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# An investigation of broadband optical nonlinear absorption and transient nonlinear refraction in a fluorenone-based compound†

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A novel fluorenone derivative, FO52, is designed and synthesized. The fluorenone group is introduced to provide the central  $\pi$ -conjugated system in the molecule and triphenylamine is substituted at both sides. Intramolecular Charge Transfer (ICT) from the terminal groups to the molecular center is confirmed *via* DFT calculations. Ultrafast optical nonlinearities are investigated *via* Z-scan and transient absorption spectroscopy (TAS) studies with a 190 fs laser. Reverse saturable absorption, two-photon induced excited-state absorption, and pure two-photon absorption are observed at 532 nm, 650 nm, and 800 nm, respectively. The different mechanisms at these wavelengths are discussed and interpreted with assistance from the results from TAS. Furthermore, strong excited-state refraction and ultrafast negative refraction from the bound electron response are resolved and discussed in phase object pump probe (POPP) experiments. The results suggest that the ICT-enhanced optical nonlinearities provide FO52 with strong optical limiting capabilities at visible wavelengths and ultrafast refraction with tiny attenuation in the near infrared region. The combination of these properties in one compound could be attractive for applications like laser protection and low-loss all-optical switching.

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## 1. Introduction

Research into nonlinear optics began decades ago after the invention of laser in 1960s. Various applications such as harmonic generation, optical limiting and all optical switching (AOS) have been developed.<sup>1,2</sup> To realize these applications, nonlinear optical (NLO) materials are indispensable. Among all the NLO materials, organic materials have attracted substantial attention due to their large molecular optical nonlinearity, relatively easy synthesis and tailoring.<sup>3</sup> To optimize the NLO response in organic NLO materials, one feasible way is to introduce electron donor or acceptor groups into the molecule as the intramolecular charge transfer (ICT) could enhance optical nonlinearity in numerous organic molecular systems.<sup>4</sup> The relationship between ICT and molecular optical nonlinearity has been studied in many  $\pi$ -conjugated systems such as phthalocyanines,<sup>5</sup> pyrene derivatives,<sup>6–10</sup> thiophene derivatives<sup>11–13</sup> and benzothiadiazole derivatives.<sup>14–16</sup>

Fluorenone, as an aromatic organic group, is usually employed as an electron acceptor in ICT molecules. Due to their high photostability and bright fluorescence,<sup>17–19</sup> fluorenone dyes have been synthesized and studied as probes for cell imaging.<sup>20,21</sup> Besides, the enhanced optical nonlinearity *via* ICT provides fluorenone-based compounds with a superior multi-photon response, which further extends the excited wavelengths towards the infrared region. Taking advantage of this property, fluorenone derivatives display great potential in applications like multi-photon microscopy<sup>21–24</sup> and second harmonic generation (SHG).<sup>25,26</sup> However, although multi-photon absorption<sup>27,28</sup> in fluorenone derivatives has been reported many times, systematic investigations into excited-state associated nonlinear absorption are rare.<sup>29–31</sup> Furthermore, most of the research focuses on nonlinear absorption (NLA), little attention has been paid to the ultrafast nonlinear refraction (NLR) in fluorenone derivatives.

Here in this work, a novel NLO compound FO52 is designed and synthesized. Fluorenone is employed as the core of the molecule with triphenylamine linked at both sides (Fig. 1). The ultrafast nonlinear absorption and refraction at various wavelengths are systematically studied. The magnitudes of NLA & NLR are estimated and the ultrafast mechanisms of NLA & NLR are investigated *via* transient absorption and refraction experiments. The combined research of NLA and NLR could help us gain a more comprehensive understanding of the excited-state optical nonlinearity in fluorenone derivatives and the additional information

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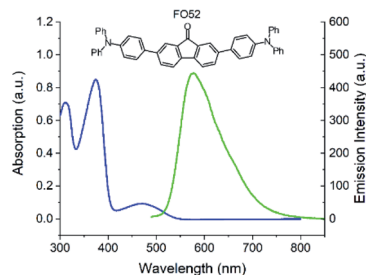


Fig. 1 UV-vis absorption (blue) and fluorescence (green) spectra of FO52 solution.

from NLR could provide a different perspective to consider the structure–property relationship in these compounds.

## 2. Experimental section

### 2.1 Synthesis and characterization

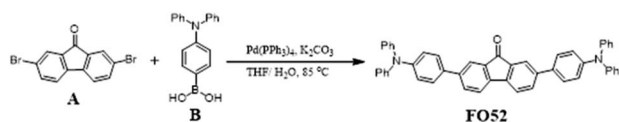
The synthetic procedure is depicted in Scheme 1.

