Chemical Science



EDGE ARTICLE

View Article Online
View Journal | View Issue



Cite this: Chem. Sci., 2022, 13, 9425

d All publication charges for this article have been paid for by the Royal Society of Chemistry

The transient-chelating-group-controlled stereoselective Rh(1)-catalyzed silylative aminocarbonylation of 2-alkynylanilines: access to (Z)-3-(silylmethylene)indolin-2-ones†

Ya-Fei Han,^a Gui-Fen Lv,^a Yang Li, *\overline{O} *\alpha Li-Jun Wu,*\overline{O} Xuan-Hui Ouyang *\overline{O} and Jin-Heng Li *\overline{O} *\alpha \text{*} \alpha \text{carde}

Received 30th May 2022 Accepted 27th July 2022

DOI: 10.1039/d2sc03009h

rsc.li/chemical-science

A new method involving mild acryl transient-chelating-group-controlled stereoselective Rh(i)-catalyzed silylative aminocarbonylation of 2-alkynylanilines with CO and silanes is presented for producing (Z)-3-(silylmethylene)indolin-2-ones. Upon using an acryl transient chelating group, 2-alkynylanilines undergo an unprecedented alkyne *cis*-silylrhodation followed by aminocarbonylation to assemble (Z)-3-(silylmethylene)indolin-2-ones. Mechanistic studies show that acryl transient chelating effects result in the key alkyne *cis*-silylrhodation process.

Introduction

Oxindoles, including methylene oxindoles, are a class of importantly coveted scaffolds for organic and medicinal chemistry purposes due to their omnipresence in natural products and biologically active molecules, and their widely established utilization as versatile synthetic building blocks. 1,2 In particular, the use of 3-methylene-indolinone scaffolds has already been successfully established for VEGFR, Trk A, CDK, and GSK3 kinase inhibition, antitumor, antibacterial, antiinflammatory, analgesic, and antimalarial applications (Fig. 1).1 As a result, developing efficient methods, especially stereoselective ones, for the synthesis of a diverse range of 3methylene-indolinones is unarguably critical for continued progress in the area of drug development.3-9 Despite this growing demand, the stereoselective construction of the substituted methylene moiety of 3-methylene-oxindoles has been a longstanding challenge and, for these reasons, highly stereoselective preparation methods remain rare to date. Classical approaches for the assembly of methylene oxindoles mainly involve the intermolecular condensation of oxindoles with aryl carbonyl compounds, including diaryl ketones and aromatic formaldehydes, but these transformations face serious stereoselective control issues and substrate scope limitations.1,3 To overcome these issues, transition-metal-catalyzed tandem annulation reactions with unsaturated hydrocarbons,4 such as cross-coupling-enabled annulation cascades of N-(2haloaryl)propiolamides,⁵ N-arylpropiolamides,⁶ or 2-(alkynyl) arylisocyanates;7 the carbonylative annulation of 2-alkynylanilines or 2-alkenylanilines;8 the chloroacylation of alkynetethered carbamoyl chlorides;9 and the cross-dehydrogenation coupling (CDC) of 2,3-diarylacrylamides or N-cinnamoylanilines, 10 have been developed. Common transition-metal catalysts (such as those containing Pd, Rh, Co₂Rh₂, and Ni) are efficient for use in these transformations to access various functionalized 3-methylene-oxindoles; however, the careful

[†] Electronic supplementary information (ESI) available. See https://doi.org/10.1039/d2sc03009h

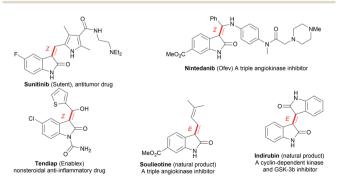


Fig. 1 Selected examples of important 3-methylene-oxindoles.

^aKey Laboratory of Jiangxi Province for Persistent Pollutants Control and Resources Recycle, Nanchang Hangkong University, Nanchang 330063, China. E-mail: liyang8825490@126.com; jhli@hnu.edu.cn

^bCollege of Sciences, Central South University of Forestry and Technology, Changsha, 410004, China. E-mail: lijunwu1105@hnu.edu.cn

Key Laboratory of Medicinal Chemistry for Natural Resource, Ministry of Education, Yunnan Provincial Center for Research & Development of Natural Products, School of Chemical Science and Technology, Yunnan University, Kunming, Yunnan 650091, China

^dState Key Laboratory of Applied Organic Chemistry, Lanzhou University, Lanzhou 730000. China

^{*}School of Chemistry and Chemical Engineering, Henan Normal University, Xinxiang, Henan 453007, China

