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Transition-metal-catalyzed remote C–H functionalization of thioethers

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In the last decade, transition-metal-catalyzed direct C–H bond functionalization has been recognized as one of most efficient approaches for the derivatization of thioethers. Within this category, both mono- and bidentate-directing group strategies achieved the remote C(sp²)–H and C(sp³)–H functionalization of thioethers, respectively. This review systematically introduces the major advances and their mechanisms in the field of transition-metal-catalyzed remote C–H functionalization of thioethers from 2010 to 2021.

1. Introduction

In the last decade, transition-metal-catalyzed C–H bond functionalization stands out as one of the most efficient and economical approaches for the C–C bond or C–X bond constructions.¹ Different kinds of mono- and bidentate-directing groups have been employed to control selectivity and enhance the efficiency of C–H bond functionalization.^{2,3} In this field, remote C–H bond functionalization is performed by using transition-metal catalysts and thioether directing groups. In this process, the transition-metals can coordinate with the internal sulfur atoms to achieve this issue. More importantly, as a traceless directing group, the thioether group is easily

removed or transformed to other useful functional groups, which dramatically improves the practicability of this method.⁴

Meanwhile, thioethers are also important ubiquitous skeletons in a variety of pharmaceutical drugs, natural products, and organic materials.⁵ Furthermore, thioethers can also be used as transition-metal ligands and organo-catalysts in synthetic chemistry.⁶ However, previous reviews on thioethers mainly focused on the C–S bond activation and transformations.⁷ In comparison, only a few related reviews highlighted the remote C–H functionalization of thioethers. In 2019, the Sun group introduced different S-containing directing groups, including thioether, thioamide, sulfoxide, thioketone and alkoxythiocarbonyl, in transition metal-catalyzed C–H functionalization reactions.^{8a} Very recently, Zhao group also provided a similar review.^{8b} However, the above two reviews only briefly describe the recent advances on thioethers, and there are no detailed mechanism discussions. In addition, the direct α -C(sp³)–H functionalization of thioethers also represents another important strategy to prepare novel thioethers.⁹

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In this review, we will systematically introduce and discuss the significant progress made in the field of transition-metal-catalyzed remote C–H functionalization of thioethers by using thioethers themselves as directing groups from 2010 to 2021. Meanwhile, the detailed mechanisms for these reactions are also disclosed. Furthermore, thioether-directed asymmetric C–H bond functionalization is firstly introduced.

2. Remote C(sp²)-H functionalization

Transition metal catalyzed/mediated selective functionalization of either C(sp²)-H or C(sp³)-H bonds by employing thioethers as directing groups *via* coordination of the internal sulfur atom to transition metal species has been well developed. This strategy enables direct remote C–H bond functionalization of thioethers to efficiently construct a series of novel thioether compounds. In 2012, the Zhang group reported a thioether-directed Pd(II)-catalyzed C(sp²)-H olefination reaction of unactivated arenes (Scheme 1a).¹⁰ This is the first example of thioether utilization as a sole guiding group. Different acrylates were selectively incorporated into thioether compounds in moderate to good yields. In this work, an acetate-bridged dinuclear cyclopalladation intermediate **1A** was isolated and

had its structure confirmed by single crystal X-ray diffraction, providing strong evidence on directed C–H bond activation. Furthermore, mechanistic studies suggest that the C(sp²)-H cleavage process may be a turnover-limiting step (Scheme 1b). This study proposes a plausible catalytic cycle involving Pd(II)/Pd(0) catalysis for this transformation (Scheme 1c). The thioether-directed C–H bond cleavage of **1a** with the Pd(OAc)₂ catalyst provides the dinuclear cyclopalladation intermediate **1A**. Next, alkene insertion takes place to produce the carbo-palladation intermediate **1B**. Then, β-hydride elimination of the intermediate **1B** and subsequent reductive elimination afford the desired product **3a** and regenerate the Pd(0) catalyst. Finally, the Pd(0) catalyst is re-oxidized to Pd(II) by AgOTFA.

On the basis of the aforementioned research, a thioether-directed C(sp²)-H arylation reaction of unactivated arenes catalyzed by a palladium(II) catalyst was also reported by the same group (Scheme 2a).¹¹ Various potassium aryltrifluoroborates were proven to be suitable coupling partners, leading to desired biaryl thioether derivatives. Notably, *p*-benzoquinone (BQ) was shown to be an important additive for the reductive elimination step in this reaction. Subsequently, they demonstrated an efficient Pd(II)-catalyzed acetoxylation of aromatic C(sp²)-H bonds by employing thioether as a sole



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Xiaoqiang Sun obtained his Ph.D. degree from Nanjing University in 1988. Then he became a postdoctoral fellow at the University of Sheffield and University of Birmingham under the supervision of Prof. J. Fraser Stoddart. After that, he became a Professor at Changzhou University in 1997. His current research interests involve pharmaceutical design and synthesis, and supramolecular chemistry.



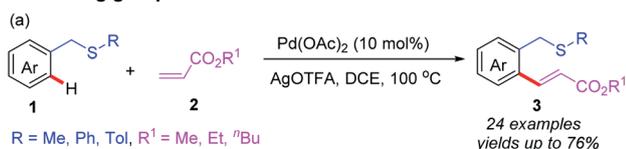
Long Tang received his master's degree in Organic Chemistry from Changzhou University. Since 2019, he began to study for a doctorate at the Chinese Academy of Forestry. At present, his research interest involves biomass organic synthetic chemistry.



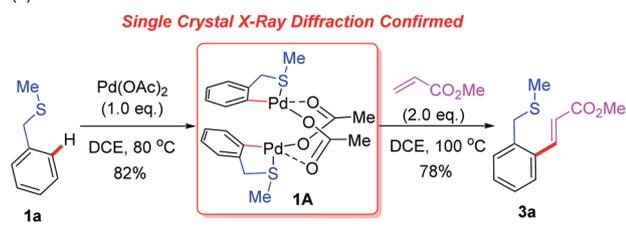
Ke Yang obtained his Ph.D. degree from Nanjing University in 2014. After that, he became a postdoctoral fellow at Nanjing University with Professor Guigen Li. In 2015, he joined in the group of Professor Haibo Ge at the Department of Chemistry and Chemical Biology, Indiana University – Purdue University Indianapolis (IUPUI) in USA. In 2017, he moved to Changzhou University and started his independent academic career. His research interests involve novel transition metal catalysis, green organic synthesis and pharmaceutical chemistry.



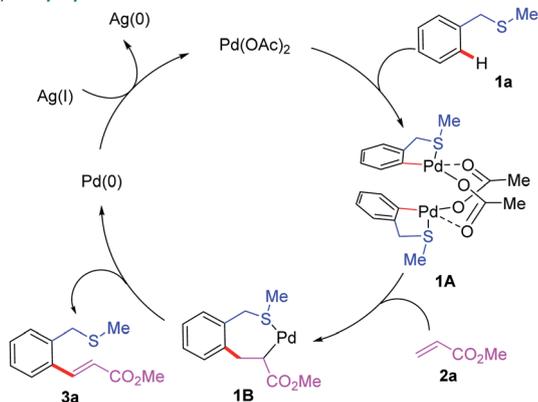
The Zhang group work:



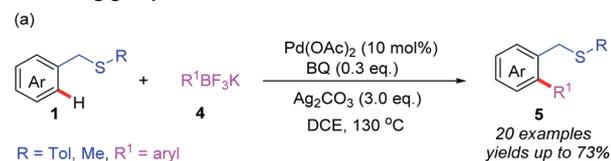
(b)



(c) The proposed mechanism

Scheme 1 (a) Thioether-directed Pd(II)-catalyzed C(sp²)-H olefination of unactivated arenes. (b) Single crystal X-ray diffraction experiment. (c) The proposed mechanism.

