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H₂O₂-independent chemodynamic therapy initiated from magnetic iron carbide nanoparticle-assisted artemisinin synergy†

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Chemodynamic therapy (CDT) is a booming technology that utilizes Fenton reagents to kill tumor cells by transforming intracellular H₂O₂ into reactive oxygen species (ROS), but insufficient endogenous H₂O₂ makes it difficult to attain satisfactory antitumor results. In this article, a H₂O₂-free CDT technique with tumor-specificity is developed by using pH-sensitive magnetic iron carbide nanoparticles (PEG/Fe₂C@Fe₃O₄ NPs) to trigger artemisinin (ART) to *in situ* form ROS. ART-loaded PEG/Fe₂C@Fe₃O₄ NPs are fabricated for the enormous release of Fe²⁺ ions induced by the acidic conditions of the tumor microenvironment after magnetic-assisted tumor enrichment, which results in the rapid degradation of the PEG/Fe₂C@Fe₃O₄ NPs and release of ART once endocytosed into tumor cells. *In situ* catalysis reaction between the co-released Fe²⁺ ions and ART generates toxic ROS and then induces apoptosis of tumor cells. Both *in vitro* and *in vivo* experiments demonstrate that the efficient Fe-enhanced and tumor-specific CDT efficacy for effective tumor inhibition based on ROS generation. This work provides a new direction to improve CDT efficacy based on H₂O₂-independent ROS generation.

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Introduction

Chemodynamic therapy (CDT), that employs highly cytotoxic reactive oxygen species (ROS) generated from endogenous hydrogen peroxide (H₂O₂) *via* the Fenton-type reactions to induce tumor cell apoptosis, has been intensively investigated for tumor therapy nowadays.^{1,2} Benefiting from the over-produced endogenous H₂O₂ which is the characteristic feature of the tumor microenvironment (TME), CDT has been recognized as an emerging therapeutic strategy with high therapeutic specificity and low invasiveness.^{3,4} However, the concentration of intratumoral H₂O₂ (50–100 μM), although at a higher level compared with normal cells, is still insufficient to maximize the Fenton catalytic efficacy, which restricts the clinical application

of CDT.^{5–7} To address this limitation, substantial efforts have been devoted to increasing the level of intratumoral H₂O₂, such as applying natural enzymes like glucose oxidase (GOx), nicotinamide adenine dinucleotide phosphate oxidase (NOX), and superoxide dismutase (SOD) as H₂O₂ self-supplying systems to *in situ* generate H₂O₂ inside tumor.^{8–10} Nevertheless, the rigorous reaction condition further confines the therapeutic efficiency of CDT utilizing these strategies. Recently, Wang and coauthors applied ferrous ion (Fe²⁺) to catalyze linoleic acid hydroperoxide (LAOOH) to produce abundant ROS through the Russell mechanism, overcoming the dependence on H₂O₂ concentration by CDT.¹¹ This suggests a potential alternative to H₂O₂-dependent CDT.

Artemisinin (ART), a class of sesquiterpene lactone extracted from *Artemisia annua*, has been broadly used to treat malaria.^{12,13} Over the last 10 years, studies have suggested that ART containing endoperoxide bridge also has a therapeutic effect on various tumor cells.^{14,15} The critical mechanism for antitumor ability of ART is that peroxide bridge structure (R–OO–R') inside of ART could be catalysis-cleaved by Fe²⁺ within tumor cells to produce toxic carbon-centered free radicals (a kind of ROS).^{16–19} Thus, a H₂O₂-independent CDT agent can be constructed based on ART. Noteworthy that Fe²⁺ plays an irreplaceable role in tumor cell killing for ART-based drugs and sufficient Fe²⁺ may make a remarkable contribution to maximize the therapy efficiency of ART.^{20–24} Whereas, the concentration of Fe²⁺ at the tumor site is generally much lower than the

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optimum concentration of Fe^{2+} which can exhaust the anti-tumor ability of ART, leading to an unsatisfactory therapeutic efficiency.²⁵ Therefore, designing a nanocarrier for co-delivery of ART and Fe^{2+} to tumor cells can be an ideal strategy for the enhancement of CDT that independent from endogenous H_2O_2 .

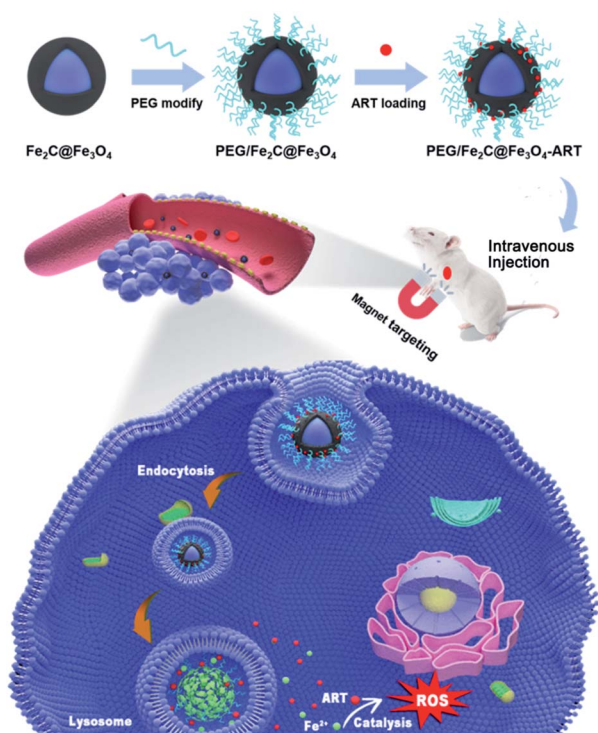
Iron carbide nanoparticles (ICNPs), a kind of nano-intermetallic iron-based compounds, have gained enormous attention for their extraordinary magnetic property which renders them to be T_2 contrast agents in magnetic resonance imaging (MRI).^{26–29} Besides, ICNPs with appropriate surface modification could be excellent drug carriers, and the high saturation magnetization of ICNPs endows them magnetic targeting property.^{30–32} Interestingly, our previous works suggested that ICNPs in which the valence state of iron tends to be zero valence can easily release Fe^{2+} under acidic condition and chronically remain stable in the neutral environment.³³ Consequently, taking advantage of their pH-sensitivity, ICNPs can be designed as 'iron storage pool' to specifically release Fe^{2+} at acidic microenvironment of the tumor to trigger ART.

