



Cite this: *Nanoscale*, 2022, **14**, 602

Received 14th September 2021,
 Accepted 20th December 2021
 DOI: 10.1039/d1nr06049j
rsc.li/nanoscale

Material strategies for function enhancement in plasmonic architectures

Melissa E. King, Maria V. Fonseca Guzman and Michael B. Ross  *

Plasmonic materials are promising for applications in enhanced sensing, energy, and advanced optical communications. These applications, however, often require chemical and physical functionality that is suited and designed for the specific application. In particular, plasmonic materials need to access the wide spectral range from the ultraviolet to the mid-infrared in addition to having the requisite surface characteristics, temperature dependence, or structural features that are not intrinsic to or easily accessed by the noble metals. Herein, we describe current progress and identify promising strategies for further expanding the capabilities of plasmonic materials both across the electromagnetic spectrum and in functional areas that can enable new technology and opportunities.

Introduction

Plasmonic materials have been actively investigated at both the fundamental and applied level for decades. Early impacts were in the areas of sensing, biomedical applications, and spectroscopy, while current advancements are rapidly influencing diverse areas that include catalysis, distillation, communications, information processing, and heat transfer.^{1–7} Continued progress in these areas, however, requires enhance-

ment of the plasmonic material functionality to be suited to the specific application. However, this is challenging because the most common plasmonic materials, noble metals, have a limited plasmonic range and narrow set of chemical properties when compared to the broader set of transition metals and soft materials. Meeting the growing demands for greater material function necessitates expansion of plasmonic functionality beyond the visible and near-infrared in addition to learning to integrate chemical or physical functionality not native to noble metals.

Localized surface plasmon resonance (LSPR), refers to the collective oscillation of conduction electrons, which is the origin of efficient light absorption in plasmonic nanoparticles.

Department of Chemistry, University of Massachusetts, Lowell, Lowell, MA 01854,
 USA. E-mail: Michael_ross@uml.edu



Melissa E. King

Melissa E. King is an AAUW Postdoctoral Fellow at the University of Massachusetts, Lowell, working with Prof. Michael Ross. She earned her Ph.D. in chemistry from Wesleyan University working under the supervision of Prof. Michelle Personick and her B.S. in chemistry at Central Connecticut State University. Melissa's research focuses on understanding the properties of metallic and composite nano-

materials at the interface of inorganic chemistry and surface science, which will be leveraged to tailor structural, optical, and electronic properties towards addressing challenges in sustainable catalysis, energy, and environmental remediation.



Maria V. Fonseca Guzman

Maria Fonseca Guzman is now a Ph.D. student in chemistry at the University of California Berkeley. She received her B.S. in chemical engineering from the University of Massachusetts Lowell in 2021. Her research interests focus on energy, optical and catalytic applications of light-active nano-materials.

Using the LSPR to access new material functionalities is dependent on the integration of new kinds of functional materials and motifs and the resultant cooperative interactions between nanoscale components. Traditionally, gold and silver nanoparticles were the primary focus for plasmonic materials due to their synthetic accessibility, robust absorption, and colloidal stability.⁸ Noble and coinage metal plasmonics, however, are sharply limited in range and application due to narrow absorption regions, a relatively weak affinity for most adsorbates, and high cost.⁶ These challenges have spurred an ongoing interest in identifying new plasmonic materials that have better absorbance and increased spectral range as well as highly active surface properties. To access these enhanced properties, which would diversify plasmonic application significantly, other metals and materials must be identified.

Compositional enhancement through the incorporation of non-archetypal metals or functional ligands affords finely tuned electronic structures, broader absorbance ranges, and increased stability. Combining fundamental knowledge from synergistic areas has led to an increased number of functional plasmonic materials and an expansion of the applicable wavelengths for well-known but limited plasmonic materials. Herein, we describe progress toward identifying non-archetypal plasmonic materials with enhanced spectral activity or functional motifs that can enable the next generation of light-enhanced functional materials.

Emerging materials that support high and low energy plasmon resonances

To widen the functional plasmonic electromagnetic range that is accessible the development and integration of new plasmonic components and functional motifs is necessary. Accessing higher and lower energy regions of the spectrum towards



Michael B. Ross

Michael Ross is an Assistant Professor in the Chemistry Department at University of Massachusetts Lowell. Before that, he was awarded a CIFAR Bio-Inspired Solar Energy Postdoctoral Fellowship working at University of California, Berkeley. Michael earned his Ph. D. in chemistry at Northwestern University as a National Defense Science and Engineering (NDSEG) Fellow and his B.S. in biochemistry at Providence College. His research interests focus on understanding the unique optical, chemical, and catalytic properties of metallic nano-materials, and leveraging those properties to address challenges in energy, environmental detection, and photonics.

College. His research interests focus on understanding the unique optical, chemical, and catalytic properties of metallic nano-materials, and leveraging those properties to address challenges in energy, environmental detection, and photonics.

increased material functionality must include non-traditional components including semiconductors, transition metals, organic polymers, and two-dimensional (2D) materials. Both theoretical and experimental works have relied on using gold and silver as model systems, thus providing a comparative analysis that is thorough but limited to the visible.^{3,6} As such, the understanding of these materials, their dielectric properties, and their agreement with classical models is well understood. While not the focus of this review, for many applications it is essential to decrease the inherent loss of the LSPR for performance; we direct the reader to other reviews that discuss this in depth.^{1,6} For emerging materials, however, a better understanding of both the intrinsic material properties and their influence on LSPR is needed. This is particularly true for emerging metal nanoparticle platforms, such as Al and Bi, as well as a wide variety of potential alloys and intermetallics.^{6,9} Better understanding of the complex dielectric functions, surface oxidation, confinement, and interfacial effects will all provide guidance toward which compositions are best suited for a given application.

Ultraviolet (UV) plasmonic materials are typically metals that do not have interfering interband transitions from approximately 200–400 nm (Fig. 1 and Table 1). Interest in UV plasmonic materials is primarily driven by the opportunity for taking advantage of higher energy resonances compared to the visible; the relatively higher energy of UV photons can promote enhanced sensing due to molecular resonance effects, as well as photocatalysis and enhanced photodegradation processes that use high-energy photons for bond breaking.¹⁰ Non-traditional

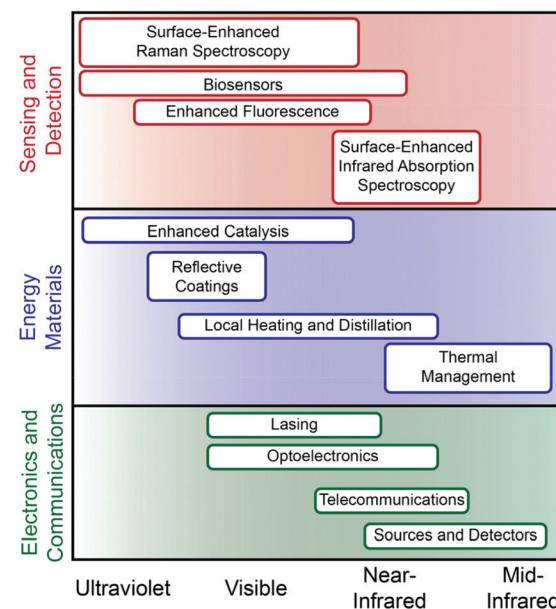


Fig. 1 Representation of functional versatility in plasmonic materials across the electromagnetic spectrum. The integration of non-archetypal plasmonic materials has greatly expanded the region of spectral activity and enhanced functionality in the areas of sensing and detection, energy materials, and electronics and communication.

