with the absence of the Ag-rich or Sb-deficient region. In contrast, samples without thermal annealing and quenching tend to show mixed phases, as shown in Fig. S2 and Table S1.†

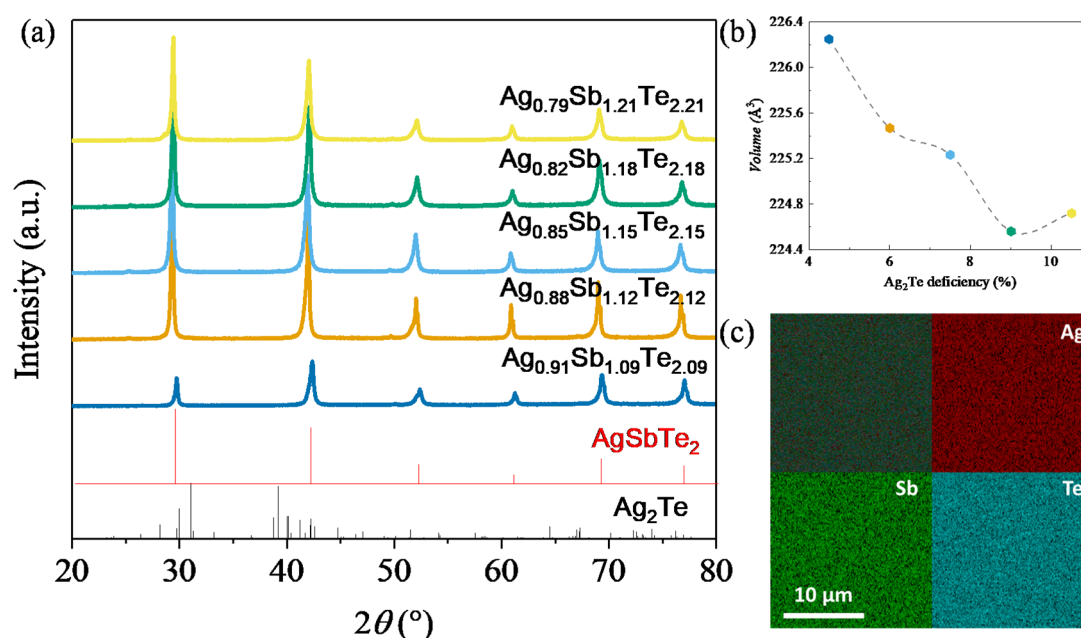


Fig. 2 Structural analysis of samples with various levels of Ag₂Te deficiency. (a) XRD (X-ray diffraction) showing the absence of Ag₂Te peaks in all samples. (b) Lattice parameters of all samples showing a decreasing trend with increasing Ag₂Te deficiency. (c) SEM EDS analysis showing a homogeneous distribution and absence of Ag-rich nanoprecipitates.

To systematically study the impact of different compositions and processing conditions on the thermoelectric properties, first, it is worth noting the Seebeck coefficient shown in Fig. 3(a). With the exception of $\text{Ag}_{0.88}\text{Sb}_{1.12}\text{Te}_{2.12}$, all other samples show a lower magnitude of Seebeck coefficients than pristine AgSbTe_2 . This can be ascribed to the lower electrical resistivity, as shown in Fig. 3(b). It is well known that the Seebeck coefficient and electrical resistivity are positively correlated *via* carrier concentrations.¹ In addition, samples annealed at 773 K generally show lower electrical resistivity than the non-annealed samples. This can be partially explained by the more ordered crystal structure due to atom diffusion at elevated temperatures and improved defect conditions within the materials, that is better crystallinity after annealing, which helps to improve the carrier mobility. In addition, annealing at 773 K helps to ensure a high purity single phase with no secondary phase precipitates, which also results in higher carrier mobility. As a result, most of the annealed samples show higher weighted mobility (Fig. 3(c)). It is worth noting that the hump in weighted mobility in Fig. 3(c) for pristine AgSbTe_2 can be associated with the presence of Ag_2Te precipitates, which has a phase transition temperature around 400–450 K. Consequently, a drastically higher power factor can be observed in most

