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Introduction

In recent years, recreational and medicinal cannabis (*Cannabis sativa*) use has been growing in popularity, with countries increasingly removing restrictions on its consumption and sale.¹ The psychotropic effects commonly associated with cannabis are caused by cannabinoids produced in the plant, most notably Δ^9 -tetrahydrocannabinolic acid (THCA) that, upon heating, decarboxylates to Δ^9 -tetrahydrocannabinol (THC), which is known to cause psychogenic effects in consumers.² THC is chemically very similar to other

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Chloro aluminum phthalocyanine-based organic thin-film transistors as cannabinoid sensors: engineering the thin film response[†]

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Cannabis producers, retailers, and law enforcement increasingly need low-cost point-of-source cannabinoid sensors. Organic thin-film transistor (OTFT) based sensors are a promising technology that can provide rapid speciation and detection of Δ^9 -tetrahydrocannabinol (THC) while maintaining low manufacturing costs and ease of use. Herein, chloro aluminum phthalocyanine (Cl-AIPc) OTFTs were optimized through engineering film thickness (30, 50 or 100 nm) and the device source-drain geometry (*W*/*L* = 100, 200, 400, 800 and 1000), as these parameters have been shown to strongly influence OTFT performance. Optimized Cl-AIPc OTFT based sensors were exposed to both THC solution and THC vapor, demonstrating that improved device performance was not directly correlated with increased sensitivity. Grazing-incidence wide-angle X-ray scattering (GIWAXS) and atomic force microscopy (AFM) were used to interrogate thin-film morphology. While little change in molecular orientation resulted from film thickness or exposure to THC, the data suggests that the improved sensing response of Cl-AIPc-derived devices is directly related to increased thin-film surface area resulting from increased roughness and reduced film thickness.

cannabinoids in cannabis products, including some that are potentially therapeutic, such as cannabidiol (CBD).² Due to this chemical similarity, it is difficult to differentiate THC and CBD in consumer products or human biological samples, which presents a challenge for a variety of cannabis-related fields, including cannabis producers, government regulatory bodies, law enforcement, and medical professionals. The current state-of-the-art detection methods used in industry, such as high-performance liquid chromatography (HPLC), have a high level of sensitivity but are expensive, timeconsuming, require specialists to interpret the results, and cannot be performed on-the-spot.3 Thus, there is a present need for low-cost point-of-source sensors that maintain both accuracy and a high level of sensitivity. Organic thin-film transistor (OTFT)-based sensors are a promising alternative that offer fast and accurate detection of cannabinoids while being accessible, disposable, and inexpensive.⁴⁻⁶ OTFTs have the potential to achieve a lower manufacturing cost than conventional electronics, as OTFTs utilize more affordable organic materials and are manufactured at ambient temperatures. The simple device structure of an OTFT also makes them conducive to high-volume manufacturing processes, like roll-to-roll printing.7-14 OTFTs are threeelectrode solid state electrical devices where current travels at

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the interface between an organic semiconductor and a dielectric. Changes in electrical behaviour can result from several mechanisms, including oxidation/reduction, and doping or physical changes in the active semiconducting material.^{15–20} Understanding which mechanisms contribute to sensing and how these mechanisms interact will enable the optimization and enhancement of OTFT based sensors.

Metal and metalloid phthalocyanines (MPcs) are a class of conjugated macrocyclic molecules that have found success as semiconducting layers in OTFTs^{21,22} and organic photovoltaics (OPVs).^{23,24} MPcs are thermally and chemically stable, inexpensive to manufacture on the ton scale with facile metal substitution providing opportunities to tune integration molecular interactions for into OTFT sensors.^{6,21,25-30} While many MPcs are insoluble or sparingly soluble in organic solvents, they can be modified with solubilizing groups making them potentially compatible with low-cost solution processing techniques. Previously, we demonstrated copper phthalocyanine (CuPc) OTFT sensors ratiometric detection for and differentiation of cannabinoids.6 Observed by spectroelectrochemistry and 2D NMR, cannabinoids interact strongly with MPcs in solution, and in thin-films, causing changes in Pc-Pc π -stacking and increasing π -electron delocalization.³¹ These interactions can induce recrystallization of MPc thin-films, altering thin-film morphologies and nanostructures, resulting in OTFT performance changes which enhance the detection of THC and lead to better sensitivity.26 The central aluminum chloride of chloro aluminum phthalocyanine (Cl-AlPc, Fig. 1) bends the MPc ring, resulting in unique packing structures that are susceptible to cannabinoid induced physical effects, demonstrating greater interactions both in solution and as a thin-film compared to CuPc.26,31 In bottom-gate bottomcontact (BGBC) OTFTs, the thickness of the deposited semiconductor film has been observed to effect OTFT charge transport properties.³²⁻³⁴ For evaporated small molecules, surface coverage, growth mode, and thin-film morphology can change with thickness, leading to altered charge transport properties.^{25,35-38} Thus, to increase the sensitivity



