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9, 9547An NADH-selective and sensitive fluorescence  
probe to evaluate living cell hypoxic stress†Mingzhe Li,<sup>‡a</sup> Chang Liu,<sup>‡b</sup> Wenjuan Zhang,<sup>b</sup> Longfei Xu,<sup>c</sup> Miaomiao Yang,<sup>a</sup>  
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Cellular disease and senescence are often accompanied by an imbalance in the local oxygen supply. Under hypoxia, mitochondrial NADH and FADH<sub>2</sub> cannot be oxidized by the mitochondrial electron transport chain, which leads to the accumulation of reducing equivalents and subsequent reduction stress. Detecting changes in intracellular NADH levels is expected to allow an assessment of stress. We synthesized a red fluorescent probe, DPMQL1, with high selectivity and sensitivity for detecting NADH in living cells. The probe **DPMQL1** has strong anti-interference abilities toward various potential biological interferences, such as metal ions, anions, redox species, and other biomolecules. In addition, its detection limit can reach the nanomolar level, meaning it can display small changes in NADH levels in living cells, so as to realize the evaluation of cell-based hypoxic stress.

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## Introduction

The cellular redox state is the central regulator of energy production and intermediary metabolism, playing a crucial role in health and disease.<sup>1</sup> The nicotinamide adenine dinucleotide (NAD) derivatives NADH and NADPH (general term: NAD(P)H) play an important role in cellular metabolism and energy systems as electron carriers.<sup>2</sup> NADH is generated by the tricarboxylic acid (TCA) cycle, with three NADH molecules produced per TCA cycle. In recent years, studies have shown that NADH also plays an important role in gene expression,<sup>3</sup> immune function,<sup>4</sup> cellular senescence<sup>5</sup> and oxidative stress injury,<sup>6</sup> *etc.* Therefore, it is important to evaluate the concentration of NADH for an understanding of overall cellular energy metabolism.

We pay special attention to the indicative role of NADH changes in cellular redox stress (oxidative stress and reductive stress). Although the mechanism of hypoxia producing stress in living cells is still controversial, a large number of studies have found that hypoxia can induce redox stress in living cells. Studies have confirmed that an important part of cell damage

caused by chemical hypoxia-induced reductive stress seems to be the formation of ROS.<sup>7</sup> Park *et al.* found that hypoxic perfusion of rat and rabbit hearts can cause significant increases in lipid peroxidation products, protein carbonyls and glutathione disulfide, which appear within 10 minutes of exposure, and continue to rise after 90 minutes of hypoxia.<sup>8</sup> In addition, there are similar findings in various cell preparations using ROS-sensitive fluorescent probes.<sup>9</sup> Accumulating evidence shows that hypoxic stress increases the levels of cellular NADH in most mammalian cells,<sup>7,10</sup> and excessive NAD(P)H is considered to be a sign of reducing stress.<sup>11,12</sup> In addition, there is also evidence that the increase of ROS accumulation in mitochondria is positively correlated with an increase of the NADH/NAD<sup>+</sup> ratio.<sup>7,13</sup> In hypoxia, mitochondrial NADH and FADH<sub>2</sub> cannot be oxidized by the mitochondrial electron transport chain (ETC), which leads to the accumulation of reducing equivalents and subsequent reduction stress.<sup>7,13,14</sup> Reduction stress causes the single electron reduction of oxygen to form O<sub>2</sub><sup>•-</sup>, which leads to an increase in the production of reactive oxygen species under hypoxia.<sup>7,13,14</sup> Oldham *et al.* found that, compared with normoxic cells (21% O<sub>2</sub>), primary human lung fibroblasts cultured under hypoxia (0.2% O<sub>2</sub>) had increased NADH levels and produced significantly higher levels of mitochondrial ROS.<sup>13</sup> In bovine coronary artery smooth muscle cells, hypoxia increased the levels of mitochondrial ROS and NAD(P)H, and at the same time increased the NADH/NAD<sup>+</sup> ratio.<sup>15</sup> Therefore, it is expected to realize the evaluation of cell stress by detecting the change of NADH level. However, NAD(P)H is a highly reducible dynamic molecule with a high reactivity and instability, and it is difficult to detect in cells and

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animal models. Therefore, it is necessary to establish an ultra-sensitive, rapid, and selective method for measuring NAD(P)H *in vivo* to study reducing stress.<sup>12</sup>

