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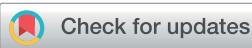


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EDGE ARTICLES

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Enantiomeric β -sheet peptides from A β form homochiral pleated β -sheets rather than heterochiral rippled β -sheets
Visualizing the mode of action and supramolecular assembly of teixobactin analogues in *Bacillus subtilis*



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Enantiomeric β -sheet peptides from $\text{A}\beta$ form homochiral pleated β -sheets rather than heterochiral rippled β -sheets†

Xingyue Li, ^a Stephanie E. Rios ^a and James S. Nowick ^{ab}

In 1953, Pauling and Corey postulated "rippled" β -sheets, composed of a mixture of D - and L -peptide strands, as a hypothetical alternative to the now well-established structures of "pleated" β -sheets, which they proposed as a component of all- L -proteins. Growing interest in rippled β -sheets over the past decade has led to the development of mixtures of D - and L -peptides for biomedical applications, and a theory has emerged that mixtures of enantiomeric β -sheet peptides prefer to co-assemble in a heterochiral fashion to form rippled β -sheets. Intrigued by conflicting reports that enantiomeric β -sheet peptides prefer to self-assemble in a homochiral fashion to form pleated β -sheets, we set out to address this controversy using two β -sheet peptides derived from $\text{A}\beta_{17-23}$ and $\text{A}\beta_{30-36}$, peptides **1a** and **1b**. Each of these peptides self-assembles to form tetramers comprising sandwiches of β -sheet dimers in aqueous solution. Through solution-phase NMR spectroscopy, we characterize the different species formed when peptides **1a** and **1b** are mixed with their respective D -enantiomers, peptides *ent*-**1a** and *ent*-**1b**. ^1H NMR, DOSY, and ^1H , ^{15}N -HSQC experiments reveal that mixing peptides **1a** and *ent*-**1a** results in the predominant formation of homochiral tetramers, with a smaller fraction of a new heterochiral tetramer, and mixing peptides **1b** and *ent*-**1b** does not result in any detectable heterochiral assembly. ^{15}N -edited NOESY reveals that the heterochiral tetramer formed by peptides **1a** and *ent*-**1a** is composed of two homochiral dimers. Collectively, these NMR studies of $\text{A}\beta$ -derived peptides provide compelling evidence that enantiomeric β -sheet peptides prefer to self-assemble in a homochiral fashion in aqueous solution.

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Introduction

Do enantiomeric β -sheet peptides prefer to self-assemble in a homochiral fashion or co-assemble in a heterochiral fashion? In the early 1950s, Pauling and Corey introduced the terms "pleated" β -sheets and "rippled" β -sheets to describe two types of β -sheet assembly.¹⁻³ In both a pleated β -sheet and a rippled β -sheet, adjacent peptide strands hydrogen bond through edge-to-edge interactions (Fig. 1). Pleated β -sheets are composed of peptide strands of the same chirality (all L -peptide strands or all D -peptide strands), while peptide strands of opposite chirality (L -peptide strands and D -peptide strands) are required to form rippled β -sheets. In a pleated β -sheet, the side chains of sequential residues are oriented up-down-up-down and those of the adjacent peptide strands are also oriented up-down-up-down (Fig. 1A). In a rippled

β -sheet, however, the side chains of sequential residues are oriented up-down-up-down and those of the adjacent peptide strands are oriented down-up-down-up (Fig. 1B). Although pleated β -sheets are a near ubiquitous feature of proteins, rippled β -sheets are not found in nature, because ribosomal proteins and peptides are composed only of L -amino acids.

Rippled β -sheets, formed by mixing D - and L -peptides, have recently attracted considerable interest as biomaterials and for other biomedical applications.⁴⁻⁶ Schneider and co-workers demonstrated the effects of chirality with the hydrogel-forming peptide MAX1. When MAX1 was mixed with an equimolar amount of its enantiomer, the resulting hydrogel showed four times greater rigidity than that of the enantiopure MAX1 peptide.^{7,8} Nilsson and co-workers demonstrated by isotope-edited IR spectroscopy and FRET studies that mixtures of enantiomeric peptides L -Ac-(FKFE)₂-NH₂ and D -Ac-(FKFE)₂-NH₂ form rippled β -sheet fibrils. The authors further demonstrated by isothermal titration calorimetry (ITC) that the resulting heterochiral assembly is more thermodynamically favored than the homochiral assembly.⁹ In a subsequent paper, the authors reported that the hydrogel formed by the heterochiral rippled β -sheets is stronger and more resistant to proteolytic degradation than the hydrogel formed by the homochiral L -pleated β -sheets.¹⁰

^aDepartment of Chemistry, University of California Irvine, 4126 Natural Sciences I, Irvine, CA 92697-2025, USA. E-mail: jsnowick@uci.edu

^bDepartment of Pharmaceutical Sciences, University of California Irvine, 4126 Natural Sciences I, Irvine, CA 92697-2025, USA

† Electronic supplementary information (ESI) available: Details of peptide synthesis, Fmoc-protection of amino acids, NMR spectroscopic studies, and peptide characterization data. See <https://doi.org/10.1039/d2sc02080g>



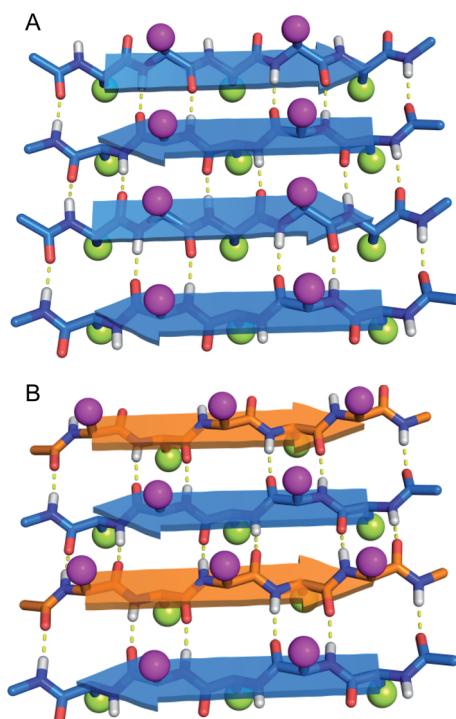


Fig. 1 Molecular models of (A) pleated β -sheets formed from L-peptide strands (blue) and (B) rippled β -sheets formed from a mixture of L-peptide and D-peptide strands (orange). Amino acid side chains are depicted as sticks, with magenta balls on the top face and yellow balls on the bottom face of the β -sheets.

