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Structure modulation on fluorescent probes for biothiols and the reversible imaging of glutathione in living cells†

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The detection of small molecular biothiols (cysteine, homocysteine and glutathione) is of great importance, as they involve in a series of physiological and pathological processes and are associated with many diseases. To realize the real-time monitoring of a specific biothiol, a rapid and reversible probe is required. Therefore, three probes, namely, *o*-MNPY, *m*-MNPY and *p*-MNPY, with pyridine substituted α , β -unsaturated ketone as the recognition site, were reported here, and the reactivity of the recognition site was finely tuned by the connection mode of the pyridine unit. To single out the optimal one, the response performances of three probes toward each biothiol were systematically studied, taking the differences of the intracellular contents of three biothiols into account during the evaluation. Biothiols reacted with the probes through Michael addition, and results showed that the slight structural variations could affect the performances of the probes obviously. *p*-MNPY with the pyridine unit connected to the recognition site through the *para*-position of the nitrogen atom, revealed the best sensing ability among the three probes. It demonstrated rapid response, good selectivity and sensitivity, excellent pH adaptability to Cys and GSH, and displayed reversible detection toward GSH. Finally, *p*-MNPY was successfully applied to track the GSH fluctuations under the oxidative stress stimulated by H_2O_2 in living cells.

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1 Introduction

Nonprotein biothiols, including cysteine (Cys), homocysteine (Hcy) and glutathione (GSH), are the most important small-molecular sulfur containing amino acids in living cells, which play vital roles in a series of physiological and pathological activities.^{1–3} The abnormal fluctuations of the endogenous biothiol levels are usually valuable signals for malfunction of cells and organs, and therefore are closely related to some symptoms and diseases. The deficiency of Cys is associated with symptoms such as slow growth, hair pigmentation, lethargy, *etc.*^{4–8} The excess of Hcy in serum is an independent risk factor for cardiovascular disease.^{9,10} GSH is the most abundant biothiol, which acts as an antioxidant to defend against attack from free

radicals and sustains the intracellular redox homeostasis,^{11,12} and an imbalance in its expression is implicated with cancer, HIV, Parkinson, *etc.*^{13,14} Therefore, it is significant to discriminate the biothiols and monitor the level changes in living cells, which may offer meaningful information for the diagnosis and treatment of related diseases.

Fluorescent probes have attracted considerable attentions in the determination of the variety and content of bio-species, ascribed to their advantages of high sensitivity, facile operation, visualization and noninvasiveness.^{15–22} A variety of fluorophores have been utilized to construct fluorescent probes for biothiol detection, which covered a wide range of emission signals from green to near-infrared fluorescence that could satisfy the requirements in different situations.^{23–29} And some probes could discriminate Cys, Hcy and GSH selectively.^{30–34} However, most of the reported probes are not reversible, which means the probes could respond to biothiols only once and cannot track the changes of biothiols dynamically in living cells. This defect may be not favorable for providing pathophysiological insights. In order to solve this problem, reversible probes that can implement the real-time quantitative detection of biothiols are valuable.^{35–38} From the perspective of molecular design, the reaction between the probe and biothiols should be reversible. Among the extensively studied reaction mechanisms for biothiol detection, such as nucleophilic substitution,^{39–41}

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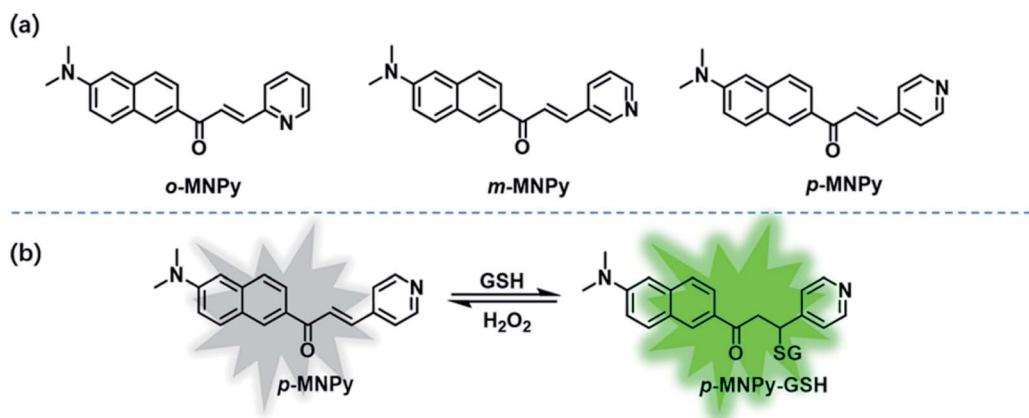
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Scheme 1 The molecular structures of (a) *o*-MNPy, *m*-MNPy and *p*-MNPy in this work, and (b) the schematic illustration of the reversible recognition of GSH by *p*-MNPy.

nucleophilic addition and cyclization,^{42–44} Michael addition,^{36–38} and multisite reaction strategy,^{45–47} Michael addition is more reasonable to build a reversible probe credited to its nature of reversibility. Wang and coworkers reported the first fluorescent probe ThiolQuant Green (TQ Green) for quantitative detection of GSH in living cells through carefully adjusting the properties of the Michael acceptor.^{48,49} Urano and coworkers obtained a mitochondrion targeted rapid fluorescent probe QuicGSH (QG3.0) for GSH quantification through rational molecular design, which was based on fluorescence resonance energy transfer (FRET) mechanism with rhodamine as the platform.³⁶ These two probes are representative examples for ratiometric and reversible detection of GSH. We have reported a lysosome-targeted two-photon fluorescent probe **MNPO** for rapid and selective detection of cysteine in living cells.⁵⁰ We mainly focused on the targeting ability and high sensitivity of this two-photon activatable probe, whereas the structure of the recognition site “pyridine substituted α , β -unsaturated ketone” was not carefully studied. The capability of this probe to reversibly sense biothiols was not taken seriously either at that time. However, the Michael addition mechanism of this probe makes the reversible detection possible. Considering that the fluorogen “amino naphthalene ketone” has demonstrated high brightness during the imaging and the modulation of the Michael acceptor is crucial for optimizing the performance of the probe,⁴⁸ we adopted the same fluorogen skeleton as **MNPO** and tried to regulate the property of the Michael acceptor by altering the connection mode of the pyridine unit, in order to pick out the optimal probe for reversible detection of biothiols.

