



REVIEW

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Enhancement of MXene optical properties towards medical applications *via* metal oxide incorporation

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MXenes have garnered research attention in the field of biomedical applications due to their unique properties, such as a large surface area, low toxicity, biocompatibility, and stability. Their optical behavior makes them versatile for a wide range of biomedical applications, from diagnostics to therapeutics. Nonetheless, MXenes have some minor limitations, including issues with restacking, susceptibility to oxidation, and a non-semiconducting nature. These limitations have prompted researchers to explore the incorporation of metal oxides into MXene structures. Metal oxides possess advantageous properties such as a high surface area, biocompatibility, intriguing redox behavior, catalytic activity, semiconducting properties, and enhanced stability. Incorporating metal oxides into MXenes can significantly improve their conductivity, surface area, and mechanical strength. In this review, we emphasize the importance of incorporating metal oxides into MXenes for light-influenced biomedical applications. We also provide insights into various preparation methods for incorporating metal oxides into MXene structures. Furthermore, we discuss how the incorporation of metal oxides enhances the optical behavior of MXenes. Finally, we offer a glimpse into the future potential of metal oxide-incorporated MXenes for diverse biomedical applications.

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1. Introduction

The use of nanomaterials in biomedical applications is becoming increasingly crucial, offering significant support to the healthcare system and advancements in diagnostics and therapeutics. Many nanomaterials are used in biomedical applications like biosensors, drug delivery, bioimaging, and implants.^{1,2} Different dimensional nanomaterials are employed in various biomedical applications since nanomaterials show many beneficial properties in different dimensions.³ Two-dimensional nanomaterials have gained more attention in various biomedical applications due to their tunable surface area and electrical and optical properties.⁴ In particular, graphene,⁵ antimonene,⁶ black phosphorus,⁷ dichalcogenides,⁸ and oxides⁹ are widely used 2D nanomaterials in biomedical applications. Nevertheless, MXenes have gained immense attention in the 2D nanomaterials research¹⁰ and have an idiosyncratic two-dimensional (2D) structure embedded with transition metal carbides, nitrides,

and carbo-nitrides. These structures evolved in 2011 when Prof. Michel W. Barsoum and Prof. Yury Gogotsi from Drexel University, Philadelphia discovered the first-ever MXene–titanium-carbide layers by exfoliating 3D titanium aluminium carbide (Ti₃AlC₂) *via* the acid etching technique.³ The geometrical structures of MXenes are symbolized as M_{n+1}X_nT_x, where M represents an early transition metal (such as Sc, Ti, Zr, Hf, V, Nb, Ta, Cr, Mo, Mn), X is C and N, T is an exterior windup unit especially hydroxyl, oxygen or fluorine groups, and *n* = 1, 2, or 3. Akin to graphene, MXenes are generally formed by exfoliating their 3D precursors, represented as a top–down approach.⁴ The preparation and fabrication modalities of MXenes involve both the top and bottom scaling approaches, more specifically the top scaling methods are vastly applied relative to the bottom–up scaling methods that have seen rare utilization. The top–down scaling methods mainly comprise liquid-based chemical and mechanical exfoliation from the bulk crystal of the MAX phase. There are different types of MAX phases available like Ti₂AlC, Ti₂AlN, and V₂GeC, Ti₃SiC₂, Ti₃GeC₂, Ti₃AlC₂, Ti₃SnC₂, Ta₃AlC₂, Ti₄AlN₃, Ti₄SiC₃, Ti₄GeC₃, Ta₄AlC₃, Nb₄AlC₃, V₄AlC₃, and Ti₄G₂C₃.¹¹ The conversion of the MAX phase into MXene is generally performed *via* wet chemistry methods by means of etching the atomic (A) layers from their corresponding layered precursors (MAX phase) with the use of strong etchants

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namely HF,¹² LiF,¹³ NH₄HF₂,¹⁴ and NH₄F.¹⁵ In addition, non-HF etchants such as Lewis acids¹⁶ and molten salts¹⁷ (LiF, NaF, KF, NaOH) based electrochemical and hydrothermal methods have been introduced.¹⁸

As depicted in Fig. 1, the 3D precursors for MXenes are termed MAX phases, ternary carbides, or nitrides described with the general formula M_{n+1}AX_n, in which A stands for an A group element (mostly leading group IIIA or IVA). The A layers in MXenes are more active than the M–X layers. Because M–X bonds are much stronger than M–A bonds, A layers can be selectively etched using a strong acid (*i.e.*, hydrofluoric acid, HF) to produce M_{n+1}X_n layers further detached by sonication.¹⁹ The exterior of MXenes is typically winded up with fluorine (–F), hydroxide (–OH), and oxygen (–O) groups by the etching process, which provides high surface energy.³ Hence, the final geometrical symbol of MXenes is displayed as M_{n+1}X_nT_x, where T_x is the surface functional group.⁶ In addition, MXenes have various properties that can be employed in their fundamental prospects, such as structural, chemical, optical,²⁰ electronic, and even biological.¹⁰ Many experimental and theoretical studies²⁰ explored the stupendous potential of MXenes for various applications in optoelectronics, photonics, catalysis, and many other areas.^{19,21} MXenes emerge as new 2D nano-platforms for advanced optoelectronics to be explored in various biomedical applications like biosensing,²² bio-imaging,²³ and therapeutic applications.^{24,25} The essential mechanical and electronic properties can be attained by tailoring the elemental composition and chemical functionalization.²⁶ The exclusive traits of MXenes involve a high surface area, high atomic numbers (some transition metals), the presence of hydrophilic functional groups, and paramagnetic behaviour. The functional groups on MXenes also contribute to their rigidity and flexibility, which is crucial in thin film formation as part of bio-electronic devices.²⁷

Most importantly, the high surface area is beneficial for drug loading and delivery for theranostic applications in various synergistic diseases. Furthermore, hydrophilic functional groups are also crucial for drug loading and targeted drug delivery, as modification or functionalization increases biocompatibility for living cells/tissues.^{28,29} MXenes, comprising high atomic number and paramagnetic transition metals,

are more relevant for biomedical imaging because they show superior X-ray attenuation for computed tomography (CT) scans and can also be employed as magnetic resonance imaging (MRI) contrast agents.²⁹ Distinct types of MXenes could be prepared based on different approaches; therefore, the biocompatibility assessment of MXene-based biomaterials is essential for general biomedical applications.³⁰ Metal oxides are widely employed in various biomedical applications in the diagnostic and therapeutic fields as they possess biocompatibility, a high surface area, good chemical and mechanical stability, and a unique structure.³¹ Besides, the unique properties of metal oxides, such as enhanced electron transfer kinetics, strong adsorption capability, high sensitivity, and high optical absorption ability, make them more suitable for incorporation into MXenes. This enhances the optical properties for biomedical applications.^{32,33} Metal oxides like TiO₂,³⁴ ZnO,³⁵ ZrO₂,³⁶ Fe₂O₃,³⁷ MnO₂,³⁸ MgO,³⁹ CeO₂,⁴⁰ and CuO⁴¹ are broadly used in biomedical applications. In the part of the set, various metal oxide-based hybrid and composite formulations of MXenes were reported elsewhere for biomedical applications.^{42–44} In recent years, the incorporation or coating of metal oxides over MXenes has gained prominence due to the substantial enhancement of their optical properties for various light-influenced medical applications.^{45–47} This has sparked significant interest in understanding the underlying mechanisms of metal oxide incorporation into MXenes. However, to date, a comprehensive review of how the technology of metal oxide incorporation alone improves MXenes' optical properties and their applications in medicine has not been published.

This review emphasized the crucial part of tuning the optical properties of MXenes *via* incorporating metal oxides for different photo-influenced biomedical applications, as presented in Fig. 2. Also, the unique enhancement of the optical properties of MXenes using metal oxides resulting in advanced biomedical applications has been covered with recent tenets and trends. This review discussed the importance of incorporating metal oxides into MXenes to enhance their optical behavior. The different preparation methods for incorporating metal oxides into MXenes were covered, along with their advantages and disadvantages. This review emphasized the scope and challenges of metal oxides masked with MXenes for future research.

