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Does electrophilic activation of nitroalkanes in polyphosphoric acid involve formation of nitrile oxides?†

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The mechanistic rationale involving activation of nitroalkanes towards interaction with nucleophilic reagents in the presence of polyphosphoric acid (PPA) was re-evaluated. Could nitrile oxide moieties be formed during this process? This experiment demonstrates that at least in some cases this could happen, as generated nitrile oxides were successfully intercepted as adducts of [3 + 2] cycloadditions.

Introduction

Nitroalkanes possessing acidic α-CH bonds are traditionally employed as versatile nucleophilic components in a variety of base-assisted C-C bond-forming reactions. 1-8 Another mode of activation of nitroalkanes enables acid-promoted tautomerization into nitronic acid (aci form), which is electrophilic in nature and can interact with a number of nucleophiles. In aqueous media, facile acid-assisted reaction with water takes place, commonly known as the Nef reaction, 9-12 resulting in the formation of aldehydes. In anhydrous media, alternative reactivity patterns are possible, involving interaction with various Cand N-based nucleophiles. Over the last decade, our group has been instrumental in the development of multistep cascade transformations triggered by initial electrophilic activation of nitroalkanes in the presence of polyphosphoric acid (PPA). 13-17 Proposed mechanistic rationales for all these processes involve formation of a phosphorylated nitronate species as a key electrophilic intermediate. Herein, we wish to disclose our recent studies proving that formation of nitrile oxide species is also possible under the same conditions. This suggests that alternative mechanisms could be potentially considered, at least for some of the earlier described processes.

Results and discussion

As discussed above, it was previously stated that interaction of nitroalkane 1 with PPA leads to the formation of electrophilic nitronate species 2, enabling subsequent reactions with various nucleophiles. The resulting phosphorylated N-hydroxy-N-alkyl hydroxylamine 4 may undergo further elimination of the ortho-phosphoric acid molecule to afford species 5 (oximes or amidines resulting from reaction with carbon or nitrogen-based nucleophiles, respectively, Scheme 1). Oximes 5 (Nu = Ar), as

$$\begin{array}{c} \overset{\oplus}{\mathbf{R}^1} \overset{\ominus}{\mathbf{3}} & \overset{(\mathsf{HO})_2\mathsf{OPO}}{\mathbf{6}} & \overset{\mathsf{N}}{\mathbf{6}} & \overset{\mathsf{N}}{\mathbf{1}} & \overset{\mathsf{N}}{\mathbf{7}} & \overset{\mathsf{N}}{\mathbf{6}} & \overset{\mathsf{N}}{\mathbf{1}} & \overset{\mathsf{N}}{\mathbf{7}} & \overset{\mathsf{N}}{\mathbf{1}} & \overset{\mathsf{N}}{\mathbf{7}} & \overset{\mathsf{N}}{\mathbf{1}} & \overset{$$

Scheme 1

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Table 1 Optimization of the reaction conditions for generation of nitrile oxide 3a and its subsequent dimerization into furoxan 8a

m (PPA) per mmol of 1a (g)	<i>T</i> , °C (time, h)	Yield ^a , %
1.5	50 (2)	23
2	50 (2)	30
2	50 (2)	33
2	70 (0.5)	51
2	90 (0.3)	25
	of la (g) 1.5 2 2 2	of 1a (g) (time, h) 1.5 50 (2) 2 50 (2) 2 50 (2) 2 70 (0.5)

^a NMR yields are reported.

was previously demonstrated, may participate in Beckmann-type rearrangements, which was employed in acetamidation and carboxyamination reactions. ^{18,19} Generation of amidines 5 (Nu = NHR) were utilized in design of cascade transformations leading to various heterocyclic scaffolds. ^{13,20,21} We pondered if elimination of H_3PO_4 entity can take place in species 2 to form nitrile oxide 3 under the described reaction conditions. ²² This species is also inherently electrophilic and reactions with nucleophiles should provide the same type of product 5 (Scheme 1).

Furthermore, reaction of 3 with *ortho*-phosphoric acid in the absence of other nucleophiles could be envisioned, producing *N*-hydroxyalkanimidic phosphoric anhydride **6**, which after hydrolytic cleavage, should provide hydroxamic acid 7. It should be noted that the isomerization of nitroalkanes into hydroxamic acids taking place in the presence of PPA is well documented.²³ On the other hand, in most typical reactions nitrile oxides act as 1,3-dipoles. If nitrile oxides are indeed generated from nitroalkanes in the presence of PPA, then in the absence of external dipolarophiles, they should undergo [3 + 2]-dimerization to provide 1,2,5-oxadiazole 2-oxides (furoxans) **8** (Scheme 1). Alternatively, cross-cycloaddition reactions with alkynes **9** and chalcones **11** could be performed to generate isoxazoles **10** or 4,5-dihydroisoxazoles **12**, respectively (Scheme 1).

First, we attempted to conduct the experiment involving interaction of 1-nitropropane (1, $R^1 = \text{Et}$, Scheme 1) with PPA in the absence of any other reagents by targeting detection for furoxan 8 ($R^1 = \text{Et}$) resulting from [3 + 2]-homodimerization. However, formation of such product was not detected, as almost the entire amount of starting nitroalkane was transformed into *N*-hydroxypropionamide 7 ($R^1 = \text{Et}$). Evidently, even if formed under the described reaction conditions, propionitrile oxide (3, $R^1 = \text{Et}$) is much too reactive and quickly interacts with the medium. To address this issue, we decided to test the same idea by employing 2-nitro-1-phenylethan-1-one (1a), producing greater concentration of *aci*-form, which should translate in higher concentration of benzoyl cyanide *N*-oxide (3a). Therefore, this starting material should be more prompt towards the desired dimerization reaction.

Initial tests were carried out in PPA with 80% content of P2O5 (corresponding to diphosphoric acid, H₄P₂O₇). A mixture of 1a (1 mmol) with PPA (1.5 g) was stirred and heated at 50 °C for 2 hours, stopping only after TLC analysis proved complete consumption of starting material. Delightfully, some nitroalkane was indeed converted into furoxan 8a, albeit in marginal yield (Table 1, entry 1). Increasing the amount of PPA (2.0 g) improved the yield, though not significantly (entry 2). We thought that switching to 87% PPA might enhance the reaction performance, since this medium has less nominal content of water and thus less nucleophilic; however, the reaction proceeded sluggishly and at 50 °C, the yield of 8a was essentially the same as in 80% PPA (entry 3). Further improvement was achieved upon heating to a higher temperature, which also allowed to cut down reaction time. At 70 °C, a maximum yield of 51% was obtained (entry 4) and further increases of temperature proved detrimental (entry 5). The purified product 8a was isolated in a 46% yield (Scheme 2). With optimized reaction conditions in hand, we moved towards scope and limitation studies, the results of which are shown in Scheme 2. Generally, the reactions of substituted 2-nitro-1-phenylethan-1-ones 1b-h proceeded smoothly, well tolerating alkyl and aromatic halide substituents and affording the corresponding furoxans in moderate yields. We also managed to execute successful transformations of ethyl 2-nitroacetate (1i) into furoxan 8i, bearing two ester functions under the same reaction conditions (Scheme 2).