Here, three probes, *o*-MNPy, *m*-MNPy and *p*-MNPy, were reported (Scheme 1). In these molecules, dimethylaminonaphthalene moiety acted as the signal reporting unit, and pyridine modified α , β -unsaturated ketone moiety which was a Michael acceptor acted as the recognition site for biothiols. As the pyridine unit could be attached to the molecular skeleton through three positions, namely, the *ortho*, *meta*, and *para* position of the nitrogen atom, three probe molecules mentioned above were designed and synthesized accordingly. The different connection modes of pyridine would affect the

reactivity of the Michael acceptor, and hence lead to different response performances of the probes, including signal to noise ratio, response speed, and even selectivity. The structure–property relationships of three probes were systematically investigated, and *p*-MNPy showed better performances than the other two. Then the reversibility of *p*-MNPy was carefully studied, as well as the sensitivity and selectivity during the detection of GSH. Finally, *p*-MNPy was utilized to image the GSH fluctuations under the oxidative stress in living cells.

2 Experimental

2.1 Materials, instruments and measurements

The starting materials and dry *N,N*-dimethylformamide (**DMF**) were commercially obtained from Energy Chemical Corporation. Phosphate buffered saline (**PBS**) was purchased from Maokangbio Corporation. Other organic solvents were purchased from Sinopharm Chemical Reagent Co., Ltd HeLa cells were obtained from iCell Bioscience Inc. The detailed information of instruments, sample preparation and test condition, cytotoxicity assay, cell culture and cell imaging process are all listed in the ESI.† The experiments in solution were performed at room temperature, and after the addition of Cys/Hcy/GSH, the probe solution was let stand for 10 min before any spectra measurements.

2.2 Synthesis

1-(6-(Dimethylamino)naphthalen-2-yl)ethenone (MN): 1-(6-aminonaphthalen-2-yl)ethenone (150 mg, 0.8 mmol) and K_2CO_3 (552 mg, 4 mmol) were mixed in 15 mL of DMF at room temperature, and then iodomethane (0.249 mL, 4 mmol) was added. 10 min later, the mixture was heated to 60 °C and stirred for 12 h. After cooling to room temperature, the mixture was poured to ethyl acetate and washed with water for three times, then the organics were dried over anhydrous Na_2SO_4 . After the solvent was removed under reduced pressure, the residue was purified by a silica-gel column chromatography using petroleum ether/ethyl acetate (v/v, 8/1) as eluent to give the product



MN as a brown solid (124 mg, 72.84%). ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ [ppm]: 8.44 (s, 1H), 7.92 (d, J = 8 Hz, 1H), 7.81 (dd, J = 8 Hz, 1H), 7.68 (d, J = 8 Hz, 1H), 7.29 (dd, J = 8 Hz, 1H), 6.96 (d, J = 4 Hz, 1H), 3.07 (s, 6H), 2.62 (s, 3H).

(E)-1-(6-(Dimethylamino)naphthalen-2-yl)-3-(pyridine-2-yl)prop-2-en-1-one (***o*-MNPy**): **MN** (250 mg, 1.17 mmol) and picolinaldehyde (375.71 mg, 3.50 mmol) were dissolved in a mixture of ethanol (5 mL) and THF (5 mL) under an argon atmosphere, and piperidine (0.3 mL) and acetic acid (0.6 mL) were slowly added to the solution separately. Then the mixture was heated to 80 °C and stirred for 96 h. After cooling to room temperature, the mixture was poured to ethyl acetate, and washed with Na_2CO_3 aqueous solution (10%). The organic layer was dried over anhydrous Na_2SO_4 . After the solvent was removed under reduced pressure, the residue was purified by silica-gel column chromatography using petroleum ether/ethyl acetate (v/v, 8/1) to give product ***o*-MNPy** as a solid (152 mg, 43%). ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ [ppm]: 8.72–8.69 (m, 2H), 8.34 (d, J = 16 Hz, 1H), 8.02 (d, J = 8 Hz, 1H), 7.98–7.92 (m, 3H), 7.76–7.72 (m, 2H), 7.47–7.44 (m, 1H), 7.31 (dd, J = 12 Hz, 1H), 6.99 (s, 1H), 3.09 (s, 6H). ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$) δ [ppm]: 188.38, 153.56, 150.85, 150.43, 142.27, 137.88, 137.61, 131.38, 137.27, 130.89, 126.78, 125.76, 125.22, 125.05, 125.02, 124.85, 116.89, 105.24, 40.62. HRMS for $\text{C}_{20}\text{H}_{18}\text{N}_2\text{O}$ [$\text{M} + \text{H}$]⁺, calculated: 303.1497; found: 303.1499. Mass error: 0.66 ppm.