Herein, core-shell structured ICNPs ($\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ nanoparticles), which were capable of pH-triggered Fe^{2+} -release, were designed as the nanocarriers for ART. Benefiting from the high saturation magnetization, an enrichment of $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs together with ART in tumor sites was feasible by magnetic targeting. Combined with the acidic microenvironment of tumor, the designed $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ nanocarrier was able to resolve at tumor, release both Fe^{2+} and ART synchronously for yielding quantity of toxic ROS *in situ* to damage proteins and nucleic

acid, and consequently induce cell death which significantly enhance the CDT efficiency (Scheme 1). This study paves a new direction to improve CDT efficacy based on H_2O_2 -free ROS generation.

Results and discussion

$\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs were synthesized by a two-step synthetic method.³⁴ Based on transmission electron microscopy (TEM) image (Fig. 1a), the as-synthesized NPs showed monodisperse core-shell structure with uniform size at approximately 18 nm, which was consistent with the data of dynamic light scattering (DLS) (Fig. S1†). The core size of $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs was measured to be 14 ± 2 nm, while the shell thickness was approximately 1 ± 0.5 nm. The crystallization of the NPs could be identified in the high-resolution transmission electron microscopy (HRTEM) image (Fig. 1b). The lattice spacing in the core was 0.206 nm, corresponding to the (101) plane of highly crystallized Fe_2C , while the shell structure showed an amorphous feature of Fe_3O_4 .³⁴ The crystal structure of the NPs was further determined by X-ray diffraction (XRD) (Fig. 1c). All the diffraction peaks could be well-indexed to the Fe_2C core (JCPDS no. 37-0999), while no obvious peak of Fe_3O_4 was observed, due to its low crystallization. To verify the composition of the $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs, X-ray photoelectron spectroscopy (XPS) was applied, and the result indicated the coexistence of carbon, oxygen, and iron (Fig. 1d). The energy-dispersive spectroscopy (EDS) further confirmed that obtained $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs were composed of Fe, C, and O elements (Fig. S2†). Besides, the magnetic property of $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs was tested by using a vibrating sample magnetometer (VSM) at room temperature. The saturation magnetization (M_s) value of $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs was 65.483 emu



Scheme 1 Schematic illustration of ART loaded PEG/ $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs to tumor cells assisted by an externally applied magnetic field and the antitumor mechanism of the ART delivery system.

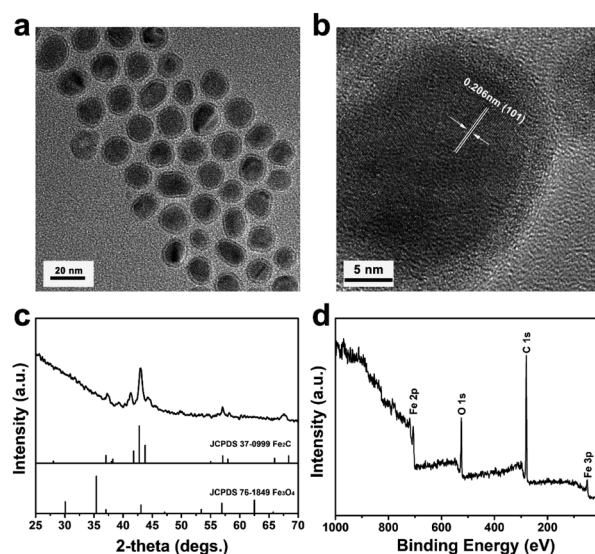


Fig. 1 Characterization of $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs. (a) TEM image of $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs. (b) HRTEM image of $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs. (c) XRD pattern of $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs (JCPDS no. 37-0999, 76-1849). (d) XPS survey spectrum of $\text{Fe}_2\text{C}@ \text{Fe}_3\text{O}_4$ NPs.



g^{-1} (Fig. S3†), suggesting their potential as T_2 contrast agents as well as nanocarriers with magnetic targeting.

To endow a hydrophilic nature as well as a long-circulation property to $Fe_2C@Fe_3O_4$ NPs, DSPE-PEG was applied to form PEG/ $Fe_2C@Fe_3O_4$ NPs. The data of FT-IR and macrophotograph showed the successful modification of PEG on $Fe_2C@Fe_3O_4$ NPs (Fig. S4 and S5†). TEM image of PEG/ $Fe_2C@Fe_3O_4$ NPs obtained from their aqueous dispersions, indicating that there was no obvious morphology change after the ligand addition process (Fig. S6†). In addition, hydrodynamic diameter of PEG/ $Fe_2C@Fe_3O_4$ NPs was increased to 96.1 nm and zeta potential was measured to be -21.5 mV (Fig. S7†), which was suitable for enhanced permeability and retention (EPR) effect to achieve practical applications. Furthermore, $Fe_2C@Fe_3O_4$ NPs after PEGylation could keep long-term stability in the aqueous phase (Fig. S8†).

The antitumor activity of ART was related to the concentration of Fe^{2+} . Hence, Fe^{2+} release characteristic of PEG/ $Fe_2C@Fe_3O_4$ NPs was firstly explored. Three solutions with pH value of 7.4, 6.5, and 5.4 were applied that corresponding to the environment of normal tissue, tumor site, and lysosomes. As shown in Fig. 2a, the degradation of PEG/ $Fe_2C@Fe_3O_4$ NPs in different solutions showed an obvious pH-sensitivity. Iron ions released in pH 5.4 solution increased significantly in the first 4 h, with the amount reached to 63.42% after 24 h, while only 14.02% and 22.04% of iron ions could be released under pH 7.4 and 6.5, respectively. TEM images of PEG/ $Fe_2C@Fe_3O_4$ NPs after being dispersed in pH 5.4 and 7.4 media for 24 h, further confirming that the NPs could be degraded only in pH 5.4 solution (Fig. 2b and c). In addition, strong hypo-intensities in T_2 -weighted magnetic resonance imaging (T_2 -MRI) of PEG/ $Fe_2C@Fe_3O_4$ NPs, which induced from their high magnetization, was reduced by altering the incubation solution from

