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Intermolecular difunctionalization of alkenes: synthesis of β -hydroxy sulfides

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Direct difunctionalization of carbon–carbon double bonds is one of the most powerful tools available for concomitant introduction of two functional groups into olefinic substrates. In this context, vicinal hydroxysulfenylation of unactivated alkenes has emerged as a novel and straightforward strategy for the fabrication of β -hydroxy sulfides, which are extremely valuable starting materials in constructing various natural products, pharmaceuticals, and fine chemicals. The aim of this review is to summarize the most representative and important reports on the preparation of β -hydroxy sulfides through intermolecular hydroxysulfenylation of the corresponding alkenes with special emphasis on the mechanistic features of the reactions.

1. Introduction

Organosulfur compounds are ubiquitous in natural products, pharmaceuticals, and agrochemicals.^{1–3} In particular, organic

sulfides widely exist in many useful commercialized drugs.⁴ In this family of organosulfur compounds, β -hydroxy containing sulfides have recently attracted significant attention due to their fascinating biological activities, such as antimicrobial, anti-tumor, anti-HIV, and anti-inflammatory activities (Scheme 1).^{4,5} One of the most straightforward synthetic procedure for the preparation of β -hydroxy sulfides is the thiolysis of epoxides with thiols⁶ or disulfides.⁷ However, this pathway generally suffers from certain disadvantages, such as low yields, poor regioselectivity, harsh reaction conditions, and the formation of undesirable by-products *via* rearrangement of epoxides and oxidation of thiols. Thus, the development of a convenient and

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truly efficient synthetic strategy for the preparation of the titled compounds remains a challenge.

Direct difunctionalization of C–C unsaturated bonds is one of the most powerful synthetic methodologies toward polyfunctionalized molecules in a very simple manner.⁸ In this context, 1,2-hydroxysulfenylation of alkenes with various sulfonyl sources provides an excellent route to the synthesis of β -hydroxy sulfides in one step (Fig. 1). High atom- and step-economy efficiency, high functional group tolerance, good regioselectivity, and the use of inexpensive easily available starting materials are the main advantages motioned for this page of β -hydroxy sulfide synthesis. Recently, Kine and co-workers highlighted this synthetic strategy in their interesting review paper entitled “ β -hydroxy sulfides and their syntheses”.⁹ However, several important examples were omitted and generally the explanation of mechanistic pathways of the reactions were ignored. In connection with our recent works on organo-sulfur chemistry¹⁰ and modern organic synthesis,¹¹ herein we

will attempt to provide a comprehensive overview of the synthesis of β -hydroxy sulfides through intermolecular difunctionalization of the respective alkenes with particular emphasize on the mechanistic aspect of reactions which may allow possible new insights into catalyst improvement.

2. Metal-catalyzed/mediated reactions

The fabrication of β -hydroxy sulfides through metal-mediated direct hydroxysulfenylation of alkenes was accomplished first in 1978 by Trost *et al.*¹² They showed that treatment of aliphatic alkenes **1** with aromatic disulfides **2** in the presence of $\text{Pb}(\text{OAc})_4$ as an oxidant and $\text{CF}_3\text{CO}_2\text{H}$ as the source of the hydroxyl oxygen in DCM furnished trifluoroacetoxy sulfide intermediate **I** that after hydrolysis in the work-up converted to the corresponding β -hydroxy sulfides **3** in moderate to almost quantitative yields (Scheme 2). The reaction is notable in that both terminal and internal alkenes were tolerated. However, the regioselectivity was poor and in the cases of unsymmetrical alkenes a mixture of two possible isomers were obtained. It is worthwhile to note that under the identical conditions, alkenes bearing a carboxylic acid group underwent sulfonyllactonization to give sulfenylated lactones with good to excellent yields. Eight years later, Mellor's research team improved the efficiency of this process in the terms of regioselectivity and product yields by replacing of $\text{Pb}(\text{OAc})_4$ with $\text{Mn}(\text{OAc})_3$ and using allylic esters and amides as the substrates.¹³ Afterwards, El-Samii successfully applied this procedure in the hydroxysulfenylation of a small series of butadiene derivatives with diphenyldisulfane and di-*p*-tolylidysulfane.¹⁴ Of note, $\text{Pb}(\text{OAc})_4$ was also found to be effective oxidant for this transformation, however $\text{Cu}(\text{OAc})_2$ and $\text{Fe}(\text{OAc})_2$ proved to be completely ineffective.

