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Introduction of cadmium chloride additive to improve the performance and stability of perovskite solar cells

 Mustafa K. A. Mohammed,^a  ^{*aj} Majid S. Jabir,^b Haider G. Abdulzahraa,^c Safa H. Mohammed,^d Waleed Khaild Al-Azzawi,^e Duha S. Ahmed,^{*b} Sangeeta Singh,^{*f} Anjan Kumar,^{fg} S. Asaithambi^h and Masoud Shekargoftar 

With the increase in the importance of using green energy sources to meet the world's energy demands, attempts have been made to push perovskite solar cell technology toward industrialization all around the world. Improving the properties of perovskite materials as the heart of PSCs is one of the methods to fabricate favorable photovoltaic (PV) solar cells based on perovskites. Here, cadmium chloride (CdCl₂) was used as an additive source for the perovskite precursor to improve its PV properties. Results indicated CdCl₂ improves the perovskite growth and tailors its crystalline properties, suggesting boosted charge transport processes in the bulk and interfaces of the perovskite layer with electron–hole transport layers. Overall, by incorporation of 1.0% into the MAPbI₃ layer, a maximum power conversion efficiency of 15.28% was recorded for perovskite-based solar cells, higher than the 12.17% for the control devices. The developed method not only improved the PV performance of devices but also boosted the stability behavior of solar cells due to the passivated domain boundaries and enhanced hydrophobicity in the CdCl₂-based devices.

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1. Introduction

Due to their exceptional photoelectric performance and easy fabrication processes, hybrid perovskite solar cells (HPSCs) have attracted considerable attention as a promising candidate for new-generation photovoltaic (PV) devices.^{1–3} This could be related to the high absorption coefficient, high charge transfer phenomenon, and adjustable bandgap along with high charge diffusion length, which are all important requirements.^{4–6} The efficiency of HPSCs has substantially increased from a low of 3.8% (in 2009)⁷ to a verified 25.2% (in 2021),⁸ which has encouraged researchers in the HPSC's future development. Typically, according to the direction of the photocurrent, the

fundamental architecture of HPSCs can be categorized into conventional (n–i–p) and inverted (p–i–n) architectures.^{9,10} It's worth mentioning that the perovskite material is the heart of an HPSC device, and its overall quality determines how well it performs. The power conversion efficiency (PCE), hysteresis, and long-term stability of HPSCs are all influenced by the morphology, crystalline nature, and surface covering of the perovskite film.^{11–13} A high content of pinholes and grain boundaries (GBs) in the perovskite film, particularly, produces severe shorting and trap centers for non-radiative recombination, which is one of the primary sources of PCE degradation in HPSCs.^{14,15} Charge recombination aided by pinholes and GBs is also associated with abnormal hysteresis and instability concerns. As a result, enhancing the perovskite properties *via* passivating defects and optimizing perovskite quality has become a critical technique for decreasing recombination rates, removing unfavorable hysteresis, prolonging HPSC longevity, and increasing PCE.^{16,17}

To date, several ways of controlling perovskite morphology have been explored, involving manipulating the deposition method, additive engineering, interfacial engineering, and compositional engineering.^{18–22} Additive engineering has been shown to be effective in improving the quality of the perovskite layer since it is a more simple and reproducible approach. Some agents, including copper chloride, *N*,1-diiodoformamidine, potassium hexafluorophosphate, formamidine acetate salt, *etc.*, can form an intermediate phase with perovskite precursors,

^aRadiology Techniques Department, Dijlah University College, Al-Masafi Street, Baghdad 00964, Iraq. E-mail: mustafa_kareem97@yahoo.com; Tel: +964-7719047121

^bApplied Science Department, University of Technology, Iraq

^cDepartment of Prosthodontic, Dijlah University College, Al-Masafi Street, Baghdad, Iraq

^dRadiological Techniques Department, Al-Mustaqbal University College, Babylon, Iraq

^eDepartment of Medical Instruments Engineering Techniques, Al-Farahidi University, Baghdad, Iraq

^fMicroelectronics Lab, National Institute of Technology, Patna 800005, India

^gVLSI Research Lab, GLA University, Mathura-281406, India

^hDepartment of Physics, Alagappa University, Karaikudi, 630 003, Tamil Nadu, India

ⁱLaboratory for Biomaterials and Bioengineering (CRC-I), Department of Min-Met-Materials Engineering, Laval University, Quebec City, QC G1V0A6, Canada

