

## CORRECTION

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## Correction: Solid oxide fuel cells for ammonia synthesis and energy conversion

Valentina Goldstein,<sup>a</sup> Manasa Kumar Rath,<sup>\*a</sup> Alexey Kossenko,<sup>ab</sup> Natali Litvak,<sup>a</sup> Alexander Kalashnikov<sup>c</sup> and Michael Zinigrad<sup>ab</sup>Cite this: *Sustainable Energy Fuels*, 2022, 6, 5199

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Correction for 'Solid oxide fuel cells for ammonia synthesis and energy conversion' by Valentina Goldstein et al., *Sustainable Energy Fuels*, 2022, 6, 4706–4715, <https://doi.org/10.1039/D2SE00954D>.

Table 2 in the original article shows the incorrect unit for the rate of NH<sub>3</sub> in two locations. The correct unit of the rate of NH<sub>3</sub> is 10<sup>−9</sup> mol cm<sup>−2</sup> s<sup>−1</sup>. Below is the correct version of Table 2.

**Table 2** The parameters of the tested cells, including the current and voltage used for the NH<sub>3</sub> synthesis, the NH<sub>3</sub> synthesis rate (mol cm<sup>−2</sup> s<sup>−1</sup>), energy consumption, the fraction of energy expended for NH<sub>3</sub> synthesis, the faradaic efficiency ( $\mathcal{A}_F$ ), and the energy efficiency ( $\mathcal{A}_E$ )

#	Parameter/cell	CH <sub>4</sub>  cell ① air	CH <sub>4</sub>  cell ④ air	CH <sub>4</sub>  cell ② air	H <sub>2</sub>  cell ⑥ air
1	Wet, %	20	14	5	20
2	Current (mA)	16	16	1.3	3.36
3	Voltage (V)	0.039	0.05	0.9	0.9
4	Mass of NH <sub>3</sub> (mg)	2.166	0.437	0.640	3
5	Time (min)	107	12	46	60
6	Energy consumption (kJ mol <sub>NH<sub>3</sub></sub> <sup>−1</sup> )	22.7	22.4	85.78	61.7
7	Fraction of energy applied for NH <sub>3</sub> synthesis (%)	3.6	7	24.6	7
8	Rate of NH <sub>3</sub> (10 <sup>−9</sup> mol cm <sup>−2</sup> s <sup>−1</sup> )	1.2	1.4	0.85	3
9	$\mathcal{A}_F$ , Faraday efficiency	—	—	3.0	4
10	$\mathcal{A}_E$ , energy efficiency	2	2.8	0.72	1
	Rate of NH <sub>3</sub> at OCV (10 <sup>−9</sup> mol cm <sup>−2</sup> s <sup>−1</sup> )	1.4	1.4	0.97	2.6

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

<sup>a</sup>Materials Research Center, Ariel University, Ariel 40700, Israel. E-mail: manas.physics@gmail.com

<sup>b</sup>Department of Chemical Engineering, Biotechnology and Materials, Faculty of Engineering, Ariel University, Ariel 40700, Israel

<sup>c</sup>Rock Capital Partners, Moscow, Russia

