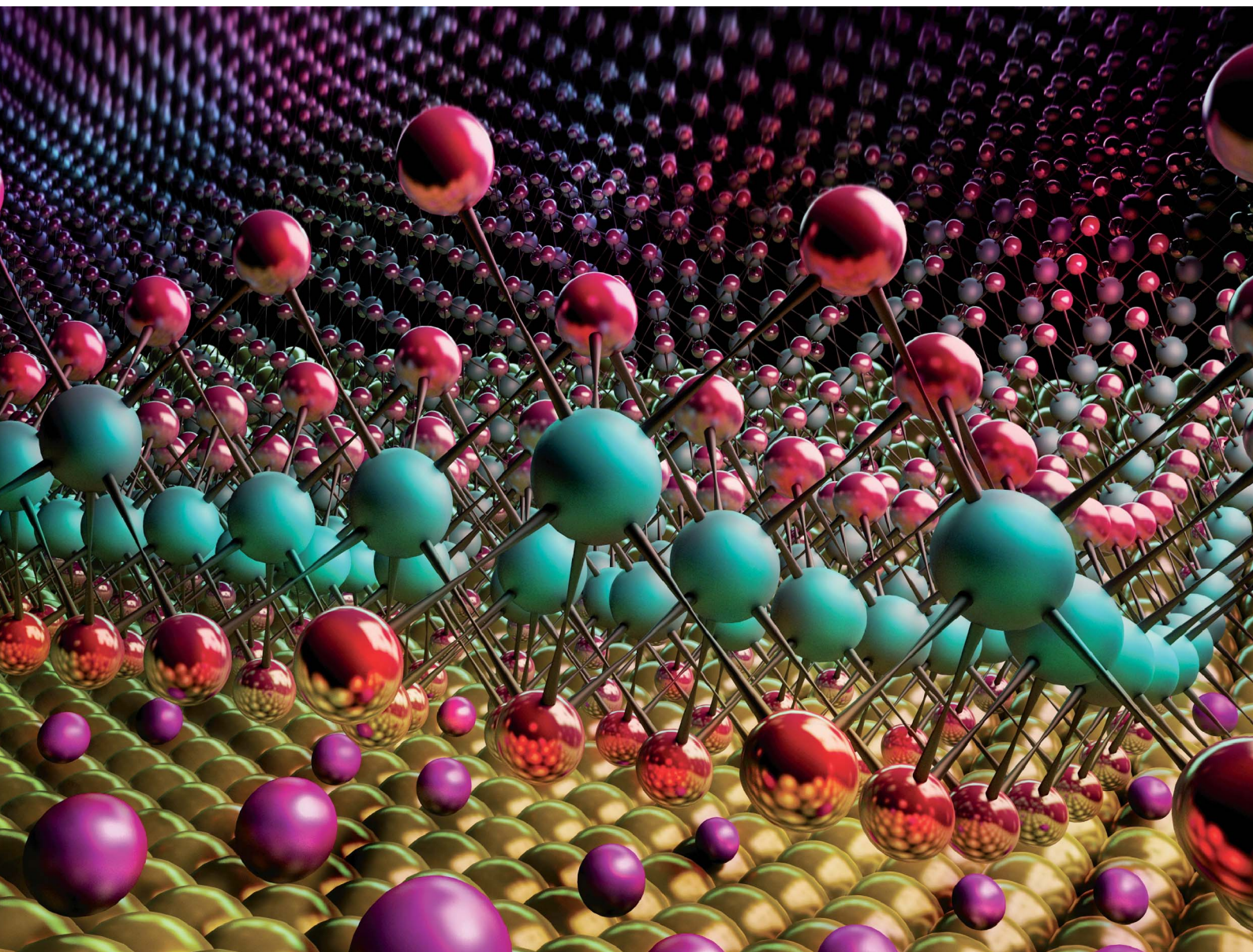


Nanoscale Advances

Volume 6
Number 1
7 January 2024
Pages 1–304

rsc.li/nanoscale-advances



ISSN 2516-0230

PAPER

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Cite this: *Nanoscale Adv.*, 2024, 6, 92

Structural and electronic properties of MoS₂ and MoSe₂ monolayers grown by chemical vapor deposition on Au(111)[†]

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The exceptional electronic and photonic properties of the monolayers of transition metal dichalcogenides including the spin-orbit splitting of the valence and conduction bands at the *K* points of the Brillouin zone make them promising for novel applications in electronics, photonics and optoelectronics. Scalable growth of these materials and understanding of their interaction with the substrate is crucial for these applications. Here we report the growth of MoS₂ and MoSe₂ monolayers on Au(111) by chemical vapor deposition at ambient pressure as well as the analysis of their structural and electronic properties down to the atomic scale. To this aim, we apply ultrahigh vacuum surface sensitive techniques including scanning tunneling microscopy and spectroscopy, low-energy electron diffraction, X-ray and angle-resolved ultraviolet photoelectron spectroscopy in combination with Raman spectroscopy at ambient conditions. We demonstrate the growth of high-quality epitaxial single crystalline MoS₂ and MoSe₂ monolayers on Au(111) and show the impact of annealing on the monolayer/substrate interaction. Thus, as-grown and moderately annealed (<100 °C) MoSe₂ monolayers are decoupled from the substrate by excess Se atoms, whereas annealing at higher temperatures (>250 °C) results in their strong coupling with the substrate caused by desorption of the excess Se. The MoS₂ monolayers are strongly coupled to the substrate and the interaction remains almost unchanged even after annealing up to 450 °C.

Received 29th June 2023
Accepted 24th October 2023

DOI: 10.1039/d3na00475a

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1 Introduction

In the last decade, monolayers of transition metal dichalcogenides (TMDs) such as MoS₂ and MoSe₂ have come into the focus of research. These two-dimensional (2D) materials possess unique properties including valley polarization^{1,2} or high photoluminescence yield^{3,4} making them ideal candidates for electronic,^{5–10} optoelectronic,^{9,11,12} photonic,^{13,14} memory,^{10,15} and sensing¹⁶ devices. For such applications, the metal-semiconductor junction and thus also the interaction between TMDs and the substrate plays an essential role. Different fabrication methods have been developed to prepare high-quality TMD monolayers including chemical vapor deposition (CVD),^{7,17,18} molecular beam epitaxy,¹⁹ or mechanical exfoliation.²⁰ Among these techniques, the CVD method is considered as the most viable route for scalable growth of TMDs.²¹ In this respect gold single crystal substrates enable even their epitaxial growth.^{1,22–34}

