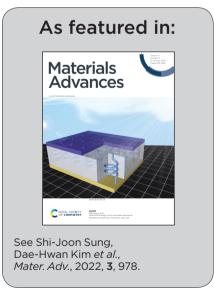


Showcasing research from Dr Kim & Dr Sung's Group, Division of Energy Technology, Convergence Research Institute, DGIST, Daegu, Korea.

Facile growth of a  ${\rm Sb_2Se_3}$  nanorods array induced by a  ${\rm MoSe_2}$  interlayer and its application in 3D p-n junction solar cells

Co-evaporation process was used to deposit uniformly grown  $\mathrm{Sb}_2\mathrm{Se}_3$  nanorods array using a  $\mathrm{MoSe}_2$  interlayer. The  $\mathrm{MoSe}_2$  interlayer acted as a crucial role as a seed layer for the preferential growth of the  $\mathrm{Sb}_2\mathrm{Se}_3$  nanorods and also improved the contact quality between the  $\mathrm{Sb}_2\mathrm{Se}_3$  nanorods array and a  $\mathrm{Mo}$  substrate. The application of the  $\mathrm{Sb}_2\mathrm{Se}_3$  nanorods array to 3D p-n junction high-efficiency  $\mathrm{Sb}_2\mathrm{Se}_3$  solar cells was also presented.





# Materials Advances



PAPER
View Article Online
View Journal | View Issue



Cite this: *Mater. Adv.*, 2022, **3** 978

Received 3rd September 2021, Accepted 19th December 2021

DOI: 10.1039/d1ma00804h

rsc.li/materials-advances

# Facile growth of a Sb<sub>2</sub>Se<sub>3</sub> nanorod array induced by a MoSe<sub>2</sub> interlayer and its application in 3D p-n junction solar cells†

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A uniformly grown  $Sb_2Se_3$  nanorod array, with the introduction of a  $MoSe_2$  interlayer, obtained by a coevaporation process and its application in three-dimensional (3D) p-n junction high-efficiency  $Sb_2Se_3$  solar cells were investigated in this study. The  $MoSe_2$  interlayer played a crucial role as a seed layer for the preferential growth of  $Sb_2Se_3$  crystals, which facilitated the formation of a  $Sb_2Se_3$  nanorod array regardless of the process conditions. 3D p-n junction between the  $Sb_2Se_3$  nanorod array and the CdS buffer layer improved the short-circuit current of  $Sb_2Se_3$  solar cells due to improved carrier transportation from the  $Sb_2Se_3$  absorber to the CdS buffer. The  $MoSe_2$  interlayer also improved the contact quality between the  $Sb_2Se_3$  nanorod array and the Mo substrate by forming a quasi-ohmic contact, which resulted in a higher open-circuit voltage due to a reduced contact barrier and series resistance in  $Sb_2Se_3$  solar cells. The crystal growth rate of  $Sb_2Se_3$  was controlled by the source evaporation rate and substrate temperature to tune the final nanostructure and crystalline orientation of the co-evaporated  $Sb_2Se_3$  nanorods array. 3D p-n junction solar cells based on an ordered and (hk1) preferentially oriented  $Sb_2Se_3$  nanorod array showed a power conversion efficiency of  $Sb_2Se_3$  solar cells.

#### Introduction

Antimony selenide  $(Sb_2Se_3)$  is one of the most promising semiconductor materials for thin-film solar cells because of its outstanding optoelectronic properties. Additionally, it is a non-toxic, relatively low-cost material with high absorption coefficient and low processing temperature. Therefore, significant research has been conducted in the past decade to replace the conventional multinary compound semiconductors in thin-film solar cells for  $Sb_2Se_3$ . However, the best power conversion efficiency (PCE) reported so far for  $Sb_2Se_3$  solar cells is approximately 10%, which is similar to those of thin-film solar cells made of conventional materials. In order to overcome the efficiency limitation of the  $Sb_2Se_3$  solar cells, the application of three-dimensional (3D) nanostructured  $Sb_2Se_3$ 

Carrier transportation and p-n junction formation in Sb<sub>2</sub>Se<sub>3</sub> solar cells are closely related to the Sb<sub>2</sub>Se<sub>3</sub> nanostructure. Therefore, the latter is one of the critical factors determining the photovoltaic performance of Sb<sub>2</sub>Se<sub>3</sub> solar cells. However, there is little study on the nanostructure control of the Sb<sub>2</sub>Se<sub>3</sub> thin films for high efficiency solar cells. Li et al. improved the short-circuit current ( $I_{SC}$ ) of a solar cell using a 1000 nm-high Sb<sub>2</sub>Se<sub>3</sub>-nanorod array with superior light-harvesting capacity and reduced optical reflection.<sup>16</sup> In our previous work, we investigated the evolution of Sb<sub>2</sub>Se<sub>3</sub> nanorods by controlling the process temperature of co-evaporation and proposed the photovoltaic performance enhancement of Sb<sub>2</sub>Se<sub>3</sub> solar cells based on 3D p-n junction formation.<sup>18</sup> However, the temperature window for the formation of Sb<sub>2</sub>Se<sub>3</sub> nanorods was narrow and the orientation of the Sb<sub>2</sub>Se<sub>3</sub> nanorods was random. From these previously reported studies, it was found that the application of 3D nanostructured Sb<sub>2</sub>Se<sub>3</sub> absorber layers would be very useful to overcome of the performance limitation of Sb<sub>2</sub>Se<sub>3</sub> solar cells based on the 3D p-n junction solar cell concept. However, to the best of the authors' knowledge, there is no detailed investigation of the nanostructure evolution of Sb<sub>2</sub>Se<sub>3</sub> thin films reported in the literature.

thin films for the formation of the 3D p-n junction solar cells could be one of the promising solutions. <sup>13-15</sup>

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 $<sup>\</sup>dagger\,$  Electronic supplementary information (ESI) available. See DOI: 10.1039/d1ma00804h