physiological condition (pH 7.4) to acidic solution (pH 5.4), with the transverse relaxivity (r_2) decreased from $205.11 \text{ mM}^{-1} \text{ s}^{-1}$ to $58.29 \text{ mM}^{-1} \text{ s}^{-1}$ (Fig. 2d). This result further proved the dissolution of the NPs in acidic environment. Color of the solution turned to bright yellow at pH of 5.4, and after adding potassium ferricyanide (a Fe^{2+} indicator), the color became blue, suggesting an efficient release of Fe^{2+} at acidic environment (Fig. S9†). The excellent pH-sensitive Fe^{2+} release ability of PEG/ $Fe_2C@Fe_3O_4$ NPs was attributed to successful introduction of amorphous Fe_3O_4 shells which performed stronger pH sensitivity after amorphization.^{33,35,36} The above results demonstrated that PEG/ $Fe_2C@Fe_3O_4$ NPs were good Fe^{2+} donors in low pH environment.

PEG was modified onto hydrophobic $Fe_2C@Fe_3O_4$ NPs through ligand addition method, endowing the interspace for loading ART by hydrophobic-hydrophobic interaction. The appearance of the characteristic peaks at the wavelength of 1737 cm^{-1} , between 800 cm^{-1} and 1200 cm^{-1} confirming the loading of ART onto PEG/ $Fe_2C@Fe_3O_4$ NPs, which were ascribed to the C-O stretching vibrations, and the O-O and C-O modes of O-O-C (peroxide), respectively (Fig. S10†).³⁷ The ART loading efficiency was calculated to be 21.9 wt% by using ultraviolet and visible (UV-vis) spectra. Result from TEM image showed the ART loaded PEG/ $Fe_2C@Fe_3O_4$ (PEG/ $Fe_2C@Fe_3O_4$ -ART) NPs could still keep their morphology with good dispersity in aqueous solution (Fig. S11†).

The result from EPR spectrometry showed ART was able to generate carbon-centered free radicals (a kind of ROS) in Fe^{2+} -riched environment (Fig. S12†).^{21,23} Considering the acidic environment of lysosome and the efficient release of Fe^{2+} under low pH solution, locating PEG/ $Fe_2C@Fe_3O_4$ NPs as an ART activator in lysosomes after being cellular uptake was important. After labelling PEG/ $Fe_2C@Fe_3O_4$ NPs with fluorescein isothiocyanate (FITC), cellular uptake assay was carried out by staining nuclei with 4,6-diamidino-2-phenylindole (DAPI) and lysosomes with Lyso-Tracker red. As shown Fig. 3a, green fluorescence from the FITC-labeled NPs was well overlapped with the red fluorescence of Lyso-Tracker, indicating that PEG/ $Fe_2C@Fe_3O_4$ NPs were internalized and located in the acidic lysosomal compartment, and should produce ROS consequently.

To verify the capability of PEG/ $Fe_2C@Fe_3O_4$ -ART NPs in generating ROS in tumor cells, 2,7-dichlorofluorescein diacetate (DCFH-DA), a fluorescent ROS-detection probe was used. Fluorescence emitted from PEG/ $Fe_2C@Fe_3O_4$ -ART NPs increased with the incubation time, and the fluorescence intensity after 9 h of incubation was 1.8 and 3.6-fold higher compared to the ART alone and PEG/ $Fe_2C@Fe_3O_4$ NPs treated group respectively (Fig. 3b). Similarly, observed under a confocal laser scanning microscope (CLSM), a bright green fluorescence could be clearly observed in cells incubated with PEG/ $Fe_2C@Fe_3O_4$ -ART NPs within 4 h, while in contrast, fluorescence from the cells incubated with ART alone or PEG/ $Fe_2C@Fe_3O_4$ NPs was much weaker (Fig. 3c). These results demonstrated that intracellular PEG/ $Fe_2C@Fe_3O_4$ -ART NPs could boost the generation of ROS *in situ* for tumor therapy.

Cell growth inhibition was then evaluated through MTT assay. As shown in Fig. 3d and S13,† ART alone and PEG/

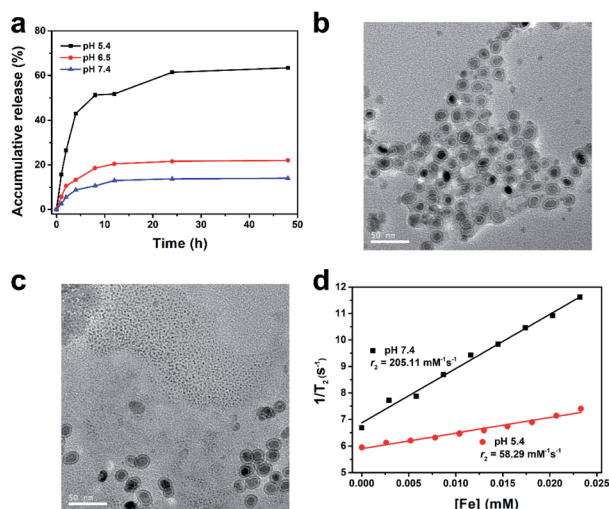


Fig. 2 (a) Accumulative iron ions release of PEG/ $Fe_2C@Fe_3O_4$ NPs in PBS with pH 7.4, 6.5 and 5.4. (b) TEM image of PEG/ $Fe_2C@Fe_3O_4$ NPs by dispersing in pH 7.4 solution for 24 h. (c) TEM image of PEG/ $Fe_2C@Fe_3O_4$ NPs by dispersing in pH 5.4 solution for 24 h. (d) The longitudinal relaxation rates of PEG/ $Fe_2C@Fe_3O_4$ NPs dispersing at pH 5.4 or pH 7.4 environment.



