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Magnetic response properties of carbon nano-onions[†]

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The magnetic response of a number of double- and triple-layer carbon nano-onions (CNOs) is analyzed by calculating the magnetically induced current density and the induced magnetic field using the pseudo- π model. Qualitatively the same magnetic response was obtained in calculations at the all-electron level. The calculations show that the CNOs exhibit strong net diatropic (paratropic) ring currents when the external magnetic field points perpendicularly to one of the six-membered (five-membered) rings. They are deshielded inside and shielded outside the CNO; the latter dominates for larger CNOs. The magnetic response originates from a combination of spherical paratropic current densities on the inside of each carbon layer and diatropic current densities on the outside of them. The quantitative differences in the aromaticity of the CNOs as compared to single fullerenes are discussed in terms of ring-current strengths. The magnetic response of some of the CNOs is approximately the sum of the magnetic response of the individual layers, whereas deviations are significant for CNOs containing C₈₀. For the largest CNOs, the deviation from the sum of the fullerene contributions is larger, especially when the external magnetic field is perpendicular to a six-membered ring.

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1 Introduction

Aromaticity is commonly associated with properties of cyclic conjugated compounds such as high stability, high symmetry, low reactivity, and a diamagnetic response to external magnetic fields.¹ Originally the aromaticity concept originates from planar organic compounds such as benzene.² However, since the discovery of buckminsterfullerene (C₆₀),³ aromaticity has been extended to non-planar molecular structures making its definition and quantification challenging. According to Hückel's rule a planar molecule is aromatic (antiaromatic) when it possesses $4n + 2$ ($4n$) delocalized π electrons.⁴ For approximate spherical molecules such as fullerenes, Hirsch's rule predicts that those with $2(n+1)^2$ π electrons are spherically aromatic.^{5,6} However, it is not possible to judge whether a fullerene is aromatic only based on this electron-counting rule. Hirsch's rule suggests that C₆₀ is non-aromatic, because it does not fulfil the rule.⁵

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† Electronic supplementary information (ESI) available: Fig. S1–S3 contain a schematic representation of the integration planes, a comparison of the ring-current profiles calculated at the all-electron and pseudo- π levels, and the differences in the ring-current profiles of the carbon nano-onions and the sum of the ring current profiles of the individual fullerenes, respectively. Cartesian coordinates of the optimized molecular structures and the current-density animations for C₆₀, C₅₄₀, C₂₄₀@C₅₄₀ and C₆₀@C₂₄₀@C₅₄₀. See DOI: <https://doi.org/10.1039/d2cp02718f>

The molecular magnetic response to an external magnetic field has been successfully used for elucidating and quantifying the aromatic character of planar and three-dimensional conjugated molecules. In aromatic molecules, the magnetic response is diamagnetic, characterized by a ring current flowing in the classical (diatropic) direction leading to a shielding of the external magnetic field. Paramagnetic response is associated with antiaromatic rings, which sustain a ring current in the opposite (paratropic) direction leading to a deshielding of the external magnetic field.^{7–9} For instance, C₆₀ responds to an external magnetic field by sustaining spherical current densities flowing in opposite directions inside and outside the molecular surface.¹⁰ Integration of the ring-current strength in C₆₀ yields a very small positive value when the external field points perpendicularly to a six-membered ring (6-MR), suggesting that it is weakly aromatic or non-aromatic.^{10,11} However, when the external magnetic field points perpendicularly to a five-membered ring (5-MR), the ring-current strength is paratropic suggesting that it is an antiaromatic compound.^{10,11} This has been confirmed by calculating the induced magnetic field, where the 6- and 5-MRs of fullerene have shielding and deshielding cones, respectively.¹² The classification of the aromaticity in fullerenes is not unambiguous. We recently showed that the magnetic response of the π electrons of C₆₀ has an antiaromatic character to the strong paratropic current density on the inside,¹¹ which promotes a deshielding cone,^{11,12} whereas recent calculations point out that internal cage shielding seems to be caused by σ delocalization.¹³



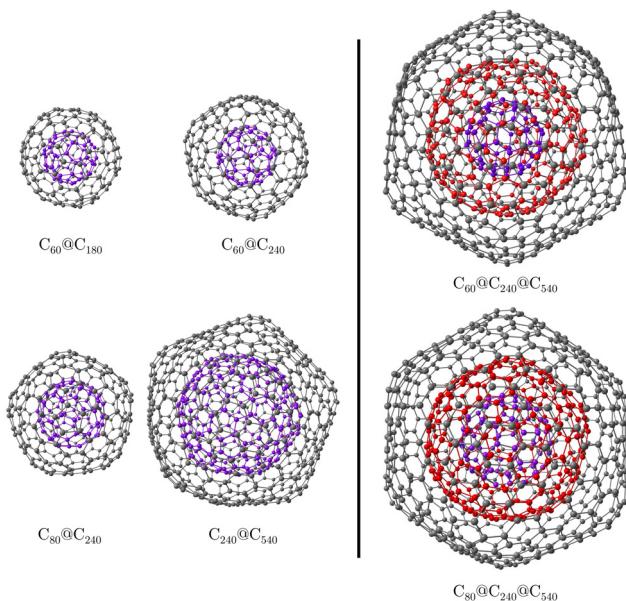


Fig. 1 The molecular structures of the studied double- (left) and triple-layer (right) CNOs. Each layer is shown in a different color.

Multilayer fullerenes, also known as carbon nano-onions (CNOs), are a class of concentric fullerenes with spacing distances between the carbon layers of about 3 to 4 Å.^{14–16} Since fullerenes have generally the 5-MRs surrounded by 6-MRs. The distance between the 5-MRs increases for larger fullerenes resulting in a pronounced sp^2 -hybridization of the carbon atoms.¹⁶ Recently, Hashmi and Lein scrutinized the structural properties of a number of double- and triple-layer CNOs.¹⁵ They suggested that the number of layers plays an important role for detecting these structures. They also noted that the highest occupied orbital (HOMO) and the lowest unoccupied orbital (LUMO) are alternately localized on different shells. The HOMO is located on an inner layer, while the LUMO is on an outer layer and they do not mix.^{15,17} The concentric structures allow rotation of the inner fullerenes with respect to the outer ones. This can be considered as a requirement for having a true CNO because when the distance between adjacent pairs is not large enough, C–C interlayer bonds can be formed.¹⁵

CNOs are of scientific interest due to their potential applications in electronic devices, in energy storage systems, as photosensors, as drug-deliverers, and as containers of heavy elements.^{14,18,19} Since magnetic interactions in CNOs are crucial for several of these applications, studying aromaticity by addressing the magnetic response of an external magnetic field is of interest.^{14,18–20} Here, we study the magnetic response properties of the CNOs proposed by Hashmi and Lein¹⁵ by calculating and visualizing magnetically induced current densities and the induced magnetic field. The studied CNOs are the $C_{60} @ C_{180}$, $C_{60} @ C_{240}$, and $C_{60} @ C_{540}$ double-layer systems as well as the $C_{60} @ C_{240} @ C_{540}$ and $C_{80} @ C_{240} @ C_{540}$ triple-layer CNOs shown in Fig. 1.

Calculating the magnetic response is computationally demanding for large compounds as the studied CNOs.

