

Cite this: *Chem. Sci.*, 2021, **12**, 6406

All publication charges for this article have been paid for by the Royal Society of Chemistry

Received 18th February 2021
Accepted 25th March 2021

DOI: 10.1039/d1sc00972a
rsc.li/chemical-science

Introduction

Enantioenriched amines are among the most important structural motifs in the pharmaceutical industry.¹ It has been estimated that chiral amines are substructures in 40% of current pharmaceuticals.^{2,3} Among amine-containing molecules, allylic amines are an important sub-class in the pharmaceutical industry (cruentaren B, naftifine, terbinafine and abamine, Fig. 1). Moreover, allylic amines are fundamental building blocks in synthesis.^{4,5} Enantioenriched allylic amines, however, are often difficult to synthesize using asymmetric catalysis.^{6,7}

In recent years, practical synthetic routes toward enantioenriched amines have been advanced by Ellman,^{8–10} Carreira^{11–15} and others.^{16–21} Enantioenriched allylic amines are desirable targets because of their utility, and several methods have been reported that involve the asymmetric addition of organometallic reagents to activated imines in the presence of enantioenriched catalysts. Early work on the rhodium catalyzed asymmetric arylation of imines by Hayashi's group²² inspired the use of vinyl trifluoroborates, as exemplified by the work of

Lin and Wu (Scheme 1a).^{23,24} Other approaches based on inexpensive metals, such as Trost's alkyne hydrozirconation followed by zinc-Pro-phenol-based catalyzed asymmetric addition to *N*-Boc activated aldimines have received attention (Scheme 1b).²⁵ An impressive asymmetric imine vinylation reaction was reported by Krische starting with imine and alkyne in the presence of hydrogen and a chiral iridium-based catalyst (Scheme 1c).²⁶ The intermediate vinyl iridium species is diverted from hydrogenation to the asymmetric vinylation process. Other interesting approaches have also been documented.²⁷

With an interest in the synthesis of amines, several groups have focused on the Umpolung reactivity of *N*-benzyl ketimine derivatives. Upon deprotonation under mild conditions, *N*-benzyl ketimines form 2-azaallyl anions that can be functionalized in transition metal catalyzed processes, or under transition metal-free conditions, to provide various amines.^{28,29} This strategy of deprotonation of *N*-benzyl ketimines to generate intermediate 2-azaallyl anions as reactive nucleophiles benefits from its avoidance of preformed organometallics that are common reagents in C–C bond forming reactions through cross-coupling processes.

^aKey Laboratory of Medicinal Chemistry for Natural Resource, Ministry of Education, Yunnan Provincial Center for Research & Development of Natural Products, School of Chemical Science and Technology, Yunnan University, Kunming, 650091, P. R. China. E-mail: xdyang@ynu.edu.cn; zhanghb@ynu.edu.cn

^bRoy and Diana Vagelos Laboratories, Penn/Merck Laboratory for High-Throughput Experimentation, Department of Chemistry, University of Pennsylvania, 231 South 34th Street, Philadelphia, PA, USA. E-mail: pwalsh@sas.upenn.edu; liminyan@sas.upenn.edu

† Dedicated to Prof. Madeleine Joullié (Penn) on her birthday for many years of inspirational leadership in chemistry.

‡ Electronic supplementary information (ESI) available: CCDC 2058299. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/d1sc00972a

§ These authors contributed equally to this work.

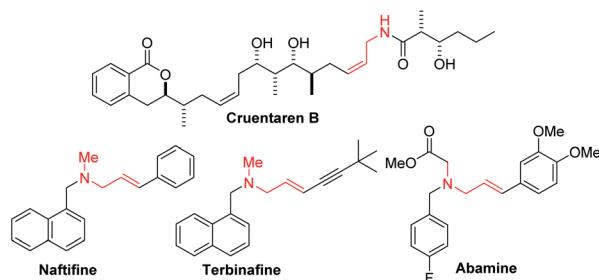
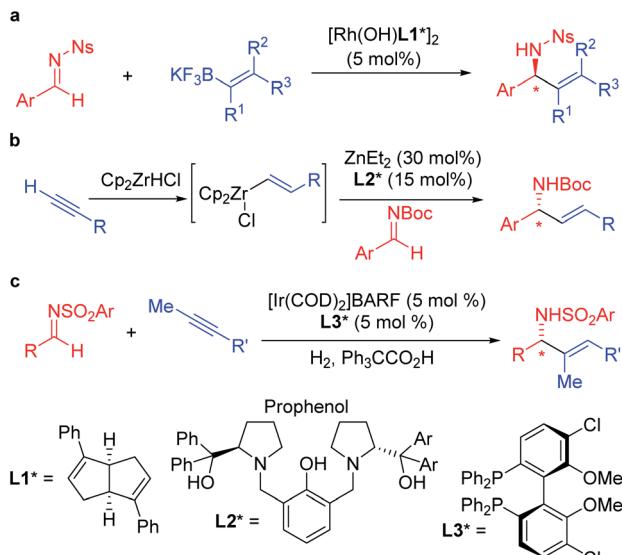


Fig. 1 Examples of allylic amine-containing pharmaceuticals.