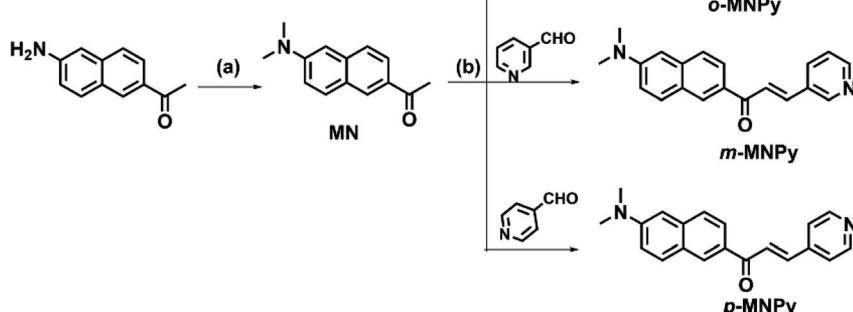
(E)-1-(6-(Dimethylamino)naphthalen-2-yl)-3-(pyridine-3-yl)prop-2-en-1-one (***m*-MNPy**): similar procedures to ***o*-MNPy** except that the starting materials were **MN** (122 mg, 0.572 mmol) and nicolinaldehyde (183.68 mg, 1.71 mmol). ***m*-MNPy** was obtained as an orange solid (35 mg, 20%). ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ [ppm]: 9.06 (s, 1H), 8.79 (s, 1H), 8.62 (d, J = 4 Hz, 1H), 8.39 (d, J = 8 Hz, 1H), 8.25 (d, J = 16 Hz, 1H), 8.00–7.94 (m, 2H), 7.79–7.73 (m, 2H), 7.52 (dd, J = 8 Hz, 1H), 7.33 (dd, J = 8 Hz, 1H), 6.99 (d, J = 4 Hz, 1H), 3.09 (s, 6H). ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$) δ [ppm]: 187.98, 151.19, 150.82, 150.62, 139.61, 137.86, 135.46, 131.45, 131.27, 131.21, 130.87, 126.67, 125.03, 124.97, 124.59, 124.33, 116.90, 105.29, 40.61. HRMS for $\text{C}_{20}\text{H}_{18}\text{N}_2\text{O}$ [$\text{M} + \text{H}$]⁺, calculated: 303.1497; found: 303.1498. Mass error: 0.33 ppm.

(E)-1-(6-(Dimethylamino)naphthalen-2-yl)-3-(pyridine-4-yl)prop-2-en-1-one (***p*-MNPy**): similar procedures to ***o*-MNPy** except that the starting materials were **MN** (273 mg, 1.281 mmol) and isonicolinaldehyde (379.33 mg, 3.54 mmol). ***p*-MNPy** was obtained as an orange solid (97 mg, 25.04%). ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ [ppm]: 8.79 (s, 1H), 8.69 (d, J = 8 Hz, 2H), 8.32 (d, J = 16 Hz, 1H), 8.00 (dd, J = 8 Hz, 1H), 7.96 (d, J = 8 Hz, 1H), 7.87 (d, J = 8 Hz, 2H), 7.75–7.67 (m, 2H), 7.33 (dd, J = 8 Hz, 1H), 6.99 (d, J = 4 Hz, 1H), 3.09 (s, 6H). ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$) δ [ppm]: 188.06, 150.93, 150.81, 142.61, 140.21, 137.99, 131.75, 131.31, 130.69, 127.05, 126.77, 125.00, 124.96, 122.94, 116.96, 105.31, 40.62. HRMS for $\text{C}_{20}\text{H}_{18}\text{N}_2\text{O}$ [$\text{M} + \text{H}$]⁺, calculated: 303.1497; found: 303.1499. Mass error: 0.66 ppm.

3 Results and discussion

3.1 Design and synthesis

The rigid and planar structure of the donor- π -acceptor (D- π -A) type **MN** (Scheme 2) had shown high brightness during the cell imaging,⁵⁰ therefore, it was selected as the signal reporting unit of the probe. Pyridine substituted α , β -unsaturated ketone, making up of the Michael acceptor, acted as the recognition site. The different connection modes of the pyridine group in ***o*-MNPy**, ***m*-MNPy** and ***p*-MNPy** would affect the electronic properties of the Michael acceptor in different degrees and hence alter the reactivities of the probes with biothiols. So, the whole three probe molecules were designed and synthesized to single out the best one for biothiol detection. The synthetic routes of the intermediates and the target molecules are shown in Scheme 2. **MN** was prepared according to the literature with some modifications in a high yield using K_2CO_3 as the catalyst for the nucleophilic substitution.⁵¹ The probe molecules were synthesized by aldol condensation between **MN** and the corresponding pyridine aldehyde, where weak base and acid were chosen as the co-catalyst. The reactions lasted for 4 days, while full conversion of the reactants was not observed. The yields were only around 20% to 40%. The probe molecules were fully characterized by ^1H NMR, ^{13}C NMR and HRMS (Fig. S1–S5†).



Scheme 2 The synthetic routes of ***o*-MNPy**, ***m*-MNPy** and ***p*-MNPy**. Conditions: (a) CH_3I , K_2CO_3 , DMF, 60 °C; (b) piperidine, acetic acid, ethanol/THF, 80 °C, 96 h.



