

Cite this: *Chem. Sci.*, 2021, 12, 11515

All publication charges for this article have been paid for by the Royal Society of Chemistry

CO/light dual-activatable Ru(II)-conjugated oligomer agent for lysosome-targeted multimodal cancer therapeutics†

Min Yang,^{‡a} Hao Zhao,^{‡b} Ziqi Zhang,^a Qiong Yuan,^a Qian Feng,^a Xinrui Duan,^{‡a} Shu Wang,^{‡b} and Yanli Tang,^{‡*a}

Stimuli-activatable and subcellular organelle-targeted agents with multimodal therapeutics are urgently desired for highly precise and effective cancer treatment. Herein, a CO/light dual-activatable Ru(II)-oligo-(thiophene ethynylene) (Ru-OTE) for lysosome-targeted cancer therapy is reported. Ru-OTE is prepared via the coordination-driven self-assembly of a cationic conjugated oligomer (OTE-BN) ligand and a Ru(II) center. Upon the dual-triggering of internal gaseous signaling molecular CO and external light, Ru-OTE undergoes ligand substitution and releases OTE-BN followed by dramatic fluorescence recovery, which could be used for monitoring drug delivery and imaging guided anticancer treatments. The released OTE-BN selectively accumulates in lysosomes, physically breaking their integrity. Then, the generated cytotoxic singlet oxygen (¹O₂) causes severe lysosome damage, thus leading to cancer cell death via photodynamic therapy (PDT). Meanwhile, the release of the Ru(II) core also suppresses cancer cell growth as an anticancer metal drug. Its significant anticancer effect is realized via the multimodal therapeutics of physical disruption/PDT/chemotherapy. Importantly, Ru-OTE can be directly photo-activated using a two-photon laser (800 nm) for efficient drug release and near-infrared PDT. Furthermore, Ru-OTE with light irradiation inhibits tumor growth in an MDA-MB-231 breast tumor model with negligible side effects. This study demonstrates that the development of an activatable Ru(II)-conjugated oligomer potential drug provides a new strategy for effective subcellular organelle-targeted multimodal cancer therapeutics.

Received 5th March 2021
Accepted 21st July 2021

DOI: 10.1039/d1sc01317c

rsc.li/chemical-science

Introduction

Cancer, with its high mortality and low therapeutic treatment efficiency, is one of the most serious healthcare problems that threatens human health worldwide.¹ The development of highly precise and effective agents, as well as therapeutic methods, is urgently desired for cancer treatments.^{2,3} Photodynamic therapy (PDT) has been clinically approved as a non-invasive and effective alternative approach to traditional treatments for various types of cancer due to its negligible drug resistance and minimal side effects.⁴ In PDT, photosensitizers (PSs) can be excited by light and sensitize the surrounding oxygen to produce cytotoxic reactive oxygen species (ROS), which lead to cell death through vasculature damage, apoptosis and/or

necrosis pathways, and acute inflammatory reactions.^{5–8} However, conventional PSs may suffer low effectiveness because of their strict dependence on an abundant oxygen supply, undesirable nonspecific photodamage and simple therapeutic model.⁹ Stimuli-activatable PSs for multimodal therapeutics provide a new opportunity to cancer therapy, aimed at improving their therapeutic precision and efficiency.

Stimuli-activatable agents combined with PSs and drugs, have attracted increasing attention in cancer therapy in virtue of their selective activation, targeted delivery, high systemic safety, and enhanced therapeutic efficiency.^{10–13} Upon selective activation by external and/or internal stimulations, stimuli-activatable agents release active therapeutic moieties (PSs and drugs) that exercise multi-modal PDT and chemotherapy functions, thus improving selectivity and reducing toxicity towards non-triggered normal tissues. Importantly, subcellular organelle-targeted PDT can enhance the therapeutic efficiency as ROS work robustly in a small radius in biological systems.^{9,14–16} In particular, lysosomes have emerged as a target for PDT owing to their close relationship with apoptosis. It has been demonstrated that lysosome damage caused by PDT is more efficacious in leading to cell death rather than other organelles.^{17–20} Lysosome-targeted PSs are promising agents for

^aKey Laboratory of Analytical Chemistry for Life Science of Shaanxi Province, Key Laboratory of Applied Surface and Colloid Chemistry, Ministry of Education, School of Chemistry and Chemical Engineering, Shaanxi Normal University, Xi'an, Shaanxi Province, 710119, P. R. China. E-mail: yltang@snnu.edu.cn

^bBeijing National Laboratory for Molecular Sciences, Key Laboratory of Organic Solids, Institute of Chemistry, Chinese Academy of Sciences, P. R. China

† Electronic supplementary information (ESI) available. See DOI: 10.1039/d1sc01317c

‡ These authors contributed equally to this work.



highly efficient and precise PDT. To date, the reported stimuli-activatable PSs were prepared *via* non-covalent interactions (electrostatic interactions, hydrophobic interactions and van der Waals force) and were initiated by only one type of stimulation (*e.g.*, light or pH). PSs that are dual-activated *via* internal gaseous signaling molecules and external light with robust spatiotemporal precision further increases the biosafety and reduces the risk of systemic toxicity. The construction of covalently stable and dual-activatable systems for lysosome-targeted multimodal cancer treatment remains a big challenge to augment therapeutics controllability, precision and efficiency.

Ru(II) complexes have proven to be promising anticancer metal drugs by taking advantages of their improved selectivity, preferable biocompatibility, and better therapeutic effects.^{21,22} Generally, some Ru(II) complexes can produce ¹O₂ and undergo ligand substitution under light irradiation to release Ru(II) species and ligands for chemotherapy and PDT.^{23–29} In terms of Ru(II) complexes, an activated fluorescence system as a therapeutic platform for imaging cancer cells, monitoring drug delivery and multimodal cancer treatment has been rarely reported. Recently, cationic conjugated oligomers (including cationic oligo(*p*-phenylene vinylene), oligo(*p*-phenylene ethynylene)s, and oligo-(thiophene ethynylene) (OTE)), have been shown to exhibit high antibacterial and antitumor activity on account of their noticeable PDT effect.^{30–32} In addition, cationic conjugated oligomers usually exhibit strong light-harvesting ability and high fluorescence quantum yield,^{33,34} which form a solid base for investigating the dynamic distribution of drugs inside cells.³⁵ In particular, OTE can physically break the membrane structure of microorganisms efficiently.³⁶ Thus, it would be appealing to design and construct a cationic conjugated oligomer-based dual-activatable potential drug system for lysosome-targeted multimodal cancer treatment together with fluorescence monitored drug delivery.

In this work, we designed and synthesized a new dual-activatable Ru(II)-conjugated oligomer agent (Ru-OTE) through the coordination-driven self-assembly of a cationic conjugated oligomer ligand (OTE-BN) and a Ru(II) center. Carbon monoxide (CO) is an intracellular gaseous signaling molecule and an intrinsic metal ligand in enzymes,³⁷ the production of which will increase in the body under certain pathophysiological conditions (*e.g.* inflammation).³⁸ Herein, CO and light as internal and external stimulants, respectively, are applied to trigger the release of OTE-BN (PSS) and Ru(II) agents. As shown in Scheme 1, Ru-OTE has weak fluorescence due to the heavy atom effect, and can be uptaken by cancer cells. Subsequently, OTE-BN is released from Ru-OTE *via* the dual-activation of CO and light. The released OTE-BN then emits bright fluorescence and selectively accumulates in lysosomes, physically breaking their integrity. Besides this, the generated cytotoxic ¹O₂ causes severe lysosome damage, thus leading to cancer cell death *via* PDT. Meanwhile, the release of Ru(II) anticancer agents also suppresses cancer cell growth as anticancer metal drugs.^{39–41} Significant anti-cancer effect was realized by the multimodal therapeutics of physical disruption/PDT/chemotherapy. This study reports a cationic conjugated oligomer-based dual-activatable system for lysosome-targeted multimodal cancer

treatment together with fluorescence monitored drug delivery, which paves the way to the development of a more secure, effective and intelligent therapeutic potential drug.

