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New dimeric chromanone derivatives from the mutant strains of *Penicillium oxalicum* and their bioactivities†

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Three new chromanone dimer derivatives, paecilins F–H (1–3) and ten known compounds (4–13), were obtained from the mutant strains of *Penicillium oxalicum* 114-2. Their structures were elucidated by extensive analysis of spectroscopic data and comparison with reported data, and the configurations of 1–3 were resolved by quantum chemical calculations of NMR shifts and ECD spectra. Compounds 5 and 11 showed significant anti-influenza A virus activities with IC₅₀ values of 5.6 and 6.9 μ M, respectively. Compounds 8 and 9 displayed cytotoxic activities against the MIA-PaCa-2 cell line with IC₅₀ values of 2.6 and 2.1 μ M, respectively. Compound 10 exhibited antibacterial activities against *Bacillus cereus* with a MIC value of 4 μ g mL⁻¹.

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Introduction

Fungi have been demonstrated to be an important resource which could produce biologically active compounds with diverse chemical structures.1-4 The Penicillium genus composed of over 200 species is one of the largest groups of fungi,5 and a large quantity of secondary metabolites with varied architectural features and potent biological activities are isolated from Penicillium, including polyketides,6-8 terpenoids,9-11 alkaloids, 12-14 and macrolides. 15-17 The chromanone dimer is a structurally complex polyketide widely distributed in several genera of fungi, exemplified by Aspergillus, 18,19 Xylaria, 20 Penicillium, 21,22 Setophoma, 23 and Gonytrichum, 24,25 which display antimicrobial,23 cytotoxic, 18,19 and innate promoting^{24,25} activities. In the course of investigating the biosynthesis of oxalicine B, tetrahydroxanthone dimer, secalonic acid D was isolated from the oxalicine B deletion mutant unexpectedly, which is well known for impressive anti-tumor activity. As part of our ongoing research for bioactive tetrahydroxanthone dimer derivatives from the mutant strains of Penicillium oxalicum 114-2, the extracts of the mutant strains of P. oxalicum 114-2 were investigated, which led to the isolation of thirteen polyketides, including three new dihydrochromone

Results and discussion

Compound 1 was isolated as yellow powder, and HRESI mass spectrum gave a quasi-molecular ion peak at m/z 679.1600 [M + Na]⁺, corresponding to a molecular formula of C₃₂H₃₂O₁₅ with 17 degrees of unsaturation. The IR spectrum displayed absorption bands at 3422, 1783, 1676, and 1624 cm⁻¹, which indicated the existence of hydroxyl, carbonyl and aromatic moieties. The ¹H NMR spectrum (Table 1) of 1 indicated the signals of two phenolic hydroxyl protons at $\delta_{\rm H}$ 11.91 (1H, s) and 11.82 (1H, s), one carboxyl protons at $\delta_{\rm H}$ 12.13 (1H, br s), two sets of doublets of ortho-coupled aromatic protons at $\delta_{\rm H}$ 7.49 (1H, d, J = 8.4 Hz), 6.68 (1H, d, J = 8.4 Hz), 7.47 (1H, d, J = 8.4 Hz)Hz), 6.61 (1H, d, J = 8.4 Hz), one hydroxyl protons at $\delta_{\rm H}$ 5.73 (1H, d, J=6.6 Hz), two oxymethines at $\delta_{\rm H}$ 4.94 (1H, d, J=6.6 Hz) and 3.89 (1H, d, J = 5.4 Hz), two methoxyls at $\delta_{\rm H}$ 3.70 (3H, s) and 3.64 (3H, s), and two methyls at δ_H 1.18 (3H, d, J = 7.2 Hz), 0.93 (3H, d, J = 6.0 Hz). The ¹³C NMR and DEPT spectra (Table 1) exhibited 32 carbon resonances, including sixteen nonprotonated carbons ($\delta_{\rm C}$ 197.2, 195.3, 175.5, 173.6, 170.4, 169.1, 159.2, 158.3, 158.2, 158.2, 140.8, 140.5, 107.2, 107.2, 87.7, and 83.8, including six carbonyl, four oxygenated aromatic, four aromatic, and two oxygenated), eight methine carbons ($\delta_{\rm C}$ 140.8, 140.5, 107.2, 107.2, 81.8, 74.8, 32.6, and 30.8, including four aromatic and two oxygenated), four methylene carbons ($\delta_{\rm C}$

dimer derivatives: paecilin F (1), paecilin G (2), and paecilin H (3), along with ten known compounds (4–13) (Fig. 1). The isolated compounds were evaluated for antiviral, cytotoxic, and antibacterial activities. Herein, we describe the isolation, structural elucidation and bioactivities of these secondary metabolites from the mutants of *Penicillium oxalicum* 114-2.

