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A ratiometric fluorescent probe for the detection of β -galactosidase and its application[†]

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Herein, a coumarin fluorescent probe (Probe 1) was developed for the ratiometric detection of β -galactosidase (β -gal) activity. The detection range was 0–0.1 U mL⁻¹ and 0.2–0.8 U mL⁻¹, and the limit of detection (LOD) was 0.0054 U mL⁻¹. Moreover, the luminous intensity of Probe 1 increased gradually with increase in β -gal activity. It could be observed under 254 nm UV irradiation by the naked eye. Furthermore, this method only required a small amount of sample (20 µL) and a short analytical time (30 min) for the detection of β -gal activity with a low LOD. Probe 1 was successfully used to detect β -gal activity in real fruit samples, and can be applied to the quantitative and qualitative detection of β -gal activity.

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1. Introduction

β-Galactosidase (β-gal, EC 3.2.1.23) is part of the glycoside hydrolase family and has many microbial sources.¹ In addition to its hydrolysis activity, β-gal from some sources also exhibits glycosylation activity. Food-processing applications of β-gal mainly include the following: allowing lactose-intolerant individuals to consume dairy; improving the sweetness of dairy products; preventing dairy products from crystallizing during freezing; the production of galactooligosaccharides; applications in fermented dairy products; whey processing; the analysis of lactose content in dairy products; and promoting the softening and ripening of fruit and vegetables.^{2–4} β-gal has also been used for improving the sweetness, digestibility, flavor, and solubility of dairy products.⁵

β-Galactosidase is widely found in many kinds of plants. Generally, the content of β-gal increases when plants mature,⁶ and the activity of β-gal has a positive correlation with the maturity of fruit, including tomato,⁷ papaya,⁸ apple,⁹ persimmon,¹⁰ kiwi fruit,¹¹ avocado,¹² pear,¹³ peach,¹⁴ and mango.¹⁵ It can degrade cell-wall polysaccharides and release free galactose, which can promote – for example – pepper ripening and ethylene production in tomatoes.⁷ However, many enzymes exist in different types of fruit. Therefore, the development of a simple, selective, and rapid detection method for β-gal activity in fruit is important.

To date, many methods have been used for the detection of β -gal activity. These include chemiluminescence,¹⁶ HPLC,¹⁷ colorimetric methods,¹⁸ magnetic resonance,¹⁹ UV

spectrophotometry,²⁰ positron emission tomography,²¹ enzymelinked immunosorbent assay technology,²² and electrochemical methods.²³ However, there are certain disadvantages associated with these methods, such as long experimental duration, complex operation conditions, and high experimental costs. Thus, developing a selective, rapid, and simple detection method for β -gal activity is necessary for applications in fruit production. To this end, the development of novel fluorescent probes is very promising,^{24–29} and many kinds of β -gal probes have been reported for use in biological imaging,^{30–32} including two-photon fluorescence probes,^{33,34} ratiometric probes,^{35,36} and turn-on probes.^{37–39} However, fluorescent probes used in fruit detection are extremely rare in scientific literature.^{40,41}

To develop a simple detection method for β -gal activity in fruit, a ratiometric fluorescent probe (Probe 1) with high accuracy and detection precision was developed. This ratiometric fluorescent probe has the characteristics of high accuracy and strong anti-interference. Further, the luminous intensity of the Probe 1 solution increased with increasing β -gal activity under 254 nm UV irradiation. In addition, Probe 1 was successfully applied to the quantitative and qualitative detection of β -gal activity in fruit.

2. Materials and methods

2.1 General methods

The β -gal, sodium chloride (NaCl), magnesium chloride (MgCl₂), hydrogen peroxide (H₂O₂), ammonium chloride (NH₄Cl), sodium bromide (NaBr), glutathione (GSH, 98%), glycine (Gly), D-leucine (Leu), potassium iodide (KI), histidine (His), potassium chloride (KCl), L-valine (Val), β -glucuronidase (from *Escherichia coli*), lysozyme (from chicken egg whites), lipase (from porcine pancreas), and α -galactosidase (α -gal) were purchased from Bailingwei Co., Ltd, China.