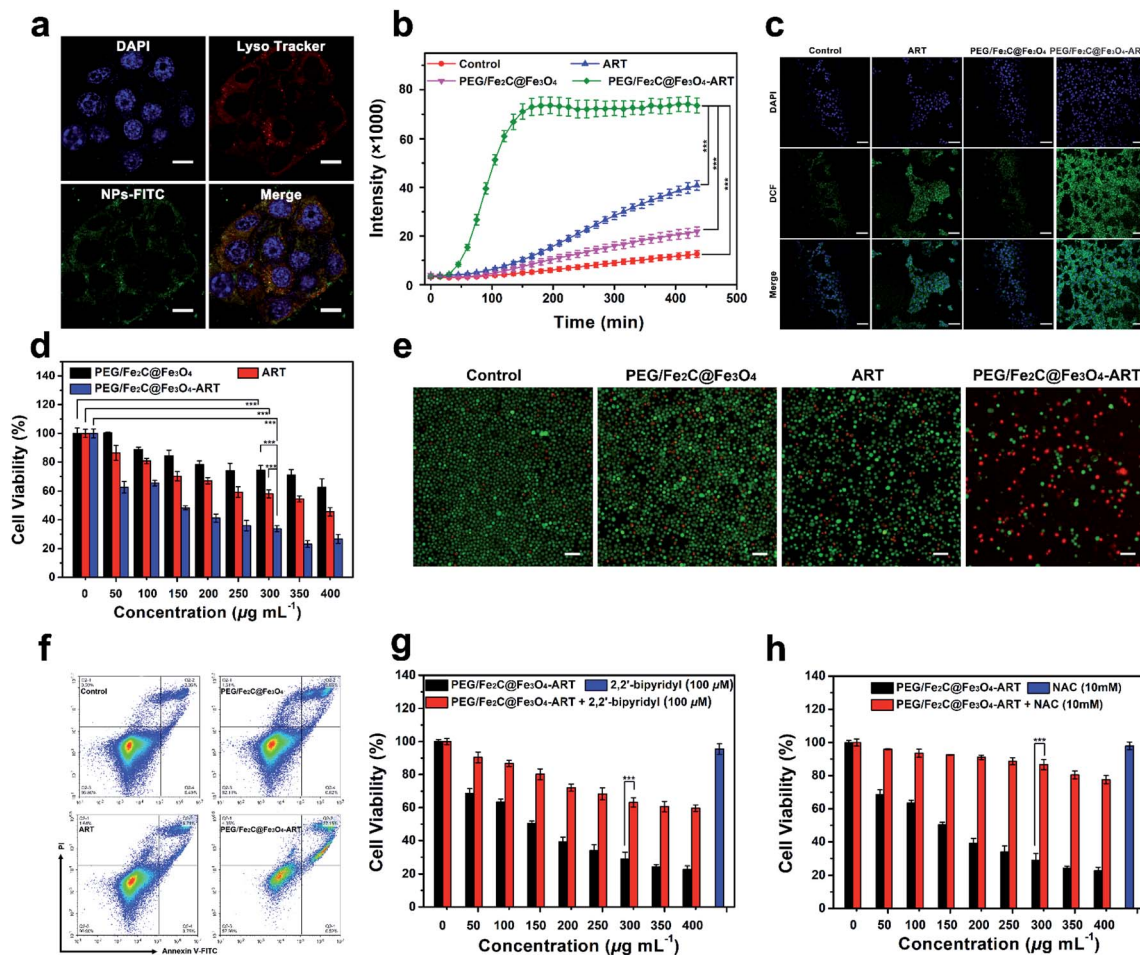


Fig. 3 *In vitro* experiments with PEG/Fe₂C@Fe₃O₄-ART NPs. (a) Fluorescence images of 4T1 cells treated with PEG/Fe₂C@Fe₃O₄-FITC NPs for 4 h. Nuclei of live cells were stained with DAPI and lysosomes were stained with Lyso-Tracker red (scale bar 15 μm). (b) Time-dependent fluorescent intensity from DCFH-DA labeled 4T1 cells by ART, PEG/Fe₂C@Fe₃O₄ and PEG/Fe₂C@Fe₃O₄-ART ($n = 3$, mean \pm s.d., *** $p < 0.001$). (c) Fluorescence images of DCFH-DA labeled 4T1 cells treated by ART, PEG/Fe₂C@Fe₃O₄ and PEG/Fe₂C@Fe₃O₄-ART (scale bar 75 μm). (d) Viability of 4T1 cells after different treatments under various conditions for 24 h ($n = 6$, mean \pm s.d., *** $p < 0.001$). (e) Fluorescence images of calcein AM (green, live cells) and PI (red, dead cells) contained 4T1 cells after incubation with ART, PEG/Fe₂C@Fe₃O₄ NPs or PEG/Fe₂C@Fe₃O₄-ART for 24 h. The scale bar represents 100 μm. (f) Flow cytometry analysis in 4T1 cells treated with ART, PEG/Fe₂C@Fe₃O₄ NPs or PEG/Fe₂C@Fe₃O₄-ART for 24 h. (g) Viability of 4T1 cells incubated with PEG/Fe₂C@Fe₃O₄-ART alone or PEG/Fe₂C@Fe₃O₄-ART plus 2,2-bipyridyl (100 μM) for 24 h ($n = 6$, mean \pm s.d., *** $p < 0.001$). (h) Viability of 4T1 cells incubated with PEG/Fe₂C@Fe₃O₄-ART alone or PEG/Fe₂C@Fe₃O₄-ART plus NAC (10 mM) for 24 h ($n = 6$, mean \pm s.d., *** $p < 0.001$).

Fe₂C@Fe₃O₄ NPs were not obviously toxic to 4T1, MDA-MB-231, and HeLa cells. Whereas, the cytotoxicity of PEG/Fe₂C@Fe₃O₄-ART NPs was significantly enhanced, demonstrating Fe₂C@Fe₃O₄ NPs could improve the antitumor efficiency of ART. Results from live/dead cell staining assay further confirmed this conclusion. By staining live cells with calcein-AM and dead cells with propidium iodide (PI), red fluorescence from PEG/Fe₂C@Fe₃O₄-ART NPs 24 h-treated 4T1 cells indicated the induction of cell death (Fig. 3e). It has been reported that cell killing through ROS-mediated pathway could mainly induce cell apoptosis. Flow cytometry was further applied to identify the type of cell death by using the Annexin V-FITC/PI apoptosis detection kit. Obviously, the PEG/Fe₂C@Fe₃O₄-ART-treated groups had higher ratios of apoptotic tumor cells compared with other groups (Fig. 3f).

