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# KHA model comprising $\text{MoS}_4$ and $\text{CoFe}_2\text{O}_3$ in engine oil invoking non-similar Darcy–Forchheimer flow with entropy and Cattaneo–Christov heat flux†

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**Objective:** Nanoliquid flows are widely utilized in industrial, petroleum, engineering, and pharmaceutical applications including electric cooling, drug delivery, nuclear reactor cooling, solar collectors, heat exchangers, magnetohydrodynamic power generators, aerospace, porous media, thermal storage systems, and many others. Darcy–Forchheimer magnetized hybrid nanoliquid subjected to a stretchable cylinder was addressed, and the Cattaneo–Christov heat flux analysis was considered. Herein, disulfido (dithioxo) molybdenum ( $\text{MoS}_4$ ) and cobalt ferrite ( $\text{CoFe}_2\text{O}_4$ ) were considered as nanoparticles, and engine oil as a conventional liquid. The thermal relationship of heat generation and radiation was discussed, and the influence of the entropy rate was addressed. **Methodology:** Governing expressions were transformed into dimensionless forms. Simulation by the ND-solve technique was implemented. **Conclusions:** Features for the entropy rate, liquid flow, and temperature against emerging variables for nanoliquid ( $\text{MoS}_4$ /engine oil) and hybrid nanoliquid ( $\text{MoS}_4 + \text{CoFe}_2\text{O}_4$ /engine oil) were explored. The numerical results of the coefficient of skin friction and thermal transport rate for nanoliquid ( $\text{MoS}_4$ /engine oil) and hybrid nanoliquid ( $\text{MoS}_4 + \text{CoFe}_2\text{O}_4$ /engine oil) were examined. Reduction in velocity clearly occurred through a magnetic field, whereas the reverse impact held for the entropy rate. The thermal field and entropy rate against the curvature parameter were enhanced. A decrease in liquid flow occurred for higher porosity variables. An enhancement in the entropy rate was witnessed for radiation and porosity parameters. Higher radiation and thermal relaxation time variables resulted in enhancement of the thermal transport rate.

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## 1. Introduction

Considerable research has been conducted on nanomaterials due to their significance in pharmaceutical, industrial, chemical, and engineering activities. Nanomaterial is basically created by the insertion of small-size (1–100 nm) particles in conventional liquid, which results in thermal conductivity enhancement. Nanomaterials play a key role in heat-related equipment, cooling and heating systems, radiators, nuclear reactors, fuel chambers, space technology, and caloric control. The initial investigations into nanofluids were conducted by Choi<sup>1</sup> and Eastman *et al.*<sup>2</sup> The distributed ultrafine particles efficiently strengthen the viscosity and thermal conductivity of nanomaterial and enhance its competence in energy exchange.<sup>3,4</sup> Thermal conductivity and viscosity may vary with

the temperature of the nanofluid mixture, nanoparticle size, and volume fraction.

The thermal conductivity of metallic nanoparticles is higher than that of conventional materials, such as water and nanoparticles with a lower density and thermal conductivity than that of traditional heat transfer fluids (HTFs). Sivasankaran *et al.*<sup>5</sup> addressed magnetohydrodynamic hybrid nanoliquid flow between stretchable parallel plates. Heat transport in the electromagnetohydrodynamic flow of Casson nanoliquid considering the heat source/sink was explored by Hussain *et al.*<sup>6</sup> Mixed convective flow of hybrid nanomaterial was explored by Patil and Shankar.<sup>7</sup> Çiçek *et al.*<sup>8</sup> explored the convective flow of hybrid nanofluid considering particle deposition inside a square cavity. There have been numerous investigations that have been conducted to explore nanofluid flow.<sup>9–13</sup>

A hybrid nanoliquid is the combination of two or more small size metallic nanoparticles with different chemical characteristics in conventional materials. Hybrid nanomaterials are useful to augment the advantages of nanoliquid such as heat transport transmission and enhancement processes. Because hybrid nanomaterials are advantageous for thermal conductivity enhancement, they are widely used in nuclear reactor

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cooling processes, biomedical applications, solar energy, the automotive industry, automobile generators, space technology, and heat exchangers.

Thermal enhancement of convective material by insertion of single particles and hybrid nanoparticles was given in Jana *et al.*<sup>14</sup> Bhatti *et al.*<sup>15</sup> discussed the magnetohydrodynamic flow of hybrid nanomaterial by a circular non-Darcy surface. Wahid *et al.*<sup>16</sup> analyzed heat transfer in hybrid nanomaterial subject to a Riga plate. Slip impacts in the convection stagnation point flow of hybrid nanoliquid were reported by Zangoee *et al.*<sup>17</sup> Khan *et al.*<sup>18</sup> explored entropy in a radiating flow of hybrid nanoliquid through the Darcy–Forchheimer relation. Zainal *et al.*<sup>19</sup> deliberated the influence of Arrhenius kinetics in hybrid nanomaterial flow by a shrinking and stretching surface with radiation. It should be noted that there have been few attempts to study hybrid nanoliquid flow.<sup>20–30</sup>

Entropy generation is the measurement of energy wastage during any thermal process. Entropy always increases in an irreversible process, and therefore, entropy generation is a positive quantity. Entropy generation is also a non-conserved property. There is no existence of conservation of entropy. Therefore, the entropy of the universe is continuously increasing. Joule heating, fluid flow friction between solid surfaces, molecular vibration, liquid viscosity, and diffusion are sources that produce entropy generation. In fact, Bejan discovered entropy optimization for convective flow with thermal convection.<sup>31,32</sup> Entropy for the radiative magnetized flow of a hybrid nanoliquid due to a stretchable rotating disk is presented in Khan *et al.*<sup>33</sup> Kumawat *et al.*<sup>34</sup> analyzed magnetized entropy-optimized flow with variable viscosity. Nonlinear

radiation in hybridized nanoliquid flow with a convectively heated surface was explored by Ashwinkumar *et al.*<sup>35</sup> Rajkumar *et al.*<sup>36</sup> examined Cattaneo–Christov flux in the radiative flow of micropolar nanofluid with entropy optimization. Similar studies on entropy have been conducted.<sup>37–42</sup>

The main theme here is to communicate the Cattaneo–Christov heat flux for the magnetohydrodynamic flow of hybrid ( $\text{MoS}_4 + \text{CoFe}_2\text{O}_4/\text{engine oil}$ ) nanoliquid towards a stretchable cylinder. The Darcy–Forchheimer relation is used for porous space. Disulfido (dithio) molybdenum ( $\text{MoS}_4$ ) and cobalt ferrite ( $\text{CoFe}_2\text{O}_4$ ) are considered as nanoparticles, and engine oil as a conventional fluid. Cattaneo–Christov heat flux was deliberated, and radiation and heat generation were considered. Entropy optimization was also considered. Related equations were converted into dimensionless versions. The ND-solve technique was implemented for solution of nonlinear non-dimensional systems. Fluid flow, entropy rate, and temperature *via* the involved variables for nanoliquid ( $\text{MoS}_4/\text{engine oil}$ ) and hybrid nanoliquid ( $\text{MoS}_4 + \text{CoFe}_2\text{O}_4/\text{engine oil}$ ) were graphically analyzed. The numerical results illustrating the impacts of influential variables on physical quantities for nanoliquid ( $\text{MoS}_4/\text{engine oil}$ ) and hybrid nanoliquid ( $\text{MoS}_4 + \text{CoFe}_2\text{O}_4/\text{engine oil}$ ) are presented.