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2.2 Instruments

Fluorescence spectra were obtained using the Rili F-4600 fluorescence spectrometer. NMR spectra were obtained using the Bruker AV 300 MHz NMR machine. HRMS was performed using a Bruker Apex IV FTMS.

2.3 Synthesis of Probe 1

7-Hydroxy-4-methylcoumarin (0.18 g, 1.00 mmol), Cs_2CO_3 (1.63 g, 5.00 mmol), Na_2SO_4 (0.36 g, 2.50 mmol), galactopyranosyl-1-bromide (3068-32-4, 0.31 g, 0.75 mmol), and CH_3CN (30 mL) were added to a flask (Scheme 1) and reacted for 1 h at 25 °C to obtain compound **2** (0.35 g, 91% yield).

Compound 2 (0.23 g, 0.4 mmol) was dissolved in CH_3OH (20 mL). A solution of K_2CO_3 (0.13 g, 0.09 mmol) and CH_3OH (80 mL) was added and reacted for 4 h at 25 °C, after which the mixture was adjusted to pH 7 using an aqueous H_2SO_4 solution (0.05 M). The precipitate was removed by filtration and recrystallized from ethanol to obtain Probe 1 (0.11 g, 81% yield).

¹H NMR (300 MHz, CDCl₃) δ (ppm): 7.70 (d, J = 9.4 Hz, 1H), 7.03 (d, J = 7.0 Hz, 2H), 6.24 (s, 1H), 5.21 (s, 1H), 4.98 (d, J =7.6 Hz, 1H), 4.87 (s, 1H), 4.66 (s, 1H), 4.52 (s, 1H), 3.72 (s, 1H), 3.66 (d, J = 6.1 Hz, 1H), 3.60 (d, J = 8.1 Hz, 1H), 3.53 (s, 2H), 3.45 (s, 1H), 2.41 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ (ppm): 160.23, 160.06, 154.39, 153.28, 126.35, 113.99, 113.40, 111.62, 103.17, 100.64, 75.69, 73.20, 70.09, 68.11, 60.37, 18.08. HRMS (ESI): calcd for [M – H]⁻ 337.092891, found 337.0931.

2.4 Preparation of analytes

Probe 1 was dissolved using DMSO. β -gal, β -glucuronidase, lysozyme, lipase, and α -gal were dissolved in Tris–HCl (pH 7.3, 50 mM) and frozen in a -20 °C refrigerator. The stock solution was diluted using a certain concentration gradient with Tris–HCl (pH 7.3, 50 mM) before each use. NaCl, KCl, MgCl₂, KI, NH₄Cl, NaBr, H₂O₂, GSH, Gly, Leu, His, and Val, were dissolved in deionized water.

2.5 Preparation of samples

Pears, apples, grapes, strawberries, and kiwis were bought from a local supermarket. Twenty grams of the abovementioned fruit

were centrifuged for 10 min, at 12 000 rpm, after grinding; subsequently, the supernatant was obtained, which was filtered and reserved.

2.6 Fluorescence detection assays

The Probe 1 solution (1 mM, 0.02 mL) was added to a cuvette, to which 2 mL of water was added. Then, β -gal was added; after 40 min, the mixture was analyzed using fluorescence spectra (slit width = 5 nm, λ_{ex} = 327 nm, voltage = 500 V, and temperature = 37 °C).

