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## A ratiometric fluorescent probe for the detection of $\beta$ -galactosidase and its application<sup>†</sup>

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Herein, a coumarin fluorescent probe (Probe 1) was developed for the ratiometric detection of  $\beta$ -galactosidase ( $\beta$ -gal) activity. The detection range was 0–0.1 U  $\text{mL}^{-1}$  and 0.2–0.8 U  $\text{mL}^{-1}$ , and the limit of detection (LOD) was 0.0054 U  $\text{mL}^{-1}$ . Moreover, the luminous intensity of Probe 1 increased gradually with increase in  $\beta$ -gal activity. It could be observed under 254 nm UV irradiation by the naked eye. Furthermore, this method only required a small amount of sample (20  $\mu\text{L}$ ) and a short analytical time (30 min) for the detection of  $\beta$ -gal activity with a low LOD. Probe 1 was successfully used to detect  $\beta$ -gal activity in real fruit samples, and can be applied to the quantitative and qualitative detection of  $\beta$ -gal activity.

### 1. Introduction

$\beta$ -Galactosidase ( $\beta$ -gal, EC 3.2.1.23) is part of the glycoside hydrolase family and has many microbial sources.<sup>1</sup> In addition to its hydrolysis activity,  $\beta$ -gal from some sources also exhibits glycosylation activity. Food-processing applications of  $\beta$ -gal mainly include the following: allowing lactose-intolerant individuals to consume dairy; improving the sweetness of dairy products; preventing dairy products from crystallizing during freezing; the production of galactooligosaccharides; applications in fermented dairy products; whey processing; the analysis of lactose content in dairy products; and promoting the softening and ripening of fruit and vegetables.<sup>2–4</sup>  $\beta$ -gal has also been used for improving the sweetness, digestibility, flavor, and solubility of dairy products.<sup>5</sup>

$\beta$ -Galactosidase is widely found in many kinds of plants. Generally, the content of  $\beta$ -gal increases when plants mature,<sup>6</sup> and the activity of  $\beta$ -gal has a positive correlation with the maturity of fruit, including tomato,<sup>7</sup> papaya,<sup>8</sup> apple,<sup>9</sup> persimmon,<sup>10</sup> kiwi fruit,<sup>11</sup> avocado,<sup>12</sup> pear,<sup>13</sup> peach,<sup>14</sup> and mango.<sup>15</sup> It can degrade cell-wall polysaccharides and release free galactose, which can promote – for example – pepper ripening and ethylene production in tomatoes.<sup>7</sup> However, many enzymes exist in different types of fruit. Therefore, the development of a simple, selective, and rapid detection method for  $\beta$ -gal activity in fruit is important.

To date, many methods have been used for the detection of  $\beta$ -gal activity. These include chemiluminescence,<sup>16</sup> HPLC,<sup>17</sup> colorimetric methods,<sup>18</sup> magnetic resonance,<sup>19</sup> UV

spectrophotometry,<sup>20</sup> positron emission tomography,<sup>21</sup> enzyme-linked immunosorbent assay technology,<sup>22</sup> and electrochemical methods.<sup>23</sup> However, there are certain disadvantages associated with these methods, such as long experimental duration, complex operation conditions, and high experimental costs. Thus, developing a selective, rapid, and simple detection method for  $\beta$ -gal activity is necessary for applications in fruit production. To this end, the development of novel fluorescent probes is very promising,<sup>24–29</sup> and many kinds of  $\beta$ -gal probes have been reported for use in biological imaging,<sup>30–32</sup> including two-photon fluorescence probes,<sup>33,34</sup> ratiometric probes,<sup>35,36</sup> and turn-on probes.<sup>37–39</sup> However, fluorescent probes used in fruit detection are extremely rare in scientific literature.<sup>40,41</sup>

To develop a simple detection method for  $\beta$ -gal activity in fruit, a ratiometric fluorescent probe (Probe 1) with high accuracy and detection precision was developed. This ratiometric fluorescent probe has the characteristics of high accuracy and strong anti-interference. Further, the luminous intensity of the Probe 1 solution increased with increasing  $\beta$ -gal activity under 254 nm UV irradiation. In addition, Probe 1 was successfully applied to the quantitative and qualitative detection of  $\beta$ -gal activity in fruit.

