



Cite this: *RSC Adv.*, 2022, 12, 33936

Development of a β -cyclodextrin-modified gold nanoparticle-assisted electromembrane extraction method followed by capillary electrophoresis for methadone determination in plasma

Mir Saleh Hoseininezhad-Namin,^{ab} Sibel Aysil Ozkan,^c Elaheh Rahimpour^{id} *^{bd} and Abolghasem Jouyban^{ib} ^{be}

In this study, gold nanoparticles (AuNPs) modified with β -cyclodextrin (β -CD) were used to assist with electromembrane extraction (EME) and were coupled with capillary electrophoresis (CE) and ultraviolet (UV) detection (CE-UV) for the extraction and measurement of methadone from plasma samples. A β -CD-modified AuNP-reinforced hollow fiber (HF) was utilized in this work. The β -CD-modified AuNPs act as an absorbent and provide an extra pathway for the analyte extraction. For obtaining the effect of the presence of β -CD-modified AuNPs in the HF pores, the extraction efficiency of the EME and β -CD-modified AuNPs/EME techniques were compared. Different parameters influencing the extraction efficacy of the EME and β -CD-modified AuNPs/EME methods were optimized. Optimal extractions were performed with 1-octanol as the organic solvent in the supported liquid membrane (SLM), with an applied voltage of 10 V as the driving force across the SLM, and with pH 7.0 in the donor solutions with a stirring speed of 1000 rpm after 20 min and 25 min for the β -CD-modified AuNPs/EME and EME methods, respectively. Under optimal conditions, compared with the EME method, the β -CD-modified AuNPs/EME method exhibited increased extraction efficacy in a short time. The β -CD-modified AuNPs/EME technique demonstrated a lower limit of detection (5.0 ng mL^{-1}), higher extraction recovery (68%), and a more optimal preconcentration factor (135). Furthermore, this method was successfully utilized for measuring methadone in real plasma samples.

Received 11th October 2022
Accepted 11th November 2022

DOI: 10.1039/d2ra06419g

rsc.li/rsc-advances

1 Introduction

Opioid maintenance treatment (OMT) drugs can be used for the treatment of opioid addiction.^{1,2} Methadone is an OMT drug used primarily for addiction and pain treatment (Fig. 1). Nevertheless, it is an addictive drug that can be abused. Moreover, the overdose and nonmedical use of methadone can result in death.^{3,4} Thus, developing a reliable and practical method for measuring its concentration in biological samples is essential.

Various techniques, including high-performance liquid chromatography-tandem mass spectrometry (HPLC-MS/MS),⁵ gas chromatography-mass spectrometry (GC-MS),⁶ gas

chromatography-flame ionization detector (GC-FID),⁷ flow-injection analysis,⁸ capillary electrophoresis (CE),⁹ and HPLC-UV¹⁰ have been used for methadone identification. There are many problems associated with drug analysis of biological samples due to the complexities of biological

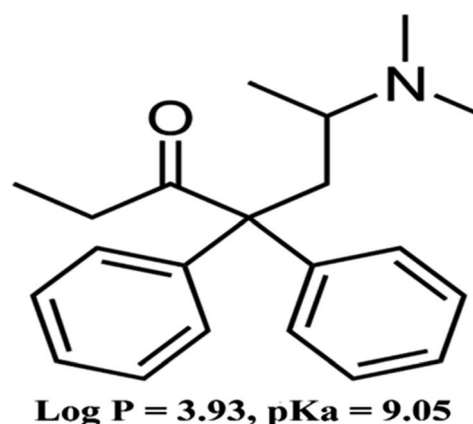


Fig. 1 Chemical structure of methadone.

^aLiver and Gastrointestinal Diseases Research Center, Tabriz University of Medical Sciences, Tabriz, Iran

^bPharmaceutical Analysis Research Center and Faculty of Pharmacy, Tabriz University of Medical Sciences, Tabriz, Iran. E-mail: Rahimpour_e@yahoo.com

^cAnkara University, Faculty of Pharmacy, Department of Analytical Chemistry, Ankara, Turkey

^dInfectious and Tropical Diseases Research Center, Tabriz University of Medical Sciences, Tabriz, Iran

^eFaculty of Pharmacy, Near East University, P.O. Box 99138 Nicosia, North Cyprus, Mersin 10, Turkey



matrices. They contain salts, different proteins, and other components that can interfere with the analytical method.¹¹ Thus, sample pre-treatment before analysis is essential in bioanalytical chemistry.

In the last few decades, scientists focused on miniaturization, efficiency, sensitivity, and environmentally friendly pre-treatment techniques such as solid-phase extraction (SPE), solvent bar microextraction, headspace solid-phase microextraction, liquid-phase microextraction, and electromembrane extraction (EME).^{12,13} Among these emerging techniques, fast, excellent sample clean-up and highly selective extraction can be obtained with EME.¹⁴ With this method, charged analytes are extracted across a supported liquid membrane (SLM) on a microporous polypropylene hollow fiber (HF) based on electrokinetic migration. The use of an electrical field over the SLM increases the extraction efficacy.¹⁵ In the EME method, the flux of analytes across an SLM is described by the modified Nernst-Planck equation. Based on this equation, the magnitude of the applied potential influences the flux of the analyte.¹⁴ Using EME, clean extracts have been obtained from biological samples such as whole blood,¹⁶ saliva,¹⁷ urine,¹⁸ breast milk,¹⁹ and plasma.¹⁸ This method has been developed for different types of acidic and basic drug extraction.^{20–22}

Recently, nanostructures such as carbon nanotubes, silver nanoparticles, maghemite nanoparticles, copper nanoparticles, and titanium dioxide nanoparticles have been utilized in the SLM to increase the efficacy of the EME method.^{23–26} Gold nanoparticles (AuNPs) are used widely in diagnosis, medical investigations, and analytical chemistry.^{27,28} Functionalization can be utilized to produce stable, smaller, efficient, and highly active nanoparticles. Thiols, mercaptans, citrates, polymers, and amines are some capping agents that can increase the efficacy of nanoparticles and prevent their aggregation.^{29,30} β -Cyclodextrin (β -CD) is a known functionalizing agent that is utilized for self-assembly and surface modification of AuNPs.^{31,32} Modification with β -CD enhances the chemical and physical stability of AuNPs and prevents their aggregation. The functionalization of nanoparticles leads to ease in biofunctionalization.^{33,34}

In the current work, a novel strategy was developed to promote the EME method for the extraction of methadone by inserting β -CD-modified AuNPs into the pores of a HF. The presence of β -CD-modified AuNPs in the HF wall can enhance the surface area and increase the analyte migration. Herein, the β -CD-modified AuNP-assisted EME method coupled with capillary electrophoresis was fully validated for the determination of methadone in a plasma sample, and its efficiency was compared with an unmodified EME method for methadone extraction.