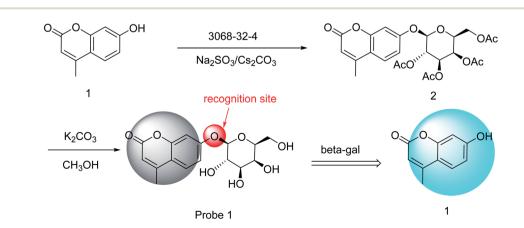
3. Results and discussion

3.1 Probe preparation

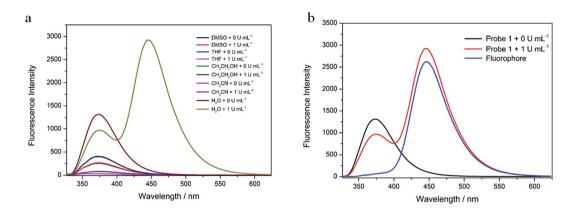
Probe **1** was synthesized in two steps. First, compound **2** was obtained through the nucleophilic substitution of 7-hydroxy-4-methylcoumarin with galacto-pyranosyl-1-bromide (3068-32-4). Second, the acetyl groups of compound **2** were hydrolyzed to obtain Probe **1** (Scheme 1). Probe **1** was purified by recrystallization from ethanol, and NMR and HRMS were used to characterize this probe (Fig. S1–S3, ESI†). The preparation process was carried out at 25 °C under mild conditions; thus, the synthesis of Probe **1** was a simple process.

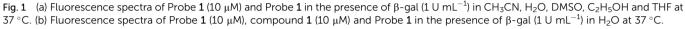
3.2 Fluorescence properties

First, the fluorescence response of Probe **1** with β -gal in different solutions (CH₃CN, H₂O, DMSO, C₂H₅OH, and THF) was determined (Fig. 1a). After the addition of β -gal, the fluorescence intensity was decreased at 374 nm in CH₃CN, DMSO, C₂H₅OH, and THF, with changes in only one fluorescent emission. In H₂O, however, the fluorescence intensity was decreased at 374 nm and significantly increased at 444 nm. As shown in Fig. 1b, Probe **1** displayed a fluorescence emission peak at 374 nm in an H₂O solution. After the addition of β -gal, the fluorescence intensity at 374 nm decreased and a peak appeared at 444 nm. The fluorophore (7-hydroxy-4-methylcoumarin, compound **1**) had a peak at 444 nm in an H₂O solution. This confirmed that 374 nm was the peak of Probe **1** and 444 nm was the fluorescence-emission peak of the fluorophore. These results indicated that Probe **1** was



Scheme 1 Synthesis of Probe 1 and the recognition mechanism of Probe 1 to β -gal.





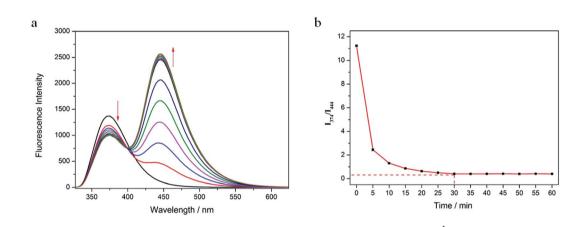


Fig. 2 (a) Time-dependent fluorescence spectra of Probe 1 (10 μ M) in the presence of β -gal (1 U mL⁻¹) in water at 37 °C. (b) The fluorescence emission ratio ($l_{444 \text{ nm}}/l_{374 \text{ nm}}$) of Probe 1 in the presence of β -gal form 0 min to 60 min. Tests were performed in triplicate.

a ratiometric probe, facilitating the detection of β -gal by determining the ratio of fluorescence intensity at two different emission wavelengths.

Second, the time-response relationships of Probe **1** toward β gal in water were tested (Fig. 2a). The fluorescence intensity at 374 nm decreased slowly with the addition of β -gal. Simultaneously, the fluorescence intensity at 444 nm increased rapidly. The fluorescence emission ratio ($I_{444 \text{ nm}}/I_{374 \text{ nm}}$) decreased rapidly from 0 to 10 min, and reached an equilibrium in 30 min (Fig. 2b). The emission ratio remained unchanged from 30 to 60 min. This shows that 30 min were required for the identification of β -gal by Probe **1**, which was set as the duration for the subsequent experiments.

