














Cite this: *Chem. Soc. Rev.*, 2023, 52, 879

The design of small-molecule prodrugs and activatable phototherapeutics for cancer therapy

Hai-Hao Han, ^{aceh} Han-Min Wang, ^{ce} Paramesh Jangili, ^d Mingle Li, ^d Luling Wu, ^{*b} Yi Zang, ^{ck} Adam C. Sedgwick, ^{*f} Jia Li, ^{*ceh} Xiao-Peng He, ^{*aj} Tony D. James ^{*bg} and Jong Seung Kim ^{*d}

Cancer remains as one of the most significant health problems, with approximately 19 million people diagnosed worldwide each year. Chemotherapy is a routinely used method to treat cancer patients. However, current treatment options lack the appropriate selectivity for cancer cells, are prone to resistance mechanisms, and are plagued with dose-limiting toxicities. As such, researchers have devoted their attention to developing prodrug-based strategies that have the potential to overcome these limitations. This tutorial review highlights recently developed prodrug strategies for cancer therapy. Prodrug examples that provide an integrated diagnostic (fluorescent, photoacoustic, and magnetic resonance imaging) response, which are referred to as theranostics, are also discussed. Owing to the non-invasive nature of light (and X-rays), we have discussed external excitation prodrug strategies as well as examples of activatable photosensitizers that enhance the precision of photodynamic therapy/photothermal therapy. Activatable photosensitizers/photothermal agents can be seen as analogous to prodrugs, with their phototherapeutic properties at a specific wavelength activated in the presence of disease-related biomarkers. We discuss each design strategy and illustrate the importance of targeting biomarkers specific to the tumour microenvironment and biomarkers that are known to be overexpressed within cancer cells. Moreover, we discuss the advantages of each approach and highlight their inherent limitations. We hope in doing so, the reader will appreciate the current challenges and available opportunities in the field and inspire subsequent generations to pursue this crucial area of cancer research.

Received 6th August 2022

DOI: 10.1039/d2cs00673a

rsc.li/chem-soc-rev

Key learning points

1. The importance of developing new and effective anticancer agents with different mechanism of action
2. The design strategies used for the development of prodrugs
3. The importance of introducing a diagnostic component during cancer therapy
4. The advantages and disadvantages of each therapeutic approach (chemotherapy, photodynamic therapy, and photothermal therapy).
5. Future perspectives in the area of prodrug development.

^a Key Laboratory for Advanced Materials and Joint International Research Laboratory of Precision Chemistry and Molecular Engineering, Feringa Nobel Prize Scientist Joint Research Center, Frontiers Center for Materiobiology and Dynamic Chemistry, School of Chemistry and Molecular Engineering, East China University of Science and Technology, 130 Meilong Rd., Shanghai 200237, P. R. China. E-mail: xphe@ecust.edu.cn

^b Department of Chemistry, University of Bath, Bath, BA2 7AY, UK. E-mail: t.d.james@bath.ac.uk, wllcyl@126.com

^c State Key Laboratory of Drug Research, Molecular Imaging Center, Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai 201203, China. E-mail: jli@simm.ac.cn

^d Department of Chemistry, Korea University, Seoul 02841, Republic of Korea. E-mail: jongskim@korea.ac.kr

^e University of Chinese Academy of Sciences, No. 19A Yuquan Road, Beijing 100049, P. R. China

^f Chemistry Research Laboratory, University of Oxford, Mansfield Road, OX1 3TA, UK. E-mail: adam.sedgwick@chem.ox.ac.uk

^g School of Chemistry and Chemical Engineering, Henan Normal University, Xinxiang 453007, China

^h Shandong Laboratory of Yantai Drug Discovery, Bohai Rim Advanced Research Institute for Drug Discovery, Yantai, Shandong 264117, P. R. China

ⁱ The International Cooperation Laboratory on Signal Transduction, Eastern Hepatobiliary Surgery Hospital, Shanghai 200438, China

^j National Center for Liver Cancer, Shanghai 200438, China

^k Lingang laboratory, Shanghai 201203, China



1. Introduction

Cancer is defined as a disease caused by the abnormal proliferation of cells. This uncontrollable cell growth not only accumulates to form solid tumours but leads to subsequent invasion to adjacent and distal tissues (and organs), a phenomenon which is known as cancer metastasis.^{1,2} Metastasis is the leading cause of cancer morbidity and mortality and responsible

for ~90% of cancer-related deaths.³ According to the International Agency for Research on Cancer, 19.29 million new cancer cases and 9.96 million cancer-related deaths were reported worldwide in 2020; unfortunately, these numbers are only expected to continue to increase.¹ Thus, research efforts are extensively being devoted to developing new and effective treatment protocols. A primary focus is on overcoming the inherent limitations of current chemotherapeutics, such as poor aqueous solubility, poor



Hai-Hao Han

Hai-Hao Han is an Associate Professor at SIMM (CAS). He received his PhD in 2020 from ECUST under the supervision of Prof. Xiao-Peng He. His research interests include fluorescent probes for disease theranostic applications and targeted drug development.



Han-Min Wang

Han-Min Wang is currently pursuing a PhD degree at SIMM (CAS) with Prof. Yi Zang. Her research focuses on the biological research of new target for NASH treatment and development of new chemical probe tools for monitoring the development of NASH.



Paramesh Jangili

Paramesh Jangili received his PhD in 2016 from Jawaharlal Nehru Technological University-Hyderabad, India. He subsequently joined in Ewha Womans University, Korea and in 2017 joined Prof. Jong Seung Kim lab at Korea University.



Yi Zang

Yi Zang obtained her PhD from SIMM (CAS) in 2008 and is currently a professor at Lingang Laboratory. Her research mainly focuses on the biological research of AMPK, drug discovery for organ fibrosis and development of new chemical biology probe tools.



Jia Li

Jia Li received his PhD from SIMM (CAS) in 2000 and was promoted to professor in 2005. He has been the director of SIMM (CAS) since 2019. His research interests are centered on the investigation of mechanisms of metabolic diseases and medicinal chemical biology.



Xiao-Peng He

Xiao-Peng He is a professor at Feringa Nobel Prize Scientists Research Center, ECUST. He obtained his BSc (2006) and PhD (2011) from ECUST. He conducted postdoctoral research with Kaixian Chen (SIMM, CAS) from 2011 to 2013 at ECUST. His research interests are chemical probes for glycobiology and sugar-based drug discovery.



selectivity toward cancer cells, the intrinsic or acquired resistance of tumours (multidrug resistance), and the dose-limiting toxicities (low maximum tolerated doses, MTD).^{4–6} Select examples of routinely used FDA-approved chemotherapeutics (Fig. 1) include doxorubicin (DOX, **1**), camptothecin [CPT (**2**) CPT-11 (**3**) and its active metabolite, SN-38, (**4**)], gemcitabine (**5**), 5-fluorodeoxyuridine (FDU, **6**), pazopanib (**7**), panobinostat (**8**), cisplatin (**9**), carboplatin (**10**), and oxaliplatin (**11**), all of which are discussed in this review. These therapeutics follow a similar mode of action that involves the inhibition of DNA replication, interference with RNA transcription, blockage of cell division, and inhibition of topoisomerases, which prevent cell division and tumour growth.^{7–9} Unfortunately, these therapies cannot distinguish between the uncontrollable growth of cancer cells and fast replicating healthy cells. Therefore, therapeutics that can target and treat cancers *via* new modes of action are particularly sought after. Crucial to their effective construction, aqueous solubility, cell uptake and tumour specificity are important factors that need to be considered.

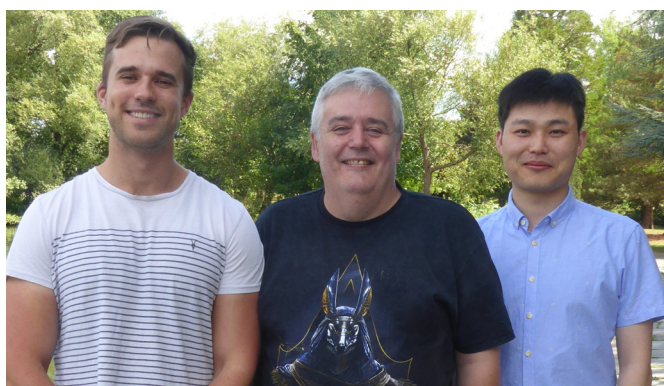
In recent years, a significant focus has been directed towards developing prodrug-based strategies for cancer therapy. Anticancer prodrugs are “masked” inactive chemotherapeutics designed to have little to no pharmacological activity. Beneficially, in the

presence of a cancer-specific biomarker, the “masked to unmasked” conversion takes place to produce a therapeutic effect at the desired location. This enhanced specificity is designed to minimize off-target toxicities and improve therapeutic efficacies. Several prodrug examples have been reported with high tumour specificities, minimal off-target toxicities, and higher maximum tolerated doses.^{10–13} Prodrugs have also been developed to improve aqueous solubility and cell permeability of a therapeutic.¹⁴ Select examples of prodrug-based cancer treatments include FDA-approved Zytiga (hormone-refractory prostate cancer) and Ixazomib Citrate (multiple myeloma). Over recent decades, design strategies for small-molecule prodrugs with improved therapeutic efficacy and minimal side effects during disease treatments have been reviewed.^{15–17} The aforementioned review articles focus on the design and application strategies of prodrugs for individual chemotherapy or phototherapy, with minimal elaboration on the commonalities in the design of these therapeutic modalities. This tutorial review focuses on prodrugs that can be activated by specific biomarkers found in the tumour microenvironment or by external stimuli, as well as theranostics for precision-enhanced chemo- and phototherapy. In this review, we discuss recently reported prodrug strategies for chemo- and phototherapeutic applications. Each prodrug design and its advantages are



Jong Seung Kim (Left) and Mingle Li (Right)

Jong Seung Kim (Left) received his PhD from the Department of Chemistry and Biochemistry at Texas Tech University in 1993. Currently he is a full professor in the Department of Chemistry at Korea University in Seoul. Mingle Li (Right) obtained his PhD degree in 2019 from the Dalian University of Technology under the supervision of Prof. Xiaojun Peng. Currently he is a Research Professor in Prof. Jong Seung Kim's lab at Korea University in Seoul.



Adam C. Sedgwick (Left), Tony D. James (Centre) and Luling Wu (Right)

Adam C. Sedgwick (Left) is a Glasstone Research Fellow at the University of Oxford and is a Junior Research Fellow at Jesus College, Oxford. His research focus is on developing new chemical tools for molecular imaging, sensing, and theranostic applications. His h-index is 29 (Google Scholar). Tony D. James (Centre) is a Professor at The University of Bath and Fellow of the Royal Society of Chemistry. His research interests include many aspects of Supramolecular chemistry, including probes for redox imbalance and theranostic systems. His h-index is 80 (Google Scholar). Luling Wu (Right) was awarded scholarships by the China Scholarship Council and University of Bath to carry out a PhD at the University of Bath. His research focuses on fluorescent probes/prodrugs and imaging. His h-index is 17.



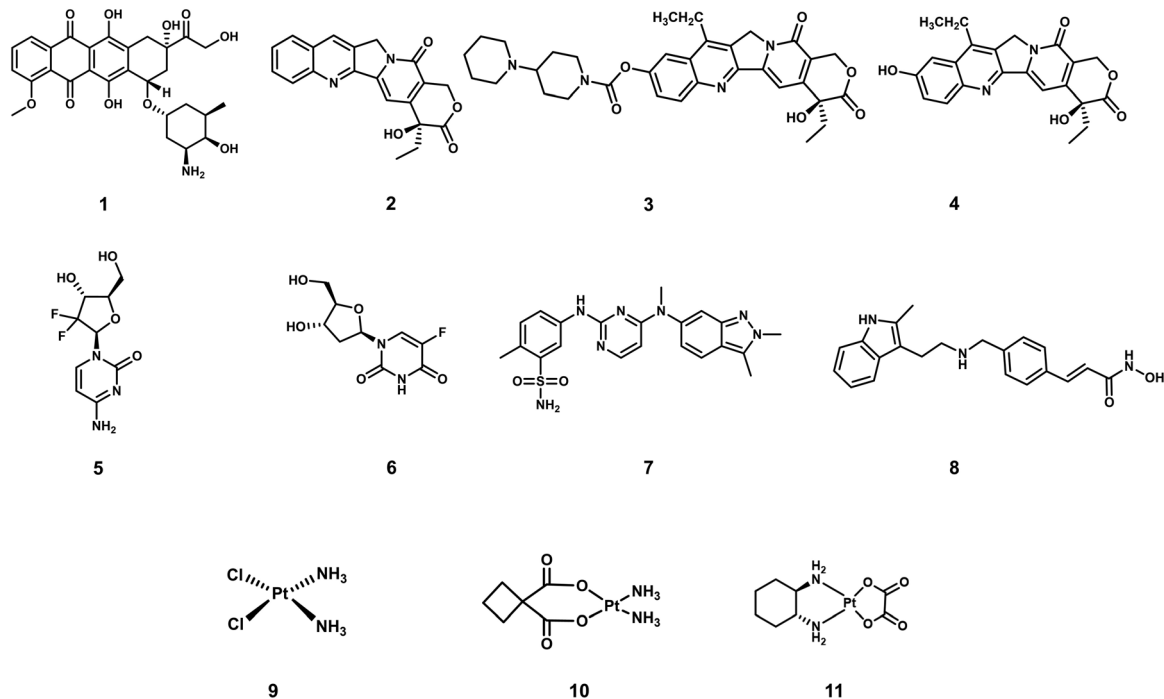
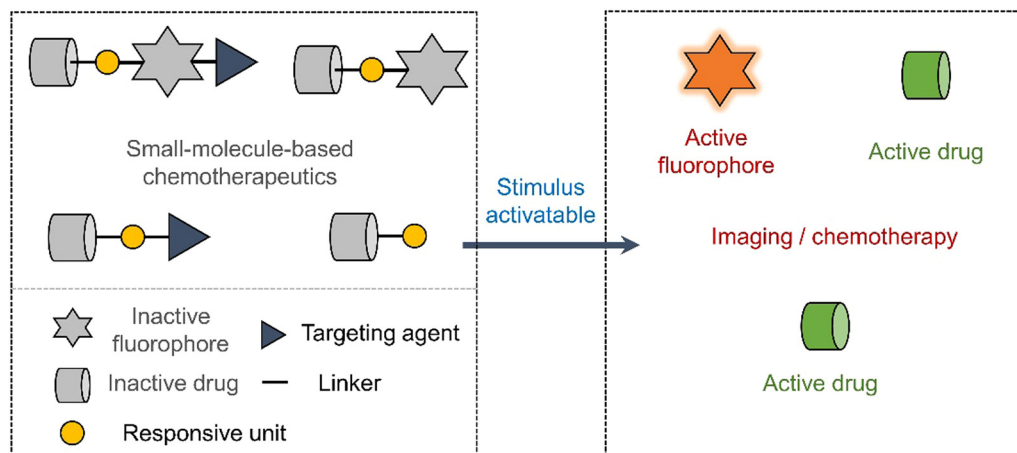


Fig. 1 Chemical structures of select examples of FDA-approved chemotherapeutics.

compared to conventional chemotherapeutics and by describing each design strategy, we believe the importance of understanding factors that support the tumour microenvironment (TME) and aid cancer cell proliferation will become apparent to the reader. Moreover, several reported prodrug examples that provide an integrated diagnostic response and the advantages of these theranostic systems are also discussed. The current limitations associated with each therapeutic strategy are discussed, and in addition, we will provide our own perspectives on the future directions for this important research area. This article provides a general overview on the design of prodrugs and activatable phototherapeutics which we believe will facilitate the development of improved therapies.

2. The design of small-molecule prodrugs and activatable phototherapeutics

Early prodrug examples were designed to improve solubility, cell permeability, and chemical stability. In recent years, the design of prodrugs has been focused towards overcoming the toxicity issues surrounding chemotherapeutics.^{18,19} Developing prodrugs usually requires synthetic modification of a therapeutic to “mask” the cytotoxic properties; this is often achieved by introducing a biomarker responsive “protecting group” (Scheme 1). Other prodrugs strategies include the use of

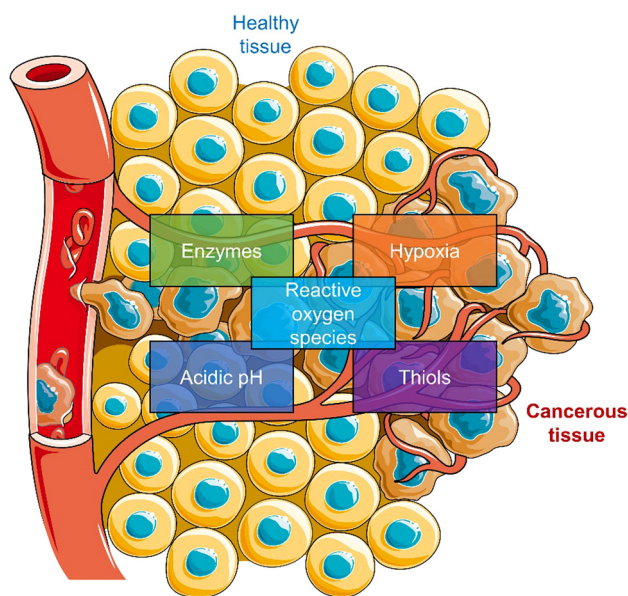


Scheme 1 Basic schematic diagram of the various design strategies used to create small-molecule prodrug-based anticancer agents.



protecting groups that are responsive to external stimuli such as light or ultrasound irradiation (Scheme 1).^{20,21} In some cases, the drugs themselves are fluorescent (*e.g.*, doxorubicin,²² camptothecin²³ and deferasirox²⁴); as a result, a fluorescence “OFF-ON” switching can be used to highlight disease specific regions (*i.e.*, diagnosis) while providing the real-time visualization of drug activation (Scheme 1). This is often referred to as theranostics (theragnostics).²⁵

Most reported prodrug strategies employ protecting groups that respond to specific biomarkers found within the tumour microenvironment (TME). TME is the environment around the tumour consisting of the blood vessels, rapidly proliferating cancer cells, endothelial cells, immune cells, fibroblasts, and an extracellular matrix.^{26,27} Several physiological processes are upregulated to sustain this microenvironment, which can be exploited for the design of prodrugs. A notable example is the enhanced cellular metabolism referred to as the Warburg effect, which leads to a lower extracellular pH being observed in tumour tissues (between 6.5 to 7.2) compared to that of normal tissues (pH around 7.4).²⁸ Other examples include inflammatory biomarkers, such as reactive oxygen species (ROS) and enzymes, matrix metalloproteinases (MMPs), and hyaluronidase (HADase). Hypoxia is a particular trait of the TME, resulting from increased oxygen consumption by cancer cells combined with an inadequate blood supply throughout the tumour (Scheme 2).²⁹ Intracellular biomarkers such as glutathione (GSH), γ -glutamyltranspeptidase (GGT), and β -galactosidase (β -Gal)^{30,31} are upregulated in cancers; for example, the concentration of glutathione (GSH) in cancer cells is ~ 20 mM, much higher than that in healthy cells (~ 5 mM).³²

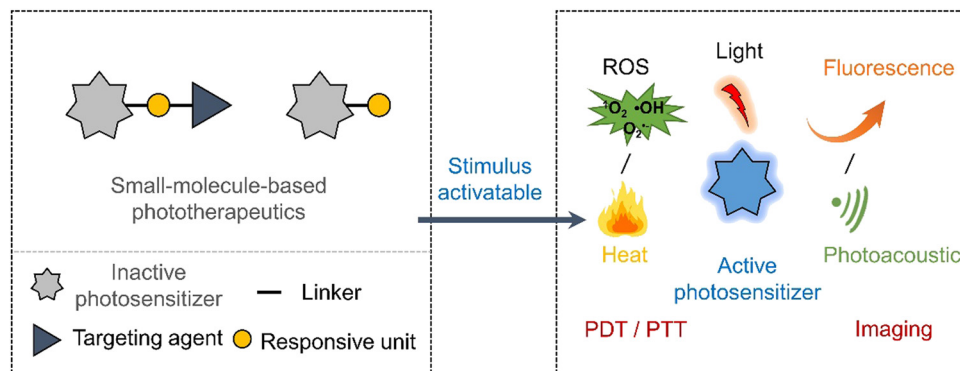


Scheme 2 Schematic illustration of the characteristics of TME used to develop TME-responsive small-molecule prodrugs and activatable “smart” phototherapeutics (parts of the scheme were drawn by using pictures from Servier Medical Art. Servier Medical Art by Servier is licensed under a Creative Commons Attribution 3.0 Unported License (<https://creativecommons.org/licenses/by/3.0/>)).³³

Despite the promise shown by currently reported prodrug strategies,¹¹ the highly cytotoxic nature often remains and therefore impedes clinical translation. Low maximum tolerated doses are often seen due to their inability to target the tumour site effectively. Light-based treatments such as photodynamic therapy (PDT) offer a high precision, non-genotoxic and non-invasive approach to eradicate cancerous tissues.^{34,35} PDT is an FDA-approved therapeutic modality with examples including Photofrin[®] (porfimer sodium), Visudyne (Verteporfin or BPD-MA), and 5-aminolevulinic acid (ALA).³⁶ In brief, PDT relies on the activation of photosensitizers by irradiation with a specific wavelength of light to produce cytotoxic ROS. The light-mediated production of ROS eradicates the surrounding diseased tissue. There are two types of photosensitized reactions, *i.e.*, type I and type II. For type I, the light-activated photosensitizer undergoes electron transfer with surrounding biomolecules to afford free radicals, such as superoxide radicals and hydroxyl radicals, which can induce a cytotoxic effect.^{37,38} For type II, the triplet excited state of the light-activated photosensitizer undergoes direct energy transfer with ground-state oxygen ($^3\text{O}_2$) to generate singlet oxygen ($^1\text{O}_2$), thus directly leading to apoptosis or necrosis of cancer cells.³⁹ Therefore, the hypoxic environment of a solid tumour often leads to poor efficacies being observed for Type II PDT agents.^{40–42} New photosensitizers that exhibit excellent tumour localising properties, high singlet oxygen quantum yields, and high therapeutic efficacies are continuously being reported.⁴³ With the advancement in imaging technologies, photosensitisers are viewed as “all in one” in phototheranostics.⁴³ However, light irradiation of the diseased tissue often results in damaging the surrounding healthy tissue. Efforts to improve the precision of PDT have led to the development of activatable photosensitizers (Scheme 3), in which the PDT excitation wavelength only exists when it is activated by a disease specific biomarker. This strategy exploits the changes in photo-physical properties of a photosensitizer,⁴⁴ therefore they can be seen as analogous to prodrugs or as fluorescent probes.⁴⁵ Their simultaneous use as fluorescence probes creates the additional ability to define tumour margins between cancerous and healthy regions *via* imaging-guided PDT.⁴⁶

In recent years, photothermal therapy (PTT) has been of great interest to researchers because of its minimal invasiveness, ability to overcome the oxygen requirements of PDT, and capacity to eradicate malignant tumours.⁴⁷ PTT, which relies on the production of heat, is an attractive alternative approach to PDT.⁴⁸ The photothermal agents generate heat upon light irradiation and induces cell death by the apoptosis or necrosis pathways.⁴⁹ Reported PTT agents range from inorganic materials (*e.g.*, silver, gold, transition-metal, and platinum nanoparticles and rare earth ions doped nanocrystals) to small organic molecules (*e.g.*, cyanines, croconaines, porphyrins, and diketopyrrolopyrroles).⁵⁰ However, compared to inorganic-based PTT agents, organic-based PTT agents are more promising due to their excellent biocompatibilities and ease of modification.^{48,51} An effective PTT agent is needed to accumulate at the solid tumour, and upon the light irradiation, the tumour temperatures exceeds >48 °C. Since PTT exploits the non-radiative decay of molecules, high molar absorptivity





Scheme 3 Basic schematic diagram of the various design strategies used to create small-molecule activatable “smart” phototherapeutics.

and low fluorescent quantum yield are desirable for PTT agents. To avoid the off-target phototoxicities for PTT agents, researchers have focused on developing activatable systems that undergo changes to photophysical properties at the region of interest (Scheme 3). This approach provides the potential to achieve differential cytotoxicity between cancer and nearby-normal cells. For the design of an activatable PTT agent, the presence of disease-related biomarkers needs to induce a change in UV absorption. These are similar design requirements to photoacoustic (PA) probes,⁵² PTT is therefore often used in combination with PA to facilitate real-time tracking *in vivo*.^{53–56} For a more extensive overview on the area of PTT and PA, the reader can be directed to an excellent review by Liu *et al.*⁵¹

Unfortunately, even though an experimental therapeutic can show promise *in vitro*, it is often not translated into small animal studies. A common issue mainly due to the factors such as limited tumour specificity, poor aqueous solubilities, rapid metabolism/excretion, and poor diffusion of prodrugs into the tumour mass. Drug permeation throughout the tumour is crucial since residual cancer cell survival can promote tumour regrowth and drug resistance. To overcome this hurdle, researchers have focused on introducing targeting units onto known chemotherapeutics.⁵⁷ Such targeting functionalities exploit known interactions between small molecules and protein receptors overexpressed on the membrane surface of tumour cells or active transport mechanisms.⁵⁷ This review will be broken down into sections by method of activation of a chemotherapeutic or phototherapeutic with cancer-related biomarkers, including enzymes, ROS, thiols, and other biomarkers specific to the TME and external stimulants, such as light and X-rays.

3. Small-molecule prodrugs and activatable phototherapeutics

3.1 Enzyme-responsive prodrugs and enzyme-mediated activation of phototherapeutics

Enzymes are critical to a number of metabolic processes and are highly specific with regards to the type of substrate, and their catalytic function.⁵⁸ It is now well-understood that several enzymes are overexpressed in a number of types of cancer cell

lines. Elevated enzymatic activities are believed to facilitate several pathological processes, ranging from tumour angiogenesis, cell invasion to metastasis.^{59,60} With the help of biologists, the biological substrates of these enzymes have been disclosed. This knowledge has then been used to functionalise chemotherapeutics, fluorescent probes and photosensitisers with enzyme-cleavable/activatable motifs and achieve controlled intracellular release of the active molecules. In this section, we discuss select examples of enzyme-based activatable chemotherapeutics and activatable “smart” phototherapeutic strategies.

Most previous enzyme-based research has focused on the substrate specific properties of hydrolases (such as ester bonds that can be broken by esterases or short peptide substrates that can be broken by proteases) to construct prodrugs or “smart” phototherapeutic agents that enable controlled release and accumulation of drugs at specific biological targets. Most enzymatic reactions are fast and efficient, specific and mild compared to other stimulation conditions, and enzymatic-based therapeutic agents are able to exhibit higher reaction efficiencies (Table 1).

Carboxylesterase (CE) enzyme is known to hydrolyse several carboxyl esters and amides catalytically, and it is seen as a useful tumour biomarker for patient staging.^{61,62} In light of this, a number of CE-responsive prodrugs have been developed for theranostic applications.^{63–65} Sharma *et al.* developed the prodrug **12** (Fig. 2a).⁶⁶ **12** consists of a DOX that possessed both fluorescence and anti-cancer properties, a dichloroacetic acid (DCA) and a lipophilic cationic triphenylphosphonium (TPP) mitochondrial targeting unit. DOX is an FDA-approved chemotherapeutic with the nickname “the Red Devil” attributed to its fluorescent red appearance in solution. It is routinely used in the clinic to treat a range of cancers due to its broad anti-tumour activity; however, debilitating side effects, ranging from hair loss, bone marrow suppression, vomiting, and cardiotoxicity, are observed from its use. In addition, DOX-induced tumour cell resistance has limited its use as a first-line agent.^{67,68} DCA is a mitochondrial PDK inhibitor that promotes glucose oxidation rather than glycolysis by inhibiting pyruvate dehydrogenase kinase and increasing the flux of pyruvate into the mitochondria, helping overcome drug resistance.⁶⁹ **12** was therefore designed to rationalise the fight against drug resistance by combining subcellular organelle (mitochondria) targeting with the modulation of tumour



Table 1 Summary of enzyme-responsive prodrugs and enzyme-mediated activation of phototherapeutics discussed in this review

	Stimuli	Active anti-cancer therapeutic agent	Treatment	<i>In vitro</i> models	<i>Ex vivo/in vivo</i> models	Ref.
12	CE	DOX	Chemotherapy	A549, HepG2, MCF7 and MCF7/DOX cells	DOX-resistant MCF7/DOX-derived tumour-bearing mice	66
13	Esterase and caspase-3	DOX	Chemotherapy	U-87 MG cells	U-87 MG tumour-bearing mice	74
14	Cathepsin B	TPECM	PDT	MDA-MB-231 cells	—	77
15	GGT	Selenium-based rhodamine derivative	PDT	SHIN3 cells	Tumour-bearing CAM model	84
16	β -Gal	Hemicyanine dye	PTT	SKOV3 cells	SKOV3 tumour-bearing mice	89
17	β -Gal	MMAE	Chemotherapy	KB and HeLa cells	KB tumour-bearing mice	30

Note: CE: carboxylesterase, GGT: γ -glutamyltranspeptidase, β -Gal: β -galactosidase, DOX: doxorubicin, MMAE: monomethyl auristatin E, CAM: chick chorioallantoic membrane, PDT: photodynamic therapy, PTT: photothermal therapy, —: not mentioned.

cell metabolism *via* DCA. As shown in Fig. 2a, the CE-mediated amide-bond hydrolysis of **12** resulted in the simultaneous release of DCA and subsequent release of DOX, which was confirmed by the enhanced fluorescence emission of free DOX characteristic fluorescence at 560 nm. Such CE-mediated transformation to release free DOX was further proved by high-performance liquid chromatography (HPLC) and liquid chromatography-mass spectroscopy (LC-MS). This change in fluorescence permitted the use of fluorescence microscopy to visualise the CE-mediated release of DOX in live cells. Cellular imaging and cytotoxicity studies were performed in a series of different cancer cell lines

(*i.e.*, A549-human lung adenocarcinoma epithelial cell; HepG2-human hepatic cancer cell) and normal cell lines (*i.e.*, NHDFs-normal human dermal fibroblasts; IMR90-normal lung fibroblast cell line). In accordance with the design expectations, **12** was shown to target the mitochondria of cancer cells rather than normal cells because of the higher mitochondrial membrane potentials of cancer cells ($\Delta\psi_m = \sim -220$ mV) compared with normal cells (~ -140 mV). The cytotoxicity was dependent on the CE activity, which was confirmed using a known CE inhibitor bis-(4-nitrophenyl)phosphate (BNPP). As a result the cytotoxicity of **12** towards healthy cells was mitigated. Interestingly, DOX

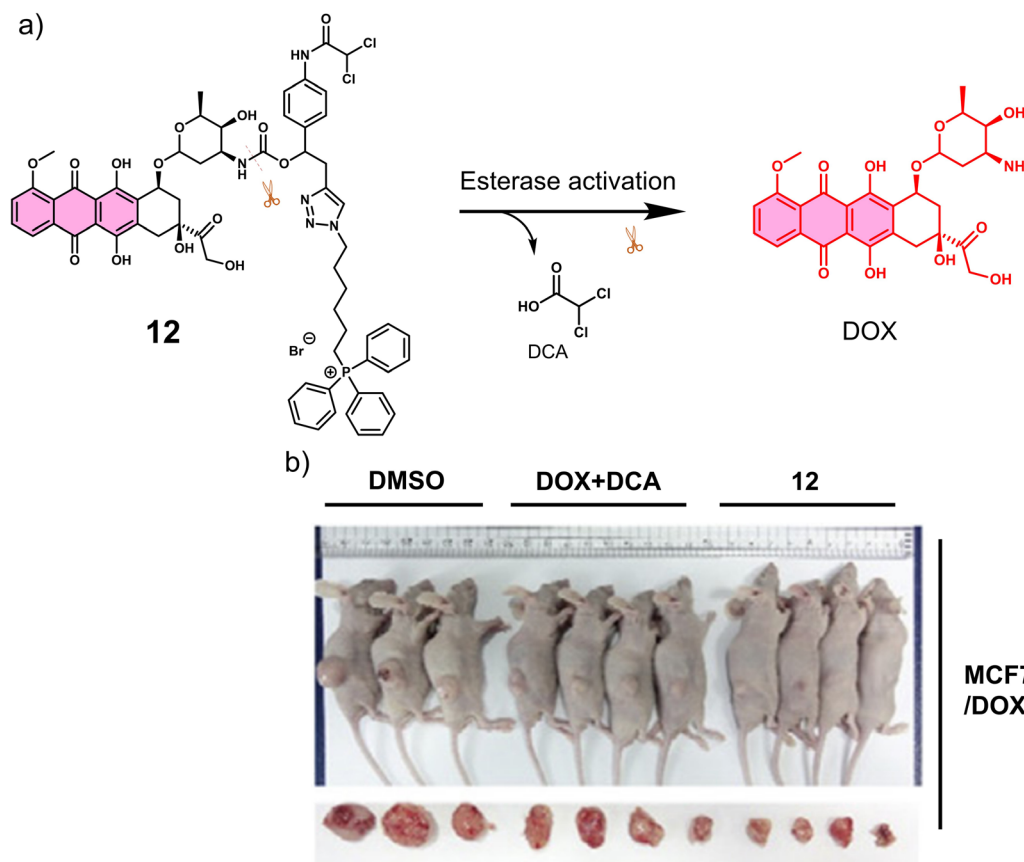


Fig. 2 (a) Chemical structure of **12** and its CE-mediated release of DOX in MCF7/DOX-resistant cell lines. (b) Representative images of DOX-resistant MCF7/DOX-derived tumour-bearing mice treated with DMSO control, DOX + DCA, and **12**. Reproduced with permission from ref. 66. Copyright (2018) Elsevier Inc.



was translocated to the nucleus from its initial mitochondrial localisation, thus reducing the efflux of DOX from cancer cells. Generally, ATP-binding cassette (ABC) transporter-mediated enhanced drug efflux (a term describing the expression of a transport protein that moves a drug from the intracellular to the extracellular space, thereby reducing the intracellular drug concentration) is known to contribute to drug resistance.⁷⁰ Tumour cells normally produce ATP by converting pyruvate to lactate through aerobic glycolysis, which contributes to tumour cell proliferation and promotes tumour metastasis.⁷¹ However, the released DCA fragment from **12** was shown to modulate the pyruvate metabolic pathway in cancer cells and trigger mitochondrial dysfunction by reducing lactate accumulation, glucose uptake and intracellular ATP levels. The above properties were believed to enable **12** to overcome drug resistance in MCF7 and DOX-resistant MCF7/DOX cell lines. The intraperitoneal administration of **12** in DOX-resistant MCF7/DOX-derived tumour-bearing mice models significantly inhibited tumour growth compared to the tumour volumes of control groups (DMSO and a simple mixture of DOX + DCA) (Fig. 2b). This work provides a new approach to overcome drug resistance by reversing cancer cell metabolism prior to drug activation, thereby increasing cancer cell apoptosis and inhibiting the regeneration of drug-resistant tumours.