**A** (338 mg, 1 mmol), **B** (723 mg, 2.5 mmol),  $K_2CO_3$  (1.38 g, 10 mmol) and THF/ $H_2O$  ( $v/v = 20$  mL/10 mL) were added into a 120 mL pressure flask in order.  $Pd(PPh_3)_4$  (81 mg, 0.07 mmol) was added after filling with nitrogen (5 min) and the reaction proceeded at 85 °C for 16 h. After that, saturated salt solution and dichloromethane were used for extraction. All solvent was removed through decompression concentration, purification was carried out with a silica gel column with  $V(\text{petroleum ether}) : V(\text{dichloromethane}) = 4 : 1$  used as the eluent. A red solid product was obtained (502 mg, 76%). The  $^1H$  NMR,  $^{13}C$  NMR and MS characterization can be found in the ESI.†

**FO52**:  $^1H$  NMR (600 MHz,  $CDCl_3$ , 298 K),  $\delta$ : 7.84 (d, 2H,  $J = 1.2$  Hz), 7.65–7.63 (dd, 2H,  $J = 7.8$  Hz), 7.49–7.48 (d, 2H,  $J = 7.8$  Hz), 7.46–7.45 (d, 4H,  $J = 9.0$  Hz), 7.25–7.23 (m, 8H), 7.11–7.10 (m, 12H), 7.03–7.01 (t, 4H);  $^{13}C$  NMR (150 MHz,  $CDCl_3$ , 298 K),  $\delta = 194.06, 147.92, 147.61, 142.72, 141.61, 135.35, 133.49, 132.67, 129.48, 127.53, 124.79, 123.62, 123.35, 122.52, 120.77$ . MS (MALDI-TOF,  $m/z$ ): Calc. for  $C_{49}H_{34}N_2$ : [ $m/z$ ] 666.27, found: [ $m/z$ ] 666.27. HR-MS ( $m/z$ ): Calc. for  $C_{49}H_{34}N_2$ : [ $m/z$ ] 666.2671, found: [ $m/z$ ] 666.2662.

### 2.2 Nonlinear optical experiments

**Z-scan experiment.** The 190 fs, wavelength tunable laser pulses used in the Z-scan<sup>32</sup> experiment are extracted from an optical parametric amplifier (OPA) (ORPHEUS, Light Conversion), which is pumped by a femtosecond fiber laser system (Pharos, Light Conversion). The input laser is focused into the sample *via* a 250 mm lens. The beam radius at focus is measured to be 25  $\mu\text{m}$ . The sample solution ( $1.5 \times 10^{-3}$  M) contained in a 2 mm cuvette is placed on a translation stage, which is controlled by a program and moves along the z-



Scheme 1 The synthesis of FO52.

direction. Experimental data is recorded by energy detectors (RjP-765a, Laser Probe Inc.)

**Transient absorption spectroscopy.** The pump beam in TAS comes from the same OPA in the Z-scan experiment. The output pulses have a 190 fs pulse-width and the wavelength is continuously tunable in a wide spectral range. White light is generated after focusing part of the 1030 laser into a sapphire crystal and is used as a probe beam. A figure showing the experimental apparatus in detail can be found in the ESI.† In our experiment, the wavelength of the pump beam is set to 400 nm. The sample solution is contained in a 2 mm cuvette.

**Phase object pump–probe.** The ultrafast dynamics of NLR are investigated by the phase object pump–probe (POPP) method, the details can be found elsewhere.<sup>33</sup> The POPP technique is a combination of the conventional pump probe and the 4f phase object imaging system, which is capable of investigating the sign and magnitude of NLA and NLR simultaneously. The output laser of OPA is employed as a pump beam, while a part of the 1030 nm laser is separated in front of the OPA as a probe beam (fundamental or doubled frequency).

**Nanosecond optical limiting.** The fluence (intensity)-dependence transmittance of the sample is recorded. A 2 mm cuvette is employed to contain the sample solution and is placed at the focal spot. The concentration of the sample solution is  $0.75 \times 10^{-3}$  M to guarantee 50% transmittance. The fluence (intensity) of the input laser is attenuated by a rotating variable ND filter. The laser source is a Q-switched YAG system (Surelite II, Continuum) which provides 532 nm, 4 ns pulses.

## 3. Results and discussion

### 3.1 UV-vis absorption and fluorescence

The UV-Vis absorption and fluorescence spectra of the FO52 toluene solution ( $\sim 10^{-5}$  M) are recorded and displayed in Fig. 1. Several absorption peaks can be found in the FO52 solution, as well as the absorption peaks at 310 nm and 374 nm, there is a relatively weak absorption peak at 472 nm, which dominates the linear absorption in the visible region. The emission spectrum of the FO52 solution covers a wide spectral range from around 500 nm to 800 nm, the peak wavelength is located at 577 nm.