Due to the vital importance of NAD(P)H, several analytic methods are currently available for the determination of NADH, including the enzyme-based method,<sup>16</sup> chromatography,<sup>17</sup> capillary electrophoresis,<sup>18</sup> and the electrochemical method,<sup>19</sup> and even for the indirect determination of various analytes, like glucose, ethanol and lactic acid, owing to its oxido-reductive properties.<sup>20</sup> Furthermore, the intrinsic fluorescence of NADH has been used in the quantification and imaging of biological samples;<sup>20</sup> however, its short excitation/emission wavelength can be easily interfered with by other biomolecules, such as proteins. Thus, colorimetric and fluorescent chemical indicators *via* trapping the change of optical output induced by the target analyte are widely used. To date, a number of indicators, including gold nanoparticles,<sup>21</sup> quantum dot (QD) conjugates,<sup>22</sup> and genetically encoded fluorescent sensors,<sup>23</sup> have been applied for the determination of NADH. However, to the best of our knowledge, only several small molecular fluorescence probes for NADH have been reported by several research groups, such as the teams of Pan, Wang, Komatsu, *etc.*, based on the reaction of 1-methyl-quinoline that responds to the increased NADH by enhancement of the fluorescence intensity;<sup>12,24,25</sup> Podder *et al.* reported a naphthimide-based dual-channel fluorescent probe (NQN) for locating NADH levels in living cells through multicolor images.<sup>26,27</sup> Moreover, among these molecular probes, those containing dicyanoisophorone/TCF hemicyanine probes (Scheme 1a) exhibited some unique superiorities as the NIR fluorescent chromophore. Thus, the development of a small molecule fluorescent probe for the

detection and imaging of NADH in living cells is an interesting topic in the field of chemical biology and it is also worth trying to realize the assessment of hypoxic stress by detecting NADH in living cells.

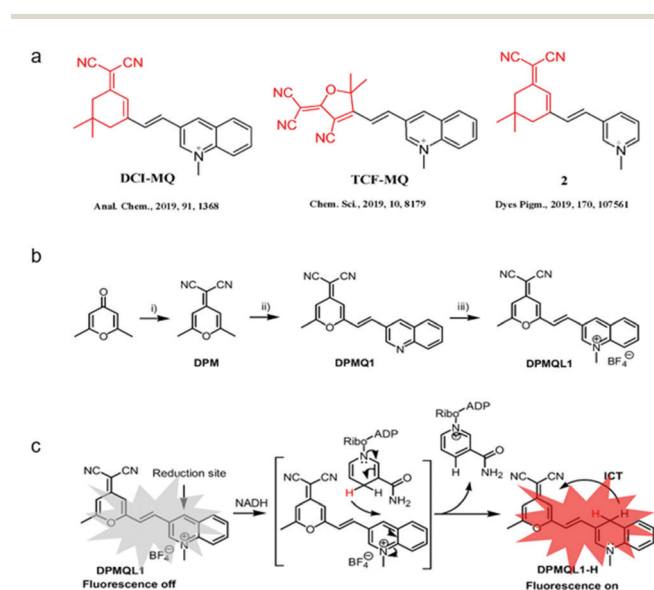
## Results and discussion

### Construction of a NADH fluorescent probe with high selectivity and sensitivity

The synthetic routes for the probe **DPMQL1** are shown in Scheme 1b. The compound **DPM** (2,6-(dimethyl-4*H*-pyran-4-ylidene) malononitrile) was prepared according to the published procedures.<sup>28</sup> **DPMQ1** was synthesized by the condensation reaction of **DPM** and quinolone-3-carboxaldehyde in the presence of a catalytic amount of piperidine in CH<sub>3</sub>CN. Finally, the probe **DPMQL1** was obtained by the *N*-methylation reaction of **DPMQ1** and CH<sub>3</sub>I in dry acetonitrile. The chemical structures of **DPMQ1** and **DPMQL1** were confirmed by <sup>1</sup>H, <sup>13</sup>C NMR, and HRMS spectroscopy (ESI,† Fig. S1–S7).