The co-assembly of enantiomeric β -sheet peptides is not limited to designed biomaterials, and has also been used to characterize and study fibril formation of $\text{A}\beta_{40}$ and $\text{A}\beta_{42}$. Nilsson and co-workers demonstrated by isotope-edited IR spectroscopy and solid-state NMR spectroscopy that L- and D- $\text{A}\beta_{16-22}$ heptapeptides co-assemble to give rippled β -sheets and showed that the heterochiral assembly is more thermodynamically favorable.¹¹ Raskatov and co-workers observed that mixing D- $\text{A}\beta_{42}$ with L- $\text{A}\beta_{42}$ led to accelerated non-toxic fibril formation and attenuated cytotoxicity by suppressing oligomer formation.¹² Raskatov subsequently proposed “ $\text{A}\beta$ chiral inactivation” as a potential therapeutic strategy for Alzheimer’s disease.^{6,13} Recently, structures of the rippled β -sheet assembly have been reported. Tycko and Raskatov used solid-state NMR spectroscopy to elucidate ¹⁵N, ¹³C-labeled D,L- $\text{A}\beta_{40}$ fibril polymorphs in rippled β -sheets consisting of three different registrations in the hydrogen-bonded antiparallel alignment.¹⁴ Raskatov and co-workers also reported the X-ray crystallographic structure of a rippled β -sheet formed from a mixture of L- and D-triphenylalanine.¹⁵ DFT calculations have further supported a model in which heterochiral rippled β -sheets are favored over homochiral pleated β -sheets.^{16,17} From these studies, a theory has emerged in which mixtures of enantiomeric β -sheet peptides are thought to prefer to co-assemble in a heterochiral fashion to form rippled β -sheets, rather than self-assemble in a homochiral fashion to form pleated β -sheets.

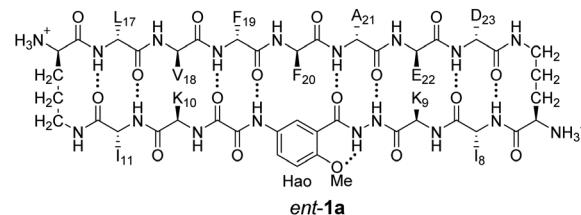
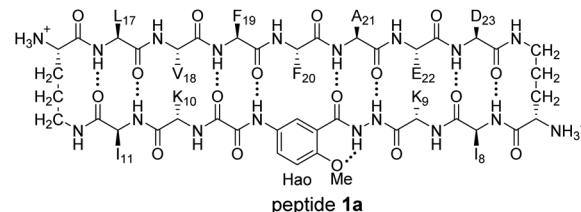
In 2004, our laboratory reported that enantiomeric β -sheet pentapeptides strongly prefer to form homochiral pleated β -sheet dimers in CDCl_3 solution, rather than heterochiral rippled β -

sheet dimers, with a selectivity of 3.1–4.2 kcal mol⁻¹.¹⁸ Recently Gellman and co-workers have studied homochiral and heterochiral β -sheet formation in aqueous solution using a β -hairpin model system and have found that peptides containing homochiral peptide strands fold to form β -hairpins, while peptides containing heterochiral peptide strands do not.¹⁹ Intrigued by the conflicting reports of preferred homochiral and heterochiral β -sheet assembly, we set out to reconcile these findings using a minimal aqueous model system that recapitulates both the edge-to-edge hydrogen-bonding interactions that occur in β -sheet formation and additional face-to-face packing interactions that occur in gel and fibril formation. The model system consists of two well characterized β -sheet peptides derived from $\text{A}\beta_{17-23}$ and $\text{A}\beta_{30-36}$, peptides **1a** and **1b**.^{20,21} Peptides **1a** and **1b** both form tetramers comprising sandwiches of β -sheet dimers. Using NMR spectroscopy, we identify and characterize the different tetramers formed by mixing peptides **1a** and **1b** with their respective D-enantiomers, *ent*-**1a** and *ent*-**1b**. Through these studies, we find that homochiral pairing to form pleated β -sheets is preferred over heterochiral pairing to form rippled β -sheets.

Results and discussion

¹H NMR spectroscopy shows that mixing peptides **1a** and *ent*-**1a** gives a new assembly

Enantiomerically pure peptide **1a** forms a homochiral tetramer in aqueous solution at millimolar concentrations. Peptide **1a** is a macrocyclic β -hairpin peptide containing two heptapeptide strands linked by two δ Orn turn units.²² The upper strand of peptide **1a** is derived from $\text{A}\beta_{17-23}$, and the lower strand contains a Hao amino acid flanked by two dipeptides to promote solubility and prevent uncontrolled aggregation.²³ The tetramer formed by peptide **1a** consists of a sandwich of β -sheet dimers. Hydrogen-bonding interactions between the edges of the β -strands stabilize the dimers, and hydrophobic packing of the side chains further stabilizes the tetrameric assembly.



The ¹H NMR spectrum of peptide **1a** at 8.0 mM in D_2O at 298 K displays a predominant set of resonances associated with a homochiral tetramer, and a smaller set of resonances (4%) associated with the monomer. When peptides **1a** and *ent*-**1a** are



