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Paired electrolysis enabled annulation for the quinolyl-modification of bioactive molecules†

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A paired electrolysis enabled cascade annulation that enables the efficient synthesis of highly functionalized quinoline-substituted bioactive molecules from readily available starting materials is reported. Using this methodology, two goals, namely, the direct synthesis of quinolines and the introduction of quinoline moieties to bioactive molecules, can be simultaneously achieved in one simple operation. The use of electroreduction for the activation of isatin, together with the further anodic oxidation of KI to catalytically result in a cascade annulation, highlight the unique possibilities associated with electrochemical activation methods. This transformation can tolerate a wide range of functional groups and can also be used as a functionalization tactic in pharmaceutical research as well as other areas.

Introduction

The quinoline moiety is the central core of a wide variety of naturally occurring and manmade compounds, many of which exhibit extraordinary biological and pharmaceutical properties (e.g. antimalarial, antifungal, antibacterial, antiprotozoal, antitubercular, anticancer, antitumor, anti-HIV, antiproliferative, antiinflammatory, antioxidant, DNA binding and antihypertensive and antiviral).¹ Alternatively, quinolines provide frameworks for industrial uses including organic light-emitting diodes (OLEDs) and photovoltaic cells, as well as solvents for terpenes and resins.² Consequently, a variety of novel and expeditious approaches for the preparation of a diverse range of substituted quinolines have been developed.³ The quinoline scaffold has even been described as a “privileged structure”, since members of this structural class are able to interact with different types of bioactive targets and have attracted the interest of investigators in a broad spectrum of sciences from chemistry and biology to medicine.⁴ Hence, introducing quinoline derivatives into bioactive molecules in the late-stage functionalization of natural products and drugs under mild conditions is of equal importance to the direct synthesis of

quinolines in the field of chemical biology and pharmacology. Achieving these two goals in one simple operation would have great appeal.

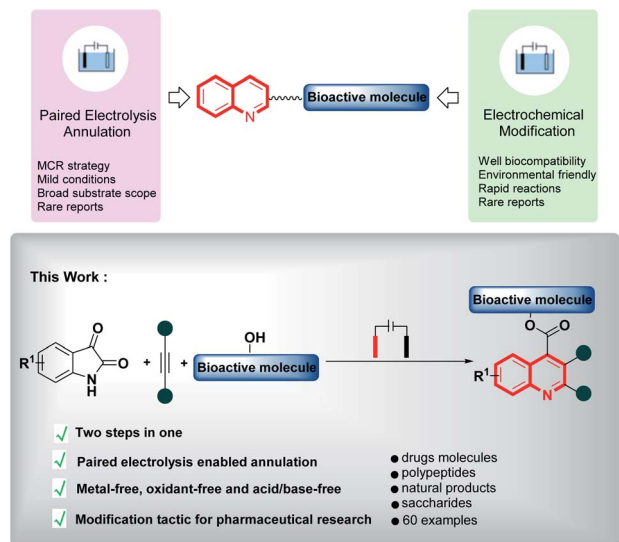
Organic electrosynthesis, as an environmentally friendly synthetic strategy has witnessed a renaissance.⁵ Compared to traditional methods that often proceed at strong oxidative or reductive conditions with elevated temperature or pressure, electrochemical reactions are usually carried out under milder conditions by precisely varying the applied electrode potential. Therefore, electrosynthesis is usually compatible with highly functionalized substrates and has displayed great potential in both synthetic and bioconjugation chemistry.⁶ Many efforts have been made to develop efficient electrochemical reactions for the synthesis of various heterocyclic structures, in which electrochemical multicomponent reactions have emerged as a versatile and efficient tool for the construction of heterocycles, and has attracted considerable attention.⁷ However, the vast majority of these methodologies have focused on electro-generated based Michael addition reactions, a strategy that requires substrates bearing an acidic methylene hydrogen, e.g. malononitrile or malonate, but examples of the synthesis of bioactive molecule-bearing heterocycles using these multicomponent reaction methodologies are rare. Herein, we introduce a ‘one-stone, two-birds’ strategy for accessing valuable quinolines using a paired electrolysis enabled multi-component reaction approach and the simultaneous late-stage functionalization of various bioactive molecules (Scheme 1). We started from commercially available isatins, alkynes, and hydroxy-containing bioactive molecules, and various quinoline-substituted target molecules were isolated in moderate to excellent yields. The traditional routes to quinolines started from isatins usually suffer from harsh conditions and poor tolerance of functional groups, the use of electroreduction for