control of stereoselectivity sometimes remains a problem, with most configurations being unknown before the conclusion of the reaction. Moreover, reports detailing the deliberate control of stereoselectivity are dominated by the introduction of halogen atoms to construct 3-(halogenated methylene) scaffolds; as a result, there is an urgent need to discover conceptually novel stereoselectivity-control strategies for building diverse functionalized scaffolds other than halogenated ones. For example, our group has reported the palladium-catalyzed carbonylative annulation of 2-alkynylanilines with CO for producing 3-(halomethylene)-indolin-2-ones using stoichiometric CuX₂ (X = Br, Cl) as both the halogen source and oxidant (Scheme 1a).8a The stereoselectivity mainly depended on the substrate choice, and the assembly of (E)-3-(halomethylene)indolin-2-ones is limited to 2-(alkylalkynyl)-anilines and sterically bulky 2-(3-substituted arylalkynyl)anilines. Lautens, Schoenebeck, and coworkers disclosed the Pd(0)-catalyzed trans-selective intramolecular chloroacylation of alkynetethered carbamoyl chlorides for assembling (halomethylene)indolin-2-ones, in which both sterically bulky silyl alkynyl substituents (such as TIPS and TBS) and bulky phosphorus ligands (such as phenyl phosphaadamantanes (PA-Ph)) are necessary to precisely direct the stereoselectivity toward (E)-isomers. 9a,b Very recently, Lautens and coworkers found that the use of hexafluoroisopropanol at high temperature (about 100 °C) allowed for the cycloisomerization of alkyne-tethered carbamoyl chlorides to forge only (E)-3-(chloromethylene)oxindoles. 9d This method has the advantage of simple operation and stereospecificity under metal-free conditions, but it is not applicable to sterically bulky TIPS alkynyl substituents. The same group developed a Pd(II)-catalysis-based method to shift the stereoselectivity of the intramolecular chloroacylation of alkyne-tethered carbamoyl chlorides mainly toward the corresponding (Z)-isomers, with Z/E ratios ranging from 3.8:1 to >99: 1.9c Subsequently, they employed a similar Pd(II) catalysis

Our group (2007) PdX₂, CuX₂ C₆H₆/THF, rl NHR³ R1 = alkvl. arv 2 7·1 to > 99·1 F/7 Lautens group (2015, 2016, 2021, 2022) (CF₃)₂CHOH (8-16 equiv) Me (0.2 M), 100 °C PhMe, 50 °C Pd(PPh₃)₄, K₂CO₃ MeCN (0.1 M), 80 °C :1 to >99:1 Z/E b) This work: Stereoselective controlled by transient acryl chelating group (1 atm) За [Rh(cod)Cl]₂ (2 mol%) MeCN, rt, 18 h R3 = CH2=CHCO

Scheme 1 The stereoselective construction of functionalized 3-methylene-oxindoles.

strategy to allow the domino cyclization of alkyne-tethered carbamoyl chlorides with 2-ethynylanilines through linked $C(sp^2)-C(sp^2)$ bond stereospecific formation to access (*Z*)-3-((1*H*-indol-3-yl)methylene)indolin-2-ones. ^{9e} By comparing these findings, ^{8,9} steric hindrance effects and, especially, cooperative ligand/substrate coordination with transition-metal catalysts unarguably play important roles in the stereoselectivity control.

On that basis, we envisioned that if a transient chelating group 11 was present to coordinate with transition-metal catalysts, it may be possible to carefully control the corresponding stereoselectivity. Herein, we report a new method involving the acryl-transient-chelating-group-controlled stereoselective [RhI(-cod)Cl]₂-catalyzed silylative aminocarbonylation of 2-alkynylanilines with CO and silanes, 12 enabling the synthesis of (Z)-3-(silylmethylene)indolin-2-ones in moderate to good yields and with >99:1 Z/E stereoselectivity (Scheme 1b). The method utilizes an in situ generated acryl group on the nitrogen atom as the transient chelating group to coordinate with the active RhI species, thus resulting in unprecedented alkyne cis-silylrhodation followed by aminocarbonylation, providing (Z)-3-(silyl-methylene)indolin-2-ones.

Results and discussion

We began to test our hypothesis that a transient chelating group could control the stereoselectivity during the silylative aminocarbonylation reaction with the use of N-(4-bromobenzyl)-2ethynylaniline 1a, CO, triethylsilane 2a, and acryloyl chloride 3a as starting materials (Table 1). In the presence of 2 mol% [Rh¹(cod)Cl]₂, 2 equiv. of K₂CO₃, and 1 equiv. of chloride 3a, the silylative aminocarbonylation of substrate 1a with CO (1 atm) and triethylsilane 2a at room temperature after 18 h was effiperformed, giving the desired (Z)-1-methyl-3-((triethylsilyl)methylene)indolin-2-one 4aa with 58% yield and >99: 1 Z/E stereoselectivity (Table 1, entry 1). However, omitting the chloride 3a led to a lower yield (30%) and stereoselectivity inversion (1:5 Z/E) (Table 1, entry 2). Gratifyingly, the reaction could be efficiently executed to deliver 75% yield of 4aa in the absence of both K₂CO₃ and chloride 3a, but the stereoselectivity was shifted to 1: 5 Z/E (Table 1, entry 3). Decreasing (0.5 equiv.) or increasing (1.5 equiv.) the amount of chloride 3a resulted in diminished yields (Table 1, entries 4 and 5). Both K2CO3 and [Rh^I(cod)Cl]₂ are essential for this reaction, since the omission of either resulted in no detectable desired product 4aa (Table 1, entries 6 and 11). A brief assessment of the effects of the loading of K₂CO₃ and the effects of the base (K₂CO₃, Na₂CO₃, Cs₂CO₃, $NaHCO_3$, or Et_3N) revealed that the reaction with 2 equiv. of K_2CO_3 afforded the best results (Table 1, entries 1 and 7–10). An evaluation of the Rh loading showed that 2 mol% [Rh^I(cod)Cl]₂ was the best option (Table 1, entry 1 versus entries 12 and 13). A series of Rh salts (Table 1, entries 14-17), including Rh^I(cod)₂-BF₄, Rh^{II}(OAc)₂, [Rh^{II}(CH₃(CH₂)₆CO₂)₂]₂, and [{CP*Rh^{III}Cl₂}₂], were examined; they displayed high catalytic activity but all were less effective than [Rh^I(cod)Cl]₂, partly due to the need for triethylsilane to reduce them and form active RhI species. The solvent (MeCN, CH2Cl2, or DMF) was found to affect the yield

