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Diarylheptanoid analogues from the rhizomes of *Zingiber officinale* and their anti-tumour activity†

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Eight previously undescribed diarylheptanoids (1–8), together with fifteen known analogues (9–23), were isolated from the rhizomes of *Zingiber officinale*. Their structures were unambiguously determined by comprehensive spectroscopic analyses and electronic circular dichroism (ECD) calculations. It is worth mentioning that 1–3 are the first reported structures of diaryl ether heptanoids in *Z. officinale*, whereas 15–17 were isolated from *Zingiber* for the first time. Furthermore, a gene enrichment analysis of the interacting targets indicated that diarylheptanoids were mainly associated with the anti-tumor activity by affecting DNA damage signaling pathway. The results showed that 6, 16–19 had remarkable inhibitory effects against five tumor cell lines (A549, HepG2, HeLa, MDA-MB-231, and HCT116) with IC₅₀ values ranging from 6.69–33.46 μM, and showed down-regulating the expression of ATR (ataxia telangiectasia mutated and RAD3-related) and CHK1 (checkpoint kinase 1) levels in HCT116 and A549 cell lines. Our studies not only enrich the structural diversity of diarylheptanoids in nature, but also discover several natural compounds for anti-tumor agents.

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1. Introduction

Ginger (*Zingiber officinale* Roscoe; family: Zingiberaceae) has long been used worldwide as an important cooking spice, condiment, and dietary supplement.^{1,2} It is also valued as a medicinal herb for its prevention and treatment of a wide range of diseases, such as colds, stomachaches, headaches, nausea, diarrhoea, indigestion, rheumatic, cardiopathy and cholera.^{3–7} Gingerol-related compounds as well as diarylheptanoids are the major classes of biologically active natural products in ginger. In the past decades, the extensive investigations have been made on ginger and gingerols derivatives.^{8,9} Diarylheptanoids, known as the other significant phenolic components in ginger, have recently attracted widespread attentions for their structural diversity and potential pharmacological effects.

Diarylheptanoids are structurally diverse and have the skeletal structure of two aromatic rings conjugated with seven carbon chains, and is categorized into linear, cyclic, and dimeric diarylheptanoids, or diarylheptanoids with special moieties. Numerous pharmaceutical research revealed that diarylheptanoids possess anti-inflammatory,¹⁰ anti-oxidant,¹¹ anti-tumor,¹² leishmanicidal,¹³ melanogenesis,¹⁴ hepatoprotective,¹⁵ and neuroprotective activities.¹⁶ A variety of diarylheptanoids have been isolated from seeds, fruits, leaves and roots of plants of different families.¹⁷ Ginger is one of the most important producers of diarylheptanoids in nature. In our previous study, gingerol derivatives including six new compounds were reported from ginger rhizomes.¹⁸ As a part of a continuing search for new bioactive agents,²¹ diarylheptanoids including eight undescribed ones (1–8), were identified from the ginger. To rapidly reveal the biological activity, a network pharmacology-based approach was constructed to guide the discovery of targets and biological functions of diarylheptanoid compounds. The results demonstrated that the isolated diarylheptanoids were mainly associated with the antitumor activity by affecting DNA damage signalling pathway. Consequently, cytotoxicity and enzyme activity assays were designed to validate the predictions, suggesting that compounds 6, 17, and 18 may exert anticancer effects by regulating the ATR/CHK1 signalling pathway. Herein, the isolation, structural elucidation, targets prediction, cytotoxic evaluation, enzymatic activity assays, and molecular simulation of these compounds are discussed.

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2. Materials and methods

2.1. Instrumentation and reagents

Ultraviolet spectra (UV) were recorded on a JASCO V-550 UV spectrometer (JASCO, Tokyo, Japan). Optical rotations were determined in CHCl₃ using a JASCO P-1020 polarimeter (JASCO, Tokyo, Japan). Circular dichroism (CD) spectra were tested by JASCO J-810 circular dichroism spectrometer (Jasco, Tokyo, Japan). Infrared spectra (IR) were measured with a JASCO FT/IR-480 plus spectrometer (JASCO, Tokyo, Japan). 1D and 2D NMR spectra were acquired on a Bruker AV 600 (Bruker Co. Ltd, Bremen, German) using solvent signals (CDCl₃: δ_{H} 7.26/ δ_{C} 77.2; CD₃OD: δ_{H} 3.31/ δ_{C} 49.0) as internal references. Deuterated solvents were purchased from Cambridge Isotope Laboratories, Inc. (Saint Louis, Missouri, USA). The HR-ESI-MS spectra were obtained on a Micromass Q-TOF mass spectrometer (Waters Corporation, Milford, USA). Analytical high-performance liquid chromatography (HPLC) was carried out on a Shimadzu HPLC system (Shimadzu, Kyoto, Japan) with an LC-20AB solvent delivery system and an SPD-20A UV/vis detector using a Phenomenex Gemini C₁₈ column (5 μm , Φ 4.6 \times 250 mm; Phenomenex Inc., Los Angeles, USA). Semi-preparative HPLC was performed on a Shimadzu HPLC system (Shimadzu, Kyoto, Japan) equipped with LC-6AD solvent delivery system and a SPD-20A detector on a Phenomenex Gemini C₁₈ column (5 μm , Φ 10.0 \times 250 mm; Phenomenex Inc., Los Angeles, USA).

