

REVIEW

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Fluorine in metal-catalyzed asymmetric transformations: the lightest halogen causing a massive effect

Samuel Lauzon and Thierry Ollevier *

This review aims at providing an overview of the most significant applications of fluorine-containing ligands reported in the literature starting from 2001 until mid-2021. The ligands are classified according to the nature of the donor atoms involved. This review highlights both metal–ligand interactions and the structure–reactivity relationships resulting from the presence of the fluorine atom or fluorine-containing substituents on chiral catalysts.

1. Introduction

Fluoroorganic molecules have received widespread interest in recent research in view of the fact that their synthesis is virtually missing from any biological processes. Being the 13th most abundant element in the lithosphere, fluorine is mostly present in water-insoluble minerals, *i.e.*, fluor spar (CaF_2), cryolite (Na_3AlF_6), and fluoroapatite ($\text{Ca}_5(\text{PO}_4)_3\text{F}$), which limits its uptake into living organisms.¹ Also, the nucleophilicity of fluoride is diminished by its high hydration energy, and therefore, this anion is inadequate for any nucleophilic substitution

reactions in aqueous media. As a result, only a dozen of naturally occurring fluorometabolites, *i.e.*, fluoroacetic acid, ω -fluorinated fatty acids, (2*R*,3*R*)-2-fluorocitric acid, (2*S*,3*S*,4*R*,5*R*)-nucleocidin, and (2*S*,3*S*)-4-fluorothreonine, have been identified so far.² Since then, such scarcity has been compensated for by man-made fluorine-containing pharmaceuticals, of which many have been welcomed as blockbuster drugs on the market.³ Molecular conformation, membrane permeability, and metabolic stability are all properties affected by fluorine substitution, but the impact on the pharmacodynamics and pharmacokinetics of a lead remains quite unpredictable.⁴ The

Département de Chimie, Université Laval, 1045 Avenue de la Médecine, Québec, QC G1V 0A6, Canada. E-mail: thierry.ollevier@chm.ulaval.ca



Samuel Lauzon was born in Québec, Canada. He joined the research group of Professor Thierry Ollevier as an undergraduate intern student in 2015. He obtained his MSc in 2018 working on asymmetric iron catalysis. He received his PhD in organic chemistry at Université Laval (Québec, Canada) under the supervision of Professor Thierry Ollevier in 2022. His work involved the development of designer 2,2'-bipyridinediol ligands for applications in metal-catalyzed asymmetric transformations. He is currently interested in ligand design, crystallization of chiral catalysts, transition state models, and environmentally benign catalysis.

Thierry Ollevier was born in Brussels and obtained his Licence (1991) and PhD (1997) at the University of Namur (Belgium) and was a post doctorate fellow at the Université catholique de Louvain (Belgium), under István E. Markó (1997), a NATO and BAEF post-doctorate fellow at Stanford University under Barry M. Trost (1998–2000), and then a post doctorate fellow at the Université de Montréal under André B. Charette (2000–2001). After an Assistant Professor appointment (2001) at Université Laval (Québec, Canada), he became Associate (2006) and is currently Full Professor. Current research in his group aims at designing novel catalysts, developing catalytic reactions and applying these methods to chemical synthesis. He is active in the areas of asymmetric catalysis, iron catalysis, diazo chemistry, fluorine chemistry, and flow chemistry. He has served as a member of the Advisory Board of SynOpen since 2019, as an Associate Editor of RSC Advances since 2015 and was admitted as a Fellow of the Royal Society of Chemistry (2016).



Table 1 Electronegativities of selected atoms and functional groups

Atom or group	χ_P	χ_e	χ_H	χ_J
H	2.20	2.28	—	—
F	3.98	3.95	—	—
Cl	3.16	3.03	—	—
Br	2.96	2.80	—	—
I	2.66	2.47	—	—
NO ₂	—	3.40	—	—
CN	—	3.30	—	—
CF ₃	—	3.35	3.46	3.29
CHF ₂	—	—	3.00	2.94
CH ₂ F	—	—	2.61	2.61
CH ₃	—	2.30	2.27	2.30

specific behaviour of fluorinated compounds arises from the short, strong, and highly polarized C–F bond that electrostatically pairs with the neighbouring atoms, bonds, and lone pairs.⁵ Instead of being targeted only for biological activity purposes, various chiral organofluorine compounds were employed as catalysts in asymmetric transformations.⁶ Among the known strategies for designing enhanced stereodiscriminating catalysts,⁷ the conversion of known catalytic systems into F-containing ones unlocked new interesting features ensuing from the fluorine effect. Chiral fluorinated catalysts benefit from (i) electronically and (ii) sterically customized properties and (iii) being used in fluorous biphasic systems (FBS). Since an increased reactivity in acid catalysts is often a synonym of electronic deficiency, Pauling (χ_P), empirical (χ_e), Huheey (χ_H), and Jaffé (χ_J) electronegativities values are provided for a selection of atoms and groups (Table 1).⁸ Fluorine is the most electronegative element on the Pauling scale ($\chi_P = 3.98$) and is most frequently used as an electronically impoverishing substituent in ligands. The CF₃ group is also electronically deficient, much more than other C-based substituents, and its χ value is similar to the ones of the CN and NO₂ groups, known as strongly electron withdrawing groups (EWG). A set of steric parameters,