Proteases are enzymes that hydrolyse peptide bonds in proteins or peptides and play an important role in many diseases and biological processes, such as fetal and postnatal development, reproduction, signal transduction, immune response, many autoimmune and degenerative diseases, and cancer. Most proteases are specific in their mode of action, with hydrolysis usually occurring at specific amino acid residues, sequences or peptide bonds in the vicinity of the substrate protein or peptide.⁷² Caspase-3 is a cysteine-aspartic acid protease that cleaves various cellular targets (*e.g.*, DNA) to mediate programmed cell death (apoptosis).⁷³ **13**, referred to as a “self-triggered apoptosis enzyme prodrug,” is a caspase-3 responsive prodrug (Fig. 3a).⁷⁴ **13** consists of the cancer cell targeting integrin $\alpha_v\beta_3$ -RGD (Arg-Gly-Asp) tripeptide, caspase-3 cleavable linker (DEVD tetrapeptide, Asp-Glu-Val-Asp), cellular esterase cleaved ester linkage and DOX. Notably, DOX is known to induce apoptosis-related markers, such as caspase-3.⁷⁵ Therefore, it can activate more DOX, leading to the release of more caspase-3, and the cycle repeats itself leading to enhanced and widespread anti-cancer activity. As shown in Fig. 3a, the presence of caspase-3 results in the cleavage of the DEVD linker, followed by subsequent ester hydrolysis affords free DOX. Therefore, in this strategy, the release of DOX is designed to increase the intracellular expression of caspase-3

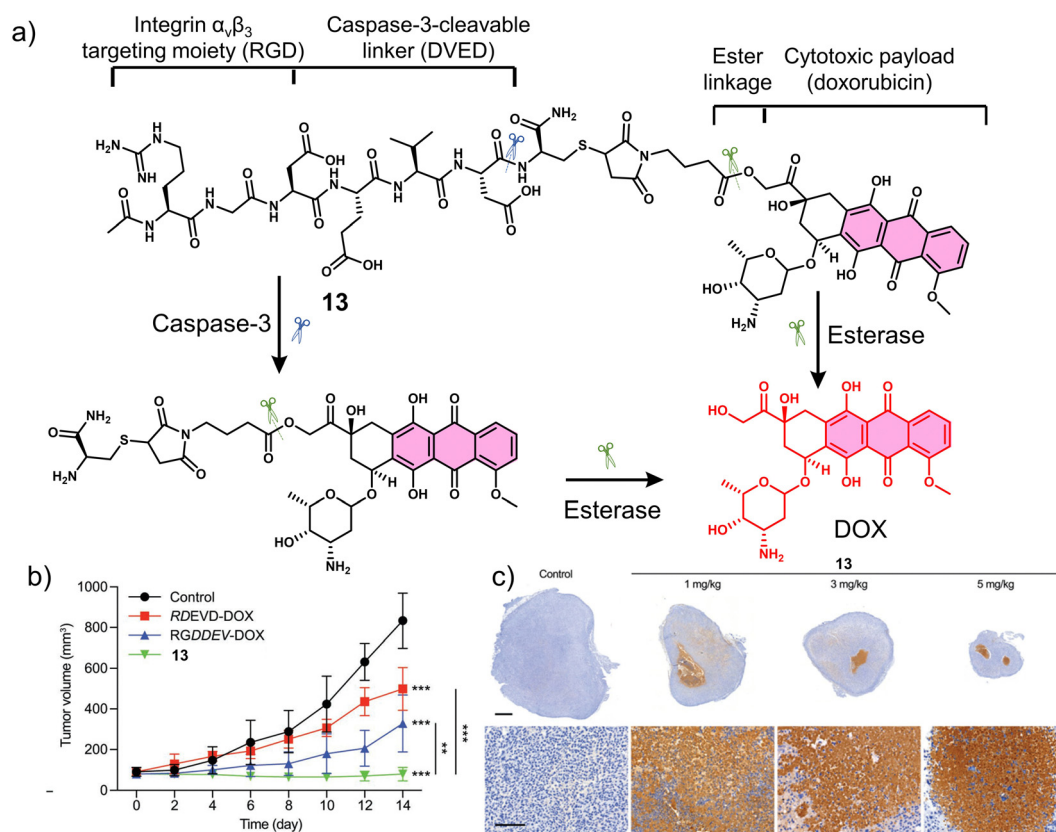


Fig. 3 (a) Chemical structure of **13** and its enzymatic products (caspase-3 and esterase). (b) Tumour growth profiles of U-87 MG tumour-bearing mice that received saline as control, RDEVD-DOX (3 mg kg⁻¹), RGDDEV-DOX (3 mg kg⁻¹), or **13** (3 mg kg⁻¹); *n* = 6. Data are mean \pm s.d. ***P* < 0.01, ****P* < 0.001. (c) Immunostaining of caspase-3 in the tumour sections from U-87 MG tumour-bearing mice treated with control and **13** (1 mg kg⁻¹, 3 mg kg⁻¹, and 5 mg kg⁻¹); scale bar: 1 mm (upper panels) and 100 μ m (lower panels). This is an open access article distributed under the terms of the Creative Commons CC BY license, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.



and enhance the enzymatic turnover of **13** (self-triggered). Esterase- and caspase-3-mediated hydrolysis was confirmed by HPLC (in solution) and confocal laser scanning microscopy (CLSM) images (*in vitro*). The selective cytotoxicity of **13** was found only in U-87 MG (human glioma cells, integrin $\alpha_v\beta_3$ -positive) cell line rather than HT-29 (human colon cancer cells, integrin $\alpha_v\beta_3$ -negative) cell lines. This selectivity was attributed to integrin-mediated endocytosis by the RGD targeting peptide, confirmed using a fluorescence imaging study. Corroborating the “self-triggered” properties, the release of DOX from **13** treated cells was found to induce a 154-fold increase in caspase-3 activity. Using *in vivo* models, compared to controls (RDEV-D-DOX (RGD-deficient) and RGDDEV-DOX (DVED-deficient)), **13**-administered U-87 MG tumour-bearing mice were found to have significantly reduced tumour volumes (near 0 mm³ vs. greater than 200 mm³ (control)) after 7 days of treatment (3 mg kg⁻¹, once a day for seven days) with 14 days of monitoring (Fig. 3b and c). This report demonstrates both the importance of cancer-targeting units and a prodrug strategy to achieve ideal antitumour activity *in vivo*.

Cathepsin B is a lysosomal protease overexpressed in a range of cancers. It is responsible for promoting the spatial expansion of tumours, angiogenesis, and the metastasis of tumour cells within and outside the blood vessels through protein hydrolysis. Since the discovery of its specificity for the peptide sequence Gly-Phe-Leu-Gly (GFLG), numerous groups have developed cathepsin B-responsive systems.⁷⁶ Liu and co-workers developed the multifunctional PDT agent, **14**. This system consisted of

an AIEgen (TPECM) that possessed both fluorescence and PDT properties, GFLG as the cathepsin B-responsive linker, cRGD for actively targeting $\alpha_v\beta_3$ integrin, and a hydrophilic peptide sequence (Fig. 4).⁷⁷ AIEgens are molecules that emit minimal fluorescence in a mono-dispersed state; however, the fluorescence output is significantly enhanced when the molecules are aggregated (or rigidified).^{78,79} The initial good hydrophilicity of **14** meant that the AIEgen was unable to form fluorescent aggregates and produce ROS under light irradiation. However, the cathepsin B-mediated cleavage of the GFLG linker resulted in a concomitant increase in fluorescent intensity at 615 nm and a significant increase in ROS generated in the solution. This was further confirmed through fluorescence microscopy in MDA-MB-231 cells with high signal-to-noise ratio. Dose-dependent cytotoxicity was seen for **14** upon white light irradiation (0.25 W cm⁻², 2 min) in MDA-MB-231 cells over-expressing $\alpha_v\beta_3$ integrin when compared to MCF-7 and 293T cell lines ($\alpha_v\beta_3$ integrin-negative controls). The cathepsin-B-mediated activation and phototoxicity of **14** were confirmed through the pre-treatment of MDA-MB-231 cells with cRGD, CA-074-Me (a cathepsin B inhibitor), and vitamin C (a ROS scavenger). All were shown to suppress its light-mediated cytotoxicity. Unique to this strategy is that this method does not require the use of any protecting groups or FRET pairs to deactivate the phototoxicity of the photosensitizer. Thus, it illustrates the unique advantages of AIE for the development of activatable photosensitizers. Although limited by the short

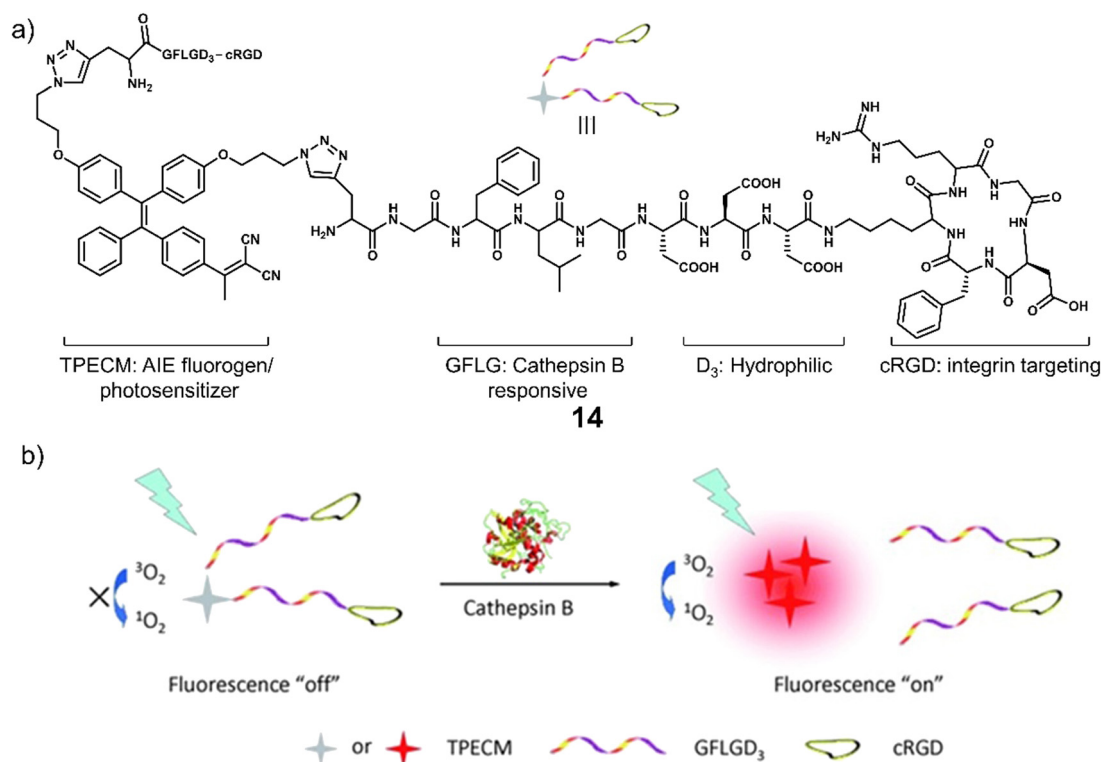


Fig. 4 (a) Chemical structure of **14** and (b) the schematic illustration of **14** being activated by cathepsin B with a fluorescence “turn-on” signal and the ability to generate ROS upon white light irradiation. Reproduced with permission from ref. 77. Copyright (2014) Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim.



emission wavelength, this study sets the basis for the use of AIE-active, enzyme-responsive PDT agents for cancer therapy.

γ -Glutamyltranspeptidase (GGT) is a cell surface N-terminal nucleophilic hydrolase that is associated with GSH metabolism. GGT is known to be overexpressed during oxidative stress and is believed to be associated with tumour progression, invasion, and the development of chemotherapeutic resistance.^{80,81} For these reasons, researchers have focused on the development of GGT-responsive systems for fluorescence imaging and therapeutic applications.⁸² Using the knowledge of prior GGT-responsive systems,⁸³ Urano and co-workers developed a selenium-based rhodamine derivative, **15**, for GGT-activated PDT (Fig. 5).⁸⁴ The replacement of the oxygen atom on rhodamine with a selenium atom was found to increase the singlet oxygen quantum yield by 10.8 fold.⁸⁵ Initially, **15** was non-fluorescent and displayed minimal phototoxicity under 532 nm light excitation. Which was attributed to the rhodamine derivative HMSeR being “masked” with a glutamate functionality. In the presence of GGT, the glutamate protecting group was hydrolysed and released photoactive HMSeR, which is confirmed *via* increased fluorescence emission at 562 nm and the change in HPLC retention times. Light irradiation (510–550 nm, 50 mW cm⁻², 1 min) of a SHIN3 cell line (human ovarian cancer cells with high expression of GGT) incubated with **15** (10 μ M) resulted in a remarkable photocytotoxicity being observed (20% cell viability). The intracellular GGT-activation of **15** was confirmed using the pre-treatment of SHIN3 cells with GGT inhibitor, GGsTop. The *in vivo* antitumour studies in a tumour-bearing chick chorio-allantoic membrane (CAM) model revealed that **15** was able to selectively eradicate GGT-positive tumours without causing unwanted phototoxicity to the surrounding healthy tissues. This study represents one of the first aminopeptidases-activatable PDT agents, which offers scope for the subsequent development of a variety of peptidase (overexpressed in a variety of different tumours)-activated PDT agents and long excitation wavelength GGT-responsive analogues.

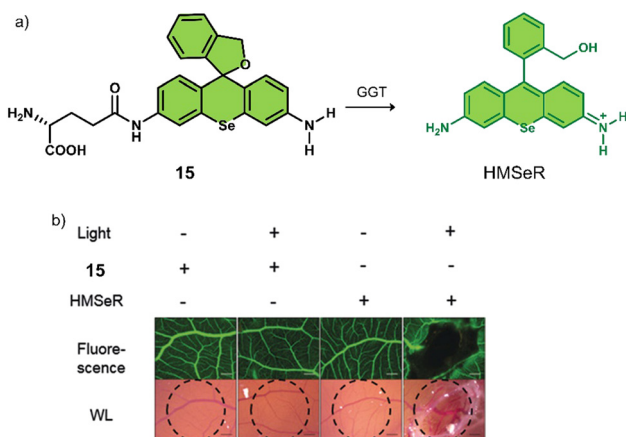


Fig. 5 (a) Chemical structure of **15** and the GGT-activatable release of HMSeR. (b) Evaluation of vessel occlusion on CAM after photoirradiation (510–550 nm, 50 mW cm⁻², 15 min) in the presence of **15** (100 μ M, 25 μ L) and HMSeR (100 μ M, 25 μ L). Reproduced with permission from ref. 84. Copyright (2017) Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim.

Glycosidases are involved in a variety of intracellular catabolic processes, such as endocytosis of glycoproteins and deglycosylation modifications following transport to lysosomes.⁸⁶ Deficiency of glycosidases creates cellular storage disorders and their abnormalities can also cause severe multisystem disorders, therefore, glycosidases have become biomarkers used in the diagnosis of cancer and in monitoring its malignant progression, recurrence and prognosis.⁸⁷ Glycosidase-responsive therapeutic agents are primarily activated by enzyme catalysis of substrates to release active therapeutic agents. β -Gal is a glycoside hydrolase that catalyses the hydrolysis of galactoside residues from glycoconjugates. The overexpression of β -Gal is associated with primary ovarian cancer and cell senescence.⁸⁸ Zhen *et al.* developed a NIR fluorescent and PA-imaging PTT agent, β -Gal-responsive phototheranostic **16** used for the image-guided PTT treatment of ovarian cancer SKOV3 tumour-bearing mice models (Fig. 6).⁸⁹ **16** consists of a galactose-functionalised NIR hemicyanine dye (CyOH) that possesses both imaging and PTT properties. A poly(ethylene glycol) (PEG) chain was added to provide good aqueous solubility. **16** initially exhibits minimal fluorescence and photoacoustic properties; however, the presence of β -Gal removes the galactosyl group to afford free CyOH-P, leading to a simultaneous enhancement in optical (fluorescent intensity and PA signal) and photothermal properties. A change in absorption from 600 nm to 688 nm and an increase in fluorescence intensity at 720 nm was observed. The co-incubation of **16** with SKOV3 (β -Gal-overexpressed ovarian cancer cells) and NIH-3T3 cells (mouse embryonic fibroblast cells, control) led to an enhanced fluorescent intensity being observed only in SKOV3 cells. Suggesting good selectivity for β -Gal in biological environments. **16** displayed significant phototoxicity towards SKOV3 cells following 680 nm laser irradiation (0.6 W cm⁻², 5 min). The NIR fluorescence and the photoacoustic (PA) signals of the β -Gal-response to **16** were found to reach a maximum value at the tumour site 1 h post-injection in SKOV3 tumour-bearing mice. This confirmed overexpression of β -Gal at the tumour and identified an optimal time point for PTT treatment. A significant reduction in tumour volume was observed compared to the control (saline) when **16** (300 μ M) was administered *via* tail vein, and the tumour region was subjected to light irradiation (680 nm laser, 0.6 W cm⁻², 5 min). **16** represents the first example of combining both optical imaging and PTT.

To improve the selectivity of β -Gal-activated prodrugs, Legigan *et al.* developed a folic acid-functionalised β -Gal responsive prodrug, **17** (Fig. 7a) for MMAE-based therapy for the treatment of KB-derived xenograft mice models.³⁰ Monomethyl auristatin E (MMAE) is a synthetic anti-tumour agent derived from a peptide-based natural product, dolastatin-10. MMAE inhibits cell division by suppressing microtubule protein polymerization. Unfortunately, due to its potent toxicity, it cannot be used as a drug on its own. MMAE has therefore been approved for use as an antibody-drug conjugate (ADCs), including Tivdak and Adcetris.⁹⁰ Due to the high production costs of ADCs, **17** was designed to offer an alternative therapeutic approach. **17** consisted of a self-immolative linker, which was connected to a folate subunit (targets folate receptors (FR) expressed on the surface of cancer cells), a galactose unit, and the antineoplastic monomethyl auristatin E (MMAE). In the



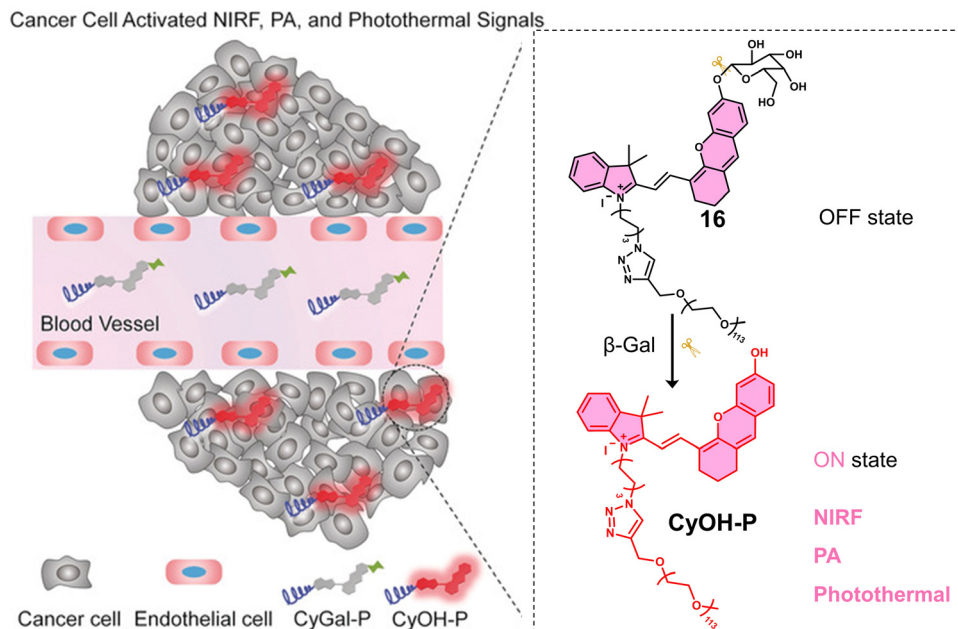


Fig. 6 Chemical structure of **16** and schematic illustration of its β -Gal activated theranostic properties. Reproduced with permission from ref. 89. Copyright (2018) Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim.

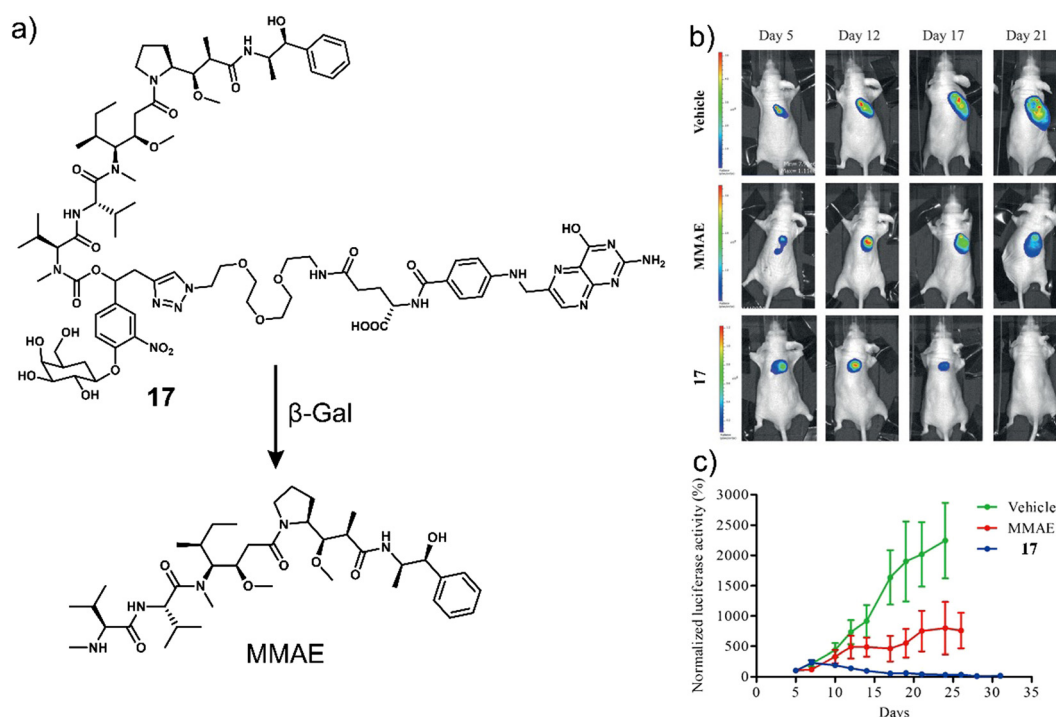


Fig. 7 (a) Chemical structure of **17** and the resultant MMAE product with β -Gal. (b) Representative bioluminescence imaging of luciferase-transfected KB tumour-bearing mice at days 5, 12, 17, and 21 post-implantation when treated with vehicle (5% DMSO in PBS buffer), MMAE (intravenous administration at days 4, 7, 10, 14, and 17) or **17** (intravenous administration at days 5, 7, 10, 12, 14, 17, 19, 21, 23, and 26). (c) Tumour growth inhibition over time (days) with the vehicle, MMAE (1 mg kg⁻¹) or **17** (5 mg kg⁻¹). Reproduced with permission from ref. 30. Copyright (2012) Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim.

presence of β -Gal, the glycosidic bond of **17** was hydrolysed, and the bioactive MMAE was released (confirmed by HPLC). As β -Gal is present in the lysosomes of both healthy and malignant cells,

the introduction of the folate moiety affords specificity towards FR-positive tumour cells, thus avoiding non-selective drug release in non-malignant tissues. High cytotoxicity was seen in



FR-positive KB (human oral epidermoid carcinoma cell line) and HeLa (human cervical carcinoma cell line) cells when compared with FR-negative A549 cells and a positive correlation with the expression of FR on the cancer cells was observed. Interestingly, it was shown that by using a co-culture study that the release of MMAE in FR-positive KB cells could spread to surrounding FR-negative A549 cancer cells and induce cell death. These results highlight the potential of **17** to overcome the significant heterogeneity of tumours, consisting of a wide range of tumour cells that may not express FR on the cancer cells. A significant reduction in the tumour volume with a high survival rate was observed with **17** in luciferase-transfected KB tumour-bearing mice models (Fig. 7b and c). This work represents one of the first generation β -Gal prodrugs that exhibits several advantages over the alternative and expensive antibody-directed enzyme prodrug therapy (ADEPT).

3.2 Prodrugs and activatable phototherapeutics responsive to either oxidative or reducing environments

The dynamic balance between redox states is essential for maintaining the physiological function of cells.⁹¹ Redox homeostasis is achieved by regulating both ROS and reductive species. An imbalance in redox states is associated with cancer, with excessive ROS production and somewhat contradictory, excessive GSH concentrations being observed in cancer cells.⁹² To date, the consensus is that elevated reductive and oxidative stress may exist in different tumours (intertumoural) and even coexist at different progression stages within the same tumour (intratumoural). This includes the suborganelle level within the same cancer cell.⁹³ For these reasons, researchers continue to explore prodrugs that respond to oxidative and reductive environments. In this section, we discuss prodrug examples that are designed to be activated in either reductive or oxidative environments (Table 2).

Several research groups have focused on the development of H₂O₂-responsive prodrugs. Examples include the use of boronate ester and α -ketoamide to construct ROS-activated therapeutic reagents. Kim *et al.* reported boronate ester-functionalized

coumarin fluorophore-SN-38 conjugate (**18**) as a hydrogen peroxide (H₂O₂)-responsive theranostic system. H₂O₂ is well-known to be elevated in cancer cells and is believed to promote their differentiation, growth, and survival.^{94,95} **18** used a boronate ester H₂O₂-responsive moiety to mask the fluorescent properties of the coumarin fluorophore, which was used as a self-immolative linker for the chemotherapeutic SN-38 (Fig. 8a).⁹⁶ The diagnostic boronate-functionalised coumarin allowed the direct monitoring of SN38 release. The addition of H₂O₂ (300 μ M) to an aqueous solution of **18** (5 μ M) was shown to oxidise the boronate moiety and release of SN-38 and fluorescent coumarin within 1 h, which was confirmed using fluorescence spectroscopy and fast atom bombardment mass spectroscopy (FAB-MS). The incubation of B16F10 and HeLa cell lines with **18** exhibited dose-dependent cytotoxicity in the presence of H₂O₂ (100 μ M) for 48 h. Whereas, in the absence of H₂O₂, the cytotoxicity of **18** was found to be lower than SN-38. More importantly, **18** exhibited greater antitumour efficacy than SN-38 and extended the survival period by 6 days when compared with the control group (saline) (Fig. 8b–d). Although moderate antitumour effects were observed, this report has inspired others to develop systems for the fluorescence imaging of drug release *in vitro* and *in vivo*.

The vast majority (>90%) of reported H₂O₂-responsive prodrugs are based on phenylboronic acid/phenylboronic acid ester fragments, but deficiencies such as the high sensitivity of boronic acid esters to hydrolysis, off-target reactions between boronic acid and bio-diols, and the moderate selectivity of boronic acid esters for H₂O₂ limit their clinical translation. The Yin group developed a novel H₂O₂-responsive prodrug, **19**, by functionalising anti-tumour active nitrogen mustard analogues with α -ketoamide unit (Fig. 9a).⁹⁷ The strong electron-withdrawing nitro group of **19** enhance the electrophilicity of the adjacent carbonyl group, allowing a significantly faster reaction with H₂O₂.⁹⁸ Considering the important role of the nitro group of **19** in H₂O₂-induced nucleophilic attack, the authors also synthesized a non-nitro analogue of **20** that is resistant to H₂O₂ as a negative control. **19** exhibited reasonable stability in the absence of H₂O₂ and displayed little cytotoxicity

Table 2 Summary of oxidative or reducing environments-responsive prodrugs and phototherapeutics discussed in this review

	Stimuli	Active anti-cancer therapeutic agent	Treatment	<i>In vitro</i> models	<i>Ex vivo/in vivo</i> models	Ref.
18	H ₂ O ₂	SN-38	Chemotherapy	B16F10 and HeLa cells	B16F10-induced metastatic lung tumour mice	96
19	H ₂ O ₂	Nitrogen mustard BrM	Chemotherapy	HL-60 cells	—	97
21	GSH	CPT	Chemotherapy	HepG2 cells	H22 tumour-bearing mice	101
22	GSH	PPa	PDT	U-87 MG cells	U-87 MG tumour-bearing mice	102
23	GSH	BODIPY fluorophore	PTT	HeLa cells	U14 tumour-bearing mice	103
24	GSH	Gemcitabine	Chemotherapy	A549, U87 and HEK 293 cells	A549 and U87 tumour-bearing mice	104
25	GSH	SN-38	Chemotherapy	LoVo and SW620 cells	SW620 tumour-bearing mice	31
26	H ₂ O ₂ and GSH sequentially	Benzothiadiazole	PDT	HeLa cells	—	108
27–29	GSH	Oxaliplatin	Chemotherapy	A549, A2780, HCT116, CT26 and EMT6 cells	A549, A2780, HCT116, CT26 and EMT6 tumour-bearing mice	115
30	GSH	Cisplatin	Chemotherapy	A2780, HeLa and MCF-7 cells	—	116
31	GSH	Carboplatin	Chemotherapy	A2780, HeLa and MCF-7 cells	—	116

Note: H₂O₂: hydrogen peroxide, GSH: glutathione, CPT: camptothecin, PPa: pheophorbide a, BTd: benzothiadiazole, PDT: photodynamic therapy, PTT: photothermal therapy, —: not mentioned.



