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Multiple valence electron detachment following Auger decay of inner-shell vacancies in gas-phase DNA

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We have studied soft X-ray photoabsorption in the doubly deprotonated gas-phase oligonucleotide [dTGGGGT-2H]²⁻. The dominating decay mechanism of the X-ray induced inner shell vacancy was found to be Auger decay with detachment of at least three electrons, leading to charge reversal of the anionic precursor and the formation of positively charged photofragment ions. The same process is observed in heavy ion (12 MeV C⁴⁺) collisions with [dTGGGGT-2H]²⁻ where inner shell vacancies are generated as well, but with smaller probability. Auger decay of a single K-vacancy in DNA, followed by detachment of three or more low energy electrons instead of a single high energy electron has profound implications for DNA damage and damage modelling. The production of three low kinetic energy electrons with short mean free path instead of one high kinetic energy electron with long mean free path implies that electron-induced DNA damage will be much more localized around the initial K-shell vacancy. The fragmentation channels, triggered by triple electron detachment Auger decay are predominantly related to protonated guanine base loss and even loss of protonated guanine dimers is tentatively observed. The fragmentation is not a consequence of the initial K-shell vacancy but purely due to multiple detachment of valence electrons, as a very similar positive ion fragmentation pattern is observed in femtosecond laser-induced dissociation experiments.

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Introduction

The production of inner-shell vacancies in biomolecular systems, in particular DNA, is one of the first steps in many types of biological radiation action.¹⁻³ Therapeutic X-ray photons and electrons, protons and heavy ions at clinically relevant kinetic energies are all likely to create inner-shell holes but also natural nuclear decay processes can do so.^{2,4-6} Their large internal excitation energy renders such vacancies of key radiobiological importance.⁷ In light atoms, which make up biological tissue, the excitation energy is released mostly in (cascades of) radiationless Auger transitions in which the excess energy from the de-excitation process is transferred to a second electron that is emitted from the atom. The emission of these

secondary electrons is considered responsible for a large part of the biological radiation damage.³ From the perspective of molecular damage, it makes a difference if one high-energy secondary electron is emitted or whether the Auger process is accompanied by many-electron detachment. In the latter case the molecular damage and break-up is expected to be more localized and prominent. Here we address the outstanding question how (un-)likely it is that inner-shell Auger decay in a single, large biomolecule is accompanied by many-electron detachment.

In radiotherapy of deep-seated tumors, one uses hard X-rays of MeV photon energies,⁸ protons of hundreds of MeV or swift heavy ions of GeV energies.⁹ For such high-energy photons and particles cross sections for the creation of inner-shell vacancies in a single molecule are very small and their beams are technically difficult to handle.¹⁰ Therefore, for the efficient production of inner-shell vacancies in light atoms such as C, N, and O we use soft X-rays. The energies of soft X rays are tuned on and over the respective K-shell absorption edges.¹¹

For investigation of the direct molecular response of DNA to inner-shell excitation and ionization without effects of the environment gas-phase studies have proven very useful.¹² Only in the gas-phase, it is directly possible to measure

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interaction products such as low energy electrons which are extremely short-lived in solution. Gas-phase studies allow to distinguish genuine molecular properties from effects that are due to the chemical environment and last but not least gas-phase studies are ideally suited for comparison to quantum chemical calculations. It has been shown that standard crossed molecular beam techniques in which the molecules are evaporated from an oven cannot be used for molecules, much larger than the nucleobases.¹³

Bari *et al.*¹⁴ and Milosavljevic *et al.*¹⁵ independently developed tandem-mass-spectrometry approaches tailored to the investigation of radiation action on complex gas-phase biomolecular systems. Here electro-sprayed biomolecular ions are mass-over-charge selected and accumulated in a radiofrequency ion trap where they can be exposed to photons or ions. The technique was quickly successfully applied to soft X-ray interaction with gas-phase protonated peptides¹⁶ and proteins.¹⁷

As DNA in biological systems is typically negatively charged, it is essential for a biologically relevant study to focus on DNA in a deprotonated state. Studies in negative ions are technically more challenging, in particular for singly or doubly deprotonated molecules, because (many-) electron detachment can lead to neutrals which are lost from the ion trap or to oppositely charged positive fragments. Many radiofrequency ion traps with linear geometries are not well-suited for simultaneous trapping of positive and negative ions. Our home-built system is based on the classical Paul-trap design allowing ion trapping independent of the sign of the electric charge.

In this study, we first present direct evidence for the X-ray induced generation of positive fragment ions from deprotonated gas-phase DNA. Multiple electron detachment Auger decay of the carbon 1s vacancy will be discussed as the underlying process. We will then present evidence for the same process in the therapeutically relevant collisions of 12 MeV C⁴⁺ ions on deprotonated DNA, in which carbon 1s vacancies are induced as well. In a last step, we will demonstrate that even though the carbon 1s vacancy is the initial step of the multiple ionization process, the subsequent fragmentation dynamics can also be triggered by multiple valence ionization, for instance by infrared multiphoton absorption in intense femtosecond laser pulses.