Ag_2Te deficient samples, especially for annealed $\text{Ag}_{0.91}\text{Sb}_{1.09}\text{Te}_{2.09}$ which has a power factor of more than $1600 \mu\text{W m}^{-1} \text{K}^{-2}$ throughout a wide range of temperatures (300 K–623 K), more than double that of the pristine AgSbTe_2 sample. It is worth noting that there was no observable change in the structural properties of all samples after the transport property characterization, as shown in Fig. S3.† This is presumably due to the transient nature of the thermal cycle during testing. Furthermore, this is further supported by the stability of the thermoelectric properties upon repeated cycles of measurement, as shown in Fig. S4.†

To further elucidate the experimentally observed electronic transport properties in Fig. 3, comparison of density functional theory (DFT) based electronic band structures of pristine stoichiometric AgSbTe_2 and $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ is shown in Fig. 4. As evident from Fig. 4(a) and (b), the deficiency in Ag_2Te in $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ results in more states near the Fermi level, which results in higher total density of states (DOS), as shown in Fig. 4(c). This can be explained by the contribution of the respective elements to the states near the Fermi level, as shown in Fig. 5(a). Since all elements strongly contribute to the total DOS near the Fermi level, $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ should result in a higher DOS since Ag-deficiency is compensated by

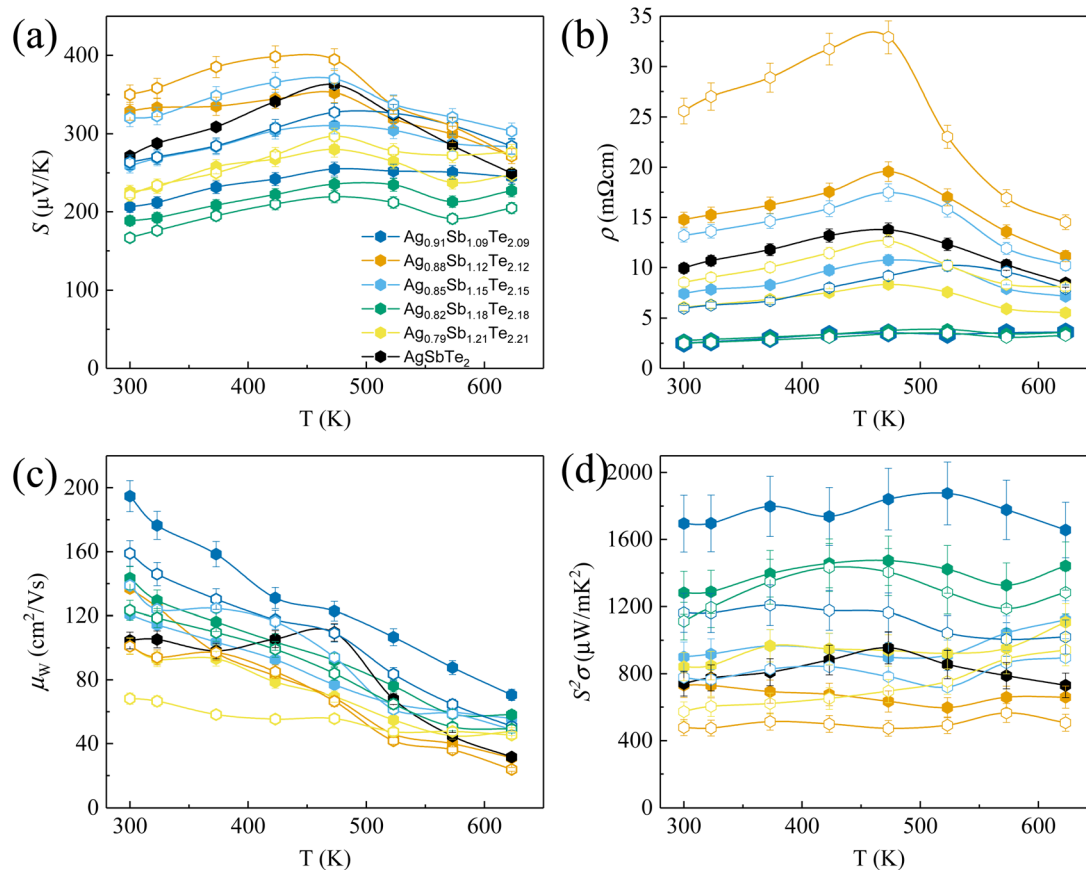


Fig. 3 Electronic transport properties of all samples showing the (a) Seebeck coefficient, (b) electrical resistivity, (c) weighted-mobility and (d) power factor as a function of temperature. Filled and hollow hexagon symbols represent samples with and without annealing at 773 K, respectively.

