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CORRECTION

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Correction: Synthesis and stereocomplex formation of enantiomeric alternating copolymers with two types of chiral centers, poly(lactic acid-alt-2-hydroxybutanoic acid)s

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Correction for 'Synthesis and stereocomplex formation of enantiomeric alternating copolymers with two types of chiral centers, poly(lactic acid-alt-2-hydroxybutanoic acid)s' by Hideto Tsuji et al., RSC Adv., 2020, 10, 39000–39007. DOI: 10.1039/D0RA08351H.

In the original manuscript, incorrect weight- and number-average molecular weights ($M_{\rm w}$ and $M_{\rm n}$, respectively) were shown for the synthesized and used polymers in the 2. Experimental section. The correct $M_{\rm w}$ and $M_{\rm n}$ values of P(LLA-*alt*-L-2HB) and P(DLA-*alt*-D-2HB) were 3.00×10^4 , 3.08×10^4 , 1.53×10^4 , and 1.54×10^4 g mol⁻¹, respectively.

Due to the correction of the molecular weight values, the authors would like to remove the following statement regarding the suggested reason for the noncrystallizability of unblended P(LLA-alt-L-2HB) and P(DLA-alt-D-2HB) in the 3.2. Wide-angle X-ray diffractometry section based on the low molecular weight values: "The noncrystallizability of P(LLA-alt-L-2HB) and P(DLA-alt-D-2HB) may be due to their lower M_w values ($M_w = 3.0 \times 10^3$ and 3.1×10^3 , g mol⁻¹, respectively) compared to those of the P(LLA-co-L-2HB) (56/44) and P(DLA-co-D-2HB) (52/48) random copolymers ($M_w = 1.4 \times 10^4$ and $M_w = 1.6 \times 10^4$, g mol⁻¹, respectively) and P(LLA-alt-GA) alternating copolymers ($M_w = 4.8 \times 10^3$ and $M_w =$

Due to the same reason, in the 3.3. Differential scanning calorimetry section, the authors would like to remove the statement "The lower Tm values of P(LLA-alt-L-2HB) and P(DLA-alt-D-2HB) can be attributable to the low molecular weights compared to those of PLLA and PDLA, P(LLA-co-L-2HB) (56/44) and P(DLA-co-D-2HB) (52/48), and P(LLA-alt-GA) and P(DLA-alt-GA).". Additionally, the authors would like to replace the statement "The $T_{\rm m}$ values for P(LLA-alt-1-2HB)/P(DLA-alt-1-2HB) ($M_{\rm w}=3.0\times10^3$ and 3.1×10^3 , g mol⁻¹, respectively) blends are higher than those for solvent-evaporated and melt-crystallized ($T_c = 70 \,^{\circ}\text{C}$) P(L-2HB)/P(D-2HB) ($M_w = 100 \,^{\circ}\text{C}$) 1.8×10^3 and 3.3×10^3 , g mol⁻¹, respectively) homopolymer blends ($T_{\rm m} = 173.0$ and 172.1 °C, respectively)⁸¹ but lower than those for melt-crystallized ($T_c = 130$ °C) PLLA/PDLA ($M_w = 4.0 \times 10^3$ and 5.4×10^3 , g mol⁻¹, respectively) homopolymer blends ($T_m = 10^{-1}$) for melt-crystallized ($T_c = 130$ °C) PLLA/PDLA ($M_w = 4.0 \times 10^3$) and $M_w = 10^{-1}$, respectively) homopolymer blends ($M_w = 10^{-1}$) and $M_w = 10^{-1}$) are the first order to the following state of the first order to the first order 197.5 °C), ⁷⁸ and solvent-evaporated and melt-crystallized ($T_c = 160$ °C) P(LLA-co-L-2HB) (56/44)/P(DLA-co-D-2HB) (52/48) ($M_w = 1.4$ \times 10⁴ and 1.6 \times 10⁴, g mol⁻¹, respectively) random copolymer blends (203.6 and 198.4 °C), ⁶³ and solvent evaporated and meltcrystallized ($T_c = 100$ °C) P(LLA-alt-GA)/P(DLA-alt-GA) ($M_w = 4.8 \times 10^3$ and 5.9×10^3 , g mol⁻¹, respectively) blends (187.8 and 187.6 °C). 67" with the revised statement "The $T_{\rm m}$ values for P(LLA-alt-L-2HB)/P(DLA-alt-D-2HB) ($M_{\rm w}=3.0\times10^4$ and 3.1×10^4 g mol⁻¹, respectively) blends are lower than those for solvent-evaporated and melt-crystallized ($T_c = 70 \, ^{\circ}\text{C}$) P(L-2HB)/P(D-2HB) ($M_w = 10 \, ^{\circ}\text{C}$) $M_w = 10 \, ^{\circ}\text{C}$ 3.1×10^4 and 3.3×10^4 g mol⁻¹, respectively) homopolymer blends ($T_{\rm m} = 218.9$ and 214.5 °C, respectively)⁸¹ and those for meltcrystallized ($T_{\rm c}=130~{\rm ^{\circ}C}$) PLLA/PDLA ($M_{\rm w}=4.0\times10^3~{\rm and}~5.4\times10^3~{\rm g~mol}^{-1}$, respectively) homopolymer blends ($T_{\rm m}=197.5~{\rm ^{\circ}C}$), 78 and solvent-evaporated and melt-crystallized ($T_c = 160$ °C) P(LLA-co-L-2HB) (56/44)/P(DLA-co-D-2HB) (52/48) ($M_w = 1.4 \times 10^4$ and 1.6 × 10⁴ g mol⁻¹, respectively) random copolymer blends (203.6 and 198.4 °C)⁶³, but similar to those of solvent evaporated and meltcrystallized ($T_c = 100$ °C) P(LLA-alt-GA)/P(DLA-alt-GA) ($M_w = 4.8 \times 10^3$ and 5.9×10^3 g mol $^{-1}$, respectively) blends (187.8 and 187.6 °C).67".

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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