### 2. Materials and methods

#### 2.1 General methods

The  $\beta$ -gal, sodium chloride (NaCl), magnesium chloride ( $\text{MgCl}_2$ ), hydrogen peroxide ( $\text{H}_2\text{O}_2$ ), ammonium chloride ( $\text{NH}_4\text{Cl}$ ), sodium bromide (NaBr), glutathione (GSH, 98%), glycine (Gly), D-leucine (Leu), potassium iodide (KI), histidine (His), potassium chloride (KCl), L-valine (Val),  $\beta$ -glucuronidase (from *Escherichia coli*), lysozyme (from chicken egg whites), lipase (from porcine pancreas), and  $\alpha$ -galactosidase ( $\alpha$ -gal) were purchased from Bailingwei Co., Ltd, China.

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## 2.2 Instruments

Fluorescence spectra were obtained using the Rili F-4600 fluorescence spectrometer. NMR spectra were obtained using the Bruker AV 300 MHz NMR machine. HRMS was performed using a Bruker Apex IV FTMS.

## 2.3 Synthesis of Probe 1

7-Hydroxy-4-methylcoumarin (0.18 g, 1.00 mmol),  $\text{Cs}_2\text{CO}_3$  (1.63 g, 5.00 mmol),  $\text{Na}_2\text{SO}_4$  (0.36 g, 2.50 mmol), galacto-pyranosyl-1-bromide (3068-32-4, 0.31 g, 0.75 mmol), and  $\text{CH}_3\text{CN}$  (30 mL) were added to a flask (Scheme 1) and reacted for 1 h at 25 °C to obtain compound 2 (0.35 g, 91% yield).

Compound 2 (0.23 g, 0.4 mmol) was dissolved in  $\text{CH}_3\text{OH}$  (20 mL). A solution of  $\text{K}_2\text{CO}_3$  (0.13 g, 0.09 mmol) and  $\text{CH}_3\text{OH}$  (80 mL) was added and reacted for 4 h at 25 °C, after which the mixture was adjusted to pH 7 using an aqueous  $\text{H}_2\text{SO}_4$  solution (0.05 M). The precipitate was removed by filtration and recrystallized from ethanol to obtain Probe 1 (0.11 g, 81% yield).

$^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm): 7.70 (d,  $J$  = 9.4 Hz, 1H), 7.03 (d,  $J$  = 7.0 Hz, 2H), 6.24 (s, 1H), 5.21 (s, 1H), 4.98 (d,  $J$  = 7.6 Hz, 1H), 4.87 (s, 1H), 4.66 (s, 1H), 4.52 (s, 1H), 3.72 (s, 1H), 3.66 (d,  $J$  = 6.1 Hz, 1H), 3.60 (d,  $J$  = 8.1 Hz, 1H), 3.53 (s, 2H), 3.45 (s, 1H), 2.41 (s, 3H).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm): 160.23, 160.06, 154.39, 153.28, 126.35, 113.99, 113.40, 111.62, 103.17, 100.64, 75.69, 73.20, 70.09, 68.11, 60.37, 18.08. HRMS (ESI): calcd for  $[\text{M} - \text{H}]^-$  337.092891, found 337.0931.

## 2.4 Preparation of analytes

Probe 1 was dissolved using DMSO.  $\beta$ -gal,  $\beta$ -glucuronidase, lysozyme, lipase, and  $\alpha$ -gal were dissolved in Tris-HCl (pH 7.3, 50 mM) and frozen in a –20 °C refrigerator. The stock solution was diluted using a certain concentration gradient with Tris-HCl (pH 7.3, 50 mM) before each use.  $\text{NaCl}$ ,  $\text{KCl}$ ,  $\text{MgCl}_2$ ,  $\text{KI}$ ,  $\text{NH}_4\text{Cl}$ ,  $\text{NaBr}$ ,  $\text{H}_2\text{O}_2$ , GSH, Gly, Leu, His, and Val, were dissolved in deionized water.

## 2.5 Preparation of samples

Pears, apples, grapes, strawberries, and kiwis were bought from a local supermarket. Twenty grams of the abovementioned fruit

were centrifuged for 10 min, at 12 000 rpm, after grinding; subsequently, the supernatant was obtained, which was filtered and reserved.

## 2.6 Fluorescence detection assays

The Probe 1 solution (1 mM, 0.02 mL) was added to a cuvette, to which 2 mL of water was added. Then,  $\beta$ -gal was added; after 40 min, the mixture was analyzed using fluorescence spectra (slit width = 5 nm,  $\lambda_{\text{ex}} = 327$  nm, voltage = 500 V, and temperature = 37 °C).