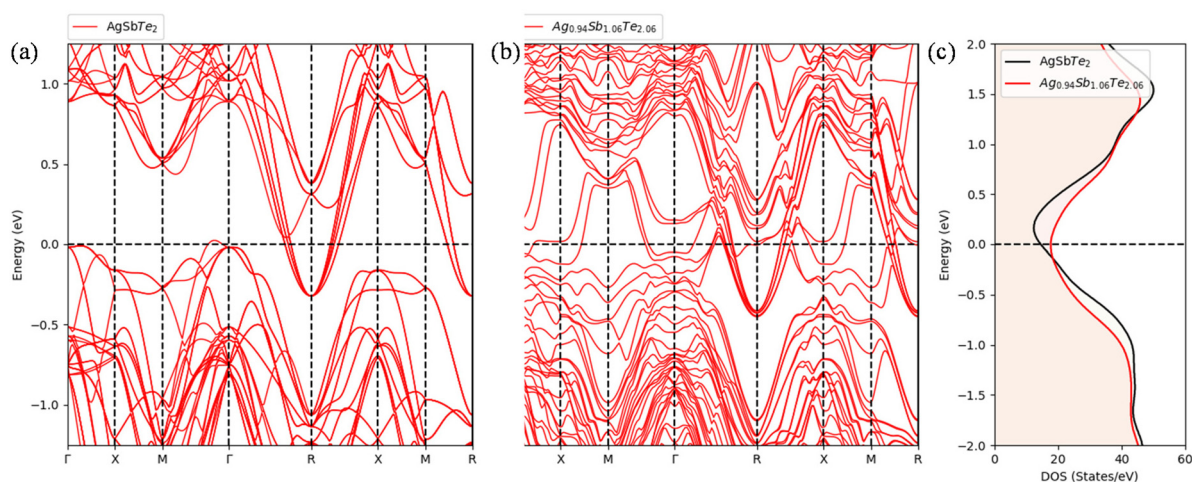


Fig. 4 Electronic band structure comparison of (a) pristine AgSbTe_2 and (b) the Ag_2Te -deficient $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ composition. (c) Total density of states (DOS) showing higher DOS near the Fermi level for $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$. The calculations were based on the conventional cell.

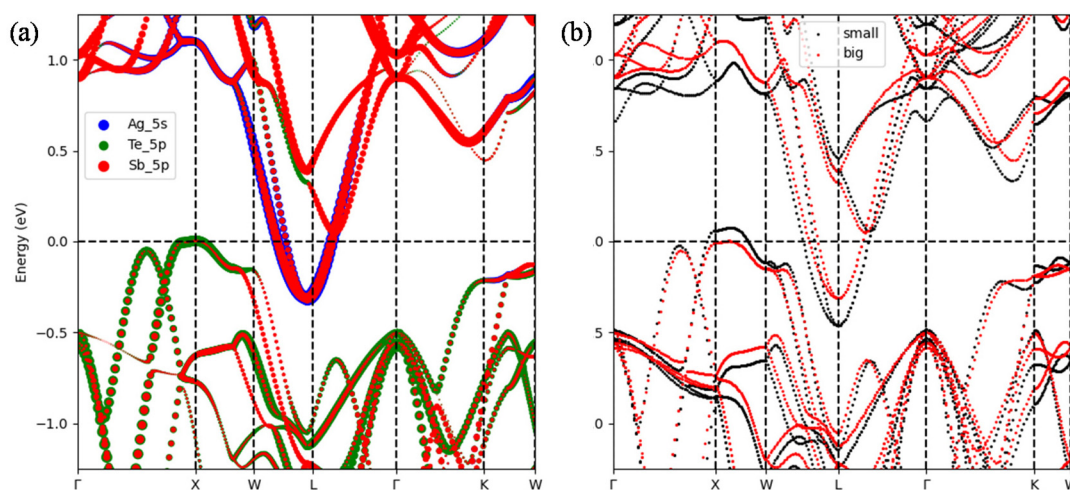


Fig. 5 Electronic band structures showing (a) the flatband of AgSbTe_2 with contributions from each element and (b) the effect of lattice parameters on the band position and the effective band gap. Both calculations were based on primitive cells.

Sb and Te excess. Consequently, this results in generally lower electrical resistivity for $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ than that for AgSbTe_2 , as evident in Fig. 3(b).