Methods

All experiments in this study were performed using a home-built tandem mass spectrometer, designed for interfacing with synchrotron or accelerator beamlines. The system has been described in detail before.^{14,18,19} For the soft X-ray experiments, we used the U49-2_PGM-1 soft X-ray beamline at BESSY II synchrotron at Helmholtz-Zentrum Berlin²⁰ (Germany) with photon energy at the C K-edge 280–310 eV (typically 4.0×10^{13} photons per s⁻¹ for a 600 μm slitwidth). The C⁴⁺ experiments were performed with the IRRSUD heavy ion beamline at the Grand Accélérateur National d'Ions Lourds (GANIL), in Caen (France). The C⁴⁺ ions at a total kinetic energy of 11.76 MeV (typically 1.4×10^{11} ions per s) passed a 2 mm diaphragm right before the ion trap.

The IR fs-laser experiments were performed using the Potsdam Coherent Astrella Ti:Sapphire femtosecond laser system (center wavelength: 800 nm, repetition rate: 1 kHz, measured pulse length: 47 fs). A lens was used to focus the beam to a measured spot size of 103.5 μm at the laser-sample interaction point.

Briefly, oligonucleotide anions (synthesized at LGC Biosearch technologies, Risskov, Denmark) were generated from a 40 μM solution (20/80 (v:v) water/methanol, UHPLC grade chemicals, Sigma-Aldrich) in an electrospray ionization (ESI) source. The anions were then phase-space compressed into a well-defined beam using an RF ion funnel. The ions were accumulated in an RF octopole ion trap/guide. Bunches of ions were extracted into a RF quadrupole mass filter to select the anions [dTGGGGT-2H]²⁻ ($m/z = 931$ Da, see Fig. 4 and 1 for the schematic structure) and [dTTGGGT-2H]²⁻ ($m/z = 918$ Da). The mass-selected ions were accumulated in an RF Paul trap and cooled by He buffer gas collisions. We present [dTGGGGT-2H]²⁻ data for the soft X-ray and MeV ion studies but [dTTGGGT-2H]²⁻ data for the IR fs-laser experiments. In the IR case, the recorded [dTGGGGT-2H]²⁻ dataset was smaller, but featured almost identical mass spectra.

In the trap, the oligonucleotide anions were exposed to soft X-rays photons at the C K-edge ($t_{\text{exposure}} \sim 1$ s), to 11.76 MeV C⁴⁺ ions ($t_{\text{exposure}} \sim 1$ s), or IR fs laser pulses (t_{exposure} can be varied, typically below 1 s). The trap content was then extracted into a time-of-flight (TOF) mass spectrometer. The sign of the extraction voltage was chosen in order to extract either negative or positive ions. A few hundred of individual TOF spectra were averaged to obtain a mass spectrum of sufficient signal to noise ratio. The typical acquisition time for single mass spectra presented in this work was between 15 and 60 minutes. It is important to realize that exposure of doubly deprotonated oligonucleotides to soft X-rays, MeV ions and IR fs laser pulses can also lead to formation of neutral precursor molecules or neutral fragments. Neutrals are always lost from the ion-trap and therefore inaccessible in our experiment. Furthermore, both trapping efficiency and detection efficiency depend strongly on the m/q of an ion. This, together with the fact that multiple charge fragments can be produced from a single precursor ion, makes it impossible to directly compare the amount of precursor loss with the total yield of positive and negative interaction products. It is however possible to compare the relevance of neutralization for the soft X-rays, MeV ions and IR fs pulses.

Results and discussion

In 2013, Gonzalez-Magaña and co-workers performed a pioneering study of a doubly protonated oligonucleotide (dGCAT, where G, C, A and T are guanine, cytosine, adenine and thymine).²¹ Mostly protonated and radical nucleobase cations as well as a prominent deoxyribose fragment were detected for X-ray photoabsorption at the various 1s absorption edges. Glycosidic bond cleavage was identified as a key mechanism and the radiation sensitivity of deoxyribose was confirmed. In biological systems and at physiological pH, the phosphate



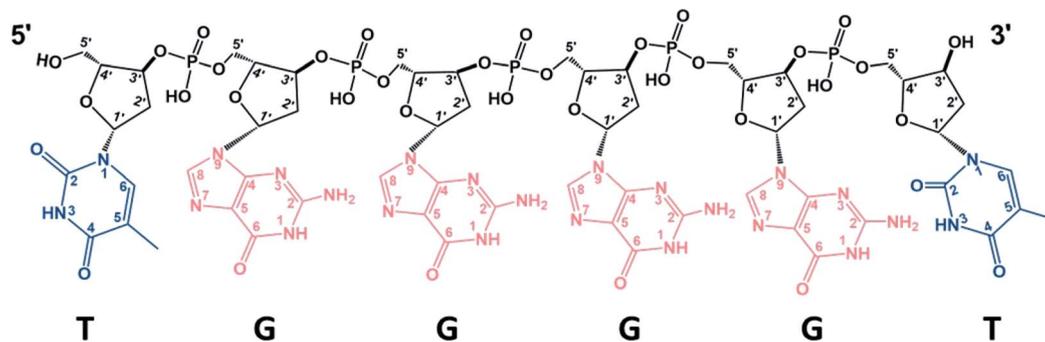


Fig. 1 Chemical structure of the dTGGGGT oligonucleotide.

groups are deprotonated and thus negatively charged. Biologically relevant gas-phase studies can take this into account by using deprotonated oligonucleotides.

