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Representative 2D-material-based nanocomposites and their emerging applications: a review

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Composites (or complex materials) are formed from two or many constituent materials with novel physical or chemical characteristics when integrated. The individual components can be combined to create a unique composite material through mechanical transfer, physical stacking, exfoliation, derivative chemical mixtures, mixtures of solid solutions, or complex synthesis processes. The development of new composites based on emerging 2D nanomaterials has allowed for outstanding achievements with novel applications that were previously unknown. These new composite materials show massive potential in emerging applications due to their exceptional properties, such as being strong, light, cheap, and highly photodegradable, and their ability to be used for water splitting and energy storage compared to traditional materials. The blend of existing polymers and 2D materials with their nanocomposites has proven to be immediate solutions to energy and food scarcity in the world. Although much literature has been reported in the said context, we tried to provide an understanding about the relationship of their mechanisms and scope for future application in a comprehensive way. In this review, we briefly summarize the basic characteristics, novel physical and chemical behaviors, and new applications in the industry of the emerging 2D-material-based composites.

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1. Introduction

A composite material is formed by mixing at least two materials, often with different properties, resulting in unique output characteristics compared to the single material, *i.e.*, a high strength-to-weight ratio, low density, high modulus, excellent fatigue resistance, improved creep rupture strength, superior corrosion

prevention, and low coefficient of thermal expansion.¹ The development and advancement of next-generation composite materials have been widely observed in the aerospace industry and the mechanical industry, construction industry, biomedical industry, automobile industry, optoelectronic industry, and electrical and electronic packaging industry.² An additional benefit of composite materials is customization because they can transform into complex shapes.

Two-dimensional (2D) materials are composed of multiple thin layers. These layers are weakly coupled through van der Waals interactions. A single-atom-thick layer usually has a thickness of few nanometres. Electrons in these layers are free to move in the 2D plane, but quantum mechanics governs their restricted motion in the third direction. These materials can be further classified as the zero-dimensional (0D), *e.g.*, quantum dots; one-dimensional (1D), *e.g.*, nanoribbons, nanotubes, and nanowires; two-dimensional (2D) monolayer materials, *e.g.*, single-atom graphene sheets; and three-dimensional (3D), *e.g.*, nanoballs and nanocones. Materials such as graphene, transition metal dichalcogenides (TMDs), quantum dots, transition metal oxides (TMOs), graphitic C₃N₄, 2D clay materials, hexagonal boron nitride (hBN), black phosphorus (BP), silicene, and germanene are made up entirely of their surface and are often only a single atom thick, falling in the 2D material category. A composite made of these 2D materials is called a 2D material composite. The typical traditional materials and composites, *i.e.*, metals, wood, reinforced plastics, fiber-reinforced, fiberglass, Kevlar, carbon fiber, and polymers,

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have been overcome by these new 2D materials. These new 2D materials have been emerged as potential candidates for next-generation devices due to their exceptional mechanical strength, flexibility, and rigidity, as well as their enhanced electrical and optical properties.³⁻²³ Notably, recent advances in growth techniques enable high-uniformity and large-area 2D materials, which lower the production cost and enhance the quality of 2D-based devices.^{24,25} The engineering of 2D composited materials needs systematic and interactive strategies to achieve the optimum material properties. This process demands the application of several methods and technologies for the (i) investigation of the specific mechanical, electrical, optical, chemical, and physical properties of composited materials; (ii) analysis of the design, manufacturing, and composition influences on the material characteristics; (iii) optimization of the material characteristics according to the appropriate working condition and application, *e.g.*, prediction of material efficiency/reliability; and (iv) development of quantitative characterization approaches and analysis in the sense of various working conditions.²⁶

Composites are gaining a considerable amount of attention due to their outstanding properties. As seen in Fig. 1, since the last decade of the 21st century, the interest of the 2D material research community in composite materials, illustrated in annual publications, has remarkably increased. In relation to semiconductors, group transition metal dichalcogenides, *e.g.*, MX₂, have been intensively studied, most likely due to their quantum confinement effect-induced variation in electric, optical, and chemical properties.^{3,4,24,25} For example, the chemical reactivity of MX₂ is site-dependent; *i.e.*, the edge of TMDs with abundant active sites is more chemically reactive than the basal plane, thus effectively producing more hydrogen.²⁵ Moreover, the high surface-to-volume ratio of 2D material is also a primary advantage when it is composited. For instance, the application of carbon-encapsulated Fe₃O₄ nanoparticles (NPs) as a high-rate lithium-ion battery anode material is known to be beneficial for next-generation batteries.²⁷

The 2D-material-based composites exhibited various boosted properties, such as improved mechanical strength and flexibility, enhanced electrical conductivity, high optical transparency, high

resistance to chemicals and fire, high strength-to-weight ratios, wear resistance, and corrosion resistance.^{4,27-31} Herein, we present a short review of 2D-material-based composites and their characterizations, classifications, and advantages regarding physics and mechanics related to recent studies. While the novelty of nanocomposites in previous reports has lied on only several proper aspects, either 2D materials or single applications that forming 2D composites such as biocomposites, fiber-reinforced composites and hybrid composites, gas sensing,²⁸ battery,^{32,33} only graphene,³⁴⁻³⁷ carbon nanotubes (CNTs) and MXenes for energy storage,³⁸ strain sensor,^{39,40} mechanics,⁴¹ metal,⁴² protein,⁴³ dentistry,⁴⁴ orthopaedic and bone,⁴⁵ polymer,^{46,47} humid sensor,⁴⁸ or biology;⁴⁹ in this review article, we would like to emphasize in other aspects that formed new nanocomposites based on polymers and metals as well as the latest achievements of new synthesized-2D materials enabling to fabricate new nanocomposite species. In addition, the representative applications on strain engineering, thermal characters, photocatalyst and water splitting, optoelectronics, solar energy, and other promising applications (*e.g.*, degradation of organic pollutants, enhanced strength and stiffness) are pointed out. In particular, the representative fabrication techniques and applications on the latest achievements of new 2D materials are addressed. Science and modern technology have been always looking for possible solutions for the ever-growing need for energy and fuel resources. The industrial revolution has brought a big challenge for technologists to provide new opportunities for energy production, harvesting, and conservation. As we discussed previously, there is a significant role of novel materials-based composites for solving major challenges faced by humanity. We will present a brief overview of the contribution of nanocomposites specifically for energy-related applications, for example, photocatalysis, water splitting, and solar energy. Today, it is necessary to discover new nanocomposite assemblies that could integrate artificial and natural materials by implementing modern manufacturing processes. The classification of 2D composites, including carbon; graphene; polymers; and ceramic, metal, and bio-based materials, is shown in Fig. 2.

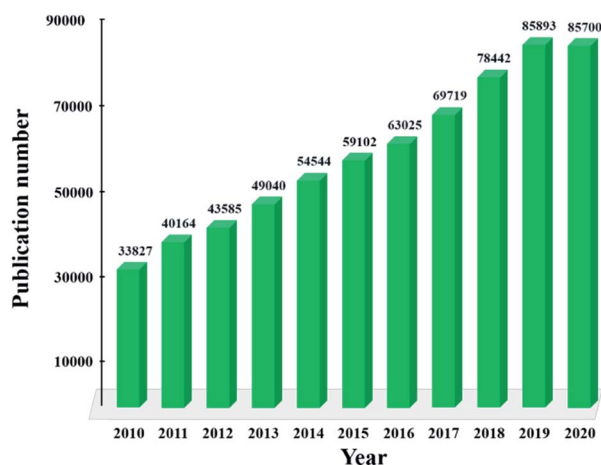


Fig. 1 Publications on 2D-material-based composites from 2010 to 2020. Source: ISI Web of Science (search: "2D Composites").

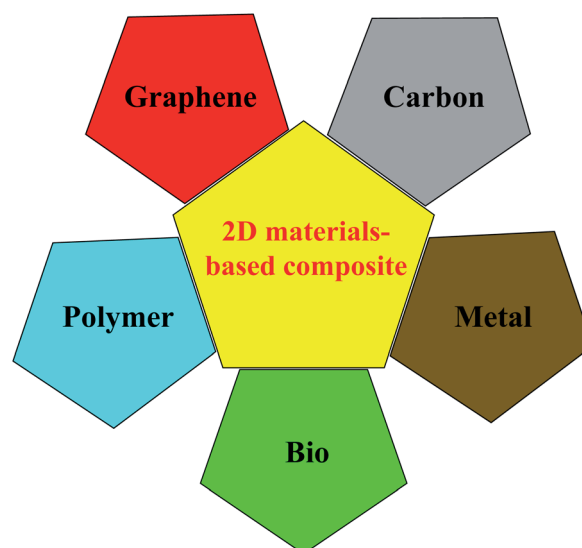


Fig. 2 Classification of 2D-Materials Based composites.



2. Composite classifications

2.1 Polymer-based composites

Polymer composite is composed of more than two phases, such as polymer/polymer or filler (metal or fiber)-based polymer composite. Compared to traditional materials such as ceramic and clay, polymer composites were developed later. They have been widely used in the electronics, food packaging industry, and medical industry due to their low cost, low density, corrosion resistance, easy design, and processing. Traditional polymer fillers such as metals and minerals are in macroscale, which have been widely used in various applications. However, these additives cannot remarkably improve the properties due to the limited interfacial structure between them and the matrix. To further enhance the performance of the composites, reducing the size of fillers into microscale and nanoscale the primary strategy. Carbon being a nano-sized material, have the novel ability to form a covalent bond with other atoms, and due to this ability, various carbon species usually include polymer composite. Graphene is a 2D material that also belongs to the carbon family, dominates the entire 2D family, and has been studied a lot for various aspects since its discovery in 2004. Thin-film graphene can be synthesized by multiple means, *e.g.*, chemical vapor deposition (CVD), chemical or plasma exfoliation from natural graphite, and mechanical cleavage using scotch tape.^{16,50} However, for graphitic oxide, Hummers *et al.* developed a rapid, relatively safe method by treating graphite with essentially a water-free mixture of concentrated sulfuric acid, sodium nitrate, and potassium permanganate graphite.⁵¹ Fig. 3 summarizes some methods, products, and applications of graphene composites.

Liu *et al.* synthesized a series of 0D, 1D, and 2D WS₂ mixed with graphene oxide (GO) and reduced graphene oxide (RGO) nanosheets and then employed vacuum filtration to form WS₂/GO and WS₂/RGO composites,³² as shown in Fig. 4A and B. The WS₂/RGO composite exhibits excellent cyclic stability (100 cycles) and a high-rate capability 697.7 mA h g⁻¹. This composite structure not only revealed higher ability than pure WS₂ but also showed high reversible capacity. On the other hand, Chen *et al.* prepared a composite by integrating WS₂ (few-layer sheets) and nitrogen-doped graphene (WS₂/NG) *via* facile surfactant-assisted synthesis under hydrothermal conditions high transportation of electrons. The performance as a high energy density lithium-ion battery anode was evaluated (from 100 to 5000 mA g⁻¹). They found that the performance of the composite was dependent on the number of WS₂ layers. The best cyclability (average of only 0.08% capacity fade per cycle for 100 cycles) was achieved when the composite formed with a surfactant : tungsten precursor ratio of 1 : 1.³³ A schematic illustration is shown in Fig. 4C. Multifunctional graphene nanocomposites exhibit more substantial enhancement at considerably low loading than conventional composites.³⁴ This leads to potential multifunctional applications.