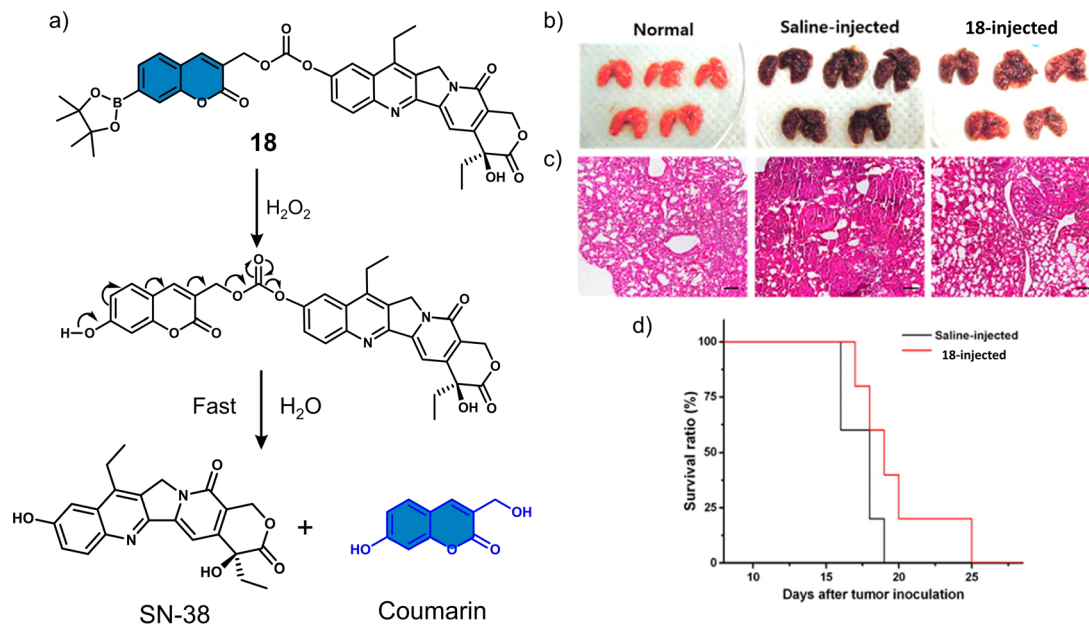


Fig. 8 (a) Chemical structure of **18** and mechanism for the H_2O_2 -mediated release of SN-38. (b) Representative images of lungs isolated from normal mice and B16F10-induced metastatic lung tumour mice that were administered intratracheally with saline or **18** (0.25 mg kg^{-1}). (c) Histological sections of lung tissues with H&E staining. Scale bars: $100 \mu\text{m}$. (d) Kaplan–Meier survival analysis of mice administered with saline or **18**. Reproduced with permission from ref. 96. Copyright (2014) American Chemical Society.

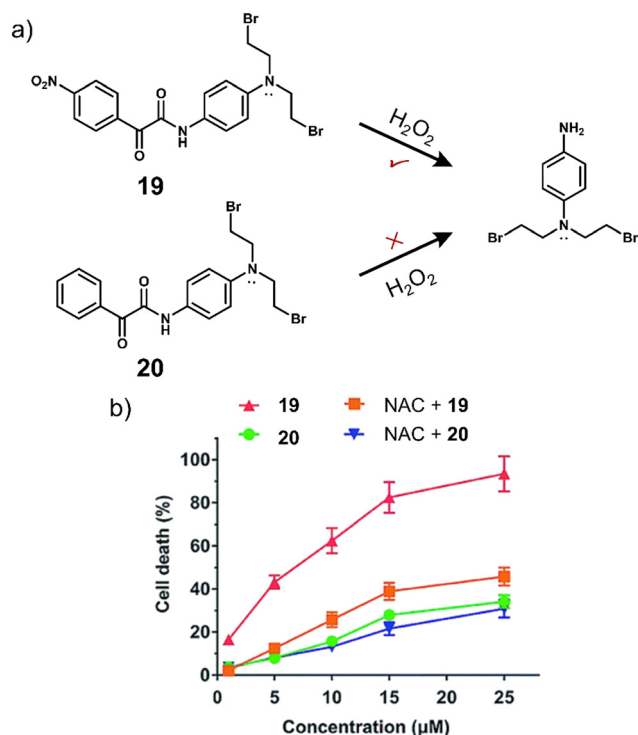


Fig. 9 (a) Chemical structure of **19**, **20** and the H_2O_2 -mediated release mechanism of nitrogen mustard BrM. (b) The effects of NAC on the cytotoxicity of **19** and **20** in HL-60 cells after 72 h incubation. Reproduced with permission from ref. 97. Copyright (2015) The Royal Society of Chemistry.

against cancer cells. HPLC, ^1H NMR and mass spectroscopic analysis confirmed the release of nitrogen mustard *N,N*-bis(2-bromoethyl)benzene-1,4-diamine (BrM) when **19** ($100 \mu\text{M}$) was exposed to H_2O_2 (2 eq.). **19** exhibited a high antiproliferative effect in HL-60 promyelocytic cell line and pretreatment with the H_2O_2 scavenger *N*-acetyl-cysteine (NAC) reduced its antiproliferative capacity. **20**, which is inert to H_2O_2 , exhibited relatively low cytotoxicity to HL-60 cells, and pretreatment with NAC did not reduce the toxicity of **20** to HL-60 cells (Fig. 9b). Mechanistic investigations indicated that **19** selectively releases nitrogen mustard by H_2O_2 activation, which effectively inhibits the proliferation of HL-60 cells through DNA cross-linking and mitochondria-dependent apoptotic pathways.

Several research groups have focused on the development of prodrugs that release a therapeutic in reducing environments. Examples include the use of disulfide bonds and dinitrobenzenesulfonyl groups to construct GSH-activated therapeutic systems. Glutathione (GSH) is a biological tripeptide (glutamic acid, cysteine, and glycine) that contains a free sulfhydryl group. It is the most abundant biological thiol in cells, regulating cellular redox by acting as a key antioxidant. As mentioned in brief earlier, elevated GSH concentrations are found in tumour cells ($2\text{--}10 \text{ mmol L}^{-1}$), which is believed to contribute to tumour progression and drug resistance.⁹⁹ For these reasons, researchers have focused on developing GSH-responsive prodrugs and strategies that can reduce intracellular GSH concentrations to overcome potential drug resistance mechanisms.¹⁰⁰

Kong *et al.* developed a fluorescence-based theranostic **21** for monitoring the GSH-mediated release of chemotherapeutic camptothecin (CPT) in a H22 tumour-bearing mice model.



CPT is a natural product isolated from the Chinese endemic medicinal plant, *Camptotheca acuminata*, and has been found to inhibit the proliferation of cancer cells by selectively inhibiting DNA topoisomerase I (Topo I). **21** was constructed by conjugating a merocyanine-based dye with the anticancer agent, camptothecin (CPT), *via* a GSH-responsive disulfide linker (Fig. 10).¹⁰¹ **21** displayed an initial weak fluorescence emission intensity due to the amino unit of the merocyanine unit being capped. The addition of GSH induced disulfide cleavage and resulted in a significant increase in fluorescence signal at 702 nm, which was ascribed to the release of free merocyanine and CPT. This was further confirmed using high-resolution mass spectroscopy (HRMS). *In vitro* studies revealed that **21** exhibited higher anti-proliferative activity in the hepatoma HepG2 cell line compared to a normal liver HL-7702 cell line. These results were attributed

to the HepG2 cell lines having elevated intracellular GSH concentrations, which was confirmed *via* fluorescence microscopy by monitoring the increase in emission at 702 nm. Using the NIR-emissive properties of merocyanine, it was confirmed that **21** exhibited good tumour localisation properties in H22 tumour-bearing mice after intravenous injection (visualised using an IVIS LuminaIII *in vivo* imaging system).

Ye and co-workers constructed the tumour-targeting GSH-activated PDT agent, **22**, that could be visualised by dual-modality NIR fluorescence/MRI imaging to allow image-guided PDT (Fig. 11).¹⁰² Combining the ability of MRI to image the whole body and the high sensitivity of fluorescence for functional imaging of tumour cells, the diagnostic accuracy and reliability can be substantially improved. **22** consisted of the photosensitizer pheophorbide a (PPa), the thiol-responsive disulfide linker,

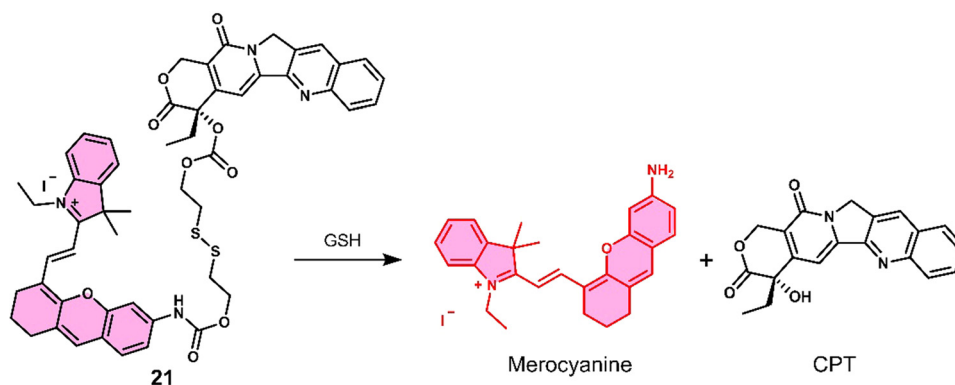


Fig. 10 Chemical structure of **21** and the GSH-mediated release mechanism of CPT and fluorescent merocyanine dye.

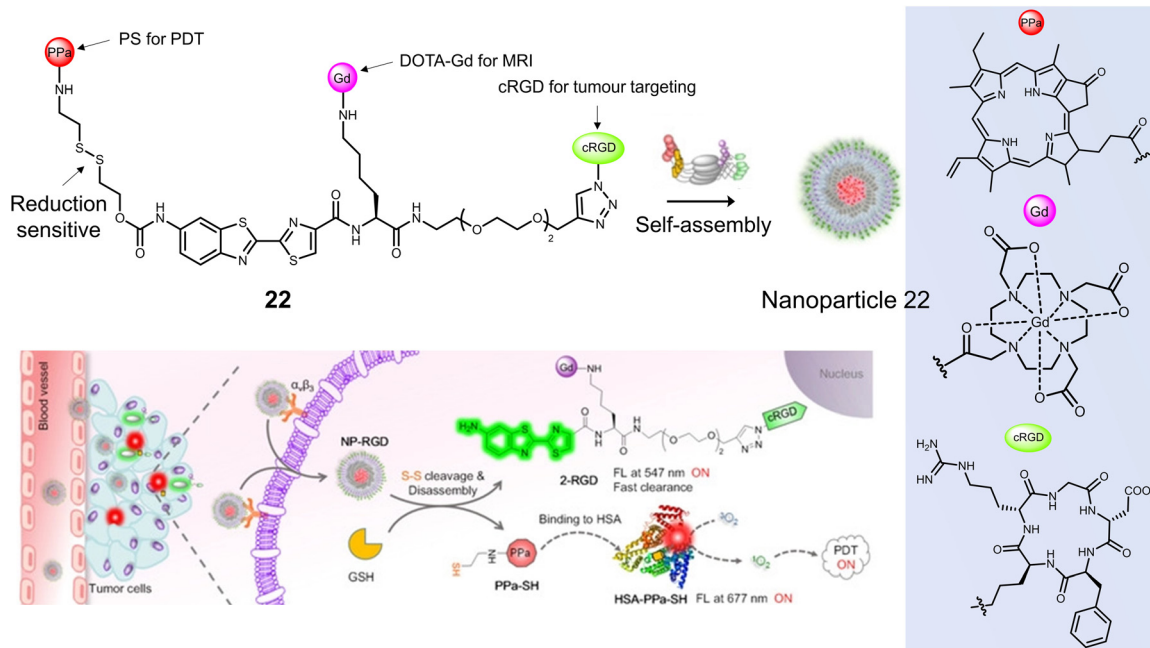


Fig. 11 The chemical structure of **22** and schematic illustration of **22** targeting tumour cells by binding $\alpha_v\beta_3$ integrin with subsequent fluorescence turn-on and light-mediated therapeutic effect on cancer cells after activation by GSH and binding to HSA. Reproduced with permission from ref. 102. Copyright (2020) Wiley-VCH Verlag GmbH.



DOTA-Gd complex, and cyclic peptide cRGD, all of which were linked *via* an amino oxyluciferin fluorophore. Owing to hydrophobic PPa and hydrophilic RGD, **22** was found to self-assemble in an aqueous solution to form nanoparticles with a hydrodynamic size of ~ 60 nm. This self-assembly resulted in high r_1 relaxivity ($20.0 \pm 1.7 \text{ mm}^{-1} \text{ s}^{-1}$), quenched fluorescence at 677 nm, and a low $^1\text{O}_2$ quantum yield. In the presence of GSH in a PBS-HSA mixture (PBS buffer containing 5% HSA), **22** was cleaved to 2-RGD and released photoactive PPa-SH. This structural transformation was confirmed by HPLC and matrix-assisted laser desorption/ionization mass spectroscopy (MALDI-MS). Hydrophobic PPa-SH was effectively bound to HSA, allowing it to disperse in water and exhibit a strong NIR fluorescence emission at 677 nm as well as enabling $^1\text{O}_2$ generation under laser irradiation (690 nm, 40 mW cm^{-2} , 5 min). Fluorescence imaging studies confirmed the cellular uptake of **22** nanoparticles in U-87 MG cells, followed by an increase in fluorescence emission at 690–750 nm when activated by intracellular GSH and HSA. An increase in T_1 -weighted MR images was observed after incubation of U-87 MG cells with **22** nanoparticles, which matched those observed by the fluorescence imaging. The light irradiation (690 nm, 40 mW cm^{-2} , 180 s) of U-87 MG cells incubated with **22** nanoparticles ($20 \mu\text{M}$), displayed a significant

photocytotoxicity ($<10\%$ cell viability); however, minimal cytotoxicity was observed for GES-1 cells under the same conditions. **22** nanoparticles using dual-modal fluorescence and MRI imaging exhibited good tumour localising ability in a U-87 MG tumour-bearing mice model. **22** nanoparticles ($200 \mu\text{L}$, $200 \mu\text{M}$) significantly reduced tumour volume compared to control groups (PBS) under laser irradiation (690 nm , 800 mW cm^{-2} , two consecutive exposures of 10 min each at an interval of 10 min). Good biosafety of **22** nanoparticles was confirmed since body weight losses were comparable to the PBS group.

Disulfide-based BODIPY scaffolds have also been used to construct promising PTT theranostic agents. Wang *et al.* reported the disulfide-based PTT agent **23** that uses a BODIPY fluorophore for both imaging and PTT. The amino unit on the phenyl group of the fluorophore was conjugated to PEG-based disulfide chain to afford good aqueous solubility and enable GSH detection (Fig. 12).¹⁰³ **23** formed stable nanoparticles in an aqueous solution *via* self-assembly and was weakly emissive due to the aggregation-caused quenching (ACQ) effect. The GSH-mediated cleavage of the disulfide bond afforded soluble products, leading to an enhancement in fluorescent intensity and an increase in the diameter of the nanoparticles from 89 nm to 350 nm. Incubation of **23** nanoparticles with HeLa

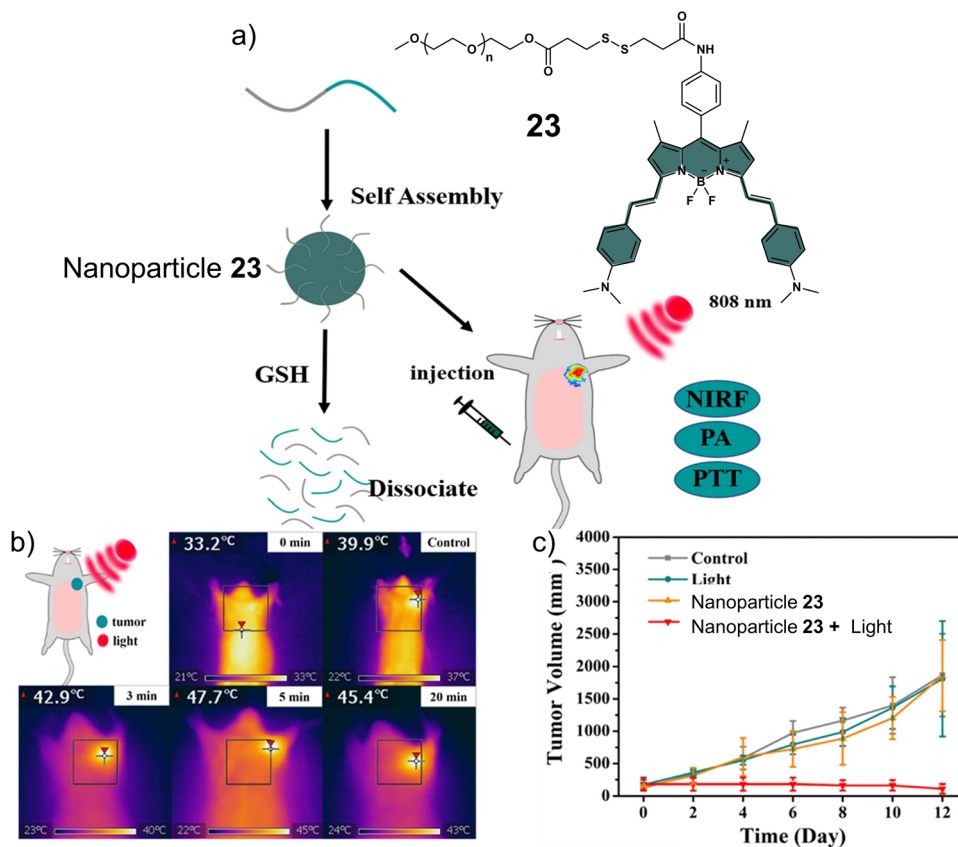


Fig. 12 (a) Chemical structure of **23** and the activation of its theranostic properties by GSH. (b) Infrared thermal imaging of the uterine cervical cancer (U14) tumour-bearing mice after treatment with **23** nanoparticles under 808 nm irradiation (0.5 W cm^{-2} , 5 min.). (c) The changing curve of real-time tumour volume of the uterine cervical cancer (U14) tumour-bearing mice 12 days after the treatment with **23** nanoparticles with or without 808 nm irradiation (0.5 W cm^{-2} , 5 min) or control. Reproduced with permission from ref. 103. Copyright (2018) Elsevier B.V.



cells resulted in a low fluorescence emission being initially observed. However, treatment with GSH resulted in a significant increase in fluorescence emission. The NIR signal of **23** nanoparticles was found to reach its maximum value at 36 h post-injection in uterine cervical cancer (U14) tumour-bearing mice. The maximum PA signal at the tumour region was identified at 12 h post-injection of **23** nanoparticles. **23** nanoparticles exhibited excellent photothermal conversion efficiency upon NIR laser irradiation (808 nm, 0.5 W cm^{-2} , 5 min), resulting in an increase of up to 47.7°C in temperature (Fig. 12b). This resulted in successful tumour ablation with over a 90% decrease in tumour volume compared to controls (light alone and **23** nanoparticles without light) (Fig. 12c). Excellent biosafety of the agent was confirmed using haematoxylin and eosin (H&E) staining.

Lucero *et al.* developed a GSH-responsive fluorescent and photoacoustic theranostic (**24**), and evaluated its properties in lung cancer A549 cells and an A549 tumour-bearing mice model.¹⁰⁴ In this study, the authors initially focused on tuning the chemical reactivity of the traditional GSH-mediated nucleophilic aromatic substitution ($\text{S}_{\text{N}}\text{Ar}$) reaction of 2,4-dinitrobenzenesulfonyl.¹⁰⁵ Since, it was believed that the high sensitivity of **24** towards GSH (micromolar) might lead to premature release *in vivo* and minimise selectivity between cancerous and healthy cells. Through extensive synthesis and evaluation, a 2-fluoro-4-nitrobenzenesulfonyl group was identified as an ideal candidate because its millimolar reactivity towards GSH was believed to provide appropriate selectivity *in vivo*. **24** was constructed using 2-fluoro-4-nitrobenzenesulfonyl, a self-immolative hemicyanine chemical probe (HD- CH_2OH), and the FDA-approved chemotherapeutic, gemcitabine (Fig. 13a). Gemcitabine is used for the

treatment of a wide range of cancers, including breast, testicular, ovarian, and non-small cell lung cancer. Its mode of action is intracellular metabolism by nucleoside kinases to active nucleotides gemcitabine diphosphate (dFdCDP) and triphosphate (dFdCTP), which lead to apoptosis by inhibiting the synthesis of DNA.¹⁰⁶ In accordance with design expectations, minimal fluorescence and photoacoustic signals were observed for **24** alone in solution (pH 7.4, with 70% PBS/MeCN). The addition of GSH (10 mM) to an aqueous solution of **24** (5 μM) was shown to afford HD- CH_2OH and result in the release of gemcitabine (confirmed by fluorescence, PA, MS, and nuclear magnetic resonance (NMR) analysis). For this system, a change in absorption from 570 nm to 680 nm and a significant increase in PA emission intensity at 690 nm was observed. **24** was then assessed *in vitro* using lung cancer A549, glioblastoma U87, and HEK 293 cells. **24** could readily distinguish GSH concentrations in lung cancer A549, glioblastoma U87, and HEK 293 using fluorescence microscopy, with the signal intensity correlating to antiproliferative activity; A549 cells displayed a greater fluorescence response and lower IC_{50} than U87 cells. A blind study was performed to demonstrate the importance of companion diagnostics,¹⁰⁷ **24** was also able to differentiate between A549 tumour-bearing mice and U87 tumour-bearing mice by the GSH-activatable PA signal with 95% accuracy (Fig. 13b). Moreover, **24** exhibited a significantly better anti-tumour effect in A549 tumour-bearing mice than in U87 tumour-bearing mice ascribed to differences in endogenous GSH concentrations between the mice models (Fig. 13c).

Considering that tumours can exist in a reductive and an oxidative state, Sharma *et al.* focused on identifying a suitable protecting group that could respond to either GSH or H_2O_2 .

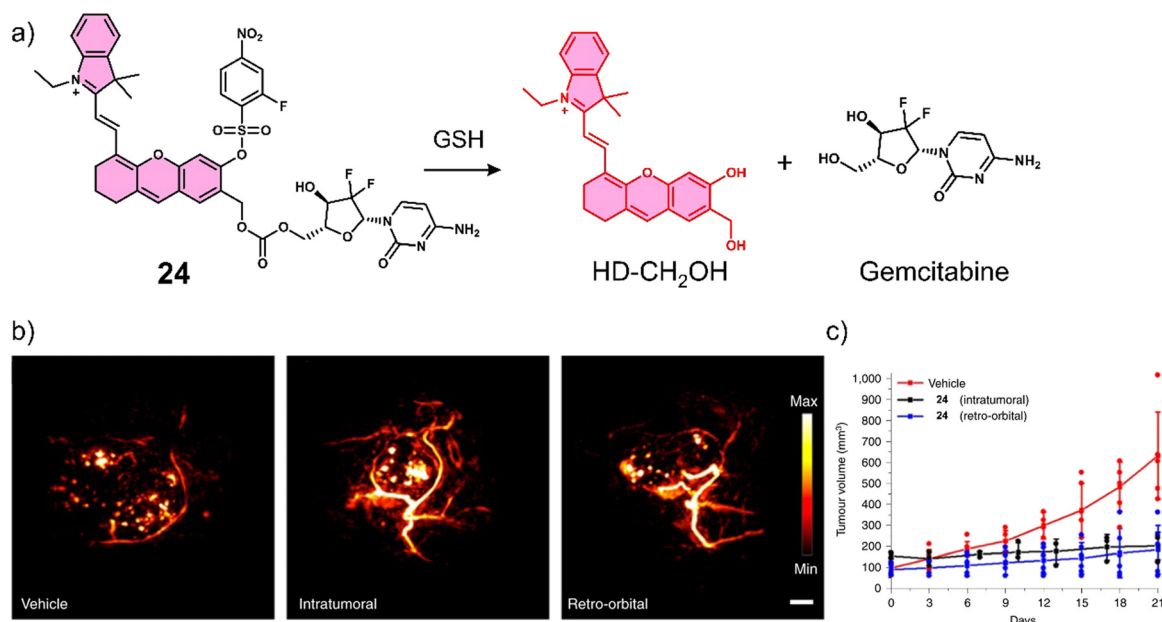


Fig. 13 (a) Chemical structure of **24** and the GSH-responsive release of Gemcitabine and HD- CH_2OH . (b) PA images of tumours after treatment with vehicle (10% DMSO/PBS), intratumoural injection of **24** (100 μM , 10% DMSO/PBS), and retro-orbital injection of PARx (400 μM , 10% DMSO/PBS). Samples were irradiated at 680 nm. Scale bar: 2 mm. (c) Average tumour volume after treatment with vehicle ($n = 6$ independent animals), intratumoural injection of **24** ($n = 3$ independent animals) and retro-orbital injection ($n = 6$ independent animals) of **24** (100 μM , 50 μL) once every 7 days for 21 days. Reproduced with permission from ref. 104. Copyright (2021) The Author (M. Y. Lucero and J. Chan). Published by Springer Nature.



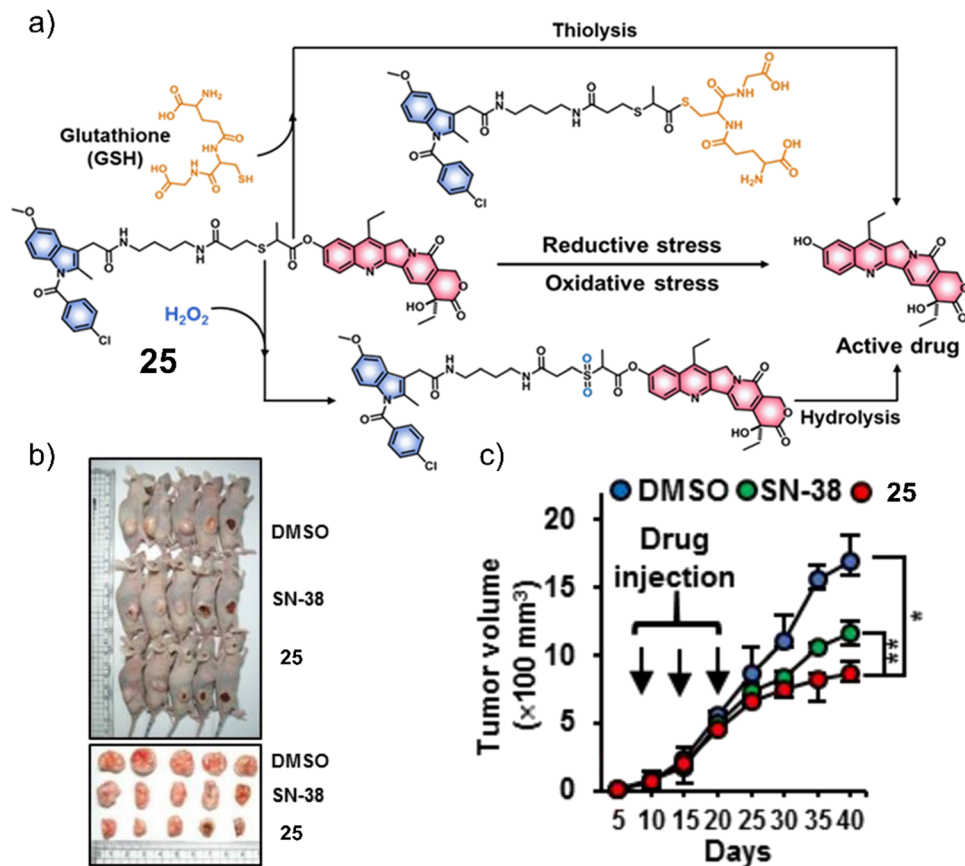


Fig. 14 (a) Chemical structure of **25** and the products of the GSH or H_2O_2 -mediated release of SN-38. (b) Representative images of SW620 tumour-bearing mice treated with control (DMSO), SN-38, and **25**. (c) Tumour growth inhibition of SW620 tumour-bearing mice over 40 days with DMSO, SN-38 ($5 \text{ mg kg}^{-1} \text{ d}^{-1}$), and **25** ($5 \text{ mg kg}^{-1} \text{ d}^{-1}$) once every five days for three weeks. $n = 5$, mean \pm SD, * $P < 0.05$, ** $P < 0.01$. Reproduced with permission from ref. 31. Copyright (2019) American Chemical Society.

As can be seen in Fig. 14a, a thioether linker was used to conjugate cancer-targeting cyclooxygenase-2 (COX-2) inhibitor, indomethacin with SN-38 to afford a theranostic **25**.³¹ The inherent fluorescent properties of SN-38 were used to determine drug release in solution-based studies, in which the thioether linker was identified to undergo thiol-mediated hydrolysis (thiolysis) by GSH to release SN-38. Moreover, H_2O_2 was shown to oxidise the thioether linker to its sulfone analogue and undergo hydrolysis to release SN-38. SN-38 release by either GSH or H_2O_2 was studied by fluorescence spectroscopy with increased fluorescence emission intensity at 540 nm. *In vitro* analysis found greater antiproliferative activities in cell lines that overexpressed COX-2 (LoVo and SW620) than in those with a lower COX-2 expression (NHDFs, normal human dermal fibroblasts; MCF10A, human breast epithelial cell lines), which illustrated the importance of the indomethacin targeting unit. Moreover, the treatment of LoVo cells with either H_2O_2 or *N*-acetylcysteine (a GSH precursor) was shown to induce the intracellular release of SN-38 *via* monitoring changes in fluorescence emission intensity. Subsequent animal-based studies were performed using a colon cancer (SW620) tumour-bearing mice model *via* intraperitoneal injection (**25** ($5 \text{ mg kg}^{-1} \text{ d}^{-1}$), SN-38 ($5 \text{ mg kg}^{-1} \text{ d}^{-1}$), or vehicle (DMSO)) once every five days

for three weeks. The tumour volume of **25**-treated mice decreased significantly relative to what was seen for the controls about 17 mm^3 and 12 mm^3 (DMSO and SN-38, respectively) (Fig. 14b and c). Moreover, using immunoblotting assays, **25** was shown to significantly inhibit the expression of pro-inflammatory factors (*e.g.*, TNF- α , IL-6, and VEGF). These results were ascribed to the anti-inflammatory properties of indomethacin. Minimal toxicity was observed for **25** with normal blood parameters (*e.g.*, AST, ALT, and serum creatine); this indicates a good safety profile using this strategy. We anticipate this approach will inspire others to develop other multi-biomarker responsive prodrugs or even exploit the GSH or H_2O_2 -responsive thioether linker for use in other applications.

