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## Extraction of flavonoids from *Citri Reticulatae Pericarpium Viride* using a deep eutectic solvent<sup>†</sup>

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Flavonoids are the main active ingredient in *Citri Reticulatae Pericarpium Viride* (CRPV). In this study, a deep eutectic solvent (DES) was explored to extract the main flavonoids from CRPV, including narirutin (NAR) and hesperidin (HES). A total of 30 DESs were prepared, and DES-20 (proline : urea) was selected as the optimal solvent. According to the single factor and response surface methodology experiments, the optimal extraction conditions for the flavonoids included a molar ratio of 1 : 2, water content of 30%, extraction time of 28 min, extraction power of 240 W, and a liquid-to-solid ratio of 70 mL g<sup>-1</sup> and the optimal extraction yields of NAR and HES were 21 ± 2 mg g<sup>-1</sup> and 60 ± 2 mg g<sup>-1</sup>, respectively. The recovery rate of NAR and HES with macroporous resin SP825 were 88 ± 3% and 86 ± 4%, respectively. Compared with traditional extraction methods, this novel method greatly improved extraction efficiency and saved time. Scanning electron microscopy results showed that this novel method could destroy the plant cell wall to enhance the extraction efficiency. Then, 21 flavonoids were identified through an ultra-high-performance liquid chromatography-quadrupole time-of-flight mass spectrometry (UPLC-Q-TOF-MS) technique and nine of them were discovered for the first time in CRPV extracts. Furthermore, a high-performance liquid chromatography (HPLC) method was simultaneously used to determine NAR and HES in CRPV extracts upon methodological validation. Finally, antioxidant and cytotoxic experiments *in vitro* showed that the flavonoids extracted from the CRPV exhibited certain antioxidant and cytotoxic activities. The above results indicated that the DES was a green solvent, which can effectively extract NAR and HES from CRPV.

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## 1. Introduction

*Citrus reticulata* Blanco, a fruit belonging to the Rutaceae family, is one of the most produced and consumed fruits in the world.<sup>1</sup> In 2021, approximately 6.82 million hectares of land were used to cultivate 98 million tons of citrus worldwide.<sup>2</sup> China is a major producer of citrus and its citrus industrial system is relatively complete, including fruits, beverages, canned goods, cosmetics, medicines, and additives.<sup>3</sup> In particular, citrus has become one of the sources of medicine in the Chinese medicine industry. *Citri Reticulatae Pericarpium Viride* (CRPV), derived from the peel of *Citrus reticulata* Blanco, has been widely used as a traditional Chinese medicine for treating chest distension, hernia, nucleus pulposus, and canker sores.<sup>4</sup> Several studies have revealed flavonoids, such as narirutin (NAR) and hesperidin (HES), as the main chemical components of CRPV.<sup>5</sup> Flavonoids exhibit pharmacological activities, including

antitumor, antioxidant, anti-inflammatory, antibacterial, and hypoglycemic.<sup>6,7</sup> In the contemporary world, the flavonoids extracted from CRPV are important precursors for drugs and health products. Therefore, it is necessary to develop an efficient extraction method for extracting flavonoids from CRPV.

Usually, it is found that some extraction factors will exert a huge impact on the extraction process of flavonoids. Therefore, one classic process design method, response surface methodology (RSM), is generally chosen by previous researchers to obtain the optimal extraction process.<sup>8</sup> The response surface methodology (RSM) refers to a statistical and mathematical tool for evaluating the relationship between factors and response values. Compared with other optimization methods (orthogonal experimental design and uniform test design), RSM can not only continuously analyze each level of the experiment, but also reduce the number of experiments and avoid wasting time.<sup>9</sup> For these reasons, RSM is widely used to optimize the extraction process of natural products. Tarun *et al.*<sup>10</sup> used RSM to optimize the extraction of phenolics in barberry fruit, and the results showed that the optimal extraction process were methanol concentration of 80%, pH of 3, liquid-solid ratio of 50 : 1 mL g<sup>-1</sup>, temperature of 80 °C, time of 30 min, and the extraction rate could reach 4659 ± 1%. In addition, in another study, RSM

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was used to optimize the extraction of flavonoids from *Angelica sinensis*, and the results showed that the optimal extraction conditions were a temperature of 80 °C, a time of 4 min, an ethanol concentration of 78%, and a liquid-solid ratio of 35 mL g<sup>-1</sup>, and the best extraction rate was 8 mg g<sup>-1</sup>.<sup>11</sup>

Generally, acetone, ethanol, and acetonitrile are widely used as extraction solvents in natural products such as flavonoids, polyphenols and styrene acrylic. However, these organic solvents are volatile, toxic and environmentally harmful.<sup>12</sup> It is necessary to explore a green, environmentally friendly and efficient extraction solvent to extract active ingredients from natural products. Deep eutectic solvent (DES), firstly proposed by Abbott in 2003, is the eutectic mixture of hydrogen bond donors (HBDs) and hydrogen bond acceptors (HBAs).<sup>13</sup> DES is characterized by easy preparation, low cost, low-vapor pressure, and good thermal stability, thus, DES is considered as an alternative to traditional solvents.<sup>14</sup> According to the excellent characteristics of DES, a large number of complex natural products are fully extracted, including flavonoids, polyphenols and polysaccharides.<sup>15-17</sup> Moreover, ultrasound-assisted DES extraction (UAE) possesses the advantages of both ultrasound extraction and DES extraction, which is able to maximize the yield of natural active ingredients as well as saving time and energy.<sup>18</sup> All of the above points suggest that DES expands the approach to natural active substance extraction.

In this study, the extraction of flavonoids from the CRPV using UAE and the biological activity of CRPV extracts were investigated. First, 30 DESs were prepared and the best performed DESs were selected. Second, the recovery of flavonoids from DES extract was investigated. Then, the extraction conditions were optimized using single factor and response surface methodology (RSM). In addition, the UAE method was compared with the conventional extraction methods (ME and PE). The morphological alterations of the CRPV powder using the UAE and conventional extraction methods were initially evaluated *via* scanning electron microscopy (SEM) characterization. Moreover, the number of flavonoids in CRPV extracts was identified using the ultra-high-performance liquid chromatography-quadrupole time-of-flight mass spectrometry (UPLC-Q-TOF-MS) technique. Subsequently, the high-performance liquid chromatography (HPLC) method was simultaneously used to determine NAR and HES upon methodological validation. Finally, the biological activities of the flavonoid extract, including *in vitro* antioxidant and cytotoxic activities, were investigated.

