

## CORRECTION

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## Correction: Solid oxide fuel cells for ammonia synthesis and energy conversion

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Correction for 'Solid oxide fuel cells for ammonia synthesis and energy conversion' by Valentina Goldstein et al., *Sustainable Energy Fuels*, 2022, **6**, 4706–4715, <https://doi.org/10.1039/D2SE00954D>.

Table 2 in the original article shows the incorrect unit for the rate of  $\text{NH}_3$  in two locations. The correct unit of the rate of  $\text{NH}_3$  is  $10^{-9} \text{ mol cm}^{-2} \text{ s}^{-1}$ . Below is the correct version of Table 2.

**Table 2** The parameters of the tested cells, including the current and voltage used for the  $\text{NH}_3$  synthesis, the  $\text{NH}_3$  synthesis rate ( $\text{mol cm}^{-2} \text{ s}^{-1}$ ), energy consumption, the fraction of energy expended for  $\text{NH}_3$  synthesis, the faradaic efficiency ( $\Lambda_F$ ), and the energy efficiency ( $\Lambda_E$ )

#	Parameter/cell	CH <sub>4</sub>  cell ① air	CH <sub>4</sub>  cell ② air	CH <sub>4</sub>  cell ② air	H <sub>2</sub>  cell ⑥ air
1	Wet, %	20	14	5	20
2	Current (mA)	16	16	1.3	3.36
3	Voltage (V)	0.039	0.05	0.9	0.9
4	Mass of $\text{NH}_3$ (mg)	2.166	0.437	0.640	3
5	Time (min)	107	12	46	60
6	Energy consumption (kJ mol <sub>NH<sub>3</sub></sub> <sup>-1</sup> )	22.7	22.4	85.78	61.7
7	Fraction of energy applied for $\text{NH}_3$ synthesis (%)	3.6	7	24.6	7
8	Rate of $\text{NH}_3$ ( $10^{-9} \text{ mol cm}^{-2} \text{ s}^{-1}$ )	1.2	1.4	0.85	3
9	$\Lambda_F$ , Faraday efficiency	—	—	3.0	4
10	$\Lambda_E$ , energy efficiency	2	2.8	0.72	1
	Rate of $\text{NH}_3$ at OCV ( $10^{-9} \text{ mol cm}^{-2} \text{ s}^{-1}$ )	1.4	1.4	0.97	2.6

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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