e.g., A values ($-\Delta G$),⁹ Taft ($-E_s$),¹⁰ Charton (ν),¹¹ and biphenyl rotational interference values (I^{X-H}),¹² and Boltzmann-weighted Sterimol parameters (wB_1 , wL , and wB_5),¹³ eases the comparison of bulkiness between selected atoms and functional groups (Table 2). Fluorine is a small element (van der Waals radius = 1.47 Å vs. 1.20 Å for H vs. 1.52 Å for O)¹⁴ and generally causes minimal steric perturbation upon H to F exchange, whereas the substitution of C–H by C–F in a methyl group successively leads to an increase in size (Me < CH₂F < CHF₂ < CF₃). Thus, fluorine is a powerful tool to increase the steric hindrance – even more in polyfluoroalkyl groups – rendering the CF₃ group as a bulky substituent in the general order Me < ⁱPr ~ CF₃ << Ph ~ ^tBu. The CF₃ group is a key component for fine-tuning simultaneously the steric and electronic properties, which is undeniably of great interest for ligand design. Following the “like dissolves like” principle, a chiral ligand bearing sufficiently long perfluoroalkyl chains, *i.e.*, ponytails (R_F), acquires a strong affinity for the fluorous phase, also called “fluorophilicity”.¹⁵ This engineered technology involves the temperature-dependent miscibility of fluorous-organic solvents leading to partition of two phases at lower temperature. The fluorous ligand (or catalyst) can be selectively separated from the reactant/product mixture and recovered using various experimental methods.

The synergistic participation of the steric, electronic, and physical properties brought up by the chiral fluorinated ligand may modulate the stereoselective event of a reaction positively vs. “fluorine-free transition states (TS)”. This review highlights both metal–ligand interactions and structure–reactivity relationship related to the presence of the fluorine atom or fluorine-containing substituents on chiral catalysts. The article focusses on the molecular architecture of the ligands rather than on the type of reaction they were employed in. The selection of chiral fluoroorganic ligands are classified according to *O*-, *N,O*-, *N*-, *P*-, *N*-, *P*- and *C*-binding modes with metals. The advantages of using catalytic systems involving fluorine for both the reactivity and the stereochemical outcomes are highlighted, compared,

Table 2 Steric parameters of selected atoms and functional groups

Atom or group	$-\Delta G$ (kcal mol ⁻¹)	$-E_s$	ν	I^{X-H} (kJ mol ⁻¹)	w Sterimol		
					wB_1 (Å)	wL (Å)	wB_5 (Å)
H	—	0.00	0.00	~4.0	—	—	—
F	0.15	0.46	0.27	19.2	1.52	3.12	2.03
Cl	0.43	0.97	0.55	38.1	1.80	3.99	2.03
Br	0.38	1.16	0.65	42.5	1.95	4.29	2.03
CN	0.17	0.51	—	25.6	—	—	—
NO ₂	1.10	1.01/2.52 ^a	—	32.4	—	—	—
Me	1.70	1.24	0.52	40.4	1.88	3.43	2.02
CH ₂ F	1.59	1.48	0.62	—	—	—	—
CHF ₂	1.85	1.91	0.68	—	—	—	—
CF ₃	2.10	2.40	0.91	50.6	1.87	3.75	2.60
ⁱ Pr	2.15	1.71	0.76	52.6	1.91	4.56	3.17
SiMe ₃	2.50	3.36	1.40	47.2	—	—	—
Ph	3.00	1.01/3.82 ^a	1.66	33.1	1.70	6.80	3.15
^t Bu	~4.20	2.78	1.24	76.6	2.76	4.54	3.17

^a Thickness/width values.



and discussed. Furthermore, the effects on the stereochemical outcome arising from using fluorinated ligands *vs.* non-fluorinated ones are often provided when both results were available in the literature. The set of examples for promoting the massive effect of fluorine in metal-catalyzed asymmetric catalysis is not exhaustive; hence fluorinated reactants, solvents, additives, and Brønsted acids-based catalysts are not reviewed herein.

2. Chiral fluoroorganometallic complexes

2.1 O-Based binding modes

2.1.1 Phosphoric acids. Chiral phosphoric acid (CPA) ligands (*R_a*)-1–3 and (*S_a*)-4 were obtained by the phosphorylation of fluorinated 1,1'-bi-2-naphthol (BINOL)-type ligands and were described as promising candidates for enantioselective applications (Fig. 1). As an example, the fluorination reaction of β -ketoesters 5–8 using the 3 : 1 (*R_a*)-1/*ScCl*₃ complex afforded the corresponding α -fluorinated products 9–12 with 78–84% ee (Scheme 1).¹⁶ The importance of the eight fluorine atoms was demonstrated based on the low 11% ee obtained for (+)-9 using the *Sc*^{III} salt of the non-fluorinated CPA. On the other hand, (*R_a*)-2¹⁷ and (*R_a*)-3¹⁸ were respectively employed with *Cu*^I salts in the synthesis of chiral polyfluoro *N*-containing compounds, albeit with modest ees.