Results and discussion

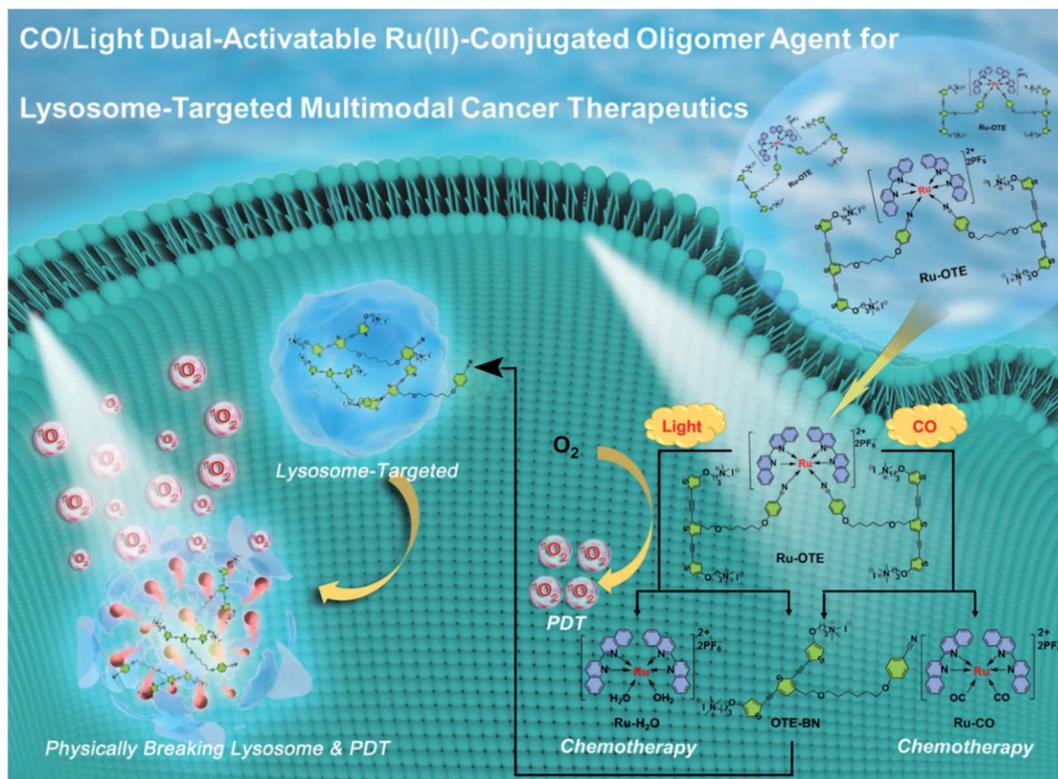
Design, synthesis and characterization of Ru-OTE

To obtain a multifunctional agent for synergistic cancer therapy, a novel cationic conjugated oligomer-Ru(II) complex Ru-OTE was designed. The synthesis route of Ru-OTE is displayed in Fig. 1 and the detailed synthesis procedures are shown in the ESI.† Conjugated oligomer **9** with a cyano group was obtained *via* a Sonogashira coupling reaction of compound **3** and **8** in a yield of 52%. Ru(biq)₂Cl₂ and Ru-H₂O were synthesized according to the literature.²⁴ The reaction of Ru-H₂O with **9** afforded Ru-OTE-pre in a yield of 11% through coordination between Ru and the cyano group on compound **9**. Then, cationic Ru-OTE was prepared *via* the quaternization reaction of Ru-OTE-Pre with CH₃I in a yield of 98%, which imparted Ru-OTE with water-solubility and the benefit of being able to be applied in biomedical applications without the involvement of harmful organic solvents. Additionally, OTE-BN was obtained *via* a quaternization reaction as a control molecule in following investigations. All intermediates and the final product Ru-OTE were fully characterized by nuclear magnetic resonance (NMR) spectroscopy and mass spectrometry (Fig. S12–S23, ESI†). The photophysical properties, ROS generation ability and optical response of Ru-OTE towards CO and light were studied prior to exploring the biomedical applications. As shown in Fig. 2a and S1,† Ru-OTE exhibits a broad metal-to-ligand charge transfer (MLCT) band, with three absorption peaks at 348, 556 and 674 nm, respectively, indicating that the photocleavage of the Ru-cyano coordination bond in Ru-OTE could be induced in the whole white light region. Besides this, Ru-OTE displays two emission peaks at 412 and 440 nm. The maximal absorption of OTE-BN is located at 371 nm, while two absorption peaks at 344 and 593 nm are observed for Ru-H₂O. The emission peaks of OTE-BN and Ru-H₂O appear at 438 and 445 nm, respectively.

CO/light dual-activatable response

Furthermore, the optical response of Ru-OTE towards CO and light was explored to prove its dual-activation properties. CORM-3 was chosen as a CO release agent in the following studies. CO coordinates with Ru(II) and undergoes ligand exchange so that OTE-BN is released from Ru-OTE. The fluorescence of OTE-BN is subsequently recovered due to the inhibited heavy atomic effect. As shown in Fig. 2c, the fluorescence intensity indeed increased after the addition of CO. A 1.76-fold intensity enhancement was observed and the fluorescence reached a plateau upon the addition of 10 μM CORM-3 (Fig. 2d). The low detection limit was calculated to be 98 nM (S/N = 3). Fig. S2† displays the rapid response of Ru-OTE to CO, where the ligand substitution of water by CO at the Ru(II) center was achieved within 15 min. Then, the selectivity of Ru-OTE towards CO was investigated by assaying the fluorescence of Ru-OTE in the presence of various species, including hydrogen





Scheme 1 Schematic illustration of CO/light dual-activatable Ru-OTE agent for lysosome-targeted multimodal cancer therapeutics.

peroxide (H₂O₂), hydrogen sulfide (H₂S), nitric oxide (NO), cysteine, glutathione (GSH), peroxyntirite anion (ONOO⁻), carbonate anionnitrite (HCO₃⁻), sodium citrate (NaCit), ascorbic acid (AA) and imidazole. As shown in Fig. 2e, negligible fluorescence changes were observed for the interferents, except for CO. This result distinctly demonstrates that Ru-OTE can be specifically activated by CO.