Table 1 Optimization of the reaction conditions^a

Edge Article

Entry	Variation from the standard conditions	Yield ^b /%
1^b	None	58 (>99:1)
2	Without 3a	30 (1:5)
3	Without 3a and K ₂ CO ₃	75 (1 : 5)
4	3a (0.5 equiv.)	40 (>99:1)
5	3a (1.5 equiv.)	11 (>99:1)
6	Without base	Trace
7	K_2CO_3 (1.5 equiv.)	45 (>99:1)
8	K ₂ CO ₃ (2.5 equiv.)	56 (>99:1)
9	Na ₂ CO ₃ instead of K ₂ CO ₃	41 (>99:1)
10	Cs ₂ CO ₃ , NaHCO ₃ , or Et ₃ N instead of K ₂ CO ₃	Trace
11	Without $[Rh(cod)Cl]_2$	0
12	$[Rh(cod)Cl]_2$ (1 mol%)	53
13	[Rh(cod)Cl] ₂ (5 mol%)	60
14	Rh(cod) ₂ BF ₄ instead of [Rh(cod)Cl] ₂	38 (>99:1)
15	Rh(OAc) ₂ instead of [Rh(cod)Cl] ₂	25 (>99:1)
16	[Rh(CH ₃ (CH ₂) ₆ CO ₂) ₂] ₂ instead of [Rh(cod)Cl] ₂	20 (>99:1)
17	$[\{CP*RhCl_2\}_2]$ (2) instead of $[Rh(cod)Cl]_2$	51 (>99:1)
18	CH ₂ Cl ₂ instead of MeCN	55 (>99:1)
19	DMF instead of MeCN	34 (15:1)

^a Standard conditions: **1a** (0.2 mmol), **2a** (0.2 mmol), **3a** (0.2 mmol), [Rh(cod)Cl]₂ (2 mol%), K_2CO_3 (0.4 mmol; 2 equiv.), and MeCN (2 mL), in argon, at room temperature, for 18 h. ^b Isolated yield. The Z/E value is given in parentheses, determined based on GC-MS analysis of the crude product.

and stereoselectivity, and MeCN was shown to be the optimal medium (Table 1, entry 1 *versus* entries 18 and 19).

With the optimized conditions in hand, we set out to further investigate the feasibility of this transient-chelating-group-based strategy (Scheme 2). Directly using N-(4-bromobenzyl)-N-(2-ethynylphenyl)acrylamide **1b** in a reaction with CO, silane **2a**, [Rh(cod)Cl]₂, and K_2CO_3 afforded (Z)-**4aa** with a lower yield

[Rh(cod)Cl]₂ (2 mol%) K₂CO₃ (2 equiv): **4aa**, 22% (>99:1 *Z/E*) vithout K2CO3: 4aa, 43% (>99:1 Z/E)) SiEt₃ [Rh(cod)Cl]₂ (2 mol%) MeCN, rt, 18 h 4ca, 80% (>99:1 Z/E)) TCG-X 3 (1 equiv [Rh(cod)Cl]₂ (2 mol% HSiEt₃ K₂CO₃ (2 equiv) MeCN, rt. 18 h R² = CH₃, **1d** CH2=CHCOCI, 3a 4ca, 84% (>99:1 Z/E) 5daa, trace R² = Bn, **1e** PhCH=CHCOCL 3h 4ea 41% (>99·1 7/F) 5eab 50% (F) R2 = CH3, 1d MeCOCI, 3c 4ca, 20% (>99:1 Z/E) 5dac, trace HCO₂H/(MeCO)₂O,^a 3d $R^2 = CH_3$, 1d 12% (>99:1 Z/E) 5dad, 62% (E)

Scheme 2 Screening the transient chelating groups (3).

(22%) (Scheme 2, eqn (1)), whereas the omission of K₂CO₃ increased the yield of (Z)-4aa to 43%. The results show that the base can improve the acylation process via the removal of chloride ions, but it suppresses the silylative aminocarbonylation. Furthermore, the in-situ-generated transient chelating group process is more efficient than the process involving the direct use of substrate 1b, probably because coordination effects relating to the acryloyl chloride may improve the catalytic activity of the Rh catalyst. Similarly, the treatment of N-(2-ethynylphenyl)-N-methylacrylamide 1c with CO, silane 2a, and $[Rh(cod)Cl]_2$ also afforded the (Z)-isomer 4ca in 80% yield (Scheme 2, eqn (2)); meanwhile acryloyl chloride was found to be the optimal transient-chelating-group reagent and it could efficiently allow the silylative aminocarbonylation of substrate 1d, stereoselectively assembling (Z)-4ca exclusively with a slightly increased yield (84%; Scheme 2, eqn (3), run 1). Using cinnamovl chloride 3b in a reaction with the N-benzylsubstituted substrate **1e** decreased the yield of (Z)-**4ea** to 41%, with 50% yield of the alkyne silylformylation product (E)-5eab (Scheme 2, eqn (3), run 2). Both acetyl chloride 3c and the formyl system 3d13 were less reactive (Scheme 2, eqn (3), runs 3 and 4). Notably, the treatment of the formyl system 3d with substrate 1d, CO, and silane 2a mainly resulted in the alkyne silylformylation product (E)-5dad in 62% yield, with a lower yield (12%) of (Z)-4ca (Scheme 2, eqn (3), run 4). These findings suggest that the cis-silyl vinyl-Rh intermediate may be initially formed via the cis-silylrhodation of the alkyne moiety, followed by the insertion of CO. However, methyl iodide 3e was inert (Scheme 2, eqn (3), run 5).