2 Materials and methods

2.1 Chemicals and reagents

Methadone was kindly supplied by Tofigh Daru Pharmaceutical Company (Tehran, Iran). All chemicals utilized in this study were of analytical grade. HCl, NaOH, and potassium dihydrogen phosphate (KH_2PO_4) were supplied from Merck (Darmstadt,

Germany). 1-Heptanol, nitrobenzene, 1-octanol, and dihexyl ether were purchased from Fluka (Buchs, Switzerland). Deionized (DI) water was supplied from Shahid Ghazi Pharmaceutical Company (Tabriz, Iran).

2.2 Stock solution

The methadone stock solution was prepared at 5000 ng mL^{-1} in methanol. The stock solution was stored at 4°C and protected from light. Sample solutions were freshly prepared from the stock solution every day.

2.3 Capillary electrophoresis conditions

An Agilent Technologies 7100 (Waldbronn, Germany) capillary electrophoresis system equipped with diode-array detection (190–600 nm) was utilized for analyses. The separation was performed on a 48.5 cm bare fused silica capillary with a 40 cm effective length and an internal diameter of $50 \mu\text{m}$ (Agilent). The background electrolyte (BGE) solution was a 100 mmol L^{-1} borate buffer adjusted to pH 3.0. Before using the CE device, the capillary was washed for 30 min with 1 mol L^{-1} sodium hydroxide. Then, DI water was used for 30 min, and after that, the capillary was washed for 30 min with BGE solution.³⁵

Before beginning electrophoresis every day, the capillary was flushed with 0.1 mol L^{-1} sodium hydroxide (10 min), DI water (10 min), and BGE solution (10 min). Between analyses, the capillary was washed using 0.1 mol L^{-1} sodium hydroxide for 3 min, DI water for 3 min, and BGE for 5 min. All samples and BGE were filtered with a polytetrafluoroethylene (PTFE) syringe filter ($0.20 \mu\text{m}$ pore size). Hydrodynamic injection at a pressure of 0.5 psi for 10 s was used for all samples. The constant voltage and capillary temperature were set at 20 kV and 25°C , respectively. The detection of methadone was accomplished at 195 nm.

2.4 Equipment for EME

A screw-capped homemade glass sample compartment was used with a volume of 7 mL (internal diameter 1 cm and height 7 cm). A porous HF with 0.2 mm wall thickness, 0.6 mm internal diameter, and a pore size of $0.2 \mu\text{m}$ was utilized for the acceptor solution (PP Q3/2 polypropylene HF (Membrana, Wuppertal, Germany)). For two donor and acceptor solution electrodes, platinum wires were used with a 3 mm interelectrode distance and a diameter of 0.2 mm.³⁶ These electrodes were coupled to a power supply with a programmable voltage range of 0 to 300 V and current range of 0 to 1000 mA (model PS858, S.KAR, Tabriz, Iran). A multimeter was applied to monitor the current during extraction (Zhangzhou Weihua Electronic Co., Ltd. China, Model M890C+). The EME system was stirred using a magnetic bar with a heater-magnetic stirrer during the extraction (model RCT basic from IKA Company, Staufen im Breisgau, Germany).

2.5 Synthesis of β -CD-modified AuNPs

β -CD-modified Au NPs were synthesized by adding 0.023 g of β -CD and $200 \mu\text{L}$ of HAuCl_4 50 mmol L^{-1} to a beaker

containing 30 mL deionized water. The mixture was stirred until complete dissolution of reagents occurred. Then, the solution was heated to 100 °C, and 2–3 drops of NaOH (0.07 g mL⁻¹) solution was added. Approximately 8 minutes later, a cherry red solution was formed, confirming the synthesis of AuNPs.³⁷ The synthesized nanoparticles were stored in darkness at 4 °C.

2.6 Procedure for β -CD-modified AuNPs/EME

The HF was cut into 4 cm parts to prepare the nanoparticle-immobilized membrane. β -CD-modified AuNPs were dispersed in the 1-octanol organic solvent. Then, the contents of the HPLC syringe were gently injected into the HF. The SLM (HF filled with the β -CD-modified AuNPs and 1-octanol mixture) was sonicated until the mixture spread and dispersed within the HF pores. The excess solvent in the HF was removed by blowing with the medical syringe. Then, utilizing the micro-syringe, 20 μ L of HCl with pH of 1 (acceptor solution) was added to the lumen of the HF. After that, the end of the HF was closed by nonconductive plastic. Then, 1 mL of plasma sample diluted with 3 mL of water with a pH of 7.2 was utilized as a donor solution. Positive and negative electrodes were inserted into the donor and acceptor solution. The positive and negative electrodes were connected to a DC power supply, and the EME system was placed on the stirrer.

After the extraction process, with the micro-syringe, the acceptor phase was collected and transferred to a micro-vial, followed by CE. The principle of the EME method is shown in Fig. 2. The desired compounds were extracted from a donor, using an SLM, into an acceptor phase. This mass transfer is affected by electrokinetic migration through a stable external electric field in the SLM. The SLM includes an organic solvent (1-octanol and β -CD-modified AuNPs) fixed on the pores of a porous polymeric membrane.

