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Highly selective cross-coupling reactions of 1,1-dibromoethylenes with alkynylaluminums for the synthesis of aryl substituted conjugated enediynes and unsymmetrical 1,3-diynes†

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A highly efficient method for the synthesis of aryl substituted conjugated enediynes and unsymmetrical 1,3-diynes *via* selective cross-coupling reactions of 1,1-dibromoethylenes with alkynylaluminums using the Pd(OAc)₂-DPPE and Pd₂(dba)₃-TFP complexes as catalysts, respectively, has been successfully developed. Though the alkyl substituted conjugated enediynes and unsymmetrical 1,3-diynes were not obtained, this case is also remarkable as the same starting materials could selectively produce either aryl substituted conjugated enediynes or unsymmetrical 1,3-diynes in moderate to excellent yields (up to 99%) in the different Pd-phosphine catalytic systems.

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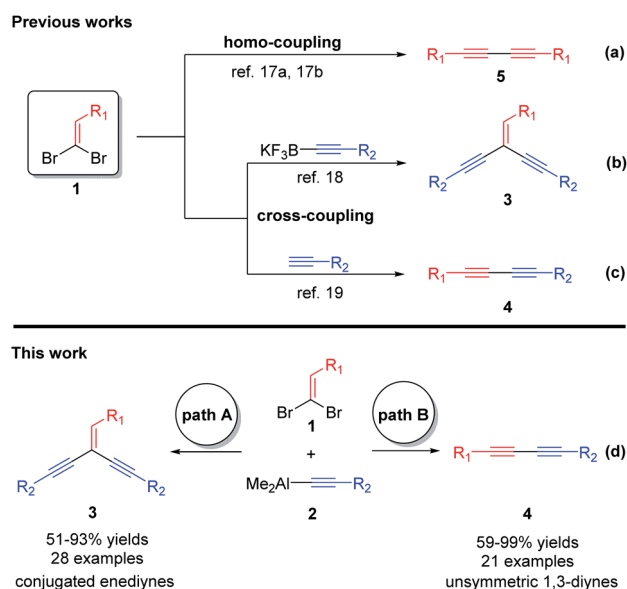
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Introduction

The conjugated enynes¹ and diynes,² which play important roles in organic synthesis, have been widely used in the preparation of natural products,³ pharmaceuticals⁴ and advanced materials.⁵ Particularly, the conjugated enediynes are usually used for synthesis of electronic and optical materials.⁶ Meanwhile, the 1,3-diynes are common structural motifs found in biologically active and pharmaceutical compounds, which are known to have anti-HIV,⁷ anticancer,⁸ antibacterial,⁹ and anti-inflammatory properties.¹⁰ Though several kinds of typical synthetic processes, including homo-¹¹ or cross-coupling,^{12,13} diynone decarbonylation,¹⁴ oxidative coupling¹⁵ and oxidative decarboxylative homo-coupling,¹⁶ have been realized, developing some efficient methods for accessing such frameworks from easily available organic compounds is very desirable and important.

The 1,1-dibromoethylenes, which are readily available from aldehydes, have attracted some attention due to their potentialities in the construction of conjugated enediynes and 1,3-diynes. Compared with the mature cases for constructing of symmetric conjugated 1,3-diynes, which are mostly from homo-coupling of 1,1-dibromoethylenes (Scheme 1a),¹⁷ the applications of 1,1-dibromoethylenes for synthesizing of conjugated enediynes and unsymmetrical 1,3-diynes are very rare. In 2005, Kabalka *et al.* reported a highly efficient Suzuki-Miyaura

coupling of 1,1-dibromo-1-olefin with potassium alkynyl trifluoroborate to provide conjugated enediynes (Scheme 1b).¹⁸ Besides, they have been successfully applied to the cross-coupling reaction of terminal alkynes to produce unsymmetrical conjugated 1,3-diynes (Scheme 1c).¹⁹ Obviously, the cross-coupling reactions of 1,1-dibromoethylenes with organometal reagents is one of the most generally useful. However, this type of reactions has been less explored due to a complication of three competitive pathways (Scheme 1a-c). A key success of this



Scheme 1 The coupling reactions involving 1,1-dibromoethylenes 1 (a-d).

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reaction relies mainly on suitable catalytic systems and/or appropriate organometallic reagents that can selectively produce either conjugated enediynes or 1,3-diyne.