To further examine the effect of Ag_2Te deficiency and the corresponding reduction in lattice parameters observed in Fig. 2(b) on the electronic transport properties, Fig. 5(a) shows the contributions of each element to the band structure near the Fermi level. Due to the p-type nature of AgSbTe_2 , the region of interest is near the edge of valence bands. Evidently, most of the contributions come from Te 5p states. In terms of composition, shifting away from AgSbTe_2 into $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ results in stronger Te 5p states, which enhances the states near the valence band edge. On the other hand, the simultaneous reduction and increase in Ag and Sb, respectively, does not appreciably change the conduction band profile. In addition, as evident in Fig. 5(b), bigger lattice parameters (red dotted

plot) result in effective band-gap opening, which is expected to reduce the bipolar contribution and hence increase the overall power factor. This is consistent with the fact that $\text{Ag}_{0.91}\text{Sb}_{1.09}\text{Te}_{2.09}$ has the highest power factor and also the highest lattice parameters amongst all $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ samples.

Fig. 6 shows the thermal transport properties of all samples. In terms of total thermal conductivity (Fig. 6(a)), many of the $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ samples show reduction compared to AgSbTe_2 . This is despite the fact that the electrical resistivity in the majority of $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ samples is lower than that in pristine AgSbTe_2 (Fig. 3(b)), which results in lower κ_e as shown in Fig. 6(b). Not surprisingly, the lattice thermal conductivity shown in Fig. 6(c) of a majority of $\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ samples is lower than that of pristine AgSbTe_2 . Consequently, drastic zT enhancements can be observed for a majority of the

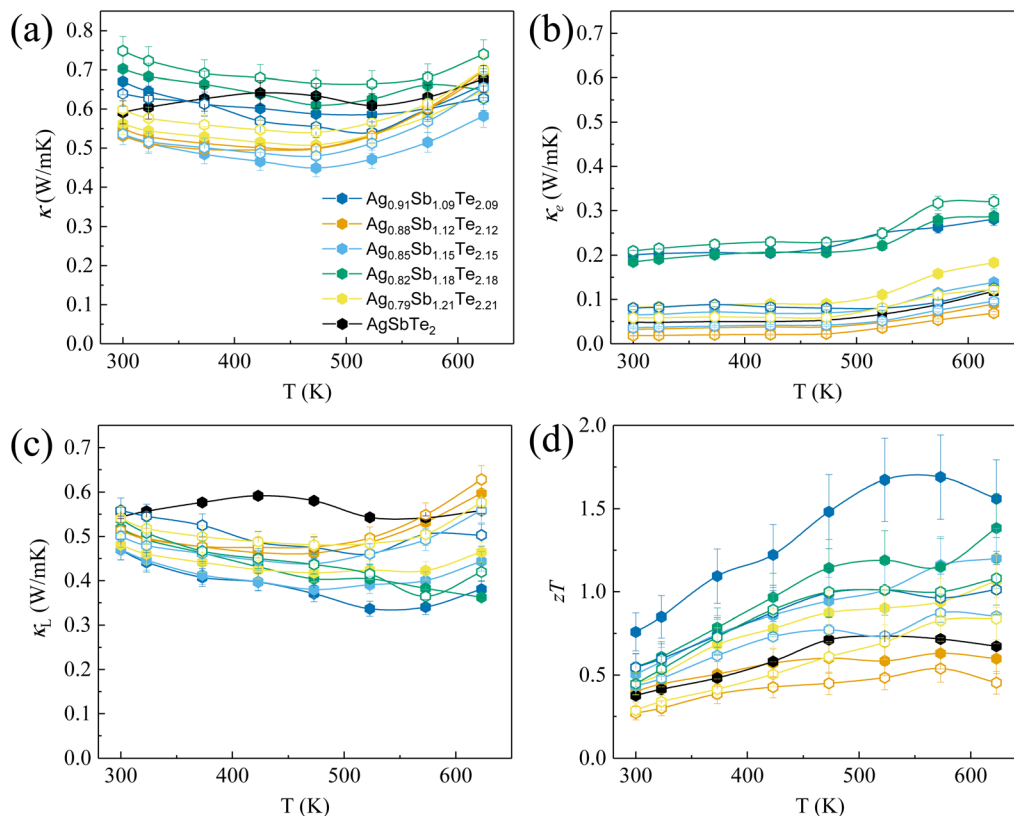


Fig. 6 Thermal transport properties of all samples showing (a) total thermal conductivity, (b) electronic contribution to thermal conductivity, (c) lattice thermal conductivity and (d) zT as a function of temperature. Filled and hollow hexagon symbols represent samples with and without annealing at 773 K, respectively.