Fig. 2a and b shows two $[dTGGGGT-2H]^{2-}$ photo-fragmentation spectra obtained at the C K-edge. The photon

energies are $E_{X\text{-ray}} = 292$ eV (in protonated peptides and proteins, this energy is within the carbon 1s excitation range;^{22,23} for deprotonated biomolecules it is most likely in the transition region between excitation and ionization) and $E_{X\text{-ray}} = 300$ eV (this energy is deep in the ionization range).

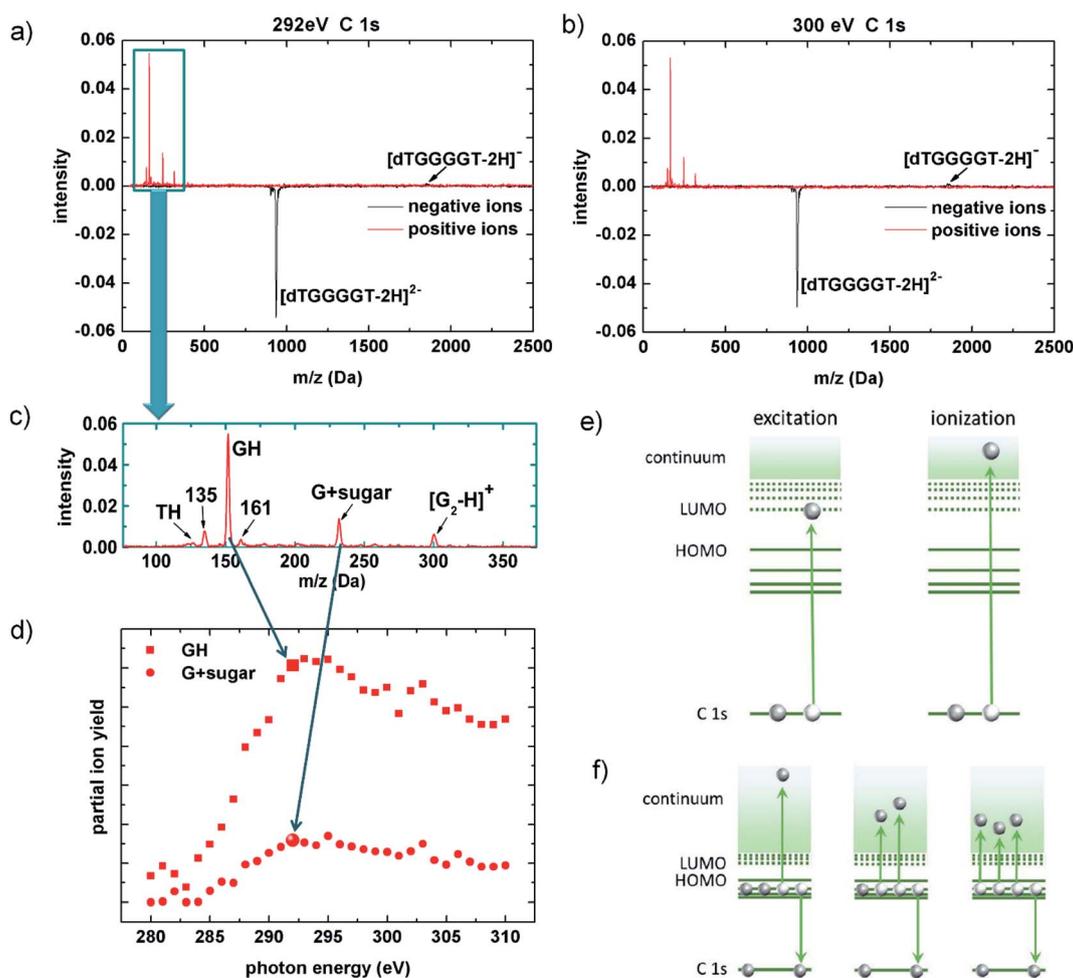


Fig. 2 Top panel: X-ray photoabsorption mass spectra of $[dTGGGGT-2H]^{2-}$ anions ($m/z = 930.6$) for $E_{X\text{-ray}} = 292$ eV (a) and 300 eV (b) (black: negative ions, red: positive ions). (c) Zoom into the cationic fragment region 75–375 Da. (d) Relative total photoabsorption cross sections as a function of photon energy for two fragments. (e) Schematics for C 1s excitation and ionization. (f) Sketch for single, double and triple valence electron detachment, associated to an Auger decay.