A theranostic system co-activated by H_2O_2 and GSH is also used in the PDT treatment of tumour cells. Wang and co-workers reported the mitochondria-targeted PDT agent, **26**, that was shown to be activated by H_2O_2 and GSH sequentially (Fig. 15a).¹⁰⁸ **26** consisted of the PS benzothiadiazole (BTB) functionalised with a TPP mitochondrial targeting unit and a boronic ester protecting group that was covalently attached to a thiol unit.^{109,110} **26** initially exhibited a low fluorescence emission at 510 nm in PBS buffer (pH 7.4), and the sequential addition of H_2O_2 and GSH was shown to afford TPP-NH₂ArSO₃H, a process initiated by



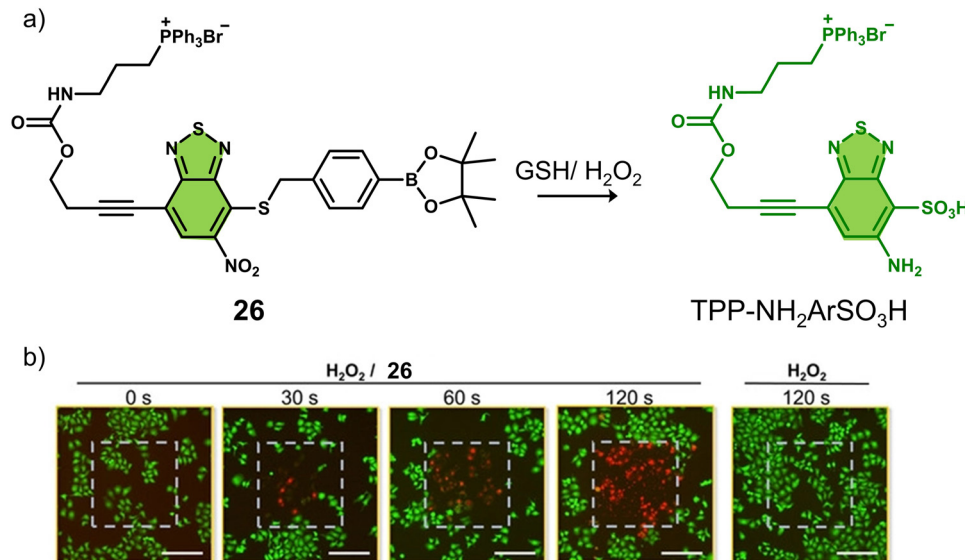


Fig. 15 (a) The chemical structure of **26** and the H₂O₂/GSH-activatable TPP-NH₂ArSO₃H release mechanism. (b) Live/dead staining of HeLa cells that were pretreated with H₂O₂ (50 μM) followed by treated with **26** (0 or 10 μM). The cells were irradiated by a two-photon laser ($\lambda = 800$ nm) for 0, 30, 60, and 120 s. The laser irradiation area (400 μm × 400 μm) in each image was labeled with a gray dashed square. Scale bars: 200 μm. Reproduced with permission from ref. 108. Copyright (2020) Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim.

H₂O₂-mediated removal of the boronic ester and subsequently completed by a series of redox reactions mediated by H₂O₂ and GSH. This sequential activation resulted in a concomitant increase in fluorescence intensity at 510 nm and increased ROS production (both ¹O₂ and O₂^{•-}). TPP-NH₂ArSO₃H exhibits a high two-photon (TP) absorption cross-section ($\delta = 41 \pm 4$ GM in methanol) at 800 nm, allowing its NIR TP activation for greater tissue penetration. Fluorescence microscopic evaluation confirmed the mitochondrial localisation of **26** in HeLa cells. The PDT efficacy of **26** was evaluated in HeLa cells, pretreated with H₂O₂ (50 μM). Significant cell death was observed after irradiating the cells with a two-photon laser at 800 nm (120 s) (Fig. 15b). Remarkably, TPP-NH₂ArSO₃H exhibited mixed type 1 and type 2 mechanisms; therefore, minimal changes to its phototoxicity were observed under hypoxic conditions. **26** represents the first H₂O₂/GSH sequentially activated mitochondria-targeted PDT agent with mixed type I and type II mechanisms.

Nearly 50% of cancer patients with solid tumours will receive platinum (Pt)-based therapy.¹¹¹ The Pt(II)-based therapeutics that have been approved to treat cancers include cisplatin, carboplatin, nedaplatin, oxaliplatin, and lobaplatin. Unfortunately, their poor tumour specificity leads to dose-limiting toxicities and a narrow therapeutic index, severely limiting their clinical efficacy. In recent years, researchers have focused on overcoming these limitations by developing Pt(IV)-based prodrugs. The key advantage of using Pt(IV)-based prodrugs is their relative non-toxic nature until reduced intracellularly to their cytotoxic Pt(II) form (see Fig. 16). In addition, the ease of modification of the axial ligands allows the ability to tailor the pharmacological properties of these molecules. For an extensive overview on platinum-based therapies, the reader can turn to an excellent review by Lippard and co-workers.¹¹²

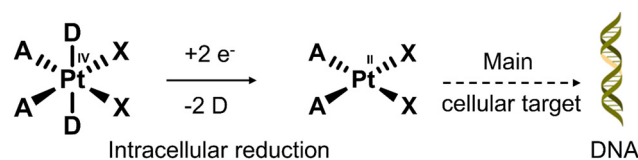


Fig. 16 The conversion mechanism of the prodrug Pt(IV) complex to the Pt(II) complex under a reducing environment. D = axial ligands that modulate the lipophilicity and the optimal redox properties; X = leaving groups that control the rate of aquation and stability in the biological medium; A = non-leaving groups that are closely related to the antiproliferative activity.

In 1988, the Sessler group introduced the world to “texaphyrins” which are penta-aza Schiff base expanded porphyrins, that form stable 1:1 complexes with an extensive array of functional metal cations.^{52,113} Notably, the paramagnetic gadolinium(III) metallotexaphyrin (MGd) was found to be more reducible than porphyrins, endowing its use as a redox mediator. Furthermore, in recent studies, the ability of MGd to catalyse the intracellular reduction of oxaliplatin-based Pt(IV) prodrugs to cytotoxic Pt(II) has been confirmed, *i.e.*, oxaliplatin.¹¹⁴ This led to the construction of a series of Pt(IV)-based texaphyrin conjugates, namely **27**, **28**, **29**.¹¹⁵ It was rationalised that the tumour-localising properties of MGd would enable the effective delivery of Pt(IV) and selective reduction of Pt(IV) to Pt(II) in solid tumours. These candidates consisted of an MGd, which is conjugated to oxaliplatin Pt(IV)-derivatives *via* a succinate linker (Fig. 17a). Each compound differed by the axial ligand (L = Cl, **27**; OAc, **28**; OH, **29**), which plays a key role in the rate of reduction and hydrolysis of the complexes. **28** was identified as the most stable candidate, and the reduction of Pt(IV) was shown by the reverse-phase (RP)-HPLC. *In vitro* analysis indicated that **28** exhibited a broad



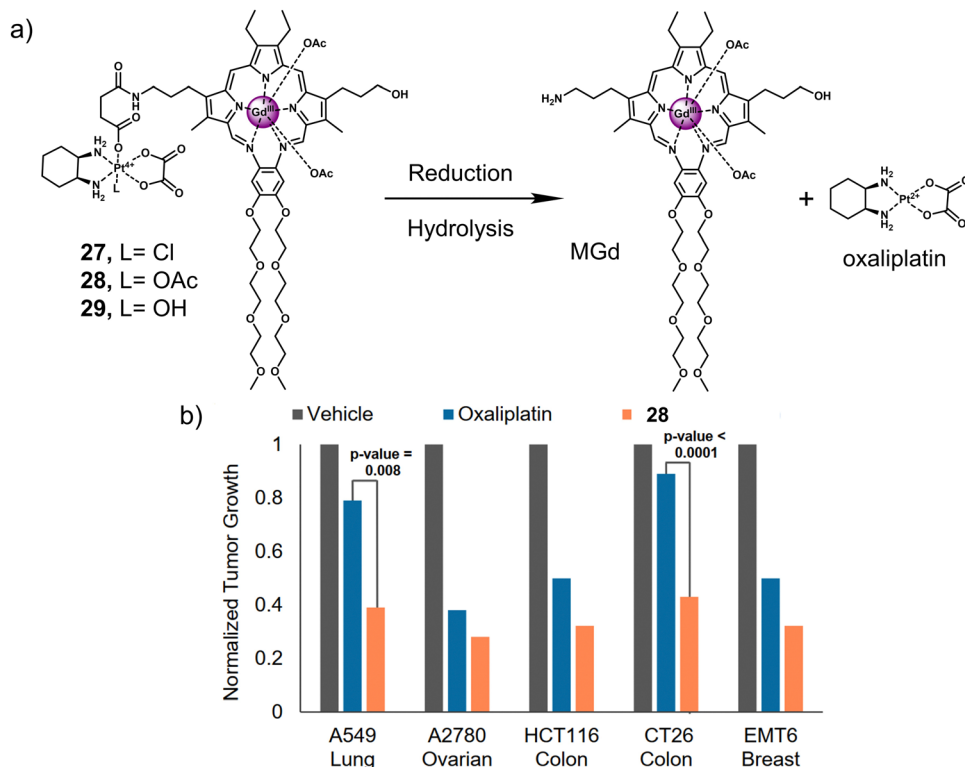


Fig. 17 (a) Chemical structure of **27**, **28**, **29** and the release of therapeutic oxaliplatin and MGd under the sequential action of reduction and hydrolysis. **27**, L = Cl; **28**, L = OAc; **29**, L = OH. (b) *In vivo* efficacy of **28** (70 mg kg⁻¹ per dose on days 1, 5, 9, 13) vs. oxaliplatin (4 mg kg⁻¹ per dose on days 1, 5, 9, 13) in A549, A2780 and HCT116 tumour-bearing mice models and syngeneic tumour (CT26, EMT6) mice models. The study endpoint for the A549 model was 30 d. The study endpoint for all other models was the day at which the vehicle-treated mice reached maximum tumour burden. Reproduced with permission from ref. 115. Copyright (2020) Published by National Academy of Sciences.

antiproliferative activity compared to oxaliplatin across several cell lines, including platinum-resistant ones. This broad anticancer effect translated to A549, A2780 (ovarian cancer), and HCT116 (human colon cancer cells) tumour-bearing mice and syngeneic tumour (CT26, murine colorectal carcinoma cells; EMT6, murine mammary carcinoma cells) mice models. **28** (70 mg kg⁻¹ per dose on days 1, 5, 9, 13) *via* tail vein injection of mice with patient-derived xenografts (PDXs) was shown to reduce the tumour volume significantly when compared to vehicle and oxaliplatin (4 mg kg⁻¹ per dose on days 1, 5, 9, 13) (Fig. 17b). Remarkably, these results confirmed that **28** is more effective in retarding and inhibiting tumour growth than the most closely related platinum-based treatments currently available. Which clearly illustrates the promise of Pt(IV)-based prodrugs for potential clinical translation.

Another two gadolinium-based theranostics, **30** and **31**, were reported for tandem MR imaging of Pt(IV) reduction in A2780 and HeLa cells (Fig. 18).¹¹⁶ The Gd(III)-DOTA MR contrast agent was conjugated to Pt(IV) cisplatin and carboplatin derivatives by the axial ligand to afford **30** and **31**, respectively. The relaxivity (r_1 and r_2) of both **30** and **31** was greater than that of the Gd(III) MR contrast agent, Gd-1, due to an increase in rotation-related time (τ_R) and change in the inner sphere hydration number (q) of Gd(III). Excellent stability was observed in the absence of

reductants. However, the relaxivity (r_1) of both **30** and **31** decreased significantly and converged with that of the parent scaffold Gd(III) MR contrast agent Gd-1 when GSH (5 mM) was added. This resulted in the conversion of both **30** and **31** to carboxylic-derived monomeric Gd(III) complexes Gd-1 and the active Pt(II) analogues, confirmed by HPLC analysis. This apparent change in r_1 before and after the reduction of **30** and **31** enabled the monitoring of intracellular reduction of the Pt(IV) derivatives. Higher IC₅₀ values were observed for **30** and **31** compared to cisplatin and carboplatin, respectively. This was ascribed to poor cell permeability. However, it is believed that higher concentrations of **30** and **31** can be used to address the lower cytotoxicity. In addition, the cell impermeable Gd(III) complex Gd-1 released by intracellular degradation of **30** and **31** can be used as a contrast agent for MR imaging. An increased r_1 was observed in **30**-treated A2780 cells and in HeLa cells compared to the untreated control group. This example demonstrates the ability to use MRI to visualise and monitor the release of Pt(IV) derivatives in cells.

3.3 Hypoxia-responsive prodrugs and phototherapeutics

Hypoxia is a common feature of all solid tumours, which arises from an inadequate blood supply within tumour tissue affording low oxygen environments. This observation is exacerbated with



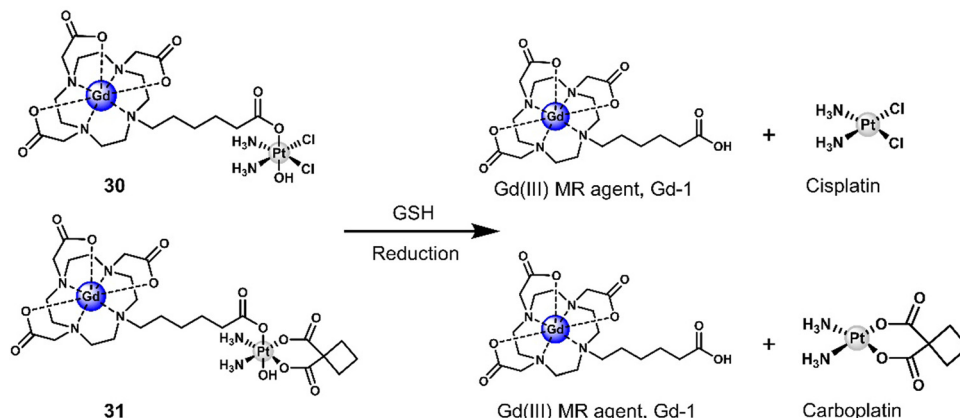
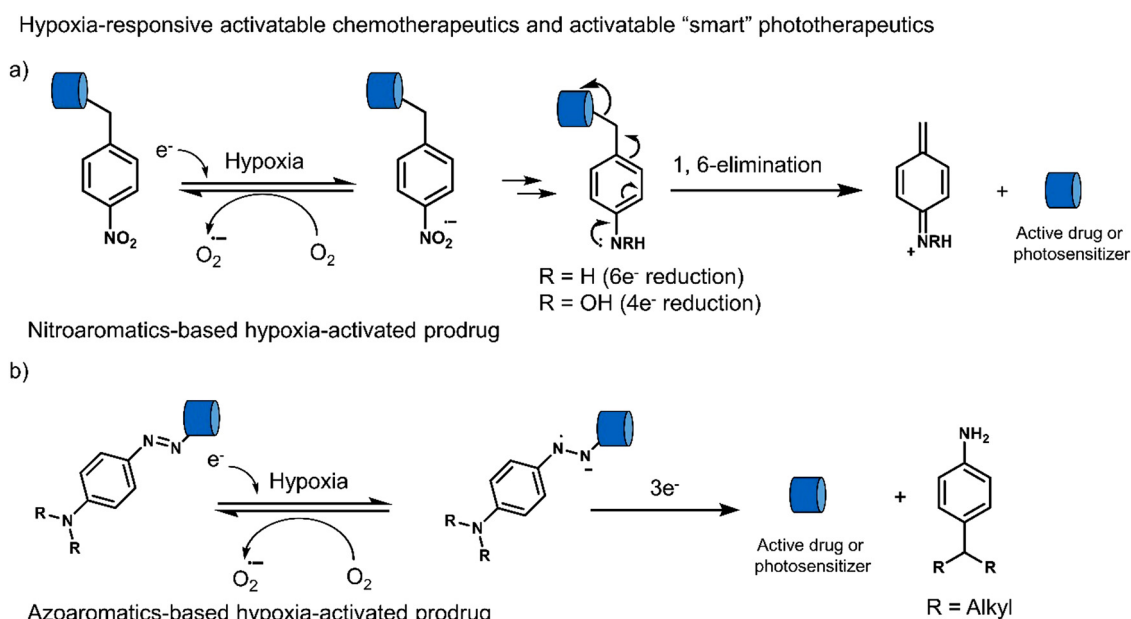


Fig. 18 Chemical structure of **30**, **31**, and the GSH-responsive release of therapeutic Pt(II) drug (cisplatin and carboplatin) and a Gd(III) MR contrast agent, Gd-1. DOTA: 1,4,7,10-tetraazacyclododecane-*N,N',N'',N'''*-tetraacetic acid.



Scheme 4 Basic mechanism of nitroaromatic-based and azoaromatic-based prodrugs activation.

tumour growth as oxygen supply is unable to meet the metabolic demands of tumour cells, thus causing the intratumoural micro-environment to exhibit a significant hypoxic phenotype.^{117,118} It is well-established that the degree of hypoxia is correlated with the local concentration of reducing enzymes, such as nitroreductase (NTR) and azoreductase. These enzymes are known for reducing nitro-aromatic and azoaromatic-derivatives, respectively. Researchers have therefore exploited nitroaromatics and azoaromatic units as hypoxia-based protecting groups to develop prodrugs that are activated in the hypoxic environment of tumours (Scheme 4) (Table 3).

A particularly illustrative example is the chemotherapeutic **32** developed by Liu *et al.*,¹¹⁹ which contains a hypoxia-responsive 4-nitrobenzyl protecting group, 4'-(diethylamino)-1,1'-biphenyl-2-carboxylate linker (and fluorescent reporter), and the chemotherapeutic 5-fluorodeoxyuridine (FDU). FDU is

commonly used to treat colorectal cancer as well as kidney and stomach cancers through the inhibition of thymidylate synthase, and ultimately the synthesis of the DNA. As shown in Fig. 19a, the nitroaromatic of **32** was reduced by the nitroreductase to release FDU and 4'-(diethylamino)-1,1'-biphenyl-2-carboxylate, which isomerises to afford fluorescent 7-(diethylamino)coumarin (CM). The reaction progress was confirmed *via* an increased fluorescence emission at 530 nm and HPLC studies. *In vitro* studies of **32** were carried out in MGC-803 (human gastric carcinoma) and MCF-7 (human breast cancer cell line) cancer cell lines under normoxic and hypoxic conditions, in which the antiproliferative effect of **32** was found to be higher in both cell lines under hypoxic conditions. Compared to FDU, the prodrug **32** exhibited minimal cytotoxicity to a normal cell line (BRL-3A, a rat liver-derived cell line), suggesting the potential to overcome the known off-target toxicities of FDU. The treatment of MCF-7



Table 3 Summary of hypoxia-responsive prodrugs and phototherapeutics discussed in this review

	Stimuli	Active anti-cancer therapeutic agent	Treatment	<i>In vitro</i> models	<i>Ex vivo/in vivo</i> models	Ref.
32	Hypoxia	FDU	Chemotherapy	MGC-803 and MCF-7 cells	MCF-7 tumour-bearing mice	119
33	Hypoxia	Iodinated heptamethine cyanine	PTT	HeLa, HepG2, and A549 cells	HeLa tumour-bearing mice	120
34	Hypoxia	3,4-Difluorobenzyl curcumin agent	Chemotherapy	CD133 ⁺ MDA-MB-231 cells and tumour spheroids	MDA-MB-231 tumour-bearing mice	122
36	Hypoxia	Panobinostat	Chemotherapy	OE21 and HCT116 cells	OE21 tumour-bearing mice	124
37	Hypoxia	IR-1048	PTT	—	A549 tumour-bearing mice	129
38	Hypoxia	SN-38	Chemotherapy	A549, HeLa, HepG2, MCF-7, and MDA-MB-231 cells	4T1 tumour-bearing mice	130
39	Hypoxia	Seleno-rosamine scaffold	PDT	A549 cells	—	131
40	Hypoxia	PPA	PDT	BEL-7402 cells	—	132

Note: FDU: 5-fluorodeoxyuridine, PPA: pyropheophorbide α , SN-38: the active metabolite of camptothecin, PDT: photodynamic therapy, PTT: photothermal therapy, —: not mentioned.

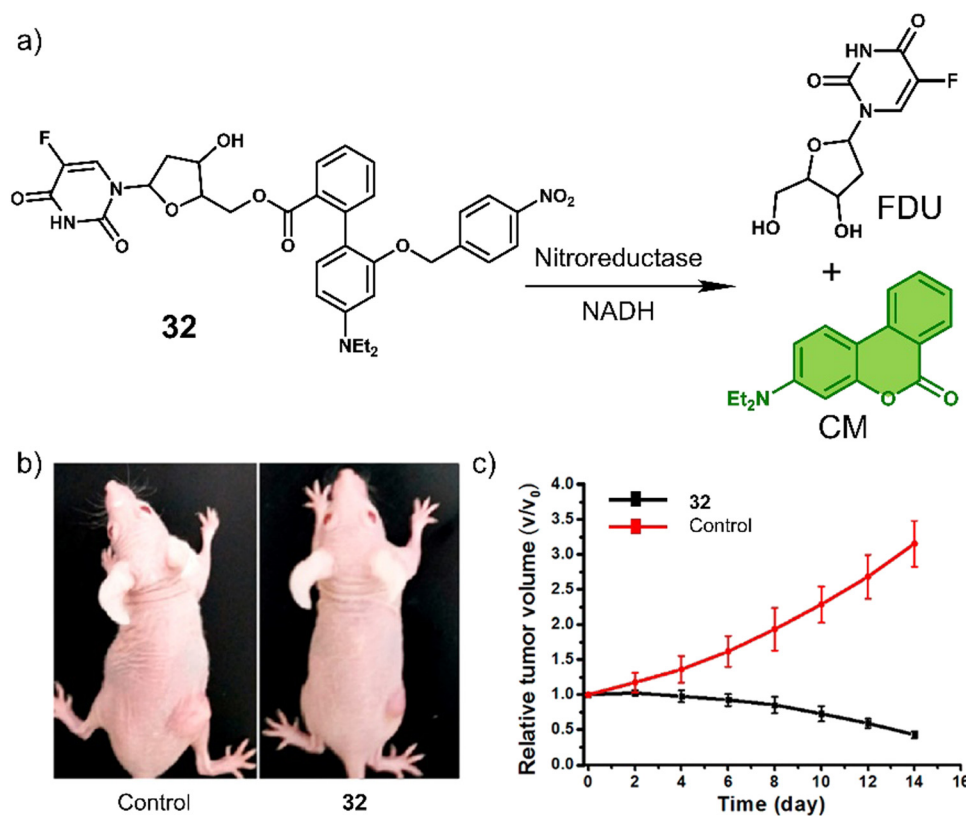


Fig. 19 (a) Chemical structure of **32** and the hypoxia-induced release of FDU and 7-(diethylamino)coumarin (CM). (b) Representative images of MCF-7 tumour-bearing mice after treatment with the control group (saline) and **32**. (c) Tumour growth inhibition over the time with the control group (saline) and **32** (10 mg kg⁻¹). Reproduced with permission from ref. 119. Copyright (2018) American Chemical Society.

tumour-bearing mice with **32** (10 mg kg⁻¹ once every four days for 12 days) was shown to reduce the tumour size compared to controls (Fig. 19b and c). The tumour reduction volume induced by **32** was 86% greater than that of the control group (saline).

In another study, the nitrobenzyl hypoxia release trigger was used in the construction of an activatable phototherapeutic. Peng and co-workers developed a hypoxia-activatable PTT agent, **33** (Fig. 20a), for the PTT treatment of HeLa tumour-bearing mice models.¹²⁰ An interesting aspect to the design of this system is that it can convert from a PDT agent to a PTT agent under hypoxic conditions, thus overcoming the oxygen

requirement of PDT. **33** exploited the hypoxia-responsive 4-nitrobenzyl group and an iodinated heptamethine cyanine dye as the PS. In the presence of NADH and NTR, the nitro-aromatic unit was reduced, followed by the release of Icy-NH₂, which was confirmed *via* fluorescence and HRMS studies. Both **33** and Icy-NH₂ exhibited high molar extinction coefficients ($2.35 \times 10^5 \text{ M}^{-1} \text{ cm}^{-1}$ and $2.21 \times 10^5 \text{ M}^{-1} \text{ cm}^{-1}$, respectively) at 837 and 835 nm, respectively. Initially, **33** displayed a high ¹O₂ quantum yield and high fluorescence emission at 837 nm. However, upon being converted to Icy-NH₂, the ¹O₂ quantum yield and fluorescence emission decreased significantly, and



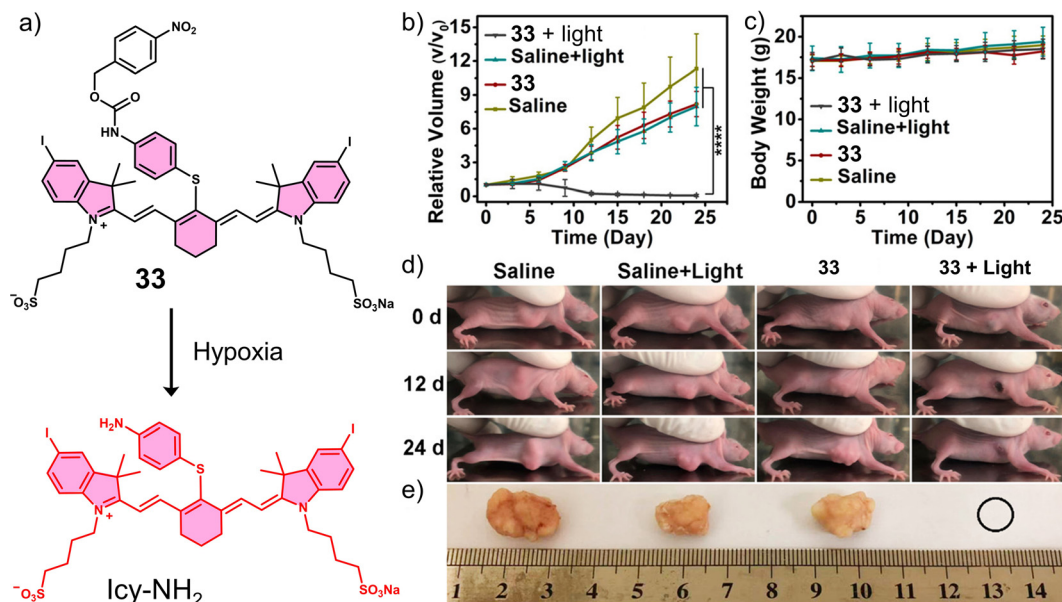


Fig. 20 (a) Chemical structure of **33** and its reduction mechanism under hypoxic environment to afford Icy-NH₂. (b) Relative tumour volume and (c) Body weights of HeLa tumour-bearing mice after the treatment of saline or **33** (15 nM, 100 μ L) with or without 808 nm irradiation (252 J cm⁻², 7 min) once. *****P* < 0.0001. (d) Images of HeLa tumour-bearing mice of (b) at 0, 12, and 24 days post-treatment. (e) Images of tumours harvested from (d) at 24 days post-treatment. Reproduced with permission from ref. 120. Copyright (2020) American Chemical Society.

the non-radiative relaxation pathway was enhanced for efficient heat production. Therefore, as the surrounding oxygen environment changed, the therapeutic function of **33** was switched from PDT to PTT under 808 nm light irradiation. **33** exhibited high cytotoxicity upon 808 nm light (240 J cm⁻²) under normoxic conditions *via* PDT and under hypoxic conditions *via* PTT. *In vivo* experiments using HeLa tumour-bearing mice models revealed the fluorescence intensity of **33** was inversely proportional to the level of NTR activity and temperature. The 808 nm light irradiation (0.6 W cm⁻², 7 min) of **33** was shown to increase the temperature of the tumour to 60 °C, which reduced the tumour size significantly compared to controls (Fig. 20b-d). This work demonstrates a unique strategy that exploits the advantageous properties of both PDT and PTT and serves as a new approach for the development of intelligent phototherapeutic agents that maximise photon efficiency.

Cancer stem cells (CSCs) contribute to tumour regeneration, metastasis, and recurrence. CSCs have therefore been a focal point of cancer research over the last decade, with several reports on their detection and treatment.¹²¹ Breast CSCs have been associated with tumourigenesis, chemoresistance, and poor patient prognosis. For these reasons, the hypoxia-responsive prodrug **34** was developed for the treatment of breast cancer stem cells (CSCs) (Fig. 21a).¹²² **34** was functionalised with the carbonic anhydrase IX (CAIX) inhibitor acetazolamide (Az), the hypoxia-responsive dimethylnitrothiophene unit, and the experimental anti-CSC drug 3,4-difluorobenzyl curcumin (CDF). Dimethylnitrothiophene has a lower reduction potential than the more widely used nitrobenzyl hypoxia release trigger. Furthermore, the presence of dimethyl results in favourable reaction kinetics. Additionally, by replacing the CDF group with a naphthalimide

fluorophore, the corresponding fluorescent analogue **35** was constructed, which could release the fluorophore in a similar manner to **34** for the targeted imaging of CSCs. **Az** was used as a targeting unit for CSCs (MDA-MB-231, human breast cancer cell line) and an inhibitor for CSC proliferation.¹²³ In solution studies, UV/vis spectroscopy and HPLC analysis, CDF was shown to be released from **35** in the presence of *E. coli* NTR and NADH. Under the same conditions, the fluorescence intensity of **35** increased at 550 nm using fluorescence microscopy. The fluorescence of **35** could be activated in CD133⁺ (a biomarker of cancer stemness) MDA-MB-231 cells but not in BJ cells (CAIX negative) under mild hypoxic conditions (3% O₂), and the fluorescence was suppressed in CAIX knock-down CD133⁺ MDA-MB-231 cells or under normoxic conditions. The dimethylnitrothiophene-based hypoxia-sensitive group was readily activated under 3% O₂, with significant cytotoxicity observed for CD133⁺ MDA-MB-231 cells under hypoxic conditions. Also, similar cytotoxicity was observed using a tumour spheroid *in vitro* assay (CD133⁺ MDA-MB-231 tumour spheroids). Single tail-vein injection of **35** in MDA-MB-231 tumour-bearing mice was shown *via* fluorescence imaging (Maestro™ *In Vivo* Fluorescence Imaging System) to predominantly accumulate at the tumour site over 144 hours. Besides, the tail vein administration of **34** (once a week for 3 weeks at a dose of 50.0 nM in 100 μ L of PBS) resulted in a significant reduction in tumour volume compared to control groups (R-Az, without drug CDF; R-CDF, without targeting group Az) (Fig. 21b). Confirming the effectiveness of **34** to target CSCs, CD133⁺ MDA-MB-231 cells, and nude mice injected with **34**-treated CD133⁺ MDA-MB-231 cells under hypoxic conditions exhibited suppressed tumourigenesis. However, for mice injected with DMSO-treated cells rapid tumour growth was observed, which demonstrated that **34** could be used to overcome tumourigenesis associated with CSCs.



