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# Cobalt-catalyzed, directed arylation of C–H bonds in *N*-aryl pyrazoles†

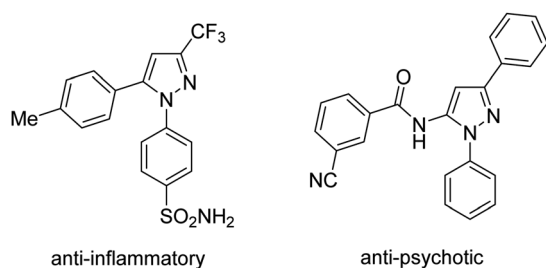
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We report a method for directed *ortho*-arylation of *N*-aryl pyrazoles with arylboronic acids. Reactions proceeded in the presence of a Co(hfacac)<sub>2</sub> catalyst, CeSO<sub>4</sub> oxidant, and HFIP solvent. Functionalities such as nitro, ester, bromo, and ketone groups were compatible with the reaction conditions. Using heterocycles including thiophene and carbazole was also feasible.

The last few decades have witnessed extensive development of methods for directed functionalization of C–H bonds.<sup>1</sup> Since many functionalities are weak coordinating groups toward C–H activation, directing groups are commonly required, thus facilitating the cleavage of C–H bonds.<sup>2</sup> Among these, using *N*-heterocycles which could bring the transition metal proximal to the targeted C–H bonds have attracted substantial attention. It should be noted that most of the general methods have relied on the use of bidentate, strongly coordinating groups. Meanwhile, monodentate, neutral, nitrogen-based directing groups have often suffered from the utilization of second- or third-row transition metals, which are relatively expensive and scarce.<sup>3</sup>

The pyrazole moiety is ubiquitously found in bio-related molecules (Scheme 1).<sup>4</sup> Functionalization of C–H bonds in pyrazoles would offer a convenient route to diversify the structures which are useful for further studies. Using the nitrogen

atom in pyrazole for directed C–H activation has been documented.<sup>5</sup> Arylation of *ortho* C–H bonds in *N*-aryl pyrazoles would afford condensed, synthetically useful anilides.<sup>5b</sup> Methods for palladium-catalyzed, pyrazole-directed arylation of sp<sup>2</sup> C–H bonds with aryl iodides have been developed.<sup>5a,5b</sup> Aryl (pseudo) halides were competent substrates for ruthenium-catalyzed, *ortho*-functionalization of *N*-aryl pyrazoles.<sup>5c–h</sup> Notably, using complexes of first-row metals for pyrazole-directed arylation is rare.<sup>5i</sup> Herein we report our attempts for arylation of *ortho* C–H bonds in *N*-aryl pyrazoles with arylboronic acids. Successes relied on the use of cobalt(II) hexafluoroacetylacetonate Co(hfacac)<sub>2</sub> catalyst, CeSO<sub>4</sub> co-oxidant, and HFIP solvent.



Scheme 1 *N*-Aryl pyrazole based bio-related molecules.

Table 1 Control experiments<sup>a</sup>

Entry	Variation from standard conditions	Yield of <b>3aa</b> (%)
1	None	82
2	Co(acac) <sub>2</sub> instead of Co(hfacac) <sub>2</sub>	65
3	No PivOH	57
4	Mn(OAc) <sub>3</sub> ·2H <sub>2</sub> O instead of CeSO <sub>4</sub>	22
5	PhI(OAc) <sub>2</sub> instead of CeSO <sub>4</sub>	0
6	TFE instead of HFIP	77
7	80 °C instead of 100 °C	52
8	No Co(hfacac) <sub>2</sub>	0

<sup>a</sup> **1a** (0.1 mmol), **2a** (0.2 mmol), solvent (1 mL), under air for 24 h. Yields are GC yields using diphenyl ether internal standard. Abbreviations: hfacac = hexafluoroacetylacetonate, TFE = trifluoroethanol, HFIP = hexafluoroisopropanol.