The spectra show the net-effect of the photoabsorption on the ion-trap content. Loss of the precursor manifests as a peak of negative intensity and photoproducts (anions or cations) appear as positive peaks. The yields are normalized to the initial trap content and to the number of photons to which the trap content has been exposed. For both photon energies, non-dissociative detachment of one electron forming $[\text{dTGGGGT-2H}]^{1-}$ is observed as a very weak feature as compared to the depletion of the $[\text{dTGGGGT-2H}]^{2-}$ precursor from the trap. Even if detection efficiency decreases with m/z , this effect cannot compensate for the lack of signal attributed to $[\text{dTGGGGT-2H}]^{1-}$. The absence of negatively charged fragments is different from what we observed at lower vacuum ultraviolet (VUV) energies, where only valence shell vacancies (that, in a radiobiological context, can be considered as oxidative damage) are induced:¹⁸ in deprotonated G-rich oligonucleotide anions, the photon-induced holes were found to migrate towards an energetically favorable G-rich site where they can weaken bonds and ultimately lead to bond cleavages and formation of anionic fragments.¹⁸

So, what is happening with the photoexcited/photoionized oligonucleotides in the soft X-ray case? As mentioned before, a straightforward explanation would be neutralization leading to loss from the ion trap. A second possibility is the formation of positive ions. K-shell ionization-induced production of positive ions from anionic precursors has only been observed for atoms and small molecules before.²⁴ For fullerene anions, multiple electron detachment upon single VUV photon absorption has been reported.²⁵ The red curves in Fig. 2a and b indeed show evidence for very efficient production of small cationic fragments with masses below 500 Da. No cationic fragments with m/q exceeding the range of 300 are observed and in particular, no non-dissociative multiple (three or more) electron detachment that would lead to positively charged dTGGGGT moieties is observed. This implies that positive ion formation due to X-ray induced multiple electron detachment induces extensive fragmentation and is thus closely related to X-ray induced DNA damage. This fact is even more clear, when zooming in on the small positive fragments (see Fig. 2c): the most intense fragment GH^+ (152 Da) is due to glycosidic bond cleavage accompanied by double H/proton transfer, a very common process also observed in photofragmentation²⁶ but also in collision-induced dissociation²⁷ of protonated oligonucleotides. The peak at 135 Da is most likely formed by NH_3 loss from GH^+ , which is a common fragmentation channel in protonated guanine.^{28,29} And the strong peak at 232 Da can be assigned to a cyclic nucleoside complex containing the guanine moiety and the five-membered sugar ring³⁰ (G + sugar). One additional strong fragment that is not commonly observed is found at 301 Da. This fragment could be assigned to a $(\text{G}_2\text{-H})^+$ -dimer, formed by double glycosidic bond scission accompanied by single H transfer.

To demonstrate that positive fragment formation is due to inner-shell excitation/ionization rather than (non-resonant) valence ionization, we choose a spectroscopic approach: by recording the partial ion yields from the peak integrals of selected photoproducts in the mass spectra in Fig. 2c as

a function of $E_{\text{X-ray}}$, the $[\text{dTGGGGT-2H}]^{2-}$ near-edge X-ray absorption mass spectrometry (NEXAMS) spectra for GH^+ and $[\text{G} + \text{sugar}]^+$ shown in Fig. 2d have been obtained. In qualitative agreement with our results for smaller, protonated oligonucleotides,²⁶ the spectra feature a strong increase between 285 eV and 292 eV that is due to resonant C 1s excitation into unoccupied molecular orbitals, mostly of π^* character (see Fig. 2e, left). At higher photon energy, direct C 1s ionization sets in (see Fig. 2e, right). In the ionization continuum between 295 eV and 300 eV the ion yields show the expected slow decrease. The transition from 1s excitation to 1s ionization implies the removal of an additional electron in the ionization case. With the presented data, we are not able to quantify the influence of this effect.

Inner shell ionization clearly is not a requirement for positive ion formation, as positive nucleobase ion formation is observed already in the excitation regime. This rules out direct photodouble detachment³¹ and even single detachment as an underlying process. The induction of a K-shell vacancy is sufficient for positive fragment ion formation, and as a consequence, Auger de-excitation (see Fig. 2f) must be an important step. The Feifel group has systematically studied double Auger decay, *i.e.*, Auger process in which two valence electrons are emitted, for a series of small molecules.³² The percentage of double Auger processes systematically increases with the number of valence electrons available in the vicinity of the core vacancy, *i.e.* with the density of valence orbitals (from 6.3% in CH_4 to 29% in CCl_4 for a carbon 1s hole).³³ If we extrapolate this trend, double Auger decay could be a dominant channel in large molecules such as oligonucleotides. However, only triple electron detachment can explain the observed positive ion formation from a doubly negative precursor. For small neutral molecules such as CO_2 , about 1% of the total Auger process are due to triple Auger decay for a C 1s vacancy, with the percentage again increasing with valence state density.³⁴ In negative ions, multiple electron detachment due to K-shell vacancy followed by Auger decay has until now only been studied for atoms and small molecules (*i.e.*, C-³⁵ and B2-/B3-³⁶). Multiple detachment is generally much stronger in negative than in neutral systems. In a very recent study on C-, it was for instance found that upon excitation of the lowest energy resonance, triple detachment was 5% and quadruple detachment was about 0.1% of the double detachment cross sections.³⁷ Triple electron detachment in a C 1s decay process could thus realistically be explained by a multiple electron detachment Auger decay, but it remains remarkable that this channel is so strong in our system under study.