## 2. Materials and methods

### 2.1. Reagents

The CRPV sample was collected from Zunyi in October 2020 and identified as genuine by the Pharmacy Teaching and Research

Table 1 The viscosity of 30 DESs

Number	Abbreviation	HBA	HBD	Molar ratio	Viscosity (Pa s)
DES-1	Chcl : Lac	Choline	Lactic acid	1 : 2	3
DES-2	Chcl : Ery	Choline	Erythritol	1 : 2	13
DES-3	Chcl : Ure	Choline	Urea	1 : 2	4
DES-4	Chcl : Gly	Choline	Glycerol	1 : 2	3
DES-5	Chcl : But	Choline	1,2,4-Butanetriol	1 : 2	4
DES-6	Chcl : Xyl	Choline	Xylitol	1 : 2	4
DES-7	Chcl : Mal	Choline	L-Malic acid	1 : 2	6
DES-8	Chcl : Cit	Chloride	Citric acid	1 : 2	7
DES-9	Chcl : Glu	Chloride	D-Glucose	1 : 2	7
DES-10	Bet : Lac	Betaine	Lactic acid	1 : 2	3
DES-11	Bet : Gly	Betaine	Glycerol	1 : 2	3
DES-12	Bet : But	Betaine	1,2,4-Butanetriol	1 : 2	5
DES-13	Bet : Ery	Betaine	Erythritol	1 : 2	5
DES-14	Bet : Mal	Betaine	L-Malic acid	1 : 2	11
DES-15	Bet : Glu	Betaine	D-Glucose	1 : 2	14
DES-16	Bet : Xyl	Betaine	Xylitol	1 : 2	13
DES-17	Bet : Cit	Betaine	Citric acid	1 : 2	10
DES-18	Bet : Fru	Betaine	D-Fructose	1 : 2	33
DES-19	Bet : Sor	Betaine	D-Sorbitol	1 : 2	6
DES-20	Pro : Ure	L-Proline	Urea	1 : 2	2
DES-21	Pro : Lac	L-Proline	Lactic acid	1 : 2	4
DES-22	Pro : But	L-Proline	1,2,4-Butanetriol	1 : 2	9
DES-23	Pro : Ery	L-Proline	Erythritol	1 : 2	7
DES-24	Pro : Gly	L-Proline	Glycerol	1 : 2	5
DES-25	Pro : Mal	L-Proline	L-Malic acid	1 : 2	20
DES-26	Pro : Cit	L-Proline	Citric acid	1 : 2	23
DES-27	Pro : Xyl	L-Proline	Xylitol	1 : 2	23
DES-28	Pro : Glu	L-Proline	D-Glucose	1 : 2	14
DES-29	Lac : Glu	Lactic acid	D-Glucose	1 : 2	8
DES-30	Lac : Fru	Lactic acid	D-Fructose	1 : 2	16



Table 2 Elution procedure

Time (min)	Acetonitrile (%)	0.1% aqueous formic acid (%)
0	20	80
15	30	70
20	50	50
35	100	0

Office of Zunyi Medical University. The samples (voucher number: CRPV-20211015) were kept at room temperature in a storage cabinet at 2–605, School of Pharmacy.

Choline chloride and betaine were purchased from Shanghai Yuanye Biotechnology Co., Ltd. (Shanghai, China, purity >98%). NAR and HES standards were obtained from Shanghai Merrill Chemical Technology Co., Ltd. (Chengdu, China, purity >98%). Chromatographic grade methanol and acetonitrile were purchased from Shuangju Chemical Co., Ltd. (Zunyi, Guizhou). CRPV was collected from Zunyi, Guizhou, China, and stored in a cool (light-free place).

## 2.2. Preparation of DES

The DESs were synthesized *via* the “one-pot method” described in previous reports.<sup>19</sup> Generally, HBAs and HBDs with an appropriate amount of water were mixed and then stirred at 80 °C until a clear mixture was formed to prepare DESs. Then, the mixture was cooled at 25 °C and stored for subsequent experiments. The 30 DESs were prepared in this experiment (Table 1) and its appearances at room temperature were displayed in Fig. S1.†

## 2.3. Quantitative analysis of CRPV extract

### 2.3.1. Determination of NAR and HES content.

NAR and HES were identified as the main flavonoids in CRPV, and their

contents were determined *via* HPLC. The HPLC (Agilent 1260, Agilent, USA) was operated under the following conditions: a Phenomenex C<sub>18</sub> column (5 μm, 250 × 4.6 mm, Guangzhou FLM Scientific Instrument Co., Ltd., China); mobile phases of acetonitrile (A) and 0.1% formic acid in water (B); flow rate, column temperature, and injection volume of 1 mL min<sup>-1</sup>, 30 °C and 20 μL, respectively; and detection wavelength of 283 nm. The elution procedure is listed in Table 2. As shown in Fig. 1, the peaks in the CRPV extract were determined according to the retention times of the HES standard (12.63 min) and NAR standard (13.86 min).

**2.3.2. Preparation of CRPV extract.** The dried CRPV was pulverized with a multifunctional pulverizer (2000C, Dongguan Fangtai Electric Co., Ltd., China) and passed through an 80-mesh sieve. Then, 1 g of CRPV powder and 70 mL of DES (30% water content) were added to an Erlenmeyer flask and the mixture was stirred. The extraction was performed with an ultrasonic cleaning instrument (KQ5200, Kunshan Ultrasonic Instruments Co., Ltd., China) at a power of 240 W for 30 min. After the extraction, the mixture was separated *via* centrifugation (3000 rpm) for 10 min. The supernatant was collected, diluted with methanol (to 100 mL), and filtered through a 0.45 μm microporous membrane for HPLC analysis. The contents of HES and NAR were determined using formula (1).

$$\text{Flavonoid content (mg g}^{-1}\text{)} = \frac{M}{M_{\text{sample}}} \quad (1)$$

where  $M$  (mg) was the mass of HES and HES,  $M_{\text{sample}}$  (mg) was the mass of CRPV extract.

## 2.4. Optimal DES screening

A total of 30 DESs listed in Table 1 were prepared to extract the flavonoids from CRPV *via* the method described in Section 2.3.2. Taking the content of flavonoids as the response, the DES with high extraction efficiency was selected through HPLC analysis.

## 2.5. Single factors experiment

Setting the content of two flavonoids in CRPV extract as indexes, the effects of DES molar ratio, DES water content, extraction time, solid-liquid ratio, and ultrasonic power on the indexes were investigated. The detailed parameters of the single-factor experiment are listed below. The extraction of flavonoids from CRPV was performed at different operating conditions: DES molar ratios (2 : 1, 1 : 1, 1 : 2, 1 : 3, and 1 : 4, HBA : HBD), DES water contents (10, 20, 30, 40, and 50%), extraction times (10, 30, 50, 70, and 90 min), solid-liquid ratios (10 : 1, 30 : 1, 50 : 1, 70 : 1, and 90 : 1 mL g<sup>-1</sup>), and ultrasonic power (160, 200, 240, 280, and 320 W).

## 2.6. Optimization of extraction process

Combined with the results of single-factor experiments, the extraction conditions of flavonoids in CRPV were optimized using Box-Behnken experimental design (BBD) experiments. DES molar ratio (1 : 1, 1 : 2 and 1 : 3), DES water content (20, 30, and 40%), and extraction time (10, 30, and 50 min) were

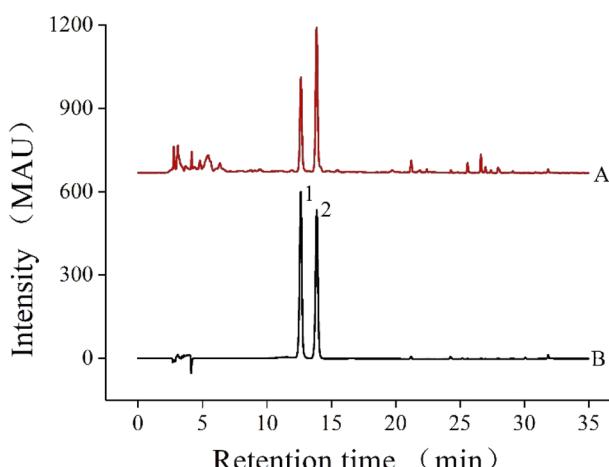


Fig. 1 The HPLC chromatograms of the extracts of CRPV and the standard mixture solution ((A) CRPV extract, (B) the standard mixture solution; (1) NAR, (2) HES).