^jUniversity of Warith Al-Anbiyaa, Karbala, Iraq


which could regulate the dynamic process of crystal formation and thereby minimize GBs.^{23–27} The incorporation of cadmium (Cd^{2+}) additive by Suneth *et al.* was a fundamental advance in controlling methylammonium lead iodide (MAPbI_3) crystallinity using a two-step deposition method. With this treatment, the efficiency of MAPbI_3 -based HPSCs was significantly boosted from 7.1 to 13.8%.²⁸ Yong *et al.* reported the addition of Cd^{2+} into organic–inorganic cation-based $\text{Cs}_x\text{FA}_{1-x}\text{PbI}_3$ lattice to enhance the crystallinity and the charge carrier lifetime of perovskite film. An efficiency up to 20.59% was achieved for HPSC with a structure of ITO glass/ $\text{SnO}_2/\text{Cs}_x\text{FA}_{1-x}\text{PbI}_3/\text{spiro-OMeTAD}/\text{Au}$.²⁹ In another study, a suppression of atomic vacancies *via* strain relaxation of CsMAFA triple-cation perovskite film was reported by Makhsud *et al.* This suppression was achieved *via* incorporation of Cd isovalent small ions, which let to boost the stability of HPSCs in ambient air.³⁰ Recently, Cd^{2+} introduced into all-inorganic perovskite for carbon-based HPSCs with a framework of $\text{FTO}/\text{SnO}_2/\text{CsPbIBr}_2/\text{carbon}$. They demonstrated that Cd ions improves the crystalline nature and morphological merits of perovskite film, resulting in better optical absorption and carrier transfer in perovskite with less recombination rates. Also, in compared to the HPSCs without Cd^{2+} doping (4.36%), the improved HPSCs attained the highest PCE of 6.79%.³¹ Yi *et al.* introduced ammonium benzenesulfonate as a passivator into the MAPbI_3 solution to assist the film crystallization and mitigate trap density. A champion PCE of 20.6% is obtained for the modified HPSCs.³² Hua *et al.* reported the incorporation of *N*,1-diiodoformamidine modifier into FAMAPbI_3 to improve the quality of perovskite with bigger grain size and fewer defects, hence improving the PCE from 19.07% to 21.22% as well as their light and thermal stabilities.²⁴ Recently, a dual additive approach for enhancing the performance of HPSCs was reported by Yang *et al.*³³ The highest PCE of 23.2% is yielded *via* a mixed modifier of MACl/CsCl with a 2 : 1 ratio, which is better than that for a single modifier of MACl or CsCl . Despite their high performance, the majority of HPSCs fabricated with organic hole transport layers (HTLs) such as spiro-OMeTAD have low stability and are excessively expensive, making them unsuitable for commercialization. Therefore, further work is needed to find a cheaper alternative to spiro-OMeTAD while simultaneously utilizing the additive engineering approach.

Herein, we take advantage of using an efficient additive of cadmium chloride (CdCl_2) alongside with using a copper phthalocyanine (CuPc) semiconductor as a model HTL for regular (n–i–p) HPSCs. CuPc is known to possess high hole mobility, excellent interfacial bonding features, and superb stability while also being relatively inexpensive, which is why it has been used as a p-type semiconductor in different optoelectronic devices.^{34–36} We investigate the impact of employing low CdCl_2 concentrations and high MAI contents during MAPbI_3 development in the anti-solvent deposition procedure. In addition to better crystalline phase and increased grain size, steady-state photoluminescence (PL) tests revealed a significant reduction in charge recombination process. The best-performing CdCl_2 -based HPSCs using a CuPc HTL yielded a PCE of 15.28% with a V_{OC} of 1.0 V and a FF of 73.2%, recorded

under reverse scanning. The modified devices demonstrated enhanced long-term stability as compared to unmodified devices fabricated with mesoscopic structures.

2. Experimental part

2.1 Materials

All solvents, titanium isopropoxide (TTIP, 97%), cadmium chloride (CdCl_2 , 99.99%), lithium bis(trifluoromethylsulfonyle) imide salt (LiTFSI, 98%), and copper phthalocyanine (CuPc , 99%) were provided from Sigma-Aldrich. Lead iodide (PbI_2 , 99.99%), methylammonium iodide (MAI, 99.5%), and 2,2',7,7'-tetrakis(*N,N*-di-*p*-methoxyphenylamino)-9,9'-spirobifluorene (spiro-OMeTAD, 99.8%) were prepared from LumTec. TiO_2 paste was prepared from Dyesol.

2.2 Solution preparation

922 mg of PbI_2 were dissolved in 1.27 mL of *N,N*-dimethylformamide (DMF, 99.8%) and 0.142 mL dimethyl sulfoxide (DMSO, 99.8%). The resultant product was mixed at 80 °C for 45 min. After cooling down the PbI_2 solution, 318 mg of MAI was inserted into it and stirred for 15 min to prepare MAPbI_3 . For incorporation of CdCl_2 into the perovskite layer, different amounts of CdCl_2 in molar ratios to MAI are added into the prepared perovskite solutions, followed by further stirring at RT for 60 min. 500 mg of TiO_2 paste is dissolved in 3.0 g ethanol (EtOH, 99.8%) and is stirred at RT for 24 h to attain mesoporous TiO_2 (mp- TiO_2) solution. 700 mg of TTIP is added into 10 mL of 2-propanol (IPA, 99.9%) containing 70 μL of 2 M HCl, followed by stirring in an ice bath for 30 min to prepare compact TiO_2 (c- TiO_2) precursor. The spiro-OMeTAD precursor was prepared as reported in the literature of (ref. 37) by dissolving 36.2 mg of spiro-OMeTAD (99%) in 500 μL chlorobenzene (CB, 99.9%) doped with LiTFSI, and 4-*tert*-butylpyridine (98%).

2.3 HPSC fabrication

Transparent conductive F-doped SnO_2 (FTO, Pilkington) glasses with a sheet resistance of 15 $\Omega \text{ sq}^{-1}$ are ultrasonically cleaned with distilled water, soap, distilled water, acetone, and IPA successively for 10 min. 45 μL of the c- TiO_2 solution is spread in each FTO piece and spin-coated at 4000 rpm, followed by annealing at 500 °C for 30 min to fabricate a c- TiO_2 layer. 60 μL of mp- TiO_2 precursor is spin-coated on each c- TiO_2 layer at 4000 rpm for 30 s, followed by baking at 450 °C for 60 min. To fabricate perovskite layer, 60 μL of perovskite precursors with or without CdCl_2 additive are spread on each substrate and spin coated at 1000 rpm for 8 s and then accelerate up to 6000 for 30 s. During the faster step, 200 μL of CB was pipetted on perovskite layer to assist crystal growth of perovskite layer. Perovskite layers were annealed for 30 min at 100 °C. The CuPc hole transport layer (HTL) with thickness of 30 nm was thermally evaporated on the pre-formed perovskite films in a vacuum of 10^{-3} Pa with deposition rate of 1 $\text{\AA} \text{ s}^{-1}$. To fabricate spiro-OMeTAD HTL, its precursor was spin-coated at 5000 rpm for 30 s on top the perovskites. Finally, a 100 nm gold thin-layer was thermally deposited the HTLs.



