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Nanoenzyme–chitosan hydrogel complex with cascade catalytic and self-reinforced antibacterial performance for accelerated healing of diabetic wounds†

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The significant disability and fatality rate of diabetes chronic wounds necessitates the development of efficient diabetic wound treatment. Ordinary wound treatment is restricted by issues such as poor penetration, inadequate supply, and absorption difficulties as well as tanglesome diabetic wound microenvironment issues such as hyperglycemia, excessive reactive oxygen species (ROS), and hypoxia. Herein, we designed a multifunctional glucose oxidase (GOx) and catalase (CAT) nanoenzyme–chitosan (GCNC) hydrogel complex to improve the microenvironment of diabetic wounds and provide continuous oxygen delivery for efficient wound healing. By simultaneously forming the GOx–CAT nanoenzyme (GCNE) composite, the GCNC hydrogel complex could effectively reduce glucose and ROS (H2O2) concentrations in diabetic wounds through cascade catalytic reactions and achieve continuous oxygen supply, which promoted cell proliferation, migration, and angiogenesis, thereby accelerating diabetic wound healing. In addition, the byproduct gluconic acid produced by the cascade reaction can activate the amino group of chitosan to reinforce the antibacterial performance and prevent microbial infection. This multifunctional GCNC hydrogel complex with continuous oxygen supply, self-reinforcing antibacterial properties, and byproduct-free features provides a general strategy for repairing the extensive tissue damage in diabetes.

1. Introduction

Diabetes is one of the most critical public health concerns impacting human health in the 21st century. It has been on the rise in the last ten years, and it is anticipated to affect 10% of the world’s population by 2030.1,2 Currently, more than 25% of diabetic patients are in danger of amputation, such as diabetic foot, or possibly death owing to chronic non-healing wounds. Ordinary wound healing requires the precise coordination and integration of complicated biological and molecular activities such as cell migration, proliferation, extracellular matrix deposition, and angiogenesis.3–5 However, diabetic wounds are typically difficult to heal fast and efficiently owing to the complicated physiological milieu of diabetic wounds, such as tissue hypoxia, hyperglycemia, and oxidative stress and are often vulnerable to pathogenic bacteria that aggravate the condition.6–9 To speed the healing of chronic diabetes wounds, a novel technique that can cope with the intricate physiological milieu of diabetic wounds is required.

Oxygen is essential for wound healing because it can effectively promote cell proliferation, metastasis and angiogenesis, and other wound repair processes.10–13 Currently, hyperbaric therapy and local gaseous oxygen therapy have been used to treat diabetes in clinics.12,14–16 However, because their oxygen supply is gaseous oxygen, they encounter issues such as low tissue permeability, and difficulties delivering and sustaining effective oxygen concentrations, resulting in inadequate therapy outcomes.17–22 In recent years, researchers have investigated a variety of oxygen delivery strategies to ensure effective oxygen delivery in wounds, including supplying oxygen-binding hemoglobin23,24 or photosynthetic algae organisms25,26

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that can dramatically expedite wound healing.\textsuperscript{23,25,27-29} Despite the fact that these treatments have effectively enhanced the oxygen flow to the wound, the high-efficiency and continuous provision of oxygen still confronts significant obstacles due to the complicated physiological milieu of diabetes. Besides hypoxia, adverse hyperglycemia and an oxidative microenvironment (such as high H\textsubscript{2}O\textsubscript{2}) slow diabetic wound healing.\textsuperscript{4,30,31} As a result, developing a technique for diabetic wound recovery that can effectively regulate blood sugar, reduce ROS (such as H\textsubscript{2}O\textsubscript{2}), and constantly deliver oxygen is very desirable.

In recent years, enzymes and their analogues have shown considerable promise in speeding wound healing.\textsuperscript{32-38} A variety of wound dressings containing enzyme or enzyme-like abilities are being developed to catalyze endogenous H\textsubscript{2}O\textsubscript{2} at the wound site, generating oxygen and improving the diabetic microenvironment.\textsuperscript{39,40} such as catalase-like polyvinyl alcohol film with hematite\textsuperscript{38} or multifunctional hydrogel containing manganese dioxide.\textsuperscript{37} It has been shown that improving the imbalanced physiological environment in diabetic wounds and increasing oxygen delivery are effective strategies for treating significant tissue damage caused by diabetes. However, it is challenging to create and maintain adequate oxygen concentration at the wound owing to insufficient endogenous H\textsubscript{2}O\textsubscript{2} and the restricted enzymatic activity of individual nanozymes. Because of its high local concentration, reduced intermediate decomposition, and high mass transfer efficiency, the enzyme cascade reaction in biological systems can significantly improve catalytic performance.\textsuperscript{41} For example, the glucose oxidase (GOx) and catalase (CAT) cascade systems have been investigated in the treatment of diabetic wounds,\textsuperscript{42-45} with GOx primarily consumes glucose to create H\textsubscript{2}O\textsubscript{2}, while CAT further catalyzes H\textsubscript{2}O\textsubscript{2} to provide sufficient oxygen. However, the majority of these systems use metal or metal oxide enzyme-like compounds, and their catalytic effectiveness, the cascade effect, and safety of byproducts such as acid or metal ions are all causes for concern. It is critical for the construction of a safe and efficient enzyme cascade system to understand how to increase enzyme stability, the cascade efficiency, and decrease unwanted by-products via rational material design. Natural enzymes have excellent cascade catalytic efficiency as well as biocompatibility. Hydrogels have high water content and cargo loading efficiency, as well as superior biocompatibility, making them popular in wound healing applications.\textsuperscript{46-49}