In 2008, Movassagh and Navidi provided a further example of β -hydroxy sulfide derivatives **6** synthesis from the corresponding styrenes **4** and aromatic disulfides **5** assisted by Zn/AlCl_3 system in $\text{MeCN}/\text{H}_2\text{O}$ (4 : 1) under an oxygen atmosphere



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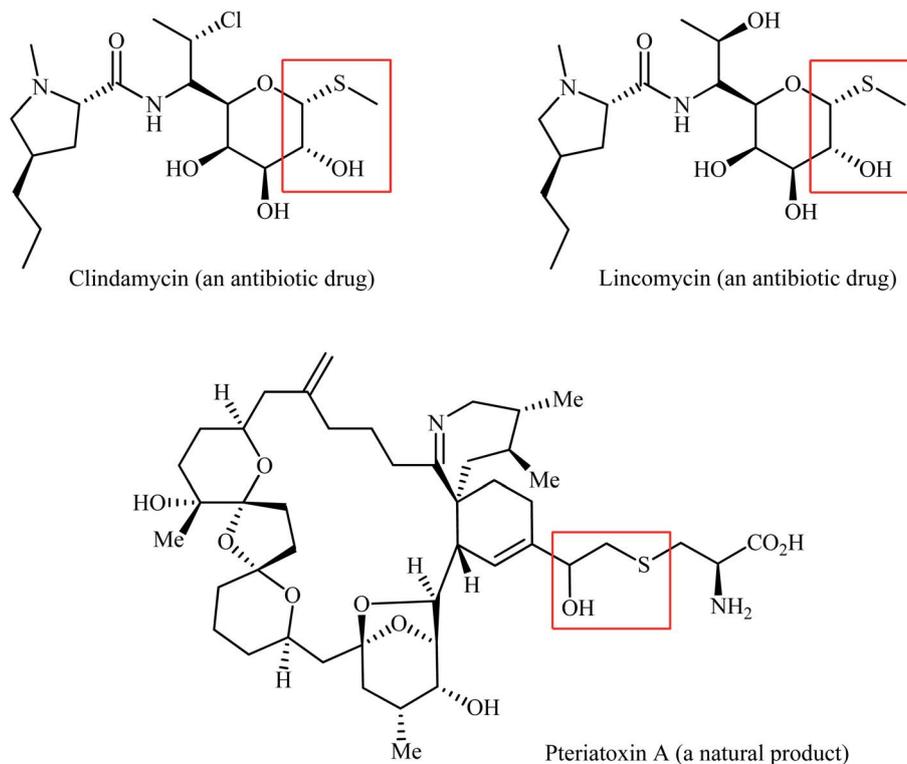
tional conferences. He has cooperation with many research project teams around world such as in China, Malaysia, and Thailand. His interests are Environmental Biotechnology, Biochemistry, and Gene pathways in phytoremediation processes.



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Scheme 1 Bioactive compounds containing a β -hydroxy sulfide unite.

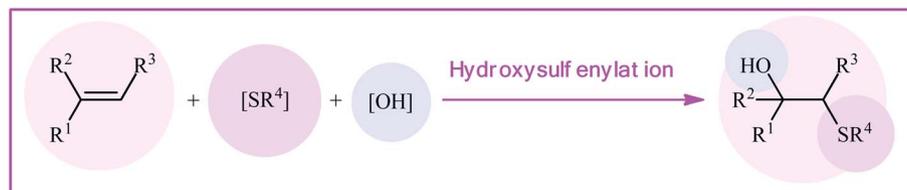


Fig. 1 Direct one-pot hydroxysulfenylation of alkenes.

(Scheme 3).¹⁵ Here, both electron-donating and electron-withdrawing substituents on the styrenes and the disulfides were compatible with the reaction condition and afforded the target products in good yields with high regioselectivities. However, the procedure was unsuccessful for aliphatic alkenes. It should be mentioned that the presence of oxygen is crucial for the success of this reaction since in its absence, the addition of hydrogen and thiolate anion across the double bond ensues, yielding the undesired sulfides.