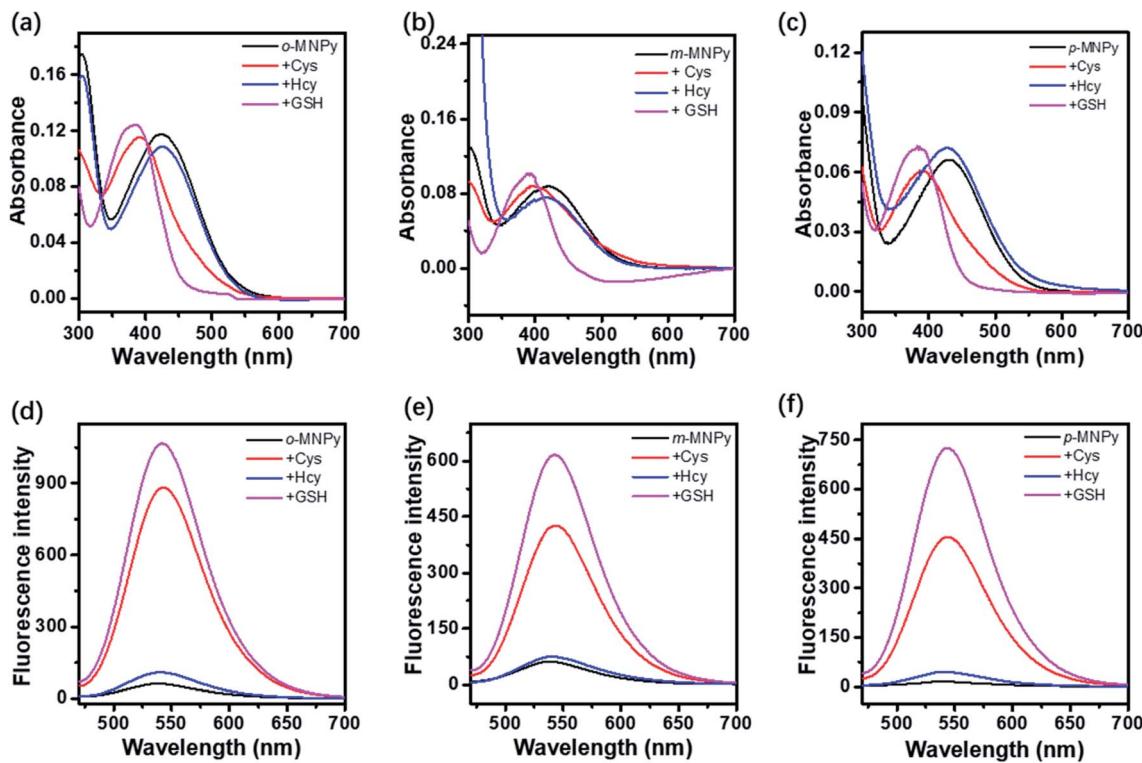


Fig. 1 The UV-vis spectra of (a) *o*-MNPy, (b) *m*-MNPy and (c) *p*-MNPy, and the fluorescence spectra of (d) *o*-MNPy, (e) *m*-MNPy and (f) *p*-MNPy before and after the addition of Cys (100 μ M), Hcy (10 μ M) and GSH (5 mM). The concentration of each probe was 10 μ M in PBS (containing 10% DMSO).

3.2 Photophysical properties

The photophysical properties of the three probes were studied in PBS. In most literatures, the concentrations of the three biothiols were set in the same level (for example 100 μ M) to examine the performances of the probes. However, considering their huge differences in intracellular contents (Cys 30–200 μ M, Hcy 5–13 μ M, and GSH 0.5–10 mM),⁵² the concentration of Cys, Hcy and GSH during this test was selected as 100 μ M, 10 μ M and 5 mM, respectively. As shown in Fig. 1, the absorption maxima of *o*-MNPy and *p*-MNPy were at 429 and 430 nm, respectively, while that of *m*-MNPy was blue shifted to 419 nm. As is known,

the conjugation degree of the compound and the strength of the intramolecular charge transfer (ICT) process could both impact the absorption spectrum of a molecule.^{24,53} Since the structures of the three probes were similar, we supposed that the 10 nm blue-shift of the absorption maximum of *m*-MNPy was ascribed to the weaker ICT process compared with the other two. This result also indicated the inferior electron-withdrawing ability of the *meta*-connected pyridine. When GSH was added to the probes severally, the absorption spectra of the three probes were all blue shifted to around 390 nm, which suggested the happening of the Michael addition reaction between the probes and GSH resulting in the interruption of the conjugation

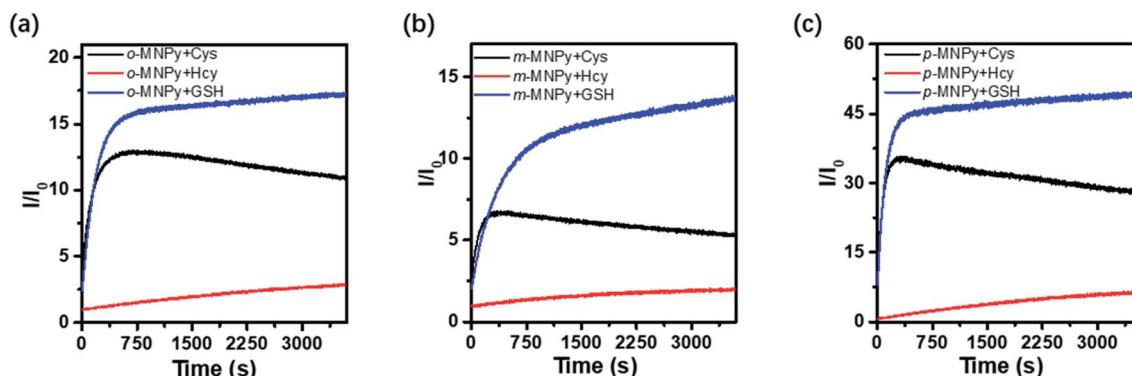


Fig. 2 The time-dependent fluorescence intensity changes of (a) *o*-MNPy, (b) *m*-MNPy and (c) *p*-MNPy in the presence of three biothiols. Concentrations: probe 10 μ M, Cys 100 μ M, Hcy 10 μ M, and GSH 5 mM.