To the best of our knowledge, the cross-coupling reaction of 1,1-dibromoethylenes with alkynylaluminums, which have been extensively used in organic synthesis,²⁰ has not been achieved. Herein, we would like to describe the novel Pd–phosphine complexes catalyzed selective cross-coupling reactions of 1,1-dibromoethylenes **1** with alkynylaluminums **2** to provide the aryl substituted conjugated enediynes **3** and unsymmetrical 1,3-diyne **4**, respectively (Scheme 1d).

Results and discussion

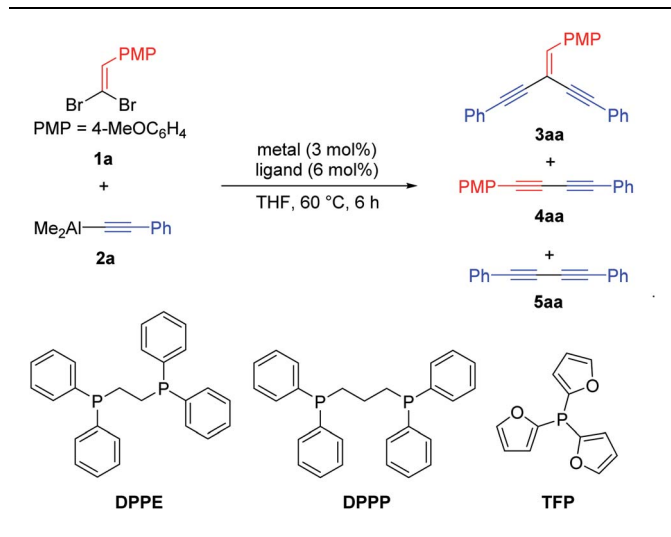
In the initial study, dimethyl(phenylethynyl)aluminum **2a** and 1,1-dibromoethylene **1a** were chosen as the model substrates for the synthesis of aryl substituted conjugated enediynes *via* cross-coupling reaction.²¹ Various palladium salts were surveyed in THF at 60 °C and PdCl₂, Pd(PPh₃)₂Cl₂ or Pd(PPh₃)₄ could afford

3aa in lower yield, meanwhile, the homo-coupling byproduct 1,4-diphenylbuta-1,3-diyne **5aa** was observed (Table 1, entries 1–3). When the reaction was performed with Pd(OAc)₂, an acceptable yield of **3aa** (36%) was obtained, and also the homo-coupling byproduct **5aa** was isolated in 16% yield (Table 1, entry 4). Unexpectedly, the cross-coupling product 1,3-diyne **4aa** as the major compound was achieved in 13% yield under Pd₂(dba)₃ catalysis (Table 1, entry 5). Further optimization of the reaction conditions was then aimed at exploring the efficiency of Pd(OAc)₂ with various P-ligands. Among them, the diphosphine ligands benefited the reactivity (Table 1, entries 6–9). As for the backbone moiety, the 1,2-bis(diphenylphosphanyl) ethane (DPPE) exhibited a slight superiority in reactivity toward this cross-coupling compared with 1,3-bis(diphenylphosphanyl) propane (DPPP) (Table 1, entry 8 *vs.* 9). To further improve the conversion, the efficiency of additives was then examined (Table 1, entries 10–12). We were delighted to find that the addition of 10 mol% of K₃PO₄ as additive could improve the yield of **3aa** to 80% and only trace of byproducts **4aa** and **5aa** were observed (Table 1, entry 11). Therefore, the optimal conditions were identified as 3 mol% of Pd(OAc)₂ with 6 mol% of DPPE, 10 mol% of K₃PO₄ as additive in THF at 60 °C for 6 h.

Under the optimal conditions (Table 1, entry 11), various 1,1-dibromoethylenes **1** and alkynylaluminum reagents **2** were evaluated, affording the corresponding aryl substituted conjugated enediynes **3** with moderate to good yields (up to 93%) and the trace of 1,3-diyne byproducts **4** and **5** were not isolated. As shown in Table 2, the reactivity of the cross-coupling was sensitive to the steric hindrance rather than to the electronic property of substituents on the phenyl ring of 1,1-dibromoethylenes **1**. The substrates **1** with *ortho*-substituents gave lower yields than those with *para* ones (Table 2, entries 2 *vs.* 1, 6 *vs.* 5 and 8 *vs.* 7). Meanwhile, the fused-ring and heteroaromatic substrates (**1m**, **1n** and **1o**) were also tolerable, giving the desired products with 51% to 93% yields (Table 2, entries 11, 12, 13, 16, 19, 20, 24 and 28). On the other hand, the reactivity of this reaction was sensitive to neither the electronic properties nor the steric hindrance of substituents on the phenyl ring of alkynylaluminums **2**. Generally, the desired conjugated enediynes **3** were isolated with good to excellent yields (up to 93%) except **3fa** and **3oa** (53% and 51% yields, Table 2, entries 6 and 13). Moreover, the 2-thienyl substituted substrate **2g** also successfully afforded the desired products with good yields (Table 2, entries 26–28).²²