2.7 Calculation of extraction recovery, preconcentration factor, and relative recovery

The extraction recovery (ER%) of methadone from the plasma samples was calculated according to eqn (1):

$$ER = \frac{n_{a,final}}{n_{s,initial}} \times 100 = \left(\frac{V_a}{V_s} \right) \left(\frac{C_{a,final}}{C_{s,initial}} \right) \times 100 \quad (1)$$

where $n_{s,initial}$ and $n_{a,final}$ denote the number of moles of methadone present in the sample solution (donor phase) and the number of moles of methadone present in the acceptor solution after extraction, respectively. $C_{s,initial}$ and $C_{a,final}$ denote the initial and final methadone concentration in the donor and acceptor solutions, respectively, whereas V_s and V_a denote the donor and acceptor volumes, respectively. The preconcentration factor (PF) was obtained utilizing eqn (2):

$$PF = \frac{C_{a,final}}{C_{s,initial}} \quad (2)$$

Percent relative recovery (% RR) was calculated with eqn (3):

$$RR = \frac{C_{found} - C_{real}}{C_{added}} \times 100 \quad (3)$$

where C_{added} , C_{real} , and C_{found} denote the concentration of the known amount of the standard added to the real plasma sample, the concentration of methadone in the real plasma sample, and the concentration of methadone following the addition of a known amount of the standard into the real plasma sample, respectively.

2.8 Ethical considerations

The sample donors signed a consent form approved by the Ethics Committee at Tabriz University of Medical Sciences (code of ethics committee: IR.TBZMED.REC.1400.426).

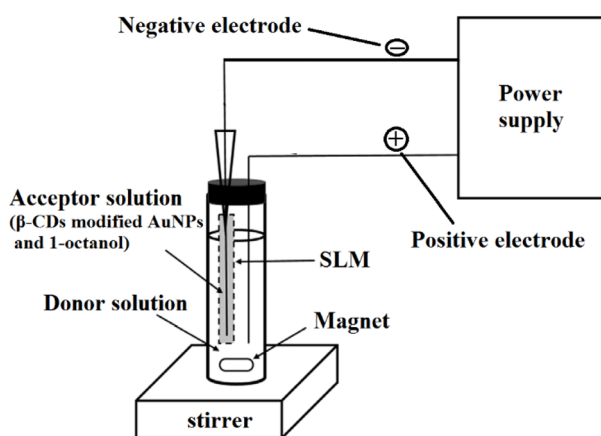
3 Results and discussion

3.1 Characterization of β -CD-modified Au NPs

TEM analysis was used to investigate the size and morphology of the prepared nanoparticles. Fig. 3A shows that the β -CD-modified Au NPs possess a spherical morphology with an average size of <30 nm that was confirmed by dynamic light scattering (DLS) measurement (Fig. 3B).

3.2 Optimization of EME and β -CD-modified AuNPs/EME conditions

Different parameters affecting the extraction process were optimized to establish excellent extraction. The initial experimental conditions during the optimization were (1) 100 ng mL⁻¹ of methadone, 750 rpm as the stirring rate, voltage of 15 V, 15 min extraction time, donor solution pH of 7.0, and 300 μ L of β -CD-modified AuNPs for effect of the SLM composition on the EME and β -CD-modified AuNPs/EME methods; (condition), (2) 100 ng mL⁻¹ of methadone, 750 rpm as the stirring rate, 15 min extraction time, donor solution pH of 7.0, and 300 μ L of β -CD-modified AuNPs for effect of applied voltage, (3) 100



Electromembrane extraction setup

Fig. 2 Principle of EME.



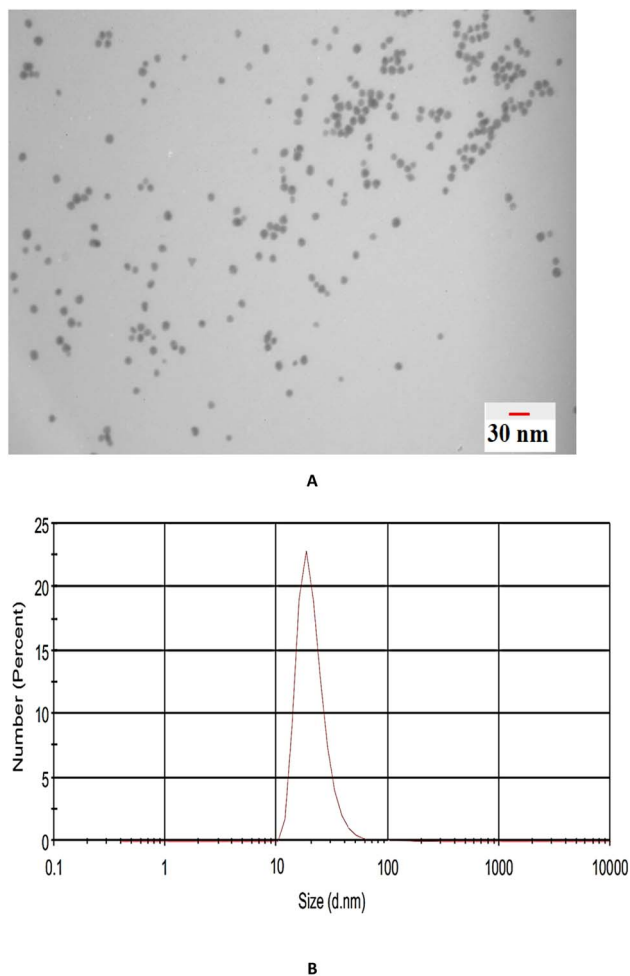


Fig. 3 (A) TEM image and (B) DLS analysis of β -CD-modified AuNPs.

ng mL⁻¹ of methadone, voltage of 10 V, 15 min extraction time, donor solution pH of 7.0, and 300 μ L of β -CD-modified AuNPs for effect of stirring rate, (4) 100 ng mL⁻¹ of methadone, voltage of 10 V, 1000 rpm as the stirring rate, donor solution pH of 7.0, and 300 μ L of β -CD-modified AuNPs for effect of extraction time, (5) 100 ng mL⁻¹ of methadone, voltage of 10 V, 1000 rpm as the stirring rate, 20 min extraction time for the β -CD-modified AuNPs/EME method and 25 min extraction time for the EME method, and 300 μ L of β -CD-modified AuNPs for effect of donor solution pH, (6) 100 ng mL⁻¹ of methadone, voltage of 10 V, 1000 rpm as the stirring rate, 20 min extraction time for the β -CD-modified AuNPs/EME method and 25 min extraction time for the EME method, and solution pH of 7.0 for effect of β -CD-modified AuNP concentration.