between **MN** and the pyridine unit. The same reaction also occurred upon Cys treatment, as the absorption peaks of the probes were blue-shifted as well. We suspected that these similar changes of the absorption spectra in the presence of Cys and GSH were probably due to the structural similarity of the two biothiols that GSH contains Cys in its structure. However, in the presence of 10 μ M of Hcy, the spectra changed little. Unlike the other two probes, **p-MNPy** could respond to higher concentration of Hcy (100 μ M) (Fig. S6†). The fluorescence of the probes was obviously turned on in the presence of Cys and GSH with the emission maxima at around 543 nm. The enhancements of the fluorescence intensity during the addition of GSH were 17, 11, and 43-fold for *o*-, *m*-, and **p-MNPy**, respectively. And similar to the results from the absorption spectra, the addition of Hcy showed little influence on the fluorescence. The quantum yields (QYs) of *o*-, *m*-, and **p-MNPy** in PBS (containing 10% DMSO) were calculated to be 0.16%, 0.19% and 0.13%, respectively, using quinine sulfate as the standard (QY = 0.53 in 0.1 M H_2SO_4), and they were significantly increased to 5.64%, 3.56% and 6.55%, respectively, in the presence of GSH. All the spectral data were summarized in Table S1.† As both the absorption and emission experiments had shown that the three molecules could respond to Cys and GSH in the range of their intracellular concentrations, they had great potentials as the fluorescent probe for biothiols in living cells.

3.3 Response speed

As an important factor, the response speed of the probe toward each biothiol was further investigated, which was associated with the efficiency of the detection. The time-dependent fluorescence changes of the probes toward Cys, Hcy and GSH were shown in Fig. 2. Agreed with the findings in the previous part, all the three probes showed much more distinct response toward Cys and GSH than Hcy. For *o*- and **p-MNPy**, their reactivities with Cys and GSH were very similar. In 10 min, the fluorescence intensity of *o*-**MNPy** could reach a plateau after the addition of Cys or GSH, and an even shorter time of 8 min was

needed for **p-MNPy** to complete the reactions. However, *m*-**MNPy** spent more than 20 min to reach its largest fluorescence intensity in the presence of GSH. We speculated that the different reactivities of three probes were probably due to the subtle structural changes of the Michael acceptors, where *m*-connected pyridine unit showed the worst electron-withdrawing ability and could not activate the adjacent double bond as efficiently as the *o*- or *p*-connection mode. It was interesting that the fluorescence intensities of three probes in the presence of Cys tended to decrease with the increase of time. Just like the probes designed for Cys with acryloyl group as the recognition site,^{23,34,54,55} we supposed that further reaction between the “-NH₂” of Cys and the “C=O” of the probe took place with a stable 7-membered ring generated after the Michael addition, which weakened the ICT process of the D-A system and made the emission decrease over time. This phenomenon might be used to differentiate Cys and GSH. Compared with the other two probes, **p-MNPy** had demonstrated both larger fluorescence on/off ratio and faster response speed, therefore, it was chosen as the fluorescent probe for Cys and GSH with its other performances studied in the following parts.

3.4 Sensitivity of **p-MNPy**

The fluorescence changes of **p-MNPy** with the addition of different amount of Cys or GSH were recorded as shown in Fig. 3 and S7.† There was a linear relationship between the concentration of Cys and the fluorescence intensity of the probe when Cys was added to the probe in the range of 1–10 μ M, and for GSH, the concentration range was 0.2–1 mM. Based on the equation of $3\sigma/k$, the limit of detection (LOD) of **p-MNPy** was calculated to be 0.029 μ M for Cys and 1.40 μ M for GSH, where σ is the standard deviation of the blank solution ($\sigma = 0.0998$) and k is the slope of the linear fitting curve ($k = 10.473$ for Cys, and 213.35 for GSH). The LOD was much lower than the corresponding intracellular biothiol contents, which demonstrated that **p-MNPy** was sensitive enough to detect the Cys or GSH in living cells.

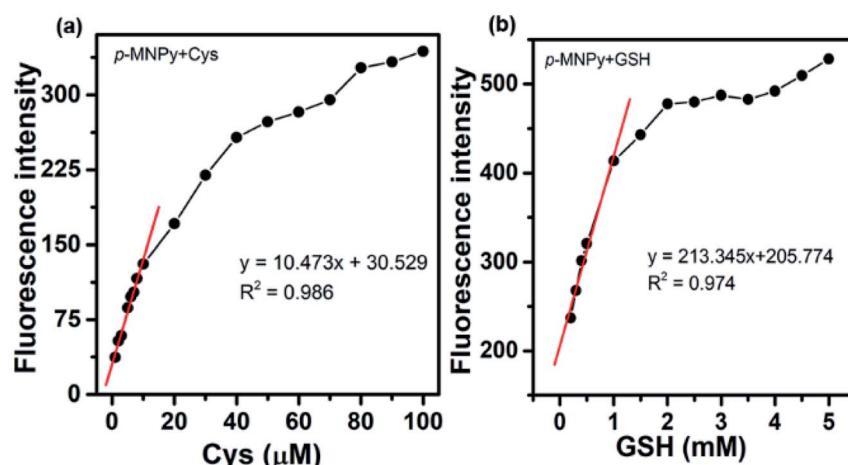


Fig. 3 The changes of fluorescence intensity of **p-MNPy** as the addition of (a) Cys, and (b) GSH. Concentrations: **p-MNPy** 10 μ M, Cys 1–100 μ M, GSH 0.1–5 mM.