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Scheme 1 Paired electrolysis enabled annulation to construct quinoline-substituted bioactive molecules.

the activation of isatin, together with the further anodic oxidation of KI to catalytically accomplish the intermolecular cascade annulation under mild pH neutral conditions, highlight the unique possibilities associated with electrochemical activation methods.

Initially, isatin (**1a**), dimethyl but-2-ynedioate (**2a**) and a serine residue (**3a**) were used as model substrates for the electroreduction-induced cascade annulation. To our delight, the desired quinoline product (**4a**) was obtained in 85% yield under a constant current of 5 mA with 20 mol% KI in a simple undivided cell (Table 1, entry 1). KI was found to be required for this cascade annulation to proceed and, in its absence, there was no reaction (Table 1, entry 2). Replacing CH₃CN with DCM

did not favor the annulation (Table 1, entry 3). The effect of electrode materials was also explored. Using a carbon rod as the cathode decreased product yield, to some extent (Table 1, entry 4). A good yield was obtained with KI as electrolyte (Table 1, entry 5). Additionally, control experiments indicated that no product was produced in the absence of electric current (Table 1, entry 6). More detailed information can be found in the ESI.†

Remarkably, this protocol also provided direct access to the late-stage derivatization of valuable drugs and natural products (Scheme 2). Tetrahydrofurfuryl alcohol, perilla alcohol, cinnamyl alcohol and isoborneol, which are important food and flavoring additives, also participated in the annulation reaction with moderate to good yields, respectively (**4aj–4am**). Additionally, menthol, geraniol and phytol could also be readily converted into the corresponding products in 42–70% yields (**4an–4ap**). Saccharides are important types of biomolecules for which late-stage functionalization would be highly desirable. Interestingly, this reaction can be used for the synthesis of various quinoline-substituted saccharide derivatives, which highlights the potential of this method for bioconjugation reactions (**4aq–4as**). Meanwhile, when various drug molecules, such as buctin (an analgesic-antipyretic drug), ospemifene (an estrogen receptor modulator) metronidazole (an antibiotic and anti-protozoan drug), picaridin (a commonly used insect repellent), dehydroepiandrosterone (a steroid agent) and testosterone were treated with isatin (**1a**) and dimethyl but-2-ynedioate (**2a**), the desired functionalized products **4at** (75%), **4au** (67%), **4av** (60%), **4aw** (73%), **4ax** (52%) and **4ay** (54%) were obtained, respectively. To further explore the utilities of this electrochemical cascade annulation for the synthesis of quinoline-substituted bioconjugates, gram scale reactions were performed for three kinds of transformations (**4av**, **4ak** and **4al**). In these cases, the reaction remained selective and good yields were obtained.

Results and discussion

We next explored the generality of the protocol for a wide range of isatin derivatives (Scheme 3). The effect of substituents on the benzene ring of the isatin molecule was first scrutinized. Molecules with electron-donating groups and halogens at different positions reacted smoothly to yield the corresponding products with satisfactory efficiencies (**4ba–4bi**). Tri-fluoromethyl and nitro groups as electron-withdrawing groups were also amenable (**4bj** and **4bk**). The scope of the reaction with respect to alkynes was also explored. The installation of the bulky *tert*-butyl group in the ester moiety had no effect on the progress of the reaction, and the corresponding product **4bm** was produced in 60% yield. Other different electron-withdrawing groups, such as, acyl and trifluoromethyl groups, all afforded the products in moderate to good yields (**4bn–4bp**).