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The objective of this work was to study the evolution of a nanostructured Sb<sub>2</sub>Se<sub>3</sub> thin film fabricated using a MoSe<sub>2</sub> interlayer and the co-evaporation process. Additionally, a detailed investigation of the controlling factor to achieve finetuned Sb<sub>2</sub>Se<sub>3</sub> nanorod arrays is presented in this work. In our work, we proposed the evolution of nanostructured Sb<sub>2</sub>Se<sub>3</sub> thin films using a MoSe2 interlayer and detailed investigation on the control factor of the co-evaporation process for the fine-tuning of the Sb<sub>2</sub>Se<sub>3</sub> nanorod array. By introducing a MoSe<sub>2</sub> interlayer between the Mo substrate and Sb<sub>2</sub>Se<sub>3</sub>, it was possible to form uniformly ordered packed Sb<sub>2</sub>Se<sub>3</sub> nanorod formation using coevaporation of Sb<sub>2</sub>Se<sub>3</sub> and Se sources. The MoSe<sub>2</sub> interlayer has locally distributed dangling bonds, which play the role of nucleation seed for Sb<sub>2</sub>Se<sub>3</sub> crystalline growth perpendicular to the Mo substrate. Because of the seeding effect of the MoSe<sub>2</sub> interlayer, Sb<sub>2</sub>Se<sub>3</sub> thin films deposited on the MoSe<sub>2</sub> interlayer dominantly showed a uniformly ordered packed Sb<sub>2</sub>Se<sub>3</sub> nanorod array. In addition, the formation of a quasi-ohmic contact between Sb<sub>2</sub>Se<sub>3</sub> and MoSe<sub>2</sub> was another improving point of open-circuit voltage (Voc) of Sb<sub>2</sub>Se<sub>3</sub> solar cells compared with Sb<sub>2</sub>Se<sub>3</sub> solar cells without the MoSe<sub>2</sub> interlayer.

The nanostructure of the Sb<sub>2</sub>Se<sub>3</sub> nanorod arrays on the MoSe<sub>2</sub> interlayer was also sensitively affected by the conditions of the co-evaporation processes, such as the evaporation rate of sources and substrate temperature. The growth behavior of Sb<sub>2</sub>Se<sub>3</sub> crystals during the co-evaporation process was closely related with the final nanostructure of Sb<sub>2</sub>Se<sub>3</sub> nanorod array, which was the main factor that determines the short-circuit current  $(J_{SC})$  of  $Sb_2Se_3$  solar cells. The crystalline orientation of co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films, which is a crucial factor for high efficiency Sb<sub>2</sub>Se<sub>3</sub> solar cells, was also affected by the substrate temperature. 18-25 After chemical bath deposition (CBD) of n-type CdS buffer layers on the Sb<sub>2</sub>Se<sub>3</sub> nanorod array, the 3D p-n junction between Sb<sub>2</sub>Se<sub>3</sub> and CdS was formed and the 3D p-n junction was effective for the enhancement of  $J_{SC}$ . By introducing the MoSe<sub>2</sub> interlayer, the growth of a uniformly ordered Sb<sub>2</sub>Se<sub>3</sub> nanorod array and the quasi-ohmic contact between Sb<sub>2</sub>Se<sub>3</sub> and MoSe<sub>2</sub> were possible and 3D p-n junction Sb<sub>2</sub>Se<sub>3</sub> solar cells with 5.637% efficiency was achieved.

## **Experimental**

#### MoSe<sub>2</sub> interlayer and Sb<sub>2</sub>Se<sub>3</sub> thin-film preparation

The structure of the studied Sb<sub>2</sub>Se<sub>3</sub> thin films consisted of a soda-lime glass (SLG) substrate and a 600 nm-thick Mo layer as the back-contact layer. Before deposition, the substrate was sequentially cleaned in an ultrasonic bath using acetone, methanol, and deionized water. The Mo layer was deposited on the SLG substrate via direct current (DC) magnetron sputtering using a Mo target with a purity of 99.99%.

To form the MoSe<sub>2</sub> interlayer on Mo substrates, the latter were treated in a tube furnace in a Se atmosphere. Vacuum  $(10^{-3} \text{ Torr})$  was applied to the furnace to remove moisture and impurities. Then, Ar gas was introduced into the furnace, allowing the pressure inside the furnace to reach the

atmospheric pressure. The furnace was heated to 430 °C with Se pellets inside for 10 min and then annealed for 10 min. The MoSe<sub>2</sub> layer obtained was approximately 100 nm-thick.

The Sb<sub>2</sub>Se<sub>3</sub> thin films were prepared on bare Mo substrates or MoSe<sub>2</sub>/Mo substrates using a co-evaporator system using Sb<sub>2</sub>Se<sub>3</sub> and Se sources. The substrate temperature was controlled in the range of 250-290 °C. The substrate was heated at  $0.53~^{\circ}\text{C s}^{-1}$  from room temperature (~25  $^{\circ}\text{C}$ ) to the desired temperature. The evaporation rate of Sb<sub>2</sub>Se<sub>3</sub> and Se sources varied from 50-200 Å s<sup>-1</sup>. Sb<sub>2</sub>Se<sub>3</sub> was deposited to obtain thin films (thickness: approximately 500 nm). After deposition, the samples were allowed to cool to room temperature naturally.

#### Solar cell fabrication

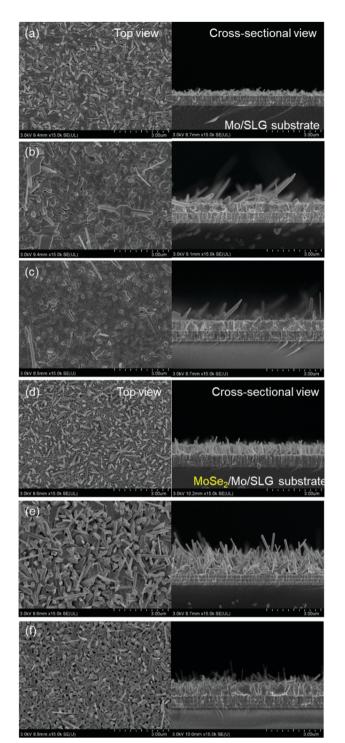
Chemical bath deposition (CBD) was used to cover the Sb<sub>2</sub>Se<sub>3</sub> thin films obtained with a 50 nm-thick n-type cadmium sulfide (CdS) buffer layer, and radiofrequency (RF) sputtering was used to sequentially deposit a 50 nm-thick intrinsic ZnO window layer and a 300 nm-thick Al-doped ZnO (AZO) A transparent conducting oxide (TCO) layer was deposited on top of the buffer layer to fabricate the solar cells. Thermal evaporation was then used to deposit a 1 µm-thick Al collection grid on top of the device.

#### Characterization of Sb<sub>2</sub>Se<sub>3</sub> solar cells

The Sb<sub>2</sub>Se<sub>3</sub> thin films were analyzed using various techniques to determine the influence of substrate temperature on the chemical composition and morphology of the films, and consequently, on the photovoltaic characteristics of the Sb<sub>2</sub>Se<sub>3</sub> solar cells. The surface morphology and chemical composition of the Sb<sub>2</sub>Se<sub>3</sub> thin films were observed using field emission scanning electron microscopy (FE-SEM, Hitachi, SU8020) and transmission electron microscopy (FE-TEM, Hitachi, HF-3300) with a focused ion beam (FIB) system (Hitachi, NB 5000). The crystalline structure of the Sb2Se3 thin films was determined using X-ray diffraction (XRD) analysis (Panalytical, Empyrean). The photovoltaic characteristics of the solar cells were characterized using a source meter (Keithley, 2400) unit and a solar simulator (Newport, 94022A) to simulate 1.5AM irradiation.