## 2. Formulation

The hydromagnetic flow of hybrid ( $\text{MoS}_4 + \text{CoFe}_2\text{O}_4/\text{engine oil}$ ) nanomaterial by stretchable porous cylinder was examined. The Darcy–Forchheimer expression was used for porous space. Cattaneo–Christov heat flux was carried out. Disulfido

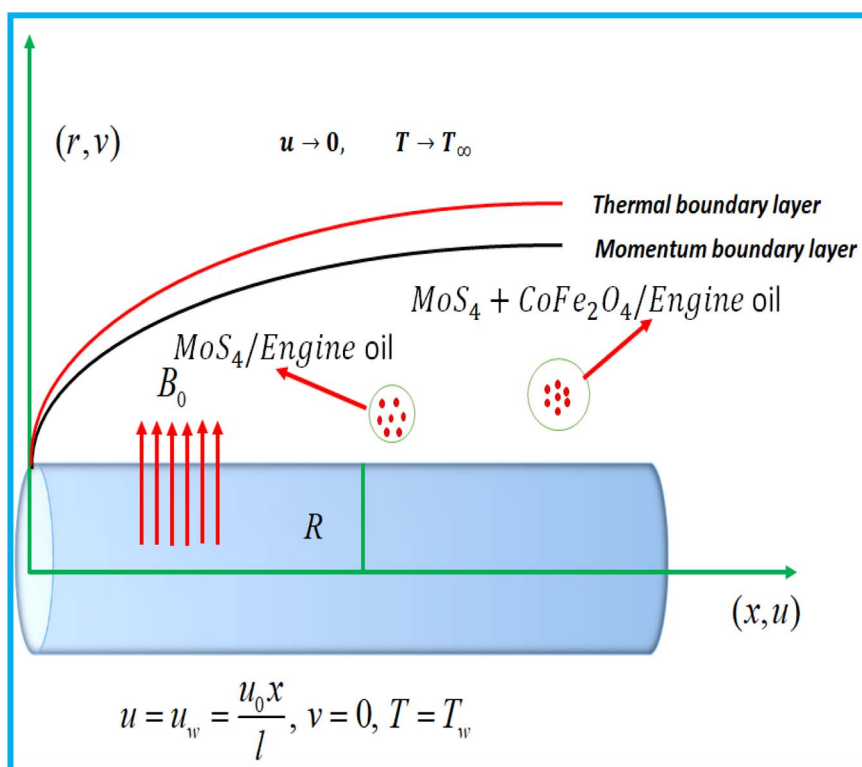


Fig. 1 Physical model.



Table 1 Values of nanoparticles and base liquid<sup>50–53</sup>

Physical property	$\rho_f$ (kg m <sup>-3</sup> )	$(C_p)_f$ (J kg <sup>-1</sup> K <sup>-1</sup> )	$\sigma_f$ (S m <sup>-1</sup> )	$k_f$ (W m <sup>-1</sup> K <sup>-1</sup> )
Engine oil	884	1910	$10^{-11}$ to $2 \times 10^{-9}$	0.1410
MoS <sub>4</sub>	5060	397.21	$2.09 \times 10^4$	904.4
CoFe <sub>2</sub> O <sub>4</sub>	4907	700	$5.51 \times 10^9$	3.7

(dithioxo) molybdenum (MoS<sub>4</sub>) and cobalt ferrite (CoFe<sub>2</sub>O<sub>4</sub>) were considered as nanoparticles, and engine oil as a conventional material. Radiation and heat generation were present in the energy equation, and entropy modeling was carried out. A constant magnetic field ( $B_0$ ) perpendicular to the liquid flow was applied. The radius of a cylinder was considered as ( $R$ ), and stretching velocity ( $u = u_w = \frac{u_0 x}{l}$  ( $l$  shows characteristics length)). The flow configuration is sketched in Fig. 1.

Under these considerations, the related expressions are:<sup>43–48</sup>

$$\frac{\partial(rv)}{\partial r} + \frac{\partial(ru)}{\partial x} = 0, \quad (1)$$

$$u \frac{\partial u}{\partial x} + v \frac{\partial u}{\partial r} = \frac{\mu_{\text{hnf}}}{\rho_{\text{hnf}}} \left( \frac{\partial^2 u}{\partial r^2} + \frac{1}{r} \frac{\partial u}{\partial r} \right) - \frac{\sigma_{\text{hnf}} B_0^2}{\rho_{\text{hnf}}} u - \frac{\mu_{\text{hnf}}}{\rho_{\text{hnf}}} \frac{1}{k_p} u - F u^2, \quad (2)$$

$$\left. \begin{aligned} & u \frac{\partial T}{\partial x} + v \frac{\partial T}{\partial r} - \frac{1}{(\rho c_p)_{\text{hnf}}} \frac{16\sigma^* T_\infty^3}{3k^*} \left( \frac{1}{r} \frac{\partial T}{\partial r} + \frac{\partial^2 T}{\partial r^2} \right) - \frac{Q_0}{(\rho c_p)_{\text{hnf}}} (T - T_\infty) \\ & + \delta_E \left( u^2 \frac{\partial^2 T}{\partial x^2} + u \frac{\partial u}{\partial x} \frac{\partial T}{\partial x} + u \frac{\partial v}{\partial x} \frac{\partial T}{\partial r} + 2uv \frac{\partial^2 T}{\partial x \partial r} + v \frac{\partial u}{\partial r} \frac{\partial T}{\partial x} + v \frac{\partial v}{\partial r} \frac{\partial T}{\partial r} + v^2 \frac{\partial^2 T}{\partial r^2} \right) \\ & - \delta_E \frac{1}{(\rho c_p)_{\text{hnf}}} \frac{16\sigma^* T_\infty^3}{3k^*} \left( u \frac{\partial^2 T}{\partial x \partial r} + u \frac{\partial^3 T}{\partial x \partial r^2} - \frac{v}{r^2} \frac{\partial T}{\partial r} + \frac{v}{r} \frac{\partial^2 T}{\partial r^2} + v \frac{\partial^3 T}{\partial r^3} \right) \\ & - \delta_E \frac{Q_0}{(\rho c_p)_{\text{hnf}}} \left( u \frac{\partial T}{\partial x} + v \frac{\partial T}{\partial r} \right) = \frac{k_{\text{hnf}}}{(\rho c_p)_{\text{hnf}}} \left( \frac{1}{r} \frac{\partial T}{\partial r} + \frac{\partial^2 T}{\partial r^2} \right) \end{aligned} \right\}, \quad (3)$$

with

$$\left. \begin{aligned} & u = u_w(x) = \frac{u_0 x}{l}, v = 0, T = T_w, \text{ at } r = R \\ & u \rightarrow 0, T \rightarrow T_\infty, \text{ as } r \rightarrow \infty \end{aligned} \right\}, \quad (4)$$

in which ( $u, v$ ) represent velocity components,  $\mu_{\text{hnf}}$  the dynamic viscosity,  $\sigma_{\text{hnf}}$  the electrical conductivity, ( $x, r$ ) the cylindrical coordinates,  $F \left( = \frac{C_b}{\sqrt{k_p}} \right)$  the inertia coefficient,  $\rho_{\text{hnf}}$  the density,  $k_p$  the porous space permeability,  $\nu_{\text{hnf}}$  the kinematic viscosity,  $C_b$  the drag force coefficient,  $T$  the temperature,  $(C_p)_{\text{hnf}}$  specific heat,  $Q_0 > 0$  the heat generation coefficient,  $\sigma^*$  Stefan Boltzmann constant,  $\alpha_f \left( = \frac{k_f}{(\rho c_p)_f} \right)$  thermal diffusivity,  $k^*$  mean absorption coefficient,  $k_{\text{hnf}}$  the thermal conductivity,  $\delta_E$  the thermal relaxation time,  $T_w$  the wall temperature,  $u_0$  the reference velocity, and  $T_\infty$  the ambient temperature.

## 2.1. Thermophysical characteristics

The thermophysical characteristics were determined by Hamilton–Crosser relations for the nanomaterial and hybrid nanomaterial (Table 1).<sup>49</sup>

### 2.1.1. Viscosity

$$\mu_{\text{nf}} = \frac{\mu_f}{(1 - \phi_1)^{2.5}} \text{ for nanoliquid}, \quad (5)$$

$$\mu_{\text{hnf}} = \frac{\mu_f}{(1 - \phi_1)^{2.5} (1 - \phi_2)^{2.5}} \text{ for hybrid nanoliquid}, \quad (6)$$

### 2.1.2. Density

$$\rho_{\text{nf}} = (1 - \phi_1)\rho_f + \phi_1\rho_{s1} \text{ for nanoliquid}, \quad (7)$$

$$\rho_{\text{hnf}} = (1 - \phi_2)[(1 - \phi_1)\rho_f + \phi_1\rho_{s1}] + \phi_2\rho_{s2} \text{ for hybrid nanoliquid}. \quad (8)$$

### 2.1.3. Electrical conductivity

$$\frac{\sigma_{\text{nf}}}{\sigma_f} = \frac{\sigma_{s1} + (n - 1)\sigma_f - (n - 1)\phi_1(\sigma_f - \sigma_{s1})}{\sigma_{s1} + (n - 1)\sigma_f + \phi_1(\sigma_f - \sigma_{s1})} \text{ for nanoliquids}. \quad (9)$$

$$\frac{\sigma_{\text{hnf}}}{\sigma_{\text{nf}}} = \frac{\sigma_{s2} + (n - 1)\sigma_{\text{nf}} - (n - 1)\phi_2(\sigma_{\text{nf}} - \sigma_{s2})}{\sigma_{s2} + (n - 1)\sigma_{\text{nf}} + \phi_2(\sigma_{\text{nf}} - \sigma_{s2})} \text{ for hybrid nanoliquid}. \quad (10)$$

