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# Nonadiabatic dynamics studies of the $\text{H}(^2\text{S}) + \text{RbH}(\text{X}^1\Sigma^+)$ reaction: based on new diabatic potential energy surfaces

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The global diabatic potential energy surfaces (PESs) that correspond to the ground ( $1^2\text{A}'$ ) and first excited states ( $2^2\text{A}'$ ) of the  $\text{RbH}_2$  system PES are constructed based on 17 786 *ab initio* points. The neural network method is used to fit the PESs and the topographic features of the new diabatic PESs are discussed in detail. Based on the newly constructed diabatic PESs, the dynamics calculations of the  $\text{H}(^2\text{S}) + \text{RbH}(\text{X}^1\Sigma^+) \rightarrow \text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)/\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  reactions are performed using the time-dependent wave packet method. The dynamics properties of these two channels such as the reaction probabilities, integral cross sections, and differential cross sections (DCSs) are calculated at state-to-state level of theory. The nonadiabatic effects are discussed in detail, and the results indicate that the adiabatic results are overestimated from the dynamics values. The DCSs of these two channels are forward biased, which indicates that the abstraction mechanism plays a dominant role in the reaction.

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## 1. Introduction

The Born–Oppenheimer (BO) approximation separates the motions of the electrons and nucleus of an atom and is the foundation of modern theoretical chemistry. However, the BO approximation fails at describing electronic state crossings or degeneracies (nonadiabatic effects), and these phenomena are ubiquitous in molecular reaction dynamics. Thus, to further our understanding of reaction mechanisms, accurate diabatic potential energy surfaces (PESs) are necessary. Solving for the mixing angle between different electronic states is a key step during the construction, and is often challenging. A rigorous approach to obtaining the mixing angle is to calculate the integral of the nonadiabatic coupling matrix elements (NACMEs) between different electronic states. However, some drawbacks are found when the NACME approach is applied to polyatomic systems. First of all, the NACMEs are not equal to zero at positions far from the conical intersection area. Secondly, singularities in the NACMEs and abrupt changes within certain ranges of configuration space result in discontinuities in the derivatives near the conical intersection area. Thirdly, the integration path through NACME is not unique for systems containing more than two atoms. As a result, some approximation methods have been developed to build diabatic PESs, such as the CI coefficients,<sup>1–3</sup> molecular properties,<sup>4–6</sup> and diabaticization by Ansatz<sup>7</sup> methods.

Reactions of  $\text{X} + \text{H}_2$  ( $\text{X} = \text{Li}, \text{Na}, \text{K}, \text{Rb}, \text{Cs}$ ) are highly endothermic, and nonadiabatic coupling occurs along the reaction path. Thus,  $\text{X} + \text{H}_2$  reactions and their reverse reactions are ideal models for the study of nonadiabatic processes. In present work, the  $\text{H} + \text{RbH}$  reaction was chosen. Because Rb is highly endergonic, it is always excited from ground state to an excited state in experiments using a laser. For example, using a crossed beam apparatus, Cuvellier and co-worker<sup>8</sup> studied the electronic-to-vibrational ( $E-V$ ) energy transfer processes of the  $\text{Rb}(7\text{S}) + \text{H}_2(j=1) \rightarrow \text{Rb}(5\text{D}) + \text{H}_2(j=3)$  reaction in 1986. The results indicated that the near-resonant process has a larger cross section than the non-resonant process at low collision velocities. In 1997, the  $E-V$  energy transfer processes between  $\text{Rb}(5^2\text{P}_{1/2,3/2})$  atom and  $\text{H}_2$  molecule were investigated by Chen *et al.*<sup>9</sup> using coherent anti-stokes Raman scattering spectroscopy. To best of our knowledge, only Fan *et al.*<sup>10</sup> studied the reaction between  $\text{Rb}(5^2\text{D}, 7^2\text{S})$  atom and  $\text{H}_2$  in 1999. The relative reactivity of different excited states of Rb with  $\text{H}_2$  followed  $\text{Rb}(7^2\text{S}_{1/2}) > \text{Rb}(5^2\text{D}_{3/2}) > \text{Rb}(5^2\text{D}_{5/2})$  and was determined by comparing the spectral intensities of  $\text{RbH}$  action spectra and Rb atomic fluorescence excitation spectra. In addition, the results also indicated that the available energy was mainly transformed into translational energy (80%).

Theoretical studies of the  $\text{H} + \text{RbH}$  reaction were mainly carried out by the group of Pascale<sup>11,12</sup> in 1987. The non-reaction collisions processes between  $\text{Rb}(5^2\text{P})$  and  $\text{H}_2$ ,  $\text{D}_2$  molecules were studied using close-coupling quantum mechanical methods based on a diabatic PES the authors constructed. The fine-structure transition cross sections of between  $\text{Rb}(5^2\text{P}_{1/2})$  and  $\text{Rb}(5^2\text{P}_{3/2})$  were calculated and compared with available experimental data.

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As far as we know, no theoretical studies of the  $\text{Rb} + \text{H}_2$  reaction and its reverse reactions until now. The main reason is that few PESs are available for this system. Thus, the aims of present paper are to present new global nonadiabatic PESs for the  $\text{RbH}_2$  system and to study the nonadiabatic  $\text{H} + \text{RbH}$  ( $v_0 = 0$ ,  $j_0 = 0$ ) reaction process. This paper is organized as follows: the theoretical methods used in the *ab initio* calculations as well as the PES fitting and dynamics calculations are shown in Section 2; the properties of the diabatic PESs and the dynamical results are presented in Section 3; and the conclusions are presented in Section 4.

## 2. Theoretical methods

### 2.1 Construction of the diabatic PES

**2.1.1 *Ab initio* calculations.** The *ab initio* energy points used in the PES fitting were calculated using the internally contracted multi-reference configuration interaction (MRCI) method<sup>13,14</sup> with the aug-cc-pVQZ<sup>15,16</sup> basis set for the H atoms and pseudopotential basis set Def2-QZVPP<sup>17</sup> for the Rb atom. The *ab initio* calculations were first performed using the Hartree-Fock calculation, and then state-averaged complete active space self-consistent field (SA-CASSCF)<sup>18,19</sup> calculations over three lowest electronic states ( $1^2\text{A}'$ ,  $2^2\text{A}'$  and  $1^2\text{A}''$ ) with equal weight were performed. The active space of the  $\text{RbH}_2$  system was composed of 12 active orbitals ( $10\text{a}' + 2\text{a}''$ ). In detail,  $3\text{a}'$  and  $1\text{a}''$  are closed and related to the 4s and 4p orbitals of Rb, and the  $7\text{a}'$  and  $1\text{a}''$  active orbitals correspond to the 1s, 2s orbitals of H and the 5s, 5p and 4d orbitals of Rb. Subsequently, the internally contracted MRCI calculation was performed based on the reference orbits determined in the SA-CASSCF calculation. The Davidson correction was applied to the final energies to compensate for the termination errors of high-order terms. Jacobi coordinates ( $R_Q$ ,  $R_{\text{HH}}$ ,  $\theta$ ) were used to cover the entire configuration space, and 17 786 *ab initio* energy points were calculated on a non-uniform grid. In detail,  $R_Q$  is the distance between Rb atom and the midpoint of  $\text{H}_2$  molecule, and varied between  $0.4 \leq R_Q/a_0 \leq 25$ ;  $R_{\text{HH}}$  is the bond length of  $\text{H}_2$  molecule, and varied between  $0.6 \leq R_{\text{HH}}/a_0 \leq 25$ ;  $\theta$  is the angle between  $R_Q$  and  $R_{\text{HH}}$  varied from  $1^\circ$  to  $89^\circ$ . The geometrical structures used in the *ab initio* calculations are presented in Table 1. In addition, all *ab initio* calculations were performed using the MOLPRO<sup>20</sup> program and using  $C_s$  symmetry.

**2.1.2 Diabatization scheme.** As discussed in the introduction, because it is very difficult to determine the mixing angle between the coupled electronic states from the integral of NACMEs, the molecule property method was used in the

present calculations. After thorough testing, the molecule property method accurately reflected the transition characteristics between coupled states in the  $\text{RbH}_2$  system. The diabaticization scheme has been widely used in previous theoretical studies.<sup>4–6</sup> Herein, a brief introduction is given, and more details can be found in previous literature.<sup>21–23</sup> In  $C_s$  symmetry, the  $1^2\text{A}'$  and  $2^2\text{A}'$  states in the  $\text{RbH}_2$  system are correlated. As the symmetry changes to  $C_{2v}$ , the  $1^2\text{A}'$  and  $2^2\text{A}'$  states change to the  $1^2\text{A}_1$  and  $1^2\text{B}_2$  states, respectively.