To obtain more information about the linear absorption and fluorescence of FO52, DFT calculations are conducted. The

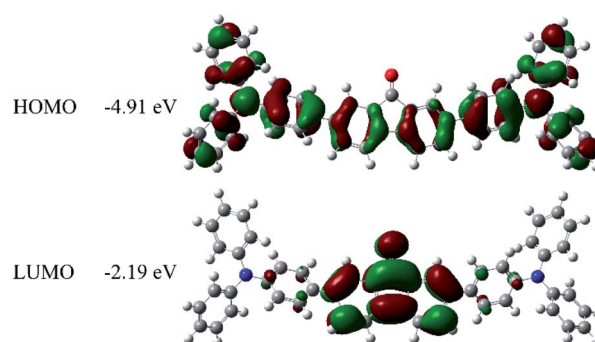


Fig. 2 Electronic distributions and energies of the frontier molecular orbitals in FO52 extracted from DFT calculations.



distribution of the frontier molecular orbitals and their corresponding energies are obtained and displayed in Fig. 2. One can find that there is an evident charge transfer towards the fluorenone core of the molecule upon excitation. The calculated energy gap between the HOMO and LUMO is 2.72 eV (456 nm), which is close to the first absorption peak of the FO52 solution at 472 nm. The results indicate that the absorption of FO52 at visible wavelengths has a significant ICT feature.

### 3.2 Z-scan experiments

To estimate the intensity-dependent NLA and NLR of FO52, femtosecond Z-scans are conducted at 532 nm, 650 nm and 800 nm, peak intensities are measured to be 24, 66 and 75 GW cm<sup>-2</sup>, respectively. The linear transmittance measured at these wavelengths are 0.43, 0.99 and 0.99, respectively. The open aperture (OA) and closed aperture (CA) curves are recorded and displayed in Fig. 3. When excited with 532 nm laser pulses, the OA Z-scan curve shows a single valley at the zero position (Fig. 3(a)), indicating that FO52 has strong reverse saturable absorption. Besides, the CA Z-scan in Fig. 3(b) displays a distinct peak-valley shape after removal of the solvent background, which suggests a negative change of the refractive index when the sample passes through the focal spot towards the +z direction and is often referred to as self-defocusing. Note that there is a significant transmittance change between the transmittance peak and valley ( $\Delta T_p - \Delta T_v = 0.4$ ), which indicates a large negative NLR in FO52. Considering that 532 nm is located within the first linear absorption band of the compound solution (Fig. 1), the reverse saturable absorption and negative nonlinear refraction under resonant excitation could both originate from the excited-state optical nonlinearity. When the incident laser wavelength is tuned away from the resonant absorption band, FO52 shows an extremely high transmittance (0.99) at 650 nm and 800 nm. The OA Z-scan records a single valley curve at both 650 nm and 800 nm as shown in Fig. 3(c)

and (d). Unlike the 532 nm situation, NLA at 650 nm and 800 nm cannot originate from the one photon induced excited-state absorption, hence we assign the NLA to the multi-photon absorption. Moreover, no sign of NLR could be found after careful comparison between the solution and solvent samples at 650 nm and 800 nm, which is different from the large negative NLR at 532 nm. Furthermore, the transmittance valley at 650 nm in the OA Z-scan is much narrower compared with the 532 nm and 800 nm cases. This narrowing of the OA Z-scan curve is further confirmed by comparing the sample solution with a ZnSe sample in the same Z-scan. This situation suggests the existence of a higher order NLA in FO52 at 650 nm.

Numerical simulation based on Sheik Bahae's theory is used to analyse the Z-scan results. The absorption coefficient  $\alpha$  and refractive index  $n$  could be expanded as:

$$\alpha = \alpha_0 + \beta I + (\gamma I^2)$$

$$n = n_0 + n_2 I$$

where  $I$  represents the laser intensity,  $\alpha_0$  is the linear absorption coefficient,  $\beta$  is the third-order absorption coefficient, and an additional fifth-order absorption coefficient  $\gamma$  is considered for the 650 nm case in our work. Analogously,  $n_0$  is the sample's linear refractive index and  $n_2$  represents the third-order refractive index. The output transmittance at each position is extracted after the entire simulation of the Z-scan. The results are listed in Table 1 and the simulated transmittance curves are plotted in Fig. 3 with solid and dashed lines.