Table 1 Application areas for commonly used plasmonic materials across the spectrum

Material	Applications
Al	SERS, enhanced fluorescence (100–400 nm)
Rh	SERS, biosensing, enhanced fluorescence (100–400 nm)
In	SERS, biosensing, reflective coatings (100–400 nm)
Au	Biosensing, photothermal therapy, enhanced catalysis (520–1800 nm)
Ag	Enhanced catalysis, SERS, biosensing, enhanced fluorescence, reflective coatings (400–800 nm)
Cu	Enhanced catalysis, SERS (400–800 nm)
CuS/Se/Te	Optical communications (1100–1700 nm)
Graphene	Telecommunications, sources and detectors, optoelectronics (~3000–8000 nm)
MoS ₂	Telecommunications, sources and detectors, optoelectronics (~3000–8000 nm)

ditional plasmonic metals such as aluminum, rhodium, indium, and magnesium have consistently been shown to support plasmon resonances within the UV region of the spectrum.^{10–14} Aluminum has been investigated as a thin film,¹³ as nanoparticle antennas,¹⁵ and patterned in arrays¹⁶ for enhanced Raman and fluorescence spectroscopies. Rhodium nanoparticles with spherical and tripod morphologies have been successfully used for SERS and enhanced fluorescence *via* charge transfer and theoretical studies show significant promise for biosensing in the deep-UV (>6 eV).^{11,17} Indium nanoparticles fabricated in arrays have been used for refractive index sensing¹⁸ and to enhance blue light photodetectors.¹⁹ Colloidally synthesized indium particles have been used for deep-UV SERS,²⁰ biomolecular imaging²¹ and anti-reflective coatings.²² Colloidally synthesized magnesium nanoparticles with a ~6 nm self-limiting oxide layer exhibit size- and shape-dependent plasmon resonance in the UV.²³ Dielectric function analysis suggests that bismuth, antimony, and gallium can have inducible plasmon resonances in the ultraviolet region that is mediated through interband transitions; this could create new tunability opportunities by manipulation of the metal band structure, rather than nanoparticle size or shape.²⁴

Traditional coinage metals have plasmon resonances that are found in the visible portion of the spectrum and have applications in sensing, photocatalysis, and photothermal therapy (Fig. 1 and Table 1). Gold nanoparticles can be tuned to support an LSPR from ~520–1800 nm; these have been used extensively for biosensing and photothermal therapies²⁵ as well as photocatalysis.^{26,27} Silver, like gold, has a similarly wide plasmonic range and generally supports stronger LSPRs; it has well-known applications in photocatalysis²⁸ and SERS.²⁹ Copper, while less well-studied than gold or silver, can be used for selective photocatalysis,³⁰ SERS,³¹ and single-particle nanospectroscopy.³² Heavily-doped semiconductor materials are an emerging plasmonic colloidal platform, however, they face a significant number of challenges in reaching carrier concentrations sufficient for resonance in the visible range, which must be $\sim 10^{22} \text{ cm}^{-3}$.³³ Transition metal nitrides, such as titanium nitride (TiN), have higher carrier concentrations

than their oxide counterparts, enabling them to support an LSPR in the visible.^{1,29}

Doped semiconductors do, however, provide an emerging and remarkably tunable platform for plasmonics in the near-infrared (NIR (Fig. 1 and Table 1)).³⁴ Transparent conducting oxides (TCO) such as aluminum-zinc-oxide and indium-tin-oxide can be doped to have resonances throughout the NIR.¹ Similarly, sulfides and transition metal oxides have high electron mobility allowing them to have observable metallic properties in the NIR.¹ Fluorine/indium co-doped cadmium oxide and fluorine/tin co-doped cadmium oxide are examples of cooperative effects that allow for increased charge carrier density in TCO's allowing for continuous tunability in the NIR.³⁵ Cu–S/Se plasmonic materials can also be manipulated through dopant concentrations which alter the carrier density and have shown further tunability in the NIR based on size and shape.³⁶ Such colloidal chalcogenide semiconductors composed of CuS or CuTe are self-doping through oxidation which allows for fine-tuning of the LSPR from 1100–1700 nm and are applicable for optical communication.³⁷ Transition metal oxides such as RuO₂, provide a functional branching point for plasmonic materials in the NIR due to their small negative permittivity and low losses making them ideal candidates for photonics applications.³⁸

2D materials provide a unique platform for mid-infrared (MIR) plasmonics owing to their electrical conductivity, strong light-matter interactions, and mechanical flexibility (Fig. 1 and Table 1).³⁹ Graphene displays metallic behavior and plasmon resonance in the MIR.⁴⁰ Graphene's plasmon has been used in this region for thermal energy transfer, imaging, and sensing that are tunable *via* carrier density and surface geometry.⁴¹ MoS₂ has lower carrier densities than graphene, however when layered its photoresponse increases from ~7.5 mA, which is similar to graphene, to ~100 mA, providing a comparable response to silicon of the same thickness thus making it a candidate for photoluminescent applications.³⁹ Cerium-doped indium oxide nanocubes show better optical mobility than ITO nanocrystals because Ce doping minimizes ionized impurity scattering producing exceptionally narrow line widths. The result of this is that these nanocubes exhibit the highest quality factors observed for nanocrystals in this spectral region and can be used to enhance molecular vibration signals in the MIR.⁴² Black phosphorus is an electronically anisotropic plasmonic material that has a resonance in the MIR and is also classified as a hyperbolic material, meaning that it has potential applications in broadband thermal sources and propagation of directional surface plasmons.⁴³

The extension from traditional plasmonic materials such as gold and silver, to non-noble metals, semiconductors, and 2D-materials expands the range of plasmonics from deep UV to the MIR. This broadening of the functional spectral window for plasmonics creates new opportunities for enhancing a wide variety of optical processes (Fig. 1). Furthermore, the increasing variety of material compositions that exhibit plasmonic functionality increases the environments and modalities through which these materials can be used. To fully utilize

these novel plasmonic materials, fundamental research must characterize intrinsic properties as well as cooperative interactions in functional materials. Future discoveries in this area require continued synthetic advances for accessing new materials and improved dielectric and resonant understanding of transition metals, alloys, and semiconductors.

Designing materials for enhanced physical and chemical functionality

To realize many of the emerging applications of plasmonic materials, expansion of the spectral plasmonic window is not sufficient. That in itself only expands the functional electromagnetic range over which one can absorb and use photons. Incorporating functional motifs that can provide access to enhanced chemical and physical functionality remains essential. Functional motifs including ligands, geometries, surface structure, assembly, and judicious metallic-mixing are emerging as consistent strategies for enhancing for catalysis, sensing, and optoelectronic applications (Fig. 2).

