RSC Advances



PAPER

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Cite this: RSC Adv., 2022, 12, 13314

Highly selective cross-coupling reactions of 1,1-dibromoethylenes with alkynylaluminums for the synthesis of aryl substituted conjugated enedignes and unsymmetrical 1,3-dignes†

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A highly efficient method for the synthesis of aryl substituted conjugated enediynes and unsymmetrical 1,3-diynes via selective cross-coupling reactions of 1,1-dibromoethylenes with alkynylaluminums using the $Pd(OAc)_2$ -DPPE and $Pd_2(dba)_3$ -TFP complexes as catalysts, respectively, has been successfully developed. Though the alkyl substituted conjugated enediynes and unsymmetrical 1,3-diynes were not obtained, this case is also remarkable as the same starting materials could selectively produce either aryl substituted conjugated enediynes or unsymmetrical 1,3-diynes in moderate to excellent yields (up to 99%) in the different Pd-phosphine catalytic systems.

Received 2nd April 2022 Accepted 25th April 2022

DOI: 10.1039/d2ra02127g

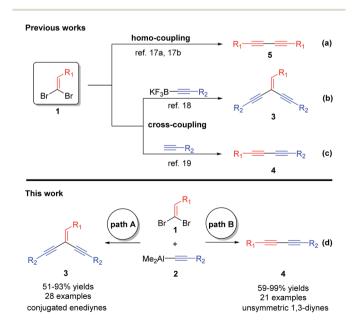
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Introduction

The conjugated enynes¹ and diynes,² which play important roles in organic synthesis, have been widely used in the preparation of natural products,³ pharmaceuticals⁴ and advanced materials.⁵ Particularly, the conjugated enediynes are usually used for synthesis of electronic and optical materials.⁶ Meanwhile, the 1,3-diynes are common structural motifs found in biologically active and pharmaceutical compounds, which are known to have anti-HIV,⁵ anticancer,⁶ antibacterial,⁶ and anti-inflammatory properties.¹⁰ Though several kinds of typical synthetic processes, including homo-¹¹ or cross-coupling,¹²,¹³ diynone decarbonylation,¹⁴ oxidative coupling¹⁵ and oxidative decarboxylative homo-coupling,¹⁰ have been realized, developing some efficient methods for accessing such frameworks from easily available organic compounds is very desirable and important.

The 1,1-dibromoethylenes, which are readily available from aldehydes, have attracted some attention due to their potentialities in the construction of conjugated enediynes and 1,3-diynes. Compared with the mature cases for constructing of symmetric conjugated 1,3-diynes, which are mostly from homocoupling of 1,1-dibromoethylenes (Scheme 1a),¹⁷ the applications of 1,1-dibromoethylenes for synthesizing of conjugated enediynes and unsymmetrical 1,3-diynes are very rare. In 2005, Kabalka *et al.* reported a highly efficient Suzuki–Miyaura

Key Laboratory of General Chemistry of the National Ethnic Affairs Commission, College of Chemistry and Environment, Southwest Minzu University, Chengdu 610041, P. R. China. E-mail: zhoulin@swun.edu.cn coupling of 1,1-dibromo-1-olefin with potassium alkynyl trifluoroborate to provide conjugated enediynes (Scheme 1b).
Besides, they have been successfully applied to the cross-coupling reaction of terminal alkynes to produce unsymmetrical conjugated 1,3-diynes (Scheme 1c).
Obviously, the cross-coupling reactions of 1,1-dibromoethylenes with organometal reagents is one of the most generally useful. However, this type of reactions has been less explored due to a complication of three competitive pathways (Scheme 1a–c). A key success of this



Scheme 1 The coupling reactions involving 1,1-dibromoethylenes 1 (a-d).

[†] Electronic supplementary information (ESI) available: Experimental details, compounds characterization and NMR spectra. See https://doi.org/10.1039/d2ra02127g

reaction relies mainly on suitable catalytic systems and/or appropriate organometallic reagents that can selectively produce either conjugated enediynes or 1,3-diynes.

To the best of our knowledge, the cross-coupling reaction of 1,1-dibromoethylenes with alkynylaluminums, which have been extensively used in organic synthesis,²⁰ has not been achieved. Herein, we would like to describe the novel Pd-phosphine complexes catalyzed selective cross-coupling reactions of 1,1-dibromoethylenes 1 with alkynylaluminums 2 to provide the aryl substituted conjugated enediynes 3 and unsymmetrical 1,3-diynes 4, respectively (Scheme 1d).