Fig. 1 Chemical structure of chloro aluminum phthalocyanine (Cl-AlPc) and Δ^9 -tetrahydrocannabinol (THC), as well as a schematic representation of the bottom-gate bottom-contact (BGBC) organic thin-film transistor (OTFT) architecture used in this study.

of cannabinoid sensors, we exploit the sensitivity of Cl-AlPc to cannabinoids by altering the semiconducting layer and device architecture of Cl-AlPc-based OTFTs.

Towards the goal of understanding how to optimize Cl-AlPc-based devices for THC sensing, we studied the molecular interactions of Cl-AlPc in OTFTs and the structural alterations that take place upon exposure to cannabinoids in both liquid and vapor phases. By synchrotron grazingincident wide-angle X-ray scattering (GIWAXS) and atomic force microscopy (AFM) we probe the nanostructures of Cl-AlPc thin-films, comparing the effects of OTFT architecture and thin-film thickness to overall OTFT sensor performance. In addition, we aim to determine if these parameters affect baseline device performance in the same way as sensor performance to differentiate the dynamics of THC sensing from normal OTFT operation. This work demonstrates that device architecture has little effect on sensor performance; instead, we find that film thickness substantially impacts both pre- and post-exposure device performance and sensitivity, particularly to THC solution. This illustrates the importance of deposition conditions and thin-film morphology.

Results and discussion

Baseline Cl-AlPc OTFT optimization

To examine the effects of device architecture on OTFT performance we prepared a series of BGBC Cl-AlPc-based OTFTs under identical evaporation conditions, with thin-film thicknesses of 30 nm, 50 nm or 100 nm and W/L ratios from 100 to 1000 on OTS treated Si/SiO₂ substrates (Fig. 1). Characteristic transfer curves can be found in Fig. 2, along with plots of the on-current (I_{on}) and μ_h with the tested W/Lratios and film thicknesses. The output curves and a summary table can be found in Fig. S1 and Table S1,† respectively. A two-way ANOVA analysis was performed on the data shown in Fig. 2D and E to determine the significance of each pair-wise data set to a confidence of 95%, and the results can be found in Table S2.† As expected from eqn (1), $I_{\rm on}$ increases with increasing W/L, from approximately 100 μ A to 5000 μ A from a W/L of 100 to 1000. Additionally, $I_{\rm on}$ increases with increasing film thickness, yielding an increase in $\mu_{\rm h}$ from 0.02 cm² V⁻¹ s⁻¹ to 0.5 cm² V⁻¹ s⁻¹, consistent with previously reported performances for Cl-AlPc devices.^{26,27}

An ideal transfer curve would have a $V_{\rm T}$ near 0, a low $I_{\rm off}$ followed by a sharp increase in current after the $V_{\rm T}$ (steep subthreshold slope), and the current would plateau at a high $I_{\rm on}$ resulting in a high $I_{\rm on/off}$, with low hysteresis throughout the transfer curve. While a higher $I_{\rm on}$ led to a higher $\mu_{\rm h}$ for the thicker films, the transfer curves for these films are less ideal. A greater $V_{\rm T}$, lower $I_{\rm on/off}$, and a smaller subthreshold slope is observed for the 100 nm devices compared to the 30 nm devices. The $I_{\rm on}$ also increases as the W/L ratio increases from 100 to 800, while the transfer curves become less ideal. A W/L of 1000 shows the most ideal transfer curves with the highest $I_{\rm on}$ and $\mu_{\rm h}$. The devices with a W/L ratio of 1000 had