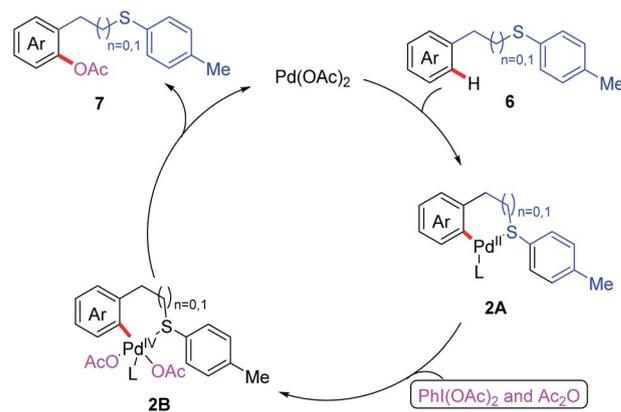
The Zhang group work:



(b)

Scheme 2 (a) Thioether-directed Pd(II)-catalyzed C(sp²)-H arylation of unactivated arenes. (b) Thioether-directed Pd(II)-catalyzed C(sp²)-H acetoxylation of unactivated arenes.

directing group (Scheme 2b).¹² Both benzyl and phenethyl sulfides reacted effectively with PhI(OAc)₂ and Ac₂O to provide the desired acetoxylation products in moderate to good yields. Furthermore, a plausible catalytic cycle involving Pd(II)/Pd(IV) catalysis is also proposed in Scheme 3. First, the coordination of

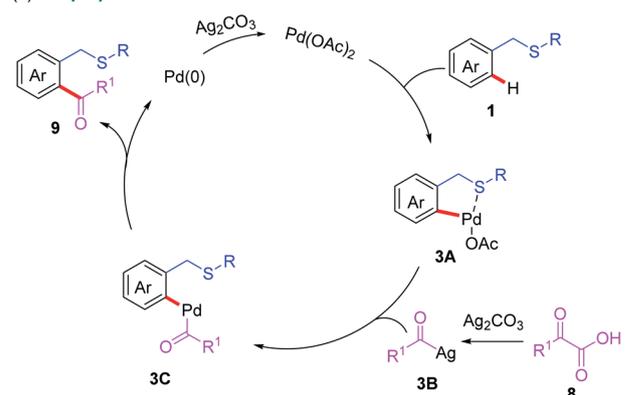
Scheme 3 Proposed mechanism of thioether-directed Pd(II)-catalyzed C(sp²)-H acetoxylation reaction.

substrate **6** to Pd(OAc)₂ followed by a thioether-directed C-H bond activation produces the Pd(II) species **2A**. Next, the oxidative addition of the intermediate **2A** with PhI(OAc)₂ and

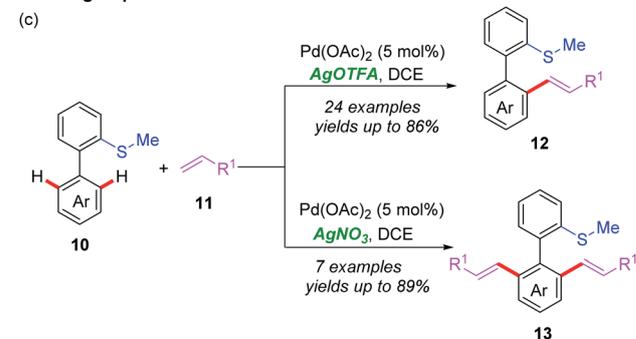
The Kuang group work:



(b) The proposed mechanism



The Shi group work:

Scheme 4 (a) Pd(II)-catalyzed C(sp²)-H acylation reactions directed by a thioether group. (b) The proposed mechanism of acylation. (c) Pd(II)-catalyzed C(sp²)-H alkenylation reactions directed by a thioether group.

Ac₂O generates the Pd(IV) complex **2B**. Finally, the reductive elimination of intermediate **2B** affords the desired product **7** and releases the active Pd(II) species.

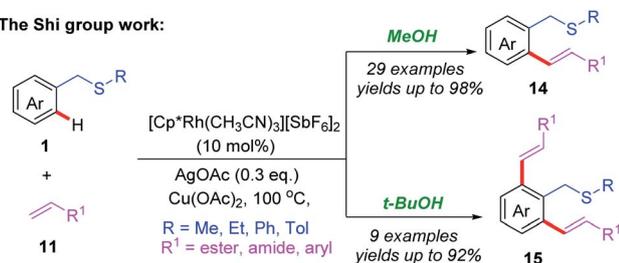
Moreover, the Kuang group also reported a similar protocol to achieve Pd(II)-catalyzed C(sp²)-H acylation directed by a thioether group (Scheme 4a).¹³ In this reaction, thioethers bearing an electron-donating (Me, *i*-Pr, or MeO) or electron-withdrawing (F or Cl) group were well-tolerated and provided the corresponding acylated products with good yields. However, only trace products were detected with thioethers containing a strong electron-withdrawing (NO₂ or CN) group. In addition, a plausible catalytic cycle is also shown in Scheme 4b. First, the thioether-directed coordination of substrate **1** to Pd(OAc)₂ with a subsequent cyclopalladation process afford the Pd(II) complex **3A**. Meanwhile, a silver-mediated decarboxylation of α -oxo-carboxylic acid **8** produces acyl silver species **3B**. After that, transmetalation of the Pd(II) complex **3A** with the acyl silver species **3B** gives the Pd(II) complex **3C**. Reductive elimination of **3C** generates the desired product **9**, and the Pd(0) species. Finally, re-oxidation of the Pd(0) species by Ag₂CO₃ regenerates the Pd(II) catalyst. In addition, the Shi group also reported

a thioether-directed and oxidant-controlled selective mono- or di-alkenylation of C(sp²)-H bonds of biaryl compounds (Scheme 4b).¹⁴ When AgOTFA was employed as an oxidant, mono-alkenylated products were obtained in moderate to good yields, whereas di-alkenylated products were generated with AgNO₃ as an oxidant instead.

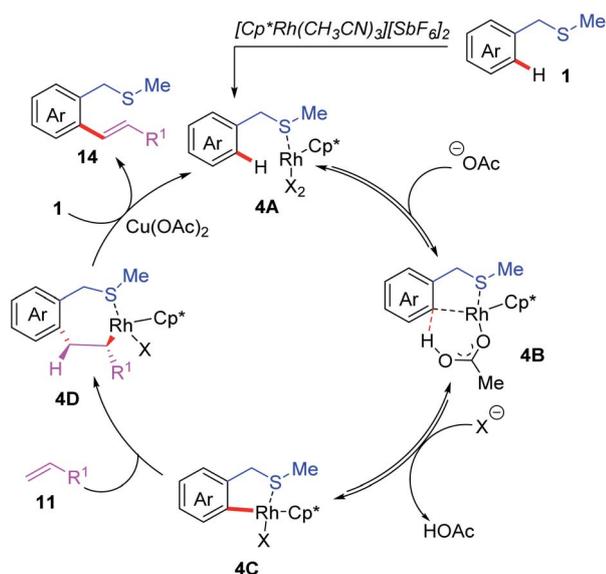
In addition to palladium catalysts, other transition metal catalysts, such as Rh, Ru and Ir, could also be utilized in the C(sp²)-H functionalization reactions.¹⁵ In 2013, the Shi group reported the first example of Rh(III)-catalyzed thioether-directed C(sp²)-H alkenylation reactions (Scheme 5).¹⁶ The substrate scope indicated that both mono- and di-substituted alkenylated products could be obtained in good yields by tuning the solvents. Later, a catalytic cycle is proposed (Scheme 6); the initial pre-coordination of the Rh(III) catalyst to the substrate **1** generates the rhodium intermediate **4A**. Next, in the presence of the OAc anion, a thioether-directed C-H bond activation step provides the 5-membered intermediate **4C** through a CMD pathway. The subsequent alkene insertion followed by a β -hydride elimination process affords the final product **14** and releases the rhodium intermediate **4A**.