In another application, cycloadducts (*S,R,R*)-15 and *endo*-(*R,S,S,S*)-16, respectively obtained from the hetero-Diels–Alder (HDA) and the Diels–Alder (DA) reactions between dienophile 13 and cyclopentadiene 14, were synthesized with excellent yields and stereoselectivities using the binary (*S_a*)-4/*In*^{III} catalytic system (Scheme 2).¹⁹ In most cases, a higher

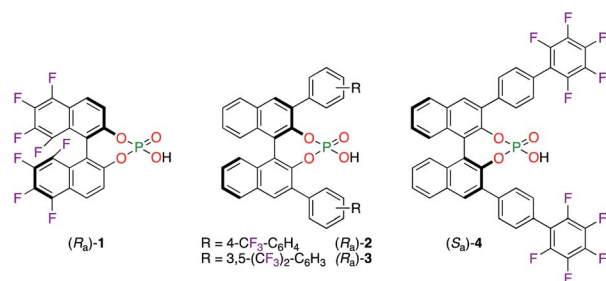
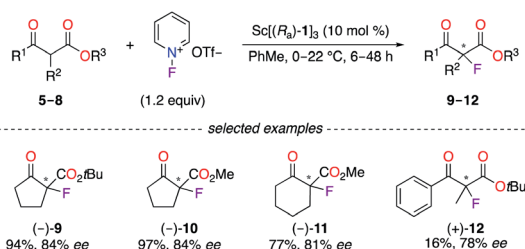
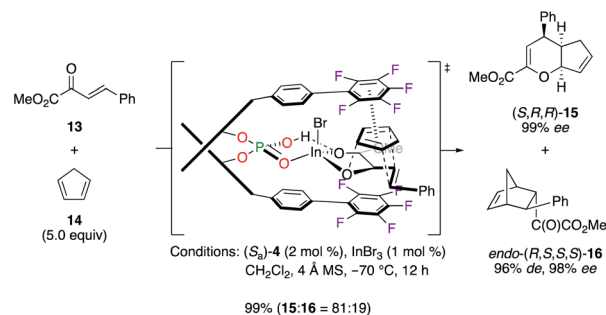


Fig. 1 Chiral fluorinated CPA ligands 1–4 in metal-catalyzed reactions.



Scheme 1 [(*R_a*)-1]₃/*Sc*^{III}-catalyzed fluorination reaction.

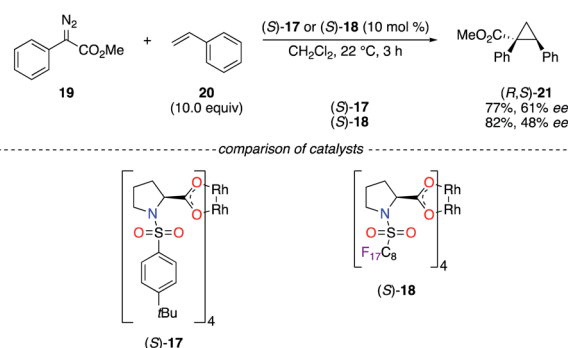


Scheme 2 (*S_a*)-4/*In*^{III}-catalyzed HDA and DA reactions.

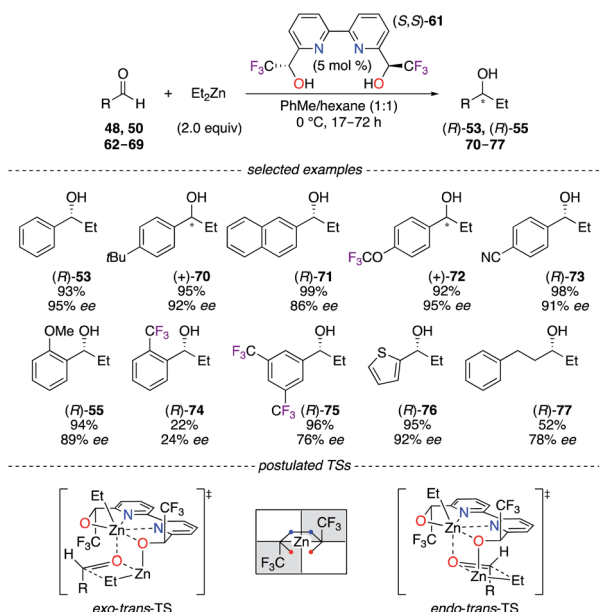
chemoselectivity was noted for the HDA cycloadduct *vs.* the DA one. The stereoselective induction was explained by the important π , π interaction, arising from an electrostatic pairing between one electronically deficient pentafluorophenyl ring and diene 14, together with highly sensitive *ortho* positions occupied by F atoms.

2.1.2 Carboxylates and alkoxides. Known as valuable catalysts in the chemistry of diazo compounds, chiral *Rh*^{II} tetrakis(carboxylates) are rather expensive to be easily used in industrial applications. The recyclable chiral fluoros *Rh*^{II} complex (*S*)-18 was then targeted in the cyclopropanation reaction of diazoester 19 and styrene 20 (Scheme 3).²⁰ Whereas it was less selective in terms of ee than (*S*)-17, the perfluoroalkyl chain allowed an efficient recovery of the catalyst using both liquid and solid fluoros phase extraction strategies. Slight improvements of the chiral induction were obtained in the insertion reaction of 19 into the C–H bond of cyclohexane using (*S*)-18 under either homogeneous or heterogeneous conditions.