As the diabatic coupling is considered between  $1^2\text{A}'$  and  $2^2\text{A}'$  states, the diabatic wave function  $\phi_i$  was obtained from the adiabatic wave function  $\psi_i$  using a unitary transformation

$$\begin{pmatrix} \phi_1 \\ \phi_2 \end{pmatrix} = \begin{pmatrix} \cos \gamma & -\sin \gamma \\ \sin \gamma & \cos \gamma \end{pmatrix} \begin{pmatrix} \psi_1 \\ \psi_2 \end{pmatrix} \quad (1)$$

where  $\gamma$  is the mixing angle. Then the diabatic energies  $V_{ii}$  were obtained from adiabatic energies following

$$V_{11} = \cos^2 \gamma E_1 + \sin^2 \gamma E_2 \quad (2)$$

$$V_{22} = \cos^2 \gamma E_2 + \sin^2 \gamma E_1 \quad (3)$$

$$V_{12} = (E_2 - E_1) \cos \gamma \times \sin \gamma \quad (4)$$

$$V_{12} = V_{21} \quad (5)$$

where  $V_{11}$  and  $V_{22}$  are diabatic energies corresponding to the diabatic states, and  $V_{12}$  and  $V_{21}$  are the coupling potential energies between  $1^2\text{A}'$  and  $2^2\text{A}'$  states. In present calculation, the dipole operator transition moments are used to obtain the mixing angle. To obtain the mixing angle, the  $\psi_3$  wave function for the  $1^2\text{A}''$  state was taken and was not coupled with the  $1^2\text{A}'$  and  $2^2\text{A}'$  states. Thus, according to eqn (1), the matrix elements  $\langle \psi_3 | \hat{P} | \psi_1 \rangle$  and  $\langle \psi_3 | \hat{P} | \psi_2 \rangle$  can be presented as

$$\langle \psi_3 | \hat{P} | \psi_1 \rangle = \cos \gamma \langle \psi_3 | \hat{P} | \phi_1 \rangle + \sin \gamma \langle \psi_3 | \hat{P} | \phi_2 \rangle \quad (6)$$

$$\langle \psi_3 | \hat{P} | \psi_2 \rangle = \cos \gamma \langle \psi_3 | \hat{P} | \phi_2 \rangle - \sin \gamma \langle \psi_3 | \hat{P} | \phi_1 \rangle \quad (7)$$

The matrix elements  $\langle \psi_3 | \hat{P} | \phi_1 \rangle = 0$  and  $\langle \psi_3 | \hat{P} | \phi_2 \rangle = 1$  at high symmetry conditions. In the present work, the approximation that  $\langle \psi_3 | \hat{P} | \phi_1 \rangle = 0$  and  $\langle \psi_3 | \hat{P} | \phi_2 \rangle = 1$  was used for all geometries, not just for the high symmetry geometries. Then, the mixing angle was obtained by the following formula

$$\gamma = \arctan \left( \left| \frac{\langle \psi_3 | \hat{P} | \psi_1 \rangle}{\langle \psi_3 | \hat{P} | \psi_2 \rangle} \right| \right) \quad (8)$$

Table 1 The geometrical structures used in the *ab initio* calculation

|                        |  |
|------------------------|--|
| $R_Q$ (bohr)           | 0.6, 0.8, 1.0, 1.2, 1.4, 1.6, 1.8, 2.0, 2.2, 2.4, 2.6, 2.8, 3.0, 3.3, 3.6, 3.9, 4.2, 4.5, 4.9, 5.3, 5.7, 6.1, 6.5, 7.0, 7.5, 8.0, 8.5, 9.0, 9.5, 10.0, 11.0, 12.0, 13.0, 14.0, 15.0, 16.0, 18.0, 20.0, 22.0, 25.0                                    |
| $R_{\text{HH}}$ (bohr) | 0.6, 0.7, 0.8, 0.9, 1.0, 1.1, 1.2, 1.3, 1.4, 1.5, 1.6, 1.7, 1.8, 1.9, 2.0, 2.2, 2.4, 2.6, 2.8, 3.0, 3.3, 3.6, 3.9, 4.2, 4.5, 4.9, 5.3, 5.7, 6.1, 6.5, 7.0, 7.5, 8.0, 8.5, 9.0, 9.5, 10.0, 11.0, 12.0, 13.0, 14.0, 15.0, 16.0, 18.0, 20.0, 22.0, 25.0 |
| $\theta$ ( $^\circ$ )  | $1^\circ$ , $15^\circ$ , $30^\circ$ , $45^\circ$ , $60^\circ$ , $75^\circ$ , $89^\circ$  |



**2.1.3 PES fitting.** Compared with traditional fitting methods, the neural network (NN) method has several numerical advantages and is more accurate. As such, the NN method has been widely used in the construction of PESs for the  $F + H_2$ ,<sup>24</sup>  $O^+ + H_2$ ,<sup>25</sup>  $Br + H_2$  (ref. 26) reactions among others. In present work, the permutation invariant polynomial neural network (PIP-NN)<sup>27,28</sup> method was used to fit the PES. Herein, we only make a brief introduction to the method. For the calculations, three bond lengths in  $RbH_2$  were transformed into  $P_{RbH_a} = \exp(-0.2 \times r_{RbH_a})$ ,  $P_{RbH_b} = \exp(-0.2 \times r_{RbH_b})$  and  $P_{HH} = \exp(-0.2 \times r_{HH})$ . To avoid the discontinuities in the derivatives, the permutation invariant polynomial method was introduced.

$$G_1 = \frac{(P_{RbH_b} + P_{RbH_a})}{2} \quad (9)$$

$$G_2 = P_{RbH_b} \times P_{RbH_a} \quad (10)$$

$$G_3 = P_{HH} \quad (11)$$

To ensure a random sample, the data for  $\{G_1, G_2, G_3\}$  were subjected to a normalization treatment

$$K_i = \frac{2(G_i - G_{i,\min})}{(G_{i,\max} - G_{i,\min})} - 1 \quad (12)$$

$$I_i = (1 + K_i) \times (G_{i,\max} - G_{i,\min}) \times \frac{1}{2} + G_{i,\min} \quad (13)$$

where  $G_{i,\min}$  and  $G_{i,\max}$  are the minimum and maximum values of  $G_i$ , and  $I_i$  is the input data. In this work, two hidden layers were used and 15 neurons were included in each layer. The Levenberg–Marquardt method<sup>29</sup> was used to train the neural network and the expansion of neural network can be written as

$$V_{RbH_2} = \sum_{k=1}^{15} w_{1,k}^3 \times f^2 \times \left( \sum_{j=1}^{15} w_{k,j}^2 \times f^1 \times \left( \sum_{i=1}^{N_{\text{tot}}} w_{j,i}^1 \times I_i + b_j^1 \right) + b_k^2 \right) + b_1^3 \quad (14)$$

where  $N_{\text{tot}}$  is the number of energy points,  $w_{j,i}^l$  are connection weights of the  $i$ th neuron in  $(l-1)$ th layer and the  $j$ th neuron in  $l$ th layer, and  $b_j^l$  are biases of the  $j$ th neuron in  $l$ th layer,  $f^l$  and  $f^2$  are training function and it can be presented as

$$f(x) = \frac{e^x - e^{-x}}{e^x + e^{-x}} \quad (15)$$

Over-fitting is a well-known problem in neural network calculations. To avoid over-fitting, the input data were randomly sampling into three portions, namely the training part (90 percent), testing part (5 percent) and validation part (5 percent).

## 2.2 Time-dependent wave packet method

The time-dependent wave packet (TDWP) method is a powerful theoretical tool for reaction scattering calculations, and it is

widely used to calculate atom-diatom collision scattering<sup>30–32</sup> as well as the reactions involving polyatomic system.<sup>33,34</sup> Herein, we only make a brief introduction to the TDWP method used in the atom-diatom reaction scattering. In body-fixed frame, the Hamiltonian of the  $H(^2S) + RbH(X^1\Sigma^+)$  reaction for a given total angular momentum  $J$  can be presented as

$$\hat{H} = -\frac{\hbar^2}{2\mu_R} \frac{\partial^2}{\partial R^2} - \frac{\hbar^2}{2\mu_r} \frac{\partial^2}{\partial r^2} + \frac{(\hat{J} - \hat{j})^2}{2\mu_R R^2} + \frac{\hat{j}^2}{2\mu_r r^2} + \begin{pmatrix} V_{11} & V_{12} \\ V_{21} & V_{22} \end{pmatrix} \quad (16)$$

where  $R$  is the distance between the H atom and the center of mass of  $RbH$  molecule, and  $r$  is the vibrational coordinate of  $RbH$ .  $\hat{J}$  and  $\hat{j}$  are angular momentum operators for the  $RbH_2$  system and  $RbH$  molecule, respectively, and  $\mu_R$  and  $\mu_r$  are the reduced masses. The  $2 \times 2$  matrix is the potential energy obtained from the diabatic PES of  $RbH_2$  system.

Before the propagation, an initial wave packet needs to be defined. Here, at  $t = 0$ , the initial wave packet can be presented as

$$\psi(R, r, t = 0) = u_{k_0}(R) \phi_{v_{j_0}}(r) Y_{j_0 K_0}^{JM\epsilon}(\hat{R}, \hat{r}) \quad (17)$$

where  $u_{k_0}(R)$  is a Gaussian function, and  $\phi_{v_{j_0}}(r)$  is the ro-vibrational eigenfunction of  $RbH$ .