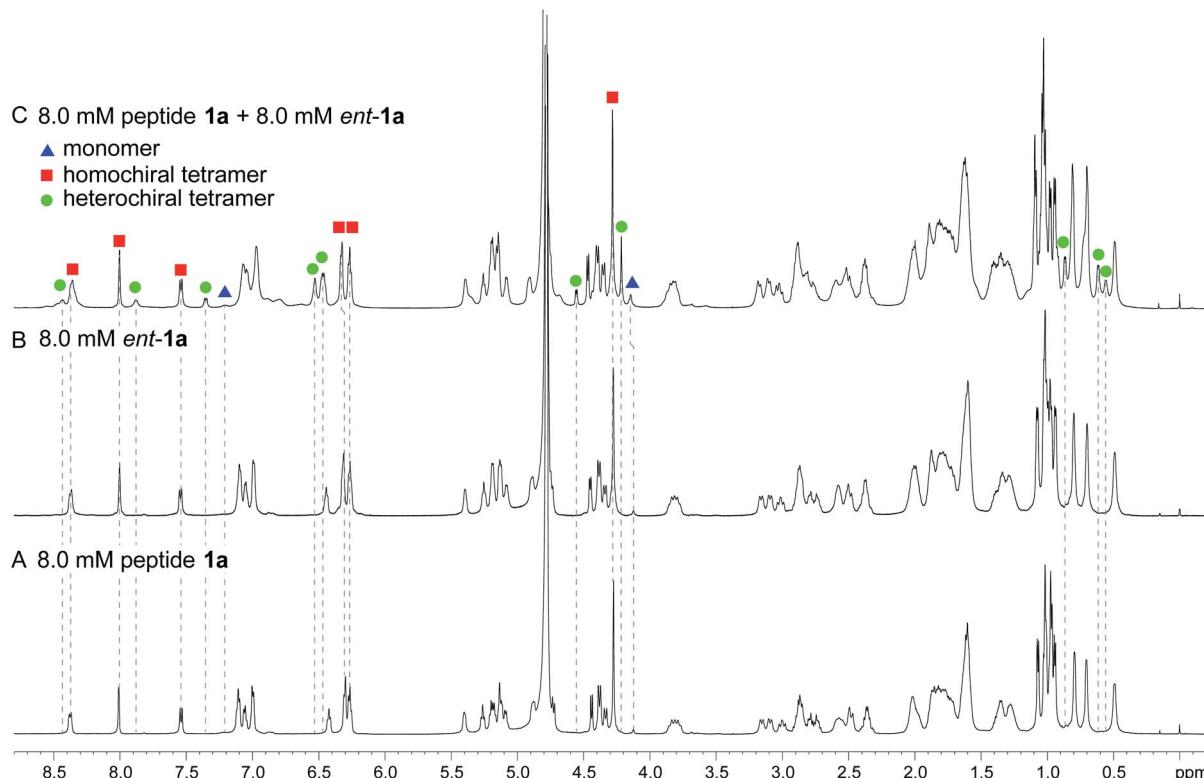
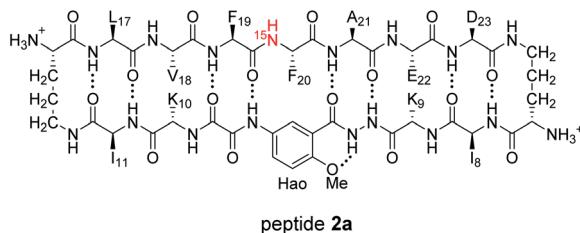


Fig. 2 ^1H NMR spectra of (A) 8.0 mM peptide **1a**, (B) 8.0 mM peptide **ent-1a**, and (C) 8.0 mM peptide **1a** and 8.0 mM peptide **ent-1a** in D_2O at 600 MHz and 298 K with 0.06 mM DSA as an internal standard.²⁵ Dashed lines mark key resonances associated with the monomer, homochiral tetramer, and heterochiral tetramer. These resonances are designated as follows: blue triangle, monomer; red square, homochiral tetramer; green circle, heterochiral tetramer.

mixed in equal concentrations (16.0 mM total), new resonances (29%) emerge that were previously unobserved for each enantiopure peptide (Fig. 2). An EXSY experiment at 328 K shows that these new resonances exchange with the homochiral tetramer and monomer and thus correspond to a new heterochiral assembly (Fig. S1–S4†).²⁴

^1H , ^{15}N HSQC and DOSY studies reveal a heterochiral tetramer

HSQC studies using ^{15}N isotopic labeling corroborate the formation of the new assembly observed in the 1D ^1H NMR spectrum. ^1H , ^{15}N HSQC experiments give a unique crosspeak for each species containing an ^{15}N isotope, readily allowing the identification of the different isotopically labeled species present.^{20,21} We thus prepared an isotopologue of peptide **1a** containing an ^{15}N label on Phe₂₀—peptide **2a**—and studied its mixture with peptide **ent-1a** by ^1H , ^{15}N HSQC.



The ^1H , ^{15}N HSQC spectrum of 8.0 mM peptide **2a** in 9 : 1 H_2O – D_2O solution shows two crosspeaks—one associated

with the homochiral tetramer and the other with the monomer (Fig. 3A).²⁰ The tetramer has a crosspeak that appears at 8.56 ppm in the ^1H dimension and 121.4 ppm in the ^{15}N dimension; the monomer has a crosspeak that appears at 8.30 ppm in the ^1H dimension and 122.8 ppm in the ^{15}N dimension. When peptide **2a** is mixed with peptide **ent-1a** (16.0 mM total), the crosspeak of the monomer is no longer observed, and a new crosspeak appears at 8.63 ppm in the ^1H dimension and 122.5 ppm in the ^{15}N dimension (Fig. 3B). This crosspeak is not observed in the ^1H , ^{15}N HSQC spectrum of the enantiomerically pure peptide **2a** and is thus associated with the formation of a heterochiral species. The weaker intensity of this new crosspeak indicates that the homochiral

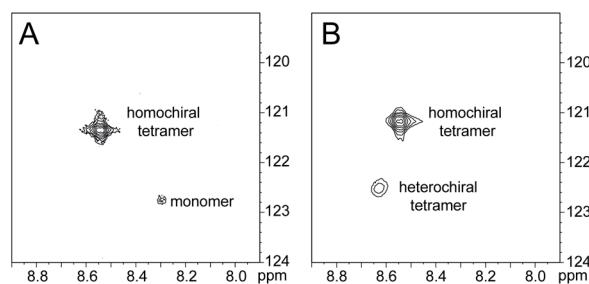


Fig. 3 ^1H , ^{15}N HSQC spectra of (A) 8.0 mM peptide **2a** and (B) 8.0 mM peptide **1a** and 8.0 mM peptide **ent-1a** in 9 : 1 H_2O – D_2O at 500 MHz and 298 K.

tetramer forms preferentially under the conditions of the experiment.

