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Mild ketyl radical generation and coupling with alkynes enabled by Cr catalysis: stereoselective access to *E*-exocyclic allyl alcohols†

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Introduction

Coupling reactions with transient radicals are one of the most powerful tools in the formation of C-C and C-heteroatom bonds in modern organic chemistry.1 In this context, carboncentered ketyl radicals or ketyl radical anions are valuable coupling intermediates, along with cross-coupling that enables the incorporation of important carbonyl groups to construct synthetically useful aliphatic alcohols.2 However, because of the high reduction potential of carbonyl motifs,3 especially for reducing aliphatic carbonyls, stoichiometrically strong reductants, such as Na, K and Ti, are usually required for the generation of the ketyl radicals via the single electron transfer (SET) process.4 Alternatively, the thermodynamic barrier for ketyl formation can be overcome by the use of stoichiometric Kagan's reagent (SmI₂), 2b,5 as well as by combining strong reductants of Zn/Hg in the catalytic formation of the ketyl radicals. Despite the progress in developing photoredox catalysis⁷ and reductive electrocatalysis,8 mild strategies using metal catalysis for the reduction of carbonyls to the ketyl radicals for couplings are

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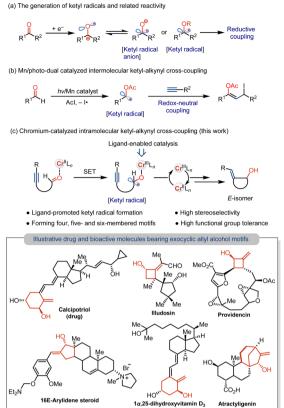
underdeveloped, particularly those that use earth-abundant 3d metal catalysts. In addition, to overcoming high reduction potential, chemoselectivity in metal-catalyzed reactions of the carbonyls without reducing resulting ketyl-coupled unsaturated adducts remains an issue. As a result, current studies focus on ketyl couplings with olefins to produce saturated products (Scheme 1a). Approaches that enable the formation of unsaturated motifs by catalytic formation of the ketyl radicals and selective couplings without over-reduction or over-addition would be value-addition to functionalized alcohol derivatives.

To address the selectivity problem, an impressive redoxneutral intermolecular coupling of the ketyl radicals with the alkynes has been reported by mechanisms involving atomtransfer radical addition (Scheme 1b).71 We anticipate that an intramolecular coupling of aliphatic ketyl radicals generated in situ with the alkynes could be a possibility using appropriate 3d metal catalysts. This would need designing a mild reductive strategy to achieve the chemoselective formation of cyclic allyl alcohols and retain the vinyl group. In addition to overcoming the high reduction potential of aliphatic carbonyls, combining the vinyl radical with metals and regeneration of reactive catalysts remain unresolved issues. Herein, we report that a low-cost chromium(II) catalyst supported by a bipyridine ligand enables the mild formation of the ketyl radicals from unactivated aliphatic aldehydes, allowing for the intramolecular coupling with the alkynes under the mild conditions (Scheme 1c).10 This Cr-catalyzed ketyl coupling occurs by carbonyl reduction with Cr

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Scheme 1 Catalytic formation of the ketyl radicals and couplings with the alkynes

catalyst and keeps the resulting vinyl functional group intact, achieving the stereo- and chemoselective formation of the Eexocyclic allyl alcohols that are ubiquitous structural motifs found in many pharmaceuticals and bioactive molecules.11

Results and discussion

At the outset, we commenced our studies by the exploration of the formation of the ketyl radicals and intramolecular coupling by using alkynyl aliphatic aldehyde 1a as a precursor (Fig. 1). Commercially available salt of CrCl₃ was chosen as a precatalyst in combination with 4,4'-di-tert-butyl-2,2'-dipyridyl (dtbpy) as the ligand, Mn as the reductant, and Cp₂ZrCl₂ with LiCl as the additive. We were pleased that the ketyl coupling occurs smoothly at ambient temperature, giving exocyclic allyl alcohol compound 2a in 67% yield and excellent E/Z-selectivity (>99:1). The replacement of CrCl₃ by CrCl₂ remarkably increased the conversion to afford allylic alcohol 2a in 87% yield. The reaction using Cr(acac)₃ led to the formation of 2a in low yield. Nevertheless, a wide range of metal catalysts, such as FeCl₂, CoCl₂, NiCl₂, CuCl₂, AuCl₃, AgCl, VCl₃, MoCl₃, WCl₅ and RuCl₃, almost showed no efficiency in promoting the transformation under present conditions. However, relatively low stereoselectivity was obtained in these reactions. Further studies suggested that ligands play important roles in promoting the reactivity of Cr and stereochemistry of the conversion. In the absence of the

