



Cite this: *Chem. Sci.*, 2021, **12**, 15632

All publication charges for this article have been paid for by the Royal Society of Chemistry

Received 10th September 2021  
 Accepted 3rd November 2021

DOI: 10.1039/d1sc05021d  
[rsc.li/chemical-science](http://rsc.li/chemical-science)

## Chiral, sequence-definable foldamer-derived macrocycles†

Toyah M. C. Warnock,<sup>1</sup> Sundaram Rajkumar,<sup>1</sup> Matthew P. Fitzpatrick,<sup>1</sup> Christopher J. Serpell,<sup>1</sup> Paul Dingwall<sup>1</sup> and Peter C. Knipe<sup>1\*</sup>

Nature's oligomeric macromolecules have been a long-standing source of inspiration for chemists producing foldamers. Natural systems are frequently conformationally stabilised by macrocyclisation, yet this approach has been rarely adopted in the field of foldamer chemistry. Here we present a new class of chiral cyclic trimers and tetramers formed by macrocyclisation of open-chain foldamer precursors. Symmetrical products are obtained *via* a [2 + 2] self-assembly approach, while full sequence control is demonstrated through linear synthesis and cyclisation of an unsymmetrical trimer. Structural characterisation is achieved through a combined X-ray and DFT approach, which indicates the tetramers adopt a near-planar conformation, while the trimers adopt a shallow bowl-like shape. Finally, a proof-of-concept experiment is conducted to demonstrate the macrocycles' capacity for cation binding.

## Introduction

Nature frequently exploits macrocycles as functional molecules within living systems,<sup>1–3</sup> and many of these and their derivatives have been exploited as therapeutics.<sup>4–6</sup> Some of the most important drugs for human health are naturally-occurring macrocycles, including vancomycin and cyclosporin. The relative success of macrocyclic peptide drugs can be attributed to several factors including stability to degradation, cyclic constraint favouring the active conformation, and high membrane permeability. Inspired by macrocycles' use in Nature and potency in the clinic, chemists have developed numerous cyclisation strategies to stabilise the conformation of small bioactive peptides, including helix "stapling"<sup>7</sup> and cross-linking to stabilise  $\beta$ -hairpin structures.<sup>8</sup> Peptides have also been replaced entirely, with artificial folded molecules dubbed "foldamers".<sup>9</sup> Entirely abiotic macrocycles have also been used in applications outside biology, particularly in host-guest chemistry.<sup>10</sup> Here, the pre-organisation of functionality within the macrocycle enables large binding constants and high guest-specificity.

Foldamers have been developed to mimic peptide secondary<sup>11–14</sup> and tertiary structures,<sup>15–17</sup> and have shown promise as therapeutic agents in their own right.<sup>9</sup> However, the field of *foldamer-derived macrocycles* – analogous to macrocyclic

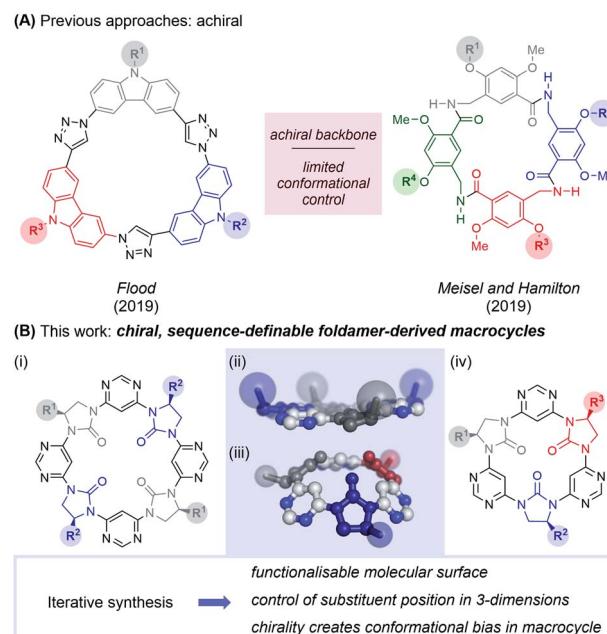


Fig. 1 Sequence-defined chiral macrocyclic oligomers. (A) Recent examples of sequence-defined macrocyclic foldamers reported by Flood, and Hamilton and Meisel. The Flood macrocycles adopt an entirely planar conformation, forming 2-dimensional crystalline arrays, while those of Meisel and Hamilton adopt enantiomeric cavitand-like conformations under rapid exchange. (B) This study describes the synthesis, structure and cation-binding properties of a novel chiral foldamer-derived macrocycle. (i) General structure of tetrameric macrocycles; (ii) X-ray structure (top) and computed energy minimum (bottom) of tetrameric and trimeric macrocycles respectively; (iii) general structure of trimeric macrocycles. Sidechains are highlighted as coloured spheres.