$\text{Ag}_{1-x}\text{Sb}_{1+x}\text{Te}_{2+x}$ samples, especially in $\text{Ag}_{0.91}\text{Sb}_{1.09}\text{Te}_{2.09}$ annealed at 773 K, as shown in Fig. 6(d). In addition, an average zT value above 1.0 was observed between 300 and 500 K, which is higher than many results reported in the literature, as shown in Fig. S5.† Furthermore, the high average zT between room temperature and 623 K for $\text{Ag}_{0.91}\text{Sb}_{1.09}\text{Te}_{2.09}$ makes it suitable for power harvesting applications. To verify this, a single-leg device based on $\text{Ag}_{0.91}\text{Sb}_{1.09}\text{Te}_{2.09}$ with an Fe diffusion barrier and Cu electrode layers on both ends was prepared. In order to enable impedance matching with the equipment (minimum circuit resistance ~ 60 m Ω), the aspect ratio of the single leg was intentionally kept high (1.5 mm \times 1.5 mm \times 12 mm), as shown in Fig. S6.†

Fig. 7(a) shows the voltage and power as a function of the current flowing through the single leg for various ΔT . The slope of the voltage and power represents the internal resistance (total resistance of the single leg and the contact resistance). Due to the minimum circuit resistance of the equipment being around 60 m Ω , it is important to ensure high enough single leg resistance. The total resistance of the single leg in this case is 84 m Ω , 101 m Ω , and 108 m Ω , for ΔT of 63 K, 164 K, and 290 K, respectively. Fig. 7(b) shows the open circuit voltage (V_{oc}) and the maximum power output as a function of ΔT . It is noteworthy that the slight non-linearity in V_{oc} can be associated with the temperature-dependency of the

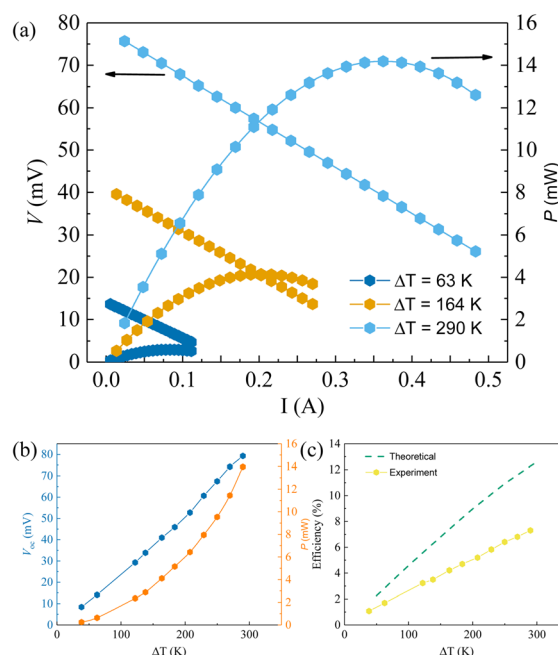


Fig. 7 Single leg power harvesting characteristics of $\text{Ag}_{0.91}\text{Sb}_{1.09}\text{Te}_{2.09}$ showing (a) voltage vs. current and output power vs. current, (b) open circuit voltage vs. temperature gradient and the corresponding maximum output power, and (c) theoretical and experimental power conversion efficiency as a function of temperature gradient.

Seebeck coefficient in the single leg. On the other hand, the maximum output power approximately follows a quadratic relation with ΔT , consistent with the relation $P_{\max} = S^2 \Delta T^2 / 4R_{\text{int}}$ where R_{int} represents the internal resistance. Last but not the least, Fig. 7(c) shows the theoretical maximum and experimental power conversion efficiency, respectively. The deviation from the maximum can be associated with parasitic contact resistance as well as uncertainty in zT estimation of the material. Peak conversion efficiency as high as 7.3% was achieved with a reasonable ΔT of only 290 K.