The properties of the TMD monolayers can be influenced by the interaction with the substrate, which is reflected in moiré structures or band gap renormalization.^{22,24,27,29} To reduce this interaction, *e.g.*, germanium can be used as the intercalant for WS₂ monolayers on Au(100).³⁴

Here we report technologically relevant ambient pressure epitaxial CVD growth of high crystalline quality MoS₂ and MoSe₂ monolayers on Au(111) substrates. We characterize these *ex situ* grown samples down to the nanoscale with complementary microscopy and spectroscopy techniques under ultrahigh vacuum (UHV) conditions. Such monolayer systems enable their structural investigation with atomic resolution as well as deep insights into their interaction with gold, a commonly used material for contacting the monolayers in devices.³⁵ Thus, MoS₂ monolayers show a moiré structure in scanning tunneling microscopy (STM) and distortion-corrected low-energy electron diffraction (LEED) measurements, which is measured in our study with very high accuracy and indicates a strong electronic coupling with Au(111) substrate. After annealing the sample at 450 °C in UHV this strong interaction with the gold substrate preserves, as confirmed by both X-ray photoelectron spectroscopy (XPS) and LEED. Interestingly, for MoSe₂ monolayers we find a weak electronic coupling, that is interaction, with the substrate after the CVD growth, which, as we find, results from

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[†] Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d3na00475a>



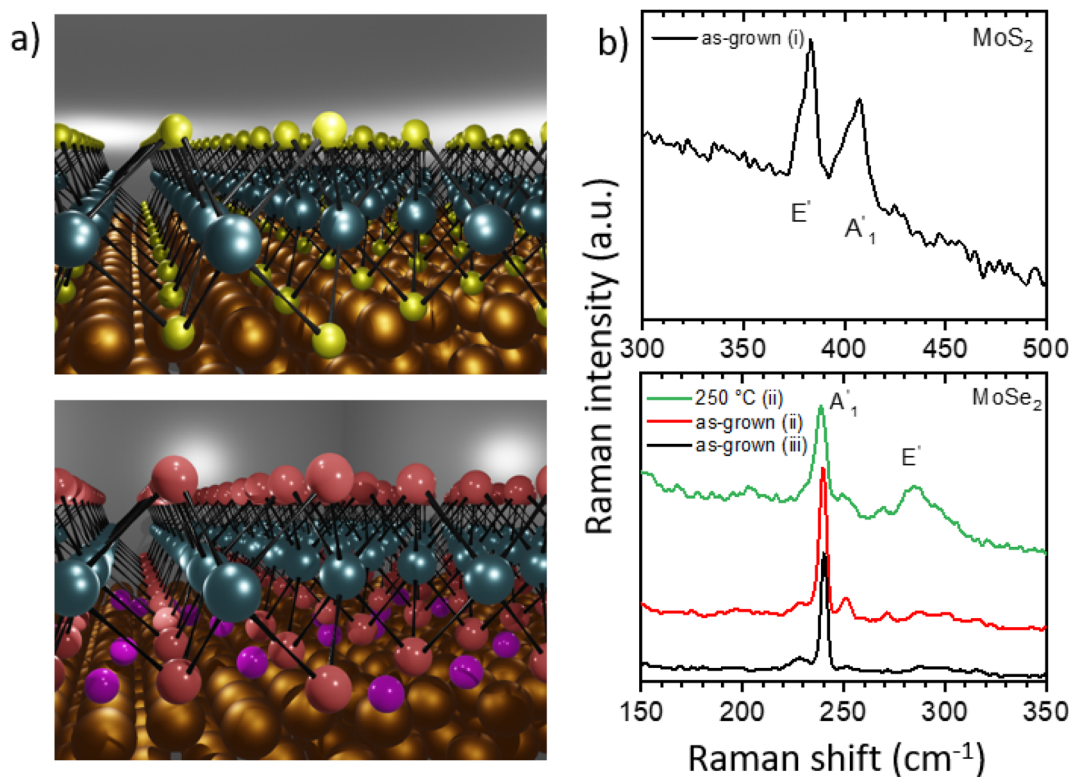


Fig. 1 (a) Schematic representation of the epitaxial growth of MoS₂ (top) and MoSe₂ (bottom) monolayers on Au(111). Mo, S, Se, and Au atoms are highlighted in cyan, yellow, red, and orange, respectively. For the MoSe₂ monolayer, additional Se atoms are shown in purple, which vanish after annealing to 250 °C. (b) Raman spectra of as-grown MoS₂ on Au(111) (top, sample (i)), MoSe₂ on Au(111) (bottom, samples (ii), (iii)) and MoSe₂ on Au(111) annealed at 250 °C (bottom, sample (iii)). The most pronounced Raman modes are marked.

the intercalation of excess Se atoms (see schematic representation in Fig. 1a). However, after annealing at 250 °C in UHV, changes in the valence band structure measured by angle-resolved ultraviolet photoelectron spectroscopy (ARUPS) and in the core level spectra reveal an increased interaction between the MoSe₂ monolayer and the Au(111) substrate resulting from desorption of the excess selenium. These for the nanotechnology interesting findings are analyzed and discussed in detail in the following sections.

2 Results and discussion

2.1 Basic characterization: Raman spectroscopy and XPS

MoS₂ and MoSe₂ monolayers were grown on gold single crystals with (111) orientation by chemical vapor deposition (CVD, see details in the Experimental section), Fig. 1a. We characterize and discuss three different samples: (i) MoS₂ grown on Au(111), and (ii) MoSe₂ grown on Au(111) using the standard CVD conditions as well as (iii) MoSe₂ grown on Au(111) using a higher amount of selenium. In the following, we refer to these sample assignments, which are also listed in Table 1 together with some spectroscopy data.

Raman spectroscopy at ambient conditions confirms the 2H structure of the TMD monolayers, Fig. 1b. For the as-grown MoS₂ sample (i) and MoSe₂ samples (ii), (iii), we measured the most pronounced Raman modes at $386 \pm 2 \text{ cm}^{-1}$ (E', in-plane)

and $408 \pm 2 \text{ cm}^{-1}$ (A'₁, out-of-plane) as well as at $240 \pm 2 \text{ cm}^{-1}$ (A'₁), respectively. Since the E' mode has a larger intensity than the A'₁ mode and the A'₁ mode shows an asymmetric shape,^{28,36} we conclude the formation of a MoS₂ monolayer strongly interacting with the Au(111) substrate.³⁷ For the MoSe₂ sample (ii) before annealing and sample (iii), we found a sharp A'₁ mode (FWHM = $4 \pm 1 \text{ cm}^{-1}$) without a noticeable E' mode which indicates a weak interaction of the monolayers with the substrate.³⁷ In contrast, the MoSe₂ sample (ii) annealed at 250 °C in UHV shows a broader A'₁ mode (FWHM = $6 \pm 1 \text{ cm}^{-1}$) and a clearly recognizable E' mode. The broadening of the A'₁ peak and the appearance of the E' peak indicate an increased interaction with the gold substrate.³⁷ In optical microscopy images (Fig. S1†), we see the formation of an almost complete continuous monolayer for MoS₂ while 2D crystals in the range from several to tens of micrometers are observed for MoSe₂.