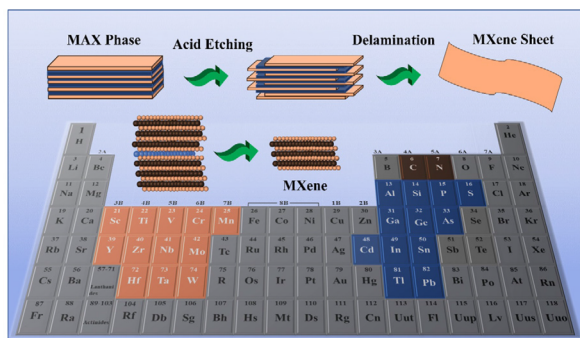


Fig. 1 Schematic representation of MAX phase elements in the periodic table and synthesis steps of MXenes from the MAX phase.

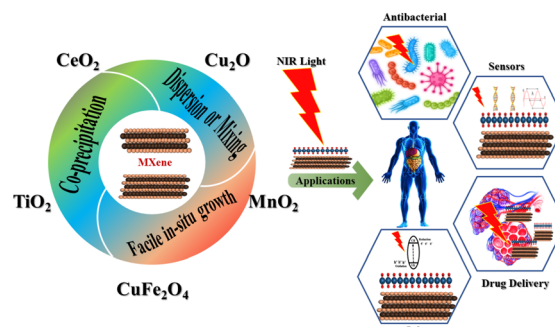


Fig. 2 Schematic overview of the review organization.

2. Complementing MXene with metal oxide composites

The optical properties of the MXene rely on the energy structures, the dispersion of linear and nonlinear dielectric function (ϵ), or the refractive index ($n = \sqrt{\epsilon\mu}$).^{48,49} MXenes have a linear absorption loss of $\sim 1\% \text{ nm}^{-1}$,⁵⁰ a nonlinear absorption coefficient of about $10\text{--}21 \text{ m}^2 \text{ V}^{-2}$,⁵¹ and a negative nonlinear refractive index of about $10\text{--}20 \text{ m}^2 \text{ W}^{-1}$.⁵¹ MXenes have exclusive characteristics, with enormous electrons gathered close to the Fermi level of MXene to serve as an electron reservoir, supplying decisive transfer of charge carriers. The beneficial properties of MXene include surface-modified functional groups that tune the bandgap of the obtained materials. These functional groups generate a negative charge on the surface of MXene, which electrostatically attracts the cations.⁵² It also furnishes a platform to enhance interfacial contact between the semiconductor and MXenes, resulting in the transfer or separation of charge carriers in composites to enhance the optical properties.⁵³ Though MXene comprises many unique properties, it shows limitations like lack of semiconducting behaviour, restacking of MXene layers after exfoliation, easy oxidation, and low electrical conductivity.⁵⁴ It is also worth mentioning that metal oxides lack conductivity and have more significant band gaps. Hence, the heterostructures of MXenes incorporated with metal oxides exhibit favourable improvement in enhancing the optical properties to be employed in the photocatalytic performance for various biomedical applications. Different metal oxides, like SnO_2 ,⁵⁵ TiO_2 ,⁵⁶ and MnO_2 ,⁵⁷ have been incorporated into MXenes to enhance their optical properties. Several studies⁵⁸ suggested that incorporating NiFe_2O_4 into MXene heterostructures cogently assisted in averting the restacking of MXene flakes to increase the surface area of the heterostructure, and the surface area of MXene was found to be 19.26 and improved to $49.1812 \text{ m}^2 \text{ g}^{-1}$ after the incorporation of NiFe_2O_4 .

3. Different methods to mask metal oxides in MXene composites

MXenes are typically synthesized by wet chemical acidic etching of MAX phases using robust etching solutions with fluoride ions (F^-) such as hydrofluoric acid (HF)⁵⁹ and ammonium bifluoride⁶⁰ and a mixture of hydrochloric acid and lithium fluoride. They can be prepared in single to few flakes or multiple stacks of MXene sheets. The morphology of MXenes can be tuned by adjusting the concentration of the etching solution, etching time, ultrasonication, and temperature. The literature also highlights exciting methods such as wet chemical reduction, self-oxidation, self-reduction, and pressure-less sintering for incorporating metal oxides into MXenes. Other techniques, including dry mixing/thermal pressing, solution blending, emulsion mixing, *in situ* polymerization, and lamination stacking, have also been utilized.

Metal oxides are also connected with MXenes using simple chemical methods like blending and precipitation.

3.1 Co-precipitation method

Co-precipitation is a simple and fastest technique to prepare metal oxide-masked MXenes. This method has advantages like high yield, high product purity, and lack of necessity for organic solvents. As shown in Fig. 3, MXenes and metal oxides are dispersed in a suitable solvent at a particular temperature with a suitable reducing agent and the pH is maintained to form intermediate precipitates resulting from nucleation and growth.^{61,62} The parameters like temperature, pressure, concentration, pH, stirring speed, and reaction time are of significant concern in controlling nanostructure size and shape distribution. First, the⁶³ MXenes are dispersed in a specified solvent, mostly in distilled water, followed by the addition of the precursor for metal oxide with either a suitable stabilizing agent or targeting moieties depending upon application usage. For instance, for anti-bacterial application, Ti_3C_2 is masked using cuprous oxide (Cu_2O).⁶³

As depicted in Fig. 4, aminated MXenes reacted with cupric nitrate trihydrate as a precursor for synthesizing Cu_2O and polyvinylpyrrolidone (PVP), and hydrazine hydrate is used as stabilizing and reducing agents, respectively.⁶³ A similar and simple synthesis approach is attempted for preparing the $\text{Cu}_2\text{O}/\text{MXene}$ composite with copper(II) sulfate, PVP, and N_2H_4 as the precursor, stabilizing, and reducing agents, respectively.⁶⁴ To support the co-precipitation synthesis route, various metal oxide-masked MXenes are synthesized, such as $\text{Ti}_3\text{C}_2/\text{Al}_2\text{O}_3/\text{Ag}$,⁶⁵ $\text{Ti}_3\text{C}_2/\text{SiO}_2/\text{Ag}$,⁶⁶ and $\text{Ti}_3\text{C}_2/\text{SiO}_2/\text{Pd}$,⁶⁶ $\text{Ti}_3\text{C}_2\text{F}/\text{FeWO}_4$,⁶⁷ $\text{TiO}_2/\text{Ti}_3\text{C}_2/\text{Cu}_2\text{O}$, and $\text{WO}_3/\text{MXenes}$.⁴⁵

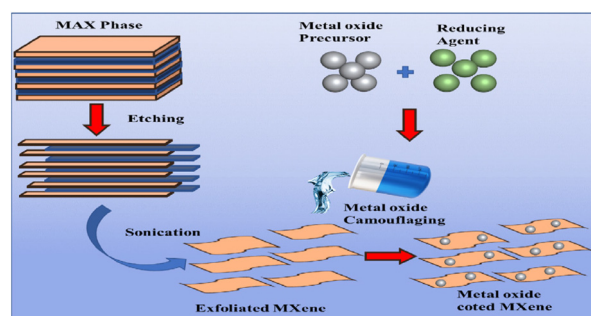


Fig. 3 Schematic illustration of MXene/metal oxide synthesis via the co-precipitation method.

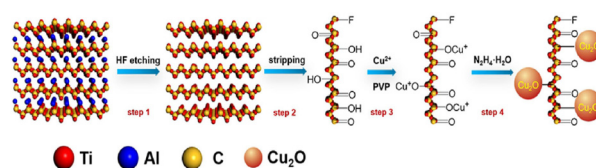


Fig. 4 Schematic illustration of the etching, stripping and reduction process of $\text{Cu}_2\text{O}/\text{MXene}$ synthesis via the co-precipitation method. (Reproduced with permission from ref. 64 Copyright, 2020 Elsevier.)

3.2 Dispersion or mixing

Another approach for preparing MXene/metal oxide nanocomposite is the dispersion of both MXene/metal oxide in a suitable solvent under ultrasonication, as depicted in Fig. 5. Ultrasonication mixing or mechanical stirring assists in properly distributing MXene/metal oxide to form a composite after removing the solvent and drying. In general, the parameters like ultrasonication power and frequency, temperature, and ultrasonication time play a crucial role in this method. Following a longer sonication time and less ultrasonication power is essential for proper dispersion.⁶⁸ As evidence, Fig. 6 shows the dispersion or mixing method of CuO-masked MXene preparation *via* the ultrasonication route for photocatalytic anti-bacterial applications.⁴⁶ The composites, NiFe₂O₄/MXene⁵⁸ and CuFe₂O₄/Ti₃C₂,⁶⁹ were also prepared by dispersion in an ultrasonicated water bath for an hour, followed by vacuum drying.