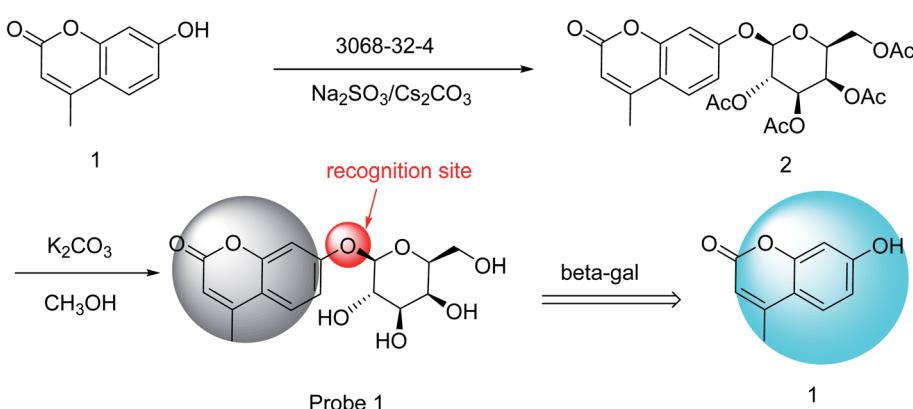
## 3. Results and discussion

### 3.1 Probe preparation

Probe 1 was synthesized in two steps. First, compound 2 was obtained through the nucleophilic substitution of 7-hydroxy-4-methylcoumarin with galacto-pyranosyl-1-bromide (3068-32-4). Second, the acetyl groups of compound 2 were hydrolyzed to obtain Probe 1 (Scheme 1). Probe 1 was purified by recrystallization from ethanol, and NMR and HRMS were used to characterize this probe (Fig. S1–S3, ESI†). The preparation process was carried out at 25 °C under mild conditions; thus, the synthesis of Probe 1 was a simple process.

### 3.2 Fluorescence properties

First, the fluorescence response of Probe 1 with  $\beta$ -gal in different solutions ( $\text{CH}_3\text{CN}$ ,  $\text{H}_2\text{O}$ , DMSO,  $\text{C}_2\text{H}_5\text{OH}$ , and THF) was determined (Fig. 1a). After the addition of  $\beta$ -gal, the fluorescence intensity was decreased at 374 nm in  $\text{CH}_3\text{CN}$ , DMSO,  $\text{C}_2\text{H}_5\text{OH}$ , and THF, with changes in only one fluorescent emission. In  $\text{H}_2\text{O}$ , however, the fluorescence intensity was decreased at 374 nm and significantly increased at 444 nm. As shown in Fig. 1b, Probe 1 displayed a fluorescence emission peak at 374 nm in an  $\text{H}_2\text{O}$  solution. After the addition of  $\beta$ -gal, the fluorescence intensity at 374 nm decreased and a peak appeared at 444 nm. The fluorophore (7-hydroxy-4-methylcoumarin, compound 1) had a peak at 444 nm in an  $\text{H}_2\text{O}$  solution. This confirmed that 374 nm was the peak of Probe 1 and 444 nm was the fluorescence-emission peak of the fluorophore. These results indicated that Probe 1 was



Scheme 1 Synthesis of Probe 1 and the recognition mechanism of Probe 1 to  $\beta$ -gal.