The optical response of Ru-OTE to light was also studied by measuring its fluorescence intensity after irradiation with white light. Light illumination breaks the Ru(II)-cyano bonds in Ru-OTE, then OTE-BN is quickly and efficiently released, leading to an enhancement in fluorescence intensity (Fig. 2f). Around 10.8-fold increase in intensity was achieved when Ru-OTE was irradiated by white light (25 mW cm⁻²) for 18 min, which was higher than that of CO. To investigate the synergistic response of Ru-OTE co-activated by CO and light, Ru-OTE was incubated with CO for 15 min followed by irradiation with light. As shown in Fig. 2g, the fluorescence intensity increased rapidly and reached maximum at 15 min, which was faster than that only by light irradiation (Fig. 2h). This result indicates that the synergistic effect of CO and light accelerates the release of OTE-BN. To directly confirm the release of OTE-BN from Ru-OTE, high-resolution mass spectra of Ru-OTE before and after activation by CO and/or light were measured. As shown in Fig. 2i, there was no peak of OTE-BN at 386.6662 (*m/z*) in the spectrum of the Ru-OTE group. While a peak for OTE-BN obviously appeared after Ru-OTE was activated by CO, light and CO + light, which solidly confirmed the valid activation of Ru-OTE by CO and light.

¹O₂ generation capacity of Ru-OTE

Additionally, the ¹O₂ generation capacity of the Ru-OTE system in the presence of CO and light was explored using a singlet oxygen sensor green reagent (SOSG). As shown in Fig. 2b, the fluorescence of SOSG dramatically increased upon the addition of Ru-OTE, which was similar to that of OTE-BN. Interestingly, higher fluorescence intensity was detected after Ru-OTE was incubated with CO. When SOSG was mixed with Ru-H₂O, the fluorescence remained the same as the control. In addition, as displayed in Fig. S3,† the ¹O₂ quantum yield of OTE-BN was measured to be 1.05 using 9,10-anthracenediyl-bis(methylene) dimalonate (ABDA) as an indicator and rose bengal (RB) as a standard reference (¹O₂ quantum yield = 0.75).^{42,43} These results reasonably confirm that Ru-OTE can be regarded as an effective PDT agent for antitumor treatment, which is probably due to the release of OTE-BN from Ru-OTE. Importantly, the synergistic release of OTE-BN *via* the dual activation of CO and light lead to the higher production of ¹O₂.

Intracellular response of Ru-OTE towards light/CO

The excellent CO/light dual-activatable system, desirable turn-on fluorescence properties and high ¹O₂ generation capacity of the activatable Ru-OTE system paved the way for exploring the bioapplications of activable drug delivery monitoring and multimodal cancer therapeutics. Firstly, the intracellular response of Ru-OTE towards light (as an external stimulus) and CO (as an internal stimulus) in MDA-MB-231 human breast



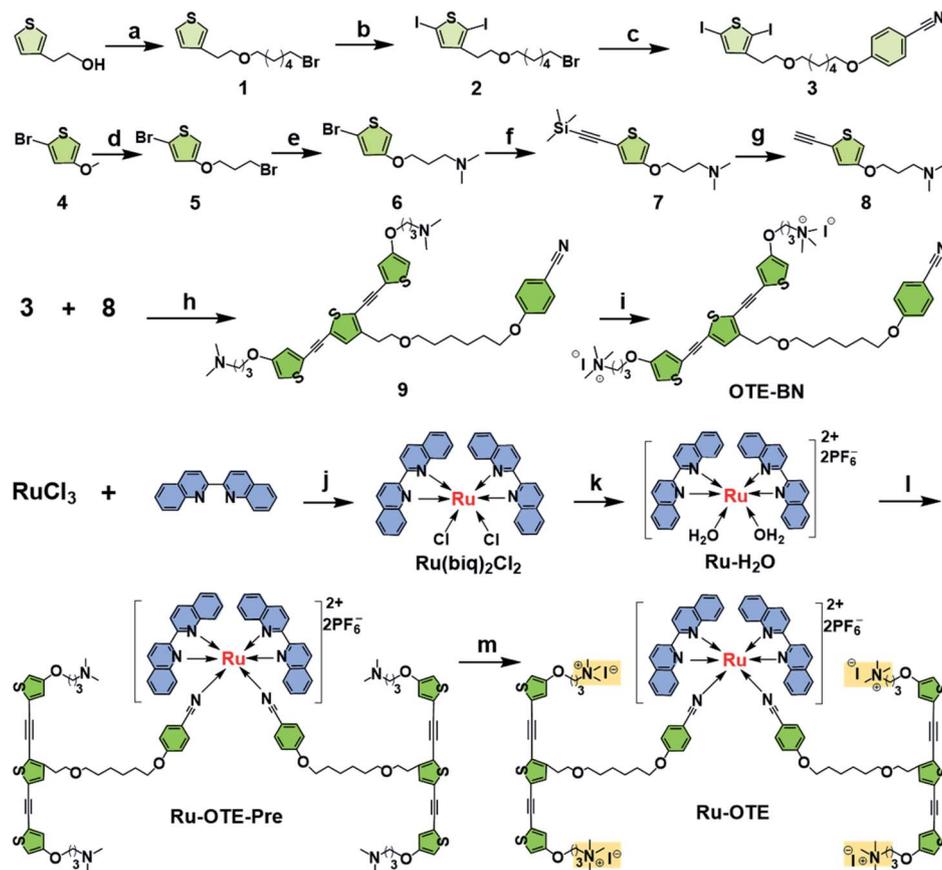


Fig. 1 The synthesis route of Ru-OTE. (a) $\text{Br}(\text{CH}_2)_6\text{Br}$, THF, KOH, 85 °C, 12 h; (b) NIS, $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{COOH}$, r.t., 8 h; (c) 4-hydroxybenzoinitrile, K_2CO_3 , 18-crown-6, acetone, 80 °C, 8 h; (d) $\text{Br}(\text{CH}_2)_3\text{OH}$, toluene, NaHSO_4 , 100 °C, 12 h; (e) $\text{NH}(\text{CH}_3)_2$, THF, r.t., 24 h; (f) TMSA, $\text{Pd}(\text{PPh}_3)_2\text{Cl}_2/\text{CuI}$, triethylamine, 100 °C, 12 h; (g) THF, CH_3OH , K_2CO_3 , r.t. 3 h; (h) TMSA, diethylamine/ CHCl_3 , $\text{Pd}(\text{PPh}_3)_2\text{Cl}_2/\text{CuI}$, 35 °C, 2 h; (i) CH_3I , CHCl_3 , 48 h; (j) LiCl, *N,N*-dimethylformamide, 135 °C, overnight; (k) $\text{AgPF}_6/\text{KPF}_6$, ethanol/ H_2O , 90 °C, 5 h; (l) OTE-BN, acetone, 56 °C, 12 h; (m) CH_3I , $\text{CHCl}_3/\text{CH}_3\text{OH}$, 48 h.

cancer cells was studied. MDA-MB-231 cells were incubated with Ru-OTE in the dark for 2, 4, 6, and 8 h, followed by irradiation with white light (25 mW cm^{-2}) for 30 min. As shown in Fig. S4a and b,[†] the signal of Ru-OTE in cells gradually increased with incubation time, reaching a maximum at 6 h, indicating the complete cellular uptake of Ru-OTE. Then, the response of Ru-OTE to endogenous CO was next investigated. As heme induces endogenous CO production, MDA-MB-231 cells were treated with heme for 2, 4, 6, and 8 h, then incubated with Ru-OTE for 6 h. As shown in Fig. S4c and d,[†] the fluorescence intensity of Ru-OTE inside cells indeed increased along with the heme treatment time, which was due to endogenous CO coordinating with Ru-OTE and releasing fluorescent OTE-BN. Furthermore, the dual-activation of Ru-OTE by CO/light was studied after cells being treated with heme and irradiated by light. As shown in Fig. S5,[†] bright blue fluorescence was observed when cells were treated by both hemin and light, indicating that dual-stimulation generates stronger fluorescence than only one treatment. These results verify that the release of OTE-BN and Ru(II) species from Ru-OTE can be dual-activated *via* external and internal stimulation, thus improving the delivery efficacy and accuracy, and reducing the possible

deficiency of single stimulus. Such release was also monitored by fluorescence imaging. Additionally, the morphological structure of the cells was gradually destroyed, which may result from the $^1\text{O}_2$ produced by OTE-BN under light irradiation (Fig. S5[†]).

