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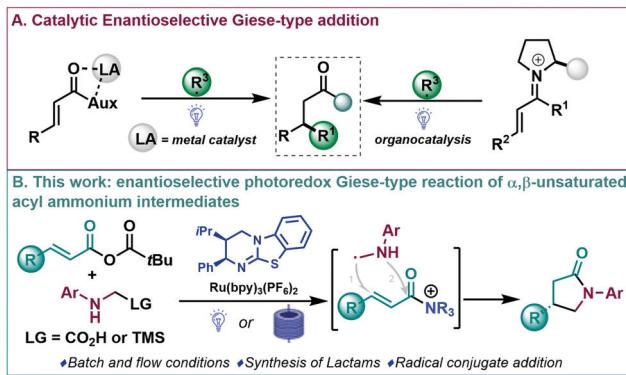
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The photocatalytic generation of α -amino radicals is combined with chiral isothiourea derived α,β -unsaturated acyl ammonium intermediates. The reaction proceeds via a [3+2] radical-polar crossover mechanism to generate γ -lactams in good yields and enantioselectivities. The enantioselective radical conjugate addition was carried out under batch and flow conditions†

The selective synthesis of enantioenriched molecules is a fast-moving research area in organic synthesis and the development of methodologies for their preparation still plays a crucial role for the synthesis of prevalent structures in natural and synthetic products.¹ Michael additions (or conjugate additions) are among the most widely studied transformations in organic synthesis and are a powerful and appealing synthetic tool for the direct and rapid modification of activated alkenes.² In this context, the addition of radicals to electron poor alkenes (Giese reaction) is a well-established alternative for the construction of C–C bonds.^{2b,3} Despite modern organic synthesis relying heavily upon radical species,⁴ efficient catalytic enantioselective versions are still limited,^{3,5} which can be pinpointed to the high background (uncatalyzed) reactivity of the highly reactive radicals involved.⁵ In this area, Sibi and coworkers⁶ described one of the first enantioselective conjugate radical additions using chiral Lewis acid catalysis. Inspired by these pioneering studies, and taking advantage of the development of dual catalytic systems

for the efficient construction of stereochemically defined motifs,⁷ a dual Lewis acid/photoredox catalysed enantioselective addition of α -amino radicals to Michael acceptors has been reported (Scheme 1A, left).⁸ In addition, other radical precursors and Lewis acid-based catalytic systems have been described in enantioselective photoredox Giese-type reactions.⁹ In these processes, these Lewis acid catalysed systems all require two-point binding of an acyl auxiliary bearing Michael acceptor to ensure bidentate coordination with chiral Lewis acids to achieve high enantioinduction (Scheme 1A, left).⁹ During the last decade, organocatalysis has proved well suited to be combined with photochemical processes for the enantioselective construction of complex molecules.^{7,10} As an overview, enantioselective Giese-type reactions of transient iminium ions of enals¹¹ and enones¹² as chiral platforms have been studied (Scheme 1A, right). Despite these advances, enantioselective organocatalytic Giese-type reactions are still underdeveloped. At the onset of these studies, to the best of our knowledge, versatile carboxylic acid derivatives had not been reported as prochiral templates in enantioselective radical conjugate additions.¹³

The use of chiral tertiary amine derived α,β -unsaturated acyl ammonium intermediates has proved an increasingly popular



Scheme 1 Previous works (A) and present work (B).

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strategy over the past decade.¹⁴ Various ionic Michael addition processes have been developed with a wide array of structural diversity accessible but radical derived approaches have remaining underexplored. Based upon our experience,^{14,15} the use of chiral isothiourea derived α,β -unsaturated acyl ammonium intermediates for the stereoselective addition of α -amino radicals was considered. In this process, the proposed conjugate radical addition would yield interesting γ -lactams through a [3+2] radical-polar crossover mechanism that would allow catalyst turnover by N-acylation that should be amenable in both batch and flow (Scheme 1B). Enantioselective routes to γ -lactams are an important and appealing synthetic goal,¹⁶ with the targeted γ -lactam motif a widespread heterocyclic scaffold found in an array of natural products, pharmaceuticals and agrochemicals.¹⁷

Initial exploration used the reaction of mixed α,β -unsaturated anhydride **1a**, accessible in a high yielding single step from the corresponding acid, and *N*-phenylglycine **2a** as the radical precursor (Table 1 and ESI†). Using (2*R*,3*S*)-hyperBTM isothiourea catalyst (**3a**) in the presence of TBAB (tetrabutylammonium bromide) as an additive and $[\text{Ru}(\text{bpy})_3](\text{PF}_6)_2$, under 23W CFL irradiation, the corresponding lactam (**4a**) was isolated in good yield and enantioselectivity (entry 1). Other isothiourea catalysts were tested (see ESI†), but product enantioselectivity was significantly decreased (entry 2 and ESI†). The use of TBAB as an additive was found to be decisive, with substitution by a base leading to diminished yield and enantioselectivity of the desired lactam (entry 3). Various photocatalysts were also tested (see ESI†), with for example *fac*-Ir(*ppy*)₃ giving **4a** in substantially lower yield and enantioselectivity (entry 4). The use of only acetonitrile as solvent, or as a mixture with other aromatic solvents or TBME, was detrimental to reaction efficiency (entries 5–7 and ESI†). Control experiments ensured that in absence of

light or photocatalyst the reaction did not take place (entries 8 and 9).