One can find that FO52 shows the largest NLA and NLR at 532 nm in our experiments. With the increasing of laser wavelength, the NLA decreases while the NLR is covered by the strong solvent background. As discussed above, the optical nonlinearity of FO52 at 532 nm is dominated by the one photon induced excited-state absorption and refraction. On the other side, the NLA at 650 nm and 800 nm are assigned to multi-photon absorption since the one-photon absorption could be neglected in such a high transmittance. Although both originate from multi-photon absorption, it is interesting to find that the OA curve at 650 nm is much narrower than the 532 nm and 800 nm case. Numerical simulation using third-order NLA cannot fit the OA data at 650 nm (dashed line in Fig. 3(c)). An excellent matching is obtained after the replacing of the third-order term ( $\beta I$ ) with a fifth-order term ( $\gamma I^2$ ) in eqn (1). Considering that the possibility of simultaneous three-photon absorption is relatively low, the fifth-order NLA at 650 nm may originate from the two-photon induced excited-state

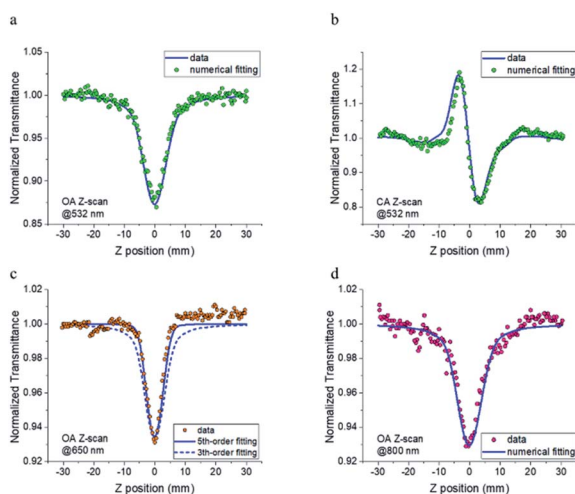


Fig. 3 Femtosecond Z-scan measurements of FO52. (a) Open aperture and (b) closed aperture results at 532 nm, and open aperture results at (c) 650 nm and (d) 800 nm. The solid and dashed lines represent the results of the numerical fittings.

Table 1 The nonlinear optical parameters used for the numerical simulations in the Z-scan studies

	532 nm, $T = 0.43$	650 nm, $T = 0.99$	800 nm, $T = 0.99$
<b>FO52</b>	$\beta = 1.1 \times 10^{-12}$ m W <sup>-1</sup>	$\beta = 0$	$\beta = 1.5 \times 10^{-13}$ m W <sup>-1</sup>
	$n_2 = -1.2 \times 10^{-19}$ m <sup>2</sup> W <sup>-1</sup>	$\gamma = 3.5 \times 10^{-28}$ m <sup>3</sup> W <sup>-2</sup>	$\sigma_{\text{TPA}} = 412 \text{ GM}$



**Table 2** The nonlinear optical parameters compared with some reported fluorenone derivatives

Sample	$\lambda$ (nm)	$\sigma_{\text{TPA}}$ (GM)	Ref.
<b>FO52</b>	800	412	This work
F-benzothiophene	710	220	34
TPA-F-FO-TPA	800	1759	27
FOVPC6	800	495	35

absorption. Recalling the femtosecond laser pulses used in the Z-scan experiment, an excited-state with ultrafast established time and large absorption cross-section is required in such a case. As for the 800 nm case, the OA data could be well fitted by a third-order NLA, indicating that two-photon absorption dominates the NLA at 800 nm. Two-photon absorption is observed in the Z-scan at both 650 nm and 800 nm, however, the promoted electrons on the excited state are likely to absorb another 650 nm photon rather than an 800 nm photon. A much stronger excited-state absorption is expected at 650 nm than 800 nm in this case, which would be investigated *via* transient absorption.

The NLO coefficients obtained in the Z-scan experiment are compared with some of the reported fluorenone derivatives in Table 2. Compared with other compounds, **FO52** displays significant nonlinear refraction and a relatively large absorption cross-section for two-photon absorption. The nonlinear absorption in **FO52** is considered to be enhanced by the symmetric  $\pi$ -conjugated system while the nonlinear refraction arises from the significant ICT between the HOMO and LUMO.

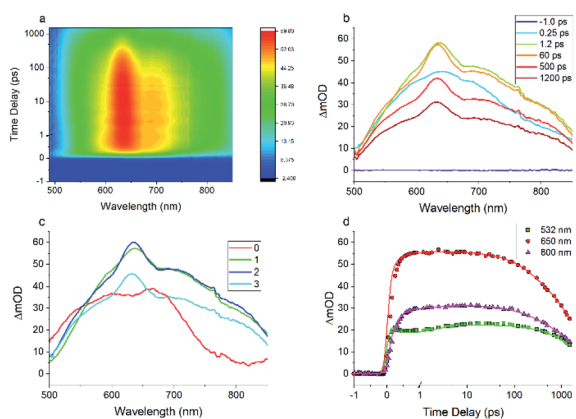
### 3.3 Transient absorption and refraction

The results of the Z-scan indicate that the excited-state plays a key role in the optical nonlinearity of **FO52**. Besides, as a single beam measuring technique, the Z-scan could only measure the NLA and NLR within a pulse width. To obtain more information about the ultrafast dynamics of the excited-state

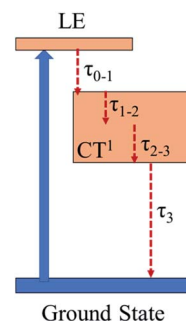
optical nonlinearity, investigation into the transient absorption and refraction is conducted.