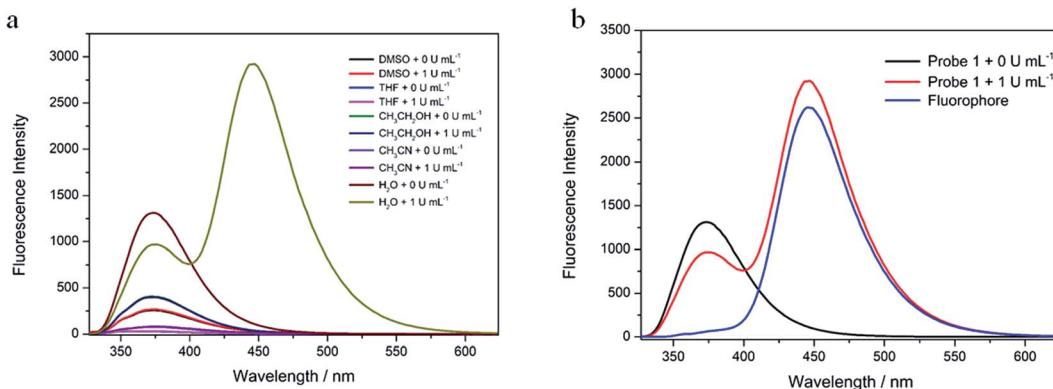


Fig. 1 (a) Fluorescence spectra of Probe 1 (10  $\mu\text{M}$ ) and Probe 1 in the presence of  $\beta$ -gal (1  $\text{U mL}^{-1}$ ) in  $\text{CH}_3\text{CN}$ ,  $\text{H}_2\text{O}$ , DMSO,  $\text{C}_2\text{H}_5\text{OH}$  and THF at 37  $^\circ\text{C}$ . (b) Fluorescence spectra of Probe 1 (10  $\mu\text{M}$ ), compound 1 (10  $\mu\text{M}$ ) and Probe 1 in the presence of  $\beta$ -gal (1  $\text{U mL}^{-1}$ ) in  $\text{H}_2\text{O}$  at 37  $^\circ\text{C}$ .

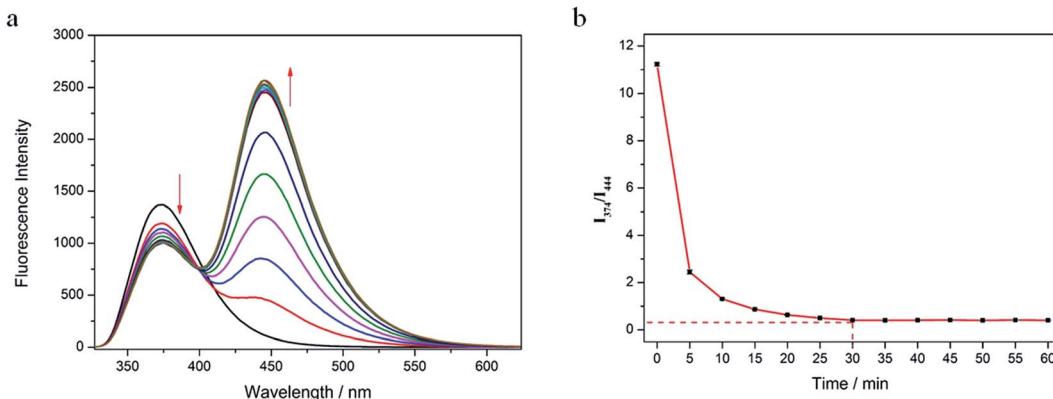


Fig. 2 (a) Time-dependent fluorescence spectra of Probe 1 (10  $\mu\text{M}$ ) in the presence of  $\beta$ -gal (1  $\text{U mL}^{-1}$ ) in water at 37  $^\circ\text{C}$ . (b) The fluorescence emission ratio ( $I_{444 \text{ nm}}/I_{374 \text{ nm}}$ ) of Probe 1 in the presence of  $\beta$ -gal from 0 min to 60 min. Tests were performed in triplicate.

a ratiometric probe, facilitating the detection of  $\beta$ -gal by determining the ratio of fluorescence intensity at two different emission wavelengths.

Second, the time-response relationships of Probe 1 toward  $\beta$ -gal in water were tested (Fig. 2a). The fluorescence intensity at 374 nm decreased slowly with the addition of  $\beta$ -gal. Simultaneously, the fluorescence intensity at 444 nm increased rapidly. The fluorescence emission ratio ( $I_{444 \text{ nm}}/I_{374 \text{ nm}}$ ) decreased rapidly from 0 to 10 min, and reached an equilibrium in 30 min (Fig. 2b). The emission ratio remained unchanged from 30 to 60 min. This shows that 30 min were required for the identification of  $\beta$ -gal by Probe 1, which was set as the duration for the subsequent experiments.