#### Results and discussion

To obtain a controlled Sb<sub>2</sub>Se<sub>3</sub> crystalline structure in Sb<sub>2</sub>Se<sub>3</sub> solar cells, an MoSe<sub>2</sub> interlayer between the Sb<sub>2</sub>Se<sub>3</sub> absorber and the Mo substrate was adopted. Additionally, the MoSe<sub>2</sub> interlayer between Sb<sub>2</sub>Se<sub>3</sub> and the Mo substrate played an important role in the growth control of the Sb<sub>2</sub>Se<sub>3</sub> crystalline structure. To investigate the role and effect of the MoSe<sub>2</sub> interlayer on Sb<sub>2</sub>Se<sub>3</sub> crystal growth, Sb<sub>2</sub>Se<sub>3</sub> thin films were deposited on bare Mo substrates and MoSe<sub>2</sub>/Mo substrates using the co-evaporation process at different source evaporation rates (50-200 Å s<sup>-1</sup>). The source evaporation rate is closely related to the Sb<sub>2</sub>Se<sub>3</sub> crystal growth mechanism. Fig. 1 shows



**Fig. 1** Top and cross-sectional SEM images of co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on bare Mo substrates or MoSe<sub>2</sub>/Mo substrates at 290 °C and different source evaporation rates: (a), 50 Å s<sup>-1</sup>; (b), 100 Å s<sup>-1</sup>; and (c), 200 Å s<sup>-1</sup> on bare Mo substrates; and (d), 50 Å s<sup>-1</sup>; (e), 100 Å s<sup>-1</sup>; and (f), 200 Å s<sup>-1</sup> on MoSe<sub>2</sub>/Mo substrates.

the cross-sectional SEM images of Sb<sub>2</sub>Se<sub>3</sub> thin films with different substrates and at different evaporation rates.

In the case of co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on bare Mo substrates, thin films deposited at a slow source evaporation

rate (50 Å s<sup>-1</sup>) showed a small and randomly oriented Sb<sub>2</sub>Se<sub>3</sub> nanorod structure (Fig. 1a). However, as the source evaporation rate increased to 100 Å s<sup>-1</sup>, Sb<sub>2</sub>Se<sub>3</sub> thin films showed a densely packed columnar structure with some long Sb<sub>2</sub>Se<sub>3</sub> nanorods (Fig. 1b). At a source evaporation rate of 200 Å s<sup>-1</sup>, a flat Sb<sub>2</sub>Se<sub>3</sub> thin-film structure comprising densely packed grains was observed (Fig. 1c). Generally, the nucleation rate is closely related to the evaporation rate.<sup>26</sup> Thus, a easily formed continuous film under a high evaporation rate was expected. For film materials having an anisotropic crystal structure, such as Sb<sub>2</sub>Se<sub>3</sub>, rod-shaped film formation is highly probable when the evaporation rate is low. Under these conditions, the impingement rate of adatoms on the substrate decreases and their diffusion length may increase, causing the formation of a noncontinuous film. Based on these results, authors believe that controlling the evaporation rate is critical to developing the Sb<sub>2</sub>Se<sub>3</sub> rod array for 3D p-n junction solar cells.

Contrary to the co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on Mo substrates, Sb<sub>2</sub>Se<sub>3</sub> thin films prepared on MoSe<sub>2</sub>/Mo substrates showed a different nanostructure. All Sb<sub>2</sub>Se<sub>3</sub> thin films on MoSe<sub>2</sub>/Mo substrates showed non-flat thin-film structures, regardless of the source evaporation rate. At an evaporation rate of 50 Å s<sup>-1</sup>, thin films showed a vertically oriented narrow Sb<sub>2</sub>Se<sub>3</sub> nanorod array of length 500 nm (Fig. 1d). The diameter and length of the Sb<sub>2</sub>Se<sub>3</sub> nanorods increased with increasing evaporation rate from 50 to 100  $\text{Å s}^{-1}$  (Fig. 1e). At an evaporation rate of 200  $\text{Å s}^{-1}$ , the thin-film showed ordered arrays of thick Sb<sub>2</sub>Se<sub>3</sub> nanorods with a diameter of 100 nm (Fig. 1f). Sb<sub>2</sub>Se<sub>3</sub> thin films deposited on MoSe<sub>2</sub> substrates showed a Sb<sub>2</sub>Se<sub>3</sub> nanorod array structure regardless of the source evaporation rate. Additionally, the Sb<sub>2</sub>Se<sub>3</sub> nanorod array nanostructure was tuned by varying the source evaporation rate. The formation of Sb<sub>2</sub>Se<sub>3</sub> nanorod array on the MoSe<sub>2</sub> interlayer might be attributed to different Sb<sub>2</sub>Se<sub>3</sub> crystal growth mechanisms (Fig. 2). Moreover, the MoSe<sub>2</sub> interlayer has unique anisotropic structural properties. For example, the (00l) plane of MoSe<sub>2</sub> has no surface dangling bond, making the nucleation process difficult while facilitating the diffusion and desorption of adatoms on the (00l) plane of MoSe<sub>2</sub>. However, the nucleation process of Sb<sub>2</sub>Se<sub>3</sub> could be easily formed on a plane having a high-dangling bond density, such as the one perpendicular to the (00l) plane. Therefore, the MoSe<sub>2</sub> interlayer was expected to provide a selective nucleation site, facilitating the formation of the Sb<sub>2</sub>Se<sub>3</sub> rod array on the MoSe<sub>2</sub> interlayer. To investigate in

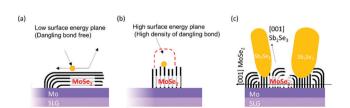


Fig. 2 Schematic diagram of the selective nucleation and growth mechanism of  $Sb_2Se_3$  on the randomly grown  $MoSe_2$  interlayer; (a) low surface energy plane, (b) high surface energy plane, and (c) growth of  $Sb_2Se_3$  nanorods on  $MoSe_2$  interlayer.

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detail the interface between Sb<sub>2</sub>Se<sub>3</sub> and MoSe<sub>2</sub>, a high-angle annular detector dark-field scanning transmission electron microscopy (HADDF-STEM) image of the Sb<sub>2</sub>Se<sub>3</sub>/MoSe<sub>2</sub>/Mo sample was observed (Fig. S3, ESI†). The MoSe<sub>2</sub> showed a two-dimensional crystal structure with weak van der Waals bonding forces and without dangling bonds in the c-plane of MoSe<sub>2</sub>. Overall, the layer structure was oriented perpendicular to the Mo substrate. Interestingly, in our case, the density of the MoSe<sub>2</sub> interlayer bunch seemed relatively low. Thus, the orientation of the flexible bunch of the MoSe2 layer was not perfectly aligned. The exposed-MoSe<sub>2</sub> layer on the surface consisted of a c-plane without dangling bonds and another plane with dangling bonds.