Next, we introduced the samples into the UHV systems and continued the analysis using surface sensitive techniques. To identify the samples' chemical composition, XPS was employed. In the as-grown XP spectra of MoS₂ (see Fig. 2a), we identify one intensive and narrow doublet in the Mo 3d region at binding energies (BEs) of 229.3 eV (Mo 3d_{5/2}) and 232.5 eV (Mo 3d_{3/2}) as well as one doublet in the S 2p region at 162.3 eV (S 2p_{3/2}) and 163.5 eV (S 2p_{1/2}) (see Table 1). These findings confirm the formation of a MoS₂ monolayer. The quantitative analysis of these peaks reveals an atomic ratio between S and Mo of $(2.6 \pm$



Table 1 Summary of the photoelectron spectroscopy data for MoS₂ sample (i), MoSe₂ sample (ii) and MoSe₂ sample (iii) after annealing at various temperatures

Sample/phase	BE of Mo 3d _{5/2} peak, eV	Intensity of Mo 3d, %	BE of S 2p _{3/2} , Se 3d _{5/2} peak, eV	BE of VBM, eV	Valence band splitting, meV
MoS ₂ (i) 100 °C	229.3(1)	100	162.3(1)	1.40(1)	120(10)
MoSe ₂ (ii) 100 °C phase (I)	228.6(1)	57(2)	54.2(1)	0.63(1)	180(10)
MoSe ₂ (ii) 100 °C phase (II)	229.2(1)	43(2)	54.7(1)	0.99(2)	150(15)
MoSe ₂ (ii) 250 °C phase (I)	228.7(1)	16(2)	54.2(1)	0.58(2)	190(15)
MoSe ₂ (ii) 250 °C phase (II)	229.0(1)	84(2)	54.6(1)	0.99(1)	150(10)
MoSe ₂ (iii) 100 °C phase (I)	228.6(1)	70(2)	54.1(1)	0.60(1)	170(10)
MoSe ₂ (iii) 100 °C phase (II)	229.0(1)	30(2)	54.7(1)	1.00(2)	150(15)

0.2) : 1, which is slightly larger than for the ideal stoichiometry of 2 : 1. A MoS₂ sample with a lower S to Mo ratio of (2.1 ± 0.2) : 1 shows similar XP features (see Fig. S2†). The additional doublets in the Mo 3d region are attributed to MoO₂ (light blue) and MoO₃ residuals (dark green) present in the sample.³⁸ After annealing at 450 °C, no significant differences in the relevant XP spectra in Fig. 2a are observed except for the reduction of adsorbed carbon-based airborne contaminations from the surface (not shown).

In contrast, the XP spectra for MoSe₂ sample (ii) show a distinctively different behavior, Fig. 2b. On the one hand, the Mo 3d and Se 3d spectra are much more complex. On the other hand, we see temperature-dependent changes in the relevant XP signals. For the as-grown and moderately annealed (100 °C) sample (ii), we can identify two doublets (MoSe₂ (I), (II)) at BE = 228.6 eV (Mo 3d_{5/2}) and 231.8 eV (Mo 3d_{3/2}) as well as 229.1 eV

(Mo 3d_{5/2}) and 232.3 eV (Mo 3d_{3/2}) related to two different MoSe₂ monolayer phases. In the following, we name them MoSe₂ phase (I) and MoSe₂ phase (II), respectively. The most intense component (phase (I)) is localized at lower BEs. Since the intensity of the MoSe₂ phase (I) is approximately 33% larger than for phase (II), the second phase is not visible in the Raman spectrum (see red curve in Fig. 1b). Also, in the Se 3d spectra we can distinguish two different doublets at BE = 54.2 eV (Se 3d_{5/2}) and 55.1 eV (Se 3d_{3/2}) as well as at 54.8 eV (Se 3d_{5/2}) and 55.7 eV (Se 3d_{3/2}) corresponding to the two different MoSe₂ monolayer phases which are also observed in the Mo 3d region. After annealing of sample (ii) at 250 °C, the intensity of the Mo 3d peaks of phase (II) is now significantly higher (~5 times) than for phase (I), allowing the second phase with distinct E' mode to be clearly seen in Raman spectroscopy (see green curve in Fig. 1b). BEs of the characteristic Mo and chalcogen XPS peaks