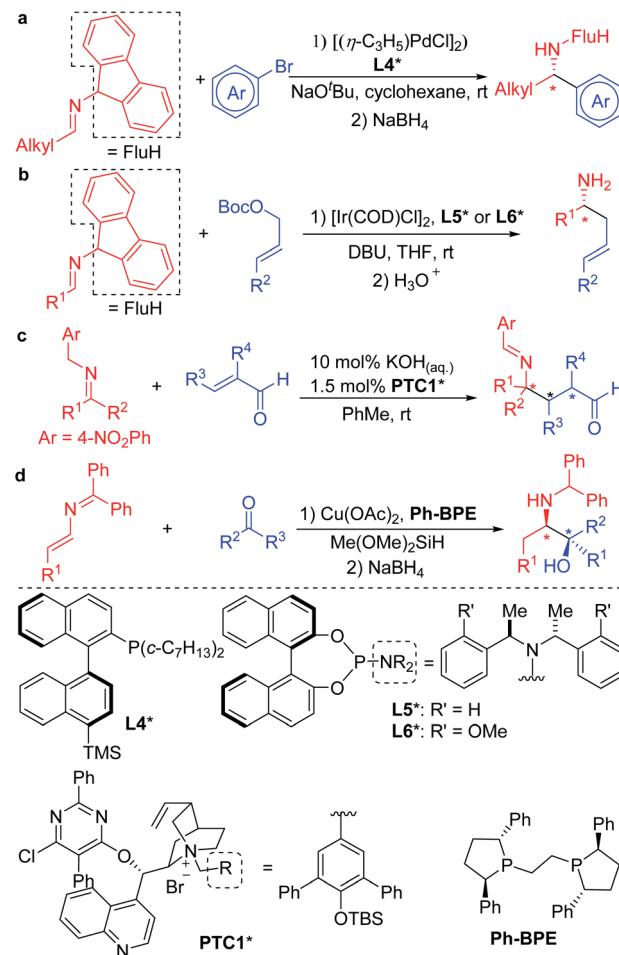




Scheme 1 Enantioenriched allylic amine synthesis via asymmetric additions to imines.

The mild nature of semi-stabilized 2-azaallyl anions has made them targets for use in enantioselective functionalization reactions. Successful examples include Buchwald and Zhu's pioneering enantioselective Pd catalyzed arylation of alkyl 2-azaallyl anions with tailored chiral phosphine **L4*** (Scheme 2a).³⁰ Enantioselective allylic substitution with 2-azaallyl anion nucleophiles has attracted the attention of groups including Niu,^{31–34} Chruma,^{35,36} You,^{37–41} and Han.⁴² Among these, the iridium catalyzed asymmetric allylic substitution with ligands **L5*** and **L6*** stand out as highly enantioselective (Scheme 2b).²⁹ Deng and coworkers^{43–46} reported an impressive functionalization of trifluoromethyl amines using asymmetric conjugate additions (Scheme 2c).⁴³ Here, the 4-nitro group proved essential to stabilize the 2-azaallyl anion, enabling the deprotonation with KOH in the presence of phase-transfer catalyst **PTC1***. The nitrobenzyl moiety is also likely responsible for the regioselectivity of the C-3 functionalization, which results in the formation of quaternary stereocenters. A novel strategy was employed by the Malcolmson's group^{47,48} who started with 2-azadienes and an enantioenriched copper catalyst. Hydrocupration generates an enantioenriched copper complex with the bound 2-azaallyl anion that adds in an enantioselective fashion to the carbonyl group (Scheme 2d).⁴⁷

Since 2014, our group has accessed a wide variety of diarylmethylamines through the functionalization of 2-azaallyl anions (Scheme 3a).^{49,50} We also discovered the unique reducing feature of 2-azaallyl anions and developed a series of methods for the efficient transition metal-free synthesis of aryl-, alkyl- and allyl-methylamines from 2-azaallyl radicals (Scheme 3b).^{51–54} Herein, we continue our journey in 2-azaallyl chemistry by developing the first enantioselective nickel-catalyzed vinylation of 2-azaallyl anions (Scheme 3c). Successful identification of $\text{Ni}(\text{COD})_2$ /chiraphos is key for the enantioselectivity. A wide range of imines and vinyl bromides are tolerated under the mild



Scheme 2 Chiral amine synthesis from enantioselective functionalization of 2-azaallyl anions.

reaction conditions with no C-3 vinylation⁵¹ or base promoted product isomerization observed. We also conducted a telescoped gram scale synthesis and product derivatization study to demonstrate the scalability and synthetic potential of the current method. It is noteworthy that the methods developed by Buchwald's and Niu's groups (Scheme 2) involve expensive precious metals and/or ligands and are not suitable for the synthesis of the enantioenriched allylic amines reported herein.