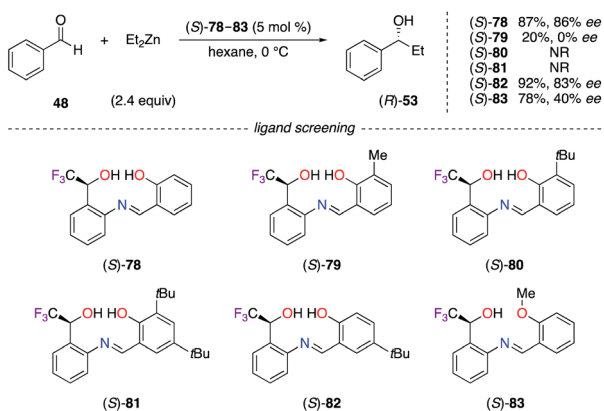
Various enantioselective organic transformations were performed using more sophisticated *CF*₃-containing BINOL-based chiral catalysts (Fig. 2). Indeed, the synergetic activation using self-assembled bifunctional chiral catalysis was shown as an efficient strategy in the hydrophosphonylation reaction of aldehydes.²¹ The multiple reactive sites of catalyst (*R_a*)-22, generated *in situ* in the presence of *Ti*(*O*^{*i*}*Pr*)₄, *CF*₃-aryl-substituted BINOL, and (–)-cinchonidine, allowed cooperative interactions between steric and electronic properties, as postulated in the transition state model of the reaction. Also,



Scheme 3 Cyclopropanation reaction using chiral dirhodium catalysts.



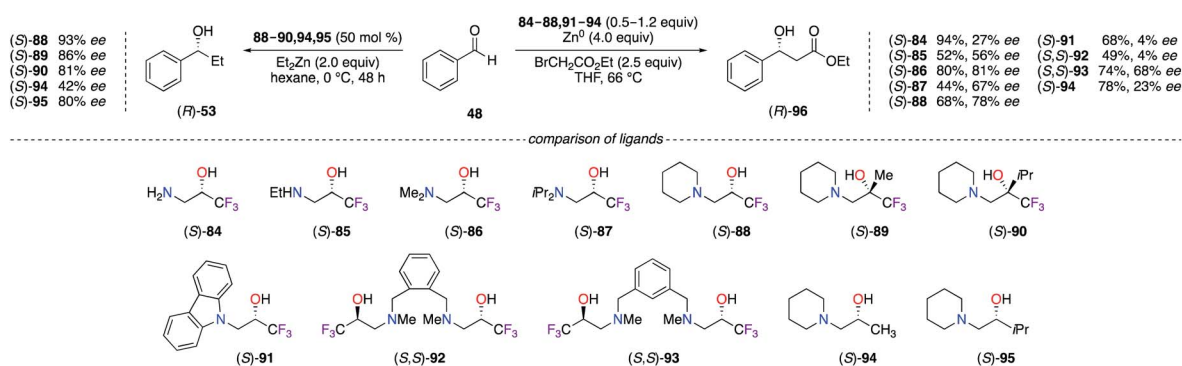
Scheme 8 2,2'-Bipyridine- α,α' -CF₃-diol/Zn^{II}-mediated ethylation reaction of aldehydes.



Scheme 9 Zn^{II}-Mediated ethylation reaction of 48 using Schiff's bases bearing an α -CF₃-alcohol.

diethylzinc to benzaldehyde 48 (Scheme 9).³⁵ Ligands (S)-79–81, substituted at the *ortho* position with a Me or a ^tBu group, failed to increase the chiral induction. Only *para*-substituted (S)-82 afforded (R)-53 with similar ee to (S)-78. A non-linear relationship with a minimum enantiomeric amplification, together with high resolution mass spectrometry (HRMS) analysis, suggested that the dimeric [(S)-78]₂/Zn^{II} complex is the active catalyst. The C-stereocentre of Schiff's base ligands (S)-78–83 was constructed by the enantioselective reduction of the *o*-nitrophenyl- α -CF₃-ketone using the (R)-CBS oxazaborolidine reagent. Surprisingly, the non-fluorinated analogues of these Schiff's base ligands have not been described in the literature so far.

2.2.2 β -Amino alcohols. Fluorinated β -amino alcohol ligands 84–95 have been used in various asymmetric transformations involving the alkylation of aldehydes (Scheme 10). In the Et₂Zn alkylation reaction of benzaldehyde,³⁶ only the organozinc catalyst prepared from (S)-88 led to a maximum enantioselectivity, notwithstanding the presence of the Me or ^tPr substituents on the ligand at the α -position of the hydroxyl group giving (S)-89 and (S)-90 with bulkier quaternary carbon stereocentres (Scheme 10, left). Correlation studies on the catalyst loading (2–50 mol%) have demonstrated a strong dependence between the increased amount of (S)-88–90 and the ee observed on (R)-53, whereas no dependency was determined for the non-fluorinated analogues (S)-94 and (S)-95; the superior degree of aggregation of CF₃-containing β -amino alcohol ligands, particularly for the (S)-88/Zn^{II} catalyst, strongly participated in the mechanism to reach higher chiral induction of alcohol (R)-53.^{36a} A wider library of β -amino α -CF₃-alcohol ligands were screened in the Reformatsky reaction of PhCHO (Scheme 10, right).³⁷ The ees obtained with ligands (S)-84 and (S)-85, containing a primary or a secondary amine, were much lower than the ones provided using tertiary amine-based ligands. Furthermore, (S)-86 possessing a *N,N*-dimethyl-amino group led to the highest 81% ee of (R)-96 in comparison with the other ligands having diisopropylamine ((S)-87), piperidine ((S)-88), and carbazole ((S)-91) motifs at the β -position. As shown on (S,S)-92 and (S,S)-93, the benzene ring bearing β -amino α -CF₃-alcohols tethered at the 1,2 and 1,3 positions were considerably less efficient than (S)-86. The aggregation effect of