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Fig. 1 Structures of compounds 1–13.

40.3, 39.8, 39.1, and 36.7), two methoxyl carbons ($\delta_{\rm C}$ 53.6 and 53.0), and two methyl carbons ($\delta_{\rm C}$ 14.7 and 13.9). The 1 H NMR and 13 C NMR spectroscopic data of **1** were similar to those of paecillin D (**5**) (Tables S10 and S11†). The obvious difference was that one γ -butyrolactone group in paecillin D was changed to a linear side chain in **1**, which was confirmed by the HMBC correlations (Fig. 2) from H-5′ to C-6′, C-7′, C-10′, C-10a′, C-11′, and C-12′; from H-7′ to C-5′, C-6′, C-8′, and C-11′; from 5′-OH to C-5′, C-6′, and C-10a′, along with the 1 H- 1 H COSY correlations

(Fig. 2) of 5'-OH/H-5'/H-6'/H-7'. Furthermore, the HMBC cross-peaks of H-3 with C-2', H-3' with C-2 indicated that two monomers had 2–2' linkage. The large coupling constant of ${}^3J_{\text{H-5,H-6}}$ (6.6 Hz) and NOESY correlation between H-5 and H-6 established their *cis* configuration in the γ-butyrolactone group. The small coupling constant of ${}^3J_{\text{H-5',H-6'}}$ (1.8 Hz) and NOESY correlation between H-5' and H-6' indicated their gauche relationship. Because the γ-butyrolactone group and flexible side chain had few effect on the ECD spectrum of 1, a simplified

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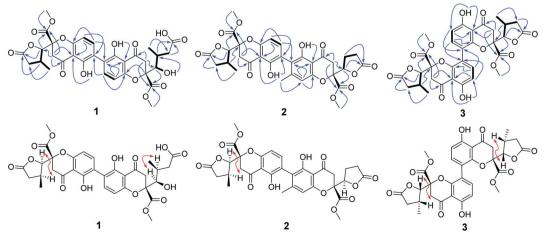


Fig. 2 Key COSY, HMBC and NOESY correlations of 1–3.