From the phenomenon above, the reason for the enhanced apoptosis of PEG/Fe₂C@Fe₃O₄-ART NPs could be ascribed to the efficient *in situ* generation of ROS by released Fe²⁺ and ART. Consequently, Fe²⁺ is sufficient for ART-based tumor therapy. Therefore, 2,2-bipyridyl, a widely used Fe²⁺ chelator was applied to confirm the dependance of Fe²⁺ in the cell killing.³⁸ Cells incubated with PEG/Fe₂C@Fe₃O₄-ART in the presence of 2,2-bipyridyl showed suppressed intracellular ROS production (Fig. S14†), and as a result, significantly increased the cell viability compared with that without 2,2-bipyridyl (Fig. 3g). It proved the involvement of the released Fe²⁺ in the cytotoxicity of PEG/Fe₂C@Fe₃O₄-ART NPs. To verify the significance of ROS, *N*-acetyl-cysteine (NAC),^{39,40} a ROS scavenger, was applied. As shown in Fig. 3h, the addition of NAC significantly reversed the cytotoxicity induced by PEG/Fe₂C@Fe₃O₄-ART NPs, further



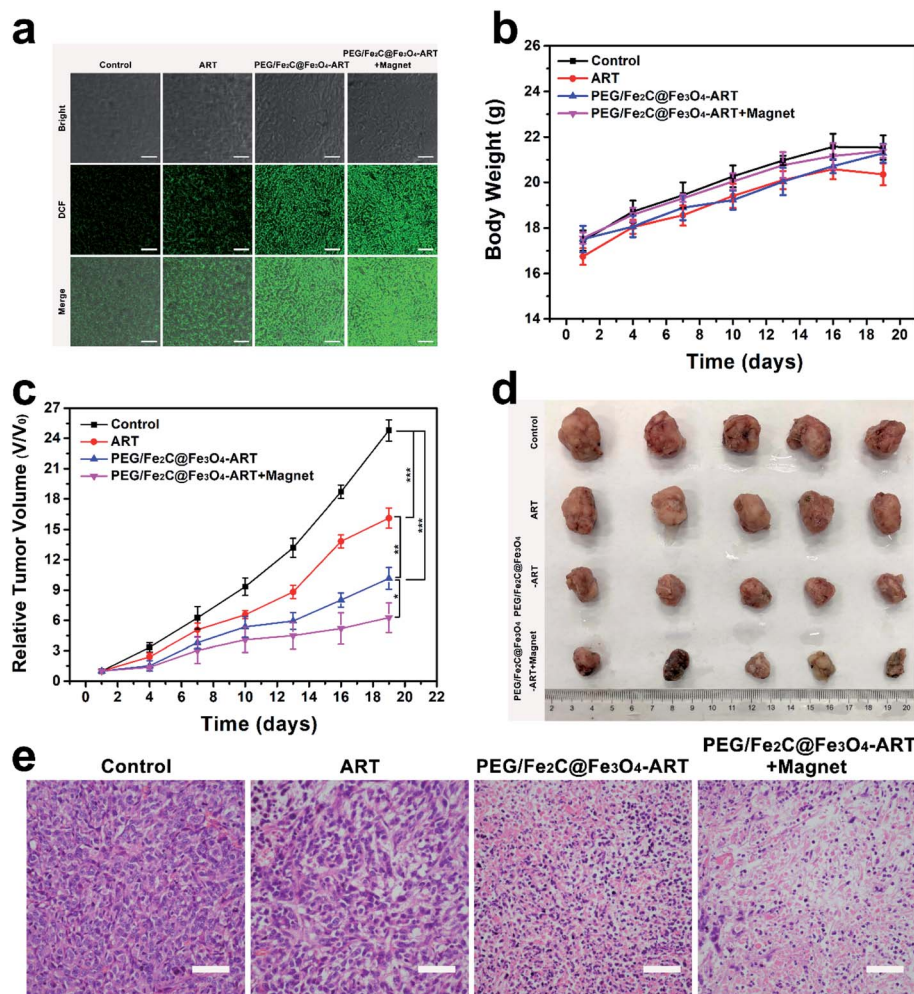


Fig. 4 *In vivo* experiments with PEG/Fe₂C@Fe₃O₄-ART NPs. (a) Fluorescence image of DCFH-DA labeled tumor slices of the mice after the different treatment (scale bar 25 μ m). (b) Time course change in the body weight after different treatments ($n = 5$, mean \pm s.d.). (c) Time course change in the relative tumor volume after different treatments ($n = 5$, mean \pm s.d., * $p < 0.05$, ** $p < 0.01$, *** $p < 0.001$). (d) Digital photograph of excised tumors after 18 days of various treatments. (e) H&E staining of 4T1 tumor sections in different groups (scale bar 100 μ m).

confirmed the cell apoptosis was primarily originated from ROS. From the above, we can conclude that PEG/Fe₂C@Fe₃O₄ NPs can provide Fe²⁺ in tumor cells and *in situ* activate ART to generate abundant ROS, which result in the cytotoxicity for cell killing.

Inspired by the excellent *in vitro* cell growth inhibition based on PEG/Fe₂C@Fe₃O₄-ART NPs, *in vivo* biological performances were then evaluated on 4T1 tumor-bearing mice. Benefiting from the magnetic property of PEG/Fe₂C@Fe₃O₄ NPs, bio-distribution of NPs was initially evaluated by measuring Fe concentration in major organs and tumors 24 h post-injection. Compared with magnetic field-free group, in which PEG/Fe₂C@Fe₃O₄ NPs were primarily distributed in liver, and a few NPs accumulated in tumor tissue *via* EPR effect, the magnetic targeting group could effectively accumulate NPs in tumor tissue by locating a magnet at tumor (Fig. S15[†]). Furthermore, *in vivo* NIR images further verified PEG/Fe₂C@Fe₃O₄ NPs could be effectively accumulated in tumor site by magnetic targeting (Fig. S16[†]). Together with the acid environment of tumor, the

effective enrichment of NPs in tumors by magnetic targeting makes PEG/Fe₂C@Fe₃O₄ NPs as good carriers to effective deliver ART and donate Fe²⁺ for the *in situ* activation of ART by ROS generation.