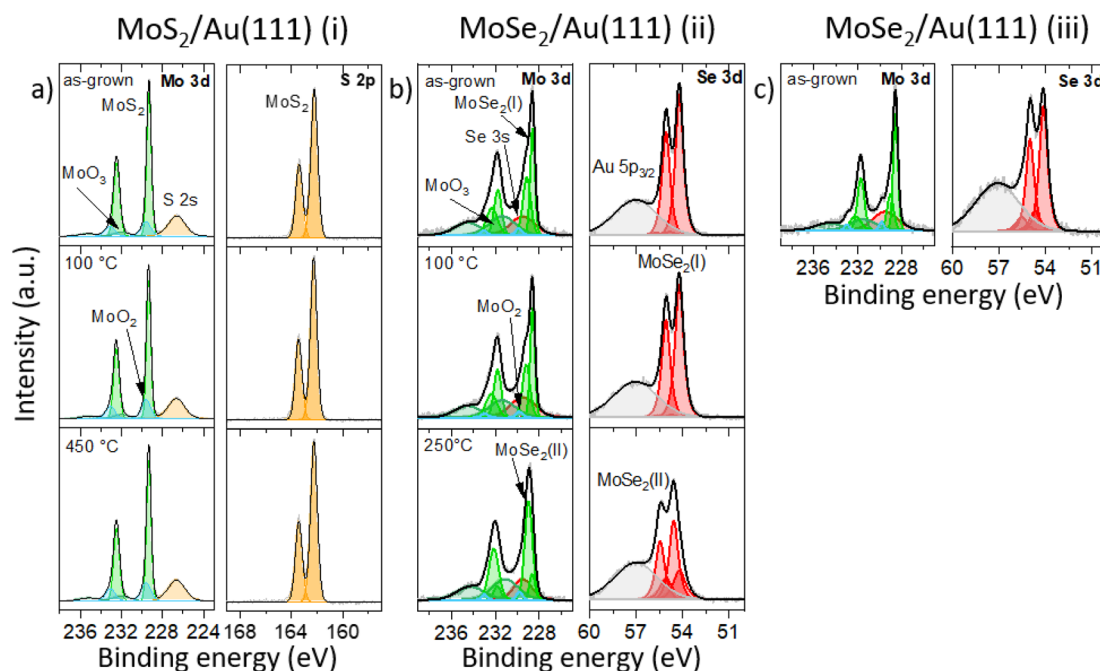


Fig. 2 High-resolution XPS spectra of (a) MoS₂ sample (i), (b) MoSe₂ sample (ii) and (c) MoSe₂ sample (iii) on Au(111). Mo 3d, S 2s and S 2p spectra of the MoS₂ sample (i) as-grown and after different annealing steps in UHV. Mo 3d, Se 3s, Se 3d and Au 5p_{3/2} spectra of the MoSe₂ sample (ii) as-grown and after different annealing steps as well as of the MoSe₂ sample (iii) as-grown. In the Mo 3d spectra the peaks related to the TMD monolayers, MoO₂, and MoO₃ are highlighted in green, light blue or dark green, respectively.



are summarized in Table 1. The Se : Mo ratio is calculated to be $(2.0 \pm 0.3) : 1$ for the as-grown samples and samples moderately annealed at 100 °C, and $(1.9 \pm 0.3) : 1$ for the sample annealed at 250 °C (ii). Note that due to the overlapping of the Se 3d peaks with the Au 5p_{3/2} peak a precise quantification of the Se amount is challenging. From Raman spectroscopy and XPS results we conclude that MoSe₂ phase (I) and MoSe₂ phase (II) can be assigned to MoSe₂ monolayer areas which are weakly or strongly interact with the substrate, respectively.

For the MoSe₂ sample (iii) annealed at 100 °C (Fig. 2c), we detect an increased Se : Mo ratio of $(2.7 \pm 0.3) : 1$ in comparison to the MoSe₂ sample (ii) annealed at 100 °C (*cf.* Fig. 2b). Here, the amount of excess Se can be quantified to approximately 35%. The relevant XP spectra (Fig. 2c) look qualitatively similar to the spectra of the as-grown and 100 °C annealed sample (ii) with less Se (see Fig. 2b). This appearance means that the MoSe₂ phase (I) is dominant in comparison to phase (II) which agrees with the Raman spectroscopy data (*cf.* black and red curve in Fig. 1b). We will refer to these results in the structural and electronic characterization parts again. As we have already described for MoS₂, the additional peaks in the Mo 3d spectra arise from MoO₂ (light blue) and MoO₃ (dark green) precursor residuals.³⁸

2.2 Structural characterization: STM and LEED

Before we start discussing the structural results, we would like to point out that all samples discussed in this section were annealed at 100 °C in UHV (unless otherwise described) to remove contaminations. First, we focus on the structure of MoS₂ on Au(111) (sample (i)). In our measurements, the moiré structure^{22,24,26–29} is observed in the STM image, the fast Fourier transform (FFT) of the STM image, and the LEED pattern (Fig. 3a). This moiré structure arises from the lattice mismatch between Au(111) (2.88 Å) and MoS₂ (3.15 Å) with a zero twist angle between the lattices.^{22,39} In the large area STM images, defects are slightly visible as black dots which originate probably from single sulfur vacancies.⁴⁰ The close-up STM image in Fig. 3a shows the topmost S layer²² with atomic resolution. Using a quantitative LEED analysis, we determine the lattice parameters of this sample precisely (see Table 2; *e.g.*, lattice constant of 3.15 ± 0.01 Å) by fitting all visible LEED spots and taking multiple scattering up to the 2nd order of the substrate lattice into account. The simulation of the fitted structure is shown in the LEED pattern with yellow circles, Fig. 3b. Also, the moiré structure (light blue unit cell in Fig. 3b) was fitted with inclusion of the same visible spots, whereby the lattice constants were determined to be 33.3 ± 0.3 Å and 33.9 ± 0.3 Å, respectively, which are in agreement with the results from our quantitative FFT analysis of the LT STM image (Table 2) and ref. 24. After annealing at 450 °C, the structure visible in STM (not shown) and the LEED pattern do not change (Fig. S3a and d†).

Next, we characterized the MoSe₂ sample (ii) on Au(111) with a lower Se amount according to XPS. Here, we also measured an atomically resolved STM image and a LEED pattern with the typical moiré pattern,^{23–25} whereby the moiré contrast in the STM image is less pronounced as compared to MoS₂ (Fig. 3b).

Similar as for the MoS₂ sample, in the STM image defects are also slightly visible as black dots. From the quantitative LEED analysis including all visible spots, we estimate the moiré lattice constant to be 23.0 ± 0.1 Å (Table 2). The moiré unit cell is shown in magenta (FFT of STM image and LEED pattern in Fig. 3b). Furthermore, from the same LEED spots, we precisely extract the lattice parameters of MoSe₂ (green unit cell) by considering multiple scattering in a similar manner as for the MoS₂ sample (i). The lattice parameters are shown in Table 2, matching very well the lattice parameters of freestanding MoSe₂.³⁹ Furthermore, after annealing, all LEED spots become more intense (see Fig. S3b and e†), most likely due to desorption of airborne contaminations, which is also reflected in a reduced intensity of the C 1s peak in XPS (not shown). However, the position of the LEED spots does not change, so we do not observe by LEED and STM two different monolayer phases concluded from the XPS data. A possible explanation is that both phases have the same orientation and a similar lattice making them difficult to distinguish in our structural analysis. For a detailed discussion of both phases, we will refer to the analysis in the ARUPS part below (Section 2.3). Since we have not observed any further LEED structure, we suggest that the excess Se, detected by XPS (see Section 1.1), form a disordered layer which vanish after annealing at elevated temperatures.

For the MoSe₂ sample with a higher Se concentration (sample (iii)), besides the MoSe₂ pattern also another pattern is observed in LEED after a moderate annealing at 100 °C (Fig. 3c), whereas only the moiré structure was faintly visible before annealing (Fig. S3f†). Using STM, it was possible to measure an atomically resolved MoSe₂ layer (Fig. 3c). In the corresponding FFT, we can clearly identify a 6-fold symmetric structure confirming the MoSe₂ structure (green structure in Fig. 3c). This structure overlaps with another structure which is visible as black dots with a nearly quadratic unit cell in the original STM image. To visualize this structure better, we filtered out the MoSe₂ structure from the STM image (Fig. S4†). Moreover, we were able to measure another structure at positions which were not covered with MoSe₂ (Fig. S5†). We attribute this structure to the Se₈ superstructure reported on Au(111).⁴¹ By comparison of the STM images in Fig. S5† and in Fig. 3c, we conclude that the MoSe₂ monolayer is on top of a Se₈ interlayer. This conclusion is supported by the quantitative FFT analysis of sample (iii) shown in Fig. 3c (blue unit cell). The obtained lattice parameters (Table 2) match the parameters of the Se₈ superstructure on Au(111)⁴¹ very well. Note that the values of this superstructure were distortion-corrected by using the known MoSe₂ structure which is also visible in the FFT (green lattice).