Inner shell vacancies can also be produced in proton and heavy ion therapy. It is thus straightforward to investigate the response of the $(\text{dTGGGGT-2H})^{2-}$ upon interactions with a therapeutically relevant ion beam. Carbon ion beams for cancer therapy typically have energies in the GeV range. This energy is however chosen such that the ions are slowed down to the 10 MeV range at the tumor location. At these kinetic energies, energy deposition per unit length is maximum (Bragg-peak),³⁸ which is why we have chosen a 11.76 MeV C^{4+} beam for our study. MeV ion collisions predominantly lead to emission of



valence electrons, and only collisions with very small impact parameters lead to inner shell ionization, as recently shown for 42 MeV C^{6+} collisions with the gas-phase adenine nucleobase, where only about 4% of the total ionization cross section were due to K shell ionization.^{39–42}

A mass spectrum obtained after 11.76 MeV C^{4+} collisions with $(dTGGGGT-2H)^{2-}$ is shown in Fig. 3a. The negative ion spectrum is very similar to the soft X-ray case, with strong loss of the $(dTGGGGT-2H)^{2-}$ precursor, that is not compensated by the small yield of $(dTGGGGT-2H)^{-}$. Non-dissociative electron detachment is relatively stronger as compared to the soft X-ray case, but it remains a minor channel also for the MeV ion case. The positive ion spectrum again shows clear peaks in the low mass region. However, relative to the precursor loss, these peaks are much weaker as compared to the soft X-ray case, in agreement with the very low probability for 1s ionization in MeV ion collisions. The comparable yields for non-dissociative detachment and the significantly lower yield of positive fragments implies that in the MeV ion case, a larger fraction of interaction products has to be neutral and remains inaccessible in our study. It is very likely that these additional neutral products are due to double valence ionization rather than K-shell ionization. Fig. 3b shows a zoom in the low mass region of the positive ion spectrum. Essentially the same fragments as those formed upon soft X-ray absorption (Fig. 2c) are observed (the additional peak for $C_4H_5N_4$ – a guanine fragment⁴³ lacked in the soft X-ray mass spectrum because of a slightly higher low-mass cutoff of the ion trap). In summary, the relatively small cross section for K-shell ionization in MeV ion collisions leads to a lower yield of positive fragments, but the 1s ionized oligonucleotides appear to decay into the same fragmentation pattern as observed for soft X-ray absorption. There might be a contribution from the much more likely MeV ion induced multiple

valence ionization, but it is difficult to disentangle these two channels.

In a last step, we demonstrate that the K-shell vacancy is only required to facilitate multiple electron detachment in Auger decay, whereas the observed actual fragmentation (e.g., glycosidic bond cleavage) is due to valence ionization, only. We have performed complementary femtosecond laser-induced ionization/dissociation (fs-LID) experiments on $(dTGGGGT-2H)^{2-}$. In fs-LID, molecules are exposed to near infrared pulses of several 10 fs duration and with peak powers of the order of $10^{13} \text{ W cm}^{-2}$. We have employed pulses of length $\tau = 47 \text{ fs}$, photon energy $E_{\text{photon}} = 1.55 \text{ eV}$ and peak power between $9.6 \times 10^{12} \text{ W cm}^{-2}$ and $8.1 \times 10^{13} \text{ W cm}^{-2}$. The ionization of molecular anions in femtosecond laser pulses has only been studied for a few small molecules such as SF_6 before.⁴⁴ Detachment of the weakly bound electrons from a doubly negative precursor requires less energy, and therefore much lower laser intensities, than positive ion formation from a neutral or cationic molecule.⁴⁵

Daly *et al.* have recently performed a ns-pulse UV-vis spectroscopic study on deprotonated gas phase oligonucleotide 6mers in different charge states.⁴⁵ For $[dGGGGG-2H]^{2-}$, they observed a gas-phase absorption spectrum very close to the one observed in solution, with a pronounced band 1 centered at about 4.9 eV and band 2 centered at about 6.4 eV. The energetic position of these maxima is only weakly dependent on nucleobase composition and deprotonation state and consists of a multitude of excited states. Besides radiative decay back to the ground-state, excitation into one of these bands can lead to the two different processes sketched in Fig. 4. In both cases, the excited oligo can decay back to the ground state *via* internal conversion (IC). The vibrational excitation that results from a single excitation process is insufficient to induce fragmentation of an oligonucleotide 6-mer.⁴⁵ However, according to Daly *et al.*'s study,⁴⁵ the sum of adiabatic detachment energy (ADE) and repulsive coulomb barrier (RCB) is just below the first absorption band (e.g. 4.1 eV for $[dGGGTTT-2H]^{2-}$), opening up an efficient electron detachment channel. Interestingly, this channel appears to be inefficient for the case of excitation into band 2, leaving IC as the dominant channel.