The effects of competitor ions and compounds were used to ascertain the selectivity of Probe 1. Various competitors were tested, including  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{I}^-$ ,  $\text{NH}_4^+$ ,  $\text{Br}^-$ ,  $\text{H}_2\text{O}_2$ ,  $\text{Cl}^-$ ,  $\text{GSH}$ ,  $\text{Gly}$ ,  $\text{Leu}$ ,  $\text{Val}$ ,  $\text{His}$ ,  $\beta$ -glucuronidase, lipase, and lysozyme (Fig. 3). In the presence of any of these competing ions and compounds, there was minimal change in the emission ratio ( $I_{444 \text{ nm}}/I_{374 \text{ nm}}$ ). In particular, Probe 1 was almost unresponsive in the presence of  $\beta$ -glucuronidase and  $\alpha$ -gal. However, with the addition of  $\beta$ -gal, the emission ratios ( $I_{444 \text{ nm}}/I_{374 \text{ nm}}$ ) of Probe 1 +  $\beta$ -gal and Probe 1 +  $\beta$ -gal + competitors were

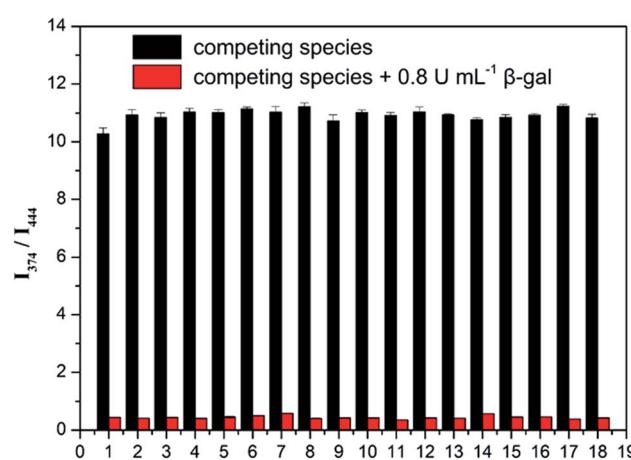
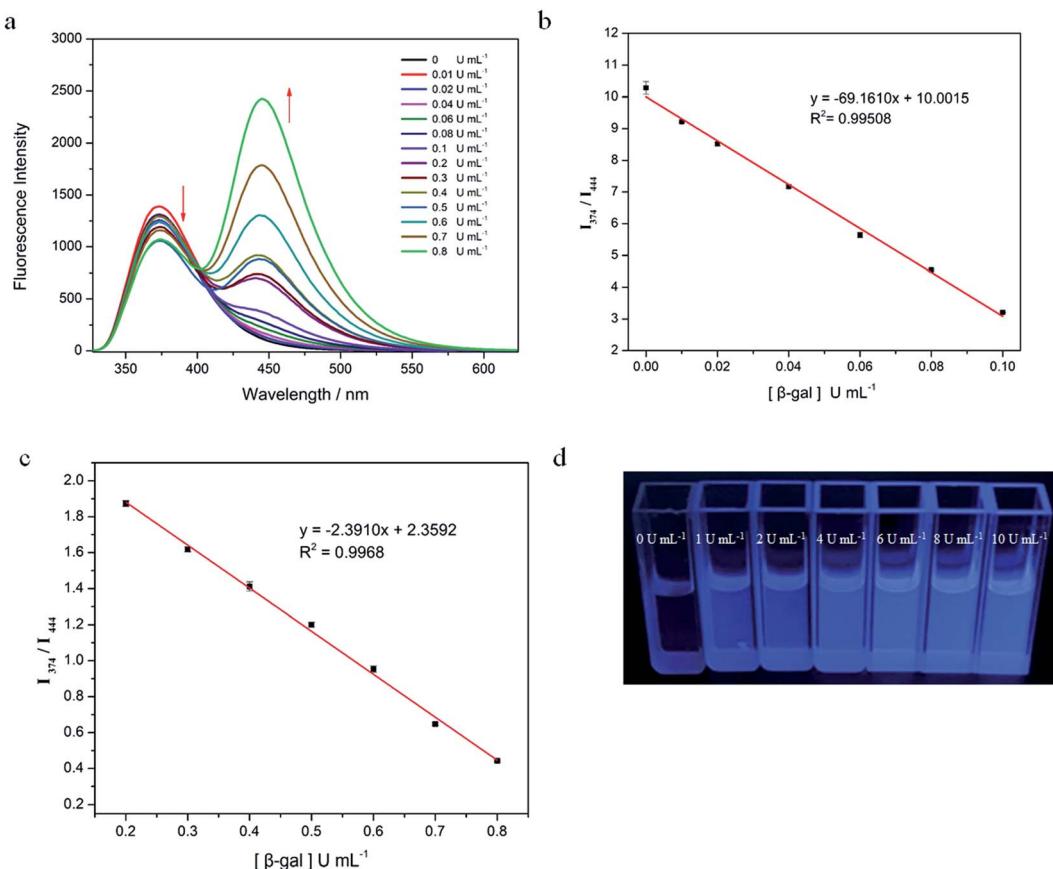