Paper

Scheme 3

The possibility for a one-pot assembly of a bicyclic structural core, including a furoxan moiety fused to pyridazine ring, was also evaluated. To this end, nitroketone 1a was treated with PPA in the presence of hydrazine hydrate. The bicyclic structure 13a was formed as a sole isolable product, although in low quantities (Scheme 3).

Next, we decided to explore the possibility to intercept the nitrile oxides generated from nitroalkanes with external dipolarophiles, such as terminal acetylenes. Treatment of various nitroalkanes with PPA was carried out in the presence of phenyl acetylene (9a), 1-heptene (9b), or propargyl bromide (9c) to yield the desired isoxazoles 10 (Scheme 4). Formation of isoxazole 10da was unambiguously confirmed by single crystal X-ray crystallography (Fig. 1).

This reaction cannot be recommended as a preparative protocol towards these materials, since most amount of nitroalkanes were converted into the corresponding hydroxamic acids. Despite that, it clearly confirms that nitrile oxides were indeed generated under the described reaction conditions.

Interception of nitrile oxides with alkenes were tested as well. Reaction of 1a in the presence of chalcone 11a gave nearly

$$\begin{array}{c} O_{2}N \\ O_{3}N \\ O_{2}N \\ O_{3}N \\ O_{2}N \\ O_{3}N \\ O_{3}N \\ O_{4}N \\ O_{5}N \\ O_{5}$$

Fig. 1 ORTEP drawing of X-ray structures of (4-fluorophenyl)(5phenylisoxazol-3-yl)methanone (10da, CCDC #2099762†). The thermal ellipsoids are shown at 50% probability.

an equimolar mixture of regioisomeric dihydroisoxazoles 12aa and 14aa, formation of which are most likely controlled by electronic and steric factors, respectively (Scheme 5). Reaction of 1f in the presence of chalcone 11b, with enhanced polarization of double bond, resulted in perfect electronicallycontrolled regiochemistry, but was accompanied by oxidative aromatization of the cycloaddition product 15fb (Scheme 5). Finally, reaction of 1a in the presence of malonic imide 16 proceeded smoothly, providing bicyclic imide 17a in good yield (Scheme 5).

Since the yields obtained were quite moderate, we decided to investigate the material balance by tracing the byproducts formed in dimerization reactions towards furoxans 8 (Scheme 2). Initially we thought we will be dealing with hydroxamic acids 7, however, benzamides 18 (sometimes in a mixture with benzoic acids 19) were observed and isolated instead. Evidently, under the listed reaction conditions hydroxamic acids 7 undergo facile Lossen rearrangement or competing hydrolysis as shown in Scheme 6. It is believed, that O-phosphorylated hydroxamic acid 6 (initially formed after nucleophilic attack on nitrile oxide 2 with ortho-phosphoric acid as shown in Scheme 1, vide infra) could be re-esterified to afford NO-phosphorylated form 20. The later would undergo Lossen rearrangement into benzoyl isocyanate 21, which can be hydrolyzed with extrusion of isocyanic acid and formation of benzoic acid 19 (Scheme 6). More commonly, an alternative pathway was operating,

Scheme 5

Scheme 4

O NO2 PPA 70 °C, 30 min R NO2 R NO3 min R NO4 R NO4 R NO5 R R NO5 R R NO5 R R NO5 R

involving reaction of isocyanate 21 with phosphoric acid to afford benzoylcarbamic phosphoric anhydride 22, which then underwent hydrolysis to afford benzamide species 18. These compounds were isolated in moderate yield from most of the reaction mixtures along with normal furoxan products 8.

Conclusions

In conclusion, we demonstrated that electrophilic activation of nitroalkanes in the presence of polyphosphoric acid can, at least in some cases, provide nitrile oxide moieties, which likely exists in equilibrium with more ordinary nitronate species. In the presence of PPA, nitrile oxides tend to decompose to hydroxamic acids, but for most stable electron-deficient dipolar species, typical [3 + 2] cycloaddition reactions can be studied. Nevertheless, formation of hydroxamic acid is a dominant process in most cases, so this reaction cannot be recommended as a preparative method for generation of nitrile oxides.

Experimental

General

NMR spectra, 1 H and 13 C was measured in solutions of CDCl $_{3}$ or DMSO- d_{6} on Bruker AVANCE-III HD instrument (at 400.40 or

100.61 MHz, respectively). Residual solvent signals were used as internal standards, in DMSO-d₆ (2.50 ppm for ¹H, and 40.45 ppm for ¹³C nuclei) or in CDCl₃ (7.26 ppm for ¹H, and 77.16 ppm for 13C nuclei). HRMS spectra was measured on Bruker maxis impact (electrospray ionization, in MeCN solutions, employing HCO₂Na-HCO₂H for calibration). IR spectra was measured on FT-IR spectrometer Shimadzu IRAffinity-1S equipped with an ATR sampling module. Reaction progress, purity of isolated compounds, and R_f values were monitored with TLC on Silufol UV-254 plates. Column chromatography was performed on silica gel (32–63 μm, 60 Å pore size). Melting points were measured with Stuart SMP30 apparatus. Polyphosphoric acid samples were prepared by dissolving precisely measured amounts of P2O5 in 85% ortho-phosphoric acid. Nitroacetophenones and chalcones were prepared according to the previously published procedures. 25-27 All other reagents and solvents were purchased from commercial venders and used as received.