With the optimized reaction in hand (Table 1, entry 1), we next investigated the scope of the enantioselective radical conjugate addition to α,β -unsaturated anhydrides using amino acids as the α -amino radical source (Table 2). Firstly, the scope of the α,β -unsaturated component was evaluated. Aliphatic α,β -unsaturated anhydrides were very well tolerated, leading to the desired γ -lactams in good yields and enantioselectivities (**4a**–**4d**). Notably, β -phenyl substitution in the anhydride showed a similar performance in the catalytic system (**4e**). Various β -aryl substituted α,β -unsaturated systems were then tested. While electron donating *para*-methoxy (**4f**) and *meta*-bromo (**4h**) substituents were well tolerated, an electron deficient *para*-CF₃ substituent delivered the lactam with diminished enantioselectivity (**4g**). A β -2-thiophene substituent led to the desired cyclised product in slightly lower yield and enantioselectivity (**4i**). A series of *N*-aryl glycines were then subjected to the optimised reaction conditions. While *para*-F (**4j**) and *para*-Cl (**4k**) substituents showed moderate yields and good enantioselectivities, *para*-methyl displayed enhanced efficiency, yielding lactam **4l** in good yield with a slight erosion in the enantiomeric ratio. The use of a sterically hindered disubstituted *N*-aryl glycine gave rise to the desired product in high yield and low enantioselectivity (**4m**).

Given the encouraging results achieved and the synthetic potential of the developed method, we decided to perform the enantioselective reaction under flow conditions (Table 3, see ESI†). The synthetic method proved compatible using a flow-photoreactor (Table 3, see ESI†), and applying similar reaction conditions when compared with the batch conditions (Table 2), **4a** was obtained in a significantly higher yield, although diminished enantioselectivity. The robustness of the flow system was proved as high yields of the γ -lactams were obtained in all cases in a shortened reaction time (1 h), which provides a reaction throughput up to 30 times higher than the batch conditions. Surprisingly, a β -*para*-CF₃ substituent delivered the lactam with improved enantioselectivity to batch (**4g**). Considering other potential α -amino radical precursors,¹⁸ the applicability of this isothiourea catalysed system was expanded to radicals generated from α -silyl secondary amines (Scheme 2). In this system, an acid additive proved essential to minimise competitive 1,2-amine addition. Using the hydrochloride salt of the corresponding isothiourea HyperBTM organocatalyst under slightly modified reaction conditions (see ESI† for more details), γ -lactam **4a** was obtained in similar yield with slightly enhanced enantioselectivity. β -Alkyl substitution was well tolerated, giving the corresponding γ -lactams with good yields and enantioselectivities (**4b** and **4c**, **4n** and **4o**). β -Aryl substituted anhydrides showed increased competitive 1,2-addition leading to reduced product yield and enantioselectivity (**4f** and **4g**). *N*-aryl substitution within the amine component was tolerated, with the *ortho*-CH₃ α -silyl amine giving γ -lactam **4p** in good yield and enantioselectivity. However, *ortho*-phenyl substitution gave negligible reactivity under the optimised conditions and a solvent mixture with a greater proportion of MeCN was necessary to deliver **4q** with good yield but reduced

Table 1 Optimization of the isothiourea catalysed radical conjugate addition to α,β -unsaturated anhydrides^a

Entry	Deviation from optimized conditions	Yields ^b (%)	e.r. ^c	Reaction conditions	
				CH ₃ CN:Toluene (1:1)	23W CFL (white light)
1	No deviation	65(59)	81:19		
2	(<i>S</i>)-BTM (3b) catalyst instead of (2 <i>R</i> ,3 <i>S</i>)-HyperBTM (3a)	24	55:45		
3	K ₂ HPO ₄	40	75:25		
4	<i>fac</i> -Ir(<i>ppy</i>) ₃ instead of Ru(<i>bby</i>) ₃ (PF ₆) ₂	30	70:30		
5	CH ₃ CN instead of CH ₃ CN:PhCH ₃	70	70:30		
6	CH ₃ CN:PhCF ₃ instead of CH ₃ CN:PhCH ₃	59	81:19		
7	CH ₃ CN:TBME instead of CH ₃ CN:PhCH ₃	35	81:19		
8	No light	n.r.	—		
9	No Ru(<i>bby</i>) ₃ (PF ₆) ₂	n.r.	—		