Results and discussion

In the initial study, dimethyl(phenylethynyl)aluminum **2a** and 1,1-dibromoethylene **1a** were chosen as the model substrates for the synthesis of aryl substituted conjugated enediynes *via* crosscoupling reaction.²¹ Various palladium salts were surveyed in THF at 60 °C and PdCl₂, Pd(PPh₃)₂Cl₂ or Pd(PPh₃)₄ could afford

Table 1 Optimization of the reaction conditions for the synthesis of conjugated enediynes a

		$\underline{\text{Yield}^b}$ (%)		
Entry	Metal/ligand/base	3aa	4aa	5aa
1	$PdCl_2$	17	Trace	22
2	$Pd(PPh_3)_2Cl_2$	15	Trace	26
3	$Pd(PPh_3)_4$	14	Trace	24
4	$Pd(OAc)_2$	36	Trace	16
5	$Pd_2(dba)_3$	Trace	13	Trace
6	Pd(OAc) ₂ /PPh ₃	56	Trace	7
7	Pd(OAc) ₂ /PCy ₃	45	Trace	5
8	Pd(OAc) ₂ /DPPE	70	Trace	Trace
9	Pd(OAc) ₂ /DPPP	55	Trace	10
10	Pd(OAc) ₂ /DPPE/Cs ₂ CO ₃	32	Trace	Trace
11	Pd(OAc) ₂ /DPPE/K ₃ PO ₄	80	Trace	Trace
12	$Pd(OAc)_2/DPPE/Et_3N$	75	Trace	Trace

 $[^]a$ Reaction conditions: 1a (0.5 mmol), 2a (1.0 mmol), metal (3 mol%), ligand (6 mol%), base (10 mol%), THF (1.0 mL), 60 °C, 6 h, under Ar. b Isolated yield.

3aa in lower yield, meanwhile, the homo-coupling byproduct 1,4-diphenylbuta-1,3-divne 5aa was observed (Table 1, entries 1-3). When the reaction was performed with Pd(OAc)2, an acceptable yield of 3aa (36%) was obtained, and also the homocoupling byproduct 5aa was isolated in 16% yield (Table 1, entry 4). Unexpectedly, the cross-coupling product 1,3-diyne 4aa as the major compound was achieved in 13% yield under Pd₂(dba)₃ catalysis (Table 1, entry 5). Further optimization of the reaction conditions was then aimed at exploring the efficiency of Pd(OAc)2 with various P-ligands. Among them, the diphosphine ligands benefited the reactivity (Table 1, entries 6-9). As for the backbone moiety, the 1,2-bis(diphenylphosphanyl) ethane (DPPE) exhibited a slight superiority in reactivity toward this cross-coupling compared with 1,3-bis(diphenylphosphanyl) propane (DPPP) (Table 1, entry 8 vs. 9). To further improve the conversion, the efficiency of additives was then examined (Table 1, entries 10-12). We were delighted to find that the addition of 10 mol% of K₃PO₄ as additive could improve the yield of 3aa to 80% and only trace of byproducts 4aa and 5aa were observed (Table 1, entry 11). Therefore, the optimal conditions were identified as 3 mol% of Pd(OAc)2 with 6 mol% of DPPE, 10 mol% of K₃PO₄ as additive in THF at 60 °C for 6 h.

Under the optimal conditions (Table 1, entry 11), various 1,1dibromoethylenes 1 and alkynylaluminum reagents 2 were evaluated, affording the corresponding aryl substituted conjugated enedignes 3 with moderate to good yields (up to 93%) and the trace of 1,3-diyne byproducts 4 and 5 were not isolated. As shown in Table 2, the reactivity of the cross-coupling was sensitive to the steric hindrance rather than to the electronic property of substituents on the phenyl ring of 1,1-dibromoethylenes 1. The substrates 1 with ortho-substituents gave lower yields than those with para ones (Table 2, entries 2 vs. 1, 6 vs. 5 and 8 vs. 7). Meanwhile, the fused-ring and heteroaromatic substrates (1m, 1n and 1o) were also tolerable, giving the desired products with 51% to 93% yields (Table 2, entries 11, 12, 13, 16, 19, 20, 24 and 28). On the other hand, the reactivity of this reaction was sensitive to neither the electronic properties nor the steric hindrance of substituents on the phenyl ring of alkynylaluminums 2. Generally, the desired conjugated enedivnes 3 were isolated with good to excellent yields (up to 93%) except 3fa and 3oa (53% and 51% yields, Table 2, entries 6 and 13). Moreover, the 2-thienyl substituted substrate 2g also successfully afforded the desired products with good yields (Table 2, entries 26-28).22