After establishing the optimal acryl transient chelating group, we next investigated the scope of this Rh-catalyzed stereoselective silylative aminocarbonylation protocol with respect to 2-ethynyl-N-acrylanilines 1 and silanes 2 for the synthesis of (Z)-3-(silylmethylene)indolin-2-ones 4 (Table 2). Various substituents, including benzyl (1e), 4-methoxybenzyl (1f), 4-(trifluoromethyl)benzyl group (1g), cyclopropylmethyl (1h), and allyl (1i) groups, on the nitrogen atom were well tolerated under the optimized conditions, affording the corresponding (Z)-isomers 4ea-ia in moderate to good yields. The substitution effects of the aniline moiety were evaluated (4jana), and the results showed that electronic effects and steric hindrance had no obvious influence on the reaction. 2-Alkynylanilines 1j-l, bearing a 5-Me, 5-F, or 5-Cl group on the aryl ring, could be stereoselectively converted to (Z)-4ja-la, respectively, with yields of 67-82%. Most importantly, the halogen atom, such as F, Cl, and Br, remains intact, so it can serve as a functional handle for further derivatization (4aa, 4ka-la). 2-Alkynylanilines 1m-n possessing an electron-donating Me group or an electron-withdrawing CN group at the 4-position were viable for obtaining (Z)-4ma-na with good yields. A variety of silanes, including trihexylsilane 2b, triisopropylsilane 2c, tert-butyldimethylsilane **2d**, allyldimethyl-silane **2e**, dimethyl(phenyl) silane 2f, methyldiphenylsilane 2g, and triphenylsilane 2h, tolerated the stereoselective silylative aminocarbonylation protocol, attaining (*Z*)-isomers 4db-dh with high yields.

To demonstrate the generality of this silylative aminocarbonylation protocol, we directly used 2-ethynyl-*N*-

 $^{^{}a}$ A solution of HCO $_{2}$ H (3.0 equiv) and (MeCO) $_{2}$ O (3.6 equiv) was stirred at 60 $^{\circ}$ C for 3 h, and then was added to the reaction mixture.

Table 2 The reaction scope in terms of the 2-alkynylaniline (1) and silane $(2)^a$

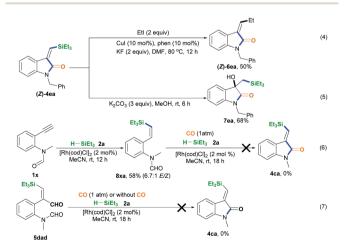
 a Reaction conditions: 1 (0.2 mmol), 2 (0.2 mmol), 3a (0.2 mmol), [Rh(cod)Cl]_2 (2 mol%), $K_2\mathrm{CO}_3$ (0.4 mmol; 2 equiv.), and MeCN (2 mL), in argon, at room temperature, and for 18 h. Some side-products, such as the alkyne silylformylation product 5 and C–N decomposition products, were observed.

acrylanilines **1** to execute silylative aminocarbonylation with CO and silanes **2** (Table 3). In the presence of [Rh(cod)Cl]₂, CO, and silanes **2**, *N*-acrylanilines **1o–r** possessing substituents such as benzyl (**1o**), 4-methoxybenzyl (**1p**), 4-(trifluoromethyl)benzyl (**1q**), cyclopropylmethyl (**1h**), and allyl (**1r**) groups on the

Table 3 The silylative aminocarbonylation of N-(2-ethynylaryl) acrylamide (1) with CO and silanes (2) a

nitrogen atom could be successfully converted to **4ea-ia** with moderate to good yields and >99:1 *Z/E* stereoselectivity. The substitution effects on the aromatic ring of the aniline moiety were investigated, and electron-donating substituents (such as Me; **4ja** and **4ma**) are more efficient than electron-withdrawing ones (such as F, Cl, and CN; **4ka-la** and **4na**). For example, the *N*-acrylanilines **1s** and **1v** possessing an electron-donating Me group at the 4- or 5-position, donating to the aniline moiety efficiently, underwent the reaction to afford **4ja** and **4ma**, respectively, with yields of 76% and 81%, whereas substrate **1w**, having an electron-withdrawing CN group, delivered **4na** with diminished yield (67%). The array of silanes **2b-h** displayed high reactivity when reacting with substrate **1c**, giving 3-(silylmethylene)indolin-2-ones **4db-dh** with good yields.

