


 Cite this: *RSC Adv.*, 2021, 11, 31433

Synthesis, cytotoxicity evaluation and molecular docking studies on 2',4'-dihydroxy-6'-methoxy-3',5'-dimethylchalcone derivatives†

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2',4'-Dihydroxy-6'-methoxy-3',5'-dimethylchalcone (DMC, **1**) was isolated from seeds of *Syzygium nervosum* A.Cunn. ex DC. exhibiting intriguing biological activities. Herein, thirty three DMC derivatives including 4'-O-monosubstituted-DMC (**2**), 7-O-acylated-4-hydroxycoumarin derivatives (**3**), stilbene-coumarin derivatives (**4**), 2',4'-disubstituted-DMC (**5**), and flavanone derivatives (**6**), were synthesised through acylation, alkylations, and sulfonylation. These semi-synthetic DMC derivatives were evaluated for *in vitro* cytotoxicity against six carcinoma cell lines. It was found that most derivatives exhibited higher cytotoxicity than DMC. In particular, 4'-O-caproylated-DMC (**2b**) and 4'-O-methylated-DMC (**2g**) displayed the strongest cytotoxicity against SH-SY5Y with IC₅₀ values of 5.20 and 7.52 μM, respectively. Additionally, 4'-O-benzylated-DMC (**2h**) demonstrated the strongest cytotoxicity against A-549 and FaDu with IC₅₀ values of 9.99 and 13.98 μM, respectively. Our structure-activity relationship (SAR) highlights the importance of 2'-OH and the derivatisation pattern of 4'-OH. Furthermore, molecular docking simulation studies shed further light on how these bioactive compounds interact with cyclin-dependent kinase 2 (CDK2).

 Received 15th July 2021
 Accepted 1st September 2021

DOI: 10.1039/d1ra05445g

rsc.li/rsc-advances

Introduction

Natural products are known nowadays as important sources for the discovery and development of new drugs.¹ Over 33 years from 1981 to 2014, more than 26% of FDA-approved drugs (320 of 1211) were natural products (NPs) or NP-derived pharmacophores.² However, there has been quite a challenge to apply NPs

to medication due to undesirable properties such as low oral bioavailability,³ metabolic instability,⁴ and less solubility in aqueous media⁵ that meant a low permeation rate from gut to blood system.⁶

Cancer is an uncontrolled proliferation of cells and a widespread cause of death among the worldwide population.⁷ According to reports from the World Health Organization, it is expected that the number of cancer patients will have dramatically increased by more than 50% in 2020. Numerous methods, such as surgery, chemotherapy, and radiation therapy, targeted therapy, and checkpoint inhibitors,⁸ are potentially used for the treatment of cancer. Nevertheless, a large number of undesirable side effects due to the low selectivity of a chemotherapeutic agent for tumour cells is a major problem, which may be the cause of death in patients.⁹ Currently, an abundance of natural substances and their derivatives have been used as anticancer agents, such as vincristine, vinblastine, docetaxel, etoposide, and seliciclib.¹⁰

Syzygium nervosum A.Cunn. ex DC. or *Cleistocalyx operculatus* (Roxb.) Merr & L. M. Perry, *aka* Ma-kiang in Thai, is distributed over tropical countries¹¹ such as Thailand, Vietnam, China, Myanmar, and Malaysia. Its fresh fruits are valuable for the Thai food and drink manufacturing industry for jam, juice, tea, yogurt, and wine. Its leaves and buds were traditionally used to treat colds, fever, inflammation,¹² gastrointestinal disorders,

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† Electronic supplementary information (ESI) available: Additional data of anticancer and antimicrobial activities, molecular docking, ¹H- and ¹³C-NMR spectra of synthetic compounds. See DOI: 10.1039/d1ra05445g



and for antiseptics.¹¹ Phytochemical investigations of *S. nervosum* disclosed a number of natural products, including terpenes,¹³ flavanone,¹⁴ flavone,¹⁵ flavonol,¹⁶ chalcone,¹⁷ anthocyanin,¹⁸ and phloroglucinol derivatives.¹⁹ These derivatives have been reported to possess a wide range of anticancer and cytotoxic activities.^{19–21}

More specifically, chalcone (1,3-diphenyl-2-propen-1-one) is an open-chain flavonoid containing an α,β -unsaturated carbonyl moiety. This class of compound is a secondary metabolite typically found in natural products and serves as a common precursor for flavone, flavanone, flavonol, and aurone syntheses.²² Recently, the chalcone scaffold has attracted considerable interest from researchers in the field of drug discovery and development.²³ Chalcone derivatives exhibited a variety of pharmaceutical activities such as anti-inflammatory,²⁴ antiviral, antimalarial, antibacterial, anticancer, anti-fungal, antioxidant, anti-inflammation, anti-HIV-1 reverse transcriptase, tyrosinase inhibitor, anti-plasmodial, α -glucosidase inhibitory, and protein kinase C inhibitor effects.²⁵ In particular, metochalcone and sofalcone are clinically approved drugs for the treatment of choleric and gastrointestinal diseases, respectively.²³ The actions of chalcone derivatives against malignant cells are exerted through multiple mechanisms such as cell cycle arrest induction,²⁶ tubulin polymerisation inhibition,²⁷ epidermal growth factor receptor tyrosine kinase (EGFR-TK) inhibition,²⁸ DNA and mitochondrial damage, cell apoptosis induction,²⁹ blockade of the nuclear factor-kappa B (NF- κ B) signalling pathway,³⁰ and angiogenesis inhibition.³¹ These pathways are associated with the cell proliferation process, which may contribute to the further development of new anticancer drugs.³²

2',4'-Dihydroxy-6'-methoxy-3',5'-dimethylchalcone (DMC, **1**) is a major bioactive compound isolated from *S. nervosum* seeds, which exhibits various bioactivities, such as anticancer, antimicrobial,³³ AMP-dependent protein kinase (AMPK) activator,³⁴ anti-influenza,^{35,36} and antiviral effects.³⁷ More specifically, DMC shows moderate to excellent anticancer activities in various cancer cell lines, such as human leukaemia (K562),³⁸ pancreatic cancer (PANC-1),³⁹ cervical cancer (HeLa), hepatocellular carcinoma (SMMC-7721),⁴⁰ oral nasopharyngeal carcinoma (KB), breast cancer (MCF-7), and lung cancer (A-549).³⁷ Furthermore, drug combination studies involving DMC and the 5-fluorouracil (5-FU) anticancer drug revealed a dramatic increase in the anticancer activity of 5-FU against 5-FU-selected drug-resistant human hepatocellular carcinoma (BEL-7402/5-FU) by inducing apoptosis and cell cycle arrest through the PI3K/AKT pathway.⁴¹ Moreover, DMC increased the sensitivity of the tumour cells and showed reversal effects on drug-resistant KB-A1 cells both *in vitro* and *in vivo*.⁴²

Herein, the structure of DMC was modified through methylation, allylation, benzoylation, benzylation, acylation, and sulfonylation under basic conditions yielding several semi-synthetic DMC derivatives, including 4'-O-monosubstituted-DMC (**2a–2n**), 7-O-acylated-4-hydroxycoumarin (**3a–3e**), stilbene-coumarin derivatives (**4a–4e**), 2',4'-O-disubstituted-DMC (**5a–5n**), and flavanone derivatives (**6g** and **6m**). These semi-synthetic DMC derivatives were evaluated for cytotoxicity

against six cancer cell lines, KKU-M213, FaDu, HT-29, MDA-MB-231, A-549, and SH-SY5Y, as well as CL as a normal cell line. The anticancer activity of bioactive DMC derivatives was further studied by *in silico* molecular docking with cyclin-dependent kinase 2 (CDK2) to understand the plausible binding modes and the corresponding key interactions with amino acid residues in protein pockets. Furthermore, the antimicrobial activities of DMC derivatives were also evaluated against three bacterial strains: *Escherichia coli*, *Staphylococcus aureus* and *Pseudomonas aeruginosa*.

Results and discussion

Isolation and characterisation of DMC

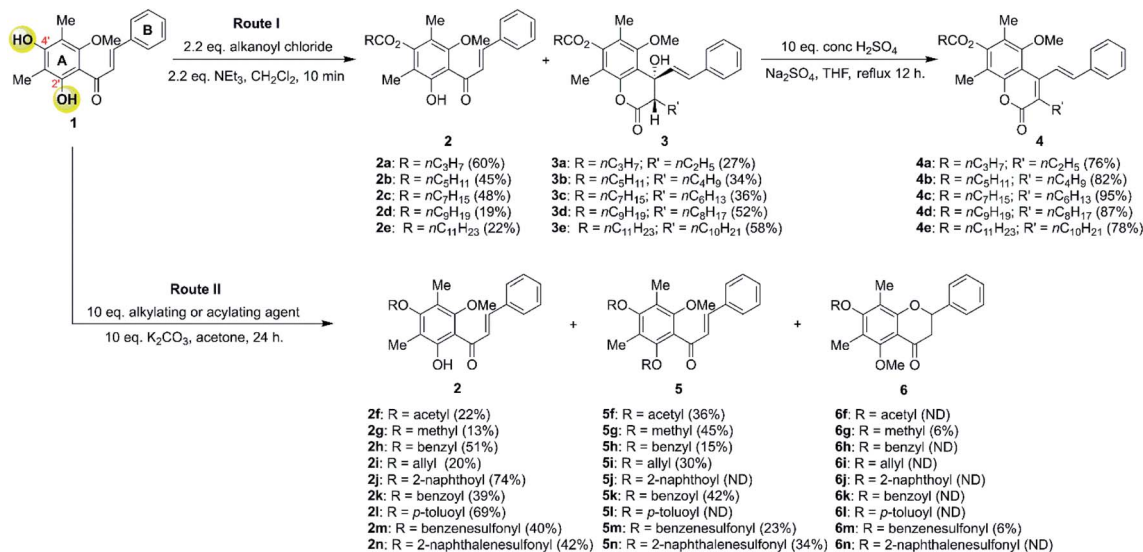
Air-dried and ground seeds of *S. nervosum* were extracted with EtOAc. The extract was subjected to flash column chromatography (FCC) on silica gel, which was eluted by gradient eluents of *n*-hexane/EtOAc. The DMC-containing fractions were further chromatographed on silica gel and eluted by isocratic elution of *n*-hexane/acetone. The DMC-containing fractions were additionally crystallised in CH₂Cl₂/*n*-hexane to obtain orange needle-shaped crystals. DMC (**1**), orange needle crystals, had a melting point value of 124.5–126.4 °C. The HRMS-ESI experiment indicated that the (M + Na)⁺ ion at *m/z* 321.1100, belonged to a molecular formula of C₁₈H₁₈O₄Na (calcd: 321.1103). The FTIR spectrum indicated characteristic absorption bands of the hydroxy group and conjugated carbonyl ketone at 3545 and 1632 cm⁻¹, respectively. The ¹H-NMR spectrum exhibited proton signals of a monosubstituted aromatic ring resonating at δ 7.36–7.45 and 7.60–7.68 ppm, *trans*-double bond protons at δ 7.84 and 7.99 ppm with a coupling constant value (*J*) of 15.7 Hz, a methoxy group at δ 3.67 ppm, and benzylic methyl groups at δ 2.13 and 2.15 ppm. Additionally, the spectrum indicated the presence of two hydroxy groups resonating at δ 5.39 and 13.61 ppm which were assigned to free hydroxy and chelated hydroxy groups, respectively. These spectroscopic data were similar to the published values.³⁷

Structural modifications of DMC

The DMC (**1**) has two hydroxy groups at C-2' and C-4' on aromatic ring A. Typically, the 2'-OH is less reactive than the 4'-OH due to intramolecular hydrogen bonding with an adjacent carbonyl group. The modification was conducted under weak basic conditions through a nucleophilic substitution reaction with various acylating, alkylating, and sulfonylating agents to facilitate provide several semi-synthetic DMC derivatives (Scheme 1).

We began the investigation by performing the *O*-acylation of DMC with various alkanoyl chlorides, as shown in Scheme 1, route I. The DMC readily reacts with alkanoyl chlorides under basic conditions which provide a mixture of 4'-*O*-acylated-DMC derivatives (**2a–2e**) and 7-*O*-acylated-4-hydroxycoumarins (**3a–3e**). The monoacylation product **2** was thought to proceed *via* the acylation at more reactive 4'-OH thus providing 4'-*O*-acylated-DMC derivatives (**2a–2e**) in moderate yields. However,





Scheme 1 Derivatisations of DMC. ND = not detected.

compound **2** can also undergo acylation at the less reactive 2'-OH and subsequent intramolecular Claisen-cyclisation to afford 7-O-acylated-4-hydroxycoumarins (**3a–3e**) in moderate yields. It should be noted that the product derived from monoacylation at 2'-OH was not observed in all cases.

The example FTIR spectrum of 4'-O-caprinoylated-DMC (**2d**) evidently indicated a new strong absorption band at 1767 cm^{-1} which was assigned to the C=O stretching of carbonyl ester. The $^1\text{H-NMR}$ spectrum showed an absence of a 4'-OH signal of DMC and it became apparent that the α -methylene proton signal of the attached alkanoyl moiety resonating at δ 2.63 ppm. Furthermore, the $^{13}\text{C-NMR}$ spectrum displayed the signal of a C=O ester carbon resonating at δ 171.0 ppm, which confirmed the completed acylation reaction.