Initially, the electron absorption and fluorescence emission spectra of **DPMQL1** and those in the presence of NADH were carried out. Due to the electron-deficient property of the fluorophore, the free probe **DPMQL1** only showed a very weak  $\pi$ - $\pi$  transition absorption band at *ca.* 525 nm ( $\epsilon = 5600 \text{ M}^{-1} \text{ cm}^{-1}$ ) and a weak fluorescence emission band with the maxima at *ca.* 600 nm, respectively (Fig. S8, ESI†). With the addition of NADH (10 equiv.) to the solution of **DPMQL1**, the maxima absorption band at 528 nm ( $\epsilon = 24\,100 \text{ M}^{-1} \text{ cm}^{-1}$ ) increased significantly. At the same time, a visual color change from light pink to bright pink was observed upon the addition of NADH (Fig. S9, ESI†), allowing for the naked-eye detection of NADH. In the presence of NADH, **DPMQL1** displayed a gradually enhanced red emission as the incubation time was prolonged at an excitation of 510 nm (Fig. S10 and S11, ESI†). The relative fluorescence intensity reached a plateau after 60 min. The probe **DPMQL1** showed a prominent enhanced emission band centered at 624 nm (Fig. S8b, ESI†). The peak-to-peak Stokes shift is over 100 nm. The large Stokes shift is favorable for bioapplications. The results of the prominent fluorescence change upon interaction with NADH suggested that **DPMQL1** could function as a fluorescence turn-on chemosensor for the detection of NADH.

Subsequently, the sensitivity of the probe **DPMQL1** was studied by fluorescence spectra. As shown in Fig. 1, **DPMQL1** emitted a weak fluorescence emission band at 624 nm. The fluorescence intensity at 624 nm gradually increased with increasing concentrations of NADH (0–55  $\mu\text{M}$ ) (Fig. S12, ESI†). The fluorescence turn-on constant *K* derived from the titrations was found to be  $1.52 \pm 0.74 \mu\text{M}$  ( $R^2 = 0.995$ ) (Fig. S13, ESI†). The detection limit (LOD) of **DPMQL1** ( $3\sigma/K$ ) was estimated to be 0.36 nM (Fig. S14, ESI†). This is particularly beneficial for applications in cell-based assays as intracellular NADH concentrations typically lie in the millimole range.<sup>29</sup> The results indicated that **DPMQL1** could serve as a highly sensitive red fluorescence probe for the detection of NADH. The proposed optical responses of the probe **DPMQL1** towards NADH are shown in Scheme 1c (Table S1 and Fig. S15, ESI†).



**Scheme 1** (a) Structures of hemicyanine probes containing dicyanoisophorone/TCF. (b) The synthesis of **DPMQL1**; reagents and conditions: (i) CH<sub>2</sub>(CN)<sub>2</sub>, (CH<sub>3</sub>CO)<sub>2</sub>O, 130 °C, reflux, 4 h; (ii) quinolone-3-carboxaldehyde, acetonitrile, 40 °C, overnight; (iii) CH<sub>3</sub>I, 50 °C, overnight. (c) The structure of the probe **DPMQL1** and the reaction of the probe **DPMQL1** with NADH.



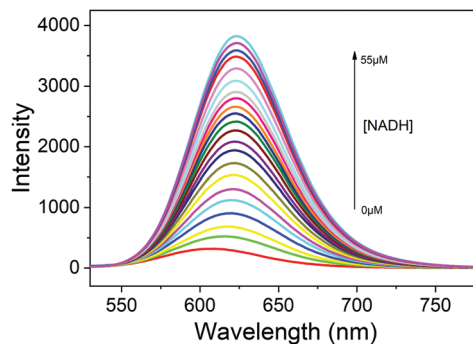


Fig. 1 Fluorescence spectra of **DPMQL1** ( $5 \mu\text{M}$ ) with different concentrations of NADH ( $0$ – $55 \mu\text{M}$ ) in PBS–EtOH (PBS,  $10 \text{ mM}$ ,  $\text{pH} = 7.4$ ,  $1:1$ ,  $\text{v/v}$ ).  $\lambda_{\text{ex}}$ :  $510 \text{ nm}$ , slit width:  $5 \text{ nm}$ .