Thus, we next investigated the *in vivo* tumor ROS concentration. As shown in Fig. 4a, bright green fluorescence was emitted only in the group that had been subjected to magnetic-assisted intravenous (*i.v.*) injection of PEG/Fe₂C@Fe₃O₄-ART NPs after staining with DCFH-DA, illustrating amounts of ROS could be generated by specifically accumulating PEG/Fe₂C@Fe₃O₄-ART NPs in tumor with Fe²⁺-ART catalysis. The high ROS intensity led to tumor cell apoptosis and even necrosis. Therefore, tumor inhibition assay was then carried out. 4T1 tumor-bearing female Balb/c mice were established and divided into four groups, which were *i.v.* injected with either saline, ART only, PEG/Fe₂C@Fe₃O₄-ART NPs or PEG/Fe₂C@Fe₃O₄-ART NPs with magnetic targeting. Tumor volume and body weight were monitored one-day post-injection. No obvious body-weight changes were observed during the overall therapeutic process



(Fig. 4b). More importantly, tumor growth in the group treated with magnetic-assisted *i.v.* injection of PEG/Fe₂C@Fe₃O₄-ART NPs was significantly suppressed (Fig. 4c and d), compared to the control group, the ART-only group, and *i.v.* injection of PEG/Fe₂C@Fe₃O₄-ART NPs group (Fig. S17†). Besides, tumor sections from mice dissected at the end of the treatment period conducted hematoxylin–eosin (H&E) and TdT-mediated dUTP nick-end labeling (TUNEL) staining, which showed that the multitude of cell necrosis and apoptosis occurred in the tumor sections from mice treated with PEG/Fe₂C@Fe₃O₄-ART NPs (Fig. 4e and S18†). By contrast, no obvious cell necrosis was observed in the control group or the ART-only group, indicating the improved chemotherapeutic efficacy of PEG/Fe₂C@Fe₃O₄-ART NPs for ART-based chemotherapy.

Eventually, systemic biotoxicity valuation is crucial for nanomaterials to clinical research. To explore long-term biosafety, the normal hematology parameters and standard blood biochemical indexes were also measured in healthy mice. As shown in the blood biochemistry and hematology analysis, no obvious changes were induced by PEG/Fe₂C@Fe₃O₄-ART NPs (Fig. S19†). H&E staining also showed no obvious pathological toxicity in the major organs, indicating high biocompatibility of PEG/Fe₂C@Fe₃O₄-ART NPs (Fig. S20†).

Conclusions

In summary, a H₂O₂-free and tumor-selective chemodynamic therapy agent was constructed by ART loaded magnetic iron carbide NP, which *in situ* generated ROS by releasing Fe²⁺ and ART under tumor specificity environment after magnetic targeting. The core–shell structured PEG/Fe₂C@Fe₃O₄ magnetic nanoparticles were synthesized to deliver ART into the tumors, where they could rapidly co-release Fe²⁺ and ART in a tumor-specific manner. The released ART molecules then transformed into toxic ROS by *in situ* catalysis of Fe²⁺, resulting in apoptosis of tumor cells. These features synchronously endow the nanocarrier with enhanced efficacy and biosafety for tumor-specific therapy. Both *in vitro* and *in vivo* experiments confirmed the high chemotherapeutic efficacy and revealed the related therapeutic mechanism. This work sets an example of iron-based magnetic nanocarrier with excellent Fe²⁺ and drug release behavior in the tumor microenvironment after magnetic-assisted tumor enrichment for *in situ* enhanced Fe²⁺-dependent chemodynamic.

Experimental section

Materials

Iron carbonyl (Fe(CO)₅, 99%), oleylamine (OAm), oleic acid (OAc, 90%), 1-octadecene (ODE, 90%) and ammonium bromide (NH₄Br, 99%) were purchased from Alfa Aesar. Hexane (C₆H₁₄), and ethanol (C₂H₆OH) of analytic grade were from the Juhua Group Factory, China. Artemisinin (ART, 98%) was purchased from J&K. 1,2-Distearoyl-*sn*-glycero-3-phosphoethanolamine-2000 (DSPE-PEG) was purchased from Toyongbio corporation, china. All chemicals were used as received without further purification.

Characterization

X-ray diffraction (XRD) patterns were recorded on a X'Pert PRO X-ray diffractometer equipped with Cu K α (λ = 1.54178 Å) radiation. Transmission electron microscopy (TEM) was carried out on a FEI Tecnai G2 F30 microscope. High-resolution TEM (HRTEM) was carried out on a FEI Tecnai G2 F30 microscope. Dynamic light scattering was measured using a particle size analyzer (ZetaPALS, Brookhaven Instruments). X-ray photoelectron spectroscopy (XPS) measurements were carried out on an Axis Ultra imaging photoelectron spectrometer (Kratos Analytical Ltd). Magnetization was measured by a superconducting quantum interference device (SQUID). The concentrations of Fe were quantified using an inductively coupled plasma-atomic emission spectrometer (ICP-AES, Profile, Leeman, USA). Fourier transform infrared radiation (FT-IR) spectrum was determined using a Nicolet 6700 spectrometer. Ultraviolet visible (UV-vis) absorption spectra were measured on a HACH DR6000 ultraviolet visible absorption spectrometer.

Synthesis of Fe₂C@Fe₃O₄ NPs

In the typical synthesis, bcc-Fe NPs were firstly synthesized by the following steps. ODE (15 mL), NH₄Br (0.05–0.1 mmol) and OAm (0.4 mL) were mixed magnetically and degassed under a gentle Ar flow for 1 h in a four-neck flask. The solution was then heated to 100 °C and kept at this temperature for 1 h before it was heated further to 180 °C. After that, Fe(CO)₅ (0.5–0.7 mL) was injected to the reaction mixture and kept there for 30 min. OAc (0.1 mL) was added *via* a syringe and the resultant solution was aged at 140 °C for another 30 min before it was cooled down to room temperature. The complex was collected through magnetic separation and rinsed three times with ethanol. After centrifugation (5 min, 10 000 rpm), the product was collected and re-dispersed in hexane. ODE (3.0 g), OAm (10 mL) was magnetically blended in the four-neck flask and degassed under a gentle Ar flow for 1 h at 120 °C. Then, the resulted bcc-Fe NPs (5 mmol, in 10 mL hexane) was added *via* a syringe and the reaction solution was heated at 280–340 °C for 15–45 min. The black-brown colored solution was cooled down to room temperature. Ethanol was added and the mixture was centrifuged (5 min, 10 000 rpm) three times. The precipitates were re-dispersed in hexane.