In contrast, 7-O-caprinoylated-4-hydroxycoumarin (**3d**) showed the characteristic peaks of hydroxy and carbonyl ester groups at 3470 and 1779 cm^{-1} , respectively. The $^1\text{H-NMR}$ spectrum showed the signals of a hydroxy group at δ 5.68 ppm and a methine proton at δ 2.84 ppm. The $^{13}\text{C-NMR}$ spectrum showed the peaks at δ 168.7 and 171.1 ppm which indicated the presence of lactone and carbonyl ester groups, respectively. To understand the relative configuration of the Claisen-cyclisation product, a NOESY experiment (see Fig. 24S[†]) was performed which suggests key correlations between 4-OH and 5-OCH₃, 6-CH₃, and 2'-CH. Furthermore, correlations between 3-CH and 1'-CH and 2'-CH were also observed as evidence of the anti-configuration between 3-CH and 4-OH, which are in agreement with the simulated structure performed by DFT calculation at B3LYP/6-311++G(d,p) level of theory in the gas phase, as depicted in Fig. 1.

Controlled experiments (as shown in Table 1) were performed to further understand the plausible mechanisms for product formation. The use of caprinoyl chloride (1.2 equiv.) and NEt_3 (1.2–2.2 equiv.) only provided 4'-O-caprinoylated-DMC (**2d**) as the sole product (entries 1 and 2). However, increasing the equivalents of caprinoyl chloride (2.2 equiv.) and NEt_3 (2.2

equiv.) yielded a mixture of 4'-O-caprinoylated-DMC (**2d**) and 7-O-capyloylated-4-hydroxycoumarin (**3d**) products in 19 and 52%, respectively. The use of excess caprinoyl chloride (5 equiv.) and NEt_3 (5 equiv.) exclusively provided Claisen-cyclisation product **3d** in 91% yield.

The plausible mechanism for the formation of Claisen-cyclised product is shown in Scheme 2. Our results suggested that 2',4'-O-diacylated DMC (INT) was presumably the key intermediate for the Claisen-cyclised product **3**. However,

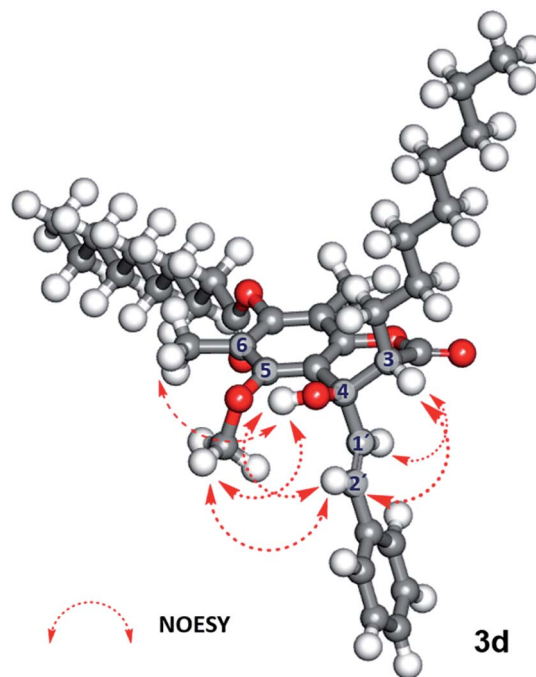


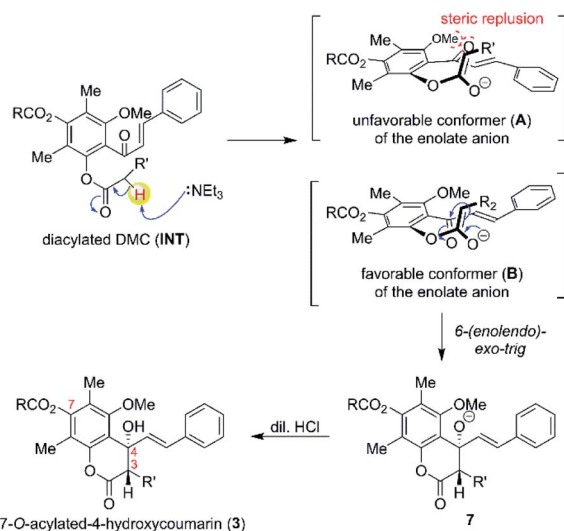
Fig. 1 Geometrically optimised structure of 7-O-caprinoylated-4-hydroxycoumarin (**3d**) performed by B3LYP/6-311++G(d,p) level of theory and its key NOESY experiment.



Table 1 Effects of caprinoyl chloride and NEt_3 equivalents on yields of **2d** and **3d**^a

Entry	Caprinoyl chloride (equiv.)	NEt_3 (equiv.)	Yield (%)	
			2d	3d
1	1.2	1.2	75	ND
2	2.2	1.2	74	ND
3	2.2	2.2	19	52
4	5	5	ND	91

^a ND = not detected.



Scheme 2 Plausible mechanism for the synthesis of 7-O-acylated-4-hydroxycoumarin derivatives (3).

attempts to spectroscopically detect and isolate a diacylated intermediate were unsuccessful. Once 2',4'-O-diacylated DMC (INT) was formed, the α -proton of INT was rapidly deprotonated by an NEt_3 base to generate an enolate anion (A and B). The anti-alignment (B) between the ketone carbonyl group and the enolate anion was favoured to minimise the steric repulsion between the ketone and methoxy group. The enolate anion (in structure B) then undergoes an intramolecular cyclisation with α,β -unsaturated ketone *via* more favourable 6-(enolendo)-exo-trig cyclisation³⁹ thus providing a stereospecific 7-O-acylated-4-hydroxycoumarin (3).

According to previous literature, the stilbene-coumarin hybrid scaffold exhibited strong cytotoxicity toward several cancer cell lines.⁴³ In our studies, 7-O-acylated-4-hydroxycoumarin derivatives (3) can serve as valuable precursors to access stilbene-coumarin derivatives. To our delight, 7-O-acylated-4-hydroxycoumarin derivatives (3) underwent dehydration by refluxing in THF with H_2SO_4 as the catalyst that delivered stilbene-coumarin derivatives **4** in excellent yields.

Spectroscopically, the FTIR spectrum of stilbene-coumarin **4d** indicated strong absorption bands at 1755 and 1712 cm^{-1} corresponding to C=O stretching of the ester and conjugated C=O of lactone, respectively. The $^1\text{H-NMR}$ spectrum exhibited

the disappearance of both hydroxy and methine proton signals. In addition, the $^{13}\text{C-NMR}$ spectrum showed the signals of the quaternary carbons of the double bond of the lactone ring as evidence of a completed dehydration reaction.

Next, we turn our attention to modifying the DMC core with the alkylation, acylation or sulfonylation, as shown in Scheme 1, route II. The DMC was dissolved in anhydrous acetone in the presence of anhydrous K_2CO_3 and then subjected to several alkylating, acylating, and sulfonylating agents. A mixture of products, including 4'-O-monosubstituted-DMC (**2f-2n**), 2',4'-O-disubstituted-DMC (**5f-5n**), and flavanone derivatives **6g** and **6m**, was obtained. The 4'-O-monosubstituted-DMC derivatives (**2f-2n**) were obtained in moderate yield, while derivatives **5f-5n** were obtained in low yield, presumably due to intramolecular hydrogen bonding with 2'-OH.

To rationalise the product formation, the DMC was first reacted at more reactive 4'-OH with alkylating, acylating or sulfonylating agents to yield 4'-O-monosubstituted-DMC (2). Next, the substitution at the less reactive 2'-OH further provided 2',4'-O-disubstituted-DMC (5). To our surprise, only flavanones **6g** and **6m** can be isolated in 6% yield. This flavanone product was derived from the intramolecular Michael addition of 2'-OH to α,β -unsaturated ketone *via* 6-*exo-trig* cyclisation. Moreover, the observed low yield of flavanone formation may result from the intramolecular hydrogen bonding between 2'-OH and the carbonyl group that decreases the acidity of the hydroxy proton. As a result, the 2'-OH was difficult to deprotonate by the weak base K_2CO_3 .⁴¹ In addition, optical rotation experiments suggested that racemic flavanones **6** were formed.

In vitro cytotoxicity

Our semi-synthetic DMC derivatives, including 4'-O-monosubstituted-DMC derivatives (**2a-2n**), 7-O-acylated-4-hydroxycoumarins (**3a-3e**), stilbene-coumarins (**4a-4e**), 2',4'-O-disubstituted-DMC derivatives (**5f-5n**), and flavanone derivatives (**6g** and **6m**), were further evaluated for cytotoxicity against six cancer cell lines, KKU-M213, FaDu, HT-29, MDA-MB-231, A-549, and SH-SY5Y with CL as a normal cell line, using sulforhodamine B (SRB) assay. DMC and anticancer drugs used in general medicine such as ellipticine were taken as the reference and positive control, respectively. Cytotoxicity was expressed as IC_{50} (the concentration which indicated 50% cell-growth inhibition) in μM units of concentration as listed in Table 2, while the residues are given in Table 1S.†

4'-O-Monosubstituted-DMC (2) exhibited the strongest cytotoxicity among the DMC derivatives. 4'-O-Caproylated-DMC (**2b**) and 4'-O-methylated-DMC (**2g**) showed the strongest cytotoxicity toward the SH-SY5Y cell line with IC_{50} values of 5.20 μM and 7.52 μM , respectively. 4'-O-Benzylated-DMC (**2h**) showed the strongest cytotoxicity toward A-549 and FaDu cell lines with IC_{50} values of 9.99 and 13.98 μM , respectively. Furthermore, the cytotoxicity of the other DMC derivatives toward HT-29 and MDA-MB-231 cell lines displayed moderate activity with IC_{50} values higher than 19.69 and 18.13 μM , respectively. The cytotoxicity on other cell lines was quite similar to that of these cell lines, which were classified as having moderate cytotoxicity.



Table 2 Cytotoxicity of some DMC derivatives against human intrahepatic cholangiocarcinoma (KKU-M213), human squamous cell carcinoma (FaDu), human colorectal adenocarcinoma (HT-29), human mammary gland adenocarcinoma (MDA-MB-231), human lung carcinoma (A-549), human neuroblastoma (SH-SY5Y), and human normal liver hepatic cell (CL)

Compound	Cytotoxicity (IC ₅₀ , μM)						CL
	KKU-M213	FaDu	HT-29	MDA-MB-231	A-549	SH-SY5Y	
2a	30.05	31.16	35.77	38.24	>50	27.55	32.82
2b	21.26	18.92	26.63	18.13	24.84	5.20	23.53
2c	24.43	20.63	26.62	25.86	>50	18.04	25.16
2d	>50	34.71	>50	36.70	>50	24.79	41.87
2f	42.98	36.49	38.63	40.28	>50	28.56	31.02
2g	23.88	24.84	19.69	26.22	25.35	7.52	25.42
2h	23.19	13.98	27.47	26.85	9.99	16.45	22.58
2i	31.26	28.96	34.63	39.10	36.67	28.25	31.68
2k	>50	37.89	32.53	>50	41.30	20.25	>50
2l	28.86	22.09	26.75	25.09	22.43	15.56	25.36
2m	23.76	27.00	26.66	26.11	26.29	20.43	29.01
2n	27.84	21.10	24.73	39.75	>50	16.37	30.38
3c	>50	>50	>50	>50	>50	20.86	35.90
4a	29.58	>50	30.87	>50	>50	>50	>50
5f	28.01	33.65	36.51	29.78	15.40	18.75	27.35
5g	23.47	19.30	21.94	29.78	22.67	12.56	18.69
5k	31.55	24.54	33.76	>50	30.93	25.31	25.31
6m	>50	>50	45.49	>50	37.86	44.69	>50
DMC (1)	29.4	37.57	38.58	37.64	19.9	13.1	15.6
Ellipticine (+) ^a	2.40	2.03	2.15	2.80	1.14	2.64	2.07

^a Ellipticine was used as a positive control for the cytotoxicity assay.

Moreover, most DMC derivatives displayed low cytotoxicity toward the normal cell line (CL) with the selectivity index (SI) to the cancer cell lines being higher than that of DMC with SI values up to 4.53 (see Table 1S[†]) for SH-SY5Y in compound 2b. Overall, our structural modifications enhanced the cytotoxicity and selectivity of DMC toward cancer cell lines.

Generally, 2',4'-O-disubstituted-DMC (5) displayed lower cytotoxicity than that of 4'-O-monosubstituted-DMC (2). In particular, 2',4'-O-dimethylated-DMC (5g) showed moderate cytotoxicity on all tested cancer cell lines with IC₅₀ value ranging from 12.56 to 29.78 μM, while other derivatives such as 2',4'-diacetylated-DMC (5f) and 2',4'-O-dibenzoylated-DMC (5k) displayed poor cytotoxicity. It should be noted that there was no cytotoxicity when derivative 5 was derivatised with allyl, benzenesulfonyl, 2-naphthalenesulfonyl, and benzyl moieties. Most 7-O-acylated-4-hydroxycoumarin (3), stilbene-coumarin (4), and flavanone (6) derivatives lacked cytotoxicity for all cell lines under tested concentrations.

Because of their intriguing cytotoxicity, it is required to build a complete structure-activity relationship (SAR) for these molecules. The summarisation of SAR studies is shown in Fig. 2. Our findings showed that functionalising DMC on 4'-OH significantly increased cytotoxicity while decreasing cytotoxicity when both 2'- and 4'-OH were derivatised. Importantly, our results suggest 2'-OH is required for anticancer activity. Moreover, among 4'-O-acylated-DMC derivatives (2a-2e), the cytotoxicity increased with the increment of the number of carbon atoms and reached a maximum at six carbons, whereas a shorter or longer chain resulted in a decrease in activity.

Additionally, methylation, benzylation, benzenesulfonylation, naphthalenesulfonylation, and naphthoylation on 4'-OH showed remarkable cytotoxicity.

