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A ratiometric fluorescent probe for the detection of β-galactosidase and its application†

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Herein, a coumarin fluorescent probe (Probe 1) was developed for the ratiometric detection of β-galactosidase (β-gal) activity. The detection range was 0–0.1 U mL^{-1} and 0.2–0.8 U mL^{-1}, and the limit of detection (LOD) was 0.0054 U mL^{-1}. Moreover, the luminous intensity of Probe 1 increased gradually with increase in β-gal activity. It could be observed under 254 nm UV irradiation by the naked eye. Furthermore, this method only required a small amount of sample (20 μL) and a short analytical time (30 min) for the detection of β-gal activity with a low LOD. Probe 1 was successfully used to detect β-gal activity in real fruit samples, and can be applied to the quantitative and qualitative detection of β-gal activity.

1. Introduction

β-Galactosidase (β-gal, EC 3.2.1.23) is part of the glycoside hydrolase family and has many microbial sources.[1] In addition to its hydrolysis activity, β-gal from some sources also exhibits glycosylation activity. Food-processing applications of β-gal mainly include the following: allowing lactose-intolerant individuals to consume dairy; improving the sweetness of dairy products; preventing dairy products from crystallizing during freezing; the production of galactooligosaccharides; applications in fermented dairy products; whey processing; the analysis of lactose content in dairy products; and promoting the softening and ripening of fruit and vegetables.[2–4] β-gal has also been used for improving the sweetness, digestibility, flavor, and solubility of dairy products.[5]

β-Galactosidase is widely found in many kinds of plants. Generally, the content of β-gal increases when plants mature,[6] and the activity of β-gal has a positive correlation with the maturity of fruit, including tomato,[7] papaya,[8] apple,[9] persimmon,[10] kiwi fruit,[11] avocado,[12] pear,[13] peach,[14] and mango.[15] It can degrade cell-wall polysaccharides and release free galactose, which can promote – for example – pepper ripening and ethylene production in tomatoes.[7] However, many enzymes exist in different types of fruit. Therefore, the development of a simple, selective, and rapid detection method for β-gal activity in fruit is important.

To date, many methods have been used for the detection of β-gal activity. These include chemiluminescence,[16] HPLC,[17] colorimetric methods,[18] magnetic resonance,[19] UV spectrophotometry,[20] positron emission tomography,[21] enzyme-linked immunosorbent assay technology,[22] and electrochemical methods.[23] However, there are certain disadvantages associated with these methods, such as long experimental duration, complex operation conditions, and high experimental costs. Thus, developing a selective, rapid, and simple detection method for β-gal activity is necessary for applications in fruit production. To this end, the development of novel fluorescent probes is very promising,[24–29] and many kinds of β-gal probes have been reported for use in biological imaging,[30–32] including two-photon fluorescence probes,[33,34] ratiometric probes,[35,36] and turn-on probes.[37–39] However, fluorescent probes used in fruit detection are extremely rare in scientific literature.[40,41]

To develop a simple detection method for β-gal activity in fruit, a ratiometric fluorescent probe (Probe 1) with high accuracy and detection precision was developed. This ratiometric fluorescent probe has the characteristics of high accuracy and strong anti-interference. Further, the luminous intensity of the Probe 1 solution increased with increasing β-gal activity under 254 nm UV irradiation. In addition, Probe 1 was successfully applied to the quantitative and qualitative detection of β-gal activity in fruit.

2. Materials and methods

2.1 General methods

The β-gal, sodium chloride (NaCl), magnesium chloride (MgCl$_2$), hydrogen peroxide (H$_2$O$_2$), ammonium chloride (NH$_4$Cl), sodium bromide (NaBr), glutathione (GSH, 98%), glycine (Gly), D-leucine (Leu), potassium iodide (KI), histidine (His), potassium chloride (KCl), L-valine (Val), β-glucuronidase (from *Escherichia coli*), lysozyme (from chicken egg whites), lipase (from porcine pancreas), and α-galactosidase (α-gal) were purchased from Bailingwei Co., Ltd, China.

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2.2 Instruments

Fluorescence spectra were obtained using the Rili F-4600 fluorescence spectrometer. NMR spectra were obtained using the Bruker AV 300 MHz NMR machine. HRMS was performed using a Bruker Apex IV FTMS.