We therefore calculate the magnetic response using the pseudo- π model,²¹ which is computationally less demanding and has been found to yield accurate π -electron contributions to the magnetic response of organic and all-carbon structures.^{11,21–26} Calculations of the magnetic response of the π electrons is sufficient, because the π electrons provide the largest contribution to the global ring-current pathways.¹¹

The calculations show that the CNOs have strong net diatropic (paratropic) ring currents when the external magnetic field is directed perpendicularly to the 6-MRs (5-MRs). The CNOs have a deshielding region inside and a shielding one outside, which dominates for the larger CNOs as a result of the inner spherical paratropic current density and the diatropic outer one of each carbon layer. The magnetic response of some CNOs is approximately the sum of the magnetic response of the individual layers. For the magnetic response of largest CNOs, the deviation from the added magnetic response of each layer is larger, especially when the external magnetic field is perpendicular to a 6-MR. The reason for this is probably due to stretching of the C–C bonds of the 6-MR leading to an increased diamagnetic response.

2 Computational details

It has recently been shown that density functional theory (DFT) calculations using long-range corrected hybrid functionals in combination with the D3(BJ) dispersion correction lead energies of stacked π – π structures that are in good agreement with results obtained in CCSD(T) calculations.²⁷ These functionals correctly describe the molecular structures of large delocalized carbon structures.^{28,29} All molecular geometries were therefore fully optimized at the CAM-B3LYP-D3(BJ)/def2-SVP level.^{30–32} Magnetic properties calculated with hybrid functionals agree rather well with those calculated at the CCSD(T) and MP2 levels.^{33–36} Nuclear magnetic resonance (NMR) shieldings were therefore calculated at the CAM-B3LYP/def2-SVP level^{30,32} using gauge-including atomic orbitals (GIAOs).^{37,38}

The magnetic response was addressed by computing the magnetically induced current density (J^{ind}),^{39–42} and the first derivative of the induced magnetic field^{43–45} (\mathbf{B}^{ind}) with respect to a uniform external magnetic field (\mathbf{B}^{ext}) in the zero-field limit. The electronic structure calculations were performed using the Turbomole program.^{46,47} In the calculations of the J^{ind} and \mathbf{B}^{ind} we generally consider contributions from all electrons, *i.e.*, the sum of the core-, σ -, π -contribution. However, in this work, the π -electron contribution to the magnetic response was calculated at the CAM-B3LYP/def2-SVP^{30–32} level using the inexpensive pseudo- π model.²¹ The current density (${}^{\text{p}}\pi J^{\text{ind}}$) and induced magnetic field (${}^{\text{p}}\pi \mathbf{B}^{\text{ind}}$) were obtained with the pseudo- π model by calculating the magnetic response for a system where the carbon atoms are replaced by hydrogen atoms in the same positions. The ${}^{\text{p}}\pi J^{\text{ind}}$ and ${}^{\text{p}}\pi \mathbf{B}^{\text{ind}}$ computations were performed using the gauge-including magnetically induced currents (GIMIC)^{39–41} and Aromagnetic⁴⁸ programs, respectively. An external magnetic field ($|\mathbf{B}^{\text{ext}}| = 1$ T) can be considered to be



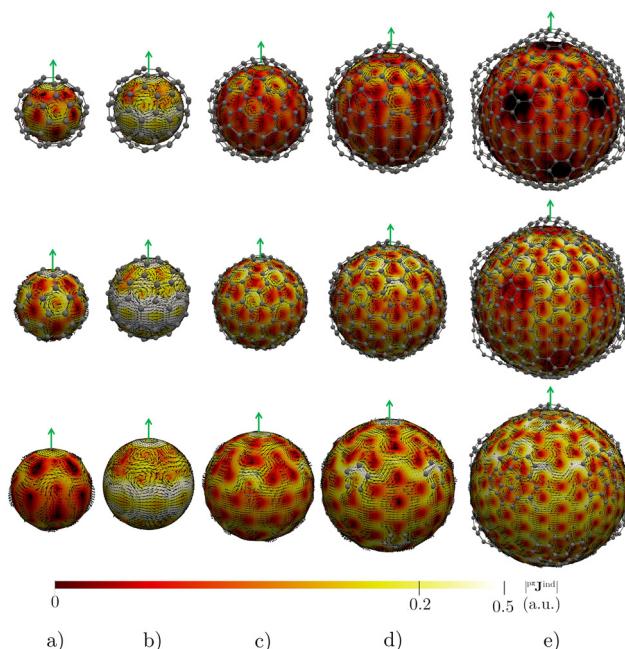


Fig. 2 ${}^{\text{pr}}\mathbf{J}^{\text{ind}}$ maps plotted at 1 bohr inside (top), at (middle), and at 1 bohr outside (bottom) the molecular surface of (a) C_{60} , (b) C_{80} , (c) C_{180} , (d) C_{240} , and (e) C_{540} calculated at the CAM-B3LYP/def2-SVP level. The $|{}^{\text{pr}}\mathbf{J}^{\text{ind}}|$ scale is given in atomic units (1 a.u. = 100.63 nA T⁻¹ Å⁻²). The direction of \mathbf{B}^{ext} is shown with the green arrow.

applied parallel to the z axis and perpendicularly to the 6- or 5-MRs.⁴⁹ The \mathbf{B}^{ind} analysis can then be made in terms of B_z^{ind} , which is equal to the zz component of the nucleus-independent chemical shift (NICS_{zz}) index.^{50–52} Since three-dimensional molecules represent a challenge when using magnetic criteria that depend on the orientation of \mathbf{B}^{ext} with respect to the molecular ring(s), isotropic NICS^{50,51} calculations using the pseudo- π model were also performed (${}^{\text{pr}}\text{NICS}$). NICS values can be understood as the average strength of \mathbf{B}^{ind} using an external magnetic field in the three Cartesian directions. The GIMIC program is also able to integrate the strength of the current density (J^{ind}) flowing through a plane intersecting one or more chemical bonds.^{39–41} The obtained strengths are particularly useful to quantify aromaticity in planar and three-dimensional molecules. Changes in the current-density flux along one coordinate of the integration plane

can be determined by calculating current-strength profiles.⁴¹ The profiles show the derivative of J^{ind} (dJ^{ind}/dr) with respect to one of the coordinates of the integration plane (r). The r coordinate can in fullerenes be chosen to begin at the symmetry axis. The plane intersects the molecular surface and extends outside the molecule where the current density vanishes (see Fig. S1 in the ESI[†]). The area under the profile is the ring-current strength. The unit of the pseudo- π ring-current strengths is nA T⁻¹, while ${}^{\text{pr}}\text{NICS}$ and ${}^{\text{pr}}B_z^{\text{ind}}$ are given in ppm. For visualization purposes, the molecular structures shown in the figures of the pseudo- π magnetic response correspond to the original carbon structure.