The synthetic utilization of the Si-containing product (Z)-4ea was conducted (Scheme 3).14 Hiyama coupling of (Z)-4ea with ethyl iodide, CuI, 1,10-phenanthroline (phen), and KF in DMF at 80 °C for 12 h was performed, successfully affording (Z)-4ea with 50% yield (Scheme 3, eqn (4)). Using K_2CO_3 , (Z)-4ea was into 1-benzyl-3-hydroxy-3-((triethylsilyl)methyl) indolin-2-one 7ea with 68% yield (Scheme 3, eqn (5)).1 Some control experiments were performed to understand the mechanism of this silylative aminocarbonylation protocol. Without CO, 2-alkynylaniline $1 \times$ underwent the alkyne hydrosilylation reaction with silane 2a to afford 8xa in 58% yield and with 6.7:1 E/Z stereoselectivity (Scheme 3, eqn (6)). However, substrate 8xa is inert toward the carbonylation reaction in the presence of CO, silane 2a, and [Rh(cod)Cl]₂ (Scheme 3, eqn (6)). These observations suggest that the Si-Rh complex intermediate is initially formed and then addition across the C-C bond generates the silyl vinyl-Rh intermediate, followed by reductive elimination and protonation, ruling out the generation of the alkynehydrosilylation intermediate 8 during this silylative aminocarbonylation process. We found that substrate 5dad treated with silane 2a and [Rh(cod)Cl]₂ in the presence or absence of CO could not afford the desired product 4ca (Scheme 3, eqn (7)), supporting the idea that the silylative aminocarbonylation reaction does not involve the formation of intermediate 5.



Scheme 3 The utilization of (Z)-4ea and control experiments.

 $[^]a$ Standard conditions: 1 (0.2 mmol), 2 (0.2 mmol), CO (1 atm), [Rh(cod) Cl]₂ (2 mol%), and MeCN (2 mL), at room temperature and for 18 h. Some side-products, such as the alkyne silylformylation product 5 and C–N decomposition products, were observed.

Edge Article Chemical Science

Scheme 4 A possible reaction mechanism.

A plausible mechanism for the silylative aminocarbonylation protocol was proposed (Scheme 4).5-12 Oxidative addition of the active RhI species to silane 2a forms the H-RhIII-Si complex intermediate A.12 Subsequently, coordination of the RhIII complex intermediate A with the N-(2-ethynylphenyl)acrylamide intermediate B, which is in situ generated from the reaction of 2alkynylaniline 1 with acryloyl chloride and K₂CO₃, affords the intermediate C. Therein, intermediate C containing an acryl transient chelating group can strongly coordinate with the Rh^{III} species, thus resulting in cis-silylrhodation across the C≡C bond to form the cis-silyl vinyl-Rh^{III} intermediate D.^{11,12} Intermediate D may undergo two pathways for the insertion of CO:5c-f,8,12 One is the direct insertion of CO into the vinyl-Rh bond with the simultaneous formation of a N-Rh bond via the reductive loss of the acryl group with the aid of the base (K₂CO₃)^{12j,k} to generate the carbonyl-Rh^{III}-N six-membered ring intermediate F; the other involves the formation of the vinyl-RhIII-N five-membered ring intermediate E through the reductive decomposition of the acryl C(sp²)-N bond with the aid of the base, 12j,k followed by the insertion of CO to generate the intermediate F. The reductive elimination of intermediate F results in the desired product (Z)-4 and regenerates the active Rh^I species.

Using cinnamoyl chloride $3\mathbf{b}$ as the transient chelating group may afford the alkyne *cis*-addition intermediate D' and the alkyne *trans*-addition intermediate D" due to steric hindrance and electron effects from the cinnamoyl group. The alkyne *cis*-addition intermediate D' undergoes CO insertion, C-N bond cleavage, and N-Rh bond formation to afford the intermediate F, whereas the alkyne classic *trans*-addition intermediate D" may undergo hydroformylation with CO to form (*E*)-5eab. This is because the *in situ* generated cinnamyl C-N bond in the intermediate D" involving conjugative effects is more stable than the acryl C-N bond, leading to no cleavage of the cinnamyl C-N bond.

Conclusions

In summary, we have developed a novel strategy involving a mild acryl transient chelating group for the stereoselective rhodium(1)-catalyzed silylative aminocarbonylation of 2-alkynylanilines with CO and silanes, enabling the formation of (*Z*)-3-

(silylmethylene)indolin-2-ones. The method involves the use of an acryl transient chelating group to enable the unprecedented *cis*-silylrhodation of alkynes and aminocarbonylation cascades to produce (*Z*)-3-(silylmethylene)indolin-2-ones; the highlights include exquisite stereoselectivity, a wide substrate scope, and excellent functional group tolerance. This acryl-transient-chelating-group-controlled stereoselectivity strategy provides a conceptually novel approach for stereoselective transformations of unsaturated hydrocarbons and it could inspire the further development of new and efficient methods for stereoselective synthesis.

Data availability

Experimental data is provided in the ESI.†

Author contributions

L.-J. W. and J.-H. L. conceived and designed the experiments. Y.-F. H. and L.-J. W. carried out most of the experiments. Y.-F. H., Y. L., X.-H. O., L.-J. W. and J.-H. L. analysed the data. Y.-F. H., L.-J. W. and J.-H. L. prepared the manuscript. Y. L., L.-J. W. and J.-H. L. directed the project.