Results and discussion

Reaction development and optimization

Prompted by our previous experience with benzophenone *N*-benzyl imines, which readily participate in single electron processes upon deprotonation (Scheme 3b), the Pd catalyzed 2-azaallyl anion arylation^{49,50,55,56} (Scheme 3a) and Ni catalyzed cross-coupling^{57–61} chemistry, we selected $\text{Ni}(\text{COD})_2$ as the nickel source and the mild base LiO^+Bu to deprotonate the imine **1a**. We wanted to avoid complication by SET steps and radical intermediates, as proposed by Ohshima's team in their recent copper-catalyzed coupling reactions synthesis of hindered amino acids using 2-azaallyl anions.⁶² To lower the



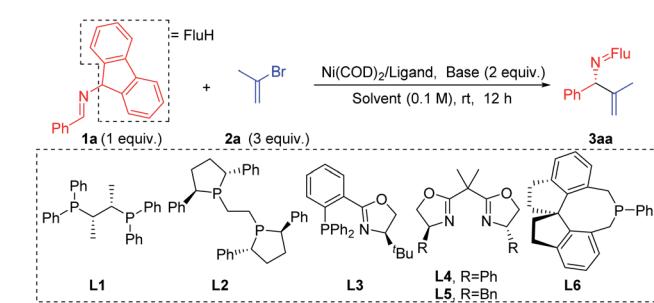
reducing tendencies of the 2-azaallyl anion intermediates, we opted to use fluorenyl amine derivatives.

We began to explore this reaction by examining 26 chiral ligands (see ESI, Table S1‡ for details) under the conditions listed in Table 1. The top hits, as judged by product enantiomeric excess, were observed with Fryzuk and Bosnich's (*S,S*)-chiraphos⁶³ (**L1**, 78% ee, 65% yield, entry 1) and Ph-BPE (**L2**, 47% ee, 52% yield, entry 2). We found that a phosphine-oxazoline ligand **L3** afforded the target product **3aa** in 55% ee and 47% yield. When BOX ligands **L4** [2,2'-(propane-2,2-diyl)bis(4-phenyl-4,5-dihydrooxazole)] and **L5** [2,2'-(propane-2,2-diyl)bis(4-benzyl-4,5-dihydrooxazole)] were used we observed formation of **3aa** in 66% and 75% yield, respectively, but were surprised to find that both gave racemic product. In addition to bidentate ligands, mono-dentate phosphine ligand **L6** afforded 45% ee, albeit in 34% yield. Based on these results, we continued to use chiraphos (**L1**), which afforded the highest product ee and yield in the initial screen.

The next variable examined in the optimization was the base. At the outset of this work, we were concerned that a base that could deprotonate the aldimine substrate, might also deprotonate the product, as was observed by Ohshima,⁶² resulting in

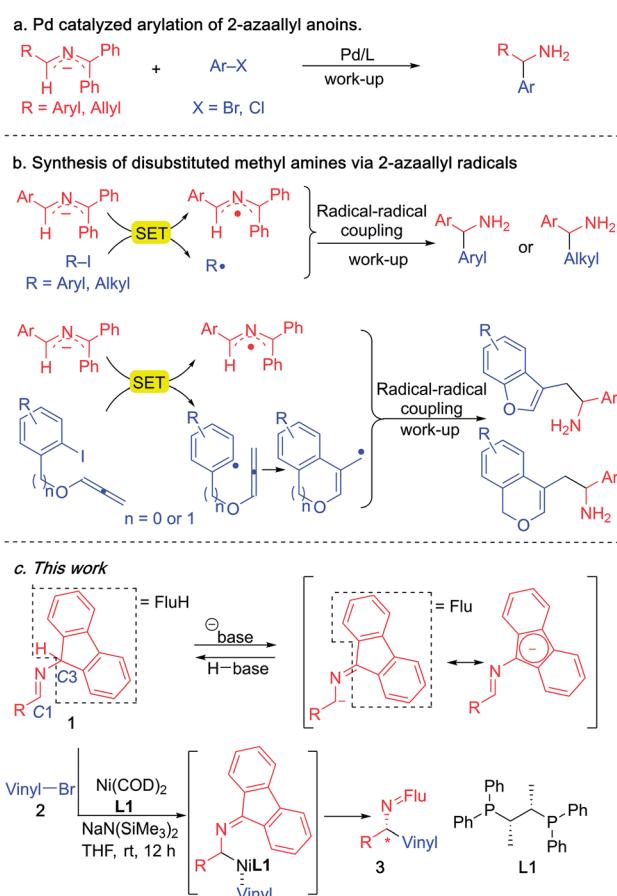
product racemization and possibly partial isomerization. We were also cognizant that transmetalation would likely be the enantiodetermining step and, if true, the nature of the main group metal associated with the 2-azaallyl anion would be important. We tested 5 different bases that could deprotonate the aldimine [NaO^tBu , KO^tBu , $\text{LiN}(\text{SiMe}_3)_2$, $\text{NaN}(\text{SiMe}_3)_2$, $\text{KN}(\text{SiMe}_3)_2$, entries 7–11]. We were delighted to discover that $\text{NaN}(\text{SiMe}_3)_2$ provided the desired product in 94% yield with 92% ee (Table 1, entry 10). We then turned our attention to probing the impact of the solvent. Three solvents were evaluated [(CPME (cyclopentyl methyl ether), MTBE (methyl *tert*-butyl ether), and diethyl ether (entries 12–14)], however none of these rivaled the results with THF in entry 10. Dropping the equivalents of base from 2 to 1.5 led to a slight increase in the yield to 95% and the product ee to 93% (entry 15). An attempt to lower the catalyst loading to 2.5 mol% afforded a synthetically acceptable yield of 62% and high enantioselectivity (92% ee, entry 16). Notably, under otherwise identical conditions to entry 15, switching from $\text{Ni}(\text{COD})_2$ to $\text{Pd}(\text{OAc})_2$ significantly decreased the yield (4%, entry 17, see ESI, Table S5‡ for details).