In silico molecular docking study

The molecular docking study was performed on AutoDock Vina to understand the binding modes of DMC derivatives 2b, 2g, 2h, and 5g on the cyclin-dependent kinase 2 (CDK2) which is related to cell cycle regulation to rationalise their promising cytotoxicity. According to previous reports, key amino acid residues of CDK2 are located at Thr14, Lys33, Glu81, Leu83,

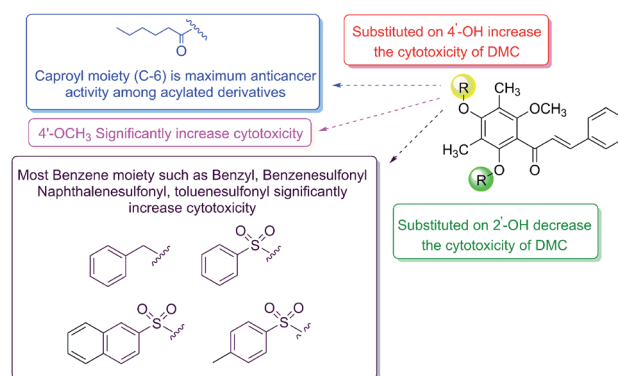


Fig. 2 The structure-activity relationships (SAR) study of 2',4'-disubstituted-6'-methoxy-3',5'-dimethylchalcone derivatives on the anticancer activity.



Asp86, Asp127, Asn132, and Asp145 of the ATP binding pocket.^{44,45}

CDK2 in complex with inhibitor RC-3-89 (PDB ID 4GCJ) was employed first. The molecular docking protocol was validated by redocking of a co-crystallised ligand to the binding pocket, as indicated by an RMSD value of 0.976 Å between the docking pose and the co-crystallised ligand with a binding energy value of $-9.7 \text{ kcal mol}^{-1}$ as well as all key interactions being identical (see Fig. 1S†).

The docking score and the protein–ligand interactions are shown in Table 3. The molecular docking study revealed that our compounds can accommodate the binding pocket of CDK2 with binding energies ranging from -7.1 to $-9.6 \text{ kcal mol}^{-1}$, which included H-bonding interactions with key amino acid residues of the receptor. Moreover, the predicted binding interactions are shown in Fig. 3.

As a control, DMC can accommodate the same ATP binding site as the co-crystallised inhibitor RC-3-89 with a binding energy value of $-8.4 \text{ kcal mol}^{-1}$. Hydrogen bonding interactions of 2'-OH with the key amino acid Asn132 ($\text{OH}\cdots\text{O}=\text{C}-\text{Asn132}$, 2.62 Å) and Asp145 ($\text{OH}\cdots\text{O}=\text{C}-\text{Asp145}$, 2.57 Å) backbones are crucial for the interactions (see Fig. 2S†).

The binding energy of 4'-O-caproylated-DMC (2b) with CDK2 was $-8.2 \text{ kcal mol}^{-1}$. Several hydrogen bonding interactions can be deduced from the docking pose, including the interaction between α,β -unsaturated ketone and Leu83 residue ($\text{C}=\text{O}\cdots\text{HN}-\text{Leu83}$, 4.61 Å), and carbon–hydrogen interactions with Asp86 ($\text{C}\cdots\text{O}=\text{C}-\text{Asp86}$, 3.63 Å) and Gln131 ($\text{C}\cdots\text{O}=\text{C}-\text{Gln131}$, 3.69 Å) residues. Furthermore, the docking studies also suggest hydrophobic interactions between the alkanoyl moiety and various residues such as Val18 and Lys33 which may play an important role. The binding energy of 4'-O-methylated-DMC (2g) with CDK2 of $-9.4 \text{ kcal mol}^{-1}$ could be calculated. Hydrogen bonding interactions between α,β -unsaturated $\text{C}=\text{O}$ ketone and the key amino acid Leu83 residue ($\text{C}=\text{O}\cdots\text{HN}-\text{Leu83}$, 2.33 Å), and the methoxy group with Asp145 residue ($\text{C}=\text{O}\cdots\text{HN}-\text{Asp145}$, 3.03 Å) were observed. Additionally, the interactions with Ile10, Val18, Ala31, Phe80, Leu134, and Ala144 residues through hydrophobic interaction were also noted. 4'-O-Benzylated-DMC (2h) showed the lowest binding energy among DMC derivatives with a value of $-9.6 \text{ kcal mol}^{-1}$. The hydrogen bonding interaction between 2'-OH and key amino acid Asp86 residue ($\text{OH}\cdots\text{OC}-\text{Asp86}$, 2.58 Å) was noticed. Besides, it was found that the phenyl ring B interacted with Glu12 through π -

anion interaction. The attached benzyl group also interacted with Phe80, Ala31, Val64, Ala144, and Leu134 residues. In 2',4'-O-dimethylated-DMC (5g), the oxygen atom of the methoxy groups interacted with key amino acids Thr14 ($\text{O}\cdots\text{HN}-\text{Thr14}$, 4.14 Å) and Lys33 ($\text{O}\cdots\text{HN}-\text{Lys33}$, 4.52 Å) through hydrogen bonding interactions. Furthermore, multiple interactions were observed including carbon–hydrogen interactions between the incorporated methoxy carbon and Asp127 residue ($\text{C}\cdots\text{O}=\text{C}-\text{Asp127}$, 3.26 Å) and electrostatic interactions between the phenyl ring and the key amino acids *i.e.* Asp 127 and Asp145 through π -anion interactions with bond distances of 4.66 and 3.17 Å, respectively.

We also found that 7-O-acylated-4-hydroxycoumarin derivatives (3) cannot accommodate the binding pocket of CDK2. The two new chiral positions resulted in a loss of flat geometry and sterically hindered access to the binding site, which is consistent with the lack of cytotoxicity (see Fig. 3S†). Most stilbene-coumarin derivatives showed excellent binding energy values to the CDK2 binding site. However, their diminished anticancer activities may be due to their low solubility (see Fig. 4S†).

In vitro antimicrobial activity

The DMC derivatives were further evaluated for antimicrobial activities against *S. aureus*, *E. coli*, and *P. aeruginosa*. In the broth microdilution method, DMC demonstrated moderate antimicrobial activity against all bacterial strains, with MIC values of 2.50, 1.25, and 1.25 mg mL^{-1} for *S. aureus*, *E. coli*, and *P. aeruginosa*, respectively, while the MBC values were higher than 5.00 mg mL^{-1} . In addition, the DMC displayed a zone of inhibition (ZOI) with diameter values of 9, 8, and 8 mm for *S. aureus*, *E. coli*, and *P. aeruginosa*, respectively. However, most DMC derivatives may decrease in antimicrobial activity in comparison to unmodified DMC and they were lacking in activity in some derivatives. For more details see Table 2S.†

Experimental section

General experimental methods

All melting points were determined on a Gallenkamp Electrothermal apparatus. Optical rotations were measured at room temperature in a 1 cm cell on a Rudolph research analytical automatic polarimeter at the sodium D-line (589 nm) reported as $[\alpha]_{589}^T$ (c in g per 100 mL, solvent). FTIR spectra were recorded on a Bruker TENSER 27 spectrometer and the wavenumbers of

Table 3 Protein–ligand interaction profile results of molecular docking of some DMC derivatives–CDK2 complexes

Compound	Binding energy (kcal mol^{-1})	Protein–ligand interactions		
		Hydrogen bonding	Electrostatic	Hydrophobic
2b	−8.2	Leu83, Asp86, Gln131	—	Ile10, Val18, Ala31, Lys33, Phe80, Leu134, Ala144
2g	−9.4	Leu83, Gln131, Asp145	—	Ile10, Val18, Ala31, Phe80, Leu134, Ala144
2h	−9.6	Asp86, Gln131	Glu12	Val18, Ala31, Val64, Phe80, Leu134, Ala144
5g	−7.1	Glu12, Thr14, Gly16, Lys33, Asp127, Gln131	Asp127, Asp145	Val18, Ala144
DMC (1)	−8.4	Asp86, Gln131, Asn132, Asp145	—	Ile10, Ala31, Leu134



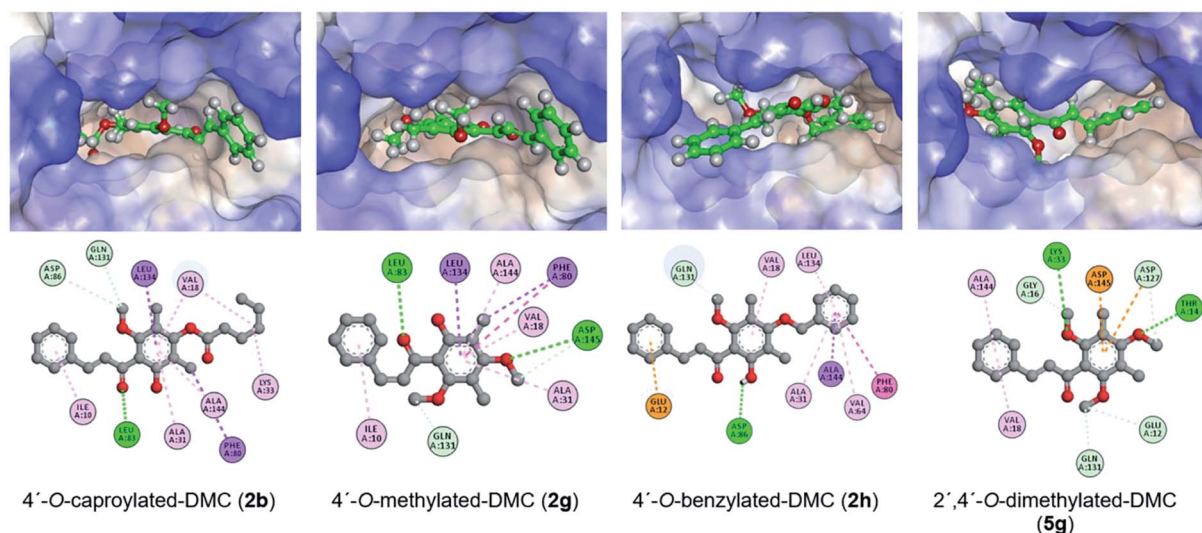


Fig. 3 Docking poses of compounds **2b**, **2g**, **2h** and **5g** at the ATP binding site of CDK2 (PDB ID 4GCJ) and ligand binding interactions.

the maximum absorption peaks are reported in cm^{-1} . The high-resolution mass spectra (HRMS, m/z values) were obtained from an ESI-QTOF-MS/MS Agilent model QTOF 6540 UHD mass spectrometer. $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ were recorded on a Bruker DRX-400 or NEOTM 500 spectrometer. Where necessary, 2D NMR experiments (HMBC and HMQC) were carried out to assist the structural elucidation and signal assignments. Chemical shifts (δ) were reported in ppm down-field from TMS as an internal reference or with the solvent resonance as the internal standard (CHCl_3 impurity in CDCl_3 , δ 7.26 and 77.16 ppm for $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$, respectively); data are reported in the following order: chemical shift, multiplicity and coupling constants (J) are given in hertz. Flash column chromatography was performed employing Merck silica gel 60H. The progress of chemical reactions was monitored by a thin layer chromatography (TLC), using Merck silica gel PF₂₅₄ aluminum plates. All the reactions were conducted in oven-dried glassware, under N_2 , with anhydrous solvents, which were dried and distilled before use according to standard procedures, and acid chlorides were freshly distilled under reduced pressure prior to use.

Extraction and isolation of DMC (1)

A previous study on the seeds of *S. nervosum* illustrated isolation and structural elucidation of 2',4'-dihydroxy-6'-methoxy-3',5'-dimethylchalcone (DMC).³⁷ A similar process was conducted to isolate DMC for the starting material in chemical modification. Briefly, 13.38 kg of air-dried ground seeds of *S. nervosum* were macerated with 40 L of EtOAc at ambient temperature (performed three times, each time for three days). The seed residues were filtered out and the filtrate was collected. Then it was concentrated under reduced pressure to afford 266.49 g of crude EtOAc extract as a dark-green viscous liquid. The extract was subjected to flash column chromatography on silica gel eluted with gradient eluents of *n*-hexane/EtOAc = 100 : 0 to 80 : 20. DMC-containing fractions were collected and further purified by flash column chromatography on silica gel with *n*-hexane/

acetone = 90 : 10 as an eluent to give the orange precipitates. The DMC was crystallised with $\text{CH}_2\text{Cl}_2/n$ -hexane to obtain 4.3615 g of orange needle-shaped crystals. The chemical structure was confirmed by comparison of its spectral data with the published values.³⁷

2',4'-Dihydroxy-6'-methoxy-3',5'-dimethylchalcone (1). Orange needle-shaped crystals; mp 124.5–126.4 °C ($\text{CH}_2\text{Cl}_2/n$ -hexane); ν_{max} (thin film) 3545 (O–H), 3033 (C–H), 2941 (–CH₃) 1632 (C=O), 1610 (C=C), 1423 (–CH₃), 1232 (C–O–C), 1170 (C–O–C) cm^{-1} ; δ_{H} (400 MHz, CDCl_3) 2.13 (s, 3H, *Me*), 2.15 (s, 3H, *Me*), 3.67 (s, 3H, *OMe*), 5.39 (s, 1H, *OH*), 7.36–7.45 (m, 3H, *ArCH*), 7.60–7.68 (m, 2H, *ArCH*), 7.84 (AB system, *d*, J = 15.7 Hz, 1H, *CH*), 7.99 (AB system, *d*, J = 15.7 Hz, 1H, *CH*), 13.61 (s, 1H, *OH*); δ_{C} (100 MHz, CDCl_3) 7.6, 8.2, 62.4, 106.6, 108.9, 109.0, 126.7, 128.4, 128.9, 130.2, 135.3, 142.9, 158.8, 159.2, 162.0, 193.4; HRMS (ESI) calcd for $\text{C}_{18}\text{H}_{18}\text{O}_4\text{Na}$ ($M + \text{Na}$)⁺: m/z 321.1103, found 321.1100.

