


Cite this: *RSC Adv.*, 2021, **11**, 15695

Tunable electron property induced by B-doping in g-C₃N₄[†]

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Graphitic carbon nitrides are a research hotspot of two-dimensional (2D) materials, which attract more and more attention from researchers. Topological properties are a focus in graphitic carbon nitrides materials. Using first-principles calculations, we modified the g-C₃N₄ (formed by tri-s-triazine) by B atoms, proposing a novel two-dimensional monolayer, g-C₆N₇B, which showed excellent stability verified by positive phono modes, molecular dynamic simulations and mechanical criteria. The valence band and conduction band touch at the Γ point. Interestingly, g-C₆N₇B is topologically nontrivial, because the valence and conduction band can be gapped by the spin-orbit coupling (SOC) effect associated with robust gapless edge states. Additionally, molecular dynamic simulations indicate that g-C₆N₇B will still maintain good geometry structure when the temperature is as high as 1500 K. The flexibility of g-C₆N₇B is confirmed by its elastic constants and Young's moduli. This work opens an avenue for graphitic carbon nitride materials with topological properties.

Received 7th January 2021
Accepted 18th April 2021

DOI: 10.1039/d1ra00149c

rsc.li/rsc-advances

Introduction

Since the successful synthesis of the graphene monolayer,¹ it has shown many excellent properties. However, another type of two-dimensional monolayer, graphitic carbon nitrides, which have a similar framework to graphene, have attracted a great deal of researchers' attention in recent years. In fact, graphitic carbon nitrides were studied by Franklin as early as 1922.² There are two prevailing units, *s*-triazine (C₃N₃) and tri-*s*-triazine (C₆N₇), forming the framework of graphitic carbon nitrides materials.^{3–7} These frameworks were synthesized by chemical synthetic approaches under relatively mild conditions.^{6,8–14} Among the graphitic carbon nitrides, the framework with the formula of C₃N₄ (named g-C₃N₄), has a major advantage, that is, high thermal and chemical stability.¹⁵ Initially, researchers found that g-C₃N₄ can be applied in many fields, such as fuel cells, hydrogen production, and photocatalysis.^{16–19} It is disappointing to note that the gap of g-C₃N₄ is 2.88 eV,²⁰ which is too wide induced the utilization of visible light in the photocatalytic reaction is too low.^{21,22} Doping or replanting element is an effective strategy to improve the performance of g-C₃N₄.^{23–32} Based on the above studies, g-C₃N₄ has been found many other exotic properties in a similar way, such as spin polarize^{33–35} and topological properties.³⁶

Topological insulators (TIs) are a novel quantum state, which with a bulk gap under spin-orbit interaction and protected by gapless edge states or surface states, leading to the quantum spin Hall (QSH) effect.³⁷ Graphene is theoretical predicted to be a two-dimensional (2D) topological insulator,³⁸ which was considered to be the first material capable of achieving the QSH effect. Then, because of special lattice structure and stable edge states of 2D materials, 2D TIs become a new research hotspot in condensed matter physics. As we all know, g-C₃N₄ has great potential in the catalytic application. However, the topological study of 2D graphitic carbon nitrides monolayer is quite rarely. A. Zhu *et al.* proposed a new 2D topological insulator material, g-C₆N₆ in which the *s*-triazines are separated *via* C–C bonds.³⁹ Therefore, we used one B atom to replace the central N atom in the g-C₃N₄, which is composed of tri-*s*-triazine unit with a chemical ratio is C₆N₈, formed g-C₆N₇B, which is a topological 2D material.

In this work, we study the electronic, topological, and mechanical properties of g-C₆N₇B on the basis of first-principles calculations. Firstly, the dynamical and mechanical stability of g-C₆N₇B was verified from the phonon spectrum, molecular dynamics simulations, and elastic constants, respectively. Our calculated results indicated that the conduction band and valence band of g-C₆N₇B touch with each other at Γ point. The conduction and valence band opened a small gap with spin-orbit coupling (SOC) interaction. Additionally, we also calculated the edge states of g-C₆N₇B, which showed that the band structure of edge states is gapless. More interestingly, molecular dynamics simulations calculations showed that the g-C₆N₇B monolayer can maintain its original configuration at temperatures up to 1500 K. And, the in-plane Young's moduli are only

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[†] Electronic supplementary information (ESI) available. See DOI: 10.1039/d1ra00149c


two-five of that of graphene, revealed the ultra-softness of g-C₆N₇B monolayer. More interesting, there has two absorption peaks in ultraviolet region.