<sup>1</sup>School of Chemistry and Chemical Engineering, Queen's University Belfast, David Keir Building, Belfast, BT9 5AG, UK. E-mail: [p.knipe@qub.ac.uk](mailto:p.knipe@qub.ac.uk)

<sup>1</sup>Almac Group Ltd., 20 Seagoe Industrial Estate, Craigavon, BT63 5QD, UK

<sup>1</sup>School of Physical Sciences, University of Kent, Ingram Building, Canterbury, Kent, CT2 7NH, UK

† Electronic supplementary information (ESI) available. CCDC 2057482, 2057483, 2057484 and 2057486. For ESI and crystallographic data in CIF or other electronic format see DOI: [10.1039/d1sc05021d](https://doi.org/10.1039/d1sc05021d)



peptides – remains in its infancy. Extant foldamer-derived macrocycles typically display a large degree of symmetry,<sup>18–32</sup> probably due to synthetic convenience (a notable exception is in mixed peptide/foldamer<sup>33–37</sup> and peptoid<sup>38,39</sup> systems). However, Nature demonstrates that diversity (rather than uniformity) of structure is crucial in enabling the use of macrocycles across a range of functions. In the past year the first two examples of sequence-defined abiotic foldamer-derived macrocycles have emerged from the laboratories of Flood,<sup>40</sup> who reported cyclic carbazole-triazole trimers and Meisel and Hamilton<sup>41</sup> who synthesised cavitand-like molecular containers (Fig. 1). Here, we present our synthesis of semi and fully sequence-definable abiotic foldamer-derived macrocycles possessing a homochiral backbone, allowing the defined positioning of sidechains in three-dimensions.

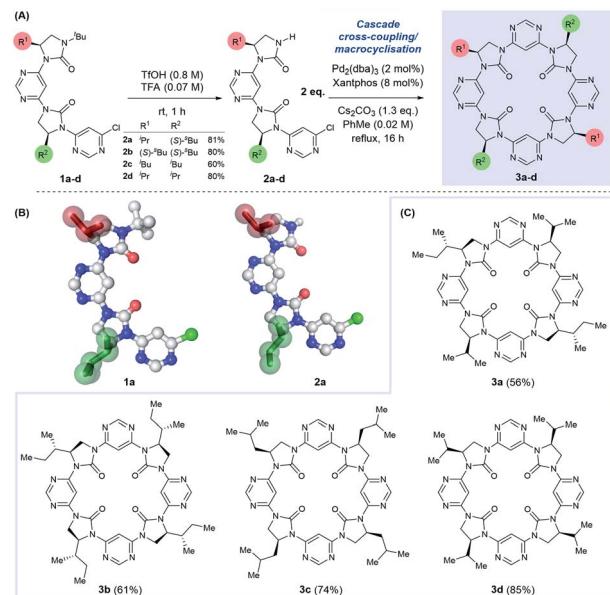
## Results and discussion

### Synthesis of macrocycles

We have previously described a foldamer architecture where dipolar repulsion between adjacent pyrimidine and imidazolidin-2-one components leads to structures with a turn per monomer of  $\sim 86^\circ$ .<sup>15</sup> The attempted synthesis of related pyridine-linked macrocycles by Meth-Cohn failed entirely, generating only oligomeric material,<sup>42</sup> but we hypothesised that the dipole-mediated pre-organisation in our putative linear precursor could favour macrocyclisation over oligomerisation. Similar preorganisation towards macrocyclisation has previously been exploited by Gong in the self-assembly of oligobenzamide foldamers.<sup>26,43</sup> We tested this hypothesis by first synthesising linear dimer **2a** by iterative deprotection/Buchwald–Hartwig coupling steps from a chiral, amino alcohol-derived monomeric precursor (see ESI† for procedures). Removal of the *N*-*tert*-butyl protecting group on **1a** under acidic conditions afforded dimer **2a** in 81% yield. Both the *N*-terminally protected and *tert*-butyl deprotected intermediates **1a** and **2a** were crystalline solids, with the latter purified conveniently by trituration with hot hexanes.