Conclusions

In this study, we have harnessed the potential of off-stoichiometry and non-equilibrium thermal annealing to significantly enhance the thermoelectric performance of undoped AgSbTe_2 -based compounds. As depicted in Fig. 1, we systematically prepared samples with varying degrees of Ag_2Te deficiency, subjecting them to different annealing temperatures. The outcome of this effort was the achievement of an impressive zT value of 1.7 at 573 K for the $\text{Ag}_{0.91}\text{Sb}_{1.09}\text{Te}_{2.09}$ composition, which had undergone annealing at 773 K. This remarkable improvement in performance can be attributed to a dual effect: the effective suppression of Ag_2Te nanoprecipitates and a reduction in lattice parameters of the Ag_2Te -deficient samples. These factors collectively led to an increase in the Seebeck coefficient and a decrease in electrical resistivity, resulting in a substantial enhancement of the power factor and consequently a significantly improved zT value over a broad temperature range. Furthermore, we demonstrated the practical application of these findings by achieving a high power conversion efficiency of 7.3% at a temperature difference (ΔT) of 290 K for a single-leg $\text{Ag}_{0.91}\text{Sb}_{1.09}\text{Te}_{2.09}$ thermoelectric device, utilizing Fe and Cu encapsulation. The off-stoichiometry and non-equilibrium strategies presented in this study hold promise for broader applications in other ABX_2 material systems, offering the potential for similarly enhanced thermoelectric performance in a variety of contexts.

Author contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Conflicts of interest

The authors declare no competing financial interest.

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References

- 1 N. Jia, J. Cao, X. Y. Tan, J. Dong, H. Liu, C. K. I. Tan, J. Xu, Q. Yan, X. J. Loh and A. Suwardi, *Mater. Today Phys.*, 2021, **21**, 100519.
- 2 J. Cao, X. Y. Tan, J. Dong, H. Liu, Y. Zheng, Q. Zhu, J. Xu, G. Zhang, J. Wu and A. Suwardi, *Mater. Today Phys.*, 2023, **34**, 101071.
- 3 J. Dong, A. Suwardi, X. Y. Tan, N. Jia, K. Saglik, R. Ji, X. Wang, Q. Zhu, J. Xu and Q. Yan, *Mater. Today*, 2023, **66**, 137–157.
- 4 J. Mao, G. Chen and Z. Ren, *Nat. Mater.*, 2021, **20**, 454–461.
- 5 J. Mao, Z. Liu, J. Zhou, H. Zhu, Q. Zhang, G. Chen and Z. Ren, *Adv. Phys.*, 2018, **67**, 69–147.
- 6 J. Zheng, S. F. D. Solco, C. J. E. Wong, S. A. Sia, X. Y. Tan, J. Cao, J. C. C. Yeo, W. Yan, Q. Zhu and Q. Yan, *J. Mater. Chem. A*, 2022, **10**, 19787–19796.
- 7 J. Cao, J. Zheng, H. Liu, C. K. I. Tan, X. Wang, W. Wang, Q. Zhu, Z. Li, G. Zhang and J. Wu, *Mater. Today Energy*, 2022, **25**, 100964.
- 8 S. Hao, V. P. Dravid, M. G. Kanatzidis and C. Wolverton, *npj Comput. Mater.*, 2019, **5**, 58.
- 9 J. Cao, Y. Sim, X. Y. Tan, J. Zheng, S. W. Chien, N. Jia, K. Chen, Y. B. Tay, J. F. Dong and L. Yang, *Adv. Mater.*, 2022, **34**, 2110518.
- 10 J. Cao, X. Y. Tan, N. Jia, D. Lan, S. F. D. Solco, K. Chen, S. W. Chien, H. Liu, C. K. I. Tan and Q. Zhu, *Nanoscale*, 2022, **14**, 410–418.
- 11 H. K. Ng, D. Xiang, A. Suwardi, G. Hu, K. Yang, Y. Zhao, T. Liu, Z. Cao, H. Liu and S. Li, *Nat. Electron.*, 2022, **5**, 489–496.
- 12 Y. Pei, X. Shi, A. LaLonde, H. Wang, L. Chen and G. J. Snyder, *Nature*, 2011, **473**, 66–69.
- 13 F. Yang, J. Wu, A. Suwardi, Y. Zhao, B. Liang, J. Jiang, J. Xu, D. Chi, K. Hippalgaonkar and J. Lu, *Adv. Mater.*, 2021, **33**, 2004786.
- 14 A. Suwardi, J. Cao, Y. Zhao, J. Wu, S. W. Chien, X. Y. Tan, L. Hu, X. Wang, W. Wang and D. Li, *Mater. Today Phys.*, 2020, **14**, 100239.
- 15 N. Jia, J. Cao, X. Y. Tan, J. Zheng, S. W. Chien, L. Yang, K. Chen, H. K. Ng, S. S. F. Duran and H. Liu, *J. Mater. Chem. A*, 2021, **9**, 23335–23344.
- 16 B. Zhu, X. Su, S. Shu, Y. Luo, X. Y. Tan, J. Sun, D. Sun, H. Zhang, Q. Zhang and A. Suwardi, *ACS Appl. Energy Mater.*, 2022, **5**, 2002–2010.
- 17 B. Zhu, Y. Luo, H. Wu, D. Sun, L. Liu, S. Shu, Z.-Z. Luo, Q. Zhang, A. Suwardi and Y. Zheng, *J. Mater. Chem. A*, 2023, **11**, 8912–8921.