The feature of selectivity over potentially competing species is very important for the application of a probe. Therefore, we chose commonly found metal ions in the living cell, certain reactive nitrogen species, ROS, thiols, and various amino acids as references to investigate the NADH selectivity of the probe **DPMQL1** (Fig. S9, ESI<sup>†</sup>). The addition of possible disturbing relevant species, including  $\text{Zn}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Fe}^{3+}$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Na}^+$ ,  $\text{H}_2\text{O}_2$ , HOCl, NADH, NAD(P)H, NO,  $\text{NO}_2^-$ ,  $\text{O}_2^{\bullet-}$ ,  $\text{ONOO}^-$ , TBHP, Arg, Cys, Glu, GSH, Hcy, His, Leu, Lys, Met, Phe, Pro, Ser, Thr, Try, Tyr, Val,  $\text{H}_2\text{S}$ , did not induce any major change in the fluorescence intensity at  $624 \text{ nm}$ . However, as shown in Fig. 2, the presence of GSH, Cys, and Hcy can cause a slight increase in the fluorescence signal at  $624 \text{ nm}$ , which may be due to the strong reduction of thiol groups. Notably,  $\text{SO}_3^{2-}$  had a considerably higher fluorescence intensity change than that of other thiols, because 1-methyl-quinoline can reaction with  $\text{SO}_3^{2-}$  as the recognition unit.<sup>30</sup> But it is still negligible compared with that of NADH because of its low *in vivo* concentration.<sup>31,32</sup> In contrast, **DPMQL1** showed an obvious increase in fluorescence at  $624 \text{ nm}$  with the addition of NADH. The results of the fluorescence data confirmed that **DPMQL1** had a high selectivity with a specific response to NADH, not markedly influenced by biologically relevant species.

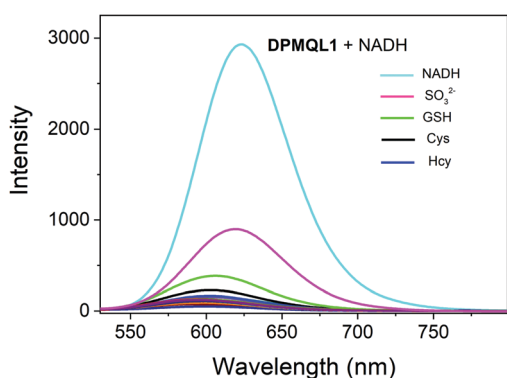


Fig. 2 Fluorescence response of **DPMQL1** ( $5 \mu\text{M}$ ) to various biological species ( $50 \mu\text{M}$ ) in PBS–EtOH (PBS,  $10 \text{ mM}$ ,  $\text{pH} = 7.4$ ,  $1:1$ ,  $\text{v/v}$ ).  $\lambda_{\text{ex}}$ :  $510 \text{ nm}$ , slit width:  $5 \text{ nm}$ .

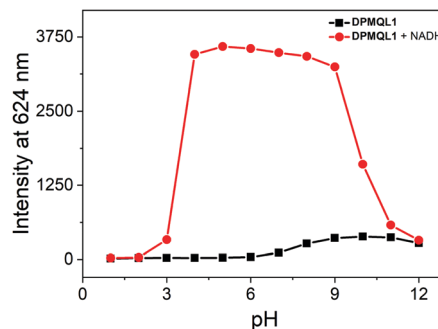


Fig. 3 The pH dependence of **DPMQL1** ( $5 \mu\text{M}$ ) at  $624 \text{ nm}$  in the absence and presence of NADH ( $50 \mu\text{M}$ ) in PBS–EtOH (PBS,  $10 \text{ mM}$ ,  $1:1$ ,  $\text{v/v}$ ) solution.  $\lambda_{\text{ex}}$ :  $510 \text{ nm}$ , slit width:  $5 \text{ nm}$ .

To further establish the workability of **DPMQL1** in competitive medium, anti-interference studies were also carried out by adding  $50 \mu\text{M}$  of other related species to the solution of **DPMQL1**. After the addition of  $50 \mu\text{M}$  NADH, other analyte species added showed negligible interference with the detection of NADH (Fig. S16, ESI<sup>†</sup>), which clearly authenticated that the probe **DPMQL1** can selectively determine NADH in complex samples. The pH experimental results showed that the probe displayed a distinct increase in the fluorescence signal at  $624 \text{ nm}$  in the pH range of  $4$ – $9$  in the presence of NADH (Fig. 3). The results indicated that **DPMQL1** could be applied for the detection of NADH changes in complex biological environments.