Fig. 3 The fluorescence emission ratio ( $I_{444 \text{ nm}}/I_{374 \text{ nm}}$ ) of Probe 1 (10  $\mu\text{M}$ ) upon addition of various species (0.8  $\text{U mL}^{-1}$  for  $\alpha$ -gal, lipase and 100  $\mu\text{M}$  for others. 1, blank; 2,  $\text{Na}^+$ ; 3,  $\text{K}^+$ ; 4,  $\text{Mg}^{2+}$ ; 5,  $\text{I}^-$ ; 6,  $\text{NH}_4^+$ ; 7,  $\text{Br}^-$ ; 8,  $\text{H}_2\text{O}_2$ ; 9,  $\text{GSH}$ ; 10,  $\text{Gly}$ ; 11,  $\text{Leu}$ ; 12,  $\text{His}$ ; 13,  $\text{Val}$ ; 14,  $\beta$ -glucuronidase; 15, lysozyme; 16, lipase; 17,  $\alpha$ -gal; 18,  $\text{Cl}^-$ . 0.8  $\text{U mL}^{-1}$  for  $\beta$ -gal). Tests were performed in triplicate.





**Fig. 4** (a) Fluorescence spectra of Probe **1** (10  $\mu\text{M}$ ) with  $\beta\text{-gal}$  (0, 0.01, 0.02, 0.04, 0.06, 0.08, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8  $\text{U mL}^{-1}$ ). (b) Plot of fluorescence intensity differences with 0–0.1  $\text{U mL}^{-1}$   $\beta\text{-gal}$ . (c) Plot of fluorescence intensity differences with 0.2–0.8  $\text{U mL}^{-1}$   $\beta\text{-gal}$ . Tests were performed in triplicate. (d) Photograph of Probe **1** (10  $\mu\text{M}$ ) luminescent intensity subjected to  $\beta\text{-gal}$  (0, 1, 2, 4, 6, 8, 10  $\text{U mL}^{-1}$ ) under 254 nm UV light.

almost identical. This indicates that Probe **1** has high selectivity in detecting  $\beta\text{-gal}$ .

The fluorescence-intensity changes of Probe **1** with various activities of  $\beta\text{-gal}$  (0, 0.01, 0.02, 0.04, 0.06, 0.08, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, and 0.8  $\text{U mL}^{-1}$ ) were recorded and are shown in Fig. 4a. The emission ratio ( $I_{444 \text{ nm}}/I_{374 \text{ nm}}$ ) exhibited two linear sections in response to  $\beta\text{-gal}$  activity: 0–0.1  $\text{U mL}^{-1}$  ( $R^2 = 0.9951$ , Fig. 4b) and 0.2–0.8  $\text{U mL}^{-1}$  ( $R^2 = 0.9968$ , Fig. 4c). The Probe **1** limit of detection (LOD) for  $\beta\text{-gal}$  activity was 0.0054  $\text{U mL}^{-1}$ , based on  $\text{LOD} = 3 \text{ SD/B}$ . This indicated that Probe **1** could be used to detect  $\beta\text{-gal}$  activity with a low LOD in water. In addition, the luminous intensity of Probe **1** gradually increased with increases in  $\beta\text{-gal}$  activity, as was observed by the naked eye under 254 nm UV irradiation (Fig. 4d). All results showed that Probe **1** could be used as a quantitative and qualitative tool to detect  $\beta\text{-gal}$  activity.

### 3.3 Recognition mechanism

After the addition of  $\beta\text{-gal}$  (1  $\text{U mL}^{-1}$ ), a new peak appeared, which was previously proven to be compound **1** (Fig. S4, ESI<sup>†</sup>). Mass spectrometry provided further evidence that the  $\beta\text{-galactosides}$  were hydrolyzed (Fig. S5, ESI<sup>†</sup>), with the peak at  $m/z = 174.56.00$  corresponding to compound **1** ( $\text{M} - \text{H}$ ); the peak at  $m/z = 337.04$  was that of Probe **1** ( $\text{M} - \text{H}$ ). These results show that Probe **1**'s mechanism for  $\beta\text{-gal}$  recognition is the  $\beta\text{-gal}$  enzymatic hydrolysis of  $\beta\text{-galactosides}$ .