1-(4-Ethylphenyl)-2-nitroethan-1-one (1c). 4-Ethylbenzaldehyde (1.34 g, 10.0 mmol), nitromethane (610 mg, 10 mmol) and ethanol (10 mL) were combined in 25 mL round bottom flask. The reactor was placed into ice bath and aqueous solution of NaOH (425 mg, 10.5 mmol in 40% aqueous solution) was added dropwise in such a rate so temperature did not exceed 15 °C. The reaction mixture was stirred for 15 minutes and then guenched with 5% aqueous solution of acetic acid. The solution was extracted with EtOAc (5 × 20 mL) and concentrated in vacuo. Resulting nitroalcohol was employed at the next step without additional purification. Concentrated sulfuric acid (1.5 mL) was slowly added to an Erlenmeyer flask charged with acetic acid (14 mL), potassium bichromate (3.0 g), water (28 mL), and nitroalcohol cooled in ice bath. The solution was stirred for 24 hours at room temperature, formed precipitate was filtered, washed with cold water, dried, and recrystallized from EtOH. The titled compound was obtained as colorless solid, mp 129.3-131.1 °C (EtOH); yield 1.47 g (7.6 mmol, 76%). R_f 0.29, EtOAc/hexane (1 : 5, v/v). ¹H NMR (400 MHz, DMSO) δ 7.87 (d, J = 8.3 Hz, 2H), 7.44 (d, J = 8.2 Hz, 2H), 6.50 (s, 2H), 2.71 (q, J = 7.6 Hz, 2H), 1.20 (t, J = 7.6 Hz, 3H). ¹³C NMR (101 MHz, DMSO) δ 187.9, 151.6, 131.2, 128.7 (2C), 128.5 (2C), 82.8, 28.3, 15.1. IR, $v_{\text{max}}/\text{cm}^{-1}$: 3027, 2974, 2935, 1961, 1933, 1694, 1607, 1549, 1328, 1229, 1185. HRMS (ES TOF) calcd for $C_{10}H_{11}NNaO_3$ ([M + Na]⁺): 216.0631, found 216.0633 (-0.7 ppm).

3,4-Dibenzoyl-1,2,5-oxadiazole 2-oxide (8a). General method for synthesis of furoxans. 2-Nitroacetophenone $1a^{27}$ (165 mg, 1.00 mmol), was combined with 2.0 g of polyphosphoric acid (87 wt% of P_2O_5). The reaction mixture was stirred at 70 °C for 30 minutes. Then the mixture was diluted with water (40 mL) and quenched with aqueous ammonia (20% solution in H_2O , 15 mL). The material was extracted with ethyl acetate (4 × 15 mL) and combined organic phases were washed with brine. The solution was concentrated *in vacuo*, and the residue was purified by column chromatography on silica gel eluting EtOAc/PE, (1:4 v/v) to give pure furoxan. Colorless oil, R_f 0.5, EtOAc/hexane (1:4, v/v). Yield 72 mg (0.24 mmol, 49%). ¹H NMR (400 MHz, CDCl₃) δ 8.25–8.18 (m, 2H), 7.93–7.80 (m, 2H), 7.76–

Paper

7.65 (m, 2H), 7.60-7.49 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 182.2, 180.9, 154.8, 135.9, 135.8, 134.3, 134.3, 131.0 (2C), 130.1 (2C), 129.7 (2C), 129.5 (2C), 112.1. IR, $v_{\text{max}}/\text{cm}^{-1}$: 1735, 1684, 1602, 1559, 1455, 1419, 1325, 1246, 1178, 1101. HRMS (ES TOF) calcd for $C_{16}H_{10}N_2NaO_4$ ([M + Na]⁺): 317.0533, found 317.0532 (0.4 ppm).

Benzamide (18a)28 was also isolated from the same reaction mixture as colorless solid. Yield 53 mg (0.44 mmol, 44%). ¹H NMR (400 MHz, CDCl₃) δ 7.84 (dt, J = 7.1, 1.4 Hz, 2H), 7.59–7.50 (m, 1H), 7.46 (t, J = 7.5 Hz, 2H), 6.35 (s, 2H). ¹³C NMR (101 MHz, CDCl₃) δ 170.2, 133.8, 132.4, 129.1 (2C), 127.8 (2C).

3,4-Bis(4-methylbenzoyl)-1,2,5-oxadiazole Product 8b was obtained via the method described for compound 8a, employing 2-nitro-4'-methylacetophenone 1b²⁷ (179 mg, 1.00 mmol), and purified by column chromatography (eluent EtOAc/PE 1:5). Yellow solid, mp 118-119.5 °C (EtOH), lit²⁹ mp 124 °C (EtOH); yield 83 mg (0.26 mmol, 52%). R_f 0.46, EtOAc/hexane (1 : 5, v/v). ¹H NMR (400 MHz, CDCl₃) δ 8.11 (d, J = 8.3 Hz, 2H, 7.76 (d, J = 8.3 Hz, 2H), 7.33 (dd, J = 13.4, 8.3 Hz,4H), 2.45 (d, J = 10.1 Hz, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 181.3, 180.0, 154.6, 147.0, 146.8, 131.5, 131.4, 130.7 (2C), 130.0 (2C), 129.9 (2C), 129.8 (2C), 111.9, 22.0 (2C). IR, $v_{\text{max}}/\text{cm}^{-1}$: 1675, 1646, 1600, 1470, 1328, 1251, 1181, 1123, 1104. HRMS (ES TOF) calcd for C₁₈H₁₄N₂NaO₄ ([M + Na]⁺): 345.0846, found 345.0840 (1.7 ppm).

4-Methylbenzamide (18b)28 was also isolated from the same reaction mixture as colorless solid. Yield 28 mg (0.21 mmol, 21%). ¹H NMR (400 MHz, DMSO- d_6) δ 7.90 (s, 1H), 7.77 (d, I =8.2 Hz, 2H), 7.29 (s, 1H), 7.24 (d, J = 8.0 Hz, 2H), 2.34 (s, 3H). ¹³C NMR (101 MHz, DMSO- d_6) δ 167.8, 141.1, 131.5, 128.8 (2C), 127.6 (2C), 21.0.

3,4-Bis(4-ethylbenzoyl)-1,2,5-oxadiazole 2-oxide (8c). Product 8c was obtained via the method described for compound 8a, employing 2-nitro-4'-ethylacetophenone 1c (193 mg, 1.00 mmol), and purified by column chromatography (eluent EtOAc/ PE, 1:4). Colorless oil, R_f 0.60, EtOAc/hexane (1:5, v/v). Yield 86 mg (0.25 mmol, 49%). ¹H NMR (400 MHz, CDCl₃) δ 8.18–8.07 (m, 2H), 7.83-7.70 (m, 2H), 7.36 (dd, J = 13.6, 8.1 Hz, 4H), 2.75(dq, J = 9.8, 7.6 Hz, 4H), 1.28 (q, J = 7.8 Hz, 6H). ¹³C NMR (101) MHz, CDCl₃) δ 181.5, 180.1, 154.7, 153.2, 152.9, 131.8, 131.7, 131.0 (2C), 130.1 (2C), 128.9 (2C), 128.7 (2C), 112.1, 29.4, 29.3, 15.1, 15.1. IR, $v_{\text{max}}/\text{cm}^{-1}$: 2964, 2877, 1733, 1595, 1460, 1427, 1383, 1337, 1272, 1147, 1096. HRMS (ES TOF) calcd for $C_{20}H_{18}N_2NaO_4$ ([M + Na]⁺): 373.1159, found 373.1151 (2.0 ppm).