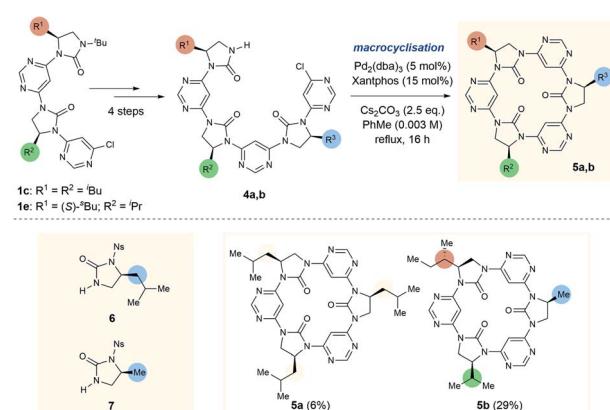
Examination of their single crystal X-ray structures revealed that the expected dipole-opposed conformation was adopted, giving both molecules an overall crescent shape (Scheme 1B). We were therefore confident that treatment of dimer **2a** under Buchwald–Hartwig cross-coupling conditions would lead directly to the *C*<sub>2</sub>-symmetrical macrocycle **3a**, since the initial tetrameric product of homo-coupling would be pre-organised *via* dipolar repulsion for a second coupling reaction to generate the macrocycle. Thus, treatment of **2a** with  $\text{Pd}_2(\text{dba})_3$ , Xantphos and  $\text{Cs}_2\text{CO}_3$  under reflux in toluene afforded macrocycle **3a** in 56% isolated yield, with no evidence of the formation of larger macrocycles or polymeric material. With this synthetic strategy validated we proceeded to synthesise a further three dimeric precursor molecules **2b–d**.<sup>44</sup> Upon treatment under cross-coupling conditions all were converted to the corresponding *C*<sub>4</sub>-symmetrical macrocycles **3b–d** in isolated yields of 61–85%.

To further test the limits of the macrocyclisation approach we synthesised linear trimers **4a** and **4b** (Scheme 2). While the cyclic tetramers were expected to be approximately planar and



**Scheme 1** (A) Synthesis of tetrameric macrocycles *via* *N*-*tert*-Bu deprotection of dimer **1a–d** to afford **2a–d**, and cascade dimerisation-cyclisation under Buchwald–Hartwig conditions. (B) Single crystal X-ray structures of **1a** and **2a**, sidechains are highlighted in green and red. (C) Synthetic scope of tetrameric macrocycle formation.

free from torsional strain, we anticipated that the trimers would be more strained, with the backbone forced out of the ideal planar conformation. To obtain the open-chain trimers it was necessary to add an additional pyrimidine and imidazolidin-2-one sub-unit to linear dimers **1c** and **1e**, which was achieved in four steps. In accordance with greater conformational strain, yields of the corresponding trimeric macrocycles were low



**Scheme 2** Synthesis of *C*<sub>3</sub>- (5a) and *C*<sub>1</sub>-symmetrical (5b) trimeric macrocycles. *Synthesis of 4a:* (1) **6** (1.3 eq.),  $\text{Pd}_2(\text{dba})_3$  (5 mol%), Xantphos (15 mol%),  $\text{K}_2\text{CO}_3$  (2.5 eq.), PhMe, reflux, 16 h, 73% brsm; (2) PhSH (1.5 eq.),  $\text{K}_2\text{CO}_3$  (3 eq.), *N,N*-DMF, rt, 3 h, 90%; (3) 4,6-dichloropyrimidine (9 eq.),  $\text{Pd}_2(\text{dba})_3$  (5 mol%), Xantphos (15 mol%),  $\text{K}_2\text{CO}_3$  (2.5 eq.), PhMe, reflux, 16 h, 62%; (4)  $\text{TfOH:TFA}$  4:1 *v:v*, rt, 1 h, >99%. *Synthesis of 4b:* (5) **7** (1.5 eq.),  $\text{Pd}_2(\text{dba})_3$  (5 mol%), Xantphos (15 mol%),  $\text{K}_2\text{CO}_3$  (2.5 eq.), reflux, 16 h, 95%; (6) as step 2, 52%; (7) as step 3, 90%; (8) as step 4, >99%. **5b** is an example of an entirely sequence-defined chiral foldamer-derived macrocycle.



relative to the tetrameric homologues, with **5a** and **5b** obtained in 6% and 29% yields respectively. Macrocycle **5b** is especially noteworthy as it is entirely “sequence-defined”, with the cyclic ordering of the monomers predetermined by the order of monomer addition in the preparation of the linear precursor.