3.2.1 Selection of the SLM organic solvent. During the optimization process, the first parameter is the type of organic solvent utilized in the SLM for electrokinetic migration across the membrane. The solvent should have electrical conductivity to establish the electric field between the donor and acceptor phases. To prevent solvent loss in the extraction process, the organic solvent vapor pressure must be low.³⁸ Furthermore, the analyte tendency should be higher in the distribution into the SLM compared with the donor phase for the transfer of the

analyte to the acceptor phase. Moreover, the β -CD-modified Au NPs must be well dispersed in the organic solvent when using the β -CD-modified AuNPs/EME method.

In this work, we used solvents tested in previous works for extracting methadone and opioids with the EME method.^{7,39} To determine the most optimal organic solvent, 1-heptanol, nitrobenzene, 1-octanol, and dihexyl ether were investigated for extraction efficiency using both methods. The results are shown in Fig. 4A, and they indicate that the greatest efficiency was obtained with 1-octanol. 1-Octanol showed the greatest extraction recovery with effective extraction of the analyte. Thus, in this work, 1-octanol was chosen for further experiments.

3.2.2 Effect of voltage on EME. The applied voltage is the driving force for the electrokinetic migration of analytes in the EME technique. Therefore, various voltages were used to

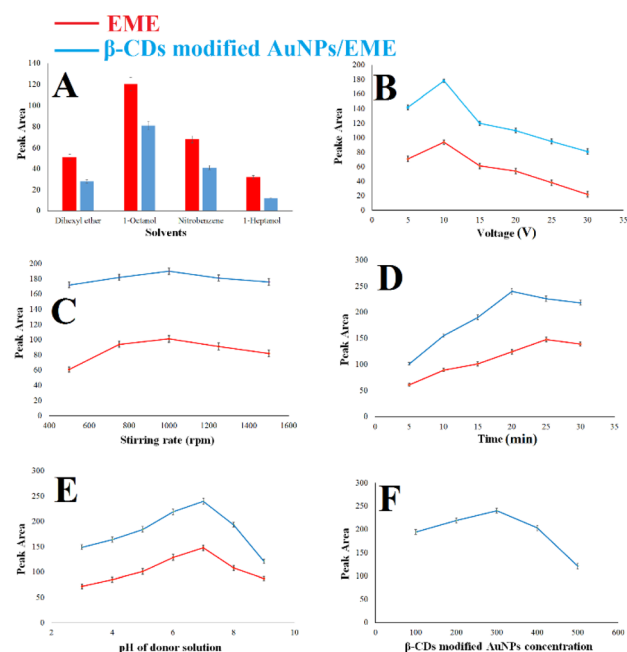


Fig. 4 (A) Effect of the SLM composition on the EME and β -CD-modified AuNPs/EME methods; (condition: 100 ng mL⁻¹ of methadone, 750 rpm as the stirring rate, voltage of 15 V, 15 min extraction time, donor solution pH of 7.0, and 300 μ L of β -CD-modified AuNPs), (B) effect of applied voltage; (condition: 100 ng mL⁻¹ of methadone, 750 rpm as the stirring rate, 15 min extraction time, donor solution pH of 7.0, and 300 μ L of β -CD-modified AuNPs), (C) effect of stirring rate; (condition: 100 ng mL⁻¹ of methadone, voltage of 10 V, 15 min extraction time, donor solution pH of 7.0, and 300 μ L of β -CD-modified AuNPs), (D) effect of extraction time; (condition: 100 ng mL⁻¹ of methadone, voltage of 10 V, 1000 rpm as the stirring rate, donor solution pH of 7.0, and 300 μ L of β -CD-modified AuNPs), (E) effect of donor solution pH; (condition: 100 ng mL⁻¹ of methadone, voltage of 10 V, 1000 rpm as the stirring rate, 20 min extraction time for the β -CD-modified AuNPs/EME method and 25 min extraction time for the EME method, and 300 μ L of β -CD-modified AuNPs). (F) Effect of β -CD-modified AuNP concentration; (condition: 100 ng mL⁻¹ of methadone, voltage of 10 V, 1000 rpm as the stirring rate, 20 min extraction time for the β -CD-modified AuNPs/EME method and 25 min extraction time for the EME method, and solution pH of 7.0). The red color indicates the EME method parameters, and the blue color indicates the β -CD-modified AuNP method.



investigate the optimum condition (between 5 to 30 V). Fig. 4B shows that the peak area increases with increasing voltage up to 10 V for the EME and β -CD-modified AuNPs/EME methods. However, a decrease in signal was observed when 10 V was exceeded. Two theories can illustrate this. First, electrolysis of water in the negative electrode occurred by increasing the voltage. Thus, hydronium ions increase in the acceptor solution, and the pH of the acceptor solution gently rises.⁴⁰ Second, bubbles formed in the SLM at high voltages, which can hinder analyte migration.⁴¹ Therefore, 10 V as an optimum voltage was chosen.

3.2.3 Effect of stirring rate. Stirring decreases the time until thermodynamic equilibrium and increases mass transfer.⁴² Thus, stirring rates between 500 and 1500 rpm were studied. The extraction signal increased until 1000 rpm due to convection effects and then remained constant. These results indicated that 1000 rpm is an excellent stirring rate for the extraction of methadone using both methods (Fig. 4C).

3.2.4 Effect of extraction time. Another parameter that can affect the mass transfer is extraction time. Thus, extraction times of 5–30 min were studied to investigate the migration of the analyte to the acceptor solution. As shown in Fig. 4D, the extraction signal increased at 20 and 25 min for the EME and β -CD-modified AuNPs/EME methods, respectively, and then it decreased. A decrease in efficiency at extraction times longer than 20 and 25 min may be due to the Joule heating phenomenon, which can result in the evaporation or dissolution of the organic phase in the sample solution. This effect can be attributed to the destabilization of the transport of the analyte or can result in the back-extraction of target analytes to the SLM.⁴³ Finally, 25 and 20 min were selected as optimal extraction times for these methods, respectively.