Herein, we designed an injectable GOx–CAT nanoenzyme–chitosan (GCNC) hydrogel system for wound healing in diabetes mellitus, as shown in Scheme 1. GOx–CAT nanoenzyme (GCNE) composites were prepared by polymerization using acryloylated GOx and CAT, acrylamide (AAM), and N,N\textsuperscript{+}-methylenebisacrylamide (BIS). Due to the close spatial distance between GOx and CAT, H\textsubscript{2}O\textsubscript{2} generated after GOx catalyzes glucose in diabetic wounds and endogenous H\textsubscript{2}O\textsubscript{2} can be rapidly catalyzed by CAT to produce O\textsubscript{2}. Meanwhile, the generated O\textsubscript{2} can in turn improve the catalytic activity of GOx, allowing it to efficiently regulate blood glucose while continually producing O\textsubscript{2}. In addition, the gluconic acid catalyzed by the GOx catalytic reaction can activate the amino groups of chitosan in the hydrogel, which further improves the antibacterial activity of chitosan while consuming acidic by-product gluconic acid. To the best of our knowledge, this multi-functional diabetic wound dressing hydrogel is described for the first time with multiple functions of controlling blood glucose, antibacterial, hemostasis, improving the oxidative microenvironment of the diabetic wound, and circulating oxygen supply, demonstrating a versatile strategy for healing a broad range of tissue damage caused by diabetes.

### 2. Results and discussion

**2.1. Synthesis and characterization of multifunctional GOx–CAT nanoenzyme–chitosan (GCNC) hydrogel system**

The GOx–CAT nanoenzyme and chitosan-modified injectable hydrogel were physically mixed to obtain a multifunctional GOx–CAT nanoenzyme–chitosan (GCNC) hydrogel complex system (Scheme 1).\textsuperscript{50,51} TEM was used to observe the morphology of the nanoenzyme composite. As shown in Fig. 1A, the spherical structure with a size of about 90 nm is clearly visible, and the DLS image also shows that the average particle size is about 128 nm, which is attributed to the water swelling ability of nanoenzyme. Besides, SEM was used to observe the morphology of the GCNC hydrogel complex. As shown in Fig. 1B, the hydrogel structure of the porous network is clearly visible, facilitating the delivery of oxygen and nutrients. The injectable and adhesive properties of the GCNC hydrogel...
complex are also shown in Fig. S1D.† When at room temperature, the hydrogel is in a liquid state; when at 37° C, the hydrogel solidified, which illustrates the thermos-sensitive properties of the hydrogel. Then, the stability of the GCNC hydrogel complex was evaluated. As shown in Fig. S1A,† the GCNC hydrogel complex under PBS and the acidic, oxidative, and glucose-containing microenvironment remained stable (weight remained above 80%) within 48 h. The weight of hydrogel was monitored, as shown in Fig. S1B.† The weight showed a slight decrease, which may be due to the degradation and release behavior of hydrogels. Importantly, the slow release behavior of the GCNC hydrogel complex was evaluated, as shown in Fig. 1C. In solution 1 (PBS), the protein amount of hydrogel released was less than that in solution 2 (PBS containing glucose, acid, and H₂O₂), which may be due to the faster degradation of hydrogels in solution 2.

