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Strain engineering of lateral heterostructures based on group-V enes (As, Sb, Bi) for infrared optoelectronic applications calculated by first principles†

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In this work, the electronic structure, and optical properties of As/Sb and Sb/Bi lateral heterostructures (LHS) along armchair and zigzag interfaces affected by strain were investigated by density functional theory. The LHSs presented strain-dependent band transformation characteristics and sensitivity features. And a reduction and transition of the bandgap was observed when the As/Sb and Sb/Bi LHS existed under compressive strain. The density of states and the conduction band minimum-valence band maximum characteristics exhibited corresponding changes under the strain. Then a spatial charge-separation phenomenon and strong optical absorption properties in the mid-infrared range can also be observed from calculated results. Theoretical research into As/Sb and Sb/Bi LHSs has laid a solid foundation for As/Sb and Sb/Bi LHS device manufacture.

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1. Introduction

The electronic and optical properties of semiconductors are critical features that determine their performance in optoelectronic applications. Strain engineering has been explored and implemented as an efficient approach to improve the performance of photonic and electronic devices.^{1–4} The electronic, optical, mechanical, magnetic, superconducting, and thermal properties are controlled effectively through strain. For instance, band and electronic structures can be modulated under a unique strain, by an increase or decrease or induction of a direct-to-indirect (or the reverse) transition.^{5,6} Many strain sources exist, including lattice mismatch between the hetero-junction and epitaxy interfaces or deposition processes^{7–9} and thermal mismatch from the difference in thermal expansion coefficient.^{9–11} External strain, such as material bending or

compression, can also induce strain^{12–14} In addition, a reduction in dimensions also results in an increased sensitivity of the physicochemical characteristics of materials to strain.

As a novel semiconductor type, two-dimensional (2D) materials have superior electronic properties, an optical transparency, a specific surface area, and controllability.^{15–20} Moreover, 2D materials have a lower bulk defect density and stronger covalent bonding of atoms in the in-plane direction,^{20–23} compared with bulk materials. Under compressive (or tensile) strain, the lattice structure can be tuned easily and without fracture.^{24,25} This strong deformation capacity indicates great potential in strain engineering. For example, the phonon structure of 2D materials changes because of lattice structure distortion by uniaxial or biaxial strain, and tensile strain leads to a softening of the phonon mode, and the opposite behavior for compressive strain.^{25–28} Strain can change the electronic structure of single materials and, thus, increase or decrease carrier mobility. The optical properties of MoS₂ are determined from the band gap, which can be adjusted by strain, and the optical properties are reflected by changes in the PL spectrum intensity and the shift in peak position.^{29,30} Strain is a promising approach to induce and modulate the magnetic properties of monolayer MoS₂ through the competitive behavior of through-bond and through-space interactions.^{31,32} The phase transition of MoTe₂ from 2H to 1T involves overall movement of atomic positions, which can be achieved by applying strain.^{33,34} In the so-called strain state, 2D materials can exhibit extraordinary optical, thermal, electronic, and other properties because of

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changes in the distance between atoms.^{14,35–37} Therefore, an investigation of the effect of strain on optoelectronic properties, especially for 2D materials, is important, either from theoretical or experimental aspects.³⁸

Among the numerous monolayer 2D materials, a lateral monolayer heterostructure has been predicted and realized in recent years.^{38–42} A lateral heterostructure (LHS) can have different atom configuration at the interface, which induces neoteric properties. In this work, we constructed Group-V enes (As, Sb, Bi) monolayer LHSs of As/Sb and Sb/Bi. Two interface atoms that were arranged in the armchair (AC) and zigzag (ZZ) direction were considered. We used a large range of tensile and compressive strains and investigated the characteristics of the energy band and electronic structures under a strain effect for the two LHSs using a first-principles method.

2. Computational details

Density functional theory calculations were carried out by using the Vienna *Ab initio* Simulation Package (VASP).⁴³ The electronic interaction was described by the projector-enhanced wave (PAW) potential.⁴⁴ In the generalized gradient approximation (GGA), the Perdew–Burke–Ernzerhof (PBE) functional⁴⁵ was selected to deal with the exchange-related interaction of electrons. The long-range van der Waals interaction between the atomic layers was handled by the semi-empirical dispersion correction D3 scheme that was proposed by Grimme.⁴⁶ The Brillouin zone used a k-point grid sampling with uniform

intervals of $2\pi \times 0.04 \text{ \AA}^{-1}$. The cutoff energy that was selected for the plane wave foundation was 600 eV. To confirm its identity as an energy minimum, vibrational analysis was performed at each stationary point with no imaginary vibrational frequency. The convergence criteria for energy (converged to 1.0×10^{-5} eV per atom) and force (converged to 0.01 eV \AA^{-1}) were set for the geometry optimization. To avoid the interaction between adjacent layers, a vacuum spacing of 15 Å was added along the direction perpendicular to the 2D sheet. The Perdew–Burke–Ernzerhof functional was used to calculate the electronic energy band structure, density of state (DOS), light absorption spectrum, valence band maximum (VBM), and conduction band minimum (CBM) of each 2D LHS.