Fig. 2 Transfer curves for Cl-AlPc devices with 30 nm (A), 50 nm (B), and 100 nm (C) thickness. I_{on} (D) and μ_h (E) are plotted with W/L and Cl-AlPc thickness. In panels D and E, points represent the means of a minimum of 4 devices and error bars represent standard deviation. Significance of data has been analyzed by two-way ANOVA and can be found in Table S2.†

the same $L(10 \text{ }\mu\text{m})$ but a greater W(10 mm) than the devices with a W/L ratio of 200 (W = 2 mm), thus, the dependence of these metrics on W/L is likely a result of contact resistance, which becomes more dominant as L decreases.39,40 This increased contact resistance would lead to the less optimal transfer characteristics that we observed for higher W/L ratios. Additionally, there is a reduced effect from channel resistance in devices with a lower $L_{,}^{39,40}$ leading to higher operating currents. This resistance dominates in devices that have a greater L, which explains the lower performance of the devices with a lower W/L ratio. The design of the devices with the W/L of 1000 balances the channel and contact resistances give high operating current with ideal transfer to characteristics. This balance is illustrated by the $V_{\rm T}$ of the devices in Fig. 2, where the $V_{\rm T}$ increases from a W/L of 100 to 800 but drops for the W/L of 1000.

The contact resistance for each Cl-AlPc thickness was calculated as a function channel length using the modified-transfer length method (M-TLM)(eqn (2)), the results of which can be found in Table S3.† The contact resistance decreased with increasing film thickness. This supports the resulting higher $I_{\rm on}$ and $\mu_{\rm h}$ observed for the 100 nm devices, as its contact resistance was the lowest, allowing for easier charge injection and higher mobilities.

Thin-films of Cl-AlPc, with thicknesses of 30 nm, 50 nm, or 100 nm, were characterized by GIWAXS with the resulting spectra shown in Fig. 3. These spectra reveal that the films are moderately semicrystalline and correspond well to the single crystal as demonstrated by the diffraction patterns shown in Fig. 3G. The high intensity (001) peak at approximately q = 0.5 Å⁻¹, along the q_z axis observed in all spectra indicates a major preferential orientation of molecules aligned parallel to the substrate. The distinct

feature at $\chi = -30^{\circ}$ observed in Fig. 3H indicate Cl-AlPc molecules aligned $\sim 30^{\circ}$ to the substrate in a face-on configuration for all thin-film thicknesses. Although no change in molecular orientation is observed with film thickness, a loss of molecular order is exhibited in the thicker 100 nm films. Characterizing the surface morphology of the Cl-AlPc films by AFM, we observe the ordering of the grains improves with decreasing film thickness, supporting the observations from the GIWAXS spectra. The 30 nm film shows the most ordered grains and sharpest signals by GIWAXS, indicating the highest level of molecular order.

The 30 nm films also demonstrated the best transfer characteristics in their corresponding devices, with high I_{on/-} $_{\rm off}$ ratios, low $V_{\rm T}$, and greater subthreshold slopes (Fig. 2). In contrast, the 100 nm film showed more disordered grains by AFM and broad GIWAXS signals, indicating a disordered morphology (Fig. 3). Though these films had high operating currents and low contact resistance, which yielded a high $\mu_{\rm h}$, they had the worst transfer characteristics with a high $V_{\rm T}$, hysteresis, and low subthreshold slope (Fig. 2). The 50 nm films appear significantly rougher than the other thicknesses, suggesting surface-ordered film growth up to \sim 30 nm (RMS = 5.69 nm) which transitions to island growth up to \sim 50 nm (RMS = 9.49 nm), yielding the observed roughness and grain order (Fig. 3). The islands fill in as the thickness approaches 100 nm (RMS = 5.06 nm), yielding the smoothest, but most disordered, films. Consequentially, the more ordered films will have improved charge transport pathways, however, the number of pathways available are likely lower. This will hinder charge injection and limit the operating current and $\mu_{\rm h}$ as a result. Alternatively, the 100 nm films experience an increased number of charge transport pathways, making charge injection easier, giving rise to the higher $\mu_{\rm h}$ and