3.2.5 Effect of sample pH. In the next step, the sample solution pH was optimized. The ionic concentration of the analyte should be high to achieve electrokinetic migration. Thus, different sample solution pH values (from 1 to 10) were studied to achieve optimum conditions. As indicated in Fig. 4E, pH 7.0 is more efficient for extracting methadone from the plasma in the sample solution. As is clear, the maximum peak area was achieved for the highest ion concentration in the acceptor phase in comparison with the donor phase. In the donor phase, potent ionization of the basic methadone drug ($pK_a = 9.05$) is necessary for reliable electrokinetic migration through the SLM. Due to the presence of tertiary amine, at pH 7, it almost completely ionizes (near 100%) and gives a positive charge. Raising the pH increases the competition between hydronium and analyte ions, and thus decreases the extraction performance.⁴⁴ Thus, pH 7.4 (physiological pH) was chosen for further investigations of both systems.

The pH of the acceptor phase was also optimized by the same method to investigate the effect of pH in the acceptor phase on recovery percentage. The results indicated that extraction performance decreases with an increase in the pH. Thus, a pH value of 1.0 was required for the acceptor phase to obtain effective extraction.

3.2.6 Effect of β -CD-modified AuNP concentration. To investigate the influence of the β -CD-modified AuNP concentration on the extraction efficiency, the range of 100 to 500 μ L of nanoparticles in 4.0 mL of 1-octanol was investigated. The results indicated that 300 μ L is an optimum concentration for the β -CD-modified AuNPs in 1-octanol (Fig. 4F). The β -CD-modified Au NPs act as a nanoscale solid-phase extracting agent with a high surface area. They provide sites where the analyte molecules can transfer from the donor to the acceptor phase. The analytes were adsorbed on the surface of the β -CD-modified Au NPs, and were desorbed to the organic solvent in the acceptor phase. At concentrations higher than 300 μ L for the β -CD-modified Au NPs, the analytes were not desorbed completely from the surface of the β -CD-modified Au NPs. Therefore, recoveries decreased. Also, if the concentration of the β -CD-modified Au NPs is very high, accumulation will occur that will close the pores of the HF.

3.3 Analytical performance

In the following, the obtained optimization conditions were used to validate the EME and β -CD-modified AuNPs/EME techniques for determining and extracting methadone from plasma samples with CE-diode array detector (DAD). It should be noted that FDA guidelines were utilized to validate the method.

3.3.1 Linearity and calibration curves. Table 1 shows linearity ranges obtained from 10–1000 ng mL⁻¹ and 5–1000 ng mL⁻¹ for the proposed EME and β -CD-modified AuNPs/EME methods in the plasma matrix. The equations of representative regression for the EME and β -CD-modified AuNPs/EME methods were calculated as $Y = 1832X_m + 26.32$ with R^2 of 0.9997 and $Y = 2244X_m + 0.16$ with R^2 of 0.9998, respectively. The limit of detection (LOD) for methadone was obtained based on a signal-to-noise (S/N) ratio of 3 and was at 3.3 and 1.5 ng mL⁻¹ for the EME and β -CD-modified AuNPs/EME methods, respectively. The upper limits of quantification (ULOQ) and lower limits of quantification (LLOQ) are the highest and lowest calibration curve points with acceptable uncertainty (precision within $\pm 20\%$) for both methods. The PF and ER% values were obtained from eqn (1) and (2), respectively. The PF (ER%) values for the EME and β -CD-modified AuNPs/EME methods were calculated to be 107 (53%) and 135 (68%), respectively.

Table 1 Analytical characteristics of the proposed methods

Method	Calibration equation	Linear range ^a	R^2	LOD ^a	LOQ ^a	PF ^b	% ER ^c
EME	$Y = 1832.9X_m + 27.02$	10–1000	0.9997	3.3	10	107	53
β -CD-modified AuNPs/EME	$Y = 2244X_m + 0.16$	5–1000	0.9998	1.5	5	135	68

^a ng mL⁻¹. ^b Pre-concentration factor. ^c Extraction recovery.



3.3.2 Comparison of the EME and β -CD-modified AuNPs/EME methods. To investigate the effect of β -CD-modified AuNPs in the HF pores for extraction of methadone, the results of the EME and β -CD-modified AuNPs/EME methods were compared. The EME and β -CD-modified AuNPs/EME methods were investigated under similar conditions. As indicated in Table 1, there was increased efficacy for the β -CD-modified AuNPs/EME method (low LOD, high PF, and high % ER) for extraction in a shorter time. Furthermore, this method indicated excellent linear response and repeatability. In contrast, the EME method may not be able to measure low concentrations of methadone.

It can be concluded that the presence of β -CD-modified AuNPs increased the surface area, increased the analyte migration, and consequently increased the extraction efficiency. Moreover, the β -CD-modified AuNPs/EME method is not very complicated, and the method can be developed by adding a certain amount of nanostructure to the organic solvent. Thus, continuous precision, accuracy, recovery, and stability parameters were investigated for the β -CD-modified AuNPs/EME method.

3.3.3 Precision and accuracy. Precision and accuracy are utilized to gauge a technique's repeatability and closeness of the obtained values to the nominal (actual) values. The index of precision is the relative standard deviation (% RSD), and the index of accuracy is the relative error (% RE). The concentration levels should be in the range of $\pm 15\%$ for RSD% and RE% values, and this range for the LLOQ level should be $\pm 20\%$. For the validation of the technique, the intra-day and inter-day assays were measured utilizing LLOQ, low, middle, and high concentrations in the calibration range. There were five quality control (QC) replicates to determine the amount of methadone in plasma samples. As indicated in Table 2, the accuracy and precision values are in acceptable ranges.