In 2018, the Miura group reported the first example of rhodium(I)-catalyzed alkenylation of 1-(methylthio)naphthalene with alkynes and alkenes through a thioether-directed C-H bond functionalization process (Scheme 7a).¹⁷ This reaction was

The Shi group work:

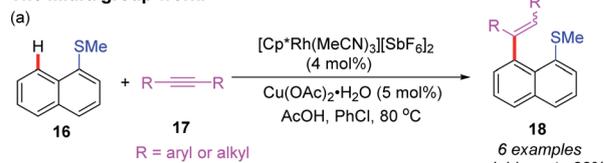


Scheme 5 Rh(III)-catalyzed and thioether-directed C(sp²)-H alkenylation reactions.



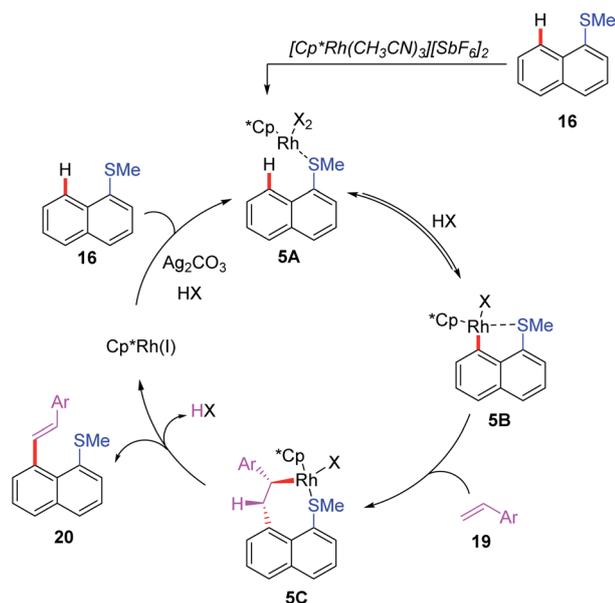
Scheme 6 Proposed mechanism of Rh(III)-catalyzed and thioether-directed C(sp²)-H alkenylation reactions.

The Miura group work:



Scheme 7 (a) Rh(III)-catalyzed and thioether-directed C-H alkenylation of 1-(methylthio)naphthalene. (b) Rh(III)-catalyzed and thioether-directed C-H arylation of 1-(methylthio)naphthalene. (c) Rh(III)-catalyzed and thioether-directed C-H alkenylation of indole derivatives.

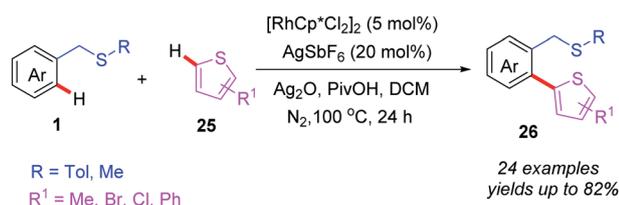




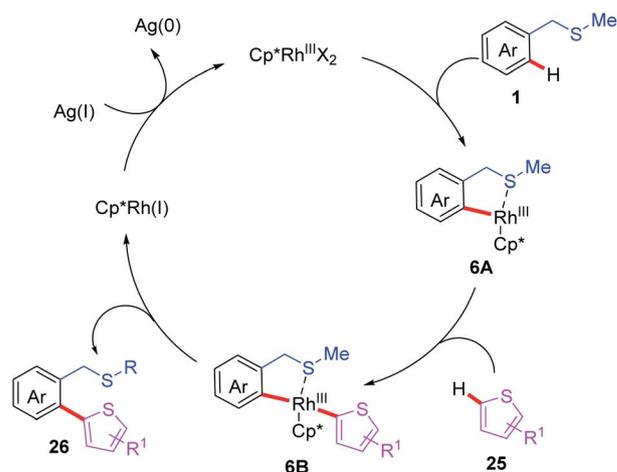
Scheme 8 Proposed mechanism of Rh(III)-catalyzed and thioether-directed C–H alkenylation of 1-(methylthio)naphthalene.

selectively carried out in the *peri*-position of the naphthalene ring. When alkynes were used as coupling counterparts, a mixture of *E/Z*-isomers were produced, while alkene couples afforded the *E*-configuration products. Later, they reported a Rh(I)-catalyzed direct C–H arylation reaction of naphthalene derivatives by employing a thioether directing group (Scheme 7b).¹⁸ The reaction exhibited an excellent *peri*-selectivity and broad functional group compatibility. Furthermore, the arylated product was very susceptible to the sulfur-containing poly aromatic compound. Additionally, the same group demonstrated a thioether-directed selective C–H alkenylation of indoles under rhodium catalysis (Scheme 7c).¹⁹ Using different thioether directing groups including MeS, CyS, and PhS, more than 36 examples of C4-alkenylated indoles were isolated in excellent yields (up to 97%). The authors speculated that the formation of 5-membered metallacycle intermediates was the key for this transformation. Moreover, a plausible catalytic cycle involving Rh(III)/Rh(I) catalysis is depicted in Scheme 8. First of all, the initial coordination of substrate **16** and Cp*Rh(III) catalyst gives the intermediate **5A**. Next, the C–H bond activation assisted by thioether group provides the intermediate **5B**.

The You group work:



Scheme 9 Rh(III)-catalyzed and thioether-directed C–H/C–H cross coupling of benzylthioethers with thiophenes.

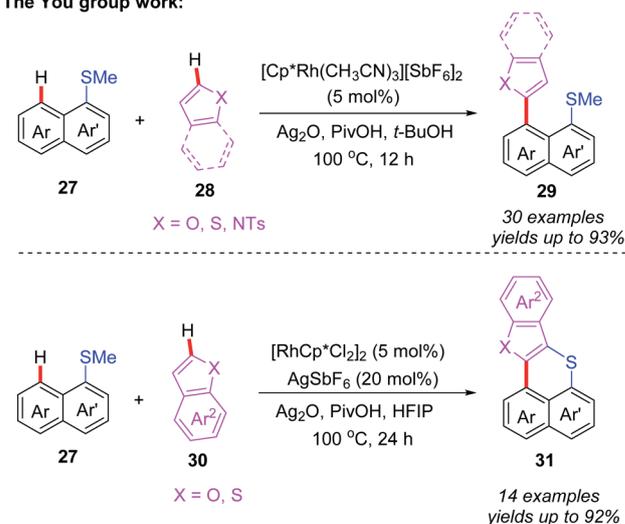


Scheme 10 The proposed mechanism of Rh(III)-catalyzed and thioether-directed C–H/C–H cross coupling of benzylthioethers with thiophenes.

Subsequently, the alkene insertion with a β -hydride elimination generates the desired product **20** with Cp*Rh(I) catalyst.

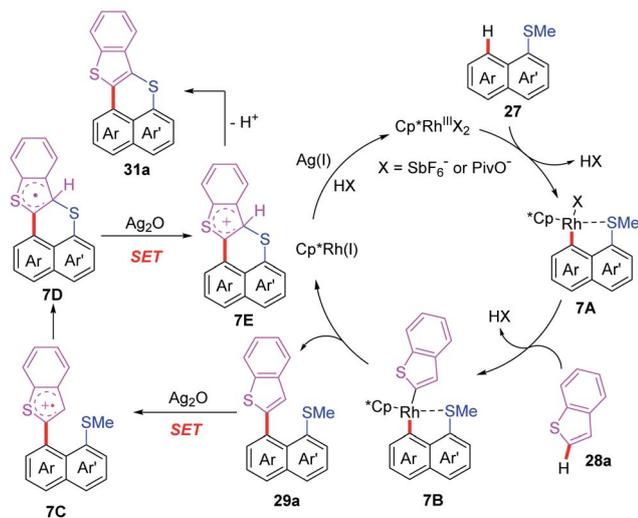
Recently, the You group reported an efficient Rh(III)-catalyzed thioether-directed C–H/C–H cross coupling of benzylthioethers with thiophenes to construct various *ortho*-functionalized 2-aryl thiophenes (Scheme 9).²⁰ The substrate scope studies indicated that both *ortho*- and *meta*-substituted substrates produce mono-heteroarylated benzylsulfides products, while *para*-substituted substrates mainly gave rise to the di-heteroarylated products. Moreover, a plausible catalytic cycle involving Rh(III)/Rh(I) catalysis is depicted (Scheme 10). First, coordination of the Rh(III) catalyst to the thioether **1** followed by thioether-directed C–H bond activation provides the rhodium intermediate **6A**. Then, the intermediate **6A** reacts with the thiophene to generate

The You group work:



Scheme 11 Rh-catalyzed *peri*-selective heteroarylation/Ag-mediated SET intramolecular cyclization sequence of 1-(methylthio)naphthalenes and heteroatom-embedded analogues.