Scheme 10 Et₂Zn alkylation (left) and Reformatsky (right) reactions of benzaldehyde using amino-alcohol derivatives.



described as highly enantioselective using phosphino(imidazoline) (StackPHIM) (R_a,R,R)-**274** (Scheme 32).¹⁰⁰ Since the imidazole analogue (S_a)-**273** (StackPhos ligands)¹⁰¹ and (R,R)-**275** (having no axial chirality) both led to 2-aminoalkyl furan **280** (R or S) with lower enantiomeric enrichments, the complementary between the stereocentres and the chiral axis to reach higher ee was demonstrated. Further fine-tuning of the **274**/Cu^I catalyst was highlighted by the combination of the (R,R)-DPEN scaffold with the R_a or the S_a atropisomer, resulting in an increased chiral induction of **280** from 82% ee (R) to 94% ee (S). Noteworthy, atropisomerism of configurationally stable P,N -ligands **273** and **274** arises from π,π -stacking interactions between the naphthyl and C_6F_5 moieties. Being determined experimentally at 50 °C, rotational energy barriers (ΔG^\ddagger) of 26.8 kcal mol⁻¹ (R_a into S_a) and 27.5 kcal mol⁻¹ (S_a into R_a) have proven that both atropisomers of **274** could be synthesized, separated and successfully employed as chiral ligands in metal catalysis. However, the absence of fluorine atoms considerably lowered the ΔG^\ddagger values of R_a - S_a (or S_a - R_a) interconversion, and epimerization of **281** occurred easily even at room temperature.

2.4.4 Ferrocenes. The popular 3,5-bis(trifluoromethyl) phenyl scaffold was incorporated into chiral ferrocenyl-derived P,N -containing ligands (Fig. 13). Indeed, imine-, amine-, and oxazoline-based phosphine ligands **282–284** were successfully used in the Pd^{II}-catalyzed allylic alkylation,¹⁰² the Rh^{II}-catalyzed hydrogenation,¹⁰³ and the Cu^I-catalyzed 1,3-dipolar cycloaddition reactions.¹⁰⁴ In the last case, the divergent *exo/endo* selectivities observed from the experimental results were rationalized through computational studies. Interestingly, the postulated TS models suggested that two different chelation modes of the substrate were arising from the electron-deficient Ar_F substituents of (S_p,S)-**284** vs. the electron-rich phenyl rings of its non-fluorinated analogue. Moreover, closely related ferrocenyl-based bis(perfluoroalkyl)phosphino(oxazoline) ligands were designed in a series of bulky ^tBu, Ph and Bn substituents at the C -stereocentre.^{89,105}

In brief, the widespread use of 4- CF_3 and 3,5-(CF_3)₂ motifs on the aromatic scaffold of the ligands was highlighted in this section. Importantly, the C_6F_5 group has been demonstrated as highly efficient when used in axially chiral phosphines, whereas its electrostatic pairing with the π system of the naphthyl moiety has induced a rotational energy barrier allowing atropisomerism. Also, the use of *o,o'*- CF_3 within the TF-BIPHAM architecture was proved valuable in this strategy.

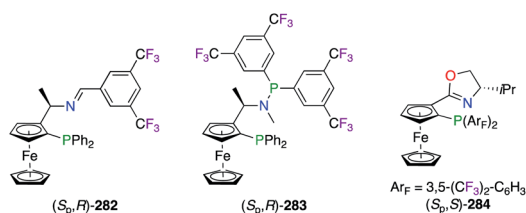


Fig. 13 Chiral ferrocenyl-derived ligands containing the 3,5-(CF_3)₂- C_6H_3 group.

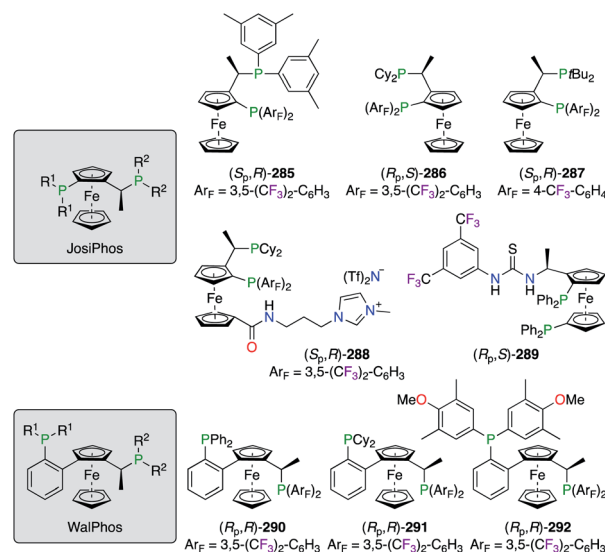
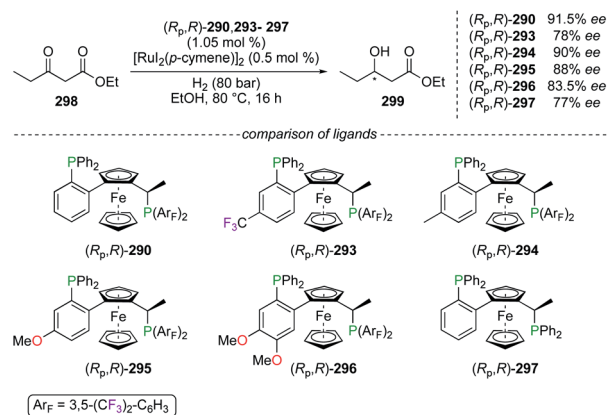