3.3 Facile *in situ* growth

Fig. 7 shows the facile *in situ* technique; redox reactions grow the metal oxide *in situ* over the MXene exterior. For example, Fig. 8 shows the *in situ* growth of iron-oxide on the surface of Ta₄C₃ MXene according to the redox reaction of MXene for photoactivated cancer theranostic applications. The exterior is stabilized or modified with soybean phospholipids for

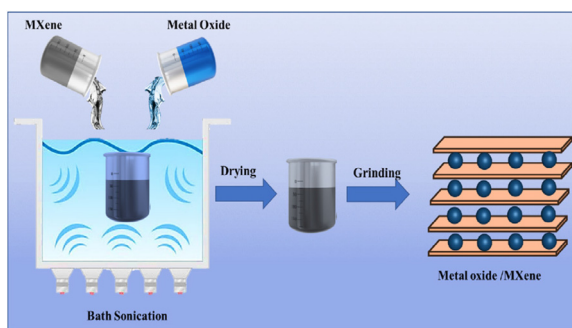


Fig. 5 Schematic representation of MXene/metal oxide synthesis *via* the dispersion or mixing method.

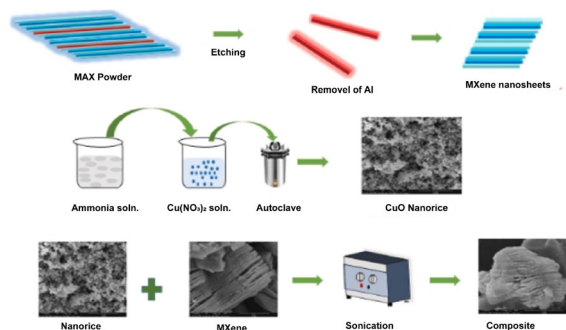


Fig. 6 Schematic illustration of CuO/MXene synthesis *via* the dispersion/mixing method. (Reproduced with permission from ref. 46 Copyright, 2022 Elsevier.)

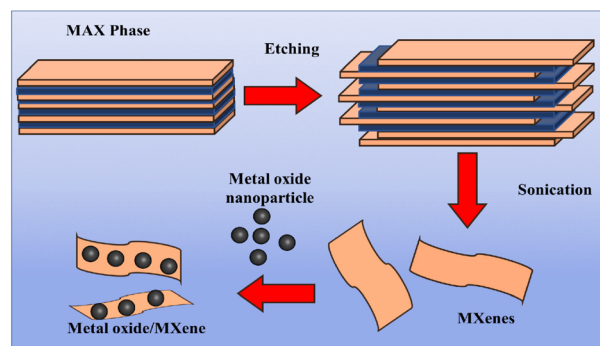


Fig. 7 Schematic illustration of *in situ* growth of metal oxide over MXene nanosheets.

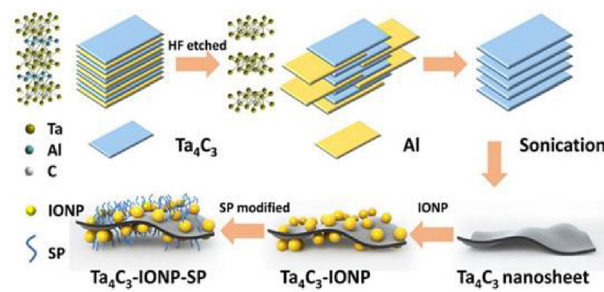


Fig. 8 Schematic illustration of *in situ* growth of iron oxide over Ta₄C₃ MXene. (Reproduced with permission from ref. 91 Copyright, 2018 IVY Spring.)

enhanced delivery of nanoparticles at the site of action. Akin to the abovementioned technique, manganese oxide is grown over Ti₃C₂⁷⁰ by a simple redox reaction for cancer therapeutics. Briefly, the KMnO₄ aqueous solution was mixed with Ti₃C₂ aqueous solution under magnetic stirring at room temperature, followed by centrifugation resulting in the composite nanosheet. The challenge is that many MXenes need to be fabricated from the MAX phase before mixing, and sometimes it requires retaining the stratified architecture to supply more cavities for metals or metal compounds to be deposited.

The kinds of composites reported to date have been synthesized in four ways: (i) *ex situ* mixing, (ii) self-oxidation mixing, (iii) one-step *in situ* growth, and (iv) multistep *in situ* conversion. The choice of an appropriate method varies according to the target material and its precursors. Different synthetic ways result in additional product structures and performances (Table 1).

4. Applications of metal oxide masked MXenes

The enhanced surface area, biocompatibility, hydrophilicity, and strong plasmonic effects of MXenes have gained more attention in biomedical applications. MXenes are widely

Table 1 Advantage and disadvantage of different metal oxide incorporation methods

| Methods | Advantages | Disadvantages | Ref. |
|------------------------------|---|--|-------------|
| Co-precipitation | Purity, scalability, versatility, homogeneous incorporation of metal oxides, easy to functionalize with organic and inorganic materials | Limited choice of precursors, challenge to control the layer thickness | 128 and 129 |
| Dispersion or mixing | Uniform thickness, high surface area, tunability of size, thickness, surface chemistry, and stability | Solvent selection is critical, reproducibility, time consuming | 130 and 131 |
| Facile <i>in situ</i> growth | Controlled morphology, homogeneous distribution of metal oxide in MXene matrix, reduced agglomeration, tailored composition | Undesired materials or by products, longer reaction times | 132 and 133 |

employed in biomedical applications such as drug delivery, biosensors, cancer theranostics, and antimicrobial activity. The incorporation of metal oxides has enhanced the catalytic, photothermal, therapeutic, and diagnostic performance of MXenes. This section discusses the antibacterial, anticancer, and sensing applications of metal oxide-masked MXenes.

4.1 Anti-bacterial activity

2D nanosheets of MXenes exhibit anti-bacterial activity owing to their high surface area, hydrophilicity, and negatively charged surface.⁷¹ As it has smaller electrons at the Fermi level, incorporating metal oxide can increase the number of electrons. The surface of MXene has the dexterity to accept electrons from the metal oxide resulting in a substantial number of electrons that ease the charge transfer between bacterial cell walls. The formation of MXene-based heterostructures like $\text{NiFe}_2\text{O}_4/\text{Ti}_3\text{C}_2$ (NiFe) proved their anti-bacterial activity against Gram-negative *E. coli* bacteria using the colony count method. It is revealed that NiFe/MXene composite phases showed the best anti-bacterial performance compared to individual NiFe and MXene flakes. The addition of NiFe facilitated a cross-linked network, high surface area, and enhanced magnetic properties.⁵⁸

The same Ti_3C_2 is also anchored with cuprous oxide (Cu_2O) for analyzing photocatalytically triggered anti-bacterial activity.⁶⁴ The anti-bacterial studies shown in Fig. 9 revealed that the $\text{Cu}_2\text{O}/\text{MXene}$ nanosheet exhibits excellent anti-bacterial activity against *S. aureus* and *P. aeruginosa* in comparison with MXene, Cu_2O , and their mixture (MXene and Cu_2O), thus resulting in the admirable increase in the bacteriostasis efficiencies to around 97.04% and 95.59% against both the bacteria, respectively.