Table 1 $\,^{1}$ H (600 MHz) and $\,^{13}$ C NMR (150 MHz) data of compounds 1–3

	1^a		2^b	2^b		3^a	
	$\delta_{ m C}$	δ_{H} (J in Hz)	$\delta_{ m C}$	δ_{H} (J in Hz)	$\delta_{ m C}$	δ_{H} (J in Hz)	
1	158.2, s		159.3, s		160.2, s		
2	117.0, s		116.6, s		109.5,	6.49, d (8.4)	
3	140.8	7.49, d (8.4)	1/11 7	7.37, d (8.4)	d 141.0	7.75, d (8.4)	
3	d	7.45, u (6.4)	d	7.57, u (6.4)	d	7.73, (1 (0.4)	
4	107.2,	6.68, d (8.4)	107.9,	6.67, d (8.4)	114.0, s		
	d		d				
4a	158.3, s		159.3, s		155.4, s		
5		4.94, d (6.6)		4.80, d (6.6)		4.85, d (6.6)	
6		2.98, m	,	3.00, m	-	2.88, m	
7				2.69, dd (17.4, 8.4), 2.50, dd (17.4, 8.4)			
8	175.5, s		175.0, s		174.9, s		
9	195.3, s		194.2, s		194.6, s		
9a	107.1, s		107.7, s		107.3, s		
10 10a	-	3.61, d (17.4), 3.06, d (17.4)	-	3.27, d (17.4), 3.20, d (17.4)		3.56, d (18.0), 3.17, d (18.0)	
10a 11	83.8, s	1.18, d (7.2)	84.7, s	1.36, d (7.2)	85.3, s	1.00, d (7.2)	
12	169.1, s		169.1, s		169.1, s		
13	53.6, q		53.7, q		53.6, q		
1'	158.2, s	-	159.6, s	•	160.2, s		
2'	115.5, s		117.6, s		-	6.49, d (8.4)	
3′	140.5,	7.47, d (8.4)	151.5, s			7.75, d (8.4)	
	d				d		
4'	107.2,	6.61, d (8.4)	109.1,	6.57, s	114.0, s		
	d		d				
4a′	159.2, s		158.9, s		155.4, s		
5'		3.89, d (5.4)		4.88, d (8.4, 6.0)		4.85, d (6.6)	
6′	-	2.22, m		2.45, m	-	2.88, m	
7 8′	-	2.41, m, 2.19, m	-	2.72, m, 2.60, m	-	2.45, dd (16.8, 8.4), 1.72, dd (16.8, 9.0)	
8' 9'	173.6, s 197.2, s		175.7, s 193.2, s		174.9, s 194.6, s		
9 9a'	197.2, s 106.9, s		195.2, s 105.8, s		194.0, s		
10'		3.50, d (17.4), 3.01, d (17.4)	,	3.11, d (16.8), 2.97, d (16.8)		3.56, d (18.0), 3.17, d (18.0)	
10a'	87.7, s	3.30, a (17.4), 3.01, a (17.4)	84.2, s	3.11, u (10.0), 2.57, u (10.0)	85.3, s	3.30, a (10.0), 3.17, a (10.0)	
11'		0.93, d (6.0)	21.5, q	2.12. s	,	1.00, d (7.2)	
12'	170.4, s		169.1, s	· · · · · · · · · · · · · · · · · · ·	169.1, s		
13'	53.0, q		53.9, q		53.6, q		
1-OH		11.82, s		11.74, s		11.57, s	
1'-		11.91, s		11.73, s		11.57, s	
OH							
5'-		5.73, d, (6.6)					
OH							

^a Recorded in DMSO-d₆. ^b Recorded in CDCl₃.

structure of 1A (Fig. S1†), in which two hydroxymethyls replaced γ -butyrolactone group and butanoic acid in 1, was used for the ECD calculations to determine the absolute configurations of C-10a and C-10a'.27 Three stereoisomers, (10aR,10a'R)-1Aa, (10aR,10a'S)-1Ab, (10aS,10a'S)-1Ac existed on the basis of the relative configuration (Fig. S1†). The calculated ECD spectrum of 1Aa was consistent with the experimental one (Fig. 3), suggesting the 10aR,10a'R configuration of 1. Based on abovementioned relative and absolute configurations analyses, four possible configurations there (5S,6R,10aR,5'S,6'R,10a'R)-1a, (5S,6R,10aR,5'R,6'S,10a'R)-1b, (5R,6S,10aR,5'S,6'R,10a'R)-1c, (5R,6S,10aR,5'R,6'S,10a'R)-1d (Fig. S3†). Subsequently, we performed computational

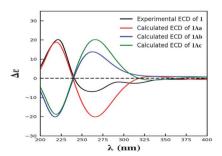


Fig. 3 The experimental ECD spectrum of 1, and the calculated ECD spectra of 1Aa-1Ac.

predictions of NMR chemical shifts of both the isomers 1a-1d using the gauge-independent atomic orbital (GIAO) method at the mPW1PW91/6-311G(2d,p) level in DMSO (PCM). ^{28,29} Compared of the calculated 1H and ^{13}C NMR data for 1a-1d and the experimental 1H and ^{13}C NMR data for 1 (Tables S1 and S2†) indicated that the possibility of structure 1b can be excluded firstly. In addition, the correlation coefficient (R^2) obtained by linear regression analysis, corrected mean absolute deviation (CMAD), corrected largest absolute deviation (CLAD) of 1d were superior to those of 1a and 1c (Table S3†), which indicated that isomer 1d is the most reasonable configuration. Therefore, the structure of 1 was identified as paecilin F.