3,4-Bis(4-fluorobenzoyl)-1,2,5-oxadiazole (8d).Product 8d was obtained via the method described for compound 8a, employing 2-nitro-4'-fluoroacetophenone 1d27 (183 mg, 1.00 mmol), and purified by column chromatography (eluent EtOAc/PE, 1:4). Yellow oil, R_f 0.53, EtOAc/hexane (1:5, v/v). Yield 58 mg (0.18 mmol, 35%). ¹H NMR (400 MHz, CDCl₃) δ 8.40–8.34 (m, 2H), 8.03–7.97 (m, 2H), 7.32 (m, J = 11.1, 8.4 Hz, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 180.2, 178.9, 167.30 (d, J =259.7 Hz), 167.28 (d, J = 259.3 Hz), 154.3, 133.7 (d, J = 9.9 Hz, 2C), 132.8 (d, J = 10.3 Hz, 2C), 130.3 (d, J = 2.9 Hz), 130.2 (d, J = 2.9 Hz) 2.9 Hz), 116.9 (d, J = 22.4 Hz, 2C), 116.6 (d, J = 22.0 Hz, 2C), 111.7. IR, $v_{\text{max}}/\text{cm}^{-1}$: 1769, 1684, 1663, 1593, 1508, 1472, 1414,

1325, 1236, 1157, 1101. HRMS (ES TOF) calcd for C₁₆H₈F₂N₂- NaO_4 ([M + Na]⁺): 353.0344, found 353.0338 (1.7 ppm).

4-Fluorobenzamide (18b)28 was also isolated from the same reaction mixture as colorless solid. Yield 36 mg (0.26 mmol, 26%). ¹H NMR (400 MHz, DMSO- d_6) δ 8.01 (s, 1H), 7.97–7.87 (m, 2H), 7.41 (s, 1H), 7.32-7.19 (m, 2H). ¹³C NMR (101 MHz, DMSO) δ 166.0 (d, J = 164.9 Hz), 162.7, 130.8 (d, J = 3.0 Hz) 130.2 (d, J = 9.0 Hz, 2C), 115.2 (d, I = 21.6 Hz, 2C).

3,4-Bis(2-fluorobenzoyl)-1,2,5-oxadiazole 2-oxide (8e). Product 8e was obtained via the method described for compound 8a, employing 2-nitro-2'-fluoroacetophenone 1e²⁶ (183 mg, 1.00 mmol), and purified by column chromatography (eluent EtOAc/PE 1:4). Yellow oil, R_f 0.32, EtOAc/hexane (1:4, v/v). Yield 81 mg (0.25 mmol, 49%). ¹H NMR (400 MHz, CDCl₃) δ 7.92 (td, I = 7.8, 1.8 Hz, 1H), 7.82 (td, I = 7.4, 1.8 Hz, 1H), 7.70-7.61 (m, 2H), 7.36–7.29 (m, 2H), 7.17 (m, J = 25.1, 11.0, 8.4, 1.2 Hz, 2H). 13 C NMR (101 MHz, CDCl₃) δ 180.1, 177.1, 162.0 (d, $J = 259.0 \,\mathrm{Hz}$, 161.6 (d, $J = 256.4 \,\mathrm{Hz}$), 154.4, 136.8 (d, $J = 7.6 \,\mathrm{Hz}$), 136.7 (d, J = 7.4 Hz), 131.5, 131.1, 125.2 (d, J = 3.3 Hz), 124.8 (d, J = 3.3 Hz), 124.8 (d, J = 3.3 Hz), 124.8 (d, J = 3.3 Hz) = 3.7 Hz, 123.4 (d, J = 10.3 Hz), 123.2 (d, J = 11.0 Hz), 117.0 (d, J= 21.6 Hz), 116.6 (d, J = 22.0 Hz), 112.7. IR, $v_{\text{max}}/\text{cm}^{-1}$: 1655, 1602, 1458, 1439, 1330, 1238, 1181, 1159, 1099, 1041. HRMS (ES TOF) calcd for $C_{16}H_8F_2N_2NaO_4$ ([M + Na]⁺): 353.0344, found 353.0334 (3.0 ppm).

3,4-Bis(4-chlorobenzoyl)-1,2,5-oxadiazole 2-oxide (8f). Product 8f was obtained via the method described for compound 8a, employing 2-nitro-4'-chloroacetophenone 1f²⁷ (199 mg, 1.00 mmol), and purified by column chromatography (eluent EtOAc/PE 1:4). Colorless solid, mp 120-121 °C (EtOH), lit³⁰ mp 124 °C; R_f 0.57, EtOAc/hexane (1 : 6, v/v). Yield 92 mg (0.25 mmol, 51%). ¹H NMR (400 MHz, CDCl₃) δ 8.18 (d, J =8.7 Hz, 2H), 7.80 (d, J = 8.6 Hz, 2H), 7.53 (m, J = 12.0, 8.7 Hz, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 180.5, 179.2, 154.0, 142.41, 142.39, 132.01, 132.00, 131.9 (2C), 131.0 (2C), 129.8 (2C), 129.5 (2C), 111.4. IR, $v_{\text{max}}/\text{cm}^{-1}$: 1773, 1682, 1610, 1586, 1472, 1405, 1323, 1243, 1176, 1089. HRMS (ES TOF) calcd for C₁₆H₈Cl₂N₂- NaO_4 ([M + Na]⁺): 384.9753, found 384.9760 (-1.8 ppm).

4-Chlorobenzamide (18f)28 and 4-chlorobenzoic acid (19f)31 were also isolated from the same reaction mixture as colorless solid. 18f: yield 25 mg (0.16 mmol, 16%). ¹H NMR (400 MHz, DMSO- d_6) δ 8.06 (br. s, 1H), 7.93–7.84 (m, 2H), 7.55–7.50 (m, 2H), 7.49 (br. s, 1H). 13 C NMR (101 MHz, DMSO- d_6) δ 166.9, 136.1, 133.1, 129.5 (2C), 128.4 (2C). 19f: yield 41 mg (0.26 mmol, 26%). ¹H NMR (400 MHz, DMSO- d_6) δ 13.19 (br. s, 1H), 7.93 (d, J = 8.6 Hz, 2H, 7.55 (d, J = 8.5 Hz, 2H). ¹³C NMR (101 MHz, DMSO- d_6) δ 166.5, 137.8, 131.2 (2C), 129.7, 128.8 (2C).