In our experiment, band 1 can be reached by resonant absorption of 3 IR photons ($3 \times 1.55 \text{ eV} \approx 4.65 \text{ eV}$) and band 2 can be reached by resonant absorption of 4 IR photons ($4 \times 1.55 \text{ eV} \approx 6.2 \text{ eV}$). Variation of the fs-laser intensity allows for variation of the probability for multiphoton absorption. In line with the expectations, below a threshold peak power (here about $10^{13} \text{ W cm}^{-2}$), the laser beam has no effect on the trapped $[dTGGGGT-2H]^{2-}$ anions. In the threshold region, where we expect 3 photons to be absorbed, only electron detachment into $[dTGGGGT-2H]^{1-}$ is observed (see Fig. 5c, 1st panel). It is likely, that many molecules decay to the precursor ground state by IC but the excitation energy is insufficient for fragmentation.

Doubling the laser power to $2 \times 10^{13} \text{ W cm}^{-2}$ leads to the mass spectrum in the 2nd panel of Fig. 5c. The electron detachment channel is still dominant, but a number of negative fragment ions appear which could be explained by increased vibrational energy in the system after IC from additional photon

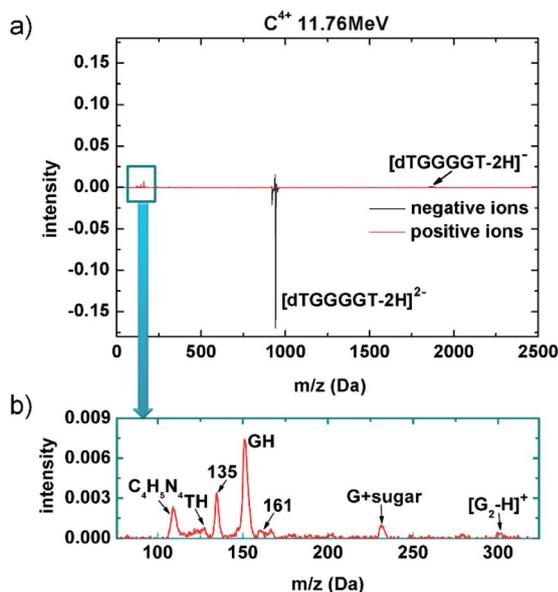


Fig. 3 Mass spectra for 10 MeV C^{4+} collisions with $(dTGGGGT-2H)^{2-}$. (a) Both negative (black) and positive (red) ion data are displayed. (b) Zoom into the region of small positive fragments.



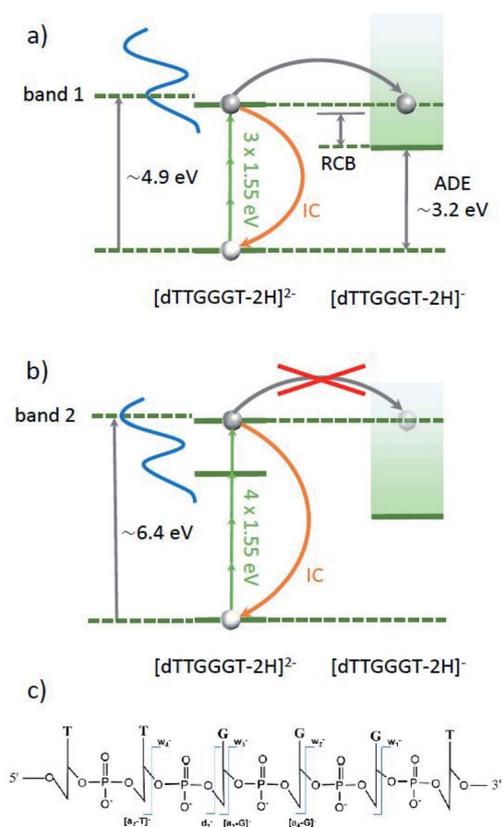


Fig. 4 Energy schemes for resonant (3 IR-photon) excitation to band 1 (a) and for resonant (4 IR-photon) excitation to band 2 (b) of [dTTGGGT-2H]²⁻. In both cases, the oligonucleotide can decay radiatively back to the ground state (not shown) or the excitation energy is transferred into vibrational energy by internal conversion (IC). In both cases, electron detachment into [dTTGGGT-2H]¹⁻ is energetically allowed. For band 1, this channel is expected to be strong whereas for band two, electron detachment is very weak.⁴⁵ Panel (c) shows the schematic structure of the oligonucleotide with the observed cleavage sites.

absorption (*i.e.* 3+). It is well established that moderately deprotonated oligonucleotides predominantly fragment by loss of a base B (here: G or T) followed by scission of the corresponding sugar 3' C–O bond (see Fig. 5c).⁴⁶ This process leads to w_n^- and $(a-B)^-$ and both types of fragments are abundant in the mass spectrum (see Fig. 5b for a zoom into a mass spectrum recorded at higher laser power, the negative-ion fragments are qualitatively similar at different laser powers). A similar pattern has also been observed in VUV photoionization of similar oligonucleotides.¹⁸ [Gn:Gn]⁻ and [T2:G3]⁻ are internal fragments.