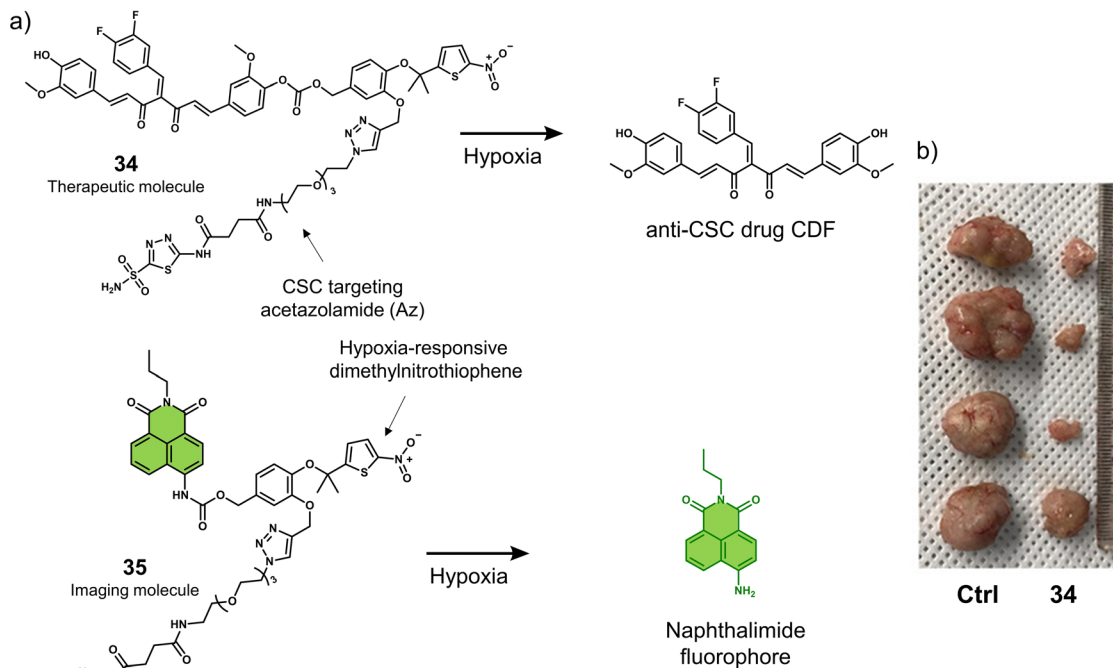


Fig. 21 (a) Chemical structures of **34** and **35** and the reaction mechanism under hypoxic conditions. (b) Representative images of excised tumours from MDA-MB-231 tumour-bearing mice treated with control (DMSO), and **34** (50.0 nM in 100 μ L of PBS) once per week for 3 weeks. Reproduced with permission from ref. 122. Copyright (2021) American Chemical Society.

Nitroimidazole groups have also been incorporated into activatable chemotherapeutics. Skwarska *et al.* functionalised the lysine deacetylase (KDAC) inhibitor panobinostat with 1-methyl-2-nitroimidazole-unit to develop a hypoxia-responsive prodrug, **36** (Fig. 22a).¹²⁴ KDACs are enzymes that promote tumour malignancy by deacetylating lysine residues found on histones or non-histones. Overexpressed class I KDACs are

found in various cancer types (*e.g.*, lung, breast, pancreatic, and gastric cancer) and are associated with poor prognosis.^{125–127} Several KDAC inhibitors have been approved for the treatment of hematological malignancies, such as panobinostat.¹²⁸ **36** was found stable under normoxic conditions and displayed little inhibitory activity against KDAC. LC-MS analysis revealed the release of panobinostat when **36** was exposed to NADPH-CYP reductase (CYP004) under hypoxic conditions ($<0.1\%$ O_2). **36** exhibited a higher antiproliferative effect in OE21 and HCT116 cancer cell lines under hypoxic conditions ($<0.1\%$ O_2) than under normoxic (21% O_2) conditions (Fig. 22b). In addition it exhibited excellent growth inhibition of HCT116 tumour spheroids. The treatment of OE21 tumour-bearing mice with **36** (50 mg kg^{-1} , three doses on days 1, 3 and 5) was shown to reduce the tumour size compared to the control (vehicle) and significantly increased the survival (range 8–28 days) of the mice evaluated.

Meng *et al.* reported a hypoxia-responsive PTT agent, **37**, that contains a nitroimidazolyl scaffold, for the near-infrared II (NIR-II) fluorescence/PA imaging and PTT treatment of A549 tumour-bearing mice models (Fig. 23).¹²⁹ **37** was synthesised through a simple one-step conjugation reaction between 2-(2-nitroimidazolyl)ethylamine and commercially available NIR-II dye, IR-1048. It displayed minimal NIR-II fluorescence and had an absorption in the range of 900–1060 nm. Upon activation by nitroreductase (NTR) and NADH, **37** produced a NIR-II fluorescence emission at 1046 nm, an increase in photoacoustic intensity, and a high photothermal conversion efficiency (20.2%). *In vivo* studies using an A549 tumour-bearing mice model

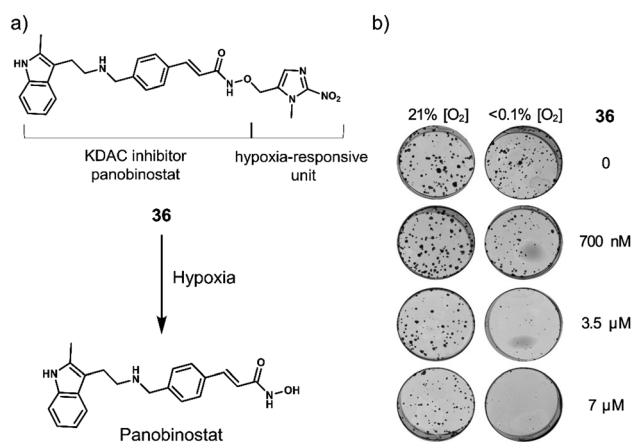


Fig. 22 (a) Chemical structure of **36** and the hypoxia-mediated release of the KDAC inhibitor panobinostat. (b) Images of OE21 cells treated with **36** (0–7 μ M) under hypoxic ($<0.1\%$ O_2) and normoxic (21% O_2) conditions for 24 hours, after which **36** was removed, and the cells were allowed to form colonies for 7 days under normoxic conditions. Reproduced with permission from ref. 124. Copyright (2021) Elsevier Inc.



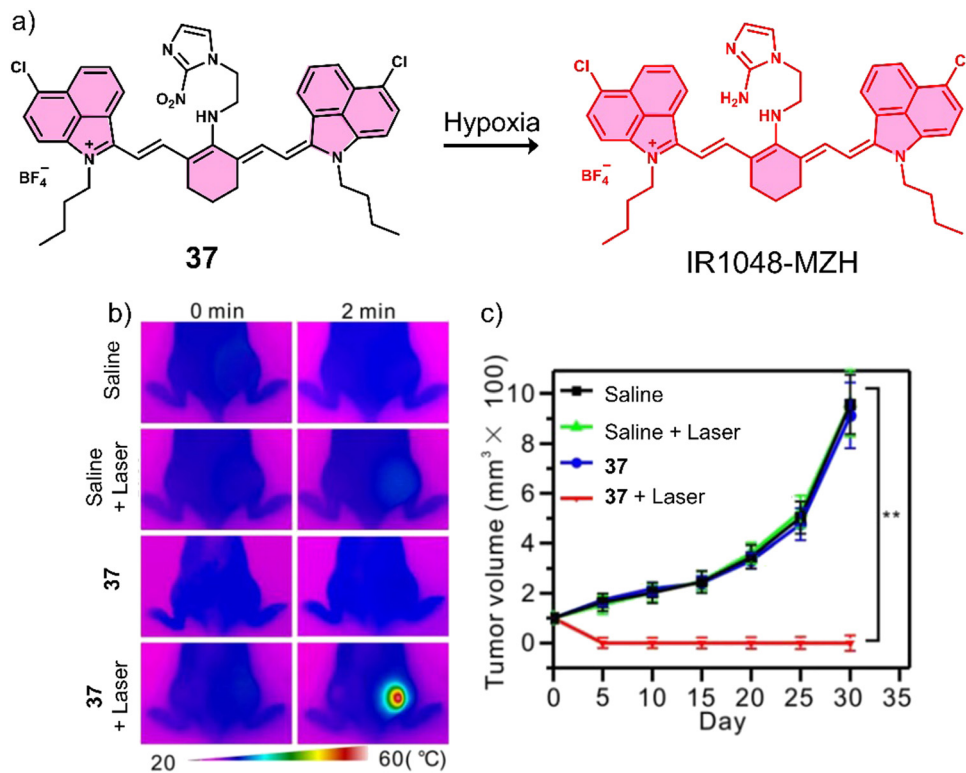


Fig. 23 (a) Chemical structure of **37** and the phototheranostic properties under hypoxic conditions in tumour cells. (b) Infrared thermal imaging of the A549 tumour-bearing mice injected with saline or **37** with or without 980 nm NIR laser irradiation (0.1 W cm^{-2} , 2 min). (c) The A549 tumour growth curves over 30 days after treatment with saline or **37** ($200 \mu\text{L}$, $40 \mu\text{g mL}^{-1}$) with or without NIR laser irradiation (980 nm, 0.1 W cm^{-2} , 2 min). $n = 5$, $**P < 0.01$. Reproduced with permission from ref. 129. Copyright (2018) Published by Ivyspring International Publisher.

revealed strong NIR-II fluorescence emission and PA signal at the tumour site. This was indicative of hypoxia-mediated activation of IR1048-MZ. Remarkably, a rapid temperature increase from $30 \text{ }^{\circ}\text{C}$ to $58 \text{ }^{\circ}\text{C}$ and complete tumour ablation was observed with no tumour recurrence being seen after 30 days.

Motivated to exploit the hypoxic environment of solid tumours, Zhou *et al.* developed the theranostic **38** that employs an aryl-azo benzyl alcohol motif to mask both the pharmacological activity and fluorescence emission of SN-38 (Fig. 24a).¹³⁰ Under hypoxic conditions, the addition of rat liver microsomes ($0\text{--}240 \mu\text{g mL}^{-1}$) to an aqueous solution of **38** (0.1 mM NADH as a coenzyme) resulted in an increase in fluorescence intensity at 560 nm , which was ascribed to the release of SN-38 (confirmed by MS-spectroscopic analysis). Fluorescence microscopy was used to visualise the release of SN-38 in HeLa cells under glutathione ethyl ester-induced hypoxic conditions and in living *C. elegans*. **38** exhibited selective toxicity for several cancer cell lines (A549, HeLa, HepG2, MCF-7, and MDA-MB-231) over normal cells (WI-38, human embryonic fibroblasts and BJ, human skin fibroblasts) under hypoxic conditions. Good stability was observed for **38** under normoxic conditions. The tail vein injection of **38** ($10 \mu\text{M}$, $20 \mu\text{M}$ in $100 \mu\text{L}$ PBS, respectively, twice a week for three weeks) in the murine mammary carcinoma 4T1 tumour-bearing mice model exhibited a dose-dependent reduction in tumour growth compared to the control group (Fig. 24b and c). This work demonstrated the utility of the

aryl-azo benzyl alcohol motif for the development of hypoxia-responsive prodrugs.

Piao *et al.* reported the hypoxia-responsive PDT agent, **39** (Fig. 25a), for the selective PDT treatment of A549 cancer cells under hypoxia.¹³¹ A seleno-rosamine scaffold (SeR) was used as the fluorophore as well as the photosensitizer with a high $^1\text{O}_2$ quantum yield ($\Phi^1\text{O}_2 = 0.56$). The amino unit was functionalised with an azo group to afford hypoxia-responsive **39**. This azo unit served to quench the fluorescence emission at 557 nm and $^1\text{O}_2$ production at 532 nm . Under hypoxic conditions, **39** was shown to be readily reduced by NADPH ($50 \mu\text{M}$) rat liver microsomes to afford fluorescent SeR. The phototoxicity of **39** was measured using a live/dead cell viability assay with ethidium homodimer (EthD, staining dead cells) and Calcein AM (staining live cells). The phototherapeutic effect of **39** in A549 cells was shown to be significantly higher under hypoxia (from extreme hypoxia 0.1% to mild hypoxia 8% oxygen concentrations) than normoxic conditions (Fig. 25b). Although this work exhibits great promise, the excitation wavelength is not optimal for *in vivo* applications as such further development is needed to afford long-wavelength derivatives.

Another hypoxia-responsive PDT agent (**40**) exploits FRET to achieve simultaneous tumour imaging and PDT (Fig. 26).¹³² With **40**, pyropheophorbide α (PPA, FRET donor) was covalently linked to fluorescent NIR dye SiR-665 (FRET acceptor) *via* an azo-benzene linker. Due to FRET and the presence of the azo



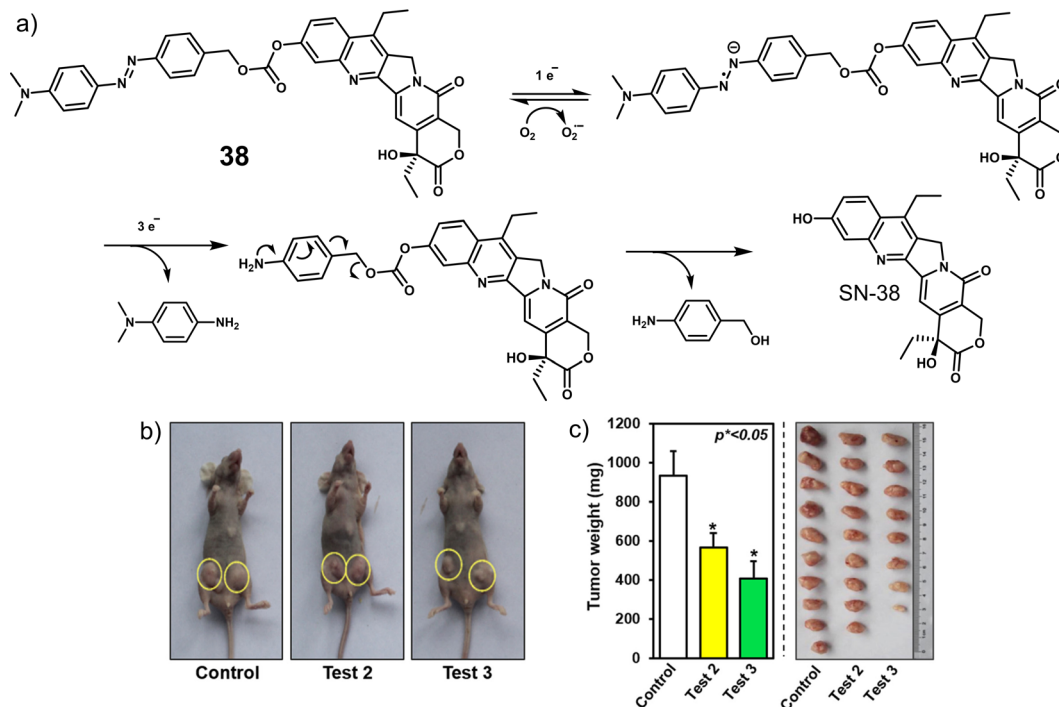


Fig. 24 (a) Chemical structure of **38** and the mechanism for the hypoxia-activated release of SN-38. (b) Photographic images of tumour-bearing xenograft mice ($n = 11$ per test group) and (c) Tumours weight with dissected tumour tissue treated with PBS, test 2 (**38**, 10 μ M, 100 μ L PBS) or test 3 (**38**, 20 μ M, 100 μ L PBS) via tail vein twice weekly for three weeks. Reproduced with permission from ref. 130. Copyright (2018) Elsevier B.V.

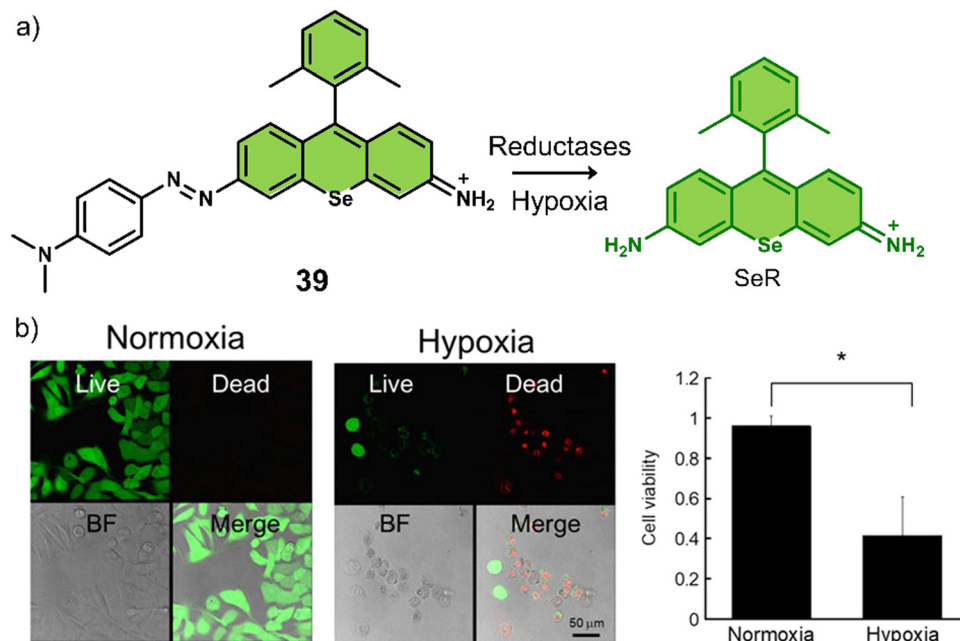


Fig. 25 (a) Chemical structure of **39** and the hypoxia-activatable release of SeR. (b) Live/dead staining images of A549 cells that were photoirradiated (535 nm, 22 mW cm^{-2} , 2 min) under normoxia and hypoxia after treatment with **39** (1 μ M). Values are represented as mean \pm SD ($n = 3$). * $P < 0.05$. Reproduced with permission from ref. 131. Copyright (2017) American Chemical Society.

group, **40** exhibited low fluorescence emission and singlet oxygen yield. However, in a hypoxic environment (NADPH and liver microsomes), an increased fluorescence emission at 680 nm and an enhanced 1O_2 -generation were observed. With

these promising results, the ability of **40** for hypoxia-specific fluorescence imaging and photodynamic ablation of cancer cells *in vitro* was demonstrated in a BEL-7402 cell line (human hepatoma cells).



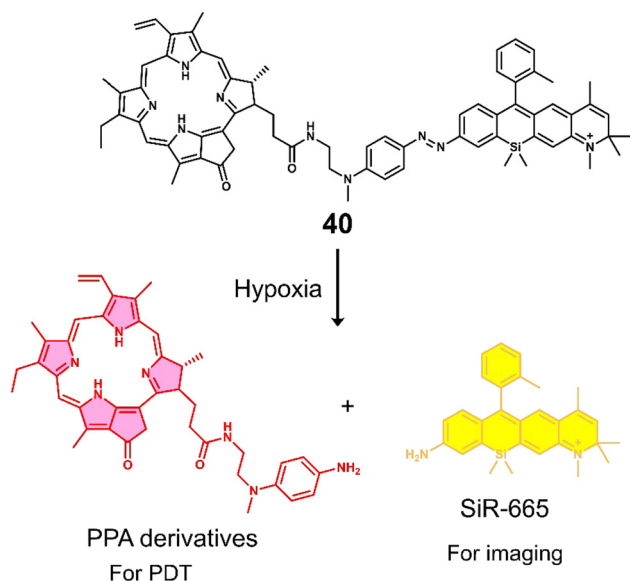


Fig. 26 Chemical structure of **40** and the hypoxia-activated release of the PS PPA derivatives and NIR-dye SiR-665.

3.4 Acidic pH-responsive activatable chemotherapeutics and activatable “smart” phototherapeutics

As stated briefly in the introduction, the TME is known to be acidic (pH 6.5 to 7.2) due to the increased glycolytic metabolism

of cancer cells (Warburg effect). The acidic TME is viewed as a hallmark trait of cancers.^{133,134} The acidic intra- and extracellular pH in solid tumour tissues is crucial for cancer cell proliferation, migration, invasion, metastasis, evasion of immune surveillance, and drug resistance.¹³⁵ This well-known property of the TME has resulted in a vast number of reports directed towards the development of acidic responsive prodrug systems.¹³⁶ In this section, we discuss some selected examples of acidic responsive prodrugs that have been proven successful for chemotherapeutic and phototherapeutic applications (Table 4).

Several research groups have focused on the development of prodrugs that release a therapeutic in acidic environments. Examples include the use of acid-sensitive breakable bonds (*e.g.*, hydrazone) and ‘ionisable’ chemical groups (*e.g.*, tertiary amines) to construct acid-sensitive therapeutic reagents (Scheme 5).

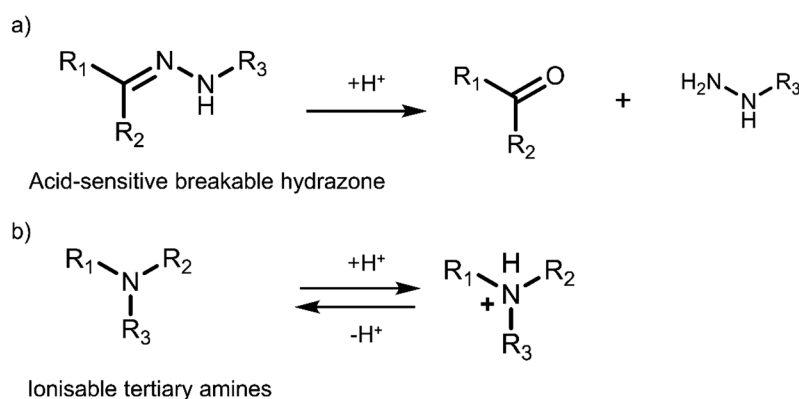
Zhao and co-workers reported the acid-responsive theranostic **41** for the targeted delivery of DOX to cancer cells. In this theranostic, the tumour-targeting decapeptide AP2H (IHGHHIISVG) was linked to DOX *via* an acid-sensitive hydrazone bond (Fig. 27).¹³⁷ AP2H is specific for the tumour-associated protein, lysosomal protein transmembrane 4 beta (LAPTM4B), that is expressed in various malignancies, including breast cancer, ovarian cancer, hepatocellular carcinoma (HCC), gastric cancer, and cervical cancer.¹³⁸ HPLC studies revealed that over a period of 48 h at pH 5.0, 89% of DOX was released from **41**, whereas at pH 6.0 and 7.4, only 31% and 10% were released under identical conditions, respectively.

Table 4 Summary of pH-responsive prodrugs and phototherapeutics discussed in this review

Stimuli	Active anti-cancer therapeutic agent	Treatment	<i>In vitro</i> models	<i>Ex vivo/in vivo</i> models	Ref.
41 Acid pH	DOX	Chemotherapy	A549 and HepG2 cells	—	137
42 Acid pH	DOX	Chemotherapy	U87 cells	—	139
43 Acid pH	DOX	Chemotherapy	A549 and CT26 cells	—	140
44 Acid pH	Iodinated-IR783	PDT	HepG2 cells	—	141
45 Acid pH	<i>p</i> -Phenyleneethynylene-based derivative in zwitterionic form	PDT	HeLa cells	—	142
46 Acid pH	Heptamethine cyanine dye, IR-822	PTT	MCF-7 cells	MCF-7 tumour-bearing mice	143
47 Acid pH	Rhodamine lactam analogue	PTT	HeLa, U-87 MG, and RAW 264.7 cells	H22 tumour-bearing mice	145

Note: DOX: doxorubicin, PDT: photodynamic therapy, PTT: photothermal therapy, —: not mentioned.

Acidic pH-responsive activatable chemotherapeutics and activatable “smart” phototherapeutics



Scheme 5 Basic mechanism of hydrazone-based and tertiary amines-based prodrugs activation.



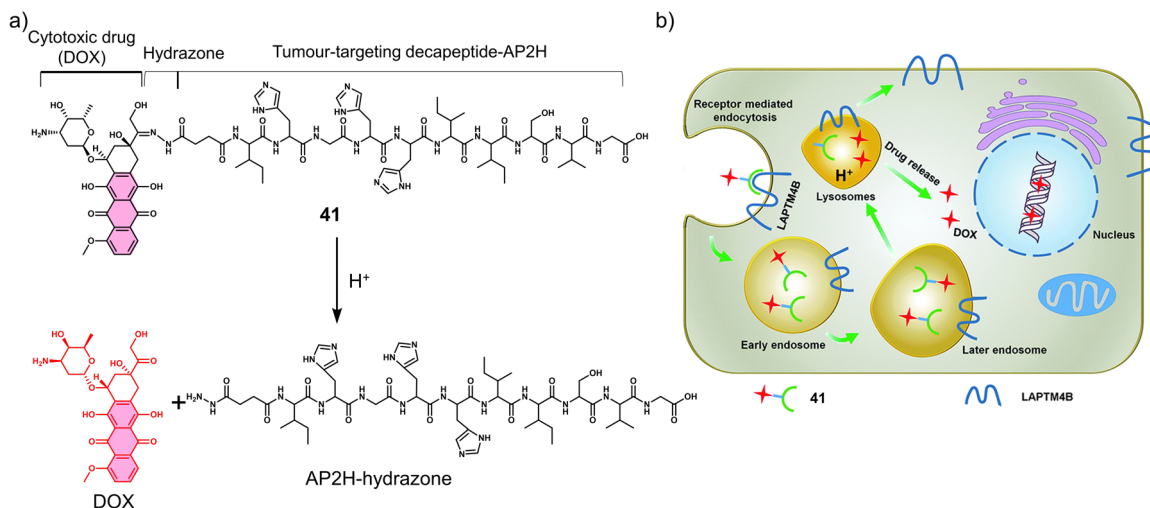


Fig. 27 (a) Chemical structure of **41** and the acidic pH-activatable release of DOX. (b) The schematic illustration of LAPTMB4B guided **41** internalization and acidic pH-mediated drug release process. Reproduced with permission from ref. 137. Copyright (2015) The Royal Society of Chemistry.

Encouraged by the excellent release profile at low pH values, **41** was shown to exhibit dose-dependent toxicity in LAPTMB4B-positive A549 and HepG2 cell lines with IC₅₀ of 1.14 μM and 4.0 μM, respectively. At the same time, minimal cytotoxicity was observed in a human embryonic kidney HEK 293 cell line (healthy cells), which illustrated the effectiveness of the AP2H targeted strategy for cancer cells. Fluorescence co-localisation experiments revealed that **41** was first localised in endosomes and lysosomes, which was ascribed to the targeting ability of AP2H for the LAPTMB4B protein. Since endosomes and lysosomes are known to have a pH range of 4.5–6.5, DOX was released efficiently, which led to its translocation to the cell nucleus to activate the anticancer effect (DNA damage).

Another acid-responsive hydrazone-based DOX prodrug (**42**) was developed by Li *et al.* for *in vitro* evaluation in U87 cells. **42** consisted of an α_vβ₃ integrin targeting moiety (GRGDS), the coumarin fluorophore, and DOX. The fluorescent peptide backbone was conjugated to DOX through a hydrazone maleimide linker (Fig. 28).¹³⁹ Under aqueous physiological conditions (pH 7.4), **42** showed a low emission intensity; however, as the pH value decreased to 5.0, a rapid fluorescence enhancement was observed at 460 nm. These results were ascribed to the hydrolysis of the hydrazone unit and the release of DOX (confirmed by HPLC analysis). Fluorescence spectroscopy results revealed that 93.8% of DOX from **42** was released at pH 5.0 after 11 h, while only about 40.7% was released at pH 7.4. Furthermore, *in vitro* studies indicated that upon incubation with α_vβ₃

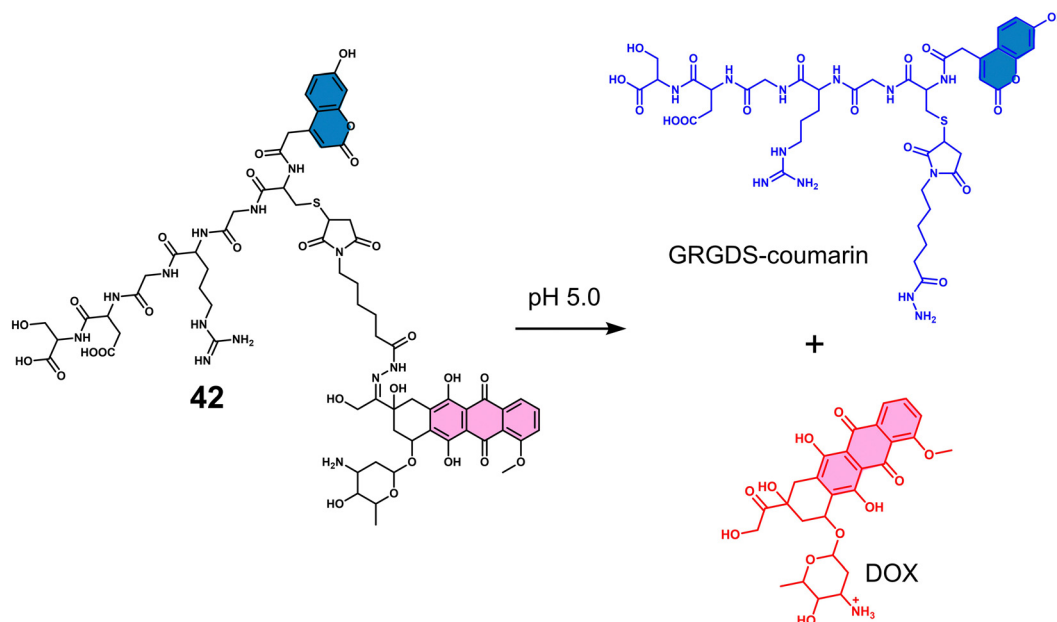


Fig. 28 Chemical structure of **42** and the acid-mediated release of DOX.



integrin-positive human glioblastoma U87 cells, **42** exhibited concentration-dependent anticancer effects ($IC_{50} = 0.19 \mu\text{g mL}^{-1}$). The result of time-dependent fluorescence imaging of **42** in U87 cells confirmed the suitability of the **42** to monitor DOX release *in vitro* in real-time.

Sessler and co-workers reported the multimodal theranostic hydrazone-based MGd and DOX conjugate (**43**, Fig. 29a) for the acid-triggered release of DOX in A549 and CT26 live cells.¹⁴⁰ **43** consisted of MGd conjugated to two molecules of DOX by an acid-responsive hydrazone linker. Each component in this strategy can be monitored by two complementary imaging methods, fluorescence spectroscopy and MR imaging. Initially, the fluorescence emission of DOX was quenched due to the paramagnetic nature of MGd. In acidic environments (pH 5.0), the hydrazone bonds of **43** underwent acid hydrolysis, resulting in the release of free DOX, which consequently showed a significant increase in the fluorescence signal at 593 nm. Relatively small changes in fluorescence emission intensity were observed at pH 7.4. This acid-triggered transformation was further confirmed by reverse-phase high-performance liquid chromatography (RP-HPLC) and electrospray ionization mass spectroscopy (ESI-MS). **43** exhibited higher antiproliferative activity in cancer cell lines (A549 and CT26) than in a normal fibroblast cell line NIH3T3 (Fig. 29b). These results were attributed to the fast uptake of **43** by cancer cells compared to normal cells, and once internalised, **43** undergoes acid-mediated hydrazone cleavage, releasing free DOX. The paramagnetic Gd(III) center of MGd was used for MR imaging of A549 and CT26 cancer cells, which revealed a strong MR

signal when either cell line was treated with $4 \mu\text{M}$ of **43**. More importantly, the MRI signal of **43** was generated without the need for hydrolysis of hydrazone to generate DOX and generate a fluorescence signal.