2.3 Synthesis of Probe 1

7-Hydroxy-4-methylcoumarin (0.18 g, 1.00 mmol), Cs_2CO_3 (1.63 g, 5.00 mmol), Na_2SO_4 (0.36 g, 2.50 mmol), galacto-pyranosyl-1-bromide (3068-32-4, 0.31 g, 0.75 mmol), and CH_3CN (30 mL) were added to a flask (Scheme 1) and reacted for 1 h at 25 °C to obtain compound **2** (0.35 g, 91% yield).

Compound **2** (0.23 g, 0.4 mmol) was dissolved in CH_3OH (20 mL). A solution of K_2CO_3 (0.13 g, 0.09 mmol) and CH_3OH (80 mL) was added and reacted for 4 h at 25 °C, after which the mixture was adjusted to pH 7 using an aqueous H_2SO_4 solution (0.05 M). The precipitate was removed by filtration and recrystallized from ethanol to obtain Probe **1** (0.11 g, 81% yield).

^{1}H NMR (300 MHz, $CDCl_3$) δ (ppm): 7.70 (d, J = 9.4 Hz, 1H), 7.03 (d, J = 7.0 Hz, 2H), 6.24 (s, 1H), 5.21 (s, 1H), 4.98 (d, J = 7.6 Hz, 1H), 4.87 (s, 1H), 4.66 (s, 1H), 4.52 (s, 1H), 3.72 (s, 1H), 3.66 (d, J = 6.1 Hz, 1H), 3.60 (d, J = 8.1 Hz, 1H), 3.53 (s, 2H), 3.45 (s, 1H), 2.41 (s, 3H). ^{13}C NMR (75 MHz, $CDCl_3$) δ (ppm): 160.23, 160.06, 154.39, 153.28, 126.35, 113.99, 113.40, 111.62, 103.17, 100.64, 75.69, 73.20, 70.09, 68.11, 60.37, 18.08. HRMS (ESI): calcd for $[M - H]^-$ 337.092891, found 337.0931.

2.4 Preparation of analytes

Probe **1** was dissolved using DMSO. β-gal, β-glucuronidase, lysozyme, lipase, and α-gal were dissolved in Tris–HCl (pH 7.3, 50 mM) and frozen in a −20 °C refrigerator. The stock solution was diluted using a certain concentration gradient with Tris–HCl (pH 7.3, 50 mM) before each use. NaCl, KCl, $MgCl_2$, KI, NH_4Cl, NaBr, H_2O_2, GSH, Gly, Leu, His, and Val, were dissolved in deionized water.

2.5 Preparation of samples

Pears, apples, grapes, strawberries, and kiwis were bought from a local supermarket. Twenty grams of the abovementioned fruit were centrifuged for 10 min, at 12 000 rpm, after grinding; subsequently, the supernatant was obtained, which was filtered and reserved.

2.6 Fluorescence detection assays

The Probe **1** solution (1 mM, 0.02 mL) was added to a cuvette, to which 2 mL of water was added. Then, β-gal was added; after 40 min, the mixture was analyzed using fluorescence spectra (slit width = 5 nm, λ_{ex} = 327 nm, voltage = 500 V, and temperature = 37 °C).

3. Results and discussion

3.1 Probe preparation

Probe **1** was synthesized in two steps. First, compound **2** was obtained through the nucleophilic substitution of 7-hydroxy-4-methylcoumarin with galacto-pyranosyl-1-bromide (3068-32-4). Second, the acetyl groups of compound **2** were hydrolyzed to obtain Probe **1** (Scheme 1). Probe **1** was purified by recrystallization from ethanol, and NMR and HRMS were used to characterize this probe (Fig. S1–S3, ESI†). The preparation process was carried out at 25 °C under mild conditions; thus, the synthesis of Probe **1** was a simple process.