Inspired by the previous discovery (Table 1, entry 5), it was envisioned that the aryl substituted unsymmetrical 1,3-diynes 4 could be achieved *via* cross-coupling reaction of 1,1-dibromoethylenes 1 with alkynylaluminums 2 in the presence of Pd₂(dba)₃. Thus, we then restarted to optimize the reaction conditions of this Pd₂(dba)₃ catalyzed cross-coupling using 1a²¹ and 2a as the model substrates, respectively, in which the conjugated enediyne 3aa and homo-coupling byproduct 1,4-diphenylbuta-1,3-diyne 5aa were not determined. Performing the reaction in THF at higher temperature afforded the desired product 4aa with higher yield [Table 3, entries 2 (80 °C) *vs.* 1 (60 °C)]. To improve the reactivity, the efficiency of solvent was then examined and it was found that the polar aprotic solvents were

Table 2 Substrate scope for the synthesis of conjugated enediynes^a

(%)

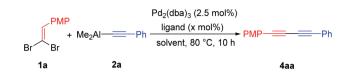
Entry	R_1	R_2	3	Yield
1	4-MeOC ₆ H ₄	Ph	3aa	80
2	2-MeOC_6H_4	Ph	3ba	61
3	$4\text{-MeC}_6\text{H}_4$	Ph	3ca	83
4	$3\text{-MeC}_6\text{H}_4$	Ph	3da	91
5	$4\text{-FC}_6\text{H}_4$	Ph	3ea	72
6	$2\text{-FC}_6\text{H}_4$	Ph	3fa	53
7	$4\text{-ClC}_6\text{H}_4$	Ph	3ga	83
8	$2,4$ - $Cl_2C_6H_3$	Ph	3ha	76
9	4 -BrC $_6$ H $_4$	Ph	3ja	75
10	$4-F_3CC_6H_4$	Ph	3la	71
11	1-Naphthyl	Ph	3ma	66
12	2-Thienyl	Ph	3na	69
13	2-Furyl	Ph	3oa	51
14	4-MeOC_6H_4	$4\text{-MeC}_6\text{H}_4$	3ab	77
15	4 -BrC $_6$ H $_4$	4-MeC_6H_4	3jb	88
16	1-Naphthyl	$4\text{-MeC}_6\text{H}_4$	3mb	74
17	4-MeOC_6H_4	$3\text{-MeC}_6\text{H}_4$	3ac	75
18	3 -BrC $_6$ H $_4$	$3\text{-MeC}_6\text{H}_4$	3kc	89
19	1-Naphthyl	$3\text{-MeC}_6\text{H}_4$	3mc	93
20	2-Thienyl	3-MeC_6H_4	3nc	69
21	$4\text{-MeC}_6\text{H}_4$	$4\text{-FC}_6\text{H}_4$	3cd	72
22	4 -BrC $_6$ H $_4$	$4\text{-FC}_6\text{H}_4$	3 jd	89
23	3 -BrC $_6$ H $_4$	$4\text{-FC}_6\text{H}_4$	3kd	84
24	2-Thienyl	$4\text{-FC}_6\text{H}_4$	3nd	83
25	4 -BrC $_6$ H $_4$	$3-FC_6H_4$	3je	66
26	$4\text{-MeOC}_6\text{H}_4$	2-Thienyl	3ag	65
27	$4\text{-MeC}_6\text{H}_4$	2-Thienyl	3cg	70
28	2-Thienyl	2-Thienyl	3ng	71

 $[^]a$ Reaction conditions: 1 (0.5 mmol), 2 (1.0 mmol), Pd(OAc)_2 (3 mol%), DPPE (6 mol%), K₃PO₄ (10 mol%), THF (1.0 mL), 60 °C, 6 h, under Ar. b Isolated yield.