CO/light dual-activation drug delivery and cell cytotoxicity

A schematic diagram of a CO/light dual-activatable Ru-OTE agent for lysosome-targeted cancer therapeutics is illustrated in Fig. 3a. Upon the CO/light dual-activation of Ru-OTE, OTE-BN and Ru(II) species were released, which was accompanied by fluorescence turn-on. OTE-BN undergoes targeted accumulation in lysosomes and physically breaks lysosomes,^{36,44,45} meanwhile $^1\text{O}_2$ is robustly generated for PDT under light irradiation. Besides this, the Ru(II) species also suppressed cancer cell growth as an anticancer metal drug. The half-maximal inhibitory concentration (IC_{50}) values of Ru-OTE, Ru- H_2O and OTE-BN under dark and light irradiation conditions were measured. As shown in Fig. 3b and S6,[†] the $\text{IC}_{50\text{dark}}$ values of Ru-OTE, Ru- H_2O and OTE-BN were determined to be 12.6, 13.5 and 8.0 μM , respectively. The $\text{IC}_{50\text{light}}$ values of Ru-OTE, Ru- H_2O and OTE-BN were as low as 1.2, 7.2 and 3.8 μM , respectively. In



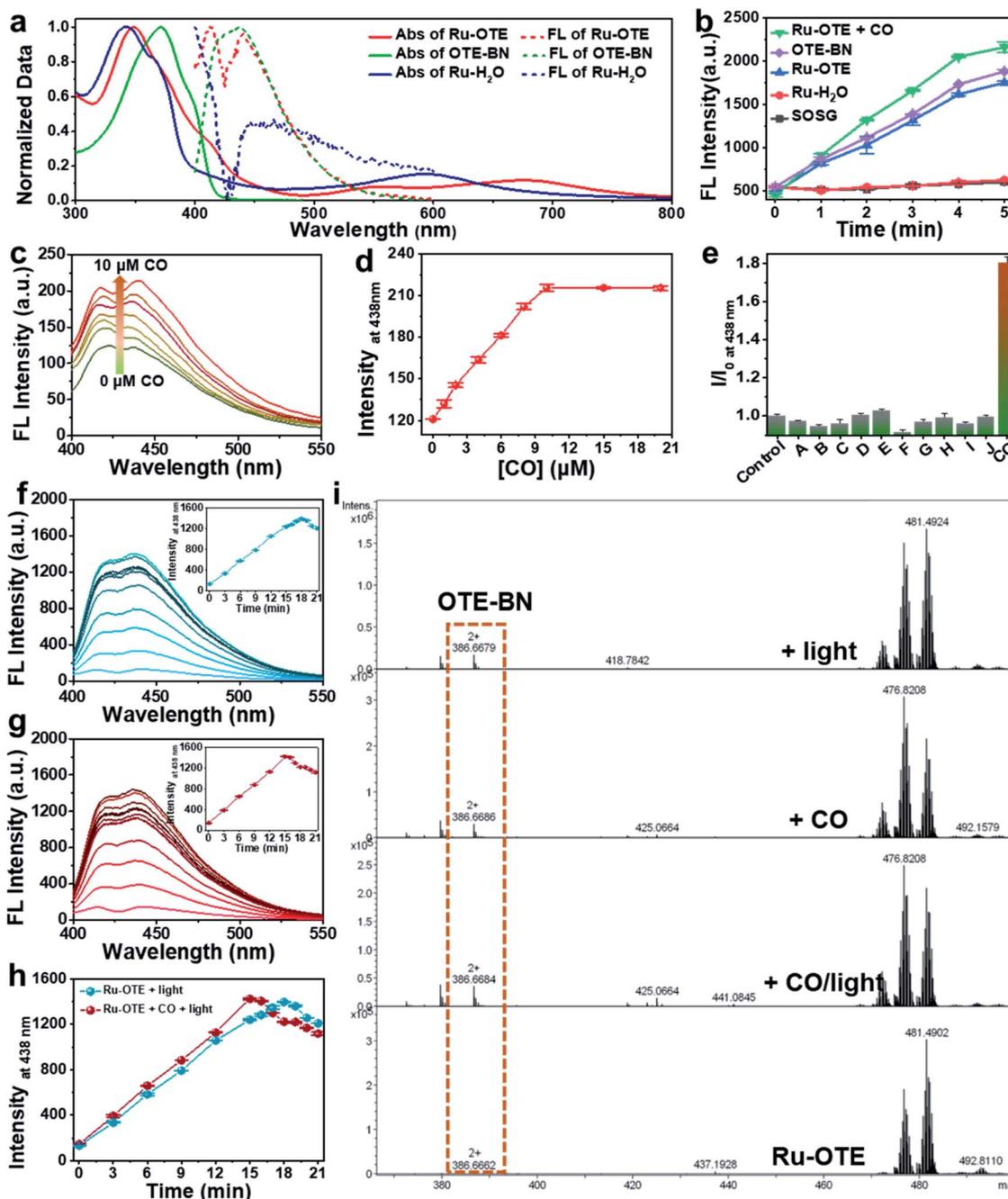


Fig. 2 The photophysical properties, ROS generation ability and optical response of Ru-OTE towards CO and light. (a) Normalized absorption and emission spectra of Ru-OTE, OTE-BN and Ru-H₂O in aqueous solution. (b) Fluorescence intensity of SOSG at 525 nm in tris (100 mM, pH 7.5) in the presence of different agents under white light irradiation (5 mW cm⁻², 5 min) with an excitation wavelength of 504 nm [SOSG] = [other agents] = 1.0 μM. (c) Fluorescence emission spectra and (d) intensity at 438 nm of Ru-OTE (1.0 μM) in PBS (10 mM, pH 7.4) in the presence of different concentrations of CO. (e) Selectivity of Ru-OTE towards CO. The concentration of all of the analytes is 10 μM. The excitation wavelength is 371 nm. The error bars represent the standard deviations of three parallel assays. (A–J) H₂O₂, H₂S, NO, Cys, GSH, ONOO⁻, HCO₃⁻, NaCit, AA and imidazole. (f) Fluorescence emission spectra of Ru-OTE (1.0 μM) in PBS after white light irradiation (25 mW cm⁻²). (g) Fluorescence emission spectra of Ru-OTE (1.0 μM) in PBS in the presence of the CO after white light irradiation (25 mW cm⁻²). (h) Fluorescence intensity at 438 nm with respect to light irradiation time. λ_{ex}: 371 nm. (i) High-resolution mass spectra of Ru-OTE before and after activation by CO and/or light.