At the end of this section we come back to the LEED pattern in Fig. 3c. Besides the MoSe₂ structure (green LEED structure) another structure was observed. We found a model describing almost all other LEED spots by considering multiple scattering, mirror and rotational domains (blue LEED structure). These spots originate from the Se₈ superstructure. The lattice parameters fitted to the LEED pattern, shown in Table 2, are different in comparison to the Se₈ superstructure observed in the FFT of the STM image. One possible explanation could be that some temperature-dependent structural changes of the Se



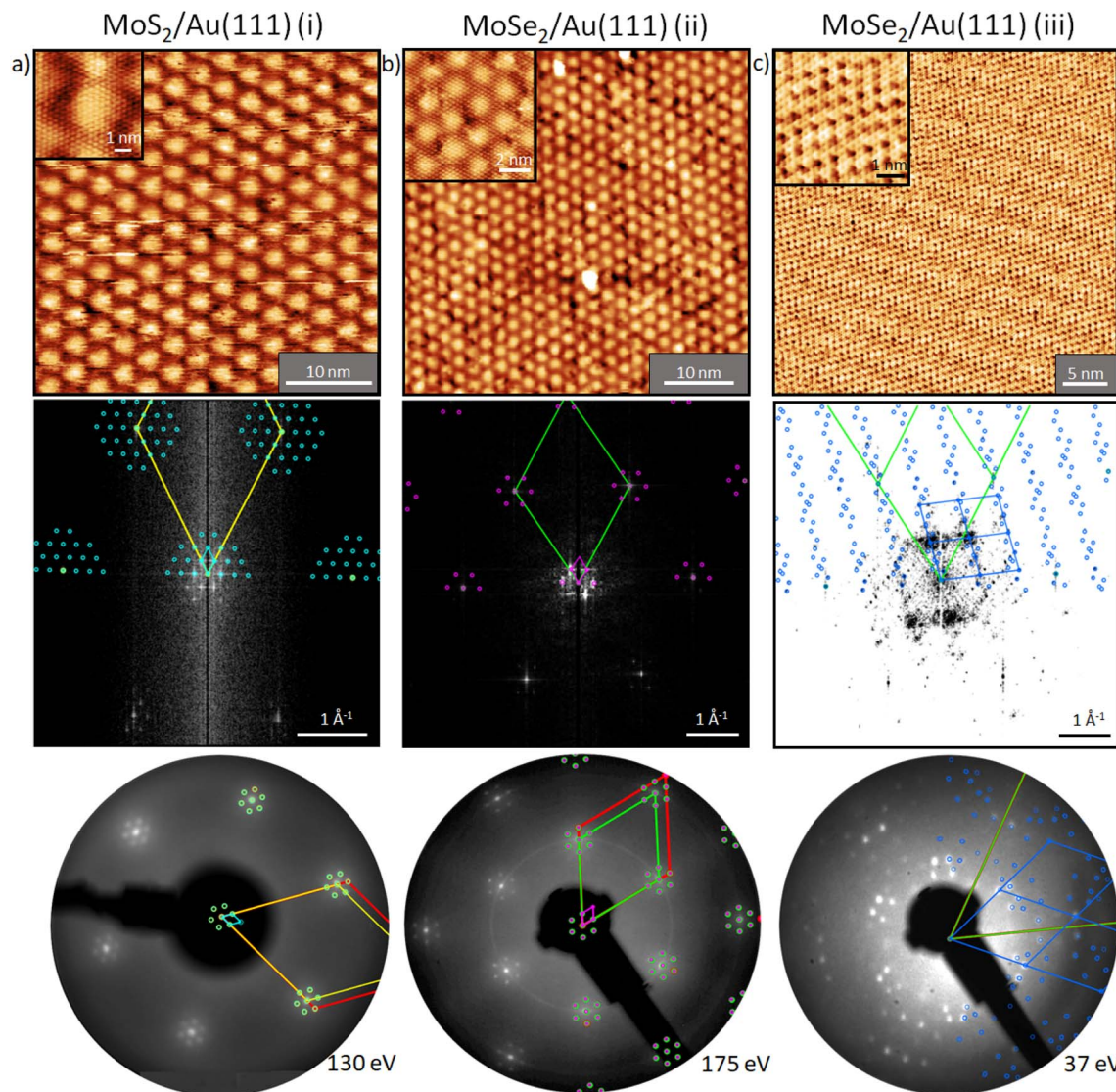


Fig. 3 STM and LEED data of (a) MoS₂ (i), (b) MoSe₂ (ii), and (c) MoSe₂ (iii) on Au(111). For MoS₂ (i) and MoSe₂ (ii), the moiré structures are visible. For MoSe₂ (iii), another structure is observed besides the MoSe₂ lattice originating from an ordered Se₈ interlayer⁴¹ discussed in the text. As insets, we show in (a) and (b) atomically resolved STM images of MoS₂ and MoSe₂ with the moiré contrast being also present and in (c) a close-up image of the MoSe₂ structure on this Se interlayer. To increase the visibility of the features, the contrast of the FFT image in (c) was inverted. In the FFT and the LEED patterns the MoS₂ (yellow), the moiré structure of MoS₂ (light blue), the MoSe₂ (green), the moiré structure of MoSe₂ (magenta), the Se₈ interlayer (blue) structure as well as the Au(111) structure (red) are simulated. Note that sample (i) and (iii) were annealed at 100 °C, sample (ii) was annealed at 250 °C and STM images in (a) are drift-corrected (see Experimental section). STM conditions: (a) 293 K, 0.5 V, 0.5 nA; (a)[inset] 293 K, 0.5 V, 1.5 nA; (b) 4.2 K, 0.4 V, 50 pA; (b)[inset] 4.2 K, 1.5, 20 pA; (c) 4.2 K, 0.1 V, 80 pA. Used LEED equipment: (a) single MCP LEED (ScientaOmicron); (b) and (c) BDL800IR MCP2 (OCI).

interlayer occur, since LEED and STM measurements were carried out at RT and 4.2 K, respectively. The unexplained spots might stem from a different Se structure present on the surface.