Compound 2 was isolated as yellow powder, and its molecular formula was established as C32H30O14 based on HR-ESI peak ion m/z 677.1267 [M + K]⁺. Its ¹H and ¹³C NMR data were similar to those of paecilin D (5) (Tables S10 and S11†). The obvious differences include the presence of an aromatic proton at $\delta_{\rm H}$ 6.57 (1H, s) and one aryl methyl moiety at $\delta_{\rm H}$ 2.12 (3H, s) in 2 instead of two *ortho*-coupled aromatic protons at $\delta_{\rm H}$ 7.54 (1H, d, J = 9.0 Hz), 6.63 (1H, d, J = 9.0 Hz) and a methyl moiety at δ_H 1.14 (1H, d, J = 7.2 Hz) in 5. The HMBC correlations from H-4' to C-2', C-4a', and C-9a'; from H-11' to C-2', C-3', and C-4' assigned the location of the aryl methyl group. Meanwhile, the typical methyl in one γ-butyrolactone group in 2 disappeared compared with 5, suggesting the existence of the γ -butyrolactone group without methyl moiety in 2, which was further determined by the ¹H-¹H COSY correlations of H-5'/H₂-6'/H₂-7' and the HMBC correlations from H-5' to C-6', C-7', C-8', and C-12'. The large coupling constant of ${}^{3}J_{\text{H-5,H-6}}$ (6.6 Hz) and NOESY correlation between H-5 and H-6 indicated their cis configuration in the γ -butyrolactone group. The similar positive Cotton effect around 214 nm and negative Cotton effect around 245 nm (Fig. S31†) compared to those of 1 suggested the 10aR,10a'R absolute configuration of 2. To assign the configurations of C-5, C-6, and C-5', quantum chemical calculations of the NMR data of diastereoisomers: (5S,6R,10aR,5'S,10a'R)-2a, four (5S,6R,10aR,5'R,10a'R)-2b, (5R,6S,10aR,5'S,10a'R)-2c, (5R,6S,10aR,5'R,10a'R)-2d (Fig. S4†), were performed using GIAO method at the mPW1PW91/6-311G(2d,p) level in chloroform (PCM).28,29 The calculated chemical shifts of 2a-2d (Tables S4 and S5†), especially the calculated NMR data of 2d, were in good agreement with the experimental values. Moreover, the R^2 , CMAD, and CLAD values of 2d were superior to those of 2a-2c (Tables S6†). The deduction was further supported by DP4+ probability analysis (Fig. S8†), which showed that isomer 2d was the most reasonable configuration with a probability of 99.84% for the combination of ¹H and ¹³C NMR data. These results indicated that the configuration of 2d was the most plausible. Accordingly, the structure of 2 was elucidated to be paecilin G.

Compound 3 was obtained as yellow powder. The molecular formula was determined to be $C_{32}H_{30}O_{14}$ by HR-ESI mass spectrum, which showed a quasi-molecular ion peak at m/z 639.1677 [M + H]⁺. Its ¹H and ¹³C NMR spectroscopic data were great similar to those of paecilin A (4).³⁰ Interpretation of its 1D and 2D NMR data of 3 established the same planar structure as 4. A literature search revealed that a total of five stereoisomers for 3 have been reported: paecilin A (4), paecilin C (3A),²¹

dimethyl (2S,2'S)-5,5'-dihydroxy-2,2'-bis[(2S,3S)-3-methyl-5-oxotetrahydrofuran-2-vl]4,4'-dioxo-[8,8'-bichroman]-2,2'-dicarboxylate (3B),31 dimethyl (2S,2'S)-5,5'-dihydroxy-2,2'-bis([2R,3R]-3methyl-5-oxotetrahydrofuran-2-yl)-4,4'-dioxo-[8,8'-bichromane]-2,2'-dicarboxylate (3C),32 dimethyl (2S,2'S)-5,5'-dihydroxy-2-([2S,3S]-3-methyl-5-oxotetrahydrofuran-2-yl)-2'-([2R,3R]-3-methyl-5-oxotetramethyl-5-oxotetrahydrofuran-2-yl)-4,4'-dioxo-[8,8'-bichromane]-2,2'-dicarboxylate (3D) (Fig. S9†).32 Detailed analysis of NMR data of these stereoisomers indicated that 3 was new chromanone dimer (Tables 1, S12, and S13†). The large coupling constant of ${}^{3}J_{\text{H-5,H-6}}$ (6.6 Hz) and ${}^{3}J_{\text{H-5',H-6'}}$ (6.6 Hz), as well as NOESY correlation between H-5 and H-6 indicated their cis configuration in the γ -butyrolactone group. The absolute configuration of C-10a and C-10a' in 3 was determined as described for 1 by comparison of its experimental and calculated ECD spectra. Similarly, a simplified structure 3A was subjected to ECD calculations.27 Only two possible stereoisomers for 3: (10aS,10a'S)-3Aa and (10aR,10a'R)-3Ab, remained on the basis of its NMR data (Fig. S5†). The calculated ECD spectrum of 3Ab was consistent with the experimental one (Fig. 4), establishing the 10aR,10a'R absolute configuration of 3. After the relative configuration analysis, there are two diastereoisomers: (5R,6S,10aR,5'R,6'S,10a'R)-3a, (5S,6R,10aR,5'S,6'R,10a'R)-3b (Fig. S7†). Chemical shifts of isomers 3a and 3b were predicted using the GIAO method at the mPW1PW91/6-311G(2d,p) level in DMSO.^{28,29} The calculated ¹H and ¹³C NMR data of 3b matched well with the experimental values (Tables S7 and S8†), and the R², CMAD, and CLAD values of calculated ¹H NMR data for 3b were also superior to those of 3a (Tables S9†), indicating the configuration of 3b was more reasonable than 3a. Thus, the structure of 3 was determined to be paecilin H.