Column chromatography (CC) was carried out on silica gel (300–400 mesh, Haiyang Chemical Co. Ltd, Qingdao, China), ODS (12 nm, S-50 μm , YMC Ltd, Tokyo, Japan) and Sephadex LH-20 (Amersham Pharmacia Biotech Co. Ltd, Atlanta, USA). Thin-layer chromatography (TLC) was performed on pre-coated silica gel plates (GF254, Qingdao Haiyang Co. Ltd, Qingdao, China). HPLC-grade methanol and acetonitrile (CH₃CN) were purchased from Oceanpak Alexative Chemicals Co. Ltd. (Göteborg, Sweden). All analytical grade reagents were from Concord Chemicals Co. Ltd (Tianjin, China).

2.2. Plant materials

Dried rhizomes of *Z. officinale* were collected from Jinxiang County, Shandong, China, in September 2016. The plant material was identified by Prof. Guang-Xiong Zhou of Jinan University. A voucher specimen (JNU-ZO-201609) was deposited at the Institute of Traditional Chinese Medicine and Natural Products, College of Pharmacy, Jinan University, Guangzhou, China.

2.3. Extraction and isolation

The dried rhizomes (30 kg) of *Z. officinale* were extracted with 70% ethanol (210 L) under reflux for 2 times at 60 °C, 2 h each time. The combined ethanol extract was evaporated under reduced pressure. The ethanol extract (5.5 kg) was suspended in 95% ethanol (16.5 L) and partitioned successively with petroleum ether and ethyl acetate. The ethyl acetate portion (ZO-2, 920 g) was divided into 22 fractions (Fr. A to Fr. V) on a silica gel column chromatograph (CC) (200–300 mesh, Φ 6.5 \times 130 cm) by gradient elution with solvent composed of cyclohexane-EtOAc (v/v, 97 : 3, 95 : 5, 9 : 1, 8 : 2, 7 : 3, 6 : 4, and 0 : 1).

Fr. O (22 g) was eluted (CH₃OH–H₂O, 5 : 5 \rightarrow 10 : 0) by ODS to yield 11 fractions (Fr. O1–O11). Fr. O7 was separated by silica gel CC (CHCl₃–acetone 95 : 5, v/v) to give 4 fractions (Fr. O7A–O7D). Compounds 1 (1.4 mg, t_{R} = 35.7 min), and 2 (1.0 mg, t_{R} = 36.5 min) were obtained from Fr. O7A by semi-preparative HPLC (45% CH₃OH). Subsequently, Fr. O9 was purified by preparative HPLC to yield compound 4 (14.1 mg, t_{R} = 84 min, 53% CH₃OH).

Fr. Q (16.0 g) was subjected to ODS eluting with CH₃OH–H₂O (5 : 5 \rightarrow 10 : 0) to afford Fr. Q1–Q17. Among them, compound 7 (2.9 mg, t_{R} = 37.5 min, 58% CH₃CN) was purified by semi-preparative HPLC from Fr. Q10. Fr. Q7 was separated by semi-preparative HPLC (29% CH₃CN) to yield 9 fractions (Fr. Q7A–Q7I), Fr. Q7I and Q7M were further treated by semi-preparative HPLC to obtain compounds 5 (1.3 mg, t_{R} = 18.2 min, 50–60% CH₃OH, 40 min) and 6 (3.0 mg, t_{R} = 19.5 min, 50–70% CH₃OH, 40 min), respectively.