2.3 Electronic properties: ARUPS and STS

Finally, we focus on the electronic properties of the TMD samples. All data discussed in this chapter were obtained after a moderate annealing step at 100 °C in UHV (unless otherwise described). The ST spectrum of MoS₂ on Au(111) (sample (i)) shows a band gap calculated from the onsets of the valence and conduction band of (1.4 ± 0.1) eV (see Fig. S6a†). This value is

smaller than the value for a freestanding MoS₂ monolayer because of hybridization with the substrate and bandgap renormalization at the Au(111) surface.^{27,29} Small variations of the onsets of valence and conduction band could be due to defect states in the band gap.^{12,40} In the ST spectrum of the MoSe₂ sample (iii) with a higher Se concentration, we observe a shift of the complete spectrum by 0.5 V compared to the spectrum of sample (ii) which is similar to the reported spectrum in ref. 29. The measured bandgap of sample (ii) (~ 1.9 eV) and sample (iii) (~ 1.7 eV) are almost the same (Fig. S6b and c†).



Table 2 Lattice parameters obtained from the analysis of the LEED patterns and the FFTs of STM images of MoS₂ (i), MoSe₂ (ii) and MoSe₂ (iii) discussed in the text and shown in Fig. 3. $|\vec{a}_1|$, $|\vec{a}_2|$, $\angle(\vec{a}_2, \vec{a}_1)$ and $\angle(\vec{a}_1, \vec{s}_1)$ are the lengths of lattice vectors, the enclosing angle of these vectors, as well as the angle between the first adsorbate vector (\vec{a}_1) and the first substrate vector (\vec{s}_1), respectively. The moiré lattice parameters are obtained from the same LEED spots as for MoS₂ and MoSe₂ (Fig. 3). For the Se superstructure analyzed by the FFT of the STM image, two different values for both angles are given. The values given in square brackets have the same convention as in ref. 41, so that $\angle(\vec{a}_2, \vec{a}_1)$ is smaller than 90°. Note, that the STM (thus also the FFT) images and the LEED patterns of sample (ii) and (iii) were measured at 4.2 K or RT, respectively. This difference can explain the deviations between the values of the Se₈ superstructure since slight temperature-dependent structural changes cannot be excluded

Structure	$ \vec{a}_1 $ [Å]	$ \vec{a}_2 $ [Å]	$\angle(\vec{a}_2, \vec{a}_1)$ [°]	$\angle(\vec{a}_1, \vec{s}_1)$ [°]
(i) MoS ₂ /Au(111) (LEED, RT)	3.15(1)	3.15(1)	120.00(4)	0.00(3)
(i) MoS ₂ /Au(111) moiré (LEED, RT)	33.3(3)	33.9(3)	120.1(6)	0.7(6)
(i) MoS ₂ /Au(111) moiré (FFT, RT)	34(2)	34(2)	119.7(9)	0(1)
(ii) MoSe ₂ /Au(111) (LEED, RT)	3.27(1)	3.28(1)	120.00(4)	0.00(2)
(ii) MoSe ₂ /Au(111) moiré (LEED, RT)	23.0(1)	23.0(1)	120.0(1)	0.0(1)
(ii) MoSe ₂ /Au(111) moiré (FFT, 4.2 K)	22(1)	22(1)	120(1)	5(1)
(iii) Se ₈ superstructure (LEED, RT)	7.54(1)	8.64(1)	116.46(7)	38.7(2)
(iii) Se ₈ superstructure (FFT, 4.2 K)	8.41(8)	8.48(8)	98.8(6) [81.2(6)]	43.0(5) [-17.0(5)]

To study the valence band structure of MoS₂ and MoSe₂ on Au(111), we carried out ARUPS along the Γ - K direction. First, we discuss the band structure of MoS₂ on Au(111), Fig. 4a. Mo 4d and S 3p orbitals generally contribute to the valence band structure. At the Γ point, bands originate from the out-of-plane Mo 4d_z and S 3p_z orbitals which are responsible for the hybridization with the substrate.^{42,43} In contrast, the Mo 4d_{x²-y²} and S 3p_{xy} orbitals contribute to the band structure at the K point leading to spin-orbit splitting.⁴³⁻⁴⁵ In our measurement, one

MoS₂ phase is observed and its valence band maximum (VBM) is localized at the K point, which is typical for the monolayer.^{39,46} The VBM is localized at a BE = 1.4 eV, matching the valence band offset of the respective ST spectrum in Fig. S6.† In the zoom-in image taken with the 2D detector, we can resolve the spin-orbit split bands with an energetic separation of (120 ± 10) meV (Fig. 4a), which corresponds roughly to the theoretical value of 148 meV.³⁹ In contrast to ARPES measurements of the CVD grown MoS₂ monolayers on highly-oriented pyrolytic graphite