Diffusion-ordered spectroscopy (DOSY) studies suggest that the new heterochiral species is a tetramer. The DOSY spectrum of 8.0 mM peptide **1a** in D₂O shows a diffusion coefficient of $11.6 \pm 0.9 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ for the tetramer (Fig. S5†). This value is similar to what we have previously reported for peptide **1a** at 8.0 mM and 298 K.²⁰ In the DOSY spectrum of the mixture of peptides **1a** and *ent*-**1a** (8.0 mM each), the resonances corresponding to the homochiral tetramer show a diffusion coefficient of $10.7 \pm 0.7 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$, and the resonances corresponding to the heterochiral tetramer show a diffusion coefficient of $10.2 \pm 0.7 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ (Fig. S6†). The small differences among the diffusion coefficients may reflect transient non-specific interactions among the tetramers at the higher concentration (16.0 mM total) of the mixing experiment leading to a lower diffusion coefficient.²⁶

Supramolecular assembly of the heterochiral tetramer

Through NOESY studies of peptide **1a**, our laboratory previously established that the tetramer formed by peptide **1a** consists of sandwiches of β -sheet dimers.^{20,21} The heterochiral tetramer formed by peptides **1a** and *ent*-**1a** can adopt a similar structure,

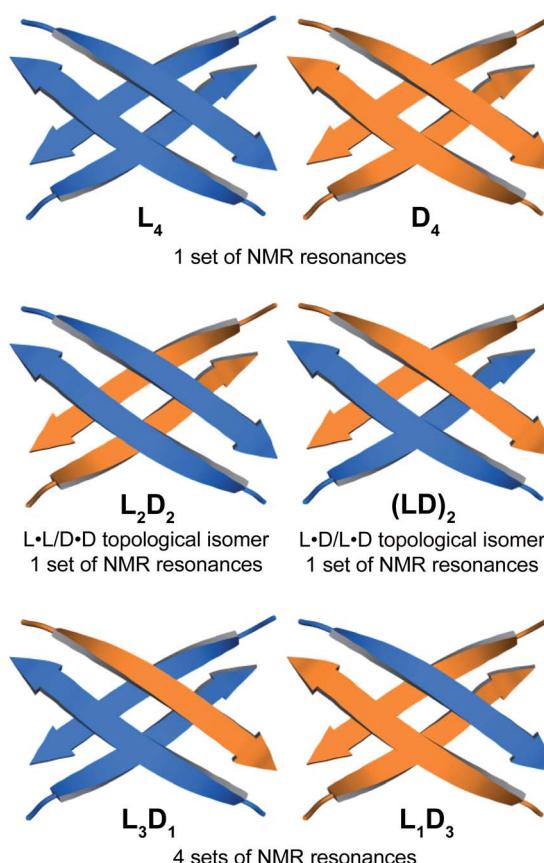


Fig. 4 Cartoon illustrations representing the possible combinations of homochiral and heterochiral tetramers formed from a mixture of peptides **1a** and *ent*-**1a**. Macrocylic peptide **1a** is represented by a blue arrow and macrocylic peptide *ent*-**1a** is represented by an orange arrow.

in which two dimers form a sandwich-like tetramer. Peptides **1a** and *ent*-**1a** can come together in two different ways to form heterochiral tetramers in 2 : 2 stoichiometry—either as an L₂D₂ topological isomer, in which the sandwich consists of L·L and D·D homochiral dimers, or as an (LD)₂ topological isomer, in which the sandwich consists of two L·D heterochiral dimers (Fig. 4). The L₂D₂ tetramer should give one set of resonances in the ¹H NMR spectrum, distinct from those of the L₄ and D₄ homochiral tetramers, which collectively should give one set of resonances. The (LD)₂ tetramer should also give one set of resonances in the ¹H NMR spectrum. Peptides **1a** and *ent*-**1a** could also come together to give heterochiral tetramers in 3 : 1 and 1 : 3 stoichiometry, L₃D₁ and L₁D₃, which should give four sets of resonances in the ¹H NMR spectrum. The observation of a single set of new resonances in the ¹H NMR spectra of the mixture thus indicates the formation of a single heterochiral tetramer with a 2 : 2 stoichiometry as either the L₂D₂ or the (LD)₂ topological isomer.

¹⁵N-edited NOESY studies reveal homochiral dimers within the heterochiral tetramer

The ¹⁵N-edited NOESY spectrum of ¹⁵N-labeled enantiomerically pure peptide **2a** shows three NOE crosspeaks associated with close contacts in the homochiral tetramer (Fig. 5B). The Phe₂₀ ¹⁵NH proton of the tetramer shows an interstrand NOE with the Ala₂₁ α -proton in its dimerization partner, as well as

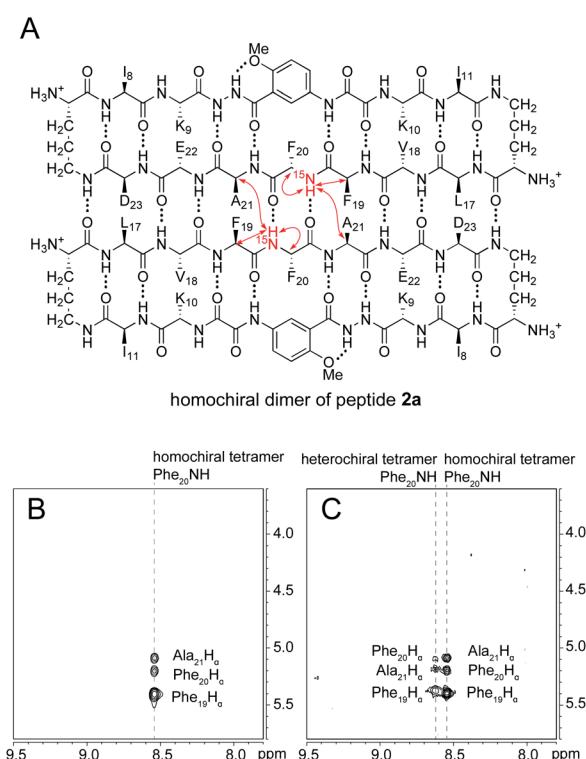


Fig. 5 (A) Structure of the homochiral dimer subunit of the tetramer formed from ¹⁵N-labeled peptide **2a**. Red arrows represent NOEs observed in the ¹⁵N-edited NOESY spectrum. (B) ¹⁵N-edited NOESY spectrum of 8.0 mM peptide **2a** in 9 : 1 H₂O-D₂O at 500 MHz and 298 K. (C) ¹⁵N-edited NOESY spectrum of 8.0 mM peptide **2a** and 8.0 mM peptide *ent*-**1a** in 9 : 1 H₂O-D₂O at 500 MHz and 298 K.



a strong intrastrand NOE with the Phe₁₉ α -proton and a weaker intrastrand NOE with the Phe₂₀ α -proton. This pattern of NOEs is characteristic of the proximities observed in β -sheet structure (Fig. 5A).^{28,29}

When ¹⁵N-labeled peptide **2a** is mixed with unlabeled peptide **ent-1a**, a new set of weaker NOE crosspeaks associated with the heterochiral tetramer emerges, in addition to the NOE crosspeaks associated with the homochiral tetramer (Fig. 5C). In the set of NOE crosspeaks from the heterochiral tetramer, the Phe₂₀ ¹⁵NH proton shows an interstrand NOE with the Ala₂₁ α -proton in its dimerization partner, as well as a relatively strong intrastrand NOE with the Phe₁₉ α -proton and a weaker intrastrand NOE with the Phe₂₀ α -proton. Although the observation of a new set of crosspeaks establishes the formation of a heterochiral tetramer, it does not distinguish between the L₂D₂ and the (LD)₂ topological isomers.