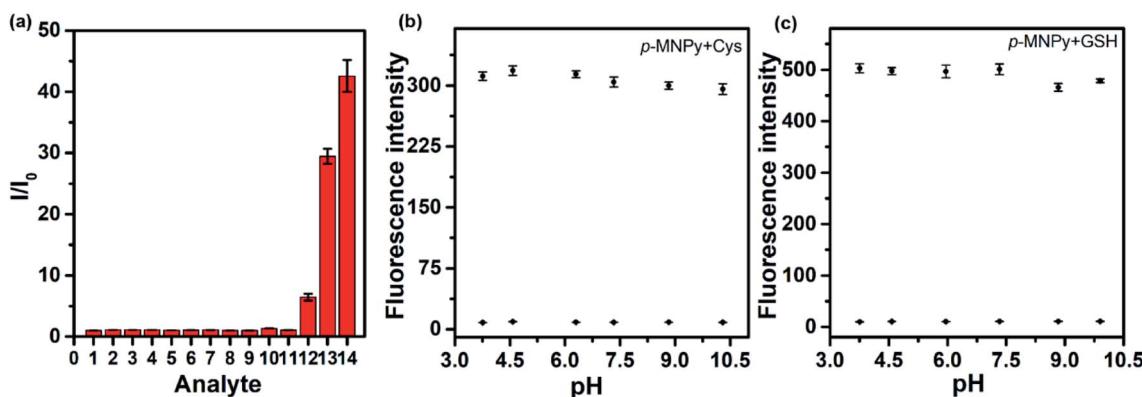


Fig. 4 (a) The fluorescence responses of *p*-MNPy (10 μ M) toward various bio-species. 1 control, 2 histidine, 3 glycine, 4 lysine, 5 proline, 6 glutamic acid, 7 glucose, 8 Na_2S , 9 Na_2SO_3 , 10 H_2O_2 , 11 NaClO, 12 Hcy, 13 Cys, 14 GSH. I_0 refers to the fluorescence intensity of the pure probe solution. The concentration of each analyte was 100 μ M, except that Hcy was 10 μ M and GSH was 5 mM. The pH-dependent fluorescence changes of *p*-MNPy toward (b) Cys and (c) GSH in PBS (containing 10% DMSO).

3.5 Selectivity and pH adaptability

Previous study had revealed that *p*-MNPy did not respond to 10 μ M of Hcy. To further confirm whether Hcy would interfere with the recognition process of Cys/GSH by the probe, the related absorption and fluorescence spectra of *p*-MNPy with Cys/GSH in the absence and presence of Hcy were recorded (Fig. S8†). As

expected, *p*-MNPy displayed satisfactory selectivity toward Cys/GSH among the biothiols. Then the responses of *p*-MNPy to other common bio-species were also investigated, such as some representative amino acids (histidine, glycine, lysine, proline, glutamic acid), glucose, strong nucleophile (Na_2S and Na_2SO_3), and reactive oxygen species (H_2O_2 and NaClO). As shown in Fig. 4a, S9 and S10,† only Cys and GSH could induce strong

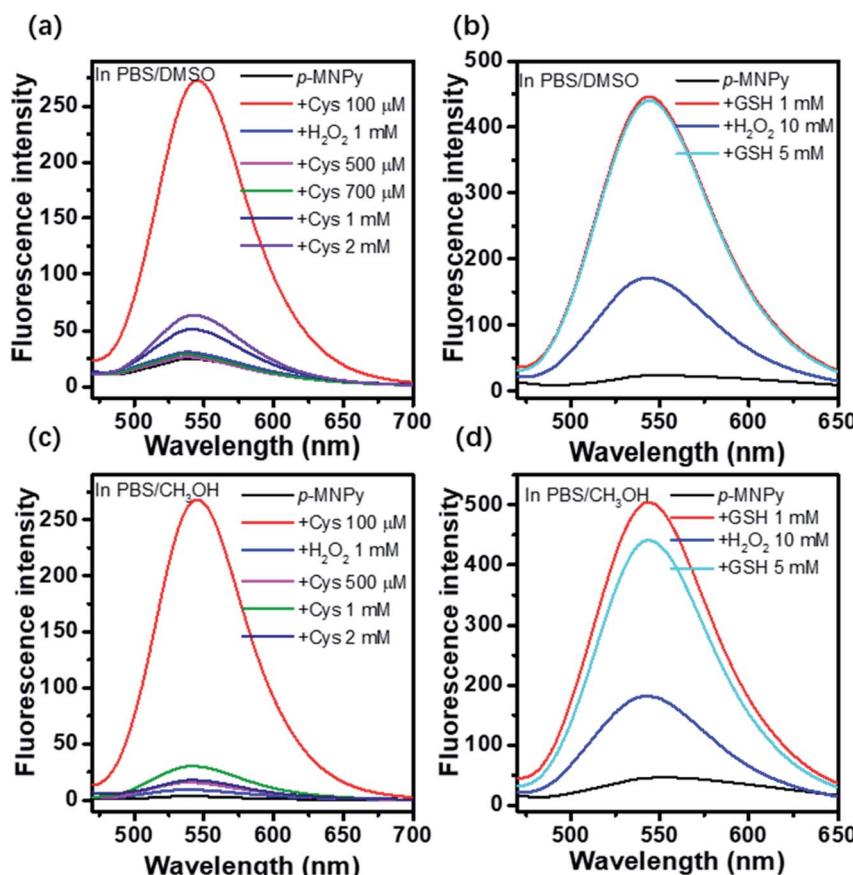


Fig. 5 The fluorescence changes of *p*-MNPy with the addition of (a) Cys or (b) GSH and H_2O_2 in PBS/DMSO (v/v, 9/1), and with the addition of (c) Cys or (d) GSH and H_2O_2 in PBS/ CH_3OH (v/v, 8/2).