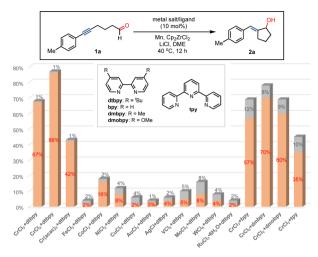


Fig. 1 Studying the effect of metal salts and ligands on ketyl coupling to produce E-exocyclic allyl alcohol.

dtbpy ligand, the ketyl radical coupling with the alkyne could almost not proceed with Cr catalysis. The use of bipyridine (bpy) as a ligand resulted in a relatively low conversion and E/Zselectivity, as well as the replacement by ligands, such as 4,4'dimethyl-2,2'-dipyridyl (dmbpy), 4,4'-dimethoxy-2,2'-dipyridyl (dmobpy) and terpyridine (tpy). In the absence of Mn or Cp₂-ZrCl₂, the ketyl coupling reaction does not occur in almost complete recovery of 1a, indicating their important roles in promoting the transformation. The reaction by the replacement of Mn by Zn or Mg produces product 2a in low yields (see ESI†), whereas the conversion is completely inhibited using Al as a reductant.

Having optimal reaction conditions in hand, we examined the scope of the radical coupling using diverse substituted alkynyl aliphatic aldehydes. As shown in Scheme 2, the treatment of 5-hexynals that contain electron-donating substituents, such as tert-butyl, methoxy, cyclopropylmethoxy, benzyloxy, phenoxy and tert-butyldimethylsilyloxy on aryls by dtbpy-ligated $Cr(\Pi)$ catalysis enables the formation of annulated (E)-2-benzylidenecyclopentanol motifs 2c-h in 81-91% yields and 94:6-99:1 stereoselectivities. The incorporation of electronwithdrawing groups of phenyl and naphthyl into the alkynyl does not affect the coupling reaction and provides access to (E)-(biphenyl-4-ylmethylene)cyclopentanol 2i and (E)-(naphthylmethylene)cyclopentanols (2j and 2k) in good yields and high stereoselectivities. It was noted that the steric hindrance caused by the mesityl substituent around the alkynyl does not impede annulated coupling and produces the sterically congested compound 21. Due to the mild conditions with Cr(II) catalyst and manganese reductant, halogen substituents, such as fluoride, polyfluoride, chloride, bromide, and iodide can be effectively compatible with the coupling system, providing a valuable strategy in the preparation of various halide-bearing 2-methylenecyclopentanols (2m-2s). It was noted that the thio group on the aryl of 5-hexynal does not hinder the catalysis, offering a strategy in the synthesis of product 2t. Using this methodology, 2-methylenecyclopentanols containing a range of

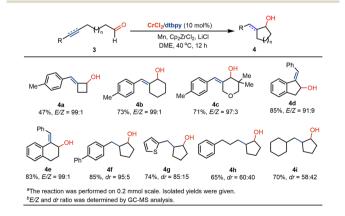
^aThe reaction was performed on 0.2 mmol scale. Isolated yields were given. ^bE/Z ratio was determined by GC-MS analysis. ^c20% yield of **2b** was detected

Scheme 2 Cr(II)-catalyzed ketyl radical coupling reactions with alkynyl aliphatic aldehydes.^{a,b}

synthetically useful groups, such as cyano, alkoxycarbonyl, trifluoromethyl, trifluoromethoxy, trimethylsilyl, boronate ester, alkenyl and alkynyl groups can be facilely accessible (2u-2ac). In addition, the reaction can be employed in the synthesis of methylenecyclopentanol derivatives that contain a range of heterocycle functional groups, such as benzofuranyl, dibenzo [b,d]furanyl, benzo[d][1,3]dioxoly, 2,2-difluorobenzo[d][1,3]dioxolyl, 9H-carbazolyl, 1H-indazolyl, 1H-pyrazolyl and thiophenyl groups (2ad-2al).

Inspired by these results, we examined whether the radical coupling could be applied in the construction of four- and sixmembered ring-containing cyclitols. Using 4-pentynal as a precursor, we were pleased that the four-member ring-based exocyclic butanol **4a** was obtained in moderate yields and high *E*-selectivity (Scheme 3). The treatment of 6-heptynal with Cr catalysis delivers six-membered (benzylidene)cyclohexanol in 73% yield and 99:1 *E/Z* selectivity (**4b**). The use of oxygen moiety as a linker in the radical coupling of alkynyl alcohol does not affect the conversion, offering a strategy in the formation of exocyclic tetrahydro-2*H*-pyranol **4c**. Moreover, benzylidenyl-based 2,3-dihydro-1*H*-indenol and 1,2,3,4-tetrahydronaphthol

motifs **4d** and **4e** can be prepared at ambient temperatures. In addition to the coupling with the alkyne, we wondered whether the ketyl radical coupling reaction with olefin could occur with Cr catalysis. By the use of alkenyl-substituted aliphatic



Scheme 3 Application in the formation of cyclobutanols, cyclopentanols and cyclohexanol derivatives. a,b

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aldehydes as precursors, we found that corresponding saturated five-membered cyclopentanols 4f-4i were formed in moderateto-good diastereoselectivity.