Plasmonic metal nanostructures have been shown to enhance efficiency in the conversion of solar energy to chemical energy. It has been shown that the integration of catalytically active materials with plasmonic materials can increase catalytic activity and selectivity (Fig. 2).⁵ These pairings result in hybrid plasmonic systems that direct energy into a catalytically relevant material that is not plasmonically active in that optical region.⁴⁴ Core-shell cubic Ag@Pt nanoparticles have been studied for preferential CO oxidation and as a model system for distinguishing plasmonic and catalytic effects of the constituent materials. It was observed that there was directed plasmon energy transfer from the silver core to the

platinum shell.⁴⁴ Antenna/reactor particles also provide a secondary way to drive plasmon-enhanced photocatalysis whereby the ‘antenna’ drives hot electron generation in a neighboring catalytic metal or semiconductor.⁴⁵ This proximity effect is based on the augmented field surrounding the plasmonic metal which is capable of driving a “forced” plasmon in the non-plasmonic metal. Such pairings include Aluminum/Palladium⁴⁶ for H₂ splitting and Aluminum/Iridium (Fig. 3e)⁴⁷ for nitrous oxide decomposition. Incorporation of semiconducting materials allows for more effective separation of charge carriers in plasmonic systems, as evidenced by gold nanoparticle decorated p-type gallium nitride photocathodes⁴⁸ and copper nanoparticle decorated p-type nickel oxide photocathodes⁴⁹ exhibiting enhanced activity and selectivity for photoelectrochemical CO₂ reduction. Moreover, the expanded understanding of plasmonic processes has enabled single metal systems where the incorporation of an ionic liquid medium in a photoelectrochemical cell opens an avenue towards the plasmonic photosynthesis of hydrocarbons.⁵⁰

Sensing and spectroscopic enhancement is one of the most well-known and longest studied applications of plasmonic materials. Indeed, plasmonic materials have been used in surface-enhanced Raman spectroscopy (SERS), biosensing and diagnostics, plasmon-enhanced fluorescence (PEF), and surface-enhanced infrared spectroscopy (SEIRAS) for sensing from the UV to NIR.^{51,52} Gold and silver have historically been used for such applications due to their strong plasmonic absorbance and synthetic accessibility, however, the introduction of UV-active metals and doped semiconducting materials have expanded the functional optical range.¹³ Colloidal synthetic techniques have led to chiral magnesium nanoparticles that show enhanced chiral sensing in aqueous media.⁵³ Furthermore, integration of distinct materials for plasmonic

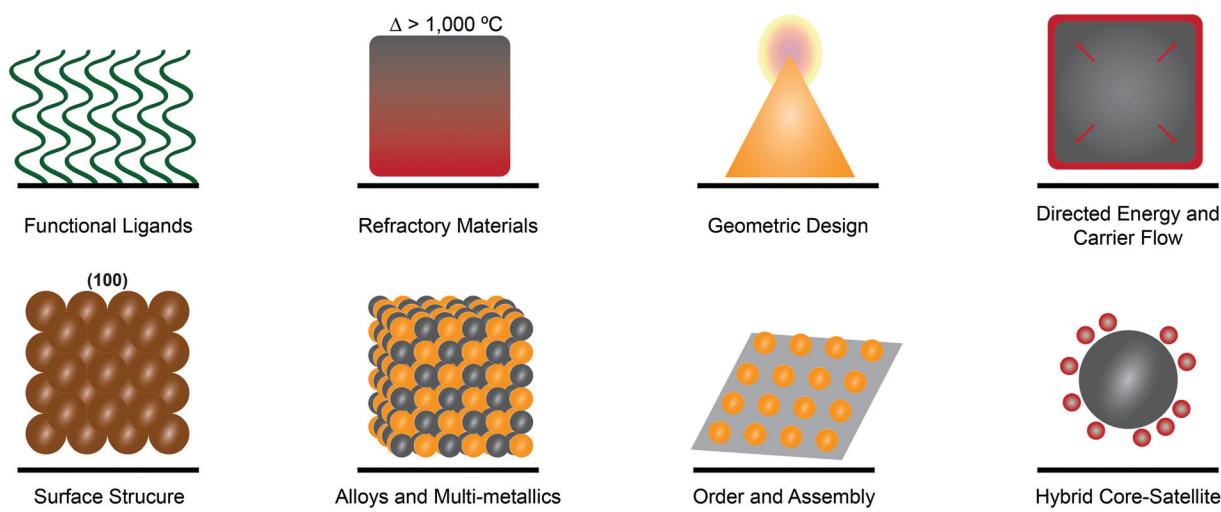


Fig. 2 Motifs and strategies for enhanced functionality in plasmonic materials. Overview of motifs that expand the functional properties of plasmonic materials. These cooperative motifs contribute to the engineering of devices which aim to harness plasmonic effects for enhancing physical and chemical processes. These include (clockwise from top left) incorporation of functional ligands, investigation of refractory materials, geometric design at the nanoscale, directing energy and carrier flow, investigating hybrid core-satellite materials, order and assembly of nanoparticles, investigating alloys and multimetallitics, and tailoring surface structure.