To confirm the effect of the substrate type on the morphology of the co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films, we also deposited Sb<sub>2</sub>Se<sub>3</sub> thin films on indium tin oxide (ITO) substrates using the same co-evaporation process. Unlike Mo substrates, all the coevaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on ITO substrates showed a similar dense thin-film morphology, regardless of the source evaporation rate (Fig. S1, ESI†) and substrate temperature (Fig. S2, ESI†). These results show that the type of substrate had a significant effect on the formation of Sb<sub>2</sub>Se<sub>3</sub> nanorods by the co-evaporation process.

To evaluate the photovoltaic characteristics of the coevaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on different Mo and MoSe<sub>2</sub>/Mo substrates, three different Sb<sub>2</sub>Se<sub>3</sub> solar cells with a planar device structure of SLG/Mo/MoSe<sub>2</sub>/Sb<sub>2</sub>Se<sub>3</sub>/CdS/i-ZnO/AZO/Al were fabricated. Fig. 3 shows the cross-sectional TEM and energy dispersive X-ray spectroscopy (EDS) mapping images of the Sb<sub>2</sub>Se<sub>3</sub> solar cells fabricated at different source evaporation rates and with different substrates. The Sb<sub>2</sub>Se<sub>3</sub> solar cells without the MoSe<sub>2</sub> interlayer showed a planar structure (Fig. 3a and b), due to the flat Sb<sub>2</sub>Se<sub>3</sub> thin-film structure shown in Fig. 1c. The CdS buffer, i-ZnO window, and AZO TCO layers were uniformly deposited on the flat Sb<sub>2</sub>Se<sub>3</sub> absorber layer. However, the Sb<sub>2</sub>Se<sub>3</sub> solar cells prepared on MoSe<sub>2</sub>/Mo substrates showed completely different nanostructures than those using the Mo substrate. The Sb<sub>2</sub>Se<sub>3</sub> absorber layer showed a complex nanorod array, and the CdS buffer layer was deposited on the Sb<sub>2</sub>Se<sub>3</sub> nanorods conformally. The ZnO and AZO layers deposited on the rough CdS/Sb<sub>2</sub>Se<sub>3</sub> interface showed irregular nanostructures. At the bottom of the Sb<sub>2</sub>Se<sub>3</sub> absorber layer, a uniformly formed MoSe<sub>2</sub> interlayer was confirmed. Moreover, the source evaporation rate showed a considerable effect on the nanostructure of the Sb<sub>2</sub>Se<sub>3</sub> absorber layer. The Sb<sub>2</sub>Se<sub>3</sub> solar cells prepared using a 200 Å s<sup>-1</sup> source evaporation rate on MoSe<sub>2</sub>/Mo substrates showed an irregular and agglomerated Sb<sub>2</sub>Se<sub>3</sub> grain structure (Fig. 3c and d). However, when the evaporation rate decreased to 100 Å s<sup>-1</sup>, the Sb<sub>2</sub>Se<sub>3</sub> absorber layer showed a more ordered and regular nanorod array (Fig. 3e and f). These different Sb<sub>2</sub>Se<sub>3</sub> nanorod array structures observed according to the source evaporation rate were consistent with the SEM images (Fig. 1e and f).

Table 1 lists the photovoltaic properties of the Sb<sub>2</sub>Se<sub>3</sub> solar cells with and without the MoSe<sub>2</sub> interlayer. Despite the planar and uniform device structure of the Sb<sub>2</sub>Se<sub>3</sub> solar cells without

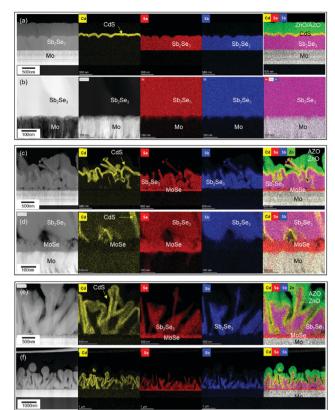


Fig. 3 Cross-sectional TEM and EDS mapping images of Sb<sub>2</sub>Se<sub>3</sub> solar cells fabricated with different source evaporation rates and at a temperature of 290 °C. (a) and (b), 200 Å s<sup>-1</sup> on a Mo substrate; (c) and (d), 200 Å s<sup>-1</sup> on a MoSe<sub>2</sub>/Mo substrate; and (e) and (f) 100 Å s<sup>-1</sup> on a MoSe<sub>2</sub>/Mo substrate.

the MoSe<sub>2</sub> interlayer, the solar cells with the MoSe<sub>2</sub> interlayer showed a superior photovoltaic performance. The higher opencircuit voltage  $(V_{OC})$  showed by the  $Sb_2Se_3$  solar cells with the MoSe<sub>2</sub> interlayer might be attributed to the quasi-ohmic contact formation between Sb<sub>2</sub>Se<sub>3</sub> and MoSe<sub>2</sub>. The quasi-ohmic contact formation of the MoSe2 interlayer is well known for copper indium gallium diselenide (CIGS) solar cells.<sup>27–31</sup> This quasi-ohmic contact could reduce the contact barrier and series resistance of  $Sb_2Se_3$  solar cell devices. The higher  $J_{SC}$  could be related to the 3D p-n junction formation between the Sb<sub>2</sub>Se<sub>3</sub> absorber and CdS buffer layers. Due to the 3D p-n junction structure, photo-generated carriers can easily escape into the CdS buffer layer. Thus, nanostructured Sb<sub>2</sub>Se<sub>3</sub> solar cells show a higher J<sub>SC</sub> than flat thin-film Sb<sub>2</sub>Se<sub>3</sub> solar cells. This superior photovoltaic performance was consistent with the results reported by Park et al. and was also attributed to the 3D p-n junction formation between the Sb<sub>2</sub>Se<sub>3</sub> and CdS buffer layers.<sup>17</sup> Between the two Sb<sub>2</sub>Se<sub>3</sub> solar cells with a MoSe<sub>2</sub> interlayer, the one fabricated at an evaporation rate of 100 Å s<sup>-1</sup> showed higher  $J_{SC}$  and fill factor (FF) than that at 200 Å s<sup>-1</sup>, which might be attributed to the more regular and ordered 3D p-n junction formation between Sb<sub>2</sub>Se<sub>3</sub> and CdS.