The abundant electrons on the surface of MXene accepted from Cu_2O facilitated better charge transfer between bacteria and $\text{Cu}_2\text{O}/\text{MXene}$ nanosheets, revealing the synergistic effect of Cu_2O and the bacteriostatic action of MXene. The production of reactive oxygen species responsible for the damage of bacterial cell membranes and cell death was analyzed by photocatalytically triggered anti-bacterial activity. The $\text{Cu}_2\text{O}/\text{MXene}$ induced by sunlight decreases the number of bacterial colonies by generating reactive oxygen species (ROS), which can ruin cell membranes and eventually cause cell death by oxidative stress. It is proved that $\text{Cu}_2\text{O}/\text{MXene}$ produced greater ROS than Cu_2O . Hence, the separation efficiency of photoinduced charges of Cu_2O was significantly ameliorated by combining with MXene nanosheets as a heterojunction

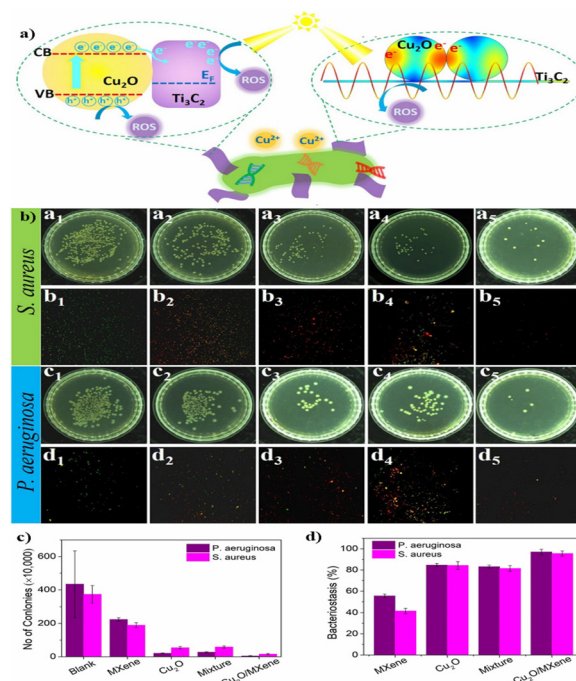


Fig. 9 Schematic representation of light induced antibacterial activity mechanisms and therapeutic effects of $\text{Cu}_2\text{O}/\text{Ti}_3\text{C}_2$ (A–C). (a) Mechanism of optical antibacterial action; (b) profiling of antibacterial activity on *S. aureus* and *P. aeruginosa*; and (c) quantified therapeutic efficiency; (d) bacteriostasis percentage. (Reproduced with permission from ref. 64 Copyright, 2020 Elsevier.)

structure. It is due to the transfer of a photoinduced electron from Cu_2O to MXene, which suppressed the recombination of electron-hole pairs because the Fermi potential level of the MXene (0.71 V vs. NHE, pH = 7) was substantially lesser than the conduction band of Cu_2O (−0.703 V vs. NHE, pH = 7). Moreover, the MXene grabbed more electrons on the surface from Cu_2O , facilitating the charge transfer between bacteria and $\text{Cu}_2\text{O}/\text{MXene}$ nanosheets. The accumulated charges of intracellular ROS, including hydrogen peroxide (H_2O_2), superoxide anions (O_2^-), or hydroxyl radicals ($-\text{OH}$), resulted in cell membrane damage and cell death.

Excitingly, studies have shown that the incorporation of different structures of metal oxides on MXenes can alter their anti-bacterial activity.⁷² For that, the anti-bacterial action of randomly oriented (*i.e.*, dispersed flower-like MnO_2 and MoS_2) versus vertically aligned MnO_2 and MoS_2 nanomaterials grown individually on GO, rGO, and Ti_3C_2 MXene substrates (*i.e.*,

MnO₂/GO, MoS₂/rGO, and MoS₂/MXene) against both Gram-positive and Gram-negative bacteria was studied. It has been proven that the sharp edges of 2D nanosheets play a role in targeting and ruining the bacterial cell wall, resulting in the loss of membrane integrity and, eventually, bacteria death.⁷³ This study was intended to evaluate the impact of 2D nanosheets on the peptidoglycan mesh (PM) in the bacterial cell wall. It was revealed that vertically aligned 2D nanosheet motifs show a higher anti-bacterial activity against both bacteria classes than 2D nanomaterials.⁷² To uphold the structural importance of metal oxide masked MXenes in optical-based anti-bacterial applications, a multifunctional hydrogel scaffold is developed using polyethyleneimine grafted Pluronic F127 (F127-PEI) and oxidized sodium alginate and incorporated MXene@CeO₂ nanocomposites and the anti-bacterial activity against both Gram-positive and Gram-negative bacteria is evaluated. The incorporation of MXene@CeO₂ nanocomposites contributed to exquisite antimicrobial efficiency in the multifunctional hydrogen scaffold (FOM) group, revealing a 100% reduction of the bacterial colonies. It is mentioned that the cationic PEI in the FOM attracted the negatively charged bacterial cell membrane, disrupting transmembrane potential, and inducing cell death. Furthermore, incorporating “nano-knife” like MXene@CeO₂ nanocomposites broke the integrity of the cell membrane and resulted in synergistic inhibition of bacterial growth.⁷⁴

Alternatively, ceramic oxide and noble metal nanoparticles are also incorporated over MXene to enhance the optical characteristics of MXene in killing bacteria.⁷⁵ In advance, Ti₃C₂ MXene was modified with Al₂O₃/Ag, SiO₂/Ag, and SiO₂/Pd nanoparticles. The bioactivities of Ti₃C₂ MXene, Ti₃C₂/Al₂O₃/Ag, Ti₃C₂/SiO₂/Ag, and Ti₃C₂/SiO₂/Pd nanocomposites were analyzed qualitatively by an agar diffusion method against both Gram-negative (*Escherichia coli*) and Gram-positive bacteria (*Bacillus* sp., *Staphylococcus aureus*, and *Sarcina lutea*). The anti-bacterial studies in Fig. 8 revealed that the modification of Ti₃C₂ MXene with Al₂O₃ + Ag, SiO₂ + Ag, and SiO₂ + Pd resulted in significantly increased anti-bacterial properties; the results of the experiment with *Staphylococcus aureus* with the application of the dilution method revealed that all the tested samples were able to inhibit the growth of potentially pathogenic bacterial strains. The considered mechanism of action for Ti₃C₂ MXene is related to the direct physical interactions between the edges of the nanosheets and the bacterial cell wall or the membrane surface. As a result, cells lose their integrity, and the internal cytoplasm is released together with DNA.

Cong Liu *et al.*⁶³ integrated a covalent organic framework with MXene structures, anchored Cu₂O nanoparticles on the surface of Ti₃C₂. The prepared Ti₃C₂/TpPa⁻¹/Cu₂O systems exhibited anti-bacterial activity against *S. aureus* and *P. aeruginosa*. It is reported that the enhancement of anti-bacterial activity is because the synergistic effect of Ti₃C₂/TpPa⁻¹/Cu₂O; tppa⁻¹ increased the contact range with bacteria and killed bacteria by generating ROS as it has a vast surface area. Ti₃C₂ assisted the flow of carriers between layers, which improved the separation efficiency of electron-hole pairs and caused more ROS generation. Besides, copper ions released by

Cu₂O denatured the bacterial DNA and exhibited toxicity to bacteria. Rather than Cu₂O metal oxide, recently, CuO was masked over Ti₃C₂ by the ultrasonication route for photocatalytic anti-bacterial applications.⁴⁶ It is revealed that the anti-bacterial activity of CuO/MXene is due to the vital force of attraction generated between the negatively charged bacterial cell membrane and positively charged heavy metal ions present on the sample surface. After penetrating the cell membrane, heavy metal ions react with proteins containing thiol (-SH) groups and inactivate the protein resulting in the death of bacteria. Under the metal oxide section, some rare-earth metal-based oxides are also masked on the surface of MXene. As a promising metal oxide gadolinium (Gd³⁺) doped V₂O₅ (GVO), nanostructures with MXenes are demonstrated to have an anti-bacterial effect on both Gram-positive (*S. aureus*) and Gram-negative (*P. vulgaris*) strains.⁷⁶ GVO/MXene creates reactive oxygen species (ROS) in the presence of light and causes the eradication of RNA, resulting in the death of bacterial strains. It is demonstrated that the efficiency of the prepared nanocomposites was higher than that of the individual nanoparticle. Akin to this, WO₃/MXene nanocomposites also presented as photoinduced anti-bacterial agents against the Gram-positive strain *S. aureus* and Gram-negative strains *E. coli*, *Pneumonia*, and *P. vulgaris* and demonstrated better activity against the Gram-positive strain *S. aureus*. The WO₃/MXene composite showed good activity at a low concentration against pneumonia in Gram-negative strains. It is stated that the WO₃/MXene composite does not exhibit anti-bacterial activity against *E. coli* and *P. vulgaris*, whereas WO₃ and MXene showed good anti-bacterial activity. It is reported that the zero anti-bacterial activity against *E. coli* bacteria and *P. vulgaris* is due to an extra outer membrane that increased the resistance to the WO₃/MXene composite. The increased concentration of composites also decreased the anti-bacterial activity owing to the size and agglomeration. CuFe₂O₄/MXene nanohybrids are also employed for evaluating the anti-bacterial activity against *E. coli*, *Pseudomonas aeruginosa*, *Klebsiella pneumonia*, *Proteus vulgaris*, and *Staphylococcus aureus*. The anti-bacterial studies revealed that composites developed better anti-bacterial performance than individual nanoparticles. It is reported that the anti-bacterial activity is due to the interaction of the positively charged metal particles with the negatively charged bacterial cells.⁶⁹ Unlike metal oxide alone, carbon-based graphite is also used in a hybrid metal oxide-carbon-MXene structure against photocatalytic and bacteriostatic applications. For example, TiO₂ is supported with graphite over Ti₃C₂ MXene, and the hybrid nanostructures revealed exquisite anti-bacterial efficiency against *E. coli*. The nanostructures expressed anti-bacterial activity under light exposure owing to the excessive radicals that cause oxidative stress in bacterial cells. It is reported that controlled oxidation of delaminated MXene structures contributes to anti-bacterial activity and assembly of hybrid nanostructures (TiO₂-Ti₃C₂) with graphitic carbon persuaded the generation of oxidative stress in bacteria to promote cell lysis.⁷⁷ List of various metal oxide-masked MXene in anti-bacterial applications is detailed in Table 2.