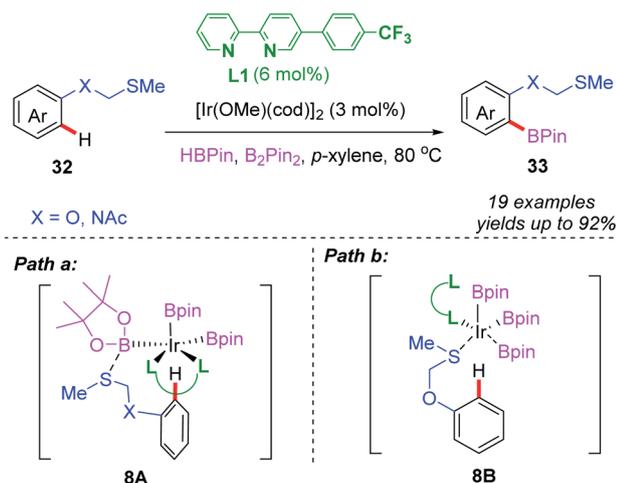


Scheme 12 Proposed mechanism of Rh-catalyzed *peri*-selective heteroarylation/Ag-mediated SET intramolecular cyclization reactions.

the rhodium intermediate **4B**. Next, reductive elimination of intermediate **6B** affords the desired product **26** and generates the Rh(I) catalyst. In the presence of a silver salt, the Rh(I) catalyst is then oxidized to Rh(III) for the following catalytic cycle.

After that, the same group also reported a Rh-catalyzed *peri*-selective heteroarylation of 1-(methylthio)naphthalenes and aromatic heterocyclic compounds (Scheme 11).²¹ Furthermore, when HFIP was instead of *t*-BuOH, benzo[*de*]thioacenes were formed through a subsequent Ag-mediated intramolecular cyclization process. Different from a traditional acid-mediated sulfoxide cyclization reaction, the detailed mechanism studies indicate that this process involves a SET cyclization of cationic radical and methanethiol group. Furthermore, a plausible mechanism is also proposed (Scheme 12). The rhodium

The Kanai and Kuninobu groups work:

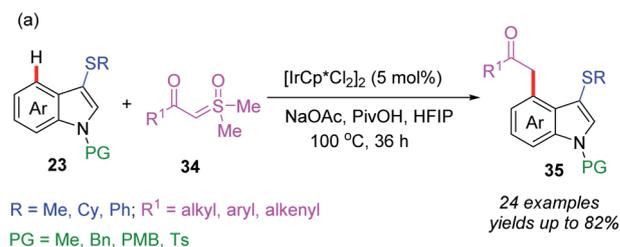


Scheme 13 Iridium/bipyridine-catalyzed *ortho*-C–H borylation of methylthiomethyl protected phenol and aniline derivatives.

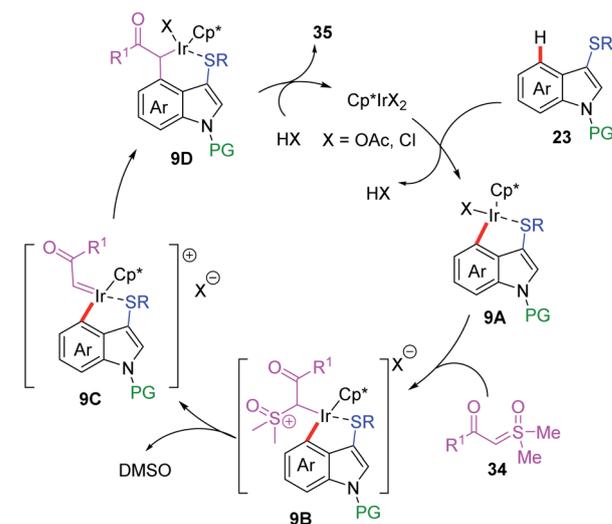
intermediate **7A** is formed through the coordination of Cp*Rh(III) catalyst and substrate **27** with a subsequent thioether-directed *peri*-C–H bond activation. Next, the reaction of benzothiophene with intermediate **7A** affords the rhodium intermediate **7B**, which can be further converted to the coupled product **29a** and Cp*Rh(I) species through a reductive elimination process. The Cp*Rh(I) species can be re-oxidized to Cp*Rh(III) catalyst for the next catalytic cycle. It's worth noting that, when HFIP is reaction solvent, the coupled product **29a** can be transformed to cation-radical intermediate **7C** through a Ag₂O-mediated SET process. The subsequent electrophilic cyclization and demethylation provide intermediate **7D**. Next, the second SET oxidation and followed by a deprotonation generate the final product **31a**.

Normally, the system of iridium/bipyridine-catalyzed C–H borylation often achieves *meta*- and *para*-(C–H) functionalization reactions.²² However, the Kanai and Kuninobu groups developed an iridium/bipyridine-catalyzed *ortho*-C–H borylation of methylthiomethyl protected phenol and aniline derivatives (Scheme 13).²³ In this work, a methylthiomethyl group was utilized as a fixed directing group, and bipyridine-type ligand **L1** as a chiral ligand. The desired borylated products were obtained in good yields with good functional group tolerance. Furthermore, the authors propose two possible reaction mechanisms. The first pathway undergoes outer spherical Lewis acid–base interaction between the boron-based ligand and the thioether

The Miura group work:



(b) The proposed mechanism



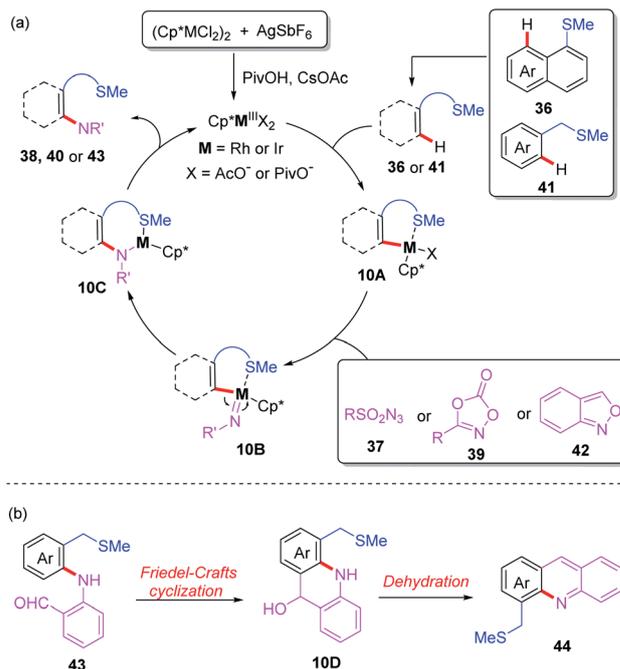
Scheme 14 (a) Ir-catalyzed and thioether-directed C4-selective acylmethylation of indoles. (b) The proposed mechanism.



group (**8A**). The second proceeds through the coordination of the thioether auxiliary, as a directing group, to the iridium catalyst (**8B**).