Method and computational details

We performed first-principles calculations based on density functional theory (DFT), implemented by the Vienna *ab initio* simulation package (VASP).⁴⁰ The generalized gradient approximation (GGA) in the form of the Perdew–Burke–Ernzerhof (PBE) was adopted for the exchange–correlation functional.⁴¹ The electron–ion interaction was described by projector-augmented-wave (PAW) potentials.⁴² The energy cutoff employed for plane-wave expansion of electron wavefunction was set to 400 eV. The Brillouin zone (BZ) integration was sampled on a grid of $6 \times 6 \times 1$ *k*-points according to the Monkhorst–Pack method for structural optimization. Two-dimensional periodic boundary conditions were employed in the *x*–*y* plane, while a vacuum space up to 20 Å was applied along the *z*-direction to exclude the interaction between neighboring images. Structural optimizations were carried out using a conjugate gradient (CG) method until the remaining force on each atom was less than 0.01 eV Å^{−1}. The phonon spectra were calculated by using the Phonopy code^{43,44} interfaced with VASP. In the whole calculation process, the correction of dispersion term is not considered. The reason is that it does not affect the properties of the system, showed in Fig. S1.†

Results and discussion

We doped one B atom in a C₆N₇ unit replaced the center N atom, which named g-C₆N₇B, the geometry structure as shown in Fig. 1(a). The optimized lattice constants are $a = b = 7.12$ Å, which slightly deviating from that of the non-doped g-C₃N₄ monolayer, 7.14 Å.³⁴ And, the optimized structure of g-C₃N₄B with slight deformation compared with the g-C₃N₄ framework. The optimized g-C₆N₇B structure has the C–N bond lengths of 1.34 Å (within a triazine ring) and 1.43 Å (between adjacent triazine rings), respectively. The bond lengths within the triazine ring are very similar to that of g-C₃N₄ (1.33 Å). However, the C–N bond lengths between the adjacent triazine rings are

shorter than that of g-C₃N₄ (1.46 Å). Additionally, the C–B bond lengths are 1.49 Å.

Next, we discussed the stability of g-C₆N₇B by different methods. First, we confirmed the dynamic stability of g-C₆N₇B from the phonon spectrum obtained by using a supercell approach within the PHONOPY code. From Fig. 1(b), it can be clearly seen that the phonon spectrum is free from imaginary frequency modes. Remarkably, the highest frequency of g-C₆N₇B reaches up to 1422 cm^{−1}, which much higher than those of MoS₂ monolayer (473 cm^{−1}),⁴⁵ silicene (580 cm^{−1})⁴⁶ and even higher than that of h-BN (1301 cm^{−1}),⁴⁷ suggesting strong bonds in the g-C₆N₇B.

To verify the stability of g-C₆N₇B at room temperature (300 K), we performed the molecular dynamics simulations with the

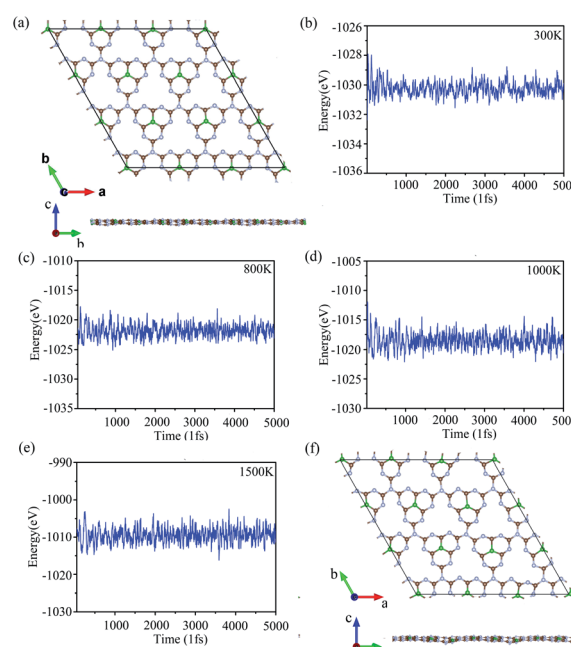


Fig. 2 (a) and (f) The geometry structure of g-C₆N₇B with respect to the simulation time at 300 K and 1500 K. (b)–(e) The total energy fluctuations with respect to the simulation time at 300 K, 800 K, 1000 K, 1500 K.