After the characterization of hydrogels, it is vital to demonstrate the cascade catalytic activity of the GOx–CAT nanoenzyme (GCNE) composite. Due to the close spatial distance of both enzymes in the nanostructure, GCNE composite can have an efficient enzyme cascade reaction, thus generating a large amount of oxygen through cyclic catalysis. Firstly, glucose (Glu) can be catalyzed by GOx to produce H₂O₂ and gluconic acid. Therefore, the pH value of the solution should decrease with the progress of the reaction. As shown in Fig. 1D, the reaction between Glu and GCNE composite proceeded quickly in a short time as the pH decreased from 7 to about 3.2 within 10 h and remained stable. However, the pH values of other groups were almost unchanged (maintaining between 6 and 7), which verified the catalytic reaction of GOx in the presence of Glu. Secondly, H₂O₂ can be catalyzed by CAT to produce oxygen and water, so the production of oxygen symbolizes the progress of the CAT reaction. As shown in Fig. 1E, a large number of bubbles were generated at the moment when 30% H₂O₂ reacted with the GCNE composite, and the concentration of dissolved oxygen (DO) in the solution increased rapidly from 5 up to 35 mg L⁻¹ within 30 s. However, the DO concentration of other groups was almost unchanged (kept around 5 mg L⁻¹), which verified the catalytic reaction of CAT in the presence of H₂O₂. The above results indicate that the catalytic performance of the two enzymes in the nanostructure was still effective and not affected by polymerization. In addition, in order to verify the cascade catalytic effect of the GCNE composite, we placed it in a solution containing only Glu. As shown in Fig. 1F, we can see that compared with single-GOx nanoenzyme materials, the GCNE composite could effectively regulate the rate of oxygen concentration reduction, because CAT in the GCNE composite could catalyze H₂O₂ to produce O₂. Finally,
we used Glu and H$_2$O$_2$ to mimic the conditions of high glucose and high oxidative stress in diabetic wounds, so as to evaluate the catalytic reaction of the GCNE composite. Because the glucose oxidation reaction consumes oxygen, CAT is required to consume endogenous H$_2$O$_2$ at the wound site to produce oxygen. With the production and consumption of H$_2$O$_2$, oxygen production gradually increases. As shown in Fig. 1G, DO can rise rapidly to 10.6 mg mL$^{-1}$ within 4 minutes because a large amount of oxygen was produced by the decomposition of H$_2$O$_2$. In the next two minutes, the oxygen concentration decreased slightly and remained at about 9.9 mg mL$^{-1}$, which is due to the excellent catalytic ability of the GCNE composite. Last but not least, the oxygen supply ability of the GCNC hydrogel complex was also evaluated. As shown in Fig. S1C,† after adding 30% H$_2$O$_2$ into the GCNC hydrogel complex, a large number of bubbles were generated instantaneously, which showed that the GCNC hydrogel complex still retains good catalytic activity of the GCNE composite. Overall, the as-designed GCNE composite possesses good catalytic activity, and it can continuously generate oxygen through a cascade catalytic reaction in the presence of glucose.

2.2. H$^+$ absorption effect of the GCNC hydrogel complex

The cytotoxicity of the GCNE composite on cell lines was tested by MTT assays. As shown in Fig. 2A and B, the GCNE composite showed the ability to promote growth at low concentrations, but with the increase in concentration, cell activity gradually decreased, which may be caused by the gluconic acid byproduct produced by GOx. Considering that gluconic acid catalyzed by GOx will produce acidic conditions in the wound, which has a negative effect on the wound healing process by reducing pH in the environment; we chose chitosan-based
(CS) hydrogel to attenuate this effect. Since chitosan has many amino groups, it has the function of adsorbing H⁺, so under the action of chitosan, the pH of the mixed solution of GCNE composite and Glu will decrease more slowly. Meanwhile, it can enhance the antibacterial effect of chitosan due to the positive charges. As shown in Fig. 2C, we found that the pH value of the mixed solution of Glu and GCNE composite could be reduced from 6.62 to about 3.18 in 16 h, but the addition of chitosan could significantly slow down this decline, and its pH value was maintained at about 6.7. Then, the cytotoxicity of the GCNC hydrogel complex was studied, as illustrated in Fig. 2D. As shown in Fig. 2E and F, the GCNE composite did not decrease the cell activity at high concentrations because of chitosan-based hydrogel-adsorbed H⁺ and avoided the decline of the environmental pH. Therefore, chitosan can avoid the adverse effect of excessive acid on wound healing, while improving the antibacterial properties to promote the process of wound healing.

2.3. Anti-oxidative stress and oxygen supply effect of GCNE composite

The above results prove that the GCNE composite had good *in vitro* oxygen-release performance, so it is of great significance to explore the effects of nanoenzyme, intracellularly. Wounds have been reported to be in a state of oxidative stress due to increased production of ROS (H₂O₂), leading to wound damage. We hypothesized that the GCNE composite could mitigate this oxidative environment because CAT could deplete H₂O₂. The ROS probe DCFH-DA was employed to detect the intracellular ROS level after L929 cells were placed into a state of oxidative stress by H₂O₂ and treated with the GCNE composite, the result of which illustrated the good ROS depletion ability of the GCNE composite (Fig. 2G and I). In addition, the intracellular oxygen level was detected to verify the O₂-release ability of the GCNE composite by using an oxygen probe, [Ru(dpp)₃]Cl₂. After the treatment, the intracellular fluorescence...
intensity of L929 cells increased significantly (Fig. 2H and J), indicating that the oxygen produced by the GCNE composite can effectively increase the oxygen supply in the cell.

2.4. Promotion of proliferation, migration, and angiogenesis of the GCNE composite in vitro

In the process of wound healing, many cells such as endothelial cells and fibroblasts, play an extremely important role in neovascularization and tissue formation. However, their life activities are inhibited by hypoxic and oxidative stress conditions of the wound site. Therefore, L929 cells and HUVECs cells pretreated with H$_2$O$_2$ were selected as model cells to verify the effects of continuous oxygen production by the GCNE composite on cell migration and proliferation. Firstly, we evaluated the promoting effect of the GCNE composite on cell proliferation using CFSE-staining. As shown in Fig. 3A and B, the fluorescence intensity of GCNE composite-treated L929 cells gradually weakened from 201.32 AU down to 55.53 AU within 72 h, while the control group weakened from 200.83 AU down to 106.04 AU. As shown in Fig. 3C and D, the fluorescence intensity of GCNE composite-treated HUVECs gradually weakened from 185.30 AU down to 49.60 AU within 72 h, while the control group weakened from 178.08 AU down to 104.76 AU. The decrease in the fluorescence intensity indicates cell division, illustrating that the oxygen produced in the glucose-containing medium can promote cell proliferation.