### 2.1.4. Specific heat capacity

$$(\rho c_p)_{\text{nf}} = (1 - \phi_1)(\rho c_p)_f + \phi_1(\rho c_p)_{s1} \text{ for nanoliquid}, \quad (11)$$

$$(\rho c_p)_{\text{hnf}} = (1 - \phi_2)[(1 - \phi_1)(\rho c_p)_f + \phi_1(\rho c_p)_{s1}] + \phi_2(\rho c_p)_{s2} \text{ for hybrid nanoliquid}. \quad (12)$$



## 2.1.5. Thermal conductivity

$$\frac{k_{nf}}{k_f} = \frac{k_{s1} + (n-1)k_f - (n-1)\phi_1(k_f - k_{s1})}{k_{s1} + (n-1)k_f + \phi_1(k_f - k_{s1})} \text{ for nanoliquid, } (13)$$

$$\frac{k_{hnf}}{k_{nf}} = \frac{k_{s2} + (n-1)k_{nf} - (n-1)\phi_2(k_{nf} - k_{s2})}{k_{s2} + (n-1)k_{nf} + \phi_2(k_{nf} - k_{s2})} \text{ for hybrid nanoliquid. } (14)$$

Considering transformations

we have

$$\left. \begin{aligned} \frac{\partial f}{\partial \eta}(\xi, 0) = 1, f(\xi, 0) = -\xi \frac{\partial f}{\partial \xi}(\xi, 0), \theta(\xi, 0) = 1 \\ \frac{\partial f}{\partial \eta}(\xi, \infty) = 0, \theta(\xi, \infty) = 0 \end{aligned} \right\} (18)$$

In the above expressions,  $\alpha \left( = \sqrt{\frac{\nu_f l}{u_0 R^2}} \right)$  indicates the curvature variable,  $M \left( = \frac{\sigma_f B_0^2 l}{U_0 \rho_f} \right)$  the magnetic parameter,  $\lambda \left( = \frac{\nu_f l}{u_0 k_p} \right)$  the porosity variable,  $\text{Pr} \left( = \frac{\nu_f}{\alpha_f} \right)$  the Prandtl

$$\left. \begin{aligned} \xi = \frac{x}{l}, \eta = \sqrt{\frac{u_0}{l \nu_f}} \left( \frac{r^2 - R^2}{2R} \right), u = u_0 \xi \frac{\partial f}{\partial \eta}(\xi, \eta), \\ v = -\frac{1}{r} \sqrt{\frac{\nu_f u_0}{l}} R f(\xi, \eta) - \frac{1}{r} \sqrt{\frac{\nu_f u_0}{l}} \xi R \frac{\partial f}{\partial \xi}(\xi, \eta), \\ \theta(\xi, \eta) = \frac{T - T_\infty}{T_w - T_\infty}, u_w = \frac{u_0 x}{l}, \end{aligned} \right\} (15)$$

$$\left. \begin{aligned} (1 + 2\alpha\eta) \frac{\partial^3 f}{\partial \eta^3} + 2\alpha \frac{\partial^2 f}{\partial \eta^2} - A_1 A_2 \left( \frac{\partial f}{\partial \eta} \right)^2 + A_1 A_2 f \frac{\partial^2 f}{\partial \eta^2} - A_1 A_2 \xi \frac{\partial^2 f}{\partial \xi \partial \eta} \frac{\partial f}{\partial \eta} + A_1 A_2 \xi \frac{\partial f}{\partial \xi} \frac{\partial^2 f}{\partial \eta^2} \\ - \lambda \frac{\partial f}{\partial \eta} - A_1 A_2 \text{Pr} \xi \left( \frac{\partial f}{\partial \eta} \right)^2 - A_1 \frac{\sigma_{hnf}}{\sigma_f} M \frac{\partial f}{\partial \eta} = 0 \end{aligned} \right\} (16)$$

$$\left. \begin{aligned} \left( \frac{k_{hnf}}{k_f} + \text{Rd} \right) \left( (1 + 2\alpha\eta) \frac{\partial^2 \theta}{\partial \eta^2} + 2\alpha \frac{\partial \theta}{\partial \eta} \right) - A_3 \text{Pr} \xi \frac{\partial f}{\partial \eta} \frac{\partial \theta}{\partial \xi} + A_3 \text{Pr} f(\xi, \eta) \frac{\partial \theta}{\partial \eta} + A_3 \text{Pr} \xi \frac{\partial f}{\partial \xi} \frac{\partial \theta}{\partial \eta} + \text{Pr} Q \theta \\ - A_3 \text{Pr} \beta_t A^* + \beta_t \text{Rd} A^{**} + \text{Pr} \beta_t Q \left( \xi \frac{\partial \theta}{\partial \xi} \frac{\partial f}{\partial \eta} - f \frac{\partial \theta}{\partial \eta} - \xi \frac{\partial \theta}{\partial \eta} \frac{\partial f}{\partial \xi} \right) = 0 \end{aligned} \right\} (17)$$

with

$$A^* = \left( \begin{aligned} & -2\xi \frac{\partial f}{\partial \xi} \frac{\partial f}{\partial \eta} \frac{\partial \theta}{\partial \eta} + \xi^2 \left( \frac{\partial f}{\partial \eta} \right)^2 \frac{\partial^2 \theta}{\partial \xi^2} + \xi^2 \frac{\partial f}{\partial \eta} \frac{\partial^2 f}{\partial \xi \partial \eta} \frac{\partial \theta}{\partial \xi} + \xi \left( \frac{\partial f}{\partial \eta} \right)^2 \frac{\partial \theta}{\partial \xi} - \xi^2 \frac{\partial f}{\partial \eta} \frac{\partial \theta}{\partial \eta} \frac{\partial^2 f}{\partial \xi^2} \\ & - 2f \xi \frac{\partial f}{\partial \eta} \frac{\partial^2 \theta}{\partial \xi \partial \eta} - 2\xi^2 \frac{\partial f}{\partial \eta} \frac{\partial f}{\partial \xi} \frac{\partial^2 \theta}{\partial \xi \partial \eta} - f \xi \frac{\partial \theta}{\partial \xi} \frac{\partial^2 f}{\partial \eta^2} - \xi^2 \frac{\partial f}{\partial \xi} \frac{\partial \theta}{\partial \xi} \frac{\partial^2 f}{\partial \eta^2} \\ & - \left( \frac{\alpha}{(1 + 2\alpha\eta)} \right) \xi f \frac{\partial f}{\partial \xi} \frac{\partial \theta}{\partial \eta} + f \xi \frac{\partial \theta}{\partial \eta} \frac{\partial^2 f}{\partial \xi \partial \eta} - \left( \frac{\alpha}{(1 + 2\alpha\eta)} \right) \xi^2 \frac{\partial \theta}{\partial \eta} \left( \frac{\partial f}{\partial \xi} \right)^2 \\ & - \left( \frac{\alpha}{(1 + 2\alpha\eta)} \right) f \xi \frac{\partial f}{\partial \xi} \frac{\partial \theta}{\partial \eta} + \xi^2 \frac{\partial f}{\partial \xi} \frac{\partial \theta}{\partial \eta} \frac{\partial^2 f}{\partial \eta \partial \xi} + f^2 \frac{\partial^2 \theta}{\partial \eta^2} + \xi \frac{\partial f}{\partial \eta} \frac{\partial \theta}{\partial \eta} \frac{\partial f}{\partial \xi} + f \frac{\partial f}{\partial \eta} \frac{\partial \theta}{\partial \eta} \\ & + 2 \left( \frac{\alpha}{(1 + 2\alpha\eta)} \right) f \xi \frac{\partial f}{\partial \xi} \frac{\partial \theta}{\partial \eta} + 2f \xi \frac{\partial f}{\partial \xi} \frac{\partial^2 \theta}{\partial \eta^2} + \xi^2 \left( \frac{\partial f}{\partial \xi} \right)^2 \frac{\partial^2 \theta}{\partial \eta^2} \\ & + \left( \frac{\alpha}{(1 + 2\alpha\eta)} \right) \xi^2 \left( \frac{\partial f}{\partial \xi} \right)^2 \frac{\partial \theta}{\partial \eta} \end{aligned} \right), (19)$$