3 Results and discussion

Single-layer fullerenes

Before addressing the aromaticity of the CNOs, we consider the magnetic response of the single-layer fullerenes (C_{60} , C_{80} , C_{180} , C_{240} , and C_{540}) to assess how their magnetic response changes when they are embedded in concentric CNO structures. None of the these fullerenes satisfies Hirsch's spherical aromaticity rule and to our knowledge no magnetic response calculations have been reported for the larger fullerenes. The exceptions are C_{60} and C_{80} .^{10–12,23,53–55} The pseudo- π model offers the possibility to determine the current-density pathways and magnetic shielding without contamination from the core and σ electrons.^{11,21–26} Although the pseudo- π model was originally proposed to address the π magnetic response of planar systems,²¹ its applicability to non-planar structures is based on the fact that the carbon atoms in fullerenes do not significantly deviate from the sp^2 -hybridization.^{11,23,56}

C_{60} is the smallest molecule in our study. As reported in many studies,^{10–12,23,53,54} its magnetic response is characterized by paratropic ring currents flowing inside the cage, while diatropic (in the classical direction) ring currents appear outside it (see Fig. 2(a)). The paratropic ring current is represented as the negative region in the inner part of the ring-current profile leading to a pronounced minimum due to the local paratropic current in the 5-MR ring when \mathbf{B}^{ext} is perpendicular to the 5-MR ring (see Fig. 3). A deshielding zone appears therefore inside the fullerene. The 5-MRs sustain local paratropic ring currents that considerably distort the outer

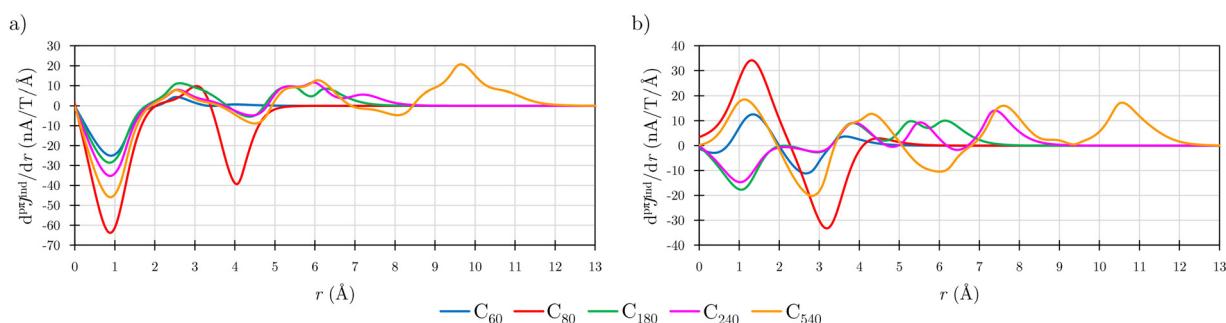


Fig. 3 The pseudo- π ring-current strength profiles ($d{}^{\text{pr}}\mathbf{J}^{\text{ind}}/dr$) of the single-layer fullerenes when \mathbf{B}^{ext} is perpendicular to (a) 5-MR and (b) 6-MR calculated at the CAM-B3LYP/def2-SVP level.

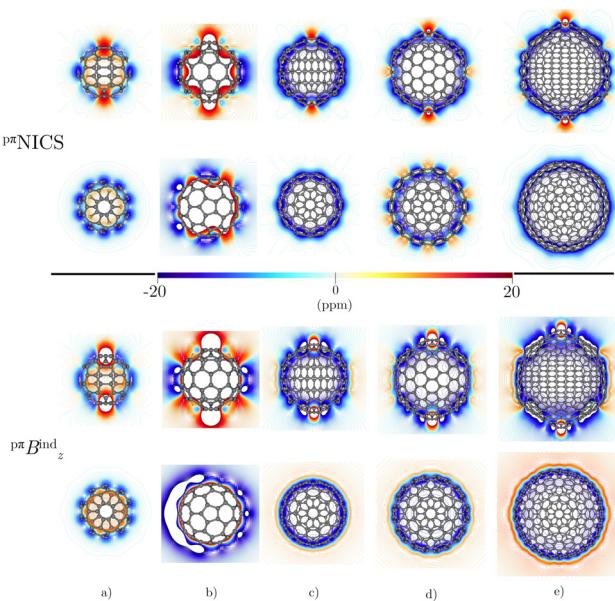


Fig. 4 Isolines of the isotropic $p\pi\text{NICS}$ and $p\pi B_z^{\text{ind}}$ plotted on a plane parallel (top) and perpendicular ($z = 0$) to \mathbf{B}^{ext} (bottom) for (a) C_{60} , (b) C_{80} , (c) C_{180} , (d) C_{240} , and (e) C_{540} calculated at the CAM-B3LYP/def2-SVP level.

spherical current. C_{60} is characterized by positive $p\pi B_z^{\text{ind}}$ values inside the cage as well as near the 5-MRs as shown in Fig. 3(a). The NICS value calculated in the cage center is close to zero,⁵³ whereas the π -component of NICS is deshielded with a value of about 20 ppm.⁵³ This is confirmed by the $p\pi B_z^{\text{ind}}$ isolines shown in Fig. 4(a). The $p\pi\text{NICS}$ function is similar to $p\pi B_z^{\text{ind}}$ because the z component dominates in B^{ind} . In fact, the $p\pi\text{NICS}$ and $p\pi B_z^{\text{ind}}$ values are identical (5.05 ppm) in the cage center.

Previous magnetic shielding studies on C_{60} at the pseudo- π level revealed that the inside deshielding calculated with the pseudo- π model differs slightly from the one obtained in all-electron π -component calculations.²² The discrepancy originates from the close proximity of the 5-MRs in C_{60} leading to a strong overlapping and mixing of the local shielding and deshielding contributions in the inner region of the cage. Since deshielding contributions of antiaromatic rings are not very well described at the pseudo- π level,^{11,21,22} a quantitative better description is obtained for the larger fullerenes with the 5-MRs spaced farther apart. The current densities calculated at the pseudo- π level are though in qualitative and quantitative agreement with the current densities sustained by the π orbitals at the all-electron level.^{21,23} The ring-current strengths calculated at the pseudo- π level can be used for determining the degree of aromaticity of fullerenes, since similar ring-current strengths are obtained in all-electron calculations.¹¹

Comparison of the ring current profile calculated at the pseudo- π model with that of the all-electron calculation shows that the profile is smoother at the pseudo- π level because contributions from the σ orbitals are not considered (see Fig. S2 in ESI[†]). The ring currents obtained at the two levels have the same tropicity with comparable strengths. Integration

Table 1 The ring-current strengths (in nA T^{-1}) of the single-layer fullerenes and their diatropic and paratropic contributions obtained by integrating the $p\pi\mathbf{J}^{\text{ind}}$ flux when \mathbf{B}^{ext} is perpendicular to a 5-MR and 6-MR (in parenthesis). $p\pi\mathbf{J}^{\text{ind}}$ is computed at the CAM-B3LYP/def2-SVP level

Molecule	Diatropic	Paratropic	Net
C_{60}	10.23 (17.63)	-32.46 (-15.26)	-22.23 (2.37)
C_{80}	17.60 (47.23)	-104.96 (-37.44)	-87.36 (9.78)
C_{180}	37.38 (28.30)	-40.02 (-23.69)	-2.64 (4.61)
C_{240}	40.14 (33.72)	-50.15 (-24.67)	-10.02 (9.04)
C_{540}	70.38 (90.36)	-75.77 (-48.97)	-5.38 (41.38)

of the strength of the pseudo- π ring current results in a weak positive value of 2.37 nA T^{-1} when \mathbf{B}^{ext} points perpendicularly to one of the 6-MRs. When \mathbf{B}^{ext} is perpendicular to 5-MR, the integration yields a strong paratropic ring-current strength of -22.23 nA T^{-1} , indicating local antiaromaticity (Table 1). These values are in qualitative agreement with the previously reported ring-current strengths of 3 nA T^{-1} and -15 nA T^{-1} that were obtained in all-electron (total) calculations with the magnetic field perpendicular to the 6-MR and 5-MR, respectively.¹⁰

Previous studies have shown that C_{80} has a singly occupied HOMO leading to a Jahn-Teller distortion of the structure and a triplet ground state.^{5,6,57,58} NICS(0) calculations yielded positive values in the middle of the 5-MRs and 6-MRs.⁵ Unconstrained structure optimization of the singlet state yielded a C_{2v} structure with a HOMO-LUMO gap of 2.18 eV at the CAM-B3LYP-D3(BJ)/def2-SVP level. This isomer has been reported to be the second most stable singlet after the highly distorted D_{5d} structure.^{57,58} The $p\pi\mathbf{J}^{\text{ind}}$ calculations reveal a strong paratropic current density flowing mainly at the equator of the fullerene both inside and outside the cage suggesting that it is antiaromatic (Fig. 2(b)) with a deep minimum in the ring-current profile originating from the local paratropic ring current of the 5-MRs. (Fig. (3)). Integration of the ring-current strength yields a value of -87.36 nA T^{-1} with a weak diatropic contribution. C_{80} is strongly deshielded both inside and outside the molecular surface (Fig. 4(b)). However, when \mathbf{B}^{ext} is perpendicular to a 6-MR, the ring current is diatropic with a strength of 9.78 nA T^{-1} .