Generally, endothelial cells and fibroblasts contribute to tissue formation and re-epithelialization during the proliferation stage of the wound healing process. Therefore, the effect of oxygen produced by the GCNE composite on the migration of L929 and HUVECs cells pretreated with H$_2$O$_2$ was investigated through in vitro wound-healing assay. As shown in Fig. 3E–H, the cell migration rates of L929 cells and HUVECs cells within 12 h were 38.37% and 74.12%, respectively. Compared with the control group (L929 13.06%; HUVECs 25.8%), the GCNE composite can significantly promote the migration of cells ($p < 0.05$), which demonstrated that the oxygen produced from the GCNE composite could promote wound healing by promoting epithelialization.

In addition, endothelial cells were found to play an essential role in vascular function and angiogenesis. Thus, a microgel tube formation assay was carried out to evaluate the capability of angiogenesis, which showed that the tube length and branch points of HUVECs after 4 h treatment were significantly higher in the GCNE composite-treated HUVECs than those in the control group, and the grid structure was also shown in the visualization (Fig. 3I–K). These consequences suggested that the enhanced vessel-forming ability revealed that a good deal of in vitro assays was due to the production of oxygen by the GCNE composite.

The above experiments show that the GCNE composite could alleviate intracellular oxidative stress and supply sufficient oxygen to cells such as L929 cells and HUVECs, which are active in the wound healing process, resulting in accelerating the process of tissue formation and re-epithelialization. Moreover, it can also use the amino group in the chitosan hydrogel to adsorb hydrogen ions to avoid the change in pH of the wound microenvironment caused by the byproduct gluconic acid, so as to reduce or eliminate the adverse effects brought by the byproduct.

2.5. Antibacterial activity of GCNC hydrogel complex

Bacterial infection is considered to be a major obstacle to the healing process of diabetic wounds. Therefore, the antibacterial behavior of chitosan-based hydrogel was comprehensively estimated in vitro. Gram-positive S. A and Gram-negative E. coli were used as model bacteria to conduct antibacterial experiments. As illustrated in Fig. 4A and B, we found the viability of bacteria in the Glu group was higher than that in the Control group ($p < 0.05$), which means the high glucose environment is conducive to the growth of bacteria. Expectantly, our chitosan-based hydrogel showed good antibacterial activity against S. A and E. coli. According to our results, the relative viability of Glu + CS hydrogel-treated S. A and Glu + GCNC hydrogel-complex-treated S. A, respectively, reached only 12.89% and 5.25% ($p < 0.05$). Besides, the relative viability of the Glu + CS hydrogel-treated E. coli and Glu + GCNC hydrogel complex-treated E. coli, respectively, reached only 31.98% and 22.18% ($p < 0.05$). These results were because the chitosan-adsorbed H$^+$ was produced by the by-product gluconic acid according to our previous experiments (Fig. 2C), which enhanced the positive charge of chitosan and thus enhanced the antibacterial effect. The results of the visual antibacterial images were consistent with the above findings (Fig. 4C). We could observe that there were a large number of colony formations in the group control, Glu and Glu + GCNE composite. On the contrary, the hydrogel groups had a small colony formation. Overall, the excellent antibacterial property of the GCNC hydrogel complex could provide a protective barrier for wound healing.

2.6. GCNC hydrogel complex accelerates diabetic chronic wound healing in vivo

In order to further verify the effect of the GOx–CAT nanoenzyme (GCNE) composite in accelerating diabetic wound healing in vivo, we established a chronic wound model of Staphylococcus aureus (S.A) infection in diabetic SD rats for verification (Fig. 5A). The blood glucose of the tail vein was monitored to ensure the rats were in a hyperglycemic state (always higher than 16.7 mmol ml$^{-1}$). According to our results, the relative viability of Glu + H$^+$-adsorbed CS, Glu + H$^+$-adsorbed GCNC, and Glu + GCNC hydrogel complex-treated S. A, respectively, reached only 12.89% and 5.25% ($p < 0.05$). These results were because the chitosan-adsorbed H$^+$ was produced by the by-product gluconic acid according to our previous experiments (Fig. 2C), which enhanced the positive charge of chitosan and thus enhanced the antibacterial effect. The results of the visual antibacterial images were consistent with the above findings (Fig. 4C). We could observe that there were a large number of colony formations in the group control, Glu and Glu + GCNE composite. On the contrary, the hydrogel groups had a small colony formation. Overall, the excellent antibacterial property of the GCNC hydrogel complex could provide a protective barrier for wound healing.