Additionally, we analyzed the coupling constant of ${}^3J_{\text{H-5,H-6}}$ for **1–6** and those of its stereoisomers reported in the literature (Table 1, S11 and S12†), ${}^{21,26,30-33}$ indicating that the large coupling constant of ${}^3J_{\text{H-5,H-6}}$ (J > 6 Hz) assigned the *cis* configuration of γ -butyrolactone group, and small coupling constant of ${}^3J_{\text{H-5,H-6}}$ (J < 5 Hz) established the *trans* configuration of γ -butyrolactone group, which can provide guidance for determining the relative configuration of C-5 and C-6 for chromanone dimer and monomer derivatives.

The known compounds were identified as paecilin A (4),³⁰ paecilin D (5),²⁶ paecilin E (6),³³ secalonic acid D (7),³⁴ 2,4′-secalonic acid D (8),³⁵ 4,4′-secalonic acid D (9),³⁶ secalonic acid

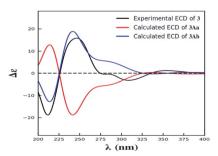


Fig. 4 The experimental ECD spectrum of 3, and the calculated ECD spectra of 3Aa and 3Ab.

Table 2 The anti-IAV activities of 1-13

Compounds	Inhibition rate (%) in 10 μ M	IC_{50} (μ M)	CC_{50} (μ M)	SI
1	23.5 ± 1.0	_	_	
2	27.1 ± 3.1	_	_	
3	18.7 ± 6.1	_	_	
4	27.1 ± 4.6	_	_	
5	90.8 ± 2.1	5.6 ± 0.7	46.2 ± 2.0	8.3
6	16.4 ± 1.5	_	_	
7	74.7 ± 7.1	_	12.2 ± 1.2	
8	65.5 ± 3.1	_	10.5 ± 3.4	
9	49.6 ± 5.4	_	13.2 ± 2.5	
10	40.2 ± 7.4	_	_	
11	97.2 ± 1.8	6.9 ± 1.1	>100	>14.5
12	15.0 ± 4.8	_	_	
13	49.9 ± 1.0	_	_	
Ribavirin	$90.3 \pm 5.2 (50 \ \mu\text{M})$	28.2 ± 4.4	>100	>3.5

H (10),²² monodictyphenone (11),³⁷ emodic acid (12),³⁸ and citreorosein (13)³⁹ by comparison of their NMR and MS data with those reported in the literatures.

Compounds 1-13 were evaluated for anti-influenza A virus (IAV), cytotoxic, and antibacterial activities. Compounds 5 and 11 exhibited significant anti-IAV (A/WSN/33, H1N1) activities with the IC₅₀ values of 5.6 and 6.9 μ M, respectively (Table 2). Compounds 8 and 9 displayed cytotoxic activities against MIA-PaCa-2 cell line with the IC₅₀ values of 2.6 and 2.1 μM, respectively (Table S14†). Compounds 1-13 were tested for antibacterial activities against Escherichia coli ATCC 25922, Enterococcus faecalis ATCC 29212, Pseudomonas syringae CPCC 101099, Bacillus cereus CPCC 101254, Acinetobacter baumannii 12-8, Pseudomonas aeruginosa 13-17, Klebsiella pneumoniae ATCC BAA-2146, Enterococcs faecium ATCC 700221 and Staphylococcus aureus ATCC 33591. Compounds 8 and 10 exhibited moderate antibacterial activities against Bacillus cereus CPCC 101254 with the MIC values of 32 µg mL⁻¹ and 4 µg mL⁻¹, respectively, and others showed no antibacterial activities (Table S15†).