General procedure for the synthesis of DMC derivatives

General procedure A for the syntheses of butyrylated DMC derivatives (2a and 3a). DMC (0.1057 g, 0.3543 mmol) was dissolved in 5 mL of dry CH_2Cl_2 . The solution was cooled down in an ice bath and then butyryl chloride (80.6 μL , 0.7794 mmol) was slowly added to the reaction mixture followed by the addition of triethylamine (108.6 μL , 0.7794 mmol). The reaction was continuously stirred for 10 minutes at ambient temperature, until the completion of reaction monitored by TLC, pH was adjusted to pH 4–5 using 6 M HCl solution and subsequently extracted several times with CH_2Cl_2 . The combined organic layer was dried over anhydrous Na_2SO_4 . After solvent removal, the crude product was purified by preparative thin-layer chromatography on silica gel using $\text{Et}_2\text{O}/\text{CH}_2\text{Cl}_2/n$ -hexane = 4 : 20 : 76 as eluent to obtain compounds **2a** and **3a** in 42% (0.0548 g) and 45% (0.0698 g) yields, respectively.

2'-Hydroxy-4'-butyryloxy-6'-methoxy-3',5'-dimethylchalcone (2a). Orange needle-shaped crystals; mp 102.0–103.9 °C (MeOH/



CH₂Cl₂); R_f (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.19; ν_{max} (thin film) 3034 (C–H), 2971 (–CH₃), 2941 (–CH₂), 2880 (–CH₃), 1762 (C=O), 1636 (C=O), 1605 (C=C), 1191 (C–O–C), 1143 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 1.08 (t, *J* = 7.4 Hz, 3H, *Me*), 1.79–1.88 (m, 2H, CH₂), 2.04 (s, 3H, *Me*), 2.05 (s, 3H, *Me*), 2.62 (t, *J* = 7.4 Hz, 2H, CH₂), 3.67 (s, 3H, *OMe*), 7.45–7.37 (m, 3H, *ArCH*), 7.62–7.68 (m, 2H, *ArCH*), 7.87 (AB system, *d*, *J* = 15.7 Hz, 1H, *CH*), 7.95 (AB system, *d*, *J* = 15.7 Hz, 1H, *CH*), 12.96 (s, 1H, *OH*); δ_C (125 MHz, CDCl₃) 9.1, 9.3, 13.9, 18.6, 36.0, 62.7, 113.3, 115.3, 115.8, 126.5, 128.7, 129.1, 130.6, 135.2, 143.9, 154.3, 158.3, 160.9, 170.8, 194.5; HRMS (ESI) calcd for C₂₂H₂₄O₅ (M)⁺: *m/z* 368.1624, found 368.1616.

3*H*-3-Ethyl-4-hydroxy-4-[(*E*)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-butyryloxy coumarin (3a). Light yellow viscous liquid; R_f (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.07; ν_{max} (thin film) 3455 (O–H), 2973 (–CH₃), 2941 (–CH₂), 2882 (–CH₃), 1776 (C=O), 1603 (C=C), 1202 (C–O–C), 1149 (C–O–C), 1102 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 1.00 (t, *J* = 7.5 Hz, 3H, CH₃), 1.09 (t, *J* = 7.5 Hz, 3H, CH₃), 1.28–1.38 (m, 1H, *CH*), 1.80–1.89 (m, 2H, CH₂), 2.06 (s, 3H, *Me*), 2.10 (s, 3H, *Me*), 2.12–2.24 (m, 1H, *CH*), 2.63 (t, *J* = 7.4 Hz, 2H, CH₂), 2.78 (dd, *J* = 11.1, 4.0 Hz, 1H, *CH*), 3.73 (s, 3H, *OMe*), 5.66 (br s, 1H, *OH*), 6.36 (d, *J* = 15.8 Hz, 1H, *CH*), 6.46 (br d, *J* = 15.8 Hz, 1H, *CH*), 7.20–7.36 (m, 5H, *ArCH*); δ_C (125 MHz, CDCl₃) 9.7, 9.9, 13.9, 14.0, 20.4, 24.8, 34.2, 52.0, 62.1, 75.3, 115.8, 116.5, 120.7, 127.1, 128.1, 128.7, 129.5, 133.4, 135.8, 146.3, 149.4, 154.7, 168.8, 171.3; HRMS (ESI) calcd for C₂₆H₃₀O₆ (M)⁺: *m/z* 438.2042, found 438.2039.

Syntheses of caproylated DMC derivatives. DMC derivatives **2b** and **3b** were synthesised following general procedure A. The starting material DMC (0.0985 g, 0.3302 mmol) and caproyl chloride (101.5 μL, 0.7264 mmol) were dissolved in 5 mL of dry CH₂Cl₂ in the presence of NEt₃ (101.2 μL, 0.7264 mmol) to afford compounds **2b** and **3b** in 21% (0.0279 g) and 67% (0.1094 g) yields, respectively.

2'-Hydroxy-4'-caproyloxy-6'-methoxy-3',5'-dimethylchalcone (2b). Orange needle-shaped crystals; mp 70.0–71.6 °C (MeOH/CH₂Cl₂); R_f (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.20; ν_{max} (thin film) 2962 (–CH₃), 2938 (–CH₂), 1765 (C=O), 1637 (C=O), 1607 (C=C), 1570 (C=C), 1284 (C–O–C), 1135 (C–O–C), 1115 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 0.94 (t, *J* = 6.8 Hz, 3H, *Me*), 1.34–1.49 (m, 4H, CH₂), 1.75–1.86 (m, 2H, CH₂), 2.04 (s, 3H, *Me*), 2.05 (s, 3H, *Me*), 2.63 (t, *J* = 7.3 Hz, 2H, CH₂), 3.67 (s, 3H, *OMe*), 7.39–7.47 (m, 3H, *ArCH*), 7.62–7.69 (m, 2H, *ArCH*), 7.87 (AB system, *d*, *J* = 15.7 Hz, 1H, *CH*), 7.95 (AB system, *d*, *J* = 15.7 Hz, 1H, *CH*), 12.97 (s, 1H, *OH*); δ_C (125 MHz, CDCl₃) 9.1, 9.3, 14.0, 22.4, 24.8, 31.5, 34.1, 62.7, 113.4, 115.3, 115.8, 126.5, 128.7, 129.1, 130.7, 135.2, 144.0, 154.3, 158.3, 160.9, 171.0, 194.5; HRMS (ESI) calcd for C₂₄H₂₈O₅ (M)⁺: *m/z* 396.1937, found 396.1935.

3*H*-3-Butyl-4-hydroxy-4-[(*E*)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-caproyloxy coumarin (3b). Light yellow viscous liquid; R_f (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.12; ν_{max} (thin film) 3469 (O–H), 2963 (–CH₃), 2937 (–CH₂), 2876 (–CH₃), 1779 (C=O), 1602 (C=C), 1219 (C–O–C), 1145 (C–O–C), 1102 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 0.87 (t, *J* = 7.0 Hz, 3H, *Me*), 0.95 (t, *J* = 6.9 Hz, 3H, *Me*), 1.18–1.51 (m, 9H, CH₂), 1.76–1.86 (m, 2H, CH₂), 2.07 (s, 3H, *Me*), 2.10 (s, 3H, *Me*), 2.10–2.16 (m, 1H, *CH*), 2.64 (t, *J* = 7.5 Hz, 2H, CH₂), 2.84 (dd, *J* = 10.7, 3.8 Hz, 1H, *CH*),

3.73 (s, 3H, *OMe*), 5.68 (br s, 1H, *OH*), 6.36 (AB system, *d*, *J* = 15.8 Hz, 1H, *CH*), 6.46 (AB system, br *d*, *J* = 15.8 Hz, 1H, *CH*), 7.19–7.40 (m, 5H, *ArCH*); δ_C (125 MHz, CDCl₃) 9.7, 9.9, 13.9, 14.0, 22.4, 22.5, 24.8, 25.8, 29.6, 31.5, 34.0, 52.0, 62.1, 75.3, 115.8, 116.4, 120.7, 127.0, 128.2, 128.7, 129.5, 133.4, 135.9, 146.3, 149.3, 154.7, 168.8, 171.2; HRMS (ESI) calcd for C₃₀H₃₈O₆ (M)⁺: *m/z* 494.2668, found 494.2666.

Syntheses of capryloylated DMC derivatives. DMC derivatives **2c** and **3c** were synthesised following general procedure A. DMC (0.0925 g, 0.3100 mmol) and capryloyl chloride (114.7 μL, 0.6821 mmol) were dissolved in dry CH₂Cl₂ in the presence of NEt₃ (95.1 μL, 0.6821 mmol) to obtain compounds **2c** and **3c** in 40% (0.0526 g) and 42% (0.0717 g) yields, respectively.

2'-Hydroxy-4'-capryloyloxy-6'-methoxy-3',5'-dimethylchalcone (2c). Orange needle-shaped crystals; mp 57.2–59.0 °C (MeOH/CH₂Cl₂); R_f (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.21; ν_{max} (thin film) 3033 (C–H), 2960 (–CH₃), 2935 (–CH₂), 2862 (–CH₃), 1766 (C=O), 1637 (C=O), 1607 (C=C), 1570 (C=C), 1284 (C–O–C), 1136 (C–O–C), 1115 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 0.90 (t, *J* = 6.7 Hz, 3H, *Me*), 1.28–1.51 (m, 8H, CH₂), 1.76–1.86 (m, 2H, CH₂), 2.04 (s, 3H, *Me*), 2.05 (s, 3H, *Me*), 2.63 (t, *J* = 7.5 Hz, 2H, CH₂), 3.67 (s, 3H, *OMe*), 7.38–7.50 (m, 3H, *ArCH*), 7.62–7.69 (m, 2H, *ArCH*), 7.88 (AB system, *d*, *J* = 15.7 Hz, 1H, *CH*), 7.95 (AB system, *d*, *J* = 15.7 Hz, 1H, *CH*), 12.97 (s, 1H, *OH*); δ_C (125 MHz, CDCl₃) 9.1, 9.3, 14.2, 22.7, 25.2, 29.0, 29.3, 29.8, 31.8, 34.1, 62.8, 113.4, 115.3, 115.8, 126.5, 128.7, 129.1, 130.7, 135.2, 144.0, 154.3, 158.3, 160.9, 171.0, 194.5; HRMS (ESI) calcd for C₂₆H₃₂O₅ (M)⁺: *m/z* 424.2250, found 424.5546.

3*H*-3-Hexyl-4-hydroxy-4-[(*E*)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-capryloyloxy coumarin (3c). Light yellow viscous liquid; R_f (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.14; ν_{max} (thin film) 3471 (O–H), 2961 (–CH₃), 2934 (–CH₂), 2863 (–CH₃), 1780 (C=O), 1603 (C=C), 1206 (C–O–C), 1144 (C–O–C), 1103 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 0.80–0.95 (m, 6H, *Me*), 1.17–1.50 (m, 17H, CH₂), 1.75–1.85 (m, 2H, CH₂), 2.07 (s, 3H, *Me*), 2.08–2.15 (m, 1H, *CH*), 2.10 (s, 3H, *Me*), 2.64 (t, *J* = 7.5 Hz, 2H, CH₂), 2.85 (dd, *J* = 10.7, 3.7 Hz, 1H, *CH*), 3.72 (s, 3H, *OMe*), 5.67 (br s, 1H, *OH*), 6.36 (AB system, *d*, *J* = 15.8 Hz, 1H, *CH*), 6.46 (AB system, br *d*, *J* = 15.8 Hz, 1H, *CH*), 7.18–7.36 (m, 5H, *ArCH*); δ_C (125 MHz, CDCl₃) 9.7, 9.8, 14.1, 14.2, 22.6, 22.7, 25.1, 26.1, 27.4, 29.0, 29.1, 29.3, 31.6, 31.7, 34.0, 52.0, 62.0, 75.3, 115.7, 116.4, 120.7, 127.0, 128.2, 128.6, 129.5, 133.4, 135.8, 146.3, 149.3, 154.7, 168.8, 171.2; HRMS (ESI) calcd for C₃₄H₄₆O₆ (M)⁺: *m/z* 550.3294, found 550.3295.

Syntheses of caprinoylated DMC derivatives. DMC derivatives **2d** and **3d** were synthesised following general procedure A. A mixture of DMC (0.1106 g, 0.3707 mmol) and caprinoyl chloride (169.2 μL, 0.8156 mmol) in 5 mL of dry CH₂Cl₂ was slowly added NEt₃ (113.7 μL, 0.8156 mmol) to obtain compounds **2d** and **3d** in 36% (0.0604 g) and 49% (0.1102 g), respectively.

2'-Hydroxy-4'-caprinoyloxy-6'-methoxy-3',5'-dimethylchalcone (2d). Orange needle-shaped crystals; mp 45.6–46.3 °C (MeOH/CH₂Cl₂); R_f (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.22; ν_{max} (thin film) 3033 (C–H), 2960 (–CH₃), 2931 (–CH₂), 2860 (–CH₃), 1767 (C=O), 1637 (C=O), 1607 (C=C), 1570 (C=C), 1135 (C–O–C), 1114 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 0.89 (t,



$J = 6.7$ Hz, 3H, Me), 1.21–1.49 (m, 12H, CH₂), 1.75–1.86 (m, 2H, CH₂), 2.03 (s, 3H, Me), 2.05 (s, 3H, Me), 2.63 (t, $J = 7.6$ Hz, 2H, CH₂), 3.67 (s, 3H, OMe), 7.38–7.46 (m, 3H, ArCH), 7.61–7.69 (m, 2H, ArCH), 7.88 (AB system, d, $J = 15.7$ Hz, 1H, CH), 7.94 (AB system, d, $J = 15.7$ Hz, 1H, CH), 12.97 (s, 1H, OH); δ_{C} (125 MHz, CDCl₃) 9.1, 9.3, 14.2, 22.8, 25.2, 29.3, 29.4, 29.4, 29.6, 32.0, 34.1, 62.7, 113.4, 115.3, 115.8, 126.5, 128.7, 129.1, 130.7, 135.2, 144.0, 154.3, 158.3, 160.9, 171.0, 194.5; HRMS (ESI) calcd for C₂₈H₃₆O₅ (M)⁺: m/z 452.2563, found 452.2558.