Fr. R (29.0 g) was applied to a ODS column using CH₃OH–H₂O (5 : 5 \rightarrow 10 : 0) as elute to obtain 11 fractions (R1–R11). Then, Fr. R4 was treated by semi-preparative HPLC eluting with 23% CH₃CN to obtain a mixture (R4G), and further purified by semi-preparative HPLC (50% CH₃OH) to yield compound 3 (4.8 mg, t_{R} = 11.9 min). Fr. R9 was treated by semi-preparative HPLC eluting with 42% CH₃CN, and compound 8 (11.6 mg, t_{R} = 25.6 min) was obtained. In addition, detailed isolation procedures for known compounds (9–23) and spectroscopic data of isolated new compounds are provided in the ESI.†

2.4. ECD calculation

The ECD spectra of (5*R*)-3 and (5*S*)-3 were calculated using Gaussian 09 software. Firstly, predominant conformations were obtained using conformation searches. Secondly, all conformations were optimized at B3LYP/6-31G(d) level. Thirdly, the B3LYP/6-311+G(d,p) level was used to calculate the compound's ECD spectra. Finally, The ECD spectra were combined after Boltzmann weighting based on the distribution of structural energy.

2.5. Compound-target network construction and analysis

The targets of the isolated compounds were predicted by using SwissTargetPrediction,¹⁹ and the cancer-related targets were extracted from GenCLIP3.²⁰ Next, these targets were merged to obtain the overlapped targets, and the compounds-antitumor targets network was built by using the overlapped targets and their corresponding compounds. The key anticancer targets were identified based on the intermediate value of degree. Furthermore, the potential biological activity of these isolated-compounds were predicted by using ClueGO program in Cytoscape 3.5 software.²¹

2.6. Assays for cell proliferation

2.6.1. Cell culture. Five tumour cell lines A549, HepG2, HeLa, MDA-MB-231, and HCT116 were obtained from ATCC and inoculated in DMEM supplemented with 10% fetal bovine serum (FBS, Gibco, NewYork, USA) under a humidified atmosphere of 95% air and 5% CO₂ at 37 °C. The culture medium was



changed every day. The cultured cell morphology was compared with the normal cell morphology in the ATCC cell bank to determine the cell status.

2.6.2. Cell viability assay. The colorimetric [3-(4,5-dimethyl-thiazol-2-yl) 2,5-diphenyltetrazolium bromide] (MTT, Aladdin, Shanghai, China) assay was employed to quantify the cell proliferation. Five tumour cells were seeded in 96-well plates (100 μL medium per well) at a concentration of 5×10^3 cells per well, and cultured for 24 h. Subsequently, the media were removed and replaced by fresh medium containing different concentrations of drugs, and the control group was incubated with drug-free medium. After culturing for 24 h, the original medium was discarded and replaced by 100 μL complete medium with 0.5 mg mL^{-1} 20 μL MTT. After incubation with 5% CO_2 -95% air at 37 $^\circ\text{C}$ for 4 h, the MTT solution was removed and followed by adding 150 μL DMSO, and the plates were shaken for 5 min to dissolve the crystals fully. The optical density of each condition was measured by the microplate reader (BioTek, Vermont, USA) at a wavelength of 490 nm. Each experiment was repeated in triplicates.

2.6.3. Western blotting. The HCT116 cells were seeded in 60 mm dishes overnight and treated with the indicated compounds. Harvested cells were disrupted with cell lysis buffer (50 mM Tris-HCl, pH 8.0; 150 mM NaCl; 0.5% Na-deoxycholate; 1% NP-40; 0.1% SDS), and with protease

inhibitor mixture. After sonication, the cell lysates were centrifuged at 14 000g and 4 $^\circ\text{C}$ for 15 min. Protein concentration of supernatant was determined. Equal amounts of lysate protein were separated by SDS-polyacrylamide gel electrophoresis and electrotransferred to polyvinylidene difluoride (PVDF) membranes. The membranes were blocked with 5% nonfat skim milk in TBST [20 mM Tris-HCl (pH 7.6), 135 mM NaCl and 0.1% Tween 20], and then incubated with specific primary antibodies to ATM, ATR, and P53 (Cell Signalling Technology, Danvers, MA) at 4 $^\circ\text{C}$ overnight. β -Actin was used as a loading control. The next day, the membrane was incubated with goat anti-mouse IgG-HRP secondary antibody (Cell Signalling Technology, Danvers, MA) at room temperature for 2 h. The signals were detected by chemiluminescence utilizing the enhanced chemiluminescent reagent and recorded on the gel imaging system (Tanon, Shanghai, China) and relative quantified using Image J software program. The experiment was repeated three times, and A549 cells were tested in the same way. The test results of A549 cell line are provided in the ESI (Fig. S72 \dagger).

2.6.4. Statistical analysis. Quantitative data were presented as mean \pm SD. The statistical analysis was performed by GraphPad Prism 4.0 (GraphPad Software Inc). The statistical significance between groups was interpreted by one-way ANOVA analysis of variance, followed by Tukey's test. All statistical tests with $p < 0.05$ were considered significantly different.