Fig. 14 Fluorinated JosiPhos- and WalPhos-type of ligands.

2.5 P-Based binding modes

2.5.1 Planar chiral diphosphines. Electron-poor diphosphine ligands were described for JosiPhos- and WalPhos-type ferrocenes **285–289** and (R_p,R)-**290–292**, respectively (Fig. 14). Chiral ligands (S_p,R)-**285**,¹⁰⁶ (R_p,S)-**286**,^{106a,107} and (S_p,R)-**287**^{106b} were tested in the Ir^I- or Rh^{III}-catalyzed hydrogenation and the Pd^{II}/Cu^I-co-catalyzed Heck/Sonogashira asymmetric reactions, but only high chiral inductions were obtained by using the (R_p,S)-**286**/Pd^{II} catalyst. An application of fluorinated JosiPhos ligands in ionic liquids was demonstrated using (S_p,R)-**288**, i.e., the imidazolium-based analogue of (S_p,R)-**286**, in combination with [Rh(norbornadiene)₂] BF_4 .^{107b} The asymmetric hydrogenation reaction of methyl acetamidoacrylate, run under biphasic *tert*-butyl methyl ether/[bmim] BF_4 conditions, afforded the corresponding product with 99% ee using either (S_p,R)-**286** or (S_p,R)-**288**. More importantly, the ionic tag on (S_p,R)-**288** allows better recyclability of the fluorinated catalyst in the chosen co-solvent system. Chiral thiourea-derived diphosphine ligand (R_p,S)-**289** was used in the Rh^I-catalyzed hydrogenation



Scheme 33 Hydrogenation reaction using electronically modified WalPhos ligands.

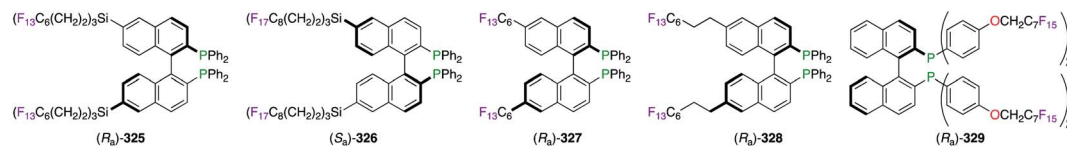


Fig. 17 Fluorinated BINAPs employed in FBS.

selected classes of ligands, the narrowest dihedral angle was attributed to DifluorPhos-type ligands ($\theta \sim 67^\circ$), whereas BINAP derivatives ($\theta \sim 86^\circ$) were located at the other end of the steric scale.^{119a,b} Overall, all electronically impoverished ligands, with specific steric profiles, showed excellent chiral inductions when complexed with the appropriate Lewis acid in various asymmetric reactions.

Strategies towards the recycling of the chiral ligand have encouraged the derivatization of BINAPs to incorporate perfluoroalkyl chains within their skeleton (Fig. 17). Fluorine substitution upon the naphthyl-backbone was then chosen for the synthesis of (*R*_a)-325. The asymmetric Heck reaction of 2,3-dihydrofuran was successfully performed in the benzene/FC-72 system using (*R*_a)-325/Pd^{II}.^{28b,124} The Ru^{II}-catalyzed hydrogenation reaction of dimethyl itaconate was described using heavier fluorinated BINAP (*S*_a)-326, which was immobilized on fluorinated silica gel.¹²⁵ Excellent retention of the fluorinated ligand within the

silica pores, *via* noncovalent interactions with the C₈F₁₇ chains, permitted its recycling without the use of conventional biphasic extraction methods. Similar to (*R*_a)-325, F-containing BINAPs (*R*_a)-327 and (*R*_a)-328 were developed for their great ability as “light” fluorinated ligands to be extracted from the other organic compounds *via* simple FRP column chromatography.¹²⁶ Introducing the perfluoroalkyl chain at the P atom, as highlighted by (*R*_a)-329, afforded poor ees in three metal-catalyzed reactions potentially due to the proximity of the fluorinated tails to the activating site.¹²⁷ The relatively low fluorine content of (*R*_a)-329 (51.5%) failed to give satisfactory chiral inductions in FBS. However, (*R*_a)-329 was separated from the reaction mixture quickly using liquid–liquid extraction by perfluorocarbons.