Table 3 Factors and levels of Box–Behnken design

Factors	Experimental level		
	Low (−1)	Medium (0)	High (+1)
$X_1$ : molar ratio	1	2	3
$X_2$ : water content, %	20	30	40
$X_3$ : extraction time, min	10	30	50

selected as the main influencing factors and coded at three levels (−1, 0, 1) (Table 3). In addition, the extraction effect of NAR ( $Y_1$ ) and HES ( $Y_2$ ) were considered as the response value. The equation between factors and response is presented in eqn (2).

$$Y = \beta_0 + \beta_1 X_1 + \beta_2 X_2 + \beta_3 X_3 + \beta_{11} X_1^2 + \beta_{22} X_2^2 + \beta_{33} X_3^2 + \beta_{12} X_1 X_2 + \beta_{23} X_2 X_3 + \beta_{13} X_1 X_3 \quad (2)$$

where  $Y$  was the response value predicted by model,  $X$  was the independent variable,  $\beta_0, \beta_1, \beta_2, \beta_3$  were the linear coefficients,  $\beta_{11}, \beta_{22}, \beta_{33}$  were the quadratic coefficients,  $\beta_{12}, \beta_{13}, \beta_{23}$  were the interaction coefficients.

## 2.7. Recovery of flavonoids from DES extracts

Recovery of flavonoids from DES extracts was achieved using macroporous resin. The details were as follows: 5 mL of CRPV extracts were adsorbed with 1 g of different pretreated macroporous resin (AB-8, D101, NAK-9, HPD600 and SP825) on a constant temperature shaker (SHA-C, Changzhou Guoyu Instrument Manufacturing Co., Ltd., China) at a speed of 100 rpm and 25 °C for 24 h. Subsequently, the filtered macroporous resin was desorbed with 20 mL of methanol under the same conditions. Finally, the mixture was filtered, and the filtrate was collected to determine the content of flavonoids through HPLC and calculate the recovery yield.

## 2.8. Comparison between traditional extraction methods

**2.8.1. Maceration extraction (ME).** In line with the previous report,<sup>20</sup> 1.0 g of CRPV powder and 70 mL of DES were blended and then soaked for 12 h. Then, the extract was collected through centrifugation (3000 rpm, 10 min) and concentrated under reduced pressure. The contents of flavonoids in the extracts were detected *via* HPLC.

**2.8.2. Percolation extraction (PE).** According to Hashemi *et al.*,<sup>21</sup> 1.0 g of CRPV powder was wetted with optimum DES and kept under seal for 30 min. Then, the mixture was percolated with DES at a flow rate of two drops per minute. After the percolation, the extract was collected and concentrated under reduced pressure, and the flavonoid contents were determined using HPLC.

## 2.9. Identification of CRPV extract

According to previous studies,<sup>22</sup> the identification of flavonoids in CRPV extracts is usually carried out using UPLC-Q-TOF-MS. The specific analytical parameters were as follows: the system

was an Acquity UPLC system (Waters Corp., Milford, MA, USA); the column was BEH C<sub>18</sub> (100 × 2.1 mm, 1.7 μm); the mobile phase was 0.1% formic acid solution (A) and 0.1% formic acid acetonitrile (B), and the elution procedure is displayed in Table 4; and the flow rate, temperature, and injection volume were 0.4 mL min<sup>−1</sup>, 45 °C, and 10 μL, respectively.

Additionally, the Q-TOF was operated in the acquisition mode of the MS<sup>E</sup> model and ionization mode of ESI positive/negative. The MS data for 0–20 min was collected within the acquisition range of 5–1000 *m/z* at a scan rate of 0.2 s. Meanwhile, other pivotal parameters were set: the positive capillary voltage, negative capillary voltage, and cone voltage were defined at 2 kV, 2 kV, and 40 V, respectively, and the desolvation gas and temperature were set as 900 L h<sup>−1</sup> and 450 °C, respectively, and cone gas and source temperatures were 50 L h<sup>−1</sup> and 115 °C, respectively.

## 2.10. Antioxidant activity of flavonoid extract

**2.10.1. 1,1-Diphenyl-2-picrylhydrazyl (DPPH) radical scavenging activity.** In line with the report of Jiang *et al.*,<sup>23</sup> 2 mL of CRPV extract at different concentrations (0–1 mg mL<sup>−1</sup>) were reacted with 0.2 mmol L<sup>−1</sup> DPPH solution for 30 min. Then, the absorbance was recorded at 517 nm, and the DPPH radical scavenging rate was calculated.  $V_c$  (in the range of 0–1 mg mL<sup>−1</sup>) were used as a standard.

$$\text{Scavenging rate (\%)} = \frac{A_0 - A_x}{A_0} \times 100\% \quad (3)$$

in this formula,  $A_0$  represented the absorbance of the blank group and  $A_x$  referred the absorbance of the sample group.

**2.10.2. 2,2-Azino-bis-3-ethybenzothiazoline-6-sulfonic acid (ABTS) radical scavenging activity.** In line with the method presented by Lei *et al.*,<sup>24</sup> 7 mmol L<sup>−1</sup> of ABTS stock solution and 2.45 mmol L<sup>−1</sup> of potassium persulfate solution were mixed in equal volumes, and the mixture was shaken at room temperature for 16 h. The ABTS solution was diluted with anhydrous ethanol until the absorbance reached 0.7 at 734 nm. Then, 2 mL of different concentrations of CRPV extract and ABTS solution were reacted in the dark for 6 min and the ABTS radical clearance was detected at 734 nm and calculated using eqn (4).  $V_c$  (in the range of 0–1 mg mL<sup>−1</sup>) were used as a standard.

$$\text{Scavenging rate (\%)} = \frac{A_0 - A_x}{A_0} \times 100\% \quad (4)$$

Table 4 Elution procedure of UPLC-Q-TOF-MS

Time (min)	0.1% formic acid (A)	0.1% formic acid acetonitrile (B)
0	95	5
3	80	20
10	0	100
12	0	100
15	95	5
20	95	5



where  $A_0$  was the absorbance of the blank group,  $A_x$  indicated the absorbance of sample group.

**2.10.3. Hydroxyl radical scavenging activity.** The hydroxyl radical scavenging activity of CRPV extract was determined according to the method of Ke *et al.*<sup>25</sup> A volume of 2 mL of CRPV extract, 0.3 mL of 1.5 mmol L<sup>-1</sup> ferrous sulfate solution, 0.4 mL of 20 mmol L<sup>-1</sup> sodium salicylate solution, and 0.3 mL of 6 mmol L<sup>-1</sup> hydrogen peroxide were mixed and reacted for 30 min at 37 °C under water bath. Then, the absorbance was measured at 734 nm, and the hydroxyl radical scavenging rate was calculated using eqn (5).

$$\text{Scavenging rate (\%)} = \frac{1 - (A_1 - A_2)}{A_0} \times 100\% \quad (5)$$

where  $A_1$  was the absorbance of sample group,  $A_2$  referred to the control group,  $A_0$  represented the blank group.