Inspired by the previous discovery (Table 1, entry 5), it was envisioned that the aryl substituted unsymmetrical 1,3-diyne **4** could be achieved *via* cross-coupling reaction of 1,1-dibromoethylenes **1** with alkynylaluminums **2** in the presence of Pd₂(dba)₃. Thus, we then restarted to optimize the reaction conditions of this Pd₂(dba)₃ catalyzed cross-coupling using **1a**²¹ and **2a** as the model substrates, respectively, in which the conjugated enediyne **3aa** and homo-coupling byproduct 1,4-diphenylbuta-1,3-diyne **5aa** were not determined. Performing the reaction in THF at higher temperature afforded the desired product **4aa** with higher yield [Table 3, entries 2 (80 °C) *vs.* 1 (60 °C)]. To improve the reactivity, the efficiency of solvent was then examined and it was found that the polar aprotic solvents were

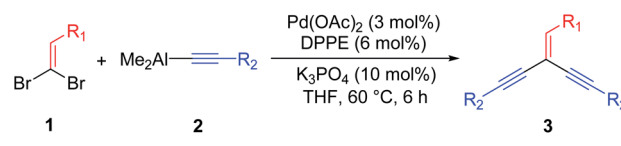
Table 1 Optimization of the reaction conditions for the synthesis of conjugated enediynes^a



Entry	Metal/ligand/base	Yield ^b (%)		
		3aa	4aa	5aa
1	PdCl ₂	17	Trace	22
2	Pd(PPh ₃) ₂ Cl ₂	15	Trace	26
3	Pd(PPh ₃) ₄	14	Trace	24
4	Pd(OAc) ₂	36	Trace	16
5	Pd ₂ (dba) ₃	Trace	13	Trace
6	Pd(OAc) ₂ /PPh ₃	56	Trace	7
7	Pd(OAc) ₂ /PCy ₃	45	Trace	5
8	Pd(OAc) ₂ /DPPE	70	Trace	Trace
9	Pd(OAc) ₂ /DPPP	55	Trace	10
10	Pd(OAc) ₂ /DPPE/Cs ₂ CO ₃	32	Trace	Trace
11	Pd(OAc) ₂ /DPPE/K ₃ PO ₄	80	Trace	Trace
12	Pd(OAc) ₂ /DPPE/Et ₃ N	75	Trace	Trace

^a Reaction conditions: **1a** (0.5 mmol), **2a** (1.0 mmol), metal (3 mol%), ligand (6 mol%), base (10 mol%), THF (1.0 mL), 60 °C, 6 h, under Ar.
^b Isolated yield.



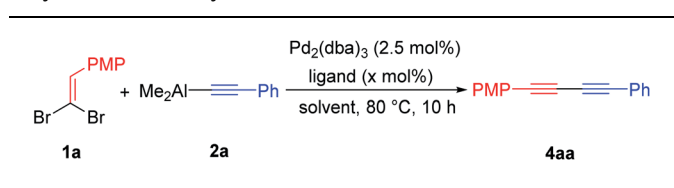
Table 2 Substrate scope for the synthesis of conjugated enediynes^a


Entry	R ₁	R ₂	3	Yield ^b (%)
1	4-MeOC ₆ H ₄	Ph	3aa	80
2	2-MeOC ₆ H ₄	Ph	3ba	61
3	4-MeC ₆ H ₄	Ph	3ca	83
4	3-MeC ₆ H ₄	Ph	3da	91
5	4-FC ₆ H ₄	Ph	3ea	72
6	2-FC ₆ H ₄	Ph	3fa	53
7	4-ClC ₆ H ₄	Ph	3ga	83
8	2,4-Cl ₂ C ₆ H ₃	Ph	3ha	76
9	4-BrC ₆ H ₄	Ph	3ja	75
10	4-F ₃ CC ₆ H ₄	Ph	3la	71
11	1-Naphthyl	Ph	3ma	66
12	2-Thienyl	Ph	3na	69
13	2-Furyl	Ph	3oa	51
14	4-MeOC ₆ H ₄	4-MeC ₆ H ₄	3ab	77
15	4-BrC ₆ H ₄	4-MeC ₆ H ₄	3jb	88
16	1-Naphthyl	4-MeC ₆ H ₄	3mb	74
17	4-MeOC ₆ H ₄	3-MeC ₆ H ₄	3ac	75
18	3-BrC ₆ H ₄	3-MeC ₆ H ₄	3kc	89
19	1-Naphthyl	3-MeC ₆ H ₄	3mc	93
20	2-Thienyl	3-MeC ₆ H ₄	3nc	69
21	4-MeC ₆ H ₄	4-FC ₆ H ₄	3cd	72
22	4-BrC ₆ H ₄	4-FC ₆ H ₄	3jd	89
23	3-BrC ₆ H ₄	4-FC ₆ H ₄	3kd	84
24	2-Thienyl	4-FC ₆ H ₄	3nd	83
25	4-BrC ₆ H ₄	3-FC ₆ H ₄	3je	66
26	4-MeOC ₆ H ₄	2-Thienyl	3ag	65
27	4-MeC ₆ H ₄	2-Thienyl	3cg	70
28	2-Thienyl	2-Thienyl	3ng	71