2.4 Characterizations

A Mira3, TESCAN field emission scanning electron microscopy (FESEM) was used to monitor the effect of the CdCl₂ additive on the micro-morphology of perovskite layers. A Bruker, D8 Advance X-ray diffractometer was used to collect XRD pattern of perovskite films. A LAMBDA 1050 spectrophotometer was used to study the light-harvesting ability of perovskite layers. A HORIBA Fluorolog-III spectrometer with an exciting wavelength at 405 nm was used to record steady-state photoluminescence (PL) responses of perovskite layers. Photovoltaic performance of MAPbI₃-based solar cells with an active area of 2 mm × 3 mm was characterized using a Keithley Model 2401 under a calibrated AM 1.5 light simulator to record current density–voltage curves. A calibrated Newport IPCE equipment was employed to record incident photon-to-electron conversion efficiency (IPCE) of solar cells.

3. Results and discussion

Fig. 1 presents the FESEM micrographs of the perovskites prepared with and without the CdCl₂ additives in the MAI precursors that were utilized to complete the MAPbI₃ conversion. The influence of the CdCl₂ incorporation is clear in that good uniformity and large grains are observed in the CdCl₂-added perovskites immediately after heat treatment (Fig. 1b and c), while film deposited without solvent additive shows low uniformity and small grains with obvious pinholes at the grain edges (Fig. 1a). As illustrated in Fig. 1e–h, the addition of small amounts of CdCl₂ enlarged the average grain size fivefold, from 280 nm to 590 nm. This is due to the inclusion of CdCl₂ into the MAI solution, which produces Cl ions, resulting in the formation of an intermediate phase that acts as a seed and enhances the development of bigger MAPbI₃ crystals.³⁸ With a further increase of CdCl₂ amount to 1.5%, the surface of the perovskite layer was identical to the 1% CdCl₂-added film but revealed small particles with obvious grain boundaries (GBs). From the FESEM results, we can argue that the grain size of the CdCl₂-modified films was not greatly enlarged. Nevertheless, smaller grains have combined into bigger ones, the gaps between grains have been passivated, and the film has become denser and more compact. This could help improve HPSC performance by reducing nonradiative charge recombination.³⁹

The performance of HPSCs is directly dependent on the crystal phase and purity of the MAPbI₃ layers.⁴⁰ Fig. 2a demonstrates the X-ray patterns of MAPbI₃ perovskites without and with 1% CdCl₂ inclusion into the MAI solution. As demonstrated, multiple main XRD peaks at 13.82°, 20.10°, 28.18°, and 31.59° were indexed as (110), (112), (220), and (310) planes of the tetragonal MAPbI₃ phase, respectively.²¹ The sharp and intensive peaks in the XRD plots of the MAPbI₃ perovskites implied that the films were well crystallized. Furthermore, the intensities of the prominent peaks all rose when the CdCl₂ solvent was incorporated, implying that a superior crystalline nature and larger crystallite size for perovskite film were developed as compared to a film prepared without CdCl₂.⁴¹ In MAPbI₃ samples prepared without and with CdCl₂, the full

width at half maximum intensity of the (110) peak was 0.108 and 0.097, respectively. According to the Scherrer equation,⁴² the average crystallite sizes of the corresponding samples were 73.46 and 81.21 nm, respectively. These findings suggest that the moderate amount of CdCl₂ in perovskite could increase both crystallinity and grain size. This result is also in line with the findings of the FESEM images.

The low XRD signal at 12.4° is indexed as the cubic PbI₂, because of the slight degradation of the MAPbI₃ to PbI₂ and/or incomplete crystallization of perovskite.²³ The presence of PbI₂ in the perovskite would be detrimental to photo-induced carrier transfer, lowering film stability and increasing the hysteresis behavior of HPSCs. Interestingly, it was apparently seen that the typical peak of PbI₂ was completely removed with CdCl₂ treatment. Fig. 2b shows the XRD patterns of perovskite samples zoomed to the major (110) peaks. The (110) peak was altered to higher values of 2θ with the addition of CdCl₂. This is most likely owing to a variation in the lattice parameters caused by the replacement of smaller Cd²⁺ species for Pb²⁺ sites,²⁸ therefore leading to the shrinkage of the MAPbI₃ lattice. The reduced lead amounts in perovskite could be beneficial for increasing the tolerance factor, improves the stability of the MAPbI₃ structure.³¹

We used UV-vis absorption and PL spectroscopy to explore the influence of the CdCl₂ solvent on light absorption and the mechanism of photoexcited carriers transport in perovskite film deposited under the same conditions unless the variation in CdCl₂ concentrations in MAI solution. The UV-vis spectra of MAPbI₃ perovskite layers with various concentrations of CdCl₂ additive are shown in Fig. 3a. The absorbance plots measured for MAPbI₃ layers with and without CdCl₂ treatments exhibited remarkably identical absorption profiles (edge-absorption position at 777 nm) thanks to the unique energy bandgap (E_g) absorption of MAPbI₃ as a sunlight harvester.⁴³ As we can see, all samples fabricated by the spin-coating clearly show strong absorbance over the full light absorption below 700 nm. When the concentration of CdCl₂ solution is raised to 1%, the absorption performance of the associated MAPbI₃ layer slightly increases within the visible range. This could be due to the increased grain size and improved crystallinity, while the film thickness remains similar with the insertion of CdCl₂.⁴⁴ However, when more CdCl₂ (1%) was introduced, the perovskite's light absorption capacity dropped because the presence of unwanted defects and GBs disrupted the film's compact structure. Fig. 3b describes the Tauc plots for all samples prepared with and without CdCl₂ additives. The value of E_g can be evaluated by extrapolating the variation between photo energy ($h\nu$) and $(\alpha h\nu)^2$.⁴⁵ The values of E_g were found to be 1.59 eV, which is similar to the values reported in the literature.^{13,40}