To evaluate the applicability of **DPMQL1** for the analysis of intracellular NADH, we further explored the detection ability of **DPMQL1** in living cells by fluorescence imaging. At first, the cellular toxicity of **DPMQL1** was evaluated by MTT and CCK-8 assays. Two kinds of cells (HeLa and PC12 cells) were incubated with **DPMQL1** ( $0$ – $50 \mu\text{M}$ ) for  $24 \text{ h}$ . As shown in Fig. 4, more than  $80\%$  of the HeLa cells still survived when  $10 \mu\text{M}$  **DPMQL1** was incubated for  $24 \text{ h}$ . When the concentration of the probe reached  $50 \mu\text{M}$ , the cell viability dropped to about  $60\%$ , but it was still better than that of the positive control. The results showed that the probe **DPMQL1** could be used for living cell imaging with concentrations lower than  $10 \mu\text{M}$ .

Then, we investigated the fluorescence imaging ability of **DPMQL1** to detect NADH in living cells. As shown in Fig. 5a1, HeLa cells incubated with **DPMQL1** ( $0.2 \mu\text{M}$ ) for  $30 \text{ min}$  at  $37^\circ\text{C}$  showed a weak fluorescence signal in the red fluorescence channel ( $582$ – $682 \text{ nm}$ ). However, after the administration of NADH ( $2 \mu\text{M}$ ) and then incubation for another  $30 \text{ min}$ , the fluorescence intensity in the red channel increased obviously (Fig. 5b1). Furthermore, after the HeLa cells were incubated with glucose ( $20 \text{ mM}$ ) to stimulate the increased levels of endogenous NADH for  $30 \text{ min}$ , the fluorescence intensity in the red channel increased markedly (Fig. 5c1). As can be seen from Fig. 5b, compared with the untreated HeLa cells, there are over 4-fold fluorescence enhancements of those being treated with exogenous NADH or the glucose stimulated one. These results indicated that **DPMQL1** was confirmed to be competent for imaging NADH in living cells in a fluorescence turn-on manner.



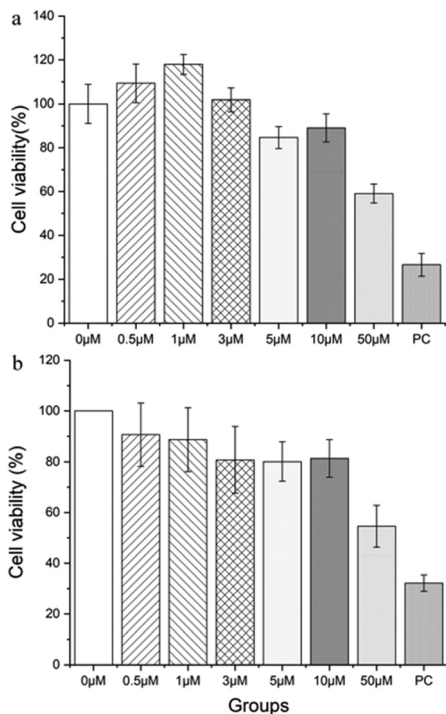


Fig. 4 (a) MTT assay results from HeLa cells incubated with different concentrations of the probe **DPMQL1** for 24 h. (b) CCK-8 assay results from PC12 cells incubated with different concentrations of the probe **DPMQL1** for 24 h. PC (positive control): 50 μM H<sub>2</sub>O<sub>2</sub>.

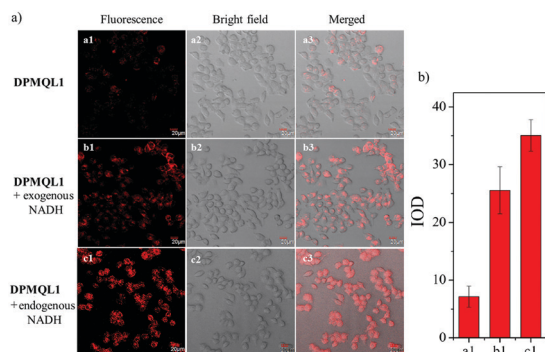


Fig. 5 (a) Confocal fluorescence images of HeLa cells incubated with the probe **DPMQL1** (0.2 μM). Cell images were obtained using an excitation wavelength of 559 nm and a band-path (582–682 nm) emission filter. (a1–a3) Fluorescence, bright field, and merged images after 30 min of incubation with **DPMQL1**; (b1–b3) fluorescence, bright field, and merged images after incubation with **DPMQL1** for 30 min followed by incubation with exogenous NADH (2 μM) for 30 min; (c1–c3) fluorescence, bright field, and merged images after incubation with glucose (20 mM) for 30 min followed by incubation with **DPMQL1** for 30 min. (b) The quantification of the fluorescence intensities collected from imaging data in (a).