### Conformational studies

In an effort to understand its conformational behavior, single crystals of macrocycle **3a** were grown by vapour diffusion (Fig. 2A). Attempts to generate diffraction-quality crystals of the remaining macrocycles were unsuccessful. As expected, all four sidechains of **3a** are projected from a single face of the macrocycle in a highly controlled manner. Inspection of the crystal packing reveals a layered, back-to-back stacking arrangement between planes of macrocycles (Fig. 2B–D), with the relative orientation of macrocycles between layers controlled in part by a dipole-opposed arrangement between adjacent imidazolidin-2-ones (Fig. 2C). Due to disorder and weak diffraction the X-ray data were insufficient to gain further structural insights, so we proceeded to explore the macrocycles' conformational behavior by DFT (Fig. 3).

The lowest energy conformation of the tetramer was found to be a shallow bowl, with all C=O groups puckered outwards from the same face of the macrocycle, and a transverse O–O distance of 5.2 Å. Two low energy conformers were identified for the trimer, corresponding to a macromolecular “ring-flip”, in

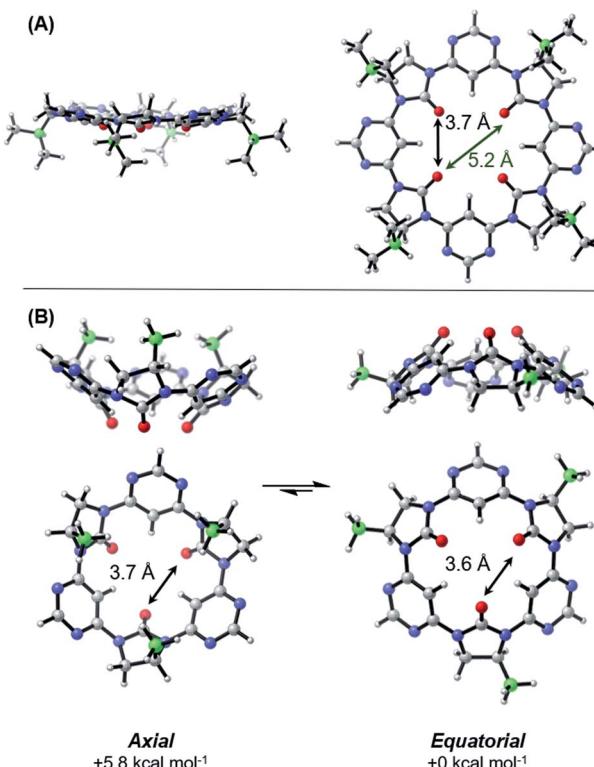


Fig. 3 Computed lowest energy conformations of tetrameric (A) and trimeric (B) macrocycles. Averaged O–O interatomic distances are indicated. Level of theory: B3LYP/6-311+G(d,p)/GD3BJ/PCM<sub>Toluene</sub>//B3LYP/6-31G(d,p)/GD3BJ. Isopropyl and methyl sidechains are shown for the tetramer and trimer respectively.