Fig. 3c shows the steady-state PL measurements of perovskite films spin-coated on glass substrates without HTL and ETL films in order to reduce charge extraction at the interfaces and enable the inherent features of the MAPbI₃ film to govern the recombination dynamics. All the steady-state PL plots of perovskites reveal a typical and strong luminescence peak at locations identical to the absorbance edges detected in UV-vis



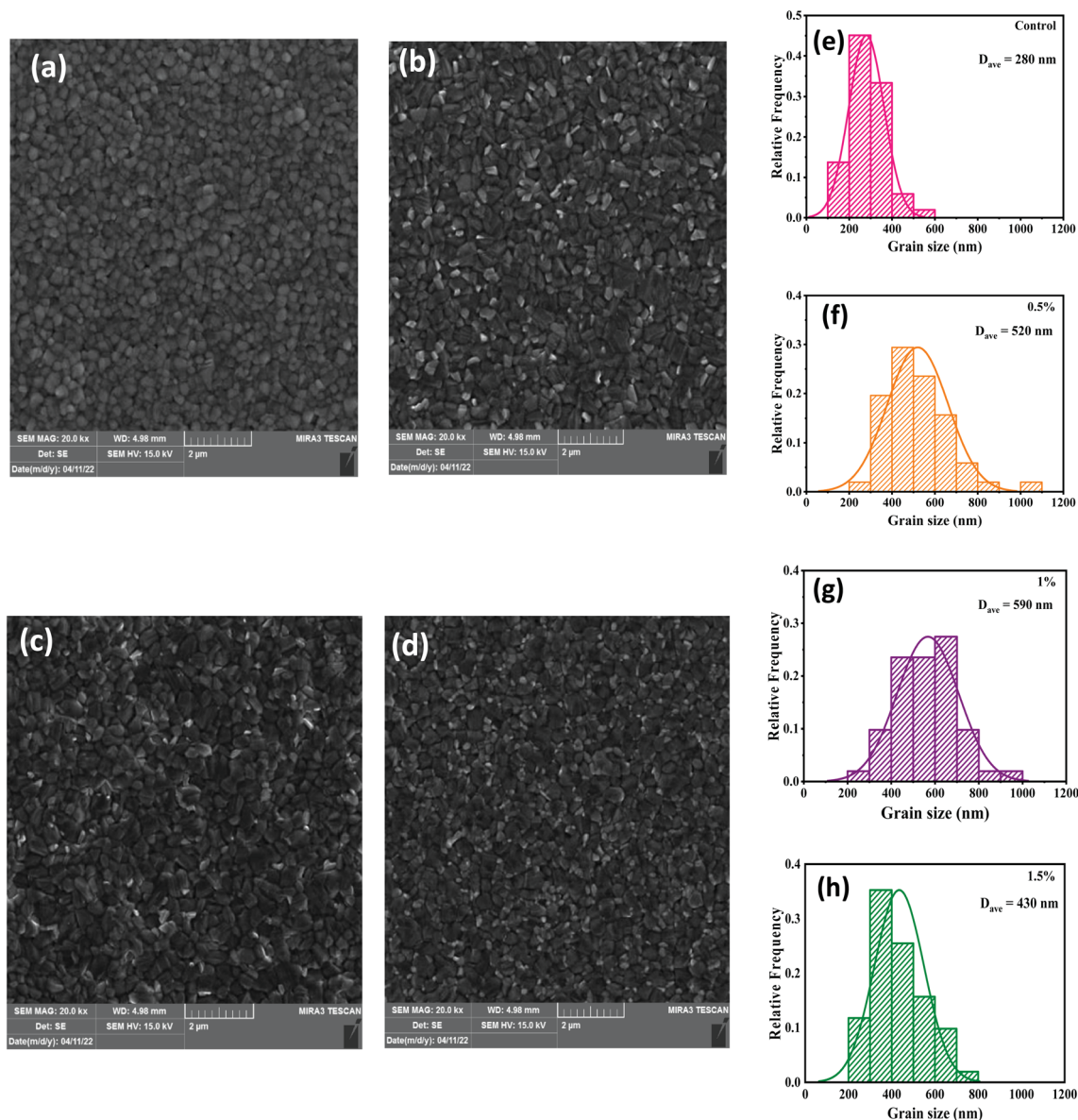


Fig. 1 The top-view FESEM visualization of perovskite layers with (a) 0%, (b) 0.5%, (c) 1%, and (d) 1.5% of CdCl₂ in perovskite precursors. Substrate for all layers was FTO/c-TiO₂/mp-TiO₂. (e–h) Particle size distributions of perovskite films evaluated by ImageJ software.

measurements. The increased PL intensity and reduced line-width of CdCl₂-treated films in comparison to the control layer indicate that the incorporation of solvent additive can significantly inhibit trap-mediated, non-radiative recombination in those perovskites.⁴⁶ This could be due to the CdCl₂ additive treatments improving the crystallinity of the developed perovskites. The enhanced crystalline structure decreased defects in the entire MAPbI₃, resulting in fewer recombination traps in the film. This inhibited the non-radiation recombination pathway, making it more suitable for optoelectronics manufacturing. Conversely, as the CdCl₂ concentration was increased to 1.5%, the PL intensity of the perovskite decreased, indicating an increase in the density of microstructural defects.