In contrast, “ionisable” groups are often introduced in the construction of phototherapeutic agents, enabling the photophysical properties of the phototherapeutic agent to be reversibly switched under different pH environments. Kamkaew and co-workers reported the synthesis of the pH-responsive PDT agent, **44**, for the treatment of hepatocellular carcinoma. The introduction of the pH-sensitive *N*-methylpiperazine unit onto iodinated-IR783 afforded an attractive system (**44**) that exhibits high $^1\text{O}_2$ quantum yield and enabled the NIR image-guided PDT of acidic environments (Fig. 30).¹⁴¹ The addition of iodine atoms to IR783 was used to enhance the PDT properties through the heavy atom effect. At neutral pH, **44** was non-fluorescent with a low $^1\text{O}_2$ production. However, in acidic environments (pH 3.0–5.0), the two nitrogen atoms of *N*-methylpiperazine unit become protonated, blocking the PeT (photoinduced electron transfer) process, thereby facilitating the fluorescence enhancement and increase in $^1\text{O}_2$ generation under NIR light irradiation (850 nm, 30 mW cm^{-2}). A live/dead cytotoxicity assay (Calcein AM and propidium iodide co-staining) of **44** indicated that the phototoxicity to HepG2 cells under 850 nm light irradiation at pH 5.0 was significantly higher than at pH 7.4. No dark toxicity was observed. Excellent penetration depth was obtained when a 5 mm piece of pork tissue was placed between the cells and the light source, attributed to the long excitation wavelengths. The above results confirm the potential of **44** for the PDT of solid tumours.

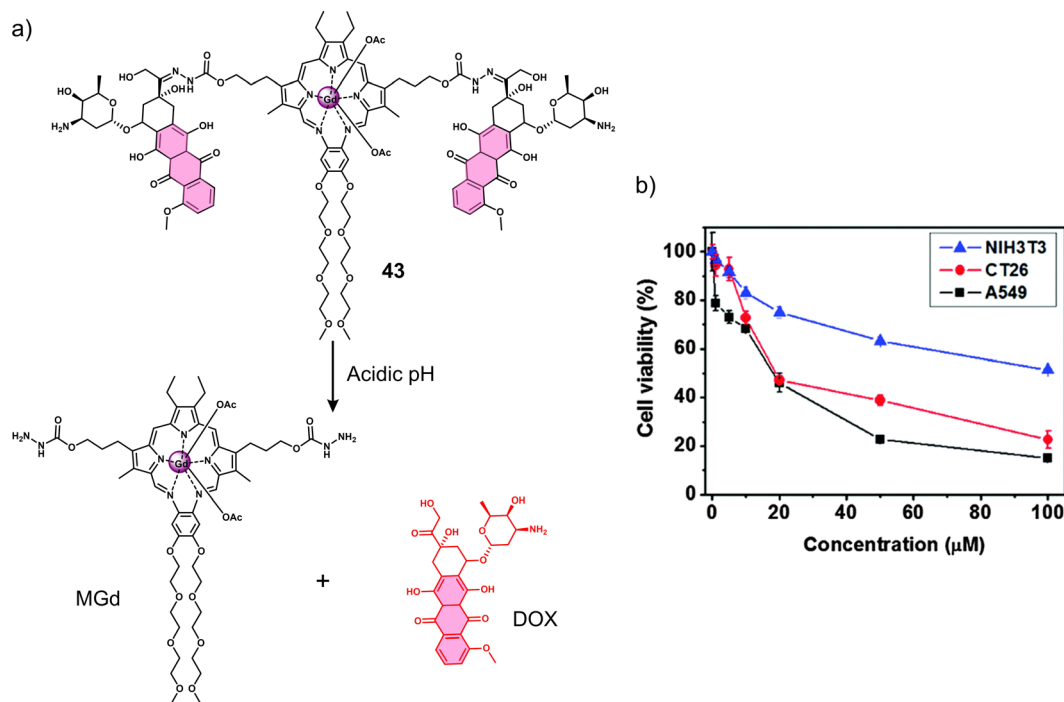


Fig. 29 (a) Chemical structure of **43** and the acid-responsive release of the Gd(III) texaphyrin and DOX. (b) Antiproliferative activity of **43** incubated at various concentrations for 48 h with A549, CT26, and NIH3T3 using a standard MTT assay. Reproduced with permission from ref. 140. Copyright (2016) The Royal Society of Chemistry.



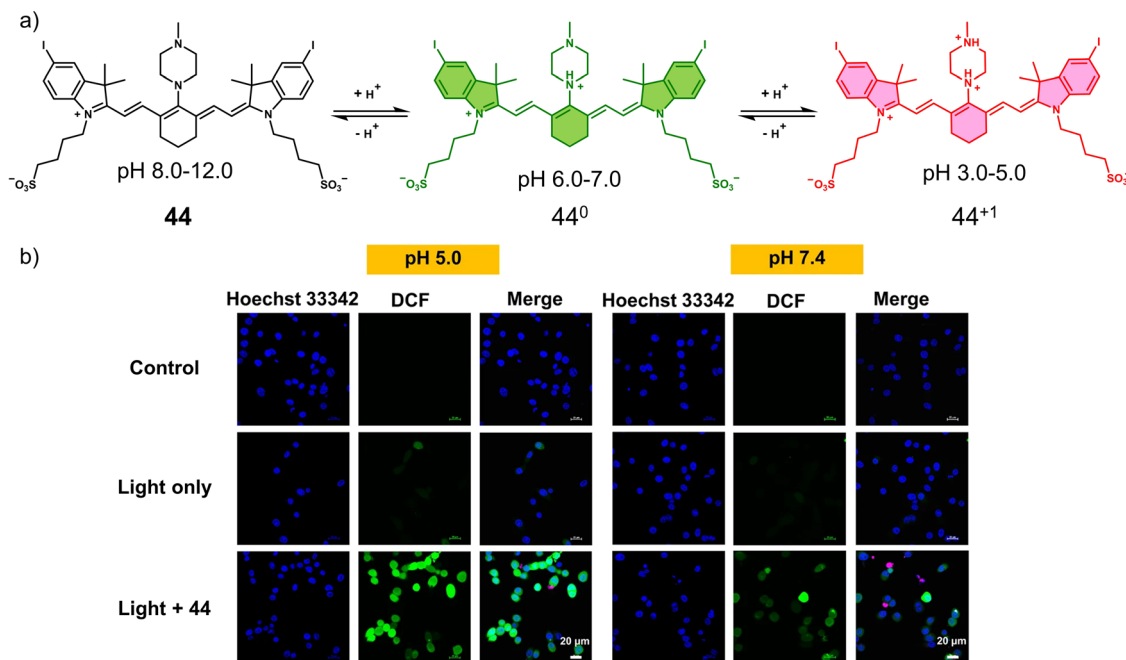


Fig. 30 (a) Chemical structure of **44** and its pH-activated mechanism at values 6.0–7.0 and 3.0–5.0. (b) Detection of intracellular ROS generated by **44** (20 μ M) at pH 5.0 and 7.4 in HepG2 cells using 2,7-dichloro-dihydro-fluorescein diacetate (DCFH-DA) cellular ROS detection assay. Scale bars: 20 μ m. Reproduced with permission from ref. 141. Copyright (2020). The Author (S. Siriwibool, N. Kaekratoke, K. Chansaenpak, K. Siwawannapong, P. Panajapo, K. Sagarik, P. Noisa, R.-Y. Lai and A. Kamkaew).

Hu *et al.* developed a unique PDT agent, **45**, where the intersystem crossing (ISC) process (singlet oxygen activation) was shown to be reversibly activated *via* a pH-induced conjugated backbone twist (CBT) (Fig. 31a).¹⁴² The twisting was achieved

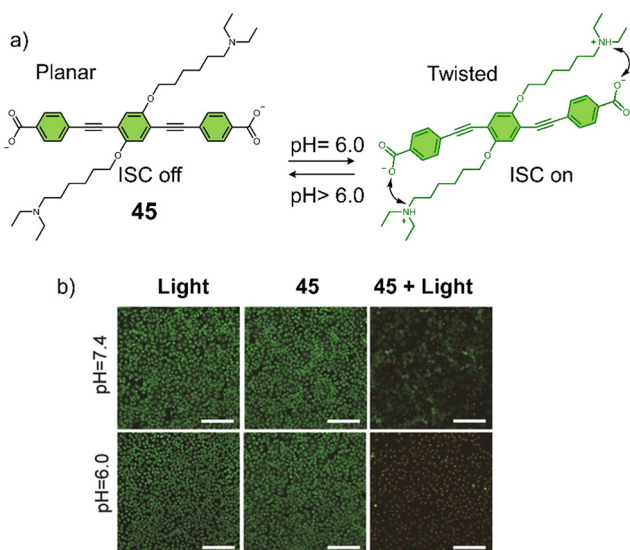


Fig. 31 (a) The chemical structure and design of **45** with pH-reversible conjugated backbone twisting (CBT) for the reversible switching of ISC. (b) Live/dead staining images of HeLa cells in media pH 6.0 and pH 7.4 upon the light irradiation after treatment with **45**. Scale bar: 160 μ m. The green color represents live cells, and the red color represents dead cells. Reproduced with permission from ref. 142. Copyright (2019) Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim.

through changes in intramolecular electrostatic interactions between positively charged quaternary ammonium functional groups and negatively charged terminal carboxyl groups. At neutral pH, the anionic form of **45** was not twisted, whereas in acidic pH, electrostatic interactions induced twisting. Techniques, including femtosecond transient absorption (fs-TA) spectroscopy and quantum chemical calculations, indicated that the ISC efficiency (Φ_{ISC}) was improved by increasing the degree of twisting of the molecule. Thus, in acidic pH (\sim 6.0) (compound's isoelectric pH), **45** exhibited a high singlet oxygen quantum yield. White light irradiation (10 mW cm^{-2} , 5 min) of **45** treated HeLa cells exhibited a good acidic-pH selectivity at pH 6.0 compared to cells at pH 7.4 (Fig. 31b). Since previously discussed activatable PDT agents are irreversible, this work provides an alternative and reversible strategy for the activation of a PS. To the best of our knowledge, this is the first time that a pathological factor (*i.e.*, pH) has been stimulated to modulate the reversible switching of ISC and thus improve the precision of photodynamic therapy. However, this concept has not yet been applied for *in vivo* studies and is still some way from practical application, so we believe that this system required further development and optimisation.

Cai and co-workers reported the acid-responsive PTT agent **46** (Fig. 32a) consisting of NIR-fluorescent heptamethine cyanine dye, IR-822, with N1-(pyridin-4-ylmethyl)ethane-1,2-diamine (PY) as the pH-responsive moiety.¹⁴³ **46** was shown to be non-fluorescent at neutral pH. However, in acidic environments, the "N" of pyridine became protonated and resulted in an increase in fluorescence emission at 792 nm. This was accompanied by a change in absorption, which led to an increase in the PA signal at



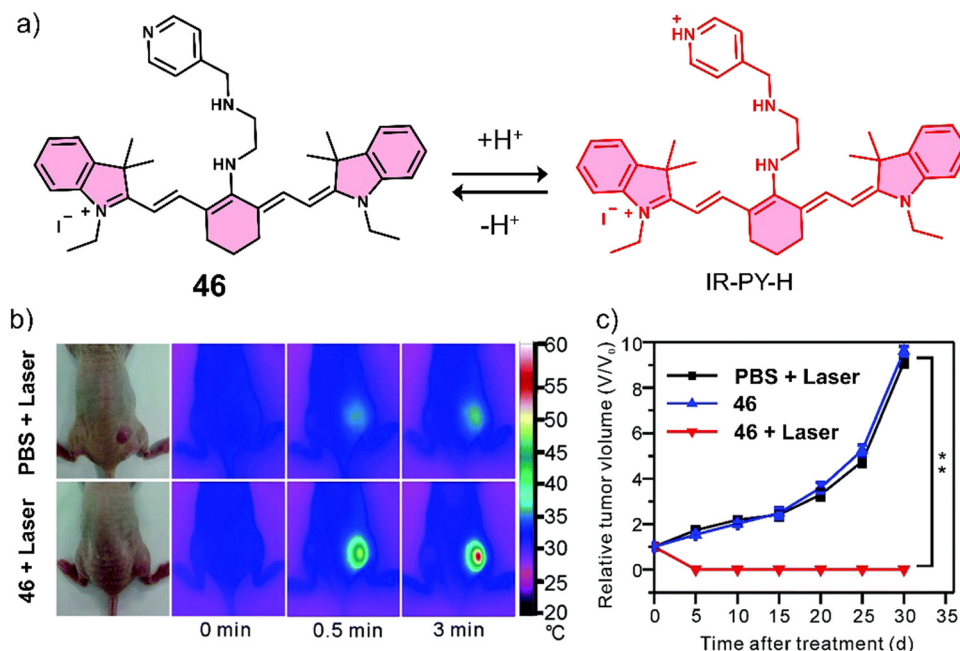


Fig. 32 (a) Chemical structure of **46** and activation at acidic pH. (b) Infrared thermal images of MCF-7 tumour-bearing mice after tail-vein injection of PBS (100 μL), or **46** (100 μL , 40 $\mu\text{g mL}^{-1}$) upon the NIR laser irradiation (808 nm, 0.4 W cm^{-2} , 3 min) once. (c) The tumour growth curve for 30 days post-treatment for the different groups. The data are shown as mean \pm SD ($n = 10$), $**P < 0.01$. Reproduced with permission from ref. 143. Copyright (2017) The Royal Society of Chemistry.

637 nm. **46** exhibited good photothermal-conversion efficiency under NIR 808 nm laser irradiation (0.4 W cm^{-2} , 3 min), with a temperature rise from 25.7 $^{\circ}\text{C}$ to 52.4 $^{\circ}\text{C}$. In MCF-7 cells, **46** exhibited a gradual increase in fluorescence intensity as the pH of the microenvironment decreased. The phototheranostic properties of **46** were evaluated using a MCF-7 tumour-bearing mice model (Fig. 32b and c). The NIR fluorescence and PA imaging indicated that the maximum imaging intensity corresponding to **46** was seen 14 h post-injection. Under low-powered laser irradiation (808 nm, 0.4 W cm^{-2} , 3 min), **46** was found to exhibit significant tumour ablation, and no tumour recurrence was seen after 30 days. The mice treated with **46** and light irradiation similarly maintained a high survival rate when compared to control groups (those treated with PBS or **46** without light irradiation).

Han and co-workers developed the PTT agent **47** that uses a sialic acid (SA) targeting unit for selective imaging and cancer treatment *in vitro* and *in vivo*. Tumour cells are well-known to have an increased metabolic demand for SA, which has led researchers to exploit this functionality as a targeting strategy.¹⁴⁴ The SA unit was linked to a pH-activatable NIR PTT agent (a rhodamine lactam analogue) (Fig. 33a).¹⁴⁵ Under physiological conditions (pH 7.2), **47** exhibited a relatively weak fluorescence emission intensity at 740 nm. However, in acidic environments (pH 4.5–6.0), **47** can be readily isomerised into highly fluorescent fluorone by proton-mediated ring opening of the intramolecular lactam,¹⁴⁶ and an enhanced NIR fluorescence emission was observed, accompanied by a strong increase in absorption at 715 nm and good photothermal conversion efficiency upon 660 nm laser illumination (0.5 W cm^{-2} , 600 s). The acid-responsiveness of **47** was confirmed in HeLa, U-87 MG, and macrophage RAW 264.7 cell lines with an

increased fluorescence intensity being observed in lysosomes. The fluorescence response was confirmed through the pre-incubation of HeLa, U-87 MG, and RAW 264.7 cell lines with a V-ATPase inhibitor, bafilomycin A1 (BFA), which causes an increase in lysosomal pH.¹⁴⁷ **47** exhibited the strongest NIR fluorescence emission at 750 nm at the tumour site 48 h post-injection in a H22 tumour-bearing mice model. Remarkably, the signal corresponding to **47** was found to last for 144 hours. Demonstrating its utility as a phototheranostic agent, significant phototoxicity was observed for HeLa, U-87 MG, and RAW 264.7 cells after NIR laser irradiation at 660 nm (0.5 W cm^{-2} , 10 min) (Fig. 33b).

3.5 The use of external stimuli for the release of therapeutics

Photocages are a class of functional groups that are designed to release “cargo” under the irradiation of light. The reader can be directed to several excellent reviews for an extensive overview of photocages.^{148–150} Owing to the non-invasive, high spatial resolution, and spatio-temporal control of light, several photocages have been developed for chemotherapeutic applications.^{151–153} Initial reports for these light-responsive strategies relied on the use of a nitrobenzyl photoresponsive group. Unfortunately, this unit requires activation using short excitation wavelengths (<500 nm), which are associated with poor tissue penetration and potential phototoxicity.¹⁵⁴ In recent years, a significant focus has been directed towards light-activatable systems that require long excitation wavelengths and that exhibit enhanced photoconversion efficiencies (Table 5). For an extensive overview on the photoremovable protecting units, the reader can be directed to the following excellent reviews.^{155,156}



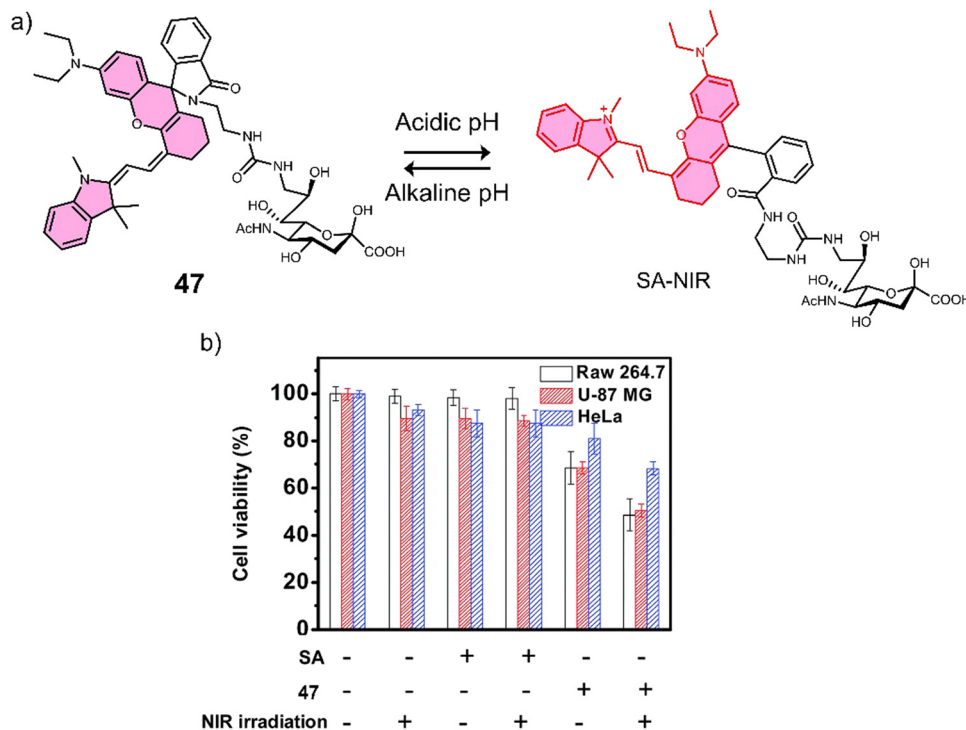


Fig. 33 (a) Chemical structure of **47** for targeted imaging of cancerous HeLa, U-87 MG, and macrophage RAW 264.7 cell lines that overexpress SA receptors and activation at acidic pH. (b) Phototoxicity study with HeLa, U87-MG, and RAW 264.7 cell lines. Cell viability was determined by MTT assay. Reproduced with permission from ref. 145. Copyright (2014) The Royal Society of Chemistry.

Recently, the work of the Winter group has led to the identification of a series of BODIPY-based photocages (*e.g.*, **48–55**) that are activated under visible or NIR light irradiation (Fig. 34).¹⁵⁷ Some of these photocages are now commercially available from Sigma Aldrich under the trade names of WinterGreen carbamoyl imidazole photocage (**51**) and WinterRed Photocage (**52**) (Fig. 34). A recent example from the Winter group for these systems was with the covalent attachment of the cell impermeable CTSSB inhibitor CA-074 with WinterGreen to form **56** (Fig. 35).¹⁵⁸ It was rationalized that cell uptake could be achieved *via* the use of a photocage *i.e.*, **56**. **56** was found to be cell-permeable, and irradiation using blue light (460–470 nm) was shown to release CA-074 and BODIPY (WinterGreen). Exhibiting cytotoxicity toward MDA-MB-231 cells at 72 h incubation with the EC₅₀ (half maximal effective concentration) of 0.78 μM under light irradiation (460–470 nm, 15 min). Also, minimal toxicity was observed towards normal MCF-10A cell

lines with/without light irradiation (EC₅₀ = 3.5 μM and 27 μM, respectively). Mechanistic studies indicated that CTSSB inhibition was enhanced by the production of ¹O₂ due to light irradiation of the BODIPY fragment. Flow cytometry confirmed that 10 μM of **56** combined with light irradiation (460–470 nm, 15 min) was sufficient to achieve similar toxicities to 4 h of treatment using 500 mM H₂O₂ (known to mediate cell necrosis).

Recently, Zhu and co-workers developed a light-activatable (650/660 nm) oxaliplatin Pt(IV) conjugate photosensitizer (**57**, Fig. 36a).¹⁵⁹ In this photosensitizer, Pyropheophorbide α (PPA) was linked to the axial position of the oxaliplatin Pt(IV) prodrug, and it was rationalised that the light activation of ROS could prove synergistic for Pt(IV) prodrugs.¹⁶⁰ Remarkably, the light irradiation of **57** using a low-intensity red light excitation (650 nm, 7 mW cm⁻²) was shown *via* MS and HPLC to reduce Pt(IV) to Pt(II) and release PPA. **57** displayed enhanced cytotoxicity towards cancer cells [(platinum-sensitive A2780 and platinum-resistant A2780cis, human

Table 5 Summary of external stimuli-responsive therapeutics discussed in this review

Stimuli	Active anti-cancer therapeutic agent	Treatment	<i>In vitro</i> models	<i>Ex vivo/in vivo</i> models	Ref.	
56	Visible light	CA-074 and WinterGreen	Chemotherapy and PDT	MDA-MB-231 cells	—	158
57	Red-light	Oxaliplatin	Chemotherapy	Platinum-sensitive A2780, platinum-resistant A2780cis and MCF-7 cells	4T1 tumour-bearing mice	159
58	X-ray	Pazopanib	Chemotherapy	HUVEC cells	HT29 tumour-bearing mice	161
59	X-ray	DOX	Chemotherapy	HeLa cells	HeLa tumour-bearing mice	161

Note: DOX: doxorubicin, PDT: photodynamic therapy, —: not mentioned.



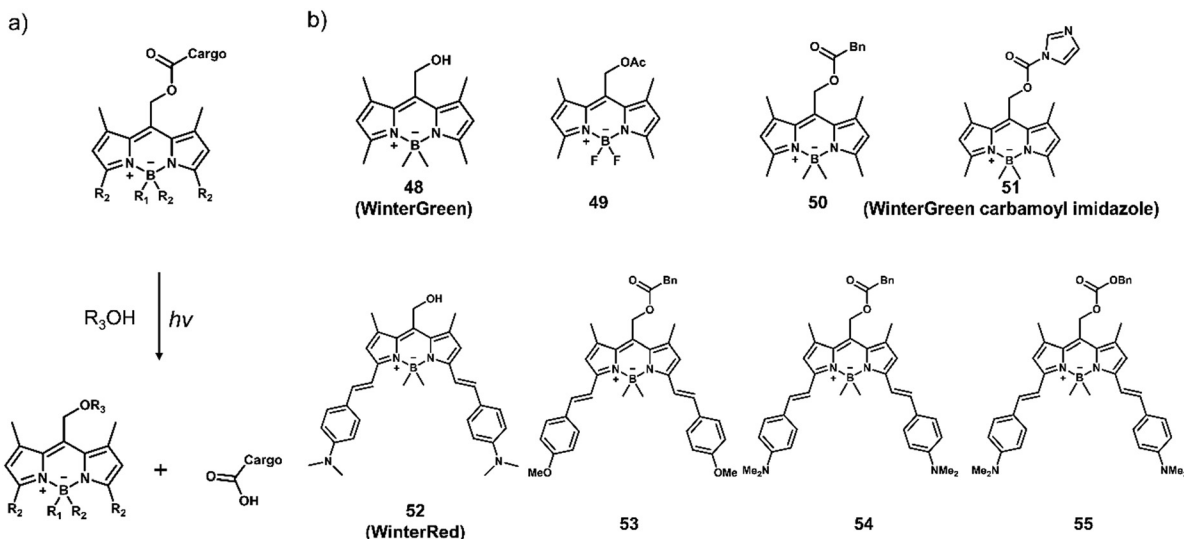


Fig. 34 (a) Light-activated release mechanism and (b) representative chemical structures of BODIPY-based photocages (**48–55**) reported by the Winter group.

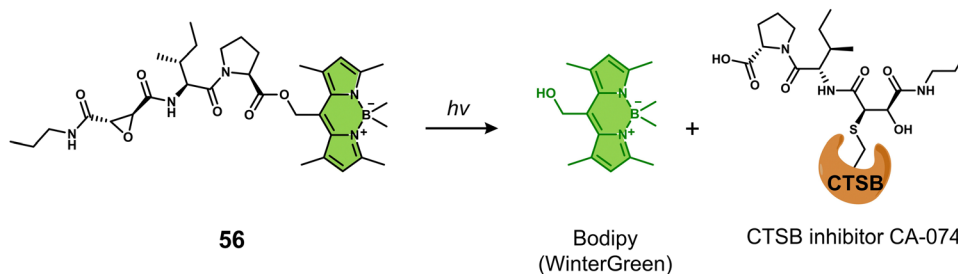


Fig. 35 Chemical structure of **56** and the light-responsive (460–470 nm) release of the CTSB inhibitor CA-074.

ovarian cancer cell; MCF-7, human breast cancer cell; 4T1, murine breast cancer cell line) ($IC_{50} < 0.2 \mu\text{M}$) with light irradiation (650 nm, 7 mW cm^{-2} , 15 min) than without light ($IC_{50} > 10 \mu\text{M}$). Minimal dark toxicity was observed towards MRC-5 cells [(MRC-5, normal human lung fibroblast) ($IC_{50} > 10 \mu\text{M}$)]. **57** combined with light irradiation (660 nm, 100 mW cm^{-2} , 10 min) was found to have greater antitumour activity *in vivo* than oxaliplatin and a mixture of oxaliplatin with PPA in a murine mammary adenocarcinoma 4T1 tumour-bearing mice model (Fig. 36b and c). This study represents the first example where red light-controlled activation of platinum prodrugs has been used *in vivo*. We anticipate that this work will motivate others to develop new strategies for the light-controlled activation of platinum (Pt(IV)) prodrugs.

Radiotherapy is often used in combination with chemotherapy to enhance the inhibition of tumour growth. However, the use of radiotherapy (X-rays) to activate prodrugs has remained elusive. The development of X-ray-activatable prodrugs provides the potential to activate therapeutics specific to the tumour area during conventional X-ray radiotherapy, potentially reducing patient side effects. In 2021, Bradley and co-workers reported the use of a medical radiotherapy X-ray source to selectively activate prodrugs to elicit their antiproliferative effects.¹⁶¹ An extensive screening process was used to identify molecules that

have the potential to undergo a chemical transformation during X-ray radiation (60 Gy). As a result 4-acetamidobenzenesulfonyl azide and 4-(hydroxymethyl)-2,3,5,6-tetrafluoroaryl azide were identified that undergo transformation into 4-acetamidobenzenesulfonamide and 4-(hydroxymethyl)-2,3,5,6-tetrafluoroaniline, respectively. To illustrate the clinical potential of this strategy, the authors synthesised a pazopanib-based prodrug (**58**, Fig. 37a) functionalised with the X-ray activatable sulphonyl azide moiety. Pazopanib is a small molecule inhibitor of the vascular endothelial growth factor receptor (VEGFR) approved for the treatment of advanced kidney cancer.¹⁶² HPLC analysis demonstrated that 90% of **58** was transformed under X-ray irradiation (60 Gy). Moreover, activation of **58** was shown to be radiation dose-dependent and was effective on HUVEC (human umbilical vein endothelial) cell lines *in vitro*. A similar antiproliferative effect was observed to that of pazopanib when cells incubated with **58** (0, 5, and $10 \mu\text{M}$) were subjected to X-ray irradiation (0, 6, 12, and 24 Gy). Intratumoural injection of **58** (100 mg kg^{-1}) in conjunction with X-ray irradiation (6 Gy) in HT29 tumour-bearing mice afforded comparable tumour inhibition as pazopanib (100 mg kg^{-1}), with prolonged survival times compared to the control groups (pazopanib or X-ray only) (Fig. 37b). To demonstrate the general nature of this strategy,



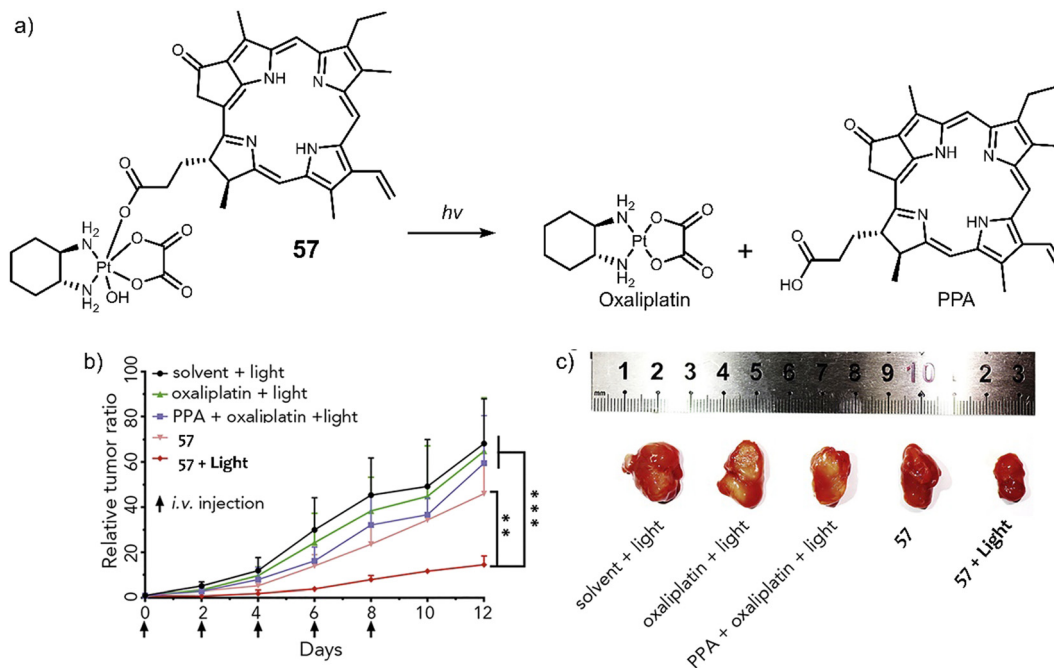


Fig. 36 (a) Chemical structure of **57** and the light-activated (650/660 nm) release mechanism of oxaliplatin. (b) Tumour growth inhibition over time with control groups (solvent + light, oxaliplatin + light, PPA + oxaliplatin + light, **57**) and **57** that i.v. injected at a Pt dose of $3.5 \mu\text{mol kg}^{-1}$ every two days for 10 days upon the light (660 nm, 100 mW cm^{-2} , 10 min). i.v.; intravenous injection. Arrows represent the time of administration and irradiation. $**P < 0.01$, $***P < 0.001$, mean \pm SD; $n = 5$. (c) Representative images of excised tumours at the study endpoint. Reproduced with permission from ref. 159. Copyright (2019) Elsevier Inc.