^a Reaction conditions: **1a** (0.15 mmol) and **2a** (0.18 mmol), TBAB (0.15 mmol), Ru(*bby*)₃(PF₆)₂ (2 mol%), (2*R*,3*S*)-hyperBTM (20 mol%), CH₃CN:PhCH₃ (1:1, 0.05 M), r.t, under N₂, 23W CFL, 20 °C, 4 h. ^b ¹H-NMR yields using 1,3,5-trimethoxybenzene as internal standard. Isolated yield in brackets. ^c Enantiomeric excess was measured by Supercritical Fluid Chromatography (SFC) using chiral columns.

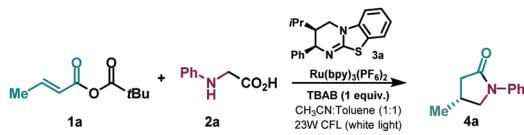
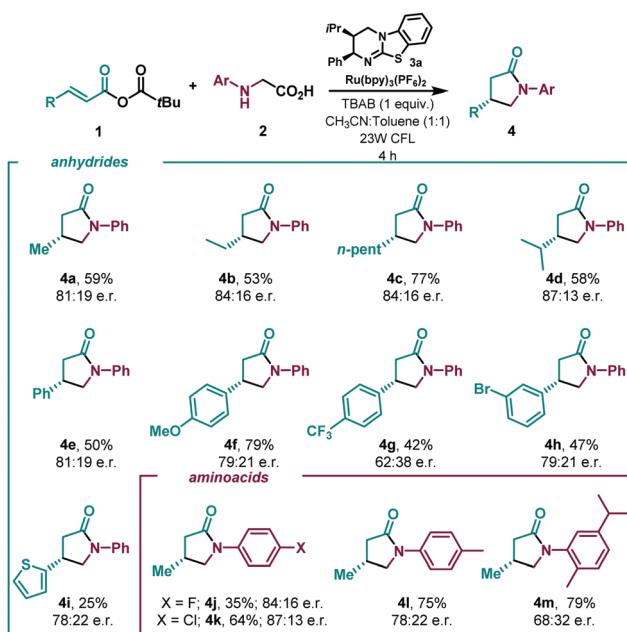
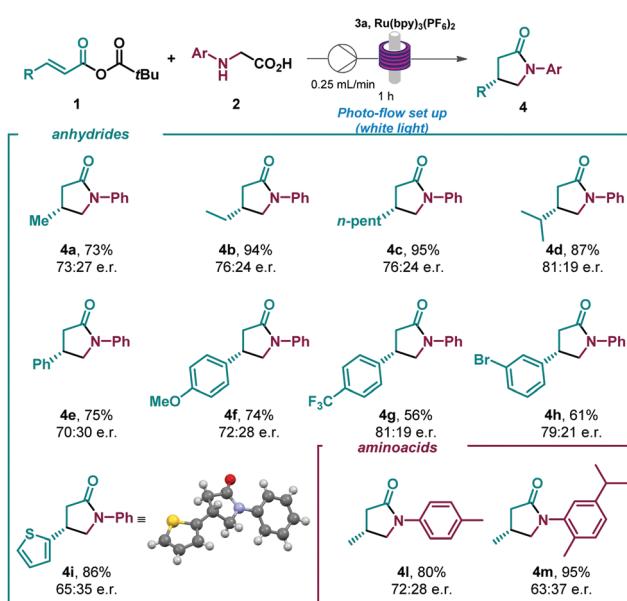


Table 2 Scope of the conjugate addition of α -amino radicals generated from α -amino acids to α,β -unsaturated anhydrides catalysed by (2R,3S)-HyperBTM^a



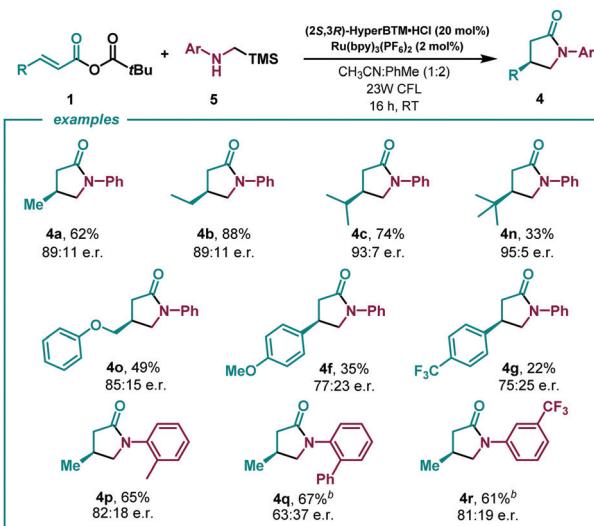
^a All reactions performed using 2 (0.15 mmol scale). Er determined by SFC analysis on a chiral stationary phase.