Preparation of ART-loaded Fe₂C@Fe₃O₄ NPs

Fe₂C@Fe₃O₄ (5 mg) was added to chloroform (10 mL) and then sonicated at room temperature. Then artemisinin (10 mg) and DSPE-PEG (20 mg) dissolving in 10 mL chloroform was slowly added to the above Fe₂C@Fe₃O₄ chloroform mixture solution and sonicated for 30 min before the solvent was evaporated by a rotary evaporator. The final products were washed with ethanol to remove the free drug. The loading efficiency of ART was measured by a UV-vis spectrophotometer. Unloaded ART in the supernatant was converted into a UV-absorbing compound through treatment with sodium hydroxide aqueous solution at 50 °C for 30 min and the detection wavelength was 292 nm.



Fe²⁺ release from PEG/Fe₂C@Fe₃O₄ NPs

PEG/Fe₂C@Fe₃O₄ NPs solution (1 mL) with Fe concentration of 2 mg mL⁻¹ was sealed in a dialysis bag (MCWO: 1000 Da), and immersed in 20 mL buffer media (PBS) at different pH values (7.4, 6.5 and 5.4) in a centrifuge tube. Centrifuge tubes were then shaken with a speed of 200 rpm at 37 °C. At given intervals, 1 mL of buffer was collected and analyzed the Fe concentration by ICP-AES. 1 mL fresh buffer medium was returned. Fe released ($R_{(Fe,n)}$) was calculated by the following equation:

$$R_{(Fe,n)} (\%) = \frac{20C_n + \sum_{i=1}^{n-1} C_i}{2 \times 1} \times 100\%$$

where C_n is the iron concentration tested in n th collection, C_i is the iron concentration in the i th collection. Unit of C_n and C_i are mg mL⁻¹.

In vitro MRI test

The *in vitro* MRI tests were conducted on a 3.0 T clinical MRI instrument (Philips). To test the *in vitro* longitudinal relaxation rate r_2 , PEG/Fe₂C@Fe₃O₄ NPs at the iron concentration from 0.003 mM to 0.021 mM with the interval of 0.003 mM were dispersed in 1 mL of solution with pH 5.4 and 7.4 containing 1 wt% agarose gel respectively and placed in centrifuge tubes (1.5 mL). MR images were acquired using a T_2 -weighted sequence with the following parameters: repetition time (T_R) = 500 ms; echo time (T_E) from 12.25 ms to 196 ms, with the interval of 12.25 ms; field of view (FOV) = 180 mm; slice thickness = 2 mm; NEX = 1.

Cell culture

4T1 cell line was obtained from the Zhejiang Provincial People's Hospital. All cell-culture-related reagents were purchased from Invitrogen. Cells were cultured in DMEM culture medium supplemented with 10% FBS and 1% penicillin/streptomycin at 37 °C under 5% CO₂ with 100% humidity.

The intracellular location of PEG/Fe₂C@Fe₃O₄ NPs in 4T1 cells

The as-prepared of PEG/Fe₂C@Fe₃O₄ NPs was labeled with fluorescein isothiocyanate (FITC) (λ_{ex} = 488 nm, λ_{em} = 517 nm) defined as PEG/Fe₂C@Fe₃O₄-FITC. 4T1 cells (5×10^4 cells per well) were seeded into a 24-well plate. When the cell confluence reached 80%, 300 μ g mL⁻¹ of PEG/Fe₂C@Fe₃O₄-FITC NPs were added into the culture medium and co-cultured with cells at 37 °C for 4 h. After that, 4T1 cells were counterstained with Lyso-Tracker red (λ_{ex} = 577 nm, λ_{em} = 590 nm) for 30 min and DAPI (λ_{ex} = 359 nm, λ_{em} = 461 nm) for 15 min. Finally, the cells were washed three times with PBS to remove the NPs that have not been endocytic. The images were taken by using a confocal microscope (Leica SP8).

Intracellular ROS burst detection

4T1 cells were plated at a density of 5×10^4 cells per well into a 24-well plate and were incubated for 12 h. Cells were loaded

with 2,7-dichlorodihydrofluorescein diacetate (DCFH-DA, 10 μ M) for 30 min and then washed with PBS for three times. Then, cells were treated by ART or PEG/Fe₂C@Fe₃O₄ NPs ([Fe] = 300 μ g mL⁻¹) or PEG/Fe₂C@Fe₃O₄-ART NPs ([Fe] = 300 μ g mL⁻¹). PEG/Fe₂C@Fe₃O₄-ART group has an equivalent ART dosage to the free ART group. The intensity of fluorescence was recorded every 15 min over a period of 9 h *via* excitation of 488 nm and emission at 525 nm by BioTek ELx800.

For the fluorescent microscopic examination, 4T1 cells with the cell density of 1×10^5 cells per well were plated onto coverslips in 12-well plates overnight. Then, cells were pre-stained with 10 μ M DCFH-DA for 30 min and washed with PBS to remove the free DCFH-DA. Later, the cells were incubated with ART or PEG/Fe₂C@Fe₃O₄ NPs ([Fe] = 300 μ g mL⁻¹) or PEG/Fe₂C@Fe₃O₄-ART NPs ([Fe] = 300 μ g mL⁻¹) for 4 h before they were washed with PBS for three times. The images were acquired by a Leica SP8 confocal microscope with excitation at 488 nm and emission from 525 nm.