Table 2 Summary of various metal oxide masked MXenes in anti-bacterial applications

| S. no. | Metal oxide | MXene | Bacteria | Remarks | Ref. |
|--------|----------------------------------|--------------------------------|---|---|------|
| 1 | NiFe ₂ O ₄ | Ti ₃ C ₂ | <i>Escherichia coli</i> (<i>E. coli</i>) | NiFe/MXene exhibited excellent antibacterial activity against Gram negative <i>E. coli</i> bacteria | 58 |
| 2 | Cu ₂ O | Ti ₃ C ₂ | <i>Staphylococcus aureus</i> (<i>S. aureus</i>) and <i>P. aeruginosa</i> | Ti ₃ C ₂ /TpPa ⁻¹ /Cu ₂ O nanocomposites demonstrated good antibacterial activity of 98.90% and 99.62%, and enhanced by 50% and 33% when compared to pure TpPa ⁻¹ -COF | 63 |
| 3 | MnO ₂ | Ti ₃ C ₂ | <i>Bacillus subtilis</i> and <i>E. coli</i> | Vertically aligned 2D MnO ₂ nanosheets on graphene oxide and Ti ₃ C ₂ MXene revealed maximal antimicrobial activity, signifying that the edges of the nanosheets weakens the bacterial cell | 72 |
| 4 | CeO ₂ | Ti ₃ C ₂ | <i>E. coli</i> (Gram-negative bacteria), <i>S. aureus</i> (Gram-positive bacteria) | MXene@CeO ₂ nanocomposite based multifunctional hydrogel exhibited distinct antibacterial behaviour against <i>E. coli</i> , <i>S. aureus</i> , MRSA and showed 100% reduction in bacterial colonies | 74 |
| 5 | SiO ₂ | Ti ₃ C ₂ | <i>E. coli</i> , <i>Sarcina lutea</i> , <i>S. aureus</i> , and <i>Bacillus</i> sp. | NiFe/MXene exhibited excellent antibacterial activity against Gram negative <i>E. coli</i> bacteria | 75 |
| 6 | TiO ₂ | Ti ₃ C ₂ | <i>E. coli</i> | TiO ₂ -Ti ₃ C ₂ T _x killed 97% <i>E. coli</i> bacteria under light exposure by producing the excessive radicals that cause oxidative stress on bacterial cells | 77 |
| 7 | Cu ₂ O | Ti ₃ C ₂ | <i>S. aureus</i> and <i>P. aeruginosa</i> | Cu ₂ O/MXene nanocomposite displayed good antibacterial activity against <i>S. aureus</i> and <i>P. aeruginosa</i> with the bacteriostasis efficiency of 97.04% and 95.59%, respectively | 63 |
| 8 | WO ₃ | Ti ₃ C ₂ | <i>S. aureus</i> , <i>E. coli</i> , <i>K. pneumonia</i> , <i>P. vulgaris</i> | WO ₃ /MXene nanocomposite expressed good antibacterial activity due to the electrostatic attraction between the nanocomposite and the bacterial cell membrane | 45 |
| 9 | CuFe ₂ O ₄ | Ti ₃ C ₂ | <i>E. coli</i> , <i>Pseudomonas aeruginosa</i> , <i>Klebsiella pneumonia</i> , <i>Proteus vulgaris</i> and <i>S. aureus</i> | CuFe ₂ O ₄ /Ti ₃ C ₂ composites showed good antibacterial activity due to the addition of CuFe ₂ O ₄ | 69 |

4.2 Anti-cancer activity

Cancer is a necrotic epidemic disease with no specific treatment. Massive research has been in progress, working on cancer therapy for decades, with some improvement, yet many drawbacks persist. Conventional treatments like surgery,⁷⁸ chemotherapy,⁷⁹ and radiation therapy⁸⁰ are the most used treatments;⁸¹ they exhibit several limitations, like incomplete removal of tumors, damage to healthy tissue, hair loss, nausea, and bowel issues. To conquer these barriers, nano-materials were predominantly studied as a targeted drug delivery method for cancer therapy.^{82,83} One of these materials is Ti₃C₂, also known as MXene, a 2D material studied for many biomedical applications, tumor detection (*i.e.*, as contrast agents), cancer therapy, and drug delivery.⁸⁴ MXenes are endowed with the following beneficial properties suitable for cancer therapy. (i) In the presence of functional groups, such as -OH, -O, or -F, MXenes are hydrophilic in nature,^{24,26} which is favourable for surface modification. (ii) The large surface area makes the MXene host massive molecules for synergistic therapy, including immune adjuvants,⁸⁵ photosensitizers (PS),⁸⁶ and chemotherapeutic drugs.⁸⁷ (iii) Several compositions of MXenes, including Ti₂C, Ti₃C₂, Nb₂C, and Mo₂C, have been tested to be non-toxic and biocompatible with living organisms. Excellent biocompatibility might be due to the metal in the "M" layer, like Ti, Mo, and Nb, being relatively inert to living organisms. Other significant elements, such as nitride and carbon, are essential in the structure of biological organisms. (iv) They exhibit substantial plasmonic effects in the NIR region, making them more favourable for both *in vivo* photothermal (PTT) and photoacoustic (PA) imaging in the

first or the second biological window. Therefore, MXenes were studied by many researchers for cancer photothermal therapy by killing cancer cells by heat, leading to protein denaturation and cell death.⁸⁸ It was reported that MXenes with a size of around 180 nm could explicitly target and reach the cancerous microenvironment by enhanced permeability and retention of EPR.⁸⁹ Although MXenes showed good photothermal efficiency that makes them a potential candidate for photothermal therapy, the internalization of MXenes into cancer cells was not studied extensively.

Lin *et al.*⁹⁰ applied MXenes to treat tumors using PTT for the first time. Typical representatives of MXenes, such as Ti₃C₂, Nb₂C, Ta₄C₃, and Ti₂C, are endorsed to have excellent photothermal properties and are used in PTT. Using Ti₃AlC₂ powder as the raw material, the 2D nanosheets of Ti₃C₂ were functionalized with soybean phospholipid (SP) for high-efficiency PTT of tumors. It is revealed that introducing SP enhanced the stability of Ti₃C₂ nanosheets in various dispersions and improved their applicability in tumor diagnosis and treatment. Furthermore, the composite nanosheet Ti₃C₂-SP had excellent PTT performance, as Ti₃C₂-SP at a low concentration of 72 µg ml⁻¹ could be heated to above 55 °C in 6 min, which fulfilled the temperature requirements of PTT for the ablation of tumor tissue. Zhuang Liu *et al.*⁹¹ prepared (Ta₄C₃)-based MXene nanosheets anchored with iron oxide nanoparticles modified using soybean phospholipid (SP) to increase the stability. The prepared Ta₄C₃-IONP-SPs were evaluated for MR/CT dual-modality imaging and photothermal hyperthermia of breast cancer. Without NIR radiation, the prepared Ta₄C₃-IONP-SP nanosheets failed to exhibit cytotoxicity on breast 4T1 cancer cells at different concentrations (0, 12.5,

25, 50, 100, 200 ppm). Upon NIR radiation for 5 min at the power density of 1.5 W cm^{-2} , the photothermal-killing effect was observed with increased Ta_4C_3 -IONP-SP concentrations, where more cancer cells were ablated and then killed under NIR irradiation. The therapeutic efficacy of the Ta_4C_3 -IONP-SP MXene *in vivo* was determined in 4T1 tumor-bearing mice and is represented in Fig. 10. It is observed that the temperature of the tumors from mice in post-injection of Ta_4C_3 -IONP-SPs increases from about 34°C to 48°C within 10 min, which is adequately high to ablate cancer cells and tumor tissues. Two days after photothermal hyperthermia, the initial tumors disappeared in the group injected with Ta_4C_3 -IONP-SPs under laser irradiation, leaving black scars at tumor sites. It was found that around 1.02% Ta_4C_3 -IONP-SPs accumulated in the tumor tissue passively due to the EPR effect. Retention of Ta_4C_3 -IONP-SPs in the liver and spleen may be associated with the significant macrophage uptake of nanoparticles in reticuloendothelial systems. Primarily, the high photothermal-conversion efficiency of Ta_4C_3 -IONP-SP composite nanosheets (η : 32.5%) has accomplished complete tumor eradication without reoccurrence, demonstrating the highly efficient breast-tumor hyperthermia performance.