The effects of competitor ions and compounds were used to ascertain the selectivity of Probe **1**. Various competitors were tested, including Na⁺, K⁺, Mg²⁺, I⁻, NH₄⁺, Br⁻, H₂O₂, Cl⁻, GSH, Gly, Leu, Val, His, β-glucuronidase, lipase, and lysozyme (Fig. 3). In the presence of any of these competing ions and compounds, there was minimal change in the emission ratio $(I_{444 \text{ nm}}/I_{374 \text{ nm}})$. In particular, Probe **1** was almost unresponsive in the presence of β-glucuronidase and α-gal. However, with the addition of β-gal, the emission ratios $(I_{444 \text{ nm}}/I_{374 \text{ nm}})$ of Probe **1** + β-gal and Probe **1** + β-gal + competitors were

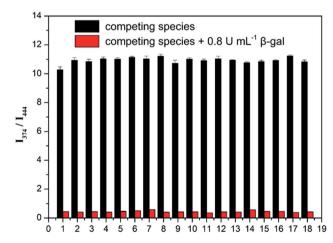


Fig. 3 The fluorescence emission ratio ($I_{444 \text{ nm}}/I_{374 \text{ nm}}$) of Probe 1 (10 μ M) upon addition of various species (0.8 U mL⁻¹ for α -gal, lipase and 100 μ M for others. 1, blank; 2, Na⁺; 3, K⁺; 4, Mg²⁺; 5, I⁻; 6, NH₄⁺; 7, Br⁻; 8, H₂O₂; 9, GSH; 10, Gly; 11, Leu; 12, His; 13, Val; 14, β -glucuronidase; 15, lysozyme; 16, lipase; 17, α -gal; 18, Cl⁻. 0.8 U mL⁻¹ for β -gal). Tests were performed in triplicate.

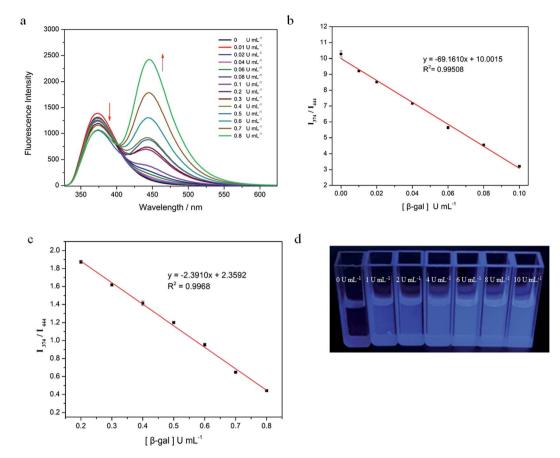


Fig. 4 (a) Fluorescence spectra of Probe 1 (10 μ M) with β -gal (0, 0.01, 0.02, 0.04, 0.06, 0.08, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8 U mL⁻¹). (b) Plot of fluorescence intensity differences with 0–0.1 U mL⁻¹ β -gal. (c) Plot of fluorescence intensity differences with 0.2–0.8 U mL⁻¹ β -gal. Tests were performed in triplicate. (d) Photograph of Probe 1 (10 μ M) luminescent intensity subjected to β -gal (0, 1, 2, 4, 6, 8, 10 U mL⁻¹) under 254 nm UV light.

almost identical. This indicates that Probe 1 has high selectivity in detecting β -gal.

The fluorescence-intensity changes of Probe 1 with various activities of β -gal (0, 0.01, 0.02, 0.04, 0.06, 0.08, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, and 0.8 U mL⁻¹) were recorded and are shown in Fig. 4a. The emission ratio ($I_{444 \text{ nm}}/I_{374 \text{ nm}}$) exhibited two linear sections in response to β -gal activity: 0–0.1 U mL⁻¹ ($R^2 = 0.9951$, Fig. 4b) and 0.2–0.8 U mL⁻¹ ($R^2 = 0.9968$, Fig. 4c). The Probe 1 limit of detection (LOD) for β -gal activity was 0.0054 U mL⁻¹, based on LOD = 3 SD/B. This indicated that Probe 1 could be used to detect β -gal activity with a low LOD in water. In addition, the luminous intensity of Probe 1 gradually increased with increases in β -gal activity, as was observed by the naked eye under 254 nm UV irradiation (Fig. 4d). All results showed that Probe 1 could be used as a quantitative and qualitative tool to detect β -gal activity.