3.2 Fluorescence properties

First, the fluorescence response of Probe **1** with β-gal in different solutions (CH_3CN, H_2O, DMSO, C_2H_5OH, and THF) was determined (Fig. 1a). After the addition of β-gal, the fluorescence intensity was decreased at 374 nm in CH_3CN, DMSO, C_2H_5OH, and THF, with changes in only one fluorescent emission. In H_2O, however, the fluorescence intensity was decreased at 374 nm and significantly increased at 444 nm. As shown in Fig. 1b, Probe **1** displayed a fluorescence emission peak at 374 nm in an H_2O solution. After the addition of β-gal, the fluorescence intensity at 374 nm decreased and a peak appeared at 444 nm. The fluorophore (7-hydroxy-4-methylcoumarin, compound **1**) had a peak at 444 nm in an H_2O solution. This confirmed that 374 nm was the peak of Probe **1** and 444 nm was the fluorescence-emission peak of the fluorophore. These results indicated that Probe **1** was
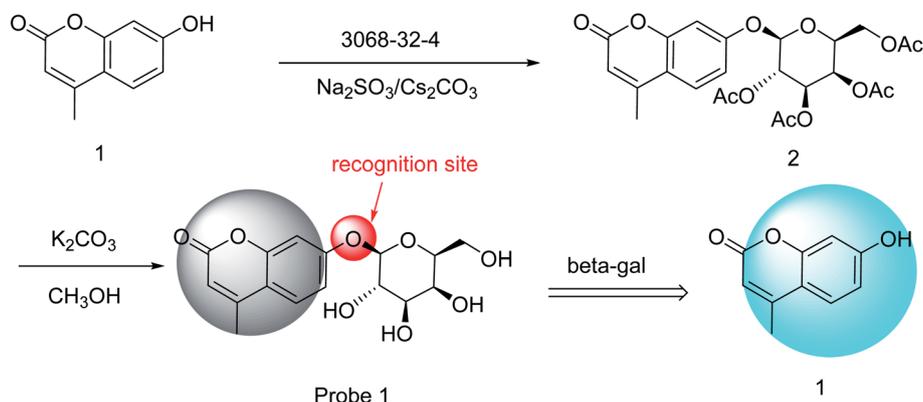
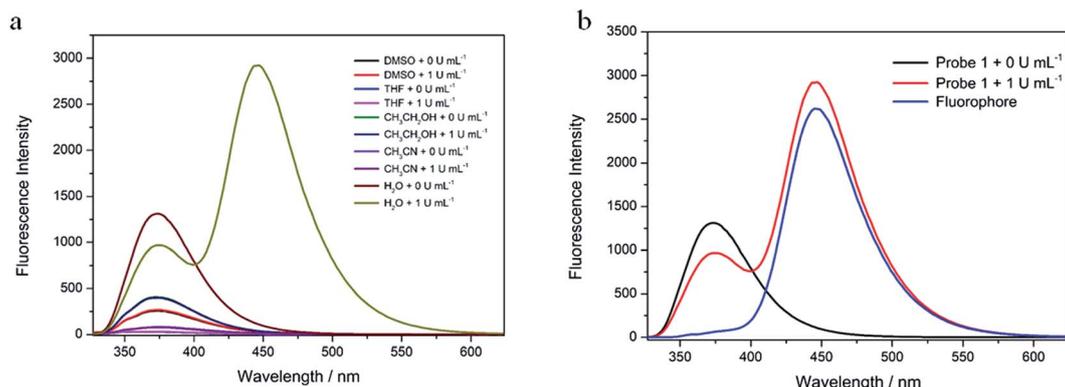**Scheme 1** Synthesis of Probe **1** and the recognition mechanism of Probe **1** to β-gal.

Fig. 1 (a) Fluorescence spectra of Probe **1** (10 μM) and Probe **1** in the presence of β-gal (1 U mL^{-1}) in CH$_3$CN, H$_2$O, DMSO, C$_2$H$_5$OH and THF at 37 °C. (b) Fluorescence spectra of Probe **1** (10 μM), compound **1** (10 μM) and Probe **1** in the presence of β-gal (1 U mL^{-1}) in H$_2$O at 37 °C.