Table 1 Optimization of vinylation of imine **1a**^{a,b}



Entry	L	Ni/L (mol%)	Base	Solvent	3aa (%)	ee (%)
1	L1	5/10	LiO^tBu	THF	65	78
2	L2	5/10	LiO^tBu	THF	52	47
3	L3	5/10	LiO^tBu	THF	47	55
4	L4	5/10	LiO^tBu	THF	66	0
5	L5	5/10	LiO^tBu	THF	75	0
6	L6	5/10	LiO^tBu	THF	34	45
7	L1	5/10	NaO^tBu	THF	23	23
8	L1	5/10	KO^tBu	THF	6	—
9	L1	5/10	$\text{LiN}(\text{SiMe}_3)_2$	THF	63	84
10	L1	5/10	$\text{NaN}(\text{SiMe}_3)_2$	THF	94	92
11	L1	5/10	$\text{KN}(\text{SiMe}_3)_2$	THF	42	82
12	L1	5/10	$\text{NaN}(\text{SiMe}_3)_2$	CPME	62	48
13	L1	5/10	$\text{NaN}(\text{SiMe}_3)_2$	MTBE	28	20
14	L1	5/10	$\text{NaN}(\text{SiMe}_3)_2$	Et_2O	32	14
15 ^c	L1	5/10	$\text{NaN}(\text{SiMe}_3)_2$	THF	95	93
16	L1	2.5/5	$\text{NaN}(\text{SiMe}_3)_2$	THF	62	92
17 ^d	L1	5/10	$\text{NaN}(\text{SiMe}_3)_2$	THF	4	—

^a Reactions conducted on a 0.2 mmol scale with 2 equiv. base. ^b Isolated yield of **3aa** after chromatographic purification; ee (enantiomeric excess) of **3aa** was determined by chiral phase HPLC. ^c $\text{NaN}(\text{SiMe}_3)_2$ (1.5 equiv.). ^d $\text{Pd}(\text{OAc})_2$ instead of $\text{Ni}(\text{COD})_2$.



Scheme 3 Reaction of 2-azaallyl anions from our team. (a) Pd catalyzed racemic arylation of 2-azaallyl anions. (b) Transition metal-free radical coupling reactions. (c) This work, the enantioselective vinylation of 2-azaallyl anions.

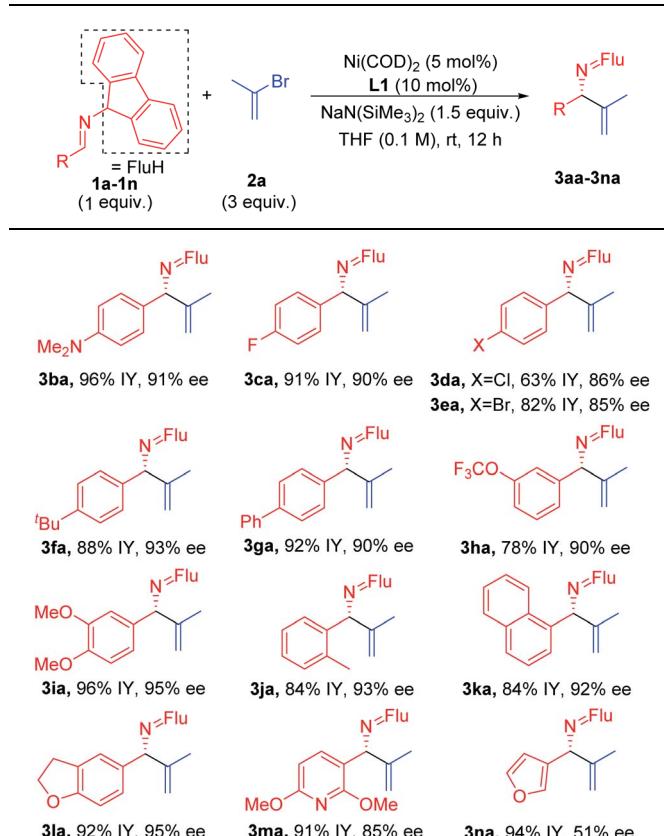


It is interesting to note that various groups,^{35,36} including ours,^{49–51} observed regioselectivity issues with 2-azaallyl anions, wherein partial substitution took place at the more hindered C-3 position of the azaallyl group. Regioselectivity issues in the functionalization of 2-azaallyl anions can be problematic in the application of the methods, because the C-1 and C-3 isomers are usually difficult to separate. We were pleased to find that the regioselectivity in our nickel catalyzed process was very high and C-3 products were not observed.

Scope of imines

With the optimized conditions in hand, a range of aldimines were subjected to the nickel catalyzed enantioselective vinylation. As shown in Table 2, isolated yields of the corresponding allylic amine derivatives were generated in >63% and most enantioselectivities are >85%. A range of *para*-substituted aldimines underwent the vinylation with 2-bromopropene regardless of the electronic nature of the substituent. For example, electron rich imine **1b** (4-NMe₂) afforded **3ba** in 96% yield and 91% ee. Despite the high aptitude of nickel complexes to undergo oxidative additions of aryl halides and other C–X bonds,⁶³ imines bearing halogens (**1c**, 4-F; **1d**, 4-Cl; and **1e**, 4-Br) furnished the allylic imine products in 91%, 63% and 82% yield with 90%, 86% and 85% ee, respectively.