Conclusions

In conclusion, thirteen secondary metabolites including three new chromanone dimers, paecilins F–H (1–3) were isolated from the mutants of *Penicillium oxalicum* 114-2. Their structures were elucidated by detailed interpretation of the spectroscopic data, and the stereochemistry of 1–3 were established by computational predictions of NMR chemical shifts and calculated ECD spectra. Compounds 5 and 11 displayed significant anti-IAV activities with IC₅₀ values of 5.6 and 6.9 μM , respectively, which were stronger than the positive control, ribavirin (IC₅₀ = 28.2 μM), indicating that 5 and 11 might be promising natural agents for antiviral drug discovery.

Experimental

General experimental procedures

Optical rotations were recorded on an Autopol IV automatic polarimeter with a 10 cm glass microcell at 25 $^{\circ}$ C (Rudolph Research Analytical, NJ, USA). The CD spectra were measured

on a JASCO J-815 spectropolarimeter using CH₃OH as the solvent at room temperature (Jasco Corporation, Tokyo, Japan). UV spectra were obtained in CH₃OH on a Persee TU-1901 UV-vis spectrometer (Beijing Purkinje General Instrument Co., Ltd, Beijing, China). IR spectra were acquired on a PerkinElmer FT-IR/NIR spectrometer (PerkinElmer, Waltham, MA, USA). 1D and 2D NMR spectra were performed at 600 MHz for ¹H NMR and 150 MHz for ¹³C NMR on Bruker ARX-600 spectrometer using solvent signals as the internal standard (Bruker, Switzerland). Chemical shifts (δ) are given in ppm, and coupling constants (J) are given in hertz (Hz). ESIMS data were recorded on a Thermo LTQ mass spectrometer (Thermo Fisher Scientific, Waltham, MA, USA). HRESIMS data were measured using a Thermo LTQ Orbitrap XL mass spectrometer (Thermo Fisher Scientific, Waltham, MA, USA). Column chromatography (CC) were carried out with silica gel (200-300 mesh, Qingdao Marine Chemical Inc. Qingdao, PR China). Analytical TLC was carried out on precoated silica gel GF₂₅₄ plates (Qingdao Marine Chemical Industry, Qingdao, China), and spots were visualized under UV light or by spraying with 10% H₂SO₄ in 90% EtOH followed by heating at 120 °C.

Fungal material

The wild type strain *P. oxalicum* 114-2 was kindly provided by Prof Qu Yinbo from National Glycoengineering Research Center, Shandong University. The oxalicine B deletion mutants derived from parental strain 114-2 including $\Delta oxaB$, $\Delta oxaL$, and $\Delta oxaK$ were obtained during our research on biosynthesis of oxalicine B. All the strains used in this study are deposited in the China Pharmaceutical Culture Collection (CPCC), Institute of Medicinal Biotechnology, Chinese Academy of Medical Sciences and Peking Union Medical College.

Fermentation, extraction and isolation

The fungal strain was spread onto slants of PDA medium (potato starch 0.4%, dextrose 2.0%, agar 1.5%; the media were autoclaved at 121 $^{\circ}$ C for 15 min), and incubated at 28 $^{\circ}$ C for 5–7 days to generate spores. The spores were inoculated in PDA plate medium at 28 $^{\circ}$ C for 5 days to prepare spore suspension (seed culture). Fermentation was carried out on plate, each plate

contained about 50 ml medium (2% malt extract broth, 2% agar and 0.2% soybean meal, autoclaving at 121 $^{\circ}$ C for 30 min). After cooling to solid medium, each plate was added into 200 μ L of the spore suspension, coating evenly, and incubated at 28 $^{\circ}$ C for 7 days