3.3.4 Recovery. According to eqn (3), recovery is shown as a percent of the added drug found. The methadone was added at low, middle, and high concentrations of the calibration range to blank plasma samples and subjected to the selected method. The C_{real} should be zero when we used the blank sample. Table 3 indicates that the recovery values were between 92.4 and 102.2%. Thus, the results show that this technique is reliable for methadone analysis in plasma samples.

Table 2 Precision and accuracy of the proposed β -CD-modified AuNPs/EME method for quantification of methadone in plasma samples

Added concentration (ng mL ⁻¹)	Intra-day assay (<i>n</i> = 5) ^a		Inter-day assay (<i>n</i> = 5)	
	RSD ^b (%)	RE ^c (%)	RSD (%)	RE (%)
5	7.81	8.83	8.71	7.51
15	5.01	-3.42	6.11	-5.73
100	1.77	8.04	2.41	8.31
500	6.01	8.79	11.21	-2.72

^a Number of replicates. ^b Relative standard deviation. ^c Relative error.

Table 3 Investigation of the method's accuracy using relative recovery values for the different added concentrations of methadone

<i>C</i> _{added} (ng mL ⁻¹)	<i>C</i> _{found} (ng mL ⁻¹)	RR (%)
15	14.6	97.3
100	92.4	92.4
500	511.0	102.2

Table 4 Stability tests of the proposed β -CD-modified AuNPs/EME method in plasma samples

Added concentration (ng mL ⁻¹)	RE (%)	RSD (%)
Freeze-thaw stability (cycle 1)		
15	-7.67	8.30
500	-5.90	6.11
Freeze-thaw stability (cycle 2)		
15	-1.72	9.19
500	3.02	5.58
Freeze-thaw stability (cycle 3)		
15	-4.70	11.01
500	7.20	5.91
Stock solution (after 4 weeks at 4 °C)		
15	-10.91	6.84
500	8.07	7.04
Bench-top stability (after 6 h at room temperature)		
15	-7.64	5.76
500	5.99	5.62

3.3.5 Stability. The freeze/thaw process, stock solution, and bench-top stability are the stability examination parameters indicated in Table 4. The results showed that the accuracies and precisions of stability examination parameters were within the acceptable range ($\pm 15\%$) of FDA guidelines.

3.3.6 Specificity. The main aims for an EME procedure such as other extraction methods are (1) extraction of the analyte from the matrix to remove the matrix or interference effects and/or (2) analyte pre-concentration to reach a low detection limit. Herein, a β -CD-modified AuNPs/EME method was used for both purposes. A pre-concentration process with PF of 135 was performed on methadone, and it was extracted from plasma samples with complicated matrices. However, to investigate possible interference from other co-administered drugs, interference studies were performed. The specificity parameters for the β -CD-modified AuNPs/EME method were obtained using ten drugs (alprazolam, acetaminophen, clonazepam, chlorthalidopoxide, diazepam, codeine, tramadol, dextromethorphan, morphine, and oxazepam) for corroborating the quantitation of methadone to develop the method in the presence of other drugs. To examine the specificity of the technique in the presence of co-administered drugs (5-fold higher than the biological concentrations), the β -CD-modified AuNPs/EME method was employed. At the migration time for methadone, no interfering peaks and no significant overlap peaks were observed.



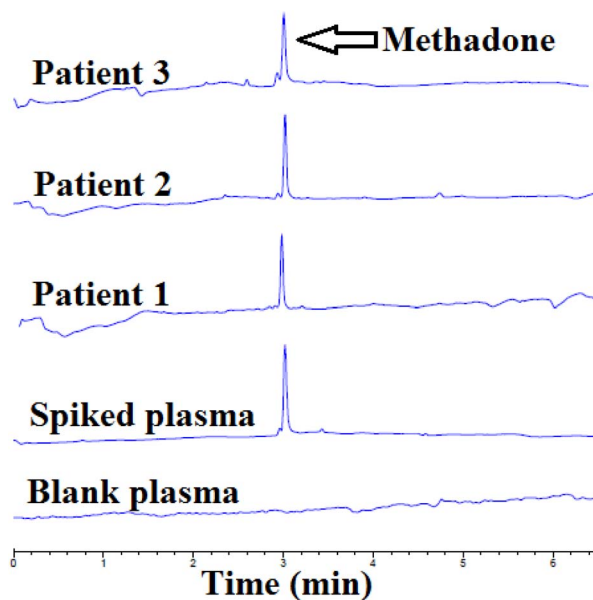


Fig. 5 Electropherograms of a blank plasma sample, plasma samples with added amounts of methadone, and real plasma samples of patients after β -CD-modified AuNPs/EME under optimal conditions.

3.4 Real sample analysis

To examine the method's efficacy in detecting real samples, the β -CD-modified AuNPs/EME method was used for the extraction and detection of methadone from plasma samples of three

Table 5 Details of patients and concentrations of methadone in plasma samples

No.	Gender	Age (year)	Interval between sampling and drug administration (hours)	Daily dose (g)	Methadone in plasma (ng mL ⁻¹)
1	Male	42	5	0.1	164 \pm 0.2
2	Male	46	4	0.2	181 \pm 0.2
3	Male	44	5	0.05	144 \pm 0.1

patients. First, 1.0 mL of plasma was diluted with 3.0 mL of DI water. For this purpose, β -CD-modified AuNPs/EME was used under optimum conditions. Then, the acceptor solution obtained from the β -CD-modified AuNPs/EME method was injected for CE, and the electropherograms of the results are indicated in Fig. 5. Table 5 provides patient details, the administration of the dose, and the concentration of methadone in the plasma samples. The concentrations of methadone in the three real plasma samples were 164 \pm 0.2, 181 \pm 0.2, and 144 \pm 0.1 ng mL⁻¹. Furthermore, to determine the method's accuracy for the investigation of methadone in real plasma samples with consideration of the matrix effect, the recovery test was performed by the addition of a known amount of methadone (100 ng mL⁻¹). The obtained recovery was between 96.0 to 102.0%, which is in the appropriate range.