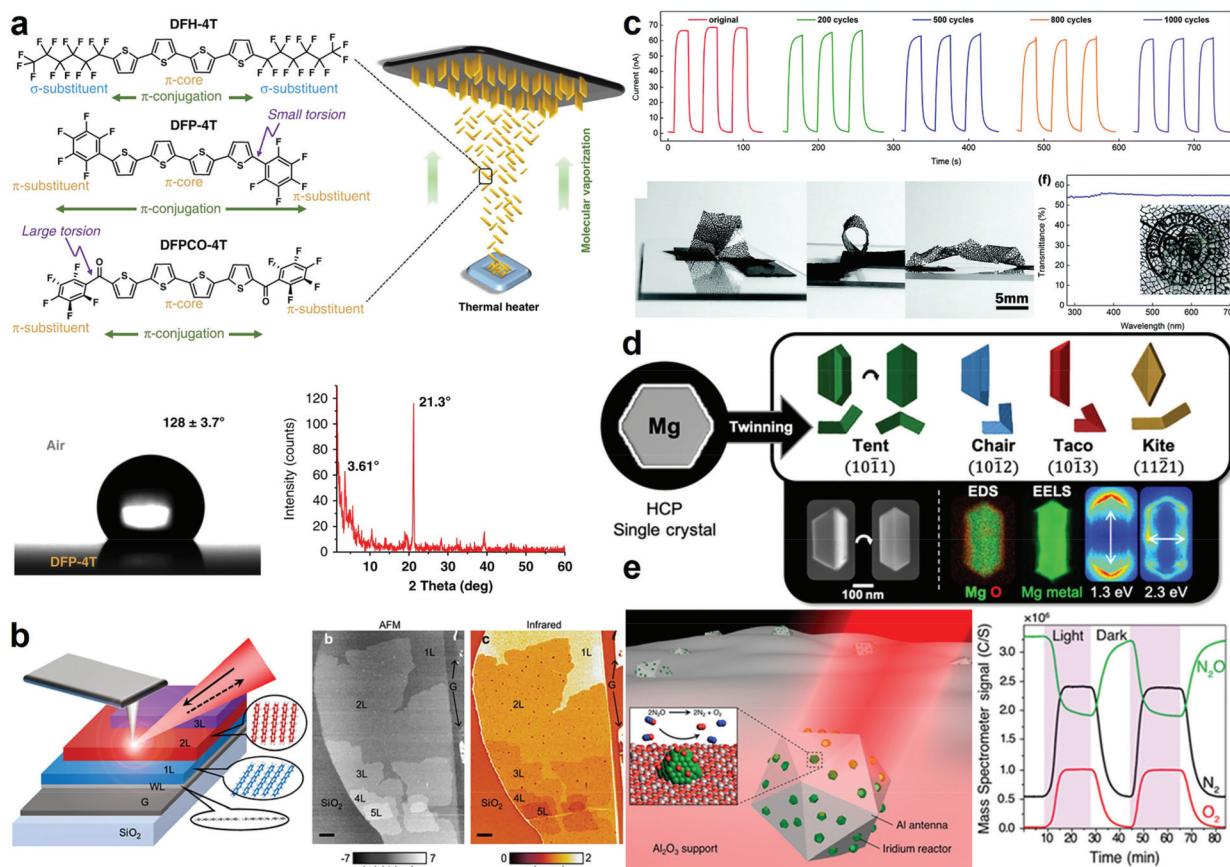


Fig. 3 Recent demonstrations of function-enhanced plasmonic materials. (a) Schematic representation of the fabrication and characterization of fully organic nanostructured SERS films. Adapted from ref. 93 under creative commons attribution 4.0 <http://creativecommons.org/licenses/by/4.0/>. (b) Illustration and AFM topography of 2D graphene/pentacene heterostructured nanomaterials. Adapted from ref. 106 with permission from the American Chemical Society. (c) Chronoamperometry measurements under 350 nm illumination of bending cycles coupled with optical images of MXene UV photodetectors. Adapted from ref. 109 with permission from the Royal Society of Chemistry. (d) Cartoon representations, SEM images, and elemental analysis of twinned Mg nanoparticles. Adapted from ref. 73. (e) Schematic representation of Al-Ir nanoparticles and reactivity under irradiation measured via mass spectrometry. Adapted from ref. 47 with permission from the American Chemical Society.

enhancement and analyte capture can further lower limits of detection (Fig. 2). Quantum dots composed of silver chalcogenides coupled with mercaptosuccinic acid have shown up to 10-fold enhancement for SEIRAS applications and are advantageous for potential applications in biological systems due to their low toxicity.⁵⁴ The inclusion of soft materials through the deliberate incorporation of functional ligands allows for increased sensing specificity, especially in biological systems.⁵⁵ Coupling peptides, which have been long used for self-assembly, with plasmonically active metallic nanoparticles produces stimuli responsive biosensing tools that have optical response that is identifiable as an SPR shift or observable color change that is enzyme specific (Fig. 2).⁵⁶

Controlled phase change, either through modification of the crystal lattice, or transition from a solid to a liquid, falls into the area of 'active plasmonics' whereby a characteristic LSPR 'signal' can be modulated in real-time.⁵⁷ Ge₃Sb₂Te₆ and Ge₂Sb₂Te₅ allow for a fast optical switch in response to thermal stimuli that originates from the pronounced changes

in the dielectric properties that occur during the phase transition between amorphous and crystalline states.⁵⁸ Single crystalline palladium nanocubes have also been shown to exhibit crystal structure phase changes from α to β as well as a redshift in the LSPR in H₂ rich environments, providing avenues towards more sensitive and specific H₂-sensing.⁵⁹ Gallium and its alloys have garnered significant interest due to near-room temperature melting points that allow for ease of phase transition and the subsequent study of metallic plasmonic properties that are free from structural defects that lead to extrinsic losses.^{60,61} The incorporation of selectively fluorinated surface ligands (Fig. 2) allows for reversible, reconfigurable assemblies of Ag nanoparticles that achieve long-term switchability and a substantial plasmonic response originating from the reconfiguration process.⁶²

Refractory materials with plasmonic properties are promising alternatives for high temperature electronic applications because of their stability at temperatures >1000 °C (Fig. 2).⁶³ Interstitial transition metal nitrides with their excellent

dopant capacities and tunable wavelengths, starting in the visible region, provide a practical solution to the problem of noble metal instability at elevated temperatures.²⁹ TiN is a well-studied refractory material that has applications in biomedical sensing, cancer therapy, photocatalysis, and optoelectronics.²⁹ The use of TiN nanoparticles embedded in Mg-based materials to transform photoexcitation to photothermal heating of a nanoparticle packed bed reactor has proved a promising alternative to the energy-intensive Haber–Bosch process for the production of ammonia.⁶⁴

The incorporation of non-traditional materials with plasmonically-active ones along with the continued improvement in nanoscale design and structure, increases opportunities for plasmonic systems. A few of the emerging motifs that enhance plasmonic properties and functionality are highlighted in Fig. 2, however this is by no means exhaustive. Unlike traditional plasmonic metal nanomaterials, such as thin films or small nanoparticles, the use of functional materials can increase catalytic activity, augment sensitivity in plasmon-enhanced sensing techniques, decrease structural defects through ease of reconfiguration, and improve stability for high temperature applications. This increased functionality coupled with spectrally versatile components leads to emergent materials with greater spectral tunability and relevance in novel photonic applications.

Emerging platforms and materials

The development of new classes and applications of plasmonic materials is rapid. These emerging areas include entirely new compositions as well as new kinds of applications in diverse fields such as quantum materials. These areas continue to build upon the themes and motifs discussed above by integrating fundamental knowledge from distinct areas. These materials combine fundamental knowledge from previous plasmonic studies to push the boundaries of this field towards plasmonic composites with broad range absorbance and greater applicability in device fabrication. Here we highlight five areas: alloys and non-archetypal materials, quantum materials, hierarchical materials, soft-matter plasmonics, and 2D materials.

Alloys and multi-metallic nanostructures have long been of interest for providing a tunable plasmonic optical response (Fig. 2).^{5,65–67} Additionally, the coupling of cooperative metals can provide access to enhanced electronic effects⁴⁴ and tunable adsorption.⁶⁸ Synthetic advances in both physical deposition and wet chemical methods have led to several new approaches. Layer-by-layer lithography and co-sputtering, for example, have led to the fabrication of supported ternary alloy arrays for hydrogen sensing⁶⁹ and ultrathin Pd–Au alloys for low-temperature catalysis,⁷⁰ respectively. Advanced synthetic techniques also provide an avenue towards efficient plasmonic materials including controlled galvanic replacement reactions for Au–Ag nanoparticles with tunable cavity size for enhanced photocatalytic hydrogen generation.^{71,72} While still novel, sig-

nificant progress has been made on the optical properties and potential applications of magnesium nanoparticles as they have sustainable plasmon resonances from the UV to NIR (Fig. 3d).^{12,73} Gallium and gallium based alloys provide a unique opportunity to manipulate the plasmon resonances and regions actively and are tunable based on composition.⁷⁴

Quantum materials combined with plasmonic ones could integrate the promise of new modes of information storage and manipulation with the intense optical enhancement provided by plasmonics.^{75,76} Incorporating plasmonic nanoparticles on quantum photonic platforms has lead faster operating speeds and functionality at room temperature.⁷⁷ Additionally, the ability for plasmonic materials to confine light can be used to create switchable and enhanced emitters, as has been shown with Ag and WS₂⁷⁸ and for metallised optical fibers.⁷⁹ The quantum materials field is rapidly advancing, and the small mode volumes and rapid optical switchability of plasmonic materials will continue to provide unique enhancement opportunities.