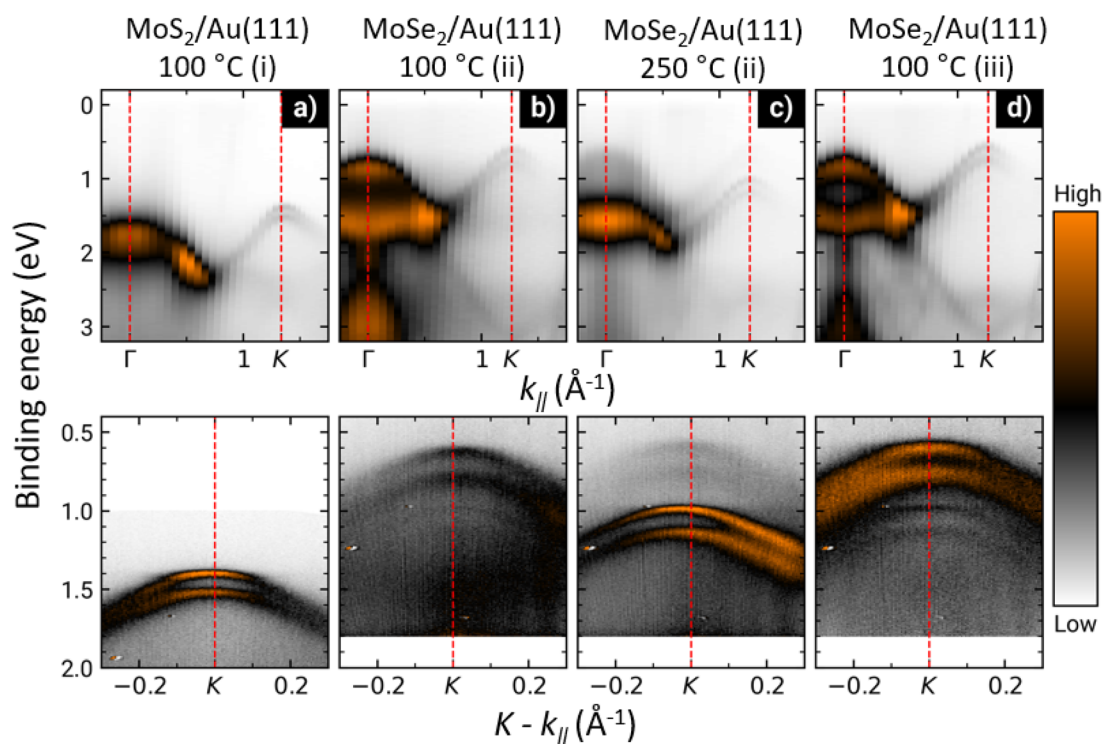


Fig. 4 Angle-resolved UP spectra of (a) MoS₂ (i) annealed at 100 °C, MoSe₂ (ii) annealed at (b) 100 °C and (c) 250 °C as well as (d) MoSe₂ (iii) on Au(111) annealed at 100 °C. In the top parts, the band structures along Γ - K direction are shown. Below, detailed parts of the band structure acquired with the 2D detector around the K points are presented.



(HOPG)⁴⁷ we clearly observe a spin–orbit splitting of the valence band in our measurements. Moreover, the band structure at the Γ point shows a strong spectral broadening due to the hybridization between the MoS₂ monolayer and the substrate.²⁹ Since all MoS₂ samples with different S to Mo ratio show the same properties according to Raman spectroscopy, LEED and STM we conclude that we do not see an influence of excess S atoms on the electronic structure of MoS₂.

For MoSe₂ sample (ii), we expect temperature-dependent changes in the ARUPS data, because Raman spectroscopy and XPS suggest a stronger substrate interaction after annealing at 250 °C compared to the annealing at lower temperatures. Generally, Mo 4d and Se 4p orbitals contribute to the valence band structure of MoSe₂. In the ARUPS measurement of the MoSe₂ sample (ii) on Au(111) directly after the preparation, it was not possible to measure any features of MoSe₂ due to contaminations on top of the sample (Fig. S7†). After annealing at 100 °C, the MoSe₂ bands become visible, Fig. 4b. The VBM is also localized at the K point, confirming the monolayer formation of MoSe₂.³⁹ In the close-up image taken with the 2D detector (Fig. 4b below), we observe two spin–orbit split bands at the K point (four bands in total) with different binding energies and the more intense bands at a lower BE (maxima at BE \approx 0.6 eV and 0.8 eV). The other spin–orbit split bands are faintly visible at BEs of the maxima of 1.0 and 1.2 eV. Since we also observed two doublets corresponding to MoSe₂ monolayers with different intensities in the Mo 3d and Se 3d core level spectra, we correlate these two different XPS doublets with the two phases observed in the valence band structure, which can be assigned to the presence of two monolayer phases differently interacting with the gold substrate. By qualitative comparison of the XPS and ARUPS intensities, the XPS doublet at smaller BEs and the spin–orbit split bands located at smaller BEs in the ARUPS data can be assigned to the MoSe₂ phase (I) which weakly interacts with the Au(111) substrate as well as the XPS and ARUPS features at higher BEs correlate with the MoSe₂ phase (II) which strongly interacts with the substrate. The spin–orbit splitting at the K point for the more intense bands (phase (I)) at lower binding energies is (180 ± 10) meV (Fig. 4b), which is in good agreement with the theoretically calculated value of 185 meV.³⁹

Next, the MoSe₂ sample (ii) was annealed at 250 °C. Based on the Raman spectroscopy and XPS results, we expect a change in the band structure and a stronger MoSe₂/substrate-interaction. In agreement with this expectation, the bands at higher binding energies (phase (II)) become more intense (Fig. 4c), and the VBM of this band structure is located at the K point at BE \approx 1.0 eV (Fig. 4c below). Therewith, we confirm that we have the same two different monolayer phases of MoSe₂ on the sample as before, but the intensity ratio between both phases is drastically changed. The band structure of phase (II) (see Fig. 4c) shows a larger broadening of the bands at the Γ point compared to the band structure of phase (I) (see Fig. 4b), resulting from the hybridization with the Au(111) substrate similar as for MoS₂ monolayer on Au(111). In addition, the splitting at the K point of phase (II) is reduced in comparison to phase (I) and determined to be (150 ± 10) meV. This change indicates that the valence bands at the K point are also influenced by a coupling effect to

the Au(111) substrate. Therefore, we suggest, on the one hand, that the band structure at higher binding energies originates from MoSe₂ which is hybridized with the Au(111) substrate. On the other hand, the band structure at lower binding energies is caused by decoupled MoSe₂ probably due to disordered Se atoms in between MoSe₂ and substrate. Both conclusions are supported by our Raman spectroscopy results discussed above. The decoupling of the MoSe₂ is achieved by residual Se atoms underneath the TMD layer. Two MoSe₂ monolayer phases are observable in XPS and ARUPS for both annealing steps at 100 °C and 250 °C. At higher annealing temperature we can remove additional Se atoms to a large extent, and phase (II) becomes dominant whereas the weakly interacting phase (I) is dominant at annealing temperatures smaller than 100 °C.

Finally, we discuss the ARUPS data of MoSe₂ sample (iii) with the Se superstructure between MoSe₂ and Au(111). The measured band structure in Fig. 4d shows a similar behavior as for the MoSe₂ sample (ii) annealed at 100 °C (*cf.* Fig. 4b). This result means that the MoSe₂ phase (I) with the VBM at \approx 0.6 eV is dominant and the splitting at the K point is (170 ± 10) meV, whereby the electronic structure is hardly or not influenced by the Se₈ superstructure. These ARUPS findings agree with the results from Raman spectroscopy and XPS and are summarized in Table 1. Finally, our experimental data suggest that the decoupling effect of the MoSe₂ monolayer is caused either by a periodic Se interlayer or by some disordered intercalated Se atoms. In contrast, for MoS₂ monolayers we do not observe intercalation of the excess S atoms. This difference in comparison to the MoSe₂ monolayers may result from a stronger chemical binding of selenium to Au(111).^{48,49} Most probably, a weaker interaction of sulfur to Au(111) is insufficient to achieve the intercalation. Note that the interatomic distance of the MoSe₂ monolayer to Au(111) is higher than that of the MoS₂,⁵⁰ which can also facilitate intercalation of the chalcogen atoms.