3,4-Bis(3-chlorobenzoyl)-1,2,5-oxadiazole 2-oxide (8g). Product 8g was obtained via the method described for compound 8a, employing 2-nitro-3'-chloroacetophenone 1g27 (199 mg, 1.00 mmol), and purified by column chromatography (eluent EtOAc/PE, 1 : 4). Yellow oil; R_f 0.49, EtOAc/hexane (1 : 4, v/v). Yield 102 mg (0.28 mmol, 56%). ¹H NMR (400 MHz, CDCl₃) δ 8.17 (m, J = 1.9 Hz, 1H), 8.12 (m, J = 7.8, 1.4 Hz, 1H), 7.84 (m, J= 1.9 Hz, 1H, 7.73-7.64 (m, 3H), 7.50 (m, J = 15.8, 7.9 Hz, 2H).¹³C NMR (101 MHz, CDCl₃) δ 180.5, 179.2, 153.8, 135.8, 135.50, 135.46, 135.3, 135.2, 135.11, 130.6, 130.4, 130.3, 129.4, 128.7, 127.8, 111.1. IR, $v_{\text{max}}/\text{cm}^{-1}$: 1754, 1687, 1614, 1571, 1467, 1419,

1323, 1236, 1079. HRMS (ES TOF) calcd for $C_{16}H_8Cl_2N_2NaO_4$ ([M + Na]⁺): 384.9753, found 384.9762 (-2.2 ppm).

3,4-Bis(4-bromobenzoyl)-1,2,5-oxadiazole 2-oxide (8h). Product 8h was obtained via the method described for compound 8a, employing 2-nitro-4'-bromoacetophenone $1h^{27}$ (244 mg, 1.00 mmol), and purified by column chromatography (eluent EtOAc/PE, 1 : 4). Yellowish solid, mp $135.9-136.7\,^{\circ}$ C (EtOAc), $R_{\rm f}$ 0.57, EtOAc/hexane (1 : 4, v/v). Yield 90 mg (0.20 mmol, 40%). 1 H NMR (400 MHz, CDCl₃) δ 8.12–8.05 (m, 2H), 7.76–7.64 (m, 6H). 13 C NMR (101 MHz, CDCl₃) δ 180.7, 179.4, 154.0, 132.7 (2C), 132.5 (2C), 132.4 (2C), 131.9 (2C), 131.3 (2C), 131.0 (2C), 111.3. IR, $\nu_{\rm max}/{\rm cm}^{-1}$: 928, 1012, 1067, 1173, 1212, 1251, 1325, 1395, 1467, 1581, 1612, 1655, 1689, 3089. HRMS (ES TOF) calcd for $C_{16}H_8{\rm Br}_2{\rm N}_2{\rm NaO}_4$ ([M + Na] $^+$): 472.8743, found 472.8741 (0.4 ppm).

4-Bromobenzamide (18h)²⁸ was also isolated from the same reaction mixture as colorless solid. Yield 56 mg (0.28 mmol, 28%). ¹H NMR (400 MHz, DMSO- d_6) δ 8.06 (s, 1H), 7.84–7.78 (m, 2H), 7.70–7.63 (m, 2H), 7.48 (s, 1H). ¹³C NMR (101 MHz, DMSO) δ 167.4, 133.8, 131.7 (2C), 130.1 (2C), 125.5.

3,4-Bis(ethoxycarbonyl)-1,2,5-oxadiazole 2-oxide (8i). Product 8i was obtained via the method described for compound 8a, employing ethyl nitroacetate (133 mg, 1.00 mmol), and purified by column chromatography (eluent EtOAc/PE, 1 : 4). Colorless oil, $R_{\rm f}$ 0.46, EtOAc/hexane (1 : 4, v/v). Yield 49 mg (0.21 mmol, 43%). 1 H NMR (400 MHz, DMSO- $d_{\rm 6}$) δ 4.45 (qd, J = 7.1, 1.0 Hz, 2H), 4.36 (qd, J = 7.1, 1.1 Hz, 2H), 1.33 (td, J = 7.1, 1.0 Hz, 3H), 1.27 (td, J = 7.1, 1.1 Hz, 3H). 13 C NMR (101 MHz, DMSO) δ 156.5, 154.8, 148.8, 107.4, 63.5, 63.1, 13.72, 13.7. IR, $\nu_{\rm max}/{\rm cm}^{-1}$: 2988, 1742, 1619, 1472, 1373, 1335, 1246, 1195, 1063, 1022. HRMS (ES TOF) calcd for $C_{\rm 8}H_{10}N_{2}{\rm NaO}_{\rm 6}$ ([M + Na] $^{+}$): 253.0431, found 253.0426 (2.1 ppm).

Procedure for preparation of cyclization products of hydrazine hydrate with 2-nitroethan-1-one, 4,7-diphenyl-[1,2,5]oxadiazolo[3,4-d]pyridazine-1-oxide (13a). 2-Nitroacetophenone 1a (165 mg, 1.00 mmol), hydrazine monohydrate (65% solution in water, 154 mg, 150 µL, 2.00 mmol) were added to 2 g of 87% polyphosphoric acid. The mixture was stirred at 110 °C for 2 h and then diluted with water (40 mL) and quenched with aqueous ammonia (20% solution in H₂O, 15 mL). Reaction mixture was extracted with ethyl acetate (4 \times 15 mL). Combined organic layers were washed with brine and concentrated in vacuo. The residue was purified by column chromatography on silica gel (eluent: EtOAc: PE, 1:4 v/v) affording the titled product as orange solid, mp 199-201 °C (EtOAc), lit³⁰ mp 210 °C; $R_{\rm f}$ 0.44, EtOAc/hexane (1 : 4, v/v). Yield 31 mg (0.11 mmol, 21%). ¹H NMR (400 MHz, CDCl₃) δ 8.68–8.60 (m, 2H), 8.00–7.90 (m, 2H), 7.68-7.56 (m, 6H). 13 C NMR (101 MHz, CDCl₃) δ 150.4, 148.3, 145.1, 132.6, 131.9, 131.8, 130.3, 129.6 (2C), 129.3 (2C), 129.2 (2C), 128.7 (2C), 106.9. IR, $v_{\text{max}}/\text{cm}^{-1}$: 2974, 2882, 1600, 1508, 1460, 1429, 1388, 1263, 1147, 1089, 1048. HRMS (ES TOF) calcd for C₁₆H₁₀N₄NaO₂ ([M + Na]⁺): 313.0696, found 313.0691

5-Phenyl-3-benzoylisoxazole. General method for synthesis of isoxazoles (10aa). 2-Nitroacetophenone $1a^{27}$ (165 mg, 1.00 mmol) and phenylacetylene 9a (102 mg, 1.00 mmol) were added to 2.0 g of 87% polyphosphoric acid. The mixture was stirred at