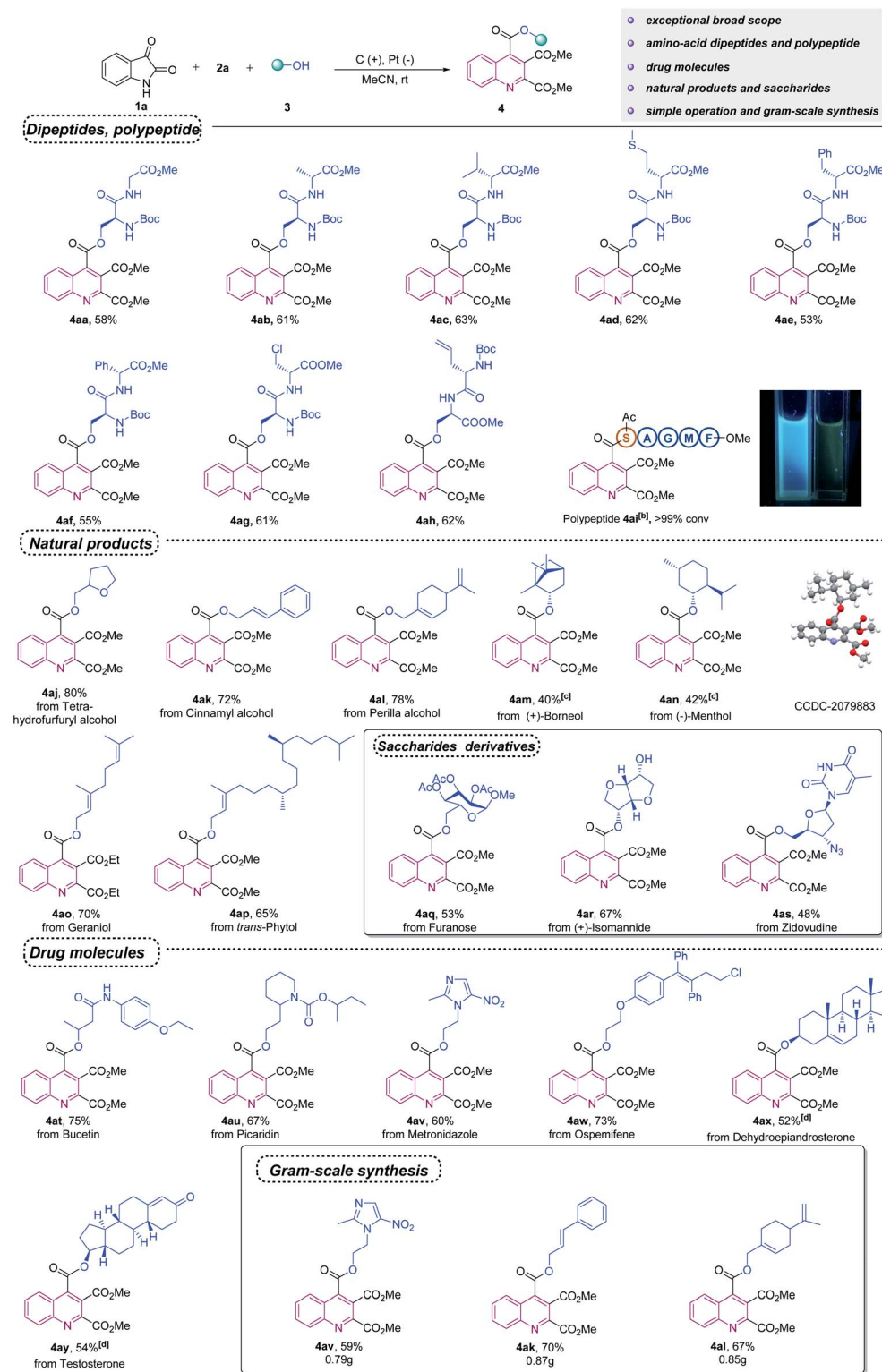
Further exploration revealed that the cascade annulation was amenable to simple alcohols (Scheme 3). The reaction also proceeded effectively with various alcohols. Several typical functional groups such as chloro, alkenyl, alkynyl, azido and trimethylsilyloxy groups were well tolerated under the standard conditions (**4ca–4cd** and **4cg**). Owing to importance of the

Table 1 Investigation of the reaction condition^a

Entry	Variation from standard conditions ^a	Yield (%)
1	Standard conditions	85
2	Without KI	N.D.
3	CH ₂ Cl ₂ as the solvent	Trace
4	Carbon as cathode	70
5	I ₂ instead of KI	N.D.
6	KI as electrolyte	89
7	Without electric current	N.R.