Nanocomposites are used to combine different nanoscale materials to achieve optimal material properties. Nanocarbon-based composite materials showed enhanced electrochemical, bacterial detection, gas sensing, chemical sensing, strain/pressure sensing, energy storage, and organic electrolyte properties, which are advantageous for various applications.^{28,38,39,52-57} Conductive graphene composites are made by adding graphene to matrix materials, such as polymers and

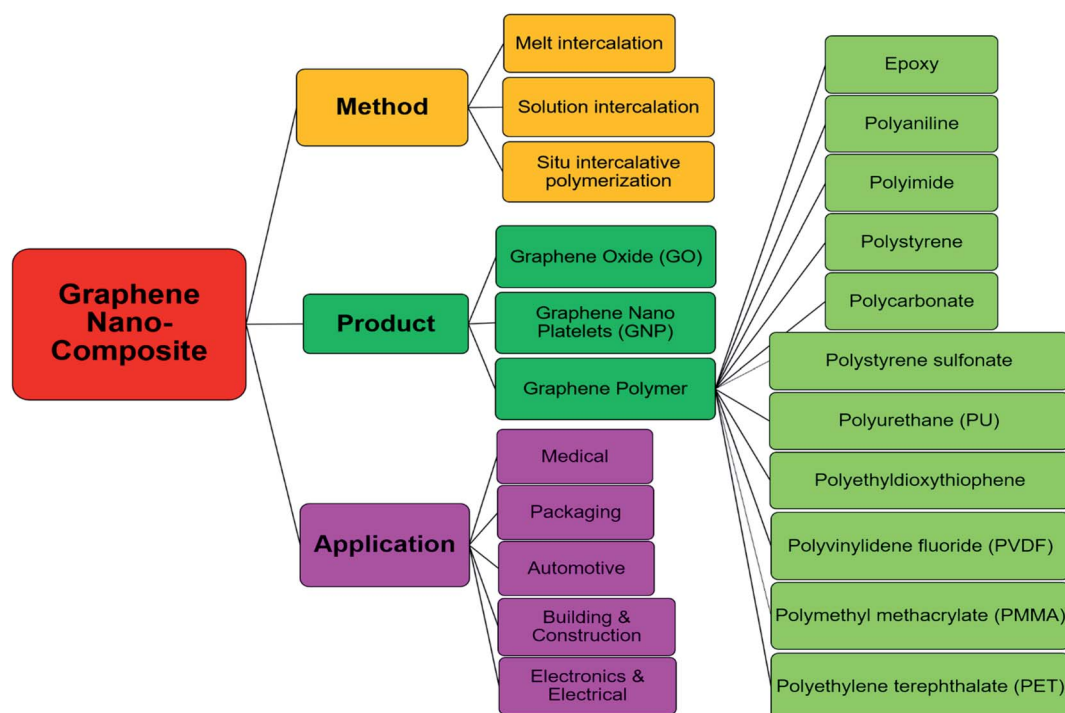


Fig. 3 Carbon-based nanocomposites possibilities.



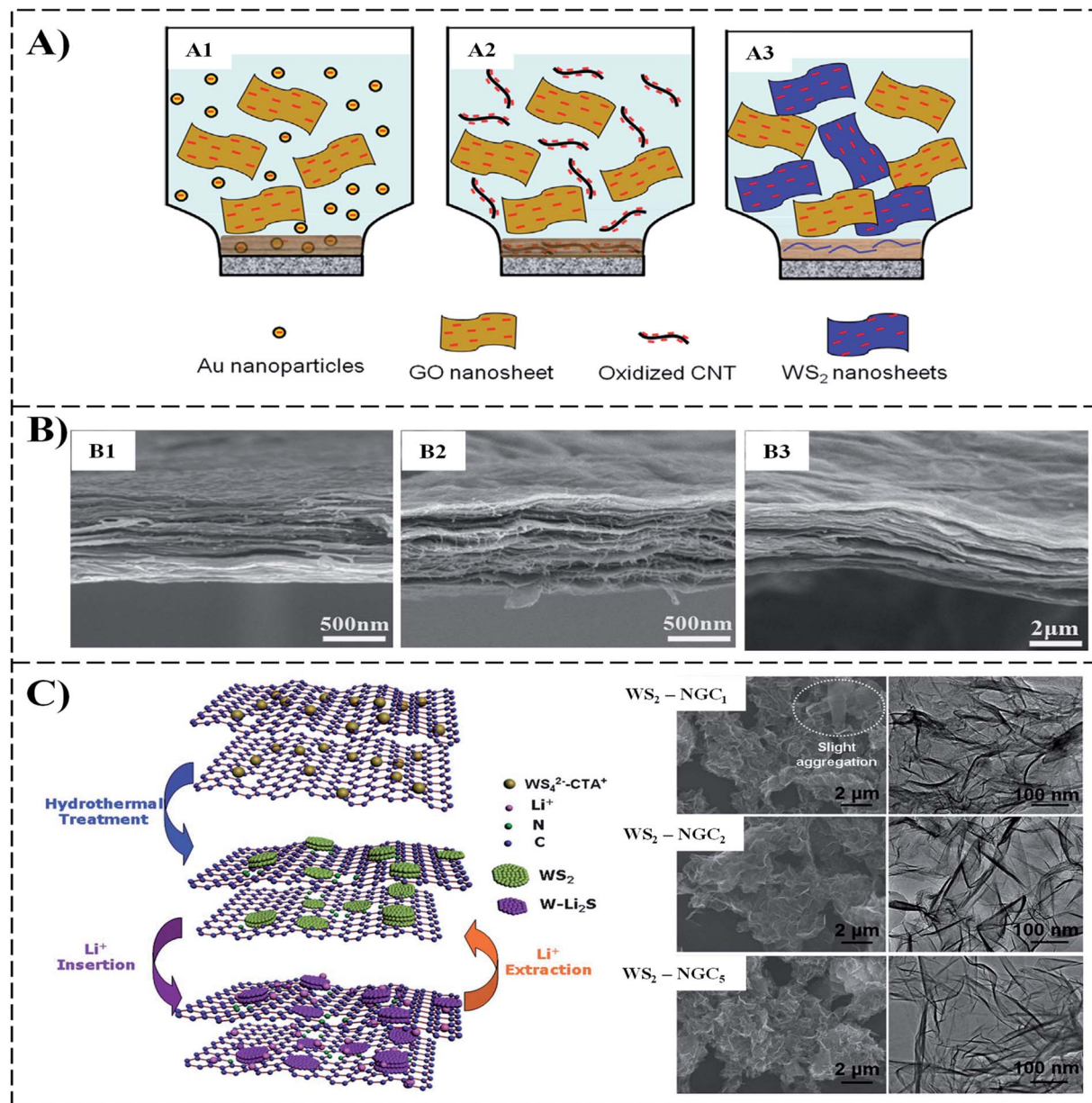


Fig. 4 (A) Schematic illustration formation negatively like-charged, (A1) Au/GO, (A2) CNT/GO, (A3) WS₂/GO lamellar composite films by direct filtration. (B) Cross-section SEM images of, (B1) Au/GO, (B2) CNT/GO and (B3) WS₂/rGO WG-2-1 films.³² (C) schematic illustration, SEM and TEM of the preparation of WS₂-NG composites.³³

metal, using a specific method to produce a conductive composite. Their dependence on the conductive filler characterizes the electrical conductivity of composites. The high content ratio of conductive fillers enhances the transport of free electrons inside the material, which leads to an increase in electrical conductivity. This also improves the physicochemical qualities of the host matrix/material. Some experimental results show that the host matrix can enhance its electrical and mechanical properties better than other composites, e.g., carbon nanotubes (CNTs) or clay.⁵⁸ Consequently, graphene-based composites reveal various advantageous properties, such as high electrical and thermal conductivity, enhanced

mechanical properties, and improved flame retardant and gas barrier properties.^{35,36,59,60}

One possible way to bond these novel properties for applications is to integrate graphene sheets into a composite material. The production of such composites requires graphene sheets to be unified and equally scattered over a sufficient scale into various matrices. Stankovich *et al.* reported a bottom-up chemical method to tune graphene sheet properties *via* a solution mixture of exfoliated (chemical reduction) phenyl isocyanate-treated graphite oxide sheets with polystyrene (PS)³⁷ (see Fig. 5). This PS-graphene composite showed a percolation threshold of ~ 0.1 vol% and conductivity of ~ 0.1 S m⁻¹ at room temperature, which was sufficient for various electrical



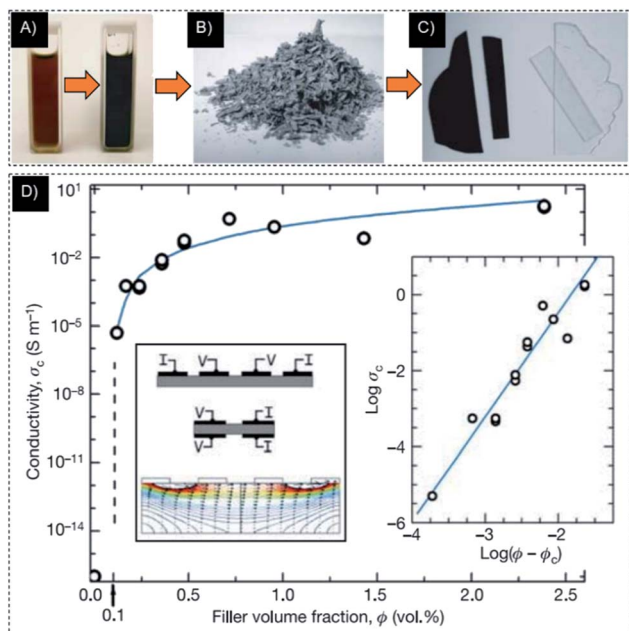


Fig. 5 (A) Suspension of phenylisocyanate-treated GO and dissolved polystyrene in DMF before/after a reduction through *N,N*-dimethylhydrazine. (B) The fabricated composite powder *via* coagulation in methanol. (C) Hot-pressed graphene-based composite and pure polystyrene processed in the same way. (D) The conductivity of composite as a function of filler volume fraction. Right inset, $\log \sigma_c$ plotted against $\log(\phi - \phi_c)$.³⁷

applications. Xie *et al.* prepared a graphene-based conductive polymer composite *via* the mixture of a solution containing GO and a PVA precursor. The RGO was fabricated by employing the γ -ray irradiation of GO through the water/ethanol mixture. The composite films revealed good electrical conductivity and visible transmittance.