To differentiate between the L₂D₂ and the (LD)₂ topological isomers, we strategically incorporated two deuterated residues (*d*₈-Phe₁₉ and *d*₄-Ala₂₁) into ¹⁵N-labeled peptide **2a**, to create peptide **3a**, and we studied its interaction with unlabeled **ent-1a** by ¹⁵N-edited NOESY experiments. A homochiral dimer in which peptide **3a** is paired with itself should not exhibit an interstrand NOE between the ¹⁵NH proton of Phe₂₀ and the α -proton of *d*₄-Ala₂₁, because the α -proton has been replaced with deuterium (Fig. 6A). In contrast, a heterochiral dimer in which peptide **3a** is paired

with peptide **ent-1a** should exhibit an interstrand NOE between the ¹⁵NH proton of Phe₂₀ in peptide **3a** and the α -proton of Ala₂₁ in peptide **ent-1a** (Fig. 6B).

¹⁵N-Edited NOESY studies of the mixture of peptides **3a** and **ent-1a** reveal a weak crosspeak associated with a heterochiral tetramer composed of homochiral dimers, in addition to crosspeaks associated with the homochiral tetramer. Enantiomerically pure peptide **3a** exhibits the expected intrastrand NOE crosspeak between the ¹⁵NH proton and the α -proton of Phe₂₀ and an unexpected weaker NOE crosspeak to the α -proton of *d*₈-Phe₁₉ (Fig. 7A). This weaker NOE results from incomplete deuterium labeling at the α -position of the *d*₈-Phe₁₉.³⁰ In the mixture of peptides **3a** and **ent-1a**, a new NOE crosspeak between the ¹⁵NH proton and the α -proton of Phe₂₀ associated with the heterochiral tetramer is observed (Fig. 7B). No additional NOE crosspeaks are observed for the heterochiral tetramer. The presence of only an intrastrand NOE crosspeak in the ¹⁵N-edited NOESY spectrum of the mixture indicates that peptides **3a** and **ent-1a** are not dimerization partners and shows that the heterochiral tetramer is composed of two homochiral dimer subunits.

Collectively the 1D, ¹H, ¹⁵N HSQC, DOSY, and ¹⁵N-edited NOESY studies establish that peptides **1a** and **ent-1a** preferentially form the L₄ and D₄ homochiral tetramers, in addition to a smaller amount of the L₂D₂ heterochiral tetramer (Fig. 4). The formation of the L₂D₂ heterochiral tetramer rather than the (LD)₂ heterochiral tetramer demonstrates that even within heterochiral assemblies, enantiomeric β -sheet peptides prefer to self-assemble in a homochiral fashion. Thus, the formation of the L₂ and D₂ pleated β -sheets is preferred over the formation of the LD rippled β -sheets.

¹H NMR spectroscopy shows that mixing peptides **1b** and **ent-1b** does not result in heterochiral assembly

To further assess the preferences for homochiral or heterochiral assembly using a different β -sheet peptide, we studied the assembly of peptides **1b** and **ent-1b**.²⁰ Peptide **1b** is a homologue of peptide **1a** that contains Ab_{30–36} instead of Ab_{17–23}. We had previously found that peptide **1b** also assembles to form a tetramer at millimolar concentrations, albeit with an equilibrium that favors the tetramer less strongly.

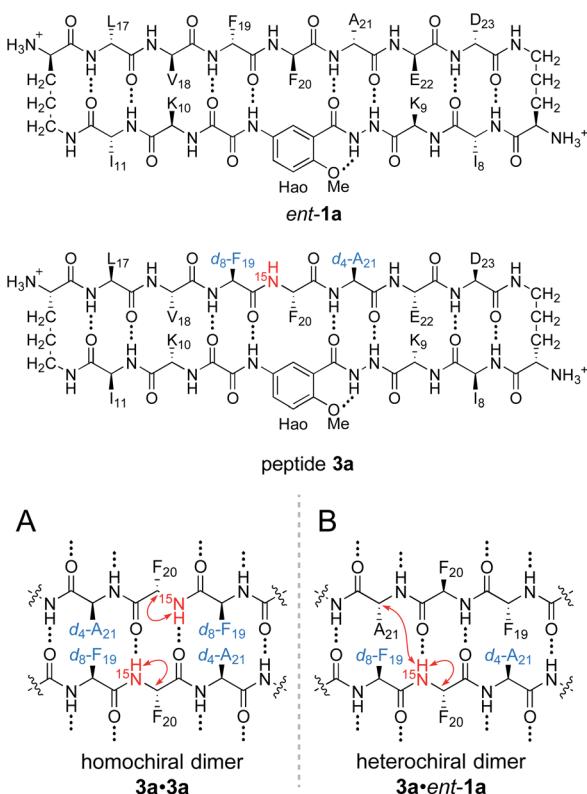


Fig. 6 Structures of peptide **ent-1a** and triply labeled peptide **3a**. (A) Expected NOEs in the ¹⁵N-edited NOESY spectrum of a homochiral dimer subunit. (B) Expected NOEs in the ¹⁵N-edited NOESY spectrum of a heterochiral dimer subunit.

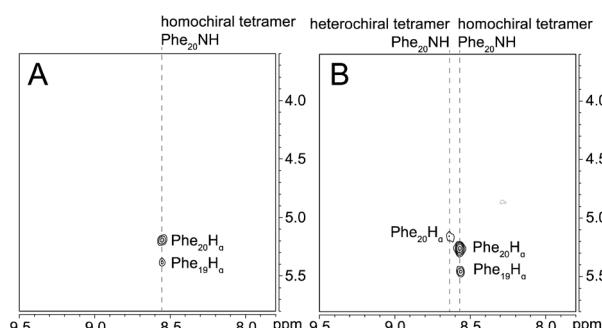
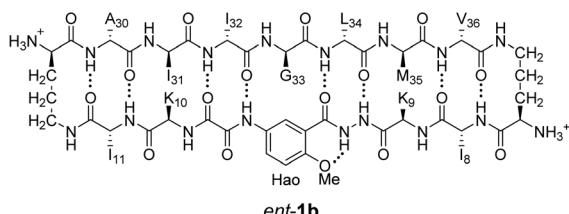
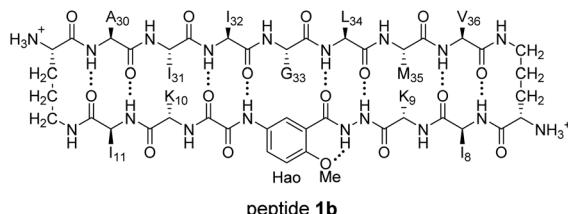


Fig. 7 ¹⁵N-Edited NOESY spectra of (A) 8.0 mM triply labeled peptide **3a** and (B) 8.0 mM peptide **3a** and 8.0 mM unlabeled peptide **ent-1a** in 9 : 1 H₂O–D₂O at 500 MHz and 298 K.