^a Standard conditions: carbon rod anode, platinum plate cathode (15 mm × 15 mm × 0.3 mm), constant current = 5 mA, **1a** (0.3 mmol), **2a** (1.5 equiv.), **3a** (1.5 equiv.). ^a tBuNBF₄ (0.5 equiv.), KI (0.2 equiv.), 6 mL MeCN, undivided cell, 100 min. Isolated yields were shown.

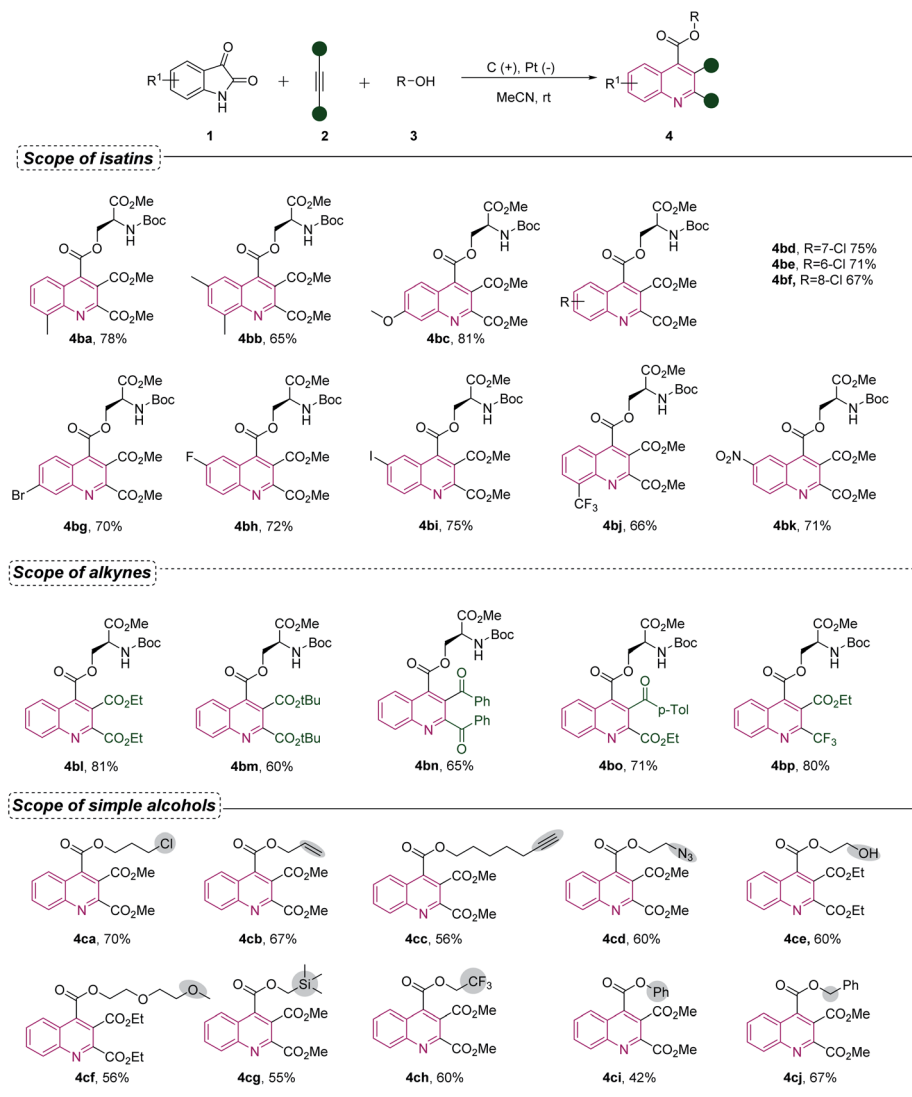




Scheme 2 Scope of bioactive molecules^a. ^aStandard condition as shown in Table 1. ^bReaction conditions: graphite rod anode, platinum plate cathode, constant current = 5 mA, polypeptides 3ai (5 mg), 1a (10 mg), 2a (12 μ L), ⁿBu₄NBF₄ (5 mg), KI (2 mg), CH₃CN (0.75 mL), room temperature, 30 min. Conversion of products are shown, as determined by HPLC. ^c4.0 equiv. of isoborneol and menthol were used. ^d3.0 equiv. of DHEA and testosterone were used.

fluorine atom in drug molecules, we attempted to introduce fluorine atoms into products. To our delight, good yields of the targeted molecules, containing a trifluoromethyl group (4ch)

were successfully obtained under the electrochemical conditions. Alkoxy alcohols were well tolerated, affording the corresponding products (4cf). When a molecule containing a glycol



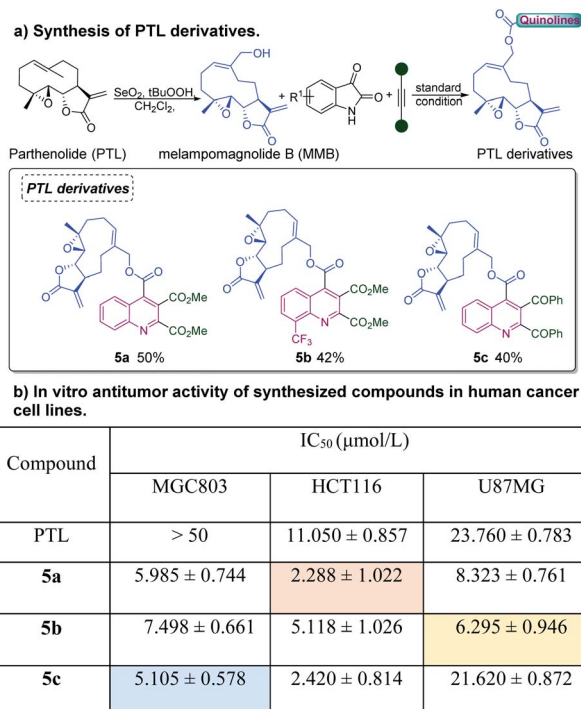
Scheme 3 Substrate scope for isatins, alkynes and selected simple alcohols^a. ^aStandard condition as shown in Table 1.

group was used as a substrate, one hydroxyl group is retained, providing the opportunity for further derivatization (**4ce**). Phenol and benzyl alcohol could also be readily converted into the corresponding products in acceptable yields (**4ci** and **4cj**). Unfortunately, the electron rich alkynes and terminal alkynes, such as phenylacetylene, 1,2-diphenylethyne and methyl propiolate were not suitable reaction partners for this approach, and the substrates containing amine group were incompatible with this transformation (for more details, see ESI†).