2D niobium carbide (Nb_2C) MXene was first reported by Huijing Xiang *et al.*⁹² via a two-step chemical exfoliation strategy for photothermal conversion in the NIR-II bio window, which demonstrated excellent *in vivo* photothermal ablation capability in the NIR-II bio window. The $\text{Nb}_2\text{C}@m\text{SiO}_2$ nanosystem reported by the same group was loaded with AIPH

molecules as the free-radical source. Upon 1064 nm NIR-II laser irradiation, the heating of nanoparticles (NPs) triggered the rapid release and decomposition of encapsulated initiators to generate free radicals independent of oxygen levels, resulting in significant *in vitro* cell apoptosis and *in vivo* tumor eradication, which is herein regarded as the cancer-therapeutic modality of “thermodynamic therapy”. The synergistic therapeutic efficiency of the photothermal effect was exhibited. Dong Yang Zhang *et al.*⁹³ prepared TiO_{2-x} decorated titanium carbide MXene for photoacoustic/photothermal bimodal imaging guided near-infrared II (NIR-II) photothermal enhanced SDT of the tumor.

Upon ultrasound/NIR-II radiation, $\text{Ti}_3\text{C}_2@\text{TiO}_{2-x}$ nanoplateform provoked substantial cellular killing *in vitro* and complete tumor eradication *in vivo* without recurrence of tumors and systemic toxicity. MXene-based multifunctional composites, $\text{Fe}_3\text{O}_4/\text{MnO}_x\text{-Nb}_2\text{C}$, were employed for the tumor-responsive T_1 and T_2 MRI-guided photothermal ablation of breast cancer in the NIR-II biowindow. The Fe_3O_4 composites act as T_2 contrast agents for T_2 -weighted MR imaging. This composite nanosheet was modified with soybean phospholipid to enhance the stability, biosafety in physiological environments, and long circulation time. It is reported that the high photothermal-conversion efficiency of $\text{Fe}_3\text{O}_4/\text{MnO}_x\text{-Nb}_2\text{C}$ composite MXenes (η = 30.9%) dexterously ablated tumor cells without further reoccurrence both *in vitro* and *in vivo* after laser irradiation in the second biological window (NIR-II, 1064 nm).⁹⁴ Two-dimensional tantalum carbide was employed for the photothermal cancer ablation. Manganese oxide nanoparticles were grown on the surface of Ta_4C_3 by the redox reaction between the MXene surface and strongly oxidative MnO_4 , and the surface of the MXene composites was modified with soybean phospholipid to enhance biosafety. The composite exhibited good photothermal conversion efficiency for ablating the tumor cells and acted as a contrast agent. It is indicated that tantalum-based and MnO_x components in the prepared composites acted as high-performance contrast agents for simultaneous computed tomography and T_1 -weighted magnetic resonance imaging, respectively. It is revealed that the composites showed significant tumor growth suppression by photothermal hyperthermia.⁷⁰ MXene composites and metal oxide composites were utilized for magnetic hyperthermia. Using the redox reaction-induced growth, MnO_x was grown on titanium carbide MXene sheets. The incorporation of MnO_x on the surface of nanosheets acts as the pH-responsive contrast agent for T_1 -weighted MR imaging. Simultaneously the ultra-thin Ti_3C_2 nanosheets revealed high photothermal-conversion performance for efficient thermal ablation of tumors *via* PTT.⁹⁰ A summary of various metal oxide-masked MXenes in anti-cancer activities is listed in Table 3.

4.3 Sensing applications

The unique physical, chemical, and ion transport properties of MXenes make them an excellent choice in many applications, exclusively in sensing^{95,96} and biosensing.^{97–99} MXenes feature excellent biocompatibility and allow secure immobilization of

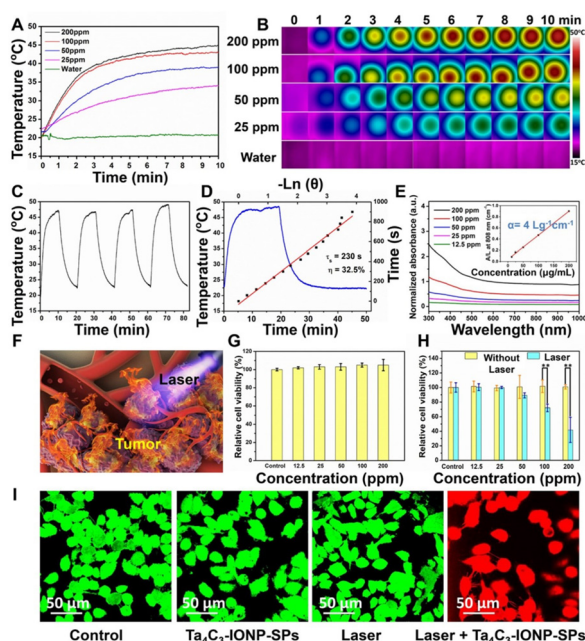


Fig. 10 (A) Photothermal and therapeutic evaluation profiles of Ta_4C_3 -IONP-SPs under a laser, (B) the corresponding infrared thermal images. (C) Heating/cooling, (D) temperature-changing profile, (E) UV-vis spectra and (F) 3D-schematic illustration of Ta_4C_3 -IONP-SPs as PTAs for 4T1 cells. (G) Relative viability and (I) confocal fluorescence images. (Reproduced with permission from ref. 91 Copyright 2018, IVY Spring.)

Table 3 Summary of various metal oxide masked MXenes in cancer therapeutics

| S. no. | Metal oxide | MXene | Light | Power | Summary | Ref. |
|--------|--------------------------------|--------------------------------|--------|------------------------|--|------|
| 1 | Iron oxide | Ta ₄ C ₃ | NIR I | 1.5 W cm ⁻² | Ta ₄ C ₃ -IONP-SPs composite MXenes achieved complete tumor eradication with high photo-thermal-conversion efficiency (η : 32.5%) | 124 |
| 2 | SiO ₂ | Nb ₂ C | NIR II | 1 W cm ⁻² | SiO ₂ /Nb ₂ C MXene showed that a better photothermal-conversion effect provoked cancer cell apoptosis | 92 |
| 3 | TiO ₂ | Ti ₃ C ₂ | NIR II | 1 W cm ⁻² | Titanium carbide, Ti ₃ C ₂ showed good photothermal efficiency and demonstrated complete tumor elimination in 4T1 tumor bearing mice | 93 |
| 4 | Fe ₃ O ₄ | Nb ₂ C | NIR II | 1.5 W cm ⁻² | Fe ₃ O ₄ /MnO _x -Nb ₂ C composite nanosheets expressed high photothermal-conversion efficiency in the NIR-II biowindow (1064 nm, η 30.9%) | 94 |
| 5 | MnO _x | Ti ₃ C ₂ | NIR I | 1 W cm ⁻² | MnO _x /Ti ₃ C ₂ composite MXenes exhibited excellent contrast-enhanced PA-imaging properties and demonstrated efficient tumor ablation and tumor-growth suppression | 90 |

enzymes and proteins on their surface.^{100–102} They possess a large specific surface area, anisotropic charge carrier conductivity, environmental friendliness, and excellent chemical stability.¹⁰³ Surface dangling bonds enable MXenes to attract various functional groups, making them unique for sensing applications. Metal oxides are used to fabricate a heterostructure or composite with MXenes to enhance sensing quality. Metal oxides have various beneficial properties like high thermal and chemical stability,³³ substantial mechanical strength,¹⁰⁴ sizeable piezoelectric coefficient,¹⁰⁵ high exciton binding energy, and optical gain,¹⁰⁶ making them an ideal material for the sensor.¹⁰⁷