Table 1 ^1H (600 MHz) and ^{13}C (150 MHz) NMR data of compounds 1–3

Pos.	1^a		2^b		3^b	
	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)
1	29.2	2.85, ddd (16.4, 8.7, 2.4) 2.95	33.6	3.01	32.9	2.96, ddd (12.9, 5.8, 2.9) 3.04, dt (12.9, 5.8)
2	42.0	2.49, ddd (15.0, 8.7, 2.4)	43.9	2.64	44.6	2.51 2.79, dt (12.2, 5.8)
3	201.1		200.8		215.6	
4	131.8	5.86, d (15.2)	131.4	5.73, dt (15.3, 1.5)	52.4	1.89, d (18.3) 2.22, dd (18.3, 10.5)
5	147.9	6.58, ddd (15.2, 8.6, 7.5)	148.2	6.56, dt (15.3, 7.0)	66.1	3.34, br t (8.4)
6	35.1	2.44 2.59, dq (12.6, 6.3)	32.8	2.34	34.8	1.65 1.33
7	34.6	2.95	32.0	2.70	29.0	2.49 2.72, ddd (16.8, 11.4, 2.1)
1'	132.6		132.7		131.4	
2'	106.5	6.38, d (1.8)	105.7	6.28, d (1.7)	105.9	6.34, br s
3'	149.6		147.3		147.9	
4'	134.1		132.9		133.1	
5'	149.9		147.9		147.6	
6'	106.3	4.99, d (1.8)	106.8	5.52, d (1.7)	107.4	5.33, br s
1''	140.8		138.5		138.0	
2''	115.8	6.80, d (1.8)	114.9	6.77, d (1.8)	116.2	6.59, d (1.8)
3''	154.0		152.7		151.9	
4''	144.1		143.5		145.2	
5''	125.1	6.92, d (8.0)	124.6	7.01, d (8.0)	124.7	7.21, d (8.0)
6''	122.5	6.73, dd (8.0, 1.8)	121.8	6.83, dd (8.0, 1.8)	120.2	6.93, dd (8.0, 1.8)
3'-OCH ₃	56.7	3.82, s	56.4	3.87, s	56.4	3.87, s
3''-OCH ₃	56.6	3.68, s	56.5	3.72, s	56.3	3.59, s

^a Measured in CD_3OD . ^b Measured in CDCl_3 .



3. Results and discussion

3.1. Structural elucidation of new compounds

Compound **1** was isolated as a yellow oil. The molecular formula was determined to be $C_{21}H_{22}O_5$ on the basis of the HR-ESI-MS (m/z 355.1539 $[M + H]^+$, calcd 355.1545). The 1H NMR spectrum showed diagnostic signals of one 1,2,4-trisubstituted benzene [δ_H 6.92 (1H, d, $J = 8.0$ Hz, H-5''), 6.80 (1H, d, $J = 1.8$ Hz, H-2''), and 6.73 (1H, dd, $J = 8.0, 1.8$ Hz, H-6'')] and one 1,2,3,5-tetrasubstituted benzene [δ_H 6.38 (1H, d, $J = 1.8$ Hz, H-2'), and 4.99 (1H, d, $J = 1.8$ Hz, H-6')], a *trans*-substituted double bond [δ_H 6.58 (1H, ddd, $J = 15.2, 8.6, 7.5$ Hz, H-5), 5.86 (1H, d, $J = 15.2$ Hz, H-4)], and two methoxys [δ_H 3.82 (3H, s, 3'-OCH₃), 3.68 (3H, s, 3''-OCH₃)]. The ^{13}C NMR and DEPT data revealed the presence of 21 carbon signals (Table 1), including eight quaternary carbons (δ_C 201.1, 154.0, 149.9, 149.6, 144.1, 140.8, 134.1, 132.6), seven methines (δ_C 147.9, 131.8, 125.1, 122.5, 115.8, 106.5, 106.3), four methylenes, and two methoxy carbon signals. The above NMR data supported **1** to be a cyclic diaryl-heptanoid.²² The structural elucidation of **1** was accomplished by analysis of COSY, HSQC and HMBC data (Fig. 2). In particular, the 1H - 1H COSY correlations of H₂-1/H₂-2 and H-4/H-5/H₂-6/H₂-7, combined with the HMBC correlations of H₂-1, H₂-2/C-3, H-4, H-5/C-3, and H₂-7/C-5 revealed the presence of a heptane chain moiety. Further, two *meta*-coupling doublets, associated with chemical shift characteristics, and key HMBC peaks allowed the assignments of aromatic ring A. In aromatic B, HMBC correlations from a broad *ortho* doublet H-5'' to C-1'', 3'', 4'', from a narrow *meta* doublet H-2'' to C-3'', 4'', 1'', 6'', and from a doublet of doublets to C-1'', 2'', 4'', completed its assignments of the signals. The linkage of alkyl and aromatic groups was built by the HMBC correlations of H₂-7 to C-1'', 2'', 6'', H₂-2'', 6'' to C-7, as well as H-2', 6'/C-1, and H₂-1/C-1', 2', 6', indicating the