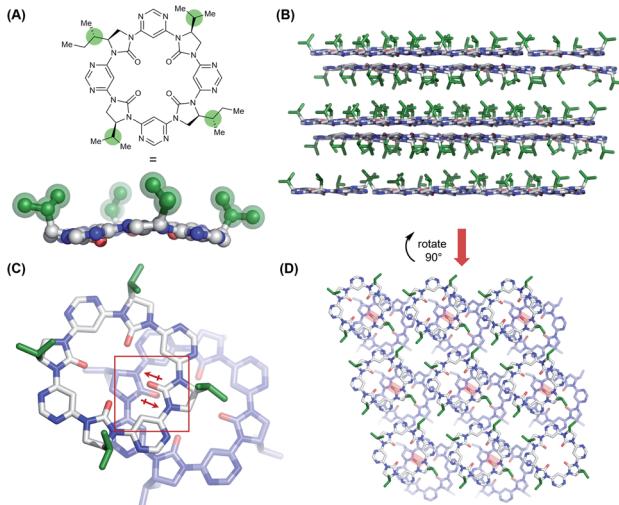


Fig. 2 Single crystal X-ray structure of  $C_2$ -symmetrical macrocycle **3a**. Sidechains are highlighted in green. Sidechains cannot be identified beyond the  $\gamma$ -position due to disorder, so all are truncated to isopropyl. Solvent molecules and hydrogen atoms are omitted for clarity. (A) Single molecule view, showing controlled projection of side-chains from a single face; (B) crystal packing arrangement (viewed along crystalline *b*-axis) showing planes of macrocycles arranged back-to-back, with side-chains projected into the interplanar void; (C) stacking arrangement between macrocycles in adjacent planes (top plane – white; bottom plane – purple) displaying a dipole-opposed orientation (dipoles indicated in red); (D) crystal packing arrangement (viewed along the crystalline *a*-axis) showing the offset arrangement between adjacent planes of macrocycles (top plane – white; bottom plane – purple). Dipole-opposed interactions are indicated by red boxes.

which the sidechain substituents are placed in *pseudo-axial* or *pseudo-equatorial* positions (Fig. 3B). The equatorial conformer is lower in energy by 5.8 kcal mol<sup>-1</sup> and represents the global energy minimum. No “mixed” conformers (in which one imidazolidin-2-one is puckered in an opposing direction to the others) are identified as minima. The less-planar structure of the trimer relative to the tetramer is also supported by <sup>1</sup>H NMR data: the inward-pointing pyrimidine hydrogens in tetramers **3a**–**3d** appear far downfield ( $\sim$ 9.9 ppm), likely due to the deshielding effect of the proximal oxygen lone-pairs, whereas the equivalent peaks in the trimers **5a** and **5b** appear at  $\sim$ 8.7 ppm indicating that these hydrogens are subject to the deshielding effect of the lone pairs to a much lesser degree.

**Guest binding.** The structural and computational data were indicative of the tetrameric macrocycles possessing a central pore with the four urea carbonyl groups directed towards its centre. We anticipated that cations would be well-stabilised within the macrocycle, as had already been observed with Cs<sup>+</sup> during the isolation of **3a**. We were particularly interested in the binding of ammonium cations since methylated lysines form part of the epigenetic histone code, and selective methods to probe them are therefore of importance.<sup>45–47</sup> The binding of macrocycle **3d** to hexadecyltrimethylammonium chloride was examined by <sup>1</sup>H NMR titration in CDCl<sub>3</sub> (Fig. 4).<sup>48</sup> Upon treatment with the ammonium salt, H<sup>B</sup> displays a small downfield shift while H<sup>A</sup> displays a larger upfield shift. The greater



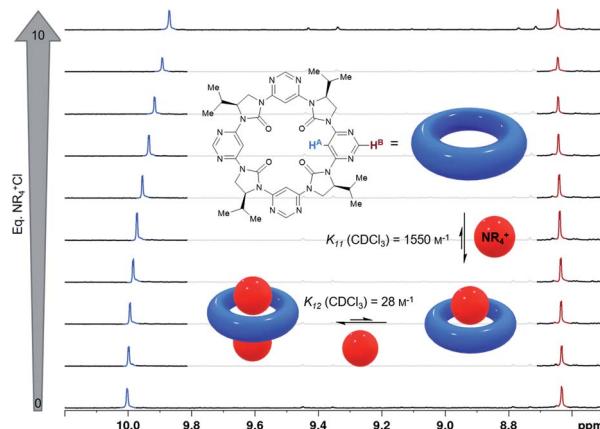


Fig. 4 Binding of hexadecyltrimethylammonium chloride (depicted as a red sphere) by  $C_4$ -symmetrical macrocycle **3d** (blue torus) determined by  $^1\text{H}$  NMR titration ( $\text{CDCl}_3$ , 600 MHz). Atoms  $\text{H}^{\text{A}}$  and  $\text{H}^{\text{B}}$  are highlighted and the corresponding  $^1\text{H}$  NMR signals are indicated in blue and red respectively. For full details refer to the ESI  $\dagger$ .