C_{180} and the larger single-layer fullerenes (C_{240} and C_{540}) exhibit diamagnetic response inside the cage. They have a spherical paratropic inner ring-current and a diatropic outer one. The 5-MRs maintain local paratropic circulations. In the larger fullerenes, the diatropic current density near the equator outside the cage is more homogeneous than in the smaller ones (Fig. 2(c)–(e)). The current-density pathways are animated for some systems in the ESI.[†] Calculation of \mathbf{B}^{ind} yielded a more negative magnetic shielding than -20 ppm . The local ring-current loops at the 5-MRs give rise to small deshielding cones that are not able to suppress the overall diamagnetic response of the molecular cage (Fig. 4(c)–(e)). The ring-current profile has a positive maximum in the vicinity of the 6-MRs when \mathbf{B}^{ext} is perpendicular to a 6-MR (Fig. 3). The integration of the ring-current strengths, for a \mathbf{B}^{ext} perpendicular to a 5-MR, leads to the paratropic values of -2.64 , -10.02 , -5.38 nA T^{-1} for C_{180} , C_{240} , and C_{540} , respectively (Table 1). When \mathbf{B}^{ext} is



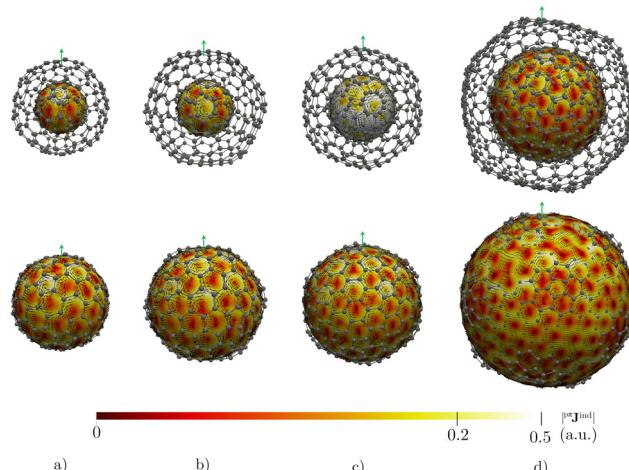


Fig. 5 πJ^{ind} maps plotted at the inner (top) and at the outer (bottom) layer surface of (a) $C_{60}@\text{C}_{180}$, (b) $C_{60}@\text{C}_{240}$, (c) $C_{80}@\text{C}_{240}$, and (d) $C_{240}@\text{C}_{540}$ calculated at the CAM-B3LYP/def2-SVP level. The $|J^{\text{ind}}$ scale is given in atomic units (1 a.u. = $100.63 \text{ nA T}^{-1} \text{ \AA}^{-2}$). The direction of \mathbf{B}^{ext} is shown with the green arrow.

perpendicular to a 6-MR, the current strengths become positive with values of 4.61, 9.04, and 41.38 nA T^{-1} for C_{180} , C_{240} , and C_{540} , respectively. The strengths of the diatropic and paratropic contributions that increase with the fullerene size almost cancel. The large fullerenes can still be considered non-aromatic, even though they shield the magnetic field inside the cage. However, C_{540} acts like a π -aromatic system when \mathbf{B}^{ext} is oriented perpendicular to a 6-MR. This is the only case where the diatropic component is very pronounced suggesting that the π delocalization increases as the fullerene size increases, at least for fullerenes belonging to the I_h point group.⁵⁹ Since the distance between the 5-MRs substantially increases, large fullerenes can sustain strong spherical diatropic ring currents that are not significantly distorted by the local paratropic ring currents of the 5-MRs. Very large fullerenes will most likely be aromatic with dominating diatropic ring currents.

Double-layer CNOs

The layers of the CNOs are concentric with a distance of 3–4 Å between them. The interaction is then dominated by van der Waals forces. Short inter-layer distances may lead to covalent bonds between the layers, whereas when they are too far apart, the inner fullerene can move significantly away from the concentricity inside the cage. The system can then be considered as a molecule inside a nanoparticle. CNOs are the intermediate case where the layers can rotate with respect to each other while remaining practically concentric.¹⁵ The π - π stacking between the layers plays an important role for the stabilization of the structures, which might also be reflected in their magnetic response.

The molecular structures of the CNOs deviate slightly from the I_h point group because the minimum-energy structures have geometries whose symmetry axis of the 5-MRs and 6-MRs of different layers do not coincide. Thus, the symmetry

Table 2 Ring-current strengths (in nA T^{-1}) divided into its diatropic and paratropic components of the CNOs obtained from πJ^{ind} calculated at the CAM-B3LYP/def2-SVP level using the minimum energy structures. \mathbf{B}^{ext} points in the direction of the z axis, which it is not exactly perpendicular to a 5-MR or a 6-MR

CNO	Diatropic	Paratropic	Net
$C_{60}@\text{C}_{180}$	51.09	-22.40	28.68
$C_{60}@\text{C}_{240}$	59.94	-58.84	1.10
$C_{80}@\text{C}_{240}$	91.37	-320.86	-229.48
$C_{240}@\text{C}_{540}$	103.47	-51.22	52.25
$C_{60}@\text{C}_{240}@\text{C}_{540}$	126.33	-78.80	47.52
$C_{80}@\text{C}_{240}@\text{C}_{540}$	226.63	-167.45	59.18

axis cannot be aligned to a C_5 or C_3 axis as for single-layer fullerenes and \mathbf{B}^{ext} cannot be oriented parallel to a common symmetry axis. The magnetic response is therefore more qualitative than quantitative because the ring-current strengths slightly vary depending on how the layers of the CNO is oriented with respect to \mathbf{B}^{ext} .

For double-layer $C_{60}@\text{C}_{180}$ and $C_{60}@\text{C}_{240}$ CNOs, the πJ^{ind} pictures show that the inner layer has practically the same magnetic response as C_{60} (see Fig. 5). The spherical current density inside the cage is paratropic and outside it is diatropic. The same holds for the magnetic response of the outer C_{180} (or C_{240}) layer. A strong diatropic current density appears outside the CNO. In the case of $C_{60}@\text{C}_{240}$, the diatropic contribution seems to dominate the global magnetic behavior.

Integration of πJ^{ind} through a plane that intersects both layers leads to a net ring-current strength of 28.68 and 1.10 nA T^{-1} for $C_{60}@\text{C}_{180}$ and $C_{60}@\text{C}_{240}$, respectively (Table 2). The diatropic component is larger. However, the paratropic one is also strong. We can associate the difference in the strengths for the different two-layered CNOs to their orientation with respect to \mathbf{B}^{ext} . The strong diatropic current density outside the CNO produces a shielding that practically covers the entire system as shown in Fig. 5. The small local paratropic circulations in the 5-MRs lead to local deshielding cones. Inside the innermost cage, the negative π NICS and πB_z^{ind} values are small because the shielding from the outer layer is suppressed due to the inner deshielding caused by the inner C_{60} shell. The similarity between π NICS and πB_z^{ind} in Fig. 5 shows that \mathbf{B}^{ind} is almost independent of the direction of \mathbf{B}^{ext} .