number,  $Fr \left( = \frac{C_b}{\sqrt{k_p}} l \right)$  the Forchheimer number,  $Rd \left( = \frac{16\sigma^* T_\infty^3}{3k^* k_f} \right)$  the radiation parameter,  $\beta_t \left( \frac{\delta_E l}{u_0} \right)$  the thermal relaxation time variable, and  $Q \left( = \frac{Q_0 l}{u_0 (\rho c_p)_f} \right)$ .  $A^*$ ,  $A^{**}$ ,  $A_1$ ,  $A_2$  and  $A_3$  are

$$A^{**} =$$

$$\left( \begin{array}{l} 2\xi\alpha \frac{\partial f}{\partial \eta} \frac{\partial^2 \theta}{\partial \xi \partial \eta} + \xi(1+2\alpha\eta) \frac{\partial f}{\partial \eta} \frac{\partial^3 \theta}{\partial \xi \partial \eta^2} - 4\alpha f \frac{\partial^2 \theta}{\partial \eta^2} - (1+2\alpha\eta) f \frac{\partial^3 \theta}{\partial \eta^3} \\ -4\alpha\xi \frac{\partial f}{\partial \xi} \frac{\partial^2 \theta}{\partial \eta^2} - (1+2\alpha\eta) \xi \frac{\partial f}{\partial \xi} \frac{\partial^3 \theta}{\partial \eta^3} \end{array} \right) \quad (20)$$

$$A_1 = (1 - \phi_1)^{2.5} (1 - \phi_2)^{2.5} \quad (21)$$

$$A_2 = (1 - \phi_2) \left[ (1 - \phi_1) + \phi_1 \frac{\rho_{s1}}{\rho_f} \right] + \phi_2 \frac{\rho_{s2}}{\rho_f} \quad (22)$$

$$A_3 = (1 - \phi_2) \left[ (1 - \phi_1) + \phi_1 \frac{(\rho c_p)_{s1}}{(\rho c_p)_f} \right] + \phi_2 \frac{(\rho c_p)_{s2}}{(\rho c_p)_f} \quad (23)$$

### 3. Quantities under interest

Skin friction coefficient  $\left( \frac{1}{2} C_{fx} Re_x^{1/2} \right)$  and Nusselt number  $(Nu_x Re_x^{-1/2})$  are

$$\left. \begin{array}{l} C_{fx} = \frac{2\tau_w}{\rho_f u_w^2} \\ Nu_x = \frac{xq_w}{k_f (T_w - T_\infty)} \end{array} \right\} \quad (24)$$

Shear stress ( $\tau_w$ ) and heat flux ( $q_w$ ) satisfy

$$\left. \begin{array}{l} \tau_w = \mu_{hnf} \left( \frac{\partial u}{\partial r} \right) \Big|_{r=R} \\ q_w = - \left( k_{hnf} + \frac{16\sigma^* T_\infty^3}{3k^*} \right) \left( \frac{\partial T}{\partial r} \right) \Big|_{r=R} \end{array} \right\} \quad (25)$$

$$\left. \begin{array}{l} \left( \frac{k_{hnf}}{k_f} + Rd \right) (2\alpha\theta' + (2\alpha\eta + 1)\theta'') + A_3 Pr f \theta' + Pr Q \theta - \beta_t Pr (f^2 \theta'' + ff' \theta') \\ -\beta_t Rd ((2\alpha\eta + 1)f \theta''' + 4\alpha f \theta'') - \beta_t Pr Q F \theta' = 0 \end{array} \right\} \quad (30)$$

One may write eqn (27) and (28) as

$$\left. \begin{array}{l} \frac{1}{2} C_{fx} Re_x^{1/2} = \frac{1}{A_1} f''(0) \\ Nu_x Re_x^{-1/2} = - \left( \frac{k_{hnf}}{k_f} + Rd \right) \theta'(0) \end{array} \right\} \quad (26)$$

where  $Re_x \left( = \frac{u_w x}{\nu_f} \right)$  shows the local Reynolds number.

### 4. Expression for entropy

Mathematically, entropy is expressed as

$$\left. \begin{array}{l} E_G = \frac{k_f}{T_\infty^2} \left( \frac{k_{hnf}}{k_f} + \frac{16\sigma^* T_\infty^3}{3k^* k_f} \right) \left( \frac{\partial T}{\partial r} \right)^2 + \frac{\mu_{hnf}}{T_\infty} \left( \frac{\partial u}{\partial r} \right)^2 \\ + \frac{\sigma_{hnf} B_0^2}{T_\infty} u^2 + \frac{\mu_{hnf}}{k_p T_\infty} u^2 \end{array} \right\} \quad (27)$$

The above expression can be reduced as

$$\left. \begin{array}{l} N_G(\eta) = \alpha_1 \left( \frac{k_{hnf}}{k_f} + Rd \right) (1 + 2\alpha\eta) \left( \frac{\partial \theta}{\partial \eta} \right)^2 + \frac{Br}{A_1} \xi^2 (1 + 2\alpha\eta) \left( \frac{\partial^2 f}{\partial \eta^2} \right)^2 + \frac{\lambda Br}{A_1} \xi^2 \left( \frac{\partial f}{\partial \eta} \right)^2 \\ + \frac{\sigma_{hnf}}{\sigma_f} M Br \xi^2 \left( \frac{\partial f}{\partial \eta} \right)^2 \end{array} \right\} \quad (28)$$

where  $N_G \left( = \frac{E_G \nu_f l T_\infty}{k_f u_0 (T_w - T_\infty)} \right)$  characterizes the entropy rate,

$\alpha_1 \left( = \frac{(T_w - T_\infty)}{T_\infty} \right)$  the temperature difference variable, and

$Br \left( = \frac{\mu_f u_0^2}{k_f (T_w - T_\infty)} \right)$  the Brinkman number.

### 5. Solution development

#### 5.1. Local similar solution

To obtain a local similar solution, we suppose that  $\frac{\partial(\cdot)}{\partial \xi} = 0$  and denote  $\frac{\partial(\cdot)}{\partial \eta}$  by prime in eqn (16)–(18). We have

$$\left. \begin{array}{l} (2\alpha\eta + 1)f'''' + 2\alpha f''' - A_1 A_2 f''^2 + A_1 A_2 f f'''' - \lambda f' - A_1 A_2 Fr \xi^2 f''^2 \\ - \frac{\sigma_{hnf}}{\sigma_f} A_1 M f' = 0 \end{array} \right\} \quad (29)$$



$$\left. \begin{aligned} f'(0) = 1, f(0) = 0, \theta(0) = 1 \\ f'(\infty) = 0, \theta(\infty) = 0 \end{aligned} \right\}. \quad (31)$$

## 5.2. Local non-similar solution

To construct the local non-similar solution, we suppose that

$$\frac{\partial f}{\partial \xi} = P, \frac{\partial^2 f}{\partial \xi \partial \eta} = \frac{\partial p}{\partial \eta}, \frac{\partial^3 f}{\partial \xi \partial \eta^2} = \frac{\partial^2 p}{\partial \eta^2}, \frac{\partial^2 f}{\partial \xi^2} = \frac{\partial p}{\partial \xi}, \frac{\partial \theta}{\partial \xi} = q, \frac{\partial^2 \theta}{\partial \xi \partial \eta} = \frac{\partial q}{\partial \eta},$$

$$\frac{\partial^3 \theta}{\partial \xi \partial \eta^2} = \frac{\partial^2 q}{\partial \eta^2}, \frac{\partial^2 \theta}{\partial \xi^2} = \frac{\partial q}{\partial \xi} \text{ in eqn (16)–(18). One obtains}$$

$$\left. \begin{aligned} (2\alpha\eta + 1) \frac{\partial^3 f}{\partial \eta^3} + 2\alpha \frac{\partial^2 f}{\partial \eta^2} - A_1 A_2 \left( \frac{\partial f}{\partial \eta} \right)^2 + A_1 A_2 f \frac{\partial^2 f}{\partial \eta^2} - A_1 A_2 \xi \frac{\partial p}{\partial \eta} \frac{\partial f}{\partial \eta} + A_1 A_2 \xi P \frac{\partial^2 f}{\partial \eta^2} \\ - \lambda \frac{\partial f}{\partial \eta} - A_1 A_2 \text{Fr} \left( \frac{\partial f}{\partial \eta} \right)^2 - A_1 \frac{\sigma_{\text{hnf}}}{\sigma_f} M \frac{\partial f}{\partial \eta} = 0 \end{aligned} \right\}, \quad (32)$$

$$\left. \begin{aligned} \left( \frac{k_{\text{hnf}}}{k_f} + \text{Rd} \right) \left[ (2\alpha\eta + 1) \frac{\partial^2 \theta}{\partial \eta^2} + 2\alpha \frac{\partial \theta}{\partial \eta} \right] - A_3 \text{Pr} \xi \frac{\partial f}{\partial \eta} q + A_3 \text{Pr} f \frac{\partial \theta}{\partial \eta} + A_3 \text{Pr} \xi p \frac{\partial \theta}{\partial \eta} \\ + \text{Pr} Q \theta - A_3 \text{Pr} \beta_t A_{11} + \beta_t \text{Rd} A_{22} + \text{Pr} \beta_t Q \left( \xi q \frac{\partial f}{\partial \eta} - f \frac{\partial \theta}{\partial \eta} - \xi p \frac{\partial \theta}{\partial \eta} \right) = 0 \end{aligned} \right\}, \quad (33)$$

$$\left. \begin{aligned} \frac{\partial f}{\partial \eta}(\xi, 0) = 1, f(\xi, 0) = -\xi p(\xi, 0), \theta(\xi, 0) = 1 \\ \frac{\partial f}{\partial \eta}(\xi, \infty) = 0, \theta(\xi, \infty) = 0 \end{aligned} \right\}. \quad (34)$$