3. Results and discussion

The As, Sb, and Bi monolayer had a puckered honeycomb structure like black phosphorus. Fig. 1(a) shows the top view of the As, Sb or Bi monolayer structure model, in which the AC and ZZ edges of the single layer structure are visible clearly. Compressive or tensile strain application to a single-layer structure results in a slight deformation of the structure. And the related deformation features of the single-layer structure are also presented as shown in Fig. 1(b). Later, we designed and constructed the As₂/Sb₂ and Sb₂/Bi₂ LHSs along the ZZ and AC interface, where “2” represents the number of atomic columns on the two sides of the interface.⁴⁷ The top view of the LHS structure constructed along the ZZ and AC interface is shown in

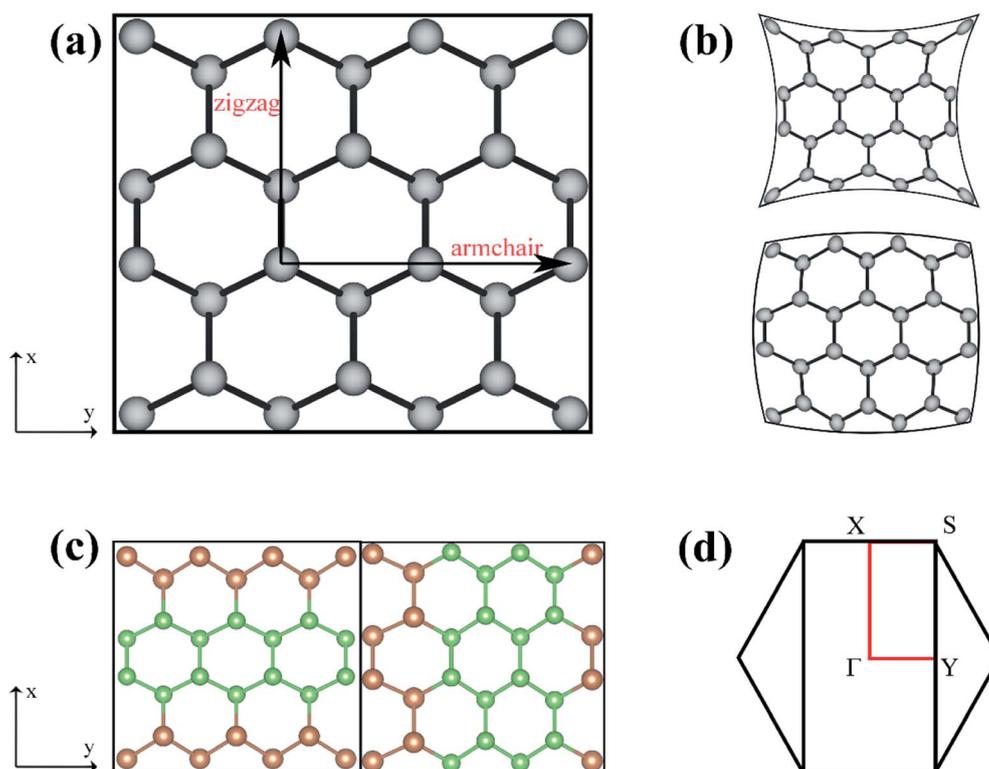


Fig. 1 (a) Top view of As, Sb, Bi single-layer structure, (b) top view of single-layer structure under compressive and tensile strains, (c) top view of lateral heterostructure structure constructed along the zigzag and armchair interface, (d) Brillouin area map of single-layer structure.



Fig. 1(c). To construct an LHS with a seamless interface, the Brillouin zone was transformed from a hexagon to a rectangle, as shown in Fig. 1(d). The Γ point represents the position of the high symmetry point in the Brillouin zone.

The formation energy (E_f) can be used as an evaluation index to ascertain the structural and energetic stability of the constructed LHSs. The E_f is calculated from:

$$E_f = \frac{E_{AB} - E_A - E_B}{16}$$

where E_A and E_B are the total energy of atom A and B, respectively (A and B represent As/Sb or Sb/Bi, respectively) and E_{AB} represents the energy of the As/Sb or Sb/Bi LHS. We calculated the formation energy of the As/Sb and Sb/Bi LHSs with and without strain (strain range: -12% – $+12\%$), and the results are shown in Fig. 2. The formation energy of both LHSs was less than zero, which indicated that the predicted As/Sb and Sb/Bi LHSs were energetically favorable and expected to be synthesized under controlled experimental conditions, even effected by large level strain. The formation energy of As/Sb LHS is less than that of Sb/Bi LHS, which indicates that As/Sb LHS is easier to synthesize than Sb/Bi LHS. When strain is applied to the LHS, the formation energy of LHS becomes larger, which shows that the strained LHS is less advantageous in energy than the unstrained.