**3.4 Application**

As  $\beta\text{-gal}$  activity is positively correlated with the maturity of fruit, the development of a simple and highly selective method for  $\beta\text{-gal}$  activity detection in fruit is crucial. Therefore, the ability of Probe **1** to detect  $\beta\text{-gal}$  activity in fruit must be demonstrated.

Pear, apple, grape, strawberry, and kiwi (20  $\mu\text{L}$ ) samples were tested using Probe **1**. The  $\beta\text{-gal}$  activity of the kiwi (ripe) sample was  $0.0938 \pm 0.0027 \text{ U mL}^{-1}$ , and those of the kiwi (unripe), pear, apple, grape, and strawberry samples were 0  $\text{U mL}^{-1}$  (Table 1). The ripeness of the kiwi fruit was mainly determined by the softness or hardness of the fruit (Fig. S6, ESI<sup>†</sup>).

To validate this method, the  $\beta\text{-gal}$  activity in these samples was tested using the  $\beta\text{-galactosidase}$  spectrophotometric method.<sup>40,41</sup> The  $\beta\text{-gal}$  activity of all samples was 0  $\text{U mL}^{-1}$ . No  $\beta\text{-gal}$  activity was detected in the kiwi (ripe) sample by the GB/T 33409-2016 method; this could be because the  $\beta\text{-gal}$  activity in kiwis (ripe) and the LOD of this method are of the same order of magnitude. Then, the addition of  $\beta\text{-gal}$  with different activities (0.02, 0.04, 0.2, and 0.4  $\text{U mL}^{-1}$ ) to the samples showed that the



Table 1 Determination of  $\beta$ -gal activity in real fruit samples

Sample	$\beta$ -gal level found ( $\text{U mL}^{-1}$ )	Added ( $\text{U mL}^{-1}$ )	Found ( $\text{U mL}^{-1}$ )	Recovery/%	RSD/% ( $n = 3$ )
Pear	0	0.02	0.022	108.00	0.11
		0.04	0.043	106.61	0.21
		0.2	0.221	101.32	1.69
		0.4	0.405	101.33	1.44
Apple	0	0.02	0.022	110.29	0.07
		0.04	0.044	108.94	0.33
		0.2	0.192	96.13	2.09
		0.4	0.387	96.71	6.31
Grape	0	0.02	0.022	108.80	0.26
		0.04	0.044	109.83	0.26
		0.2	0.217	108.32	1.40
		0.4	0.402	100.58	0.81
Strawberry	0	0.02	0.021	104.55	0.13
		0.04	0.041	101.86	0.19
		0.2	0.195	97.28	3.68
		0.4	0.376	94.07	1.33
Kiwi (unripe)	0	0.02	0.021	106.4	0.06
		0.04	0.041	103.39	0.08
		0.2	0.203	101.65	0.69
		0.4	0.403	100.63	1.89
Kiwi (ripe)	0.0938 $\pm$ 0.0027	—	—	—	—

recoveries ranged between 94.07–110.29% (Table 1). These results indicate that Probe 1 could accurately and quickly determine the  $\beta$ -gal activity in fruit.

## 4. Conclusions

In this study, a coumarin fluorescent probe (Probe 1) was developed for the ratiometric detection of  $\beta$ -gal activity. The mechanism of  $\beta$ -gal recognition by Probe 1 involves the  $\beta$ -gal enzymatic hydrolysis of  $\beta$ -galactosides. The quantitative range of  $\beta$ -gal activity detected was 0–0.1  $\text{U mL}^{-1}$  and 0.2–0.8  $\text{U mL}^{-1}$ , with an LOD of 0.0054  $\text{U mL}^{-1}$ . This method exhibited good linearity and specificity, had a short analytical time (30 min), required a small amount of sample (20  $\mu\text{L}$ ), and had a low LOD. Additionally, the luminous intensity of Probe 1 gradually increased with increasing enzyme activity. This phenomenon could be observed by the naked eye under 254 nm UV irradiation. Furthermore, Probe 1 was a useful tool for the qualitative and quantitative determination of  $\beta$ -gal activity in fruit.

## Conflicts of interest

There are no conflicts of interest to declare.

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