To verify the ability of **DPMQL1** to detect changes in NADH levels in the mitochondrial respiratory chain, we constructed a cell with mitochondrial gene deletion<sup>33</sup> ( $\rho^0$  PC12 cell, Fig. 6a and 7) and further explored the difference in fluorescence imaging monitoring capabilities of **DPMQL1** in PC12

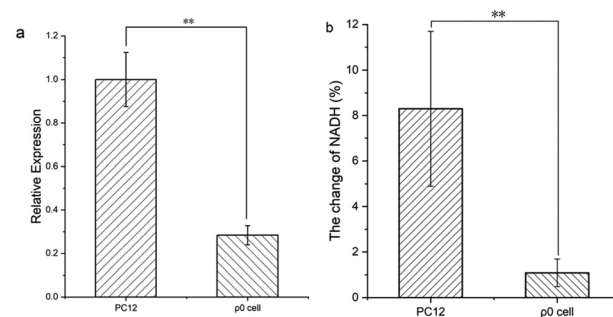


Fig. 6 (a) To confirm the lack of mtDNA, RT-PCR was used to detect the mitochondrial single-copy gene cyclooxygenase 2 (COX2) to represent the mtDNA copy number. Amplifying the COX2 gene, the primer size is 156 bp, the forward primer sequence is 5'-TGGCTTACAAGACGCCACAT-3', and the reverse primer sequence is 5'-TGGCGTCTATTGTGCTTGT-3', while the nuclear coding gene  $\beta$ -actin is used as the internal reference, the primer size is 262 bp, the forward primer sequence is 5'-TTACTGCCCTGGCTCCTAG-3', and the reverse primer sequence is 5'-CGTACTCCTGCTTGCTGATC-3'. (b) Changes in the NADH levels of wild-type PC12 cells and  $\rho^0$  cells after 30 minutes of incubation with glucose (20 mM). \*:  $p < 0.05$ , \*\*:  $p < 0.01$ .

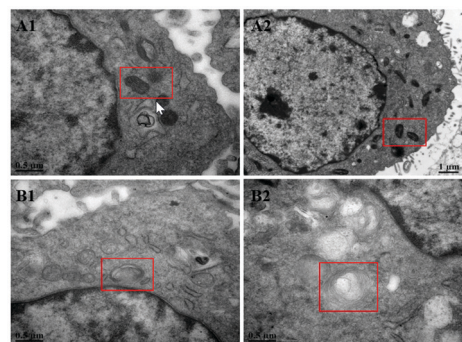


Fig. 7 An ultra-structural investigation of wild-type and  $\rho^0$  cell mitochondria via TEM. Electron micrographs of ultra-thin sections are shown. (A): The mitochondria of wild-type cells, showing an interconnected network structure with numerous regularly arranged cristae; (B):  $\rho^0$  cell mitochondria, demonstrating single vesicular organelles with distorted cristae.

wild-type cells and  $\rho^0$  cells. As shown in Fig. 6b, wild-type PC12 cells and  $\rho^0$  cells were incubated with glucose (20 mM) for 30 minutes, and then the endogenous NADH level increased, but the endogenous NADH level of the  $\rho^0$  cells changed less than that of the PC12 cells. It can be seen that the fluorescence imaging effect of **DPMQL1** is related to the function of mitochondria, and **DPMQL1** can detect the level of NADH on the mitochondrial respiratory chain.

#### Application of the probe: evaluating the levels of hypoxic stress at the cellular level

We used **DPMQL1** to detect the changes in cellular NADH levels after hypoxic stress, and then preliminarily evaluated the degree of cell stress. Three types of cells (PC12, H9C2, Scc-25) were incubated in a hypoxic incubator (37 °C, 5.0% CO<sub>2</sub>, 1.0% O<sub>2</sub>) for 8 hours, 12 hours, and 24 hours, respectively, and then