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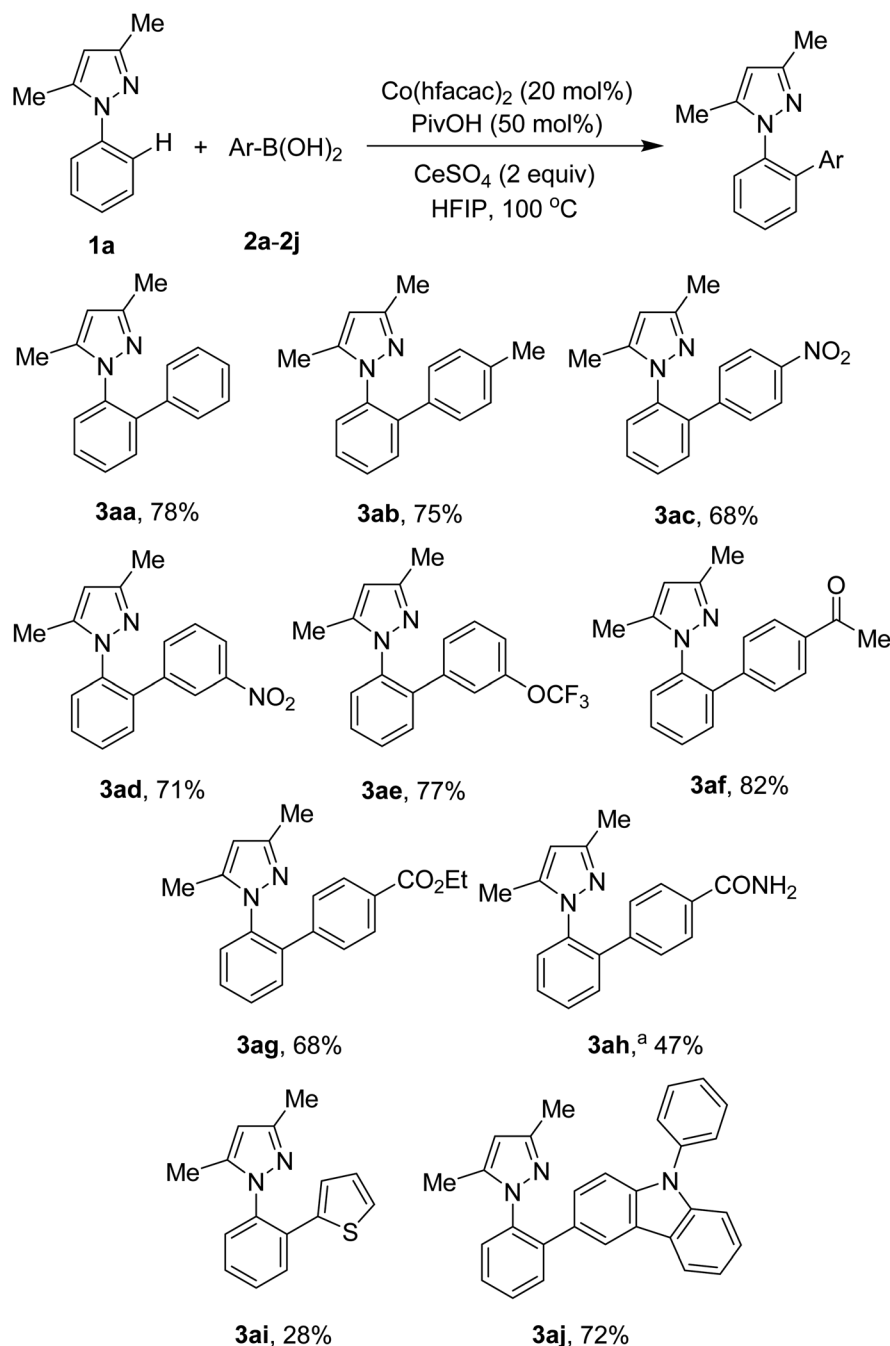
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Reaction of *N*-phenyl pyrazole **1a** and phenylboronic acid **2a** was firstly discovered. Results and control experiments are shown in Table 1. Using  $\text{Co}(\text{hfacac})_2$  catalyst afforded the *ortho*-arylation product **3aa** in 82% yield (entry 1). That  $\text{Co}(\text{acac})_2$  was an inferior catalyst (entry 2) to  $\text{Co}(\text{hfacac})_2$  somewhat implied the importance of electronic properties of cobalt(II) complexes. The additive pivalic acid  $\text{PivOH}$  was crucial to obtain an acceptable yield of **3aa** (entry 3). Other co-oxidants rather than  $\text{CeSO}_4$  were not effective for this directed arylation (entries 4 and 5). Trifluoroethanol could be used as solvent for the