^a Reaction conditions: **1** (0.5 mmol), **2** (1.0 mmol), Pd(OAc)₂ (3 mol%), DPPE (6 mol%), K₃PO₄ (10 mol%), THF (1.0 mL), 60 °C, 6 h, under Ar.
^b Isolated yield.

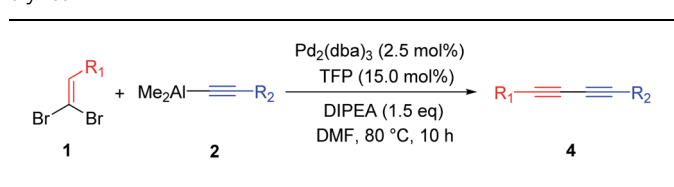
beneficial (Table 3, entries 3 and 4). Further optimization of the reaction conditions was then aimed at exploring the efficiency of Pd₂(dba)₃ with various P-ligands. The addition of DPPE, which had been proved to be the most effective ligand in the synthesis of conjugated enediyne **3aa**, could hardly provide the target product **4aa** (Table 3, entry 5). Delightedly, when the reaction was carried out with 5.0 mol% of tri(2-furyl)phosphine (TFP, Table 1) as ligand, the desired 1,3-diyne **4aa** could be isolated in 45% yield (Table 3, entry 8). Increasing the amount of TFP to 15.0 mol% could greatly improve the yield to 60% (Table 3, entry 9). The addition of 1.5 equiv. of diisopropyl ethylamine (DIPEA) could further enhance the yield to 74% (Table 3, entry 10). Thence, the optimal conditions were identified as 2.5 mol% of Pd₂(dba)₃ with 15.0 mol% of TFP, 1.5 equiv. of DIPEA in DMF at 80 °C for 10 h.

With the optimal reaction conditions in hand (Table 3, entry 10), the substrate scope of 1,1-dibromoethylenes **1** with alkylaluminums **2** was next examined and also the conjugated enediynes **3** and homo-coupling byproducts 1,3-diynes **5** were

Table 3 Optimization of the reaction conditions for the synthesis of unsymmetrical 1,3-diynes^a

Entry	Ligand	x (mol%)	Solvent	Yield ^b (%)
1 ^c	—	5	THF	12
2	—	5	THF	18
3	—	5	DMSO	26
4	—	5	DMF	28
5	DPPE	5	DMF	Trace
6	PPh ₃	5	DMF	19
7	PCy ₃	5	DMF	21
8	TFP	5	DMF	45
9	TFP	15	DMF	60
10 ^d	TFP	15	DMF	74

^a Reaction conditions: **1a** (0.5 mmol), **2a** (0.8 mmol), Pd₂(dba)₃ (2.5 mol%), ligand (x mol%), solvent (3.0 mL), 80 °C, 10 h, under Ar.
^b Isolated yield. ^c Reaction was performed at 60 °C. ^d DIPEA (0.75 mmol) was added as additive.