In the above equations,  $A_{11}$  and  $A_{22}$  are defined as

$$A_{11} = \left[ \begin{aligned} -2\xi P \frac{\partial f}{\partial \eta} \frac{\partial \theta}{\partial \eta} + \xi^2 \left( \frac{\partial f}{\partial \eta} \right)^2 \frac{\partial q}{\partial \xi} + \xi^2 q \frac{\partial f}{\partial \eta} \frac{\partial p}{\partial \eta} + \xi q \left( \frac{\partial f}{\partial \eta} \right)^2 - \xi^2 \frac{\partial f}{\partial \eta} \frac{\partial \theta}{\partial \eta} \frac{\partial p}{\partial \xi} \\ - 2\xi f \frac{\partial f}{\partial \eta} \frac{\partial q}{\partial \eta} - 2\xi^2 p \frac{\partial f}{\partial \eta} \frac{\partial q}{\partial \eta} - \xi f q \frac{\partial^2 f}{\partial \eta^2} - \xi^2 p q \frac{\partial^2 f}{\partial \eta^2} - \left( \frac{\alpha}{2\alpha\eta + 1} \right) \xi f p \frac{\partial \theta}{\partial \eta} \\ + \xi f \frac{\partial \theta}{\partial \eta} \frac{\partial p}{\partial \eta} - \left( \frac{\alpha}{2\alpha\eta + 1} \right) \xi^2 p^2 \frac{\partial \theta}{\partial \eta} - \left( \frac{\alpha}{2\alpha\eta + 1} \right) \xi f p \frac{\partial \theta}{\partial \eta} + \xi^2 p \frac{\partial \theta}{\partial \eta} \frac{\partial p}{\partial \eta} \\ + f^2 \frac{\partial^2 \theta}{\partial \eta^2} + \xi p \frac{\partial f}{\partial \eta} \frac{\partial \theta}{\partial \eta} + f \frac{\partial f}{\partial \eta} \frac{\partial \theta}{\partial \eta} + 2 \left( \frac{\alpha}{2\alpha\eta + 1} \right) \xi f p \frac{\partial \theta}{\partial \eta} + 2\xi f p \frac{\partial^2 \theta}{\partial \eta^2} \\ + \xi^2 p^2 \frac{\partial^2 \theta}{\partial \eta^2} + \left( \frac{\alpha}{2\alpha\eta + 1} \right) \xi^2 p^2 \frac{\partial \theta}{\partial \eta} \end{aligned} \right], \quad (35)$$

$$A_{22} = \left[ \begin{aligned} 2\alpha\xi \frac{\partial f}{\partial \eta} \frac{\partial q}{\partial \eta} + \xi(2\alpha\eta + 1) \frac{\partial f}{\partial \eta} \frac{\partial^2 q}{\partial \eta^2} - 4\alpha f \frac{\partial^2 \theta}{\partial \eta^2} - (2\alpha\eta + 1) f \frac{\partial^3 \theta}{\partial \eta^3} \\ - 4\alpha\xi p \frac{\partial^2 \theta}{\partial \eta^2} - (2\alpha\eta + 1) \xi p \frac{\partial^3 f}{\partial \eta^3} \end{aligned} \right] \quad (36)$$

We take the derivative of eqn (32)–(34) with respect to “ $\xi$ ” and representing  $\frac{\partial(\cdot)}{\partial \eta}$  by prime. Suppose that  $\frac{\partial p}{\partial \xi}, \frac{\partial^2 p}{\partial \xi \partial \eta}, \frac{\partial^3 p}{\partial \xi \partial \eta^2}, \frac{\partial q}{\partial \xi}, \frac{\partial^2 q}{\partial \xi \partial \eta}, \frac{\partial^3 q}{\partial \xi \partial \eta^2}, \frac{\partial^2 g}{\partial \xi \partial \eta}, \frac{\partial^3 g}{\partial \xi \partial \eta^2}$  become zero. We obtain



$$\left. \begin{aligned} (2\alpha\eta + 1)p'''' + 2\alpha p'' - 2A_1A_2f'p' + A_1A_2pf'' + A_1A_2fp'' - A_1A_2p'f' - A_1A_2\xi p'^2 \\ + A_1A_2pf'' + A_1A_2\xi pp'' - \lambda p' - A_1A_2Fr f'^2 - 2A_1A_2Fr\xi f'p' - A_1\frac{\sigma_{\text{hnf}}}{\sigma_f}MP' = 0 \end{aligned} \right\} \quad (37)$$

$$\left. \begin{aligned} \left(\frac{k_{\text{hnf}}}{k_f} + \text{Rd}\right) [(2\alpha\eta + 1)q'' + 2\alpha q'] - A_3Prf'q - A_3Pr\xi p'q + A_3Prp\theta' + A_3Prfq' \\ + A_3Prp\theta' + A_3Pr\xi pq' + PrQq - A_3Pr\beta_1A_{33} + \text{Rd}\beta_1A_{44} \\ + PrQ\beta_1(qf'' + \xi qp' - p\theta' - fq' - p\theta' - \xi pq') = 0, \end{aligned} \right\} \quad (38)$$

$$\left. \begin{aligned} p'(\xi, 0) = 0, p(\xi, 0) = 0, q(\xi, 0) = 0 \\ p'(\xi, \infty) = 0, q(\xi, \infty) = 0 \end{aligned} \right\} \quad (39)$$

variable for velocity ( $f'(\eta)$ ). The flow was enhanced by a higher approximation of the porosity parameter.

In the above equation,  $A_{33}$  and  $A_{44}$  are expressed as

$$A_{33} = \left[ \begin{aligned} -7\xi Pf'q' + 4\xi f'p'q + \xi^2 p'^2 q + f'^2 q - 2fp'q' - \xi fp'q' - \xi^2 pp'q' - fqf'' - 3\xi pqf'' + f^2 q'' \\ - \xi fqp'' - \xi^2 pqp'' + 2fp'\theta' + 2\xi pp'\theta' + 4fp\theta'' + ff'q' + 4\xi p^2\theta'' + 2\xi fpq'' + \xi^2 p^2q'' \end{aligned} \right] \quad (40)$$

$$A_{44} = \left[ \begin{aligned} 2\alpha f'q' + 2\alpha \xi p'q' + (2\alpha\eta + 1)f'q'' + (2\alpha\eta + 1)\xi p'q'' - 8\alpha p\theta'' - (2\alpha\eta + 1)p\theta'''' \\ - 4\alpha fq'' - (2\alpha\eta + 1)fq'''' - 4\alpha \xi pq'' - (2\alpha\eta + 1)pf'''' - (2\alpha\eta + 1)\xi pp'''' \end{aligned} \right] \quad (41)$$

## 6. Discussion

Fluid flow, temperature, and entropy rate for sundry parameters regarding nanoliquid ( $\text{MoS}_4/\text{engine oil}$ ) and hybrid nanoliquid ( $\text{MoS}_4 + \text{CoFe}_2\text{O}_4/\text{engine oil}$ ) are presented. In these graphs, the solid lines denote the nanoliquid impact, and the dashed lines represent hybrid nanoliquid behavior. The numerical results for coefficient of skin friction ( $C_{fx}$ ) and the thermal transport rate ( $\text{Nu}_x$ ) for nanoliquid ( $\text{MoS}_4/\text{engine oil}$ ) and hybrid nanoliquid ( $\text{MoS}_4 + \text{CoFe}_2\text{O}_4/\text{engine oil}$ ) were examined.