3.5 Comparison with other methods

Based on Table 6, the β -CD-modified AuNPs/EME method was compared with other methods used for the determination of methadone. The results indicated that the current method has lower detection limits than previously used methods, such as EME/CE-UV,⁴⁵ CNTs-EME/GC-FID,³⁹ EME-SFME/CD-IMS,⁴⁶ LV-ESSM-LLME/HPLC-UV,⁴⁷ and LV-EME/CE-UV.⁴⁸ Furthermore, some instruments used to determine methadone (such as LC-MS and GC-FID) are relatively complicated and expensive. However, the β -CD-modified AuNPs/EME-CE method could provide the same results or better for methadone analysis.

4 Conclusions

EME has been introduced as an effective research field since 2006, and its applicability is rapidly increasing due to the numerous superior features of EME. Most significantly, the sensitivity and selectivity can be adjusted by the modified chemical properties of the SLM, for example, by modifying the SLM with nanostructures. The presence of nanostructures in the SLM results in a more effective surface area and enhances the total partition coefficient of the desired analytes.

In this work, β -CD-modified AuNPs/EME as a modified approach of the EME method coupled with CE-UV was applied

Table 6 Comparison of the β -CD-modified AuNPs/EME method with other methods for extraction of methadone

Method	Sample preparation	Sample type	LOD (ng mL ⁻¹)	Linear range (ng mL ⁻¹)	RSD (%)	PF	% ER	Ref.
CE-UV	EME	Urine	—	15–500	7.5	130	—	45
GC-FID	CNTs-EME	Plasma	3	10–1000	4.4	255	—	39
LC-MS	EME	Plasma	0.4	—	9.0–10.0	—	—	49
GC-FID	SA-PEME ^a	Water	0.6	2–1000	12.7	—	—	7
CD-IMS ^b	EME-SFME ^c	Plasma	2	8–750	10.2–10.9	406.1	63.9	46
HPLC-UV	LV-ESSM-LLME ^d	Plasma	2	10–800	6.5	21.5	57	47
CE-UV	LV-EME ^e	Plasma	—	25–1000	2.9–18.7	4.1	10	48
CE-UV	β -CD-modified AuNPs/EME	Plasma	1.5	5–1000	1.8–11.2	135	68	This study

^a Surfactant-assisted pulsed two-phase electromembrane extraction. ^b Corona discharge-ion mobility spectrometry. ^c Electromembrane extraction and slug flow microextraction. ^d Low-voltage electrically stimulated stir membrane liquid–liquid microextraction. ^e Electromembrane extraction at low voltage.



for monitoring the methadone in plasma samples. A comparison of the β -CD-modified AuNPs/EME method with the EME method showed that the former was the more optimal technique for enhancing the methadone extraction efficacy. This technique exhibits excellent outcomes (lower LOD, higher % ER, and higher PF) in a shorter time, and can be considered an alternative to existing methods. A comparison of this method with previous methods showed that the β -CD-modified AuNPs/EME-CE method provided more optimal results. Finally, the validated technique was successfully utilized for monitoring methadone in real plasma samples. This method with different nanostructural modifiers such as CNTs, GO, RGO, ND-G, AgNPs, CuNPs, and maghemite nanoparticles can be employed for other analytes in samples of different matrices (such as urine or exhaled breath condensate) in future studies.

Conflicts of interest

The authors declare no conflicts of interest.

Acknowledgements

This report is a part of M. S. Hoseininezhad-Namin's PhD thesis submitted to the Faculty of Pharmacy, Tabriz University of Medical Sciences, Tabriz, Iran. The work is financially supported by the Pharmaceutical Analysis Research Center under grant number 67214.