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Fig. 3 2D scattering patterns ($\theta = 0.3^{\circ}$) of (A) 30 nm, (B) 50 nm, and (C) 100 nm thick Cl-AlPc. Atomic force microscopy (AFM) images of (D) 30 nm, (E) 50 nm, and (F) 100 nm thick Cl-AlPc with scale bars of 500 nm. Diffraction pattern of Cl-AlPc (G) predicted from single crystal (CCDC #1134071)⁴¹ and 100 nm, 50 nm and 30 nm thick films determined by grazing-incidence wide-angle X-ray scattering (GIWAXS). Linecut profiles (H) of 100 nm, 50 nm, and 30 nm thick Cl-AlPc films with respect to χ using a *q* range between 1.9–2.1 Å⁻¹. Molecular orientation (I) of Cl-AlPc relative to the substrate.

operating current in their corresponding devices. The disordered nature of the films and these pathways causes worse transfer characteristics. The 50 nm films strike a balance somewhere in between, experiencing both an increased number of charge transport pathways compared to the 30 nm film, and improved charge transport pathways compared to the 100 nm film. This leads to a high μ_h and operating current, while maintaining favorable transfer characteristics. Though thin-film morphology is conserved for all thicknesses, the degree of molecular order, and thus the OTFT characteristics, varies, with a >5-fold increase in mobility at the expense of other semiconducting properties. Thus, it is important to control the thickness of the Cl-AlPc film to control the grain morphology, charge transport properties, and baseline performance of Cl-AlPc OTFTs.

Cl-AlPc OTFT sensor optimization

In our previous studies,^{26,31} we identified that Cl-AlPc strongly interacts with THC, leading to large changes in OTFT performance. Thus, by exposing BGBC OTFTs with different Cl-AlPc film thicknesses (30 nm, 50 nm, and 100 nm) and two geometries (W/L = 200 and 1000) to both a liquid solution of THC and THC vapor, we could examine thin-film processing conditions and device architecture of Cl-AlPc OTFTs with the goal of establishing the greatest sensor response. Differences in $V_{\rm T}$, $\mu_{\rm h}$ and $I_{\rm on/off}$ between OTFTs preand post-exposure are displayed in Fig. 4. Characteristic OTFT output curves and transfer curves post-THC exposure can be found in Fig. S2 and S3,† respectively. A two-way ANOVA was performed on the results to analyze their



Fig. 4 Cl-AlPc OTFT based sensor characteristics when exposed to THC solution (A, C, and E) and THC vapor (B, D and F). The device performance changes are reported difference from the baseline device and the exposed device for V_{T} , (A and B), and a ratio from the baseline and the exposed device for $I_{on/offr}$ (C and D), and μ_{hr} (E and F). The axes of (C) and (D) have been plotted on different scales to better frame the data. Points represent the means of a minimum of 4 devices and error bars represent standard deviation. Significance of data has been analyzed by two-way ANOVA and can be found in Table S3.†

statistical significance to a 95% confidence level, and the results of the pair-wise comparisons can be found in Table S4.†

As illustrated in Fig. 4, exposure to THC solution led to a reduction in $V_{\rm T}$, $\mu_{\rm h}$ and $I_{\rm on/off}$ for all devices, however a significantly larger change was observed for the 30 nm devices, indicating that a thinner semiconductor layer is more sensitive to THC in solution. The devices were less sensitive to THC vapor; although a slight reduction in Ion/off current was still observed, with no trend as a function of film thickness. A reduction in $\mu_{\rm h}$ was also observed in response to THC vapor, with a greater response from the 30 nm film compared to the 100 nm film. This result was expected, as previous OTFT gas sensors have shown greater sensitivities films.32,42,43 with thinner semiconductor Channel architecture (W/L) did not appear to influence OTFT performance when exposed to THC solution. Contrastingly, the W/L appears to have a slight effect on the sensitivity of the OTFTs to vapor THC. The W/L of 1000 exhibited marginally larger changes in $V_{\rm T}$, $I_{\rm on/off}$, and $\mu_{\rm h}$, with this effect more pronounced in the thinner films than the 100 nm film. This was determined to be statistically significant. This effect has been demonstrated previously, where channels with larger widths lead to more sensitive gas sensors due to their larger surface area for sensing interactions to occur.⁴⁴ Exposing the OTFTs to pure hexanes as a blank solvent control leads to no change in device performance, as we have previously illustrated.⁶ This suggests that the Cl-AlPc OTFTs are sensitive not only to the analyte but also the form in which the analyte is exposed to the sensor. In both forms of THC exposure, vapor and solution, the devices with the more ideal baseline transfer curves led to the greater response; the 30 nm devices or a *W/L* of 1000.