To study the electronic properties of the LHS, we calculated the energy bands of the As/Sb and Sb/Bi LHS. Fig. 3(a) shows the trend of band gap variation with strain, and the band structures of the As/Sb and Sb/Bi LHSs that were constructed along the AC and ZZ at -4% and $+4\%$ strain are shown in Fig. 3(b–i) (the band structures at -12% , -8% , 0% , 8% , and 12% strains are shown in Fig. S1 ESI†). The main results can be divided into three parts: (I) the As/Sb LHS had a bandgap transition from indirect to direct through the stitching of different interfaces, as shown in Fig. 3(a). Under the condition of no strain, the As/Sb LHS structure that was constructed along the AC interface had an indirect bandgap of 1.1 eV, where the VBM was located at Γ point and CBM was between Γ point and X point, as shown in Fig. S1.† The As/Sb LHS structure that was constructed along the ZZ interface had a direct bandgap of 0.57 eV, and the VBM

and CBM was located at Γ point, respectively. (II) The band structure changed with the change in strain, as shown in Fig. 3(a). For a tensile strain, the AC As/Sb LHS had an indirect-to-direct bandgap transition. The ZZ As/Sb LHS had a direct-to-indirect bandgap transition in the case of a compressive strain. A similar transformation occurred for the Sb/Bi LHS. The compressive strain caused a direct-to-indirect bandgap transition of the Sb/Bi LHS. Regardless of whether a tensile or compressive strain was applied, VBM was located at Γ point, whereas CBM shifted from point Γ point to between T point and X point under a compressive strain. (III) The strain led to shifts toward a Fermi energy of the conduction and valence bands, and a reduction in the band gap of the As/Sb and Sb/Bi LHSs. When a large strain was applied, the As/Sb and Sb/Bi LHSs had a zero band gap state, and at this time, the LHS assumed a metallic state.

We conducted PDOS and CBM/VBM studies on As/Sb and Sb/Bi LHS to understand the mechanism of energy band structure change. The PDOS study illustrated the behavior of each atom in the As/Sb and Sb/Bi LHS under different strains, the position of the conduction and valence bands relative to the Fermi level, and the overall electronic properties in this type of heterostructure. The PDOS calculation results at -4% , 0% , and $+4\%$ strains are shown in Fig. 4 (the PDOS for the -12% , -8% , 0% , 8% , and 12% strains are shown in Fig. S2 in the ESI†). For the AC As/Sb LHS, the electrons in the 5p states of Sb were dominant in the conduction and valence band, no matter which type of strain applied (as shown in Fig. 4(a) and (b)), and the occupied states of the ZZ As/Sb LHS conduction bands were mainly contributed to by the 4p states of As, as shown in Fig. 4(c) and (d). The indirect-to-direct band structure transition may be caused by p-orbital hybridization of the As and Sb atoms when the LHS interface changed from the AC to the ZZ. For the Sb/Bi LHS, the electrons in the 6p states of Bi were dominant in the conduction and valence band, no matter which type of strain applied (as shown in Fig. 4(e)–(h)). After tensile strain application, the contribution of the 4p state of As in the conduction and valence bands of the As/Sb LHS increased, and the contribution of the 5p state of Sb in the conduction and valence bands of Sb/Bi LHS increased. When strain was applied to the As/Sb

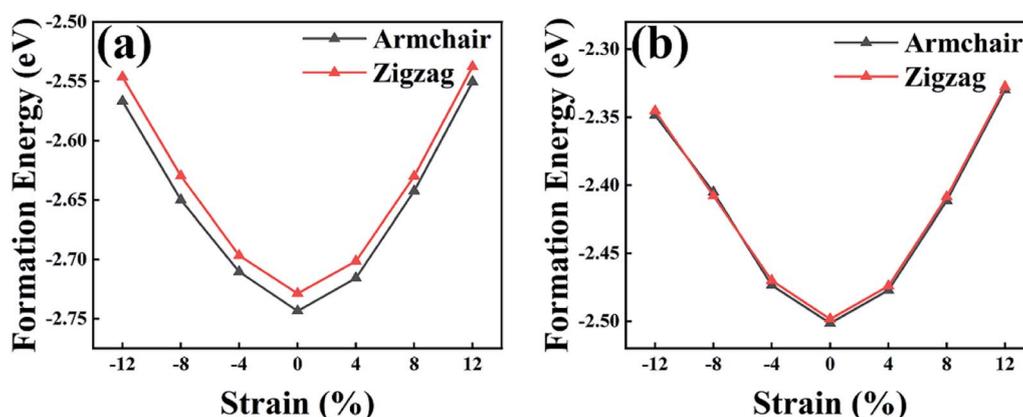


Fig. 2 Formation energy by atom for (a) As/Sb and (b) Sb/Bi lateral heterostructures with strain.