Xuejiao *et al.*⁶¹ obtained novel flexible and conductive polylactic acid–cellulose nanofibril–graphene/polypyrrole (PLA–CNF–GR/PPy) composite films by adding cellulose nanofibril (CNF) and graphene powder into polylactic acid (PLA)/polypyrrole (PPy) composite through inexpensive, eco-friendly, low-temperature, and polymerization synthesis. The proposed conductive film exhibited many unique properties, *e.g.*, high enhanced tensile strength (5.6%), thermal stability, and high conductivity. In addition, the electrical conductivity of the proposed composite film was improved from 0.12 to 1.06 S cm⁻¹ and demonstrated high flexible stability, with only 7.5% deviation after \sim 100 bendable cycles. The novel conductive composite represents a new field of potential applications of biodegradable materials and could be considered for use in sensors, flexible electrodes, and flexible displays in future electronics.

Graphene-based devices and composites are subjected to the doping of graphene to enhance the material's electrical conductivity. In this regard, Fan *et al.*⁶² proposed highly conductive few-layer graphene (FLG)-based composites, exhibiting vastly enhanced electrical conductivity from 1.4×10^3 to 2.1×10^3 S m⁻¹ by adding 4.42–5.1 vol% FLG. The dopant type

and level could be manipulated by incorporating FLG as matrixes. Here, three metal oxides with various oxygen vacancies (α -Al₂O₃, 3% mol yttria-stabilized zirconia, and 8% mol yttria-stabilized zirconia) were employed as matrix materials to form this nanocomposite.

For electrochemical energy storage devices, enlightening the availability of ions in the electrodes is important for rate performance and charge storage. Gao *et al.*³⁸ synthesized Mxene@CNTs by using knotted CNTs to break 2D layers of Mxene Ti₃C₂, as shown in Fig. 6A. This structure can improve the capacitance (reaching 130 F g⁻¹ (276 F cm⁻³)) in organic electrolytes. Zhao *et al.*⁵³ successfully prepared the LiMn₂O₄-based composite microspheres using a spray-drying technique with nanosized LiMn₂O₄ particles as host materials and VXC-72R NPs as a dispersing agent embedded with nanocarbon black (VXC-72R) particles and exhibiting a high initial discharge capacity of 127.4 mA h g⁻¹ with good cycling stability at 0.5 °C. Recently, nanocarbons, including nanosheets (CNShs), carbon nanotubes (CNTs), and nanoboxes (CNB_xs), have attracted the attention of composite researchers. Xu *et al.* synthesized carbon-coated troilite FeS (C@FeS) of different morphologies (NPs, nanosheets, and nanoplates) with the surfactant-assisted solution-based method,⁶⁴ as shown in Fig. 6B. The concentration of 1-dodecanethiol and temperate treatment are used to control the morphologies that form NPs, nanosheets, and nanoplates with a low concentration at 180 °C, a high

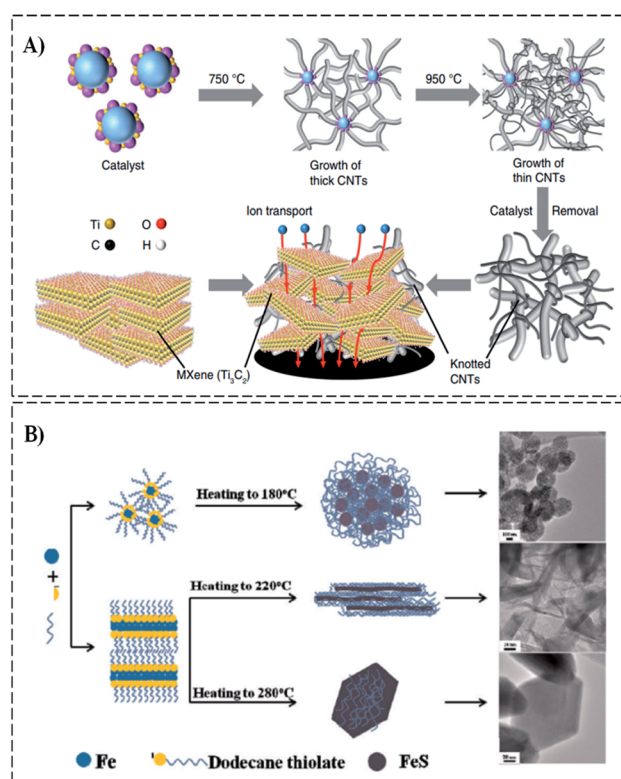


Fig. 6 (A) Schematic growth of CNTs and the formation of MXene-knotted CNT composite electrode.³⁸ (B) Schematic growth mechanism of C@FeS composites with the carbon layer and FeS nanoparticles (NPs), nanosheets, and nanoplates.⁶³



concentration at 220 °C, and a higher concentration at 280 °C, respectively. The C₂FeS nanosheet exhibited exceptional Li storage properties with good stability (charge/discharge cyclability) and a high specific capacity that reaches 233 mA h g⁻¹ during one hundred cycles at fast charge/discharge rates. General polymers usually synthesize thermo- or thermosetting-plastics with metal fiber, carbon, glass, and so on.^{30,31,65,66} Thermoset composites are manufactured by mixing hard resin to achieve higher strength and higher temperature resistance than thermoplastics.^{41,67,68} Zhi *et al.*⁶⁹ fabricated polymethyl methacrylate/boron nitride nanosheet (PMMA/BNNS) transparent composites. As a result, the elastic modulus and strength of PMMA were improved to 22% and 11% with 0.3 wt% BNNS fractions, respectively.

Recently, some conductive polymers, such as poly(3-hexylthiophene-2,5-diyl) (P3HT), phenyl-C₆₁-butyric-acid-methyl ester (PCBM), polythiophene (PTH), and polyaniline (PANI), have been studied for application in heterojunction devices. Cho *et al.*⁷⁰ successfully synthesized edge-selectively functionalized graphene/polyamide 6 (EFG/PA6) composites. The electrical conductivity of EFG/PA6 was strongly improved compared with initial graphite, ball-milled graphene, and

nanoplatelet graphene. Liu *et al.*⁷¹ synthesized mesoporous conductive polymer using mixture block copolymer micelles and 2D electrochemically exfoliated graphene, GO, MoS₂, and titania nanosheets. Fig. 7 illustrates the schematic of the polymer-based composite using a single GO layer. These conductive polymers exhibit improved electrochemical capacitance and rate performance.

2.2 Metal-based composites

The matrix material uses metallic nanomaterials such as Mg, Cu, Ag, Al, Ni, and Ti to form a 2D metal composite.⁷² Al-based composites are popularly applied in the automobile and aerospace industries due to their outstanding properties, such as low density and thermal expansion, superior stiffness strength, and improved wear resistance.^{42,73,74} In addition, Ag-based composites are the most widely used in electric devices due to their high electric conductivity.⁷² Deng *et al.*⁷⁵ obtained a sandwich-stacked SnO₂/Cu hybrid nanosheet by rolling and compressing a SnO₂/Cu layer with carbon black. The stacked SnO₂/Cu nanosheet (Fig. 8) reveals the tight clamp between the layers. Pan *et al.*⁷⁶ prepared a composite of Ag particles and MoS₂ using

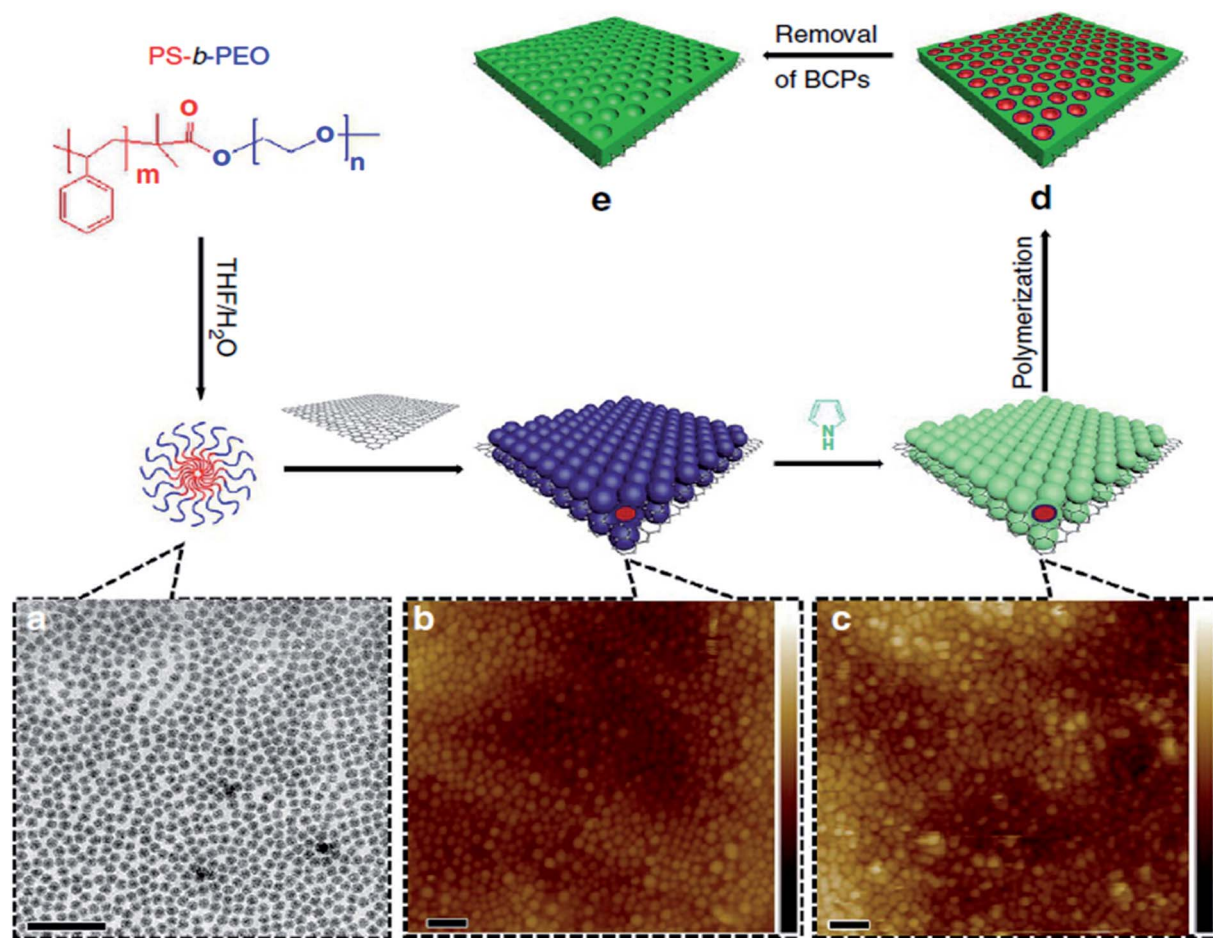


Fig. 7 Schematic of fabricating of 2D mesoporous conductive polymers. (a) TEM image and formation process of spherical BCP. (b) AFM image and the self-assembled BCP micelles on GO surface. (c) AFM image and co-assembly of BCP micelles and pyrrole monomers on GO surface. (d) Polymerization of pyrrole monomers on the addition of ammonium persulfate initiator. (e) Mesoporous polypyrrole (PPy) nanosheets.⁷¹