The magnetic response of $C_{80}@\text{C}_{240}$ differs from the sum of the magnetic response of its layers. The current density is strongly paratropic at the C_{80} surface and diatropic at the C_{240} shell (Fig. 5 and 6) leading to a very strong paratropic ring current of -229.5 nA T^{-1} .

The magnetic response of $C_{240}@\text{C}_{540}$ originates from a paratropic current density inside and a diatropic one outside as shown in Fig. 5. The π NICS and πB_z^{ind} calculations reveal a shielded CNO where the most negative values appear in the interior of C_{240} (Fig. 6). The ring-current strength integration yielded a strongly diatropic net value of 52.25 nA T^{-1} .

The data in Table 2 show that it is not straightforward to find a clear trend and they could even be interpreted as the opposite to what the πB_z^{ind} analysis shows because the

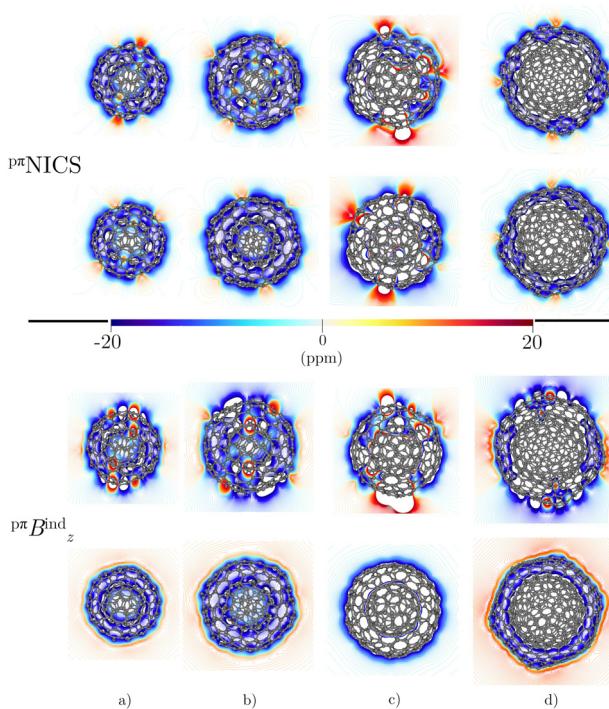


Fig. 6 Isolines of the isotropic $p\pi$ NICS and $p\pi B_z^{\text{ind}}$ on a plane parallel (top) and a perpendicular plane ($z = 0$) to \mathbf{B}^{ext} (bottom) for (a) $C_{60}@\mathcal{C}_{180}$, (b) $C_{60}@\mathcal{C}_{240}$, (c) $C_{80}@\mathcal{C}_{240}$, and (d) $C_{240}@\mathcal{C}_{540}$ calculated at the CAM-B3LYP/def2-SVP level.

Table 3 Ring-current strengths (in nA T^{-1}) divided into their diatropic and paratropic components of the C_5 -CNOs and C_3 -CNOs (in parenthesis) obtained from $p\pi \mathbf{J}^{\text{ind}}$ computed at the CAM-B3LYP/def2-SVP level

CNO	Diatropic	Paratropic	Net
$C_{60}@\mathcal{C}_{180}$	47.2 (4.8)	-73.7 (-38.4)	-26.4 (6.5)
$C_{60}@\mathcal{C}_{240}$	50.3 (63.9)	-82.8 (-50.1)	-31.9 (13.8)
$C_{80}@\mathcal{C}_{240}$	121.2 (654.6)	-320.1 (-558.1)	-198.9 (96.4)
$C_{240}@\mathcal{C}_{540}$	110.5 (133.7)	-125.6 (-50.5)	-15.1 (83.2)
$C_{60}@\mathcal{C}_{240}@\mathcal{C}_{540}$	120.6 (157.2)	-158.0 (-87.3)	-37.5 (69.9)
$C_{80}@\mathcal{C}_{240}@\mathcal{C}_{540}$	173.0 (273.6)	-446.0 (-184.8)	-273.0 (88.8)

ring-current strengths depend on the orientation of the molecule relatively to \mathbf{B}^{ext} . Information about the cooperative

magnetic response of the concentric CNOs is obtained by comparing the magnetic response of the CNOs with the sum of the magnetic response of the single-layer fullerenes. We oriented the fullerene layers such that their C_5 (and C_3) axes coincide and \mathbf{B}^{ext} is oriented parallel to a common symmetry axis. The symmetric structures are denoted C_5 -CNO and C_3 -CNO, respectively.

Details in the ring-current profiles are studied for the C_n -CNO structures (Fig. 7). The CNOs containing C_{60} have similar ring-current profiles in the C_{60} part of the CNOs. In C_3 -CNO, the paratropic component is sustained in the inner part of the structure due to C_{60} , while at long distances the magnetic response is dominated by the one of the outer layer(s). The pseudo- π calculations show that $p\pi \mathbf{J}^{\text{ind}}$ of the C_n -CNOs qualitatively agrees with the one obtained in all-electron calculations. The profiles of the ring current of $C_{60}@\mathcal{C}_{240}$ calculated at the two levels are shown in Fig. S2b of the ESI.† Integration of the two ring-current profiles yields ring-current strengths with the same tropicity and largely the same strengths. Despite the rapidly oscillating contributions from the σ orbitals to the ring-current profile in the all-electron calculation, the ring-current profiles calculated at the pseudo- π and all-electron levels display largely the same trend.

Integration of $p\pi \mathbf{J}^{\text{ind}}$ with planes that intersect the common 5-MRs and 6-MRs yielded the ring-current strengths of C_3 -CNO and C_5 -CNO shown in Table 3. The current density obtained with \mathbf{B}^{ext} along the C_5 axis is paratropic for all double-layer CNOs, whereas the current density is diatropic when the \mathbf{B}^{ext} points along the C_3 axis. The ring-current profile of $C_{80}@\mathcal{C}_{240}$ has pronounced minima and maxima leading to a ring-current strength that differs significantly from the sum of the contributions of the individual fullerenes (Fig. 7). The integrated ring-current strength of the C_3 -CNO and C_5 -CNO structures of $C_{80}@\mathcal{C}_{240}$ deviates by almost 100 nA T^{-1} from the sum of the ring-current strengths of the individual fullerenes (Table 3).

Calculations on the C_3 -CNO and C_5 -CNO structures of $C_{240}@\mathcal{C}_{540}$ show that the ring-current strength can be expressed as the sum of the individual ring-current strengths when the magnetic field is oriented along the C_5 axis of the C_5 -CNO, whereas the ring-current strength differs by about 30 nA T^{-1} in