110 °C for 2 h and then diluted with water (40 mL) and guenched with agueous ammonia (20% solution in H₂O, 15 mL). Reaction mixture was extracted with ethyl acetate (4 imes 15 mL). Combined organic layers were washed with brine and concentrated in vacuo. The residue was purified by column chromatography on silica gel (eluent: benzene/hexane, 1 : 2 v/v) to afford the titled compound as yellowish solid, mp 85.4-86 °C (EtOH), lit³² mp 88–90 °C, R_f 0.46, benzene/hexane (1 : 2, v/v). Yield 65 mg (0.26 mmol, 26%). ¹H NMR (400 MHz, CDCl₃) δ 8.38-8.32 (m, 2H), 7.86 (m, J = 7.8, 1.9 Hz, 2H), 7.67 (m, J =7.4 Hz, 1H), 7.53 (m, I = 17.5, 7.9 Hz, 5H), 7.06 (s, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 185.9, 170.8, 162.4, 135.8, 134.1, 130.8, 130.7 (2C), 129.2 (2C), 128.6 (2C), 126.7, 126.0 (2C), 100.3. IR, $\nu_{\text{max}}/\text{cm}^{-1}$: 1655, 1598, 1573, 1451, 1426, 1236, 1181, 1147, 1101, 1075. HRMS (ES TOF) calcd for $C_{16}H_{11}NNaO_2$ ([M + Na]⁺): 272.0682, found 272.0678 (1.3 ppm).

5-Pentyl-3-benzoylisoxazole (10ab). Product 10ab was obtained *via* the method described for compound 10aa, employing 2-nitroacetophenone 1a²⁷ (165 mg, 1.00 mmol) and hept-1-yne 9b (96 mg, 1.00 mmol) and purified by column chromatography (eluent EtOAc/PE 1 : 4). Colorless oil; $R_{\rm f}$ 0.72, EtOAc/hexane (1 : 4, v/v). Yield 68 mg (0.28 mmol, 28%). ¹H NMR (400 MHz, CDCl₃) δ 8.33–8.26 (m, 2H), 7.67–7.61 (m, 1H), 7.54–7.48 (m, 2H), 6.52 (s, 1H), 2.86–2.80 (m, 2H), 1.80–1.71 (m, 2H), 1.42–1.34 (m, 4H), 0.95–0.89 (m, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 185.2, 173.8, 160.8, 134.9, 132.9, 129.6 (2C), 127.5 (2C), 100.6, 30.1, 26.1, 25.6, 21.3, 12.9. IR, $\nu_{\rm max}/{\rm cm}^{-1}$: 2930, 1774, 1660, 1602, 1455, 1434, 1258, 1239, 1181, 1104. HRMS (ES TOF) calcd for C₁₅H₁₇NNaO₂ ([M + Na]⁺): 266.1151, found 266.1145 (2.6 ppm).

5-(Bromomethyl)-3-benzoylisoxazole (10ac). Product 10ac was obtained via the method described for compound 10aa, employing 2-nitroacetophenone $\mathbf{1a}^{27}$ (165 mg, 1.00 mmol) and propargyl bromide $\mathbf{9c}$ (119 mg, 1.00 mmol) and purified by column chromatography (eluent EtOAc/PE 1 : 4). Yellow solid, mp 62–65 °C; R_f 0.73, EtOAc/hexane (1 : 4, v/v). Yield 82 mg (0.31 mmol, 31%). 1 H NMR (400 MHz, CDCl₃) δ 8.31–8.28 (m, 2H), 7.68–7.63 (m, 1H), 7.55–7.50 (m, 2H), 6.85 (d, J=0.6 Hz, 1H), 4.54 (d, J=0.6 Hz, 2H). 13 C NMR (101 MHz, CDCl₃) δ 185.3, 168.5, 162.3, 135.6, 134.4, 130.8 (2C), 128.8 (2C), 104.9, 18.1. IR, $\nu_{\text{max}}/\text{cm}^{-1}$: 3152, 2959, 1766, 1645, 1599, 1578, 1452, 1291, 1224, 1187, 999. HRMS (ES TOF) calcd for $C_{11}H_8$ BrNaNO₂ ([M + Na] $^+$): 287.9631, found 287.9624 (2.2 ppm).

3-(4-Ethylbenzoyl)-5-phenylisoxazole (**10ca**). Product **10ca** was obtained via the method described for compound **10aa**, employing 2-nitro-4'-ethylacetophenone **1c** (193 mg, 1.00 mmol) and phenylacetylene **9a** (102 mg, 1.00 mmol) and purified by column chromatography (eluent EtOAc : PE 1 : 4). White solid, mp 84–85 °C (EtOH), $R_{\rm f}$ 0.49, benzene/hexane (1 : 2, v/v). Yield 105 mg (0.38 mmol, 38%). ¹H NMR (400 MHz, CDCl₃) δ 8.28 (d, J = 8.3 Hz, 2H), 7.85 (dd, J = 7.8, 1.8 Hz, 2H), 7.55–7.47 (m, 3H), 7.36 (d, J = 8.3 Hz, 2H), 7.04 (s, 1H), 2.75 (q, J = 7.6 Hz, 2H), 1.29 (t, J = 7.6 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 184.4, 169.6, 161.6, 150.3, 132.4, 129.9 (2C), 129.7, 128.1 (2C), 127.2 (2C), 125.8, 125.0 (2C), 99.3, 28.1, 14.1. IR, $v_{\rm max}/{\rm cm}^{-1}$: 2964, 1653, 1602, 1566, 1439, 1296, 1255, 1183, 1151, 1046. HRMS (ES TOF)

calcd for $C_{18}H_{15}NNaO_2$ ([M + Na]⁺): 300.0995, found 300.0997 (-0.6 ppm).

3-(4-Fluorobenzoyl)-5-phenylisoxazole (10da). Product 10da was obtained via the method described for compound 10aa, employing 2-nitro-4'-fluoroacetophenone $1d^{27}$ (183 mg, 1.00 mmol) and phenylacetylene 9a (102 mg, 1.00 mmol) and purified by column chromatography (eluent benzene/hexane, 1 : 2 v/v). White solid, mp 111–113 °C (EtOH), lit³² mp 112–114 °C; yield 77 mg (0.29 mmol, 29%). $R_{\rm f}$ 0.49, benzene/hexane (1 : 2, v/v). ¹H NMR (400 MHz, CDCl₃) δ 8.48–8.39 (m, 2H), 7.89–7.82 (m, 2H), 7.55–7.48 (m, 3H), 7.25–7.18 (m, 2H), 7.05 (s, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 183.0, 169.9, 165.4 (d, J = 256.8 Hz), 161.4, 132.5 (d, J = 9.5 Hz, 2C), 131.1 (d, J = 2.8 Hz), 129.8, 128.2, 125.6, 125.0, 114.8 (d, J = 21.6 Hz, 2C), 99.2. IR, $\nu_{\rm max}/{\rm cm}^{-1}$: 3133, 1655, 1600, 1506, 1441, 1296, 1251, 1231, 1147, 1104. HRMS (ES TOF) calcd for $C_{16}H_{10}{\rm FNNaO}_2$ ([M + Na]*): 290.0588, found 290.0582 (1.9 ppm).