## 2.11. Cytotoxic activity of flavonoids extract

The MTT method was adopted to examine the inhibitory effect of CRPV extract on CT-26, HCT-116, HT-29, and MCF-7 cells. The procedures were as follows: first, cancer cells at a logarithmic growth stage were spread in 96-well plates (cell density of  $5 \times 10^3$  cells per well, 100  $\mu$ L per well). The cells were grouped (control, 10, 100, 500, 1000, and 2000  $\mu$ g mL<sup>-1</sup> groups) when they reached approximately 50% growth. Then, CRPV extract was mixed with the cells and reacted for 48 h. A 20  $\mu$ L of MTT (5 mg mL<sup>-1</sup>) was added to each well and incubated for 4 h. After the incubation, the culture medium was removed, and 150  $\mu$ L of dimethyl sulfoxide (DMSO) was added to each well and shaken at low speed for 15 min. The absorbance was measured at

490 nm, and the cell inhibition rate was calculated through eqn (6).

$$\text{Cell inhibition rate (\%)} = \frac{\text{OD}_{\text{control group}} - \text{OD}_{\text{blank group}}}{\text{OD}_{\text{test group}} - \text{OD}_{\text{blank group}}} \times 100\% \quad (6)$$

where control group was the group of cells without adding CRPV extract, the blank group referred to the culture medium without cells and the test group is the group of cells with addition of the CRPV extract.

## 2.12. Statistical analysis

All of the experiments were repeated three times in this work, and the experimental data were expressed as mean  $\pm$  standard deviation ( $\pm$ SD). Design Expert 8.0.6 was used to optimize the extraction process, and SPSS18.0 statistical software was used for statistical analysis.  $P < 0.05$  indicated a statistical difference.

## 3. Results and discussion

### 3.1. Optimal DES screening

Various DESs possessed distinct physical properties, such as viscosity, density, which greatly impacted the extraction of natural products.<sup>26</sup> Among them, viscosity is the key factor affecting the extraction effect. To this end, we further measured the viscosity of 30 DESs. As shown in Table 1, the viscosity of DES-20 (pro : ure) was the lowest among the 30 DESs, indicating a positive performance of DES-20 in extraction process. To verify this theory, 30 DESs were used to extract flavonoids from CRPV. As shown in Fig. 2, DES-20 exhibited the strongest extraction efficiency of NAR and HES among the 30 DESs. The high

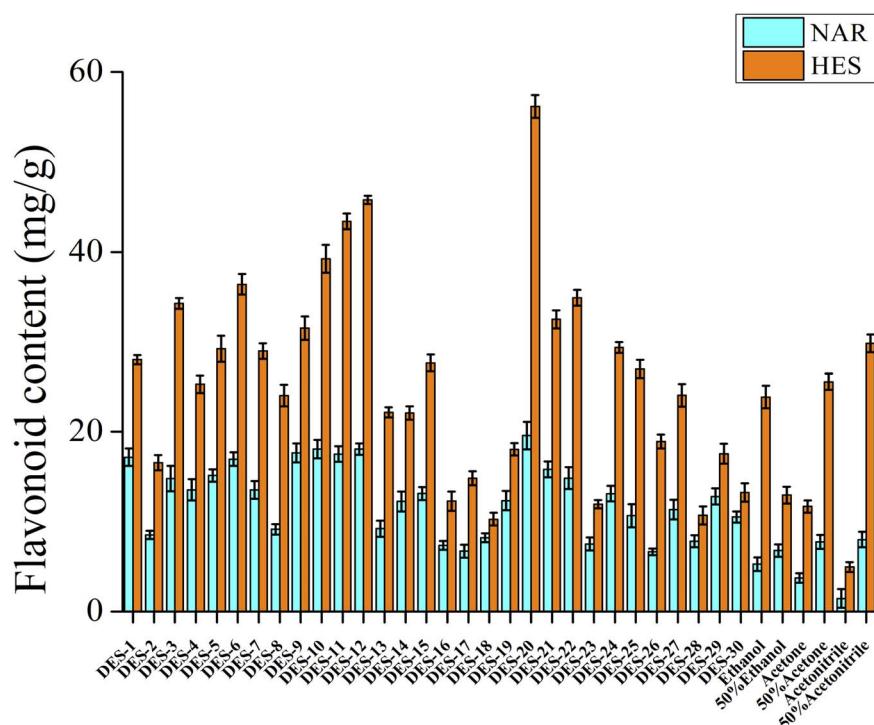


Fig. 2 Effects of different DESs on the contents of NAR and HES in CRPV.



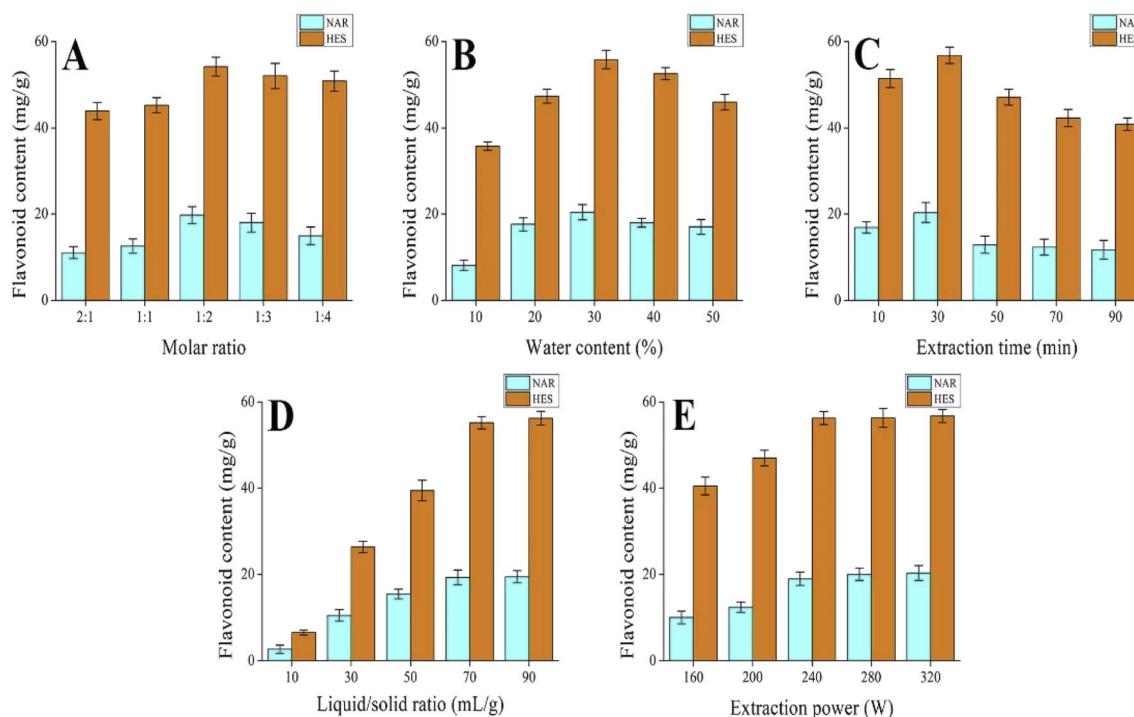


Fig. 3 The effect of extraction parameters on the extraction rate of NAR and HES in CRPV (A) molar ratio, (B) water content, (C) extraction time, (D) solid–liquid ratio and (E) extraction power.

efficiency of DES-20 can be attributed to its polarity, similar to that of flavonoids. Moreover, the DES-20 exhibited a stronger extraction effect compared to traditional solvents (ethanol, 50% aqueous ethanol, acetone, 50% aqueous acetone, acetonitrile, and 50% aqueous acetonitrile). Thus, DES-20 was selected as the preferred solvent for the extraction of flavonoids from CRPV.