Table 4 Substrate scope for the synthesis of unsymmetrical 1,3-diynes^a

Entry	R ₁	R ₂	4	Yield ^b (%)
1	4-MeOC ₆ H ₄	Ph	4aa	74
2	4-MeC ₆ H ₄	Ph	4ca	97
3	3-MeC ₆ H ₄	Ph	4da	78
4	2-FC ₆ H ₄	Ph	4fa	82
5	2-ClC ₆ H ₄	Ph	4ia	78
6	1-Naphthyl	Ph	4ma	69
7	4-MeC ₆ H ₄	4-MeC ₆ H ₄	4cb	68
8	3-MeC ₆ H ₄	4-MeC ₆ H ₄	4db (4cc)	99
9	4-FC ₆ H ₄	4-MeC ₆ H ₄	4cb (4cd)	91
10	2-FC ₆ H ₄	4-MeC ₆ H ₄	4fb	81
11	1-Naphthyl	4-MeC ₆ H ₄	4mb	66
12	2-Furyl	4-MeC ₆ H ₄	4ob	59
13	4-MeC ₆ H ₄	3-MeC ₆ H ₄	4cc (4db)	74
14	3-MeC ₆ H ₄	3-MeC ₆ H ₄	4dc	89
15	2-FC ₆ H ₄	3-MeC ₆ H ₄	4fc (4df)	77
16	2-Furyl	3-MeC ₆ H ₄	4oc	61
17	4-MeC ₆ H ₄	4-FC ₆ H ₄	4cd (4eb)	64
18	3-MeC ₆ H ₄	4-FC ₆ H ₄	4dd	69
19	3-MeC ₆ H ₄	2-FC ₆ H ₄	4df (4fc)	70
20	2-FC ₆ H ₄	2-FC ₆ H ₄	4ff	63
21	4-MeC ₆ H ₄	2-Thienyl	4cg	61

^a Reaction conditions: **1** (0.5 mmol), **2** (0.8 mmol), Pd₂(dba)₃ (2.5 mol%), TFP (15.0 mol%), DIPEA (0.75 mmol), DMF (3.0 mL), 80 °C, 10 h, under Ar. ^b Isolated yield.



not determined. As shown in Table 4, the electronic or positional nature of the substituents either in 1,1-dibromoethylenes **1** or in alkynylaluminums **2** had nearly no effect on the efficiency of this cross-coupling reaction, affording the aryl substituted unsymmetrical 1,3-diynes **4** with good to excellent yields (up to 99%). For the fused-ring substrate **1m**, the expected products **4ma** and **4mb** were obtained in good yields (Table 4, entries 6 and 11). Especially, the 2-thienyl or 2-furyl substituted substrates successfully afforded the unsymmetrical 1,3-diynes **4ob**, **4oc** and **4cg** in good yields (Table 4, entries 12, 16 and 21).²²

According to the previous studies on the palladium catalyzed cross-coupling reactions^{11–13} and our previous works about organoaluminums,^{20g–l} two reasonable catalytic cycles are proposed in Scheme 2. The possible mechanism of the cross-coupling for producing enediynes **3** is shown in path A. First, the oxidative additions of 1,1-dibromoethylenes **1** to Pd–DPPE complex generate the organopalladium(II) bromide intermediates **I**. Then, the transmetalations of alkynylaluminums **2** with intermediates **I** give complex intermediates **II** and Me₂AlBr. The intermediates **II** with another alkenyl bromide could isomerize into complex intermediates **III** *via* intramolecular Pd^{II}-translocation. Next, the transmetalation of alkynylaluminums **2** with

intermediates **III** provided complex intermediates **IV** and another Me₂AlBr again. Finally, intermediates **IV** under goes reductive eliminations to afford the desired aryl substituted conjugated enediynes **3** and regenerate the active Pd–DPPE species for the next catalytic cycle. Similar to path A, the possible mechanism of the cross-coupling for producing unsymmetrical 1,3-diynes **4** is shown in path B. Oxidative additions of 1,1-dibromoethylenes **1** to Pd–TFP generate the organopalladium(II) bromide intermediates **V** and transmetalations of alkynylaluminums **2** with intermediates **V** give intermediates **VI** and Me₂AlBr. Elimination of the β-H in intermediates **VI**, in which the acidities of β-H of intermediates **VI** could probably be stronger than those of intermediates **II** so that this elimination could be promoted by an equiv. amount of DIPEA, generate intermediates **VII** and HBr·DIPEA. Finally, intermediates **VII** also under goes reductive eliminations to afford the desired aryl substituted unsymmetrical 1,3-diynes **4** and regenerate the active Pd–TFP species for the next catalytic cycle.

Conclusions

Though the specific mechanism and reason why the same starting materials could selectively produce either conjugated enediynes or unsymmetrical 1,3-diynes in analogous Pd–phosphine catalytic system were unclear, we have successfully developed a highly efficient method for the synthesis of aryl substituted conjugated enediynes and unsymmetrical 1,3-diynes *via* selective cross-coupling reactions of 1,1-dibromoethylenes with alkynylaluminums using Pd(OAc)₂–DPPE and Pd₂(dba)₃–TFP complexes as catalysts, respectively. A series of aryl substituted conjugated enediynes **3** and unsymmetrical 1,3-diynes **4** have been obtained in moderate to excellent yields (up to 99%). To the best of our knowledge, there is no precedent for the application of 1,1-dibromoethylenes and alkynylaluminums in cross-coupling reaction to date. Further mechanistic studies of these selective cross-coupling reactions are still in progress.