connection of C-1/C-1' and C-7/C-1''. The HMBC correlations between δ_H 3.82 (3'-OCH₃) with δ_C 149.6 (C-3') and δ_H 3.68 (3''-OCH₃) with δ_C 154.0 (C-3''), demonstrated that the methoxy groups were located at C-3' and C-3'', respectively. Moreover, the NMR data the H-6' (δ_H 4.99) signal appeared abnormally upfield from other aromatic proton signals, which is the distinguished characteristic of diphenyl ether type cyclic diaryl-heptanoid. Combined with HR-ESI-MS data, it is confirmed that **1** is diphenyl ether type cyclic diaryl-heptanoid.²³ Furthermore, **1** showed positive optical rotation ($[\alpha]_D^{25} +22.7$). Since **1** has a chiral plane in the molecule.²⁴⁻²⁶ Thus, **1** was elucidated as shown in Fig. 1 and named (+)-cyclogingerenone A. Unfortunately, the absolute stereochemistry remained undetermined.

Compound **2** was separated as a yellow oil. The molecular formula of **2** was deduced as $C_{21}H_{22}O_5$ according to HR-ESI-MS ion peak at m/z 355.1553 $[M + H]^+$ (calcd for $C_{21}H_{23}O_5$, 355.1545). Comprehensive analysis of the NMR data (see Table 1) demonstrated that the structure of **2** was quite similar to **1**, and the difference between them was the connection sites of the benzenes and heptane moiety. The HMBC correlations from H₂-1 to C-3, 2'', 6'', H₂-7 to C-2', 6', 5, and H-4, H-5/C-3, 7, inferred that C-1 and C-7 were connected to C-1'' and C-1', respectively. In addition, **2** showed negative optical rotation ($[\alpha]_D^{25} -17.5$). Therefore, **2** was elucidated as a new diaryl ether heptanoid and named (-)-cyclogingerenone B.

The molecular formula of compound **3** was determined as $C_{21}H_{24}O_6$ by HR-ESI-MS (m/z 373.1654 $[M + H]^+$, calcd 373.1651). It was disclosed that the only structural difference between **3** and **2** was concerned the change of the heptane chain moiety. The NMR data of **3** revealed the presence of one more oxymethine [δ_H 3.34 (1H, br t, $J = 8.4$ Hz, H-5); δ_C 66.1 (C-5)] and one more methylene [δ_H 1.89 (1H, d, $J = 18.3$ Hz, H-4a), 2.22 (1H, dd, $J = 18.3, 10.5$ Hz, H-4b); δ_C 52.4 (C-4)], but one less *trans*-double bond by comparing with **2**. Consequently, the detailed analysis of the spectroscopic data allowed the planar structure of **3**.

The absolute configuration of C-5 in **3** was established by comparison of the calculated ECD spectra [(5*R*)-**3** and (5*S*)-**3**] with the experimental counterpart (see ESI†). The calculated ECD spectrum (Fig. S70 and 71†) (5*R*)-**3** showed good consistency with the experimental one. In addition, **3** showed positive optical rotation ($[\alpha]_D^{25} +2.7$). Thus, the structure of **3** was assigned as (+)-cyclogingerenone C.

Compound **4** was obtained as a yellow gum. The HR-ESI-MS of **4** showed a quasimolecular peak at m/z 403.2110 $[M + H]^+$ (calcd for $C_{23}H_{30}O_6$, 403.2121), corresponding to the molecular formula of $C_{23}H_{30}O_6$. Careful analysis of its 1H and ^{13}C NMR