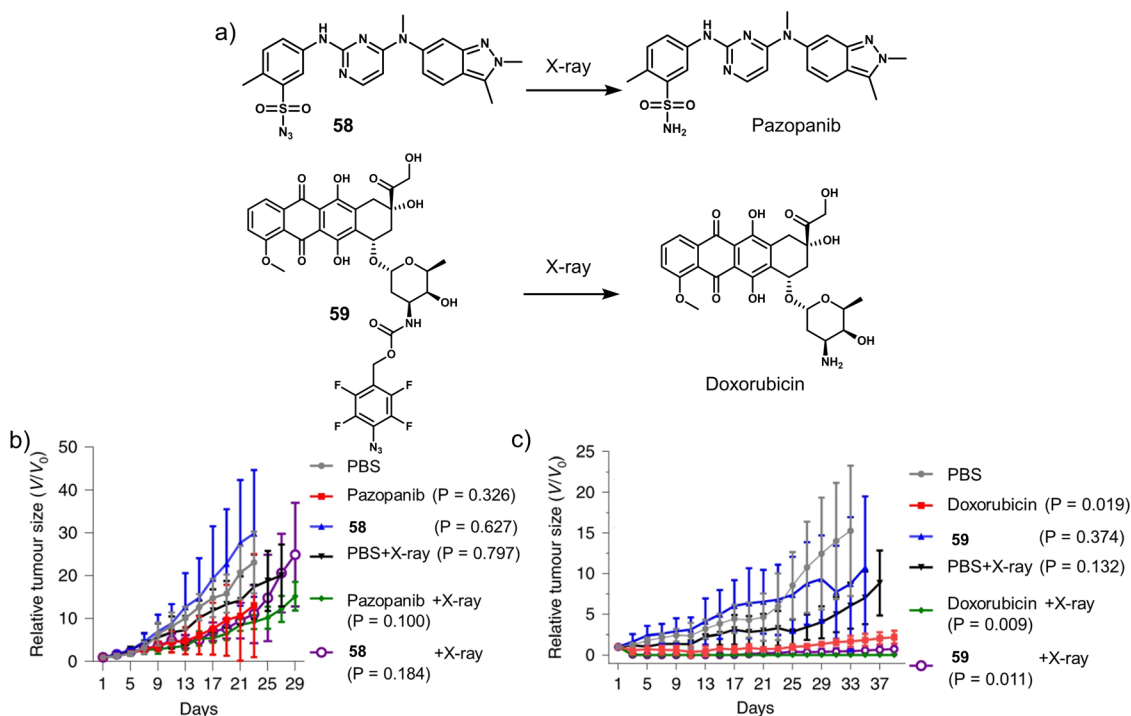


Fig. 37 (a) Chemical structures of **58**, **59**, and the X-ray-activated release of pazopanib and doxorubicin (DOX). (b) Tumour growth inhibition of HT29 tumour-bearing mice models over the time with PBS, pazopanib (100 mg kg^{-1}), **58** (100 mg kg^{-1}) with or without X-ray irradiation (6 Gy) after 4 h of post intratumoural injection. (c) Tumour growth inhibition of HeLa tumour-bearing mice models over the time with PBS, doxorubicin (10 mg kg^{-1}), **59** (10 mg kg^{-1}) with or without X-ray irradiation (6 Gy) after 4 h of post intratumoural injection. Mean \pm SD; $n = 5$. Reproduced with permission from ref. 161. Copyright (2021) The Author (J. Geng, Y. Zhang, Q. Gao, K. Neumann, H. Dong, H. Porter, M. Potter, H. Ren, D. Argyle and M. Bradley), under exclusive licence to Springer Nature Limited.



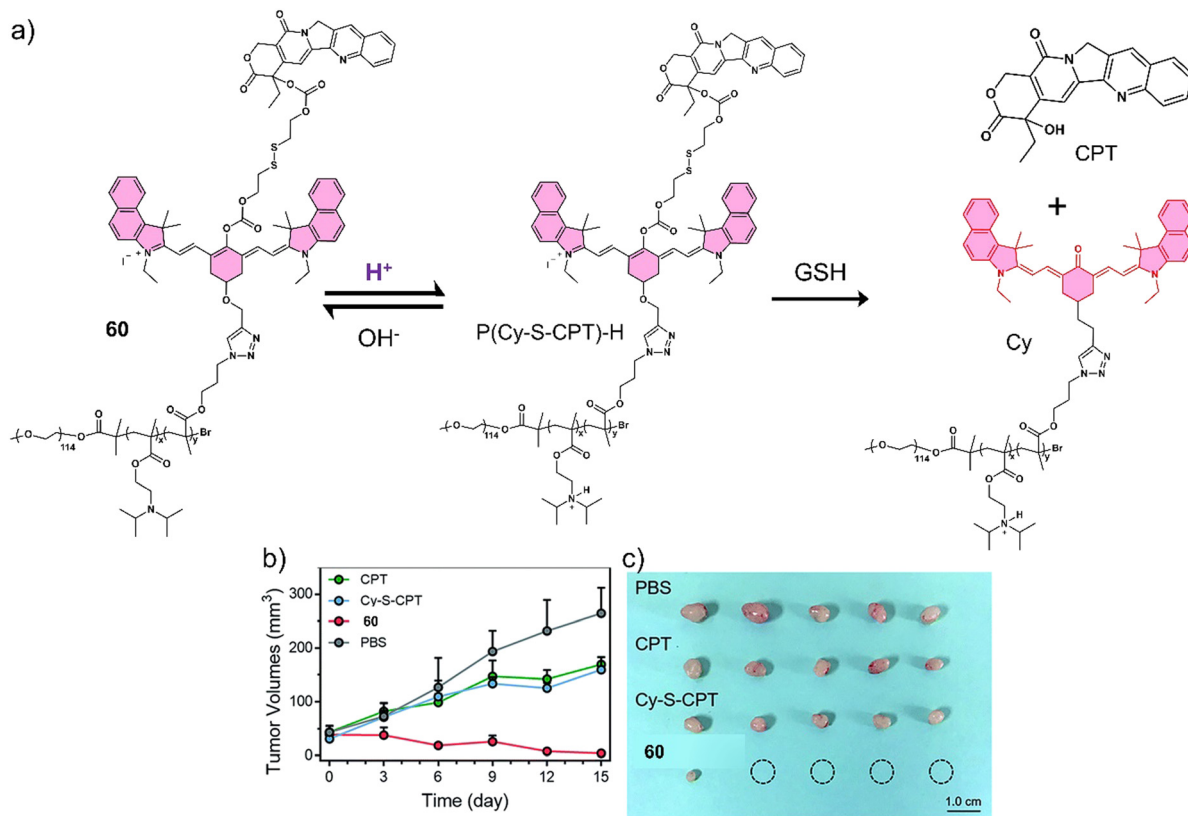


Fig. 38 (a) Chemical structures of **60** and the sequentially-activatable release mechanism of CPT. (b) Tumour volumes over 15 days and treatment with PBS, CPT, Cy-S-CPT, and **60** administered at a CPT-equivalent dose of 10 mg kg^{-1} every 3 days *via* intravenous injections and (c) representative images of excised tumours of the A549 tumour-bearing mice exposed to PBS, CPT, Cy-S-CPT, and **60**. Note: Cy-S-CPT is not functionalised with the diblock copolymer and serves as the control to **60**. Reproduced with permission from ref. 163. Copyright (2018) The Royal Society of Chemistry.

the authors further developed **59** through the X-ray activatable 4-(hydroxymethyl)-2,3,5,6-tetrafluoroaryl azide moiety conjugated to DOX *via* a carbamate linker. **59** exhibited promising results both *in vitro* and *in vivo* ($0, 5, \text{ and } 10 \mu\text{M}$ *in vitro*; 10 mg kg^{-1} *in vivo*) upon X-ray radiation (6, 12, 24, 36, 48, and 60 Gy *in vitro*; 6 Gy *in vivo*) in HeLa cell lines and HeLa tumour-bearing mice models (Fig. 38c). This pioneering work illustrated a unique prodrug design that has potential for clinical translation where the selective activation of antitumour agents can overcome off-target toxicities of currently available chemotherapeutics.

3.6 Prodrugs and phototherapeutics that respond to more than one stimuli

Due to the complexity and high heterogeneity of the human body, it is difficult to selectively target tumours and avoid unwanted toxicity issues with just one method of activation.

Therefore, researchers have been actively developing dual-responsive prodrugs and phototherapeutics to substantially enhance the precision of their activation *in vivo* (Table 6).

Zhu and co-workers reported sequentially activatable prodrug, **60** (Fig. 38a), for the programmable drug release of CPT in A549 lung cancer cells and tumours.¹⁶³ **60** consisted of an acid-sensitive tertiary amine-containing diblock copolymer conjugated to a cyanine fluorophore (Cy). The Cy fluorophore was functionalised with a disulfide linker to allow its conjugation to CPT and to enable the GSH-mediated release of CPT. In neutral aqueous conditions, **60** was shown to self-assemble to form nanomicelles and quench the fluorescence emission of Cy at 830 nm. In an acidic environment ($\text{pH} < 6.0$), the tertiary amine group of **60** became protonated, and subsequently, the nanomicelles disassembled, resulting in an increase in fluorescence intensity at 830 nm. Moreover, the presence of GSH was

Table 6 Summary of multi-stimuli-responsive prodrugs and phototherapeutics discussed in this review

Stimuli	Active anti-cancer therapeutic agent	Treatment	<i>In vitro</i> models	<i>Ex vivo/in vivo</i> models	Ref.
60 Acid pH and GSH	CPT	Chemotherapy	A549 cells	A549 tumour-bearing mice	163
61 Acid pH and caspase-3	DOX	Chemotherapy	U87 cells	—	164
62 Acid pH and thiol	Heptamethine cyanine IR765	PDT	A549 and MCF-7 cells	MCF-7 tumour-bearing mice	166
63 Acid pH and GSH	Cy7	PTT	—	4T1 tumour-bearing mice	167

Note: CPT: camptothecin, DOX: doxorubicin, GSH: glutathione, PDT: photodynamic therapy, —: not mentioned.



shown to hydrolyse the disulfide bond and release CPT, which resulted in a simultaneous change in fluorescence emission intensity to 650 nm. **60** exhibited a dose-dependent cytotoxicity in A549 cells, while no remarkable cytotoxicity was observed in normal QSG-7701 cells, which are typically known to have low intracellular concentrations of GSH. A549 cells, which were pre-incubated with NEM (*N*-ethylmaleimide, GSH inhibitor) and NaHCO₃, separately, showed a significant attenuation of the antiproliferative activity of **60**. These results revealed that the release of CPT could only be achieved at low pH and in the presence of a high concentration of GSH, both of which are present in cancer lesions. Furthermore, an *in vivo* fluorescence imaging (IVIS Spectrum CT imaging system) study demonstrated that **60** predominantly accumulated at the tumour site in A549 tumour-bearing mice. Moreover, **60** significantly reduced the tumour volume compared to control groups (PBS, CPT, and Cy-S-CPT) (Fig. 38b and c), with no loss in the body weight. This work highlights the attractive nature of employing a sequential activation strategy to enhance antitumour activity and alleviate any potential adverse effects *in vivo*.

Owing to the low pH of the TME and the presence of caspase-3 during cytotoxic drug-induced apoptosis, Li *et al.* developed **61** designed to release DOX and confirm the induction of apoptosis (Fig. 39).¹⁶⁴ **61** was functionalised with $\alpha_v\beta_3$ integrin (overexpressed in cancer cells)-targeting group Arg-Gly-Asp (RGD) peptide, a fluorophore 5(6)-carboxylfluorescein (FAM), a fluorescent quencher 4-(dimethylamino)benzene-4-carboxylic acid (Dabcyl), and a caspase-3-cleavable peptide Asp-Glu-Val-Asp (DEVD), all of which were linked to DOX *via* an acid-labile hydrazone bond. Owing to the targeting nature of

RGD tripeptide, **61** was shown to successfully target human glioblastoma U87 cells that overexpress $\alpha_v\beta_3$ integrin when compared to African green monkey kidney COS-7 cells. **61** exhibited significant antiproliferative activity with IC₅₀ values of 4.3 μ M in $\alpha_v\beta_3$ integrin-positive U87 cells. Initially, the fluorescence of both DOX and FAM were quenched by Dabcyl *via* Förster resonance energy transfer (FRET).¹⁶⁵ However, in acidic environments, the hydrazone bond was cleaved, and the fluorescence emission of DOX at 590 nm was restored. Confocal laser scanning microscopy (CLSM) was used to monitor the release of DOX in U87 cells in real-time. The release of DOX-induced apoptosis resulted in the overexpression of apoptotic biomarker caspase-3 and the hydrolysis of DEVD sequence. Which restored the fluorescence emission of FAM in U87 cells monitored using CLSM. Overall, this strategy enabled monitoring of both DOX release and the induction of apoptosis.

Cai and co-workers reported the dual-responsive (acid- and thiol-responsive) activatable phototheranostic, **62**, for tumour-targeted NIR/PA imaging and PDT *in vitro* and *in vivo* (Fig. 40).¹⁶⁶ **62** was developed by covalently linking fluorescent dye 5'-carboxyrhodamine (Rho) with NIR fluorescent photosensitizer heptamethine cyanine IR765 (Cy) using a disulfide linker. With increasing concentrations of GSH (500 μ M), the fluorescence emission intensity at 580 nm increased, which was ascribed to the cleavage of the disulfide linker and release of Rho. Decreasing the pH of the solution led to the protonation of the amino unit on Cy and an increase in fluorescence emission intensity at 765 nm. Fluorescence imaging indicated the preferential cellular uptake of **62** in cancerous A549 and MCF-7 cells rather than in normal 293-T (human embryonic kidney) cells. Concentration-dependent

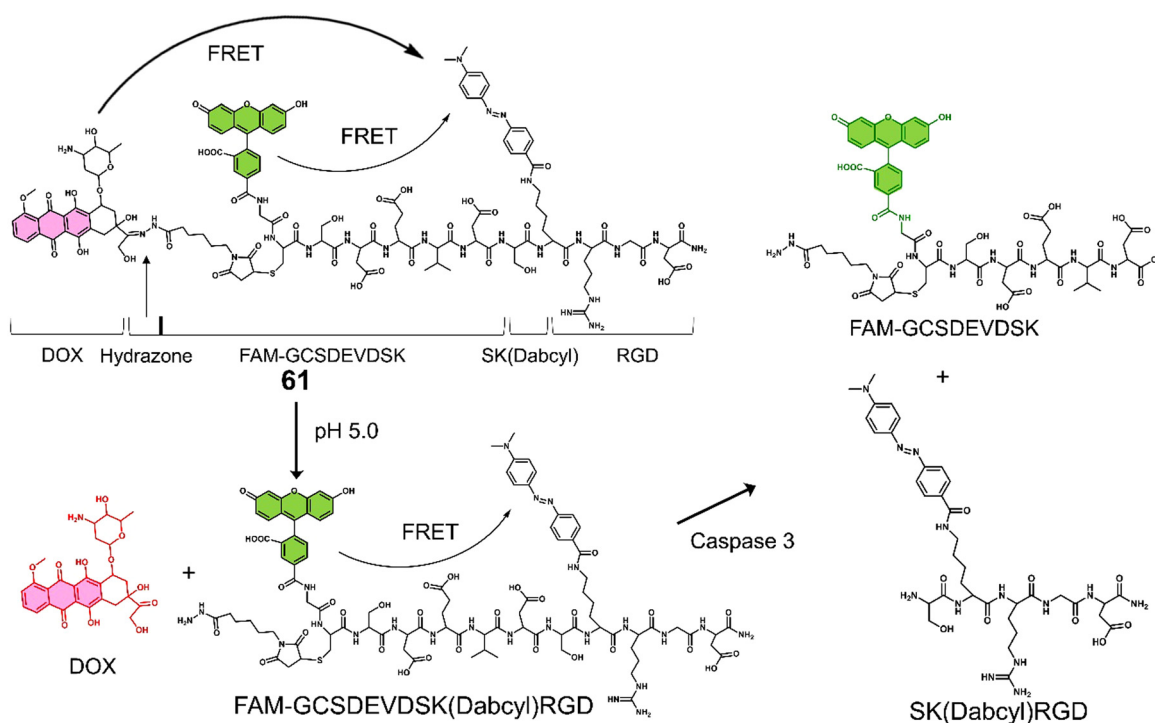


Fig. 39 Chemical structure of **61** and the low pH/caspase-3 sequentially-activated release of DOX.



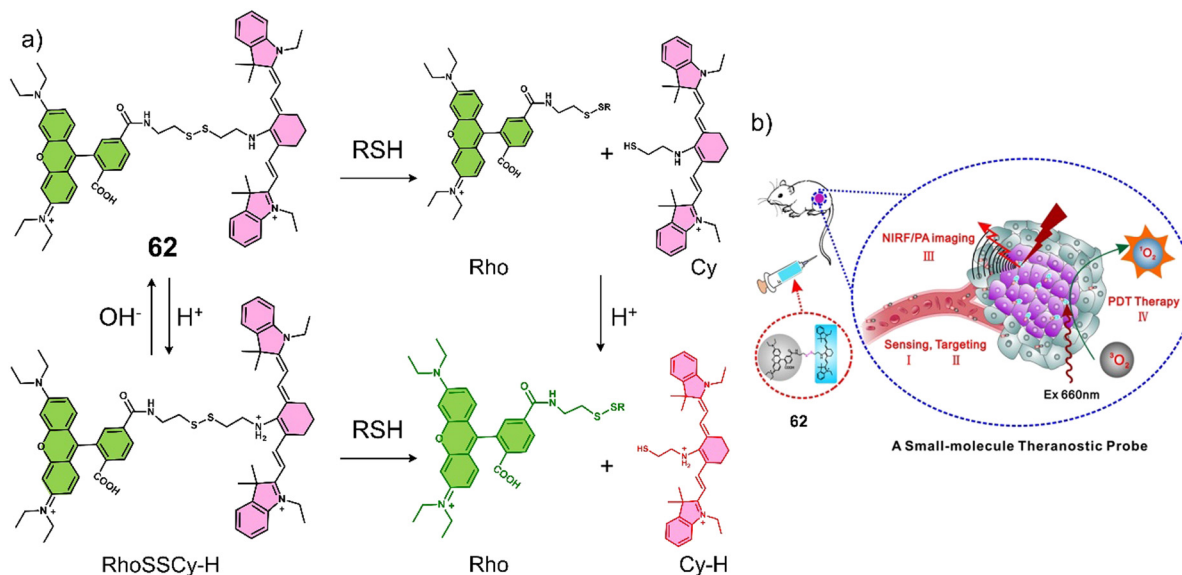


Fig. 40 (a) The chemical structure of **62** and its dual low pH and GSH-mediated release of Rho and Cy-H. (b) Illustrated action of the mechanism of **62** in tumour of the mice. Reproduced with permission from ref. 166. Copyright (2017) Published by Ivyspring International Publisher.

phototoxicity was observed for MCF-7 cells under 660 nm laser irradiation (30 mW cm^{-2} , for 5 min), and minimal dark toxicity was observed. NIR fluorescence and PA imaging exhibited good tumour specificity as well as activation by thiols and acidic pH.

Maximum imaging intensity values were seen at 15 h after tail vein injection, and this identified time point was used for light irradiation. **62** (150 μL , 150 μM , tail vein) with laser irradiation (660 nm, 100 mW cm^{-2} , 30 min) resulted in a significant

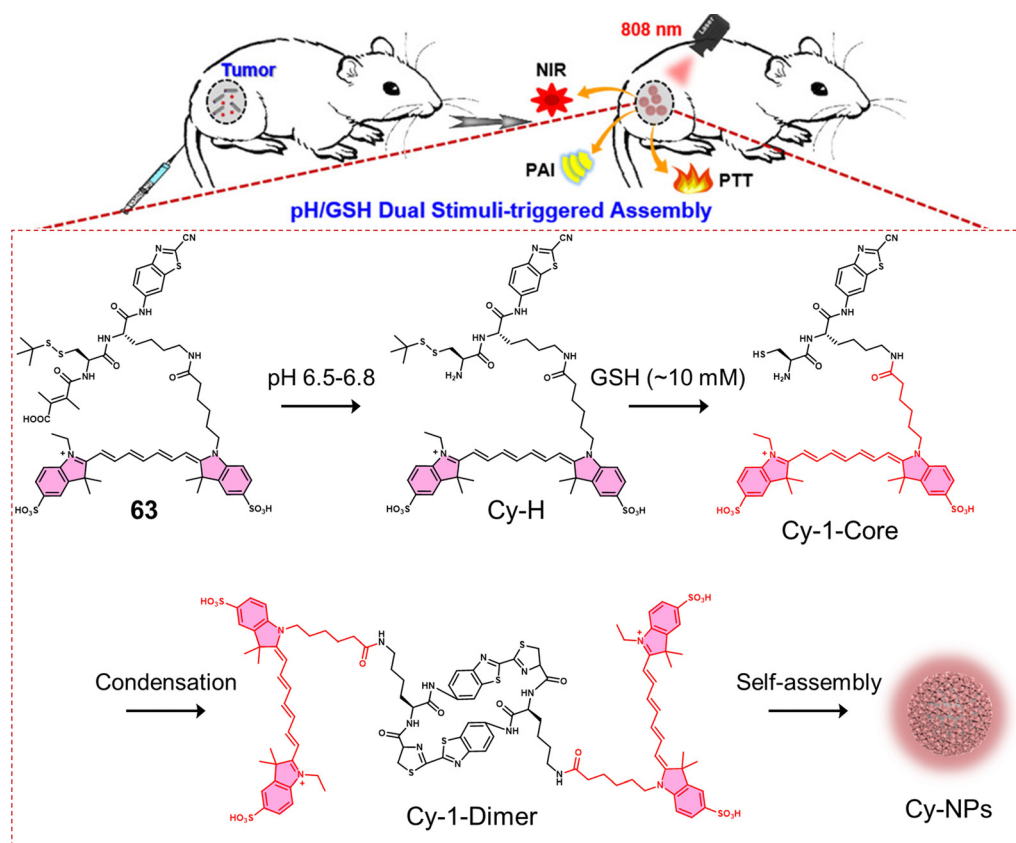


Fig. 41 Chemical structures of **63** and the low pH/high GSH sequential activated molecular assembly for enrichment in and eradication of tumour cells. Reproduced with permission from ref. 167. Copyright (2020) American Chemical Society.



reduction in tumour volume, with no obvious side effects being observed by haematoxylin and eosin (H&E) staining. This confirmed a good biosafety profile with no off-target dark toxicities. As such, this work represents the first thiol/pH dual-responsive OFF-ON PDT agent with multimodal imaging and tumour-targeting properties.

Given that a low pH and high GSH concentration are two of the main biomarkers observed in tumour regions, Wang *et al.* developed a dual-activatable PTT agent, **63**, that was activated in the presence of both to provide precise NIR/PA imaging and PTT (Fig. 41).¹⁶⁷ **63** used a citraconic group as the acidic pH-responsive moiety, thiol-protected cysteine (Cys) was used as the GSH-responsive unit and Cy7 as the fluorescent/PA reporter and PTT agent. In an acidic environment (pH 6.5) and the presence of GSH conditions, the citraconic group in **63** was hydrolysed, and a GSH-induced disulfide reduction occurred simultaneously. The free thiol intermediate underwent an intermolecular condensation to form a cyclic dimer, which then self-assembled into homogeneous nanoparticles. **63** exhibited enhanced NIR fluorescence and PA signals in the tumour site of 4T1 tumour-bearing mice after intravenous injection, attributed to the enhanced permeability and retention (EPR) effect¹⁶⁸ of the nanoparticles, and achieved maximum values at 4 h and 12 h post-injection, respectively. **63** exhibited excellent photothermal efficacy when activated by low intratumoural pH and high levels of GSH upon 808 nm NIR laser irradiation (1 W cm⁻², 10 min), resulting in a rapid local temperature increase of 17 °C. This then led to tumour ablation in more than 60% of mice after 30 days of treatment without causing any acute or systemic toxicity as determined by measuring the body weights and by H&E staining.

4. Conclusion & perspective

In this tutorial review, we focused our attention on prodrugs that can be activated by specific biomarkers found in the tumour microenvironment or by external stimuli, as well as theranostics for the precision-enhanced chemo- and phototherapy of cancer. Tumour microenvironment biomarkers are typically acidic pH, hypoxia, an oxidative and reductive environment, and different enzymes that could be more abundant in cancer cells and readily activate responsive moieties present in prodrugs. The direct chemical modification of clinically approved drugs helps significantly reduce the excessive time burden involved in the drug development process and provides the potential to improve drug efficacy and overcome any adverse side effects. As seen throughout the review, activatable chemotherapeutics as well as phototherapeutic systems (*i.e.*, PDT and PTT) provide improved selectivity for cancer cells and enable the ability to define tumour margins.

Although significant progress has been made, the clinical translation of these agents remains challenging. For example, in terms of pH activation, current systems require relatively acidic environments (pH ≤ 5) to be effective *in vivo*. Therefore, since the tumour microenvironment is a mildly acidic

environment (*e.g.*, pH 6.5–7.2), prodrugs with improved pH sensitivity are required. Furthermore, the refinement of the chemical structures is essential to identify optimal pharmacokinetics and pharmacodynamics. This optimization is needed to minimise off-target toxicity and improve tumour specificity. As such, we anticipate that dual or multi-activatable prodrugs are prime targets to provide enhanced tumour specificity for targeted cancer therapy. We are confident that through sustained research effort that in the near future improved activatable chemotherapeutics and activatable “smart” phototherapeutics will be developed for clinical use as effective agents for cancer treatment.

Abbreviations

ADC	Antibody-drug conjugate
ADEPT	Antibody-directed enzyme prodrug therapy
Az	Acetazolamide
BFA	Bafilomycin A1
BODIPY	Boron-dipyrromethenes
BTD	Benzothiadiazole
CAIX	Carbonic anhydrase IX
CAM	Chorioallantoic membrane
CBT	Conjugated backbone twist
CBT	2-Cyanobenzothiazole
CDF	3,4-Difluorobenzyl curcumin
CE	Carboxylesterase enzymes
CLSM	Confocal laser scanning microscopy
CM	7-(Diethylamino)coumarin
CPT	Camptothecin
CSCs	Cancer stem cells
Cy	Cyanine
Cys	Cysteine
Dabcy1	4-(Dimethylaminoazo)benzene-4-carboxylic acid
DCA	Dichloroacetic acid
DDS	Drug delivery systems
DEVD	Asp-Glu-Val-Asp
DOX	Doxorubicin
ESI-MS	Electrospray ionization mass spectroscopy
FAM	5(6)-Carboxylfluorescein
FDU	5-Fluorodeoxyuridine
FRET	Förster resonance energy transfer
FRs	Folate receptors
fs-TA	Femtosecond transient absorption
GFLG	Gly-Phe-Leu-Gly
GGT	γ-Glutamyltranspeptidase
GSH	Glutathione
H&E	Haematoxylin and eosin
H ⁺	Hydrogen ion
H ₂ O ₂	Hydrogen peroxide
HAdase	Hyaluronidase
HCC	Hepatocellular carcinoma
HPLC	High-performance liquid chromatography
HRMS	High-resolution mass spectroscopy



ISC	Intersystem crossing
LAPTM4B	Lysosomal protein transmembrane 4 beta
LC-MS	Liquid chromatography-mass spectrometry
MMAE	Monomethyl auristatin E
MMPs	Matrix metalloproteinases
MRI	Magnetic resonance imaging
MTD	Maximum tolerated doses
NADH	Nicotinamide adenine dinucleotide
NAFLD	Non-alcoholic fatty liver disease
NEM	N-Ethylmaleimide
NHDFs	Normal human dermal fibroblasts
NIR	Near-infrared fluorescence
NTR	Nitroreductase
O ₂ ^{•-}	Superoxide anion radicals
PA	Photoacoustic
PDT	Photodynamic therapy
PEG	Poly(ethylene glycol)
PeT	Photoinduced electron transfer
PPA	Pyropheophorbide α
PPa	Pheophorbide a
PS	Photosensitiser
PTT	Photothermal therapy
Pyro	Pyropheophorbide α
RGD	Arg-Gly-Asp
Rho	5'-Carboxyrhodamines
RP-HPLC	Reverse-phase high-performance liquid chromatography
SA	Sialic acid
SAHA	Suberoylanilide hydroxamic acid
TME	Tumour microenvironment
TPP	Triphenylphosphine
VEGFR	Vascular endothelial growth factor receptor
XIAP	X-linked inhibitor of apoptosis protein
β -Gal	β -Galactosidase
$\Phi^1\text{O}_2$	¹ O ₂ quantum yields
Φ_{ISC}	Intersystem crossing efficiency
¹ O ₂	Singlet oxygen
³ O ₂	Ground state oxygen

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

X.-P. H. thanks the National Natural Science Foundation of China (No. 21788102, 91853201, 82130099 and 9185920077), the Shanghai Municipal Science and Technology Major Project (No. 2018SHZDZX03), the National Science Foundation of Shanghai (No. 21XD1404600, 21JC1406600, and 22140901000), the International Cooperation Program of Shanghai Science and Technology Committee (No. 17520750100), the Fundamental Research Funds for the Central Universities (222201717003) and the Programme of Introducing Talents of Discipline

to Universities (B16017) for financial support. J. L. and Y. Z. thank the National Natural Science Foundation of China (No. 82130099, 82151219, 31871414 and 81971265) and the Shanghai Municipal Science and Technology Major Project (No. 22ZR1415200). H.-H. H. thanks the National Natural Science Foundation of China (No. 22107029) and Project funded by the China Postdoctoral Science Foundation (No. 2020M681196). T. D. J. wishes to thank the Royal Society for a Wolfson Research Merit Award and the Open Research Fund of the School of Chemistry and Chemical Engineering, Henan Normal University for support (2020ZD01). L. W. wishes to thank the China Scholarship Council and the University of Bath for supporting his PhD in the UK. J. S. K. thank the financial support received from the National Research Foundation of Korea (CRI project no. 2018R1A3B1052702, 2019M3E5D1A01068998). M. L. wishes to thank the support of the Brain Pool Program through the National Research Foundation of Korea (NRF) funded by the Ministry of Science and ICT (Grant No. 2020H1D3A1A02080172, M. L.). A. C. S. would like to thank the Glasstone Research fellowship (University of Oxford) and Jesus College, Oxford for support.

References

- J. Fares, M. Y. Fares, H. H. Khachfe, H. A. Salhab and Y. Fares, *Signal Transduction Targeted Ther.*, 2020, **5**, 28.
- D. Hanahan, *Cancer Discov.*, 2022, **12**, 31–46.
- T. N. Seyfried and L. C. Huysentruyt, *Crit. Rev. Oncog.*, 2013, **18**, 43–73.
- R. Haddad, N. Alrabadi, B. Altaani and T. Li, *Polymers*, 2022, **14**, 658.
- K. Wu, N. A. Yee, S. Srinivasan, A. Mahmoodi, M. Zakharian, J. M. Mejia Oneto and M. Royzen, *Chem. Sci.*, 2021, **12**, 1259–1271.
- K. E. Arnst, S. Banerjee, H. Chen, S. Deng, D.-J. Hwang, W. Li and D. D. Miller, *Med. Res. Rev.*, 2019, **39**, 1398–1426.
- D. S. Shewach and R. D. Kuchta, *Chem. Rev.*, 2009, **109**, 2859–2861.
- H. Kitao, M. Iimori, Y. Kataoka, T. Wakasa, E. Tokunaga, H. Saeki, E. Oki and Y. Maehara, *Cancer Sci.*, 2018, **109**, 264–271.
- J. L. Nitiss, *Nat. Rev. Cancer*, 2009, **9**, 338–350.
- F. Kratz, I. A. Müller, C. Ryppa and A. Warnecke, *ChemMedChem*, 2008, **3**, 20–53.
- J. Rautio, N. A. Meanwell, L. Di and M. J. Hageman, *Nat. Rev. Drug Discovery*, 2018, **17**, 559–587.
- V. Abet, F. Filace, J. Recio, J. Alvarez-Builla and C. Burgos, *Eur. J. Med. Chem.*, 2017, **127**, 810–827.
- X. Zhang, X. Li, Q. You and X. Zhang, *Eur. J. Med. Chem.*, 2017, **139**, 542–563.
- J. Rautio, H. Kumpulainen, T. Heimbach, R. Oliyai, D. Oh, T. Järvinen and J. Savolainen, *Nat. Rev. Drug Discovery*, 2008, **7**, 255–270.
- Y. Chen, L. Li, W. Chen, H. Chen and J. Yin, *Chin. Chem. Lett.*, 2019, **30**, 1353–1360.