The same experiments were performed with solution and vapor CBD, the results of which can be found in Fig. S4.† The sensitivity to CBD was similarly strongly dependent on film thickness, particularly to solution samples, while the channel architecture had little to no effect on sensitivity. Paper



Fig. 5 2D scattering patterns ($\theta = 0.3^{\circ}$) of (A) unexposed baseline, (B) THC solution exposed, and (C) THC vapor exposed 30 nm thick Cl-AlPc. AFM images of (D) unexposed baseline, (E) THC solution exposed, and (F) THC vapor exposed 30 nm thick Cl-AlPc with scale bars of 500 nm.

Cl-AlPc films were further characterized by GIWAXS postexposure to THC solution and vapor with their spectra shown in Fig. 5 and S5.† For the 30 nm films, the primary alignment of the molecules remains at 30° to the substrate surface, and no new molecular orientations appear upon exposure to THC. However, the corresponding AFM images in Fig. 5, show a slight increase in grain sizes and disordering after exposure to THC, with some slight roughening of the films from RMS = 5.69 nm to RMS = 6.57 nm for samples exposed to THC solution and RMS = 7.11 nm for samples exposed to THC vapor, with similar trends observed for the 100 nm films (Fig. S5†).

Unlike the 30 nm and 100 nm films, the 50 nm films by GIWAXS demonstrated some molecular reorganization upon exposure to a liquid THC solution and THC vapor as shown in Fig. 6. Upon exposure to liquid THC solution, the film appeared to become more ordered, reflected by the sharper signals in the GIWAXS spectra and diffraction patterns. This can also be observed in the AFM images. The roughness of the film decreased from a baseline of RMS = 9.49 nm to 8.02 nm after exposure, resulting in a more uniform, ordered film upon exposure to THC in solution. The vapor-exposed film also experienced a slight decrease in surface roughness (RMS = 8.86 nm) and increase in grain order. Addtionally, the 50 nm vapor-exposed films demonstrated small changes in the molecular orientation of Cl-AlPc, with a slight shift of the (004) peak from a baseline of q = 1.92 Å⁻¹ to q = 1.85 Å⁻¹ (Fig. S6[†]). This suggests that the rougher grains of the 50 nm films are more susceptible to THC vapor-induced molecular reorganization, resulting in a new polymorph which is not oberved in the thinner 30 nm or thicker 100 nm films.

Notably, small shifts in molecular orientation of the 50 nm films did not produce significant changes in the performance of OTFT sensors. Additionally, the absence of interactions between the 30 nm and 100 nm films and THC by GIWAXS further suggests that the detection mechanism of THC is not purely physical. AFM results demonstrated a slight change in film roughness, more substantial for THC solution, which is accompanied by an increase in detection sensitivity (Fig. 4). The results from the OTFT sensors emphasize that film thickness plays an important role in sensitivity to THC in solution, while W/L imparts a negligible effect in comparison. Jiang et al. demonstrated that sensitivity to NO2 increased with decreasing CuPc film thickness, suggesting that thinner films afford increased analyte interaction with the semiconductive/dielectric interface.32 This has been observed for different semiconductor materials as well.43 As a much larger, piconjugated organic molecule, THC is less likely to permeate throughout the film and therefore the process is likely a charge transfer interaction as THC interacts between grain boundaries and the top layer of the Cl-AlPc film. Thinner films facilitate quicker diffusion of charge and as the surface roughness increases, there are more opportunities for the Pcanalyte interaction on the top layer of the film.

For liquid solution THC sensing, the most sensitive devices were those with the most ideal baseline transfer characteristics and most ideal transport of charge through

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Fig. 6 2D scattering patterns ($\theta = 0.3^{\circ}$) of (A) unexposed baseline, (B) THC solution exposed, and (C) THC vapor exposed 50 nm thick Cl-AlPc. AFM images of (D) unexposed baseline, (E) THC solution exposed, and (F) THC vapor exposed 50 nm thick Cl-AlPc with scale bars of 500 nm.

the device. The fastest baseline charge transport did not necessarily result in the most sensitive THC sensors. This effect was less pronounced, but still present for vapor THC sensing. This finding suggests that we should prioritize the reduction of channel and contact resistances during channel architecture design and device fabrication, and the formation of ordered grains in the thin film to produce quality charge transport pathways for high-performing and sensitive THC sensors. Upon exposure to THC, the changes to more ideal transport pathways are more pronounced, especially in thinner films where less transport pathways are available. These results suggest that THC OTFT sensors can be engineered to be more effective for both solution and vapor THC testing, and that the environment or type of THC exposure will lead to different changes in device response.