### 6.1. Velocity

Fig. 2 shows the variation in the curvature variable for liquid flow. A larger approximation of curvature ( $\alpha$ ) variable leads to a decreased radius of curvature, which shrinks the region of the cylinder in contact with liquid. Consequently, resistance decreases, which hence boosts fluid flow. Fig. 3 indicates the impact of Forchheimer number on velocity. An augmentation in the Forchheimer number corresponds to a decrease in the liquid flow for nanoliquid ( $\text{MoS}_4/\text{engine oil}$ ) and hybrid nanoliquid ( $\text{MoS}_4 + \text{CoFe}_2\text{O}_4/\text{engine oil}$ ). Fig. 4 depicts the influence of the magnetic effect on liquid flow ( $f'(\eta)$ ). A physically higher magnetic field leads to resistance to liquid flow, and as a result, velocity decreases. Fig. 5 exhibits the outcomes of the porosity

### 6.2. Temperature

Fig. 6 indicates the temperature variation for thermal relaxation time parameter ( $\beta_1$ ). An increase in the thermal field occurs

versus the higher thermal relaxation time ( $\beta_1$ ) parameter. The behavior of ( $\theta(\eta)$ ) versus a higher radiation variable is shown in Fig. 7. Physically higher radiation leads to an additional energy

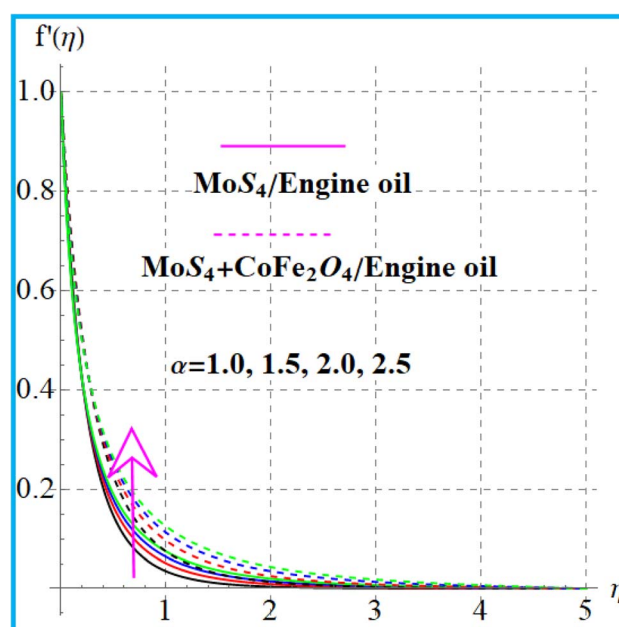
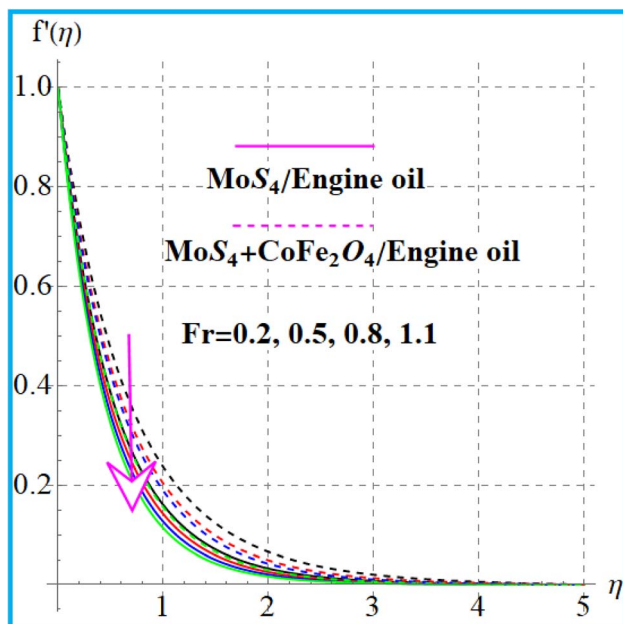
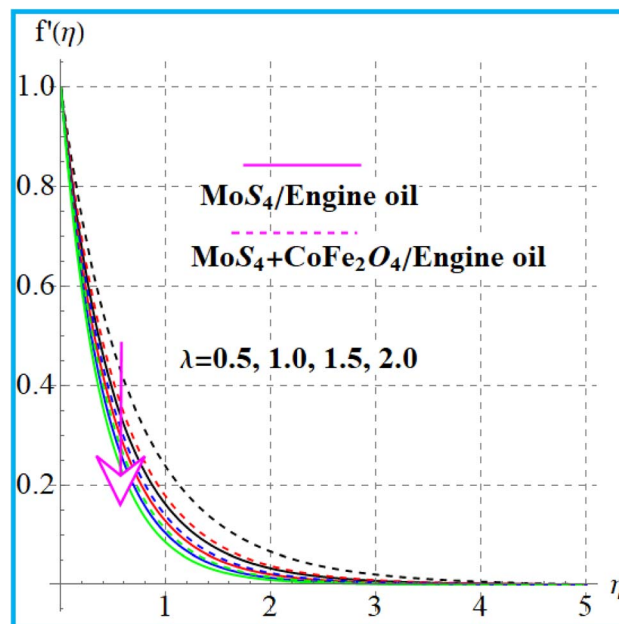
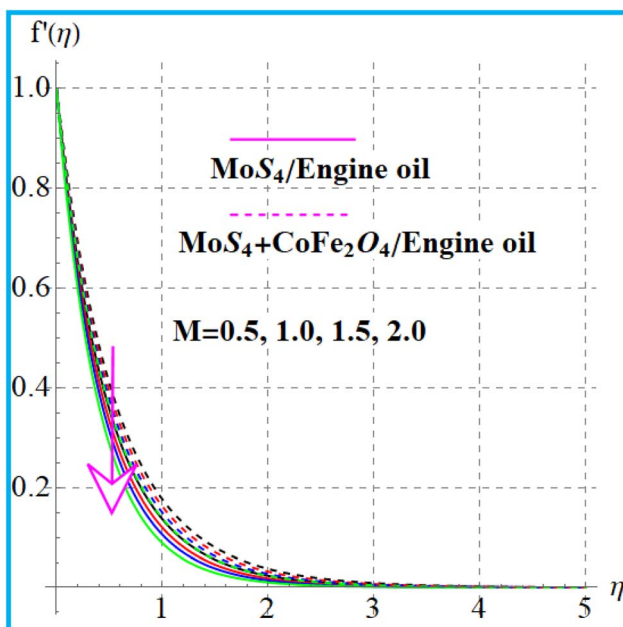
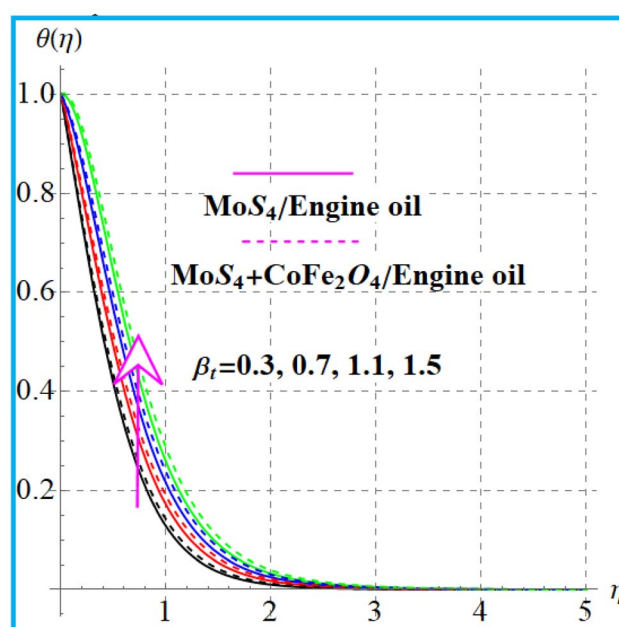


Fig. 2  $f'(\eta)$  variation versus  $\alpha$ .