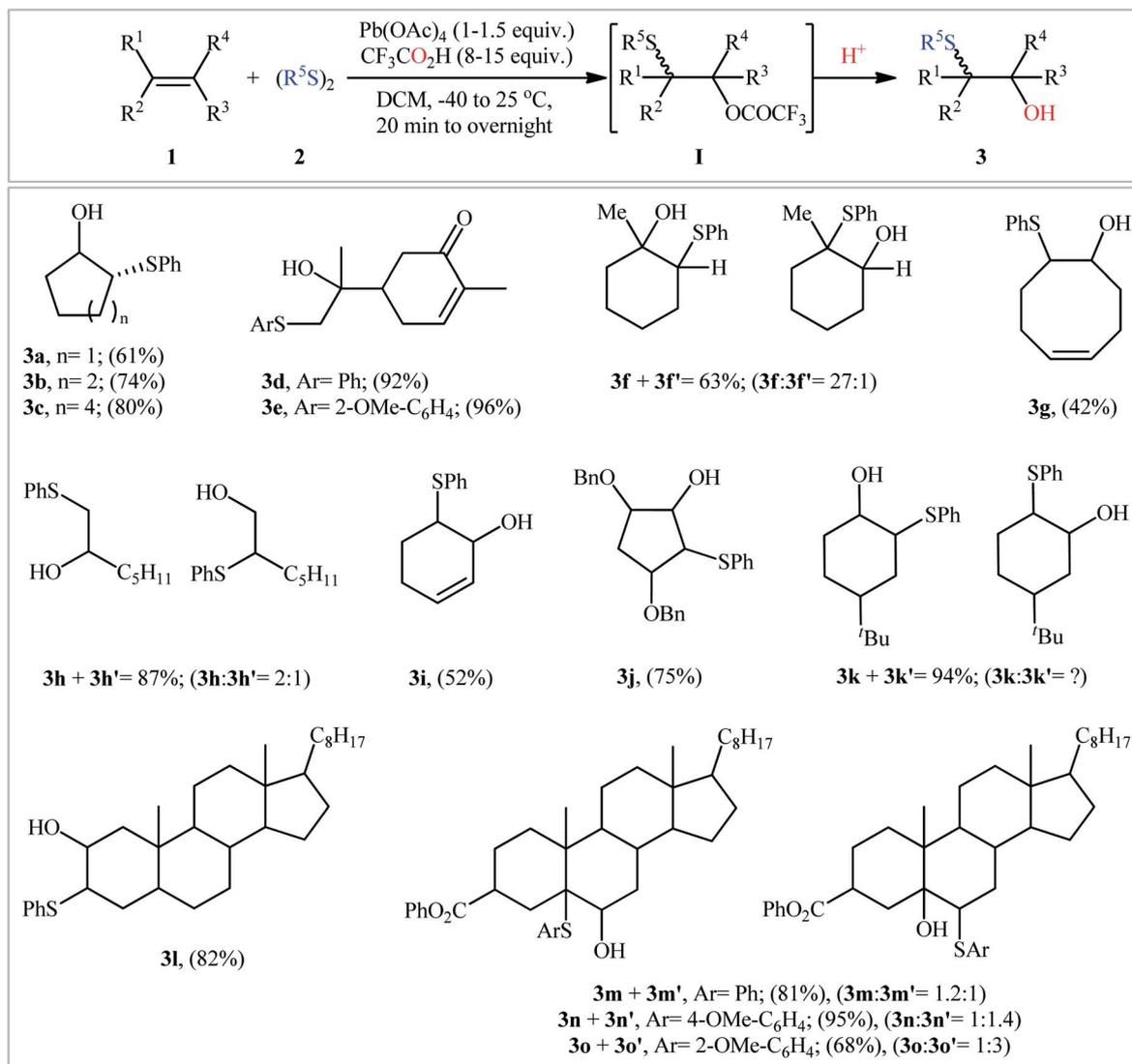
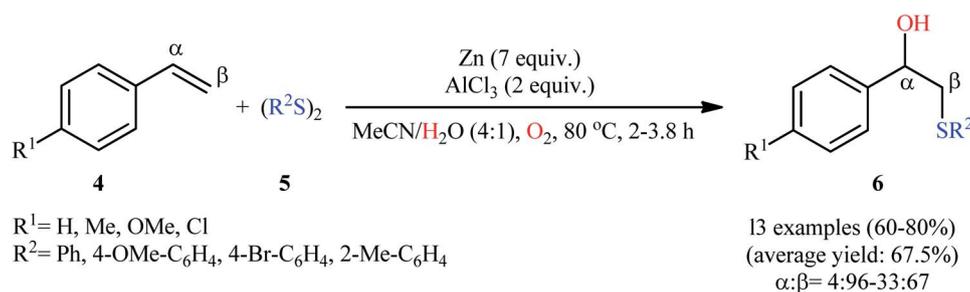
Following these works, Li's research group presented an efficient Cu(II)-catalyzed hydroxysulfenylation of a wide range of alkenes **7** and aromatic thiols **8** into the corresponding β -hydroxy sulfides **9**.¹⁶ The reaction proceeded under an oxygen atmosphere using 5 mol% of Cu(OAc)₂ as catalyst and 2 equiv. of PhCO₂H as additive in DCE at 50 °C, tolerated a variety of common functional groups (*e.g.*, fluoro, bromo, nitro, hydroxyl, methoxy, ketone, ester, and amine functionalities) and provided the expected products in moderate to excellent yields and outstanding regioselectivities, in which SR group predominantly added to the less hindered carbon atom of the C=C bond (Scheme 4). Noteworthy, this synthetic procedure was also