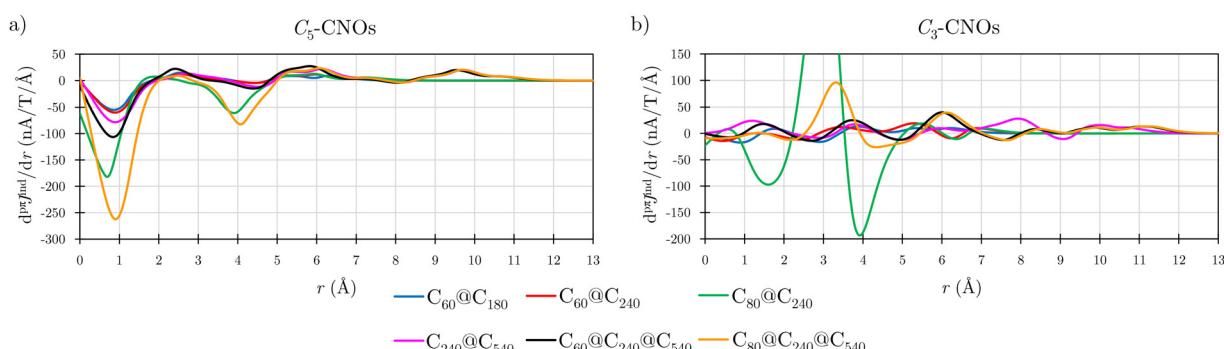


Fig. 7 The pseudo- π ring-current strength profiles ($d p\pi \mathbf{J}^{\text{ind}}/dr$) in (a) C_5 -CNOs and (b) C_3 -CNOs corresponding to the cases when \mathbf{B}^{ext} is perpendicular to common 5-MRs and 6-MRs of the layers, respectively. The current density is calculated at the CAM-B3LYP/def2-SVP level. The green line corresponding to C_3 - $C_{80}@\mathcal{C}_{240}$ has a maximum of more than $400 \text{ nA T}^{-1} \text{ Å}^{-1}$ exceeding the range of the graph.



Table 4 The net ring-current strengths (in nA T^{-1}) of the C_5 -CNOs and C_3 -CNOs are compared to the sum of the ring-current strengths of the individual fullerenes. The strengths are calculated at the CAM-B3LYP/def2-SVP level

CNO	C_5 -CNOs	Sum	C_3 -CNOs	Sum
$C_{60}@\text{C}_{180}$	-26.4	-24.9	6.5	7.0
$C_{60}@\text{C}_{240}$	-31.9	-32.2	13.8	11.4
$C_{80}@\text{C}_{240}$	-198.9	-97.4	96.4	18.8
$C_{240}@\text{C}_{540}$	-15.1	-15.4	83.2	50.4
$C_{60}@\text{C}_{240}@\text{C}_{540}$	-37.5	-37.6	69.9	52.8
$C_{80}@\text{C}_{240}@\text{C}_{540}$	-273.0	-102.8	88.8	60.2

the corresponding calculation on the C_3 -CNO structure of $\text{C}_{240}@\text{C}_{540}$ (Table 4). The ring-current strengths of most of the C_5 -CNOs are almost equal to the sum of the single-layer components (Table 1). $C_5\text{-}C_{60}@\text{C}_{240}$ has a net strength of -31.90 nA T^{-1} , which agrees well with the sum of the ring-current strengths of C_{60} (-22.23 nA T^{-1}) and C_{240} (-10.02 nA T^{-1}). A small cooperative effect can also be seen in the almost vanishing difference in the ring-current profiles (Fig. S3 in ESI[†]). However, the

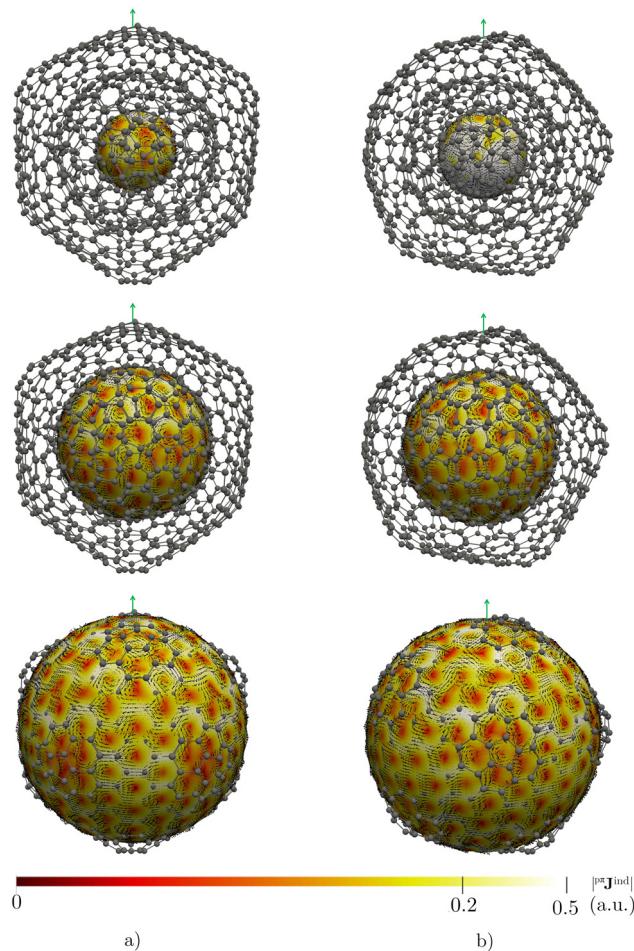


Fig. 8 prJ^{ind} maps plotted at the inner (top), at the middle (in the middle), and at the outer (bottom) layer surface of (a) $\text{C}_{60}@\text{C}_{240}@\text{C}_{540}$ and (b) $\text{C}_{80}@\text{C}_{240}@\text{C}_{540}$ calculated at the CAM-B3LYP/def2-SVP level. The $|\text{J}^{\text{ind}}|$ scale is given in atomic units (1 a.u. = $100.63 \text{ nA T}^{-1} \text{ \AA}^{-2}$). The direction of B^{ext} is shown with a green arrow.

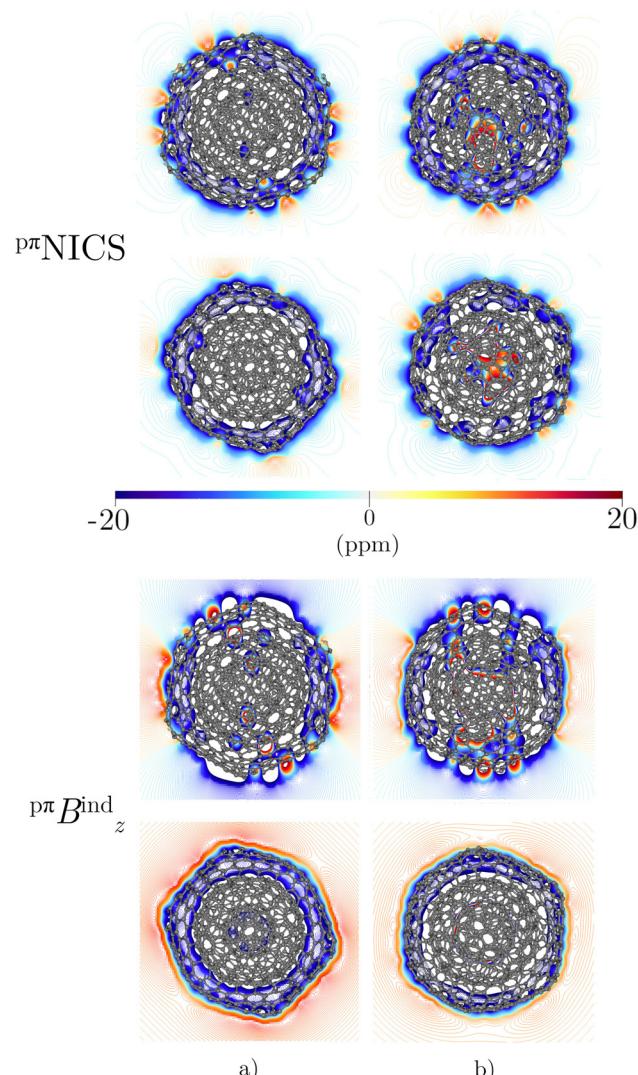


Fig. 9 Isolines of isotropic prNICS and $\text{prB}_z^{\text{ind}}$ on a plane parallel (top) and a perpendicular plane ($z = 0$) to B^{ext} (bottom) for $\text{C}_{60}@\text{C}_{240}@\text{C}_{540}$ and (b) $\text{C}_{80}@\text{C}_{240}@\text{C}_{540}$ calculated at the CAM-B3LYP/def2-SVP level.

magnetic response of $\text{C}_{80}@\text{C}_{240}$ differs completely from the magnetic response of the individual layers (Fig. S3, ESI[†]) probably because the position of the HOMO and LUMO orbitals do not alternate as the other CNOs.¹⁵ Similar additive features have been found in organic compounds where the global magnetic response could be reproduced by adding individual ring contributions.^{60,61} The cooperative magnetic response is larger for the C_3 -CNOs, whose B^{ext} is oriented perpendicular to the 6-MRs maximizing their diamagnetic contribution to the magnetic response. The CNOs have longer C-C bonds in the 6-MRs than in fullerenes, whereas the 5-MRs have practically the same bond lengths as in the fullerenes.