A previous report showed that high substrate temperatures induce the evolution of Sb<sub>2</sub>Se<sub>3</sub> nanorod arrays, demonstrating the close relationship between the nanostructure of

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Table 1 Photovoltaic characteristics of the Sb<sub>2</sub>Se<sub>3</sub> solar cells using MoSe<sub>2</sub>/Mo substrates at 290 °C. Open-circuit voltage (V<sub>OC</sub>), short-circuit current  $(J_{SC})$ , fill factor (FF), power conversion efficiency (PCE), and series and shunt resistances  $(R_s, R_{sh})$ 

Substrate	Evaporation rate $[\mathring{A} \text{ s}^{-1}]$	$V_{\mathrm{OC}}\left[\mathbf{V}\right]$	$J_{ m SC}[{ m mA~cm}^{-2}]$	FF [%]	Eff [%]	$R_{\rm s} \left[\Omega \ {\rm cm}^2\right]$	$R_{\rm sh} \left[\Omega \ {\rm cm}^2\right]$
Mo	200	0.20297 $0.43951$ $0.44999$	8.4189	27.2369	0.465	18.105	26.703
MoSe <sub>2</sub> /Mo	200		21.1832	36.0716	3.358	9.847	51.383
MoSe <sub>2</sub> /Mo	100		23.6606	40.2727	4.288	8.488	70.319

co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films and substrate temperature.<sup>17</sup> Therefore, we have tried to tune the nanostructure of coevaporated Sb<sub>2</sub>Se<sub>3</sub> thin films by controlling the substrate temperature at a source evaporation rate of 100 Å s<sup>-1</sup> on MoSe<sub>2</sub>/Mo substrates. We also prepared co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on bare Mo at 315 °C, which is the requirement for Sb<sub>2</sub>Se<sub>3</sub> nanorod array formation on bare Mo substrates. Fig. 4 shows

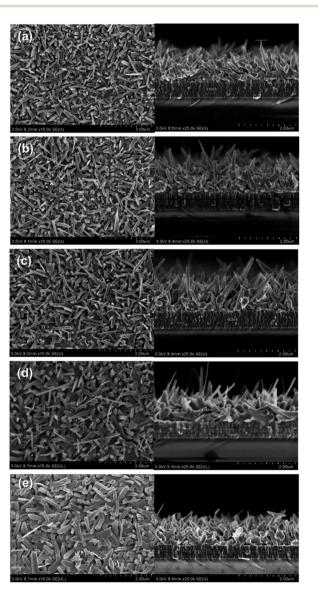


Fig. 4 Top and cross-sectional SEM images of Sb<sub>2</sub>Se<sub>3</sub> thin films at a source evaporation rate of 100 Å s<sup>-1</sup> and different deposition temperatures: (a), 250; (b), 260; (c), 270; and (d), 290 °C using MoSe<sub>2</sub>/Mo substrates; and (e), 315 °C using a bare Mo substrate.

the top and cross-sectional SEM images of co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films at different substrate temperatures on bare Mo or MoSe<sub>2</sub>/Mo.

The co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on the MoSe<sub>2</sub>/Mo substrate did not show the abrupt morphology change, from the flat thin-film structure to a nanorod array, observed in our previous report on Mo substrates. 17 Because the MoSe2 interlayer already easily induces the formation of the Sb<sub>2</sub>Se<sub>3</sub> nanorod array, the substrate temperature showed little effect on the formation of the Sb<sub>2</sub>Se<sub>3</sub> nanorod array. However, the detailed nanostructure of the Sb<sub>2</sub>Se<sub>3</sub> nanorods was affected by the substrate temperature. As the substrate temperature decreased from 290 to 250 °C, the shape regularity and the degree of ordering of the Sb<sub>2</sub>Se<sub>3</sub> nanorods improved. As the substrate temperature decreases, the Sb<sub>2</sub>Se<sub>3</sub> nuclei density increases, suppressing the rapid growth of Sb<sub>2</sub>Se<sub>3</sub> nanorods. Therefore, the growth of Sb<sub>2</sub>Se<sub>3</sub> nanorods could be controlled effectively to form an ordered Sb<sub>2</sub>Se<sub>3</sub> nanorod array by adjusting the substrate temperature on MoSe<sub>2</sub>/Mo. Moreover, using the MoSe<sub>2</sub> interlayer, it was possible to control the nanostructure of co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films by varying both the evaporation rate and substrate temperature. Co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on a bare Mo substrate at 315 °C also showed a similar Sb<sub>2</sub>Se<sub>3</sub> nanorod structure to that on MoSe<sub>2</sub>/Mo substrates, which is consistent with the results from our previous work.<sup>17</sup> The bare Mo substrate required a higher substrate temperature to form a similar Sb<sub>2</sub>Se<sub>3</sub> nanorod array to the one obtained using MoSe<sub>2</sub>/Mo substrates and the coevaporation process.

The crystalline orientation of Sb<sub>2</sub>Se<sub>3</sub> thin films is well known to be crucial for determining the photovoltaic properties of Sb<sub>2</sub>Se<sub>3</sub> solar cells due to the intrinsic one-dimensional ribbon structure of Sb<sub>2</sub>Se<sub>3</sub>. 32-34 Therefore, the crystalline structure of co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on MoSe<sub>2</sub>/Mo substrate at different substrate temperatures was analyzed by XRD analysis. All the co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films showed characteristic XRD peaks of Sb<sub>2</sub>Se<sub>3</sub>. However, the relative intensity of specific peaks corresponding to different crystalline orientations was different for each substrate temperature. Particularly, at 260 °C where the (020) and (120) peaks were not observed, and the intensity of the (211) and (221) peaks increased (Fig. 5a). To further investigate the crystalline orientation of the coevaporated Sb<sub>2</sub>Se<sub>3</sub> thin films, their texture coefficient was calculated.<sup>35–37</sup> At 260 °C the texture coefficient of the (211) peak was remarkably higher than that at other substrate temperatures (Fig. 5b). Additionally, all other Sb<sub>2</sub>Se<sub>3</sub> thin films showed relatively higher texture coefficients of (hk0) than those at 260 °C. The (hk1) preferential orientation of the Sb<sub>2</sub>Se<sub>3</sub> thin

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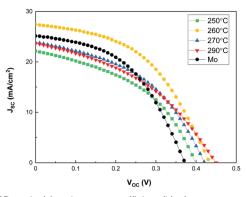


Fig. 5 XRD peaks (a) and texture coefficient (b) of co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films at different substrate temperatures and at a source evaporation rate of 100  $\rm \mathring{A}~s^{-1}$ 

film at a substrate temperature of 260 °C could help in improving the photovoltaic performance of Sb<sub>2</sub>Se<sub>3</sub> solar cells.