As depicted in Fig. 11,¹⁰⁸ hybrid composites as heterojunctions using partially-oxidized Ti₃C₂ sheets and photo-active NiWO₄ nanoparticles (NPs) are prepared for selective sensing of prostate-specific antigens (PSA). Electrochemical and photo-electrochemical (PEC) studies revealed a remarkable electrochemical response, with a redox couple shift to 0.27 V/0.15 V and a two-fold rise in the measured current output (1.2 μ A). It is mentioned that the predominant performance of MX-NiWO₄ is due to the increased surface area, high conductivity of Ti₃C₂, and exposed surface-active sites of MX-NiWO₄. The ideal interfacial arrangement of MX-NiWO₄ assisted in the electrocatalytic mechanism-based PEC immuno-sensing of PSA. The developed PEC biosensor's detection range was from 1.2 fg mL⁻¹ to 0.18 mg mL⁻¹ with a detection limit of 0.15 fg

mL⁻¹. In another study,¹⁰⁹ a TiO₂/Ti₃C₂/Cu₂O heterostructure was fabricated by *in situ* generations of TiO₂ NPs on Ti₃C₂ MXenes and the introduction of Cu₂O *via* a one-step hydrothermal reaction. The impact of Ti₃C₂, as a mediator between TiO₂ and Cu₂O, adequately accelerated the charge transfer and prolonged photoelectrons lifetime, which made the fabricated sensor extremely sensitive and conductive. It is proven that heterojunction demonstrated good PEC activity with dissolved O₂ as electron scavengers, during which H₂O₂ is produced as an intermediate to further trap the photogenerated electrons. Glucose detection was observed in the change in the photocurrent of TiO₂/Ti₃C₂/Cu₂O, decreasing photocurrent with the increased glucose concentration in a wide range from 100 nM to 10 μ M with a low detection limit of 33.75 nM (S/N = 3).

To uphold the superior sensing characteristics of MXenes, metal oxide-based MXene heterojunctions⁶⁷ are fabricated by electrospinning FeWO₄ bimetallic nanofibers tagged on delaminated single-layered Ti₃C₂ MXenes for the detection of rutin. The heterostructure revealed favourable electrocatalytic properties for rutin oxidation due to its electroactive interconnecting network of single-layered MXene with metal-oxide fibers having high defective edge/plane sites. Compared to bare glassy carbon electrodes (GCE) and composites, the prepared structures exhibited superior performance. Volumetric studies of the fabricated heterostructure displayed an ultralow detection limit of 0.42 nM, an ultrawide linear determination range from 4 to 147 nM, and a high sensitivity of 0.3799 μ A nM⁻¹ cm⁻². The electrochemical studies of the MXene-FeWO₄ nanocomposite showed favourable stability and persistent anti-interference ability. The prepared nanocomposite revealed excellent activity in detecting RT in human serum, orange juice, and black tea samples.

Interestingly, hybrid combinations of¹¹⁰ titanium dioxide/MXene and polyvinyl alcohol/graphene oxide (TiO₂/MXene-PVA/GO) composites are prepared to modify a screen-printed carbon electrode (SPCE) for urinary norepinephrine (NE) detection. The as-prepared composite substantially upgraded the sensor performances due to the favourable electrocatalytic activity of TiO₂, high conductivity of MXene, and auto-sample preconcentration *via* the PVA/GO hydrogel. Cyclic voltammetry and amperometry studies revealed that the composites expressed their electrochemical response at +0.4 V, and oxidation of NE is proportional to NE. The contribution of the

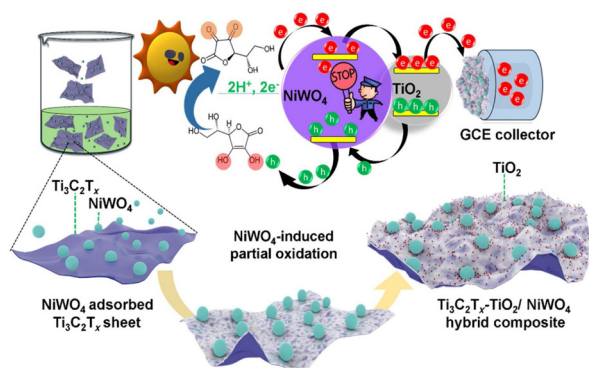


Fig. 11 Schematic illustration of mechanisms of surface adsorption of NiWO₄ NPs over ultra-thin Ti₃C₂T_x sheets. (Reproduced with permission from ref. 108 Copyright 2022, Elsevier.)

Table 4 Summary of various metal oxide masked MXenes in sensor applications

| S. no. | Metal oxide | MXene | Analyte | Remarks | Ref. |
|--------|--------------------------------------|--------------------------------|-------------------------------|--|------|
| 1 | NiWO ₄ & TiO ₂ | Ti ₃ C ₂ | Prostate-specific antigens | Ti ₃ C ₂ T _x -TiO ₂ /NiWO ₄ hybrid composite showed photoelectrochemical sensing of prostate specific antigen, over a wide detection range from 1.2 fg mL ⁻¹ to 0.18 mg mL ⁻¹ , with a detection limit of 0.15 fg mL ⁻¹ | 108 |
| 2 | TiO ₂ | Ti ₃ C ₂ | H ₂ O ₂ | TiO ₂ nanoparticle modified Ti ₃ C ₂ MXene nanocomposite used for the detection of H ₂ O ₂ with a low detection limit of 14 nM, a wide linear range of 0.1–380 μM for H ₂ O ₂ , and especially, an excellent long-term stability | 111 |
| 3 | TiO ₂ /Cu ₂ O | Ti ₃ C ₂ | Glucose | The TiO ₂ /Ti ₃ C ₂ T _x /Cu ₂ O composite was probed as a photoelectrochemical sensing system for glucose detection in a linear range from 100 nM to 10 μM with a low detection limit of 33.75 nM | 109 |
| 4 | FeWO ₄ | Ti ₃ C ₂ | Rutin | The MXene/FeWO ₄ composite showed excellent selectivity for Rutin detection among a series of interfering ions and biomolecules | 67 |
| 5 | TiO ₂ | Ti ₃ C ₂ | Norepinephrine | The TiO ₂ /MXene-PVA/GO based hydrogel used for the detection of norepinephrine with a detection limit (3σ) of 6 nM | 110 |
| 6 | TiO ₂ | Ti ₃ C ₂ | Glucose | TiO ₂ -Ti ₃ C ₂ was employed as an electrogenerated chemiluminescence biosensor for the detection of glucose in the wide concentration range of 20 nM–12 mM with a low detection limit of 1.2 nM | 125 |
| 7 | TiO ₂ | Ti ₃ C ₂ | Dopamine & uric acid | TiO ₂ /Ti ₃ C ₂ demonstrated good selectivity and reproducibility for dopamine and uric acid detection in urine and serum samples with recoveries of 98.4 to 100.9% | 112 |
| 8 | TiO ₂ | Ti ₃ C ₂ | Glutathione | The TiO ₂ /Ti ₃ C ₂ heterostructure film exhibited high stability, sensitivity, and selective detection of glutathione | 126 |
| 9 | ZnO | Ti ₃ C ₂ | Glucose | ZnO/MXene was employed to prepare a skin-attachable and stretchable electrochemical enzymatic sensor for the detection of glucose and exhibited enhanced sensitivity in sweat samples (29 μA mM ⁻¹ cm ⁻²), low limit of detection (LOD ≈ 17 μM), and a broad linear detection range (LDR = 0.05–0.7 mM) that satisfies glucose detection application in human sweat | 127 |

PVA/GO hydrogel is shown in the increased surface area and sample volume absorption, increasing sensor electrochemical signals and sensitivity.