Conflicts of interest

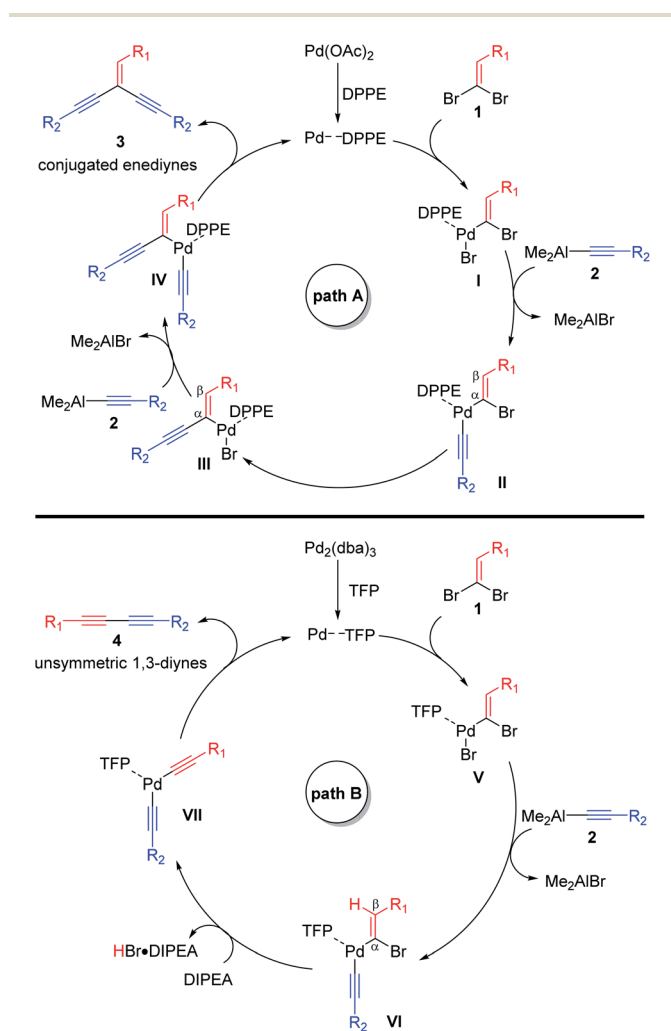
There are no conflicts to declare.

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Notes and references

- (a) K. C. Nicolaou, M. W. Dai, S. C. Tsay, V. A. Estevez and W. Wrasidlo, *Science*, 1992, **256**, 1172; (b) Y. Y. Lin, Y. J. Wang, J. H. Chen and C. F. Lee, *Synlett*, 2012, **23**, 930; (c) V. Sandrine, D. Etienne, A. Muriel, A. Corinne and P. Marc, *Adv. Synth. Catal.*, 2013, **355**, 2584; (d) H. K. H. Fong, J. M. Brunel, A. Longeon, M.-L. Bourguet-



Scheme 2 Proposed catalytic cycles.