fluorescence when each analyte was added to the probe separately, and the fluorescence of the probe with Cys or GSH did not change apparently in the presence of each analyte, which indicated that the probe **p-MNPy** was free from the interferences of the above bio-species. The sensing ability of **p-MNPy** toward Cys and GSH was further investigated in different pH environment (Fig. 4b, c and S11†). The reaction between biothiol and **p-MNPy** was supposed to be Michael addition, namely, the nucleophilic sulphydryl group of Cys or GSH would attack the carbon–carbon double bond in the Michael acceptor of **p-MNPy**. Normally, the nucleophilicity of sulphydryl group would reduce as the decrease of the pH of the solution. However, the sensing capacity of **p-MNPy** toward both Cys and GSH seemed to be independent with the environmental pH in the range of pH 3.5–10.5. We speculated that the reactivity of the Michael acceptor would be increased in acidic condition owing to the protonation of the pyridine unit, which may offset the inhibition of the reactivity of the sulphydryl group in low pH environment. In summary, the probe **p-MNPy** was appropriate for the application in cells with excellent pH tolerance.

3.6 Reversible recognition toward GSH

As the Michael addition has inherent reversibility, the reversible detection ability of **p-MNPy** toward Cys and GSH was studied. H_2O_2 , an intracellular reactive oxygen species, was utilized to induce the oxidative stress, which can oxidize GSH to its oxidative form (GSSG) and therefore decrease the content of GSH. As shown in Fig. 5a, Cys turned on the fluorescence of the probe in PBS/DMSO (v/v, 9/1), and excessive H_2O_2 could efficiently quench the fluorescence as expected. However, the continuous addition of Cys again could not relight up the probe,

even though much excessive amount of Cys was applied. In the contrast, the addition of GSH could turn on the fluorescence of the probe again after quenched by H_2O_2 (Fig. 5b). To exclude the effect of solvents, the reversibility of the probe was further studied in another solvent system, PBS/CH₃OH (v/v, 8/2). As shown in Fig. 5c and d, similar results were found as those in PBS/DMSO that **p-MNPy** still showed reversible recognition ability to GSH, but not to Cys. We supposed that these different response behaviors of the probe may be due to the higher reactivity of Cys to H_2O_2 , which might be confirmed by the fact that 10 eq. of H_2O_2 have quenched the fluorescence induced by Cys more efficiently than that induced by GSH. In addition, dilution experiments and HRMS were also employed to validate the reversible mechanism of the probe. As shown in Fig. S12,† further addition of GSH to the 10-fold diluted solution of **p-MNPy** + GSH made the absorbance at 390 nm and the fluorescence intensity at 542 nm obviously increased, and in Fig. S13 and S14† the fragment peaks of the adduct **p-MNPy**-GSH and the recovered **p-MNPy** by H_2O_2 could both be found in the HRMS. All these results manifested that **p-MNPy** could respond to GSH reversibly. At last, it may need to be noted that the effect of GSH and H_2O_2 on the probe was not proportionable, as 10 mM of H_2O_2 was necessary to offset the impact of 1 mM of GSH on **p-MNPy**, and only 5 mM of additional GSH could recover the fluorescence.

3.7 Fluorescence imaging in living cell

As the previous experiments displayed that **p-MNPy** showed the potential to monitor GSH reversibly, its sensing performance in living cells was further investigated. Before the live cell imaging, the biocompatibility of **p-MNPy** was first studied by MTT assay.

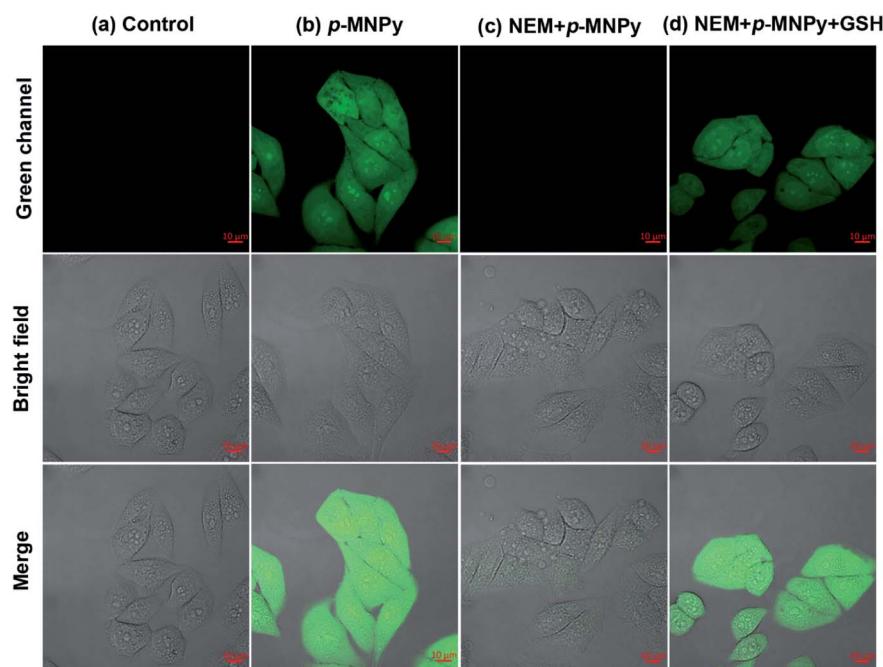


Fig. 6 The confocal images of live HeLa cells: (a) control group; (b) incubated with the probe **p-MNPy** for 30 min; (c) pretreated with NEM for 60 min, and then incubated with the probe for 30 min; (d) pretreated with NEM for 60 min, with the probe for 30 min, and then treated with GSH for 60 min. Concentration: **p-MNPy**, 10 μ M; NEM, 1 mM; GSH 1 mM. Excitation at 405 nm. Emission collected at 490–580 nm. Scale bar: 10 μ m.