The entire transient absorption spectroscopy (TAS) spectrum ranging from 500 nm to 850 nm is recorded and displayed in a 2D colormap in Fig. 4(a). For more detail, TAS spectra at several selected delay times are extracted and plotted in Fig. 4(b). One can find that the excited-state absorption could be built within a pulse width (<250 fs) and lasts for several nanoseconds. Besides, broadband reverse saturable absorption ( $\Delta m_{\text{OD}} > 0$ ) covering wavelengths from 500 nm to 850 nm is observed in the entire delay time window. The recorded ultrafast broadband excited-state absorption in TAS agrees with the measured reverse saturable absorption in the Z-scan at 532 nm. Moreover, one can find that the TAS maintains a similar shape from 0.5 ps to 1.8 ns in Fig. 4(a) and (b), this dominant excited-state absorption in **FO52** is assigned to a charge transfer singlet state.

Global and target analysis is employed to numerically reconstruct the TAS data (details of the numerical simulation could be found in the ESI†). To obtain a reasonable fit, four spectral components in total are used and labelled from 0 to 3 in Fig. 4(c) and the results of the numerical simulation at selected wavelengths are plotted in Fig. 4(d). Note that the extracted absorption spectroscopy of the four components can provide information about the evolution of TAS, component 0 represents the ultrafast absorption within a pulse duration while component 3 represents the absorption of the last excited state before the electrons relax back to ground state. Considering the 0 component and the results from the femtosecond Z-scan, reverse saturable absorption could be found at both 532 nm and 650 nm while the absorption at 650 nm is much stronger. The large absorption cross-section at 650 nm and the ultrafast establishing time agrees with the two-photon induced excited-state absorption at 650 nm in the Z-scan results. Furthermore, little absorption could be found in the 0 component at 800 nm, which could be the reason that the two-photon absorption at 800 nm cannot trigger subsequent excited-state absorption as in the 650 nm case. Besides, the lifetimes for the spectral components 0 to 3 are extracted from the simulation as 0.2 ps, 2.1 ps, 240 ps and 2500 ps, respectively (Fig. 5). The 0.2 ps lifetime is assigned to the conversion time from a locally excited (LE) state to the charge transfer state. Since all the 1, 2, and 3 components share a similar spectrum and hence all originate from the charge transfer state, 2.1 ps and 240 ps are



**Fig. 4** Results of transient absorption spectroscopy (TAS) studies of **FO52** solution. (a) A 2D color map of the TAS results; (b) TAS at various delay times; (c) spectral components extracted from global and target analysis, providing information about the evolution of TAS; and (d) kinetic traces with numerical fitting at various wavelengths.



**Fig. 5** An energy level diagram for the ultrafast dynamics in **FO52**.





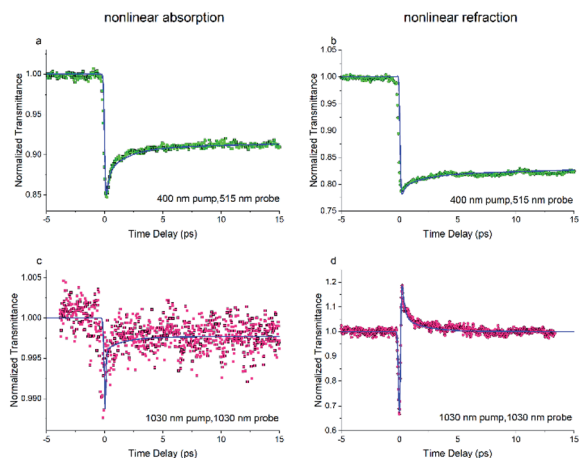


Fig. 6 The transient ((a), (c)) nonlinear absorption and ((b), (d)) nonlinear refraction of FO52 probed at 515 nm and 1030 nm. Solid lines represent the results of the numerical simulations.

assigned to the vibrational cooling processes inside the charge transfer state and 2500 ps is assigned to its lifetime.

The dynamics of NLR are investigated *via* the phase object pump-probe (POPP) technique. The results of the POPP are shown in Fig. 5 when the sample solution is pumped at 400 nm and probed at 515 nm. The dynamics of NLA in Fig. 6(a) display a strong reverse saturable absorption at zero time, after that the NLA decreases with a lifetime of 2 ps and then maintains for a relatively long lifetime, the probed dynamics of NLA agrees with the above results in Z-scan and TAS experiments. Correspondingly, the dynamics of NLR in Fig. 6(b) show a significant drop at zero time and then maintains for a long time. The results indicate the strong negative excited-state NLR in FO52, which is consistent with the above discussion.