Recently, the Miura group also reported an Ir-catalyzed, thioether-directed, C4-selective acyl-methylation of indoles employing α -carbonyl sulfoxonium ylides as carbene precursors (Scheme 14a).²⁴ Both electron-withdrawing and electron-donating groups on the aromatic rings of indoles and α -carbonyl sulfoxonium ylides were well-tolerated in this reaction. A catalytic cycle is proposed in Scheme 14b. The initial cyclo-metallation of indole substrate **23** with Ir catalyst generates the intermediate **9A** together with HX. Next, the coordination of sulfoxonium ylide affords the species **9B**. A reactive carbene species **9C** is then formed undergoes α -elimination of DMSO. The subsequent migratory insertion followed by a protonolysis process gives the desired product **35** and regenerates the active catalyst.

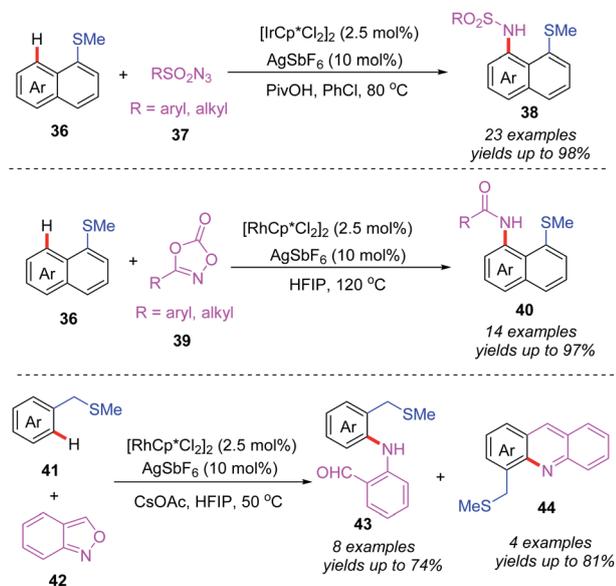
Very recently, the Zhang group reported thioether-directed, Cp*Ir(III)- and Cp*Rh(III)-catalyzed C(sp²)-H amination of unactivated arenes by using different nitrogen sources (Scheme 15).²⁵ Three types of nitrene precursor reagents inducing dioxazolone, anthranil and azide compounds, were converted to the desired amidated products in moderate to good yields. It should be noted that it is first time to use Cp*Ir(III)- and Cp*Rh(III) catalysts for thioether-directed C-H amination reactions. The further mechanistic experiments show that the C-H bond cleavage process may not be involved in a rate-determining step. In addition, a plausible catalytic mechanism is outlined in Scheme 16. The initial ligand exchange and thioether-directed C-H bond activation produce the cyclometallated intermediate **10A**. Then, the resulting intermediate **10A** reacts with nitrene precursors, affording the intermediate **10B**. After that,



Scheme 16 (a) Proposed mechanism of thioether-directed Cp*Ir(III)- and Cp*Rh(III)-catalyzed C(sp²)-H amination reactions. (b) Friedel-Crafts cyclization and subsequent dehydration.

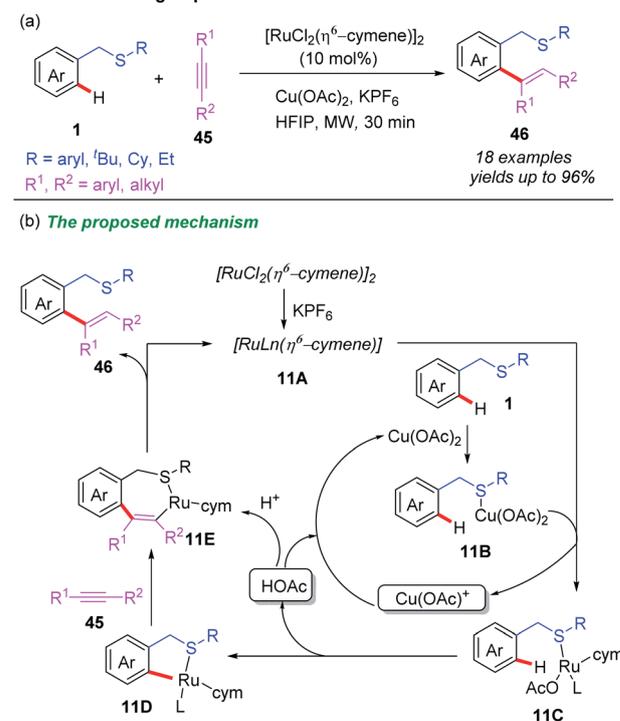
α -migratory insertion followed by a protonation process provides the final product. Besides, compound **43** can be transformed to product **44** through a Friedel-Crafts cyclization and subsequent dehydration.

The Zhang group work:



Scheme 15 Thioether-directed, Cp*Ir(III)- and Cp*Rh(III)-catalyzed C(sp²)-H amination of unactivated arenes by using different nitrogen sources.

The Urriolabeitia group work:



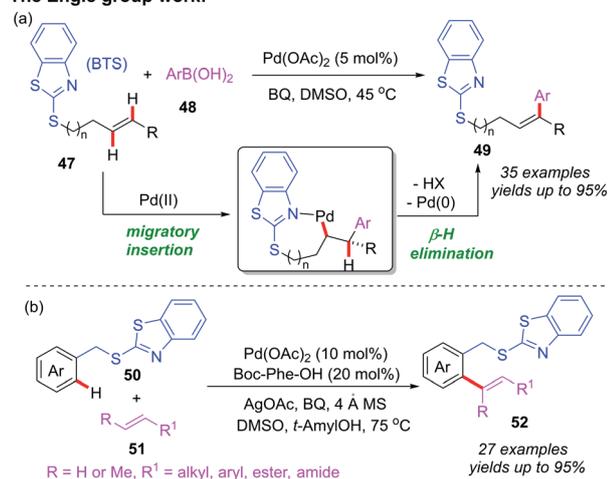
Scheme 17 (a) Ru-catalyzed and thioether-directed regioselective C-H hydroarylation of alkynes with benzylthioethers. (b) The proposed mechanism.



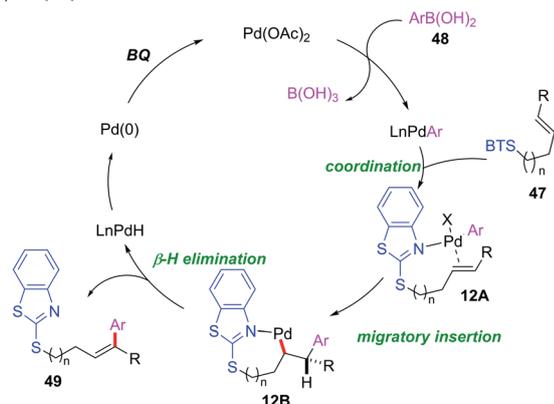
In 2015, Ru-catalyzed, thioether-directed regioselective C–H hydroarylation of alkynes with benzylthioethers was developed by Urriolabeitia and Villuendas (Scheme 17a).²⁶ A variety of alkenylated benzylthioethers were prepared from thioether substrates and electron-rich alkynes in good yields under microwave irradiation and short reaction times. A plausible catalytic cycle is depicted in Scheme 17b. First, the pre-coordination of KPF_6 to the Ru catalyst provides the Ru species **11A**. Meanwhile, the coordination of the substrate **1** to $Cu(OAc)_2$ forms intermediate **11B** which prevented the poisoning of the Ru species. Next, the reaction of intermediate **11B** and Ru species **11A** followed by a selective C–H bond cleavage produces the Ru intermediate **11D**, which reacts with the internal alkyne through π -bonding and migratory *syn*-insertion, leading to the formation of Ru intermediate **11E**. Finally, the desired product **46** and Ru species **11A** were yielded through the reductive elimination of **11E**.