Digital photographs of the wounds showed that the wound healing rates of S. A + hydrogel, S. A + GCNE composite, and S. A + GCNC hydrogel complex groups were faster than that of the control (73.94%) and S. A group (69.09%), with 74.81%, 82.59%, and 92.68% closure area, respectively (Fig. 5C–F). In addition, we randomly measured the blood glucose values of the wound site in healthy and diabetic rats, the results of which showed that the wound in diabetic mice was also a high glucose environment (Fig. S2f).
in the other groups, while wounds observed in the S. A + GCNC hydrogel complex group had already healed (black dotted line). Besides, S. A infection was observed to affect hair follicle growth (blue arrow), but hair follicles continued to grow normally on the skin treated with the hydrogel and GCNE composite. Furthermore, a large number of blood vessels were observed in the S. A + GCNE composite and S. A + GCNC hydrogel complex groups (red arrow), owing to oxygen production by the GCNE composite at the wound site. During the remodeling phase, proper collagen deposition is important to promote wound healing. Therefore, Masson’s trichrome staining was performed to observe newly formed collagen (Fig. S3†). The results showed that the S. A + GCNE composite and S. A + GCNC hydrogel complex group had more collagen deposition (blue). These results are attributed to the fact that the GCNE composite can generate oxygen by utilizing blood glucose in the microenvironment to promote the physiological functions of various cells, such as enhanced cell proliferation, migration, and angiogenesis, resulting in accelerating wound healing in diabetic wounds.

To verify the in vivo antibacterial effect of the GCNC hydrogel complex, the skin tissue homogenate was used for LB agar plate coating, and the results showed that S. A + CS hydrogel and S. A + GCNC hydrogel complex group had better in vivo
antibacterial effect (Fig. S4†). Additionally, in order to explore the biological safety of the system, we monitored the weight changes of rats and performed a frozen section analysis on their main tissues and organs. From Fig. 5E, it can be seen that the GOx-CAT nanoenzyme (GCNE) composite did not show a significant effect on the body weight of the rat, and the H&E staining analysis (Fig. S5†) did not demonstrate any pathological changes in the main organs of the rat, indicating that the system has good biocompatibility and no obvious toxicity.

2.7. Oxidative stress depletion and angiogenesis effect of the GCNC hydrogel complex in vivo

We detected the elimination of ROS in tissue sections to verify the in vivo alleviation of wound oxidative stress. As shown in Fig. 6A and B, the results showed lower fluorescence in the GCNE composite group than that in other groups, demonstrating that the GCNE composite could effectively eliminate ROS at the wound and alleviate oxidative stress. In addition, neovascularization was observed by CD31 immunostaining (Fig. 6C and D). There was a large amount of red fluorescence in the S. A + GCNE composite and S. A + GCNC hydrogel complex groups, which was significantly higher than that in other groups, indicating many new blood vessels, which provided stimulative factors to accelerate wound healing. Overall, by simultaneously eliminating oxidative stress, supplying oxygen, and facilitating angiogenesis, our GCNC hydrogel complex significantly accelerated the wound closure rate.

3. Materials and methods

3.1. Materials

Chitosan (deacetylation value 75.0 to 85.0%) was purchased from Shanghai Aladdin Biochemical Technology Co., Ltd.
Phosphate-buffered solution (PBS) was supplied by Beyotime Biotechnology Co., Ltd. Biological materials included D-glucose, anhydrous (Solarbio), yeast extract powder (OXOID), tryptone (OXOID), agar powder (Solarbio), Dulbecco’s Modified Eagle’s Medium (DMEM, Gibco), Endothelial Cell Medium (ECM) containing 500 ml basal medium, 25 ml FBS, 5 ml endothelial cell growth supplement and 5 ml antibiotic solution (ScienCell), penicillin–streptomycin (Yeasen), fetal bovine serum (FBS; Thermo Fisher Scientific), Matrigel Matrix Basement Membrane (BD Biosciences), Hematoxylin and Eosin Staining Kit (Yeasen), tris (4,7-diphenyl-1,10-phenanthroline) ruthenium(II) dichloride complex ([Ru(dpp)2Cl2], CAS 36309-88-3), DCFH-DA (Beyotime Biotechnology) DAPI-containing sealant (Yeasen) and 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2-H-tetrazolium bromide (MTT; Yeasen). Equipment included a blood glucose meter (Sinocare Inc.), JPB-607A portable dissolved oxygen tester (Shanghai INESA Scientific Instrument Co., Ltd), and PHS-3C pH meter (Shanghai INESA Scientific Instrument Co., Ltd). The instruments used in the experiment were all from the Core Facility of Biomedical Sciences, Xiamen University.

3.2. Characterization of GCNE composite

The GCNE composite and (chitosan-g-poly (N-isopropylacrylamide) [CS-g-PNIPAAm]) hydrogel were synthesized according to our previous research. The GCNC hydrogel complex was formed by physically mixing 5% w/v chitosan-based hydrogel with the GCNE composite. A transmission elec-
Sample pretreatment: the sample was weighted and dissolved in a small volume of double distilled water (dd water), then it was added to the total volume of double distilled water dropwise while stirring (the final concentration 1 mg ml\(^{-1}\)), stirring it overnight. 10 μl of it was dropped onto the 400 mesh copper wire and kept for adsorption (10 min).

Dynamic light scattering (DLS) measurements were conducted to investigate the size of the GCNE composite. For the injectability, adhesion, and stability of the hydrogel, 5% w/v chitosan-based hydrogel containing 1 mg ml\(^{-1}\) GCNE composite was used. Two solutions (PBS and pH = 5 PBS containing 3 mg ml\(^{-1}\) glucose, 3 mM \(\text{H}_2\text{O}_2\), and acetic acid) were added to the hydrogel, then the hydrogels were weighed at different time points.