3H-3-Octyl-4-hydroxy-4-[(E)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-caprinoyloxycoumarin (3d). Light yellow viscous liquid; R_{f} (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.17; ν_{max} (thin film) 3470 (O–H), 2960 (–CH₃), 2931 (–CH₂), 2860 (–CH₃), 1779 (C=O), 1603 (C=C), 1198 (C–O–C), 1143 (C–O–C), 1101 (C–O–C) cm^{–1}; δ_{H} (500 MHz, CDCl₃) 0.81–0.92 (m, 6H, Me), 1.15–1.50 (m, 25H, CH₂), 1.75–1.85 (m, 2H, CH₂), 2.06 (s, 3H, Me), 2.07–2.14 (m, 1H, CH), 2.10 (s, 3H, Me), 2.64 (t, $J = 7.5$ Hz, 2H, CH₂), 2.84 (dd, $J = 10.7$, 3.8 Hz, 1H, CH), 3.73 (s, 3H, OMe), 5.68 (br s, 1H, OH), 6.36 (AB system, d, $J = 15.7$ Hz, 1H, CH), 6.46 (AB system, br d, $J = 15.7$ Hz, 1H, CH), 7.18–7.38 (m, 5H, ArCH); δ_{C} (125 MHz, CDCl₃) 9.8, 9.9, 14.2, 22.7, 22.8, 25.1, 26.1, 27.4, 29.2, 29.3, 29.4, 29.5, 29.6, 31.9, 32.0, 34.0, 52.0, 62.1, 75.2, 115.6, 116.3, 120.6, 126.8, 128.1, 128.6, 129.4, 133.3, 135.7, 146.1, 149.2, 154.6, 168.7, 171.1; HRMS (ESI) calcd for C₃₈H₅₄O₆ (M)⁺: m/z 606.3920, found 606.3923.

Syntheses of lauroylated DMC derivatives. DMC derivatives **2e** and **3e** were synthesised following general procedure A. To a mixture of starting material DMC (0.0806 g, 0.2702 mmol) and lauroyl chloride (137.4 μ L, 0.5944 mmol) dissolved in 5 mL of dry CH₂Cl₂ was slowly added NEt₃ (82.8 μ L, 0.5944 mmol) to obtain compounds **2e** and **3e** in 28% (0.0364 g) and 63% (0.1128 g), respectively.

2'-Hydroxy-4'-lauroyloxy-6'-methoxy-3',5'-dimethylchalcone (2e). Orange needle-shaped crystals; mp 53.6–55.0 °C (MeOH/CH₂Cl₂); R_{f} (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.24; ν_{max} (thin film) 3032 (C–H), 2959 (–CH₃), 2930 (–CH₂), 2859 (–CH₃), 1767 (C=O), 1637 (C=O), 1607 (C=C), 1570 (C=C), 1136 (C–O–C), 1113 (C–O–C) cm^{–1}; δ_{H} (500 MHz, CDCl₃) 0.88 (t, $J = 6.7$ Hz, 3H, Me), 1.18–1.50 (m, 16H, CH₂), 1.75–1.86 (m, 2H, CH₂), 2.04 (s, 3H, Me), 2.05 (s, 3H, Me), 2.63 (t, $J = 7.6$ Hz, 2H, CH₂), 3.67 (s, 3H, OMe), 7.38–7.46 (m, 3H, ArCH), 7.61–7.69 (m, 2H, ArCH), 7.88 (AB system, d, $J = 15.7$ Hz, 1H, CH), 7.94 (AB system, d, $J = 15.7$ Hz, 1H, CH), 12.97 (s, 1H, OH); δ_{C} (125 MHz, CDCl₃) 9.1, 9.3, 14.3, 22.8, 25.2, 29.4, 29.5, 29.6, 29.7, 32.0, 34.1, 62.7, 113.3, 115.3, 115.8, 126.5, 128.7, 129.1, 130.6, 135.2, 144.0, 154.3, 158.3, 160.9, 171.0, 194.5; HRMS (ESI) calcd for C₃₀H₄₀O₅ (M)⁺: m/z 480.2876, found 480.2873.

3H-3-Decyl-4-hydroxy-4-[(E)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-lauroyloxycoumarin (3e). Light yellow viscous liquid; R_{f} (30% CH₂Cl₂/2% Et₂O/*n*-hexane) 0.18; ν_{max} (thin film) 3479 (O–H), 2961 (–CH₃), 2926 (–CH₂), 2860 (–CH₃) 1784 (C=O), 1604 (C=C), 1204 (C–O–C), 1143 (C–O–C), 1107 (C–O–C) cm^{–1}; δ_{H} (500 MHz, CDCl₃) 0.83–0.93 (m, 6H, Me), 1.15–1.50 (m, 33H, CH₂), 1.75–1.85 (m, 2H, CH₂), 2.06 (s, 3H, Me), 2.07–2.14 (m, 1H, CH), 2.10 (s, 3H, Me), 2.64 (t, $J = 7.5$ Hz, 2H, CH₂), 2.84 (dd, $J = 10.7$, 3.8 Hz, 1H, CH), 3.72 (s, 3H, OMe), 5.67 (br s, 1H, OH), 6.36 (AB system, d, $J = 16.1$ Hz, 1H, CH), 6.46 (AB system, br d, $J =$

16.1 Hz, 1H, CH), 7.17–7.37 (m, 5H, ArCH); δ_{C} (125 MHz, CDCl₃) 9.7, 9.8, 14.2, 22.7, 22.8, 25.1, 26.1, 27.4, 29.3, 29.3, 29.4, 29.4, 29.5, 29.6, 29.6, 29.7, 29.8, 32.0, 32.1 34.0, 52.0, 62.1, 75.3, 115.7, 116.4, 120.7, 126.9, 128.2, 128.6, 129.4, 133.4, 135.8, 146.2, 149.2, 154.7, 168.8, 171.1; HRMS (ESI) calcd for C₄₂H₆₂O₆ (M)⁺: m/z 662.4546, found 662.4542.

General procedure B for syntheses of acetylated DMC derivatives (2f and 5f). DMC (0.1100 g, 0.3687 mmol) was dissolved in 10 mL of acetone in the presence of anhydrous K₂CO₃ (0.5069 g, 3.6871 mmol). The mixture was stirred for 15 minutes followed by the addition of acetyl chloride (0.2631 mL, 3.6871 mmol) and continuously stirred for 24 h at ambient temperature. Until the completion of the reaction, the K₂CO₃ residue was filtered out. The filtrate was subsequently concentrated under reduced pressure and acidified with 6 M HCl followed by extraction several times with CH₂Cl₂. The combined organic layer was dried over anhydrous Na₂SO₄. After solvent removal, the crude product was subjected to column chromatography on silica gel using EtOAc/*n*-hexane = 1 : 10 as eluent to obtain compounds **2f** and **5f** in 22% yield (0.0276 g) and 36% (0.0507 g) yields, respectively.

2'-Hydroxy-4'-acetyloxy-6'-methoxy-3',5'-dimethylchalcone (2f). Orange needle-shaped crystals; mp 131.5–132.7 °C (CH₂Cl₂/MeOH); R_{f} (5% EtOAc/7.5% acetone/*n*-hexane) 0.55; ν_{max} (thin film) 2943 (–CH₃), 1760 (C=O), 1635 (C=O), 1607 (C=C), 1562 (C=C), 1211 (C–O–C), 1112 (C–O–C) cm^{–1}; δ_{H} (500 MHz, CDCl₃) 2.05 (s, 3H, Me), 2.06 (s, 3H, Me), 2.38 (s, 3H, Me), 3.67 (s, 3H, OMe), 7.37–7.46 (m, 3H, ArCH), 7.62–7.69 (m, 2H, ArCH), 7.88 (AB system, d, $J = 15.7$ Hz, 1H, CH), 7.94 (AB system, d, $J = 15.7$ Hz, 1H, CH), 12.95 (s, 1H, OH); δ_{C} (125 MHz, CDCl₃) 9.1, 9.3, 20.6, 62.8, 113.5, 115.3, 115.8, 126.5, 128.8, 129.2, 130.7, 135.2, 144.0, 154.2, 158.3, 160.9, 168.2, 194.5; HRMS (ESI) calcd for C₂₀H₂₀O₅ (M)⁺: m/z 340.1311, found 340.1308.

2',4'-Diacyloxy-6'-methoxy-3',5'-dimethylchalcone (5f). Yellow viscous liquid; R_{f} (5% EtOAc/7.5% acetone/*n*-hexane) 0.50; ν_{max} (thin film) 3062 (C–H), 2945 (–CH₃), 1773 (C=O), 1653 (C=O), 1607 (C=C), 1190 (C–O–C), 1180 (C–O–C), 1105 (C–O–C) cm^{–1}; δ_{H} (500 MHz, CDCl₃) 1.94 (s, 3H, Me), 2.10 (s, 3H, Me), 2.17 (s, 3H, Me), 2.38 (s, 3H, Me), 3.70 (s, 3H, OMe), 7.02 (d, $J = 16.1$ Hz, 1H, CH), 7.36–7.42 (m, 3H, ArCH), 7.46 (d, $J = 16.1$ Hz, 1H, CH), 7.53–7.58 (m, 2H, ArCH); δ_{C} (125 MHz, CDCl₃) 9.9, 10.3, 20.5, 20.6, 62.7, 120.8, 123.0, 125.8, 127.4, 128.8, 129.1, 130.9, 134.5, 144.9, 146.3, 150.2, 154.8, 168.3, 168.7, 192.7; HRMS (ESI) calcd for C₂₂H₂₂O₆ (M)⁺: m/z 382.1416, found 382.1414.

Syntheses of methylated DMC derivatives. Compounds **2g**, **5g**, and **6a** were synthesised following general procedure B. The substrate DMC (0.1004 g, 0.3365 mmol) was stirred with methyl iodide (0.2095 mL, 3.3653 mmol) and K₂CO₃ (0.4651 g, 3.3653 mmol) dissolved in 5 mL of acetone to obtain compounds **2g**, **5g**, and **6a** in 13% (0.0137 g), 45% (0.0494 g), and 6% (0.0064 g) yields, respectively.

2'-Hydroxy-4',6'-dimethoxy-3',5'-dimethylchalcone (2g). Orange viscous liquid; R_{f} (15% CH₂Cl₂/15% acetone/*n*-hexane) 0.42; ν_{max} (thin film) 3032 (C–H), 2940 (–CH₃), 2862 (–CH₃), 1635 (C=O), 1567 (C=C), 1453 (–CH₃), 1145 (C–O–C), 1114 (C–O–C) cm^{–1}; δ_{H} (500 MHz, CDCl₃) 2.17 (s, 3H, Me), 2.18 (s, 3H, Me),



3.66 (s, 3H, OMe), 3.76 (s, 3H, OMe), 7.39–7.46 (m, 3H, ArCH), 7.62–7.68 (m, 2H, ArCH), 7.87 (AB system, d, $J = 15.6$ Hz, 1H, CH), 7.97 (AB system, d, $J = 15.6$ Hz, 1H, CH), 13.09 (s, 1H, OH); δ_C (125 MHz, CDCl₃) 8.7, 8.8, 60.1, 62.3, 111.9, 115.7, 115.8, 126.6, 128.4, 129.0, 130.4, 135.2, 143.4, 158.7, 161.7, 163.6, 194.1; HRMS (ESI) calcd for C₁₉H₂₀O₄ (M)⁺: m/z 312.1362, found 312.1359.

2',4',6'-Trimethoxy-3',5'-dimethylchalcone (5g). Orange viscous liquid; R_f (5% CH₂Cl₂/10% acetone/*n*-hexane) 0.74; ν_{\max} (thin film) 3064 (C–H), 2944 (–CH₃), 1652 (C=O), 1584 (C=C), 1456 (–CH₃), 1197 (C–O–C), 1151 (C–O–C), 1110 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 2.21 (s, 6H, Me), 3.69 (s, 6H, OMe), 3.76 (s, 3H, OMe), 7.04 (d, $J = 16.1$ Hz, 1H, CH), 7.35–7.38 (m, 4H, ArCH and CH), 7.52–7.55 (m, 2H, ArCH); δ_C (125 MHz, CDCl₃) 9.4, 60.2, 62.2, 120.9, 125.0, 128.6, 128.7, 129.0, 130.7, 134.7, 145.7, 154.6, 159.4, 195.1; HRMS (ESI) calcd for C₂₀H₂₂O₄ (M)⁺: m/z 326.1518, found 326.1517.

5,7-Dimethoxy-6,8-dimethylflavanone (6g). Yellow viscous liquid; R_f (10% EtOAc/5% acetone/*n*-hexane) 0.41; ν_{\max} (thin film) 3070 (C–H), 2939 (–CH₃), 2861 (–CH₂) 1687 (C=O), 1591 (C=C), 1203 (C–O–C), 1157 (C–O–C), 1116 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 2.17 (s, 6H, Me), 2.85 (ABX system, dd, $J = 16.6$, 3.1 Hz, 1H, CH), 2.99 (ABX system, dd, $J = 16.6$, 13.1 Hz, 1H, CH), 3.75 (s, 3H, OMe), 3.82 (s, 3H, OMe), 5.42 (ABX system, dd, $J = 13.1$, 3.1 Hz, 1H, CH), 7.35–7.51 (m, 5H, ArCH); δ_C (125 MHz, CDCl₃) 8.7, 9.2, 45.9, 60.2, 61.2, 78.7, 111.9, 115.8, 118.9, 125.9, 128.6, 128.9, 139.3, 157.8, 159.8, 163.6, 190.4; HRMS (ESI) calcd for C₁₉H₂₀O₄ (M)⁺: m/z 312.1362, found 312.1360.