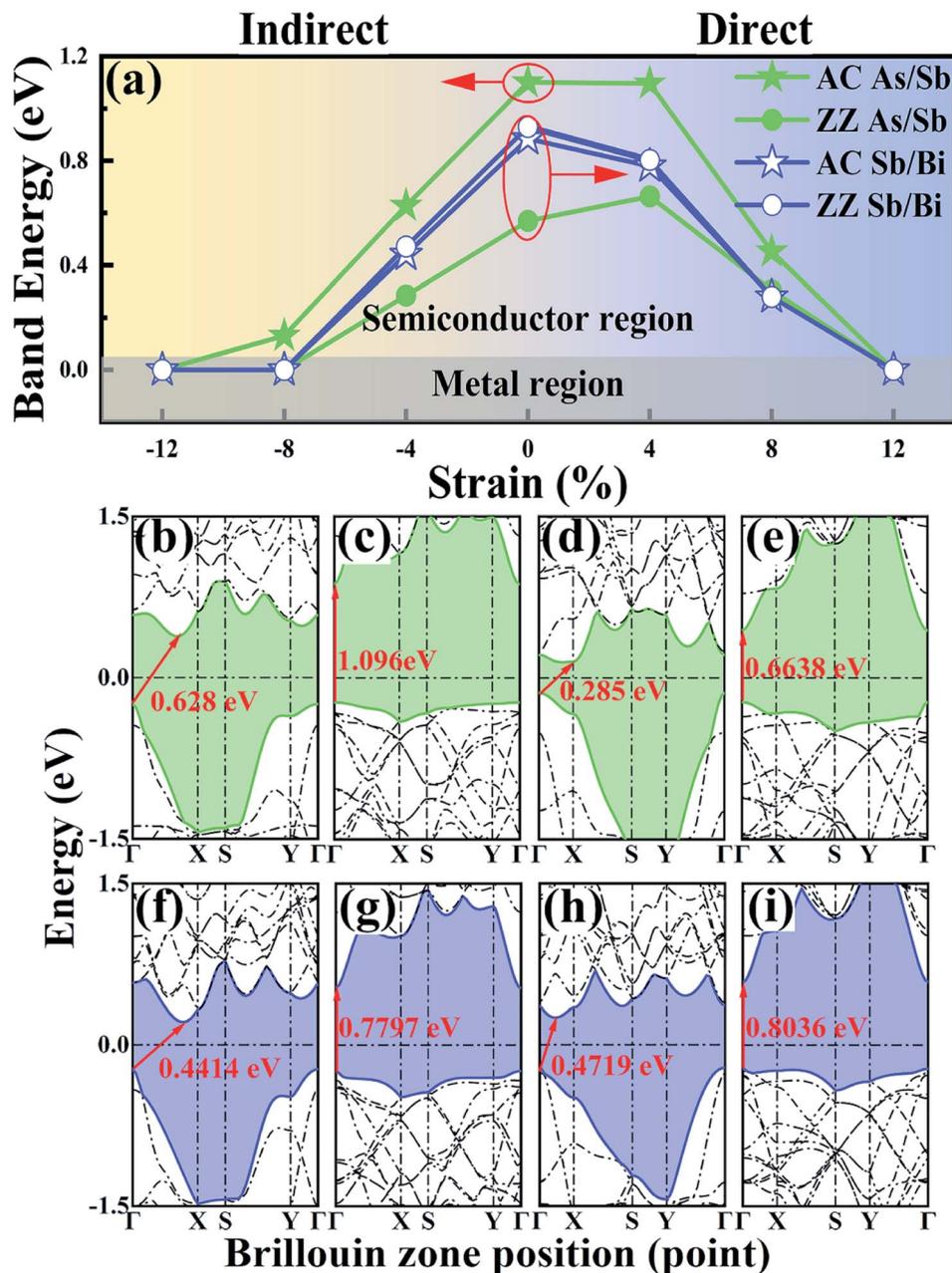


Fig. 3 (a) The change of band energy with strain. Band structure of (b) armchair (AC) As/Sb lateral heterostructure (LHS) structure at -4% strain, (c) AC As/Sb LHS structure at 4% strain, (d) zigzag (ZZ) As/Sb LHS structure at -4% strain, (e) ZZ As/Sb LHS structure at 4% strain, (f) AC Sb/Bi LHS structure at -4% strain, (g) AC Sb/Bi LHS structure at 4% strain, (h) ZZ Sb/Bi LHS structure at -4% strain, (i) ZZ Sb/Bi LHS structure at 4% strain (Fermi energy represented by a dash-dotted line and set to zero).

and Sb/Bi LHS, the state moved to the Fermi level, which resulted in a gradual decrease in band gap and an increase in local density of states at the Fermi level. At a strain of -12% , -8% , or 12% , the atomic orbital passed through the Fermi level, and the As/Sb and Sb/Bi LHS band gap was reduced to zero, which shows the characteristics of the metal.