In the calculation, the split operator scheme<sup>35</sup> was used to propagate the initial wave packet. Finally, the reactant coordinates-based method<sup>36,37</sup> was used to extract the state-to-state information. Using the state-to-state  $S$ -matrix, the state-to-state reaction probability was obtained as

$$P_{v'j' \leftarrow v_0j_0}^J = \frac{1}{(2j+1)} \sum_{K',K} |S_{v'j'K' \leftarrow v_0j_0K_0}^J|^2 \quad (18)$$

the state-to-state integral cross section (ICS) was obtained as

$$\sigma_{v'j' \leftarrow v_0j_0}^J = \frac{\pi}{(2j_0+1)k_{v_0j_0}^2} \sum_{K'} \sum_{K_0} \sum_J (2J+1) |S_{v'j'K' \leftarrow v_0j_0K_0}^J|^2 \quad (19)$$

and the state-to-state differential cross section (DCS) was obtained as

$$\frac{d\sigma_{v'j' \leftarrow v_0j_0}^J(\vartheta, E)}{d\Omega} = \frac{1}{2j_0+1} \sum_{K'} \sum_{K_0} \left| \frac{1}{2ik_{v_0j_0}^2} \sum_J (2J+1) d_{K'K_0}^J(\vartheta) S_{v'j'K' \leftarrow v_0j_0K_0}^J \right|^2 \quad (20)$$

where  $S_{v'j'K' \leftarrow v_0j_0K_0}^J$  is the state-to-state  $S$ -matrix,  $d_{K'K_0}^J(\vartheta)$  is the reduced rotation matrix element, and  $\vartheta$  is the scattering angle.

The rate constant of the  $H(^2S) + RbH(X^1\Sigma^+)$  reaction was obtained by thermally averaging the translational energy of the corresponding ICS, and the formula is presented as

$$k_{v_0j_0}(T) = \sqrt{\frac{8k_b T}{\pi \mu}} (k_b T)^{-2} \int_0^\infty E e^{-E/k_b T} \sigma_{v_0j_0}(E) dE \quad (21)$$



where  $k_b$  is Boltzmann's constant.

### 3. Results and discussion

#### 3.1 The features of the diabatic PES

In present work, the  $V_{11}$ ,  $V_{22}$ , and  $V_{12}$  terms in the diabatic potential were individually fitted using the PIP-NN method. The root-mean-square-errors (RMSE) for  $V_{11}$ ,  $V_{22}$ , and  $V_{12}$  terms are 4.95, 6.45 and 9.17 meV, respectively. Fig. 1 shows the fitting errors for  $V_{11}$ ,  $V_{22}$ , and  $V_{12}$  as a function of the diabatic energy. Because the sign of  $V_{12}$  in the PES abruptly changes with diabatic energy, the absolute values of  $V_{12}$  were used in fitting process. The energy derivatives using the absolute values were discontinuity and nonzero values emerged for high symmetry geometries ( $C_{2v}$ ,  $D_{2h}$ ); however, these effects on the final quantum dynamics results are negligible. The relatively large fitting error was mainly due to the complex morphology of  $V_{12}$ . Due to the intrude of excited states, the fitting error for the PES of  $V_{22}$  is larger than the PES of  $V_{11}$ .

Fig. 2 shows a comparison of the fitting values and the *ab initio* calculation results as a function of  $R_Q$  at several selected angles. The potential energy curves (PECs) in the diabatic representation crossed smoothly near the conical intersection area and the PECs in the adiabatic representation overlapped in the asymptotic region. These results indicated that potential coupling term  $V_{12}$  was zero in the asymptotic region. In addition, the position of crossing point becomes small with the increase of bond angle. The fitting values were in good agreement with the *ab initio* data which indicated that present PESs were well fitted.

The accuracy of long-range interaction potential will affect the dynamics results for the H + RbH reaction, especially in the low collision energy region. Therefore, long-range interaction potential of H close to RbH in collinear structures is shown in Fig. 3, when the RbH bond length is fixed at equilibrium. To test the accuracy of the fitting results for the long-range interaction

potential, the *ab initio* results are also shown in Fig. 3. As shown in the figure, the fitting results are in good agreement with the *ab initio* values. However, when zoom in the long-range region, the differences between fitting results and *ab initio* values are more obvious. The fitting error is about 1 meV for the long-range region.

The potential energies for an H atom moving around an RbH molecule on the  $V_{22}$  and  $V_{11}$  PESs are displayed in Fig. 4(a) and (b), respectively, for a fixed RbH bond length of 4.78 bohr.  $R_Q$  is the distance between the H atom and the center of the RbH molecule. As seen in the figure, a deep well was observed near the H atom in both Fig. 4(a) and (b). This well suggests that at low collision energies, if an H atom approaches the RbH molecule from the Rb side, it will move around the RbH molecule to form an  $H_2$  molecule. Thus, the sideward and backward scattering signals may be more obvious at low collision energies. Meanwhile, at high collision energies, the incoming H atom will be repelled and the H atom in RbH does not form chemical bond with the incoming H atom. If the H atom approach the RbH molecule from H atom side, the incoming H atom will be trapped in the deep well and an  $H_2$  molecule will form.

The potential energies of the coupling term  $V_{12}$  are shown in Fig. 5(a) and (b) for  $R_Q$  equal to 4.4 and 5.4 bohr, respectively. As seen in the figure, the largest value for the  $V_{12}$  is about 0.6 eV in the collinear structure. In addition,  $V_{12}$  is almost equal to zero beyond the cross region. It is very clear that the largest values of  $V_{12}$  are mainly focused in the region  $0 \leq \theta (^{\circ}) \leq 30$ ,  $1.8 \leq R_{HH}$  (bohr)  $\leq 2.7$  when  $R_Q = 4.4$  bohr and in the region  $0 \leq \theta (^{\circ}) \leq 45$ ,  $1.8 \leq R_{HH}$  (bohr)  $\leq 2.5$  when  $R_Q = 5.4$  bohr. The regular shape of  $V_{12}$  indicates that the present diabatic PESs are a reasonable description of cross region. Fig. 6 presents the mixing angles at several selected angles for  $R_Q = 5.4$  bohr and correspond to the data in Fig. 5(b). The mixing angle is very small in the region beyond the cross area and increases in the vicinity of the cross region where it reaches a maximum value of  $45^{\circ}$ , which corresponds to the cross point. At the conical intersection at  $90^{\circ}$ , the mixing angle is equal to zero, because the coupling term  $V_{12} = 0$ .

The Rb +  $H_2$  reaction studied here is very similar to the previous studied Li +  $H_2$  reaction as both are highly exothermic reactions between an alkali metal and hydrogen. The minimum energy paths from the RbH<sub>2</sub> diabatic PESs calculated here and that from the LiH<sub>2</sub> diabatic PESs reported by He *et al.*<sup>38</sup> are compared in Fig. 7. As seen in the figure, the energy of the  $H(^2S) + RbH(X^1\Sigma^+)$  channel was set as zero. For convenient comparison, the energy of the  $H(^2S) + LiH(X^1\Sigma^+)$  channel was also set as zero. For the RbH<sub>2</sub> system, it is very clear that the  $H(^2S) + RbH(X^1\Sigma^+)$  reaction is highly exothermic and the energies are about  $-1.65$  and  $-3.0$  eV for the  $Rb(5^2P) + H_2(X^1\Sigma_g^+)$  and  $Rb(5^2S) + H_2(X^1\Sigma_g^+)$  channels, respectively. As the zero-point energy considered, the exoergic energies were  $-1.435$  and  $-2.775$  eV for these two channels, respectively. The  $H(^2S) + LiH(X^1\Sigma^+)$  reaction is also an exothermic reaction and the exoergic energies were about  $-1.2$  and  $-2.4$  eV for the  $Li(2^2P) + H_2(X^1\Sigma_g^+)$  and  $Li(2^2S) + H_2(X^1\Sigma_g^+)$  channels, respectively. This comparison clearly shows that the exoergic energies of the  $H(^2S) + RbH(X^1\Sigma^+)$  reaction are larger than that of  $H(^2S) + LiH(X^1\Sigma^+)$

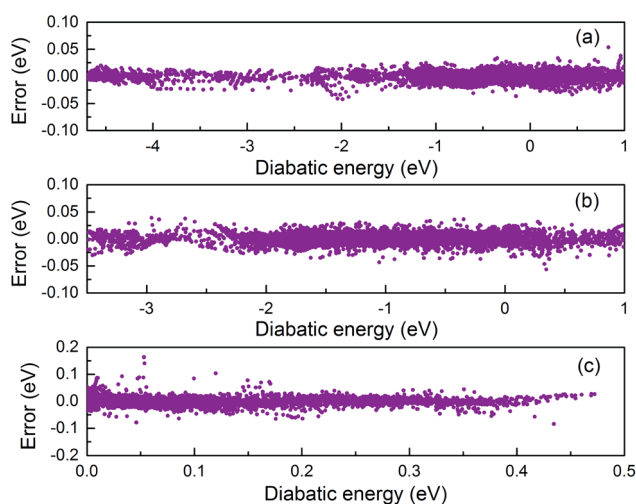


Fig. 1 The distributions of fitting errors as a function of diabatic energy for  $V_{11}$  (a),  $V_{22}$  (b) and  $V_{12}$  (c).