We next explored the utilities of this electrochemical cascade annulation for the synthesis of quinoline-substituted bioconjugates. Parthenolide (PTL, shown in Scheme 4a), a naturally occurring sesquiterpene lactone derived from medicinal plants, drew extensive scientific research due to its potency to induce the apoptosis of cancer stem cells (CSCs). Studies have found that PTL, cotreated with other bioactive molecules, showed a broad spectrum of cancer inhibition, including leukemia, lymphoma, bone cancer, and breast cancer.⁸

Therefore, the modification of PTL with quinoline scaffolds would be necessary and meaningful to improve their pharmacological properties. To our delight, a series of PTL derivatives were easily obtained under our electrochemical conditions. The *in vitro* antiproliferative activity of synthesized compounds was also evaluated in three human cancer cell lines using the CCK-8 assay, including HCT116 (colorectal carcinoma), U87MG (glioblastoma) and MGC803 (gastric cancer). It could be seen in the table that all synthesized compounds showed moderate to significant potency toward these cancer cell lines, which highlighted the potential of this method to pharmaceutical research (Scheme 4b). Furthermore, some highly functionalized quinoline-substituted bioconjugates could also be further linked with more biomolecules by using operationally simple techniques (Scheme 5). Initially, we treated the valine residue substituted isatin **6** with dipeptides under the standard conditions, quinoline derivatives connected to two different amino acid residues or dipeptides such as **7** was obtained in 48%