addition, the phototoxicity indexes (PI) of Ru-OTE, Ru-H₂O and OTE-BN were calculated be 10.5, 1.9 and 2.1. It should be noted that the PI of Ru-OTE is 10.5, which is significantly higher than that of the photoreleased products Ru-H₂O and OTE-BN,

indicating the good performance of Ru-OTE in CO/light dual-activatable multimodal cancer therapeutics. Subsequently, the anticancer efficiency of Ru-OTE after activation by CO and light was explored. As shown in Fig. 3c, the Ru-OTE + CO group only



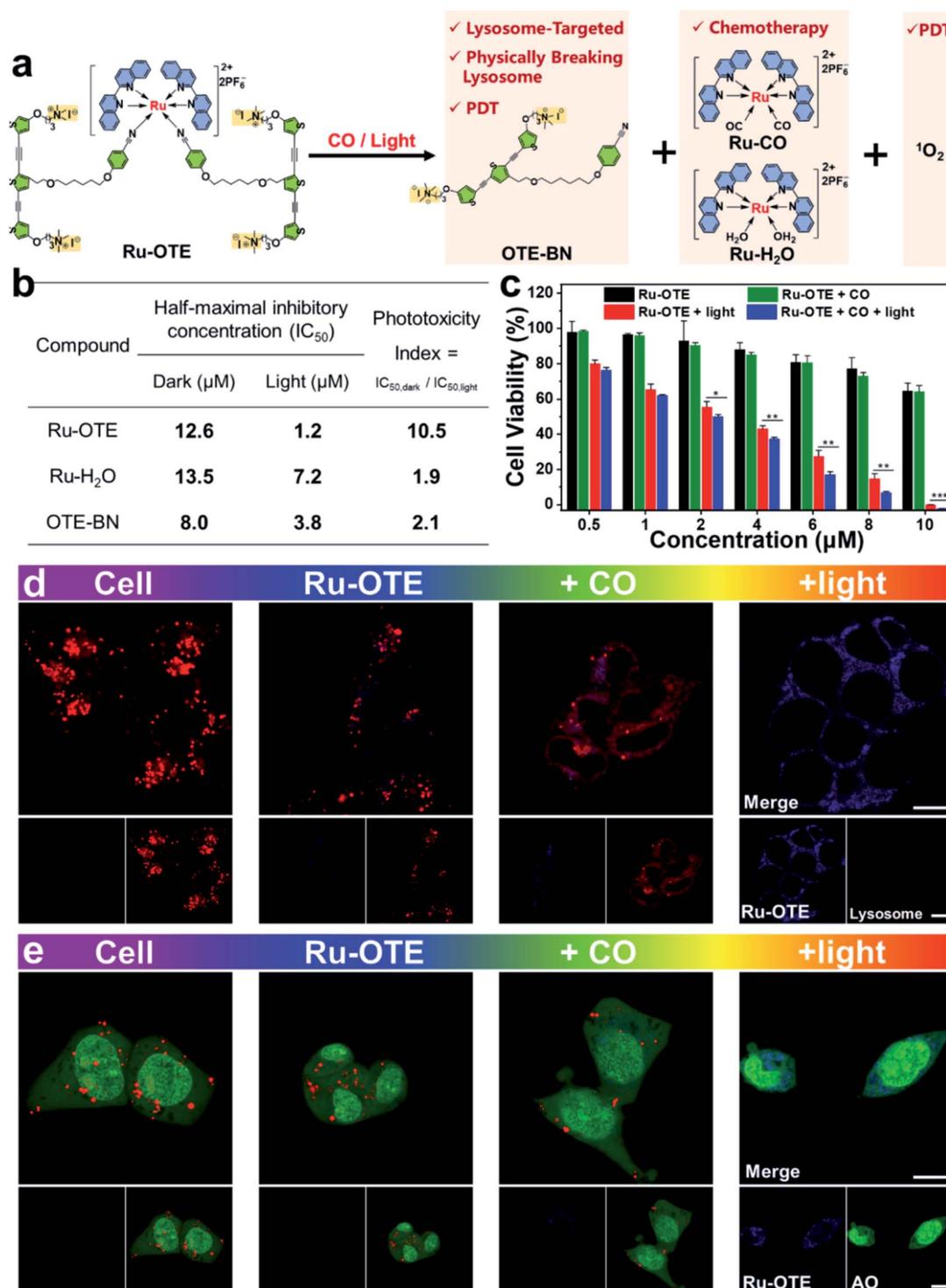


Fig. 3 Schematic illustration, anticancer efficiency and therapeutic mechanism of the CO/light dual-activatable Ru-OTE system. (a) Illustration of the chemical changes after CO/light dual-activation and explanation of the multimodal anticancer therapeutics. (b) Half-maximal inhibitory concentration (IC₅₀) values of Ru-OTE, Ru-H₂O and OTE-BN against MDA-MB-231 cells in dark or white light irradiation conditions. (c) Cytotoxicity of Ru-OTE against MDA-MB-231 in the absence and presence of CO, light, or CO + light. $n = 6$, mean \pm SD, * $P < 0.05$; ** $P < 0.01$; *** $P < 0.001$. (d) CLSM images of MDA-MB-231 cells incubated with Ru-OTE (4.0 μ M) and stained by LysoTracker® Red DND-99 in the absence and presence of CO or light. The fluorescence imaging of Ru-OTE and LysoTracker® Red DND-99 were collected at 420–460 nm (λ_{ex} : 405 nm), 570–670 nm (λ_{ex} : 559 nm), respectively. Scale bar: 10 μ m. (e) CLSM images of MDA-MB-231 cells after incubation with Ru-OTE and stained by AO in the absence and presence of CO or light. Fluorescence imaging of AO was collected at 515–545 nm in the green channel and 610–640 nm in the red channel (λ_{ex} : 488 nm). Scale bar: 10 μ m.



demonstrated slightly stronger cytotoxicity than the Ru-OTE group, which resulted from the partial delivery and good biocompatibility of OTE-BN at lower concentration. When cells were incubated with Ru-OTE and then illuminated with light for 30 min, the cell viability dramatically decreased. Under the conditions of Ru-OTE (8.0 μM), the cell viabilities of Ru-OTE + light group and Ru-OTE + light + CO were $15 \pm 2.0\%$ and $6.9 \pm 0.4\%$, respectively, around an 8% improvement in the cell killing rate. Besides, the data differences of the mentioned two groups were statistically significant. It should be noted that the improved cell killing rates were higher than the data difference between the Ru-OTE group and Ru-OTE + CO. So, the improved cell killing rate was found to be due to the synergistic effect of light/CO, rather than an additive effect, indicating the multimodal anticancer effect of Ru-OTE. Finally, Ru-OTE showed good biocompatibility towards human normal liver HL7702 cell over some concentration ranges (Fig. S7 \dagger), representing potential for use in biomedical applications.