We fabricated regular (n-i-p) HPSC devices using the architecture of glass/FTO/c-TiO₂/mp-TiO₂/MAPbI₃/CuPc/Au, as

illustrated in Fig. 4a. The ETL films are fabricated based on our previously prepared mp-TiO₂ *via* the spin-coating method, while HTLs utilize a p-type inorganic film named CuPc. The mixed solutions of MAI : PbI₂ were used to form perovskite film with an anti-solvent assistance approach. An additive engineering method was employed in our perovskite films to improve the HPSC performance. The CdCl₂ additive was selected as a modifier to assist perovskite crystallization during the deposition process. We have investigated the PV parameters of our designed HPSCs with reverse *J-V* sweep being performed under the AM 1.5 sunlight simulator. Fig. 4b demonstrates the best-performing *J-V* plots of the HPSCs fabricated with and without additive treatments, and the summarized cell parameters are displayed in Table 1. The performance of HPSCs with and without CdCl₂ additives varies significantly. Without CdCl₂,



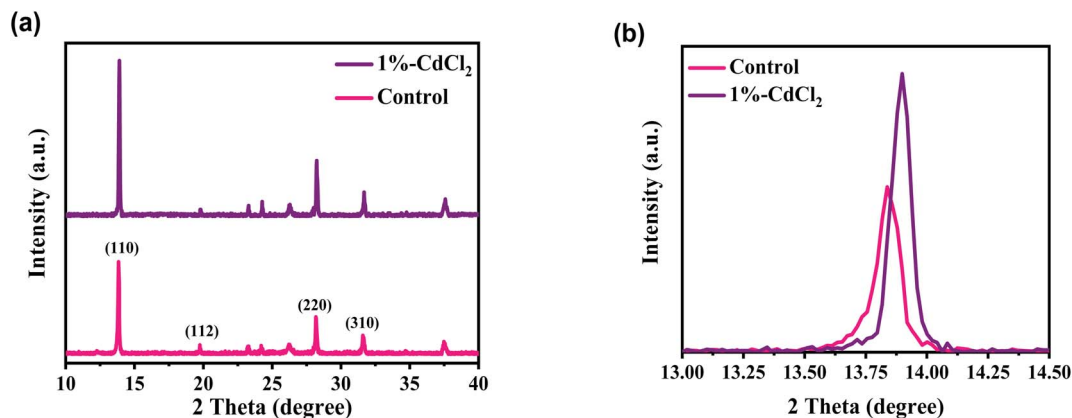


Fig. 2 (a) XRD patterns of perovskite layers without (control) and with 1% of CdCl_2 deposited on FTO/c- TiO_2 /mp- TiO_2 substrate. (b) Zoomed XRD patterns around $2\theta = 13.9^\circ$.

the HPSCs show low PV performances with a J_{SC} of 19.78 mA cm^{-2} , V_{OC} of 0.962 V, FF of 65.70% and, a champion PCE of 12.17%. The HPSC has an enhanced PCE of 15.28% with a high J_{SC} of 20.87 mA cm^{-2} , a V_{OC} of 1.00 V, and a good FF of 73% after adding 1% CdCl_2 . When the amount of CdCl_2 is increased to 1.5%, the PCE of the HPSC has a notable reduction to 14.10% with reduced J_{SC} and practically unchanged V_{OC} . As previously mentioned, an adequate dose of CdCl_2 modifier can assist in

removing PbI_2 residual without forming a non-perovskite phase, resulting in a compact high-quality layer with large crystals, which can greatly reduce interior recombination and enhance HPSC performance. The integrated J_{SC} values of the unmodified and 1% CdCl_2 -based HPSCs are 19.08 mA cm^{-2} and 20.14 mA cm^{-2} , respectively, as shown in Fig. 4c, which match with the J_{SC} values determined from the detailed J - V scan

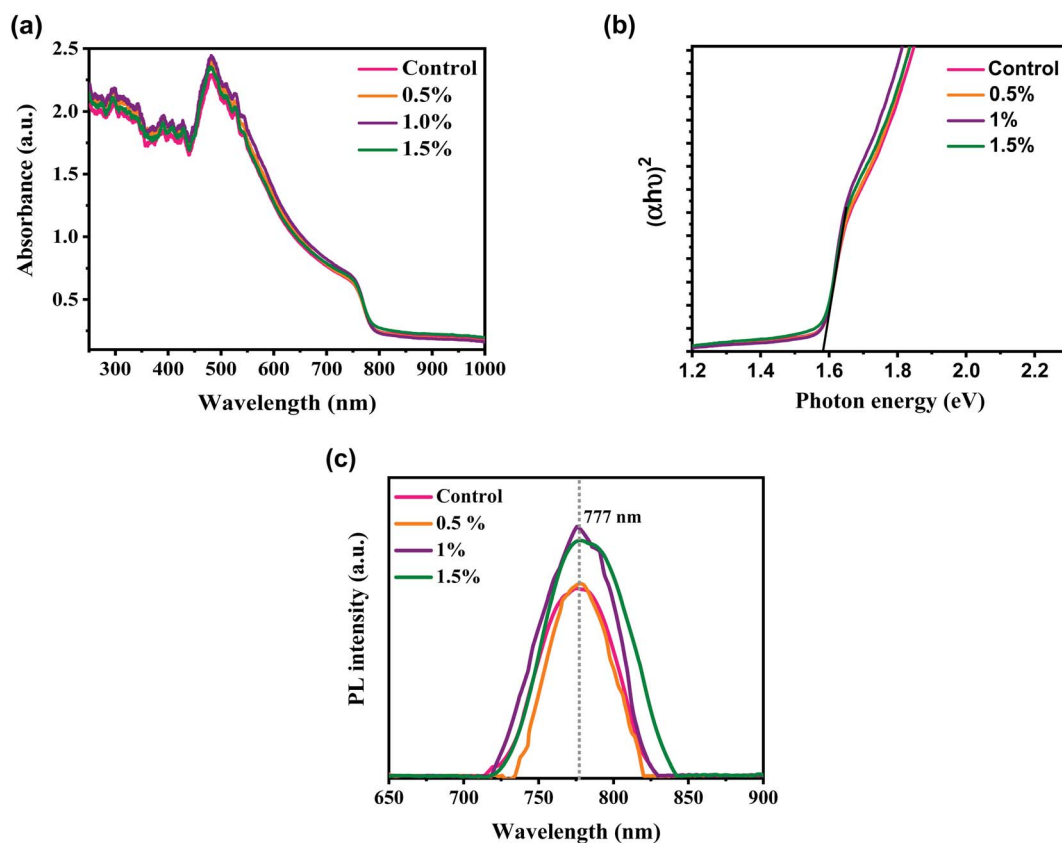


Fig. 3 (a) UV-vis absorption spectra, (b) Tauc plots, and (c) photoluminescence spectra of control and modified perovskite layers with different amounts of CdCl_2 , deposited on glass substrates.