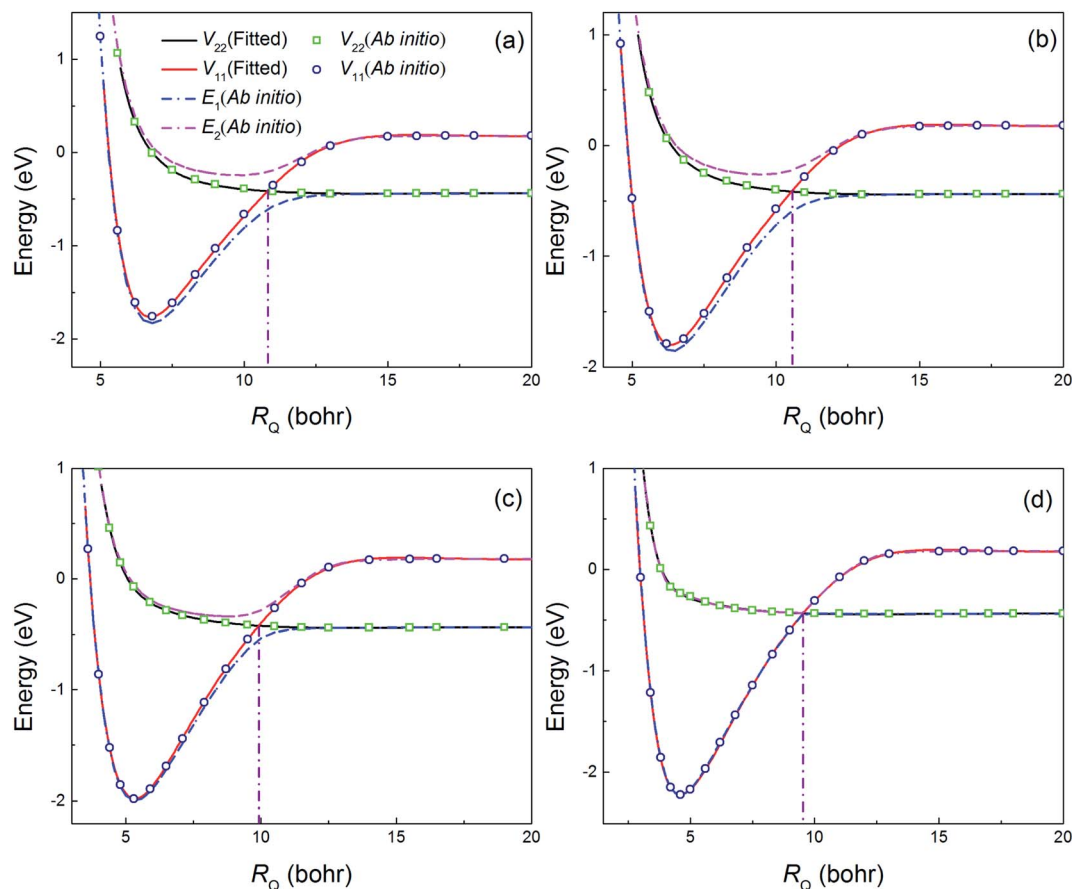


Fig. 2 Comparisons of the fitting values and the *ab initio* data at several selected angles ((a), (b), (c) and (d) correspond to 1°, 30°, 60°, and 89°, respectively) at a fixed H<sub>2</sub> molecular bond length of 3.4 bohr.

reaction. The Rb + H<sub>2</sub> and Li + H<sub>2</sub> reactions had other similar features including a well on the  $V_{22}$  PES and a well and a barrier on the  $V_{11}$  PES. Additional features of these stationary points on the RbH<sub>2</sub> diabatic PESs, including the geometrical structure, energy, and harmonic vibrational frequency are presented in Table 2. As seen in the table, these stationary points belong to  $C_{2v}$  symmetry.

### 3.2 Dynamics results and discussion

The optimal numerical parameters obtained from numerous numerical simulations of the reaction probabilities at total angular momentum  $J = 0$  are shown in Table 3 and were used in all  $J > 0$  nonadiabatic TDWP calculation.

The total reaction probabilities of the  $H(^2S) + RbH(X^1\Sigma^+) \rightarrow Rb(^5S) + H_2(X^1\Sigma_g^+)/Rb(^5P) + H_2(X^1\Sigma_g^+)$  reaction for several selected total angular momentum of  $J = 0, 30, 60$  and  $90$  are displayed in Fig. 8 over a collision energy range from 0.001 to 1.0 eV. As shown in the figure, the reaction probabilities of these two channels are similar at  $J = 0$  and both decrease with an increase in the collision energy. This is a typical phenomenon seen in exothermic reaction without barrier. In fact, it is unusual that the reaction probabilities for these two reactions are almost equal at same collision energy. Before propagation, the initial wave packet is located in the  $H(^2S) + RbH(X^1\Sigma^+)$  reactant channel. After propagation, the wave packets are distributed between the  $Rb(^5S) + H_2(X^1\Sigma_g^+)$  and  $Rb(^5P) + H_2(X^1\Sigma_g^+)$  channels according to the adiabatic-to-diabatic transformation matrix which is composed of the mixing angle. If the wave packet is evenly distributed between the two channels, the reaction probability of  $Rb(^5P) + H_2(X^1\Sigma_g^+)$  channel will be larger than that of  $Rb(^5S) + H_2(X^1\Sigma_g^+)$  channel. The exoergic energy of  $Rb(^5P) + H_2(X^1\Sigma_g^+)$  channel is less than

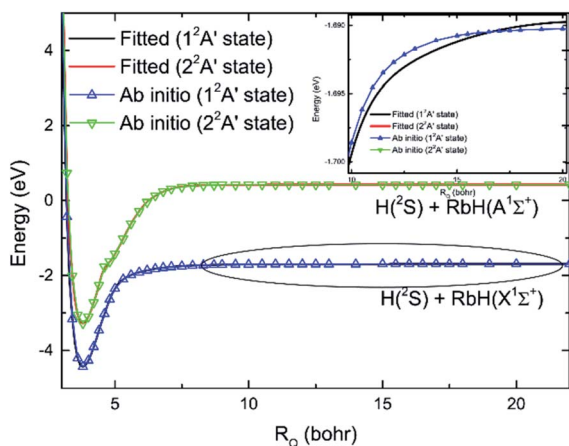


Fig. 3 The long-range interaction potential as H atom approach to the RbH molecule in collinear structures.

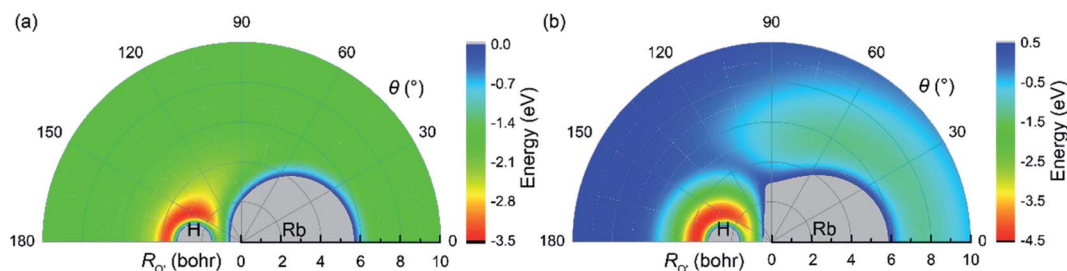


Fig. 4 Color plot of an H atom moving around an RbH molecule. The bond length of RbH is fixed at the equilibrium distance (4.78 bohr). (a) Denotes the  $V_{22}$  PES (b) denotes the  $V_{11}$  PES.

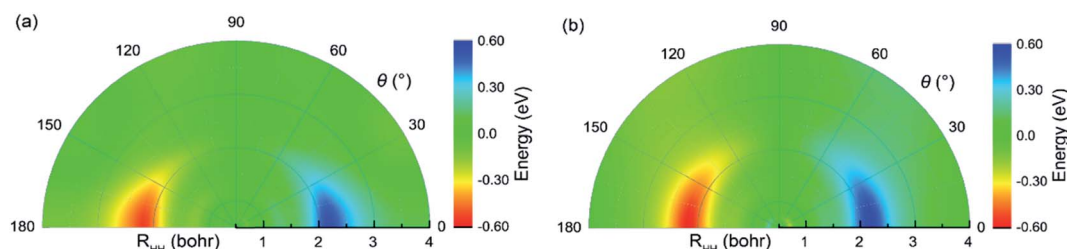


Fig. 5 Color plot for the coupling term  $V_{12}$  as a function of internal coordinates when the  $R_Q$  equal to 4.4 bohr (a) and 5.4 bohr (b), respectively.