Localization and anticancer mechanism of Ru-OTE

To determine the anticancer mechanism of Ru-OTE, cell colocalization experiments were conducted accordingly. MDA-MB-231 cells were cultured with Ru-OTE at 37 $^{\circ}\text{C}$ for 6 h, then they were dyed with LysoTracker $^{\circledR}$ Red DND-99. As shown in Fig. 3d and S8a, \dagger after CO activation, blue fluorescence started to be emitted and the red fluorescence of the lysosome tracker weakened compared to the cell and Ru-OTE groups, indicating that OTE-BN was released and interacted with lysosomes, leading to the physical disruption of lysosomal structures. Notably, when the cells (incubated with Ru-OTE) were illuminated with white light, bright luminescence in the blue channel and a negligible signal in the red channel were observed, implying that more OTE-BN was photoreleased and cytotoxic $^1\text{O}_2$ almost completely destroyed the lysosomes in the cells. To confirm these two phenomena, the synthesized OTE-BN was incubated with cells for cell imaging. Fig. S9 \dagger showed that the fluorescence signal in the red channel became weaker after incubation with OTE-BN. Importantly, the red fluorescence even disappeared completely upon white light irradiation. However, when the cells were incubated with Ru- H_2O , the bright red fluorescence of the lysosome tracker was observed in the absence of irradiation. The red color still remained, even after illumination with light (Fig. S8a \dagger), which demonstrated that Ru- H_2O has no influence on the structure of the lysosomes. These results confirm that the CO/light dual-activation of Ru-OTE was lysosome-targeted and the disruption of lysosomal structure mainly resulted from the released OTE-BN and its generated $^1\text{O}_2$. The ROS generation ability of Ru-OTE within tumor cancers was verified by incubating it with 2',7'-dichlorofluorescein diacetate (DCFH-DA) probe. After white light irradiation for 30 min, intracellular green emission originating from the oxidation of DCFH-DA by $^1\text{O}_2$ was observed, while no red fluorescence of LysoTracker was observed (Fig. S10 \dagger), which further suggested the generation of $^1\text{O}_2$ inside the cells *via* activation of OTE-BN, leading to the breakage of lysosomes.

To gain deep insight into the influence that Ru-OTE has on the integrity of lysosomes, acridine orange (AO) staining was subsequently performed.^{46,47} Generally, AO emits intense red fluorescence in acidic lysosomes, while it generates green fluorescence in cytoplasm and nuclei. As shown in Fig. 3e and S8b, \dagger when cells were incubated with Ru-OTE and dyed with AO, a large number of red dots in the cytoplasm and green fluorescence in the nuclei were observed, validating the existence of intact lysosomes. After CO activation, there were fewer red dots, while the red dots disappeared after light activation, implying the effective disruption of the lysosomal structures in MDA-MB-231. Moreover, Fig. S11 \dagger shows that there were fewer red dots in the cells incubated with OTE-BN and these even completely disappeared after light irradiation because of the production of $^1\text{O}_2$. The existence of red dots in the cells incubated with Ru- H_2O in the presence or absence of light irradiation confirmed the integrity of the lysosomes (Fig. S8b \dagger). According to the results, it can be concluded that CO/light activates Ru-OTE and triggers the release of the OTE-BN and Ru(II) species. OTE-BN specifically accumulates in lysosomes and physically breaks lysosomes, and the generated $^1\text{O}_2$ completely destroys lysosome structures and leads to the uncontrolled release of lysosomal proteases, thus inducing cell apoptosis.^{14,48}

NIR two-photon activatable PDT

To further expand the bioapplications of the CO/light dual-activatable Ru-OTE system for anticancer therapeutics and increase the tissue penetration depth of the used light, a two-photon laser ($\lambda = 800$ nm) was applied for NIR light activatable synergistic PDT. As shown in Fig. 4a, in the Ru-OTE treated MDA-MB-231 cell group, the two-photon fluorescence signal was clearly observed in the selected area after two-photon laser

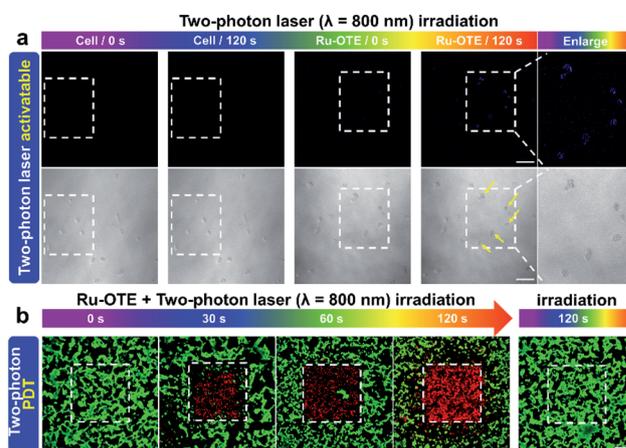


Fig. 4 NIR two-photon laser activatable Ru-OTE for PDT of cancer cells. (a) Two-photon laser activatable and fluorescence images of MDA-MB-231 cells treated with Ru-OTE (4 μM) in a selected area. Scale bar: 100 μm . (b) Live/dead staining of MDA-MB-231 cells treated with Ru-OTE (4 μM) for 6 h and then irradiated using a two-photon laser ($\lambda = 800$ nm) for 0, 30, 60, and 120 s, respectively. The laser irradiation area was 600 $\mu\text{m} \times 600$ μm . The fluorescence images of Calcein AM and PI were collected at 500–540 nm (λ_{ex} : 488 nm) and 570–620 nm (λ_{ex} : 559 nm), respectively. Scale bars: 200 μm .



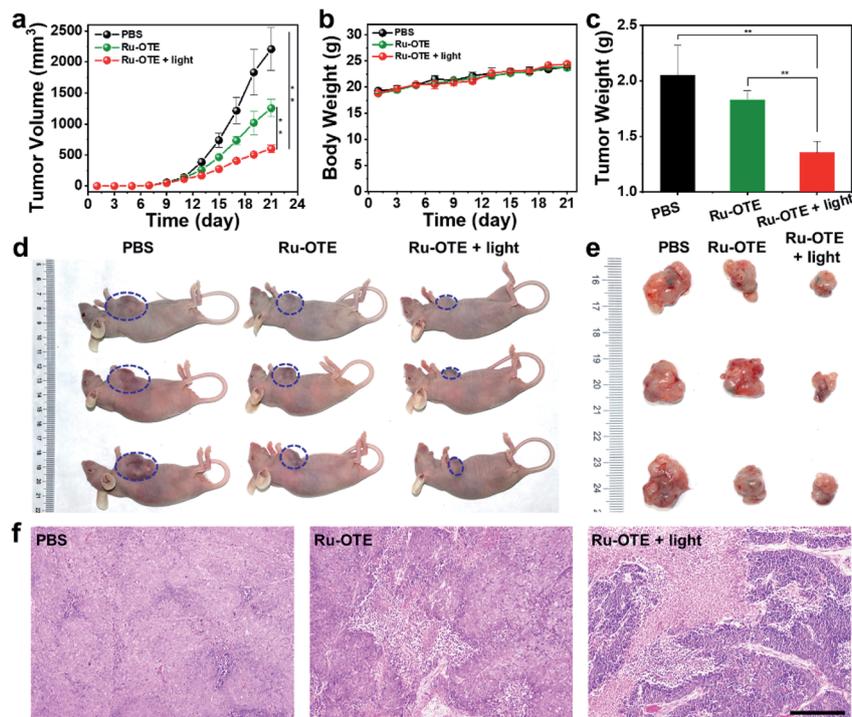


Fig. 5 Evaluation of the *in vivo* anticancer efficiency of the Ru-OTE system in MDA-MB-231 tumor-bearing BALB/c-nude mice. (a) Tumor volume, (b) body weight, and (c) tumor weight variations of the mice with various treatments. $n = 4$, mean \pm SD, * $P < 0.05$; ** $P < 0.01$; *** $P < 0.001$. (d) Representative photographs of the mice and (e) tumors extracted from the mice in the various groups at the end of the treatments. (f) Hematoxylin and eosin (H&E)-stained tumor sections from tumor-bearing mice at the end of treatment. Scale bar: 300 μ m.