Hierarchical materials, meanwhile, provide opportunities to arrange plasmonic components into two- and three-dimensions, such that the arrangement of multiple compositions and their interparticle coupling can be finely controlled (Fig. 2). A number of strategies exist for the arrangement of plasmonic nanoparticles, too many to detail here.⁸ 2D gold nanoparticle superlattices provide an opportunity to better understand and manipulate nanolasers while providing insight into controllable multi-modal lasing capabilities through deliberate symmetry breaking.⁸⁰ Meanwhile, 2D Cu–Pt nanoparticle lattices can significantly enhance hydrogen evolution reaction due to long-range coupling through the lattice.⁸¹ DNA-programmed crystalline superlattices can access a wide variety of colors when using Ag and Au nanoparticles, in addition to providing fine control over the collective refractive index of the composite material.^{82,83} Composite materials of gold nanoparticles and liquid organic crystals enable cooperative interactions that lead to tunable hierarchical structures with helical motifs and inherent optical chirality.⁸⁴ More exotic techniques have also led to the coupling of gold and graphene for plasmon enhanced detection of bacteria,⁸⁵ 3D microreactors composed of aluminum with directed deposition of silver or gold particles decorating the interior of the cavity through plasmonic manipulation,⁸⁶ and tubular nanomembrane arrays with engineered hotspots for ultrasensitive SERS.⁸⁷ Thin films of cadmium oxide doped with n-type plasmonic oxide nanocrystals, comprised of indium, aluminum, and tin, exhibit tunable plasmon modes from the NIR to the MIR that are dictated by dopant concentration.⁸⁸ Metamaterials fabricated through wide-area nanoimprinting with colloid Au nanocrystal building blocks afford superstructures with plasmon resonances from the visible to NIR.⁸⁹ While the approaches to arrange plasmonic materials continue to grow, the applications that are accessible by multi-dimensional arrangement are still emerging.

Organic or “soft” materials provide an inexpensive and exceptionally tunable alternative or accompaniment to tra-

ditional plasmonic metals.⁹⁰ The key to manipulating plasmonic activity in these structures lies in continuous conjugation such that metallic properties can emerge in these systems. For example, the incorporation of additional conjugated rings into polycyclic aromatic hydrocarbons induces a red-shift in the observed plasmon resonance.⁹¹ Advances in electrochromic devices for display technologies have been accomplished through the integration of graphene derivatives through stabilization in ion-gel membranes.⁹² Other π -conjugated systems, such as fluoroarene-modified oligothiophenes, have been shown to provide SERS enhancement factors $>10^5$ (Fig. 3a).⁹³ Interestingly, conductive organic materials can behave as composites whereby a polymer matrix acts as the plasmonic material and as a redox-active, dynamically tunable nano-antenna.⁹⁴ Fully organic plasmonic materials provide an avenue for flexible plasmonic devices such as wearable biosensors or flexible display technologies that do not require metal constituents, potentially making them more cost-effective and structurally deformable.

2D materials beyond graphene offer a unique opportunity for controlled volume plasmonic materials due to their tunable thickness and unique electronic properties that enable plasmonic response over a wide frequency range (Vis – THz).⁹⁵ Indeed, coupling complementary 2D materials that harness phonon-polariton contributions to incident irradiation such as hBN, MoO₃, and V₂O₅ provide access to greater optical tunability with lower loss potential.^{96–99} Graphene analogs, such as hexagonal boron nitride can be stacked to achieve heterostructures that have plasmon-phonon hybridization which can be controlled simultaneously with the total plasmon contribution.^{100–103} MoO₃ and V₂O₅ have been recently shown to have directionally controllable phonon-polaritons along the axis of irradiation, making these materials ideally positioned for nanophotonic applications in communications and energy transmission.^{104,105} Interlayer electron transfer has been shown in similar heterostructures of pentacene and graphene through plasmon-driven tunneling-type electron transfer that is highly sensitive to molecular orientation thus, making it ideal for structural characterization of bionanoparticles (Fig. 3b).¹⁰⁶ Photonic crystal cavity lasers, which can be applied to “on-chip” optical communications, have been realized for tungsten and molybdenum sulfides, selenides, and tellurides and are active in the visible region of the spectrum with precise control over the exciton emissions at the surface of the material.¹⁰⁷ Transition metal carbides, such as InSe, are another plasmonic 2D material that have applications in electrodes and photodetectors (Fig. 3c).^{108,109} Finally, 2D transition-metal chalcogenides were recently shown to have significant SERS enhancement due to chemical effects, as opposed to the typically more significant electromagnetic effect.¹¹⁰

Outlook

The combination of new building blocks and synthetic techniques enables the creation of novel nanostructured plasmonic

materials with optical properties ranging across the electromagnetic spectrum and enhanced chemical and physical functionality. The cooperative integration of functional motifs gives rise to advanced material properties, we have highlighted several relevant in fields including catalysis, detection, and communications.

The materials highlighted herein could provide a basis for the improvement or enhancement of plasmonic materials and photonic devices, particularly in areas that include:^{64,68,71,85,87,91,95,100,106}

- Lowering the energy barriers for catalytic reactions
- Electronic structure control for sensing enhancement and surface sensitivity
- Hybridization of plasmon-phonon interaction for next-generation communications devices

Meanwhile, the combination of refractory materials with non-archetypal metals, for example, has led to the light-driven production of ammonia.⁶⁴ Moreover, the manipulation of active plasmonic materials provides new platforms for the real-time manipulation of dielectric properties⁵⁸ and all-organic reversible photoswitches.⁹⁴ 2D materials provide access to terahertz communication frequencies at room temperature.⁷⁸ Finally, material advances are changing our perception of SERS, a relatively mature application of plasmonics, where organic semiconductors⁹³ and 2D molybdenum telluride¹¹⁰ can provide significant chemical enhancement effects that may enable highly specific detection platforms.

Future technologies that rely on plasmonic materials will continue to require materials innovation that provides high quality response across the electromagnetic spectrum while incorporating and increasing the diversity of functional motifs. Integrating materials, such as metals and oxides, continues to be important both for intrinsic performance and for complementary metal oxide semiconductor (CMOS)-compatibility. Continued effort should be made to investigate non-archetypal plasmonic materials, including multi-metallic structures, non-noble metals such as Ga, In, and Mg, and organic and 2D materials. A particular challenge in this area is the lack of established and consistent dielectric function information, due to variability in the electronic properties due to differences in processing conditions and synthesis. Because many applications of plasmonic materials require consistent performance in harsh conditions, such as catalysis, distillation, and thermal management, continued investigation of refractory materials and plasmonic response in extreme conditions is essential. Overall, novel functionality in plasmonic nanomaterials continues to be driven by nanoscale integration, robust characterization, and thoughtful combination of functional components. Continued breakthroughs in these areas will expand the spectral and functional versatility of plasmonic materials.

Conflicts of interest

The authors declare no competing financial interests.

Acknowledgements

This work relates to Department of Navy award N00014-20-1-2858 issued by the Office of Naval Research. The United States Government has a royalty-free license throughout the world in all copyrightable material contained herein. This material was also supported by the University of Massachusetts Lowell and the Commonwealth of Massachusetts. M. E. K. gratefully acknowledges support from the American Association of University Women American Postdoctoral fellowship. M. V. F. G. gratefully acknowledges support from the RIST Institute for Sustainability and Energy.