Research towards the development of greener synthetic methods focused on the replacement of commonly used organic solvents by less hazardous and more environmentally benign alternatives, *e.g.*, supercritical carbon dioxide (scCO₂). The Rh^I-catalyzed hydroformylation reaction of mono-substituted alkenes **20**, **331–333** was performed in supercritical fluids using phosphine–phosphite BINAPHOS (*R*_a,*S*_a)-330 (Scheme 35).¹²⁸ The aldehydes **334–337** were obtained with good regioselectivities and enantioselectivities using the (*R*_a,*S*_a)-330/Rh^I catalytic system. Considered as CO₂-philic, the perfluoroalkyl chains permitted sufficient solubility of the ligand to be used under homogeneous reaction conditions. Moreover, C₄F₉, C₆F₁₃, and C₈F₁₇ ponytails were incorporated into the 1,1'-binaphthyl core to generate [R_F(CH₂)₃]-BINAPHOS analogues.¹²⁹

2.5.3 Phosphoramidites. A wide library of monodentate phosphoramidite ligands was designed for numerous metal-catalyzed asymmetric reactions (Fig. 18).¹³⁰ Following the long-arm approach, substituted phenyl rings were incorporated at the 3,3'-positions of the binaphthol skeleton in order to enhance chiral inductions. Indeed, the space surrounding the ligated metal centre was considerably restricted by bulky 3,5-(CF₃)₂ and 4-NO₂ aromatic rings, as observed in (*S*_a)-342/(*R*_a)-343 and (*S*_a)-344/(*R*_a)-345, respectively. An additional fine-tuning of the chiral ligands was possible at the amine moiety,

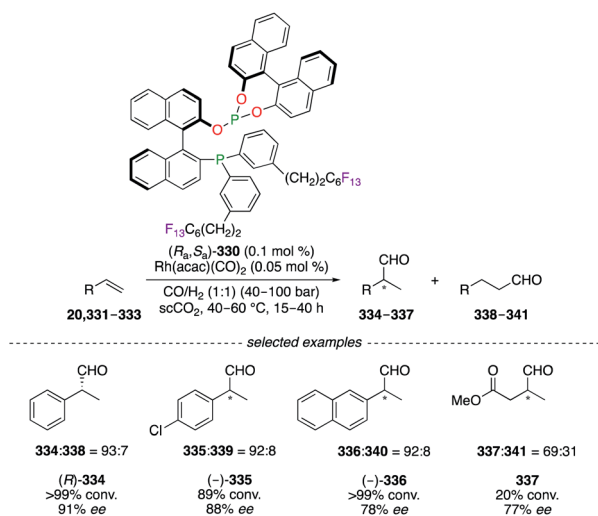
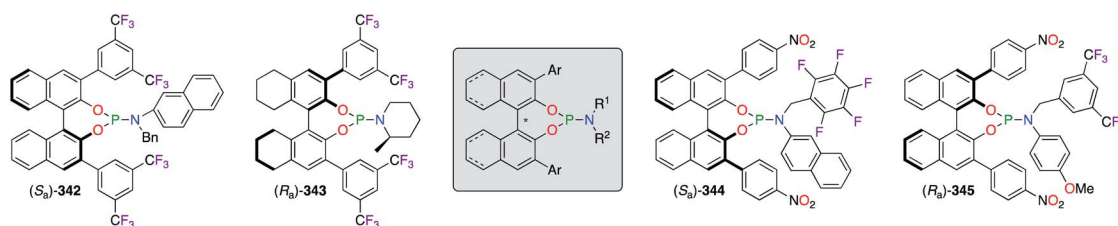
Scheme 35 (*R*_a,*S*_a)-330/Rh^I-Catalyzed hydroformylation reaction of alkenes.

Fig. 18 Selected examples of chiral phosphoramidite ligands.

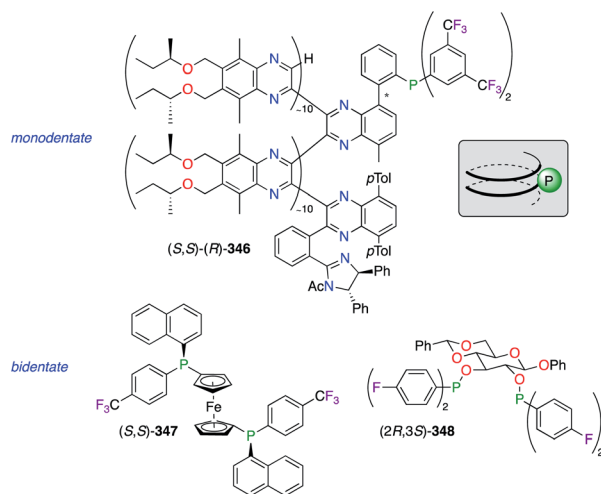


Fig. 19 Miscellaneous mono- and bidentate *P*-ligands containing fluorine atoms.

where electron-deficient benzyl substituents ((*S_a*)-344/(*R_a*)-345) or hindered piperidine ((*R_a*)-343) were generally more beneficial as observed from the obtained ees. A set of (*R,R*)-TADDOL-derived phosphoramidite ligands were screened in the Pd^{II}/Cu^I-catalyzed alkylation reaction, but only the aryl substituents bearing bulky TMS groups and one electron withdrawing F atom afforded the optimal chiral induction.¹³¹

(*R*)-BINOL-based phosphite ligands, generated from 3,5-bis(trifluoromethyl)phenol and 2,2,2-trifluoroethanol, were considered as promising ligands in the synthesis of Rh^I complexes and their use in enantioselective catalysis in ionic liquids.¹³²