- 18 Z. Danwei, S. Seng Ann, S. Samantha Faye Duran, X. Jianwei and S. Ady, *Soft Science.*, 2023, **3**, DOI: [10.20517/ss.2022.29](https://doi.org/10.20517/ss.2022.29).
- 19 D. Zhang, X. J. G. Lim, X. Li, K. Saglik, S. F. D. Solco, X. Y. Tan, Y. Leow, W. Zhai, C. K. I. Tan and J. Xu, *ACS Energy Lett.*, 2022, **8**, 332–338.
- 20 D. Zhang, W. Y. S. Lim, S. S. F. Duran, X. J. Loh and A. Suwardi, *ACS Energy Lett.*, 2022, **7**, 720–735.
- 21 Y. Wu, Z. Chen, P. Nan, F. Xiong, S. Lin, X. Zhang, Y. Chen, L. Chen, B. Ge and Y. Pei, *Joule*, 2019, **3**, 1276–1288.
- 22 A. Suwardi, J. Cao, L. Hu, F. Wei, J. Wu, Y. Zhao, S. H. Lim, L. Yang, X. Y. Tan and S. W. Chien, *J. Mater. Chem. A*, 2020, **8**, 18880–18890.
- 23 A. Suwardi, B. Prasad, S. Lee, E.-M. Choi, P. Lu, W. Zhang, L. Li, M. Blamire, Q. Jia and H. Wang, *Nanoscale*, 2016, **8**, 8083–8090.
- 24 D. Zhang, M. Xie, D. Safanama, K. Saglik, X. Y. Tan, S. F. D. Solco, J. Cao, C. K. I. Tan, H. Liu and S. Wang, *Adv. Energy Sustainability Res.*, 2023, 2300069.
- 25 J. Cao, S. W. Chien, X. Y. Tan, C. K. I. Tan, Q. Zhu, J. Wu, X. Wang, Y. Zhao, L. Yang and Q. Yan, *ChemNanoMat*, 2021, **7**, 476–482.
- 26 Y. Zheng, H. Xie, Q. Zhang, A. Suwardi, X. Cheng, Y. Zhang, W. Shu, X. Wan, Z. Yang and Z. Liu, *ACS Appl. Mater. Interfaces*, 2020, **12**, 36186–36195.
- 27 J. Cao, X. Y. Tan, N. Jia, J. Zheng, S. W. Chien, H. K. Ng, C. K. I. Tan, H. Liu, Q. Zhu and S. Wang, *Nano Energy*, 2022, **96**, 107147.
- 28 J. Cao, J. Dong, K. Saglik, D. Zhang, S. F. D. Solco, I. J. W. J. You, H. Liu, Q. Zhu, J. Xu and J. Wu, *Nano Energy*, 2023, **107**, 108118.
- 29 S. Roychowdhury, T. Ghosh, R. Arora, M. Samanta, L. Xie, N. K. Singh, A. Soni, J. He, U. V. Waghmare and K. Biswas, *Science*, 2021, **371**, 722–727.
- 30 R. Pathak, L. Xie, S. Das, T. Ghosh, A. Bhui, K. Dolui, D. Sanyal, J. He and K. Biswas, *Energy Environ. Sci.*, 2023, **16**, 3110–3118.
- 31 J. Dong, Y. Jiang, J. Liu, J. Pei, X. Y. Tan, H. Hu, A. Suwardi, N. Jia, C. Liu and Q. Zhu, *Nano Energy*, 2022, **103**, 107809.
- 32 X. Wang, H. He, L. Yin, J. Cheng, J. Sui, X. Liu, J. Mao, F. Cao and Q. Zhang, *Mater. Today Energy*, 2022, **26**, 100994.
- 33 Z. Zhang, K. Zhao, T.-R. Wei, P. Qiu, L. Chen and X. Shi, *Energy Environ. Sci.*, 2020, **13**, 3307–3329.
- 34 J. Jia, Y. Zhou, X. Chen, W. Xue, H. Yu, J. Li, S. Zhi, C. Chen, J. Wang and S. Hou, *Research*, 2022, 9875329.
- 35 D. Liu, D. Wang, T. Hong, Z. Wang, Y. Wang, Y. Qin, L. Su, T. Yang, X. Gao and Z. Ge, *Science*, 2023, **380**, 841–846.