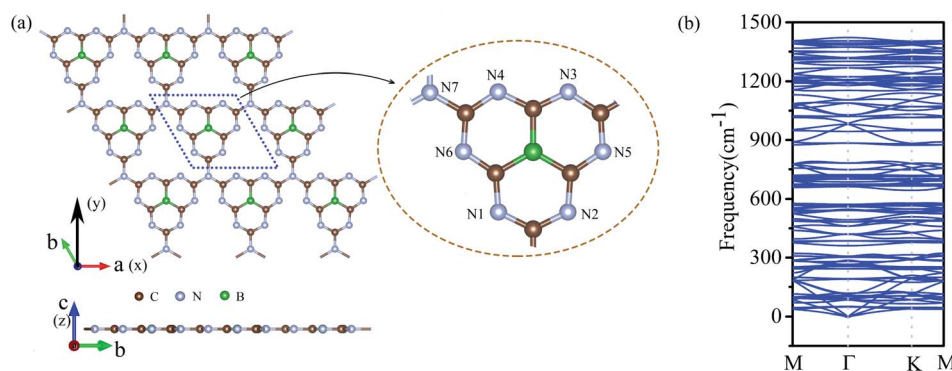


Fig. 1 (a) The top (upper) and the side (lower) view of the geometric structure of g-C₆N₇B monolayer. (b) The phonon spectrum of g-C₆N₇B along the highly symmetric directions in the BZ.



Nose–Hoover thermostat at 300 K and a time step of 1.0 fs. After running 5000 steps, the geometry of g-C₆N₇B was well preserved without any structure collapse and the total energy of the system was converged in this time scale, as shown Fig. 2 (a) and (b). Due to previous reports, we found that the non-doped g-C₃N₄ and B/S/P doped g-C₃N₄ is synthesized at high temperature^{25,26,48} and g-C₆N₇B has the same geometry with g-C₃N₄. We performed the molecular dynamic simulation at 800 K, 1000 K, 1500 K, the total energy of the system was converged after running 5000 fs, showed in Fig. 2(c)–(e). We also checked the geometries structure of g-C₆N₇B, found that the geometries structure maintained very well compared with the original the configuration of g-C₆N₇B at temperatures up to 1500 K, as shown in Fig. 2(f). The result showed that g-C₆N₇B is a high temperature resistant material. The above results indicated that the g-C₆N₇B exhibits high thermal stability.

To investigate the electronic property of g-C₆N₇B, we calculated its electronic band structure by using the first-principle theory without spin–orbit coupling (SOC). From Fig. 3(a), it is very clear to see that the conduction and valence bands meet each other at Γ point featured a gapless semi-conduction, which is different from the linear dispersion at K point of valence and conduction band in the band structure of graphene. In addition, the characteristic can also be verified the total density of states (TDOS) of g-C₆N₇B, showed in Fig. S2.† And, due to the flat conduction band, there has a significant peak near the Fermi level, showed in Fig. S2.† To show the features clearly, we plotted the electronic states near the meeting point in a 2D wave vector mesh, as shown in Fig. 3(c), which indicated the conduction and valence bands touched at Γ point. Then, by considering SOC, our DFT calculated results showed a bandgap of 5 meV opened at the Fermi level, as shown in Fig. 3(b). More detail information with the difference between the band structure with and without SOC shown in Fig S3.† The small gap is due to the weak SOC effect of the light C, N, and B elements. However, this value is much higher than the magnitude of

graphene ($\sim 10^{-3}$ meV).⁴⁹ Furthermore, we used the maximally localized Wannier functions (MLWFs) method to fit the band structure within energy range $[E_f - 0.3, E_f + 0.4]$ eV. It can be seen from Fig. 3(b) that the SOC gap of DFT calculations are good to agree with that by employing Wannier interpolation.

The occurrence of the energy gap due to the SOC effect reminds us of the topological nontriviality of g-C₆N₇B. To further demonstrate the topological nontriviality, we constructed the edge Green's function of the semi-infinite g-C₆N₇B lattice from MLWFs by using the Wannier90 package.⁵⁰ The local density of states (LDOS) of the edge calculated from the MLWFs is plotted in Fig. 3(d). From the figure, we can see that the gapless edge state appears in the gap between the valence band and conduction band. The existence of an edge state confirms the topological nontriviality of g-C₆N₇B monolayer.

To analyzed the origin of electronic states near the Fermi level at Γ point, we plotted the orbital-resolved band structure projected onto different atomic orbitals, as shown in Fig. 4(b)–(k). The p_x orbital of N1/N5, p_y orbital of N3/N4, and $p_x + p_y$ orbitals of N2/N6/B mainly to contribute the bands near the Fermi level at Γ point. However, the contribution of C atoms and joint N atoms can be neglected. The orbital-resolved band structure of other atoms was showed in Fig. S4.† The Dirac cone electronic states near Fermi level of graphene come from the p_z orbital of each C atom. We also calculated the electron density profile of the Kohn–Sham wave functions at the Γ point near the Fermi level, as shown in Fig. 4(1). It is very clearly to see that the electronic states near the Fermi level are almost contributed by p orbital of N atoms and B atoms, which are in good agreement with the orbital-resolved band structure.