The cumulative release of protein was detected by the BCA assay kit (SparkJade). 5% w/v chitosan-based hydrogel containing 2 mg ml\(^{-1}\) GCNE composite was used. Two groups were tested: (1) GCNC hydrogel complex + PBS; (2) GCNC hydrogel complex + pH = 5 PBS containing 3 mg ml\(^{-1}\) glucose, 3 mM \(\text{H}_2\text{O}_2\) and acetic acid. When tested, solution (2) was used as a control to eliminate the interference of the solvent.

3.3. In vitro reactivity evaluation of GCNE composite

Measurement of pH: different groups (dd water, Glu, GCNE composite, Glu + GCNE composite) were placed in a 37 °C incubator and their pH values were measured at different time points. The glucose concentration was 3 mg ml\(^{-1}\), and all the GCNE composite concentrations in this article are referred to GOx, which is 25 μg ml\(^{-1}\). The measurement of dissolved oxygen (DO): in order to verify the reaction activity of CAT and the reaction activity of the whole system, three groups of experiments were set up. The first group included dd water, 30% \(\text{H}_2\text{O}_2\), GCNE composite, and 30% \(\text{H}_2\text{O}_2\) + GCNE composite. The second group includes Glu, GCNE composite, GOx, Glu + GOx, and Glu + GCNE composite. The third group includes Glu + 250 μM \(\text{H}_2\text{O}_2\), Glu + 250 μM \(\text{H}_2\text{O}_2\) + GCNE composite. Different groups were placed in a 37 °C incubator and their dissolved oxygen content was measured at different time points. Photographs of bubble production were taken after adding 30% \(\text{H}_2\text{O}_2\) into the pre-warmed GCNC hydrogel complex.

![Fig. 6](image)

(A) Representative fluorescence images of ROS levels with DHE staining (green) and DAPI (blue). Scale bars, 200 μm. (B) The relative fluorescence intensity of ROS was quantified using ImageJ software (n = 3). (C) Representative immunofluorescence images of flaps highlighting blood vessel CD31-positive endothelial cells (red) and DAPI (blue). Scale bars, 200 μm. (D) The relative fluorescence intensity of CD31 was quantified using ImageJ software (n = 3). Error bars represent mean ± s.d. (n = 3, ***p < 0.001, **p < 0.01, *p < 0.05).
3.4. Verification of H⁺ adsorption by chitosan

In order to verify the adsorption of chitosan on H⁺, pH measurement and solubility observations were carried out. At 37 °C, the samples and glucose were dissolved in PBS, and chitosan was added to the experimental group, the final concentrations of Glu, GCNE composite and chitosan were 3 mg ml⁻¹, 25 μg ml⁻¹, and 2 mg ml⁻¹, respectively. The pH values of the three groups at different time points were detected.

3.5. Cell culture

Human umbilical vein endothelial cells (HUVECs), and L929 cells were kindly obtained from the School of Pharmaceutical Sciences, Xiamen University. HUVECs were incubated in ECM containing 5% FBS, 1% endothelial cell growth supplement, and 1% penicillin–streptomycin. L929 cells were incubated in DMEM supplemented with 10% FBS in the incubator at 37 °C and 5% CO₂.

3.6. Cytotoxicity assay

L929 cells and HUVECs were seeded in 96-well plates at an initial cell density of 1.0 × 10⁴ cells per well and incubated overnight. Afterward, different concentrations of the GCNE composite were incubated with the cells for 24 h. Then, the medium was replaced with MTT dissolved in a serum-free medium and incubated for 4 h. Finally, cell viabilities were quantified by an absorbance microplate reader (CMAX PLUS, Molecular Devices, USA) with the UV absorption at 490 nm. As for the hydrogel test, the L929 cells and HUVECs were seeded at a density of 1.0 × 10⁴ cells per well into the lower chamber of a 24-well transwell plate with 8 μm pore-sized filter inserts (LABSELECT, 14341), overnight. Then, different concentrations of the GCNE composite were added to the 5% w/v chitosan-based hydrogel (GCNC hydrogel complex: 0, 10, 25, 50 μg ml⁻¹) and the gels were pre-added in the inserts to gel at 37 °C. The inserts were added to the lower chamber for 24 h co-incubation. Next, the cell viability of L929 and HUVEC was tested by MTT assay, which was mentioned above.