Interestingly, also positive fragments are already formed at $2 \times 10^{13} \text{ W cm}^{-2}$. At this laser power, detachment of multiple electrons and sufficient energy transfer into vibrational modes *via* IC into the same molecule is not yet a likely combination of processes to happen. An alternative scenario would be a nucleobase-centered fs absorption process in which an electron is detached and the glycosidic bond is broken before the base can be neutralized. As a result, a positively charged nucleobase as

well as 2 negative w_n^- and $(A-B)^-$ fragments could be formed. To our knowledge, such a process has not been observed before, though.

Another doubling of the laser power to $3.8 \times 10^{13} \text{ W cm}^{-2}$ (3rd panel of Fig. 5a and c) leads to a moderate reduction in intensity for both, non-dissociative detachment and formation of negative fragments. The most dramatic change is observed for the positive fragment yields which increase in a non-linear fashion by almost one order of magnitude. The positive ion yield now clearly exceeds the negative ion yield, implying multiple electron detachment as an underlying process. Depending on the vertical detachment energy of the system, the Keldish parameter⁴⁷ could be below a value of 1 and tunneling ionization of outer valence electrons (in addition to (resonant-enhanced) multi-photon ionization processes) could become possible.⁴⁸ Tunneling ionization has occasionally been used in mass spectrometry before, in particular for protonated peptides,^{49,50} as no chromophores are required and new, non-ergodic fragmentation channels are known to open up. In any case, electron tunneling is strongly sensitive to electron binding energies which are dependent on detachment site and charge state. A thorough investigation of the respective dynamics is beyond the scope of this article, but it is clear that the mass spectrum results from removal of multiple valence electrons together with substantial electronic (and ultimately vibrational) excitation of the system. As observed for soft X-ray absorption and MeV ion impact, the positive ion spectrum is dominated by GH^+ , $(\text{G} + \text{sugar})^+$ and the tentative $(\text{G}_2\text{-H})^+$ fragment (see Fig. 5b).

A further doubling of the laser power to $8.1 \times 10^{13} \text{ W cm}^{-2}$ leads to the positive and negative ion mass spectra in the 4th panel of Fig. 5c. Negative ion yields further decrease and the positive fragment ion yields approximately double with respect to the $3.8 \times 10^{13} \text{ W cm}^{-2}$ case. The positive ion fragmentation pattern very much resembles the one for the soft X-ray case (Fig. 2a). This confirms the proposed scenario in which valence ionization is responsible for the positive fragment formation.

It is important to note that for this spectrum the laser power is well in the tunneling-regime and the increase in positive ion yields has already leveled off. From this we conclude that the formation of neutral photoproducts is probably negligible but certainly not dominating. As the spectrum is very similar to Fig. 2a, the same is probably true for the soft X-ray case and only for MeV ions, where positive ion formation is relatively weaker, neutralization may play a bigger role.

Note that in the fs-laser experiments, different power regimes can contribute to the same mass spectrum; *e.g.* electron detachment is observed at both 9.6×10^{12} and $2.0 \times 10^{13} \text{ W cm}^{-2}$, and positive ions are observed at 2.0×10^{12} and $3.8 \times 10^{12} \text{ W cm}^{-2}$. This is because the beam has a Gaussian intensity profile: precursor ions are exposed to the nominal intensity solely at the center of the beam. Oligonucleotides can be multiply ionized and positive fragments form. Away from the center, intensities are lower and single electron detachment may still dominate. At the moment, we take care to mention that these proposed mechanisms are unfortunately still somewhat speculative due to lack of similar experiments in the literature.



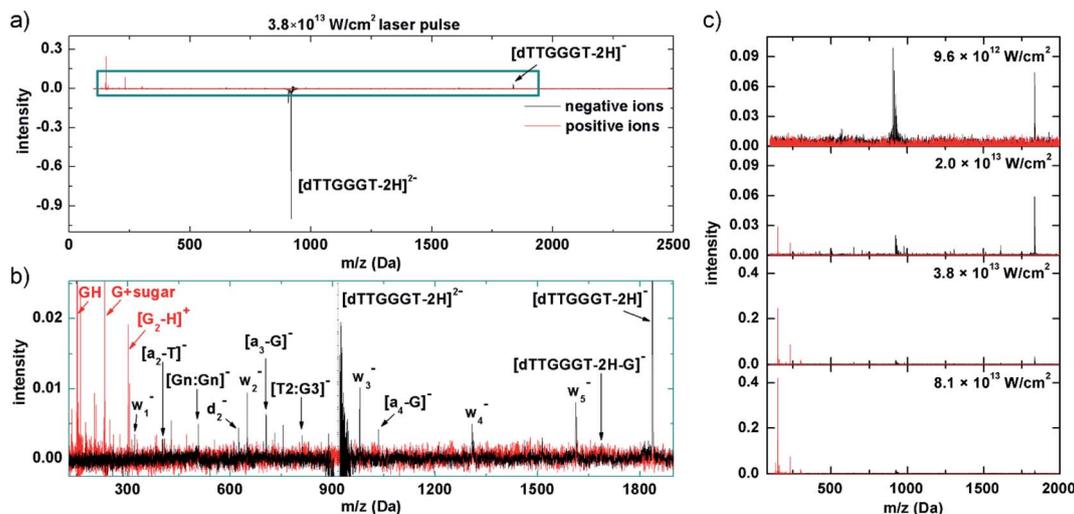


Fig. 5 Mass spectra of the fs-LID products from $(\text{dTTGGGT-2H})^{2-}$ for 47 fs 800 nm laser pulses. (a) Data for $3.8 \times 10^{13} \text{ W cm}^{-2}$ pulses (black: negative ions, red: positive ions); (b) zoom into the fragment region; (c) spectral evolution as a function of pulse energy; the strong positive peak observed for $9.6 \times 10^{12} \text{ W cm}^{-2}$ is an artifact at the m/q of the precursor ion.