Mechanical property

Then, we verified the mechanical stability of g-C₆N₇B by using the standard Voigt notation^{51,52} and studied its mechanical properties. The elastic energy per unit area for a 2D sheet can be expressed as:

$$U = \frac{1}{2}C_{11}\varepsilon_{xx}^2 + \frac{1}{2}C_{22}\varepsilon_{yy}^2 + C_{12}\varepsilon_{xx}\varepsilon_{yy} + 2C_{66}\varepsilon_{xy}^2$$

where ε_{xx} and ε_{yy} represent the uniaxial in-plane stains along x- and y-direction and ε_{xy} is the shear strain. The parameters C_{11} , C_{22} , C_{12} and C_{66} represent the components of the elastic modulus tensor, corresponding to the second partial derivation of strain energy with respect to strain. The elastic constants can be obtained by fitting the energy curves associated with uniaxial and equi-biaxial strain, which are $C_{11} = 164.53$ GPa nm, $C_{22} = 169.65$ GPa nm, $C_{12} = 74.73$ GPa nm, $C_{66} = 40.44$ GPa nm, respectively. The elastic constants satisfy the criteria of the mechanical stability of 2D sheet: $C_{11} \times C_{22} - C_{12}^2 > 0$, $C_{66} > 0$, indicating that the g-C₆N₇B monolayer is mechanically stable. To study the mechanical properties of g-C₆N₇B, we calculated Young's modulus. The direction-dependent in-plane Young's moduli (E_x and E_y) along the x- and y-direction derived from the

elastic constants using the formulas: $E_x = \frac{C_{11} \times C_{22} - C_{12}^2}{C_{22}}$ and $E_y = \frac{C_{11} \times C_{22} - C_{12}^2}{C_{11}}$ are 131.61 N m^{-1} and 131.71 N m^{-1} ,

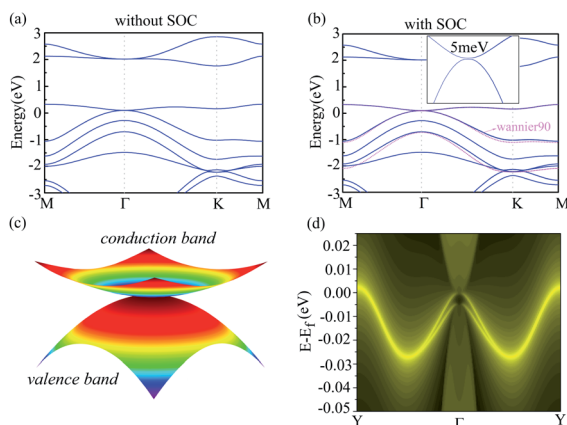


Fig. 3 (a) and (b) Electronic band structure of g-C₆N₇B without SOC and with SOC respectively. (c) The 3D band structure of conduction band and valence band. The Fermi level was set to zero. (4) The local electronic density (LDOS) calculated from edge Green's function of the semi-infinite g-C₆N₇B lattice.