magnitude shift of  $\text{H}^{\text{A}}$  is consistent with our assumption that binding occurs to the inner edge of the macrocycle, in closer proximity to  $\text{H}^{\text{A}}$  than  $\text{H}^{\text{B}}$ , likely in the manner of “perching” complexes described by Cram, and mediated by  $\text{CH}\cdots\text{O}$  hydrogen bonding between the ammonium  $\alpha$ -C-H bonds and imidazolidin-2-one carbonyl groups.<sup>49–51</sup> The data are suggestive of 1:2 host:guest complex formation, with fitted equilibrium constants of  $K_{11} = 1550 \text{ M}^{-1}$  and  $K_{12} = 28 \text{ M}^{-1}$ . The ability to alter the size and side-chains of the macrocycle raises the prospect of tailoring the selectivity of this cation binding to the desired guest in future studies.

Lastly, an attempt was made to achieve diastereoselective recognition of a chiral guest using dibenzoyl tartaric acid, chosen for its solubility in  $\text{CDCl}_3$ , availability in both enantiomeric forms, and ability to form hydrogen bonds to the acceptor-rich environment of the macrocycle interior. However, no difference in binding was observed between enantiomers (see ESI $\dagger$ ). We attribute this to the ability of guests to bind the unsubstituted face of the macrocycles, where little stereo-differentiation is feasible; studies are ongoing to synthesise macrocycles incorporating substituents on both faces to address this issue, either by use of  $C_2$ -symmetrical 1,2-disubstituted diamines, or through the use of alternating (*R*)- and (*S*)-configured imidazolidine-2-ones in adjacent monomers.

## Conclusions

To summarise, we have synthesised and examined the properties of six chiral foldamer-derived macrocycles. The strategy enables their synthesis in entirely sequence-defined manner when required, through synthesis of a linear precursor, or in a convenient semi-defined manner through generation of linear dimers which undergo spontaneous cyclisation upon coupling to form the tetramer. DFT indicates the trimeric macrocycles adopt a bowl-like shape with substituents in a *pseudo-equatorial* position, while tetramers adopt a planar conformation, as demonstrated by single crystal X-ray diffraction, and bind metal

and ammonium cations. The high level of conformational control means these macrocycles are an excellent platform for the controlled positioning of sidechain groups. Current work is ongoing to examine applications of the macrocycles in molecular recognition and catalysis, and to develop a solid-supported second-generation synthetic approach.

## Data availability

Crystallographic data for **1a**, **2a**, **2d** and **3a** has been deposited at the CCDC under accession numbers 2057484, 2057483, 2057482 and 2057486 respectively, and can be obtained from <http://www.ccdc.cam.ac.uk>. DFT data for this paper can be found at <https://pure.qub.ac.uk> with DOI: 10.17034/8953acf-c4b8-41f4-8b4b-176fd07a8956.

## Author contributions

TMCW: data curation; formal analysis; investigation; methodology; validation; visualization; writing – original draft; writing – review & editing. SR: data curation; formal analysis; investigation; methodology; writing – review & editing. MPF: investigation; writing – review & editing. CJS: formal analysis (XRD); writing – review & editing; PD: formal analysis; investigation (DFT); visualization; writing – review & editing. PCK: conceptualization; data curation; formal analysis; funding acquisition; methodology; project administration; resources; supervision; visualization; writing – original draft; writing – review & editing.

## Conflicts of interest

There are no conflicts to declare.

## Acknowledgements

We acknowledge financial support from Queen’s University Belfast (PCK, TW), the Northern Ireland Department for the Economy (TW) and the EPSRC (EP/R021481/1, PCK, SR). LCMS facilities were funded through an EPSRC block equipment grant for early career researchers (EP/S018077/1). We thank the EPSRC UK National Crystallography Service at the University of Southampton for the collection of the crystallographic data for **3a**.<sup>52</sup>

## Notes and references

- 1 L. A. Wessjohann, R. Bartelt and W. Brandt, in *Practical Medicinal Chemistry with Macrocycles*, John Wiley & Sons, Inc., Hoboken, NJ, USA, 2017, pp. 77–100.
- 2 D. J. Newman and G. M. Cragg, in *Macrocycles in Drug Discovery*, ed. J. Levin, The Royal Society of Chemistry, 2015, pp. 1–36.
- 3 C. T. Walsh, *ACS Infect. Dis.*, 2018, **4**, 1283–1299.
- 4 E. M. Driggers, S. P. Hale, J. Lee and N. K. Terrett, *Nat. Rev. Drug Discovery*, 2008, **7**, 608–624.
- 5 A. K. Yudin, *Chem. Sci.*, 2015, **6**, 30–49.
- 6 A. A. Vinogradov, Y. Yin and H. Suga, *J. Am. Chem. Soc.*, 2019, **141**, 4167–4181.