- Kondracki, D. Barker and B. R. Copp, *Org. Biomol. Chem.*, 2017, **15**, 6194; (e) D. Qian and J. Zhang, *Acc. Chem. Res.*, 2020, **53**, 2358; (f) D. Campeau, D. F. L. Rayo, A. Mansour, K. Muratov and F. Gagosz, *Chem. Rev.*, 2021, **121**, 8756; (g) T. Zhao, X. Pu, W. Han and G. Gao, *Org. Lett.*, 2021, **23**, 1199.
- 2 (a) J. Z. Liu, J. W. Y. Lam and B. Z. Tang, *Chem. Rev.*, 2009, **109**, 5799; (b) W. Shi and A. Lei, *Tetrahedron Lett.*, 2014, **55**, 2763; (c) Y. Matsuda, S. Naoe, S. Oishi, N. Fujii and H. Ohno, *Chem.–Eur. J.*, 2015, **21**, 1463; (d) J. S. Lampkowski, D. M. Uthappa, J. F. Halonski, J. C. Maza and D. D. Young, *J. Org. Chem.*, 2016, **81**, 12520; (e) H. L. Sang, C. Wu, G. G. D. Phua and S. Ge, *ACS Catal.*, 2019, **9**, 10109; (f) K. Dhananjaya, V. Nagaraju, G. Raghuram, K. K. Arup, N. Subhashree and C. M. Chandni, *Tetrahedron Lett.*, 2020, **61**, 151775; (g) M. W. Sebastian and H. Gerhard, *Front. Chem.*, 2021, **9**, 635826.
- 3 A. L. K. Shi Shun and R. R. Tykwinski, *Angew. Chem., Int. Ed.*, 2006, **45**, 1034.
- 4 H.-Y. Cao, X.-F. Guo, X.-F. Zhu, S.-S. Li and Y.-S. Zhen, *Oncol. Rep.*, 2017, **37**, 3329.
- 5 M. Gholami and R. R. Tykwinski, *Chem. Rev.*, 2006, **106**, 4997.
- 6 G. T. Hwang, H. S. Son, J. K. Ku and B. H. Kim, *J. Am. Chem. Soc.*, 2003, **125**, 11241.
- 7 M. L. Lerch, M. K. Harper and D. J. Faulkner, *J. Nat. Prod.*, 2003, **66**, 667.
- 8 S. Morandi, F. Pellati, S. Benvenuti and F. Prati, *Tetrahedron*, 2008, **64**, 6324.
- 9 D. Lechner, M. Stavri, M. Oluwatuyi, R. Perda-Miranda and S. Gibbons, *Phytochemistry*, 2004, **65**, 331.
- 10 (a) R. Schmidt, R. Thorwirth, T. Szuppa, A. Stolle, B. Ondruschka and H. Hopf, *Chem.–Eur. J.*, 2011, **17**, 8129; (b) R. A. Xiao, R. Y. Yao and M. Z. Cai, *Eur. J. Org. Chem.*, 2012, **22**, 4178; (c) A. Narani, R. K. Marella, P. Ramudu, K. S. R. Rao and D. R. Burri, *RSC Adv.*, 2014, **4**, 3774.
- 11 (a) Z. Chen, H. Jiang, A. Wang and S. Yang, *J. Org. Chem.*, 2010, **75**, 6700; (b) C. Alexis, C. François and E. Gwillherm, *Synthesis*, 2010, **9**, 1500; (c) C. Longrui, E. L. Betsegaw, S. R. Jenna and M. James, *Green Chem.*, 2014, **16**, 1101; (d) R.-W. Orestes, C. Subrata, J. W. S. Linda, B.-E. Yehoshua and M. David, *Angew. Chem., Int. Ed.*, 2016, **55**, 6942.
- 12 (a) W. Shi, Y. D. Luo, X. C. Luo, L. Chao, H. Zhang, J. Wang and A. Lei, *J. Am. Chem. Soc.*, 2008, **130**, 14713; (b) K. Park, G. Bae, J. Moon, J. Choe, K. H. Song and S. Lee, *J. Org. Chem.*, 2010, **75**, 6244; (c) X. Feng, Z. Zhao, F. Yang, T. Jin, Y. Ma and M. Bao, *J. Organomet. Chem.*, 2011, **696**, 1479; (d) D. Saha, T. Chatterjee, M. Mukherjee and B. C. Ranu, *J. Org. Chem.*, 2012, **77**, 9379; (e) X. Jie, Y. Shang, P. Hu and W. Su, *Angew. Chem., Int. Ed.*, 2013, **52**, 3630; (f) S. Ahammed, D. Kundu and B. C. Ranu, *J. Org. Chem.*, 2014, **79**, 7391.
- 13 (a) Y. Kim, A. Park, K. Park and S. Lee, *Tetrahedron Lett.*, 2011, **52**, 1766; (b) Z. Y. Ma, X. Y. Wang, S. Y. Wei, H. L. Yang, F. W. Zhang, P. Wang, M. Xie and J. T. Ma, *Catal. Commun.*, 2013, **39**, 24; (c) M. NasrEsfahani, I. Mohammadpoor-Baltork, A. R. Khosropour, M. Moghadam, V. Mirkhani, S. Tangestaninejad, V. Agabekov and H. A. Rudbaria, *RSC Adv.*, 2014, **4**, 14291; (d) A. L. Stein, F. N. Bilheri and G. Zeni, *Chem. Commun.*, 2015, **51**, 15522; (e) X. Li, X. Xie, N. Sun and Y. Liu, *Angew. Chem., Int. Ed.*, 2017, **56**, 6994; (f) J. Schrgenhummer and M. Waser, *Org. Biomol. Chem.*, 2018, **16**, 7561.