Sb<sub>2</sub>Se<sub>3</sub> solar cells were fabricated at different substrate temperatures to investigate the effect of substrate temperature of co-evaporated Sb<sub>2</sub>Se<sub>3</sub> thin films on their photovoltaic performance. Fig. 6 shows the I-V curves of the Sb<sub>2</sub>Se<sub>3</sub> solar cells at a source evaporation rate of 100 Å s<sup>-1</sup> and different substrate temperatures, and Table 2 lists the photovoltaic properties obtained for the Sb<sub>2</sub>Se<sub>3</sub> solar cells. All the Sb<sub>2</sub>Se<sub>3</sub> solar cells using the MoSe<sub>2</sub> interlayer showed a  $V_{\rm OC} > 0.4$  V, regardless of the substrate temperature. Because the MoSe<sub>2</sub> interlayer already improved the  $V_{\rm OC}$ , the structural change in the Sb<sub>2</sub>Se<sub>3</sub> nanorods had little effect on the VOC. The Sb2Se3 solar cell using a bare Mo substrate showed a slightly lower  $V_{\rm OC}$  (0.37 V) than that with MoSe<sub>2</sub>/Mo substrates. These results show that the introduction of the MoSe<sub>2</sub> interlayer significantly improves the V<sub>OC</sub> of Sb<sub>2</sub>Se<sub>3</sub> solar cells. The Sb<sub>2</sub>Se<sub>3</sub> solar cell prepared at a substrate temperature of 260  $^{\circ}$ C showed the highest  $J_{SC}$ . This might be attributed to the (hk1) preferential orientation of the Sb<sub>2</sub>Se<sub>3</sub> absorber layer. Moreover, the ordered and regular Sb<sub>2</sub>Se<sub>3</sub> nanorod array could help increase the area of the 3D p-n junction between Sb<sub>2</sub>Se<sub>3</sub> and CdS, improving the J<sub>SC</sub> of the  $Sb_2Se_3$  solar cells. By the way, it showed comparable  $J_{SC}$ (25.138 mA cm<sup>-2</sup>) to those of the Sb<sub>2</sub>Se<sub>3</sub> solar cell with MoSe<sub>2</sub>/Mo substrates because of the densely packed Sb<sub>2</sub>Se<sub>3</sub> nanorod array. Interestingly, Sb<sub>2</sub>Se<sub>3</sub> solar cells using a

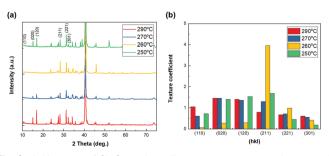


Fig. 6 I-V curves of Sb<sub>2</sub>Se<sub>3</sub> solar cells at a source evaporation rate of 100 Å s $^{-1}$  and different substrate temperatures: (a), 250; (b), 260; (c), 270; and (d), 290 °C using MoSe<sub>2</sub>/Mo and bare Mo substrates

Table 2 Photovoltaic characteristics of the Sb<sub>2</sub>Se<sub>3</sub> solar cells using MoSe<sub>2</sub>/Mo and bare Mo substrates at different temperatures (T). Opencircuit voltage ( $V_{OC}$ ), short-circuit current ( $J_{SC}$ ), fill factor (FF), power conversion efficiency (PCE), and series and shunt resistances ( $R_s$ ,  $R_{sh}$ )

Substrate	Temperature $[^{\circ}C]$		$J_{ m SC}$ [mA cm $^{-2}$ ]	FF [%]	Ef f[%]	$R_{\rm s}$ $[\Omega~{ m cm}^2]$	$R_{\rm sh}$ $[\Omega \ { m cm}^2]$
MoSe <sub>2</sub>	250	0.40	22.167	43.99	3.886	5.48	67.59
$MoSe_2$	260	0.43	27.430	47.35	5.637	5.89	118.26
$MoSe_2$	270	0.42	23.834	44.47	4.464	6.04	80.89
$MoSe_2$	290	0.45	23.661	40.27	4.288	8.49	70.32
Mo	315	0.37	25.138	47.11	4.412	5.69	112.25

MoSe<sub>2</sub>/Mo substrate at 250, 270 and 290 °C showed a relatively smaller  $J_{SC}$  than bare Mo substrate. The nanostructure changes of the co-evaporated Sb<sub>2</sub>Se<sub>3</sub> nanorod array by temperature is remarkable (Fig. 4), which might be closely related to the  $J_{SC}$ and FF of Sb<sub>2</sub>Se<sub>3</sub> solar cells. At 250 °C, the width of Sb<sub>2</sub>Se<sub>3</sub> nanorods is smaller than that at 260 °C, which is disadvantageous for sufficient light absorption. The length uniformity of Sb<sub>2</sub>Se<sub>3</sub> nanorods at 270 and 290 °C is inferior compared to that at 260 °C. The irregular nanostructure of the Sb<sub>2</sub>Se<sub>3</sub> nanorods induces insufficient light absorption and photo-generated carrier transportation. The thinner rods (250 °C) and increased empty space (270 °C) could deteriorate the FF by increasing the shunt paths too.

To investigate the correlation between the 3D p-n junction structure and the photovoltaic properties of Sb<sub>2</sub>Se<sub>3</sub> solar cells, TEM analysis of the best Sb<sub>2</sub>Se<sub>3</sub> solar cell, with the evaporation rate of 100  $\text{Å s}^{-1}$  and the substrate temperature of 260  $^{\circ}$ C, was performed. Fig. 7 shows the TEM-EDS mapping images of the Sb<sub>2</sub>Se<sub>3</sub> solar cell device with a nanorod array with a PCE of 5.637% (Table 2). The Sb<sub>2</sub>Se<sub>3</sub> solar cell using a MoSe<sub>2</sub>/Mo substrate showed a 1.225% higher PCE than the one using a bare Mo substrate, which might be attributed to the improved V<sub>OC</sub>. A densely packed, uniformly distributed, Sb<sub>2</sub>Se<sub>3</sub> nanorod array and the 3D p-n junction between CdS and Sb<sub>2</sub>Se<sub>3</sub> nanorods were observed. These properties are advantageous for the enhancement of  $J_{SC}$ . Additionally, the uniform MoSe<sub>2</sub> interlayer between the Mo substrate and the Sb<sub>2</sub>Se<sub>3</sub> nanorod array is beneficial for the formation of a quasi-ohmic contact between the layers.

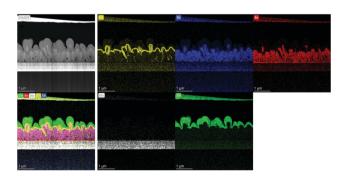


Fig. 7 STEM-EDS mapping images of Sb<sub>2</sub>Se<sub>3</sub> nanorod array solar cell with PCE of 5.637%.