Triple-layer CNOs

The triple-layer $\text{C}_{60}@\text{C}_{240}@\text{C}_{540}$ is dominated by the diamagnetic response judged from the current density (Fig. 8(a)) and the magnetic shielding (Fig. 9(a)). Animation of prJ^{ind} confirms

a bridging magnetic response between layers of large double- and triple-layer CNOs (see the ESI†). $C_{60}@C_{240}@C_{540}$ is aromatic due to the enhanced diatropic contributions along the ring-current profile compared to sum of the individual fullerene contributions (Fig. 7(a)). For the C_5 -CNO and C_3 -CNO structures, the obtained ring-current strengths are -37.5 and 69.9 nA T^{-1} , respectively. The sum of the ring-current strengths of the individual fullerenes yields the same ring-current strength as obtained for C_5 -CNO, whereas ring-current strength of the C_3 -CNO structure of $C_{60}@C_{240}@C_{540}$ is 17 nA T^{-1} larger than the sum of the fullerene contributions. The deviation is also seen in the difference of the ring current profile (in Fig. S3, ESI†). $C_{80}@C_{240}@C_{540}$ is deshielded inside the innermost layer, while the region between the second and third layers is shielded (Fig. 9(b)) because the C_{80} layer sustains a strong paratropic ring current near its surface, while the C_{240} and C_{540} layers are dominated by diatropic current densities. The ring-current strengths of the C_5 -CNO and C_3 -CNO structures (Table 3 and Fig. S3, ESI†) deviate significantly from the sum of the ring-current strengths of the individual layers (Table 1), which confirms that there is a strong interaction between the fullerenes in the CNO structures containing C_{80} .

4 Conclusions

We have studied the magnetic response of double- and triple-layer CNOs and the individual fullerenes by calculating magnetically induced current densities and the induced magnetic field using the pseudo- π model. We have considered CNOs consisting of C_{60} , C_{80} , C_{180} , C_{240} and C_{540} fullerenes. Since each layer consists of pentagonal and hexagonal rings, the overall magnetic response of the single-layer fullerenes have significant paratropic and diatropic current-density contributions. The pentagonal rings sustain local paratropic ring currents. The fullerenes also sustain inner paratropic and outer diatropic spherical ring currents. The main exception is C_{80} , which has a strong paramagnetic response because of the open-shell character of its ground state.

The distance between the twelve pentagons increases with increasing size of the fullerenes leading to a dominating diatropic current density and diamagnetic response for the larger ones. The diatropic ring current is seen in the induced magnetic field, which is deshielding for C_{60} and C_{80} . The larger ones are shielded on the inside and on the outside suggesting that the π delocalization increases as the fullerene size increases. In the limit of infinitely big fullerenes, the 6-MRs are connected by graphene-like flakes that are weakly perturbed near the 5-MRs.

Multilayer fullerenes that can be considered to be true CNOs must have an inter-layer distance that is large enough to prevent formation chemical bonds between them. The HOMO and LUMO are located in different layers allowing the layers to rotate with respect to each other.¹⁵ Fully optimization of the molecular structures of the CNOs does lead to layers whose pentagons coincide with the C_5 symmetry axis. Therefore, we

have also studied the magnetic response of CNOs whose pentagons are aligned and the external magnetic field is applied along the C_5 and the C_3 axis.

The weak coupling between the layers also results in a global magnetic response of the CNOs that can to first order be obtained by adding the magnetic response of the individual layers, especially when the magnetic field is oriented along the C_5 axis. The magnetic coupling between the layers is stronger for the larger CNOs when the magnetic field is parallel to the C_3 axis. The magnetic response of the CNOs containing C_{80} cannot be accurately expressed as the sum of the magnetic response of the individual fullerenes. It is hard to accurately assess the magnetic coupling based on calculated ring-current strengths because the ring-current strengths depend on the orientation of the layers with respect to the direction of the external magnetic field. For the largest CNOs, calculations on the fully optimized structures yield ring-current strengths in good agreement with the ones obtained for CNOs with symmetry aligned layers and the magnetic field along the C_3 axis. Larger CNOs favor π delocalization giving rise to a pronounced diamagnetic response and a strong ring current.

The magnetic response of the CNOs containing C_{80} cannot be expressed as a sum of the magnetic response of the individual fullerenes. The magnetic coupling between C_{80} and the other fullerenes in the CNO is very strong suggesting that C_{80} -containing CNOs cannot be considered as true CNOs. Thus, a weak magnetic coupling between the layers is an additional requirement for CNOs. The CNOs have longer C–C bonds in the hexagonal rings as compared to those of the corresponding fullerenes, whereas the pentagons have practically the same bond lengths as in the fullerenes.