In vitro cytotoxicity assays

The *in vitro* cytotoxicity was evaluated by standard MTT assay. 4T1 cells were seeded into 96-well cell culture plates at 5×10^3 cells per well and incubated overnight at 37 °C under 5% CO₂. After that, cells were treated with various formulations of free ART, PEG/Fe₂C@Fe₃O₄, and PEG/Fe₂C@Fe₃O₄-ART at the same Fe₂C@Fe₃O₄ and ART concentrations. After further incubation for 24 h, the culture media were replaced by medium containing 0.5 mg mL⁻¹ 3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide (MTT). Finally, the MTT solution was replaced by 100 μ L of dimethyl sulfoxide (DMSO) after co-incubation for 4 h. By using an ELISA reader (Tecan m200), cell proliferation was measured. Furthermore, after 4T1 cells were treated under the aforementioned conditions, cells were stained with calcein acetoxymethyl ester (calcein-AM, 2 μ M, Yeason Biotech, China) and propidium iodide (PI, 4.5 μ M, Yeason Biotech, China) to distinguish the live cells from dead cells through a fluorescence microscope (Nikon ECLIPSE Ti).

Assessment of apoptosis

Apoptotic status was determined by FITC-conjugated Annexin-V/PI assay kit (Sony) using flow cytometry following the manufacturer's instructions. Briefly, 5×10^4 per well 4T1 cells were seeded in 6-well plates for 12 h and treated with drugs for a further 24 h. The cells were rinsed with PBS and detached using EDTA-free trypsin. The detached cells were resuspended in 100 μ L binding buffer containing FITC-conjugated Annexin-V/PI and incubated at RT for 15 min. The cells were diluted in 400 μ L of buffer and analyzed by flow cytometry (NovoCyt 3130, USA). Apoptosis and necrosis were evaluated using PI *vs.* Annexin V plots. The cells stained with Annexin V only were classified as early apoptosis and the Annexin V and PI double-stained cells were classified as late apoptosis or necrosis.

Animal modal

All animal procedures were performed in accordance with the Guidelines for Care and Use of Laboratory Animals of Zhejiang



Chinese Medical University and experiments were approved by the Animal Ethics Committee of Zhejiang Chinese Medical University. 6 week female Balb/c mice, with an average weight of 17 g, were provided by the Shanghai Laboratory Animal Center, Shanghai, China. Mice were injected with 1×10^6 4T1 cells (0.1 mL cells in PBS) subcutaneously at the right axillary region.

Biodistribution studies

To qualitatively investigate the biodistribution of PEG/Fe₂-C@Fe₃O₄ NPs, 6 mice were assigned into two groups randomly. When the tumor volume reached to about 150 mm³, mice were intravenously injected of PEG/Fe₂C@Fe₃O₄ NPs (10 mg_{Fe} kg⁻¹). In the magnetic targeting group, a magnet was located at the tumor site. Mice were sacrificed after heart perfusion at time point of 24 h with major organs (heart, liver, spleen, lung and kidneys) and tumors dissected, rinsed and weighted. The *in vivo* biodistribution of Fe element was then measured by ICP-AES and calculated as Fe percentage over administrated dose per gram of tissues.

In vivo tumor chemotherapy of PEG/Fe₂C@Fe₃O₄-ART

When the tumor volume reached to about 100 mm³, 20 mice were randomly divided into 4 groups: (1) mice were intravenously injected of saline only as control; (2) mice were intravenously injected of saline with free ART; (3) mice were intravenously injected of saline with PEG/Fe₂C@Fe₃O₄-ART NPs; (4) mice were intravenously injected of saline with PEG/Fe₂C@Fe₃O₄-ART NPs as well as a magnet located at the tumor site. Both PEG/Fe₂C@Fe₃O₄-ART group and PEG/Fe₂C@Fe₃O₄-ART + magnet group have an equivalent ART dosage to the free ART group and have an equivalent particles dosage. All groups were injected once daily every three days in the period of 18 days with the dosage of 50 mg_{Fe} kg⁻¹ and tumor volume was monitored 1 day post injection. Tumor volumes and body weight were monitored 1 day post injection during the treatment ($n = 5$). Tumor volume was calculated according to the formula of $(a \times b^2)/2$, where a and b are the long and short diameters of the tumor, respectively.

At given intervals of 19 days, mice from each group were euthanized and major visceral organs (heart, liver, spleen, lung, and kidney) and tumor were recovered, followed by fixing with 10% neutral buffered formalin. After the organs were embedded in paraffin and sectioned at 5 μ m, hematoxylin and eosin (H&E) and TdT-mediated dUTP nick-end labeling (TUNEL) staining was performed. The slides were observed under optical microscope (Nikon ECLIPSE Ti).

Quantitative analysis of the ROS generation in tumors

At the 3rd day post injection, one mouse from each group was euthanized, and tumors were harvested from the necropsy. The tumor tissues were firstly embedded and frozen by an OCT (Jung, Tissue freezing medium, Leica). Cross sections of 10 μ m thickness were cut using a cryomicrotome (Leica), staining with DCFH-DA, 10 μ M for 30 min and mounted on the glass slides. The slides were then observed under a confocal microscope (Leica SP8, $\lambda_{\text{ex}} = 488$ nm; $\lambda_{\text{em}} = 525$ nm).

Statistical analysis

Quantitative data were expressed as mean \pm s.d. One-way ANOVA statistical analysis was used to analyze differences between datum. Values with $p < 0.05$ were considered statistically significant (* means $p < 0.05$, ** means $p < 0.01$, *** means $p < 0.001$).

Author contributions

Fan Zhao: contributed equally to this work, carried out all experiments and performed the statistical analysis, contributed to discussion, writing – original draft. Jing Yu: contributed equally to this work, carried out all experiments and performed the statistical analysis, conceptualization, writing – review & editing. Weiliang Gao: participated in the animal studies and molecular biology experiments. Xue Yang: participated in the animal studies and molecular biology experiments. Liying Liang: participated in molecular biology experiments. Xiaolian Sun: supervision. Dan Su: supervision. Juan Li: supervision. Yao Ying: supervision. Wangchang Li: supervision. Liang Qiao: supervision. Jingwu Zheng: supervision. Wei Cai: supervision. Shenglei Che: conceptualization, supervision, writing – reviewing and editing. Xiaozhou Mou: conceptualization, supervision, writing – reviewing and editing.

Conflicts of interest

There are no conflicts to declare.

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