Table 2 Scope of aldimines^{a,b,c}



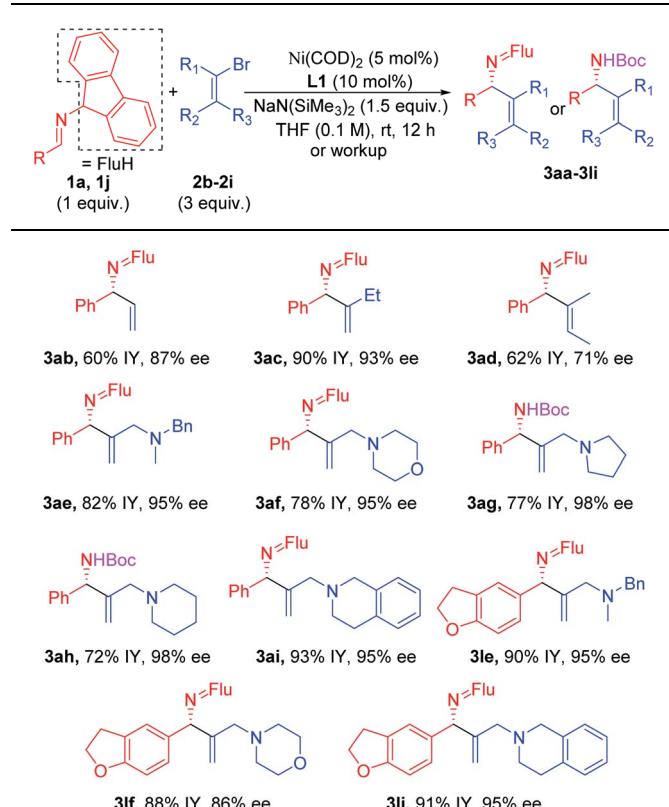
^a Reactions conducted on a 0.4 mmol scale using 1 equiv. **1** and 3 equiv. **2a** at 0.1 M. ^b Isolated yields after chromatographic purification. Flu = 9-fluorenyl. ^c ee's of imine products were determined by chiral phase HPLC analysis.

yields with 90%, 86% and 85% ee, respectively. Thus, the catalyst displays a high degree of chemoselectivity in the oxidative addition of $\text{C}(\text{sp}^2)\text{–Br}$ bonds.

Aldimines possessing electronically neutral substituents, including 4-*t*Bu and 4-Ph, performed well, providing the desired products (**3fa** and **3ga**) in 88% and 92% yields with 93% and 90% ee, respectively. Substrates bearing *meta*-substituents, like 3-OCF₃, resulted in 78% yield of **3ha** with 90% ee. A 3,4-dimethoxy substituted imine (**2i**) led to target product **3ia** in 96% yield with 95% ee. The sterically hindered 2-tolyl imine and 1-naphthyl imine did not impact the catalyst enantioselectivity, giving **3ja** and **3ka** both in 84% yield with 92–93% ee.

Heterocycle-containing structures are of great value to the pharmaceutical industry.^{64,65} With this in mind, selected heterocycles were incorporated into the imine substrates. The dihydrobenzofuran derived imine was converted to the corresponding product **3la** in 92% yield with 95% ee. Pyridines are among the most prevalent heterocycles in medicinal chemistry⁶⁶. To our delight, the pyridyl-based substrate **1m** underwent the vinylation in 91% yield with 85% ee. An imine bearing a 3-furyl group provided the product **3na** in 94% yield, but ee dropped to 51%. We were worried that the product **3na** might have undergone racemization *via* deprotonation by base followed by re-protonation. As such, we monitored the product ee

Table 3 Scope of vinyl bromides^{a,b,c}



^a Reactions conducted on a 0.4 mmol scale using 1 equiv. **1a**, **1j** and 3 equiv. **2** at 0.1 M. ^b Isolated yields after chromatographic purification. Flu = 9-fluorenyl. ^c ee's of imine products were determined by HPLC analysis.



as a function of time by analyzing samples from the reaction at 3.0, 6.0 and 9.0 h. The ee of **3na**, however, remain 51% over the time course of the reaction (see ESI, Table S6‡ for details).

Scope of the vinyl bromide coupling partners

Substituted vinyl bromides possessing aliphatic groups, heterocycles, and extended ring systems were next explored. As shown in Table 3, diverse vinyl bromides were amenable to the asymmetric additions. Use of the parent 1-bromo ethylene (**2b**) enabled the isolation of the vinylation product **3ab** with 87% ee and 60% yield. Replacing the methyl group of 2-bromo propene with an ethyl group (**2c**) did not impact the yield (90%) or the product ee (93%) compared to the model reaction. In contrast, the isomeric *trans*-2-bromo-2-butene (**2d**), containing a trisubstituted alkene, was more challenging and furnished the product in 62% yield with 71% ee.

We next examined vinyl bromide substrates bearing amino groups to prepare diamine derivatives. Thus, coupling of **1a** with *N*-benzyl-2-bromo-*N*-methylprop-2-en-1-amine (**2e**) delivered the diamine derivative **3ae** in 82% yield and 95% ee. Cyclic analogs 4-(2-bromoallyl)morpholine (**2f**), 1-(2-bromoallyl)pyrrolidine (**2g**) and 1-(2-bromoallyl)piperidine (**2h**) were next subjected to the optimized reaction conditions, affording the products in 72–78% yields and 95–98% ee. The efficiency of the reaction was maintained when a vinyl bromide bearing extended ring system on the methylene carbon (**2i**) was

employed, furnishing heterocyclic diamine derivative **3ai** (93% yield, 95% ee).