3.7. Intracellular ROS (H₂O₂) depletion and O₂ evaluation assay

L929 cells were used to evaluate the intracellular ROS scavenging ability of the GCNE composite. For the ROS depletion assay, L929 cells were seeded in 48 wells plates, overnight, and incubated with FBS-free DMEM and GCNE composite (10 μg ml⁻¹ in FBS-free DMEM). After 24 h of incubation, cells were incubated with DCFH-DA (10 μM in FBS-free DMEM) for 20 min. Then, the fluorescence was detected using an inverted fluorescence microscope (Dmi8, Leica, Germany) and quantified using ImageJ software. For intracellular O₂ evaluation, L929 cells were seeded in a 48-well plate overnight and incubated with H₂O₂ (100 μM in FBS-free DMEM) and H₂O₂ + GCNE composite (10 μg ml⁻¹ in FBS-free DMEM) for 6 h. Then, the cells were incubated with [Ru(dpp)₃]Cl₂ (10 μg ml⁻¹) for another 12 h and rinsed with PBS three times to remove free [Ru(dpp)₃]Cl₂. Then, the fluorescence was detected using an inverted fluorescence microscope (Dmi8, Leica, Germany) and quantified using ImageJ software.

3.8. Cell proliferation assay

CFSE (5(6)-carboxyfluorescein diacetate succinimidyl ester; MCE) was used for visualization of cell proliferation in vitro. L929 cells and HUVECs were plated at a density of 5 × 10⁴ per well into 24-well plates and treated with H₂O₂ (50 μM) for 12 h, then, they were washed several times and 1/10 medium volume of CFSE working solution (50 μM) was added to each well and incubate for 20 min. After washing several times to remove CFSE with PBS, the cells were treated with the GCNE composite solution (dilute with medium) and the solution was changed every 12 h to prevent toxicity caused by long-term acidity (and the control group was given the corresponding medium). At different time points (0 h, 12 h, 24 h, 48 h, 72 h), the fluorescence intensity of the cells was observed and photographed under an inverted fluorescence microscope (Dmi8, Leica, Germany) and quantified using ImageJ software.

3.9. Cell migration assay

L929 cells and HUVECs were seeded in 6-well plates at a density of 5 × 10⁴ per well. After 24 h, cells filled every well. A micro-injury was created using sterile 10 μl pipette tips and washed with PBS. Then, the attached cells were treated with H₂O₂ (50 μM) for 12 h. After that, cells were treated with the GCNE composite (10 μg ml⁻¹), and the control group was treated with a serum-free medium. The cells were photographed at 0, 4, and 12 h after wounding using the inverted fluorescence microscope. The wound area was quantified using ImageJ software to evaluate the migration rate.

3.10. Tube formation assay

For the tube formation assay, pre-cooled Matrigel was added to the 24-well plates with 250 μl per well and incubated for 30 min at 37 °C to solidify the Matrigel. Afterward, HUVECs were seeded into the Matrigel-coated 24-well plates with 5 × 10⁴ per well and treated with GCNE composite (10 μg ml⁻¹) and H₂O₂ (50 μM) for 4 h (and the control group was treated with serum-free medium). The tube formation was photographed under an inverted fluorescence microscope and quantified using ImageJ software.

3.11. In vitro antibiotic activity of chitosan-based hydrogel

Gram-negative Escherichia coli (E. coli) and Gram-positive Staphylococcus aureus (S. A) were employed to evaluate the antibiotic activity of the chitosan-based hydrogel (chitosan-g-poly[N-isopropylacrylamide] [CS-g-PNIPAAm]), which had been synthesized in our previous study and treated with the following process: prepare 5% (w/v) solution with acetic acid–sodium acetate buffer (pH 6.0; Sigma) in an ice bath and store at 4 °C overnight. The bacterial suspension (100 μl, 10⁶ CFU ml⁻¹) was resuspended in a 10 ml sterilized Luria–Bertani (LB) medium. The thermo-sensitive solution was added to a 96-well plate to fill the bottom and the plate was maintained at 37 °C to form the hydrogel. Then, 150 μl of the above-mentioned suspension
was added to each well. The LB medium was used as a control and the experimental group was divided into the Glu, Glu + GCNE composite, Glu + hydrogel, and Glu + GCNE composite. The final concentrations of the Glu, GCNE composite and hydrogel were 3 mg ml⁻¹, 10 µg ml⁻¹, and 5% w/v. After that, the 96-well plate was fixed in an incubator at 37 °C with a relative humidity of not less than 90% and thoroughly shaken for 16 h. Finally, the absorbance of each well was measured at 600 nm.

For the visualization of antibacterial activity, 100 µl suspension was taken from a 96-well plate and diluted 100 times with the LB medium. Then, the diluted suspension was plated on an LB agar surface uniformly coated with E. coli (100 µl, 10⁶ CFU ml⁻¹) and S. A (100 µl, 10⁶ CFU ml⁻¹) for 18 h. Viable bacteria were observed and imaged.

3.12. Animals and diabetic models

Sprague-Dawley (SD) rats (180–220 g) were obtained from Xiamen University Laboratory Animal Center and all animal experiments were approved by the Animal Management and Ethics Committee of Xiamen University (XMULC20210054). After a week of observation, the SD rats were used in the experiment.

For the establishment of the Type 1 diabetes model, SD rats were given a single intraperitoneal injection of streptozotocin (STZ) (50 mg kg⁻¹; Yeasen), which was dissolved in sterile citrate buffer (pH 4.2–4.5). After 2 days, the blood glucose of the rat tail vein was measured using a blood glucose meter, and the unqualified rats were supplemented with STZ (30 mg kg⁻¹). Seven days later, the blood glucose of rats was measured again and rats with a plasma glucose level of higher than 16.7 mmol L⁻¹ were considered diabetic. The animals were maintained in a diabetic state throughout the wound healing process.