### 3.2. Single-factor experiment

To improve the extraction effect of flavonoids from CRPV, the DES molar ratio, DES water content, extraction time, liquid–solid ratio, and ultrasonic power were further examined. As shown in Fig. 3A–E, the DES molar ratio, DES water content, extraction time, liquid–solid ratio, and ultrasonic power significantly influenced the extraction of NAR and HES. In addition, DES molar ratio (1 : 1, 1 : 2, and 1 : 3), DES water content (20%, 30%, and 40%), extraction time (10, 30, and 50 min), liquid–solid ratio (70 mL g<sup>−1</sup>), and ultrasonic power (240 W) were applied in the RSM.

### 3.3. Optimization of extraction process

Based on the single-factor experimental results, the interaction effects between molar ratio ( $X_1$ ), water content ( $X_2$ ), and extraction time ( $X_3$ ) were performed *via* BBD in Design-Expert software (Stat-Ease Inc., Minneapolis, MN, USA) and the results were displayed in Table 5. BBD experiment method was adopted as an excellent tool to optimize the extraction parameter of the target components, which explored the interaction effects between variables with fewer experiments.<sup>27</sup> Therefore,

the BBD experimental method has been adopted by a large number of scholars to optimize the extraction process of natural products, including flavonoids, polyphenols, polysaccharides, etc.

The ANOVA analysis result (see Tables S1–S2†) showed that the  $R^2$  and  $R_{\text{Adj}}^2$  of the two models were 0.9897 and 0.9920 and 0.9764 and 0.9818, respectively, indicating that these models were reasonable agreement and reliable, reflecting the relationship between response values and independent variables. Meanwhile, the high  $F$  values (74.56 and 96.92) and low  $P$  values

Table 5 Results of Box–Behnken experiment

Run	$X_1$	$X_2$ (%)	$X_3$ (min)	$Y_1$ (mg g <sup>−1</sup> )	$Y_2$ (mg g <sup>−1</sup> )
1	2	30	30	19	56
2	1	40	30	17	50
3	3	40	30	10	36
4	2	40	10	15	45
5	2	20	50	16	47
6	1	30	50	16	48
7	1	30	10	17	49
8	2	40	50	13	40
9	3	20	30	14	43
10	2	30	30	19	56
11	1	20	30	15	48
12	2	30	30	19	56
13	2	30	30	20	57
14	3	30	50	12	38
15	3	30	10	14	43
16	2	30	30	20	57
17	2	20	10	15	44



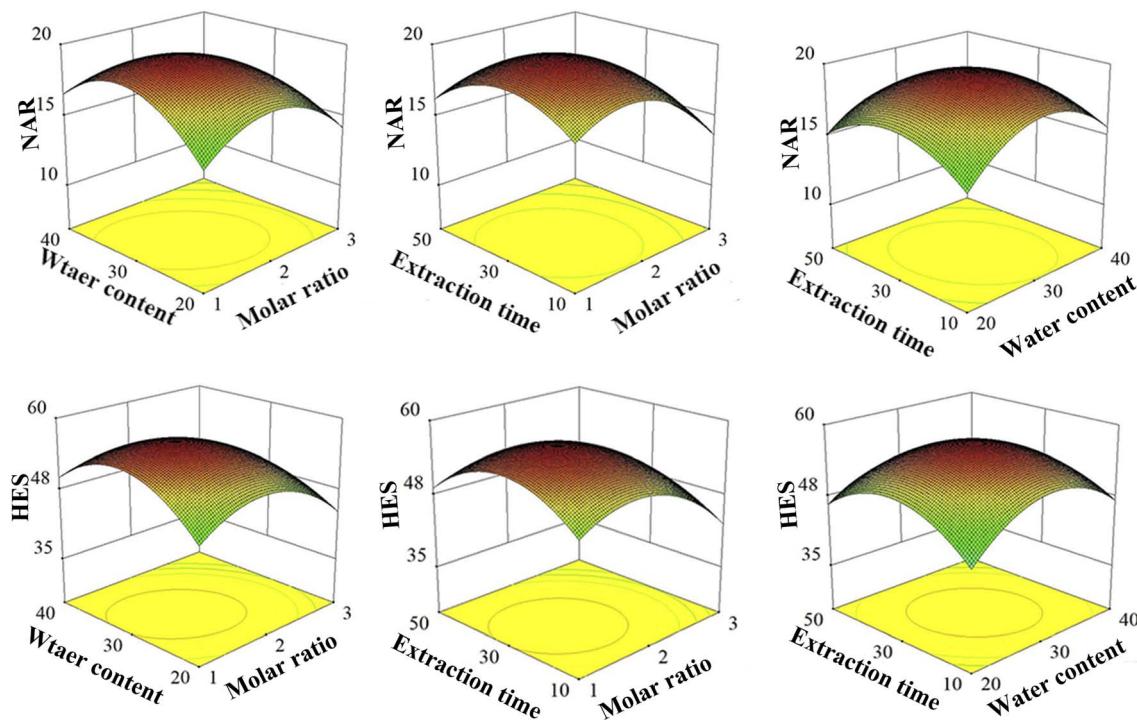


Fig. 4 3D response surface plots of NAR and HES.

Table 6 Optimal extraction parameters for NAR and HES

	Molar ratio	Water content (%)	Extraction time (min)	Predicted values (mg g <sup>-1</sup> )	Experimental values (mg g <sup>-1</sup> )
NAR	1 : 2	30	28	20	21 ± 2
HES	1 : 2	30	28	57	60 ± 2

(<0.05) indicated that the models were stable and significant. Moreover, all the three factors significantly influenced response values, with the *P* values less than 0.05. Additionally, the interaction factors  $X_1X_2$  and  $X_2X_3$  were significant ( $P < 0.05$ ), illustrating that they could significantly impact the DES extraction process. Furthermore, the *P* values of lack of fit were all greater than 0.05 (insignificant), illustrating the reliability of

the model. Moreover, multiple regression of the independent variable ( $X_1$ ,  $X_2$ , and  $X_3$ ) and the dependent variable ( $Y_1$  and  $Y_2$ ) were analyzed to obtain second-order equations.

$$Y_1 = 19.34 - 1.68X_1 - 0.53X_2 - 0.41X_3 - 1.35X_1X_2 - 0.24X_1X_3 - 0.62X_2X_3 - 2.56X_1^2 - 2.77X_2^2 - 2.08X_3^2$$

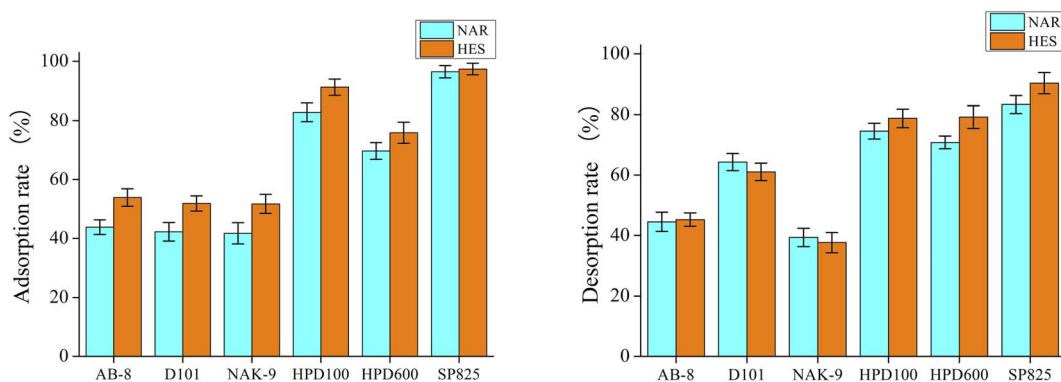


Fig. 5 Adsorption and desorption diagrams of macroporous resins.