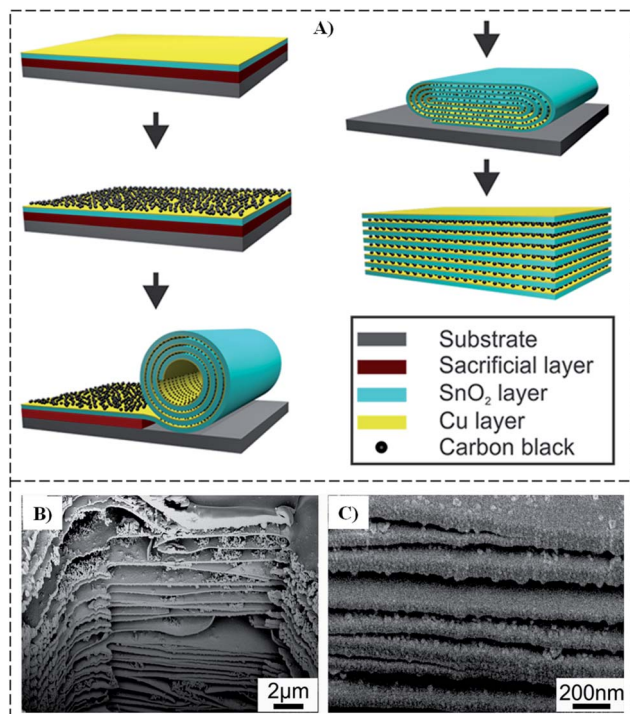


Fig. 8 (A) Schematic illustration and (B and C) cross-sectional SEM images of SnO₂/Cu hybrid nanosheets.⁷⁵

the sonication method and a multifunction organic ligand. The MoS₂ sheets were obtained by exfoliating MoS₂ powder under sonication. SEM and TEM image results show that the Ag particles were anchored and uniformly distributed on the surface of MoS₂ nanosheets.

3. Bio-composites

Novel composites from the equivalent combination of 2D materials and biomaterials in biotechnology have attracted

considerable interest because they can possess the desired properties and optimized performance.⁴³ In particular, biomedical nanocomposites have a potential application in medicine, *e.g.*, dental treatment,^{44,77} bone tissue engineering,⁴⁵ drug delivery,⁴⁶ and wound dressings.^{45,46} The precise drug delivery to sites of tumors can improve treatment and reduce injury by minimizing adverse effects.⁷⁸ Moreover, recent studies demonstrate that biocomposite materials exhibit thermal stability and enhanced tensile, tearing strength, and durability.^{49,79–82} Fig. 9A shows some 2D materials for biocomposite applications.⁸³ In addition, the demand for biodegradable, environmentally friendly materials in areas such as bio-packaging has driven researchers to develop biocomposites.^{84–88} Cheng *et al.*⁸⁹ investigated the interaction between graphene and several peptide structures of silk fibroin. The results reveal that graphene can affect the structure of all four types of silks, namely, pure amorphous, pure crystalline, a segment from the N-terminal, a combined amorphous and crystalline segment (Fig. 9B).

The advancements in materials composition always come with some merits and demerits. Firstly, discussing the merits of the composites. The nanocomposites outperformed excellent strength-weight and stiffness-weight ratios, usually called specific strength and specific modulus characteristics. We come up with a reduced production cost as compared to conventional materials with tunable mechanical characteristics. Composite materials exhibit excellent resistance to chemicals attack, corrosion, and outdoor weathering when used for aviation and paint industries. Some general advantages of composites include their light weightiness, improved fatigue life, reduced assembly designs, and repeatable features for the high-tech industry. There are some inherited demerits for the composite materials includes susceptibility to impact damage, difficulty in reconfiguring constituents, poor strength in the out-of-plane direction, and provision of an extra safe environment to ensure extended life span.

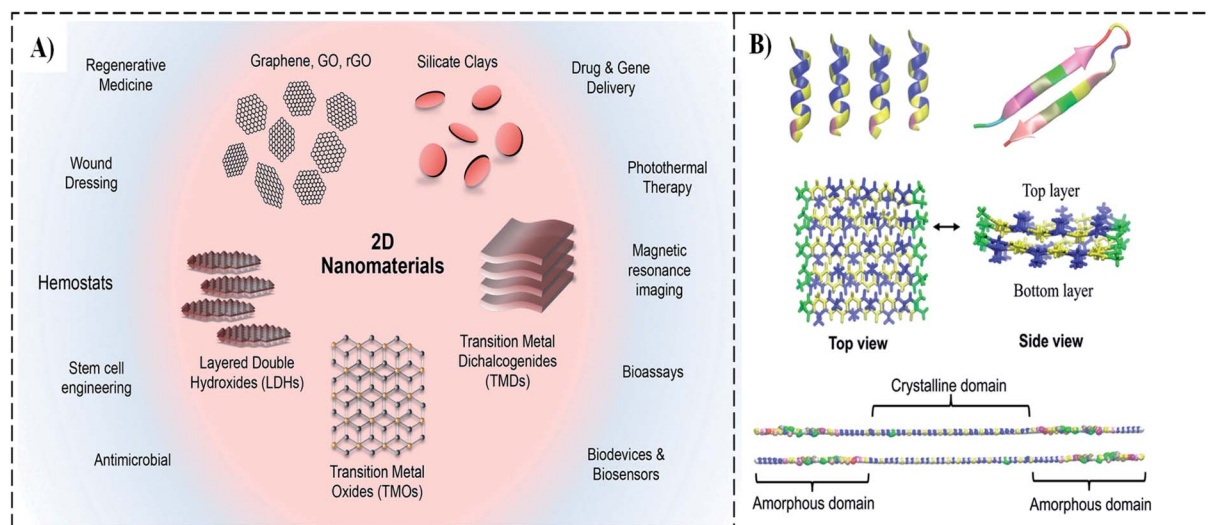


Fig. 9 (A) The 2D nanomaterials for biological applications.⁸³ (B) Simulations of peptide structure without graphene in water, including pure amorphous, pure crystalline, a segment from N-terminal, integrated amorphous, and crystal segment.⁸⁹



4. Applications of 2D materials and their composites

Composite products are appropriate for applications where high-performance is essential, such as race cars, aerospace, sporting goods, boats, and industrial applications. For example, because of its durability, lightweight, heat resistance, and supreme strength, carbon fiber has been used for decades for world-changing machines like high-speed sports cars, formula 1 race cars, aircraft engines, and aerospace equipment. Another example of the most widely used composite material is fiberglass. This composite has applications in utility poles, recreational and industrial equipment. Last but not least, fiberglass strut also has applications in external environments where a structure is subject to chemical exposure. These include water treatment plants, outdoor walkways, decks, and utility poles. Some other ideal applications of the composite are constructions, infrastructure, large complex-shaped structures such as retaining walls and floating roofs.

On the other hand, low-dimensional (2D) materials and their composites have shown exceptional chemical, optical, electrical, and physical properties since their emergence. For example, conventional fuel sources are non-renewable and environmentally unfriendly. Therefore, the demand for the storage of these new converted renewable energies, such as wind energy and solar energy, into electrical power, is high nowadays. One promising solution is to store these energies into an electrochemical device. The low-dimensional (2D) materials either used as heterostructures or composites have been widely used in electrochemical devices and significantly extended the performance limits for two main reasons: first, more opportunities for ion binding to active sites due to large specific surface area; secondly, 2D materials restacking enhance interlayer spacing, allowing more room for insertion and extraction of ions.⁹⁰ Materials used in conventional electronic devices are mainly categorized as semiconductors, insulators, and conductors. As a counterpart, 2D materials also demonstrated exemplary performance in the classification mentioned above for the device-level operation. However, competing with the silicon (Si) industry for scalability, uniformity, and repeatability is still challenging, but, notably, stacking structures of different 2D materials leads to outstanding performance enhancement of electronic devices. In this section, we briefly review the representative properties of 2D materials and their nanocomposites for potential applications and challenges in various fields of science.

4.1 Strain engineering for 2D composites

Strain in 2D composites is applied to modify electronic properties such as the bandgap and charge transport mechanism. Strain-modulated sensors fabricated by the nanoinclusion of CNTs in cement show great potential for futuristic smart city sensors in the construction industry. The stated sensor's resolution is significantly improved compared to that of the conventional strain gauge or accelerometers.⁴⁰ Graphene, having zero bandgap, when strained by about 15–30%, opens up

a bandgap of few hundreds of meV as per theoretical calculations.⁹¹ A novel nanocomposite graphene nanoplatelets (GNP) modifies the behavior of epoxy under different tensile stresses, which may contribute to the development of more durable epoxies.⁹² Various studies related to bandgap tuning by the application of strain suggest that more exploration is warranted. Table 1 illustrates physical properties of common 2D materials.

Along with graphene, TMD-based nanocomposites have attracted a considerable amount of attention for strain-modulated applications. Theoretical calculations suggest that the direct quasi-particle (QP) energy gap of MoS₂ reduces to ~0.8 eV with the application of 9% biaxial strain. The indirect QP can also be decreased by applying the strain on MoS₂, and the same result can be anticipated for other TMDCs. On the other hand, in single-layer black phosphorous, an increase of 0.65 eV was observed at an optical gap of 4.5% under the uniaxial strain applied in the ZZ direction.¹⁴⁴ This significant correlation between bandgap and strain allows for other 2D materials and their composites to perform well in emerging multidisciplinary technology.

Usually, uniaxial strain is applied through stretching or bending the substrate or producing ripples at the surface of the substrate and then transferring any 2D material. In Fig. 10, strain engineering for MoS₂ can be seen with uniaxial and biaxial strain. The photo luminescence (PL) test shows a reduction in intensity with the application of uniaxial strain in monolayer MoS₂. The A exciton PL peak in the bottom panel of Fig. 10A, related to the direct bandgap of MoS₂, shifted to relatively lower energies with the application of tensile uniaxial strain, pointing towards the closing of the bandgap. The uniaxial strain has been reported in other 2D semiconductors such as in WSe₂ and MoSe₂, which shows a very close phonon–excitation correlation. In Fig. 10A, a different approach is used to introduce strain in the material. The expansion and contraction are caused by a polypropylene substrate that carries 2D material flakes. *Via* the thermal development of piezoelectric substrate, biaxial strain can also be induced in MoS₂ flakes on such substrates or by creating a difference in pressure across the suspended MoS₂ membrane.