References

- 1 T. Clausen, K. Anchersen and H. Waal, *Drug Alcohol Depend.*, 2008, **94**, 151–157.
- 2 M. T. Brugal, A. Domingo-Salvany, R. Puig, G. Barrio, P. De García Olalla and L. De La Fuente, *Addiction*, 2005, **100**, 981–989.
- 3 K. W. Simonsen, P. Kriikku, G. Thelander, H. M. E. Edvardsen, S. Thordardottir, C. U. Andersen, A. K. Jönsson, J. Frost, D. J. Christoffersen, G. J. M. Delaveris and I. Ojanperä, *Forensic Sci. Int.*, 2020, **313**, 110343.
- 4 A. Casati, R. Sedefov and T. Pfeiffer-Gerschel, *Eur. Addict. Res.*, 2012, **18**, 228–245.
- 5 R. George, M. Lobb, A. Haywood, S. Khan, J. Hardy, P. Good, S. Hennig and R. Norris, *Talanta*, 2016, **149**, 142–148.
- 6 A. M. Bermejo, A. C. S. Lucas, M. J. Tabernero and P. Fernández, *Anal. Lett.*, 2000, **33**, 739–752.
- 7 P. Zahedi, S. S. H. Davarani, H. R. Moazami and S. Nojavan, *J. Pharm. Biomed. Anal.*, 2016, **117**, 485–491.
- 8 R. Montero, M. Gallego and M. Valcárcel, *Anal. Chim. Acta*, 1990, **234**, 433–437.
- 9 S. Hamidi, M. Khoubnasabjafari, K. Ansarin, V. Jouyban-Gharamaleki and A. Jouyban, *Anal. Methods*, 2017, **9**, 2342–2350.
- 10 S. Taheri, F. Jalali, N. Fattahi, R. Jalili and G. Bahrami, *J. Sep. Sci.*, 2015, **38**, 3545–3551.
- 11 S. Seidi, Y. Yamini, A. Heydari, M. Moradi, A. Esrafil and M. Rezazadeh, *Anal. Chim. Acta*, 2011, **701**, 181–188.
- 12 M. Havlikova, R. Cabala, V. Pacakova and Z. Bosakova, *J. Sep. Sci.*, 2019, **42**, 303–318.
- 13 M. Havlikova, R. Cabala, V. Pacakova, M. Bursova and Z. Bosakova, *J. Sep. Sci.*, 2019, **42**, 273–284.
- 14 K. S. Hasheminasab and A. R. Fakhari, *Anal. Chim. Acta*, 2013, **767**, 75–80.
- 15 S. Pedersen-Bjergaard and K. E. Rasmussen, *Anal. Bioanal. Chem.*, 2007, **388**, 521–523.
- 16 A. Gjelstad, K. E. Rasmussen and S. Pedersen-Bjergaard, *Anal. Bioanal. Chem.*, 2009, **393**, 921–928.
- 17 S. Seidi, Y. Yamini, A. Saleh and M. Moradi, *J. Sep. Sci.*, 2011, **34**, 585–593.
- 18 S. Pedersen-Bjergaard and K. E. Rasmussen, *J. Chromatogr. A*, 2006, **1109**, 183–190.
- 19 I. J. Ø. Kjelsen, A. Gjelstad, K. E. Rasmussen and S. Pedersen-Bjergaard, *J. Chromatogr. A*, 2008, **1180**, 1–9.
- 20 A. Gjelstad, K. E. Rasmussen and S. Pedersen-Bjergaard, *J. Chromatogr. A*, 2006, **1124**, 29–34.
- 21 M. Balchen, A. Gjelstad, K. E. Rasmussen and S. Pedersen-Bjergaard, *J. Chromatogr. A*, 2007, **1152**, 220–225.
- 22 M. R. Payán, M. A. B. López, R. F. Torres, M. V. Navarro and M. C. Mochón, *Talanta*, 2011, **85**, 394–399.
- 23 M. Ramos-Payán, R. Fernández-Torres, J. L. Pérez-Bernal, M. Callejón-Mochón and M. T. Bello-López, *Anal. Chim. Acta*, 2014, **849**, 7–11.
- 24 M. S. Hoseininezhad-Namin, E. Rahimpour, S. A. Ozkan and A. Jouyban, *Anal. Methods*, 2022, **14**, 212–221.
- 25 Z. Tahmasebi, S. S. H. Davarani and A. A. Asgharinezhad, *Biosens. Bioelectron.*, 2018, **114**, 66–71.
- 26 M. Khajeh, S. Pedersen-Bjergaard, M. Bohlooli, A. Barkhordar and M. Ghaffari-Moghaddam, *J. Sci. Food Agric.*, 2017, **97**, 1517–1523.
- 27 A. Tomar and G. Garg, *Global J. Pharmacol.*, 2013, **7**, 34–38.
- 28 T. Khan, N. Ullah, M. A. Khan, Z. U. R. Mashwani and A. Nadhman, *Adv. Colloid Interface Sci.*, 2019, **272**, 102017.
- 29 S. Alex and A. Tiwari, *J. Nanosci. Nanotechnol.*, 2015, **15**, 1869–1894.
- 30 T. John, A. Gladysz, C. Kubeil, L. L. Martin, H. J. Risselada and B. Abel, *Nanoscale*, 2018, **10**, 20894–20913.
- 31 S. Pande, S. K. Ghosh, S. Praharaj, S. Panigrahi, S. Basu, S. Jana, A. Pal, T. Tsukuda and T. Pal, *J. Phys. Chem. C*, 2007, **111**, 10806–10813.
- 32 T. Huang, F. Meng and L. Qi, *J. Phys. Chem. C*, 2009, **113**, 13636–13642.
- 33 M. R. Bindhu, K. Ancy, M. Umadevi, G. A. Esmail, N. A. Al-Dhabi and M. V. Arasu, *J. Photochem. Photobiol., B*, 2020, **210**, 111965.
- 34 C. Burda, X. Chen, R. Narayanan and M. A. El-Sayed, *Chem. Rev.*, 2005, **105**, 1025–1102.
- 35 B. Seyfinejad, M. Khoubnasabjafari, S. E. Ziaei, S. A. Ozkan and A. Jouyban, *Daru, J. Pharm. Sci.*, 2020, **28**, 615–624.
- 36 B. Seyfinejad, A. Meshkini, P. Habibolah, S. A. Ozkan and A. Jouyban, *Electrophoresis*, 2020, **41**, 666–677.
- 37 R. Rajamanikandan, A. D. Lakshmi and M. Ilanchelian, *New J. Chem.*, 2020, **44**, 12169–12177.
- 38 L. Xu, P. C. Hauser and H. K. Lee, *J. Chromatogr. A*, 2008, **1214**, 17–22.



- 39 K. S. Hasheminasab, A. R. Fakhari and M. H. Koruni, *J. Sep. Sci.*, 2014, **37**, 85–91.
- 40 M. S. Hoseininezhad-Namin, B. Seyfinejad, S. A. Ozkan, J. Soleymani, M. Khoubnasabjafari, V. Jouyban-Gharamaleki, E. Rahimpour and A. Jouyban, *J. Pharm. Biomed. Anal.*, 2022, **219**, 114959.
- 41 M. Balchen, L. Reubsaet and S. Pedersen-Bjergaard, *J. Chromatogr. A*, 2008, **1194**, 143–149.
- 42 A. Sarafray-Yazdi, A. H. Amiri and Z. Es'haghi, *Chemosphere*, 2008, **71**, 671–676.
- 43 S. Asadi and S. Nojavan, *Anal. Chim. Acta*, 2016, **923**, 24–32.
- 44 S. Seidi, Y. Yamini and M. Rezazadeh, *J. Chromatogr. B: Anal. Technol. Biomed. Life Sci.*, 2013, **913–914**, 138–146.
- 45 M. H. Koruni, H. Tabani, H. Gharari and A. R. Fakhari, *J. Chromatogr. A*, 2014, **1361**, 95–99.
- 46 M. Behpour, M. Maghsoudi and S. Nojavan, *J. Chromatogr. A*, 2022, **1678**, 463355.
- 47 K. M. Ara and F. Raofie, *Talanta*, 2017, **168**, 105–112.
- 48 N. C. Domínguez, A. Gjølstad, A. M. Nadal, H. Jensen, N. J. Petersen, S. H. Hansen, K. E. Rasmussen and S. Pedersen-Bjergaard, *J. Chromatogr. A*, 2012, **1248**, 48–54.
- 49 C. Huang, L. E. E. Eibak, A. Gjølstad, X. Shen, R. Trones, H. Jensen and S. Pedersen-Bjergaard, *J. Chromatogr. A*, 2014, **1326**, 7–12.