successfully applied in the high yielding preparation of biclutamide, an anticancer drug. It is worthwhile to note that other metal catalysts, such as FeCl₂, FeCl₃, Fe(OAc)₂, Mn(OAc)₂, CoCl₂, CuCl, CuCl₂, and CuBr were also found to promote this difunctionalization reaction; albeit at lower efficiencies. Based on a series of control experiments, the authors suggested that the reaction starts with the formation of thiyl radical **II** *via* the oxidation of thiol **8** with O₂, which after addition to alkene **7** leads to the formation of carbon-centered radical **III**. Next, this intermediate undergoes reaction with the *in situ* generated peroxy-copper species **IV** to afford the Cu^{III}OOR' species **V** that, after protodemetalation under acidic conditions forms hydroperoxide intermediate **VI**. Finally, reduction of **VI** by the copper catalyst leads to the final product **9** (Scheme 5).

3. Halogen-catalyzed/mediated reactions

In 2017, Peddinti's research team described an interesting metal-free molecular iodine-catalyzed direct

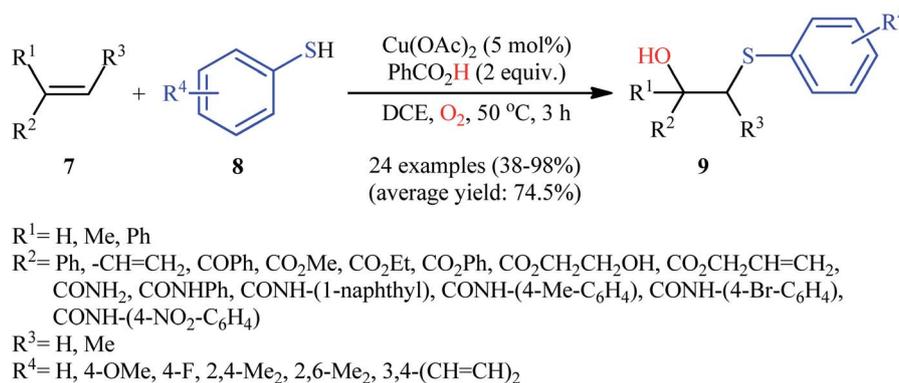


Scheme 2 Pb-catalyzed three component reaction between alkenes **1**, disulfides **2**, and CF₃CO₂H.Scheme 3 Movassagh's synthesis of β -hydroxy sulfides **6**.

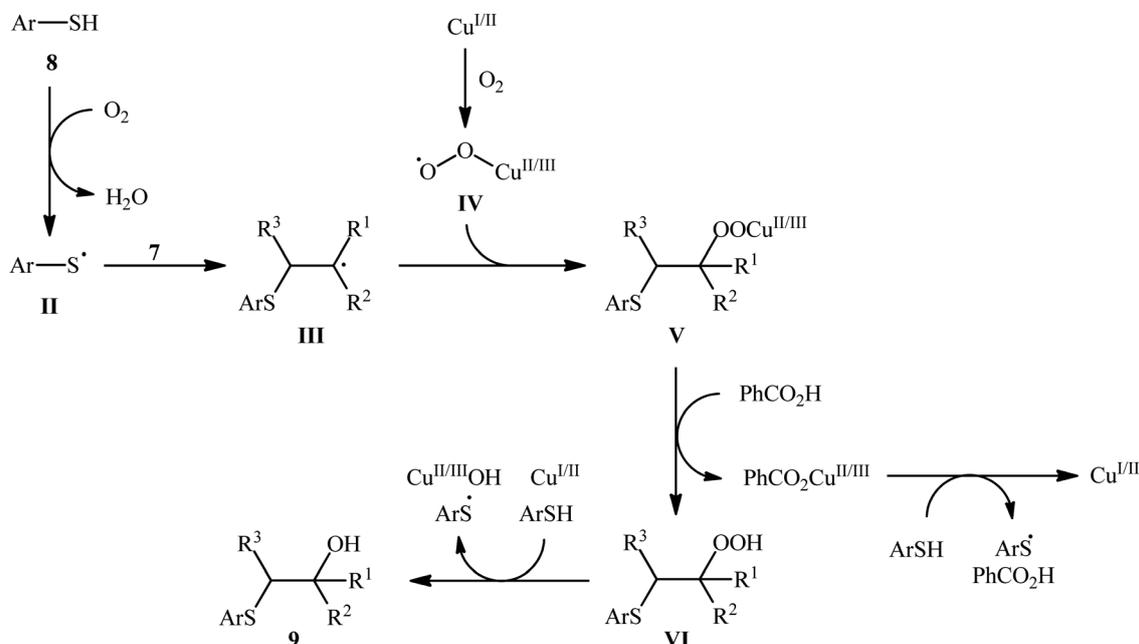
hydroxysulfenylation of styrene derivatives **10** with thiophenols **11** using DMSO as the oxygen source as well as the solvent.¹⁷ The reaction proceeded slowly at 60 °C under additive-free conditions, tolerated both electron-rich and electron-poor substrates, and provided the desired β -hydroxy sulfides **12** in moderate to excellent yields (Scheme 6a). Importantly, the reaction showed

outstanding regioselectivity, in which SAR group predominantly added to the terminal carbon atom of the double bond. The radical trapping experiments with 2,2,6,6-tetramethyl-1-piperidinyloxy (TEMPO) pointed toward the radical pathway of the reaction. The plausible mechanistic cycle proposed by the authors for this difunctionalization reaction is illustrated in