Syntheses of benzylated DMC derivatives. Compounds **2h** and **5h** were prepared following general procedure B. DMC (0.1003 g, 0.3362 mmol) and benzyl bromide (0.3993 mL, 3.3620 mmol) were dissolved in 5 mL of acetone in the presence of K₂CO₃ (0.4646 g, 3.3620 mmol) to obtain compounds **2h** and **5h** in 51% (0.0666 g) and 15% (0.0241 g) yields, respectively.

2'-Hydroxy-4'-benzyloxy-6'-methoxy-3',5'-dimethylchalcone (2h). Orange viscous liquid; R_f (5% CH₂Cl₂/5% acetone/*n*-hexane) 0.37; ν_{\max} (thin film) 3036 (C–H), 2937 (–CH₃), 2873 (–CH₂), 1636 (C=O), 1603 (C=C), 1567 (C=C), 1152 (C–O–C), 1114 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 2.17 (s, 3H, Me), 2.19 (s, 3H, Me), 3.67 (s, 3H, OMe), 4.85 (s, 2H, CH₂), 7.34–7.51 (m, 8H, ArCH), 7.63–7.69 (m, 2H, ArCH), 7.87 (AB system, d, $J = 15.7$ Hz, 1H, CH), 7.98 (AB system, d, $J = 15.7$ Hz, 1H, CH), 13.11 (s, 1H, OH); δ_C (125 MHz, CDCl₃) 9.2, 9.4, 62.5, 74.8, 112.1, 116.1, 116.2, 126.7, 128.1, 128.4, 128.7, 128.8, 129.1, 130.5, 135.4, 137.0, 143.6, 158.9, 161.8, 162.5, 194.2; HRMS (ESI) calcd for C₂₅H₂₄O₄ (M)⁺: m/z 388.1675, found 388.1672.

2',4'-Dibenzoyloxy-6'-methoxy-3',5'-dimethylchalcone (5h). Yellow viscous liquid; R_f (10% CH₂Cl₂/5% acetone/*n*-hexane) 0.39; ν_{\max} (thin film) 3032 (C–H), 2938 (–CH₃), 2858 (–CH₂), 1632 (C=O), 1564 (C=C), 1170 (C–O–C), 1114 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 2.21 (s, 3H, Me), 2.24 (s, 3H, Me), 3.72 (s, 3H, OMe), 4.84 (s, 2H, CH₂), 4.85 (s, 2H, CH₂), 7.06 (d, $J = 16.0$ Hz, CH), 7.22–7.56 (m, 16H, ArCH and CH); δ_C (125 MHz, CDCl₃) 9.8, 10.1, 62.3, 74.5, 76.8, 121.5, 121.7, 125.6, 128.0, 128.2, 128.2, 128.3, 128.5, 128.6, 128.7, 128.9, 130.6, 134.7, 137.0, 137.2, 145.7, 153.3, 154.8, 158.1, 195.1; HRMS (ESI) calcd for C₃₂H₃₀O₄ (M)⁺: m/z 478.2144, found 478.2140.

Syntheses of allylated DMC derivatives. Compounds **2i** and **5i** were synthesised from starting material DMC (0.1002 g, 0.3359 mmol) and allyl bromide (0.2906 mL, 3.3586 mmol) dissolved in 5 mL of acetone in the presence of K₂CO₃ (0.4642 g, 3.3586 mmol) following general procedure B to yield compounds **2i** and **5i** in 20% (0.0273 g) and 30% (0.0381 g), respectively.

2'-Hydroxy-4'-allyloxy-6'-methoxy-3',5'-dimethylchalcone (2i). Orange viscous liquid; R_f (5% CH₂Cl₂/5% acetone/*n*-hexane) 0.45; ν_{\max} (thin film) 3088 (C–H), 3031 (C–H), 2937 (–CH₃), 2869 (–CH₂), 1634 (C=O), 1606 (C=C), 1568 (C=C), 1412 (–CH₃), 1153 (C–O–C), 1113 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 2.16 (s, 3H, Me), 2.18 (s, 3H, Me), 3.66 (s, 3H, OMe), 4.34 (d, $J = 5.6$ Hz, 2H, CH₂), 5.30 (d, $J = 10.4$ Hz, 1H, CH₂), 5.45 (m, 1H, CH₂), 6.10 (m, 1H, CH), 7.38–7.45 (m, 3H, ArCH), 7.61–7.70 (m, 2H, ArCH), 7.87 (AB system, d, $J = 15.8$ Hz, 1H, CH), 7.97 (AB system, d, $J = 15.8$ Hz, 1H, CH), 13.09 (s, 1H, OH); δ_C (125 MHz, CDCl₃) 9.2, 9.3, 62.5, 73.8, 112.0, 116.0, 116.1, 118.0, 126.7, 128.7, 129.1, 130.5, 133.5, 135.4, 143.6, 158.8, 161.8, 162.7, 194.2; HRMS (ESI) calcd for C₂₁H₂₂O₄ (M)⁺: m/z 338.1518, found 338.1520.

2',4'-Diallyloxy-6'-methoxy-3',5'-dimethylchalcone (5i). Yellow viscous liquid; R_f (5% CH₂Cl₂/5% acetone/*n*-hexane) 0.45, ν_{\max} (thin film) 3088 (C–H), 2943 (–CH₃), 2871 (–CH₂), 1652 (C=O), 1631 (C=C), 1605 (C=C), 1454 (–CH₃), 1189 (C–O–C), 1136 (C–O–C), 1112 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 2.20 (s, 6H, Me), 3.69 (s, 3H, OMe), 4.31 (d, $J = 5.6$ Hz, 2H, CH₂), 4.34 (d, $J = 5.4$ Hz, 2H, CH₂), 5.15 (d, $J = 10.6$ Hz, 1H, CH₂), 5.24–5.35 (m, 2H, CH₂), 5.46 (d, $J = 17.1$ Hz, 1H, CH₂), 5.89–6.01 (m, 1H, CH), 6.05–6.17 (m, 1H, CH), 7.03 (d, $J = 16.1$ Hz, 1H, CH), 7.33–7.40 (m, 4H, ArCH and CH), 7.50–7.56 (m, 2H, ArCH); δ_C (125 MHz, CDCl₃) 9.7, 10.0, 62.3, 73.6, 75.7, 117.7, 121.2, 121.4, 125.2, 128.6, 128.7, 129.0, 130.6, 133.7, 133.8, 134.8, 145.6, 153.5, 154.6, 158.2, 195.1; HRMS (ESI) calcd for C₂₄H₂₆O₄ (M)⁺: m/z 378.1831, found 378.1835.

Synthesis of 2-naphthoylated DMC derivatives. Compound **2j** was synthesised following general procedure B. DMC (0.0500 g, 0.1676 mmol) and 2-naphthoyl chloride (0.3195 g, 1.6759 mmol) were dissolved in acetone in the presence of K₂CO₃ (0.2316 g, 1.6759 mmol) to afford compound **2j** in 74% yield (0.0561 g).

2'-Hydroxy-4'-(2-naphthoyloxy)-6'-methoxy-3',5'-dimethylchalcone (2j). Orange viscous liquid; R_f (5% CH₂Cl₂/5% EtOAc/*n*-hexane) 0.55; ν_{\max} (thin film) 3010 (C–H), 2928 (–CH₃), 1760 (C=O), 1638 (C=O), 1606 (C=C), 1564 (C=C), 1206 (C–O–C), 1136 (C–O–C), 1114 (C–O–C) cm^{–1}; δ_H (500 MHz, CDCl₃) 2.13 (s, 3H, Me), 2.14 (s, 3H, Me) 3.72 (s, 3H, OMe), 7.40–7.49 (m, 3H, ArCH), 7.59–7.63 (m, 1H, ArCH), 7.64–7.70 (m, 3H, ArCH), 7.88–8.06 (m, 5H, ArCH and CH), 8.19–8.26 (m, 1H, ArCH), 8.85 (br s, 1H, ArCH), 13.01 (s, 1H, OH); δ_C (125 MHz, CDCl₃) 9.3, 9.4, 62.8, 113.6, 115.6, 116.2, 125.6, 126.0, 126.6, 127.2, 128.1, 128.7, 128.8, 129.0, 129.2, 129.7, 130.7, 132.3, 132.7, 135.3, 136.2, 144.1, 154.5, 158.4, 161.1, 164.1, 194.6; HRMS (ESI) calcd for C₂₉H₂₄O₅ (M)⁺: m/z 452.1624, found 452.1622.

Syntheses of benzoylated DMC derivatives. Compounds **2k** and **5k** were synthesised from DMC (0.1004 g, 0.3365 mmol) and



benzoyl chloride (325.6 μL , 3.3653 mmol) dissolved in 5 mL of acetone in the presence of K_2CO_3 (0.4651 g, 0.3653 mmol) following general procedure B to afford compounds **2k** and **5k** in 39% (0.0528 g) and 42% (0.0716 g) yields, respectively.

2'-Hydroxy-4'-benzoyloxy-6'-methoxy-3',5'-dimethylchalcone (2k). Yellow viscous liquid; R_f (5% CH_2Cl_2 /5% acetone/*n*-hexane) 0.37; ν_{max} (thin film) 3067 (C–H), 2938 ($-\text{CH}_3$), 1743 (C=O), 1637 (C=O), 1606 (C=C), 1568 (C=C), 1137 (C–O–C), 1115 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 2.10 (s, 3H, *Me*), 2.11 (s, 3H, *Me*), 3.71 (s, 3H, *OMe*), 7.40–7.47 (m, 3H, *ArCH*), 7.56 (m, 2H, *ArCH*), 7.64–7.72 (m, 3H, *ArCH*), 7.91 (AB system, d, $J = 15.7$ Hz, 1H, *CH*), 7.98 (AB system, d, $J = 15.7$ Hz, 1H, H), 8.25 (d, $J = 7.7$ Hz, 2H, *ArCH*), 12.99 (s, 1H, *OH*); δ_{C} (125 MHz, CDCl_3) 9.2, 9.4, 62.8, 113.5, 115.6, 116.1, 126.5, 128.7, 128.8, 128.9, 129.1, 130.4, 130.7, 134.1, 135.2, 144.0, 154.4, 158.4, 161.0, 163.9, 194.5; HRMS (ESI) calcd for $\text{C}_{25}\text{H}_{22}\text{O}_5$ (M^+): m/z 402.1467, found 402.1465.

2',4'-Dibenzoyloxy-6'-methoxy-3',5'-dimethylchalcone (5k). Yellow viscous liquid; R_f (10% CH_2Cl_2 /10% acetone/*n*-hexane) 0.53; ν_{max} (thin film) 3068 (C–H), 2954 ($-\text{CH}_3$), 1744 (C=O), 1655 (C=O), 1606 (C=C), 1250 (C–O–C), 1182 (C–O–C), 1103 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 2.03 (s, 3H, *Me*), 2.19 (s, 3H, *Me*), 3.76 (s, 3H, *OMe*), 6.98 (d, $J = 16.2$ Hz, 1H, *CH*), 7.32–7.72 (m, 12H, *ArCH* and *CH*), 8.05 (d, $J = 7.6$ Hz, 2H, *ArCH*), 8.27 (d, $J = 7.6$ Hz, 2H, *ArCH*); δ_{C} (125 MHz, CDCl_3) 10.1, 10.5, 62.7, 121.3, 123.4, 126.1, 127.6, 128.6, 128.7, 128.8, 128.8, 128.9, 130.4, 130.5, 130.8, 133.8, 134.1, 134.6, 145.1, 146.8, 150.3, 154.8, 164.0, 164.4, 192.9; HRMS (ESI) calcd for $\text{C}_{32}\text{H}_{26}\text{O}_6$ (M^+): m/z 506.1729, found 506.1727.

Synthesis of *p*-toluoylated DMC derivatives. Compound **2l** was synthesised following general procedure B. DMC (0.0500 g, 0.1676 mmol) and *p*-toluoyl chloride (216.2 μL , 1.6759 mmol) were dissolved in 10 mL of acetone in the presence of K_2CO_3 (0.2316 g, 1.6759 mmol) to obtain compound **2l** in 69% yield (0.0482 g).

2'-Hydroxy-4'-(*p*-toluoyloxy)-6'-methoxy-3',5'-dimethylchalcone (2l). Orange viscous liquid; R_f (5% CH_2Cl_2 /5% acetone/*n*-hexane) 0.65; ν_{max} (thin film) 3065 (C–H), 2934 ($-\text{CH}_3$), 1723 (C=O), 1654 (C=O), 1604 (C=C), 1507 (C=C), 1185 (C–O–C), 1137 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 2.08 (s, 3H, *Me*), 2.09 (s, 3H, *Me*), 2.47 (s, 3H, *Me*), 3.70 (s, 3H, *OMe*), 7.35 (d, $J = 7.6$ Hz, 2H, *ArCH*), 7.40–7.45 (m, 3H, *ArCH*), 7.63–7.70 (m, 2H, *ArCH*), 7.90 (AB system, d, $J = 15.6$ Hz, 1H, *CH*), 7.98 (AB system, d, $J = 15.6$ Hz, 1H, *CH*), 8.13 (d, $J = 7.6$ Hz, 2H, *ArCH*), 13.00 (s, 1H, *OH*); δ_{C} (125 MHz, CDCl_3) 9.2, 9.4, 21.9, 62.8, 113.4, 115.6, 116.1, 126.0, 126.6, 128.7, 129.1, 129.6, 130.5, 130.7, 135.2, 144.0, 145.0, 154.5, 158.3, 161.0, 163.9, 194.5; HRMS (ESI) calcd for $\text{C}_{26}\text{H}_{24}\text{O}_5$ (M^+): m/z 416.1624, found 416.1619.