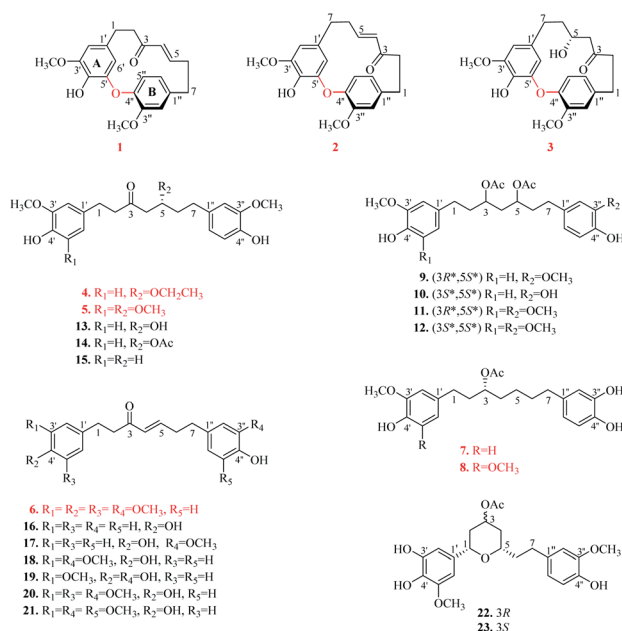
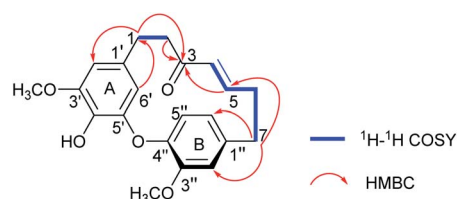


Fig. 1 Structures of compounds 1–23.

Fig. 2 Key 1H - 1H COSY and HMBC correlations of compound 1.

data (Table 2) indicated that **4** had a similar structure to **13**. The main difference found between them concerned the substituent group at C-5. A hydroxyl group at C-5 in **13** was changed to an ethoxy group in **4**. This deduction was supported by ^1H - ^1H COSY correlations of H_2 -8/ H_3 -9 and HMBC cross peaks from H-8 [δ_{H} 3.42 (2H, qd, $J = 7.1, 2.0$ Hz)] to C-5 (δ_{C} 76.3). The absolute configuration of C-5 of **4** was determined by comparing the optical rotation with analogues.^{27–30} The negative optical rotation ($[\alpha]_{\text{D}}^{25} -10.7$) of **4** indicated 5*R* configuration. On the basis of the above evidence, the structure of **4** was determined (Fig. 1) and named (5*R*)-5-ethoxyhexahydrocurcumin.

Compound **5** showed a molecular formula of $\text{C}_{23}\text{H}_{30}\text{O}_7$ by HR-ESI-MS (m/z 419.2060 [$\text{M} + \text{H}$]⁺, calcd 419.2070). A comparison of the spectral data (Table 2) of **5** and **4** revealed that an ethoxy group in **4** was changed to a methoxyl group in **5**, which was confirmed by HMBC correlations from H_3 -8 (δ_{H} 3.31, s) to C-5 (δ_{C} 76.8). Moreover, an additional methoxyl group was substituted at C-5' due to the equivalent aromatic protons [δ_{H} 6.39 (2H, s, H-2', 6')] and a symmetrical substituted aromatic ring (δ_{C} 132.3, 147.1 \times 2, 105.1 \times 2, 133.1). Similarly to **4**, according to the specific optical rotation of **5** as $[\alpha]_{\text{D}}^{25} -8.3$, the absolute configuration of C-5 is assigned to *R*. To sum up, **5** was identified as (5*R*)-5-methoxy-1-(4-hydroxy-3,5-dimethoxyphenyl)-7-(4-hydroxy-3-methoxyphenyl) heptan-3-one.

Compound **6** possessed a molecular formula of $\text{C}_{22}\text{H}_{26}\text{O}_6$ based on HR-ESI-MS (m/z 387.1794 [$\text{M} + \text{H}$]⁺, calcd 387.1808). The ^1H and ^{13}C NMR data (Table 2) of **6** was high similarity to those of **20**. The only difference between them was that a hydroxyl group in **20** was changed to a methoxyl group at C-4' in **6**, which was further deduced by HMBC correlations between 3.82 (3H, s, 4'-OCH₃) and C-4' (δ_{C} 136.3). Therefore, the structure of **6** was defined as (*E*)-7-(3,4-dihydroxyphenyl)-1-(3,4,5-trimethoxyphenyl) hept-4-en-3-one.