When peptide **1b** is mixed with peptide *ent*-**1b**, no evidence of heterochiral tetramer formation is observed. The ^1H NMR spectrum of peptide **1b** at 4.0 mM in D_2O at 298 K displays sets

of resonances associated with both the monomer and the homochiral tetramer (Fig. 8A). The 4.0 mM ^1H NMR spectrum of peptide *ent*-**1b** is identical to that of peptide **1b** (Fig. 8B). At 8.0 mM, the spectrum of peptide **1b** displays a shift in equilibrium toward the tetramer (Fig. 8C). The spectrum broadens slightly, suggesting exchange between the monomer and tetramer on an intermediate timescale (*ca.* 10^{-1} s) or additional non-specific interactions.

When peptides **1b** and *ent*-**1b** are mixed (4.0 mM of each), no new peaks form, and only peaks associated with the monomer and homochiral tetramer are observed (Fig. 8D). As with the 8.0 mM spectrum of peptide **1b**, there is slight broadening of the spectrum of the mixture of peptides **1b** and *ent*-**1b**, suggesting exchange on an intermediate timescale or additional non-specific interactions.

The DOSY spectrum of 1.0 mM peptide **1b** in D_2O shows a monomer with a diffusion coefficient of $19.5 \pm 0.7 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ (Fig. S7†). This value is similar to that which we have previously reported for peptide **1b** at 1.0 mM and 298 K.²⁰ At

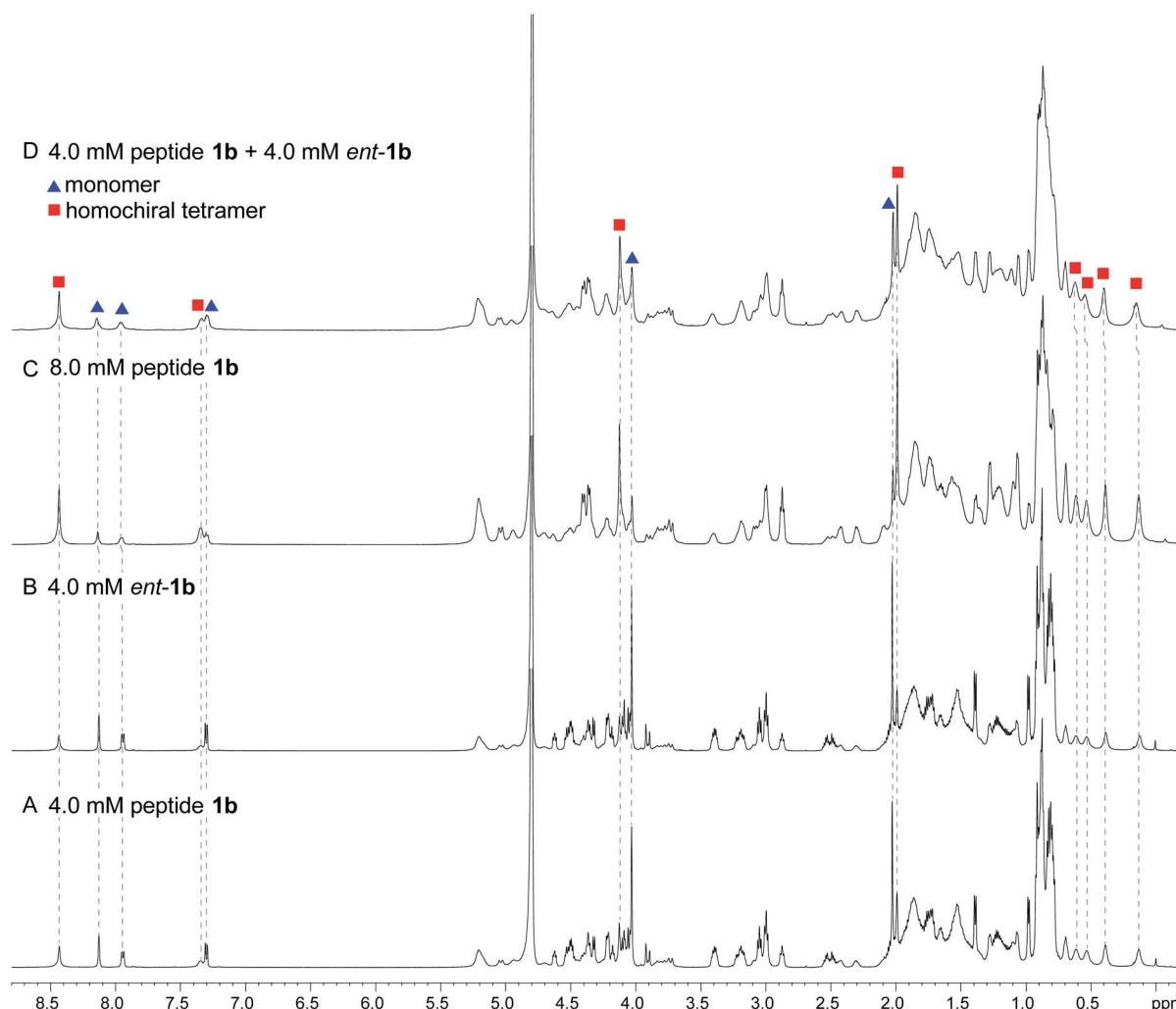


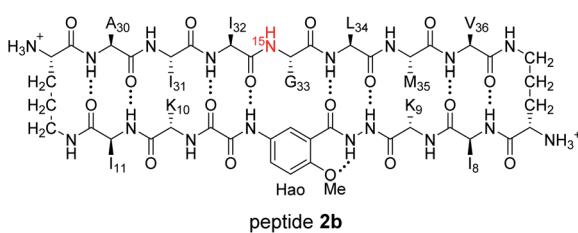
Fig. 8 ^1H NMR spectra of (A) 4.0 mM peptide **1b**, (B) 4.0 mM peptide *ent*-**1b**, (C) 8.0 mM peptide **1b**, and (D) 4.0 mM peptide **1b** and 4.0 mM peptide *ent*-**1b** in D_2O at 600 MHz and 298 K with 0.06 mM DSA.²⁵ (Spectra were referenced against an external DSA standard.) Dashed lines mark key resonances associated with the monomer and homochiral tetramer. These resonances are labeled as follows: blue triangle, monomer; red square, homochiral tetramer.