Scheme 4 Cu-catalyzed hydroxysulfenylation of alkenes **7** with aromatic thiols **8** in the presence of PhCO_2H .



Scheme 5 Plausible reaction mechanism for the formation of β -hydroxy sulfides **9**.

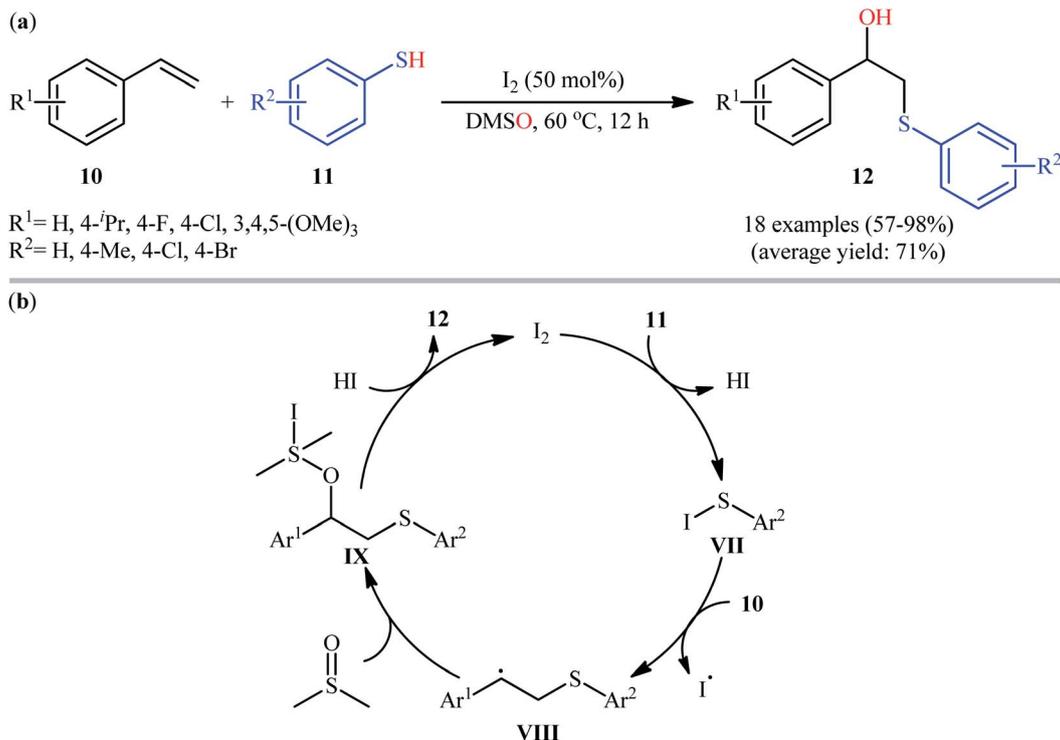
Scheme 6b. Initially, the nucleophilic thiophenol **11** attacks the electrophilic iodine centre leading to the generation of hypoiodothioite intermediate **VII**. Next, this intermediate liberates thiyl radical (ArS^\bullet); which reacts with electron-rich styrene **10** to deliver carbon-centered radical **VIII** and iodine free radical (I^\bullet). Subsequently, with the aim of iodine free radical, DMSO attacks at benzylic position of intermediate **VIII** to yield intermediate **IX**. Finally, the attack of *in situ* generated HI on **IX** affords the expected β -hydroxy sulfides **12** and regenerates molecular iodine.