The fermented material was extracted repeatedly with EtOAc $(3 \times 3 \text{ L})$, and the organic solvent was evaporated to dryness under vacuum to yield the crude extract, which was initially subjected to silica gel (Qingdao Marine Chemical Inc. Qingdao, China) column chromatography (CC) eluting with dichloromethane-acetone gradient (100:0-0:100, v/v) to produce six fractions (Fr.1-Fr.6) based on the HPLC analysis. Fr.2-Fr.4 and Fr.6 were subjected to reversed-phase silica gel column chromatography eluting with CH₃CN-H₂O gradient (10:90-100:0, v/v), respectively. Fr.2.2 (30 mg) was subjected to Sephadex LH-20 CC to give compound 11 (3 mg). Fr.2.4 (16 mg) was purified by reversed-phase semi-preparative HPLC eluting with CH₃CN- H_2O (0.1% TFA) (45 : 55) at 4 ml min⁻¹ to yield 13 (2.0 mg, t_R = 8.5 min) and 12 (3.0 mg, $t_R = 18$ min). Fr.3.2 (7 mg) was purified by reversed-phase semi-preparative HPLC eluting with CH₃CN- H_2O (0.1% TFA) (40 : 60) at 4 ml min⁻¹ to yield 1 (1.2 mg, t_R = 22.6 min). Fr.3.4 (40 mg) was subjected to Sephadex LH-20 CC to produce four fractions (Fr.3.4.1-Fr.3.4.4). Fr.3.4.2 (9 mg) was purified by reversed-phase semi-preparative HPLC eluting with CH_3CN-H_2O (0.1% TFA) (40 : 60, v/v) at 4 ml min⁻¹ to obtain 2 $(1.4 \text{ mg}, t_R = 33.4 \text{ min})$. Fr.3.4.3 (22 mg) was purified by reversed-phase semi-preparative HPLC eluting with CH3CN- H_2O (0.1% TFA) gradient (40 : 60–100 : 0, v/v) at 4 ml min⁻¹ to obtain 3 (4 mg, $t_R = 20.9$ min). Fr.4.2 (46 mg) was subjected to Sephadex LH-20 CC to produce four fractions (Fr.4.2.1-Fr.4.2.4). Fr.4.2.1 (15 mg) was purified by reversed-phase semipreparative HPLC eluting with CH₃CN-H₂O (50:50, v/v) at 4 ml min⁻¹ to obtain 5 (8 mg, $t_R = 17.0$ min). Fr.4.2.3 (10 mg) was purified by reversed-phase semi-preparative HPLC eluting with CH₃CN-H₂O (0.1% TFA) gradient (45:55-100:0, v/v) at 4 ml ${\rm min}^{-1}$ to obtain **10** (3.5 mg, $t_{\rm R} = 18.3$ min). Fr.4.2.4 (11 mg) was purified by reversed-phase semi-preparative HPLC eluting with CH₃CN-H₂O (0.1% TFA) gradient (45:55-100:0, v/v) at 4 ml min⁻¹ to obtain 4 (1.5 mg, $t_R = 17.0$ min). Fr.4.3 (14 mg) was purified by reversed-phase semi-preparative HPLC eluting with $CH_3CN-H_2O(45:55, v/v)$ at 4 ml min⁻¹ to obtain 6 (3.3 mg, $t_R =$ 32.4 min). Fr.5 subjected to recrystallization to obtain 7 (1.1 g). Fr.6.2 (380 mg) was purified by reversed-phase semi-preparative HPLC eluting with CH₃CN-H₂O (0.1% TFA) gradient (50:50-100 : 0 for, v/v) at 4 ml min⁻¹ to obtain 9 (15 mg, $t_R = 17.9$ min) and 8 (26 mg, $t_R = 18.0$ min).

Paecilin F (1). Yellow power. $C_{32}H_{32}O_{15}$; $[\alpha]_D^{25}$ –195.45 (*c* 0.044, CHCl₃); UV (MeOH) $\lambda_{\rm max}$ (log ε): 206 (2.03), 257 (1.46) and 364 (0.49) nm; IR $\nu_{\rm max}$: 3422, 2925, 1783, 1734, 1676, 1624, 1433, 1357, 1202, 1135, 800 and 722 cm⁻¹; CD (MeOH) $\lambda_{\rm max}$ (Δε): 221.5 (+20.17), 261.5 (-6.98), 313.5 (-2.59) nm; ESIMS m/z: 657.46 [M + H]⁺; HR-ESIMS m/z: 679.16003 [M + Na]⁺ (calcd for $C_{32}H_{32}$ NaO₁₅, 679.16334); ¹H and ¹³C NMR data, see Table 1.