4.0 mM, an additional set of smaller resonances associated with the homochiral tetramer appears (Fig. S8†); the monomer shows a diffusion coefficient of $17.7 \pm 0.6 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$, and the tetramer shows a diffusion coefficient of $13.0 \pm 0.4 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$. The decrease in diffusion coefficient of the monomer, as well as the somewhat higher than expected diffusion coefficient of the tetramer—typically about $0.6 \times$ that of the monomer,²⁶ *ca.* $12 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ —suggest intermediate exchange between the monomer and the tetramer on the 75 ms time scale of the DOSY experiment.

When the concentration of peptide **1b** is doubled to 8.0 mM, the resonances associated with the homochiral tetramer predominate (Fig. S9†); the monomer shows a diffusion coefficient of $15.9 \pm 0.7 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$, and the tetramer shows a diffusion coefficient of $12.5 \pm 0.4 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$. The further decrease in diffusion coefficient of the monomer is consistent with intermediate exchange. The DOSY spectrum of the mixture of peptides **1b** and *ent*-**1b** (4.0 mM of each) shows diffusion coefficients of the monomer and tetramer of $14.4 \pm 1.1 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ and $10.7 \pm 0.8 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$, respectively (Fig. S10†). The low value of the monomer is consistent with intermediate exchange between the monomer and the tetramer. The value of the tetramer is somewhat lower than expected, suggesting additional transient non-specific interactions among the tetramers.^{26,27} The absence of any additional new peaks in the spectrum of the mixture, not present in the spectra of the enantiomerically pure peptides, provides good evidence that mixing peptides **1b** and *ent*-**1b** does not result in any detectable heterochiral assembly.

¹H, ¹⁵N HSQC studies corroborate the presence of only monomer and homochiral tetramer in the mixture of peptides **1b** and *ent*-**1b**. To identify and confirm the monomer and homochiral tetramer by ¹H, ¹⁵N HSQC, we prepared an isotopeologue of peptide **1b** containing an ¹⁵N label on Gly₃₃—peptide **2b**—and studied its mixture with peptide *ent*-**1b**. We studied increasing concentrations of enantiomerically pure peptide **2b** (1.0 mM, 4.0 mM, and 8.0 mM), and compared the crosspeaks to those found in the mixture of peptides **2b** and *ent*-**1b** (8.0 mM total).



The $^1\text{H}, ^{15}\text{N}$ HSQC spectrum of 1.0 mM peptide **2b** in 9 : 1 $\text{H}_2\text{O}-\text{D}_2\text{O}$ solution shows only a crosspeak associated with the monomer, at 8.35 ppm in the ^1H dimension and 112.2 ppm in the ^{15}N dimension (Fig. 9A). At 4.0 mM peptide **2b**, the monomer is still present and a crosspeak associated with the homochiral tetramer appears 115.8 ppm in the ^{15}N dimension and 9.31 ppm in the ^1H dimension (Fig. 9B).²⁰ When the concentration of peptide **2b** is doubled to 8.0 mM, the relative intensity of the tetramer crosspeak increases (Fig. 9C). When peptide **2b**

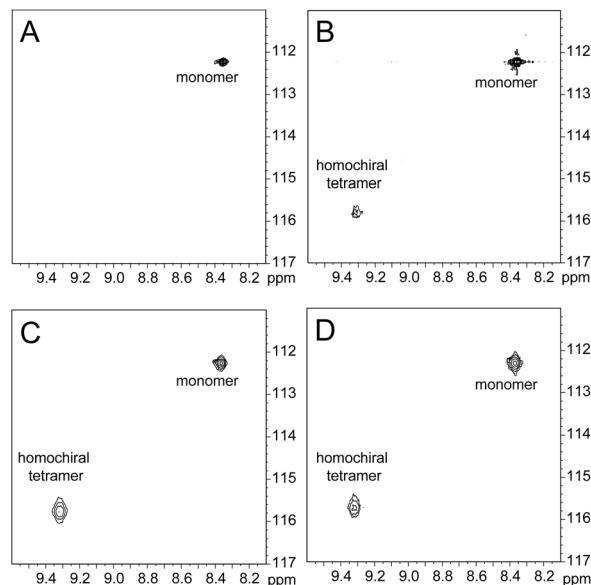


Fig. 9 $^1\text{H}, ^{15}\text{N}$ HSQC spectra of (A) 1.0 mM peptide **2b**, (B) 4.0 mM peptide **2b**, (C) 8.0 mM peptide **2b**, and (D) 4.0 mM peptide **2b** and 4.0 mM peptide **ent-1b** in 9 : 1 $\text{H}_2\text{O}-\text{D}_2\text{O}$ at 500 MHz and 298 K.

is mixed with peptide **ent-1b** (4.0 mM each), crosspeaks associated with the monomer and homochiral tetramer are still present and no new crosspeaks are observed (Fig. 9D). The lack of new crosspeaks further establishes that enantiomeric β -sheet peptides prefer to self-assemble in a homochiral fashion.

Conclusion

The solution-phase NMR studies of A β -derived peptides **1a** and **1b** and the corresponding enantiomers and isotopologues provide further evidence that enantiomeric β -sheet peptides prefer to self-assemble in a homochiral fashion. These studies recapitulate our laboratory's findings that small β -sheet peptides strongly prefer to form homochiral dimers in chloroform solution,¹⁸ as well as Gellman's findings in homo- and heterochiral β -hairpin systems.¹⁹ How then, do we reconcile these findings with the findings of other researchers where heterochiral assembly is preferred?