Similarly,¹¹¹ TiO₂ nanoparticle-modified organ-like Ti₃C₂ MXene (TiO₂-Ti₃C₂) nanocomposite is used to immobilize hemoglobin (Hb). Spectroscopic and electrochemical studies demonstrated that the TiO₂-Ti₃C₂ nanocomposite is an excellent immobilization matrix biocompatible with redox proteins, affording good protein bioactivity and stability. The direct electron transfer of Hb due to the unique organ-like hybrid structure of TiO₂-Ti₃C₂ improved the detection of H₂O₂ with a wide linear range of 0.1–380 μM for H₂O₂ (sensitivity of 447.3 μA mM⁻¹ cm⁻²), and a meager detection limit of 14 nM for H₂O₂. It is reported that incorporating TiO₂ nanoparticles enhanced biocompatibility, bioactivity, and prolonged stability. In addition to TiO₂, masked MXene sensors,¹¹² TiO₂/Ti₃C₂ nanocomposite, along with covalent organic polymers (NUF), are prepared for the detection of dopamine (DA) and uric acid (UA). The differential pulse voltammetry studies revealed that TiO₂/TiCT/NUF expressed high sensing activity with low detection limits of 0.2 and 0.18 nM (S/N = 3) in the concentration ranges from 0.002 to 100 μM and 0.001 to 60 μM for simultaneous determination of DA and UA, respectively. The simultaneous detection is achieved due to the abundant redox sites, large electroactive area, and heterostructure's enhanced electron capacity. Different combinations of metal oxide masked MXenes for sensor applications are listed in Table 4.

4.4 Other biomedical applications

Like other fascinating applications, tissue engineering and regenerative medicine also extended the fields of MXenes due

to their remarkable features like conductivity, large surface area, hydrophilicity, two-dimensional geometry, and particle size regulation. MXenes combined with polymers and hydrogels are widely explored in tissue engineering, regenerative medicine and wound dressing because of their immunomodulatory and anti-inflammatory properties.^{113,114} Jacek Wychowaniec *et al.* developed an rGO-MXene hydrogel as a cellular network for tissue engineering applications. It is reported that the rGO-based hydrogel without MXene had a hydrophobic surface with an elastic modulus of 40k Pa and rGO-MXene-based hydrogel expressed a hydrophilic surface, softer with an elastic modulus of 20 kPa. MXene Ti₂C, along with cryogen, was reported for cardiac tissue engineering, *i.e.* Ti₂C proportion provided enough mechanical properties and an appropriate conductivity to match the natural myocardium and promoted the repair of myocardial infarction. MXenes are used in wound healing along with hydrogels under electric stimulation as they have the ability to enhance the electrophysiological properties of wound dressing materials.^{115,116} Besides, MXene coupled with CeO₂-based nanozyme was reported to have a synergistic effect in hyperthermia-enhanced tumour nano catalytic therapy.¹¹⁷ Though MXenes were employed in all these applications, metal oxide-masked MXenes have not been explored much.

As described above, various metal oxides were utilized for effectively improving the optical properties of MXenes for various biomedical applications. However, depending on the metal oxide type, the applications vary. For example, TiO₂ is a known photocatalyst and is effectively utilized for various ROS-generating applications, such as anti-bacterial and anti-cancer, since its Ti²⁺ ionic form exhibits ROS-generation activities.

Similarly, copper and cobalt-masked MXenes are also used since they help generate ROS in the presence of external light. On the other hand, metal oxides such as cerium and manganese exhibit concentration-dependent ROS scavenging activities; these nanoformulations were used in anti-oxidant applications. So, the type of application was determined by the type of metal oxide incorporation.

5. Scope and perspectives

MXenes have attractive and tunable properties, but limitations like easy oxidation and restacking are encountered. In the context of MXenes and metal oxide incorporated MXenes, stability issues often arise due to their sensitivity to oxidation. The terminal surface functional groups of MXenes ($-\text{OH}$, $-\text{F}$) are particularly susceptible to oxidation. The high reactivity of MXenes, coupled with the presence of defects and impurities, may contribute to this lack of stability. Additionally, improper mixing of metal oxides with MXenes can result in a lack of structural stability. To enhance and improve the stability of MXenes and metal oxide incorporated MXene, various approaches can be employed, including surface modifications, protective coatings, and suitable composites.

Overcoming these limitations is crucial to exploring the optical properties of MXenes for biomedical applications. Metal oxides have a high surface area, biocompatibility, interesting redox, catalytic, and semiconducting activity, and good mechanical stability. The use of metal oxide masked MXenes for clinical applications is still growing. Some studies and factors must be explored for improved diagnostic and therapeutic applications. (i) Enhancing biocompatibility, including biodegradability, biodistribution, and toxicity, using the incorporation of metal oxides based on surface modification, concentration, and structure. (ii) The interaction between metal oxide masked and biological organisms, *i.e.*, the immunological effect of MXenes and the immune system, *in vivo* studies need to be conducted to optimize the size, shape, surface, and physical and chemical properties of metal-oxide masked MXenes. (iii) The incorporation or encapsulation of the drug in metal oxide masked MXenes, and their performance must be studied for theranostic applications. (iv) Optimizing the optical properties and catalytic reactions of MXenes using metal oxides must be explored widely for enhancing the sensing applications. (v) The impact of incorporating different structures of metal oxides like rods, spheres, sheets, and quantum dots towards clinical applications can be prioritized in future research. More research will develop a different combination of metal oxide-masked MXenes to enhance diagnostic and therapeutic performances. With the exceptional properties of MXenes and the potential of metal oxides, enormous improvement can be implemented in developing cancer therapies, drug delivery platforms, biosensors, and bioelectronics.

Like most inorganic nanosystems, the poor biodegradability of 2D MXene nanosheets is the pivotal issue, which may hinder their further oncological research and clinical trans-

lation. Therefore, it is essential to systematically evaluate the biodegradation performance of MXene nanosheets in a complex physiological environment. To date, several MXenes have undergone individual studies and case-by-case phenomenon evaluations, including Ti_3C_2 ,¹¹⁸ Ta_4C_3 ,¹¹⁹ Nb_2C ,¹²⁰ MoC_2 ,¹²¹ V_2C ,¹²² and TiN .¹²³ However, the biological effects of MXenes are only preliminarily assessed at the cellular and animal level. Currently, reliable data on the long-term biological effects and biosafety are still highly lacking, which requires further systematic and in-depth evaluation and assessment. Although many studies have demonstrated the great potential of MXenes in biomedical applications, their clinical translation is still in its infancy. Therefore, the systematic investigation of the interaction between MXenes and biological microenvironments should be systematically investigated in the following fundamental research.

Additionally, the effect of surface functionalization on the biological behavior of MXenes requires further investigation. Significantly, the scale-up fabrication of 2D MXenes is a significant aspect for their design and engineering. A further and deeper exploration of materials science and clinical translation requires cooperation among researchers of interdisciplinary and industrial sectors. It is highly expected that the development of nanotechnology and bioscience will achieve more fundamental and technological breakthroughs to afford the limitless application of MXenes in varied biomedical areas shortly, provided the challenges facing and critical issues are adequately solved.

6. Conclusion

This review has emphasized the significance of incorporating metal oxides into the MXene framework for biomedical applications. According to the literature survey, metal oxides have played a crucial role in enhancing the high surface area, electrical conductivity, stability, and sensitivity of MXenes. It is evident that the incorporation of metal oxides into the MXene structure holds great promise for future biomedical applications. However, the mechanism of interaction between metal oxides and MXenes is still in its infancy, requiring further investigation. This compilation underscores the need for more research on utilizing metal oxide/MXene nanocomposites for various biomedical applications. The biomedical applications of 2D MXenes face limitations such as easy aggregation, complex surface engineering, inadequate stability, and potential toxicity. Factors such as lateral size, the number of layers, and the degree of oxidation significantly influence degradation and stability during *in vivo* circulation.

The interaction between MXene nanosheets and the physiological environment, the analysis of metabolic pathways, and the potential side effects of MXene nanosheets remain unclear. There exists a notable paradox where rapid degradation may affect our desired therapeutic outcomes while meeting biological safety requirements. Therefore, the appropriate degradation rate should be considered in the design

and synthesis of MXenes to simultaneously pursue ideal therapeutic outcomes and low toxicity.

This review lays a solid foundation for further research by providing technical details and in-depth insights that warrant focused attention. A thorough investigation of these aspects will provide excellent solutions to the challenges faced in biomedical applications. Additionally, further research in the rapid assessment of nanomaterials for biomedical applications within a microfluidic platform is anticipated.

Conflicts of interest

There are no conflicts to declare.

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