In this context, Lin and Yan along with their co-workers developed an interesting HBr-catalyzed regioselective hydroxysulfenylation of styrenes **13** employing thiosulfates **14** as thiol sources.¹⁸ Using 50 mol% of HBr as catalyst and 2 equiv. of H_2O_2 as oxidant, a library of terminal styrenes **13** underwent regioselective hydroxysulfenylation with *S*-aryl thiosulfates **14** to afford the corresponding β -hydroxy sulfides **15** in good to high

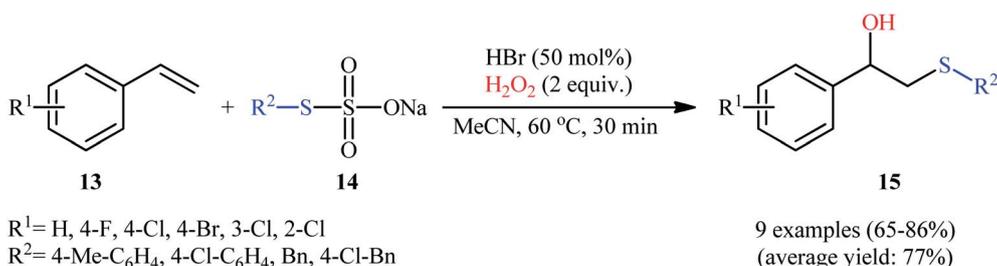
yields within 30 min (Scheme 7). However, when aliphatic alkenes and *S*-alkyl thiosulfates were employed under the identical conditions, only trace amounts of target products were obtained. Noteworthy, the optimized condition was also effective for the site-selective sulfenylation of 4-hydroxycoumarins at the C3-position.

Recently, Li and Chen described a unique NH_4I -promoted hydroxymethylthiolation of various terminal alkenes **16** using DMSO as an easy-handling methanesulfonyl source and water as an oxygen source.¹⁹ The reaction was carried out in a 1 : 1 mixture of DMSO and H_2O under additive-free conditions and afforded the desired hydroxysulfenylation products **17** in moderate to high yields and outstanding regioselectivities (Scheme 8a). Although the reaction tolerated various functional groups, the need for elevated temperature ($130\text{ }^\circ\text{C}$) may limit its range of application. It is worthwhile to note that when the reaction was carried out in dry DMSO, bis-methylsulfanes were





Scheme 6 (a) Three component reaction between styrenes **10**, thiophenols **11** and DMSO using I_2 as the catalyst; (b) mechanistic insights on I_2 -catalyzed hydroxysulfenylation of styrenes **10** with thiophenols **11** and DMSO.



Scheme 7 HBr-catalyzed hydroxysulfenylation of styrenes **13** employing thiosulfates **14** as thiol sources.

obtained as the sole products. The mechanism of this reaction probably involves the initial formation of the iodine radical (I^\cdot) and methanthiol **X** through a series of transformations with NH_4I and DMSO at high temperature, which after reaction with each other leads to the radical **XI** and HI. Subsequently, regioselective addition of this intermediate to the terminal C=C double bond of alkene **16** gives carbon-centered radical **XII** which undergoes further single-electron oxidation to form a β -MeS-substituted carbocation **XIII**. Next, intramolecular cyclization of carbocation intermediate **XIII** furnishes thiiranium ion **XVI**. Finally, the nucleophilic attack of H_2O on the thiiranium ion **XVI** produces the observed hydroxysulfenylation product **17** (Scheme 8b).

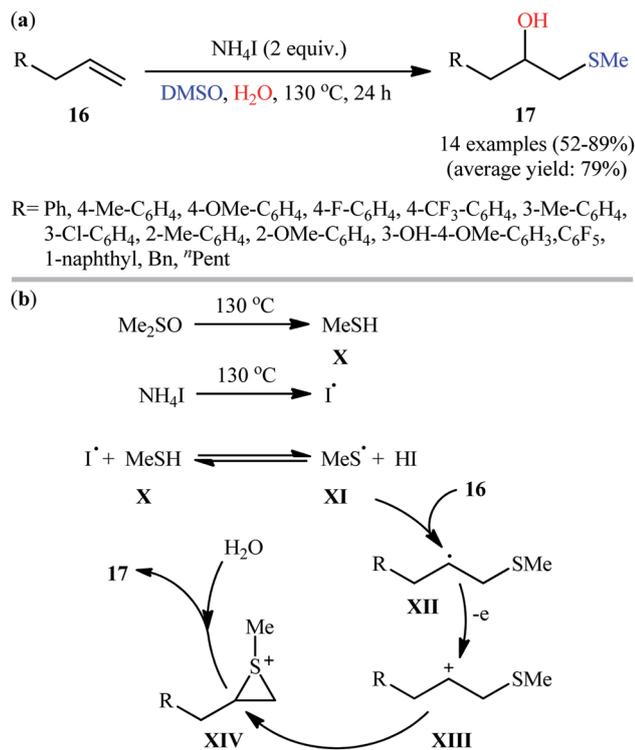
Very recently, Ni and Niu along with their colleagues extended the substrate scope of this chemistry to disulfides.²⁰ Thus, a large number of β -hydroxy sulfides **20** were selectively synthesized in moderate to high yields *via* hydroxysulfenylation of the corresponding alkenes **18** with symmetrical aromatic

disulfides **19** employing molecular iodine as the catalyst and air as the oxidant (Scheme 9). Although various terminal and internal aromatic and aliphatic alkenes bearing both electron-donating and electron-withdrawing groups were well tolerated under the reaction conditions, the scope of disulfides was limited to the electron-rich and slightly electron-poor aromatic disulfides.