References

- 1 G. V. Naik, V. M. Shalaev and A. Boltasseva, *Adv. Mater.*, 2013, **25**, 3264–3294.
- 2 M. I. Stockman, K. Kneipp, S. I. Bozhevolnyi, S. Saha, A. Dutta, J. Ndukaife, N. Kinsey, H. Reddy, U. Guler, V. M. Shalaev, A. Boltasseva, B. Gholipour, H. N. S. Krishnamoorthy, K. F. MacDonald, C. Soci, N. I. Zheludev, V. Savinov, R. Singh, P. Groß, C. Lienau, M. Vadai, M. L. Solomon, D. R. Barton, M. Lawrence, J. A. Dionne, S. V. Boriskina, R. Esteban, J. Aizpurua, X. Zhang, S. Yang, D. Wang, W. Wang, T. W. Odom, N. Accanto, P. M. de Roque, I. M. Hancu, L. Piatkowski, N. F. van Hulst and M. F. Kling, *J. Opt.*, 2018, **20**, 043001.
- 3 P. R. West, S. Ishii, G. V. Naik, N. K. Emani, V. M. Shalaev and A. Boltasseva, *Laser Photonics Rev.*, 2010, **4**, 795–808.
- 4 Y. Zhong, S. D. Malagari, T. Hamilton and D. Wasserman, *J. Nanophotonics*, 2015, **9**, 093791.
- 5 S. Linic, S. Chavez and R. Elias, *Nat. Mater.*, 2021, **20**(7), 916–924.
- 6 M. G. Blaber, M. D. Arnold and M. J. Ford, *J. Phys.: Condens. Matter*, 2010, **22**, 143201.
- 7 J. M. Luther, P. K. Jain, T. Ewers and A. P. Alivisatos, *Nat. Mater.*, 2011, **10**, 361–366.
- 8 M. R. Jones, K. D. Osberg, R. J. Macfarlane, M. R. Langille and C. A. Mirkin, *Chem. Rev.*, 2011, **111**, 3736–3827.
- 9 M. W. Knight, N. S. King, L. Liu, H. O. Everitt, P. Nordlander and N. J. Halas, *ACS Nano*, 2014, **8**, 834–840.
- 10 J. M. McMahon, G. C. Schatz and S. K. Gray, *Phys. Chem. Chem. Phys.*, 2013, **15**, 5415–5423.
- 11 A. Ahmadivand, R. Sinha, S. Kaya and N. Pala, *Plasmonics*, 2016, **11**, 839–849.
- 12 E. Ringe, *J. Phys. Chem. C*, 2020, **124**, 15665–15679.
- 13 B. Sharma, M. F. Cardinal, M. B. Ross, A. B. Zrimsek, S. V. Bykov, D. Punihale, S. A. Asher, G. C. Schatz and R. P. Van Duyne, *Nano Lett.*, 2016, **16**, 7968–7973.
- 14 M. V. Fonseca Guzman and M. B. Ross, *J. Phys. Chem. C*, 2021, **125**(35), 19428–19437.
- 15 M. W. Knight, L. Liu, Y. Wang, L. Brown, S. Mukherjee, N. S. King, H. O. Everitt, P. Nordlander and N. J. Halas, *Nano Lett.*, 2012, **12**, 6000–6004.
- 16 G. Maidecchi, G. Gonella, R. Proietti Zaccaria, R. Moroni, L. Anghinolfi, A. Giglia, S. Nannarone, L. Mattera, H.-L. Dai, M. Canepa and F. Bisio, *ACS Nano*, 2013, **7**, 5834–5841.
- 17 S. Kundu, Y. Chen, W. Dai, L. Ma, A. M. Sinyukov and H. Liang, *J. Mater. Chem. C*, 2017, **5**, 2577–2590.
- 18 X. Xu, X. Hu, X. Chen, Y. Kang, Z. Zhang, K. B. Parizi and H. S. P. Wong, *ACS Appl. Mater. Interfaces*, 2016, **8**, 31871–31877.
- 19 Y. Wang, C.-W. Ge, Y.-F. Zou, R. Lu, K. Zheng, T.-F. Zhang, Y.-Q. Yu and L.-B. Luo, *Adv. Opt. Mater.*, 2016, **4**, 291–296.
- 20 Y. Kumamoto, A. Taguchi, M. Honda, K. Watanabe, Y. Saito and S. Kawata, *ACS Photonics*, 2014, **1**, 598–603.
- 21 Y. Kumamoto, A. Taguchi and S. Kawata, *Adv. Opt. Mater.*, 2019, **7**, 1801099.
- 22 W.-J. Ho, H.-Y. Yang, J.-J. Liu, P.-J. Lin and C.-H. Ho, *Appl. Surf. Sci.*, 2020, **508**, 145275.
- 23 J. S. Biggins, S. Yazdi and E. Ringe, *Nano Lett.*, 2018, **18**, 3752–3758.
- 24 J. Toudert and R. Serna, *Opt. Mater. Express*, 2016, **6**, 2434–2447.
- 25 X. Huang, P. K. Jain, I. H. El-Sayed and M. A. El-Sayed, *Lasers Med Sci.*, 2007, **23**, 217.
- 26 C. Wang, X.-G. Nie, Y. Shi, Y. Zhou, J.-J. Xu, X.-H. Xia and H.-Y. Chen, *ACS Nano*, 2017, **11**, 5897–5905.
- 27 S. Mukherjee, F. Libisch, N. Large, O. Neumann, L. V. Brown, J. Cheng, J. B. Lassiter, E. A. Carter, P. Nordlander and N. J. Halas, *Nano Lett.*, 2013, **13**, 240–247.
- 28 P. Christopher, H. Xin and S. Linic, *Nat. Chem.*, 2011, **3**, 467–472.
- 29 U. Guler, V. M. Shalaev and A. Boltasseva, *Mater. Today*, 2015, **18**, 227–237.
- 30 A. Marimuthu, J. Zhang and S. Linic, *Science*, 2013, **339**, 1590–1593.
- 31 N. E. Markina, E. K. Volkova, A. M. Zakharevich, I. Y. Goryacheva and A. V. Markin, *Microchim. Acta*, 2018, **185**, 481.
- 32 Q.-C. Sun, Y. Ding, S. M. Goodman, H. H. Funke and P. Nagpal, *Nanoscale*, 2014, **6**, 12450–12457.
- 33 G. V. Naik, J. Kim and A. Boltasseva, *Opt. Mater. Express*, 2011, **1**, 1090–1099.
- 34 A. Agrawal, R. W. Johns and D. J. Milliron, *Annu. Rev. Mater. Res.*, 2017, **47**, 1–31.
- 35 X. Ye, J. Fei, B. T. Diroll, T. Paik and C. B. Murray, *J. Am. Chem. Soc.*, 2014, **136**, 11680–11686.
- 36 S.-W. Hsu, W. Bryks and A. R. Tao, *Chem. Mater.*, 2012, **24**, 3765–3771.
- 37 I. Kriegel, C. Jiang, J. Rodríguez-Fernández, R. D. Schaller, D. V. Talapin, E. da Como and J. Feldmann, *J. Am. Chem. Soc.*, 2012, **134**, 1583–1590.
- 38 L. Wang, C. Clavero, K. Yang, E. Radue, M. T. Simons, I. Novikova and R. A. Lukaszew, *Opt. Express*, 2012, **20**, 8618–8628.
- 39 Y. Li, Z. Li, C. Chi, H. Shan, L. Zheng and Z. Fang, *Adv. Sci.*, 2017, **4**, 1600430.

40 F. J. García de Abajo, *ACS Photonics*, 2014, **1**, 135–152.

41 V. W. Brar, M. C. Sherrott, M. S. Jang, S. Kim, L. Kim, M. Choi, L. A. Sweatlock and H. A. Atwater, *Nat. Commun.*, 2015, **6**, 7032.

42 E. L. Runnerstrom, A. Bergerud, A. Agrawal, R. W. Johns, C. J. Dahlman, A. Singh, S. M. Selbach and D. J. Milliron, *Nano Lett.*, 2016, **16**, 3390–3398.