($\lambda = 800$ nm) irradiation, while no signal was detected in the unirradiated region, confirming that OTE-BN was successfully two-photon-activated followed by its release in a precise optically-controlled manner. The cells without Ru-OTE treatment did not emit blue fluorescence neither in the presence nor absence of laser irradiation. Importantly, the morphology of the cells after selective laser irradiation was indeed destroyed, implying the NIR PDT potential of Ru-OTE. Furthermore, the two-photon PDT of Ru-OTE was evaluated by live/dead cell staining after two-photon laser ($\lambda = 800$ nm) irradiation. As shown in Fig. 4b, the cells incubated with Ru-OTE without laser irradiation and only irradiated with the laser for 120 s were almost kept alive. However, Ru-OTE treated cells upon laser irradiation for 30, 60 and 120 s gradually died and all the cells in the selected region were killed after 120 s of laser irradiation, confirming that the Ru-OTE system could be directly activated by a two-photon laser for efficient PDT of cancer cells.

In vivo anticancer therapy

In vivo tumor suppression experiments were carried out to further evaluate the validity of Ru-OTE based on a MDA-MB-231 breast tumor model. The tumor-bearing mice were divided into three groups: PBS as a control, Ru-OTE, and Ru-OTE with white light irradiation. As shown in Fig. 5a, mice in the Ru-OTE group exhibited fractional tumor suppression due to the partial release of OTE-BN from Ru-OTE under CO activation. Expectedly, the Ru-OTE with light irradiation (10 mw cm^{-2}) group

demonstrated significant hindrance of tumor growth, in which the tumor volume was suppressed greatly over 14 days. The average tumor weight in the Ru-OTE + light group was much lighter than that in the other two groups (Fig. 5c). The superior antitumor effect of Ru-OTE in the presence of light was mainly due to synergistic PDT and chemotherapy by CO/light dual-activation. Representative photos of the mice and tumors extracted from the mice also confirmed these results (Fig. 5d and e). Moreover, the body weight was monitored for these groups to estimate toxicity. As shown in Fig. 5b, no obvious fluctuations in body weight were observed for all of the groups, indicating the negligible side effects of Ru-OTE and white light irradiation. To further verify the dramatic therapeutic efficiency, histological hematoxylin and eosin (H&E) staining assays were carried out after 14 days of treatment (Fig. 5f). The results showed large areas of apoptosis and necrosis, indicating that Ru-OTE exposure to light irradiation exhibits remarkable therapeutic effectiveness.

Conclusions

In summary, a CO/light dual-activatable Ru-OTE agent for lysosome-targeted multimodal cancer therapeutics was firstly reported. Upon the dual-triggering of CO and light, Ru-OTE undergoes ligand substitution and releases OTE-BN, followed by dramatic fluorescence recovery for monitoring drug delivery. Multimodal cancer therapeutics was achieved relying on the following models: (1) the released OTE-BN selectively



accumulated in the lysosome, physically breaking its integrity. (2) Then, the generated cytotoxic $^1\text{O}_2$ causes severe lysosome damage, thus leading to cancer cell death *via* PDT. (3) The release of the Ru(II) agent also suppressed cancer cell growth as an anticancer metal drug. (4) Importantly, Ru-OTE was directly photo-activated using a two-photon laser (800 nm) for efficient NIR PDT. This study demonstrates that the development of a dual-activatable Ru(II)-conjugated oligomer potential drug provides a new strategy for highly precise and effective subcellular organelle-targeted multimodal cancer therapeutics.

Author contributions

M. Y. and Y. L. T. conceived the project. M. Y., H. Z., Z. Q. Z., Q. Y. and Q. F. performed the experiments. Y. L. T. supervised the research. M. Y., H. Z. and Y. L. T. wrote and revised the paper. X. R. D., S. W. and Y. L. T. validate, review and edit the writing. All the authors discussed the results and contributed to the preparation of the manuscript.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was financially supported by the National Natural Science Foundation of China (Grants 21974084 and 21675106), the Innovation Capability Support Program of Shaanxi (Program no. 2021TD-42), and the Fundamental Research Funds for the Central Universities (No. GK201901003, GK202101001). Animal experiments were approved by the China Committee for Research and Animals Ethics in compliance with the law on experimental animals.