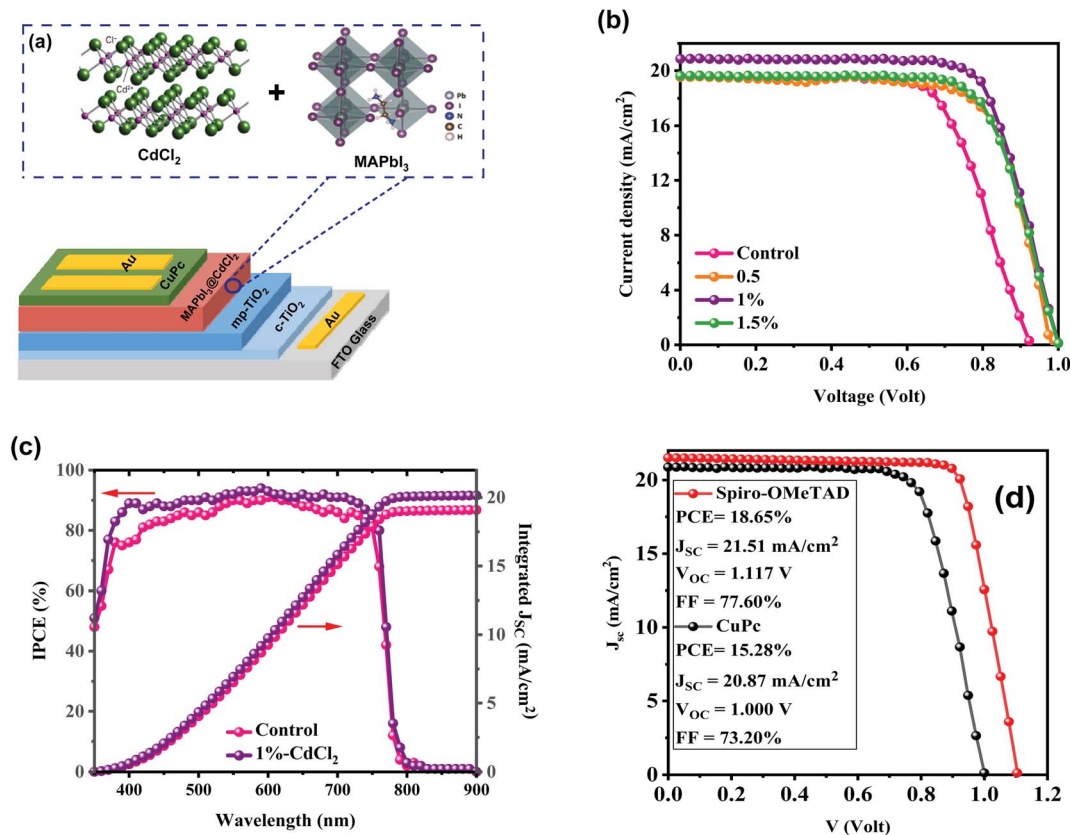


Fig. 4 (a) Schematic diagram of the device architecture of CdCl₂-assisted deposition process for MAPbI₃ film. (b) *J*-*V* curves of the best-performing HPSCs with different amounts of CdCl₂ additive. (c) Recorded IPCE responses of the control and modified device with 1% of CdCl₂ and their integrated *J*_{SC} from the IPCE. (d) *J*-*V* curves of perovskite solar cells with copper phthalocyanine (CuPc) and 2,2',7,7'-tetrakis(*N,N*-di-*p*-methoxyphenylamino)-9,9'-spirobifluorene (spiro-OMeTAD) as hole transport layers.

characteristics. The IPCE measurements confirm the accuracy of our HPSC efficiency characterizations.

To compare the performance of CuPc-based PSCs with conventional spiro-OMeTAD PSCs, we fabricated PSCs with spiro-OMeTAD HTL. A champion PCE of 18.65% with a *J*_{SC} of 21.51 mA cm⁻², *V*_{OC} of 1.117 mV, and FF of 77.60% was obtained (Fig. 4d). It notes that CuPc can be considered a low-cost HTL in perovskite-based solar cells. The CuPc HTL recorded a *J*_{SC} near to spiro-OMeTAD devices, with lower *V*_{OC} and FF than spiro-OMeTAD HTL one. Finding refers that recording higher PCE with CuPc HTL is reachable by more engineering at the

interface of perovskite/CuPc, which will be targeted in our future experiments.