3.3 Recognition mechanism

After the addition of β -gal (1 U mL⁻¹), a new peak appeared, which was previously proven to be compound 1 (Fig. S4, ESI†). Mass spectrometry provided further evidence that the β -galactosides were hydrolyzed (Fig. S5, ESI†), with the peak at m/z = 174.56.00 corresponding to compound 1 (M – H); the peak at m/z

z = 337.04 was that of Probe 1 (M – H). These results show that Probe 1's mechanism for β -gal recognition is the β -gal enzymatic hydrolysis of β -galactosides.

3.4 Application

As β -gal activity is positively correlated with the maturity of fruit, the development of a simple and highly selective method for β gal activity detection in fruit is crucial. Therefore, the ability of Probe **1** to detect β -gal activity in fruit must be demonstrated.

Pear, apple, grape, strawberry, and kiwi (20 μ L) samples were tested using Probe 1. The β -gal activity of the kiwi (ripe) sample was 0.0938 \pm 0.0027 U mL⁻¹, and those of the kiwi (unripe), pear, apple, grape, and strawberry samples were 0 U mL⁻¹ (Table 1). The ripeness of the kiwi fruit was mainly determined by the softness or hardness of the fruit (Fig. S6, ESI[†]).

To validate this method, the β -gal activity in these samples was tested using the β -galactosidase spectrophotometric method.^{40,41} The β -gal activity of all samples was 0 U mL⁻¹. No β gal activity was detected in the kiwi (ripe) sample by the GB/T 33409-2016 method; this could be because the β -gal activity in kiwis (ripe) and the LOD of this method are of the same order of magnitude. Then, the addition of β -gal with different activities (0.02, 0.04, 0.2, and 0.4 U mL⁻¹) to the samples showed that the

Table 1 Determination of β -gal activity in real fruit samples

Sample	β -gal level found (U $mL^{-1})$	Added (U mL^{-1})	Found (U mL^{-1})	Recovery/%	RSD/% ($n = 3$
Pear	0	0.02	0.022	108.00	0.11
		0.04	0.043	106.61	0.21
		0.2	0.221	101.32	1.69
		0.4	0.405	101.33	1.44
Apple	0	0.02	0.022	110.29	0.07
		0.04	0.044	108.94	0.33
		0.2	0.192	96.13	2.09
		0.4	0.387	96.71	6.31
Grape	0	0.02	0.022	108.80	0.26
		0.04	0.044	109.83	0.26
		0.2	0.217	108.32	1.40
		0.4	0.402	100.58	0.81
Strawberry	0	0.02	0.021	104.55	0.13
		0.04	0.041	101.86	0.19
		0.2	0.195	97.28	3.68
		0.4	0.376	94.07	1.33
Kiwi (unripe)	0	0.02	0.021	106.4	0.06
		0.04	0.041	103.39	0.08
		0.2	0.203	101.65	0.69
		0.4	0.403	100.63	1.89
Kiwi (ripe)	0.0938 ± 0.0027	_	_	_	_

recoveries ranged between 94.07–110.29% (Table 1). These results indicate that Probe 1 could accurately and quickly determine the β -gal activity in fruit.

4. Conclusions

In this study, a coumarin fluorescent probe (Probe 1) was developed for the ratiometric detection of β -gal activity. The mechanism of β -gal recognition by Probe 1 involves the β -gal enzymatic hydrolysis of β -galactosides. The quantitative range of β -gal activity detected was 0–0.1 U mL⁻¹ and 0.2–0.8 U mL⁻¹, with an LOD of 0.0054 U mL⁻¹. This method exhibited good linearity and specificity, had a short analytical time (30 min), required a small amount of sample (20 μ L), and had a low LOD. Additionally, the luminous intensity of Probe 1 gradually increased with increasing enzyme activity. This phenomenon could be observed by the naked eye under 254 nm UV irradiation. Furthermore, Probe 1 was a useful tool for the qualitative and quantitative determination of β -gal activity in fruit.

Conflicts of interest

There are no conflicts of interest to declare.

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