Since we are more interested in the optical nonlinearity of FO52 in the near infrared, the pump and probe beams are then switched to 1030 nm. The dynamics of NLA and NLR are recorded and displayed in Fig. 6(c) and (d), respectively. The normalized transmittance in Fig. 5(c) drops swiftly at zero time and then is maintained at 0.996 after a fast relaxation. Analogously, the ultrafast absorption at zero time is assigned to the two-photon absorption while the absorption with a relatively long lifetime is assigned to the two-photon induced excited-state absorption. The energy of two 1030 nm photons equals that of a 515 nm photon, while 515 nm is located at the edge of

the first absorption band (Fig. 1), hence the two-photon absorption at 1030 nm is very weak. Moreover, the weak two-photon absorption results in fewer excitons in the excited-state, which further suppresses the excited-state absorption. One can find that the maximum change of the normalized transmittance in Fig. 5(c) is about 0.01, indicating that both the two-photon absorption and excited-state absorption are extremely weak at 1030 nm. In contrast, a strong NLR is found in the same experiment and the result is shown in Fig. 6(d). A dramatic drop of transmittance at zero time indicates the strong ultrafast negative NLR in FO52. When the impact of the pump pulse finishes, the NLR experiences an ultrafast switching of sign (from negative to positive) and this positive NLR disappears after a relaxation in about 5 ps. The ultrafast negative NLR in a pulse width is assigned to the response of bound electrons while the positive NLR is assigned to the excited-state refraction. The modulation in intensity ( $I_p$ ) and phase ( $\varphi_p$ ) of the probe beam can be represented as:<sup>36</sup>

$$\frac{dI_p}{dz} = -\left(\sum_{n \geq 1} \sigma_{p,S_n} N_{S_n} + 2\beta I_e\right) I_p$$

$$\frac{d\varphi_p}{dz} = k \left(\sum_{n \geq 1} \Delta\eta_{p,S_n} N_{S_n} + 2n_2 I_e\right)$$

where  $z$  is the propagation length within the sample,  $N_{S_n}$  represents the population density of the effective energy state  $S_n$ ,  $\sigma_{p,S_n}$  is the absorptive cross section of the excited-state  $S_n$ ,  $\beta$  here is the two-photon absorption coefficient, and  $I_e$  and  $I_p$  stand for the intensity of the pump and probe beam, respectively.  $n_2$  is the optical nonlinear refractive index of the bound electron response and  $\Delta\eta_{p,S_n}$  (defined as  $\eta_{p,S_n} - \eta_{p,S_0}$ ) is the change of the refractive volume between the effective state  $S_n$  and ground state  $S_0$ . Like the discussion in the TAS experiment, we use  $S_0$ ,  $S_1$  and  $S_2$  to represent the ground state, the charge transfer state and the locally excited (LE) state, respectively. Note that the lifetime of the first excited state (2500 ps as measured in TAS) cannot be resolved here due to the limited delay time. After a series of numerical simulations,  $\beta = 0.4 \times 10^{-13} \text{ m W}^{-1}$  and  $n_2 = -5 \times 10^{-19} \text{ m}^2 \text{ W}^{-1}$  are finally obtained. The details of the rate equations can be found in the ESI† and the parameters are listed in Table 3.

Table 3 Parameters extracted from the numerical simulations of POPP results

$n$	400 pump, 515 probe		1030 pump, 1030 probe		$\tau_n$ (ps)
	$\sigma_{p,S_n}$ ( $\text{m}^2$ )	$\Delta\eta_{p,S_n}$ ( $\text{m}^3$ )	$\sigma_{p,S_n}$ ( $\text{m}^2$ )	$\Delta\eta_{p,S_n}$ ( $\text{m}^3$ )	
0	$7.4 \times 10^{-23}$	0	$5.5 \times 10^{-24}$	0	—
1	$7.7 \times 10^{-21}$	$-2.4 \times 10^{-21}$	$5 \times 10^{-21}$	0	250
2	$7.7 \times 10^{-21}$	$-2.6 \times 10^{-21}$	$8 \times 10^{-21}$	$0.8 \times 10^{-19}$	1.8
3	$8.5 \times 10^{-21}$	$-2.9 \times 10^{-21}$	$10 \times 10^{-21}$	$6 \times 10^{-19}$	0.2
$\beta$	0		$0.4 \times 10^{-13} \text{ m W}^{-1}$		
$n_2$	0		$-5 \times 10^{-19} \text{ m}^2 \text{ W}^{-1}$		



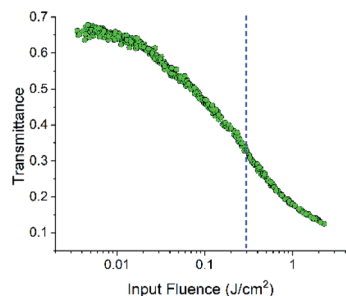


Fig. 7 Transmittance variations in the optical limiting measurements; the dashed line indicates the position of the threshold value.