- 14 A. Dermenci, R. E. Whittaker and G. Dong, *Org. Lett.*, 2013, **15**, 2242.
- 15 (a) M. S. Maji, S. Murarka and A. Studer, *Org. Lett.*, 2010, **12**, 3878; (b) S. L. Zhang, X. Y. Liu and T. Q. Wang, *Adv. Synth. Catal.*, 2011, **353**, 1463; (c) K. Yin, C. J. Li, J. Li and X. S. Jia, *Green Chem.*, 2011, **13**, 591; (d) F. Alonso and M. Yus, *ACS Catal.*, 2012, **2**, 1441; (e) Y. G. Zhu and Y. A. Shi, *Org. Biomol. Chem.*, 2013, **11**, 7451; (f) H. L. Li, M. Yang, X. Zhang, L. Yan, J. Li and Y. X. Qi, *New J. Chem.*, 2013, **37**, 1343; (g) H. Peng, Y. Xi, N. Ronaghi, B. Dong, N. G. Akhmedov and X. Shi, *J. Am. Chem. Soc.*, 2014, **136**, 13174; (h) Y. G. Zhu, T. Xiong, W. Y. Han and Y. A. Shi, *Org. Lett.*, 2014, **16**, 6144; (i) J. R. Suarez, D. Collado-Sanz, D. J. Cardenas and J. L. Chiara, *J. Org. Chem.*, 2015, **80**, 1098; (j) L. Su, J. Dong, L. Liu, M. Sun, R. Qiu, Y. Zhou and S.-F. Yin, *J. Am. Chem. Soc.*, 2016, **138**, 12348.
- 16 D. X. Liu, F. L. Li, H. X. Li, J. Gao and J. P. Lang, *Tetrahedron*, 2014, **70**, 2416.
- 17 (a) H. Jin and C. Kuang, *Chin. J. Chem.*, 2011, **29**, 592; (b) Z. Huang, R. Shang, Zi-R. Zhang, X.-D. Tan, X. Xiao and Y. Fu, *J. Org. Chem.*, 2013, **78**, 4551; (c) M. L. N. Rao, P. Dasgupta, B. S. Ramakrishna and V. N. Murty, *Tetrahedron Lett.*, 2014, **55**, 3529.
- 18 G. W. Kabalka, G. Dong and B. Venkataiah, *Tetrahedron Lett.*, 2005, **46**, 763.
- 19 (a) W. Shen and S. A. Thomas, *Org. Lett.*, 2000, **2**, 2857; (b) M. L. N. Rao, S. S. Islam and P. Dasgupta, *RSC Adv.*, 2015, **5**, 78090.
- 20 (a) O. V. Larionov and E. J. Corey, *Org. Lett.*, 2010, **12**, 300; (b) Q. H. Li and H. M. Gau, *Synlett*, 2012, **5**, 747; (c) D. F. Crepin and J. P. A. Harrity, *Org. Lett.*, 2013, **15**, 4222; (d) Q. H. Li, J. W. Liao, Y. L. Huang, R. T. Chiang and H. M. Gau, *Org. Biomol. Chem.*, 2014, **12**, 7634; (e) Q. H. Li, J. Y. Jeng and H. M. Gau, *Eur. J. Org. Chem.*, 2014, 7916; (f) B. Shrestha, S. Thapa, S. K. Gurung, R. A. S. Pike and R. Giri, *J. Org. Chem.*, 2016, **81**, 787; (g) S. Mo, X.-B. Shao, G. Zhang and Q.-H. Li, *RSC Adv.*, 2017, **7**, 27243; (h) Q.-H. Li, Y. Ding, G. Zhang, Z. Zhang and S. Mo, *Curr. Org. Chem.*, 2017, **14**, 462; (i) X.-B. Shao, C. Wen, G. Zhang, K. Cao, L. Wu and Q.-H. Li, *J. Organomet. Chem.*, 2018, **870**, 68; (j) X.-B. Shao, Z. Zhang, Q.-H. Li and Z.-G. Zhao, *Org. Biomol. Chem.*, 2018, **16**, 4797; (k) Q.-H. Li, X.-B. Shao, Y. Ding, C. Wen and Z.-G. Zhao, *Curr. Org. Chem.*, 2018, **22**, 1523; (l) X.-B. Shao, X. Jiang, Q.-H. Li and Z.-G. Zhao, *Tetrahedron*, 2018, **74**, 6063.
- 21 The *p*-methoxyphenyl substituted 1,1-dibromoethylene **1a** was selected as the model substrate to facilitate purification and distinguish from the homo-coupling product.
- 22 Under the optimal reaction conditions, no products were observed for alkyl or cycloalkyl substituted alkynylaluminums.