coupling of **1a** and **2a**, thus furnishing **3aa** in 77% yield (entry 6). The reaction should not be run at the temperature lower than 100 °C (entry 7). Lastly, omitting the presence of  $\text{Co}(\text{hfacac})_2$  gave no arylation product (entry 8), confirming the crucial role of the cobalt complex.

Scope of the arylation with respect to arylboronic acids was next studied. The results are presented in Scheme 2. Electron-neutral and electron-rich arylboronic acids were competent substrates (**3aa**, **3ab**). In contrast to palladium-catalyzed methods,<sup>5a,5b</sup> our conditions were tolerant of arylboronic acids



**Scheme 2** Scope of arylboronic acids. Reaction conditions: **1a** (0.5 mmol), **2a-2j** (1 mmol),  $\text{Co}(\text{hfacac})_2$  (0.1 mmol),  $\text{PivOH}$  (0.25 mmol),  $\text{CeSO}_4$  (1 mmol), HFIP (2.5 mL), under air, at 100 °C for 24 h. Yields are isolated yields. <sup>a</sup>4-Cyanophenylboronic acid was used.



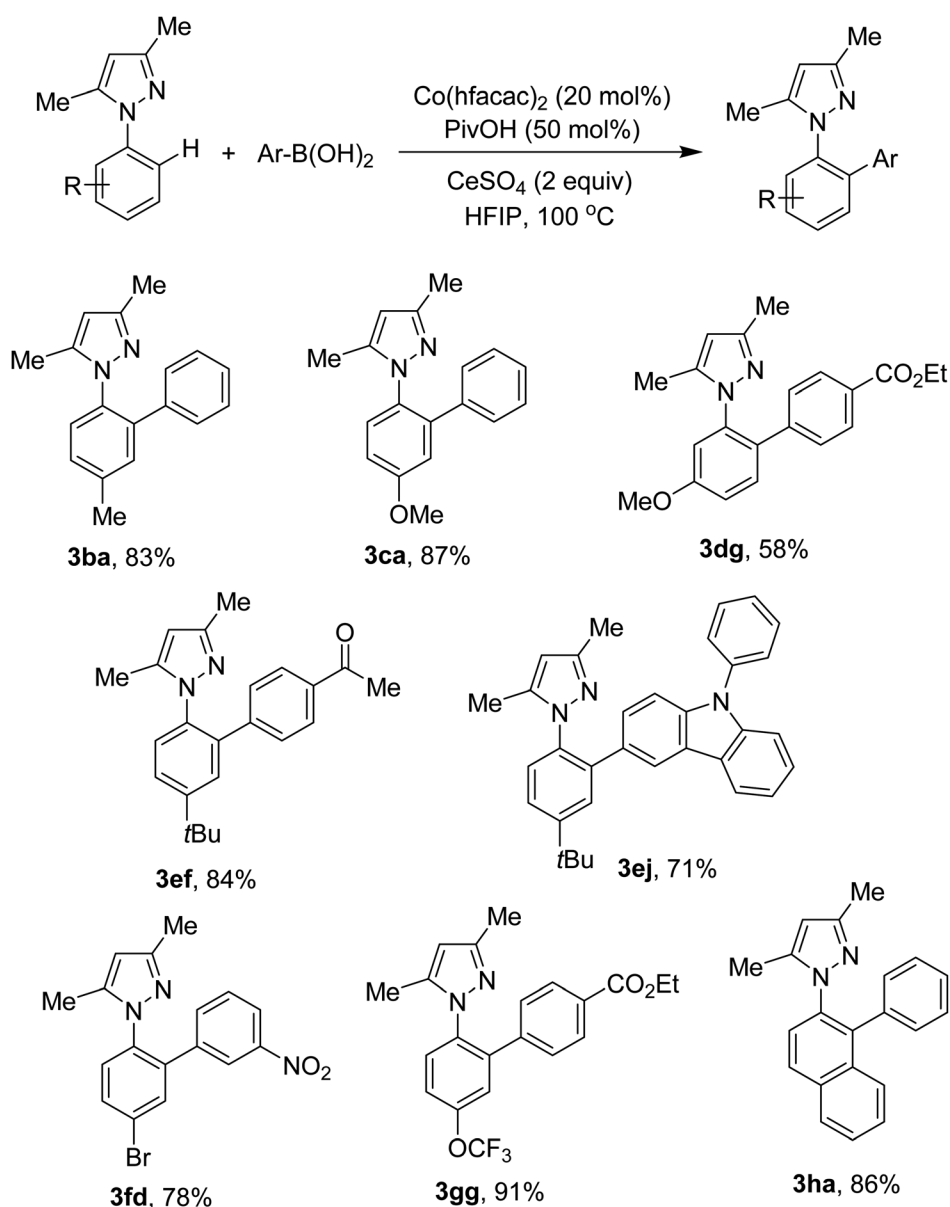
containing nitro functionality (**3ac**, **3ad**). Ketone (**3af**) and ester (**3ag**) groups were still intact during the course of the arylation. However, the nitrile functional group was hydrolyzed, affording the arylation product in moderate yield (**3ah**). Heteroaryl boronic acids were also attempted. Arylation of C–H bonds with 2-thienylboronic acid gave only 28% yield of the desired product (**3ai**). Meanwhile, a 72% yield of a carbazole-derived pyrazole (**3aj**) was successfully isolated. It should be noted that *ortho*-substituted arylboronic acids were not competent substrates.