Imine **1l**, with a heteroaromatic scaffold, was selected for coupling with three vinyl bromides (**2e**, **2f** and **2i**), producing **3le**, **3lf** and **3li** in excellent yields (88–91%) and enantioselectivities (86–95% ee). It is noteworthy that these enantioenriched diamine derivatives would be difficult to prepare by other methods.

Gram scale synthesis and product derivatization

In order for a method to be useful, it must be scalable. To test the scalability of our enantioselective vinylation, we explored the telescoped imine formation/asymmetric vinylation procedure on gram scale (Scheme 4). To our delight, product **3li** was successfully prepared in overall 92% yield (1.33 g) with 93% ee, demonstrating the potential application on larger scales. In order to determine the facial selectivity of the reaction and the absolute configuration of the asymmetric vinylation, we hydrolyzed **3ka** to the parent amine, then re-protected with TsCl to increase the crystallinity. The configuration of product **4ka** was determined to be (*R*) by single crystal X-ray analysis (Scheme 5a, CCDC 2058299‡).

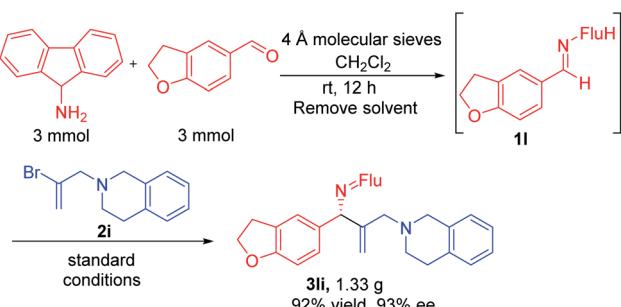
To demonstrate the synthetic utility of the allylic amine products, we explored derivatization. Cyclopropyl amines are common building blocks in the pharmaceutical industry.⁶⁷ Thus, conversion of **3la** to the corresponding sulfonamide **4la** was readily accomplished in 81% yield. Subjecting the resulting sulfonamide to diethylzinc and diiodomethane led to cyclopropyl derivative **5la** in 95% yield (Scheme 5b). Importantly, the ee of **4la** and **5la** were preserved through these transformations. Hydrogenation of the allylic double bond was also conducted using **3la** with Pd/C. The hydrogenated and deprotected amine was isolated in 70% yield. This result bodes well for the synthesis of enantioenriched amines with aliphatic substituents that are otherwise difficult to access but are of great value in pharmaceutical industry (Scheme 5c).

Conclusions

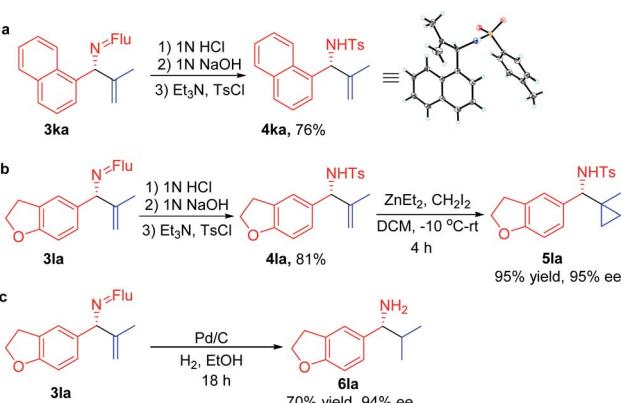
In summary, we describe the first development of a highly enantioselective, convenient and practical vinylation of 2-azaallyl anions. The current method enables the synthesis of a wide variety of enantiomerically enriched allylic amines, including highly functionalized 1,3-diamine derivatives. A telescoped procedure has been introduced that is applicable to the gram scale preparation of a highly enantioenriched allylic 1,3-diamine derivative. The catalyst is based on a commercially available diphosphine, chiraphos, and a widely used nickel source. Overall, this method constitutes a straightforward and practical contribution to the asymmetric functionalization of 2-azaallyl anions.

Author contributions

S. D. & G. D. contributed equally to this work. X. Y. conceived of the project. M. L., H. Z. and P. J. W. designed the experiments. S.



Scheme 4 Gram-scale sequential one-pot imine synthesis/vinylation.



Scheme 5 Transformation of the products.



D., G. D., Y. Z., X. W., X. T. and Z. L. performed the research. M. L., X. Y. and P. J. W. wrote the manuscript.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by grants from the National Key R&D Program of China (2019YFE0109200), NSFC (U1702286), the China Postdoctoral Science Foundation (2019M663581), the Ling-Jun Scholars (202005AB160003) and NSF (2019FY003010 and 2019FI018) of Yunnan Province, the Program for Yun-Ling Scholars and IRTSTYN. P. J. W. thanks the US National Science Foundation (CHE-1902509) for financial support.