Atoms in the As/Sb and Sb/Bi LHSs showed a strong interaction because of the covalent bond connection; therefore, the electronic state between the interfaces was the focus of research. To illustrate the charge distribution on the LHS, we

plotted the VBM and CBM 2D band-decomposed charge densities of the As/Sb and Sb/Bi LHS under different levels of strain, as shown in Fig. 5. For metallic state of LHS (applied with large levels of strain), when compressive strain is applied, the state of As/Sb LHS is distributed around Sb atoms, and the state of Sb/Bi LHS is distributed around Bi atoms; on the contrary, the state of LHS The states are distributed around two kinds of atoms when tensile strain is applied. The VBM states for the As/Sb and Sb/Bi LHSs were distributed at the bonds, whereas the CBM states were distributed around the atoms of LHSs. After



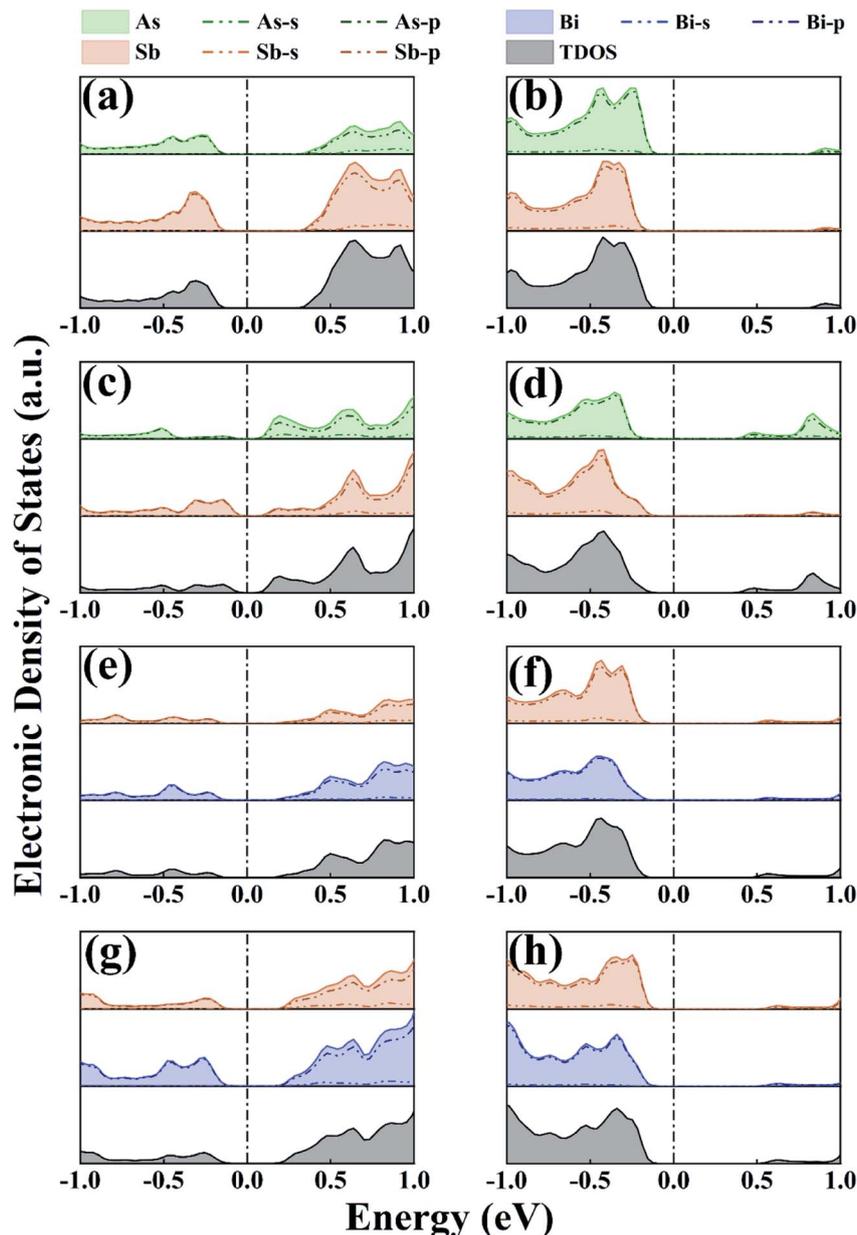


Fig. 4 PDOS for (a) armchair (AC) As/Sb lateral heterostructure (LHS) at -4% strain, (b) AC As/Sb LHS at 4% strain, (c) zigzag (ZZ) As/Sb LHS at -4% strain, (d) ZZ As/Sb LHS at 4% strain, (e) AC Sb/Bi LHS at -4% strain, (f) AC Sb/Bi LHS at 4% strain, (g) ZZ Sb/Bi LHS at -4% strain, (h) ZZ Sb/Bi LHS at 4% strain.

tensile strain application, the VBM state of the AC As/Sb LHS changed to being localized around the As monolayers, whereas the CBM state of the AC As/Sb LHS changed to being distributed around the As and Sb atoms. At a $+8\%$ strain, the VBM state of the ZZ As/Sb LHS changed to being located around the As monolayers. A similar situation occurred for the Sb/Bi LHS. Under a tensile strain, the VBM state of the Sb/Bi LHS changed to being localized around the Sb monolayers.