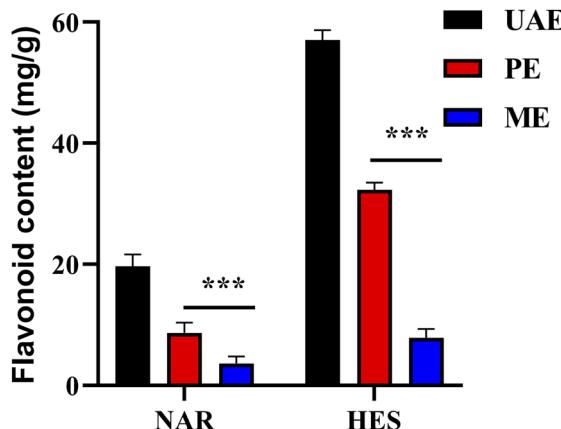


Fig. 6 Comparison of different extraction methods.

$$Y_2 = 56.46 - 4.15X_1 - 1.18X_2 - 1.06X_3 - 2.54X_1X_2 - 1.01X_1X_3 - 1.77X_2X_3 - 5.86X_1^2 - 6.48X_2^2 - 5.95X_3^2$$

To obtain optimal extraction effects of NAR and HES, 3D response surface plots were observed. As shown in Fig. 4, all of the 3D response surface curves were upper-phase convex with maximum values, which suggested that it was reliable for the level range. Based on the RSM model, the optimal extraction

process was selected with the predicted maximum extraction effects (Table 6). Upon further verification, the experimental and predicted values were in reasonable agreement, indicating that the models were reliable and significant.

### 3.4. Recovery of flavonoids from DES extracts

The recovery of target components from DES extracts is vital and urgent. Previously, some materials, such as macroporous resins, have been applied to recover target substances from DES extracts.<sup>28</sup> In this work, different macroporous resin (AB-8, D101, NAK-9, HPD600 and SP825) were explored for the recovery of flavonoids from DES extracts. As shown in Fig. 5, the adsorption and desorption of NAR and HES by the macroporous resin SP825 were the greatest among all the macroporous resins (higher than 80%). Thus, the macroporous resin SP825 was used as a material for the recovery of flavonoids from DES extracts and the recovery rates of NAR as well as HES reached  $88 \pm 3\%$  and  $86 \pm 4\%$ , respectively. Moreover, the recovered DES could extract NAR and HES for the second time, the extraction rates were  $18 \pm 1 \text{ mg g}^{-1}$  and  $53 \pm 2 \text{ mg g}^{-1}$ .

### 3.5. Comparison between traditional extraction methods

The traditional methods were compared to confirm the efficiency of flavonoid extractions using UAE. As shown in Fig. 6, the efficiency of NAR and HES extraction using UAE was higher

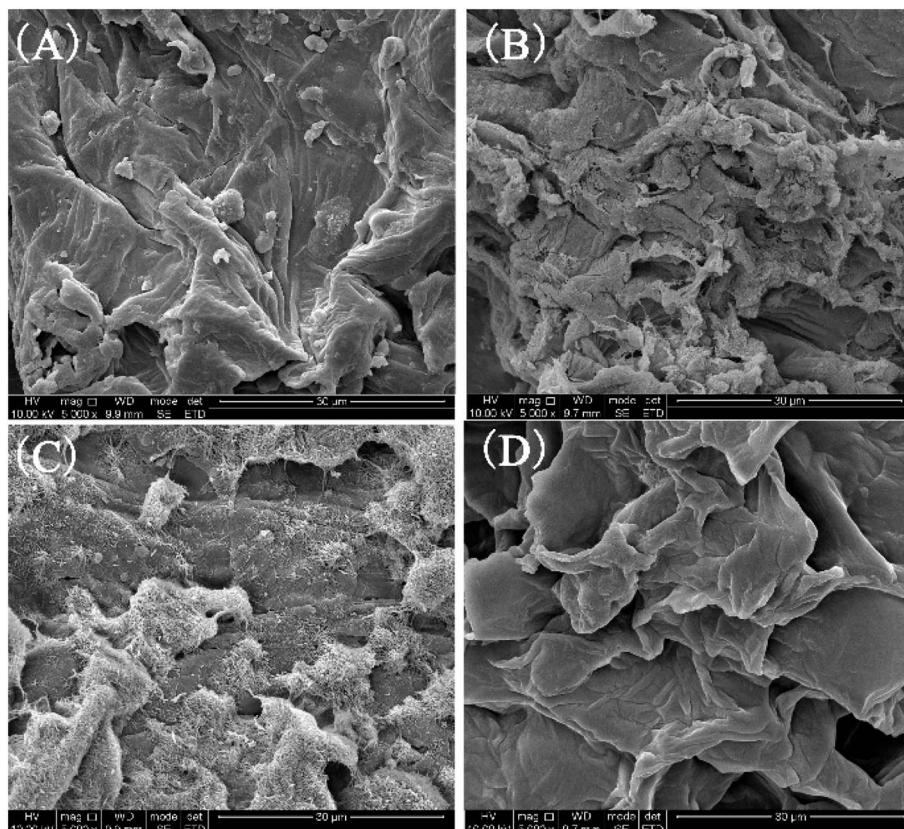


Fig. 7 SEM images of green skin ((A) was unextracted CRPV powder, (B) was the CRPV powder extracted by DES method, (C) was the ME method extracted CRPV powder, (D) was the powdered CRPV extracted by PE method).

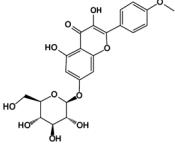
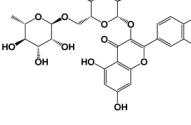
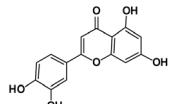
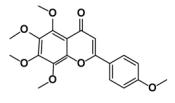
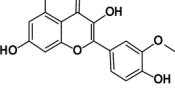
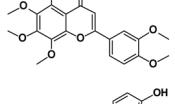
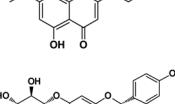
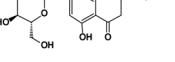


Table 7 The chemical constituents from CRPV

Retention time No. (min)	Analyte	Chemical structure	Neutral mass (Da)	Determined [M – H] <sup>–</sup> (m/z)	Product ions (m/z)	Mass error (mDa)
1 3.51	Quercetin		302.1533	301.1463	138.2652, 137.8510, 41.2310	–0.5
2 4.25	Hesperetin		302.0790	301.0714	150.2851, 148.3691, 120.8501	–0.2
3 4.55	Quercetin 3-O-neohesperidoside		610.1533	609.1463	444.8574, 268.6530, 152.2010, 136.9202, 108.9541	0.2
4 4.73	Naringenin		272.0684	271.0608	154.2134, 119.8540, 126.8412	–0.4
5 4.82	Myricetin		318.0375	317.0275	155.2341, 156.8541, 124.2845, 43.1201	–2.8
6 5.88	Narirutin		580.1792	579.1719	427.3851, 250.3612, 152.1025, 134.8463, 123.3601	–0.2
7 6.31	Quercetin 3-O-robinobioside		610.1533	609.1461	446.2150, 284.1236, 155.2140, 137.8450, 110.3201	1.7
8 6.47	Petunidin		448.1005	447.0935	133.0219, 151.2214, 285.0511	0.6
9 6.64	Apigenin		270.0528	269.0452	117.0256, 151.2310, 123.3265	–0.3
10 7.05	Isoliquiritin apioside		550.1686	595.167	417.1418, 297.0184, 255.0312, 254.0321, 135.9845, 119.7456	0.1
11 7.83	Liquiritigenin		256.0735	255.0659	135.0023, 119.0224, 107.8112	–0.4
12 8.39	Isoliquiritigenin		256.0735	255.0708	153.1214, 124.2826, 120.3210	–0.2
13 9.34	Hesperidin		610.1897	609.1819	450.3211, 288.1210, 152.1236, 149.2530, 124.0121	–0.6



Table 7 (Contd.)