To gain insight into the reaction pathway, mechanistic experiments were performed. By the addition of 2,2,6,6tetramethylpiperidine-1-oxyl (TEMPO) as a radical scavenger into the reaction system, we found that the coupling was completely inhibited without the production of related compound 2a (Fig. 2a). The analysis of the residue by the high-resolution mass spectrum (HRMS) technique suggested that the TEMPO-trapped compound 5a or 5b might be formed. Interestingly, the replacement of TEMPO by 1,1-diphenylethylene in the reaction led to the formation of dialkynyl-containing diol 6 in 20% yield, combined with a small amount of adduct 7 by the addition of aldehyde to olefin. These results indicate that the formation of the ketyl radicals could be considered in the reaction. The experiments of cyclic voltammetry (CV) were carried out to study the activation of carbonyl by Cr catalysis. A high reduction potential for 5-hexynal **1a** ($E_p = -2.46 \text{ V}$) was observed (Fig. 2b). The coordination with the dtbpy ligand leads to the promotion of the reduction potential of $CrCl_2$ ($CrCl_2$ /dtbpy: $E_p = -1.70$ V). Interestingly, the combination of the CrCl₂/dtbpy complex with 1a leads to a great decrease in the reduction potential of 1a

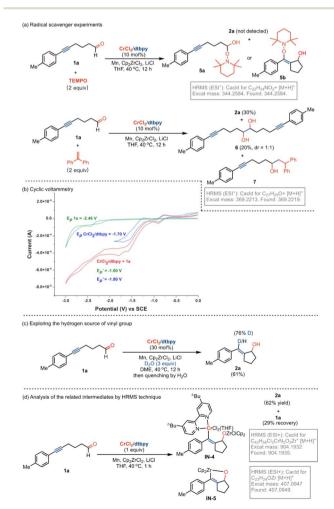
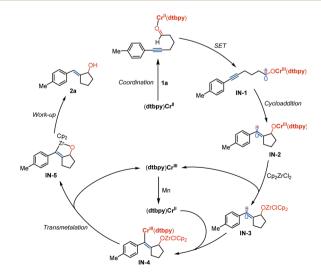


Fig. 2 Mechanistic studies.

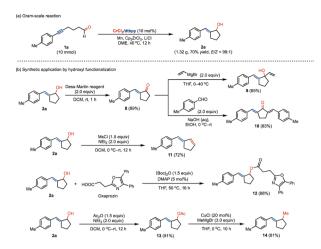
 $(E_{\rm p}^{'}=-1.80~{\rm V})$ with the increased potential of the complex $(E_{\rm p}^{\dagger} = -1.80 \text{ V})$. It indicates that the coordination of CrCl₂/dtbpy with 1a might promote the reduction process in giving the related ketyl radical. Subsequently, we explored the source of hydrogen in the vinyl scaffold of 2a. By the addition of 3.0 equivalent D₂O in the reaction, we found that the coupling occurs smoothly to afford the compound 2a in 61% yield, with the incorporation of around 76% of deuterium into the vinyl group (Fig. 2c). The related metalated vinyl intermediate may be formed during the process. We explored related intermediates in the stoichiometric reaction by HRMS analysis (Fig. 2d). Both the vinylated chromate species IN-4 that contained the Cp2ZrCl moiety and oxazirconiumacycle IN-5 were detected, indicating that the ketyl radical coupling with the alkynyl, followed by combination with dtbpy-Cr(II), might be considered, which may undergo a transmetalation process by a further reaction with Cp₂ZrCl₂ in forming oxazirconiumacycle.

Based on the preliminary results and studies of Cr/Zrcatalyzed organic transformations, 10e-k we hypothesized that the reaction is initiated by a single electron transfer process between dtbpy-Cr complex and 1a, resulting in the formation of ketyl radical IN-1 (Scheme 4). It may deliver the related vinyl radical IN-2 by the process of annulation and transmetalation with Cp₂ZrCl₂ to produce IN-3 and the (dtbpy)Cr(III) complex. The reactive Cr(II) catalyst can be regenerated by reduction with manganese and combination with vinyl radical by transmetalation between (dtbpy)Cr(III) and Zr species to produce oxazirconiumacycle IN-5. Exocyclic E-allyl alcohol compound 2a is formed by following the work-up process.