- 16 M. H. Lee, A. Sharma, M. J. Chang, J. Lee, S. Son, J. L. Sessler, C. Kang and J. S. Kim, *Chem. Soc. Rev.*, 2018, **47**, 28–52.
- 17 X. Li, S. Kolemen, J. Yoon and E. U. Akkaya, *Adv. Funct. Mater.*, 2017, **27**, 1604053.
- 18 J. B. Zawilska, J. Wojcieszak and A. B. Olejniczak, *Pharmacol. Rep.*, 2013, **65**, 1–14.
- 19 R. Mahato, W. Tai and K. Cheng, *Adv. Drug Delivery Rev.*, 2011, **63**, 659–670.
- 20 J. Guo, X. Pan, C. Wang and H. Liu, *Bioconjugate Chem.*, 2022, **33**, 993–1010.
- 21 J. M. Silva, E. Silva and R. L. Reis, *J. Controlled Release*, 2019, **298**, 154–176.
- 22 N. S. H. Motlagh, P. Parvin, F. Ghasemi and F. Atyabi, *Biomed. Opt. Express*, 2016, **7**, 2400–2406.
- 23 J. Dey and I. M. Warner, *J. Lumin.*, 1997, **71**, 105–114.
- 24 A. Steinbrueck, A. C. Sedgwick, H.-H. Han, M. Y. Zhao, S. Sen, D.-Y. Huang, Y. Zang, J. Li, X.-P. He and J. L. Sessler, *Chem. Commun.*, 2021, **57**, 5678–5681.
- 25 L. H. Goetz and N. J. Schork, *Fertil. Steril.*, 2018, **109**, 952–963.
- 26 M. Wang, J. Zhao, L. Zhang, F. Wei, Y. Lian, Y. Wu, Z. Gong, S. Zhang, J. Zhou, K. Cao, X. Li, W. Xiong, G. Li, Z. Zeng and C. Guo, *J. Cancer*, 2017, **8**, 761–773.
- 27 Y. Yuan, Y.-C. Jiang, C.-K. Sun and Q.-M. Chen, *Oncol. Rep.*, 2016, **35**, 2499–2515.
- 28 G. Hao, Z. P. Xu and L. Li, *RSC Adv.*, 2018, **8**, 22182–22192.
- 29 H. Alimoradi, S. S. Matikonda, A. B. Gamble, G. I. Giles and K. Greish, *Curr. Pharm. Des.*, 2016, **22**, 2808–2820.
- 30 T. Legigan, J. Clarhaut, I. Tranoy-Opalinski, A. Monvoisin, B. Renoux, M. Thomas, A. Le Pape, S. Lerondel and S. Papot, *Angew. Chem., Int. Ed.*, 2012, **51**, 11606–11610.
- 31 A. Sharma, M.-G. Lee, M. Won, S. Koo, J. F. Arambula, J. L. Sessler, S.-G. Chi and J. S. Kim, *J. Am. Chem. Soc.*, 2019, **141**, 15611–15618.
- 32 C. Lin, H. He, Y. Zhang, M. Xu, F. Tian, L. Li and Y. Wang, *RSC Adv.*, 2020, **10**, 3084–3091.
- 33 Servier Medical Art, <https://smart.servier.com/>, accessed February 2022.
- 34 Z. Zhou, J. Song, L. Nie and X. Chen, *Chem. Soc. Rev.*, 2016, **45**, 6597–6626.
- 35 P. Agostinis, K. Berg, K. A. Cengel, T. H. Foster, A. W. Girotti, S. O. Gollnick, S. M. Hahn, M. R. Hamblin, A. Juzeniene, D. Kessel, M. Korbelik, J. Moan, P. Mroz, D. Nowis, J. Piette, B. C. Wilson and J. Golab, *Ca-Cancer J. Clin.*, 2011, **61**, 250–281.
- 36 X. Li, J. F. Lovell, J. Yoon and X. Chen, *Nat. Rev. Clin. Oncol.*, 2020, **17**, 657–674.
- 37 M. Li, T. Xiong, J. Du, R. Tian, M. Xiao, L. Guo, S. Long, J. Fan, W. Sun, K. Shao, X. Song, J. W. Foley and X. Peng, *J. Am. Chem. Soc.*, 2019, **141**, 2695–2702.
- 38 M. Li, J. Xia, R. Tian, J. Wang, J. Fan, J. Du, S. Long, X. Song, J. W. Foley and X. Peng, *J. Am. Chem. Soc.*, 2018, **140**, 14851–14859.
- 39 M. Li, S. Long, Y. Kang, L. Guo, J. Wang, J. Fan, J. Du and X. Peng, *J. Am. Chem. Soc.*, 2018, **140**, 15820–15826.
- 40 J. Xie, Y. Wang, W. Choi, P. Jangili, Y. Ge, Y. Xu, J. Kang, L. Liu, B. Zhang, Z. Xie, J. He, N. Xie, G. Nie, H. Zhang and J. S. Kim, *Chem. Soc. Rev.*, 2021, **50**, 9152–9201.
- 41 M. Li, K. H. Gebremedhin, D. Ma, Z. Pu, T. Xiong, Y. Xu, J. S. Kim and X. Peng, *J. Am. Chem. Soc.*, 2022, **144**, 163–173.
- 42 M. Li, Y. Shao, J. H. Kim, Z. Pu, X. Zhao, H. Huang, T. Xiong, Y. Kang, G. Li, K. Shao, J. Fan, J. W. Foley, J. S. Kim and X. Peng, *J. Am. Chem. Soc.*, 2020, **142**, 5380–5388.
- 43 Z.-S. Yang, Y. Yao, A. C. Sedgwick, C. Li, Y. Xia, Y. Wang, L. Kang, H. Su, B.-W. Wang, S. Gao, J. L. Sessler and J.-L. Zhang, *Chem. Sci.*, 2020, **11**, 8204–8213.
- 44 H.-B. Cheng, B. Qiao, H. Li, J. Cao, Y. Luo, K. M. K. Swamy, J. Zhao, Z. Wang, J. Y. Lee, X.-J. Liang and J. Yoon, *J. Am. Chem. Soc.*, 2021, **143**, 2413–2422.
- 45 A. C. Sedgwick, L. Wu, H.-H. Han, S. D. Bull, X.-P. He, T. D. James, J. L. Sessler, B. Z. Tang, H. Tian and J. Yoon, *Chem. Soc. Rev.*, 2018, **47**, 8842–8880.
- 46 P. Sarbadhikary, B. P. George and H. Abrahamse, *Theranostics*, 2021, **11**, 9054–9088.
- 47 D. Jaque, L. Martínez Maestro, B. del Rosal, P. Haro-Gonzalez, A. Benayas, J. L. Plaza, E. Martín Rodríguez and J. García Solé, *Nanoscale*, 2014, **6**, 9494–9530.
- 48 H. S. Jung, P. Verwilst, A. Sharma, J. Shin, J. L. Sessler and J. S. Kim, *Chem. Soc. Rev.*, 2018, **47**, 2280–2297.
- 49 M. R. K. Ali, M. A. Rahman, Y. Wu, T. Han, X. Peng, M. A. Mackey, D. Wang, H. J. Shin, G. Chen Zhuo, H. Xiao, R. Wu, Y. Tang, D. M. Shin and M. A. El-Sayed, *Proc. Natl. Acad. Sci. U. S. A.*, 2017, **114**, E3110–E3118.
- 50 Y. Xu, Y. Wang, J. An, A. C. Sedgwick, M. Li, J. Xie, W. Hu, J. Kang, S. Sen, A. Steinbrueck, B. Zhang, L. Qiao, S. Wageh, J. F. Arambula, L. Liu, H. Zhang, J. L. Sessler and J. S. Kim, *Bioact. Mater.*, 2022, **14**, 76–85.
- 51 Y. Liu, P. Bhattarai, Z. Dai and X. Chen, *Chem. Soc. Rev.*, 2019, **48**, 2053–2108.
- 52 Y. Ren, A. C. Sedgwick, J. Chen, G. Thiabaud, C. V. Chau, J. An, J. F. Arambula, X.-P. He, J. S. Kim, J. L. Sessler and C. Liu, *J. Am. Chem. Soc.*, 2020, **142**, 16156–16160.
- 53 J. Chen, A. C. Sedgwick, S. Sen, Y. Ren, Q. Sun, C. Chau, J. F. Arambula, T. Sarma, L. Song, J. L. Sessler and C. Liu, *Chem. Sci.*, 2021, **12**, 9916–9921.
- 54 H.-H. Han, H. Tian, Y. Zang, A. C. Sedgwick, J. Li, J. L. Sessler, X.-P. He and T. D. James, *Chem. Soc. Rev.*, 2021, **50**, 9391–9429.
- 55 P. S. Low, S. Singhal and M. Srinivasarao, *Curr. Opin. Chem. Biol.*, 2018, **45**, 64–72.
- 56 W.-T. Dou, H.-H. Han, A. C. Sedgwick, G.-B. Zhu, Y. Zang, X.-R. Yang, J. Yoon, T. D. James, J. Li and X.-P. He, *Sci. Bull.*, 2022, **67**, 853–878.
- 57 M. Srinivasarao and P. S. Low, *Chem. Rev.*, 2017, **117**, 12133–12164.
- 58 M. Vellard, *Curr. Opin. Biotechnol.*, 2003, **14**, 444–450.
- 59 M. H. Baig, M. Adil, R. Khan, S. Dhadi, K. Ahmad, G. Rabbani, T. Bashir, M. A. Imran, F. M. Husain, E. J. Lee, M. A. Kamal and I. Choi, *Semin. Cancer Biol.*, 2019, **56**, 1–11.



- 60 A. Sreedhar and Y. Zhao, *Biomed. Rep.*, 2018, **8**, 3–10.
- 61 S. E. Pratt, S. Durland-Busbice, R. L. Shepard, K. Heinz-Taheny, P. W. Iversen and A. H. Dantzig, *Clin. Cancer Res.*, 2013, **19**, 1159–1168.
- 62 L. Her and H.-J. Zhu, *Drug Metab. Dispos.*, 2020, **48**, 230–244.
- 63 H. Dong, L. Pang, H. Cong, Y. Shen and B. Yu, *Drug Delivery*, 2019, **26**, 416–432.
- 64 J. Zhuang, N. Li, Y. Zhang, B. Li, H. Wen, X. Zhang, T. Zhang, N. Zhao and B. Z. Tang, *CCS Chem.*, 2021, **4**, 1028–1043.
- 65 C. Perez, K. B. Daniel and S. M. Cohen, *ChemMedChem*, 2013, **8**, 1662–1667.
- 66 A. Sharma, M.-G. Lee, H. Shi, M. Won, J. F. Arambula, J. L. Sessler, J. Y. Lee, S.-G. Chi and J. S. Kim, *Chem*, 2018, **4**, 2370–2383.
- 67 C. F. Thorn, C. Oshiro, S. Marsh, T. Hernandez-Boussard, H. McLeod, T. E. Klein and R. B. Altman, *Pharmacogenet. Genomics*, 2011, **21**, 440–446.
- 68 C. Carvalho, R. X. Santos, S. Cardoso, S. Correia, P. J. Oliveira, M. S. Santos and P. I. Moreira, *Curr. Med. Chem.*, 2009, **16**, 3267–3285.
- 69 E. D. Michelakis, L. Webster and J. R. Mackey, *Br. J. Cancer*, 2008, **99**, 989–994.
- 70 W. Li, H. Zhang, Y. G. Assaraf, K. Zhao, X. Xu, J. Xie, D.-H. Yang and Z.-S. Chen, *Drug Resistance Updates*, 2016, **27**, 14–29.
- 71 X. Lin, Z. Xiao, T. Chen, S. H. Liang and H. Guo, *Front. Oncol.*, 2020, **10**, 317.
- 72 C. López-Otín and J. S. Bond, *J. Biol. Chem.*, 2008, **283**, 30433–30437.
- 73 K. G. Ponder and L. H. Boise, *Cell Death Discovery*, 2019, **5**, 56.
- 74 S. W. Chung, J. U. Choi, Y. S. Cho, H. R. Kim, T. H. Won, P. Dimitrion, O.-C. Jeon, S. W. Kim, I.-S. Kim, S. Y. Kim and Y. Byun, *Adv. Sci.*, 2018, **5**, 1800368.
- 75 S. Bae, P. M. Siu, S. Choudhury, Q. Ke, J. H. Choi, Y. Y. Koh and P. M. Kang, *Am. J. Physiol. Heart Circ. Physiol.*, 2010, **299**, H1374–H1381.
- 76 Y. Li, T. Mei, S. Han, T. Han, Y. Sun, H. Zhang and F. An, *Chin. Chem. Lett.*, 2020, **31**, 3027–3040.
- 77 Y. Yuan, C.-J. Zhang, M. Gao, R. Zhang, B. Z. Tang and B. Liu, *Angew. Chem., Int. Ed.*, 2015, **54**, 1780–1786.
- 78 A. C. Sedgwick, K.-C. Yan, D. N. Mangel, Y. Shang, A. Steinbrueck, H.-H. Han, J. T. Brewster, X.-L. Hu, D. W. Snelson, V. M. Lynch, H. Tian, X.-P. He and J. L. Sessler, *J. Am. Chem. Soc.*, 2021, **143**, 1278–1283.
- 79 H. Zhang, Z. Zhao, A. T. Turley, L. Wang, P. R. McGonigal, Y. Tu, Y. Li, Z. Wang, R. T. K. Kwok, J. W. Y. Lam and B. Z. Tang, *Adv. Mater.*, 2020, **32**, 2001457.
- 80 A. Corti, M. Franzini, A. Paolicchi and A. Pompella, *Anti-cancer Res.*, 2010, **30**, 1169–1181.
- 81 M. H. Hanigan, H. F. Frierson, P. E. Swanson and B. R. De Young, *Hum. Pathol.*, 1999, **30**, 300–305.
- 82 B. He, X. Sui, B. Yu, S. Wang, Y. Shen and H. Cong, *Drug Delivery*, 2020, **27**, 1474–1490.
- 83 F. Zhou, S. Yang, C. Zhao, W. Liu, X. Yao, H. Yu, X. Sun and Y. Liu, *Theranostics*, 2021, **11**, 7045–7056.
- 84 M. Chiba, Y. Ichikawa, M. Kamiya, T. Komatsu, T. Ueno, K. Hanaoka, T. Nagano, N. Lange and Y. Urano, *Angew. Chem., Int. Ed.*, 2017, **56**, 10418–10422.
- 85 M. W. Kryman, G. A. Schamerhorn, J. E. Hill, B. D. Calitree, K. S. Davies, M. K. Linder, T. Y. Ohulchanskyy and M. R. Detty, *Organometallics*, 2014, **33**, 2628–2640.
- 86 J. Garbe and M. Collin, *J. Innate Immun.*, 2012, **4**, 121–131.
- 87 E. Waidely, A.-R. O. Al-Yuobi, A. S. Bashammakh, M. S. El-Shahawi and R. M. Leblanc, *Analyst*, 2016, **141**, 36–44.
- 88 X. Chai, H.-H. Han, A. C. Sedgwick, N. Li, Y. Zang, T. D. James, J. Zhang, X.-L. Hu, Y. Yu, Y. Li, Y. Wang, J. Li, X.-P. He and H. Tian, *J. Am. Chem. Soc.*, 2020, **142**, 18005–18013.
- 89 X. Zhen, J. Zhang, J. Huang, C. Xie, Q. Miao and K. Pu, *Angew. Chem., Int. Ed.*, 2018, **57**, 7804–7808.
- 90 A. Abawi, X. Wang, J. Bompard, A. Bérot, V. Andretto, L. Gudimard, C. Devillard, E. Petiot, B. Joseph, G. Lollo, T. Granjon, A. Girard-Egrot and O. Maniti, *Int. J. Mol. Sci.*, 2021, **22**, 4103.
- 91 D. Trachootham, W. Lu, M. A. Ogasawara, N. R.-D. Valle and P. Huang, *Antioxid. Redox Signaling*, 2008, **10**, 1343–1374.
- 92 X. Guo, Y. Cheng, X. Zhao, Y. Luo, J. Chen and W.-E. Yuan, *J. Nanobiotechnol.*, 2018, **16**, 74.
- 93 S. Reuter, S. C. Gupta, M. M. Chaturvedi and B. B. Aggarwal, *Free Radical Biol. Med.*, 2010, **49**, 1603–1616.
- 94 C. Lennicke, J. Rahn, R. Lichtenfels, L. A. Wessjohann and B. Seliger, *Cell Commun. Signaling*, 2015, **13**, 39.
- 95 H. Sies, *Redox Biol.*, 2017, **11**, 613–619.
- 96 E.-J. Kim, S. Bhuniya, H. Lee, H. M. Kim, C. Cheong, S. Maiti, K. S. Hong and J. S. Kim, *J. Am. Chem. Soc.*, 2014, **136**, 13888–13894.
- 97 T. Meng, J. Han, P. Zhang, J. Hu, J. Fu and J. Yin, *Chem. Sci.*, 2019, **10**, 7156–7162.
- 98 X. Xie, X. Yang, T. Wu, Y. Li, M. Li, Q. Tan, X. Wang and B. Tang, *Anal. Chem.*, 2016, **88**, 8019–8025.
- 99 A. Bansal and M. C. Simon, *J. Cell Biol.*, 2018, **217**, 2291–2298.
- 100 L. Kennedy, J. K. Sandhu, M.-E. Harper and M. Cuperlovic-Culf, *Biomolecules*, 2020, **10**, 1429.
- 101 F. Kong, Z. Liang, D. Luan, X. Liu, K. Xu and B. Tang, *Anal. Chem.*, 2016, **88**, 6450–6456.
- 102 R. An, X. Cheng, S. Wei, Y. Hu, Y. Sun, Z. Huang, H.-Y. Chen and D. Ye, *Angew. Chem., Int. Ed.*, 2020, **59**, 20636–20644.
- 103 X. Wang, W. Lin, W. Zhang, C. Li, T. Sun, G. Chen and Z. Xie, *J. Colloid Interface Sci.*, 2019, **536**, 208–214.
- 104 M. Y. Lucero and J. Chan, *Nat. Chem.*, 2021, **13**, 1248–1256.
- 105 A. C. Sedgwick, J. E. Gardiner, G. Kim, M. Yevglevskis, M. D. Lloyd, A. T. A. Jenkins, S. D. Bull, J. Yoon and T. D. James, *Chem. Commun.*, 2018, **54**, 4786–4789.
- 106 W. Plunkett, P. Huang, Y. Z. Xu, V. Heinemann, R. Grunewald and V. Gandhi, *Semin. Oncol.*, 1995, **22**, 3–10.
- 107 N. Papadopoulos, K. W. Kinzler and B. Vogelstein, *Nat. Biotechnol.*, 2006, **24**, 985–995.
- 108 J. Sun, K. Du, J. Diao, X. Cai, F. Feng and S. Wang, *Angew. Chem., Int. Ed.*, 2020, **59**, 12122–12128.



- 109 M. Weber, H.-H. Han, B.-H. Li, M. L. Odyneec, C. E. F. Jarman, Y. Zang, S. D. Bull, A. B. Mackenzie, A. C. Sedgwick, J. Li, X.-P. He and T. D. James, *Chem. Sci.*, 2020, **11**, 8567–8571.
- 110 H.-H. Han, A. C. Sedgwick, Y. Shang, N. Li, T. Liu, B.-H. Li, K. Yu, Y. Zang, J. T. Brewster, M. L. Odyneec, M. Weber, S. D. Bull, J. Li, J. L. Sessler, T. D. James, X.-P. He and H. Tian, *Chem. Sci.*, 2020, **11**, 1107–1113.
- 111 M. Miller, A. Mellul, M. Braun, D. Sherill-Rofe, E. Cohen, Z. Shpilt, I. Unterman, O. Braitbard, J. Hochman, E. Y. Tshuva and Y. Tabach, *iScience*, 2020, **23**, 101262.
- 112 T. C. Johnstone, K. Suntharalingam and S. J. Lippard, *Chem. Rev.*, 2016, **116**, 3436–3486.
- 113 J. L. Sessler, T. Murai, V. Lynch and M. Cyr, *J. Am. Chem. Soc.*, 1988, **110**, 5586–5588.
- 114 G. Thiabaud, R. McCall, G. He, J. F. Arambula, Z. H. Siddik and J. L. Sessler, *Angew. Chem., Int. Ed.*, 2016, **55**, 12626–12631.
- 115 G. Thiabaud, G. He, S. Sen, K. A. Shelton, W. B. Baze, L. Segura, J. Alaniz, R. M. Macias, G. Lyness, A. B. Watts, H. M. Kim, H. Lee, M. Y. Cho, K. S. Hong, R. Finch, Z. H. Siddik, J. F. Arambula and J. L. Sessler, *Proc. Natl. Acad. Sci. U. S. A.*, 2020, **117**, 7021–7029.
- 116 C. J. Adams and T. J. Meade, *Chem. Sci.*, 2020, **11**, 2524–2530.
- 117 A. E. Nejad, S. Najafgholian, A. Rostami, A. Sistani, S. Shojaeifar, M. Esparvarinha, R. Nedaeinia, S. H. Javanmard, M. Taherian, M. Ahmadlou, R. Salehi, B. Sadeghi and M. Manian, *Cancer Cell Int.*, 2021, **21**, 62.
- 118 J.-N. Liu, W. Bu and J. Shi, *Chem. Rev.*, 2017, **117**, 6160–6224.
- 119 W. Liu, H. Liu, X. Peng, G. Zhou, D. Liu, S. Li, J. Zhang and S. Wang, *Bioconjugate Chem.*, 2018, **29**, 3332–3343.
- 120 X. Zhao, S. Long, M. Li, J. Cao, Y. Li, L. Guo, W. Sun, J. Du, J. Fan and X. Peng, *J. Am. Chem. Soc.*, 2020, **142**, 1510–1517.
- 121 L. Yang, P. Shi, G. Zhao, J. Xu, W. Peng, J. Zhang, G. Zhang, X. Wang, Z. Dong, F. Chen and H. Cui, *Signal Transduction Targeted Ther.*, 2020, **5**, 8.
- 122 J. H. Kim, P. Verwilst, M. Won, J. Lee, J. L. Sessler, J. Han and J. S. Kim, *J. Am. Chem. Soc.*, 2021, **143**, 14115–14124.
- 123 F. E. Lock, P. C. McDonald, Y. Lou, I. Serrano, S. C. Chafe, C. Ostlund, S. Aparicio, J.-Y. Winum, C. T. Supuran and S. Dedhar, *Oncogene*, 2013, **32**, 5210–5219.
- 124 A. Skwarska, E. D. D. Calder, D. Sneddon, H. Bolland, M. L. Odyneec, I. N. Mistry, J. Martin, L. K. Folkes, S. J. Conway and E. M. Hammond, *Cell Chem. Biol.*, 2021, **28**, 1258–1270.E13.
- 125 Z. Zhang, H. Yamashita, T. Toyama, H. Sugiura, Y. Ando, K. Mita, M. Hamaguchi, Y. Hara, S. Kobayashi and H. Iwase, *Breast Cancer Res. Treat.*, 2005, **94**, 11–16.
- 126 T. Sudo, K. Mimori, N. Nishida, R. Kogo, T. Iwaya, F. Tanaka, K. Shibata, H. Fujita, K. Shirouzu and M. Mori, *Oncol. Rep.*, 2011, **26**, 777–782.
- 127 W. Feng, B. Zhang, D. Cai and X. Zou, *Cancer Lett.*, 2014, **347**, 183–190.
- 128 T. C. S. Ho, A. H. Y. Chan and A. Ganesan, *J. Med. Chem.*, 2020, **63**, 12460–12484.
- 129 X. Meng, J. Zhang, Z. Sun, L. Zhou, G. Deng, S. Li, W. Li, P. Gong and L. Cai, *Theranostics*, 2018, **8**, 6025–6034.
- 130 Y. Zhou, M. Maiti, A. Sharma, M. Won, L. Yu, L. X. Miao, J. Shin, A. Podder, K. N. Bobba, J. Han, S. Bhuniya and J. S. Kim, *J. Controlled Release*, 2018, **288**, 14–22.
- 131 W. Piao, K. Hanaoka, T. Fujisawa, S. Takeuchi, T. Komatsu, T. Ueno, T. Terai, T. Tahara, T. Nagano and Y. Urano, *J. Am. Chem. Soc.*, 2017, **139**, 13713–13719.
- 132 C. Wang, S. Wang, Y. Wang, H. Wu, K. Bao, R. Sheng and X. Li, *Sci. Rep.*, 2020, **10**, 12127.
- 133 N. Piasentin, E. Milotti and R. Chignola, *Sci. Rep.*, 2020, **10**, 13613.
- 134 B. Lin, H. Chen, D. Liang, W. Lin, X. Qi, H. Liu and X. Deng, *ACS Appl. Mater. Interfaces*, 2019, **11**, 11157–11166.
- 135 E. Boedtkjer and S. F. Pedersen, *Annu. Rev. Physiol.*, 2020, **82**, 103–126.
- 136 J.-T. Hou, W. X. Ren, K. Li, J. Seo, A. Sharma, X.-Q. Yu and J. S. Kim, *Chem. Soc. Rev.*, 2017, **46**, 2076–2090.
- 137 Y. Jin, Y. Huang, H. Yang, G. Liu and R. Zhao, *Chem. Commun.*, 2015, **51**, 14454–14457.
- 138 L. Wang, Y. Meng and Q.-Y. Zhang, *BMC Cancer*, 2019, **19**, 293.
- 139 S.-Y. Li, L.-H. Liu, H.-Z. Jia, W.-X. Qiu, L. Rong, H. Cheng and X.-Z. Zhang, *Chem. Commun.*, 2014, **50**, 11852–11855.
- 140 M. H. Lee, E.-J. Kim, H. Lee, S. Y. Park, K. S. Hong, J. S. Kim and J. L. Sessler, *Chem. Commun.*, 2016, **52**, 10551–10554.
- 141 S. Siriwibool, N. Kaekratoke, K. Chansaenpak, K. Siwawannapong, P. Panajapo, K. Sagarik, P. Noisa, R.-Y. Lai and A. Kamkaew, *Sci. Rep.*, 2020, **10**, 1283.
- 142 W. Hu, T. He, H. Zhao, H. Tao, R. Chen, L. Jin, J. Li, Q. Fan, W. Huang, A. Baev and P. N. Prasad, *Angew. Chem., Int. Ed.*, 2019, **58**, 11105–11111.
- 143 X. Meng, W. Li, Z. Sun, J. Zhang, L. Zhou, G. Deng, P. Gong and L. Cai, *J. Mater. Chem. B*, 2017, **5**, 9405–9411.
- 144 S. T. Teoh, M. P. Ogrodzinski, C. Ross, K. W. Hunter and S. Y. Lunt, *Front. Oncol.*, 2018, **8**, 174.
- 145 X. Wu, M. Yu, B. Lin, H. Xing, J. Han and S. Han, *Chem. Sci.*, 2015, **6**, 798–803.
- 146 Z. Li, Y. Song, Y. Yang, L. Yang, X. Huang, J. Han and S. Han, *Chem. Sci.*, 2012, **3**, 2941–2948.
- 147 T. Yoshimori, A. Yamamoto, Y. Moriyama, M. Futai and Y. Tashiro, *J. Biol. Chem.*, 1991, **266**, 17707–17712.
- 148 L. Josa-Culleré and A. Llebaria, *ChemPhotoChem*, 2021, **5**, 296–314.
- 149 D. C. F. Monteiro, E. Amoah, C. Rogers and A. R. Pearson, *Acta Crystallogr., Sect. D: Struct. Biol.*, 2021, **77**, 1218–1232.
- 150 P. Dunkel and J. Ilaš, *Cancers*, 2021, **13**, 3237.
- 151 S. Bonnet, *Dalton Trans.*, 2018, **47**, 10330–10343.
- 152 M. M. Dcona, K. Mitra and M. C. T. Hartman, *RSC Med. Chem.*, 2020, **11**, 982–1002.
- 153 B. M. Vickerman, E. M. Zywoot, T. K. Tarrant and D. S. Lawrence, *Nat. Rev. Chem.*, 2021, **5**, 816–834.
- 154 Y. Hou, Z. Zhou, K. Huang, H. Yang and G. Han, *ChemPhotoChem*, 2018, **2**, 1005–1011.
- 155 C. Brieke, F. Rohrbach, A. Gottschalk, G. Mayer and A. Heckel, *Angew. Chem., Int. Ed.*, 2012, **51**, 8446–8476.
- 156 P. Klán, T. Šolomek, C. G. Bochet, A. Blanc, R. Givens, M. Rubina, V. Popik, A. Kostikov and J. Wirz, *Chem. Rev.*, 2013, **113**, 119–191.



- 157 J. A. Peterson, C. Wijesooriya, E. J. Gehrman, K. M. Mahoney, P. P. Goswami, T. R. Albright, A. Syed, A. S. Dutton, E. A. Smith and A. H. Winter, *J. Am. Chem. Soc.*, 2018, **140**, 7343–7346.
- 158 N. P. Toupin, K. Arora, P. Shrestha, J. A. Peterson, L. J. Fischer, E. Rajagurubandara, I. Podgorski, A. H. Winter and J. J. Kodanko, *ACS Chem. Biol.*, 2019, **14**, 2833–2840.
- 159 Z. Wang, N. Wang, S.-C. Cheng, K. Xu, Z. Deng, S. Chen, Z. Xu, K. Xie, M.-K. Tse, P. Shi, H. Hirao, C.-C. Ko and G. Zhu, *Chem*, 2019, **5**, 3151–3165.
- 160 X. Feng, D. Jiang, T. Kang, J. Yao, Y. Jing, T. Jiang, J. Feng, Q. Zhu, Q. Song, N. Dong, X. Gao and J. Chen, *ACS Appl. Mater. Interfaces*, 2016, **8**, 17817–17832.
- 161 J. Geng, Y. Zhang, Q. Gao, K. Neumann, H. Dong, H. Porter, M. Potter, H. Ren, D. Argyle and M. Bradley, *Nat. Chem.*, 2021, **13**, 805–810.
- 162 B. Rini and M. Y. Al-Marrawi, *Expert Opin. Pharmacother.*, 2011, **12**, 1171–1189.
- 163 C. Yan, Z. Guo, Y. Liu, P. Shi, H. Tian and W.-H. Zhu, *Chem. Sci.*, 2018, **9**, 6176–6182.
- 164 S.-Y. Li, L.-H. Liu, L. Rong, W.-X. Qiu, H.-Z. Jia, B. Li, F. Li and X.-Z. Zhang, *Adv. Funct. Mater.*, 2015, **25**, 7317–7326.
- 165 L. Wu, C. Huang, B. P. Emery, A. C. Sedgwick, S. D. Bull, X.-P. He, H. Tian, J. Yoon, J. L. Sessler and T. D. James, *Chem. Soc. Rev.*, 2020, **49**, 5110–5139.
- 166 X. Meng, Y. Yang, L. Zhou, L. Zhang, Y. Lv, S. Li, Y. Wu, M. Zheng, W. Li, G. Gao, G. Deng, T. Jiang, D. Ni, P. Gong and L. Cai, *Theranostics*, 2017, **7**, 1781–1794.
- 167 A. Wang, Q. Mao, M. Zhao, S. Ye, J. Fang, C. Cui, Y. Zhao, Y. Zhang, Y. Zhang, F. Zhou and H. Shi, *Anal. Chem.*, 2020, **92**, 16113–16121.
- 168 H. Maeda, *Bioconjugate Chem.*, 2010, **21**, 797–802.