### 3.4 Optical limiting

The strong transient NLR and tiny NLA observed in the infrared region make **FO52** an ideal compound for all optical applications in which ultrafast modulation of light and low attenuation of signals are required. Furthermore, the strong excited-state NLA and NLR in the visible region with a relatively long lifetime are also attractive for NLO applications with picosecond or nanosecond laser pulses since more excitons could accumulate in the excited-state. Hence the optical limiting property of **FO52** is investigated under a 4 ns, 532 nm laser source and displayed in Fig. 7.

A significant decrease in the transmittance with the increasing input fluence could be observed in **FO52**. The threshold of optical limiting (defined as the corresponding input fluence when the transmittance decreases to half of the linear transmittance) reaches  $0.3 \text{ J cm}^{-2}$  as displayed in Fig. 7. With the further increase of the input fluence, a major part of the input fluence is absorbed. The results of the optical limiting test indicate that **FO52** is a promising material for laser protection, which further enriches the applications of this fluorenone based compound.

## 4. Conclusions

A novel fluorenone compound **FO52** is synthesized and the ultrafast optical nonlinearities are investigated. Strong reverse saturable absorption with a negative nonlinear refraction at 532 nm, two-photon induced excited-state absorption at 650 nm, and pure two-photon absorption at 800 nm are characterized with the Z-scan technique. The strong nonlinear refraction and multi-photon absorption are assigned to significant ICT and the symmetric  $\pi$ -conjugated system in **FO52**, which are revealed *via* DFT calculations. The transient absorption in the excited state is then investigated with TAS; a charge-transfer state with broadband excited-state absorption and a nanosecond-scale lifetime is observed. The ultrafast dynamics are then resolved *via* global and target analysis. The different nonlinear refraction responses originating from the excited state and the bound electron response are distinguished in POPP experiments. The strong optical nonlinearities enhanced by ICT and the high-contrast ultrafast nonlinear refraction in **FO52** make it a potential material for laser protection in the visible region and low-loss optical applications at near-infrared wavelengths.

## Conflicts of interest

There are no conflicts to declare.

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## References

- 1 R. W. Boyd, *Nonlinear optics*, Academic Press, 2020.
- 2 B. Gu, C. Zhao, A. Baev, K.-T. Yong, S. Wen and P. N. Prasad, *Adv. Opt. Photonics*, 2016, **8**, 328–369.
- 3 H. S. Nalwa, *Adv. Mater.*, 1993, **5**, 341–358.
- 4 D. Dini, M. J. F. Calvete and M. Hanack, *Chem. Rev.*, 2016, **116**, 13043–13233.
- 5 G. de la Torre, P. Vaquez, F. Agullo-Lopez and T. Torres, *Chem. Rev.*, 2004, **104**, 3723–3750.
- 6 J. Sung, P. Kim, Y. O. Lee, J. S. Kim and D. Kim, *J. Phys. Chem. Lett.*, 2011, **2**, 818–823.
- 7 M. Planells, M. Pizzotti, G. S. Nichol, F. Tessoro and N. Robertson, *Phys. Chem. Chem. Phys.*, 2014, **16**, 23404–23411.
- 8 X. Wu, J. Xiao, R. Sun, J. Jia, J. Yang, G. Shi, Y. Wang, X. Zhang and Y. Song, *Dyes Pigm.*, 2017, **143**, 165–172.
- 9 J. Jia, X. Wu, Y. Fang, J. Yang, X. Guo, Q. Xu, Y. Han and Y. Song, *J. Phys. Chem. C*, 2018, **122**, 16234–16241.
- 10 J. Wei, Y. Li, P. Song, Y. Yang and F. Ma, *Spectrochim. Acta, Part A*, 2021, **245**, 118897.
- 11 T. Jaunet-Lahary, A. Chantzis, K. J. Chen, A. D. Laurent and D. Jacquemin, *J. Phys. Chem. C*, 2014, **118**, 28831–28841.
- 12 N. B. Teran, G. S. He, A. Baev, Y. Shi, M. T. Swihart, P. N. Prasad, T. J. Marks and J. R. Reynolds, *J. Am. Chem. Soc.*, 2016, **138**, 6975–6984.
- 13 J. Jia, X. Zhang, Y. Wang, Y. Shi, J. Sun, J. Yang and Y. Song, *J. Phys. Chem. A*, 2020, **124**, 10808–10816.
- 14 J. E. Barnsley, G. E. Shillito, C. B. Larsen, H. van der Salm, L. E. Wang, N. T. Lucas and K. C. Gordon, *J. Phys. Chem. A*, 2016, **120**, 1853–1866.
- 15 K. D. Belfield, M. V. Bondar, S. Yao, I. A. Mikhailov, V. S. Polikanov and O. V. Przhonska, *J. Phys. Chem. C*, 2014, **118**, 13790–13800.
- 16 X. Wu, J. Xiao, R. Sun, J. Jia, J. Yang, G. Ao, G. Shi, Y. Wang, X. Zhang and Y. Song, *Opt. Laser Technol.*, 2018, **102**, 93–99.
- 17 F. Xu, H. Wang, X. Du, W. Wang, D.-E. Wang, S. Chen, X. Han, N. Li, M.-S. Yuan and J. Wang, *Dyes Pigm.*, 2016, **129**, 121–128.
- 18 F. Lincker, B. Heinrich, R. De Bettignies, P. Rannou, J. Pécaut, B. Grévin, A. Pron, B. Donnio and R. Demadrille, *J. Mater. Chem.*, 2011, **21**, 5238–5247.
- 19 L. A. Estrada and D. C. Neckers, *J. Org. Chem.*, 2009, **74**, 8484–8487.