Scheme 4 (a) Synthesis of PTL derivatives. (b) *In vitro* antitumor activity of PTL derivatives in human cancer cell lines.

yields. A bromine functional group in quinolines can be used in Sonogashira reactions with drug molecules to afford the product **10**.

With the method being established, efforts were then made to understand the reaction mechanism. We initiated our study by conducting cyclic voltammetry (CV) experiments. As shown in Fig. 1, using KI as the electrolyte (Fig. 1), the cyclic voltammograms showed irreversible reduction waves for isatin (**1a**). This result indicates that isatin is reduced by the cathode with the generation of a radical-anion intermediate. On the other

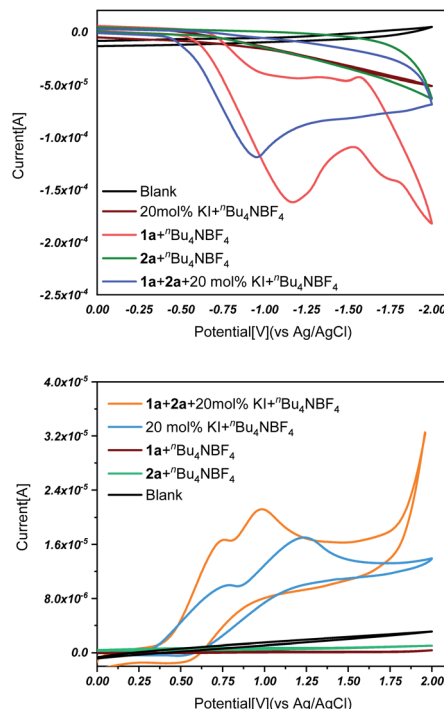
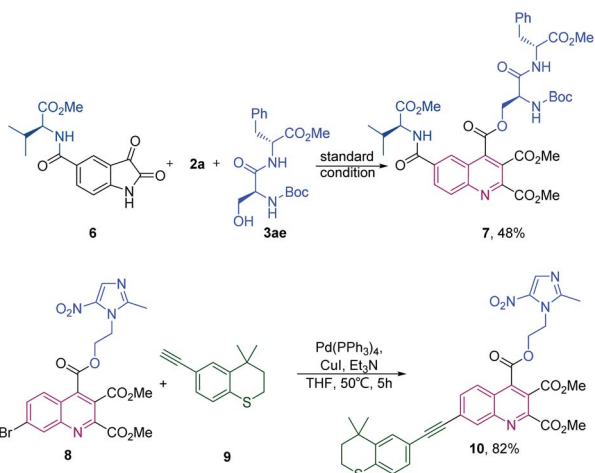


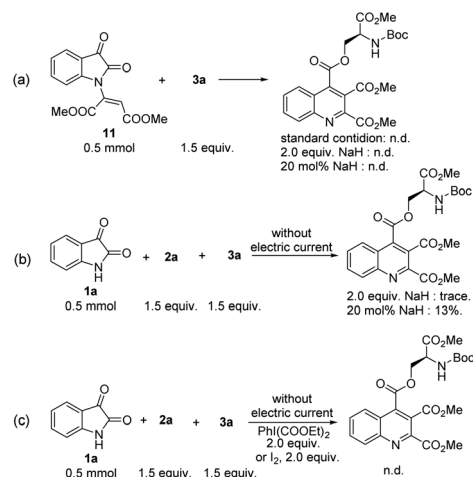
Fig. 1 Cyclic voltammograms studies.

side, the potassium iodide produced a dual oxidation peak at 0.82 V and 1.22 V *versus* Ag/Ag⁺, respectively. No other oxidation peak was observed under these chemical conditions. These results indicate that in this electrochemical transformation, both isatin (**1a**) and KI are simultaneously reduced/oxidized with the generation of reactive intermediates.