Retention time No. (min)	Analyte	Chemical structure	Neutral mass (Da)	Determined [M – H] <sup>–</sup> (m/z)	Product ions (m/z)	Mass error (mDa)
14 9.69	Kaempferide-7-glucoside		462.1162	461.1091	399.2354, 154.9321, 136.3820, 108.5240	–0.1
15 10.32	Rutin		580.1792	579.1463	463.2744, 301.2213, 155.2181, 138.2540, 110.254	0
16 10.64	Luteolin		286.0477	285.0403	155.9123, 134.2201, 106.2134	–0.2
17 10.85	Tangeretin		372.9654	371.9854	243.9541, 133.8541, 215.5489	–0.3
18 11.67	Isorhamnetin		462.1162	461.0509	153.2018, 152.0841, 124.2816, 42.3261	–0.1
19 11.89	Nobiletin		402.3912	401.2586	244.1234, 216.3250, 148.1020	1.1
20 12.25	Genkwanin		284.0684	283.0666	168.4562, 119.8630, 141.2111	0.2
21 12.51	Naringenin 7-O- $\beta$ -D-glucopyranoside		434.1213	433.1143	271.8540, 151.2564, 124.6590, 120.9201	–1.8

than that of ME and PE. Meanwhile, the extraction time of UAE was significantly lesser than that of ME and PE. The results indicated that UAE was more efficient than the conventional method and possessed great potential value for the extraction of NAR and HES from CRPV.

### 3.6. Surface morphological characterization of CRPV

It is widely known that DES can accelerate the leaching of active substances through the disruption of plant cell walls.<sup>29</sup> Thus, to investigate the morphological alterations of the CRPV powder, the CRPV powder and the extracts using different methods (UAE, ME, and PE) were analyzed through SEM. Fig. 7A showed the SEM image of CRPV powder without extraction, where cell walls were tightly arranged. However, the CRPV powder after UAE extraction was filled with holes and debris, indicating that a large number of cell walls were damaged (see Fig. 7B). Nevertheless, the cell surface of CRPV powder after ME and PE extraction shrank and caused slight damage (Fig. 7C and D).

The above results implied that the mechanism of UAE extraction was to dissolve the active substances by destroying the plant cell wall.

### 3.7. UPLC-Q-TOF-MS

Among the well-known techniques, UPLC-Q-TOF-MS has been developed as a reliable technique to identify the target components of natural products.<sup>30</sup> In this study, the flavonoids in CRPV extracts were identified using the UPLC-Q-TOF-MS technique.

A total of 21 flavonoids, including two chalcones, six dihydroflavonoids, six flavonoids, and seven flavonols, were identified from CRPV extracts based on retention times, mass spectrometry excimer ion peaks, and SciFinder database, as presented in Table 7 and Fig. 8. In particular, chalcones comprised isoliquiritin apioside and isoliquiritigenin; dihydroflavonoids comprised hesperetin, naringenin, narirutin, liquiritigenin, hesperidin, and naringenin 7-O- $\beta$ -D-

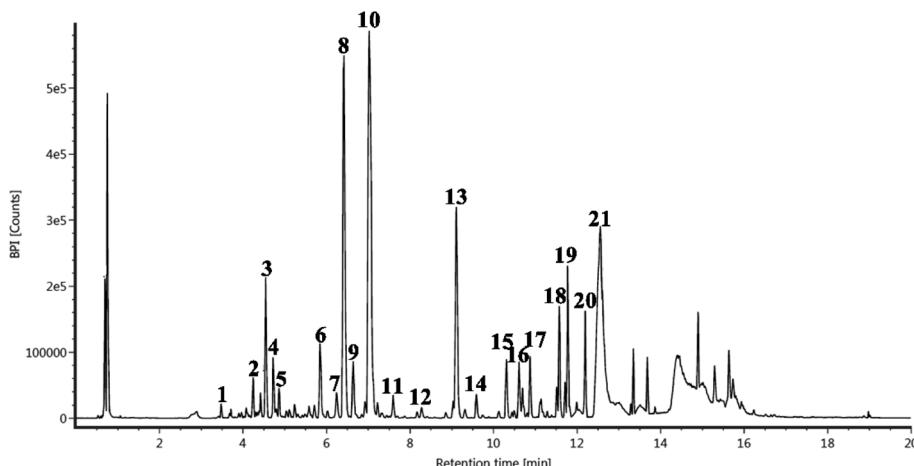


Fig. 8 UPLC-Q-TOF-MS extracted chromatograms of the compounds from CRPV.

Table 8 Method validation for NAR and HES

Analyte	Calibration curve	$R^2$	LOD ( $\mu\text{g mL}^{-1}$ )	LOQ ( $\mu\text{g mL}^{-1}$ )	Intraday precision ( $n = 6$ ) <sup>a</sup>	Interday precision ( $n = 9$ ) <sup>a</sup>	Stability	Repeatability ( $n = 6$ ) <sup>a</sup>	Recovery ( $n = 9$ ) <sup>a</sup>
NAR	$Y = 388.91X - 80.579$	0.9996	0.61	1.87	1.12	1.94	1.55	1.81	1.51
HES	$Y = 347.39X + 194.48$	0.9993	0.58	1.69	0.77	1.95	1.46	1.68	1.62

<sup>a</sup> Intraday precision, interday precision, stability, repeatability and recovery are expressed as the RSD (%) of peak area.

glucopyranoside; flavonoids comprised apigenin, luteolin, tangeretin, nobiletin, genkwanin; and flavonols comprised quercetin, quercetin 3-O-neohesperidoside, myricetin, quercetin 3-O-robinobioside, kaempferide-7-glucoside, rutin, and isorhamnetin. Notably, nine of these components were found for the first time in CRPV extracts, and they were quercetin 3-O-neohesperidoside, quercetin 3-O-robinobioside, petunidin, isoliquiritin apioside, liquiritigenin, isoliquiritigenin, kaempferide-7-glucoside, genkwanin, naringenin 7-O- $\beta$ -D-glucopyranoside, respectively.

### 3.8. Method validation

Fig. 1 shows that the main active components of the CRPV extract are NAR and HES. The methodological validation of NAR

and HES contributed to the quality determination of the CRPV extract. The standard equations for NAR and HES were  $Y = 388.91X - 80.579$  and  $Y = 347.39X + 194.48$ , respectively. As listed in Table 8, the NAR and HES showed favorable linearity ranging from 5–200  $\mu\text{g mL}^{-1}$  ( $R^2 > 0.999$ ).