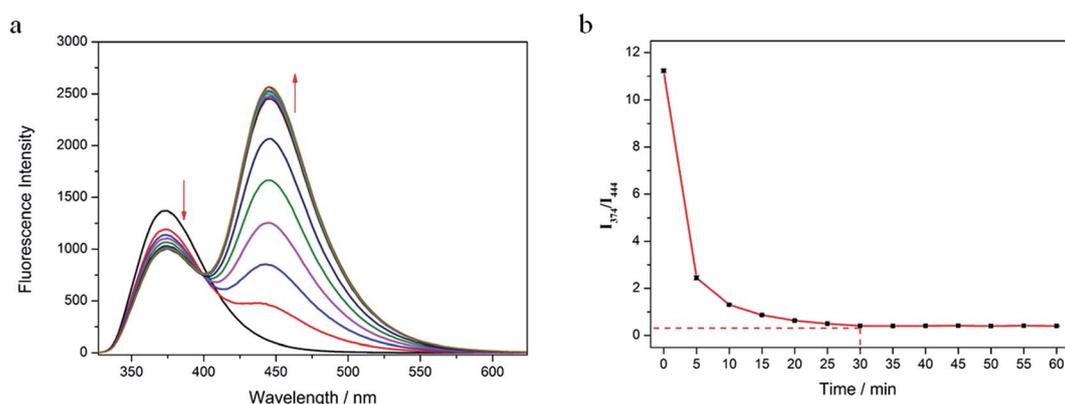
Fig. 2 (a) Time-dependent fluorescence spectra of Probe **1** (10 μM) in the presence of β-gal (1 U mL^{-1}) in water at 37 °C. (b) The fluorescence emission ratio ($I_{444\ nm}/I_{374\ nm}$) of Probe **1** in the presence of β-gal form 0 min to 60 min. Tests were performed in triplicate.

a ratiometric probe, facilitating the detection of β-gal by determining the ratio of fluorescence intensity at two different emission wavelengths.

Second, the time-response relationships of Probe **1** toward β-gal in water were tested (Fig. 2a). The fluorescence intensity at 374 nm decreased slowly with the addition of β-gal. Simultaneously, the fluorescence intensity at 444 nm increased rapidly. The fluorescence emission ratio ($I_{444\ nm}/I_{374\ nm}$) decreased rapidly from 0 to 10 min, and reached an equilibrium in 30 min (Fig. 2b). The emission ratio remained unchanged from 30 to 60 min. This shows that 30 min were required for the identification of β-gal by Probe **1**, which was set as the duration for the subsequent experiments.

The effects of competitor ions and compounds were used to ascertain the selectivity of Probe **1**. Various competitors were tested, including Na$^+$, K$^+$, Mg^{2+}, I$^-$, NH$_4^+$, Br$^-$, H$_2$O$_2$, Cl$^-$, GSH, Gly, Leu, Val, His, β-glucuronidase, lipase, and lysozyme (Fig. 3). In the presence of any of these competing ions and compounds, there was minimal change in the emission ratio ($I_{444\ nm}/I_{374\ nm}$). In particular, Probe **1** was almost unresponsive in the presence of β-glucuronidase and α-gal. However, with the addition of β-gal, the emission ratios ($I_{444\ nm}/I_{374\ nm}$) of Probe **1** + β-gal and Probe **1** + β-gal + competitors were
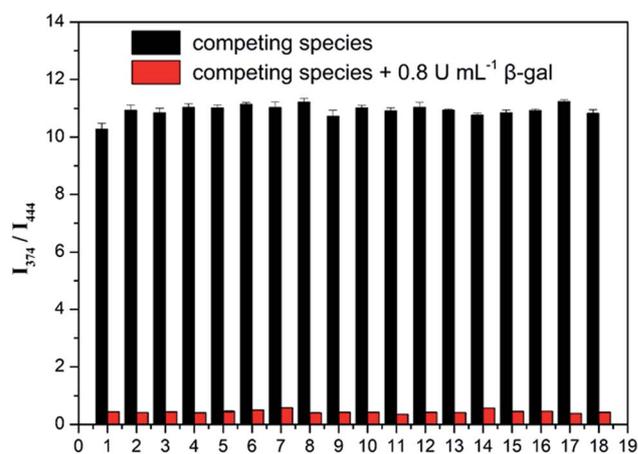