Despite basing our proposed mechanisms on similar power ranges scans and results performed by Daly *et al.*, we do note that there are differences between the two experiments in terms of the photon energy used (IR vs. UV) and the duration of the laser pulses used (fs vs. ns), which will massively affect which states are populated, which mechanisms are involved as well as the timescales in which ionisation/dissociation occurs. More work is therefore needed to elucidate how exactly (1) photon energy, (2) pulse duration and (3) laser intensity all affect the dissociation of these systems in to both positive and negative fragments. The preliminary observations presented in this article are simply the first step into these new studies.

Conclusion

We have shown evidence for strong three (or more) electron detachment in the Auger decay of a K-shell vacancy in gas-phase deprotonated oligonucleotides. This process is characteristic for soft X-ray absorption around the carbon K-edge, but can also be observed in heavy ion induced DNA damage where inner shell vacancies are generated as well. The DNA damage processes that are triggered by triple electron detachment Auger decay are predominantly related to glycosidic bond cleavage accompanied by H transfer (base loss). An indication for loss of protonated guanine dimers is observed as well. The actual damage mechanism clearly is not related to the initial K-level vacancy but due to the induced multiple detachment of valence electrons, as a very similar positive ion fragmentation pattern is observed in fs-LID experiments.

Triple electron detachment Auger decay of K-shell vacancies in DNA has profound consequences for DNA damage and for damage modeling. A single Auger electron emitted after decay of a carbon 1s vacancy at about 300 eV has a penetration range of about 10 nm whereas triple electron detachment decay would imply substantially smaller electron energies, with smaller

penetration depths ($\sim 3\text{--}4$ nm for electron energies below 100 eV).⁵¹ As a result, damage due to secondary electrons from inner shell vacancy decay in DNA very likely is much more localized than previously thought. The localized damage of three or more low energy electrons emitted after the decay of a single K-shell vacancy in DNA could be sufficient to induce a double strand break or possibly even a cluster of DNA damage.⁵²

In future experiments, we plan to measure the energy distribution of the emitted electrons and to investigate the influence of charge state, length and sequence of the deprotonated oligonucleotide on positive ion formation. This knowledge is crucial to understand how far the results of our gas-phase studies can be transferred to the (biologically relevant) liquid phase. Of particular interest would be the investigation of nanosolvated systems, *i.e.* deprotonated gas-phase oligonucleotides with attached water molecules. Furthermore, in particular electron spectroscopy will give much deeper insight into the actual Auger decay processes associated with individual excited states, the precise number of electrons emitted and possibly answer the question whether the soft-X-ray excited 1s electron participates actively in the subsequent Auger decay or whether it is merely a spectator.

Data availability

The data for Fig. 2, 3 and 5 can be found in the ESI.

Author contributions

According to CASRAI CRediT (contribution role taxonomy). (1) Conceptualization; Wen Li, Thomas Schlathöler. (2) Data curation; Thomas Schlathöler. (3) Formal analysis; Wen Li, Xin Wang, Thomas Schlathöler. (4) Funding acquisition; Thomas Schlathöler, Ronnie Hoekstra, Markus Gühr, Jean Christophe Pouilly. (5) Investigation; Wen Li, Wessel Douma, Xin Wang,



Dennis Mayer, Matthew S. Robinson, Markus Gühr, Mathieu Lalande, Marwa Abdelmouleh, Michal Ryszka, Jean Christophe Pouilly, Thomas Schlathölder. (6) Methodology; Thomas Schlathölder, Markus Gühr, Jean Christophe Pouilly. (7) Project administration; Wen Li, Thomas Schlathölder. (8) Resources; Thomas Schlathölder, Jean Christophe Pouilly, Ronnie Hoekstra, Markus Gühr. (9) Software; Thomas Schlathölder. (10) Supervision; Thomas Schlathölder, Markus Gühr, Jean Christophe Pouilly. (11) Validation; Wen Li, Matthew S. Robinson, Thomas Schlathölder. (12) Visualization; Wen Li, Oksana Kavatsyuk, Thomas Schlathölder. (13) Writing – original draft; Wen Li, Oksana Kavatsyuk, Thomas Schlathölder. (14) Writing – review & editing; Wen Li, Oksana Kavatsyuk, Matthew S. Robinson, Ronnie Hoekstra, J. C. Pouilly, Thomas Schlathölder.

Conflicts of interest

There are no conflicts to declare.

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