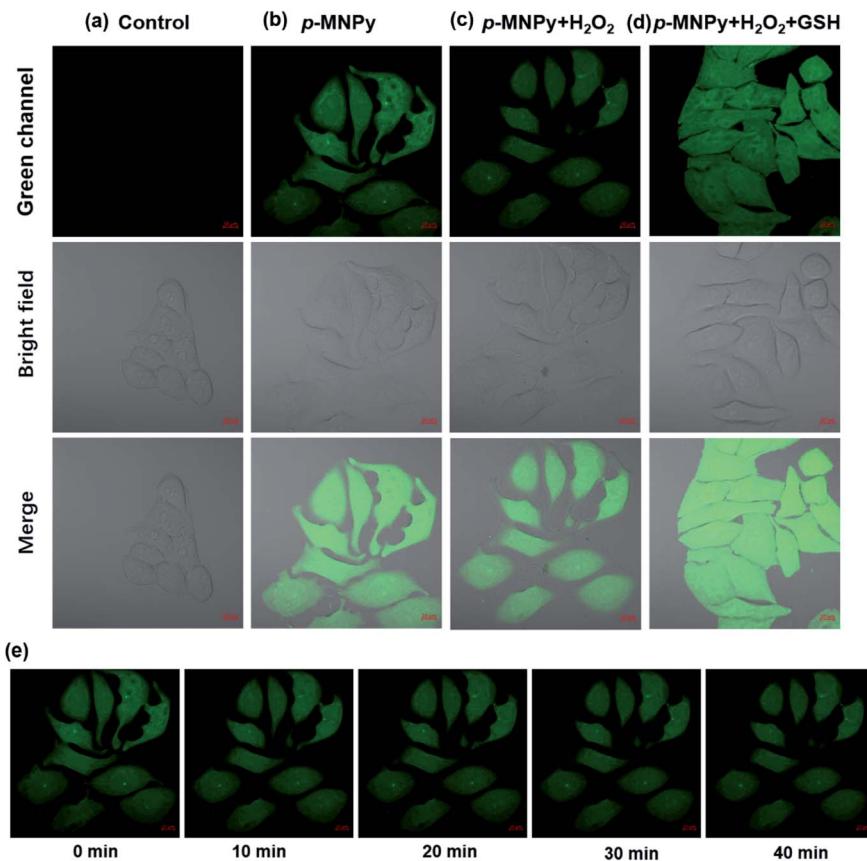


Fig. 7 The confocal images of live HeLa cells: (a) control group; (b) incubated with *p*-MNPY for 30 min; (c) incubated with *p*-MNPY for 30 min, and then treated with H₂O₂ for 40 min; (d) treated with *p*-MNPY for 30 min, with H₂O₂ for 40 min, and then with GSH for 60 min. (e) Real-time GSH imaging with *p*-MNPY in HeLa cells upon H₂O₂ treatment. Concentration: *p*-MNPY, 10 μ M; H₂O₂, 1 mM; GSH 1 mM. Excitation at 405 nm. Emission collected at 490–580 nm. Scale bar: 10 μ m.

As shown in Fig. S15,[†] HeLa cells were incubated with *p*-MNPY in different concentrations for 24 h. The cell viability was all above 90% when the probe concentration was in the range of 0–20 μ M. Considering 10 μ M of the probe was utilized in cell experiments, it was safe enough for the live cells. The confocal fluorescence images of the cells were shown in Fig. 6. When the cells were incubated with *p*-MNPY for 30 min, strong green fluorescence could be observed. However, almost no fluorescence was found when the cells were pretreated with NEM (*N*-ethyl maleimide) to scavenge the intracellular biotools, and then incubated with *p*-MNPY, which manifested that the green fluorescence induced by *p*-MNPY was probably ascribed to the selective interactions with biotools. And when this group was subsequently treated with GSH, the green fluorescence recovered markedly. It was now almost certain that the probe *p*-MNPY could specifically respond to GSH. To study the reversibility of the probe, the cells were stimulated by H₂O₂ to change the levels of intracellular GSH. As shown in Fig. 7, upon H₂O₂ treatment, the fluorescence images of the cells pretreated by *p*-MNPY became noticeably darker over time, while the brightness could restore again when the cells were treated with additional GSH. This revealed that the probe successfully sensed the fluctuations of the GSH. In a word, *p*-MNPY could image the GSH selectively and reversibly in living cells.

4 Conclusion

In conclusion, three probes, *o*-MNPY, *m*-MNPY and *p*-MNPY, were designed and synthesized with minor structural differences in the Michael acceptor through the regulation of the connection mode of the pyridine group. The response performances of these probes toward the biotools were evaluated taking the true intracellular biotool contents into account, and *p*-MNPY showed a better comprehensive performance with higher fluorescence intensity enhancement and faster recognition speed than the other two, demonstrating the advantages of the *p*-connection mode of pyridine group that could afford the probe both higher reactivity and better fluorescence properties. Further studies showed *p*-MNPY revealed good selectivity, satisfactory sensitivity, and excellent pH adaptability during the detection of Cys and GSH. What's more, it could respond to GSH reversibly. Finally, the probe *p*-MNPY was utilized to image the endogenous GSH and track the GSH fluctuations under the oxidative stress in living cells.

Conflicts of interest

There are no conflicts of interest to declare.

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