Table 3 Scope of the conjugate addition of α -amino radicals generated from aminoacids to α,β -unsaturated anhydrides under flow conditions^a



^a All reactions performed using 2 (0.15 mmol scale) in presence of K_2HPO_4 (0.075 mol). Er determined by SFC analysis on a chiral stationary phase.

enantioselectivity. These conditions gave the *meta*-CF₃ substituted lactam **4r** in good yield and enantioselectivity.

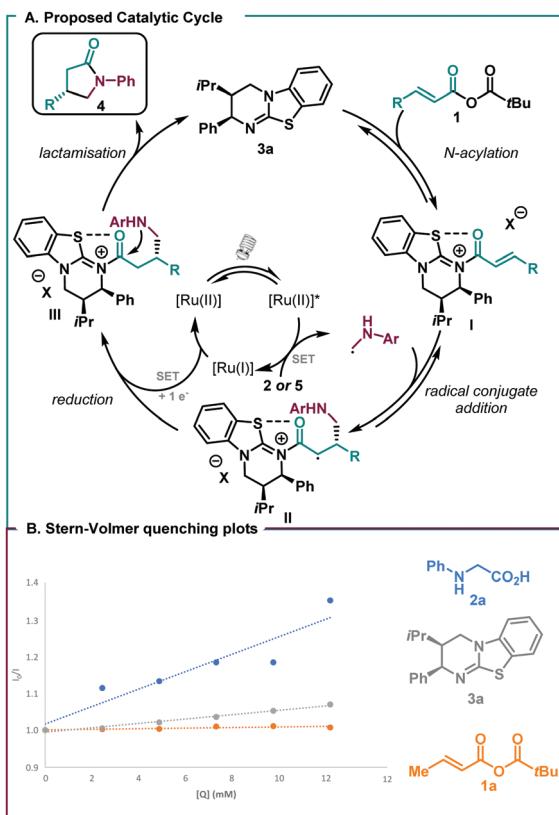


The absolute configuration of **4i** (Table 3) was unequivocally assigned as (*R*) by X-ray crystallographic analysis¹⁹ and further corroborated by the specific rotation of (*R*)-**4a**⁸ (see ESI†). The same stereochemical outcome was assigned by analogy for all the compounds included in Tables 2 and 3 and Scheme 2.

A mechanism is proposed for the transformation (Scheme 3A). N-acylation of the mixed anhydride **1** by the isothiourea leads to the α,β -unsaturated acyl ammonium ion pair **I**, adopting an *s-cis* conformation with a *syn*-coplanar 1,5 S···O chalcogen bonding interaction providing a conformational lock.²⁰ Meanwhile, the excited state photocatalyst oxidises a molecule of the α -amino radical precursor. Fluorescence quenching experiments of the excited photocatalyst prove that hypothesis. The data (Scheme 3B) confirm the lack of interaction between the Ru(bpy)₃(PF₆)₂ and the anhydride, and indicate slight potential oxidation of the organocatalyst.

This presents the possibility of the α -amino radical precursor quenching the excited state of Ru(bpy)₃(PF₆)₂, suggesting that the initial step of the photocatalytic cycle would be the oxidation of the α -amino acid or α -trimethylsilyl amine by the excited state photocatalyst. Subsequent removal of the electrofugal substituent (CO₂ or TMS) generates the α -amino radical that subsequently undergoes enantioselective radical conjugate addition into the α,β -unsaturated acyl ammonium intermediate, with facial selectivity dictated by the stereodiscriminating phenyl substituent. The reduced state photocatalyst (Ru(*i*)) then reduces the α -carbonyl radical **II** to the corresponding C(1)-ammonium enolate, regenerating the ground state photocatalyst, which upon protonation leads to the acyl ammonium **III**. Intramolecular N-acylation gives the corresponding γ -lactam **4** with concomitant release of the isothiourea catalyst.

In summary, we have described the combination of an isothiourea organocatalyst and a photocatalyst for the development of an enantioselective [3+2] radical-polar crossover process



Scheme 3 Proposed mechanism of the [3+2] radical-polar crossover reaction to generate γ -lactams.

to generate γ -lactams under batch and flow conditions. The developed system relies on the conjugate addition of α -amino radicals to transient isothiourea derived α,β -unsaturated acyl ammonium intermediates formed *in situ*.²¹

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Conflicts of interest

There are no conflicts to declare.

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