3.13. Chronic wound establishing, infecting, and healing

The diabetic rat S. A-infected cutaneous wound model was used to evaluate the chronic wound healing effect of the GCNC hydrogel complex. In brief, the SD rats were anesthetized with 10% chloral hydrate (3 ml kg⁻¹; MACKLIN) and shaved on their back. A round full-thickness cutaneous wound (9 mm × 9 mm) area was created on the back, after which, 50 µl of the S. A solution (10⁷ CFU ml⁻¹) was dropped onto the wound. When the wound was made, blood from the wound was collected and blood glucose was measured. The diabetic rats were randomly divided into 5 groups (n = 15) including (1) 50 µl PBS without S. A infected as control; (2) 50 µl PBS; (3) 50 µl chitosan hydrogel (5% w/v); (4) 50 µl GCNE composite solution (25 µg ml⁻¹); (5) 50 µl GCNE composite (25 µg ml⁻¹) + chitosan–hydrogel (5% w/v). In addition, another two groups of healthy SD rats were made to explore the function of the GCNE composite, which included: (1) 50 µl PBS; (2) 50 µl GCNE composite solution (25 µg ml⁻¹). Afterward, the rats were individually separated in cages and their dressings were changed every other day. The weight and blood glucose levels were monitored at different time points. Before dropping the experimental solution onto the wound surface, all the rats were anesthetized. In order to monitor the wound healing process, a digital camera and a caliper were used to measure, and photograph wounds every other day. The equation used for calculating was;

\[
\text{The wound size decrease rate} = \frac{\text{wound area at day 0} - \text{wound area at certain day}}{\text{wound area at day 0}}
\]

All the rats were sacrificed on day 14, while the healing skins and organs were extracted for the next step. In vivo anti-bacterial experiments were carried out using skin tissue homogenate. In brief, the skin tissue was weighed and 10 times the volume of PBS solution was added and homogenized in a tissue homogenizer for 5 min. The supernatant was taken and diluted 10–1000 times. 100 µl of tissue abrasive was coated on an LB plate. Colonies formed after incubation at 37 °C for 12 h were counted.

3.14. Histological analysis

For histological analysis, tissues from the skin (day 14), heart, liver, spleen, lung, and kidney were harvested and soaked in 15% and 30% sucrose solution for 24 h and 12 h, respectively. Thereafter, tissue sections were embedded in an optimal cutting temperature (OCT) compound and sliced into tissue sections with a thickness of 6 µm by using a freeze slicer (CM1900, Leica, Germany). Then, they were stained with hematoxylin & eosin (H&E) and Masson’s Trichrome stain kit (Solarbio) for observation and analysis under a microscope.

3.15. Immunofluorescence staining

After dehydration, skin tissue samples were made into frozen slices for fluorescence immunostaining. After being fixed with ice-cold acetone, the slice was rinsed in PBS and then blocked with a 5% Bovine Serum Albumin (BSA) blocking solution for 1 h. To evaluate the blood vessels, after 14 days of treatment, the skin slice was incubated with the PE anti-mouse CD31 Antibody (1 : 100; BioLegend, 102407) and dihydroethidium (DHE) (1 : 500) at 4 °C for 12 h. Afterward, the sections were washed several times with PBS to remove the staining solution. Finally, the sections were stained with DAPI-containing sealant and imaged with an inverted fluorescence microscope. The fluorescence intensity was quantified using the ImageJ software.

3.16. Statistical analysis

All experimental data and plots were calculated using Graphpad Prism 8.0. All data were presented as mean ± standard deviation (SD), and the difference between the two groups was studied using the Student’s t-test. The P-value below 0.05 was considered statistically significant differences when *P < 0.05, **P < 0.01, and ***P < 0.001 versus the indicated group.

4. Conclusion

We successfully designed and prepared a multifunctional GOx–CAT nanoenzyme–chitosan (GCNC) hydrogel complex
with advanced functions of antibacterial, rapid hemostasis, blood sugar regulation, anti-inflammatory, and continuous oxygen supply for diabetic wound healing applications. Through the high-efficiency cascade effect of GOx and CAT in the nanostructure, the system can effectively regulate the blood glucose concentration of diabetic wounds, improve the wound's oxidative micro-environment, and continuously generate oxygen to promote cell proliferation, migration, and angiogenesis, which promoted diabetic wound healing. In addition, the temperature-sensitive chitosan hydrogel can quickly stop bleeding from the wound, and the chitosan can be activated by the product of GOx catalyzed glucose (gluconic acid) to enhance the antibacterial ability, preventing bacterial infection from worsening the wound surface, thereby effectively reducing the pain of diabetic patients and increasing their compliance and tolerance. This highly efficient enzyme cascade system with good biocompatibility can also be used for other tissue injuries related to diabetes. We believe this multifunctional GCNC hydrogel complex will have a wide range of applications in translational medicine and regenerative medicine, such as organ transplantation and regeneration, or surgical tissue repair.

**Conflicts of interest**

There are no conflicts to declare.

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