The above reports indicate that thioether can act as a directing group for C–H bond functionalization through the coordination of the internal sulfur atom to transition metal

The Engle group work:



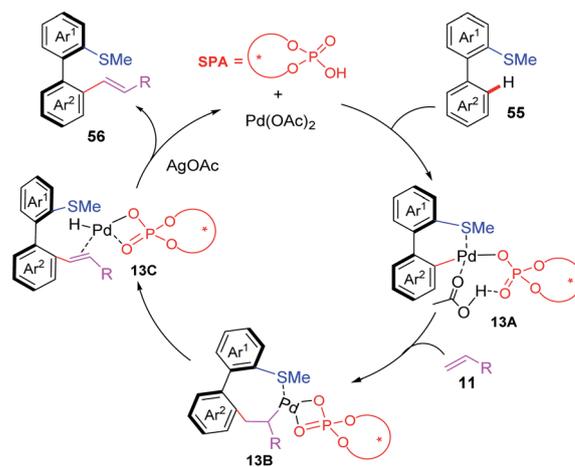
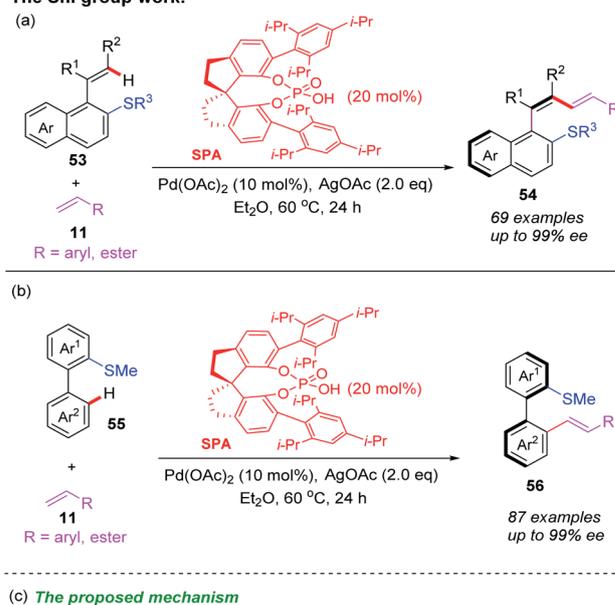
(c) The proposed mechanism



Scheme 18 (a) Palladium-catalyzed directed oxidative Heck reactions assisted by a BTS directing group. (b) Palladium-catalyzed directed olefination reactions assisted by a BTS directing group. (c) The proposed mechanism of Heck reaction.

centers, however, this is not the only possible catalysis pathway. In 2019, Engle and co-workers developed a Pd-catalyzed directed oxidative Heck reaction assisted by a weakly coordinating benzothiazole thioether (BTS) directing group (Scheme 18a).²⁷ Employing this directing group, the Pd(II)-catalyzed $C(sp^2)$ -H olefination of arenes has also been reported (Scheme 18b). The control experiments and DFT calculations suggested that the Pd catalyst was coordinated to the nitrogen atom of the benzothiazole thioether directing group. A general catalytic cycle has been proposed (Scheme 18c). It is believed that the initial transmetalation of Pd(II) with arylboronic acid provides the aryl-Pd species, which is further converted into intermediate **12A** *via* a substrate binding step. Next, the subsequent migratory insertion process generates

The Shi group work:



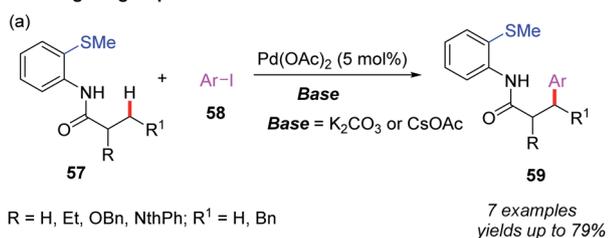
Scheme 19 (a) Palladium-catalyzed and thioether-directed atroposelective C–H olefination by employing a SPA ligand for preparing chiral styrenes with a conjugated 1,3-diene scaffolds. (b) Palladium-catalyzed and thioether-directed atroposelective C–H olefination by employing a SPA ligand for preparing chiral biaryls. (c) The proposed mechanism for preparing chiral biaryls.



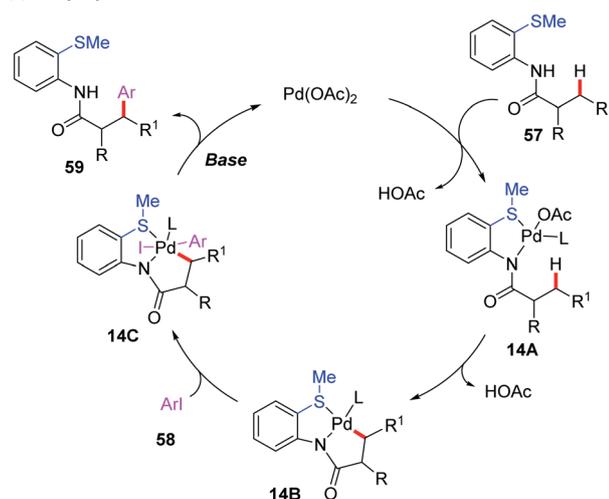
the palladacycle intermediate **12B**, upon which, the β -H elimination step affords the desired product **49** and the Pd-H species. Finally, Pd-H is transformed Pd(II) *via* reductive elimination and reoxidation.

In the past ten years, directing group assisted palladium-catalyzed asymmetric C-H bond functionalization has attracted considerable attention.²⁸ Very recently, Shi and co-workers developed a palladium-catalyzed and thioether-directed atroposelective C-H olefination by employing a novel chiral spiro phosphoric acid ligand **SPA** (Scheme 19a).²⁹ More than 69 examples of axially chiral styrenes with a conjugated 1,3-diene scaffolds were obtained in good yields (up to 99% yields) with excellent enantioselectivities (up to 99% ee) and complete *Z*-selectivity control. Subsequently, they also used the same chiral spiro phosphoric acid ligand **SPA** to prepare axially chiral biaryls through thioether-directed Pd-catalyzed atroposelective C-H olefination (Scheme 19b).³⁰ A variety of axially chiral biaryls containing thioether motifs were isolated in excellent yields (up to 99% yields) and enantioselectivities (up to 99% ee). In this process, the authors used DFT calculations to optimize reaction conditions and propose possible mechanism (Scheme 19c). The selective C-H bond activation of biaryl **55** with chiral spiro phosphoric acid ligand **SPA** and Pd(OAc)₂ catalyst generates the Pd(II) intermediate **13A**. The subsequent *R*-configuration alkene insertion forms the intermediate **13B**. Next, the β -H elimination and a rapid reductive elimination produce the final product **56**.

The Daugulis group work:



(b) The proposed mechanism



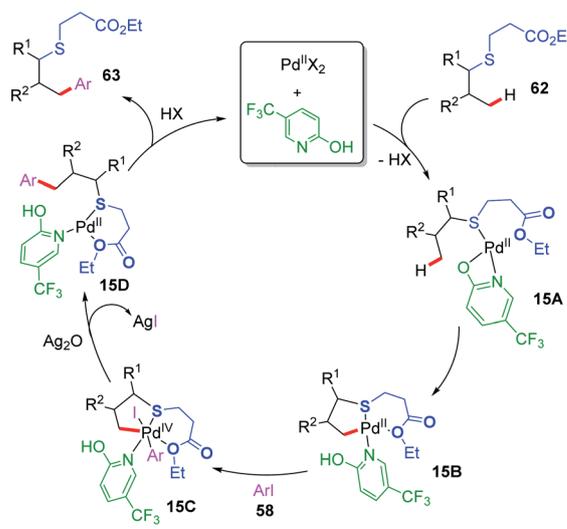
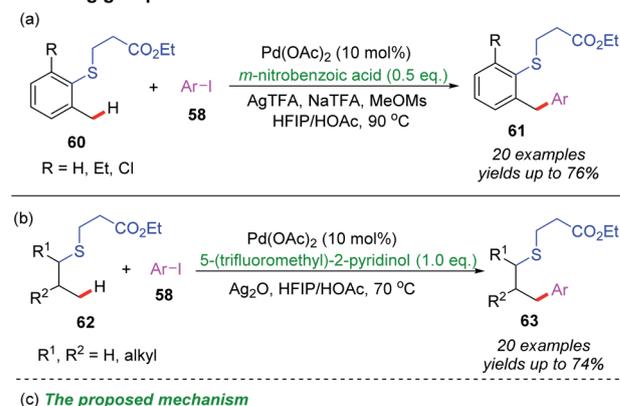
Scheme 20 (a) Palladium-catalyzed and 2-methylthioaniline-directed arylation of C(sp³)-H bonds. (b) The proposed mechanism.