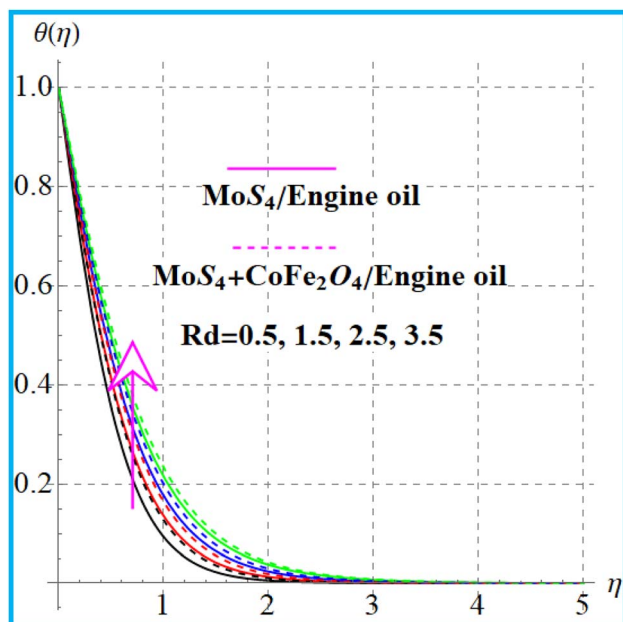
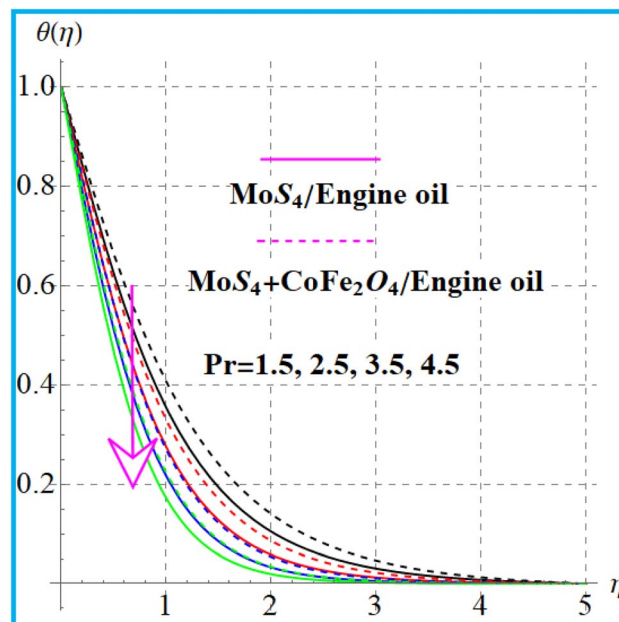
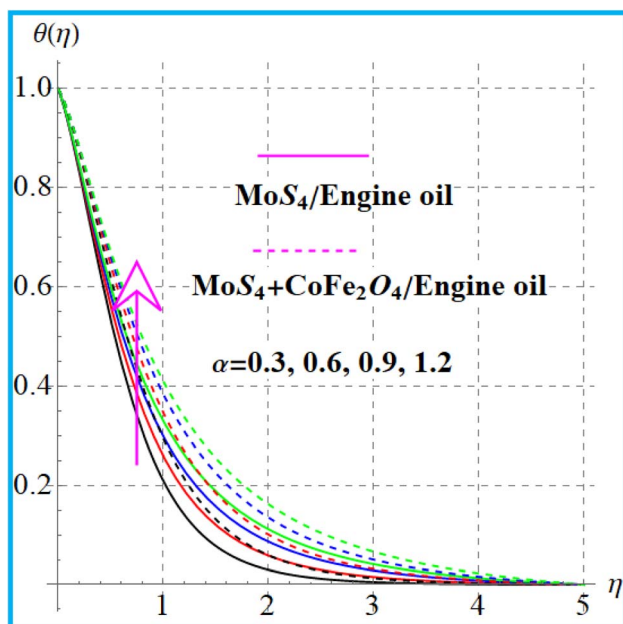
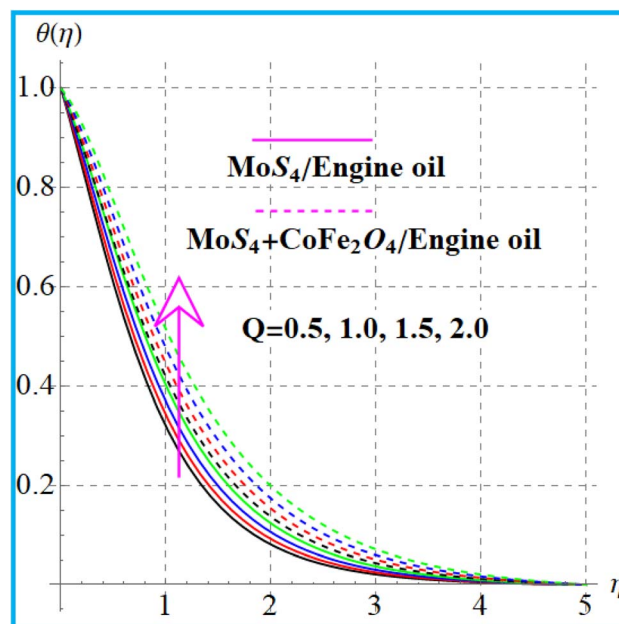
Fig. 3  $f'(\eta)$  variation versus  $Fr$ .Fig. 5  $f'(\eta)$  variation versus  $\lambda$ .Fig. 4  $f'(\eta)$  variation versus  $M$ .Fig. 6  $\theta(\eta)$  variation versus  $\beta_t$ .

in the system, which increases the thermal field. Fig. 8 displays outcomes of the curvature ( $\alpha$ ) variable for temperature. Temperature enhances *versus* a higher curvature parameter. Fig. 9 shows the Prandtl (Pr) number impact for  $(\theta(\eta))$ . Physically, (Pr) decreases the thermal diffusivity, and therefore, the temperature decreases. Fig. 10 illustrates the outcomes of heat generation for a thermal field. An increase in temperature against heat generation is possible.

### 6.3. Entropy rate

Fig. 11 depicts the behavior of entropy against a higher magnetic field. A physically higher magnetic force leads to an increasing resistive force, which gives rise to greater energy in the system, and thus, the entropy rate increases. Fig. 12 shows the variation of the porosity ( $\lambda$ ) variable against the entropy rate. An increase in the entropy production occurs against a larger approximation of the porosity parameter. Fig. 13 displays the