An exciton peak in the differential reflectance spectra shifts to lower energies under a biaxial tensile strain (~0.9%) and to higher energies under a compressive strain (~0.1%) not only in monolayers; strain engineering also influences interlayer coupling when applied to multilayered 2D stacking, as reported for a few layers of BP and tuning of the energy of interlayer excitons in bilayer MoS₂ straintronic devices can be fabricated based on the strain manipulation of 2D semiconducting materials and their composites.^{115–117} By controlling the biaxial strain level induced by thermal expansion, a single-layer MoS₂-based photodetector can be tuned. The photoresponsivity can be enhanced up to three orders of magnitude, with a response time of from 1.5 s to 80 ms.¹¹⁸ In-plane anisotropy of the MoS₂-based photodetector fabricated on PDMS (polydimethylsiloxane) also exhibited a uniaxial strain dependency. It showed a linear polarization anisotropic ratio, which could reach a value higher than two under 4.5% stretching of the



Table 1 Physical properties of common 2D materials

Material	Number of layers	Young's modulus (GPa)	Pretension (mN m^{-1})	Breaking stress (GPa)	Breaking strain (%)	Characterization method	Ref.
Graphene (mechanical exfoliated)	1	1000 ± 100	70–740	130 ± 10	~12	Indentation on circular membrane	93
	23–43	~1000	N/A	N/A	N/A	Electrostatic force	94
	4	930 ± 48	N/A	N/A	N/A	Pressurizing membranes	95
	1–5	1000 ± 31	N/A	N/A	N/A	Pressurizing membranes	96
	1, 2	2400 ± 400 (1 L), 2000 ± 500 (2 L)	N/A	N/A	N/A	Pressurizing membranes	97
	3–14	~800	N/A	N/A	N/A	Indentation on beam	98
Graphene (mechanical exfoliated + Ar plasma irradiation)	1	~1550	200–800	80–94	N/A	Indentation on circular membrane	99
Graphene (GO reduced)	1	250 ± 150	N/A	N/A	N/A	Spring constant of beam	100
Raphene (CVD growth)	1	~157	~85	~35	N/A	Indentation on circular membrane	101
	1	~1000	N/A	~121 (small grains), ~140 (large grains)	N/A	Indentation on circular membrane	102
	1	~800	~100	~55	N/A	Indentation on circular membrane	103
MoS ₂ (mechanical exfoliated)	1, 2	270 ± 100 (1 L), 200 ± 60 (2 L)	20–100	22 ± 4 (1 L), 21 ± 6 (2 L)	6–11	Indentation on circular membrane	104
	5–25	300 ± 10	0.15 ± 0.15	N/A	N/A	Spring constant of circular membrane	105
	5–25	330 ± 70	50 ± 20	N/A	N/A	Indentation on circular membrane	106
MoS ₂ (CVD growth)	1, 2	260 ± 18 (1 L), 231 ± 10 (2 L)	110 ± 40 (1 L)	N/A	N/A	Indentation on circular membrane	107
WS ₂ (CVD growth)	1	272 ± 18	150 ± 30	N/A	N/A	Indentation on circular membrane	107
WSe ₂ (mechanical exfoliated)	5, 6, 12, 14	170 ± 7 (5 L), 166 ± 6 (6 L), 168 ± 7 (12 L), 165 ± 6 (14 L)	638 ± 22 (5 L), 691 ± 37 (6 L), 499 ± 34 (12 L), 137 ± 26 (14 L)	>12	>7.3	Indentation on circular membrane	108
BP (mechanical exfoliated)	17–35	27 ± 4 (armchair direction), 59 ± 12 (zigzag direction)	N/A	2.2 (armchair direction), 4.2 (zigzag direction)	7.2 (armchair direction), 6.5 (zigzag direction)	Spring constant of beam	109
	17, 37	276 ± 32 (17 L), 90 ± 6.4 (37 L)	180–1200	>25	>8	Indentation on circular membrane	110
GO (solution-based deposition)	1–3	208 ± 23 (1 L), 224 ± 18 (2 L), 230 ± 27 (3 L)	54 ± 14 (1 L), 32 ± 6 (2 L), 28 ± 4 (3 L)	N/A	N/A	Indentation on circular membrane	111
Mica (mechanical exfoliated)	2–14	202 ± 22	140 ± 80	4–9	2–4.5	Indentation on circular membrane	112
h-BN (CVD growth)	2, 4, 5	279 ± 20 (2 L), 269 ± 13 (4 L), 252 ± 15 (5 L)	8.8 ± 1.2 (2 L), 12.8 ± 1.3 (4 L), 15.7 ± 1.5 (5 L)	~9	2.2	Indentation on circular membrane	113

polymer substrate. This anisotropic behaviour has the potential to mimic neuromorphic-based sensors that utilize weight updating upon polarization stimulus.

It has been reported that inhomogeneous local strain distributions in a single structured device provoked significant performance enhancements by the creation of wrinkles or pseudomagnetism and single-photon emissions by exciton-funneling effects¹¹⁹ as shown in Fig. 10B, in which a tilted SEM image of nanocores is presented. Localization of exciton

within a structure is called exciton funneling, and this controls the motion of exciton as a result of the inhomogeneous distribution of strain, as seen in Fig. 10C (top panel). The left side of the top panel shows the exciton funneling of monolayer MoS₂, and the inverse funneling effect for monolayer black phosphorus is shown on the right.

Due to the sub-10 nanometer vertical profile of composites, the characterization, control, and monitoring of strain is gaining attention due to the suitability of macroscopic homogeneity



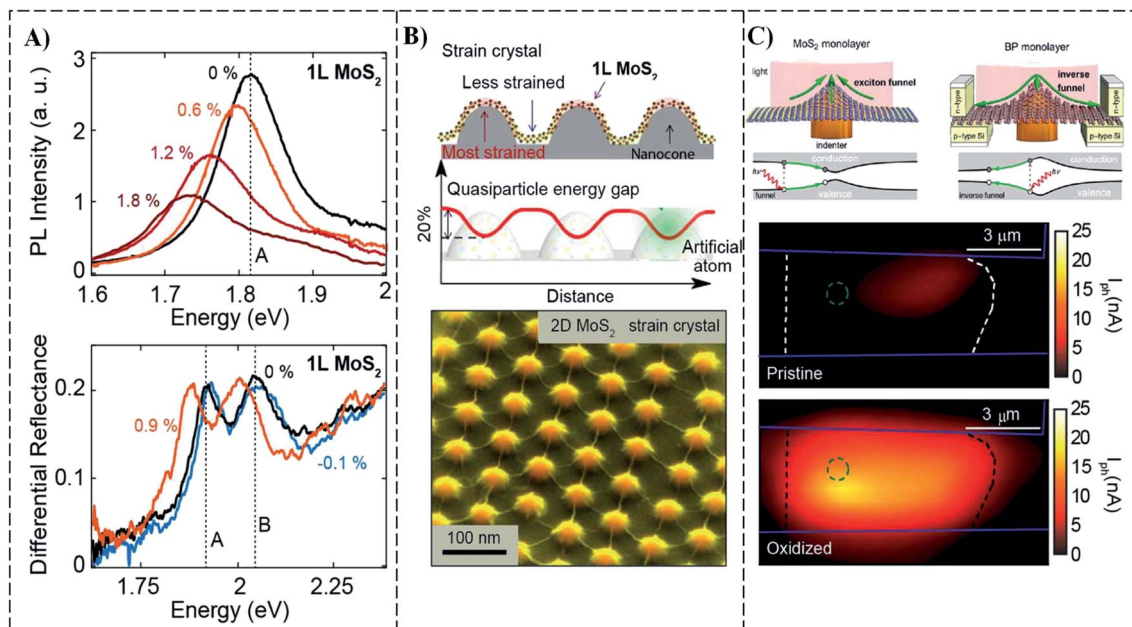


Fig. 10 (A) Bandgap tuning of uniaxial and biaxial of monolayer MoS₂ under different strain conditions. (B) Nanocones in substrate introduce periodic and local strain on top MoS₂. (C) Excitonic funneling and inverse funneling effects in 2D semiconductors. The bottom part shows inverse funneling from a strained HfS₂-based photodetector.¹¹⁹

and low density of composites.^{47,120} Traditionally, Raman spectroscopy is considered to be an efficient tool for 3D–2D material characterization.

The recently reported photoluminescence (PL) spectroscopy and Raman spectroscopy are applied to observe the non-uniform distributions of the band gap in single-layer WS₂ upon application of strain, and the analysis is extended to WS₂ thin-film nanocomposites to validate the method.¹²² The other characterization techniques involve field emission scanning electron microscopy and Fourier transform infrared spectroscopy, which are used to observe the bonding profile of nanocomposites based on multi-walled CNTs.¹²³ Vertically stacking results in interlayer coupling, which is strongly influenced by the application of strain. Interestingly, applying strain “laterally” to heterojunctions of 2D materials and their composites results in a unique class of phenomenon. The lateral heterostructure for single-layer electronics has been well studied and reported.¹²⁴ Spatial modulation was observed for a lateral PN junction fabricated with MoS₂ and WSe₂; the energy distributions of both 2D materials affect the band structure due to lattice mismatch and cause local strain distributions. Further

predictions state that band structure modification results in carrier confinement, leading to high mobility and low-power electronic devices.^{114,125,126}

The tensile strength is a reaction of the materials to resist when forces are applied in tension. It is often defined as stress needed to break the sample. It is usually expressed in Pascals or psi (pounds per sq. inch). The tensile strength is a very crucial property of polymers which are supposed to undergo a stretch or elongations for example fiber should have better tensile strength for its common applications. When compared with the conventional polymers, functional nanocomposites exhibit greater tensile strength. For example, cellophane films and nitrile rubber sheet have tensile strengths of 50–120 MPa and 20–30 MPa, respectively whereas carbon fiber reinforced polymer composites range from 1200 to 2410 MPa, depending on fiber orientation and other design factors.

4.2 Thermal properties of 2D composites

Some 2D nanocomposites with efficient transport properties were proven to be suitable candidate materials for applications involved in thermal management. Well-known 2D materials,

Table 2 The thickness-dependent thermal conductivity of different 2D materials and composites

Sample	Thickness	Method	κ (W m ⁻¹ K ⁻¹)	Comments	Ref.
MoS ₂	Monolayer	Raman	62.2	Supported	128
MoS ₂	11 layers	Raman	52	CVD, transferred, suspended	129
TaSe ₂ film	45 nm film	Raman	9	Exfoliated, suspended	130
WSe ₂	62 nm film	TDTR	0.05	Cross plane, disordered	131
TiS ₂	Bulk	Parallel thermal conductance method	0.69	Organic intercalation	132
WS ₂	Monolayer/bilayer	Raman	32/53	CVD, suspended	133



when fabricated with other composites or polymers, exhibit crossing-band features. For example, when graphene is rolled.