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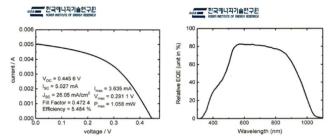


Fig. 8 Certified J-V curve (a) and external quantum efficiency (EQE) data (b) of Sb<sub>2</sub>Se<sub>3</sub> solar cells at a substrate temperature of 260 °C and a source evaporation rate of 100 Å s<sup>-1</sup> using a MoSe<sub>2</sub>/Mo substrate (Korea Institute of Energy Research, KIER).

The results presented in this work reveal that the MoSe<sub>2</sub> interlayer between a Mo substrate and a Sb<sub>2</sub>Se<sub>3</sub> absorber layer improves the performance of Sb<sub>2</sub>Se<sub>3</sub> solar cells in two relevant ways. First, the contact quality between Sb<sub>2</sub>Se<sub>3</sub> and Mo layers improved due to quasi-ohmic contact formation, which was confirmed by the higher V<sub>OC</sub> values of Sb<sub>2</sub>Se<sub>3</sub> solar cells using MoSe<sub>2</sub>/Mo substrates compared to those using a bare Mo substrate; second, a facile formation of the Sb<sub>2</sub>Se<sub>3</sub> nanorod array and a lower substrate temperature were achieved using the co-evaporation process, which was followed by the 3D p-n junction Sb<sub>2</sub>Se<sub>3</sub> solar cell fabrication. Moreover, the MoSe<sub>2</sub> interlayer played a key role in improving the photovoltaic performance of the Sb<sub>2</sub>Se<sub>3</sub> nanorod array solar cell by increasing both  $V_{\rm OC}$  and  $J_{\rm SC}$ . The quasi-ohmic contact caused by the MoSe<sub>2</sub> interlayer effectively improved the contact quality between Sb<sub>2</sub>Se<sub>3</sub> layers and Mo substrates. Moreover, the MoSe<sub>2</sub> interlayer also played a critical role as a seed layer for the uniform growth of the Sb<sub>2</sub>Se<sub>3</sub> nanorod array.

The Sb<sub>2</sub>Se<sub>3</sub> solar cells prepared by co-evaporation at a source evaporation rate of 100 Å s<sup>-1</sup> and substrate temperature of 260 °C on a MoSe<sub>2</sub>/Mo substrate were analyzed by an external certified laboratory to verify our PCE laboratory measurements. The certified J-V curve (Fig. 8a) and external quantum efficiency data (Fig. 8b) showed similar results (PCE of 5.484%) to our labscale photovoltaic measurements. We calculated integrated  $J_{\rm SC}$ from EQE data and compared with the  $J_{SC}$  of J-V curve. The  $J_{SC}$ from integrated EQE (25.89 mA cm<sup>-2</sup>) is similar to the  $J_{SC}$  of the J-V curve (26.05 mA cm<sup>-2</sup>).

#### Conclusions

This work investigated the role of the MoSe<sub>2</sub> interlayer on the growth of the Sb<sub>2</sub>Se<sub>3</sub> nanorod array and the contact quality between Sb<sub>2</sub>Se<sub>3</sub> thin films and Mo substrates. By introducing a MoSe<sub>2</sub> interlayer between Sb<sub>2</sub>Se<sub>3</sub> layers and Mo substrates, uniformly ordered densely packed Sb<sub>2</sub>Se<sub>3</sub> nanorods were successfully prepared. The MoSe<sub>2</sub> interlayer, with locally distributed high surface energy planes, played a crucial role as a seed layer for the preferential growth of Sb<sub>2</sub>Se<sub>3</sub> crystals perpendicular to the substrate, which facilitated the formation of a Sb<sub>2</sub>Se<sub>3</sub> nanorod array on the MoSe<sub>2</sub> interlayer. The latter also significantly improved the photovoltaic performance of Sb<sub>2</sub>Se<sub>3</sub> solar cell devices by forming a quasi-ohmic contact between Sb<sub>2</sub>Se<sub>3</sub> absorber layers and Mo substrates, resulting in higher  $V_{\rm OC}$  and  $J_{\rm SC}$ . The nanostructure and crystalline orientation of the Sb<sub>2</sub>Se<sub>3</sub> nanorods array were affected by the substrate temperature and the source evaporation rate during the coevaporation process. 3D p-n junction solar cells based on an ordered and (hk1) preferentially oriented Sb2Se3 nanorods array showed a PCE of 5.637%.

### Conflicts of interest

There are no conflicts to declare.

## Acknowledgements

This work was supported by the DGIST R & D Programs of the Ministry of Science and ICT, Republic of Korea (21-CoE-ET-01, 21-ET-08). We thank Mr Eun and Mr Cheon at the Center for Core Research Facilities (CCRF) of DGIST for FIB sampling and STEM measurements.

### Notes and references

- 1 Y. Zhou, M. Y. Leng, Z. Xia, J. Zhong, H. B. Song, X. S. Liu, B. Yang, J. P. Zhang, J. Chen, K. H. Zhou, J. B. Han, Y. B. Cheng and J. Tang, Adv. Energy Mater., 2014, 4, 1301846.
- 2 G. X. Liang, Z. H. Zheng, P. Fan, J. T. Luo, J. G. Hu, X. H. Zhang, H. L. Ma, B. Fan, Z. K. Luo and D. P. Zhang, Sol. Energy Mater. Sol. Cells, 2018, 174, 263-270.
- 3 C. C. Yuan, L. J. Zhang, W. F. Liu and C. F. Zhu, Sol. Energy, 2016, 137, 256-260.
- 4 M. Birkett, W. M. Linhart, J. Stoner, L. J. Phillips, K. Durose, J. Alaria, J. D. Major, R. Kudrawiec and T. D. Veal, APL Mater., 2018, 6, 84901.
- 5 X. M. Wang, R. F. Tang, C. Y. Wu, C. F. Zhu and T. Chen, J. Energy Chem., 2018, 27, 713-721.
- 6 K. Shen, C. Z. Ou, T. L. Huang, H. B. Zhu, J. J. Li, Z. Q. Li and Y. Mai, Sol. Energy Mater. Sol. Cells, 2018, 186, 58-65.
- 7 S. Chen, T. Liu, Z. Zheng, M. Ishaq, G. Liang, P. Fan, T. Chen and J. Tang, J. Energy Chem., 2022, 67, 508-523.
- 8 P. Fan, G. J. Chen, S. Chen, Z. H. Zheng, M. Azan, N. Ahmad, Z. H. Su, G. X. Liang, X. H. Zhang and Z. G. Chen, ACS Appl. Mater. Interfaces, 2021, 13, 46671-46680.
- 9 G. X. Liang, Y. D. Luo, S. Chen, R. Tang, Z. H. Zheng, X. J. Li, X. S. Liu, Y. K. Liu, Y. F. Li, X. Y. Chen, Z. H. Su and X. H. Zhang, Nano Energy, 2020, 73, 104806.
- 10 Y. D. Luo, R. Tang, S. Chen, J. G. Hu, Y. K. Liu, Y. F. Li, X. S. Liu, Z. H. Zheng, Z. H. Su, X. F. Ma, P. Fan, X. H. Zhang, H. L. Ma, Z. G. Chen and G. X. Liang, Chem. Eng. J., 2020, 393, 124599.
- 11 R. Tang, S. Z. H. Zheng, Z. H. Su, X. J. Li, Y. D. Wei, X. H. Zhang, Y. Q. Fu, J. T. Luo, P. Fan and G. X. Liang, Nano Energy, 2019, 64, 103929.
- 12 R. F. Tang, X. M. Wang, W. T. Lian, J. L. Huang, Q. Wei, M. L. Huang, Y. W. Yin, C. H. Jiang, S. F. Yang, G. C. Xing,