The intraday precision was investigated by repeating the same sample test six times in one day, and interday precision was investigated by measuring the same sample for three consecutive days (three replicate injections per injection). The stability of the DES was analyzed by conducting the sample tests at 0, 2, 4, 8, 12, and 24 h. The reproducibility of the experiment was determined by repeating the preparation of six samples on the same day. Three concentration grades (80%, 100%, and 120%) of NAR and HES standards were spiked into nine

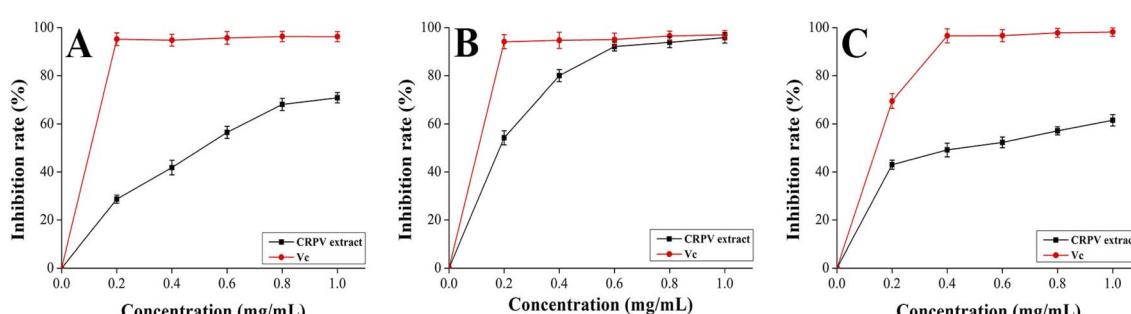


Fig. 9 Antioxidant activities of flavonoids extract ((A) DPPH radical scavenging activity, (B) ABTS radical scavenging activity, (C) hydroxyl radical scavenging activity).



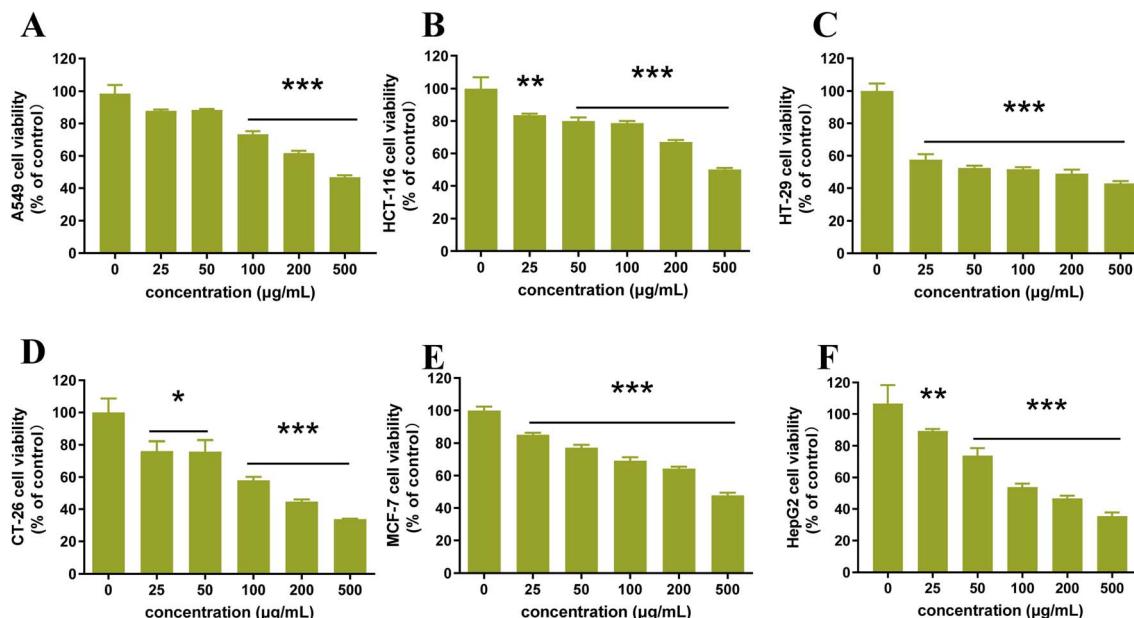


Fig. 10 Antitumor activity of flavonoids extract ((A) A549 cell, (B) HCT-116 cell, (C) HT-29 cell, (D) CT-26 cell, (E) MCF-7 cell, (F) HepG2 cell).

samples to evaluate their recovery rates. The result revealed that the method exhibited good precision ( $RSD < 2\%$ , Table 8). In addition, the stability result showed that the samples were stable at  $25\text{ }^{\circ}\text{C}$  for 12 h ( $RSD < 2\%$ ). Moreover, the RSD of reproducibility was less than two, which indicated that this method showed good reproducibility. Furthermore, the recovery rate, limits of detection, and limits of quantification of the method are presented in Table 8. The above results indicated that the method was fit for determining NAR and HES in CRPV extract.

### 3.9. Antioxidant activity

To explore natural antioxidants, three methods (DPPH, ABTS, and hydroxyl radical scavenging assay) were adopted to evaluate the antioxidant activity of flavonoid extracts. Fig. 9A–C showed that the flavonoids extract exhibited relatively strong DPPH, ABTS, and hydroxyl radical scavenging abilities at a concentration of  $0.2\text{--}1.0\text{ mg mL}^{-1}$ . At  $1.0\text{ mg mL}^{-1}$ , the DPPH, ABTS, and hydroxyl radical scavenging rates of the samples were maximum at  $71 \pm 2\%$ ,  $96 \pm 2\%$ , and  $62 \pm 2\%$ , respectively. Hence, the samples exhibited great scavenging rates on DPPH, ABTS, and hydroxyl radicals.

### 3.10. Cytotoxic activity

Cancer has become a critical disease impacting the health of humans across the world.<sup>31</sup> To find natural cytotoxic agents, the cytotoxic activity of flavonoid extracts was further investigated using the MTT method. The influence of flavonoid extract on the viability of the six cancer cells was presented in Fig. 10. As extract concentration increased from  $0\text{ }\mu\text{g mL}^{-1}$  to  $500\text{ }\mu\text{g mL}^{-1}$ , cancer cell viability decreased significantly for HCT-116, HT-29, CT-26, MCF-7 and HepG2 cells. In particular, for HT-29 cell line,

cell survival was below 50% when the extract concentration was  $100\text{ }\mu\text{g mL}^{-1}$ . However, it is not significant for the inhibition of A549 cancer cells when the extract concentration was increased from  $0\text{ }\mu\text{g mL}^{-1}$  to  $50\text{ }\mu\text{g mL}^{-1}$ , but while the sample concentration continued to increase, the cell survival rate decreased significantly. All the results showed that the flavonoid extract from CRPV has a good inhibitory effect on the six cancer cells selected in the experiment.

## 4. Conclusion

In this paper, a novel, green, and efficient method with DES was developed and successfully applied to extract flavonoids from CRPV. DES-20 (proline : urea) was screened as the preferred extraction solvent, and it was green, safe, and meets the development needs of green chemistry. The single-factor and RSM experiments were employed to obtain the optimal extraction conditions with the extraction yields of NAR and HES as index. Meanwhile, the UAE-DES method has various advantages including short time-consuming, solvent saving, environmental protection, *etc.* compared to conventional extraction methods (ME and PE). In addition, UAE-DES was able to disrupt the plant cell wall to enhance the solubilization of active ingredients, as revealed *via* SEM characterization. Moreover, 21 flavonoids were detected using the UPLC-Q-TOF-MS technique and nine of them were newly identified in CRPV extract. Furthermore, HPLC methods were used simultaneously to determine NAR and HES, and the macroporous resin SP825 was used to recover flavonoids from DES extracts. Finally, CRPV extracts exhibited positive antioxidant and cytotoxic activities. Hence, the DES as an extractant was developed and implemented successfully for the extraction of flavonoids in CRPV.



## Data availability

Research data are not shared.

## Conflicts of interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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