- 20 A. L. Capodilupo, V. Vergaro, G. Accorsi, E. Fabiano, F. Baldassarre, G. A. Corrente, G. Gigli and G. Ciccarella, *Tetrahedron*, 2016, **72**, 2920–2928.
- 21 S. Chi, L. Li and Y. Wu, *J. Phys. Chem. C*, 2016, **120**, 13706–13715.
- 22 J. Liu, G. Li and Y. Wang, *J. Phys. Chem. A*, 2012, **116**, 7445–7451.
- 23 A. L. Capodilupo, V. Vergaro, E. Fabiano, M. De Giorgi, F. Baldassarre, A. Cardone, A. Maggiore, V. Maiorano, D. Sanvitto, G. Gigli and G. Ciccarella, *J. Mater. Chem. B*, 2015, **3**, 3315–3323.
- 24 W. Zhang, W. Liu, P. Li, J. kang, J. Wang, H. Wang and B. Tang, *Chem. Commun.*, 2015, **51**, 10150–10153.
- 25 J. Xu, S. Semin, D. Niedzialek, P. H. J. Kouwer, E. Fron, E. Coutino, M. Savoini, Y. Li, J. Hofkens, H. Uji-I, D. Beljonne, T. Rasing and A. E. Rowan, *Adv. Mater.*, 2013, **25**, 2084–2089.
- 26 X. Jialiang, S. Sergey, C. Jonathan, W. Linjun, S. Matteo, F. Eduard, C. Eduardo, C. Thibault, W. Chunliang, L. Yongjun, L. Huibiao, L. Yuliang, T. Paul, P. H. J. Kouwer, T. W. Ebbesen, H. Johan, B. David, A. E. Rowan and R. Theo, *Adv. Opt. Mater.*, 2015, **3**, 948–956.
- 27 T. H. Huang, X. C. Li, Y. H. Wang, Z. H. Kang, R. Lu, E. L. Miao, F. Wang, G. W. Wang and H. Z. Zhang, *Opt. Mater.*, 2013, **35**, 1373–1377.
- 28 T. H. Huang, D. Yang, Z. H. Kang, E. L. Miao, R. Lu, H. P. Zhou, F. Wang, G. W. Wang, P. F. Cheng, Y. H. Wang and H. Z. Zhang, *Opt. Mater.*, 2013, **35**, 467–471.
- 29 L. A. Estrada, J. E. Yarnell and D. C. Neckers, *J. Phys. Chem. A*, 2011, **115**, 6366–6375.
- 30 T. Gerbich, J. Herterich, J. Köhler and I. Fischer, *J. Phys. Chem. A*, 2014, **118**, 1397–1402.
- 31 J. Köhler, P. Hemberger, I. Fischer, G. Piani and L. Poisson, *J. Phys. Chem. A*, 2011, **115**, 14249–14253.
- 32 M. Sheik-Bahae, A. A. Said, T.-H. Wei, D. J. Hagan and E. W. Van Stryland, *IEEE J. Quantum Electron.*, 1990, **26**, 760–769.
- 33 J. Y. Yang, Y. L. Song, Y. X. Wang, C. W. Li, X. Jin and M. Shui, *Opt. Express*, 2009, **17**, 7110–7116.
- 34 J. Dipold, R. J. M. B. Batista, R. D. Fonseca, D. L. Silva, G. L. C. Moura, J. V. dos Anjos, A. M. Simas, L. De Boni and C. R. Mendonca, *Chem. Phys. Lett.*, 2016, **661**, 143–150.
- 35 S. Chi, L. Li and Y. Wu, *J. Phys. Chem. C*, 2016, **120**, 13706–13715.
- 36 G. Shi, C. Y. He, Y. B. Li, R. X. Zou, X. R. Zhang, Y. X. Wang, K. Yang, Y. L. Song and C. H. Wang, *J. Opt. Soc. Am. B*, 2009, **26**, 754–761.