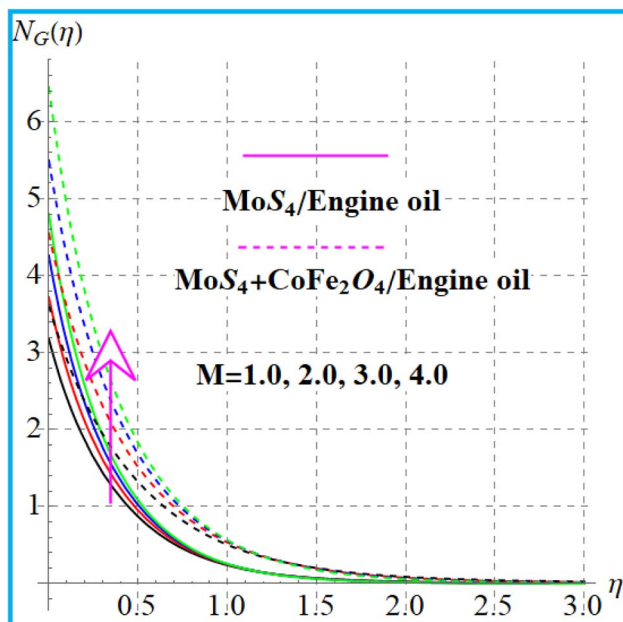
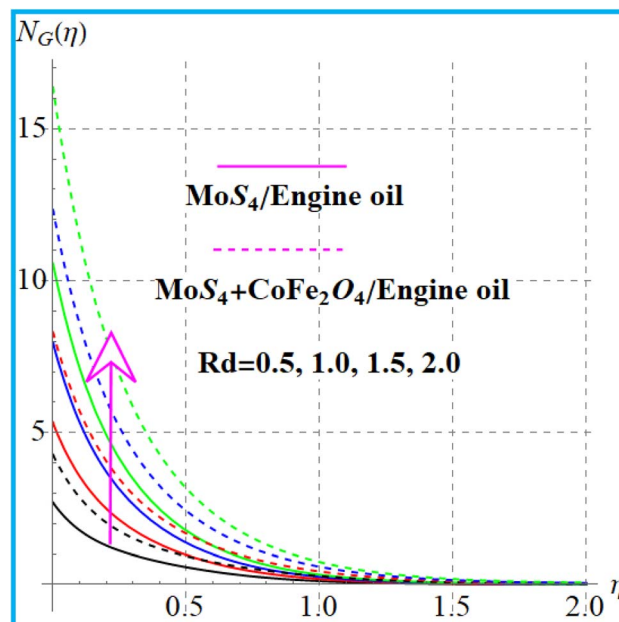
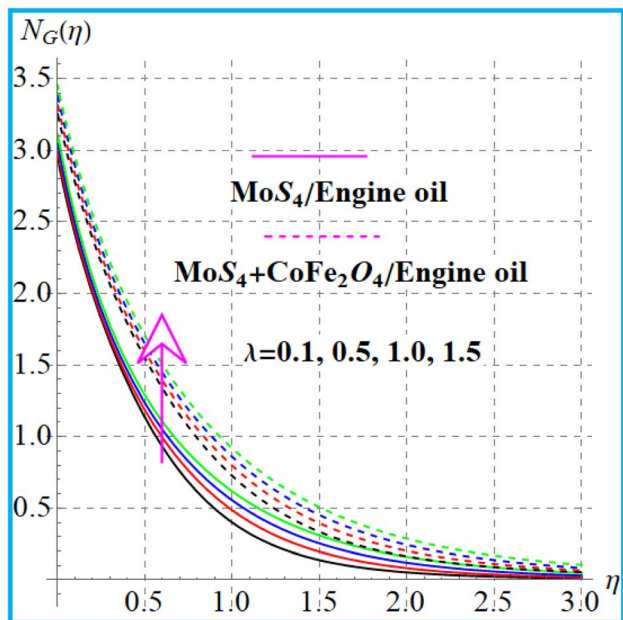
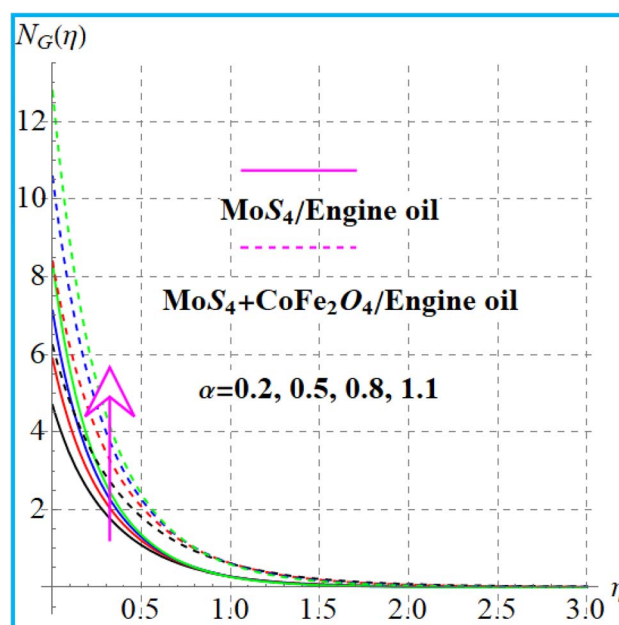
Fig. 7  $\theta(\eta)$  variation versus  $Rd$ .Fig. 9  $\theta(\eta)$  variation versus  $Pr$ .Fig. 8  $\theta(\eta)$  variation versus  $\alpha$ .Fig. 10  $\theta(\eta)$  variation versus  $Q$ .

radiation variation for the entropy rate. Physically higher radiation leads to increasing thermal emission, which creates extra energy in a thermal system. As a consequence, the entropy rate increases. Fig. 14 elucidates the impact of curvature on the entropy rate. The entropy production is enhanced for the curvature variable.

#### 6.4. Physical quantities

The numerical results for coefficient of skin friction  $\left(\frac{1}{2}C_{fx}Re_x^{1/2}\right)$  and thermal transport rate  $(Nu_x Re_x^{-1/2})$  for nanoliquid ( $MoS_4$ /engine oil) and hybrid nanoliquid ( $MoS_4 + CoFe_2O_4$ /engine oil) have been sketched in Tables 2 and 3.



Fig. 11  $N_G(\eta)$  variation versus  $M$ .Fig. 13  $N_G(\eta)$  variation versus  $Rd$ .Fig. 12  $N_G(\eta)$  variation versus  $\lambda$ .Fig. 14  $N_G(\eta)$  variation versus  $\alpha$ .

**6.4.1. Coefficient of skin friction.** Surface drag force ( $\frac{1}{2}C_{fx}Re_x^{1/2}$ ) for influential variables is examined in Table 2. Higher magnetic variables lead to intensification of the skin friction coefficient for nanomaterial ( $MoS_4$ /engine oil) and hybrid nanomaterial ( $MoS_4 + CoFe_2O_4$ /engine oil). Clearly, for a larger porosity variable and Forchheimer number, the surface drag force improved.

**6.4.2. Nusselt number.** Physical features of emerging variables of the Nusselt number ( $Nu_x Re_x^{-1/2}$ ) were constructed in Table 3. An increase in the heat transport rate ( $Nu_x Re_x^{-1/2}$ ) occurs through radiation and thermal relaxation time parameters. Higher estimation of heat generation variables intensifies the heat transport rate ( $Nu_x Re_x^{-1/2}$ ) for nanoliquid ( $MoS_4$ /engine oil) and hybrid nanoliquid ( $MoS_4 + CoFe_2O_4$ /engine oil).



Table 2 Numerical values of surface drag force  $\left(\frac{1}{2}C_{fx}Re_x^{1/2}\right)$ 

$\frac{1}{2}C_{fx}Re_x^{1/2}$					
Fr	M	$\lambda$	MoS <sub>4</sub> /engine oil	MoS <sub>4</sub> + CoFe <sub>2</sub> O <sub>4</sub> /engine oil	
0.2	0.5	0.3	1.23154	1.65829	
0.6			2.32562	3.09652	
1.0			3.09845	3.95624	
0.5	1.0	0.3	2.56234	3.56329	
			2.0	3.69532	4.032654
			3.0	4.65892	5.01985
0.2	0.5	0.5	0.96523	1.85746	
		1.0	1.75632	2.89635	
		1.5	2.65295	3.56982	

Table 3 Thermal transport rate ( $Nu_x Re_x^{-1/2}$ ) variation against emerging parameters

$Nu_x Re_x^{-1/2}$					
Rd	$\beta_t$	Q	MoS <sub>4</sub> /engine oil	MoS <sub>4</sub> + CoFe <sub>2</sub> O <sub>4</sub> /engine oil	
1.5	0.5	1.0	3.42352	3.89653	
2.0			4.12956	5.03256	
2.5			5.45638	6.36524	
1.0	1.0	1.0	2.86593	3.78654	
			1.5	3.58647	4.25369
			2.0	4.96583	5.65389
1.0	0.5	1.5	2.54892	3.96585	
		2.5	3.95862	4.65895	
		3.5	4.96852	5.87562	

## 7. Concluding remarks

The main results are listed below.

- The fluid flow variation for porosity and curvature variables is reversed.
- The velocity field decayed for a magnetic field, while the opposite holds for the entropy rate and temperature.
- The Forchheimer number results in velocity decrease.
- The curvature parameter variation for entropy and temperature for nanoliquid (MoS<sub>4</sub>/engine oil) and hybrid nanoliquid (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil) was the same.
- An enhancement of the temperature and entropy rate against radiation was observed.
- The thermal field decays for a higher Prandtl number in nanoliquid (MoS<sub>4</sub>/engine oil) and hybrid nanoliquid (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil).
- A higher thermal relaxation time enhances the temperature in nanoliquid (MoS<sub>4</sub>/engine oil) and hybrid nanoliquid (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil).
- Porosity parameter variation increases the entropy rate for nanoliquid (MoS<sub>4</sub>/engine oil) and hybrid nanoliquid (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil).
- A higher magnetic field intensifies the surface drag force for nanoliquid (MoS<sub>4</sub>/engine oil) and hybrid nanoliquid (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil).

- A larger estimation of the Forchheimer number and porosity variable leads to coefficient of skin friction enhancement for nanomaterial (MoS<sub>4</sub>/engine oil) and hybrid nanomaterial (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil).

- A higher radiation variable results in thermal transport rate enhancement for nanoliquid (MoS<sub>4</sub>/engine oil) and hybrid nanoliquid (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil).

- A higher thermal relaxation time variable improves the heat transport rate for nanoliquid (MoS<sub>4</sub>/engine oil) and hybrid nanoliquid (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil).

- The Nusselt number intensifies for higher heat generation.

- The skin friction coefficient for hybrid nanomaterial (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil) was more than that of nanoliquid (MoS<sub>4</sub>/engine oil).

- The thermal transport rate was more dominant for hybrid nanoliquid (MoS<sub>4</sub> + CoFe<sub>2</sub>O<sub>4</sub>/engine oil) than for nanoliquid (MoS<sub>4</sub>/engine oil).

## Conflicts of interest

There are no conflicts to declare.

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