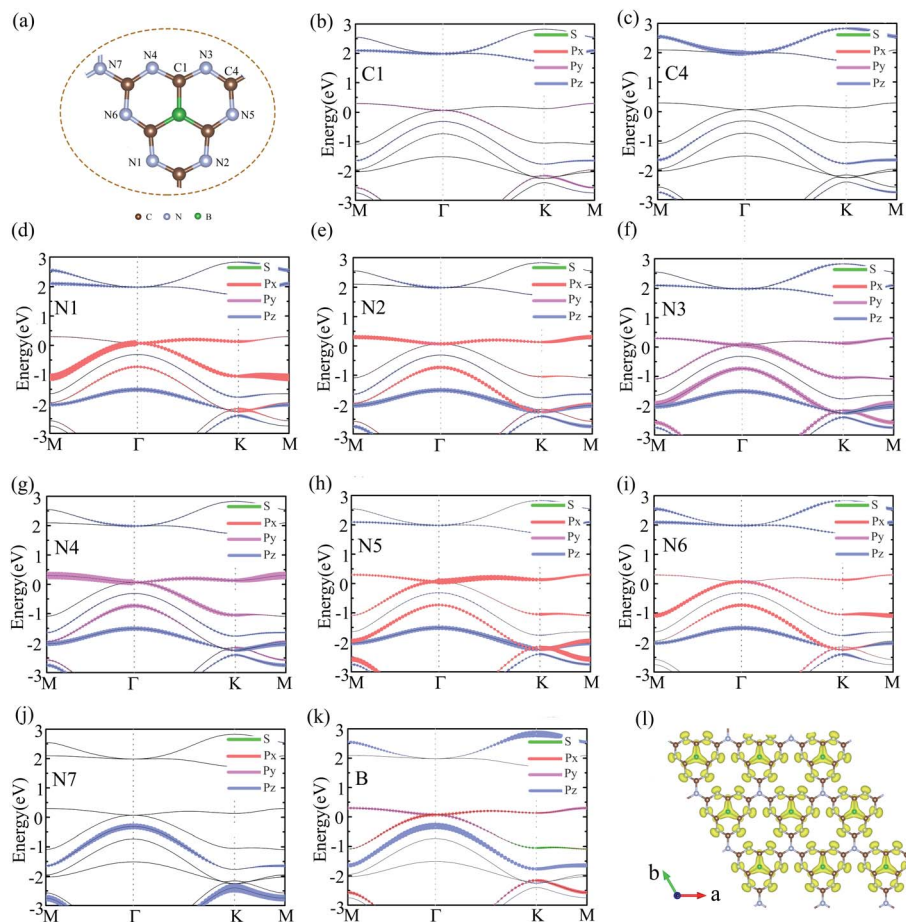


Fig. 4 (a) The geometry structure of g-C₆N₇B. (b)–(k) The orbital-resolved band structure of the g-C₆N₇B on different atoms, labelled in (a). (l) The electron density isosurfaces of the electron Kohn-Sham wave functions at Γ point near Fermi level.

respectively, which is only two-five of that of graphene (335N/m),^{53,54} suggested that the g-C₆N₇B is a promising flexible material compared with graphene.

Optical property

Finally, we calculated the light absorption spectrum of g-C₆N₇B. From the Fig. 5, it showed that the light absorption spectral

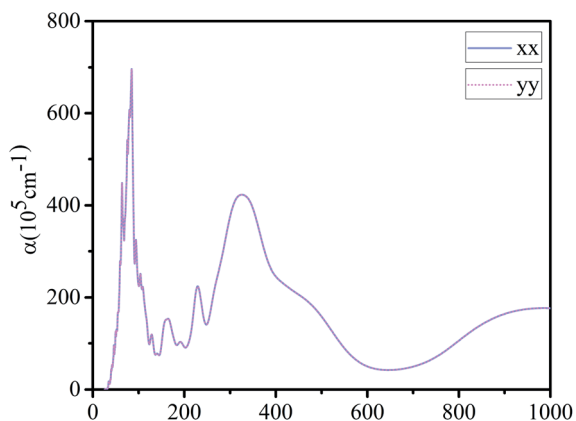


Fig. 5 Absorption coefficient of g-C₆N₇B along x and y direction.

curves of g-C₆N₇B in x and y directions are the same with each other, indicating that the absorption capacity of g-C₆N₇B for different wavelengths of light shows isotropic. There are two absorption peaks in the optical absorption spectrum at 85 nm and 327 nm, which belong to the ultraviolet region. However, the peak of the absorption coefficient of graphene (x direction) at 283 nm,⁵⁵ which also located in the ultraviolet region.

Conclusion

In this paper, we replaced the central N atom in g-C₃N₄ with the B atom to form a new 2D monolayer, named g-C₆N₇B. The introduction of the B atom leads to novel electronic properties of g-C₆N₇B. Our calculated results showed that the conduction band and valence band touch at Γ point with gapless. However, SOC opened small band gaps between the conduction band and valence band. In addition, the topological nontrivial property of g-C₆N₇B was further confirmed by the appearance of gapless edge states in the g-C₆N₇B nanoribbon. The stability and plausibility of g-C₆N₇B were verified by positive phonon modes, molecular dynamical simulations, and mechanical criteria. We also calculated the molecular dynamical simulations at high temperatures, found that g-C₆N₇B represents excellent stability



even at 1500 K. Additionally, g-C₆N₇B is a promising flexible material. There are two absorption peaks in the optical absorption spectrum at 85 nm and 327 nm, which belong to the ultraviolet region. These results not only enrich the graphitic carbon nitrides materials with topological property, but also establish fascinating properties of fundamental significance and practical interest.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work is supported by the Doctor Foundation of Shandong Jianzhu University (No. X20017Z), National Natural Science Foundation of China (No. 22005087).

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