3 Conclusion

In summary, we investigated MoS₂ and MoSe₂ monolayers epitaxially grown on an Au(111) single crystal by CVD at ambient pressure. These two-dimensional films were characterized by complementary surface sensitive methods. Raman spectroscopy, scanning tunneling microscopy, X-ray and angle-resolved ultraviolet photoelectron spectroscopy confirm the formation of high-quality monolayers of both TMDs on the Au(111) surface. For MoS₂, the moiré structure, Raman modes, as well as the band structure demonstrate strong interaction of the monolayer with the substrate. Even after annealing at 450 °C no significant changes in comparison to the non-annealed samples were observed. In contrast, MoSe₂ monolayers show a different behavior. For as-grown and moderately annealed samples (<100 °C) we observed a weak interaction of MoSe₂ with the Au(111) substrate due to the formation of a periodic as well as disordered Se interlayer. After annealing at 250 °C, due to desorption of this excess intercalated Se, we found that the interaction of the MoSe₂ monolayer with the gold substrate increases, which is reflected in both Raman spectroscopy and photoelectron spectroscopy data. Our study is a step towards the rational design of



TMD monolayer/Au interfaces which are important for the engineering of efficient electrical contacts for a variety of electronic and optoelectronic devices based on TMD monolayers.

4 Experimental section

4.1 Sample preparation

Before the growth process, the Au(111) single crystal (MaTeck, Jülich, Germany) with a purity of 99.999% was cleaned by several cycles of Ar⁺-sputtering (1 kV, 30 min, $\approx 4 \mu\text{A cm}^{-2}$) and annealing ($\approx 600 \text{ }^\circ\text{C}$, 30 min) in an ultrahigh vacuum (UHV) chamber. For the growth, we used a CVD process in a two-zone furnace outside of the UHV system at ambient pressure, as described in ref. 7. In the first zone, the S powder (purity of 99.98%, Sigma Aldrich) or the Se pellets (purity of 99.98%, Sigma Aldrich) were placed in a Knudsen cell and heated up to 200 °C (sample (i)) or 300 °C (sample (ii)), respectively. Additionally, for the MoSe₂ sample with a higher amount of Se discussed in the paper, we heated the Knudsen cell with the Se pellets up to 450 °C (sample (iii)) to provide this excess Se concentration. The MoO₃ precursor (purity of 99.97%, Sigma Aldrich) and the prepared Au(111) single crystal were placed into the second zone. Here, the temperature was held at 700 °C and 750 °C to prepare MoS₂ and MoSe₂, respectively. In addition, a mixture of Ar and H₂ (100 sccm and 10 sccm, respectively) was used as carrier gas. The subsequent annealing steps of the TMD samples were carried out in a UHV chamber to clean the sample from airborne impurities.

4.2 Optical microscopy

Optical microscopy was performed using a Zeiss Axio Imager Z1.m microscope in brightfield mode. The microscope includes a thermoelectrically cooled 5 MP CCD camera.

4.3 Raman spectroscopy

The Raman spectra were measured using a Bruker Senterra spectrometer operated in backscattering mode at ambient conditions. The utilized wavelength of 532 nm was obtained from a frequency-doubled Nd:YAG laser. A 50 \times objective and a thermoelectrically cooled CCD detector ($-65 \text{ }^\circ\text{C}$) were used. The spectral resolution of the system is 2–3 cm^{-1} .

4.4 Scanning tunneling microscopy/spectroscopy

The following techniques were carried out in UHV at a base pressure $< 3 \times 10^{-10}$ mbar. For scanning tunneling microscopy (STM) and spectroscopy (STS), we used two different systems. On the one hand, we measured with a VT SPM (Scienta Omicron) at room temperature (RT) using a W tip. On the other hand, STM images were also acquired using a SPECS JT-STM/AFM at 4.2 K with a Pt/Ir tip. All images were evaluated with Gwyddion (Version 2.56).⁵¹ The RT STM images were drift corrected using the autocorrelation function and calibrated by the lattice parameters obtained from LEED analysis. For quantitative analysis of the fast Fourier transform (FFT) of the STM images, the LEEDLab software was used.⁵² ST spectra were obtained by means of a lock-in amplifier at RT or liquid helium

temperatures. A description of the estimation of the band gap values is given in Fig. S6.†

4.5 Distortion-corrected low-energy electron diffraction

Low-energy electron diffraction (LEED) images of the TMD layers were acquired using dual microchannel plates (MCP2-SpectraLEED, Omicron; BDL800IR MCP2, OCI Vacuum Micro-engineering) or a single microchannel plate (SMCP, Scienta Omicron) LEED-system. All pictures were corrected for geometrical distortions and the energy error using the LEEDCal software.^{53,54} After that, the obtained images were analyzed with the LEEDLab software to quantify the reciprocal structures of the visible LEED spots with an accuracy of typically 1%.⁵² For the Au(111) surface a lattice constant of 2.884 Å was used.⁵⁵

4.6 X-ray and (angle-resolved) ultraviolet photoelectron spectroscopy

X-ray photoelectron spectroscopy (XPS) was performed at two different systems using monochromatized Al K _{α} radiation (SPECS Focus 500 and Scienta Omicron XM1200) and electron analyzers with a spectral energy resolution of 0.5 eV and 0.6 eV (SPECS PHOIBOS 150 and Argus CU), respectively. The spectra were calibrated to the Au 4f_{7/2} peak (84.0 eV) and fitted by Voigt functions (Gaussian : Lorentzian = 70 : 30) after subtraction of a linear (S 2p, Se 3d) or Shirley (Mo 3d) background. Ultraviolet photoelectron spectroscopy (UPS) was carried out with monochromatized and p-polarized He I _{α} excitation (SPECS ultraviolet light source (UVLS)). Band structures were measured by tilting the sample and therefore changing the photoemission angle (angle-resolved UPS (ARUPS)). The binding energy of the UP spectra was calibrated to the Fermi edge of Au(111), whereby an energy resolution of 60 meV were used. For detailed band structure measurements, we used the 2D mode of the detector. All photoelectron spectroscopy measurements were carried out at room temperature.

Author contributions

The manuscript was written with contributions of all authors.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This research was supported by the Deutsche Forschungsgemeinschaft (DFG) through a research infrastructure grant INST 275/257-1 FUGG (313713174), CRC 1375 NOA (Project B2, 398816777), SPP2244 (Project TU149/13-1, 443361515) and DFG individual grant TU149/16-1 (464283495). We acknowledge support by the German Research Foundation Projekt-No. 512648189 and the Open Access Publication Fund of the Thuringer Universitaets - und Landesbibliothek Jena. The authors thank Stephanie Höppener and Ulrich S. Schubert for enabling Raman spectroscopy studies



at the Jena Center for Soft Matter (JCSM) as well as Shahab Mohebi Fard and Emad Najafidehaghani for discussions and the artist views on the samples studied in this work.

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