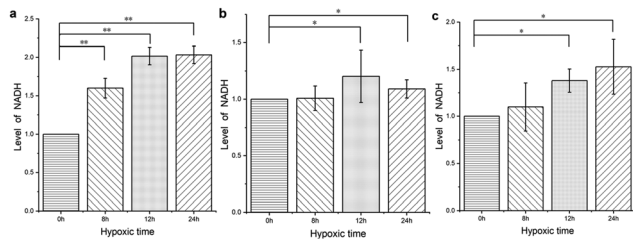


Fig. 8 DPMQL1 is used to detect changes in the NADH levels in three kinds of cells after hypoxic stress: (a). PC12 cells; (b). H9C2 cells; and (c). Scc-25 cells. \*:  $p < 0.05$ , \*\*:  $p < 0.01$ .

incubated with DPMQL1 (0.2  $\mu\text{M}$ ) at 37  $^{\circ}\text{C}$  for 30 minutes. As shown in Fig. 8a, compared with the untreated PC12 cells, the cellular NADH levels increased significantly with a prolonged period of hypoxia. As can be seen from Fig. 8b and c, short-term hypoxia intervention has no obvious effect on the NADH levels in the H9C2 and Scc-25 cells, but after prolonging the hypoxia period, the cellular NADH levels increase significantly. These results indicate that DPMQL1 can evaluate hypoxic stress by detecting changes of cellular NADH levels.

## Conclusions

In summary, a red emissive fluorescent probe DPMQL1 comprising two electron-withdrawing groups, namely a malononitrile moiety and a chinolinium moiety, for the detection of NADH was prepared. The probe displayed weak fluorescence before the reaction with NADH. Based on changes in the internal charge transfer (ICT) process triggered specifically by NADH, the probe showed a strong fluorescence turn-on response. Compared with most NADH fluorescent probes of the same type, DPMQL1 has higher sensitivity (LOD = 0.36 nM) and it can accurately detect small changes in NADH levels caused by reduction stress. DPMQL1 has high selectivity and effects from most interferent (including NADPH) that may exist in organisms are eliminated. Moreover, the probe is membrane-permeable and can detect fluctuations in NADH levels in the mitochondria of living cells. Due to hypoxia, mitochondrial NADH and other reduced equivalents cannot be oxidized by ETC, which ultimately leads to an increase in the production of reactive oxygen species. Therefore, we speculate that the level of NADH in mitochondria can reflect the degree of cellular hypoxic stress. We used the probe **DPMQL1** to evaluate the hypoxic stress in three kinds of cells, and the results showed that with an extension of the hypoxia time, the intracellular NADH level showed an upward trend, that is, cell hypoxic stress can be detected *via* detecting the intracellular NADH level. We will continue to explore the application of such probes in order to better realize the evaluation of the pathophysiological states of organisms.

## Author contributions

Mingzhe Li: Data curation, methodology, validation and roles/writing – original draft; Chang Liu: data curation, methodology and roles/writing – original draft; Wenjuan Zhang: data curation,

validation and methodology; Longfei Xu: formal analysis and software; Miaomiao Yang: formal analysis and software; Zhaoli Chen: investigation, methodology and supervision; Xinxing Wang: visualization and writing – review & editing; Lingling Pu: conceptualization, data curation and formal analysis; Weili Liu: supervision and writing – review & editing; Xianshun Zeng: funding acquisition, project administration, resources and supervision; Tianhui Wang: funding acquisition, project administration, resources, supervision and writing – review & editing.