Conclusion

We optimized Cl-AlPc OTFTs by changing the source-drain electrode configuration (W/L) and the film thickness. We found that increasing the W/L leads to improved mobilities and I_{on} , but worse transfer curve characteristics. A W/L of 1000 balances the contact and channel resistances to give the best performance with favorable transfer characteristics. It was found to have a limited effect on the sensitivity to THC. An analysis of the different film thicknesses through GIWAXS and AFM showed that the 30 nm films likely had improved charge transport pathways due to higher grain ordering, which led to improved transfer characteristics and sensitivity to THC in solution. The 100 nm films likely had an increased number of charge transport pathways leading to higher mobilities and I_{on} with disordered grains and worse transfer characteristics, resulting in less sensitive THC sensors. No change in molecular orientation was detected with exposure to THC solution and vapor for the 30 nm and 100 nm films. A slight change was observed for the 50 nm film, but this did not lead to a significant change in performance as an OTFT sensor compared to the other films, indicating that the method of detection of THC and Cl-AlPc is likely not solely dependent on physical changes in the thin film. AFM suggests a slight roughening of the Cl-AlPc film with exposure to THC solution leading to a greater surface area, which could contribute to the overall change in device performance. This study suggests that film thickness plays an important role in the quality of the charge transport pathways in Cl-AlPc OTFTs, which can influence the sensitivity to THC.

Experimental section

Materials

Cl-AlPc was obtained from TCI Chemicals and purified by train sublimation before use. (Octyl)trichlorosilane (OTS, 97%) was purchased from TCI Chemicals and used as received. All solvents were HPLC grade and purchased from Fischer Scientific. Cannabinoid standards were obtained from Toronto Research Chemicals.

Thin film formation

Prefabricated devices were purchased from Fraunhofer IPMS to make BGBC transistors. Each substrate consisted of a 230 nm thermally grown SiO₂ dielectric on a Si gate, with prepatterned gold source-drain electrodes to give five different device architectures: $W/L = 100 (W = 2 \text{ mm}, L = 20 \mu\text{m}), W/L = 200 (W = 2 \text{ mm})$ 2 mm, $L = 10 \mu$ m), $W/L = 400 (W = 2 \text{ mm}, L = 5 \mu$ m), W/L = 800 $(W = 2 \text{ mm}, L = 2.5 \text{ }\mu\text{m})$, and $W/L = 1000 (W = 10 \text{ mm}, L = 10 \text{ }\mu\text{m})$ µm). For AFM and GIWAXS characterization, substrates were purchased from Ossila with 300 nm thermally grown SiO₂ dielectric on Si. The substrates were thoroughly washed with acetone to remove the photoresist, then rinsed with isopropanol and dried under a N₂ gas stream. The substrates then underwent an oxygen plasma treatment for 15 minutes, after which time they were briefly rinsed with water, isopropanol, and dried again with N2. The substrates were submerged in a 1% v/v OTS solution in toluene at 70 °C for 1 hour. The treated substrates were then rinsed with toluene and isopropanol before being dried with a N2 gas stream and a further drying step in a 70 °C vacuum oven for 1 hour. The substrates were then heated to 140 °C in an Angstrom EvoVac thermal evaporator at a pressure below 2×10^{-6} Torr and held for 1 hour. Next, Cl-AlPc was deposited onto the surface by sublimation at a rate of 0.2 Å s⁻¹. An in situ servo shutter mask was used to progressively block the plume during fabrication so that all samples were exposed to the same plume at early stages but that some samples were blocked off at early stages and others were blocked when more material could be deposited to form the desired thicknesses of 300 Å, 500 Å, or 1000 Å. After deposition was completed, the substrates were cooled to room temperature before characterization.