Conflicts of interest

The authors declare no competing financial interests.

Acknowledgements

We thank the National Natural Science Foundation of China (No. 21871126), the Jiangxi Province Science and Technology Project (20202ACB203002) and the Open Research Fund of School of Chemistry and Chemical Engineering, Henan Normal University (No. 2021ZD01) for financial support.

Notes and references

1 For selected examples: (a) B. R. Huck, L. Kötzner and K. Urbahns, Angew. Chem., Int. Ed., 2018, 57, 4412-4428; (b) Z. Ding, M. Zhou and C. Zeng, Arch. Pharm., 2020, 353, 1900367; (c) M. M. Attwood, D. Fabbro, A. V. Sokolov, S. Knapp and H. B. Schiöth, Nat. Rev. Drug Discovery, 2021, 20, 839-861; (d) R. A. Ward, S. Fawell, N. Floc'h, V. Flemington, D. McKerrecher and P. D. Smith, Chem. Rev., 2021, 121, 3297-3351; (e) Y. M. Khetmalis, M. Shivani, Murugesan and K. V. G. C. Sekhar, Biomed. Pharmacother., 2021, 141, 111842; (f) M.-N. Paludetto, C. Bijani, F. Puisset, V. Bernardes-Génisson, C. Arellano and A. Robert, J. Med. Chem., 2018, 61, 7849-7860; (g) G. M. Amaya, R. Durandis, D. S. Bourgeois, J. A. Perkins, A. A. Abouda, K. J. Wines, M. Mohamud, S. A. Starks, R. N. Daniels and K. D. Jackson, Chem. Res. Toxicol., 2018, 31, 570-584; (h) M.-N. Paludetto, J.-L. Stigliani, A. Robert, V. Bernardes-Génisson, E. Chatelut, F. Puisset and C. Arellano, Chem. Res. Toxicol., 2020, 33, 181-190; (i) G. J. Roth, R. Binder, F. Colbatzky, C. Dallinger,

R. Schlenker-Herceg, F. Hilberg, S.-L. Wollin and R. Kaiser, *J. Med. Chem.*, 2015, **58**, 1053–1063; (*j*) T. J. Carty, F. J. Sweeney, R. J. Griffiths, J. D. Eskra, M. J. Ernest, J. S. Pillar, J. D. Cheng, L. D. Loose, P. A. Joseph, P. P. Pazoles, P. F. Moore, A. Nagahisa, S. Murase and S. B. Kadin, *Inflammation Res.*, 1997, **46**, 168–179; (*k*) L. Zhou, J.-S. Yang, X. Wu, J.-H. Zou, X.-D. Xu and G.-Z. Tu, *Heterocycles*, 2005, **65**, 1409–1414; (*l*) G. Cerchiaro and A. M. da Costa Ferreira, *J. Braz. Chem. Soc.*, 2006, **17**, 1473–1485; (*m*) M. Kaur, M. Singh, N. Chadha and O. Silakari, *Eur. J. Med. Chem.*, 2016, **123**, 858–894; (*n*) Y. S. Tingare, C. Su, M.-T. Shen, S.-H. Tsai, S.-Y. Ho and W.-R. Li, *Molecules*, 2020, **25**, 2159; (*o*) J. Wu, J. Chen, H. Huang, S. Li, H. Wu, C. Hu, J. Tang and Q. Zhang, *Macromolecules*, 2016, **49**, 2145–2152.