4. Acid-mediated reactions

In 2008, Naito and colleagues demonstrated the usefulness of triethylborane (Et_3B) as a Lewis acid promoter for the direct hydroxysulfenylation of olefinic double bond.²¹ They showed that the three-component reaction between α,β -unsaturated imines **21**, thiols **22**, and oxygen in the presence of a catalytic amount of Et_3B in DCM at room temperature afforded the corresponding β -hydroxy sulfides **23** in relatively poor to good yields and excellent regioselectivities (Scheme 10). The relative



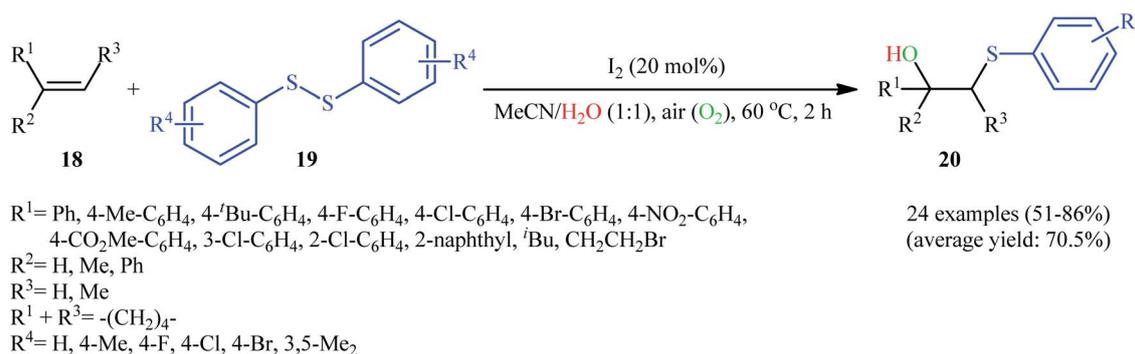


Scheme 8 (a) NH₄I-promoted hydroxymethylthiolation of terminal alkenes **16** with DMSO and water; (b) proposed mechanism for the formation of β -hydroxy sulfides **17**.

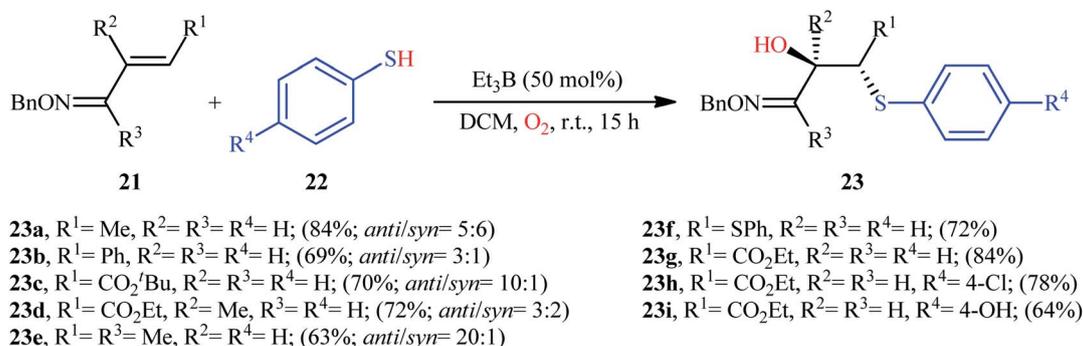
anti- and *syn*-geometric configurations of products were determined from NOESY experiments. The results indicated the preferential formation of the *anti*-isomers. The authors explained the observed *anti/syn* selectivity by invoking conformer **XX**, in which the C–S bond is eclipsed by the p orbital of the radical center, because electronic and steric effects (Scheme 11).

Eight years later, Shi's research team described an interesting Brønsted acid-promoted regioselective hydroxysulfenylation of alkenes with aromatic thiols employing air as oxygen source.²² Using a stoichiometric amount of racemic phosphoric acid **24** as the mediator, a number of styrene derivatives **25** underwent hydroxysulfenylation with *ortho*-mercaptobenzyl alcohols **26** to afford the corresponding β -hydroxy sulfides **27** in moderate to high yields within 12 h (Scheme 12). In addition, 1*H*-indene as a cyclic styrene analogue was also successfully employed under the optimized conditions in the reaction with 2-(2-mercaptophenyl)propan-2-ol, offering the desired β -hydroxy sulfide product in a considerable yield of 63%. Interestingly, when thiophenols were subjected to the reactions with styrene under the identical conditions, instead of the expected β -hydroxy sulfide products, the β -hydroxy sulfoxides were obtained as the sole products.