To collect more mechanistic information concerning this transformation. Some mechanistic experiments were performed. No product was detected when dimethyl 2-(2,3-dioxindolin-1-yl) maleate **11** was used as a substrate (Scheme 6a), indicating that **11** is not a reaction intermediate. In addition, when 2.0 equivalents of NaH were used to instead of an

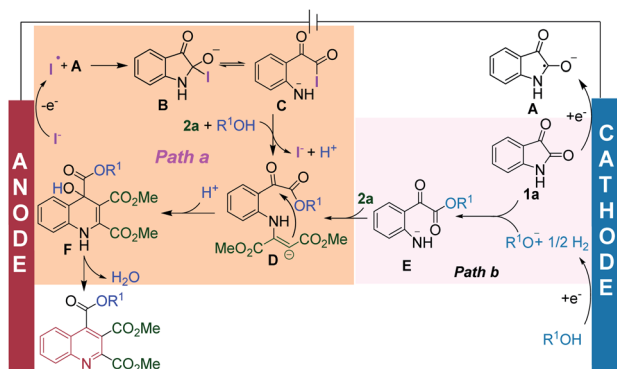


Scheme 5 Chemoselective transformations of quinoline-substituted bioactive molecule.



Scheme 6 Mechanistic experiments.





Scheme 7 Proposed mechanism.

electric current, the reaction mixture was complicated and the corresponding product was produced in trace yield, however, catalytic amount of NaH indeed led to the desired product with low yield (Scheme 6b). This indicates that alkoxide ions are not likely the mainly active species in this transformation, however, we cannot exclude the possibility that the ring-opening step involves nucleophilic attack of electrogenerated alkoxide ions. In this case, the KI would act as the anodic sacrificial agents. Moreover, when molecular iodine and a hypervalent iodine reagent were employed as the oxidant, the desired product was not obtained (Scheme 6c).

Based on the above studies, a proposed mechanism for this electrochemical cascade annulation is shown in Scheme 7. In the first step, **1a** is reduced at the cathode to generate the isatin radical anion **A**. At the same time, the anodic oxidation of KI generates an iodine radical. Radical–radical cross coupling between the isatin radical anion **A** and iodine radical affords intermediate **B**. After a ring-opening reaction, nucleophilic substitution and addition generates intermediate **D**, with the regeneration of the iodine anion to restart the catalytic cycle (Path a). In the meanwhile, we cannot exclude the possibility that the alcohol may also be reduced at the cathode to generate alkoxide ions, and then reacts with the isatin to furnish the ring-opening intermediate **E**. In this case, the KI would act as the anodic sacrificial agents (Path b). Finally, intramolecular cyclization then generates intermediate **F** with excellent regioselectivity, and dehydration gives the corresponding quinoline product.

Conclusions

In summary, we report on a paired electrolysis enabled cascade annulation that enables the efficient synthesis of highly functionalized quinoline-substituted bioactive molecules from readily available starting materials. Merging electrochemical organic synthesis with bioconjugation chemistry, two goals, namely, the direct synthesis of quinolines and introducing quinolines to biomolecules, can be simultaneously achieved in one simple operation. This transformation can tolerate a wide range of biomolecules, *e.g.* serine residues, polypeptides, saccharides, drugs and natural products, and can also be used

as a functionalization tactic for pharmaceutical research and other areas. We anticipate that advances in electrochemically induced bioconjugation will lead to an expanding library of interdisciplinary methodologies.

Data availability

All the data have been included in the ESI.†

Author contributions

M. Gao, A. Lei conceived the project. S. You, M. Ruan, C. Lu, L. Liu, Y. Weng, G. Yang, S. Wang and H. Alhumade performed the experiments, analyzed the data, and discussed the results. M. Gao and S. You wrote the manuscript, Supplementary Information and contributed other related materials.

Conflicts of interest

There are no conflicts to declare.

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