Notes and references

- 1 F. Bray, J. Ferlay, I. Soerjomataram, R. L. Siegel, L. A. Torre and A. Jemal, *Ca-Cancer J. Clin.*, 2018, **68**, 394–424.
- 2 L. Zeng, P. Gupta, Y. Chen, E. Wang, L. Ji, H. Chao and Z. S. Chen, *Chem. Soc. Rev.*, 2017, **46**, 5771–5804.
- 3 N. Rubio, I. Coupienne, E. Di Valentin, I. Heirman, J. Grooten, J. Piette and P. Agostinis, *Autophagy*, 2012, **8**, 1312–1324.
- 4 H. Chen, S. Li, M. Wu, K. Huang, C.-S. Lee and B. Liu, *Angew. Chem., Int. Ed.*, 2020, **59**, 632–636.
- 5 P. Agostinis, K. Berg, K. A. Cengel, T. H. Foster, A. W. Girotti, S. O. Gollnick, S. M. Hahn, M. R. Hamblin, A. Juzeniene, D. Kessel, M. Korbek, J. Moan, P. Mroz, D. Nowis, J. Piette, B. C. Wilson and J. Golab, *Ca-Cancer J. Clin.*, 2011, **61**, 250–281.
- 6 J. P. Celli, B. Q. Spring, I. Rizvi, C. L. Evans, K. S. Samkoe, S. Verma, B. W. Pogue and T. Hasan, *Chem. Rev.*, 2010, **110**, 2795–2838.
- 7 D. E. Dolmans, D. Fukumura and R. K. Jain, *Nat. Rev. Cancer*, 2003, **3**, 380–387.
- 8 S. Gao, G. Wang, Z. Qin, X. Wang, G. Zhao, Q. Ma and L. Zhu, *Biomaterials*, 2017, **112**, 324–335.
- 9 J. Sun, K. Du, J. J. Diao, X. T. Cai, F. D. Feng and S. Wang, *Angew. Chem., Int. Ed.*, 2020, **59**, 12122–12128.
- 10 D. Peer, J. M. Karp, S. Hong, O. C. Farokhzad, R. Margalit and R. Langer, *Nat. Nanotechnol.*, 2007, **2**, 751–760.
- 11 G. C. Yu, X. L. Zhao, J. Zhou, Z. W. Mao, X. L. Huang, Z. T. Wang, B. Hua, Y. J. Liu, F. W. Zhang, Z. M. He, O. Jacobson, C. Y. Gao, W. L. Wang, C. Y. Yu, X. Y. Zhu, F. H. Huang and X. Y. Chen, *J. Am. Chem. Soc.*, 2018, **140**, 8005–8019.
- 12 T. Senthilkumar, L. Zhou, Q. Gu, L. Liu, F. Lv and S. Wang, *Angew. Chem., Int. Ed.*, 2018, **57**, 13114–13119.
- 13 V. P. Torchilin, *Nat. Rev. Drug Discovery*, 2014, **13**, 813–827.
- 14 L. Galluzzi, J. M. Bravo-San Pedro and G. Kroemer, *Nat. Cell Biol.*, 2014, **16**, 728–736.
- 15 C. J. Zhang, Q. L. Hu, G. X. Feng, R. Y. Zhang, Y. Y. Yuan, X. M. Lu and B. Liu, *Chem. Sci.*, 2015, **6**, 4580–4586.
- 16 A. Jhaveri and V. Torchilin, *Expert Opin. Drug Delivery*, 2016, **13**, 49–70.
- 17 R. Villasenor, Y. Kalaidzidis and M. Zerial, *Curr. Opin. Cell Biol.*, 2016, **39**, 53–60.
- 18 Z. X. Zhou, J. P. Liu, J. J. Huang, T. W. Rees, Y. L. Wang, H. Wang, X. P. Li, H. Chao and P. J. Stang, *Proc. Natl. Acad. Sci. U. S. A.*, 2019, **116**, 20296–20302.
- 19 H. Y. Huang, B. L. Yu, P. Y. Zhang, J. J. Huang, Y. Chen, G. Gasser, L. N. Ji and H. Chao, *Angew. Chem., Int. Ed.*, 2015, **54**, 14049–14052.
- 20 W. K. Martins, N. F. Santos, C. D. Rocha, I. O. L. Bacellar, T. M. Tsubone, A. C. Viotto, A. Y. Matsukuma, A. B. D. Abrantes, P. Siani, L. G. Dias and M. S. Baptista, *Autophagy*, 2019, **15**, 259–279.
- 21 M. R. Gill, P. J. Jarman, S. Halder, M. G. Walker, H. K. Saeed, J. A. Thomas, C. Smythe, K. Ramadan and K. A. Vallis, *Chem. Sci.*, 2018, **9**, 841–849.
- 22 N. Dai, H. Zhao, R. L. Qi, Y. Y. Chen, F. T. Lv, L. B. Liu and S. Wang, *Chem.–Eur. J.*, 2020, **26**, 4489–4495.
- 23 C. Mari, V. Pierroz, S. Ferrari and G. Gasser, *Chem. Sci.*, 2015, **6**, 2660–2686.
- 24 W. Sun, S. Li, B. Haupler, J. Liu, S. Jin, W. Steffen, U. S. Schubert, H. J. Butt, X.-J. Liang and S. Wu, *Adv. Mater.*, 2017, **29**, 1603702.
- 25 J. Liu, Y. Chen, G. Li, P. Zhang, C. Jin, L. Zeng, L. Ji and H. Chao, *Biomaterials*, 2015, **56**, 140–153.
- 26 L. Kohler, L. Nease, P. Vo, J. Garofolo, D. K. Heidary, R. P. Thummel and E. C. Glazer, *Inorg. Chem.*, 2017, **56**, 12214–12223.
- 27 R. Li, C. Zhang, B. Xie, W. Yu, W. Qiu, H. Cheng and X. Zhang, *Biomaterials*, 2019, **194**, 84–93.
- 28 J. D. Knoll and C. Turro, *Coord. Chem. Rev.*, 2015, **282**, 110–126.
- 29 X. Zeng, Y. Wang, J. Han, W. Sun, H.-J. Butt, X.-J. Liang and S. Wu, *Adv. Mater.*, 2020, **32**, 2004766.
- 30 B. Wang, B. N. Queenan, S. Wang, K. P. R. Nilsson and G. C. Bazan, *Adv. Mater.*, 2019, **31**, 1806701–1806722.
- 31 B. Wang, M. Wang, A. Mikhailovsky, S. Wang and G. C. Bazan, *Angew. Chem., Int. Ed.*, 2017, **56**, 5031–5034.



- 32 D. G. Whitten, Y. Tang, Z. Zhou, J. Yang, Y. Wang, E. H. Hill, H. C. Pappas, P. L. Donabedian and E. Y. Chi, *Langmuir*, 2019, **35**, 307–325.
- 33 X. Duan, X. F. Jiang, D. Hu, P. Liu, S. Li, F. Huang, Y. Ma, Q. H. Xu and Y. Cao, *Nanoscale*, 2018, **11**, 185–192.
- 34 Q. Feng, Z. Q. Zhang, Q. Yuan, M. Yang, C. Zhang and Y. L. Tang, *Sens. Actuators, B*, 2020, **312**, 127981.
- 35 L. Zhou, F. Lv, L. Liu and S. Wang, *Acc. Chem. Res.*, 2019, **52**, 3211–3222.
- 36 Q. Yuan, Y. Zhao, Z. Zhang and Y. Tang, *ACS Appl. Mater. Interfaces*, 2021, **13**, 257–265.
- 37 C. Torre, A. Toscani, C. Marin-Hernandez, J. A. Robson, M. C. Terencio, A. J. P. White, M. J. Alcaraz, J. Wilton-Ely, R. Martinez-Manez and F. Sancenon, *J. Am. Chem. Soc.*, 2017, **139**, 18484–18487.
- 38 L. Wu and R. Wang, *Pharmacol. Rev.*, 2005, **57**, 585–630.
- 39 J. Karges, J. Li, L. Zeng, H. Chao and G. Gasser, *ACS Appl. Mater. Interfaces*, 2020, **12**, 54433–54444.
- 40 W. Sun, Y. Wen, R. Thiramanas, M. Chen, J. Han, N. Gong, M. Wagner, S. Jiang, M. S. Meijer, S. Bonnet, H. J. Butt, V. Mailänder, X. J. Liang and S. Wu, *Adv. Funct. Mater.*, 2018, **28**, 1804227.
- 41 Z. Deng, P. Gao, L. Yu, B. Ma, Y. You, L. Chan, C. Mei and T. Chen, *Biomaterials*, 2017, **129**, 111–126.
- 42 G. Feng, Y. Fang, J. Liu, J. Geng, D. Ding and B. Liu, *Small*, 2017, **13**, 1602807.
- 43 Y. Zhao, Z. Zhang, Z. Lu, H. Wang and Y. Tang, *ACS Appl. Mater. Interfaces*, 2019, **11**, 38467–38474.
- 44 B. Wang, H. Yuan, Z. Liu, C. Nie, L. Liu, F. Lv, Y. Wang and S. Wang, *Adv. Mater.*, 2014, **26**, 5986–5990.
- 45 B. Wang, S. L. Fronk, Z. D. Rengert, J. Limwongyut and G. C. Bazan, *Chem. Mater.*, 2018, **30**, 5836–5840.
- 46 X. Li, R. R. Tao, L. J. Hong, J. Cheng, Q. Jiang, Y. M. Lu, M. H. Liao, W. F. Ye, N. N. Lu, F. Han, Y. Z. Hu and Y. H. Hu, *J. Am. Chem. Soc.*, 2015, **137**, 12296–12303.
- 47 M. Li, T. Xiong, J. Du, R. Tian, M. Xiao, L. Guo, S. Long, J. Fan, W. Sun, K. Shao, X. Song, J. W. Foley and X. Peng, *J. Am. Chem. Soc.*, 2019, **141**, 2695–2702.
- 48 L. He, Y. Li, C. P. Tan, R. R. Ye, M. H. Chen, J. J. Cao, L. N. Ji and Z. W. Mao, *Chem. Sci.*, 2015, **6**, 5409–5418.