43 A. Nemilentsau, T. Low and G. Hanson, *Phys. Rev. Lett.*, 2016, **116**, 066804.

44 U. Aslam, S. Chavez and S. Linic, *Nat. Nanotechnol.*, 2017, **12**, 1000–1005.

45 K. Li, N. J. Hogan, M. J. Kale, N. J. Halas, P. Nordlander and P. Christopher, *Nano Lett.*, 2017, **17**, 3710–3717.

46 H. Robatjazi, M. Lou, B. D. Clark, C. R. Jacobson, D. F. Swearer, P. Nordlander and N. J. Halas, *Nano Lett.*, 2020, **20**, 4550–4557.

47 D. F. Swearer, H. Robatjazi, J. M. P. Martirez, M. Zhang, L. Zhou, E. A. Carter, P. Nordlander and N. J. Halas, *ACS Nano*, 2019, **13**, 8076–8086.

48 J. S. DuChene, G. Tagliabue, A. J. Welch, W.-H. Cheng and H. A. Atwater, *Nano Lett.*, 2018, **18**, 2545–2550.

49 J. S. DuChene, G. Tagliabue, A. J. Welch, X. Li, W.-H. Cheng and H. A. Atwater, *Nano Lett.*, 2020, **20**, 2348–2358.

50 S. Yu and P. K. Jain, *ACS Energy Lett.*, 2019, **4**, 2295–2300.

51 M. F. Cardinal, E. Vander Ende, R. A. Hackler, M. O. McAnally, P. C. Stair, G. C. Schatz and R. P. Van Duyne, *Chem. Soc. Rev.*, 2017, **46**, 3886–3903.

52 J. Langer, D. Jimenez de Aberasturi, J. Aizpurua, R. A. Alvarez-Puebla, B. Auguié, J. J. Baumberg, G. C. Bazan, S. E. J. Bell, A. Boisen, A. G. Brolo, J. Choo, D. Cialla-May, V. Deckert, L. Fabris, K. Faulds, F. J. García de Abajo, R. Goodacre, D. Graham, A. J. Haes, C. L. Haynes, C. Huck, T. Itoh, M. Käll, J. Kneipp, N. A. Kotov, H. Kuang, E. C. Le Ru, H. K. Lee, J.-F. Li, X. Y. Ling, S. A. Maier, T. Mayerhöfer, M. Moskovits, K. Murakoshi, J.-M. Nam, S. Nie, Y. Ozaki, I. Pastoriza-Santos, J. Perez-Juste, J. Popp, A. Pucci, S. Reich, B. Ren, G. C. Schatz, T. Shegai, S. Schlücker, L.-L. Tay, K. G. Thomas, Z.-Q. Tian, R. P. Van Duyne, T. Vo-Dinh, Y. Wang, K. A. Willets, C. Xu, H. Xu, Y. Xu, Y. S. Yamamoto, B. Zhao and L. M. Liz-Marzán, *ACS Nano*, 2020, **14**, 28–117.

53 H.-H. Jeong, A. G. Mark and P. Fischer, *Chem. Commun.*, 2016, **52**, 12179–12182.

54 C. F. Pereira, I. M. A. Viegas, I. G. Souza Sobrinha, G. Pereira, G. A. L. Pereira, P. Krebs and B. Mizaikoff, *J. Mater. Chem. C*, 2020, **8**, 10448–10455.

55 P. D. Howes, R. Chandrawati and M. M. Stevens, *Science*, 2014, **346**, 1247390.

56 C. Pigliacelli, R. Sánchez-Fernández, M. D. García, C. Peinador and E. Pazos, *Chem. Commun.*, 2020, **56**, 8000–8014.

57 K. F. MacDonald and N. I. Zheludev, *Laser Photonics Rev.*, 2010, **4**, 562–567.

58 N. Jiang, X. Zhuo and J. Wang, *Chem. Rev.*, 2018, **118**, 3054–3099.

59 A. Baldi, T. C. Narayan, A. L. Koh and J. A. Dionne, *Nat. Mater.*, 2014, **13**, 1143–1148.

60 S. R. C. Vivekchand, C. J. Engel, S. M. Lubin, M. G. Blaber, W. Zhou, J. Y. Suh, G. C. Schatz and T. W. Odom, *Nano Lett.*, 2012, **12**, 4324–4328.

61 M. G. Blaber, C. J. Engel, S. R. C. Vivekchand, S. M. Lubin, T. W. Odom and G. C. Schatz, *Nano Lett.*, 2012, **12**, 5275–5280.

62 M. Bagiński, E. Tomczyk, A. Vetter, R. N. S. Suryadharma, C. Rockstuhl and W. Lewandowski, *Chem. Mater.*, 2018, **30**, 8201–8210.

63 U. Guler, A. Boltasseva and V. M. Shalaev, *Science*, 2014, **344**, 263–264.

64 D. F. Swearer, N. R. Knowles, H. O. Everitt and N. J. Halas, *ACS Energy Lett.*, 2019, **4**, 1505–1512.

65 M. Rebello Sousa Dias and M. S. Leite, *Acc. Chem. Res.*, 2019, **52**, 2881–2891.

66 F. A. A. Nugroho, B. Iandolo, J. B. Wagner and C. Langhammer, *ACS Nano*, 2016, **10**, 2871–2879.

67 W.-T. Koo, J. E. Millstone, P. S. Weiss and I.-D. Kim, *ACS Nano*, 2020, **14**, 6407–6413.

68 U. Aslam, V. G. Rao, S. Chavez and S. Linic, *Nat. Catal.*, 2018, **1**, 656–665.

69 I. Darmadi, F. A. A. Nugroho, S. Kadkhodazadeh, J. B. Wagner and C. Langhammer, *ACS Sens.*, 2019, **4**, 1424–1432.

70 J. P. McClure, J. Boltersdorf, D. R. Baker, T. G. Farinha, N. Dzuricky, C. E. P. Villegas, A. R. Rocha and M. S. Leite, *ACS Appl. Mater. Interfaces*, 2019, **11**, 24919–24932.

71 X. Yue, J. Hou, H. Zhao, P. Wu, Y. Guo, Q. Shi, L. Chen, S. Peng, Z. Liu and G. Cao, *J. Energy Chem.*, 2020, **49**, 1–7.

72 P.-C. Chen, X. Liu, J. L. Hedrick, Z. Xie, S. Wang, Q.-Y. Lin, M. C. Hersam, V. P. Dravid and C. A. Mirkin, *Science*, 2016, **352**, 1565–1569.

73 J. Asselin, C. Boukouvala, E. R. Hopper, Q. M. Ramasse, J. S. Biggins and E. Ring, *ACS Nano*, 2020, **14**, 5968–5980.

74 P. Reineck, Y. Lin, B. C. Gibson, M. D. Dickey, A. D. Greentree and I. S. Maksymov, *Sci. Rep.*, 2019, **9**, 5345.

75 M. S. Tame, K. R. McEnery, S. K. Özdemir, J. Lee, S. A. Maier and M. S. Kim, *Nat. Phys.*, 2013, **9**, 329–340.

76 Z. Jacob and V. M. Shalaev, *Science*, 2011, **334**, 463–464.

77 S. I. Bogdanov, A. Boltasseva and V. M. Shalaev, *Science*, 2019, **364**, 532–533.

78 C. Han and J. Ye, *Nat. Commun.*, 2020, **11**, 713.

79 S. Keramati, A. Passian, V. Khullar, J. Beck, C. Uiterwaal and H. Batelaan, *New J. Phys.*, 2020, **22**, 083069.

80 D. Wang, A. Yang, W. Wang, Y. Hua, R. D. Schaller, G. C. Schatz and T. W. Odom, *Nat. Nanotechnol.*, 2017, **12**, 889–894.