Chemical Science

- 2 For selected examples: (a) C. Marti and E. M. Carreira, Eur. J. Org. Chem., 2003, 2209–2219; (b) G. S. Singh and Z. Y. Desta, Chem. Rev., 2012, 112, 6104–6155; (c) D. Cheng, Y. Ishihara, B. Tan and C. F. Barbas III, ACS Catal., 2014, 4, 743–762; (d) A. P. Sakia, P. Kansal and N. Shankaraiah, Eur. J. Org. Chem., 2021, 757–772; (e) G. Molteni and A. Silvani, Eur. J. Org. Chem., 2021, 1653–1675; (f) K. Babar, A. F. Zahoor, S. Ahmad and R. Akhtar, Mol. Diversity, 2021, 25, 2487–2532; (g) A. J. Boddy and J. A. Bull, Org. Chem. Front., 2021, 8, 1026–1084; (h) J. Li, W. Lu, Y. Lu, Z. Zha and Z. Wang, Chin. J. Chem., 2022, 40, 195–200; (i) C. V. Galliford and K. A. Scheidt, Angew. Chem., Int. Ed., 2007, 46, 8748–8758; (j) J. Zheng, L. Lin, K. Fu, H. Zheng, X. Liu and X. Feng, J. Org. Chem., 2015, 80, 8836–8842; (k) M. Sohail and F. Tanaka, Angew. Chem., Int. Ed., 2021, 60, 21256–21260.
- 3 For selected examples: (a) A. Millemaggi and R. J. K. Taylor, Eur. J. Org. Chem., 2010, 4527–4547; (b) E. Knoevenagel, Ber. Dtsch. Chem. Ges., 1894, 27, 2345–2346; (c) A. Wahl and P. Bayard, Comptes Rend., 1909, 148, 716; (d) W. Walter, Ber. Dtsch. Chem. Ges., 1902, 35, 1320; (e) L. Sun, N. Tran, F. Tang, H. App, P. Hirth, G. McMahon and C. Tang, J. Med. Chem., 1998, 41, 2588–2603; (f) S. Crotti, N. Di Iorio, C. Artusi, A. Mazzanti, P. Righi and G. Bencivenni, Org. Lett., 2019, 21, 3013–3017; (g) L. Marek, L. Kolman, J. Váňa, J. Svoboda and J. Hanusek, Beilstein J. Org. Chem., 2021, 17, 527–539, and references cited therein.
- 4 For reviews on transition metal catalysis: (a) A. D. Marchese, E. M. Larin, B. Mirabi and M. Lautens, Acc. Chem. Res., 2020, 53, 1605–1619; (b) M. Shrestha, X. Wu, W. Huang, J. Qu and Y. Chen, Org. Chem. Front., 2021, 8, 4024–4045; (c) J.-B. Feng and X.-F. Wu, Top. Heterocycl. Chem., 2016, 42, 25–54; (d) P. Kalck and M. Urrutigoïty, Top. Organomet. Chem., 2018, 61, 69–98; (e) A. Cunillera, C. Godard and A. Ruiz, Top. Organomet. Chem., 2018, 61, 99–144; (f) Rhodium Catalysis in Organic Synthesis: Methods and Reaction, part I, ed. K. Tanaka, Wiley-VCH, Weinheim, 2019.
- 5 (a) R. Yanada, S. Obika, M. Oyama and Y. Takemoto, Org. Lett., 2004, 6, 2825–2828; (b) R. Yanada, S. Obika, T. Inokuma, K. Yanada, M. Yamashita, S. Ohta and Y. Takemoto, J. Org. Chem., 2005, 70, 6972–6975; (c) D. M. D'Souza, F. Rominger and T. J. J. Müller, Angew. Chem., Int. Ed., 2005, 44, 153–158; (d) W. S. Cheung, R. J. Patch and M. R. Player, J. Org. Chem., 2005, 70, 3741–

- 3744; (e) W. M. Dai, J. Shi and J. Wu, *Synlett*, 2008, 2716–2720; (f) R. Shintani, T. Yamagami and T. Hayashi, *Org. Lett.*, 2006, **8**, 4799–4801; (g) M. Bararjanian, S. Hosseinzadeh, S. Balalaie and H. R. Bijanzadeh, *Tetrahedron*, 2011, **67**, 2644–2650; (h) M. Bararjanian, S. Hosseinzadeh, S. Balalaie, H. R. Bijanzadeh and E. Wolf, *Tetrahedron Lett.*, 2011, **52**, 3329–3332.
- 6 (a) A. Pinto, L. Neuville, P. Retailleau and J. Zhu, Org. Lett., 2006, **8**, 4927–4930; (b) A. Pinto, L. Neuville and J. Zhu, Angew. Chem., Int. Ed., 2007, 46, 3291-3295; (c) A. Pinto, L. Neuville and J. Zhu, Tetrahedron Lett., 2009, 50, 3602-3605; (d) R.-J. Song, Y. Liu, R.-J. Li and J.-H. Li, Tetrahedron Lett., 2009, 50, 3912-3916; (e) S. Tang, P. Peng, S.-F. Pi, Y. Liang, N.-X. Wang and J.-H. Li, Org. Lett., 2008, 10, 1179–1182; (f) S. Tang, P. Peng, Z.-O. Wang, B.-X. Tang, C.-L. Deng, J.-H. Li, P. Zhong and N.-X. Wang, Org. Lett., 2008, 10, 1875-1878; (g) S. Tang, P. Peng, P. Zhong and J.-H. Li, J. Org. Chem., 2008, 73, 5476–5480; (h) P. Peng, B.-X. Tang, S.-F. Pi, Y. Liang and J.-H. Li, J. Org. Chem., 2009, 74, 3569–3572; (i) T.-S. Jiang, R.-Y. Tang, X.-G. Zhang, X.-H. Li and J.-H. Li, J. Org. Chem., 2009, 74, 8834–8837; (j) H.-L. Wang, Z. Li, G.-W. Wang and S.-D. Yang, Chem. Commun., 2011, 47, 11336-11338.
- 7 (a) M. R. Fielding, R. Grigg and C. J. Urch, Chem. Commun., 2000, 2239–2240; (b) S. Kamijo, Y. Sasaki, C. Kanazawa, T. Schüsseler and Y. Yamamoto, Angew. Chem., Int. Ed., 2005, 44, 7718–7721; (c) T. Miaura, T. Toyoshima, Y. Takahashi and M. Murakami, Org. Lett., 2008, 10, 4887–4889; (d) T. Miura, Y. Takahashi and M. Murakami, Org. Lett., 2007, 9, 5075–5077; (e) T. Miura, T. Toyoshima, Y. Takahashi and M. Murakami, Org. Lett., 2009, 11, 2141–2143; (f) G. Cantagrel, B. Carné-Carnavalet, C. Meyer and J. Cossy, Org. Lett., 2009, 11, 4262–4265.
- 8 (*a*) S. Tang, Q.-F. Yu, P. Peng, J.-H. Li, P. Zhong and R.-Y. Tang, *Org. Lett.*, 2007, **9**, 3413–3416; (*b*) J. H. Park, E. Kim and Y. K. Chung, *Org. Lett.*, 2008, **10**, 4719–4721; (*c*) R. Shi, L. Lu, H. Zhang, B. Chen, Y. Sha, C. Liu and A. Lei, *Angew. Chem., Int. Ed.*, 2013, **52**, 10582–10585; (*d*) X.-H. Yang, K. Li, R.-J. Song and J.-H. Li, *Eur. J. Org. Chem.*, 2014, 616–623.
- 9 (a) C. M. Le, X. Hou, T. Sperger, F. Schoenebeck and M. Lautens, Angew. Chem., Int. Ed., 2015, 54, 15897–15900;
 (b) T. Sperger, C. M. Le, M. Lautens and F. Schoenebeck, Chem. Sci., 2017, 8, 2914–2922; (c) C. M. Le, T. Sperger, R. Fu, X. Hou, Y. H. Lim, F. Schoenebeck and M. Lautens, J. Am. Chem. Soc., 2016, 138, 14441–14448; (d) J. F. Rodríguez, A. Zhang, J. Bajohr, A. Whyte, B. Mirabi and M. Lautens, Angew. Chem., Int. Ed., 2021, 60, 18478–18483; (e) R. Arora, J. F. Rodríguez, A. Whyte and M. Lautens, Angew. Chem., Int. Ed., 2022, 61, e202112288.
- 10 (a) M. Wasa and J.-Q. Yu, J. Am. Chem. Soc., 2008, 130, 14058–14059; (b) S. Ueda, T. Okada and H. Nagasawa, Chem. Commun., 2010, 46, 2462–2464; (c) M. S. L. Kumar, S. K. Manna, S. Maji and G. Panda, Org. Biomol. Chem., 2017, 15, 1762–1766; (d) X. Luo, Q. Zhang, Y. Jiang, C. Wang, X. Song, J. Li, Q. Yan, A. S. C. Chan and Y. Zou, J. Org. Chem., 2021, 86, 6698–6710.