Fig. 5 depicts the distribution of PV parameters depending on the concentration of MAI to CdCl₂, where the ratio of MAI : CdCl₂ is changed from 0% to 1.5%. As shown in Fig. 5a, adjusting the ratio has a notable effect on the mean *J*_{SC} of nearly 20 mA cm⁻². *V*_{OC} improves slightly with increasing the CdCl₂ concentration (Fig. 5b). The average *V*_{OC} value of 0.946 V for the control device is increased to 0.964, 0.994, and 0.985 V as the ratio varies from 0% to 1.5%. Mean FF is also enhanced as the concentration of CdCl₂ increases, reaching 72.29% at 1% and

Table 1 Photovoltaic properties of HPSCs with various concentrations of CdCl₂ additives in MAI precursor

CdCl ₂ amount		<i>V</i> _{OC} ^a (V)	<i>J</i> _{SC} ^b (mA cm ⁻²)	FF ^c (%)	PCE (%)
Control	Average	0.946 ± 0.018	19.16 ± 0.48	63.26 ± 2.75	11.44 ± 0.63
	Best	0.962	19.78	65.70	12.17
0.5%	Average	0.964 ± 0.019	19.48 ± 0.55	69.15 ± 1.97	12.98 ± 0.52
	Best	0.987	19.54	71.70	13.82
1.0%	Average	0.994 ± 0.007	20.33 ± 0.41	72.29 ± 1.34	14.61 ± 0.43
	Best	1.000	20.87	73.20	15.28
1.5%	Average	0.985 ± 0.013	19.23 ± 0.53	69.68 ± 1.29	13.19 ± 0.50
	Best	1.000	19.64	71.80	14.10

^a *V*_{OC} is open-circuit voltage. ^b *J*_{SC} is short-circuit current density. ^c FF is fill factor.



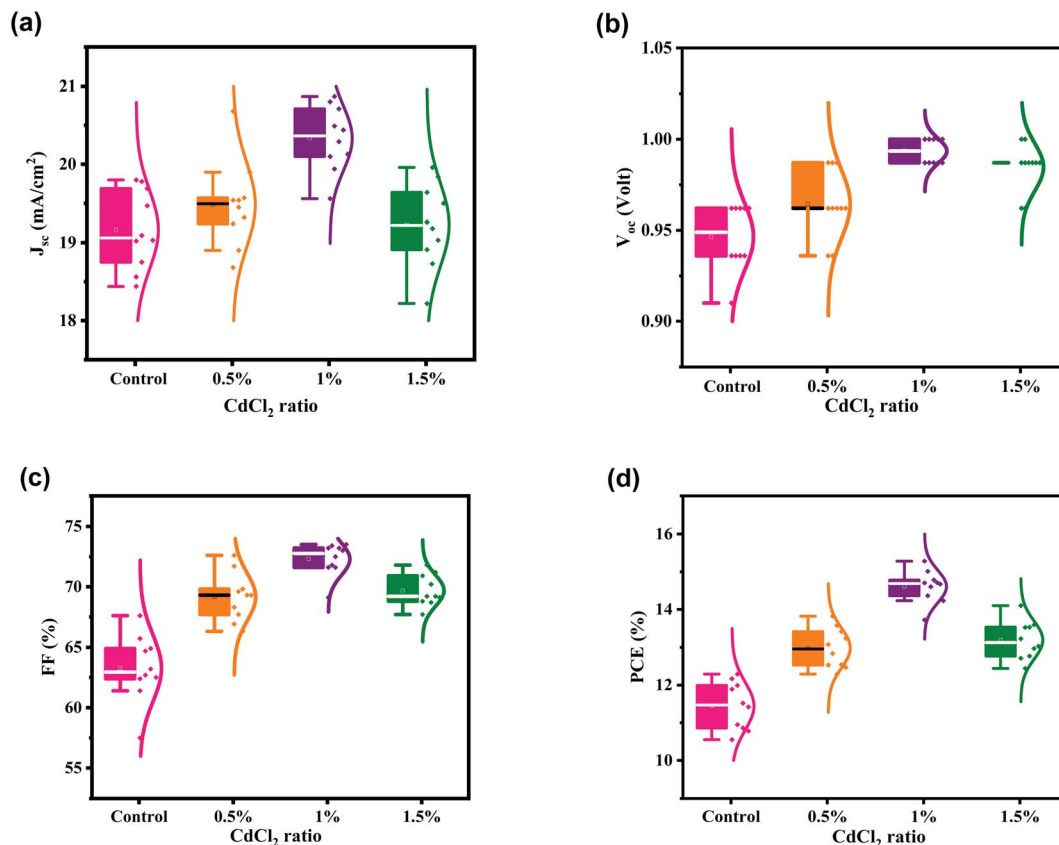


Fig. 5 (a–d) Statistics of the PCE, J_{sc} , V_{oc} and FF for HPSCs modified with different amounts of CdCl₂ additive. PV parameters collected from 10 devices for each set.

then decreasing to 69.68% at 1.5% (Fig. 5c). As a consequence, PCE is boosted from 11.44% for the control device to 14.61% for the 1% CdCl₂-modified device (Fig. 5d). Also, the statistics show the narrow distribution of PV parameters of HPSCs based on the optimized 1% CdCl₂, indicating excellent reproducibility of solar cells.

To further study the role of CdCl₂ material on the PV properties of space-charge-limited current (SCLC) measurements for electron-only devices with a structure of FTO/c-TiO₂/mp-TiO₂/MAPbI₃ with or without CdCl₂/PCBM/Au were investigated. The electron trap density (N_t) of the perovskites layers were calculated as follows:⁴⁷

$$N_t = \frac{2V_{TFL}\epsilon_r\epsilon_0}{qL^2} \quad (1)$$

where q is the electric charge, L indicates the thickness of the perovskite film, ϵ_r is the relative dielectric constant of the MAPbI₃ perovskite ($\epsilon_r = 32$), ϵ_0 refers to the free space permittivity, and V_{TFL} indicates for the trap-filled limit voltage. The V_{TFL} for the electron-only devices for the control and 1%-CdCl₂ the 2D/3D perovskite layer were found to be 0.63 V and 0.40 V respectively. The hole trap density values for the control and 1%-CdCl₂ PSCs are $13.19 \times 10^{15} \text{ cm}^{-3}$ and $8.38 \times 10^{15} \text{ cm}^{-3}$ respectively. It indicates that the CdCl₂ additive facilitates the electron transfer process and reduces the charge recombination by suppressing the carrier traps within the bulk of perovskite