Around vertically aligned with graphene films and penetrated in liquid PDMS, it shows thermal conductivity of about $614.85 \text{ W m}^{-1} \text{ K}^{-1}$, which is an enhancement of $\sim 3329\%$ compared with pristine PDMS.¹²⁷

Engineering the fabrication process of 2D materials and their composites can also lead to the modulation of thermal profiles; for example, by controlling the isotope concentration of hBN, one can modulate its thermal conductivity up to $585 \text{ W m}^{-1} \text{ K}^{-1}$ at room temperature, which is 80% higher than that previously reported in experimental observations. Moreover, various enhancements in the thermal conductivity of graphene are reported using impurity deposition,¹³⁴ dimensionality crossover, the effect of isotope manipulation,¹²⁵ chemical functionalization, and length-dependent characteristics.^{114,135}

By utilizing the van der Waals (vdWs) interaction of layered materials, thermal conductivity can be tuned according to the ingredients of the specific composite. For the case, including only a small amount of an excellent thermal conductor, for example, graphene, in the target composite, one can significantly improve its thermal conductivity with engineered

loading. The interesting aspect of h-BN is its electrically insulating ability in parallel with thermal conduction, which suggests its potential use in graphene-based heterostructures.⁴⁸ The thermal conductivity of other layered two-dimensional materials such as hBN is significantly dependent on temperature, thickness (number of layers), and type of polymers used during the fabrication process. Previously Table 2 reported results had discrepancies between theoretical estimations and experimental outcomes. This is attributed to the difference in crystal quality and/or lattice mismatches, polymer (PDMS/PMMA-assisted manufacturing) residuals, and any other contaminations that resulted in additional phonon scattering, thus decreasing thermal conductivity.¹³⁶

In the domain of layered 2D materials, TMDs play a very important role in applications where thermal management is involved. MoS₂, which is widely used in optoelectronics devices, exhibits strong layer-dependent behaviour in terms of bandgap engineering, *i.e.*, from an indirect bandgap (1.3 eV) in bulk MoS₂ to a direct bandgap (1.8 eV) in the monolayer,¹³⁷ in addition to the creation of different phonon-scattering modes.¹³⁵ An experimental study showed few layer-dependent thermal transport characteristics of CVD-grown MoS₂. The thermal

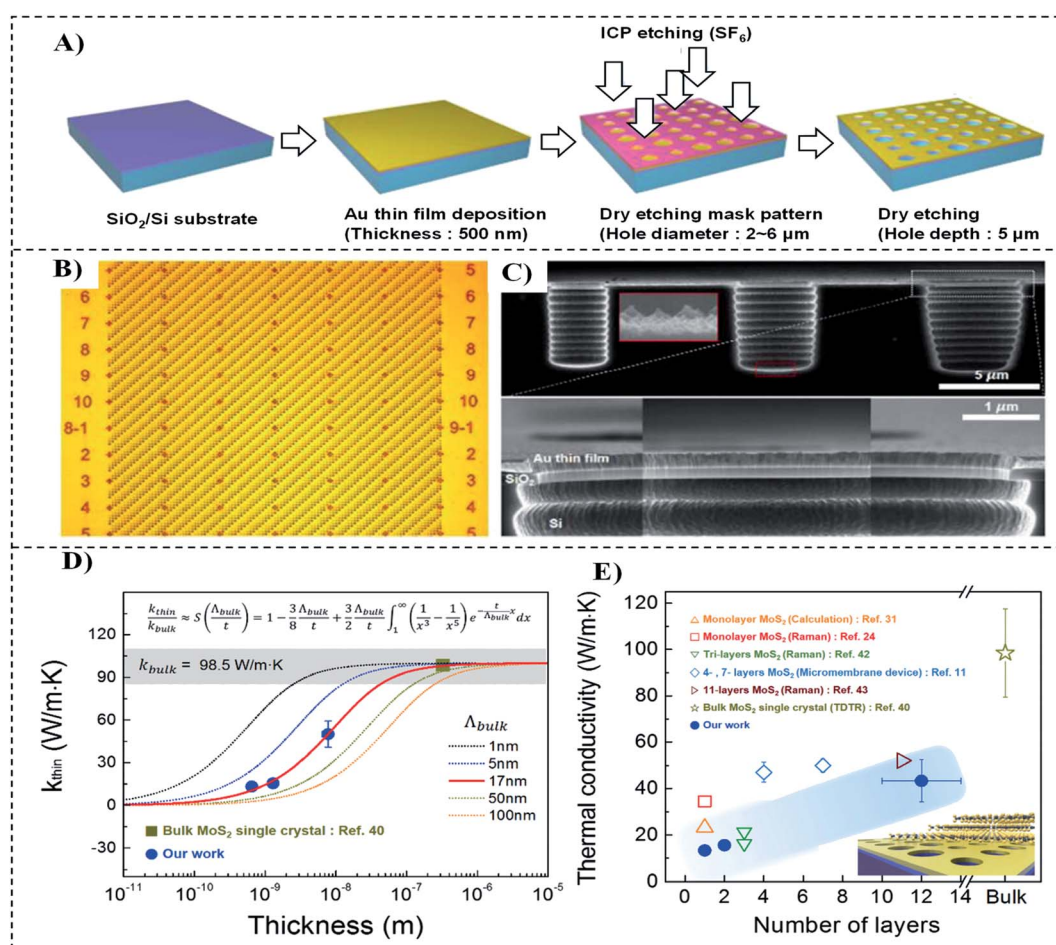


Fig. 11 (A) Fabrication steps of the Au/SiO₂/Si substrate with microcavity arrays (B) top view, (C) side view, (D) thickness-dependent thermal conductivity of MoS₂, (E) summary of thermal conductivity for MoS₂ films.¹²¹



conductivity of MoS₂ (11 layers) is measured to be about 52 W m⁻¹ K⁻¹ at room temperature. From bulk to the monolayer, thermal conductivity has a decreasing trend, as shown in Fig. 11.¹²¹ This is attributed to the higher phonon scattering induced by the change in the inharmonic force constant and the smaller group velocity of various phonon modes.

One recent study shows an anomalous behaviour of thermal conductivity in a 2D composite known as tungsten ditelluride (TdT₂).¹¹⁴ The study shows that thermal conductivity directly correlates with the number of layers of the stated 2D composite, which is attributed to the change in the phonon dispersion relations. With the addition of the layer numbers, optical phonon branches move downward, providing more channels for the Umklapp scattering and decreasing thermal conductivity.

A continuous increase in the layer number causes the low-frequency optical phonon branches to have high group velocity and leads to the increase in lattice thermal conductivity. Table 2 shows the thermal conductivity of different two-dimensional materials and composites.

BP is a well-known layered material used for many electronic and optoelectronic applications.¹¹⁴ Due to its stability issue of BP at ambient temperature, passivation and anti-oxidation must be used during the measurement of its thermal profile.¹²⁶ Initial studies on the thermal conductivity of BP identified nearly 10 W m⁻¹ K⁻¹ in bulk polycrystalline samples of BP. One interesting property of BP is anisotropy for thermal conductivity. A predictive study presented by Jain and McGaughey¹²⁷ showed a strong in-plane anisotropic ratio of thermal conductivity of 3.5 along zigzag (ZZ = 36 W m⁻¹ K⁻¹) to armchair (AM = 110 W m⁻¹ K⁻¹) orientation at a room temperature of 300 K. Zhu *et al.* exposed the 3D anisotropic thermal conductivity of BP and demonstrated results with different crystalline orientations induced by changing group velocity.¹³⁷ The contribution of ZA mode for thermal conductivity is relatively low compared with the surface scattering for 2D-BP-like materials.¹²⁷

Various types of 2D layered composites belonging to group V-VI materials are drawing significant attention towards their thermoelectric applications, such as the bismuth tellurides Bi₂Te₃ and Bi₂Se₃ and Sb₂Te₃. They consist of vdW-connected quintuple layers that are in-plane covalently bonded. There are several attractive features of these composites. For example, materials like bismuth telluride Bi₂Te₃ and others are known for their good thermoelectric properties and show potential for use as topological insulators due to their reduced electric conductivity. However, one challenging research domain for these layered composites is the isolation or decoupling of their thermal and electrical properties. Numerous attempts have been made, but no significant results have been shown. For example, the thickness-dependent thermoelectric properties of antimony telluride, Sb₂Te₃, show that with proper tuning and controlling the grain size, its electric and thermal decoupling can be achieved.

4.3 2D-materials-based composites in photocatalysts and water splitting

Recently, many efforts have been made in designing semiconductor-mediated photocatalysts. The photocatalyst is

used in a process in which acceleration of a photoreaction happens in the presence of a catalyst when solar energy (or a light source) separates the electron-hole pair within a host substrate, and these separated components contribute to oxidation and reduction simultaneously, thus producing chemical energy (hydrogen fuel). Although several studies on 2D semiconductor photocatalysts have been conducted to date, degradation, low cost, charge transfer kinetics, and recycling remain issues that obstruct the broad application of thin semiconductor-based photocatalysts. Here, the role of the 2D materials and their heterostructures are examined to address the above-mentioned shortcomings of the existing materials. The 2D-material-based photocatalyst has gained tremendous attention because of its thickness-dependent physical, mechanical, chemical, and optical properties. The mainstream domains encompass water splitting, oxidation/reduction, CO₂ reduction, N₂ fixation, and NO₂ removal. Zhu *et al.*¹³⁸ fabricated a two-dimensional/two-dimensional (2D/2D) Bi₂O₂CO₃/Bi₄O₅Br₂ (BOC/BOB) heterostructure as a photocatalyst by stacking the ultrathin nanosheets of Bi₂O₂CO₃ and Bi₄O₅Br₂ by one-step hydrothermal synthesis (Fig. 12). They concluded that, by optimizing the content of BOC, the 30% BOC/BOB composite can display superior photocatalytic activity (53.2%) for NO_x removal under simulated solar light illumination, as shown in Fig. 12, which is much sophisticated than that of single-phase BOB (37.9%) or BOC (20.4%).

With large surface-to-volume ratios, 2D materials have attracted the attention of researchers due to their large surface availability allowing co-catalysts to survive and the abundance for charge carriers.¹¹⁴ In addition, the reduced thickness of 2D layered materials, even down to a single atom, ensures less traveling from the bulk to the surface by a photo-induced charge carrier, which contributes to high yield. Although 2D materials have tremendous characteristics, there are still some challenges to be addressed. One big challenge is the scarcity of surface-active sites in 2D materials for photoactivity, even if there is an abundance of charge carriers, because these consist of relatively unsaturated atoms, which are often found on the edges of the 2D surface without surface defects or absorption states. Besides this, a wide bandgap is required for the strong reductive/oxidative ability for photocatalysis. This requires shorter wavelengths in the ultraviolet (UV) band, which is only 4% of solar energy.