In the studies of Schneider, Nilsson, Raskatov, and Tycko described in the introduction, heterochiral assembly occurs in the solid or gel state.⁵⁻¹⁷ Heterochiral assembly in the solid state is driven heavily by the packing of molecules, which in addition to edge-to-edge hydrogen bonding, drives the formation of fibrils and crystal lattices. These packing interactions involve not just the side chains within individual β -sheets, but also the packing of β -sheets together. Heterochiral packing is generally preferred over homochiral packing in the crystal state, which leads to denser solids and a preference for racemic crystal formation—a phenomenon known as “Wallach’s rule”.³¹⁻³³ Thus, it appears that packing in the solid state may drive the formation of heterochiral mixtures of β -sheet peptides, and in some cases the formation of rippled β -sheets. In the solution phase, where crystal packing forces are absent, rippled β -sheet

formation is strongly disfavored. Thus, no evidence of heterochiral pairing to form rippled β -sheets is observed with peptides **1a** and **1b** and the corresponding enantiomers.

Data availability

The data supporting this article are available in the ESI.†

Author contributions

Xingyue Li synthesized the peptides, performed the experiments, analyzed the results, and prepared the manuscript with James S. Nowick. Stephanie E. Rios assisted with the peptide synthesis. James S. Nowick supervised the project and assisted in the experimental design and writing of the manuscript.

Conflicts of interest

The authors declare no conflicts of interest.

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Notes and references

- 1 L. Pauling and R. B. Corey, *Proc. Natl. Acad. Sci. U. S. A.*, 1951, **37**, 251–256.
- 2 L. Pauling and R. B. Corey, *Proc. Natl. Acad. Sci. U. S. A.*, 1951, **37**, 729–740.
- 3 L. Pauling and R. B. Corey, *Proc. Natl. Acad. Sci. U. S. A.*, 1953, **39**, 253–256.
- 4 B. Lotz, *ChemBioChem*, 2022, e202100658.
- 5 J. A. Raskatov, J. P. Schneider and B. L. Nilsson, *Acc. Chem. Res.*, 2021, **54**, 2488–2501.
- 6 A. R. Foley and J. A. Raskatov, *Curr. Opin. Chem. Biol.*, 2021, **64**, 1–9.
- 7 K. J. Nagy, M. C. Giano, A. Jin, D. J. Pochan and J. P. Schneider, *J. Am. Chem. Soc.*, 2011, **133**, 14975–14977.
- 8 K. Nagy-Smith, P. J. Beltramo, E. Moore, R. Tycko, E. M. Furst and J. P. Schneider, *ACS Cent. Sci.*, 2017, **3**, 586–597.
- 9 R. J. Swanekamp, J. T. M. Dimaio, C. J. Bowerman and B. L. Nilsson, *J. Am. Chem. Soc.*, 2012, **134**, 5556–5559.
- 10 R. J. Swanekamp, J. J. Welch and B. L. Nilsson, *Chem. Commun.*, 2014, **50**, 10133–10136.
- 11 J. M. Urban, J. Ho, G. Piester, R. Fu and B. L. Nilsson, *Molecules*, 2019, **24**, 1983.
- 12 S. Dutta, A. R. Foley, C. J. A. Warner, X. Zhang, M. Rolandi, B. Abrams and J. A. Raskatov, *Angew. Chem., Int. Ed.*, 2017, **56**, 11506–11510.
- 13 J. A. Raskatov, *Chem.-Eur. J.*, 2017, **23**, 16920–16923.
- 14 J. A. Raskatov, A. R. Foley, J. M. Louis, W. M. Yau and R. Tycko, *J. Am. Chem. Soc.*, 2021, **143**, 13299–13313.
- 15 A. J. Kuhn, B. Ehlke, T. C. Johnstone, S. R. J. Oliver and J. A. Raskatov, *Chem. Sci.*, 2022, **13**, 671–680.
- 16 J. A. Raskatov, *ChemBioChem*, 2020, **21**, 2945–2949.
- 17 J. A. Raskatov, *Biopolymers*, 2021, **112**, e23391.
- 18 D. M. Chung and J. S. Nowick, *J. Am. Chem. Soc.*, 2004, **126**, 3062–3063.
- 19 X. Liu and S. H. Gellman, *ChemBioChem*, 2021, **22**, 2772–2776.
- 20 N. L. Truex, Y. Wang and J. S. Nowick, *J. Am. Chem. Soc.*, 2016, **138**, 13882–13890.
- 21 N. L. Truex and J. S. Nowick, *J. Am. Chem. Soc.*, 2016, **138**, 13891–13900.
- 22 J. S. Nowick and J. O. Brower, *J. Am. Chem. Soc.*, 2003, **125**, 876–877.
- 23 J. S. Nowick, D. M. Chung, K. Maitra, S. Maitra, K. D. Stigers and Y. Sun, *J. Am. Chem. Soc.*, 2000, **122**, 7654–7661.
- 24 The EXSY experiments were performed with a mixing time of 500 ms. Exchange occurs too slowly to be detected at 298 K but is observed at 328 K, suggesting that exchange occurs on the time scale of seconds at 328 K and tens of seconds at 298 K.
- 25 J. S. Nowick, O. Khakshoor, M. Hashemzadeh and J. O. Brower, *Org. Lett.*, 2003, **5**, 3511–3513.
- 26 O. Khakshoor, B. Demeler and J. S. Nowick, *J. Am. Chem. Soc.*, 2007, **129**, 5558–5569.
- 27 Alternatively, the formation of hexamers in rapid equilibrium with the tetramers cannot be precluded. We have previously observed that β -sheet peptides that form a tetramer comprising two β -sheet dimers in solution can form a hexamer comprising three β -sheet dimers in the crystal state, O. Khakshoor, A. J. Lin, T. P. Korman, M. R. Sawaya, S. C. Tsai, D. Eisenberg and J. S. Nowick, *J. Am. Chem. Soc.*, 2010, **132**, 11622–11628.
- 28 J. S. Nowick, *Org. Biomol. Chem.*, 2006, **4**, 3869–3885.
- 29 K. Wüthrich *NMR of Proteins and Nucleic Acids*, Wiley, New York, 1986, pp. 125–129.
- 30 d_8 -Phenylalanine was purchased from Cambridge Isotope Laboratories and was reported to be 98% isotopic purity. The ^1H NMR spectrum of the d_8 -phenylalanine shows a disproportionate amount of 1H isotopic impurity at the α -position. See Fig. S11† for spectral data.
- 31 C. P. Brock, W. B. Schweizer and J. D. Dunitz, *J. Am. Chem. Soc.*, 1991, **113**, 9811–9820.
- 32 T. Friščić, L. Fábián, J. C. Burley, D. G. Reid, M. J. Duer and W. Jones, *Chem. Commun.*, 2008, 1644–1646.
- 33 P. S. Navare and J. C. MacDonald, *Cryst. Growth Des.*, 2011, **11**, 2422–2428.