Paecilin G (2). Yellow power. $C_{32}H_{30}O_{14}$; $[\alpha]_D^{25}$ –21.67 (*c* 0.06, CHCl₃); UV (MeOH) $\lambda_{\rm max}$ (log ε): 206 (1.65), 276 (0.83) and 360 (0.36) nm; IR $\nu_{\rm max}$: 3421, 2923, 1787, 1677, 1646, 1456, 1435, 1203, 1136, 1027, 802 and 724 cm⁻¹; CD (MeOH) $\lambda_{\rm max}$ (Δε): 214.0

(+12.62), 245.0 (-5.00), 281.5 (-2.48), 382.0 (+0.52) nm; ESIMS m/z: 639.54 [M + H]⁺; HR-ESIMS m/z: 677.12360 [M + K]⁺ (calcd for $\rm C_{32}H_{30}KO_{14}$, 677.12671). ^{1}H and ^{13}C NMR data, see Table 1.

Paecilin H (3). Yellow power. $C_{32}H_{30}O_{14}$; $[\alpha]_D^{25}$ +51.92 (c 0.052, CHCl₃); UV (MeOH) $\lambda_{\rm max}$ (log ε): 208 (2.26), 254 (1.84) and 360 (0.52) nm; IR $\nu_{\rm max}$: 3420, 2980, 1789, 1741, 1650, 1587, 1463, 1345, 1278, 1228, 1202, 1158, 1025, 998 and 736 cm⁻¹; CD (MeOH) $\lambda_{\rm max}$ (Δ ε) 211.5 (-18.83), 246.5 (+15.85), 309.0 (-3.21), 354.5 (+1.17) nm; ESIMS m/z: 639.43 [M + H]⁺; HR-ESIMS m/z: 639.16766 [M + H]⁺ (calcd for $C_{32}H_{31}O_{14}$, 639.17083). ¹H and ¹³CNMR data, see Table 1.

ECD calculations

The simplified structures 1A and 3A (Fig. S1 and S5†) were used as the model compounds of 1 and 3, respectively. Conformational analyses of 1A and 3A were performed by using the MMFF94 molecular mechanics force field. According to the relative energy within 6 kcal mol⁻¹, the molecules of **1Aa**, **1Ab**, and 3Aa showed 9 and 7 conformers with distributions higher than 1%, respectively (Fig. S2 and S6†). The resultant conformers were further optimized and checked as the true minima of potential energy surface by the density functional theory method at the B3LYP/6-31G(d) level, and 30 lowest electronic transitions were calculated. ECD spectra of different conformers were simulated using a Gaussian function with a half-bandwidth of 0.4 eV. The overall theoretical ECD spectra were given on the basis of the Boltzmann weighting of each conformers. In the 200-400 nm region, the theoretically calculated ECD and UV spectra of 1A and 3A compared with the experimental ECD data of 1 and 3, which led the assignment of the absolute configurations of C-10a and C-10a' for 1 and 3.27

Quantum chemical calculations

All theoretical calculations were carried out using the Gaussian 09 program package. Conformation search by MMFF94 molecular force fields with 6 kcal mol⁻¹ energy window limit was performed for all possible isomers, providing corresponding stable conformers with distributions higher than 1%. The conformers were optimized using the semiempirical PM6 method available in Gaussian 09 software package. The resulting geometries were further optimized at the HF/6-31G(d) and B3LYP/6-31G9d levels of theory in the gas phase. The predominant conformers were subjected to theoretical NMR calculations at the mPW1PW91/6-311G(2d,p) level of theory. The Boltzmann distributions were performed using the Gibbs free energies as weighting factors. The linear regression approach was used to convert NMR shielding tensors into NMR chemical shifts, and the NMR chemical shifts were considered as the average values of the same atoms in the different conformers. The theoretical NMR data were statistically analyzed with experimental chemical shifts, using DP4+ probability.28,29

Anti-IAV activity assays

All the metabolites were evaluated for their anti-IAV activities as previously described methods.⁴¹

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Cytotoxic activity assays

All the metabolites were tested for their cytotoxic activities as previously reported methods.42

Antibacterial activity assays

All the metabolites were tested for their antibacterial activities as previously described methods.41

Conflicts of interest

There are no conflicts to declare.

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