Researchers conduct material engineering, such as vacancy engineering, with the 2D layered materials and composites; elemental doping; integration with other semiconductors, metal, or metal oxides;¹³⁵ and morphology and structure control. After analyzing bandgap and structural orientation/arrangements, many 2D materials and composites have shown excellent photocatalysts. Graphitic carbon nitride (g-C₃N₄) is a layered material with a special allotrope of covalent carbon nitride. It possesses a condensed and conjugated structure, exhibits great tolerance to neutral and acidic and alkaline surroundings, and possesses a bandgap of 2.7 eV, which is enough for photocatalytic hydrogen generation. Due to the weak vdW interlayer interaction can be easily transformed from bulk to nanosheets, which have shown excellent



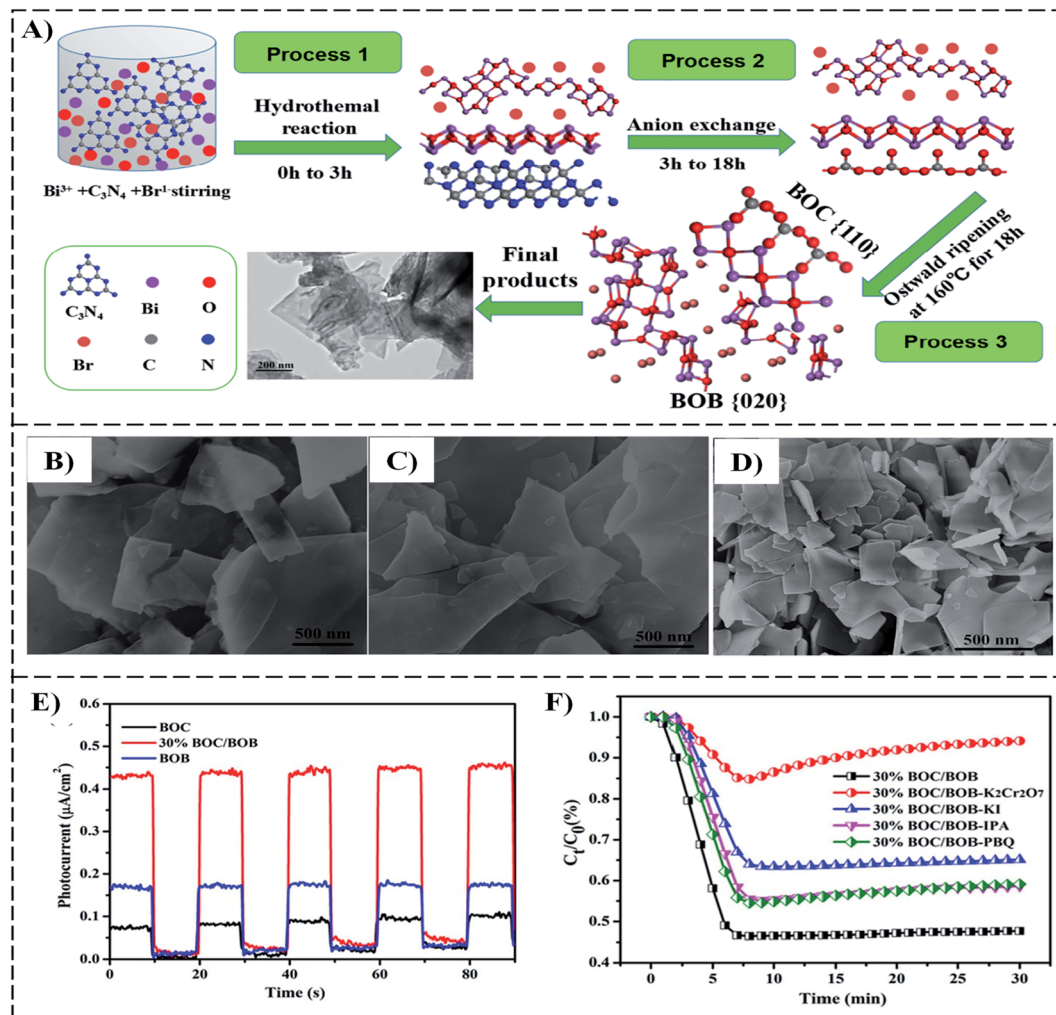


Fig. 12 (A) Schematic of the formation mechanism for 2D/2D $\text{Bi}_2\text{O}_2\text{CO}_3/\text{Bi}_4\text{O}_5\text{Br}_2$ (BOC/BOB) heterojunction; SEM images of, (B) BOB, (C) BOC, and (D) 30% BOC/BOB, respectively, (E) photocurrent densities, (F) photocatalytic activity of 30% BOC/BOB composite photocatalyst under various scavengers.¹³⁸

performance in photocatalytic water splitting.¹³⁵ Due to the abundance of its defect states, it is also observed that it provides additional sites for the activity of photocatalysts. Similarly, BP is also considered a good candidate for such a process as its bandgap is tunable by the number of layers. Zhu *et al.* reported that with ball milling with LiOH, 2D BP achieved a hydrogen evolution rate of $512 \mu\text{mol g}^{-1} \text{h}^{-1}$ under visible light, which was deemed a good value by the investigators, and they believe that it should be further examined in future research.¹²⁵

Two-dimensional-material-based composites, their derivatives, and metal-organic frameworks (MOFs) have also been widely used in photocatalysis-based water splitting research due to their tunable and unique architectures, distinctive features, and desirable structural details.⁹¹ It is believed that, amongst others, graphitic carbon nitride ($g\text{-C}_3\text{N}_4$) and graphene are two important factors in achieving the coveted properties of low charge recombination, efficient electron conductivity, fast kinetics, high photoabsorbance activity, and a large surface area for water splitting to produce hydrogen. It has been conceived

that various techniques can raise the production of hydrogen, including synthesis-controlled morphology, surface functionalization, nano-heterostructures, doping with noble metal particles, and surface decoration of nanoparticles in 2D cages. By cleverly engineering the electronic structure and utilizing hybridization, composites such as electrochemically reduced thin films (*e.g.*, $\alpha\text{-Fe}_2\text{O}_3$),¹³⁵ layered oxyhalides (FeOBr , $\text{Bi}_4\text{VO}_8\text{Cl}$, *etc.*), metal chalcogenides, thiophosphates (CoPS_3), 2D MOFs, and metal-free semiconducting materials, can be produced.¹¹⁴

Particularly, 2D-based composites exhibit multiple advantages such as larger specific surface areas, good conductivity, superior electron mobility, catalyst support, prolonging cyclability, and good optical properties. Therefore, 2D-based composites may provide a new generation of materials for excellent photocatalytic activity compared with one- and three-dimensional materials. The large specific surface areas lead to a great number of active sites on the surface. The good conductivity and high electron mobility lead to an ultrafast electron transfer platform to facilitate the transfer, enhanced



separation of photogenerated charges, and reduced recombination of electron–hole pairs. Consequently, 2D layered composites are opening up a new stream of enhanced photocatalysis for future energy harvesting applications.

4.4 2D-materials-based composites in optoelectronics

Two-dimensional-material-based composites have been widely used in optoelectronics since their emergence. Layer-dependent bandgap tunability and broadband detection make the 2D composites a suitable candidate to be used in optoelectronics devices. The electron–photon interaction is more exciting at the nanoscale, which also evolves plasmonic properties of 2D materials. By analyzing the physics behind 2D materials and their composites, researchers identified and quantified nearly different two-dimensional materials and composites trigonal and hexagonal for the maximum optical TAR (transmittance, absorption, and reflection) at different wavelengths ranging between near-infrared (NIR) to deep UV. Layer-dependent bandgap alignment and work functions give freedom of choice for desired optical response.⁹¹ Band gaps of 2D materials can be engineered through some reported ways of intercalation, control of the number of layers, heterostructure, strain engineering, and the dominant electric field effects. Recently discovered 2D materials exhibit highly tunable bandgaps ranging from infrared (0.3–1.6 eV) for bilayer graphene and black phosphorous (BP), visible (1.7–3.1 eV) for TMDs, and UV (3.2–6 eV) for hBN, thus enabling the scientific community to develop a wide range of optoelectronic devices.^{139–141} The localized hot excitonic effect in thinner 2D heterostructures gives rise to an ultrafast charge transfer rate and higher photocurrent generation.¹²⁵ The superposition (heterostructure) of monolayers of MoSe₂ and WSe₂ resulted in the localization of excitons, irrespective of negligible interlayer coupling.

Expansion in the material library for optical and optoelectronic devices is an ongoing process. Design rules using effective medium theory are being formulated for newly emerged nanocomposites to be effectively utilized in the industry.^{142,143} A meaningful effort towards the previous expansion was recently reported. Selectively mixing tunable, transparent As–S (Se) NPs and an acrylate monomer as initiators was carried out to observe the enhanced amplitude-phase modulation relax parameter the composite. This combines the chalcogenide glass composite with polymer.¹⁴⁴

4.5 Other applications

It is always advantageous to extract the maximum benefits from natural resources, such as water flow, solar energy, and coal/natural gases. Photovoltaics is a well-known process for the extraction of electricity using solar energy. The 2D nanocomposite is also used to modify filtration materials, which are further utilized in point-of-use (POU) water treatment systems. With restricted access to sanitarily safe water resources, the POU water treatment system and device are of considerable importance. The filtering materials employed in POU systems can thoroughly remove the contaminant.