3-(4-Chlorobenzoyl)-5-phenylisoxazole (10fa). Product 10fa was obtained via the method described for compound 10aa, employing 2-nitro-4'-chloroacetophenone 1f²⁷ (199 mg, 1.00 mmol) and phenylacetylene 9a (102 mg, 1.00 mmol) and purified by column chromatography (eluent benzene/hexane, 1 : 2 v/ v). Yellowish solid, mp 139–141 °C (EtOH), lit³² mp 143–145 °C; R_f 0.49, benzene/hexane (1 : 2, v/v). Yield 105 mg (0.37 mmol, 37%). ¹H NMR (400 MHz, CDCl₃) δ 8.38–8.30 (d, 2H), 7.85 (dd, J = 7.6, 2.0 Hz, 2H), 7.51 (m, J = 8.4, 2.0 Hz, 5H), 7.05 (s, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 183.4, 169.9, 161.3, 139.7, 133.0, 131.1 (2C), 129.8, 128.2 (2C), 127.9 (2C), 125.6, 125.0 (2C), 99.2. IR, $\nu_{\rm max}/{\rm cm}^{-1}$: 3133, 3060, 1930, 1884, 1648, 1588, 1443, 1395, 1251, 1181, 1089. HRMS (ES TOF) calcd for C₁₆H₁₀ClNNaO₂ ([M + Na]*): 306.0292, found 306.0295 (–0.9 ppm).

3-(4-Chlorobenzoyl)-5-pentylisoxazole (10fb). Product 10fb was obtained via the method described for compound 10aa, employing 2-nitro-4'-chloroacetophenone 1f²⁷ (199 mg, 1.00 mmol) and hept-1-yne 9b (96 mg, 1.00 mmol) and purified by column chromatography (eluent EtOAc/PE, 1 : 4). Colorless oil; $R_{\rm f}$ 0.74, EtOAc/hexane (1 : 4, v/v). Yield 86 mg (0.31 mmol, 31%). ¹H NMR (400 MHz, CDCl₃) δ 8.33–8.28 (m, 2H), 7.53–7.47 (m, 2H), 6.55 (s, 1H), 2.86 (t, J = 7.6 Hz, 2H), 1.83–1.75 (m, 2H), 1.44–1.35 (m, 4H), 0.95–0.92 (m, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 184.9, 175.1, 161.9, 140.7, 134.3, 132.2 (2C), 129.0 (2C), 101.8, 31.3, 27.3, 26.7, 22.4, 14.0. IR, $v_{\rm max}/{\rm cm}^{-1}$: 2935, 2853, 1923, 1665, 1590, 1455, 1400, 1255, 1210, 1178, 1096. HRMS (ES TOF) calcd for $C_{15}H_{16}{\rm ClNNaO}_2$ ([M + Na]⁺): 300.0762, found 300.0760 (0.7 ppm).

3-(3-Chlorobenzoyl)-5-phenylisoxazole (10ga). Product 10ga was obtained via the method described for compound 10aa, employing 2-nitro-3'-chloroacetophenone 1g²⁷ (199 mg, 1.00 mmol) and phenylacetylene 9a (102 mg, 1.00 mmol) and purified by column chromatography (eluent EtOAc/PE 1 : 4). Colorless solid, mp 117–117.5 °C (EtOH); $R_{\rm f}$ 0.46, benzene/hexane (1 : 2, v/v). Yield 99 mg (0.35 mmol, 35%). ¹H NMR (400 MHz, CDCl₃) δ 8.34 (t, J = 2.0 Hz, 1H), 8.27 (d, J = 7.8 Hz, 1H), 7.85 (m, J = 7.8, 2.0 Hz, 2H), 7.63 (m, J = 8.1, 2.2, 1.1 Hz, 1H), 7.54–7.48 (m, 4H), 7.06 (s, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 184.5, 171.1, 162.2, 137.2, 134.9, 134.0, 130.9, 130.6, 129.9, 129.2 (2C), 128.9, 126.6, 126.0 (2C), 100.2. IR, $v_{\rm max}/{\rm cm}^{-1}$: 1959, 1896, 1761, 1655,

1571, 1436, 1236, 1159, 1082, 1046. HRMS (ES TOF) calcd for $C_{16}H_{10}ClNNaO_2$ ([M + Na] $^+$): 306.0292, found 306.0285 (2.2 ppm).

5-(Bromomethyl)-3-(4-bromobenzoyl)isoxazole (10hc). Product 10hc was obtained via the method described for compound 10aa, employing 2-nitro-4′-bromoacetophenone 1h²⁷ (244 mg, 1.00 mmol) and propargyl bromide 9c (119 mg, 1.00 mmol) and purified by column chromatography (eluent EtOAc/PE, 1:4). Brown solid, mp 83–85 °C; $R_{\rm f}$ 0.40, EtOAc/hexane (1:4, v/v). Yield 66 mg (0.19 mmol, 19%). ¹H NMR (400 MHz, CDCl₃) δ 8.22–8.18 (m, 2H), 7.69–7.66 (m, 2H), 6.85 (d, J = 0.6 Hz, 1H), 4.54 (s, 2H). ¹³C NMR (101 MHz, CDCl₃) δ 184.0, 168.6, 162.1, 134.1, 132.2 (2C), 132.0 (2C), 129.8, 104.7, 17.9. IR, $\nu_{\rm max}/{\rm cm}^{-1}$: 3147, 2920, 2853, 1934, 1652, 1585, 1448, 1397, 1295, 1224, 1178, 932. HRMS (ES TOF) calcd for C₁₁H₇Br₂NaNO₂ ([M + Na][†]): 365.8736, found 365.8727 (2.4 ppm).