Paper

8, 1702888.

S. Y. Chen, C. F. Zhu, X. J. Hao, M. A. Green and T. Chen, Nat.

- Energy, 2020, 5, 587-595. 13 P. V. Kamat, K. Tvrdy, D. R. Baker and J. G. Radich, Chem. Rev., 2010, 110, 6664-6688.
- 14 M. Yu, Y. Z. Long, B. Sun and Z. Y. Fan, Nanoscale, 2012, 4, 2783-2796.
- 15 W. Yang, J. Ahn, Y. Oh, J. Tan, H. Lee, J. Park, H. C. Kwon, J. Kim, W. Jo, J. Kim and J. Moon, Adv. Energy Mater., 2018,
- 16 Z. Q. Li, X. Y. Liang, G. Li, H. X. Liu, H. Y. Zhang, J. X. Guo, J. W. Chen, K. Shen, X. Y. San, W. Yu, R. E. I. Schropp and Y. H. Mai, Nat. Commun., 2019, 10, 125.
- 17 S. N. Park, S. Y. Kim, S. J. Lee, S. J. Sung, K. J. Yang, J. K. Kang and D. H. Kim, J. Mater. Chem. A, 2019, 7, 25900-25907.
- 18 Z. Li, X. Chen, H. Zhu, J. Chen, Y. Guo, C. Zhang, W. Zhang, X. Niu and Y. Mai, Sol. Energy Mater. Sol. Cells, 2017, 161, 190-196.
- 19 Y. H. Kwon, Y. B. Kim, M. Jeong, H. W. Do, H. K. Cho and J. Y. Lee, Sol. Energy Mater. Sol. Cells, 2017, 172, 11-17.
- 20 L. Wang, D. B. Li, K. H. Li, C. Chen, H. X. Deng, L. Gao, Y. Zhao, F. Jiang, L. Y. Li, F. Huang, Y. S. He, H. S. Song, G. D. Niu and J. Tang, Nat. Energy, 2017, 2, 17046.
- 21 C. Chen, L. Wang, L. Gao, D. Nam, D. B. Li, K. H. Li, Y. Zhao, C. Ge, H. Cheong, H. Liu, H. S. Song and J. Tang, ACS Energy Lett., 2017, 2, 2125-2132.
- 22 O. S. Hutter, L. J. Phillips, K. Durose and J. D. Major, Sol. Energy Mater. Sol. Cells, 2018, 188, 177-181.
- 23 C. Chen, K. H. Li, S. Y. Chen, L. Wang, S. C. Lu, Y. H. Liu, D. B. Li, H. S. Song and J. Tang, ACS Energy Lett., 2018, 3, 2335-2341.
- 24 X. X. Wen, C. Chen, S. C. Lu, K. H. Li, R. Kondrotas, Y. Zhao, W. H. Chen, L. Gao, C. Wang, J. Zhang, G. D. Niu and J. Tang, Nat. Commun., 2018, 9, 2179.

- 25 K. H. Li, S. Y. Wang, C. Chen, R. Kondrotas, M. C. Hu, S. C. Lu, C. Wang, W. Chen and J. Tang, J. Mater. Chem. A, 2019, 7, 9665-9672.
- 26 M. Ohring, in Materials Science of Thin Films, ed. M. Ohring, Academic Press, San Diego, 2nd edn, 2002, pp. 357-415, DOI: 10.1016/B978-012524975-1/50010-0.
- 27 H.-y. Sun, P.-h. Li, Y.-m. Xue, Z.-x. Qiao and L. Sai, Optoelectron. Lett., 2019, 15, 428-434.
- 28 L. Assmann, J. C. Bernede, A. Drici, C. Amory, E. Halgand and M. Morsli, Appl. Surf. Sci., 2005, 246, 159-166.
- 29 N. Kohara, S. Nishiwaki, Y. Hashimoto, T. Negami and T. Wada, Sol. Energy Mater. Sol. Cells, 2001, 67, 209-215.
- 30 J. Chantana, T. Kato, H. Sugimoto and T. Minemoto, Appl. Phys. Lett., 2018, 112, 151601.
- 31 S. Nishiwaki, N. Kohara, T. Negami and T. Wada, Ipn. J. Appl. Phys., 1998, 37, L71-L73.
- 32 Y. Zhou, L. Wang, S. Y. Chen, S. K. Qin, X. S. Liu, J. Chen, D. J. Xue, M. Luo, Y. Z. Cao, Y. B. Cheng, E. H. Sargent and J. Tang, Nat. Photonics, 2015, 9, 409-415.
- 33 Z. Q. Li, H. B. Zhu, Y. T. Guo, X. N. Niu, X. Chen, C. Zhang, W. Zhang, X. Y. Liang, D. Zhou, J. W. Chen and Y. H. Mai, Appl. Phys. Express, 2016, 9, 052302.
- 34 X. S. Liu, J. Chen, M. Luo, M. Y. Leng, Z. Xia, Y. Zhou, S. K. Qin, D. J. Xue, L. Lv, H. Huang, D. M. Niu and J. Tang, ACS Appl. Mater. Interfaces, 2014, 6, 10687-10695.
- 35 S. Dias, B. Murali and S. B. Krupanidhi, Sol. Energy Mater. Sol. Cells, 2015, 143, 152-158.
- 36 S. Karim, M. E. Toimil-Molares, F. Maurer, G. Miehe, W. Ensinger, J. Liu, T. W. Cornelius and R. Neumann, Appl. Phys. A: Mater. Sci. Process., 2006, 84, 403-407.
- 37 Y. Q. Wang, W. Tang and L. Zhang, J. Mater. Sci. Technol., 2015, 31, 175-181.