81 S. Deng, B. Zhang, P. Choo, P. J. M. Smeets and T. W. Odom, *Nano Lett.*, 2021, **21**, 1523–1529.

82 M. B. Ross, J. C. Ku, B. Lee, C. A. Mirkin and G. C. Schatz, *Adv. Mater.*, 2016, **28**, 2790–2794.

83 M. B. Ross, J. C. Ku, V. M. Vaccarezza, G. C. Schatz and C. A. Mirkin, *Nat. Nanotechnol.*, 2015, **10**, 453–458.

84 M. Bagiński, M. Tupikowska, G. González-Rubio, M. Wójcik and W. Lewandowski, *Adv. Mater.*, 2020, **32**, 1904581.

85 R. Siavash Moakhar, T. AbdelFatah, A. Sanati, M. Jalali, S. E. Flynn, S. S. Mahshid and S. Mahshid, *ACS Appl. Mater. Interfaces*, 2020, **12**, 23298–23310.

86 Z. Wang, B. Ai, Y. Wang, Y. Guan, H. Möhwald and G. Zhang, *ACS Appl. Mater. Interfaces*, 2019, **11**, 35429–35437.

87 X. Fan, Q. Hao, M. Li, X. Zhang, X. Yang, Y. Mei and T. Qiu, *ACS Appl. Mater. Interfaces*, 2020, **12**, 28783–28791.

88 B. T. Diroll, T. R. Gordon, E. A. Gaulding, D. R. Klein, T. Paik, H. J. Yun, E. D. Goodwin, D. Damodhar, C. R. Kagan and C. B. Murray, *Chem. Mater.*, 2014, **26**, 4579–4588.

89 A. T. Fafarman, S.-H. Hong, H. Caglayan, X. Ye, B. T. Diroll, T. Paik, N. Engheta, C. B. Murray and C. R. Kagan, *Nano Lett.*, 2013, **13**, 350–357.

90 A. J. Wilson and K. A. Willets, *Annu. Rev. Anal. Chem.*, 2016, **9**, 27–43.

91 A. Lauchner, A. E. Schlather, A. Manjavacas, Y. Cui, M. J. McClain, G. J. Stec, F. J. García de Abajo, P. Nordlander and N. J. Halas, *Nano Lett.*, 2015, **15**, 6208–6214.

92 G. J. Stec, A. Lauchner, Y. Cui, P. Nordlander and N. J. Halas, *ACS Nano*, 2017, **11**, 3254–3261.

93 G. Demirel, R. L. M. Giesecking, R. Ozdemir, S. Kahmann, M. A. Loi, G. C. Schatz, A. Facchetti and H. Usta, *Nat. Commun.*, 2019, **10**, 5502.

94 S. Chen, E. S. H. Kang, M. Shiran Chaharsoughi, V. Stanishev, P. Kühne, H. Sun, C. Wang, M. Fahlman, S. Fabiano, V. Darakchieva and M. P. Jonsson, *Nat. Nanotechnol.*, 2020, **15**, 35–40.

95 Z. Dai, G. Hu, Q. Ou, L. Zhang, F. Xia, F. J. Garcia-Vidal, C.-W. Qiu and Q. Bao, *Chem. Rev.*, 2020, **120**(13), 6197–6246.

96 N. Rivera, T. Christensen and P. Narang, *Nano Lett.*, 2019, **19**, 2653–2660.

97 S. Foteinopoulou, G. C. R. Devarapu, G. S. Subramania, S. Krishna and D. Wasserman, *Nanophotonics*, 2019, **8**, 2129–2175.

98 S. Castilla, I. Vangelidis, V.-V. Pusapati, J. Goldstein, M. Autore, T. Slipchenko, K. Rajendran, S. Kim, K. Watanabe, T. Taniguchi, L. Martín-Moreno, D. Englund, K.-J. Tielrooij, R. Hillenbrand, E. Lidorikis and F. H. L. Koppens, *Nat. Commun.*, 2020, **11**, 4872.

99 Z. Tong, T. Dumitrică and T. Frauenheim, *Nano Lett.*, 2021, **21**, 4351–4356.

100 Y. Jia, H. Zhao, Q. Guo, X. Wang, H. Wang and F. Xia, *ACS Photonics*, 2015, **2**, 907–912.

101 S. Dai, Q. Ma, M. K. Liu, T. Andersen, Z. Fei, M. D. Goldflam, M. Wagner, K. Watanabe, T. Taniguchi, M. Thiemens, F. Keilmann, G. C. A. M. Janssen, S. E. Zhu, P. Jarillo-Herrero, M. M. Fogler and D. N. Basov, *Nat. Nanotechnol.*, 2015, **10**, 682–686.

102 X. Yang, F. Zhai, H. Hu, D. Hu, R. Liu, S. Zhang, M. Sun, Z. Sun, J. Chen and Q. Dai, *Adv. Mater.*, 2016, **28**, 2931–2938.

103 A. Woessner, M. B. Lundeberg, Y. Gao, A. Principi, P. Alonso-González, M. Carrega, K. Watanabe, T. Taniguchi, G. Vignale, M. Polini, J. Hone, R. Hillenbrand and F. H. L. Koppens, *Nat. Mater.*, 2015, **14**, 421–425.

104 W. Dong, R. Qi, T. Liu, Y. Li, N. Li, Z. Hua, Z. Gao, S. Zhang, K. Liu, J. Guo and P. Gao, *Adv. Mater.*, 2020, **32**, 2002014.

105 J. Taboada-Gutiérrez, G. Álvarez-Pérez, J. Duan, W. Ma, K. Crowley, I. Prieto, A. Bylinkin, M. Autore, H. Volkova, K. Kimura, T. Kimura, M. H. Berger, S. Li, Q. Bao, X. P. A. Gao, I. Errea, A. Y. Nikitin, R. Hillenbrand, J. Martín-Sánchez and P. Alonso-González, *Nat. Mater.*, 2020, **19**, 964–968.

106 F. Hu, M. Kim, Y. Zhang, Y. Luan, K. M. Ho, Y. Shi, C. Z. Wang, X. Wang and Z. Fei, *Nano Lett.*, 2019, **19**, 6058–6064.

107 S. Wu, S. Buckley, J. R. Schaibley, L. Feng, J. Yan, D. G. Mandrus, F. Hatami, W. Yao, J. Vučković, A. Majumdar and X. Xu, *Nature*, 2015, **520**, 69–72.

108 Y. Yang, J. Jeon, J.-H. Park, M. S. Jeong, B. H. Lee, E. Hwang and S. Lee, *ACS Nano*, 2019, **13**, 8804–8810.

109 J. Chen, Z. Li, F. Ni, W. Ouyang and X. Fang, *Mater. Horiz.*, 2020, **7**(7), 1828–1833.

110 J. P. Fraser, P. Postnikov, E. Miliutina, Z. Kolska, R. Valiev, V. Švorčík, O. Lyutakov, A. Y. Ganin and O. Guselnikova, *ACS Appl. Mater. Interfaces*, 2020, **12**, 47774–47783.