Compound **7** had the molecular formula of $\text{C}_{22}\text{H}_{28}\text{O}_6$ inferred from the HR-ESI-MS at m/z 411.1767 ($[\text{M} + \text{Na}]^+$, calcd 411.1784), corresponding to an index of hydrogen deficiency of 9. Its NMR data (Table 3) possessed similar signals to those of **10** except for the loss of one acetyl group located at C-5. The acetyl group was substituted at C-3, which was evidenced by HMBC correlations from H-9 (δ_{H} 2.03) to C-8 (δ_{C} 171.4), and H-3 (δ_{H} 4.90) to C-8 (δ_{C} 171.4). According to Brewster rule of secondary carbinol,^{31–33} **7** showed a negative optical rotation value of 10.5, which indicated that the acetyl moiety at C-3 is predicted to be *R* configuration. Therefore, **7** was determined and named as (3*R*)-3-acetoxy-7-(3,4-dihydroxyphenyl)-1-(4-hydroxy-3-methoxyphenyl) heptane.

The molecular formula of compound **8** was determined as $\text{C}_{23}\text{H}_{30}\text{O}_7$ by HR-ESI-MS (m/z 441.1895 [$\text{M} + \text{Na}]^+$, calcd 441.1889). Extensive comparison of the NMR data of **8** and **7** (Table 3) suggested that an additional methoxyl group was

Table 2 ^1H and ^{13}C NMR data of compounds 4–6

Pos.	4 ^{a,c}		5 ^{b,d}		6 ^{b,d}	
	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)
1	30.3	2.74	29.9	2.82, t (7.4)	30.8	2.86, t (7.0)
2	46.4	2.74	46.0	2.73	41.7	2.81, t (7.0)
3	211.7		208.9		200.1	
4	48.7	2.67, dd (15.8, 7.1) 2.55, dd (15.8, 7.1)	47.6	2.45, dd (15.9, 5.3) 2.71, dd (15.9, 7.1)	131.1	6.07, d (15.8)
5	76.3	3.72, dt (12.1, 6.1)	76.8	3.69	147.0	6.77, dd (15.8, 6.9)
6	37.5	1.71	36.2	1.76	34.3	2.47
7	32.1	2.54	31.2	2.55, ddd (15.6, 9.8, 6.6) 2.63, ddd (15.6, 9.8, 5.8)	33.8	2.64, t (7.1)
8	65.6	3.42, qd (7.1, 2.0)	57.2	3.31, s	—	—
9	15.8	1.10, t (7.1)	—	—	—	—
1'	134.0		132.3		137.3	
2'	113.1	6.75, d (1.5)	105.1	6.39, s	105.6	6.39, s
3'	148.9		147.1		153.3	
4'	145.6		133.1		136.3	
5'	116.2	6.68, d (8.0)	147.1		153.3	
6'	121.8	6.58, dd (8.0, 1.5)	105.1	6.39, s	105.6	6.39, s
1''	134.8		133.9		133.6	
2''	113.2	6.73, d (1.6)	111.0	6.67, br s	115.5	6.63, br s
3''	148.9		146.5		143.9	
4''	145.8		143.9		142.1	
5''	116.2	6.70, d (8.0)	114.4	6.82, d (7.9)	115.5	6.75, d (6.9)
6''	121.8	6.61, dd (8.0, 1.6)	121.0	6.65, br d (7.9)	120.7	6.55, br d (6.9)
3'-OCH₃	56.4	3.80, s	56.4	3.86, s	56.3	3.83, s
4'-OCH₃	—	—	—	—	61.1	3.82, s
5'-OCH₃	—	—	56.4	3.86, s	56.3	3.83, s
3''-OCH₃	56.4	3.82, s	56.0	3.87, s	—	—

^a Measured in CD₃OD. ^b Measured in CDCl₃. ^c 400 MHz for ^1H , 100 MHz for ^{13}C . ^d 600 MHz for ^1H , 150 MHz for ^{13}C .



Table 4 Cytotoxicity activities of compounds 6, and 16–19

Compound	Cytotoxic activity IC ₅₀ /μmol L ⁻¹				
	A549	HepG2	HeLa	MDA-MB-231	HCT116
6	17.49 ± 0.09	33.46 ± 0.50	30.46 ± 1.20	17.83 ± 0.11	14.70 ± 0.63
16	13.74 ± 0.48	19.67 ± 0.09	11.30 ± 0.36	10.78 ± 0.12	13.88 ± 0.18
17	11.49 ± 0.11	15.98 ± 0.14	10.41 ± 0.43	9.18 ± 0.18	10.61 ± 0.21
18	8.63 ± 0.42	13.06 ± 0.13	11.56 ± 0.21	10.51 ± 0.14	6.69 ± 0.09
19	17.27 ± 0.21	20.93 ± 0.41	12.20 ± 0.16	13.60 ± 0.05	11.64 ± 0.18
Curcumin ^a	21.86 ± 0.25	27.37 ± 0.30	19.71 ± 0.22	19.03 ± 0.11	14.88 ± 0.14

^a Paclitaxel was used as a control.

more superior cytotoxicity against HCT116 cell lines with IC₅₀ value of 6.69 ± 0.09 μM, compared to the positive control (curcumin, IC₅₀: 14.88 ± 0.14 μM).