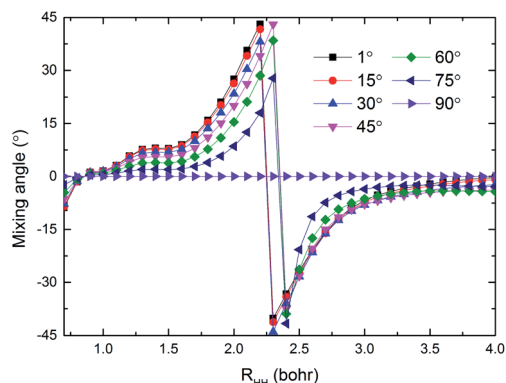


Fig. 6 The mixing angles at different selected angles when  $R_Q = 5.4$  bohr.

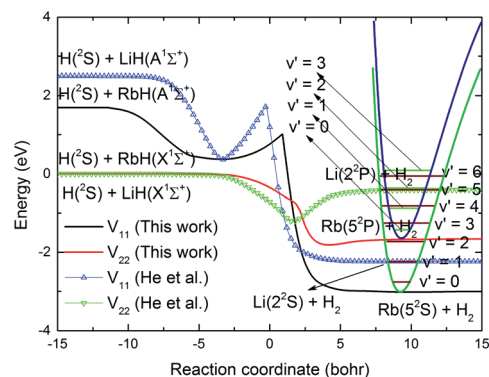


Fig. 7 Comparison of the global minimum energy paths of  $\text{RbH}_2$  diabatic PESs and that of the  $\text{LiH}_2$  diabatic PESs obtained from He *et al.*<sup>38</sup> as a function of the reaction coordinate.

that of  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel, so the velocity of H in the corresponding  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel is relatively low. Therefore, the reactivity of the  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel will be higher than that of the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel. Here the results are inconsistent with the hypothesis. Therefore, we suppose that a large proportion of wave packet is distributed in the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel, which is the main channel of the  $\text{H}(^2\text{S}) + \text{RbH}(\text{X}^1\Sigma^+)$  reaction. Moreover, the difference between the reaction probabilities of  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  and  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channels increase with increasing  $J$  value. This shows that with the increase of  $J$ , the proportion of wave packet that goes into the  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel becomes smaller and the nonadiabatic effect becomes less obvious. This conclusion is consistent with the conclusion of studies of the  $\text{H} + \text{LiH}$  reaction.<sup>39</sup>

To further understand the nonadiabatic effects in the reaction, we also performed the adiabatic dynamics calculations based on the adiabatic PES which obtained by diagonalizing the diabatic PESs. The results are presented in Fig. 9 for  $J = 0$  and 10. As seen in the figure, the reaction probabilities from the adiabatic results are larger, similar to the results seen for the  $\text{H} + \text{LiH}$  reaction.<sup>39</sup> However, the sum of two channels in the nonadiabatic results is even larger than the adiabatic results, especially in the high collision energy range. This is very different from the results for the  $\text{H} + \text{LiH}$  reaction, in which the sum of two channels in the nonadiabatic values is smaller than the adiabatic values. We suppose such difference can be attributed to the different exothermic energies of these two reactions. However, it is important to note that the conclusions based on the adiabatic results are unreliable because the



Table 2 The spectroscopic constants of the stationary points<sup>a</sup>

| $R_{\text{RbH}_e}$                                | $R_{\text{RbH}_b}$ | $R_{\text{HH}}$ | Angle ( $^\circ$ ) | $E_n$ (eV)          | $\omega_1$ ( $\text{cm}^{-1}$ ) | $\omega_2$ ( $\text{cm}^{-1}$ ) | $\omega_3$ ( $\text{cm}^{-1}$ ) |
|---|--------------------|-----------------|--------------------|---------------------|---------------------------------|---------------------------------|---------------------------------|
| <b>Global minimum (<math>V_{22}</math> PES)</b>   |                    |                 |                    |                     |                                 |                                 |                                 |
| 5.611   | 5.611              | 1.435           | 14.7               | -0.145 <sup>b</sup> | 286.47                          | 601.94                          | 4014.17                         |
| <b>Local minimum (<math>V_{11}</math> PES)</b>    |                    |                 |                    |                     |                                 |                                 |                                 |
| 5.242   | 5.242              | 8.720           | 112.56             | -1.316 <sup>c</sup> | 171.40                          | 572.42                          | 810.67                          |
| <b>Transition state (<math>V_{11}</math> PES)</b> |                    |                 |                    |                     |                                 |                                 |                                 |
| 5.761   | 5.761              | 4.128           | 42.0               | -0.393 <sup>c</sup> | 312.77                          | 421.25                          | 1847.46 <sup>i</sup>            |

<sup>a</sup> The unit is bohr for bond length. The angle is [HRbH]. <sup>b</sup> Relative to the energy of  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  asymptote. <sup>c</sup> Relative to the energy of  $\text{H} + \text{RbH}(\text{A}^1\Sigma^+)$  channel.

Table 3 The optimal numerical parameters used in the nonadiabatic TDWP calculation (the atomic unit is used unless otherwise stated)<sup>a</sup>

|                        |   |
|------------------------|---|
| Grids range and size   | $R \in [2.5, 22.0]$ , $N_R^{\text{tot}} = 179$ , $N_R^{\text{int}} = 159$<br>$r \in [2.5, 22.0]$ , $N_r^{\text{tot}} = 179$ , $N_r^{\text{asy}} = 79$                                       |
| Initial wave packet    | $j_{\text{min}} = 0 \sim j_{\text{max}} = 259$ , $N_j = 259$<br>$\exp[-(R - R_c)^2/2\Delta_R^2]\cos(k_0 R)$ , $R_c = 16.0$<br>$\Delta_R = 0.3$ , $k_0 = \sqrt{2\mu_R E_0}$ , $E_0 = 0.5$ eV |
| Matching plane         | $R_v = 16.0$  |
| K-block                | $K = 9$   |
| Total propagation time | 45 000  |
| Time step              | 10  |
| Highest $J$ value      | 90  |

<sup>a</sup>  $N_R^{\text{tot}}$  denotes the total grids in  $R$  direction,  $N_R^{\text{int}}$  denotes the grids in the interaction region of  $R$  direction,  $N_r^{\text{tot}}$  denotes the total grids in  $r$  direction,  $N_r^{\text{asy}}$  denotes the grids in the interaction region of  $r$  direction.

nonadiabatic effects were not considered for both  $\text{H} + \text{RbH}$  and  $\text{H} + \text{LiH}$  reactions.

The resonance structures were observed on the reaction probability curves. Only a shallow well was found on the  $V_{22}$  PES (seen in Fig. 7). The vibrational state-resolved reaction probabilities at  $J = 0$  of these two channels used to determine the reaction mechanism are shown in Fig. 10(a) and (b), respectively. Comparing the results in Fig. 10(a) with those in Fig. 10(b) shows that the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel had more vibrational states than the  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel due to the larger exoergic energy of  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel than that of the  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel. As shown in Fig. 7, no well was observed in the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel, but some resonance structures were observed. We believe a potential well will be formed on the reaction path when the product is in the vibrational excited state, and the potential well deepens as the excited vibrational state energy level increases. Thus, the number of bound and quasi-bound states will be trapped in the well, and the resonance structures formed due to the coupling between the bound states in the well and the continuum states on the different vibrational states. We believe that these resonance structures can be classified as “Feshbach resonance”. As shown in Fig. 10(a), the amplitude of the reaction probability first increases with an increase in the number of vibrational states, then reaches a maximum value around  $v' = 4, 5$ , and finally decreases with a further increase in the number of

vibrational states. In addition, as the number of vibrational states increases, the resonance structure becomes more obvious as we speculated. For the  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel, a shallow well was observed on the minimum energy path, however, as shown in Fig. 10(b), the situation was very similar to Fig. 10(a). The “Feshbach resonance” mechanism also plays an important role in the reaction.

The total ICSs of the two channels of the  $\text{H}(^2\text{S}) + \text{RbH}(\text{X}^1\Sigma^+)$  reaction are plotted in Fig. 11 as a function of collision energy. The ICSs of these two channels are very similar and both decrease with an increase of collision energy. In addition, the ICSs of the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel are larger than that of  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel over the collision energies range studied. This is a predictable result. Due to the influence of nonadiabatic effect, the gap between each  $J$  partial wave constituting the ICSs of the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel and each  $J$  partial wave constituting the ICSs of the  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel gradually increased with an increase in the  $J$  value.

The vibrational state-resolved ICSs of the  $\text{H}(^2\text{S}) + \text{RbH}(\text{X}^1\Sigma^+)$  reaction are displayed in the left (a) and right (b) panels of Fig. 12 and correspond to the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  and  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  reactions, respectively. For the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel, the vibrational states below 7 are opened due to the large exoergic energy (seen in Fig. 7 approx. 2.775 eV) and the higher vibrational excited states (7, 8 and 9) are gradually opened as the collision energy further increases. This trend reflects that more energy is available below 7 and transformed into internal energy. The amplitude of the vibrational state-resolved ICS was very similar to the vibrational state distribution of the reaction probabilities (Fig. 10). The amplitude first increases, reaches the maximum value around a number of vibrational states of 4, and then decreases as the number of vibrational states further increases. The amplitude of the  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel is very similar to the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel, but the number of vibrational states is relatively small, which can be attributed to the relatively small exoergic energy (approx. 1.435 eV).