3,4-Dibenzoyl-5-phenyl-4,5-dihydroisoxazole (12aa) and 3,5-dibenzoyl-4-phenyl-4,5-dihydroisoxazole (14aa). 2-Nitro-acetophenone $1a^{27}$ (165 mg, 1.00 mmol) and *trans*-chalcone 11a (312 mg, 1.5 mmol) were combined with to 2.0 g of polyphosphoric acid (80 wt% of P_2O_5). The reaction mixture was stirred at 130 °C for 3–5 h (TLC control). Reaction mixture was diluted with water and extracted with ethyl acetate (4 \times 15 mL). Combined organic layers were washed with brine, and concentrated *in vacuo*. The residue was purified by column chromatography on silica gel (eluent: EtOAc/PE, 1:4 v/v) to obtain two regioisomeric products 12aa and 14aa.

Product 12aa. Brown solid, mp 105–107 °C; $R_{\rm f}$ 0.30, EtOAe/hexane (1 : 4, v/v). Yield 138 mg (0.39 mmol, 39%). ¹H NMR (400 MHz, CDCl₃) δ 8.27 (dt, J=8.3, 1.3 Hz, 2H), 7.90–7.85 (m, 2H), 7.62 (dddd, J=7.5, 5.2, 3.9, 1.9 Hz, 2H), 7.52–7.41 (m, 7H), 7.33 (dd, J=6.3, 2.6 Hz, 2H), 5.78 (d, J=8.2 Hz, 1H), 5.58 (dd, J=8.3, 1.1 Hz, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 195.7, 185.8, 157.1, 138.3, 135.6, 135.5, 134.4, 134.1, 130.7 (2C), 129.5, 129.4 (2C), 129.3 (2C), 129.1 (2C), 128.6 (2C), 126.4 (2C), 89.5, 63.1. IR, $\nu_{\rm max}/{\rm cm}^{-1}$: 3060, 2925, 1684, 1643, 1578, 1448, 1366, 1277, 1195, 1016, 932. HRMS (ES TOF) calcd for C₂₃H₁₇NaNO₃ ([M + Na]†): 378.1101, found 378.1091 (2.6 ppm).

Product 14aa. Yellow viscous oil; $R_{\rm f}$ 0.29, EtOAc/hexane (1 : 4, v/v). Yield 121 mg (0.34 mmol, 34%). ¹H NMR (400 MHz, CDCl₃) δ 8.17 (dt, J = 8.5, 1.2 Hz, 2H), 8.01 (dt, J = 8.5, 1.2 Hz, 2H), 7.66–7.56 (m, 2H), 7.53–7.32 (m, 9H), 5.86 (dd, J = 5.4, 1.0 Hz, 1H), 5.37 (dd, J = 5.4, 1.0 Hz, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 192.1, 184.9, 159.1, 137.6, 135.9, 134.4, 134.0, 133.8, 130.5 (2C), 129.6 (2C), 129.6 (2C), 129.1 (2C), 128.5 (2C), 128.47, 127.8 (2C), 90.6, 55.4. IR, $\nu_{\rm max}/{\rm cm}^{-1}$: 3062, 2927, 1686, 1645, 1579, 1448, 1369, 1279, 1196, 1026, 932. HRMS (ES TOF) calcd for C₂₃H₁₇NaNO₃ ([M + Na]⁺): 378.1101, found 378.1092 (2.3 ppm).

3-(4-Chlorobenzoyl)-4-(2-nitrobenzoyl)-5-(2-chlorophenyl) isoxazole (15fb). Product **15gb** was obtained *via* the method described for compound **12aa**, employing 2-nitro-4'-chloroacetophenone **1f** (199 mg, 1.00 mmol), and (*E*)-2-chloro-2'-nitrochalcone **11b**²⁵ (431 mg, 1.5 mmol) and purified by column chromatography (eluent EtOAc/PE, 1 : 4). Brown viscous oil; $R_{\rm f} = 0.41$, benzene. Yield 248 mg (0.53 mmol, 53%). ¹H NMR (400 MHz, CDCl₃) δ 8.05–7.92 (m, 2H), 7.77 (dd, J = 8.5, 1.2 Hz, 1H), 7.56–7.49 (m, 3H), 7.41–7.16 (m, 6H). ¹³C NMR (101 MHz,

CDCl₃) δ 185.0, 184.3, 171.0, 160.2, 145.8, 141.6, 134.3, 134.1, 133.9, 133.9, 132.7, 131.5 (2C), 131.4, 131.3, 129.9, 129.4 (2C), 129.1, 126.8, 125.0, 123.9, 118.6. IR, $\nu_{\text{max}}/\text{cm}^{-1}$: 3089, 2925, 1732, 1674, 1587, 1522, 1346, 1224, 1084, 1014, 930. HRMS (ES TOF) calcd for $C_{23}H_{12}Cl_2N_2NaO_5$ ([M + Na]⁺): 489.0015, found 489.0013 (0.6 ppm).

3-Benzoyl-5-phenyl-3a,6a-dihydro-4H-pyrrolo[3,4-d]isoxazole-4,6(5H)-dione (17a). Product 17a was obtained via the method described for compound 10aa, employing 2-nitroacetophenone 1a27 (165 mg, 1.00 mmol) and N-phenylmaleimide 16 (173 mg, 1.00 mmol) and purified by column chromatography (eluent EtOAc/PE 1:4). Colorless solid, mp 147–150 °C, lit³³ mp 100–102 °C (EtOAc); R_f 0.16, EtOAc/hexane (1:4, v/v). Yield 201 mg (0.63 mmol, 63%). ¹H NMR (400 MHz, $CDCl_3$) δ 8.27–8.17 (m, 2H), 7.68–7.62 (m, 1H), 7.54–7.42 (m, 5H), 7.31-7.26 (m, 2H), 5.70 (d, J = 9.8 Hz, 1H), 5.24 (d, J =9.8 Hz, 1H). 13 C NMR (101 MHz, CDCl₃) δ 183.4, 170.0, 168.8, 153.2, 134.9, 134.7, 130.8, 130.6 (2C), 129.5 (2C), 129.4, 128.8 (2C), 126.2 (2C), 81.2, 54.9. IR, $v_{\text{max}}/\text{cm}^{-1}$: 3065, 1901, 1713, 1651, 1598, 1501, 1448, 1381, 1279, 1200, 1070, 1024. HRMS (ES TOF) calcd for $C_{18}H_{12}N_2NaO_4$ ([M + Na]⁺): 343.0689, found 343.0679 (2.9 ppm).

Author contributions

A. V. Aksenov – conceptualization, supervision, funding acquisition; N. A. Aksenov – methodology, formal analysis, funding acquisition; N. K. Kirilov – investigation; A. A. Skomorokhov – investigation; D. A. Aksenov – investigation; I. A. Kurenkov – investigation; E. A. Sorokina – investigation, formal analysis; M. A. Nobi – formal analysis, writing (review and editing); M. Rubin – conceptualization, supervision, writing (original draft, review, and editing).

Conflicts of interest

There are no conflicts to declare.

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