Jakubczak *et al.*¹⁴⁵ have reported a ground-breaking 2D Ti₃C₂/Al₂O₃/Ag/Cu nanocomposite-modified filtration material with the

potential in POU water treatment, which collected 99.6% of bacteria in the filter. Sher *et al.*¹⁴⁶ synthesized Cd-doped ZnO nanocomposites based on g-C₃N₄ (Cd–ZnO/g-C₃N₄ NCs) through the co-precipitation method. The products were utilized against Gram-positive and Gram-negative bacterial strains to estimate their antibacterial activities. Qamar *et al.*¹⁴⁷ also synthesized an efficient photocatalytic composite Ni/ZnO/g-C₃N₄ (NiZG), using Ni-doped ZnO NPs with graphitic carbon nitride (g-C₃N₄) and a cost-effective chemical co-precipitation approach. The catalytic proficiency of NiZG composites was evaluated by determining the UV-vis absorption spectra for methylene blue, and the bactericidal ability was tested against Gram-positive and Gram-negative microbes.¹⁴⁷

The aerospace, automotive, and electronics industries require new 2D-material-reinforced epoxy composites with excellent mechanical properties and high thermal conductivity. In this regard, Han *et al.* studied and investigated the effect of adding hBN sheets and graphene platelets (GnPs), achieving improved thermal conductivity and enhanced mechanics of the epoxy resin. As a result, GnPs improved the energy release rate of epoxy by 338%, whereas BN increased it by 260%, and both at 2 wt%. Thermal conductivity of the epoxy/GnP and epoxy/BN composites was enhanced by 135% and 64% respectively, both at 4 wt%.¹⁴⁸ Moreover, Sanchez *et al.* obtained a thermal conductivity of 0.22 W m^{−1} K^{−1} for pristine epoxy (EP) and further investigated the effect of a hybrid filler composed of zero-dimensional (0D) spherical micro-sized aluminum oxide (Al₂O₃) and 2D flake-like hBN.¹⁴⁹ Ren *et al.* placed Sb₂S₃ NPs on highly crumpled Ti₃C₂T_x nanosheets using the wet chemical reaction. The evenly distributed Sb₂S₃ NPs resolved the restack issue of the Ti₃C₂T_x nanosheets upon repetitive charging/discharging cycles, producing rich voids for rapid electron/Na⁺ transport and reducing the volume expansion effect from Sb₂S₃. The composite showed a high capacity (329 mA h g^{−1} at 100 mA g^{−1} after 100 cycles), high rate capability, and high cycling performance (215 mA h g^{−1} at 2 A g^{−1} when utilized as a sodium-ion battery (SIB) anode and also maintaining 118 mA h g^{−1} after 500 cycles).¹⁵⁰ Furthermore, Syamsai *et al.* prepared a bi-metallic titanium–tantalum carbide MXene, Ti_xTa_(4−x)C₃, by etching Al atoms from the parent Ti_xTa_(4−x)AlC₃ MAX phase, which showed an outstanding electrochemical redox performance attributed to the creation of a promising, stable bi-metallic MXene material, enabling the storage of Li-ions on the layer surface. Additionally, the Ti_xTa_(4−x)C₃ MXene anode displayed a high aptitude rate due to its good electron and Li-ion transport, indicating that it is a promising candidate as a Li-ion anode material.¹⁵¹ García *et al.* obtained MoTe₂/MoSe₂ flake composite films using an isothermal closed-space vapor transformation, with successful applications in those that demand high interfaces, favoring gas or ion exchange reactions with TMDs. The oxide precursor films were prepared from a Mo isopropoxide solution in isopropanol (IPA), and acid catalysis induced by HCl and thermal annealing at 200 °C, 400 °C, and 600 °C improved the condensation after xerogel formation.¹⁵² On the other hand, MXene/polymer nanocomposites have been investigated, and it has been found that MXene could significantly enhance the mechanical, electrical, and thermal behaviors of polymers. MXene/polymer



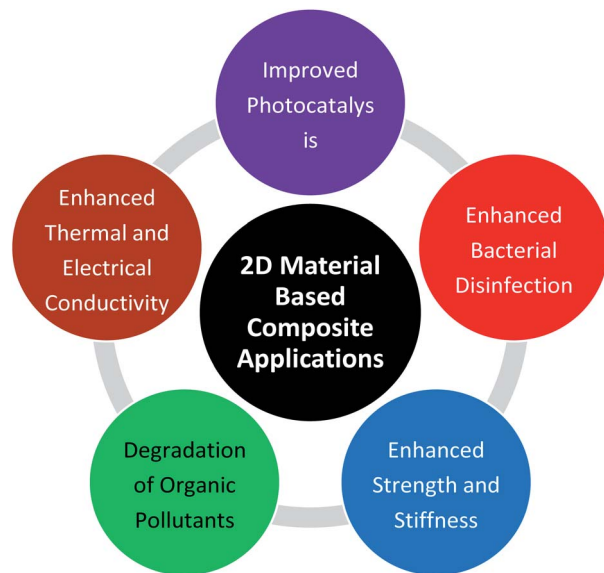


Fig. 13 Illustration of 2D-material-based composite applications.

nanocomposite structures have been observed to provide considerably good results in sensing, biomedical, environmental monitoring, energy, and other fields.¹⁵³

Finally, we would like to emphasize the characteristics of several representative 2D-layered materials, *e.g.*, graphene, $g\text{-C}_3\text{N}_4$, and MoS_2 , and the recent progress in 2D-layered composites for photocatalysts, including photocatalytic hydrogen production, bacterial disinfection, and the degradation of organic pollutants, as illustrated in Fig. 13.

5. Conclusion and perspectives

In summary, we briefly presented representative 2D-material-based composites and their characterizations, classifications, and advantages regarding physics and mechanics related to recent studies as well as emphasize other aspects that formed new nanocomposites based on polymers and metals and the latest achievements of new synthesized-2D materials enabling to fabricate of new nanocomposite species. The 2D material family and their composites widened the performance band of conventional devices supported by the synergy of modern fabrication techniques and measurement/characterization tools. Two-dimensional materials, *e.g.*, graphene, BP, h-BN, TMDs, or MXenes, contributed to a tremendous advancement in multidisciplinary science and engineering. These materials are either exploited in the form of homo- and hetero-structures or polymer-reinforced nanocomposites depending on how many performance parameters can be improved by structural engineering or through functionalization. However, interesting chemistry exists when the above-stated materials combine with other layered materials of similar groups, such as tellurium, selenium, sulphur, antimony, bismuth, and more with crystal-line structures. These 2D-material-based composites have the potential to expand the application areas of conventional 2D-based materials and devices.

Although the enormous potential of the 2D materials-based polymer composites have been demonstrated in various fields, there are still many challenges in their practical applications. First, the mass production of layered composite materials for advanced applications is a significant challenge. Secondly, the synchronization of technological advancements for both mass production and corresponding applications is one of the key factors in pursuing novel composites. The integration of polymers and a metal framework with 2D materials has broadened the scope of composites. It is believed that the composite material-oriented approach could be used to resolve some technical challenges, such as energy storage for future applications, water purification, low-energy consumption electronics, bio-inspired computing,⁹¹ biomedical applications,¹¹⁴ and efficient catalysis.¹³⁷ Many efforts have been reported for the mass production of these composites for commercialization¹⁵⁴ so that these material-based devices can be marketed as early as possible. However, the balance between quality and quantity is still a major issue, as state-of-art fabrication is needed for aliasing functionalities of similar devices in a batch of products, adding financial loading.

Nanosheet-based composites and the utilization of an MOF have made considerable contributions to broadening the sensitivity of existing devices, as sensors are one of the fundamental elements in industrial automation. During detection and sensing, two major challenges still need to be addressed. The first considers the sample matrix effect, which contains ionic contaminations, macromolecules, and particles are added in unwanted signal processing.¹³⁷ Further exploration in terms of characterization of such composites is needed to develop specific fabrication techniques that may lower these anomalies/artefacts and produce consistent detection results while working in a human-friendly environment and optimal lab conditions. The second consideration is scalable fabrication methods in parallel with controlling the quality and quantity for future technological industry.

From the in-depth exploration of nanocomposites based on 2D materials, polymers, ceramics, and many more, we have high expectations regarding the development of the future industry *via* a synergistic approach. Various scientific fields can be benefitted by expanding the nanocomposite materials library *e.g.*, (a) for the detection of specific pathogens and intelligent drug delivery systems (SDDs) in the medical field; (b) nanocomposites for the packaging and plastic industry to enhance processability, modify product properties, and provide protection against thermal or UV light influences; (c) the enhancement of mechanical properties, tensile strength, and weight reduction and improvement to flame retardants for the automotive industry, transportation, sporting goods, aviation, and the space industry. Moreover, self-lubricating composites are a kind of advanced material where solid lubricant materials, such as graphite, hBN, MoS_2 , CNT, and graphene, are used as a primary or secondary reinforcement to achieve better mechanical and tribological properties. Moreover, metal-based nanocomposite (*e.g.*, gold, silver, and copper nanoparticles; graphene; CNTs, graphene oxide; and other 2D-material-based nanocomposites) are being utilized to enhance electrically



conductive properties in the domain of flexible/wearable electronics, organic electronics, textiles, and solar cells. In addition, nanocomposites-based thin films are also used as coatings and flexible displays. Nanocomposites also have applications in the field of photocatalysts, gas sensors, supercapacitors, and broadband sensors for optical and THz band construction. It is strongly believed that the development of 2D-derived nanocomposites carries the optimized solutions of humanity's ever-existed problems of energy, food scarcity, security, and communication at an accelerated pace. Strong research collaboration and the provision of more channels for technology exchange may boost this pace.

Two-dimensional material-based composites are expected to play an essential role in future technologies. Although many significant research efforts have been made, applications using 2D material composites remain limited. Considerable challenges still exist, such as the possibility of large-scale production, enhanced cycling stability, and reduced costs for commercialization. Furthermore, the intrinsic properties of 2D materials are susceptible and dependent on many factors, including surface chemistry, neighboring materials, underlying substrate, and interfaces. Moreover, production methods also have a significant influence on heterostructure applications, *e.g.*, the composited material's performance, reproducibility, and reliability. Studies tend to focus on several properties but skip other relations. This leads to the inefficiency of using composites. For example, some investigators are interested in improving the optical–electrical properties, while other properties are not discussed. Similarly, some reports show the enhanced photocatalytic performance of 2D material composites while ignoring the possibility of cycling stability. Therefore, to enhance the performance of micro-nano devices using 2D material composites, the synchronous properties require further investigation. The combined investigation methods can help to accurately choose the components and accurately control the concentration. These conjunctions will optimize the synchronous conditions that improve the performance and present the opportunity of commercialization.

Two-dimensional composites based on multiple materials require more attention. For example, many studies have reported the production of a polymer composite by adding a single 2D material, and, as result, they observed enhancement in one or two properties; on other hand, there are only a few reports that explored the production of polymer composites by adding more than one (two or three) 2D materials together. It is apparent that each individual 2D material has its unique properties and advantages over other materials but also some downsides. For example, it is well known that graphene has high electrical conductivity, which favours being used in the energy storage device as a conductive filler. However, it has been stated that many graphene-based polymer composites exhibit very low energy density. To solve this issue, the idea of mixing graphene with other 2D materials was proposed.¹⁵⁵ In this regard, MoS₂ sheets were mixed with reduced graphene oxide (rGO) to obtain a hybrid aerogel that could be used as electrode material for supercapacitors, possessing properties of both 2D materials. We believe more effort should be made in

fundamentally researching the various aspects of 2D-material-based composites produced by mixing more than one 2D material for practical applications.

Author contributions

All authors contributed to writing the manuscript, overall draft preparation, review, and editing. All authors have read and agreed to the published version of the manuscript.

Conflicts of interest

There are no conflicts to declare.

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