7 L. K. Henchey, A. L. Jochim and P. S. Arora, *Curr. Opin. Chem. Biol.*, 2008, **12**, 692–697.

8 N. Sawyer and P. S. Arora, *ACS Chem. Biol.*, 2018, **13**, 2027–2032.

9 R. Gopalakrishnan, A. I. Frolov, L. Knerr, W. J. Drury and E. Valeur, *J. Med. Chem.*, 2016, **59**, 9599–9621.

10 Z. Liu, S. K. M. Nalluri and J. F. Stoddart, *Chem. Soc. Rev.*, 2017, **46**, 2459–2478.

11 M. K. P. P. Jayatunga, S. Thompson and A. D. Hamilton, *Bioorg. Med. Chem. Lett.*, 2014, **24**, 717–724.

12 W. A. Loughlin, J. D. A. Tyndall, M. P. Glenn and D. P. Fairlie, *Chem. Rev.*, 2004, **104**, 6085–6118.

13 T. Yamashita, P. C. Knipe, N. Busschaert, S. Thompson and A. D. Hamilton, *Chem.-Eur. J.*, 2015, **21**, 14699–14702.

14 E. A. German, J. E. Ross, P. C. Knipe, M. F. Don, S. Thompson and A. D. Hamilton, *Angew. Chem., Int. Ed.*, 2015, **54**, 2649–2652.

15 Z. Lockhart and P. C. Knipe, *Angew. Chem., Int. Ed.*, 2018, **57**, 8478–8482.

16 Z. E. Reinert, G. A. Lengyel and W. S. Horne, *J. Am. Chem. Soc.*, 2013, **135**, 12528–12531.

17 T. W. Craven, M.-K. Cho, N. J. Traaseth, R. Bonneau and K. Kirshenbaum, *J. Am. Chem. Soc.*, 2016, **138**, 1543–1550.

18 F. J. Carver, C. A. Hunter and R. J. Shannon, *J. Chem. Soc., Chem. Commun.*, 1994, 1277–1280.

19 L. He, Y. An, L. Yuan, K. Yamato, W. Feng, O. Gerlitz, C. Zheng and B. Gong, *Chem. Commun.*, 2005, **1011**, 3788–3790.

20 L. Xing, U. Ziener, T. C. Sutherland and L. A. Cuccia, *Chem. Commun.*, 2005, 5751–5753.

21 J. K. H. Hui and M. J. MacLachlan, *Chem. Commun.*, 2006, **1**, 2480–2482.

22 Z. J. Kinney and C. S. Hartley, *J. Am. Chem. Soc.*, 2017, **139**, 4821–4827.

23 Z. J. Kinney, V. C. Kirinda and C. S. Hartley, *Chem. Sci.*, 2019, **10**, 9057–9068.

24 H. Juwarker, J. M. Suk and K. S. Jeong, *Chem. Soc. Rev.*, 2009, **38**, 3316–3325.

25 V. C. Kirinda, B. R. Schrage, C. J. Ziegler and C. S. Hartley, *Eur. J. Org. Chem.*, 2020, **2020**, 5620–5625.

26 L. Yuan, W. Feng, K. Yamato, A. R. Sanford, D. Xu, H. Guo and B. Gong, *J. Am. Chem. Soc.*, 2004, **126**, 11120–11121.

27 A. Gube, H. Komber, K. Sahre, P. Friedel, B. Voit and F. Böhme, *J. Org. Chem.*, 2012, **77**, 9620–9627.

28 F. Böhme, M. Rillich and H. Komber, *Macromol. Chem. Phys.*, 1995, **196**, 3209–3216.

29 F. Böhme, C. Kunert, H. Komber, D. Voigt, P. Friedel, M. Khodja and H. Wilde, *Macromolecules*, 2002, **35**, 4233–4237.