11 For selected recent examples: (a) M. I. Lapuh, S. Mazeh and T. Besset, ACS Catal., 2020, 10, 12898–12919; (b) G. Liao, T. Zhang, Z.-K. Lin and B.-F. Shi, Angew. Chem., Int. Ed., 2020, 59, 19773–19786; (c) B. Liu, A. M. Romine, C. Z. Rubel, K. M. Engle and B.-F. Shi, Chem. Rev., 2021, 121, 14957–15074; (d) C. Jacob, B. U. W. Maes and G. Evano, Chem.-Eur. J., 2021, 27, 13899–13952.

Edge Article

- 12 For a review and selected papers: (a) Z. Xu, W.-S. Huang, J. Zhang and L.-W. Xu, Synthesis, 2015, 47, 3645–3668; (b) D. P. Drolet and A. J. Lees, J. Am. Chem. Soc., 1992, 114, 4186–4194; (c) K. Ezbiansky, P. I. Djurovich, M. LaForest, D. J. Sinning, R. Zayes and D. H. Berry, Organometallics, 1998, 17, 1455–1457; (d) N. Dunwoody, S.-S. Sun and A. J. Lees, Inorg. Chem., 2000, 39, 4442–4451; (e) T. Ureshino, T. Yoshida, Y. Kuninobu and K. Takai, J. Am. Chem. Soc., 2010, 132, 14324–14326; (f) Y. Kuninobu, K. Yamauchi, N. Tamura, T. Seiki and K. Takai, Angew. Chem., Int. Ed., 2013, 52, 1520–1522; (g) C. Cheng and
- J. F. Hartwig, *J. Am. Chem. Soc.*, 2014, **136**, 12064–12072; (h) K. Mitsudo, S. Tanaka, R. Isobuchi, T. Inada, H. Mandai, T. Korenaga, A. Wakamiya, Y. Murata and S. Suga, *Org. Lett.*, 2017, **19**, 2564–2567; (i) Y.-H. Huang, Y. Wu, Z. Zhu, S. Zheng, Z. Ye, Q. Peng and P. Wang, *Angew. Chem., Int. Ed.*, 2022, **61**, e202113052; (j) S. Sharma, S. Han, M. Kim, N. K. Mishra, J. Park, Y. Shin, J. Ha, J. H. Kwak, Y. H. Jung and I. S. Kim, *Org. Biomol. Chem.*, 2014, **12**, 1703–1706; (k) X. Wu, Y. Lu, J. Qiao, W. Dai, X. Jia, H. Ni, X. Zhang, H. Liu and F. Zhao, *Org. Lett.*, 2020, **22**, 9163–9168; (l) S. Ochiai, A. Sakai, Y. Usuki, B. Kang, T. Shinada and T. Satoh, *Chem. Lett.*, 2021, **50**, 585–588.
- 13 K. Takamoto, T. Shohei Ohno, S. Ohno, N. Hyogo, H. Fujioka and M. Arisawa, *J. Org. Chem.*, 2017, **82**, 8733–8742.
- 14 (*a*) M. Oishi, H. Kondo and H. Amii, *Chem. Commun.*, 2009, 1909–1911; (*b*) T. Komiyama, Y. Minami and T. Hiyama, *ACS Catal.*, 2017, 7, 631–651.