3. Remote C(sp³)-H functionalization

Compared to C(sp²)-H bond functionalization, C(sp³)-H bond functionalization is more challenging because the formation of alkyl-metal bonds is more difficult than aryl-metal bonds. Usually, bidentate directing groups show higher reactivity than their monodentate counterparts.³¹ Therefore, the catalytic functionalization of C(sp³)-H bonds of thioethers could be achieved by using a *S,N*- or *S,O*-bidentate directing group.

In 2010, Daugulis and Shabashov developed a method for palladium-catalyzed bidentate-directed β -arylation of C(sp³)-H bonds (Scheme 20a).³² Using 2-methylthioaniline as a bidentate directing group and aryl iodide as a coupling partner, a variety of mono-arylated products were isolated in moderate to good yields through a primary sp³ C-H bond activation process. A generalized mechanism is proposed (Scheme 20b). It is believed that this reaction is initiated by the coordination of the substrate **57** to Pd(OAc)₂, providing the palladium intermediate **14A**. Next, the C-H bond cyclometalation of intermediate **14A** produces the cyclopalladium intermediate **14B**. The subsequent

The Dong group work:



Scheme 21 (a) Pd-catalyzed benzylic C-H arylation of thioethers using a *S,O*-bidentate directing group. (b) Pd-catalyzed alkyl C-H arylation of thioethers using a *S,O*-bidentate directing group. (c) The proposed mechanism of alkyl C-H arylation.



oxidative addition of aryl iodide to intermediate **14B** affords intermediate **14C**, which is further converted to the desired product **59** through reductive elimination followed by a ligand exchange process.

Recently, Dong and co-workers reported a Pd-catalyzed γ -C(sp³)-H arylation of thioethers using a *S,O*-bidentate directing group (Scheme 21a and b).³³ Both benzylic and alkyl C-H bonds could be activated to react with various aryl iodides, providing the desired arylated thioether derivatives in moderate to good yields. In this catalytic system, the C-H metalation step could be accelerated by using 3-nitrobenzoic acid and 5-(trifluoromethyl)-pyridin-2-ol as ligands.

In additional, a proposed mechanism is also shown in Scheme 21c. The initiated coordination of thioether substrate **62**, 5-(trifluoromethyl)-pyridin-2-ol ligand and Pd(II) catalyst affords the cyclic palladium intermediate **15A**. Subsequently, 5-(trifluoromethyl)-pyridin-2-ol ligand assisted β -C(sp³)-H bond activation of the intermediate **15A** provides a [5,6]-bicyclic palladium intermediate **15B**. Then, oxidative addition of the intermediate **15B** with an aryl iodide followed by a reductive elimination process produces intermediate **15D**. Finally, the desired arylated product **63** is formed through a ligand dissociation process, and Pd(II) catalyst and 5-(trifluoromethyl)-pyridin-2-ol ligand recovered.

4. Conclusions and outlook

In this review, we provide significant recent advances on the directing group assisted C-H functionalization of thioethers enabled by different transition metals. This review mainly introduces the transition-metal-catalyzed remote C(sp²)-H functionalization of thioethers through the coordination of the internal sulfur or nitrogen atom with transition metals. Additionally, remote C(sp³)-H bond functionalization of thioethers via an *S,N*- or *S,O*-bidentate directing group strategy has also been included. While significant progress has been made in this area, there is still much room for improvement and utilization, including: (1) C-H functionalization of thioethers using photo-chemistry or electro-chemistry approaches; (2) thioether-directed transition metal-catalyzed asymmetric C(sp²)-H and C(sp³)-H bond functionalization. Finally, we hope this review will provide readers with some insights and inspire them to discover more innovative strategies in the transition metal-catalyzed remote C-H functionalization of thioethers.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- (a) U. Dutta, S. Maiti, T. Bhattacharya and D. Maiti, *Science*, 2021, **372**, 6543; (b) J. Wen and Z. Shi, *Acc. Chem. Res.*, 2021, **54**, 1723; (c) E. Nobile, T. Castanheiro and T. Besset, *Angew. Chem., Int. Ed.*, 2021, **60**, 12170; (d) F. Liu, Z. Zhang, H. Diao and Z.-J. Shi, *ChemCatChem*, 2021, **13**, 1475; (e) Q. Zhang and B.-F. Shi, *Chem. Sci.*, 2021, **12**, 841; (f) A. J. Rago and G. Dong, *Green Synthesis and Catalysis*, 2021, **2**, 216; (g) B. Li, A. I. M. Ali and H. Ge, *Chem*, 2020, **6**, 2591; (h) J. Das, S. Guin and D. Maiti, *Chem. Sci.*, 2020, **11**, 10887; (i) K. Yang, M. Song, Z. Ma, Y. Li, Z. Li and X. Sun, *Org. Chem. Front.*, 2019, **6**, 3996; (j) K. Yang, B. Niu, Z. Ma, H. Wang, B. Lawrence and H. Ge, *J. Org. Chem.*, 2019, **84**, 14045; (k) K. Yang, D. Li, L. Zhang, Q. Chen and T. Tang, *RSC Adv.*, 2018, **8**, 13671.
- (a) S. Sasmal, U. Dutta, G. K. Lahiri and D. Maiti, *Chem. Lett.*, 2020, **49**, 1406; (b) K. Ramakrishna, J. P. Biswas, S. Jana, T. K. Achar, S. Porey and D. Maiti, *Angew. Chem., Int. Ed.*, 2019, **58**, 13808; (c) S. Guin, P. Dolui, X. Zhang, S. Paul, V. K. Singh, S. Pradhan, H. B. Chandrashekar, S. S. Anjana, R. S. Paton and D. Maiti, *Angew. Chem., Int. Ed.*, 2019, **58**, 5633; (d) C. Sambigioglio, D. Schönbauer, R. Blicke, T. Dao-Huy, G. Pototschnig, P. Schaaf, T. Wiesinger, M. F. Zia, J. Wencel-Delord, T. Besset, B. U. W. Maes and M. Schnürch, *Chem. Soc. Rev.*, 2018, **47**, 6603; (e) W.-B. Ma, P. Gandeepan, J. Li and L. Ackermann, *Org. Chem. Front.*, 2017, **4**, 1435; (f) A. Maji, B. Bhaskararao, S. Singha, R. B. Sunoj and D. Maiti, *Chem. Sci.*, 2016, **7**, 3147; (g) Z.-K. Chen, B.-J. Wang, J.-T. Zhang, W.-L. Yu, Z.-X. Liu and Y.-H. Zhang, *Org. Chem. Front.*, 2015, **2**, 1107.
- (a) B. Niu, K. Yang, B. Lawrence and H. Ge, *ChemSusChem*, 2019, **12**, 2955; (b) Q. Zhao, T. Polsson, X. Pannecoucke and T. Besset, *Synthesis*, 2017, **49**, 4808; (c) P. Gandeepan and L. Ackermann, *Chem*, 2018, **4**, 199; (d) T. Bhattacharya, S. Pimparkar and D. Maiti, *RSC Adv.*, 2018, **8**, 19456; (e) S. St John-Campbell, J. Campbell and J. A. Bull, *Org. Biomol. Chem.*, 2018, **16**, 4582.
- (a) N. Barbero and R. Martin, *Org. Lett.*, 2012, **14**, 796; (b) J. F. Hooper, A. B. Chaplin, C. González-Rodríguez, A. L. Thompson, A. S. Weller and M. C. Willis, *J. Am. Chem. Soc.*, 2012, **134**, 2906; (c) A. Senthilmurugan and I. S. Aidhen, *Eur. J. Org. Chem.*, 2010, 555.
- (a) P. Li, Y. Yang, X. Wang and X. Wu, *J. Heterocycl. Chem.*, 2021, **58**, 1225; (b) K. A. Scott and J. T. Njardarson, *Top. Curr. Chem.*, 2018, **376**, 5; (c) K. L. Dunbar, D. H. Scharf, A. Litomska and C. Hertweck, *Chem. Rev.*, 2017, **117**, 5521; (d) M. Feng, B. Tang, S. H. Liang and X. Jiang, *Curr. Top. Med. Chem.*, 2016, **16**, 1200.
- (a) R. Gómez Arrayás and C. C. Carretero, *Chem. Commun.*, 2011, **47**, 2207; (b) M. Mellah, A. Voituriez and E. Schultz, *Chem. Rev.*, 2007, **107**, 5133; (c) H. Pellissier, *Tetrahedron*, 2007, **63**, 1297.
- (a) J. Lou, Q. Wang, P. Wu, H. Wang, Y.-G. Zhou and Z.-K. Yu, *Chem. Soc. Rev.*, 2020, **49**, 4307; (b) S. Otsuka, K. Nogi and H. Yorimitsu, *Top. Curr. Chem.*, 2018, **376**, 199; (c)