30 H. Jiang, J. M. Léger, P. Guionneau and I. Huc, *Org. Lett.*, 2004, **6**, 2985–2988.

31 A. Zhang, Y. Han, K. Yamato, X. C. Zeng and B. Gong, *Org. Lett.*, 2006, **8**, 803–806.

32 S. Akine, T. Taniguchi and T. Nabeshima, *Tetrahedron Lett.*, 2001, **42**, 8861–8864.

33 R. Spencer, K. H. Chen, G. Manuel and J. S. Nowick, *Eur. J. Org. Chem.*, 2013, 3523–3528.

34 C. A. Olsen, A. Montero, L. J. Leman and M. R. Ghadiri, *ACS Med. Chem. Lett.*, 2012, **3**, 749–753.

35 J. M. Rogers, S. Kwon, S. J. Dawson, P. K. Mandal, H. Suga and I. Huc, *Nat. Chem.*, 2018, **10**, 405–412.

36 J. D. Northrup, G. Mancini, C. R. Purcell and C. E. Schafmeister, *J. Org. Chem.*, 2017, **82**, 13020–13033.

37 A. Hennig, L. Fischer, G. Guichard and S. Matile, *J. Am. Chem. Soc.*, 2009, **131**, 16889–16895.

38 J. A. Schneider, T. W. Craven, A. C. Kasper, C. Yun, M. Haugbro, E. M. Briggs, V. Svetlov, E. Nudler, H. Knaut, R. Bonneau, M. J. Garabedian, K. Kirshenbaum and S. K. Logan, *Nat. Commun.*, 2018, **9**, 4396.

39 A. M. Webster and S. L. Cobb, *Chem.-Eur. J.*, 2018, **24**, 7560–7573.

40 J. R. Dobscha, H. D. Castillo, Y. Li, R. E. Fadler, R. D. Taylor, A. A. Brown, C. Q. Trainor, S. L. Tait and A. H. Flood, *J. Am. Chem. Soc.*, 2019, **141**, 17588–17600.

41 J. W. Meisel, C. T. Hu and A. D. Hamilton, *Org. Lett.*, 2019, **21**, 7763–7767.

42 O. Meth-Cohn and Z. Yan, *J. Chem. Soc., Perkin Trans.*, 1998, **1**, 423–436.

43 V. Martí-Centelles, M. D. Pandey, M. I. Burguete and S. V. Luis, *Chem. Rev.*, 2015, **115**, 8736–8834.

44 The purification of **3a** was initially hampered by the presence of an additional impurity (after trituration) which by <sup>1</sup>H NMR appeared to have the symmetry characteristic of the macrocyclic product. ESI LC-MS suggested this impurity may be the macrocycle-Cs<sup>+</sup> complex, and it was cleanly converted to the neutral “vacant” macrocycle by extensive washing with water. This raises the prospect that Cs<sup>+</sup> plays a role in templating the overall cyclization. Meth-Cohn attempted the intramolecular S<sub>N</sub>Ar cyclization of an analogous pyridine-imidazolidinone oligomer (where dipole effects would favour the extended conformation in which the termini are not in close proximity), and this reaction failed in the presence of CsF.<sup>42</sup> In our case, the dipole-opposed conformation and any metal templation would act in unison (both favouring the compact conformation with the termini in close proximity), whilst in the work of Meth-Cohn the effects are in opposition. The failure of their system to cyclise suggests that any metal templation by coordination to the imidazolidinones is insufficient in strength to overturn the dipole-enforced innate conformational preference of the linear precursor.

45 H. J. Buschmann, L. Mutihac and K. Jansen, *J. Inclusion Phenom.*, 2001, **39**, 1–11.

46 S. Lambert, K. Bartik and I. Jabin, *J. Org. Chem.*, 2020, **85**, 10062–10071.

47 M. A. Gamal-Eldin and D. H. Macartney, *Org. Biomol. Chem.*, 2013, **11**, 488–495.

48 This guest was selected since it displays good solubility in chloroform, and the host macrocycles are poorly soluble in non-chlorinated solvents.

49 D. J. Cram, *Science*, 1988, **240**, 760–767.

50 C. E. Cannizzaro and K. N. Houk, *J. Am. Chem. Soc.*, 2002, **124**, 7163–7169.

51 S. Kubik, *J. Am. Chem. Soc.*, 1999, **121**, 5846–5855.

52 S. J. Coles and P. A. Gale, *Chem. Sci.*, 2012, **3**, 683–689.