2.5.4 Other *P* ligands. Fluorinated monophosphine, stereogenic phosphine, and sugar-derived phosphinite ligands have been complexed with various noble metals (Fig. 19). The (10–1–10) copolymer poly(quinoxaline-2,3-diyl)phosphine (PQXphos) (*S,S*)-(R)-346, having *P* helicity, was employed as a screened ligand in the asymmetric Pd⁰-catalyzed Suzuki–Miyaura coupling reaction, giving only a moderate chiral induction.¹³³ Interestingly, the sense of the chirality (*R_a* or *S_a*) at the active site of 346 would be induced, according to the structure models,^{133b} by the helicity adopted by the polymer either in the right-handed (*P*) or the left-handed (*M*) helix geometry, respectively. Another monodentate (1*R*,3*R*,4*S*)-menthyl-based phosphine ligand, bearing two C₆F₁₃ and C₈F₁₇ ponytails, was reported to give fluorinated chiral catalysts when mixed with Ir^I and Rh^I salts.¹³⁴ Chiral at *P* atoms, diphosphine ligand (*S,S*)-347, belonging to the class of 1,1'-bis(diphenylphosphino)ferrocenes (dppf), was designed to modulate the stereoselective event in the Pd^{II}-catalyzed nucleophilic substitution of allylic acetates.¹³⁵ When *rac*-114 and 115 were used as substrates, the combination of the bulky 1-naphthyl substituent with the electronically deficient 4-F-C₆H₄ unit afforded a slightly lower enantioselectivity (61% ee of (*S*)-116) than the unfluorinated one (68% ee of (*R*)-116) and its electron donating analogue (4-OMe-C₆H₄; 69% ee of (*S*)-116). A Pt^{II}-catalyzed alkylation of linked secondary phosphines (HRP–PRH) in the presence of various

benzyl bromides, comprising F- and CF₃-containing ones, led to *P*-stereogenic diphosphines with low stereoselectivities.¹³⁶ The hydrogenation of a variety of dehydroamino acids was reported using chiral phosphinite/Rh^I catalysts made from various carbohydrate scaffolds.¹³⁷ Being highly dependent on the *P*-aryl substituent, the level of chiral induction was demonstrated to be considerably higher when using electron-rich bis(3,5-dimethylphenyl) groups vs. electron-poor ones, such as C2,C3-bis-(di-4-fluorophenyl)phosphinite ligand (2*R*,3*S*)-348, derived from phenyl β-D-glucopyranoside. Furthermore, a (*R,R*)-DIOP-like 4-(trifluoromethyl)phenylboronate diphosphine ligand was synthesized and mixed with Rh^I, Pd^{II}, and Pt^{II} salts to generate heterobimetallic complexes.¹³⁸

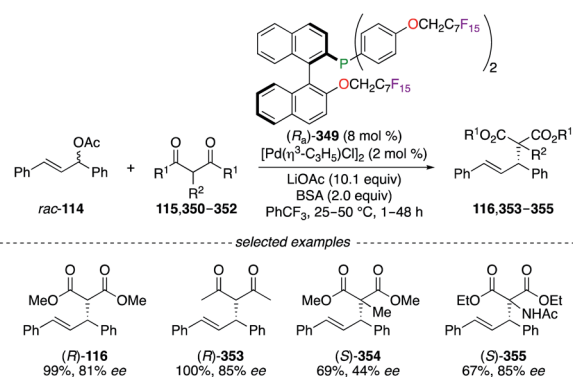
In general, the 3,5-(CF₃)₂-C₆H₃ substituent has been widely used in *P*-based chiral ligands. Various fluorinated ponytails were incorporated into BINAP ligands to give “heavy” and “light” fluorinated analogues to be used in distinct synthetic applications. Major advancements were demonstrated using CO₂-philic BINAPHOS ligands bearing perfluoroalkyl chains, in the enantioselective hydroformylation reaction performed in supercritical carbon dioxide. Noteworthy, this section has detailed numerous mono- and diphosphines incorporating a large range of structurally diverse fluorine containing groups.

2.6 *P,O*-Based binding modes

Being *O*-alkylated with perfluoroalkyl chains at the very last step of the synthesis, 2-(diphenylphosphino)-2'-alkoxy-1,1'-binaphthyl (MOP) (*R_a*)-349 was used, together with [Pd(η³-C₃H₅)Cl]₂, in the asymmetric alkylation reaction of β-dicarbonyl derivatives with 1,3-diphenyl-2-propenyl *rac*-114 (Scheme 36).^{127,139} The corresponding alkylated products (*R*)-116 and 353–355 were obtained in moderate to excellent yields, and good ees were obtained. Noteworthy, (*R_a*)-349 was completely extracted from the reaction mixture using *n*-perfluorooctane, whereas the catalytic activity of the recycled Pd^{II} complex was lost when used in subsequent reactions.

2.7 *C*-Based binding modes

2.7.1 Diaminocarbenes. Chiral Au^I complexes (*R_a*)-356–358 involving acyclic diaminocarbene ligands were described as efficient catalysts in the cyclization reaction of



Scheme 36 (*R_a*)-349/Pd^{II}-catalyzed alkylation reaction of *rac*-114.



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