Author contributions

M. O.-I. suggested pseudo- π current-density studies on CNOs. M. O.-I. and D. S. performed the calculations, discussed the results, and contributed to the final manuscript.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- 1 P. v R. Schleyer, *Chem. Rev.*, 2001, **101**, 1115–1118.
- 2 G. Merino and M. Solà, *Phys. Chem. Chem. Phys.*, 2016, **18**, 11587–11588.
- 3 H. W. Kroto, J. R. Heath, S. C. O'Brien, R. F. Curl and R. E. Smalley, *Nature*, 1985, **318**, 162–163.
- 4 K. Yates, *Hückel Molecular Orbital Theory*, Academic Press, 2012.
- 5 A. Hirsch, Z. Chen and H. Jiao, *Angew. Chem., Int. Ed.*, 2000, **39**, 3915–3917.
- 6 Z. Chen, H. Jiao, A. Hirsch and W. Thiel, *J. Mol. Model.*, 2007, **7**, 161–163.
- 7 N. F. Ramsey, *Phys. Rev.*, 1950, **78**, 699–703.
- 8 J. A. Pople, *Mol. Phys.*, 1958, **1**, 175–180.
- 9 R. McWeeny, *Mol. Phys.*, 1958, **1**, 311–321.
- 10 M. P. Johansson, J. Jusélius and D. Sundholm, *Angew. Chem., Int. Ed.*, 2005, **44**, 1843–1846.
- 11 M. Orozco-Ic, M. Dimitrova, J. Barroso, D. Sundholm and G. Merino, *J. Phys. Chem. A*, 2021, **125**, 5753–5764.
- 12 N. D. Charistos and A. Muñoz-Castro, *J. Phys. Chem. C*, 2018, **122**, 9688–9698.
- 13 M. Orozco-Ic, N. D. Charistos, A. Muñoz-Castro, R. Islas, D. Sundholm and G. Merino, *Phys. Chem. Chem. Phys.*, 2022, **24**, 12158–12166.
- 14 O. Mykhailiv, H. Zubyk and M. E. Plonska-Brzezinska, *Inorg. Chim. Acta*, 2017, **468**, 49–66.
- 15 M. A. Hashmi and M. Lein, *J. Phys. Chem. C*, 2018, **122**, 2422–2431.
- 16 M. Zeiger, N. Jäckel, V. N. Mochalin and V. Presser, *J. Mater. Chem. A*, 2016, **4**, 3172–3196.
- 17 J. A. Luque-Urrutia, A. Poater and M. Solà, *Chem. – Eur. J.*, 2020, **26**, 804–808.
- 18 M. E. Plonska-Brzezinska, *ChemNanoMat*, 2019, **5**, 568–580.
- 19 V. Dhand, M. Yadav, S. H. Kim and K. Y. Rhee, *Carbon*, 2021, **175**, 534–575.
- 20 A. I. Shames, V. Y. Osipov, A. Y. Vul, Y. Kaburagi, T. Hayashi, K. Takai and T. Enoki, *Carbon*, 2013, **61**, 173–189.
- 21 P. W. Fowler and E. Steiner, *Chem. Phys. Lett.*, 2002, **364**, 259–266.
- 22 N. D. Charistos, A. Muñoz-Castro and M. P. Sigalas, *Phys. Chem. Chem. Phys.*, 2019, **21**, 6150–6159.
- 23 A. Soncini, R. G. Viglione, R. Zanasi, P. W. Fowler and L. W. Jenneskens, *C. R. Chim.*, 2006, **9**, 1085–1093.
- 24 M. Antić, B. Furtula and S. Radenković, *J. Phys. Chem. A*, 2020, **124**, 371–378.
- 25 M. Orozco-Ic and G. Merino, *Chemistry*, 2021, **3**, 1381–1391.
- 26 M. Orozco-Ic, R. R. Valiev and D. Sundholm, *Phys. Chem. Chem. Phys.*, 2022, **24**, 6404–6409.
- 27 S. Tsuzuki and T. Uchimaru, *Phys. Chem. Chem. Phys.*, 2020, **22**, 22508–22519.
- 28 I. Casademont-Reig, R. Guerrero-Avilés, E. Ramos-Cordoba, M. Torrent-Sucarrat and E. Matito, *Angew. Chem., Int. Ed.*, 2021, **60**, 24080–24088.
- 29 D. W. Szczepanik, M. Solà, M. Andrzejak, B. Pawelek, J. Dominikowska, M. Kukulka, K. Dyduch, T. M. Krygowski and H. Szatylowicz, *J. Comput. Chem.*, 2017, **38**, 1640–1654.
- 30 T. Yanai, D. P. Tew and N. C. Handy, *Chem. Phys. Lett.*, 2004, **393**, 51–57.
- 31 S. Grimme, J. Antony, S. Ehrlich and H. Krieg, *J. Chem. Phys.*, 2010, **132**, 154104.
- 32 F. Weigend and R. Ahlrichs, *Phys. Chem. Chem. Phys.*, 2005, **7**, 3297–3305.
- 33 R. R. Valiev, H. Fliegl and D. Sundholm, *Chem. Commun.*, 2017, **53**, 9866–9869.
- 34 R. R. Valiev, I. Benkyi, Y. V. Konyshev, H. Fliegl and D. Sundholm, *J. Phys. Chem. A*, 2018, **122**, 4756–4767.
- 35 L. N. Wirz, M. Dimitrova, H. Fliegl and D. Sundholm, *J. Phys. Chem. Lett.*, 2018, **9**, 1627–1632.
- 36 S. Lehtola, M. Dimitrova, H. Fliegl and D. Sundholm, *J. Chem. Theory Comput.*, 2021, **17**, 1457–1468.
- 37 R. Ditchfield, *Mol. Phys.*, 1974, **27**, 789–807.
- 38 K. Wolinski, J. F. Hinton and P. Pulay, *J. Am. Chem. Soc.*, 1990, **119**, 8251–8260.
- 39 J. Jusélius, D. Sundholm and J. Gauss, *J. Chem. Phys.*, 2004, **121**, 3952–3963.
- 40 H. Fliegl, S. Taubert, O. Lehtonen and D. Sundholm, *Phys. Chem. Chem. Phys.*, 2011, **13**, 20500–20518.
- 41 D. Sundholm, H. Fliegl and R. J. F. Berger, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 2016, **6**, 639.
- 42 D. Sundholm, M. Dimitrova and R. J. F. Berger, *Chem. Commun.*, 2021, **57**, 12362–12378.
- 43 G. Merino, T. Heine and G. Seifert, *Chem. – Eur. J.*, 2004, **10**, 4367–4371.
- 44 T. Heine, R. Islas and G. Merino, *J. Comput. Chem.*, 2007, **28**, 302–309.
- 45 R. Islas, T. Heine and G. Merino, *Acc. Chem. Res.*, 2012, **45**, 215–228.
- 46 R. Ahlrichs, M. Bär, M. Häser, H. Horn and C. Kölmel, *Chem. Phys. Lett.*, 1989, **162**, 165–169.
- 47 F. Furche, R. Ahlrichs, C. Hättig, W. Klopper, M. Sierka and F. Weigend, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 2014, **4**, 91–100.
- 48 M. Orozco-Ic, J. L. Cabellos and G. Merino, *Aromagnetic*, Cinvestav-Merida, Mexico, 2016.
- 49 F. A. L. Anet and D. J. OLeary, *Concepts Magn. Reson., Part A*, 1991, **3**, 193–214.
- 50 P. v R. Schleyer, C. Maerker, A. Dransfeld, H. Jiao and N. J. R. van Eikema Hommes, *J. Am. Chem. Soc.*, 1996, **118**, 6317–6318.
- 51 Z. Chen, C. S. Wannere, C. Corminboeuf, R. Puchta and P. v R. Schleyer, *Chem. Rev.*, 2005, **105**, 3842–3888.
- 52 H. Fallah-Bagher-Shaidei, C. S. Wannere, C. Corminboeuf, R. Puchta and P. v R. Schleyer, *Org. Lett.*, 2006, **8**, 863–866.
- 53 Z. Chen, J. I. Wu, C. Corminboeuf, J. Bohmann, X. Lu, A. Hirsch and P. v R. Schleyer, *Phys. Chem. Chem. Phys.*, 2012, **14**, 14886–14891.
- 54 R. Zanasi and P. W. Fowler, *Chem. Phys. Lett.*, 1995, **238**, 270–280.
- 55 Z. Chen, J. Cioslowski, N. Rao, D. Moncrieff, M. Bühl, A. Hirsch and W. Thiel, *Theor. Chem. Acc.*, 2001, **106**, 364–368.
- 56 G. Monaco, F. F. Summa and R. Zanasi, *J. Chem. Inf. Model.*, 2021, **61**, 270–283.



57 C. Rios and R. Salcedo, *Molecules*, 2012, **17**, 14588–14601.

58 F. Furche and R. Ahlrichs, *J. Chem. Phys.*, 2001, **114**, 10362–10367.

59 P. Calaminici, G. Geudtner and A. M. Köster, *J. Chem. Theory Comput.*, 2009, **5**, 29–32.

60 M. Orozco-Ic, J. Barroso, N. D. Charistos, A. Muñoz-Castro and G. Merino, *Chem. – Eur. J.*, 2019, **26**, 326–330.

61 E. Paenurk and R. Gershoni-Poranne, *Phys. Chem. Chem. Phys.*, 2022, **24**, 8631–8644.