- 36 J. Mao, H. Zhu, Z. Ding, Z. Liu, G. A. Gamage, G. Chen and Z. Ren, *Science*, 2019, **365**, 495–498.
- 37 A. Suwardi, S. H. Lim, Y. Zheng, X. Wang, S. W. Chien, X. Y. Tan, Q. Zhu, L. M. N. Wong, J. Cao and W. Wang, *J. Mater. Chem. C*, 2020, **8**, 16940–16948.
- 38 J. K. Lee, M.-W. Oh, B. Ryu, J. E. Lee, B.-S. Kim, B.-K. Min, S.-J. Joo, H.-W. Lee and S.-D. Park, *Sci. Rep.*, 2017, **7**, 4496.
- 39 J. K. Lee, B. Ryu, S. Park, J. H. Son, J. Park, J. Jang, M.-W. Oh and S. Park, *Acta Mater.*, 2022, **222**, 117443.
- 40 Y. Zhang, Z. Li, S. Singh, A. Nozariasbmarz, W. Li, A. Genç, Y. Xia, L. Zheng, S. H. Lee and S. K. Karan, *Adv. Mater.*, 2023, **35**, 2208994.
- 41 Y. Wu, P. Qiu, Y. Yu, Y. Xiong, T. Deng, O. Cojocarumiredin, M. Wuttig, X. Shi and L. Chen, *J. Materiomics*, 2022, **8**, 1095–1103.
- 42 Y. Wu, Q. Liang, X. Zhao, H. Wu, P. Zi, Q. Tao, L. Yu, X. Su, J. Wu and Z. Chen, *ACS Appl. Mater. Interfaces*, 2022, **14**, 3057–3065.
- 43 T. Ghosh, S. Roychowdhury, M. Dutta and K. Biswas, *ACS Energy Lett.*, 2021, **6**, 2825–2837.
- 44 H. Wang, J.-F. Li, M. Zou and T. Sui, *Appl. Phys. Lett.*, 2008, **93**, 202106.
- 45 S. Roychowdhury, R. Panigrahi, S. Perumal and K. Biswas, *ACS Energy Lett.*, 2017, **2**, 349–356.
- 46 W. Szczypka and A. Koleżyński, *J. Alloys Compd.*, 2019, **787**, 1136–1142.
- 47 P. Wyzga and K. Wojciechowski, *J. Electron. Mater.*, 2016, **45**, 1548–1554.
- 48 P. Giannozzi, S. Baroni, N. Bonini, M. Calandra, R. Car, C. Cavazzoni, D. Ceresoli, G. L. Chiarotti, M. Cococcioni and I. Dabo, *J. Phys.: Condens. Matter*, 2009, **21**, 395502.
- 49 P. Giannozzi, O. Andreussi, T. Brumme, O. Bunau, M. B. Nardelli, M. Calandra, R. Car, C. Cavazzoni, D. Ceresoli and M. Cococcioni, *J. Phys.: Condens. Matter*, 2017, **29**, 465901.
- 50 P. Giannozzi, O. Baseggio, P. Bonfà, D. Brunato, R. Car, I. Carnimeo, C. Cavazzoni, S. De Gironcoli, P. Delugas and F. Ferrari Ruffino, *J. Chem. Phys.*, 2020, **152**, 154105.
- 51 X. Hua, X. Chen and W. Goddard, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1997, **55**, 16103.
- 52 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865.
- 53 M. Hong, Z. G. Chen, L. Yang, Z. M. Liao, Y. C. Zou, Y. H. Chen, S. Matsumura and J. Zou, *Adv. Energy Mater.*, 2018, **8**, 1702333.
- 54 S. Barabash, V. Ozolins and C. Wolverton, *Phys. Rev. Lett.*, 2008, **101**, 155704.
- 55 K. Hoang, S. Mahanti, J. R. Salvador and M. G. Kanatzidis, *Phys. Rev. Lett.*, 2007, **99**, 156403.