Fig. 3 The fluorescence emission ratio ($I_{444\ nm}/I_{374\ nm}$) of Probe **1** (10 μM) upon addition of various species (0.8 U mL^{-1} for α-gal, lipase and 100 μM for others. 1, blank; 2, Na$^+$; 3, K$^+$; 4, Mg^{2+}; 5, I$^-$; 6, NH$_4^+$; 7, Br$^-$; 8, H$_2$O$_2$; 9, GSH; 10, Gly; 11, Leu; 12, His; 13, Val; 14, β-glucuronidase; 15, lysozyme; 16, lipase; 17, α-gal; 18, Cl$^-$. 0.8 U mL^{-1} for β-gal). Tests were performed in triplicate.
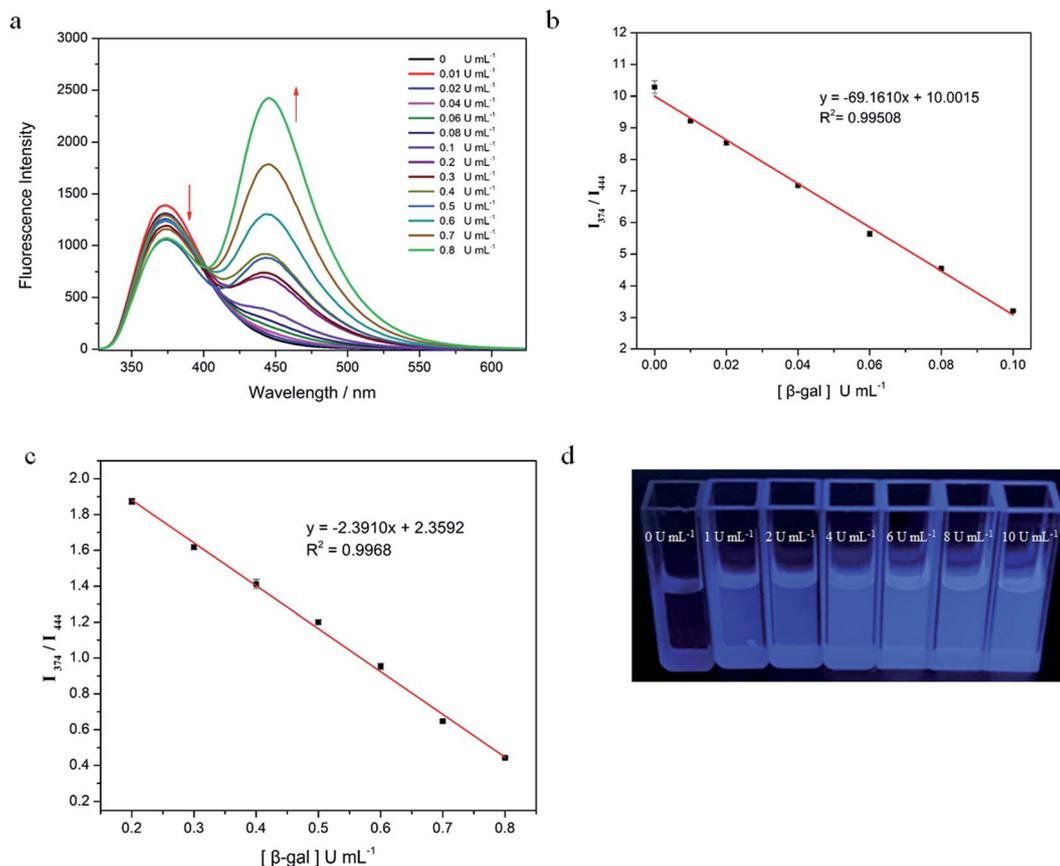
Fig. 4 (a) Fluorescence spectra of Probe **1** (10 μM) with β-gal (0, 0.01, 0.02, 0.04, 0.06, 0.08, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8 U mL^{-1}). (b) Plot of fluorescence intensity differences with 0–0.1 U mL^{-1} β-gal. (c) Plot of fluorescence intensity differences with 0.2–0.8 U mL^{-1} β-gal. Tests were performed in triplicate. (d) Photograph of Probe **1** (10 μM) luminescent intensity subjected to β-gal (0, 1, 2, 4, 6, 8, 10 U mL^{-1}) under 254 nm UV light.

almost identical. This indicates that Probe **1** has high selectivity in detecting β-gal.

The fluorescence-intensity changes of Probe **1** with various activities of β-gal (0, 0.01, 0.02, 0.04, 0.06, 0.08, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, and 0.8 U mL^{-1}) were recorded and are shown in Fig. 4a. The emission ratio ($I_{444\ nm}/I_{374\ nm}$) exhibited two linear sections in response to β-gal activity: 0–0.1 U mL^{-1} ($R^2 = 0.9951$, Fig. 4b) and 0.2–0.8 U mL^{-1} ($R^2 = 0.9968$, Fig. 4c). The Probe **1** limit of detection (LOD) for β-gal activity was 0.0054 U mL^{-1}, based on LOD = 3 SD/B. This indicated that Probe **1** could be used to detect β-gal activity with a low LOD in water. In addition, the luminous intensity of Probe **1** gradually increased with increases in β-gal activity, as was observed by the naked eye under 254 nm UV irradiation (Fig. 4d). All results showed that Probe **1** could be used as a quantitative and qualitative tool to detect β-gal activity.