Different *N*-aryl pyrazoles were examined. The results are shown in Scheme 3. Reaction conditions were compatible with electron-rich arenes (**3ba**, **3ca**, **3ef**, **3ej**). Less hindered C–H bonds were selectively activated if a biased arene was attempted (**3dg**). *N*-Aryl pyrazoles containing bromo and trifluoromethoxy

groups were successfully arylated (**3fd**, **3gg**). If 2-naphthyl derived pyrazole was used, arylation occurred at the C1–H bond (**3ha**).

Based on the results of previous studies,<sup>6</sup> our hypothesis on the reaction mechanism may include a Co(II)/Co(III) catalytic cycle. Coordination of nitrogen atom in pyrazole directing group would bring the cobalt(III) center proximal to the *ortho* C–H bond, thus facilitating the C–H activation. Transmetallation followed by reductive elimination would afford the arylation product.

In conclusion, we have developed a method for cobalt-catalyzed, pyrazole-directed arylation of sp<sup>2</sup> C–H bonds. Reaction conditions were tolerant of many functionalities including ester, ketone, nitro, and bromo groups. Ongoing projects will be



**Scheme 3** Arylation of C–H bonds in *N*-aryl pyrazoles. Reaction conditions: *N*-aryl pyrazole (0.5 mmol), arylboronic acid (1 mmol),  $\text{Co(hfacac)}_2$  (0.1 mmol),  $\text{PivOH}$  (0.25 mmol),  $\text{CeSO}_4$  (1 mmol), HFIP (2.5 mL), under air, at  $100^\circ\text{C}$  for 24 h. Yields are isolated yields.

involving investigation of other pyrazole-directed functionalization of inert C–H bonds.

## Conflicts of interest

There are no conflicts to declare.

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