Notes and references

- D. C. Blakemore, L. Castro, I. Churcher, D. C. Rees, A. W. Thomas, D. M. Wilson and A. Wood, *Nat. Chem.*, 2018, **10**, 383–394.
- M. Breuer, K. Dittrich, T. Habicher, B. Hauer, M. Kesseler, R. Sturmer and T. Zelinski, *Angew. Chem., Int. Ed.*, 2004, **43**, 788–824.
- D. Ghislieri and N. J. Turner, *Top. Catal.*, 2013, **57**, 284–300.
- P. Merino, S. Anoro, S. Franco, J. M. Gascon, V. Martin, F. L. Merchan, J. Revuelta, T. Tejero and V. Tuñon, *Synth. Commun.*, 2000, **30**, 2989–3021.
- B. M. Trost and M. L. Crawley, *Chem. Rev.*, 2003, **103**, 2921–2943.
- M. Johannsen and K. A. Jørgensen, *Chem. Rev.*, 1998, **98**, 1689–1708.
- S. K. Maurya, M. Dow, S. Warriner and A. Nelson, *Beilstein J. Org. Chem.*, 2013, **9**, 775–785.
- S. Maity, T. J. Potter and J. A. Ellman, *Nat. Catal.*, 2019, **2**, 756–762.
- S. Dongbang and J. A. Ellman, *Angew. Chem., Int. Ed.*, 2021, **60**, 2135–2139.
- M. T. Robak, M. A. Herbage and J. A. Ellman, *Chem. Rev.*, 2010, **110**, 3600–3740.
- M. Roggen and E. M. Carreira, *J. Am. Chem. Soc.*, 2010, **132**, 11917–11919.
- F. Glatz, D. A. Petrone and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2020, **59**, 16404–16408.
- S. Breitler and E. M. Carreira, *J. Am. Chem. Soc.*, 2015, **137**, 5296–5299.
- Y. Sempere, J. L. Alfke, S. L. Rossler and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2019, **58**, 9537–9541.
- S. L. Rossler, D. A. Petrone and E. M. Carreira, *Acc. Chem. Res.*, 2019, **52**, 2657–2672.
- M. D. Patil, G. Grogan, A. Bommarius and H. Yun, *ACS Catal.*, 2018, **8**, 10985–11015.
- S. Kobayashi, Y. Mori, J. S. Fossey and M. M. Salter, *Chem. Rev.*, 2011, **111**, 2626–2704.
- M. Yus, J. C. Gonzalez-Gomez and F. Foubelo, *Chem. Rev.*, 2013, **113**, 5595–5698.
- N. Kumagai and M. Shibasaki, *Bull. Chem. Soc. Jpn.*, 2015, **88**, 503–517.
- J. S. Bandar and T. H. Lambert, *J. Am. Chem. Soc.*, 2013, **135**, 11799–11802.
- T. Ma, X. Fu, C. W. Kee, L. Zong, Y. Pan, K.-W. Huang and C.-H. Tan, *J. Am. Chem. Soc.*, 2011, **133**, 2828–2831.
- N. Tokunaga, Y. Otomaru, K. Okamoto, K. Ueyama, R. Shintani and T. Hayashi, *J. Am. Chem. Soc.*, 2004, **126**, 13584–13585.
- Z. Cui, Y. J. Chen, W. Y. Gao, C. G. Feng and G. Q. Lin, *Org. Lett.*, 2014, **16**, 1016–1019.
- B. Gopula, C. W. Chiang, W. Z. Lee, T. S. Kuo, P. Y. Wu, J. P. Henschke and H. L. Wu, *Org. Lett.*, 2014, **16**, 632–635.
- B. M. Trost, C. I. Hung, D. C. Koester and Y. Miller, *Org. Lett.*, 2015, **17**, 3778–3781.
- M.-Y. Ngai, A. Barchuk and M. J. Krische, *J. Am. Chem. Soc.*, 2007, **129**, 12644–12645.
- C. Fan, X. Y. Lv, L. J. Xiao, J. H. Xie and Q. L. Zhou, *J. Am. Chem. Soc.*, 2019, **141**, 2889–2893.
- S. Tang, X. Zhang, J. Sun, D. Niu and J. J. Chruma, *Chem. Rev.*, 2018, **118**, 10393–10457.
- M. Li, O. Gutierrez, S. Berritt, A. Pascual-Escudero, A. Yesilcimen, X. Yang, J. Adrio, G. Huang, E. Nakamaru-Ogiso, M. C. Kozlowski and P. J. Walsh, *Nat. Chem.*, 2017, **9**, 997–1004.
- Y. Zhu and S. L. Buchwald, *J. Am. Chem. Soc.*, 2014, **136**, 4500–4503.
- J. Liu, C. G. Cao, H. B. Sun, X. Zhang and D. Niu, *J. Am. Chem. Soc.*, 2016, **138**, 13103–13106.
- Y. Wang, L. F. Deng, X. Zhang and D. Niu, *Org. Lett.*, 2019, **21**, 6951–6956.
- D. Niu and S. L. Buchwald, *J. Am. Chem. Soc.*, 2015, **137**, 9716–9721.
- M. Zhan, X. Pu, B. He, D. Niu and X. Zhang, *Org. Lett.*, 2018, **20**, 5857–5860.
- X. Qian, P. Ji, C. He, J. O. Zirimwabagabo, M. M. Archibald, A. A. Yeagley and J. J. Chruma, *Org. Lett.*, 2014, **16**, 5228–5231.
- S. Wang, X. Qian, Y. Chang, J. Sun, X. Xing, W. F. Ballard and J. J. Chruma, *J. Org. Chem.*, 2018, **83**, 4054–4069.
- R. Jiang, L. Ding, C. Zheng and S. L. You, *Science*, 2021, **371**, 380–386.
- W. J. Cui, Z. J. Wu, Q. Gu and S. L. You, *J. Am. Chem. Soc.*, 2020, **142**, 7379–7385.
- Q. Cheng, J. H. Xie, Y. C. Weng and S. L. You, *Angew. Chem., Int. Ed.*, 2019, **58**, 5739–5743.
- Q. Wang, Q. Gu and S. L. You, *Angew. Chem., Int. Ed.*, 2019, **58**, 6818–6825.
- Z. P. Yang, C. Zheng, L. Huang, C. Qian and S. L. You, *Angew. Chem., Int. Ed.*, 2017, **56**, 1530–1534.
- Y. L. Su, Y. H. Li, Y. G. Chen and Z. Y. Han, *Chem. Commun.*, 2017, **53**, 1985–1988.
- Y. Wu, L. Hu, Z. Li and L. Deng, *Nature*, 2015, **523**, 445–450.
- Y. Wu and L. Deng, *J. Am. Chem. Soc.*, 2012, **134**, 14334–14337.
- L. Hu, Y. Wu, Z. Li and L. Deng, *J. Am. Chem. Soc.*, 2016, **138**, 15817–15820.