Syntheses of benzenesulfonylated DMC derivatives. Compounds **2m**, **5m**, and **6m** were synthesised following general procedure B. DMC (0.0500 g, 0.1676 mmol) and benzenesulfonyl chloride (216.7 μL , 1.6759 mmol) were dissolved in 10 mL of acetone in the presence of K_2CO_3 (0.2316 g, 1.6759 mmol) to obtain compounds **2m**, **5m** and **6m** in 40% (0.0294 g), 23% (0.0223 g), and 6% (0.0045 g) yields, respectively.

2'-Hydroxy-4'-benzenesulfonyloxy-6'-methoxy-3',5'-dimethylchalcone (2m). Orange viscous liquid; R_f (15% EtOAc/5%

acetone/*n*-hexane) 0.68; ν_{max} (thin film) 3071 (C–H), 2945 ($-\text{CH}_3$), 1634 (C=O), 1602 (C=C), 1566 (C=C), 1347 (S=O), 1195 (C–O–C), 1114 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 1.89 (s, 3H, *Me*), 2.08 (s, 3H, *Me*), 3.66 (s, 3H, *OMe*), 7.39–7.46 (m, 3H, *ArCH*), 7.57–7.73 (m, 5H, *ArCH*), 7.86–7.95 (m, 2H, *CH*), 7.99 (d, $J = 7.5$ Hz, 2H, *ArCH*), 12.84 (s, 1H, *OH*); δ_{C} (125 MHz, CDCl_3) 10.2, 10.5, 62.7, 114.0, 117.3, 117.8, 126.2, 128.3, 128.8, 129.2, 129.5, 130.9, 134.5, 135.1, 137.1, 144.5, 152.5, 158.5, 160.9, 194.5; HRMS (ESI) calcd for $\text{C}_{24}\text{H}_{22}\text{O}_6\text{S}$ (M^+): m/z 438.1137, found 438.1135.

2',4'-Dibenzenesulfonyloxy-6'-methoxy-3',5'-dimethylchalcone (5m). Yellow viscous liquid; R_f (20% EtOAc/5% CH_2Cl_2 /*n*-hexane) 0.35; ν_{max} (thin film) 2935 ($-\text{CH}_3$), 2855 ($-\text{CH}_2$), 1740 (C=O), 1566 (C=C), 1453 (CH_3), 1348 (S=O), 1134 (C–O) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 1.91 (s, 3H, *Me*), 2.17 (s, 3H, *Me*), 3.66 (s, 3H, *OMe*), 6.72 (d, $J = 16.0$ Hz, 1H, *CH*), 7.35–7.50 (m, 7H, *ArCH* and *CH*), 7.55–7.66 (m, 5H, *ArCH*), 7.79–7.84 (m, 2H, *ArCH*) 7.97–8.01 (m, 2H, *ArCH*); δ_{C} (125 MHz, CDCl_3) 11.3, 12.1, 62.5, 124.9, 126.9, 127.3, 127.7, 128.3, 128.4, 128.5, 128.6, 128.9, 129.1, 129.6, 130.8, 133.6, 134.3, 134.6, 143.1, 143.2, 145.0, 149.0, 155.0, 190.6; HRMS (ESI) calcd for $\text{C}_{30}\text{H}_{26}\text{O}_8\text{S}_2$ (M^+): m/z 578.1069, found 578.1065.

5-Methoxy-7-benzenesulfonyloxy-6,8-dimethylflavanone (6m). Yellow viscous liquid; R_f (20% EtOAc/*n*-hexane) 0.25; ν_{max} (thin film) 3096 (C–H), 2938 ($-\text{CH}_3$), 2860 ($-\text{CH}_2$), 1694 (C=O), 1587 (C=C), 1302 (S=O), 1197 (C–O–C), 1146 (C–O–C), 1120 (C–O–C) cm^{-1} ; δ_{H} (400 MHz, CDCl_3) 1.94 (s, 3H, *Me*), 2.02 (s, 3H, *Me*), 2.88 (ABX system, dd, $J = 16.6$, 3.0 Hz, 1H, *CH*), 3.03 (ABX system, dd, $J = 16.6$, 13.2 Hz, 1H, *CH*), 3.80 (s, 3H, *OMe*), 5.42 (ABX system, dd, $J = 13.2$, 3.0 Hz, 1H, *CH*), 7.35–7.51 (m, 5H, *ArCH*), 7.57–7.64 (m, 2H, *ArCH*), 7.69–7.76 (m, 1H, *ArCH*), 7.96–8.02 (m, 2H, *ArCH*); δ_{C} (125 MHz, CDCl_3) 10.1, 10.9, 45.8, 61.5, 79.0, 113.9, 118.0, 120.3, 126.0, 128.3, 128.8, 128.9, 129.5, 134.6, 137.0, 138.8, 152.5, 157.6, 159.4, 190.4; HRMS (ESI) calcd for $\text{C}_{24}\text{H}_{22}\text{O}_6\text{S}$ (M^+): m/z 438.1137, found 438.1135.

Syntheses of 2-naphthalenesulfonylated DMC derivatives. Compounds **2n** and **5n** were synthesised following general procedure B. DMC (0.0500 g, 0.1676 mmol) and 2-naphthalenesulfonyl chloride (0.3812 g, 1.6759 mmol) were dissolved in 10 mL of acetone in the presence of K_2CO_3 (0.2316 g, 1.6759 mmol) to afford compounds **2n** and **5n** in 42% (0.0344 g) and 34% (0.0387 g) yields, respectively.

2'-Hydroxy-4'-(2-naphthalenesulfonyloxy)-6'-methoxy-3',5'-dimethylchalcone (2n). Orange viscous liquid; R_f (15% EtOAc/*n*-hexane) 0.63; ν_{max} (thin film) 3065 (C–H), 2941 ($-\text{CH}_3$), 1637 (C=O), 1602 (C=C), 1567 (C=C), 1349 (S=O), 1186 (C–O–C), 1113 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 1.89 (s, 3H, *Me*), 2.08 (s, 3H, *Me*), 3.65 (s, 3H, *OMe*), 7.40–7.46 (m, 3H, *ArCH*), 7.62–7.69 (m, 3H, *ArCH*), 7.72 (m, 1H, *ArCH*), 7.86–8.08 (m, 6H, *ArCH*, and *CH*), 8.53 (s, 1H, *ArCH*), 12.84 (s, 1H, *OH*); δ_{C} (125 MHz, CDCl_3) 10.4, 10.6, 62.7, 114.0, 117.4, 117.9, 122.9, 126.2, 128.1, 128.2, 128.8, 129.2, 129.7, 129.9, 130.0, 130.8, 132.1, 134.0, 135.1, 135.6, 144.5, 152.6, 158.5, 160.9, 194.5; HRMS (ESI) calcd for $\text{C}_{28}\text{H}_{24}\text{O}_6\text{S}$ (M^+): m/z 488.1294, found 488.1291.

2',4'-Di-(2-naphthalenesulfonyloxy)-6'-methoxy-3',5'-dimethylchalcone (5n). Orange viscous liquid; R_f (25% EtOAc/*n*-hexane) 0.45; ν_{max} (thin film) 3065 (C–H), 2934 ($-\text{CH}_3$), 1654



(C=O), 1604 (C=C), 1382 (S=O), 1185 (C–O–C), 1114 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 1.98 (s, 3H, *Me*), 2.18 (s, 3H, *Me*) 3.60 (s, 3H, *OMe*), 6.57 (d, $J = 16.0$ Hz, 1H, *CH*), 7.08 (d, $J = 16.0$ Hz, 1H, *CH*), 7.16–7.41 (m, 5H, *ArCH*), 7.46–7.63 (m, 2H, *ArCH*), 7.63–7.92 (m, 6H, *ArCH*), 7.93–8.10 (m, 4H, *ArCH*), 8.32 (s, 1H, *ArCH*), 8.55 (s, 1H, *ArCH*); δ_{C} (125 MHz, CDCl_3) 11.5, 12.5, 62.5, 122.8, 122.9, 125.2, 126.5, 127.3, 127.8, 128.0, 128.1, 128.2, 128.5, 128.6, 128.9, 129.5, 129.6, 129.7, 129.8, 129.9, 130.0, 130.2, 130.4, 130.8, 131.8, 132.1, 133.2, 133.7, 134.2, 135.5, 135.7, 143.6, 144.8, 149.3, 155.1, 190.5; HRMS (ESI) calcd for $\text{C}_{38}\text{H}_{30}\text{O}_8\text{S}_2$ (M^+): m/z 678.7700, found 678.7701.

General procedure C for the synthesis of 3-ethyl-4-[(*E*)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-butyryloxy coumarin (4a). 7-*O*-Butyrylated-4-hydroxycoumarin (3a) (0.0272 g, 0.0620 mmol) was dissolved in 5 mL of dry THF in the presence of anhydrous Na_2SO_4 (0.0881 g, 0.6200 mmol). Then conc. H_2SO_4 (34.6 μL , 0.6200 mmol) was slowly added to the solution with strong stirring. The reaction was refluxed for 12 hours. Until the completion of the reaction monitored by TLC, the solution was neutralised using 10% NaHCO_3 solution followed by extraction several times with CH_2Cl_2 . The combined organic layer was dried over anhydrous Na_2SO_4 . After removal of the solvent, the crude product was purified by preparative-thin layer chromatography using $\text{Et}_2\text{O}/n$ -hexane = 15 : 85 as the eluent to obtain compound 4a in 92% yield (0.0239 g).

3-Ethyl-4-[(*E*)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-butyryloxy coumarin (4a). White solid; mp 118.7–119.9 °C ($\text{MeOH}/\text{CH}_2\text{Cl}_2$); R_{f} (15% $\text{Et}_2\text{O}/n$ -hexane) 0.14; ν_{max} (thin film) 2972 ($-\text{CH}_3$), 2941 ($-\text{CH}_2$), 2880 ($-\text{CH}_3$), 1765 (C=O), 1719 (C=O), 1597 (C=C), 1118 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 1.08 (t, $J = 7.4$ Hz, 3H, *Me*), 1.22 (t, $J = 7.4$ Hz, 3H, *Me*), 1.80–1.89 (m, 2H, CH_2), 2.10 (s, 3H, *Me*), 2.22 (s, 3H, *Me*), 2.63 (t, $J = 7.4$ Hz, 2H, CH_2), 2.76 (q, $J = 7.4$ Hz, 2H, CH_2), 3.47 (s, 3H, *OMe*), 6.64 (d, $J = 16.5$ Hz, 1H, *CH*), 7.30–7.44 (m, 4H, *ArCH* and *CH*), 7.50–7.56 (m, 2H, *ArCH*); δ_{C} (125 MHz, CDCl_3) 9.4, 9.9, 13.9, 14.3, 18.7, 22.1, 35.9, 61.5, 113.0, 115.1, 120.7, 126.1, 126.7, 127.6, 128.4, 129.1, 131.5, 136.7, 146.4, 149.6, 150.3, 154.1, 161.5, 171.1; HRMS (ESI) calcd for $\text{C}_{26}\text{H}_{28}\text{O}_5$ (M^+): m/z 420.1937, found 420.1940.

Synthesis of stilbene-coumarin 4b. Following general procedure C, to a stirred solution of 7-*O*-caproylated-4-hydroxycoumarin (3b) (0.0328 g, 0.0663 mmol) and anhydrous Na_2SO_4 (0.0942 g, 0.6631 mmol) in 5 mL of dry THF, conc. H_2SO_4 (35.5 μL , 0.6631 mmol) was added and refluxed for 12 hours to afford compound 4b in 88% yield (0.0278 g).

3-Butyl-4-[(*E*)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-caproyloxy coumarin (4b). White solid; mp 112.1–113.5 °C ($\text{MeOH}/\text{CH}_2\text{Cl}_2$); R_{f} (15% $\text{Et}_2\text{O}/n$ -hexane) 0.20; ν_{max} (thin film) 2963 ($-\text{CH}_3$), 2937 ($-\text{CH}_2$), 2866 ($-\text{CH}_3$), 1766 (C=O), 1721 (C=O), 1597 (C=C), 1117 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 0.87–0.98 (m, 6H, *Me*), 1.33–1.48 (m, 6H, CH_2), 1.54–1.65 (m, 2H, CH_2), 1.77–1.86 (m, 2H, CH_2), 2.10 (s, 3H, *Me*), 2.21 (s, 3H, *Me*), 2.64 (t, $J = 7.6$ Hz, 2H, CH_2), 2.70–2.76 (m, 2H, CH_2), 3.46 (s, 3H, *OMe*), 6.63 (d, $J = 16.5$ Hz, 1H, *CH*), 7.30–7.44 (m, 4H, *ArCH* and *CH*), 7.50–7.57 (m, 2H, *ArCH*); δ_{C} (125 MHz, CDCl_3) 9.4, 9.9, 14.0, 14.1, 22.4, 23.0, 24.8, 28.3, 31.5, 31.9, 34.1, 61.5, 113.0, 115.1, 120.7, 126.2, 126.5, 126.7, 128.3, 129.1, 131.6, 136.8,

146.4, 149.6, 150.3, 154.1, 161.6, 171.3; HRMS (ESI) calcd for $\text{C}_{30}\text{H}_{36}\text{O}_5$ (M^+): m/z 476.2563, found 476.2565.

Synthesis of stilbene-coumarin 4c. Compound 4c was synthesised following general procedure C. To a stirred solution of 7-*O*-capryloylated-4-hydroxycoumarin (3c) (0.0259 g, 0.0470 mmol) and anhydrous Na_2SO_4 (0.0668 g, 0.4702 mmol) in 5 mL of dry THF, conc. H_2SO_4 (25.2 μL , 0.4702 mmol) was added and refluxed for 12 h to afford compound 4c in 89% yield (0.0222 g).