The rotational state-resolved DCSs of the  $\text{H}(^2\text{S}) + \text{RbH}(\text{X}^1\Sigma^+)$  reaction at several selected collision energies of 0.1, 0.3, 0.5 and 0.7 eV are presented in Fig. 13 and 14 for the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  and  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channels, respectively. As seen in the figure, the shapes of each vibrational state-resolved ICSs are



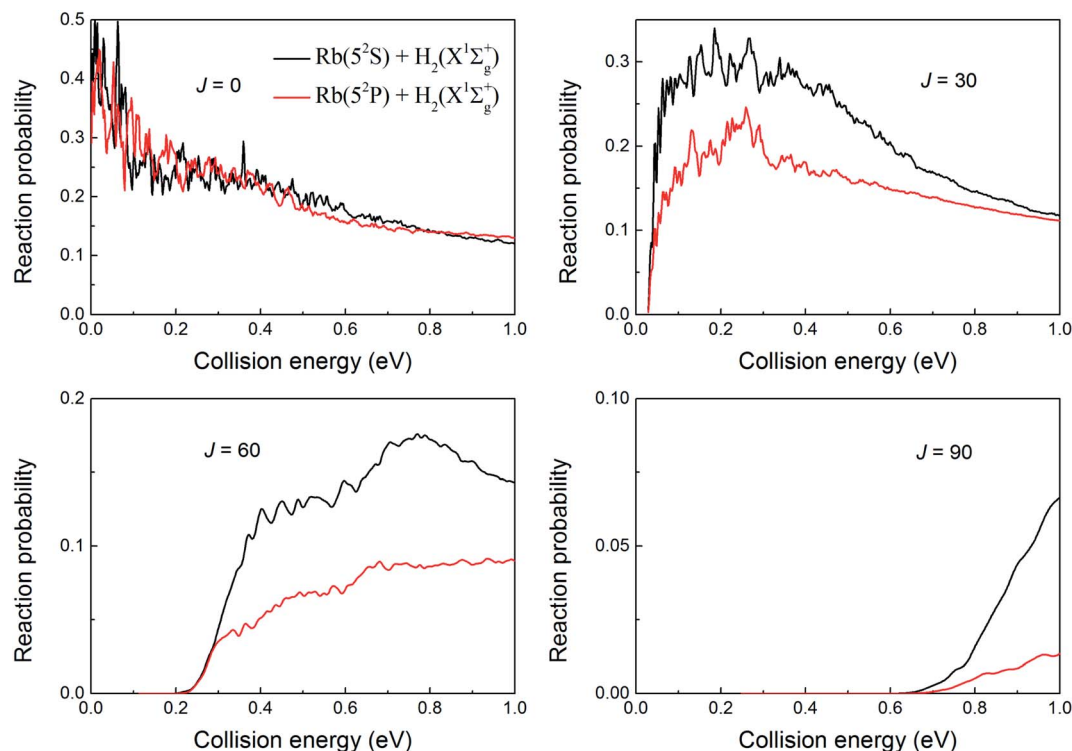


Fig. 8 Total reaction probabilities of the two channels of the  $\text{H}(^2\text{S}) + \text{RbH}(X^1\Sigma^+)$  reactions at total angular momentums of  $J = 0, 30, 60$  and  $90$ .

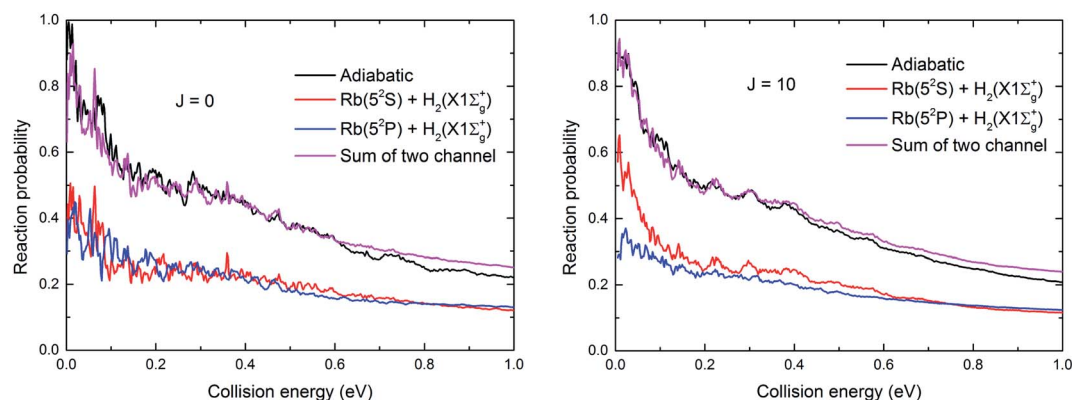


Fig. 9 The total reaction probabilities based on the adiabatic and diabatic PESs at  $J = 0$  and  $10$ , respectively.

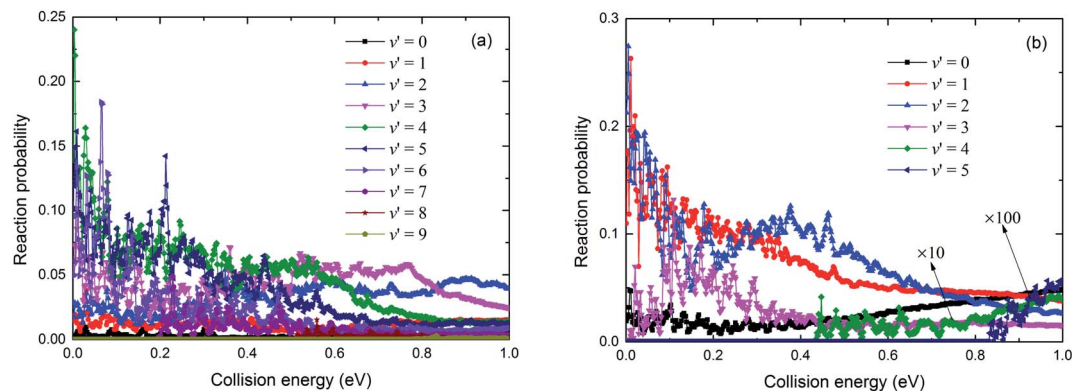


Fig. 10 The vibrational state-resolved reaction probabilities for the  $\text{Rb}(5^2\text{S}) + \text{H}_2(X^1\Sigma_g^+)$  (a) and  $\text{Rb}(5^2\text{P}) + \text{H}_2(X^1\Sigma_g^+)$  (b) channels at  $J = 0$ .





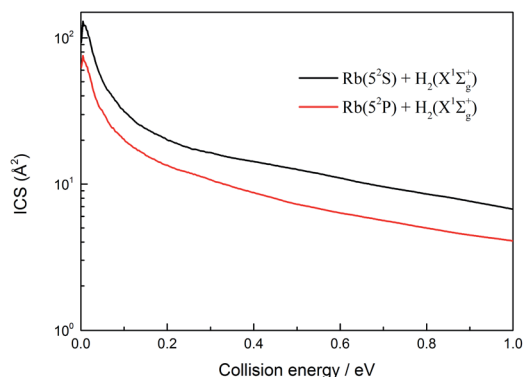


Fig. 11 Total integral cross sections of the two channels of the  $\text{H}(^2\text{S}) + \text{RbH}(X^1\Sigma^+)$  reaction.

similar. The vibrational state-resolved ICSs first increase with  $j'$ , reach a maximum value, and then decrease as  $j'$  further increases. In all cases, the rotational quantum number is the broadest in the ground state and narrows as  $v'$  increases. At a given collision energy, the total energy is fixed. Thus, the internal energy is transformed from rotational to vibrational energy as  $v'$  increases. In addition, the rotational quantum number increases with increasing collision energy, although the increases are not very apparent. In all cases, the amplitude of each vibrational state-resolved ICS increases with increasing  $v'$  before reaching a maximum value and then decreases as  $v'$  further increases.

Fig. 15 shows the DCSs of the two channels at collision energies of 0.1, 0.3, 0.5 and 0.7 eV, respectively. As seen in the figure, the forward scattering signals are very apparent at all collision energies while the sideward and backward scattering signals are not obvious. For most scattering angles, the

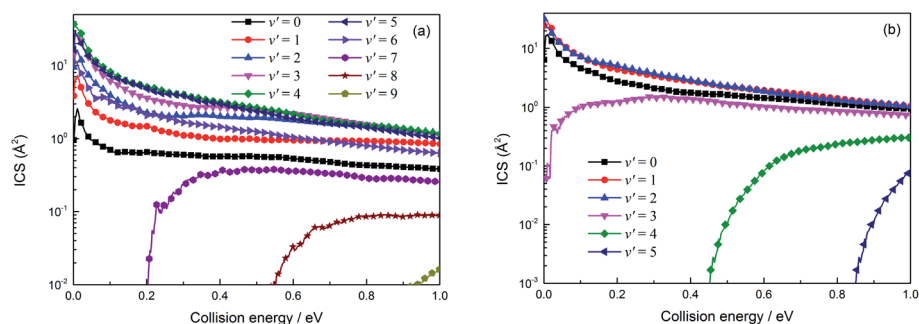


Fig. 12 Vibrational state-resolved integral cross sections for the two channels of the  $\text{H}(^2\text{S}) + \text{RbH}(X^1\Sigma^+)$  reaction (panel (a) denotes the  $\text{Rb}(5^2\text{P}) + \text{H}_2(X^1\Sigma_g^+)$  channel and panel (b) denotes the  $\text{Rb}(5^2\text{S}) + \text{H}_2(X^1\Sigma_g^+)$  channel).