The Cr-catalyzed ketyl radical coupling reaction can be conducted on a gram-scale without loss of efficiency, giving exocyclic E-allyl alcohol products with high stereoselectivity (Scheme 5a). We investigated the late-stage functionalization of the resulting exocyclic allyl alcohol in the preparation of valuable motifs. By oxidation using the Dess-Martin reagent, α,βunsaturated ketone 8 can be easily accessed, which reacts with vinyl Grignard reagent to afford vinylcyclopentanol 9 in good



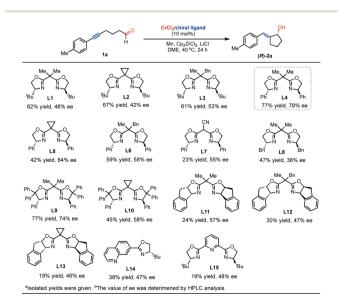
Scheme 4 Presumed reaction pathway.



Scheme 5 Gram-scale reaction and late-stage functionalization.

yields (Scheme 5b). The reaction of 8 with benzaldehyde led to the formation of functionalized bis(benzylidene)cyclopentanone motif 10. By the elimination of the hydroxyl group, the related 1,3-diene compound 11 could be obtained in 72% yield. The hydroxyl scaffold can be facilely functionalized in accessing the pharmaceutical compound of oxaprozin (12) by esterification. In addition, the transformation of the hydroxyl group by sequential esterification and methylation allowed us to prepare the methylated derivative 14.

Chiral exocyclic pentanols are synthetically intriguing motifs that serve as core structures and are widely found in numerous biologically active molecules and pharmaceuticals. ¹¹ As a result, the development of asymmetric catalysis in the stereoselective construction of these motifs is of significant interest. ¹² To probe the possibility of developing the Cr-catalyzed asymmetrical ketyl radical coupling, we studied the effect of chiral ligands on the transformation of enantioselective formation of chiral exocyclic allyl alcohols. As shown in Scheme 6, the use of chiral *tert*-butyl-



 $\begin{array}{lll} \textbf{Scheme 6} & \textbf{Exploration of Cr-catalyzed asymmetric ketyl radical coupling reaction.} \\ \end{array}$

substituted bis(oxazoline) ligand of L1, we found that the ketyl radical coupling occurs smoothly, leading to the formation of the product (R)-2a in 62% yield and 48% value of enantiomeric excess (ee). The replacement of dimethyl substituents by cyclopropanyl or methylbenzyl in the linker does not strongly promote enantioselectivity with Cr catalysis (L2 and L3). The reaction used phenyl-substituent bis(oxazoline) L4 to form the product (R)-2a with a good yield and ee value. Further studies suggested that the change of substituents of the linker and oxazolinyl groups did not greatly improve the transformation (L5-L13). While the incorporation of quinolinyl or pyridinyl as coordination groups into oxazoline (L14 and L15) led to relatively low conversions and enantioselectivity.

Conclusions

In summary, we have developed a chromium-catalyzed ligandenabled ketyl radical formation in coupling with the alkynes. This reaction was initiated by a bipyridine-supported Cr(II) catalyst for the generation of the ketyl radicals for alkyne hydrofunctionalization, offering a stereo- and chemoselective strategy in the synthesis of valuable E-exocyclic allyl alcohols. Because of mild catalytic conditions, a range of synthetically useful functionalities, including iodide, bromide, chloride, fluoride, cyano, amino, alkoxycarbonyl, alkynyl, and olefin groups, can be compatible with reaction systems. Compared with previous reports using strong reductants, this reaction employs (dtbpy)Cr(II) catalyst as a mild reagent to reduce aldehydes to the ketyl radicals, by the annulation and transmetalation with Cp₂ZrCl motif in giving oxazirconiumacycle species. Formed exocyclic allyl alcohol derivatives can be modified by hydroxyl functionalization in the construction of various synthetically useful molecules. Further improvement is underway in the enantioselectivity for the development of asymmetric ketyl radical coupling reaction by Cr catalysis.

Data availability

Experimental details and characterization of the products are available in the ESI.†

Author contributions

X. Zeng conceived and directed the overall research project. Z. Luo, X. Zhang, and Z. Li designed and conducted the experimental studies. Z. Luo, M. Luo, and X. Zeng analyzed the data and contributed to the preparation of the manuscript. X. Zeng wrote the manuscript.

Conflicts of interest

There are no conflicts to declare.

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