

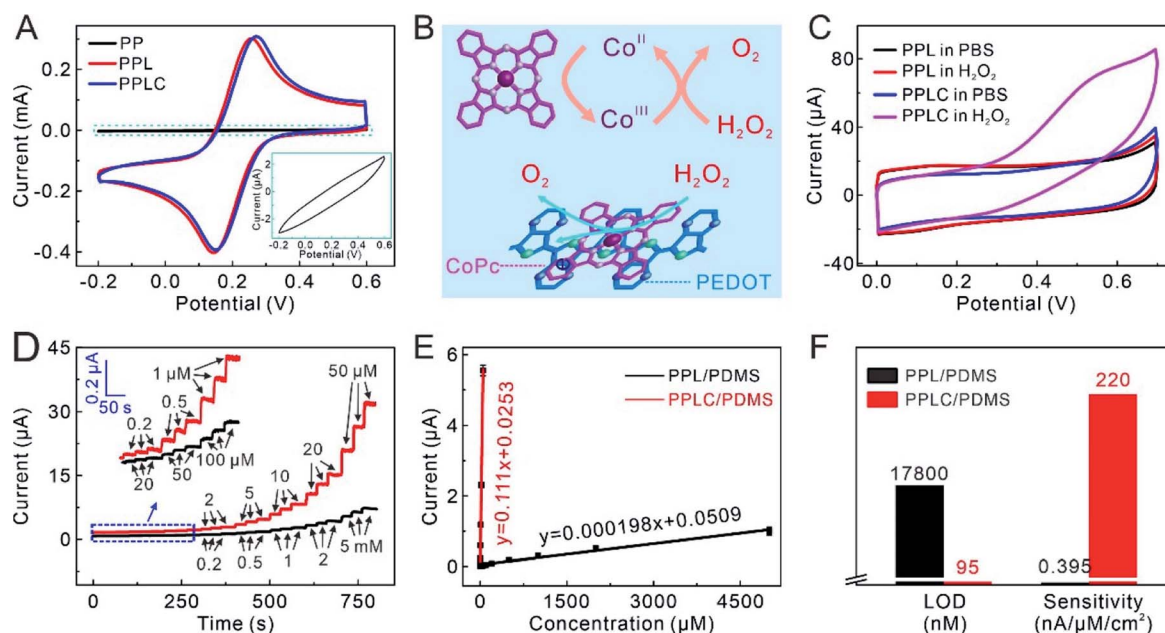
## CORRECTION

[View Article Online](#)  
[View Journal](#) | [View Issue](#)Cite this: *Chem. Sci.*, 2021, 12, 15771**Correction: Plasticizer and catalyst co-functionalized PEDOT:PSS enables stretchable electrochemical sensing of living cells**Jing Yan,<sup>a</sup> Yu Qin,<sup>a</sup> Wen-Ting Fan,<sup>a</sup> Wen-Tao Wu,<sup>a</sup> Song-Wei Lv,<sup>b</sup> Li-Ping Yan,<sup>a</sup> Yan-Ling Liu<sup>\*a</sup> and Wei-Hua Huang<sup>a</sup>

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[rsc.li/chemical-science](https://rsc.li/chemical-science)Correction for 'Plasticizer and catalyst co-functionalized PEDOT:PSS enables stretchable electrochemical sensing of living cells' by Jing Yan *et al.*, *Chem. Sci.*, 2021, 12, 14432–14440, DOI: 10.1039/d1sc04138j.

The authors regret that there was an error in the equation of the calibration curve of PPL/PDMS in Fig. 3. The correct version is shown below.



**Fig. 3** (A) CVs of different electrodes obtained in 10 mM  $K_3[Fe(CN)_6]$ . Inset: the enlarged view for CV of the PP electrode. (B) Schematic illustration of the electrocatalysis mechanism. (C) CVs of different electrodes with and without 1 mM  $H_2O_2$ . (D) Amperometric responses of PPL/PDMS (black lines) and PPLC/PDMS (red lines) electrodes to  $H_2O_2$  at a potential of +0.55 V (vs. Ag/AgCl) to increasing  $H_2O_2$  concentrations. Inset: the enlargements of amperometric responses framed in blue. (E) Calibration curves of PPL/PDMS and PPLC/PDMS electrodes to increasing  $H_2O_2$  concentrations (data presented as mean  $\pm$  standard error,  $n = 3$ ). (F) Calculated LOD and sensitivity of PPL/PDMS and PPLC/PDMS electrodes to  $H_2O_2$ .

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

<sup>a</sup>College of Chemistry and Molecular Sciences, Wuhan University, Wuhan 430072, China. E-mail: [yanlingliu@whu.edu.cn](mailto:yanlingliu@whu.edu.cn); [whhuang@whu.edu.cn](mailto:whhuang@whu.edu.cn)

<sup>b</sup>School of Pharmacy, Changzhou University, Changzhou 213164, China