GIWAXS characterization

GIWAXS experiments were completed at the Canadian Light Source (CLS) in Saskatoon, Canada using the Brockhouse Diffraction Sector (BXDS) beamline with a photon energy of 15.1 keV. A Rayonix MX300 CCD detector (73.242 μ m × 73.242 μ m pixel size) placed 417 mm from the sample was used to collect 2D scattering patterns at an angle of incidence of θ = 0.3°. Silver behenate and poly(3-hexylthiophene-2,5-diyl) standards were used to calibrate the data, which was analyzed with the GIXSGUI software package in MATLAB, where both polarization and solid-angle corrections were applied.⁴⁵

AFM

AFM measurements (2.5 μ m × 2.5 μ m) were carried out with a Bruker Dimension Icon AFM equipped with ScanAsyst-Air tips using tapping mode and a scan rate of 0.69 Hz. Image processing and editing was performed with NanoScope Analysis v.1.8 software.

OTFT characterization

Transistors with W/L ratios of 100 to 800 were characterized using a custom electrical probe station, oesProbe A10000P290 (Element instrumentation Inc. & Kreus Design Inc.), and transistors with a W/L ratio of 1000 were tested using a custom-build auto tester apparatus. Both were run using a Keithley 614B SourceMeter and Labview software, where the $V_{\rm SD}$ and $V_{\rm GS}$ were set to discrete values and the $I_{\rm SD}$ was measured. To saturate the devices, six transfer curves were obtained and the final three were averaged to yield the characteristic transfer curve for each transistor, as shown in Fig. S1 and S3.† The $V_{\rm T}$ and $I_{\rm on/off}$ was determined from these measurements, and the $\mu_{\rm h}$ was determined using the general expression relating it to the gate voltage in the saturation region, shown again in eqn (1).

$$I_{\rm SD} = \frac{\mu C_{\rm i} W}{2L} \left(V_{\rm GS} - V_{\rm T} \right)^2 \tag{1}$$

The baseline values for $V_{\rm T}$, $I_{\rm on/off}$ and $\mu_{\rm h}$ were obtained for all devices to assess which semiconductor thickness and channel architecture resulted in the best device performance, and these characteristics were averaged across a minimum of 4 devices for each condition.

Contact resistance was calculated by the M-TLM shown in eqn (2). The total resistance ($R_{\rm T}$) was obtained from the output curve at a low source-drain voltage. A plot of $R_{\rm T}W/L \nu s$. 1/*L* was generated for the channel lengths of 2.5, 5, 10, and 20 µm. The data was fit with a linear trendline, the slope of which gave the width-normalized contact resistance, $R_{\rm C}W$.

$$\frac{R_{\rm T}W}{L} = \frac{1}{\mu C_{\rm i}(V_{\rm GS} - V_{\rm T})} + R_{\rm C}W\left(\frac{1}{L}\right), \text{ where} \qquad (2)$$
$$R_{\rm T} = \frac{dV_{\rm DS}}{dI_{\rm DS}}|_{V_{\rm DS} \to 0V}$$

Following, the devices were exposed to either to a liquid THC solution or THC vapor. For solution exposure, 0.5 μ L of a 20 μ M solution of THC in hexanes was drop-cast onto the surface of each transistor channel and allowed to dry for 2 minutes. For vapor exposure, 36 mg of THC was vaporized into an 8 L bag (4.5 mg L⁻¹) using the Vapormed Volcano Medic. The bag was slowly emptied into a small chamber housing the devices where they were exposed to the vapor for a total of 90 seconds. The devices were then characterized again, and new $V_{\rm T}$, $I_{\rm on/off}$ and $\mu_{\rm h}$ values were obtained and compared to the baseline values to assess which devices and channel architectures resulted in the highest sensitivity to THC. The procedure was repeated for CBD solution and CBD vapor.

ANOVA statistical analysis

The two-way ANOVA analysis was performed using the GraphPad Prism software. All $I_{\rm on}$ and $\mu_{\rm h}$ values for baseline devices, and all changes in $V_{\rm T}$, $I_{\rm on/off}$ and $\mu_{\rm h}$ for THC-exposed devices were inputted in the software as grouped data. A two-way ANOVA analysis was performed fitting main effects. The Tukey test was used to compare all data pairs individually.

The effect of W/L and thickness was reported for the overall data set and all pairs using an α of 0.05 (95% confidence interval), and the results were tabulated.

Conflicts of interest

The authors declare no competing financial interests.

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