Previous research showed that the optical absorption rate can be used to characterize the optical properties of any system that contains 2D sheets. Fig. 6 shows the relationship between the absorption spectra of the As/Sb and Sb/Bi LHSs with

wavelength. As shown in Fig. 6(a) and (b), the optical absorption, of As/Sb LHS, in the YY direction was weak compared with that in the XX direction. The opposite happened for the Sb/Bi LHS, where the optical absorption in the YY direction was stronger compared with that in the XX direction, as shown in Fig. 6(c) and (d). Compared with the optical absorption of the LHS at no strain, the absorption spectra of LHS changed in scope and intensity with the change in strain. When a compressive strain was applied to the Sb/Bi and AC As/Sb LHSs, the light absorption intensity increased, and the absorption edge moved to a short-wave direction. However, when a tensile strain was applied to the Sb/Bi and AC As/Sb



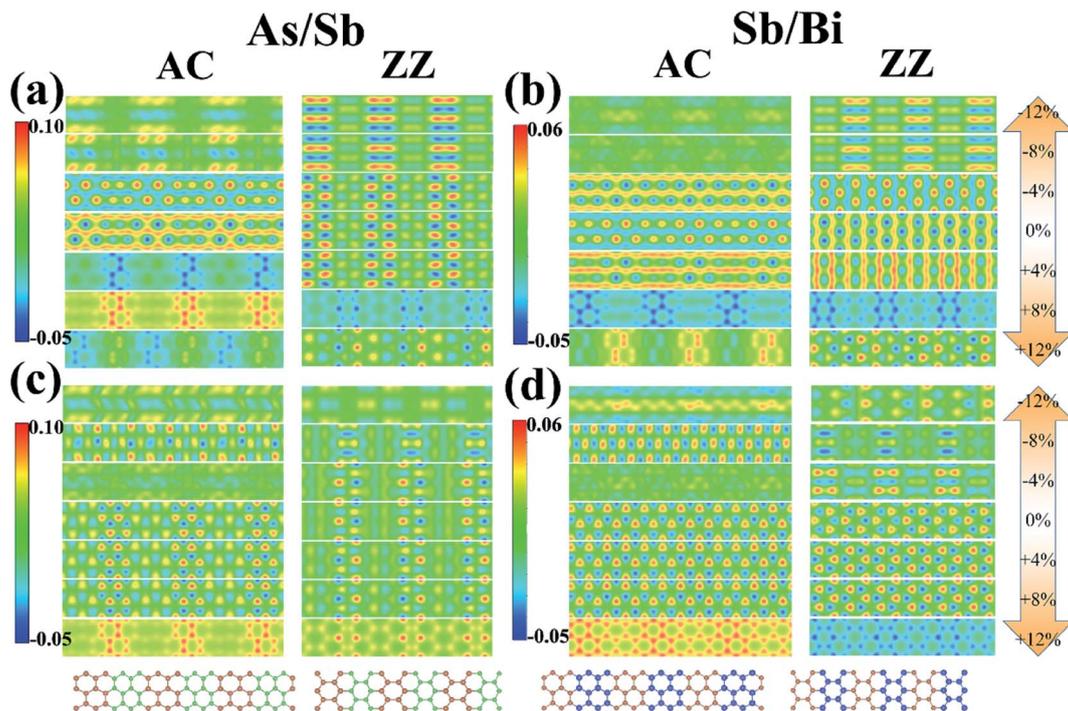


Fig. 5 (a) valence band maximum (VBM) of As/Sb lateral heterostructure (LHS), (b) VBM of Sb/Bi LHS, (c) conduction band minimum (CBM) of As/Sb LHS, (d) CBM of Sb/Bi LHS.