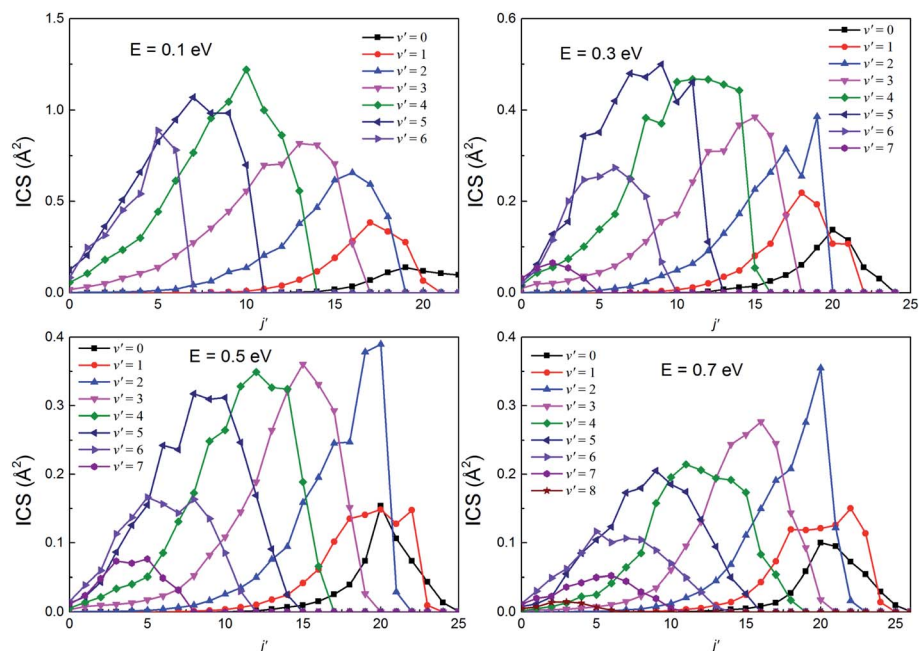
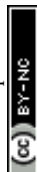


Fig. 13 The rotational state-resolved integral cross sections at several selected collision energies for the  $\text{H}(^2\text{S}) + \text{RbH}(X^1\Sigma^+) \rightarrow \text{Rb}(5^2\text{S}) + \text{H}_2(X^1\Sigma_g^+)$  reaction.



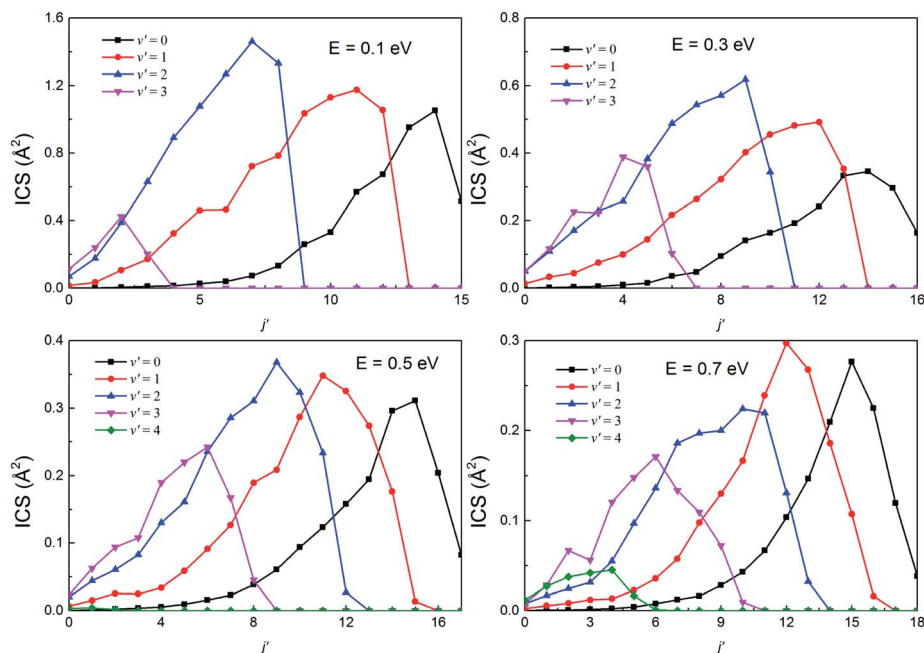


Fig. 14 The rotational state-resolved integral cross sections at several selected collision energies for the  $\text{H}(^2\text{S}) + \text{RbH}(\text{X}^1\Sigma^+) \rightarrow \text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  reaction.

scattering intensity decreases as the collision energy increases which is consistent with trends in the ICSs shown in Fig. 11. In addition, the decrements in the scattering intensity are anisotropic. At high collision energies, the  $\text{H}_2$  molecule retain a higher forward scattering signal than sideward and backward scattering signals. For example, at a scattering angle of 1 degree, the DCSs are 25.1, 22.4 and 27.7  $\text{\AA}^2 \text{sr}^{-1}$  for collision energies of 0.3, 0.5 and 0.7 eV, respectively. Whereas, at a scattering angle

of  $90^\circ$ , the DCSs are 0.77, 0.50 and 0.37  $\text{\AA}^2 \text{sr}^{-1}$  for 0.3, 0.5 and 0.7 eV, respectively. The DCSs are bias toward forward scattering, which indicates that the direct abstraction mechanism plays a dominant role in the reaction.

The vibrational state-resolved DCSs of the two channels of the title reaction are shown in Fig. 16(a) and (b), respectively, at a collision energy of 0.5 eV. Similar with the results in Fig. 10 and 12, more vibrational states were seen in the  $\text{Rb}(5^2\text{S}) +$

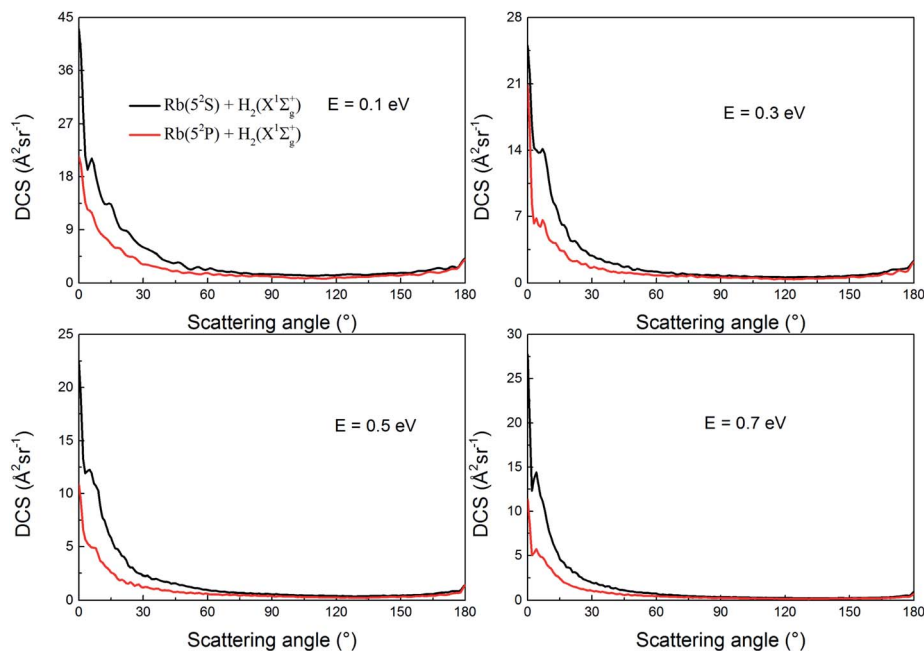


Fig. 15 Total differential cross sections of the two channels of the  $\text{H}(^2\text{S}) + \text{RbH}(\text{X}^1\Sigma^+)$  reaction at several selected collision energies.



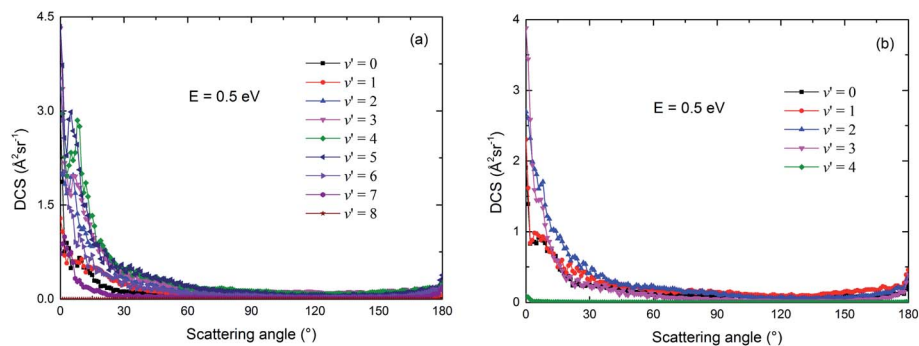


Fig. 16 The vibrational state-resolved differential cross sections for the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  (a) and  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  (b) channels when the collision energy equal to 0.5 eV.