layer. In addition, the hysteresis index (HI) of PSCs was recorded by measuring J - V curves in reverse and forward scan directions (Fig. 6b). HI was obtained using a formula of $\left(\frac{\text{PCE}_{\text{reverse}} - \text{PCE}_{\text{forward}}}{\text{PCE}_{\text{reverse}}}\right) \times 100\%$.⁴⁰ Compared with the control PSCs with a HI value of 11.99%, the 1% CdCl₂-modified PSCs showed a HI of 2.87%. The weakened HI in the CdCl₂-based PSCs can be attributed to the passivated domain boundaries in the modified perovskite layer. In addition, as SCLC results showed, CdCl₂ incorporation into the perovskite layer accelerates electron transfer within the bulk of perovskite layer and suppresses HI in the PSCs.^{48,49}

One of the most critical performance factors for future commercial uses of HPSCs is long-term stability. Therefore, the HPSC longevity of related MAPbI₃-based regular HPSCs (unmodified and with 1% CdCl₂) was assessed by tracing the PCE progression under ambient conditions with a RH of 25–40% in the dark (see Fig. 7). After 2040 h of age, the PCEs of the HPSC with 1% CdCl₂ additive demonstrate outstanding stability, maintaining 83% of the original value. In contrast, the performances of the unmodified HPSC dropped to less than 59% of their initial values after 2040 h, mainly attributable to perovskite degradation. Hence, the incorporation of CdCl₂ in MAPbI₃-based HPSCs, results in outstanding PV parameters and environmental stability.



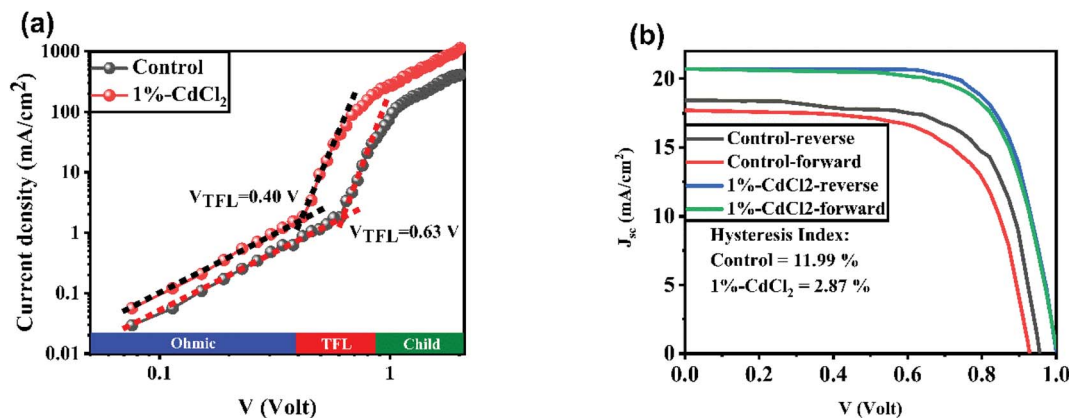


Fig. 6 (a) J - V curves of electron-only of control and 1%- CdCl_2 perovskite solar cells devices. (b) J - V curves of a perovskite solar cell device with or without of CdCl_2 additive in reverse and forward scan directions.

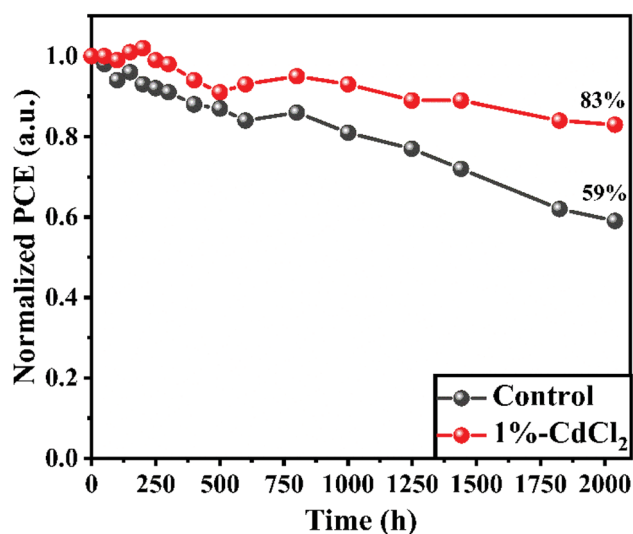


Fig. 7 Stability test of the unsealed control perovskite solar cells stored in ambient conditions with a relative humidity (RH) of 25–40%.

4. Conclusion

In conclusion, we have proven a simple method based on additive engineering to improve the PCE and stability of MAPbI_3 -based HPSCs by adding a certain concentration of CdCl_2 additive. The impacts of varying concentrations of CdCl_2 additives on perovskite layer characteristics and PV properties were studied. High crystallinity, large grain size, improved optical absorption ability, and fewer pinholes and GBs were obtained for perovskite film fabricated with the optimized CdCl_2 concentration, resulting in excellent photovoltaic performance. The synergistic influence of crystal growth control and pinhole passivation by CdCl_2 effectively decreases the charge carrier recombination. With the incorporation of 1% CdCl_2 into the MAI solution, the HPSC yielded a champion PCE of over 15.28% under the reverse scan conditions. In comparison with unmodified HPSC, the HPSC based on 1% CdCl_2 also shows excellent ambient stability, which maintained 83% of its

initial PCE after being stored for 2040 h without encapsulation. Our findings indicate that CdCl_2 is a good additive for making inexpensive HPSCs with excellent stability.

Conflicts of interest

The authors declare no conflict of interest.

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