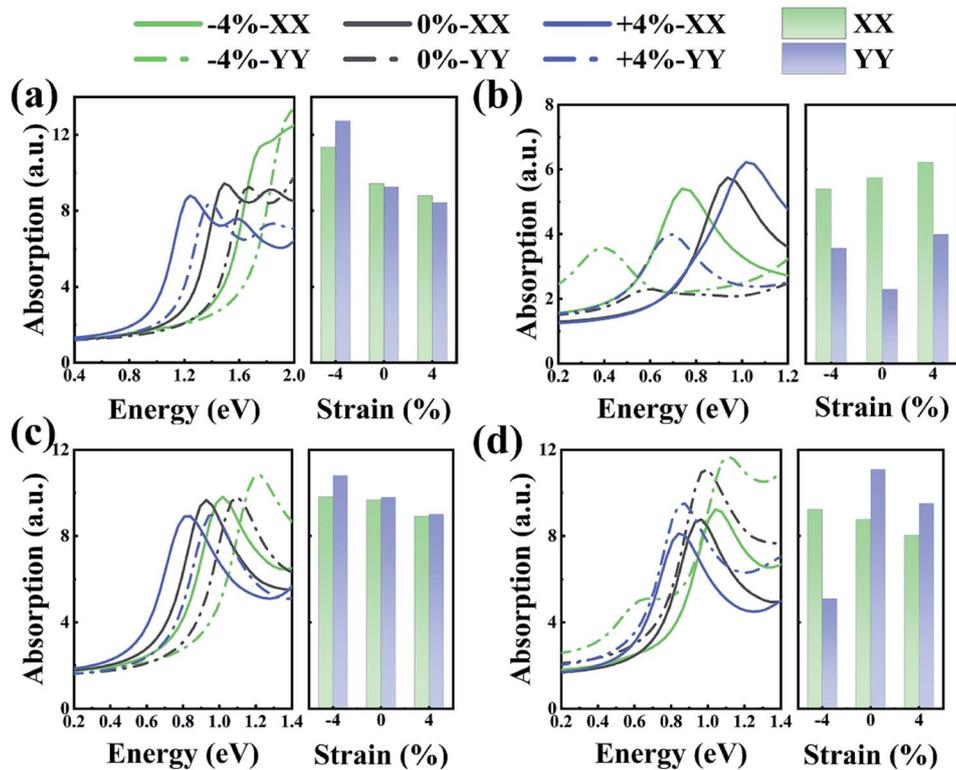


Fig. 6 Light absorption at different strains for (a) armchair (AC) As/Sb lateral heterostructure (LHS), (b) zigzag (ZZ) As/Sb LHS, (c) AC Sb/Bi LHS, (d) ZZ Sb/Bi LHS.



LHSs, the light absorption intensity decreased, and the absorption edge moved to a long wave direction. These results indicate that As/Sb and Sb/Bi LHSs have potential application in optical elements.

4. Conclusion

The structural, formation energy, electronic, and optical properties of As/Sb and Sb/Bi LHSs were investigated through density functional theory. We examined the As/Sb and Sb/Bi LHS formation energy properties of two interface structures at different strain levels. The predicted As/Sb and Sb/Bi LHSs were energetically favorable and are expected to be synthesized under controlled experimental conditions. An evaluation of the external strain for As/Sb and Sb/Bi validates that the bandgap can be engineered by the compressive or tensile strain. DOS, PDOS, and CBM/VBM studies on the As/Sb and Sb/Bi LHSs showed that the reason for the change in energy band structure was the change in electronic state. The calculated optical properties showed that the absorption rate of the spectrum changed in range and intensity with the change in strain. Our theoretical studies show that the constructed As/Sb and Sb/Bi LHSs have potential applications in nanoelectronics and optical devices.

Conflicts of interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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References

- Z. H. Dai, L. Q. Liu and Z. Zhang, *Adv. Mater.*, 2019, **31**, 1805417.
- J. L. Du, H. H. Yu, B. S. Liu, M. Y. Hong, Q. L. Liao, Z. Zhang and Y. Zhang, *Small Methods*, 2021, **5**, 2000919.
- S. X. Yang, Y. J. Chen and C. B. Jiang, *InfoMat*, 2021, **3**, 397–420.
- B. W. Zhang, D. Fang, X. Fang, H. B. Zhao, D. K. Wang, J. H. Li, X. H. Wang and D. B. Wang, *Rare Metals*, 2021, **41**, 982–991.
- Y. H. Wang, J. B. Pang, Q. L. Cheng, L. Han, Y. F. Li, X. Meng, B. Ibarlucea, H. B. Zhao, F. Yang, H. Y. Liu, H. Liu, W. J. Zhou, X. Wang, M. H. Rummeli, Y. Zhang and G. Cuniberti, *Nano-Micro Lett.*, 2021, **13**, 52.
- Q. Pei, X. C. Wang, J. J. Zou and W. B. Mi, *Nanotechnology*, 2018, **29**, 9.
- A. Ohtake, T. Mano and Y. Sakuma, *Sci. Rep.*, 2020, **10**, 7.
- E. Koo, Y. Lee, Y. Song, M. Park and S. Y. Ju, *ACS Appl. Electron. Mater.*, 2019, **1**, 113–121.
- A. Moridi, H. H. Ruan, L. C. Zhang and M. Liu, *Int. J. Solids Struct.*, 2013, **50**, 3562–3569.
- E. T. Ritz and N. A. Benedek, *Phys. Rev. Mater.*, 2020, **4**, 7.
- F. Wang, B. Zhou, H. M. Sun, A. Y. Cui, T. Jiang, L. P. Xu, K. Jiang, L. Y. Shang, Z. G. Hu and J. H. Chu, *Phys. Rev. B*, 2018, **98**, 9.
- W. Wu, J. Wang, P. Ercius, N. C. Wright, D. M. Leppert-Simenauer, R. A. Burke, M. Dubey, A. M. Dogare and M. T. Pettes, *Nano Lett.*, 2018, **18**, 2351–2357.
- J. Liang, J. Zhang, Z. Li, H. Hong, J. Wang, Z. Zhang, X. Zhou, R. Qiao, J. Xu and P. Gao, *Nano Lett.*, 2017, **17**, 7539–7543.
- S. Yang, C. Wang, H. Sahin, H. Chen, Y. Li, S. S. Li, A. Suslu, F. M. Peeters, Q. Liu and J. Li, *Nano Lett.*, 2015, **15**, 1660–1666.
- K. S. Novoselov, A. K. Geim, S. V. Morozov, D. Jiang, Y. Zhang, S. V. Dubonos, I. V. Grigorieva and A. A. Firsov, *Science*, 2004, **306**, 666–669.
- B. Radisavljevic, A. Radenovic, J. Brivio, V. Giacometti and A. Kis, *Nat. Nanotechnol.*, 2011, **6**, 147–150.
- L. Li, Y. Yu, G. J. Ye, Q. Ge, X. Ou, H. Wu, D. Feng, X. H. Chen and Y. Zhang, *Nat. Nanotechnol.*, 2014, **9**, 372–377.
- M. Dávila, L. Xian, S. Cahangirov, A. Rubio and G. L. Lay, *New J. Phys.*, 2014, **16**, 3579–3587.
- M. Pumera and Z. Sofer, *Adv. Mater.*, 2017, **29**, 1605299.
- C. V. Nguyen, *Phys. Rev. B*, 2021, **103**, 115429.
- G. G. Naumis, S. Barraza-Lopez, M. Oliva-Leyva and H. Terrones, *Rep. Prog. Phys.*, 2017, **80**, 1–62.
- M. A. Bissett, M. Tsuji and H. Ago, *Phys. Chem. Chem. Phys.*, 2014, **16**, 11124–11138.
- H. J. Jiang, L. Zheng, Z. Liu and X. W. Wang, *InfoMat*, 2020, **2**, 1077–1094.
- G. Cocco, E. Cadelano and L. Colombo, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2010, **81**, 2010–2020.
- C. Nguyen, N. V. Hoang, H. V. Phuc, A. Y. Sin and C. V. Nguyen, *J. Phy. Chem. Lett.*, 2021, **12**, 5076–5084.
- Y. Li, Z. Hu, S. Lin, S. K. Lai, J. Wei and P. L. Shu, *Adv. Funct. Mater.*, 2017, **27**, 1600986.
- T. M. G. Mohiuddin, A. Lombardo, R. R. Nair, A. Bonetti, G. Savini, R. Jalil, N. Bonini, D. M. Basko, C. Galiotis and N. Marzari, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2008, **79**, 205433.
- D. Yoon, Y. W. Son and H. Cheong, *Phys. Rev. Lett.*, 2011, **106**, 155502.
- K. He, C. Poole, K. F. Mak and J. Shan, *Nano Lett.*, 2013, **13**, 2931–2936.
- H. J. Conley, B. Wang, J. I. Ziegler, R. F. Haglund, S. T. Pantelides and K. I. Bolotin, *Nano Lett.*, 2013, **13**, 3626–3630.
- W. S. Yun and J. D. Lee, *J. Phys. Chem. C*, 2015, **119**, 2822–2827.