beneficial (Table 3, entries 3 and 4). Further optimization of the reaction conditions was then aimed at exploring the efficiency of Pd₂(dba)₃ with various P-ligands. The addition of DPPE, which had been proved to be the most effective ligand in the synthesis of conjugated enediyne 3aa, could hardly provide the target product 4aa (Table 3, entry 5). Delightedly, when the reaction was carried out with 5.0 mol% of tri(2-furyl)phosphine (TFP, Table 1) as ligand, the desired 1,3-diyne 4aa could be isolated in 45% yield (Table 3, entry 8). Increasing the amount of TFP to 15.0 mol% could greatly improve the yield to 60% (Table 3, entry 9). The addition of 1.5 equiv. of diisopropyl ethylamine (DIPEA) could further enhance the yield to 74% (Table 3, entry 10). Thence, the optimal conditions were identified as 2.5 mol% of Pd₂(dba)₃ with 15.0 mol% of TFP, 1.5 equiv. of DIPEA in DMF at 80 °C for 10 h.

With the optimal reaction conditions in hand (Table 3, entry 10), the substrate scope of 1,1-dibromoethylenes 1 with alkynylaluminums 2 was next examined and also the conjugated enedignes 3 and homo-coupling byproducts 1,3-dignes 5 were

Table 3 Optimization of the reaction conditions for the synthesis of unsymmetrical 1,3-diynes^a



ntry	Ligand	<i>x</i> (mol%)	Solvent	Yield ^b (%)
2	_	5	THF	12
	_	5	THF	18
	_	5	DMSO	26
	_	5	DMF	28
	DPPE	5	DMF	Trace
	PPh_3	5	DMF	19
	PCy_3	5	DMF	21
	TFP	5	DMF	45
	TFP	15	DMF	60
0^d	TFP	15	DMF	74
0^d	PPh_3 PCy_3 TFP TFP	5 5 5 5 5 15	DMF DMF DMF DMF DMF DMF	28 Trace 19 21 45 60

 a Reaction conditions: 1a (0.5 mmol), 2a (0.8 mmol), Pd₂(dba)₃ (2.5 mol%), ligand (*x* mol%), solvent (3.0 mL), 80 °C, 10 h, under Ar. b Isolated yield. c Reaction was performed at 60 °C. d DIPEA (0.75 mmol) was added as additive.

Table 4 Substrate scope for the synthesis of unsymmetrical 1,3-diynes a

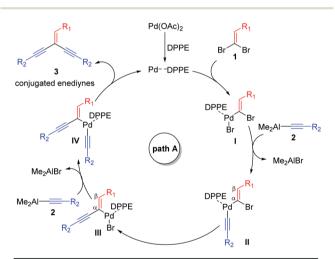
Entry	R_1	R_2	4	Yield ^b (%)
1	4-MeOC ₆ H ₄	Ph	4aa	74
2	4-MeC_6H_4	Ph	4ca	97
3	$3-MeC_6H_4$	Ph	4da	78
4	$2\text{-FC}_6\text{H}_4$	Ph	4fa	82
5	$2\text{-ClC}_6\text{H}_4$	Ph	4ia	78
6	1-Naphthyl	Ph	4ma	69
7	$4\text{-MeC}_6\text{H}_4$	$4\text{-MeC}_6\text{H}_4$	4cb	68
8	$3-MeC_6H_4$	$4\text{-MeC}_6\text{H}_4$	4db (4cc)	99
9	$4\text{-FC}_6\text{H}_4$	$4\text{-MeC}_6\text{H}_4$	4eb (4cd)	91
10	$2\text{-FC}_6\text{H}_4$	$4\text{-MeC}_6\text{H}_4$	4fb	81
11	1-Naphthyl	$4\text{-MeC}_6\text{H}_4$	4mb	66
12	2-Furyl	$4\text{-MeC}_6\text{H}_4$	4ob	59
13	4 -MeC $_6$ H $_4$	$3\text{-MeC}_6\text{H}_4$	4cc (4db)	74
14	$3-MeC_6H_4$	$3-MeC_6H_4$	4dc	89
15	$2\text{-FC}_6\text{H}_4$	$3-MeC_6H_4$	4fc (4df)	77
16	2-Furyl	$3-MeC_6H_4$	4oc	61
17	$4\text{-MeC}_6\text{H}_4$	$4\text{-FC}_6\text{H}_4$	4cd (4eb)	64
18	$3-MeC_6H_4$	$4\text{-FC}_6\text{H}_4$	4dd	69
19	$3-MeC_6H_4$	$2\text{-FC}_6\text{H}_4$	4df (4fc)	70
20	$2\text{-FC}_6\text{H}_4$	$2\text{-FC}_6\text{H}_4$	4ff	63
21	$4\text{-MeC}_6\text{H}_4$	2-Thienyl	4cg	61