3-Hexyl-4-[(*E*)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-capryloyloxy coumarin (4c). White solid; mp 93.0–94.4 °C ($\text{MeOH}/\text{CH}_2\text{Cl}_2$); R_{f} (15% $\text{Et}_2\text{O}/n$ -hexane) 0.25; ν_{max} (thin film) 2962 ($-\text{CH}_3$), 2933 ($-\text{CH}_2$), 2860 ($-\text{CH}_3$), 1755 (C=O), 1712 (C=O), 1595 (C=C), 1120 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 0.77–0.93 (m, 6H, *Me*), 1.17–1.48 (m, 14H, CH_2), 1.53–1.67 (m, 2H, CH_2), 1.77–1.85 (m, 2H, CH_2), 2.10 (s, 3H, *Me*), 2.22 (s, 3H, *Me*), 2.65 (t, $J = 7.6$ Hz, 2H, CH_2), 2.69–2.75 (m, 2H, CH_2), 3.46 (s, 3H, *OMe*), 6.62 (d, $J = 16.5$ Hz, 1H, *CH*), 7.30–7.44 (m, 4H, *ArCH* and *CH*), 7.50–7.57 (m, 2H, *ArCH*); δ_{C} (125 MHz, CDCl_3) 9.5, 9.9, 14.2, 22.7, 22.8, 25.1, 28.5, 29.0, 29.3, 29.6, 29.7, 29.8, 31.7, 31.8, 34.1, 61.5, 113.0, 115.1, 120.7, 126.2, 126.5, 126.6, 128.3, 129.0, 131.4, 136.7, 146.5, 149.5, 150.2, 154.0, 161.7, 171.4; HRMS (ESI) calcd for $\text{C}_{34}\text{H}_{44}\text{O}_5$ (M^+): m/z 532.3189, found 532.3185.

Synthesis of stilbene-coumarin 4d. Following general procedure C, to a stirred solution of 7-*O*-caprinoylated-4-hydroxycoumarin (3d) (0.0470 g, 0.0774 mmol) and anhydrous Na_2SO_4 (0.1099 g, 0.7745 mmol) in 5 mL of dry THF, conc. H_2SO_4 (42.2 μL , 0.7745 mmol) was added and refluxed for 12 h to afford compound 4d in 84% yield (0.0383 g).

3-Octyl-4-[(*E*)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-caprinoyloxy coumarin (4d). White solid; mp 84.9–86.2 °C ($\text{MeOH}/\text{CH}_2\text{Cl}_2$); R_{f} (15% $\text{Et}_2\text{O}/n$ -hexane) 0.28; ν_{max} (thin film) 2962 ($-\text{CH}_3$), 2930 ($-\text{CH}_2$), 2859 ($-\text{CH}_3$), 1755 (C=O), 1712 (C=O), 1595 (C=C), 1121 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 0.82–0.92 (m, 6H, *Me*), 1.13–1.48 (m, 22H, CH_2), 1.55–1.64 (m, 2H, CH_2), 1.76–1.85 (m, 2H, CH_2), 2.10 (s, 3H, *Me*), 2.21 (s, 3H, *Me*), 2.64 (t, $J = 7.6$ Hz, 2H, CH_2), 2.69–2.75 (m, 2H, CH_2), 3.46 (s, 3H, *OMe*), 6.62 (d, $J = 16.5$ Hz, 1H, *CH*), 7.31–7.43 (m, 4H, *ArCH* and *CH*), 7.51–7.55 (m, 2H, *ArCH*); δ_{C} (125 MHz, CDCl_3) 9.4, 9.9, 14.2, 22.7, 22.8, 25.2, 28.5, 29.3, 29.4, 29.5, 29.6, 29.7, 29.8, 32.0, 34.1, 61.5, 113.0, 115.1, 120.7, 126.2, 126.6, 126.7, 128.3, 129.0, 131.6, 136.8, 146.4, 149.5, 150.3, 154.0, 161.7, 171.3; HRMS (ESI) calcd for $\text{C}_{38}\text{H}_{52}\text{O}_5$ (M^+): m/z 588.3815, found 588.3810.

Synthesis of stilbene-coumarin 4e. Compound 4e was synthesised following general procedure C. To 7-*O*-lauroylated-4-hydroxycoumarin (3e) (0.0624 g, 0.0941 mmol) and anhydrous Na_2SO_4 (0.1337 g, 0.9412 mmol) in 5 mL of dry THF, conc. H_2SO_4 (50.4 μL , 0.9412 mmol) was added and refluxed for 12 hours to afford compound 4e in 93% yield (0.0564 g).

3-Decyl-4-[(*E*)-2'-phenylvinyl]-5-methoxy-6,8-dimethyl-7-lauroyloxy coumarin (4e). White solid; mp 76.2–77.2 °C ($\text{MeOH}/\text{CH}_2\text{Cl}_2$); R_{f} (15% $\text{Et}_2\text{O}/n$ -hexane) 0.30; ν_{max} (thin film) 2961 ($-\text{CH}_3$), 2928 ($-\text{CH}_2$), 2857 ($-\text{CH}_3$), 1755 (C=O), 1712 (C=O), 1595 (C=C), 1120 (C–O–C) cm^{-1} ; δ_{H} (500 MHz, CDCl_3) 0.84–0.92 (m, 6H, *Me*), 1.17–1.50 (m, 30H, CH_2), 1.54–1.66 (m, 2H, CH_2), 1.76–1.86 (m, 2H, CH_2), 2.10 (s, 3H, *Me*), 2.21 (s, 3H, *Me*), 2.64 (t, $J = 7.6$ Hz, 2H, CH_2), 2.68–2.75 (m, 2H, CH_2), 3.46 (s, 3H, *OMe*), 6.62 (d, $J = 16.5$ Hz, 1H, *CH*), 7.29–7.43 (m, 4H, *ArCH* and



CH), 7.51–7.55 (m, 2H, ArCH); δ_C (125 MHz, CDCl₃) 9.4, 9.9, 14.3, 22.8, 25.2, 28.5, 29.3, 29.4, 29.5, 29.6, 29.6, 29.7, 29.8, 29.9, 32.1, 34.1, 61.5, 113.0, 115.1, 120.7, 126.2, 126.6, 126.7, 128.3, 129.1, 131.6, 136.8, 146.4, 149.6, 150.3, 154.1, 161.6, 171.3; HRMS (ESI) calcd for C₄₂H₆₀O₅ (M)⁺: m/z 644.4441, found 644.4438.

Geometry optimisation

The Gaussian 16 package was used to carry out the calculations. 3D structures and conformer distributions were established using Avogadro software⁴⁶ using the Molecular Mechanics Force Field (MMFF). The lowest energy conformer was further performed by geometrical optimisation running on the Gaussian16 package^{47,48} using DFT/B3LYP as the function and 6-311++G (d,p) as the basis set for the calculation. Thermal free energy was employed to determine the relative free energy between conformers.

In silico molecular docking study

Molecular docking was performed to predict the binding poses of the DMC derivatives on cyclin-dependent kinase 2 (CDK2) which plays an important role in cell division.⁴⁹ AutoDock VinaTM was utilised in all docking experiments.⁵⁰ All the calculations were performed on an Intel Xeon Silver 4114 CPU@2.20 GHz of HP Z8 G4 workstation, with 32 GB DDR4 Ram compiled under the Ubuntu18.04 operating system. The study was performed using human cyclin-dependent kinase 2 (CDK2) in complex with an inhibitor (PDB ID 4GCJ), which is available from the protein database <http://www.rcsb.org>.⁵¹ 3D structures of the ligands were established using the Avogadro software and the energy further minimised by theoretical B3LYP/6-31G density functional theory (DFT) using the Gaussian16 package.^{47,48} The protein was prepared *via* BIOVIA discovery studio 2016 software⁵² using the protein preparation tool. All water molecules were removed, and standard protonation states were verified using the PropKa server.^{53,54} A co-crystallised inhibitor was used to determine the active site for docking. The docking protocol was validated by the redocking of the co-crystallised ligand to the protein active site obtaining a binding energy value of -9.7 kcal mol⁻¹ and an RMSD value of 0.976 Å. The grid size was set to 30 × 30 × 30xyz points. The grid centre was designated at dimensions (x, y and z): -54.5170 , 98.5972 and -59.1310 . The best score of each ligand was chosen and the protein–ligand interactions visualised in 2D and 3D plots.

Evaluation of anticancer activity

Anticancer activity was evaluated as the cytotoxicity of carcinoma cell lines. Slightly modified sulforhodamine B (SRB) was used as a standard method⁵⁵ with an ellipticine as the positive control drug. The semi-synthetic DMC analogues were dissolved in 10% DMSO solution with concentration range 0.1–50 μ M. Cell lines were grown at 10⁴ cell per well and seeded onto a 96-well plate as follows: KKKU-M213, FaDu, HT-29, MDA-MB-231, A-549, SH-SY5Y, and CL. To the cultures were added 10 μ L of cytotoxic substance solution in various concentrations and then they were incubated at 37 °C in a humidified incubator with 5%

CO₂ atmosphere for 72 h. The cultures were fixed by 10% trichloroacetic acid (TCA) and washed with slow-running tap water followed by staining with 100 μ L of 0.4% sulforhodamine B solution. The unbound dye was removed by washing with 1% acetic acid solution. Cellular bound dye was extracted and solubilised with 10 mM Trizma base solution (pH 10.5). The absorbance was measured at 510 nm using a Fluostar optima BMG plate reader. The cytotoxicity was expressed as IC₅₀ (the concentration which indicated 50% cell-growth inhibition) in the micromolar unit of concentration. The selectivity index (SI) was calculated by the ratio of IC₅₀ on a normal cell divided by that on a cancer cell.

Antibacterial activity evaluation

The agar well diffusion method was used to investigate the antimicrobial activity of synthetic compounds, as described by L. M. Cintas *et al.*⁵⁶ with a slight modification. Bacterial strains including *E. coli* (ATCC® 25922TM), *S. aureus* (ATCC® 25923TM), and *P. aeruginosa* (ATCC® 27853TM) were grown on Trypticase soy agar (TSA) at 37 °C for 18 h and adjusted to achieve a turbidity comparable to 0.5 McFarland standard (1.5 × 10⁸ CFU mL⁻¹) with sterile 0.85% NaCl solution. Then, the tested strain was swabbed on the surface of Mueller–Hinton Agar (MHA) and wells were made using a sterile cork borer (6 mm in diameter). 20 μ L of tested synthetic compounds which were previously dissolved in DMSO at 5 mg mL⁻¹ starting concentration was added to the respective wells. The plates were placed in the refrigerator for 30 minutes to let the extracts diffuse well into the agar. Then, the plates were incubated at 37 °C for 16–20 h. Antimicrobial activity was expressed as a zone of inhibition (ZOI) in mm. DMSO at a concentration of 10% was employed as a negative control.

Broth microdilution method

Broth microdilution method was achieved as described by M. Bolouri *et al.*⁵⁷ with some modifications. The inocula were prepared by adjusting the turbidity of the suspension to match the 0.5 McFarland standard and diluted to 1 : 100 for the broth microdilution procedure. Antibiotic samples were dissolved in DMSO at 5 mg mL⁻¹ starting concentration. Dilutions ranging from 5 to 0.3125 mg mL⁻¹ of the tested compounds were prepared in sterile 96-well microplates through a 2-fold serial dilution. To the well containing 1400 μ L of Mueller–Hinton Broth (MHB) was added about 50 μ L of tested compounds followed by 10 μ L of bacterial strains to a final volume of 100 μ L. The final inoculum was 5 × 10⁴ CFU mL⁻¹. DMSO was taken as a negative control. The inoculated plates were incubated in 37 °C for 16–20 hours. After incubation, 10 μ L of resazurin was added to each well and further incubated at 37 °C for 30 minutes. The MIC is the lowest concentration of the compound at which the microorganism tested does not demonstrate visible growth. The MBC values were determined by sub-culturing 10 μ L of the test dilutions onto a fresh MHA and incubated for a further 18–24 h. The highest dilution that yielded no bacterial growth was taken as the MBC. All experiments were done in triplicate.



Conclusions

In summary, DMC was isolated from a crude EtOAc extract of *S. nervosum* seeds. The DMC was further modified by acylation, alkylation, and sulfonylation accessing 4'-O-monosubstituted-DMCs (2a–2n), 7-O-acylated-4-hydroxycoumarins (3a–3e), stilbene-coumarins (4a–4e), 2',4'-O-disubstituted-DMCs (5f–5n), and flavanone derivatives (6g and 6m). These compounds were tested for cytotoxicity and antimicrobial activity against six cancer cell lines and three bacterial strains. Most semi-synthetic DMC derivatives showed higher cytotoxicity and selectivity to cancer cell lines than DMC. 4'-O-Caproylated-DMC (2b) and 4'-O-methylated-DMC (2g) exhibited the strongest cytotoxicity against SH-SY5Y with IC₅₀ values of 5.20 and 7.52 μM, respectively. 4'-O-Benzylated-DMC (2h) displayed the strongest cytotoxicity against A549 and FaDu cell lines with IC₅₀ values of 9.99 and 13.98 μM, respectively. *In silico* molecular docking revealed the poses of DMC derivatives on cyclin-dependent kinase 2 (PDB ID 4GCJ). The results suggested that 4'-O-methylated-DMC (2g) and 4'-O-benzylated-DMC (2h) displayed lower binding energy than DMC, especially compound 2h, which indicated key interactions with amino acid residues of CDK2. Most derivatised DMC, on the other hand, lacked antimicrobial activity.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This research project was supported by the National Research Council of Thailand (NRCT) and TSRI. In addition, we would also like to thank the TA/RA Scholarship of Graduate School, the Center of Excellence for Innovation in Chemistry (PERCH-CIC), the Department of Chemistry, Faculty of Science, Chiang Mai University and the Graduate School, Chiang Mai University, Center of Excellence in Materials Science and Technology, Faculty of Science, Chiang Mai University for financial support.

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