3.3 Recognition mechanism

After the addition of β-gal (1 U mL^{-1}), a new peak appeared, which was previously proven to be compound **1** (Fig. S4, ESI†). Mass spectrometry provided further evidence that the β-galactosides were hydrolyzed (Fig. S5, ESI†), with the peak at m/z = 174.56.00 corresponding to compound **1** (M − H); the peak at m/z = 337.04 was that of Probe **1** (M − H). These results show that Probe **1**'s mechanism for β-gal recognition is the β-gal enzymatic hydrolysis of β-galactosides.

3.4 Application

As β-gal activity is positively correlated with the maturity of fruit, the development of a simple and highly selective method for β-gal activity detection in fruit is crucial. Therefore, the ability of Probe **1** to detect β-gal activity in fruit must be demonstrated.

Pear, apple, grape, strawberry, and kiwi (20 μL) samples were tested using Probe **1**. The β-gal activity of the kiwi (ripe) sample was 0.0938 ± 0.0027 U mL^{-1}, and those of the kiwi (unripe), pear, apple, grape, and strawberry samples were 0 U mL^{-1} (Table 1). The ripeness of the kiwi fruit was mainly determined by the softness or hardness of the fruit (Fig. S6, ESI†).

To validate this method, the β-gal activity in these samples was tested using the β-galactosidase spectrophotometric method.[40,41] The β-gal activity of all samples was 0 U mL^{-1}. No β-gal activity was detected in the kiwi (ripe) sample by the GB/T 33409-2016 method; this could be because the β-gal activity in kiwis (ripe) and the LOD of this method are of the same order of magnitude. Then, the addition of β-gal with different activities (0.02, 0.04, 0.2, and 0.4 U mL^{-1}) to the samples showed that the

Table 1 Determination of β-gal activity in real fruit samples

Sample	β-gal level found (U mL^{-1})	Added (U mL^{-1})	Found (U mL^{-1})	Recovery/%	RSD/% ($n = 3$)
Pear	0	0.02	0.022	108.00	0.11
		0.04	0.043	106.61	0.21
		0.2	0.221	101.32	1.69
		0.4	0.405	101.33	1.44
Apple	0	0.02	0.022	110.29	0.07
		0.04	0.044	108.94	0.33
		0.2	0.192	96.13	2.09
		0.4	0.387	96.71	6.31
Grape	0	0.02	0.022	108.80	0.26
		0.04	0.044	109.83	0.26
		0.2	0.217	108.32	1.40
		0.4	0.402	100.58	0.81
Strawberry	0	0.02	0.021	104.55	0.13
		0.04	0.041	101.86	0.19
		0.2	0.195	97.28	3.68
		0.4	0.376	94.07	1.33
Kiwi (unripe)	0	0.02	0.021	106.4	0.06
		0.04	0.041	103.39	0.08
		0.2	0.203	101.65	0.69
		0.4	0.403	100.63	1.89
Kiwi (ripe)	0.0938 ± 0.0027	—	—	—	—

recoveries ranged between 94.07–110.29% (Table 1). These results indicate that Probe **1** could accurately and quickly determine the β-gal activity in fruit.

4. Conclusions

In this study, a coumarin fluorescent probe (Probe **1**) was developed for the ratiometric detection of β-gal activity. The mechanism of β-gal recognition by Probe **1** involves the β-gal enzymatic hydrolysis of β-galactosides. The quantitative range of β-gal activity detected was 0–0.1 U mL^{-1} and 0.2–0.8 U mL^{-1}, with an LOD of 0.0054 U mL^{-1}. This method exhibited good linearity and specificity, had a short analytical time (30 min), required a small amount of sample (20 μL), and had a low LOD. Additionally, the luminous intensity of Probe **1** gradually increased with increasing enzyme activity. This phenomenon could be observed by the naked eye under 254 nm UV irradiation. Furthermore, Probe **1** was a useful tool for the qualitative and quantitative determination of β-gal activity in fruit.

Conflicts of interest

There are no conflicts of interest to declare.

Acknowledgements

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