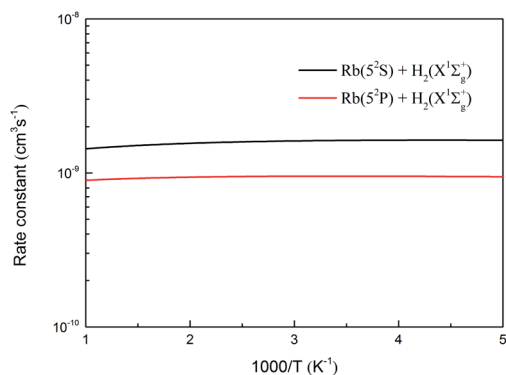


Fig. 17 Specific state rate constants of the  $\text{H}_2(\text{S}) + \text{RbH}(\text{X}^1\Sigma^+)$  reaction over a temperature range from 200 to 1000 K.

$\text{H}_2(\text{X}^1\Sigma_g^+)$  channel than in  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel, due to the relatively larger exoergic energy of the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channel. In addition, the shapes of the ICS for these vibrational states are very similar to the shape of the total ICS and are both biased towards forward scattering. This result also indicates that the direct abstraction mechanism dominates the reaction.

The specific initial state rate constants of the two channels of the  $\text{H}_2(\text{S}) + \text{RbH}(\text{X}^1\Sigma^+)$  reaction are displayed in Fig. 17 over a temperature range from 200 to 1000 K. As seen in the figure, the rate constants for the initial state ( $v = 0, j = 0$ ) of the two channels are not very sensitive to the temperature. The rate constants are around  $1.6 \times 10^{-9}$  and  $9.4 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$ , for the  $\text{Rb}(5^2\text{S}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  and  $\text{Rb}(5^2\text{P}) + \text{H}_2(\text{X}^1\Sigma_g^+)$  channels, respectively.

## 4. Conclusion

In present work, new global diabatic PESs for the ground and first excited states of the  $\text{RbH}_2$  were constructed by fitting 17 786 *ab initio* points. The SA-CASSCF and MRCI methods with aug-cc-pVQZ and Def2-QZVPP basis sets were used in the *ab initio* calculations. The mixing angle was obtained from the molecular properties of the dipole transition moment. The PIP-NN method was used to fit the diabaticization terms,  $V_{11}$  and  $V_{22}$ , as well as the coupling term  $V_{12}$  individually. These three terms

were well fitted and the RMSEs were 4.95, 6.45 and 9.17 meV for  $V_{11}$ ,  $V_{22}$  and  $V_{12}$ , respectively. The features of the new diabatic PESs were discussed in detail and dynamics studies of the  $\text{H}_2(\text{S}) + \text{RbH}(\text{X}^1\Sigma^+)$  reaction were carried out based on the new diabatic PESs. The dynamics properties, including the reaction probabilities, ICSs, DCSs, rovibrational state distributions of the product, and rate constants are reported. The forward scattering signals indicate that the abstraction mechanism plays an important role in the reaction.

## Conflicts of interest

There are no conflicts to declare.

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## References

- 1 H. J. Werner, B. Follmeg and M. H. Alexander, *J. Chem. Phys.*, 1988, **89**, 3139–3151.
- 2 B. Jiang, C. J. Xie and D. Q. Xie, *J. Chem. Phys.*, 2011, **135**, 164311.
- 3 B. Jiang and D. Q. Xie, *Chin. J. Chem. Phys.*, 2009, **22**, 601–604.
- 4 D. He, J. Yuan, H. Li and M. Chen, *Sci. Rep.*, 2016, **6**, 25083.
- 5 J. Yuan, D. He, S. Wang, M. Chen and K. Han, *Phys. Chem. Chem. Phys.*, 2018, **20**, 6638–6647.
- 6 Z. J. Yang, J. C. Yuan, S. F. Wang and M. D. Chen, *RSC Adv.*, 2018, **8**, 22823–22834.
- 7 T. Lenzen and U. Manthe, *J. Chem. Phys.*, 2017, **147**, 084105.
- 8 J. Cuvellier, L. Petitjean, J. M. Mestdag, D. Paillard, P. de Pujo and J. Berlande, *J. Chem. Phys.*, 1986, **84**, 1451–1458.
- 9 M. L. Chen, W. C. Lin and W. T. Luh, *J. Chem. Phys.*, 1997, **106**, 5972–5978.
- 10 L. H. Fan, J. J. Chen, Y. Y. Lin and W. T. Luh, *J. Phys. Chem. A*, 1999, **103**, 1300–1305.
- 11 J. Pascale, F. Rossi and W. E. Baylis, *Europhys. Lett.*, 1987, **3**, 183–188.



- 12 J. Pascale, F. Rossi and W. E. Baylis, *Phys. Rev. A*, 1987, **36**, 4219–4235.
- 13 H. J. Werner and P. J. Knowles, *J. Chem. Phys.*, 1988, **89**, 5803–5814.
- 14 P. J. Knowles and H. J. Werner, *Chem. Phys. Lett.*, 1988, **145**, 514–522.
- 15 T. H. Dunning, *J. Chem. Phys.*, 1989, **90**, 1007–1023.
- 16 R. A. Kendall, T. H. Dunning and R. J. Harrison, *J. Chem. Phys.*, 1992, **96**, 6796–6806.
- 17 T. Leininger, A. Nicklass, W. Küchle, H. Stoll, M. Dolg and A. Bergner, *Chem. Phys. Lett.*, 1996, **255**, 274–280.
- 18 H. J. Werner and P. J. Knowles, *J. Chem. Phys.*, 1985, **82**, 5053–5063.
- 19 P. J. Knowles and H. J. Werner, *Chem. Phys. Lett.*, 1985, **115**, 259–267.
- 20 H. J. Werner, P. J. Knowles, G. Knizia, F. R. Manby and M. Schutz, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 2012, **2**, 242–253.
- 21 D. W. Schwenke, S. L. Mielke, G. J. Tawa, R. S. Friedman, P. Halvick and D. G. Truhlar, *Chem. Phys. Lett.*, 1993, **203**, 565–572.
- 22 P. Halvick and D. G. Truhlar, *J. Chem. Phys.*, 1992, **96**, 2895–2909.
- 23 M. Peric, R. J. Buenker and S. D. Peyerimhoff, *Mol. Phys.*, 1990, **71**, 673–691.
- 24 J. Chen, Z. G. Sun and D. H. Zhang, *J. Chem. Phys.*, 2015, **142**, 024303.
- 25 W. T. Li, J. C. Yuan, M. L. Yuan, Y. Zhang, M. H. Yao and Z. G. Sun, *Phys. Chem. Chem. Phys.*, 2018, **20**, 1039–1050.
- 26 W. T. Li, D. He and Z. G. Sun, *J. Chem. Phys.*, 2019, **151**, 185102.
- 27 B. Jiang, J. Li and H. Guo, *Int. Rev. Phys. Chem.*, 2016, **35**, 479–506.
- 28 B. Jiang and H. Guo, *J. Chem. Phys.*, 2013, **139**, 054112.
- 29 M. T. Hagan and M. B. Menhaj, *IEEE Transact. Neural Networks Learn. Syst.*, 1994, **5**, 989–993.
- 30 Z. G. Sun, D. Q. Yu, W. B. Xie, J. Y. Hou, R. Dawes and H. Guo, *J. Chem. Phys.*, 2015, **142**, 174312.
- 31 M. L. Yuan, W. T. Li, J. C. Yuan and M. D. Chen, *Int. J. Quantum Chem.*, 2018, **118**, e25493.
- 32 W. B. Xie, L. Liu, Z. G. Sun, H. Guo and R. Dawes, *J. Chem. Phys.*, 2015, **142**, 064308.
- 33 S. Liu, X. Xu and D. H. Zhang, *J. Chem. Phys.*, 2012, **136**, 144302.
- 34 B. Zhao, Z. G. Sun and H. Guo, *J. Chem. Phys.*, 2016, **144**, 064104.
- 35 M. D. Feit, J. A. Fleck and A. Steiger, *J. Comput. Phys.*, 1982, **47**, 412–433.
- 36 S. Gómez-Carrasco and O. Roncero, *J. Chem. Phys.*, 2006, **125**, 054102.
- 37 Z. Sun, X. Lin, S.-Y. Lee and D. H. Zhang, *J. Phys. Chem. A*, 2009, **113**, 4145–4154.
- 38 D. He, J. C. Yuan, H. X. Li and M. D. Chen, *Sci. Rep.*, 2016, **6**, 25083.
- 39 W. T. Li, J. X. Sun and D. He, *Phys. Chem. Chem. Phys.*, 2020, **22**, 17587–17596.