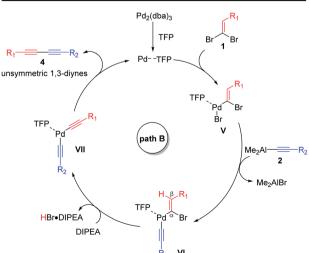
 $[^]a$ Reaction conditions: 1 (0.5 mmol), 2 (0.8 mmol), Pd₂(dba)₃ (2.5 mol%), TFP (15.0 mol%), DIPEA (0.75 mmol), DMF (3.0 mL), 80 °C, 10 h, under Ar. b Isolated yield.

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not determined. As shown in Table 4, the electronic or positional nature of the substituents either in 1,1-dibromoethylenes 1 or in alkynylaluminums 2 had nearly no effect on the efficiency of this cross-coupling reaction, affording the aryl substituted unsymmetrical 1,3-diynes 4 with good to excellent yields (up to 99%). For the fused-ring substrate 1m, the expected products 4ma and 4mb were obtained in good yields (Table 4, entries 6 and 11). Especially, the 2-thienyl or 2-furyl substituted substrates successfully afforded the unsymmetrical 1,3-diynes 4ob, 4oc and 4cg in good yields (Table 4, entries 12, 16 and 21).22

According to the previous studies on the palladium catalyzed cross-coupling reactions11-13 and our previous works about organoaluminums,20g-1 two reasonable catalytic cycles are proposed in Scheme 2. The possible mechanism of the crosscoupling for producing enedignes 3 is shown in path A. First, the oxidative additions of 1,1-dibromoethylenes 1 to Pd-DPPE complex generate the organopalladium(II) bromide intermediates I. Then, the transmetalations of alkynylaluminums 2 with intermediates I give complex intermediates II and Me₂AlBr. The intermediates II with another alkenyl bromide could isomerize into complex intermediates III via intramolecular PdII-translation. Next, the transmetalation of alkynylaluminums 2 with





Scheme 2 Proposed catalytic cycles.

intermediates III provided complex intermediates IV and another Me₂AlBr again. Finally, intermediates IV under goes reductive eliminations to afford the desired aryl substituted conjugated enediynes 3 and regenerate the active Pd-DPPE species for the next catalytic cycle. Similar to path A, the possible mechanism of the cross-coupling for producing unsymmetrical 1,3-diynes 4 is shown in path B. Oxidative additions of 1,1-dibromoethylenes 1 to Pd-TFP generate the organopalladium(II) bromide intermediates V and transmetalations of alkynylaluminums 2 with intermediates V give intermediates VI and Me₂AlBr. Elimination of the β-H in intermediates VI, in which the acidities of β -H of intermediates VI could probably be stronger than those of intermediates II so that this elimination could be promoted by an equiv. amount of DIPEA, generate intermediates VII and HBr · DIPEA. Finally, intermediates VII also under goes reductive eliminations to afford the desired aryl substituted unsymmetrical 1,3-divnes 4 and regenerate the active Pd-TFP species for the next catalytic cycle.

Conclusions

Though the specific mechanism and reason why the same starting materials could selectively produce either conjugated enediynes or unsymmetrical 1,3-diynes in analogous Pd-phosphine catalytic system were unclear, we have successfully developed a highly efficient method for the synthesis of aryl substituted conjugated enediynes and unsymmetrical 1,3diynes via selective cross-coupling reactions of 1,1-dibromoethylenes with alkynylaluminums using Pd(OAc)2-DPPE and Pd₂(dba)₃-TFP complexes as catalysts, respectively. A series of aryl substituted conjugated enediynes 3 and unsymmetrical 1,3diynes 4 have been obtained in moderate to excellent yields (up to 99%). To the best of our knowledge, there is no precedent for the application of 1,1-dibromoethylenes and alkynylaluminums in cross-coupling reaction to date. Further mechanistic studies of these selective cross-coupling reactions are still in progress.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

We are grateful to the National Natural Science Foundation of China (No. 22001219) and the Fundamental Research Funds for the Central Universities, Southwest Minzu University (No. ZYN2022030) for financial supports.

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