- 32 W. Xu, S. M. Yan and W. Qiao, *RSC Adv.*, 2018, **8**, 8435–8441.
- 33 S. Song, D. H. Keum, S. Cho, D. Perello, Y. Kim and Y. H. Lee, *Nano Lett.*, 2015, **16**, 188–193.
- 34 Y. Zhao, Y. Li, M. Liu, K. Xu and F. Ma, *J. Phys. Chem. C*, 2020, **124**, 4299–4307.
- 35 S. B. Desai, G. Seol, J. S. Kang, H. Fang and A. Javey, *Nano Lett.*, 2014, **14**, 4592–4597.
- 36 Y. L. Wang, C. X. Cong, W. H. Yang, J. Z. Shang, N. Peimyoo, Y. Chen, J. Y. Kang, J. P. Wang, W. Huang and T. Yu, *Nano Res.*, 2015, **8**, 2562–2572.
- 37 S. Yang, Y. Liu, M. Wu, L. D. Zhao, Z. Lin, H. C. Cheng, Y. Wang, C. Jiang, S. H. Wei and L. Huang, *Nano Res.*, 2018, **011**, 554–564.
- 38 C. Q. Nguyen, Y. S. Ang, S.-T. Nguyen, N. V. Hoang, N. M. Hung and C. V. Nguyen, *Phys. Rev. B*, 2022, **105**, 045303.
- 39 Y. Zhao, S. Tan and G. Ouyang, *J. Phys. D: Appl. Phys.*, 2021, **54**, 145107.
- 40 T. Huang, J.-C. Lian, K. Yang, Y. Si, H.-Y. Wu, W.-Q. Huang, W. Hu and G.-F. Huang, *Phys. E*, 2020, **118**, 113962.
- 41 J. Yuan, N. Yu, J. Wang, K.-H. Xue and X. Miao, *Appl. Surf. Sci.*, 2018, **436**, 919–926.
- 42 Z. Li, J. Zheng, Y. Zhang, C. Zheng, W.-Y. Woon, M.-C. Chuang, H.-C. Tsai, C.-H. Chen, A. Davis, Z.-Q. Xu, J. Lin, H. Zhang and Q. Bao, *ACS Appl. Mater. Interfaces*, 2017, **9**, 34204–34212.
- 43 G. G. Kresse and J. J. Furthmüller, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1996, **54**, 11169.
- 44 G. Kresse and D. Joubert, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1999, **59**, 1758–1775.
- 45 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865–3868.
- 46 S. Grimme, J. Antony, S. Ehrlich and H. Krieg, *J. Chem. Phys.*, 2010, **132**, 154104.
- 47 W. J. Li, X. Fang, D. K. Wang, F. Tian, H. Z. Wang, D. Fang, J. H. Li, X. Y. Chu, H. B. Zhao, D. B. Wang and X. H. Ma, *Phys. E*, 2021, **134**, 114933.