The cytotoxic activity of compound **13** was nearly identical to compounds **4**, **14** and **15**, which suggested the cytotoxic activity of diarylheptanoids might be less dependent on hydroxy groups of heptane chain. The results of MTT assay for compounds **6**, **16**–**19** indicated that the substituted phenyl ring might not be a critical structural determinant for the cytotoxic activities.⁴³ The superior cytotoxic effect of **18** compared with **13** might be attributed to the presence of an α, β-unsaturated carbonyl moiety in **18**, as reported earlier.^{44,45} Previously, it had been suggested that the α, β-unsaturated carbonyl compounds-mediated toxicity was attributable to formation of Michael-type adducts with the nucleophilic sulfhydryl groups of protein thiols.^{46,47} The above results were not sufficient to further clarify the structure–activity relationship between the diarylheptanoid derivatives and/or other components. More research may be required to clarify their potential selective cytotoxic activity.

3.4. Effects of diarylheptanoids on the ATR/CHK1 signalling pathway

Eukaryotic cells have such a cumbersome DNA damage response (DDR) to protect the integrity of the genome, which contains two main signal cascades, namely the ATM/CHK2 and ATR/CHK1 cascades. As the core protein and kinase of DDR, ATR can activate downstream CHK1 to form ATR/CHK1 signalling pathway, so that bodies have enough time to repair the wrong DNA before turning to the next stage or going to apoptosis. If the damage of the cell cannot be repaired, the DDR will lead the damaged cell to apoptosis.⁴⁸

If the damage of the cell cannot be repaired, the DDR will lead the damaged cell to apoptosis.⁴⁸ To understand the anti-tumour mechanism of diarylheptanoids in ginger, the effect of compounds **6**, **17**, and **18** (which showed superior suppressed cell proliferation activity) on the ATR/CHK1 signalling pathway was investigated. As shown in Fig. 5, compounds **6**, **17**, and **18** significantly suppressed the total protein levels of ATR and CHK1 in HCT116 cells. Our results indicated that diarylheptanoids had potential anti-tumour effects, and they might act through the regulation of the ATR/CHK1 signalling pathway.

4. Conclusions

In summary, 23 diarylheptanoids including eight undescribed ones (**1**–**8**), were isolated and identified from an important natural dietary ingredient ginger. It is worth mentioning that **1**–**3** are the first reported structures of diarylether heptanoids in *Z. officinale*, whereas **15**–**17** were isolated from *Zingiber* for the first time. Network pharmacology demonstrated that diarylheptanoids were mainly associated with the antitumor activity by affecting DNA damage signalling pathway. Cytotoxic and enzymatic activity assays discovered that compounds **6**, **17**, and **18** may exert anti-cancer effects by regulating the ATR/CHK1 signalling pathway.

Conflicts of interest

The authors declare no conflict of interest.

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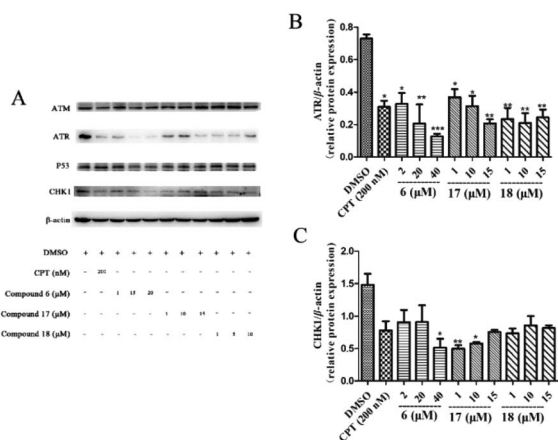


Fig. 5 Effects of compounds **6**, **17**, and **18** on the protein expressions of ATM, ATR, P53, and CHK1 in HCT116 cell line. HCT116 cells were pre-treated with different concentrations of **6**, **17**, and **18** for 24 h. The cells were lysed with RIPA buffer and the protein levels for total ATM, ATR, P53, and CHK1 were measured by using immunoblot analysis. β-Actin was used as a loading control. CPT was used as a positive control. And all of the experiments have been repeated three times independently. Data presented as mean ± SD, $n = 3$. * $p < 0.05$, ** $p < 0.01$, and *** $p < 0.001$ as compared with the DMSO group.



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