- L. Zhang, J. Dong, X. Xu and Q. Liu, *Chem. Rev.*, 2016, **116**, 22; (d) F. Pan and Z.-J. Shi, *ACS Catal.*, 2014, **4**, 280; (e) S. G. Modha, V. P. Mehta and E. V. Van der Eycken, *Chem. Soc. Rev.*, 2013, **42**, 5042; (f) L. D. Wang, W. He and Z.-K. Yu, *Chem. Soc. Rev.*, 2013, **42**, 599.
- 8 (a) K.-X. Tang, C.-M. Wang, T.-H. Gao, L. Chen, L. Fan and L.-P. Sun, *Adv. Synth. Catal.*, 2019, **361**, 26; (b) X. Tang and Y. Zhao, *Mini-Rev. Org. Chem.*, 2021, **18**, 902.
- 9 (a) L. Tang, Q. Hu, K. Yang, M. Elsaid, C. Liu and H. Ge, *Green Synthesis and Catalysis*, 2022, DOI: 10.1016/j.gresc.2022.02.002; (b) A. M. F. Phillips and A. J. L. Pombeiro, *ChemCatChem*, 2018, **10**, 3354.
- 10 M. Yu, Y. Xie, C. Xie and Y. Zhang, *Org. Lett.*, 2012, **14**, 2164.
- 11 J. Yao, M. Yu and Y. Zhang, *Adv. Synth. Catal.*, 2012, **354**, 3205.
- 12 B. Wang, C. Lin, Y. Liu, Z. Fan, Z. Liu and Y. Zhang, *Org. Chem. Front.*, 2015, **2**, 973.
- 13 B. Xu, W. Liu and C. Kuang, *Eur. J. Org. Chem.*, 2014, 2576.
- 14 X.-S. Zhang, Y.-F. Zhang, K. Chen and Z.-J. Shi, *Org. Chem. Front.*, 2014, **1**, 1096.
- 15 (a) X.-T. Cao, S.-N. Wei, H.-T. Sun, M. Li, Z.-L. Zheng and G. Wang, *RSC Adv.*, 2021, **11**, 22000; (b) H. Xu, Y. Zhu, X. Chai, J. Yang and L. Dong, *Green Synthesis and Catalysis*, 2020, **1**, 167; (c) H. Shen, D. Cheng, Y. Li, T. Liu, X. Yi, L. Liu, F. Ling and W. Zhong, *Green Synthesis and Catalysis*, 2020, **1**, 175; (d) Y. Rao, G. Shan and X. Yang, *Sci. China Chem.*, 2014, **57**, 930.
- 16 X.-S. Zhang, Q.-L. Zhu, Y.-F. Zhang, Y.-B. Li and Z.-J. Shi, *Chem.-Eur. J.*, 2013, **19**, 11898.
- 17 M. Shigeno, Y. Nishii, T. Satoh and M. Miura, *Asian J. Org. Chem.*, 2018, **7**, 1334.
- 18 S. Moon, Y. Nishii and M. Miura, *Org. Lett.*, 2019, **21**, 233.
- 19 C. N. Kona, Y. Nishii and M. Miura, *Org. Lett.*, 2018, **20**, 4898.
- 20 S. Yang, R. Cheng, T. Zhao, A. Luo, J. Lan and J. You, *Org. Lett.*, 2019, **21**, 5086.
- 21 S. Yang, R. Cheng, M. Zhang, Z. Bin and J. You, *ACS Catal.*, 2019, **9**, 6188.
- 22 (a) M. E. Hoque, R. Bisht, C. Haldar and B. J. Chattopadhyay, *J. Am. Chem. Soc.*, 2017, **139**, 7745; (b) L. Yang, K. Semba and Y. Nakao, *Angew. Chem., Int. Ed.*, 2017, **56**, 4853; (c) L. Zhu, X. Qi, Y. Li, M. Duan, L. Zou, R. Bai and Y. Lan, *Organometallics*, 2017, **36**, 2107; (d) B. E. Haines, Y. Saito, Y. Segawa, K. Itami and D. G. Musaev, *ACS Catal.*, 2016, **6**, 7536; (e) Y. Saito, Y. Segawa and K. Itami, *J. Am. Chem. Soc.*, 2015, **137**, 5193; (f) Y. Kuninobu, H. Ida, M. Nishi and M. Kanai, *Nat. Chem.*, 2015, **7**, 712.
- 23 H. L. Li, M. Kanai and Y. Kuninobu, *Org. Lett.*, 2017, **19**, 5944.
- 24 C. N. Kona, Y. Nishii and M. Miura, *Org. Lett.*, 2020, **22**, 4806.
- 25 H. Xie, M. Zhong, X.-T. Wang, J.-Q. Wu, Y.-Q. Cai, J. Liu, B. Shu, T. Che and S.-S. Zhang, *Org. Chem. Front.*, 2021, **8**, 635.
- 26 P. Villuendas and E. P. Urriolabeitia, *Org. Lett.*, 2015, **17**, 3178.
- 27 A. M. Romine, K. S. Yang, M. K. Karunananda, J. S. Chen and K. M. Engle, *ACS Catal.*, 2019, **9**, 7626.
- 28 (a) K. Yang, M. Song, H. Liu and H. Ge, *Chem. Sci.*, 2020, **11**, 12616; (b) Y. Dong, R. Liu and W. Wang, *Green Synthesis and Catalysis*, 2020, **1**, 83; (c) S. Li, Q. Chen, Z.-M. Zhang and J. Zhang, *Green Synthesis and Catalysis*, 2021, **2**, 374; (d) A. A. Zhang, C. Chen, Y. Gao, M. Mo, R.-Z. Shen, Y.-H. Zhang, N. Ishida, M. Murakami and L. Liu, *Green Synthesis and Catalysis*, 2021, **2**, 311.
- 29 L. Jin, P. Zhang, Y. Li, X. Yu and B.-F. Shi, *J. Am. Chem. Soc.*, 2021, **143**, 12335.
- 30 G. Liao, T. Zhang, L. Jin, B.-J. Wang, C.-K. Xu, Y. Lan, Y. Zhao and B.-F. Shi, *Angew. Chem., Int. Ed.*, 2022, **61**, e202115221.
- 31 (a) S. Rej, Y. Ano and N. Chatani, *Chem. Rev.*, 2020, **120**, 1788; (b) O. Daugulis, J. Roane and L. D. Tran, *Acc. Chem. Res.*, 2015, **48**, 1053; (c) G. Rouquet and N. Chatani, *Angew. Chem., Int. Ed.*, 2013, **52**, 11726.
- 32 D. Shabashov and O. Daugulis, *J. Am. Chem. Soc.*, 2010, **132**, 3965.
- 33 L. Jin, J. Wang and G. Dong, *Angew. Chem., Int. Ed.*, 2018, **57**, 12352.