46 B. Hu and L. Deng, *Angew. Chem., Int. Ed.*, 2018, **57**, 2233–2237.

47 K. Li, X. Shao, L. Tseng and S. J. Malcolmson, *J. Am. Chem. Soc.*, 2018, **140**, 598–601.

48 X. Shao, K. Li and S. J. Malcolmson, *J. Am. Chem. Soc.*, 2018, **140**, 7083–7087.

49 M. Li, S. Berritt and P. J. Walsh, *Org. Lett.*, 2014, **16**, 4312–4315.

50 M. Li, B. Yucel, J. Adrio, A. Bellomo and P. J. Walsh, *Chem. Sci.*, 2014, **5**, 2383–2391.

51 M. Li, S. Berritt, L. Matuszewski, G. Deng, A. Pascual-Escudero, G. B. Panetti, M. Poznik, X. Yang, J. J. Chruma and P. J. Walsh, *J. Am. Chem. Soc.*, 2017, **139**, 16327–16333.

52 G. Deng, M. Li, K. Yu, C. Liu, Z. Liu, S. Duan, W. Chen, X. Yang, H. Zhang and P. J. Walsh, *Angew. Chem., Int. Ed.*, 2019, **58**, 2826–2830.

53 K. Yu, M. Li, G. Deng, C. Liu, J. Wang, Z. Liu, H. Zhang, X. Yang and P. J. Walsh, *Adv. Synth. Catal.*, 2019, **361**, 4354–4359.

54 R. A. Shelp and P. J. Walsh, *Angew. Chem., Int. Ed.*, 2018, **57**, 15857–15861.

55 T. Niwa, H. Yorimitsu and K. Oshima, *Org. Lett.*, 2008, **10**, 4689–4691.

56 M. Li, M. Gonzalez-Esguevillas, S. Berritt, X. Yang, A. Bellomo and P. J. Walsh, *Angew. Chem., Int. Ed.*, 2016, **55**, 2825–2829.

57 H. Guan, Q. Zhang, P. J. Walsh and J. Mao, *Angew. Chem., Int. Ed.*, 2020, **59**, 5172–5177.

58 G. Gao, Y. Fu, M. Li, B. Wang, B. Zheng, S. Hou and P. J. Walsh, *Adv. Synth. Catal.*, 2017, **359**, 2890–2894.

59 S. C. Sha, H. Jiang, J. Mao, A. Bellomo, S. A. Jeong and P. J. Walsh, *Angew. Chem., Int. Ed.*, 2016, **55**, 1070–1074.

60 X. Cao, S. C. Sha, M. Li, B. S. Kim, C. Morgan, R. Huang, X. Yang and P. J. Walsh, *Chem. Sci.*, 2016, **7**, 611–618.

61 Z. Liu, M. Li, B. Wang, G. Deng, W. Chen, B.-S. Kim, H. Zhang, X. Yang and P. J. Walsh, *Org. Chem. Front.*, 2018, **5**, 1870–1876.

62 Y. Matsumoto, J. Sawamura, Y. Murata, T. Nishikata, R. Yazaki and T. Ohshima, *J. Am. Chem. Soc.*, 2020, **142**, 8498–8505.

63 M. D. Fryzuk and B. Bosnich, *J. Am. Chem. Soc.*, 1977, **99**, 6262–6267.

64 R. D. Taylor, M. MacCoss and A. D. Lawson, *J. Med. Chem.*, 2014, **57**, 5845–5859.

65 D. J. Newman and G. M. Cragg, *J. Nat. Prod.*, 2020, **83**, 770–803.

66 E. Vitaku, D. T. Smith and J. T. Njardarson, *J. Med. Chem.*, 2014, **57**, 10257–10274.

67 A. Narczyk, M. Pieczykolan and S. Stecko, *Org. Biomol. Chem.*, 2018, **16**, 3921–3946.