## Conflicts of interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Notes and references

- C. Cantó, K. J. Menzies and J. Auwerx, *Cell Metab.*, 2015, **22**, 31–53.
- J. H. Joo, D. Youn, S. Y. Park, D.-S. Shin and M. H. Lee, *Dyes Pigm.*, 2019, **170**, 107561.
- F. Malavasi, S. Deaglio, A. Funaro, E. Ferrero, A. L. Horenstein, E. Ortolan, T. Vaisitti and S. Aydin, *Physiol. Rev.*, 2008, **88**, 841–886.
- L. Alberghina and D. Gaglio, *Cell Death Dis.*, 2014, **5**, e1561.
- H. Teng, M. Lv, L. Liu, X. Zhang, Y. Zhao, Z. Wu and H. Xu, *Sensors*, 2017, **17**(4), 788.
- S. Li, C. Y. Lo, M. H. Pan, C. S. Lai and C. T. Ho, *Food Funct.*, 2013, **4**, 10–18.
- T. L. Clanton, *J. Appl. Physiol.*, 1985, **2007**(102), 2379–2388.
- Y. Park and J. P. Kehrer, *Free Radical Res. Commun.*, 1991, **14**, 179–185.
- T. Vanden Hoek, L. B. Becker, Z. H. Shao, C. Q. Li and P. T. Schumacker, *Circ. Res.*, 2000, **86**, 541–548.
- W. Xiao, R. S. Wang, D. E. Handy and J. Loscalzo, *Antioxid. Redox Signaling*, 2018, **28**, 251–272.
- A. C. Brewer, S. B. Mustafi, T. V. Murray, N. S. Rajasekaran and I. J. Benjamin, *Antioxid. Redox Signaling*, 2013, **18**, 1114–1127.
- X. Pan, Y. Zhao, T. Cheng, A. Zheng, A. Ge, L. Zang, K. Xu and B. Tang, *Chem. Sci.*, 2019, **10**, 8179–8186.
- W. M. Oldham, C. B. Clish, Y. Yang and J. Loscalzo, *Cell Metab.*, 2015, **22**, 291–303.
- M. Nakane, *Intensive Care Med.*, 2020, **8**, 95.
- Q. Gao and M. S. Wolin, *Am. J. Physiol.*, 2008, **295**, H978–H989.
- H. Gao, J. Li, D. Sivakumar, T. S. Kim, S. K. S. Patel, V. C. Kalia, I. W. Kim, Y. W. Zhang and J. K. Lee, *Int. J. Biol. Macromol.*, 2019, **123**, 629–636.



- 17 W. Xie, A. Xu and E. S. Yeung, *Anal. Chem.*, 2009, **81**, 1280–1284.
- 18 J. Liang, W. Wei, H. Yao, K. Shi and H. Liu, *Phys. Chem. Chem. Phys.*, 2019, **21**, 24572–24583.
- 19 M. Thiruppathi, P. Y. Lin, Y. T. Chou, H. Y. Ho, L. C. Wu and J. A. Ho, *Talanta*, 2019, **200**, 450–457.
- 20 A. Podder, S. Koo, J. Lee, S. Mun, S. Khatun, H. G. Kang, S. Bhuniya and J. S. Kim, *Chem. Commun.*, 2019, **55**, 537–540.
- 21 Z. Wang, Q. Chen, Y. Zhong, X. Yu, Y. Wu and F. Fu, *Anal. Chem.*, 2020, **92**, 1534–1540.
- 22 D. S. Bilan and V. V. Belousov, *Free Radical Biol. Med.*, 2016, **100**, 32–42.
- 23 Y. Zhao, K. Wei, F. Kong, X. Gao, K. Xu and B. Tang, *Anal. Chem.*, 2019, **91**, 1368–1374.
- 24 L. Wang, J. Zhang, B. Kim, J. Peng, S. N. Berry, Y. Ni, D. Su, J. Lee, L. Yuan and Y. T. Chang, *J. Am. Chem. Soc.*, 2016, **138**, 10394–10397.
- 25 H. Komatsu, Y. Shindo, K. Oka, J. P. Hill and K. Ariga, *Angew. Chem., Int. Ed.*, 2014, **53**, 3993–3995.
- 26 A. Podder, V. P. Murali, S. Deepika, A. Dhamija, S. Biswas, K. K. Maiti and S. Bhuniya, *Anal. Chem.*, 2020, **92**, 12356–12362.
- 27 A. Podder, N. Thirumalaivasan, Y. K. Chao, P. Kukutla, S.-P. Wu and S. Bhuniya, *Sens. Actuators, B*, 2020, **324**, 128637.
- 28 S. Cai, Y. Lu, S. He, F. Wei, L. Zhao and X. Zeng, *Chem. Commun.*, 2013, **49**, 822–824.
- 29 N. Karton-Lifshin, L. Albertazzi, M. Bendikov, P. S. Baran and D. Shabat, *J. Am. Chem. Soc.*, 2012, **134**, 20412–20420.
- 30 Z. Meng, R. Li and X. Zhang, *Inhalation Toxicol.*, 2005, **17**, 309–313.
- 31 D. Li, X. Tian, Z. Li, J. Zhang and X. Yang, *J. Agric. Food Chem.*, 2019, **67**, 3062–3067.
- 32 H. Li, *Anal. Chim. Acta*, 2015, **897**, 102–108.
- 33 D. J. Scarlett, P. Herst, A. Tan, C. Prata and M. Berridge, *BioFactors*, 2004, **20**, 199–206.