Very recently, Duan and Li along with their co-workers studied the metal-catalyzed hydroxysulfenylation of unactivated C=C double bonds with sulfonyl hydrazides as the sulfenyl sources and water as the oxygen source.²³ Thus, in the

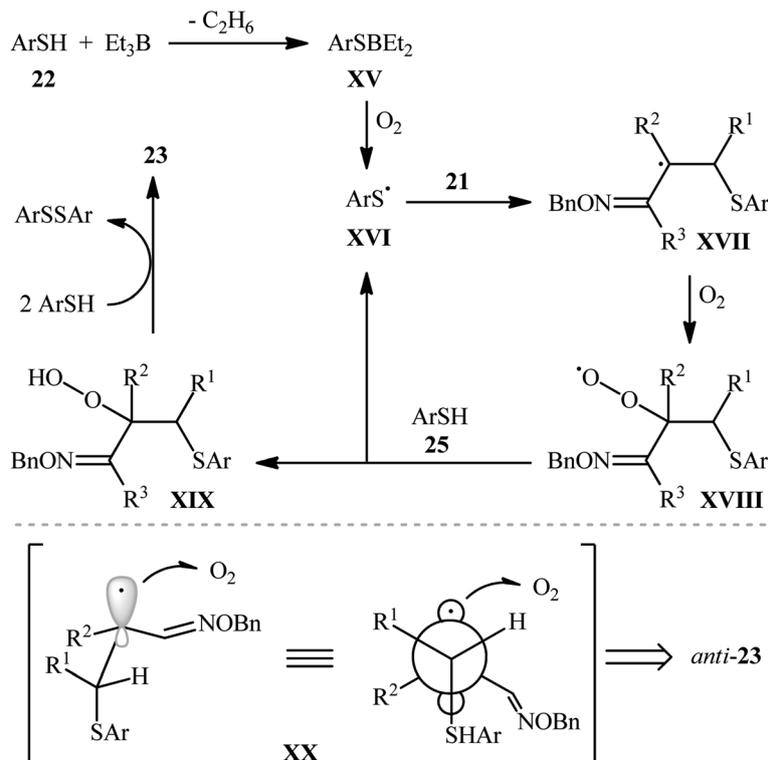


Scheme 9 I₂-catalyzed aerobic hydroxysulfenylation of alkenes **18** with disulfides **19** developed by Ni and Niu.

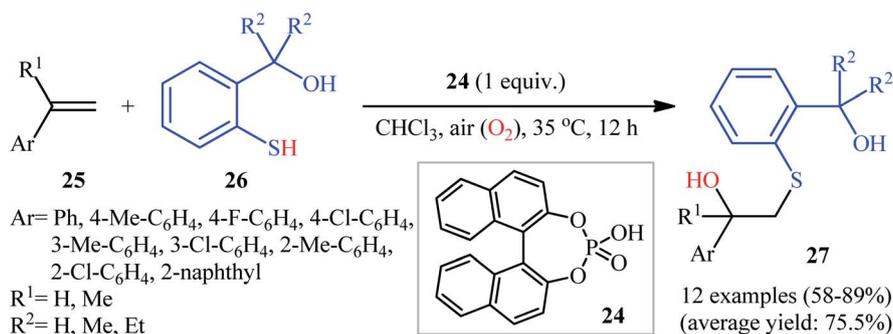


Scheme 10 Selected examples of Et₃B-catalyzed hydroxysulfenylation of α,β -unsaturated imines **21**, thiols **22**, and O₂.





Scheme 11 Proposed mechanism for the reaction in Scheme 10.



Scheme 12 Brønsted acid-catalyzed aerobic hydroxysulfenylation of styrenes 25 with mercaptobenzyl alcohols 26.

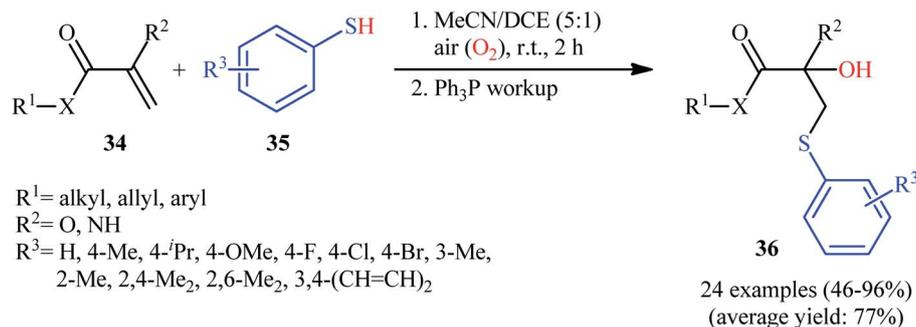
presence of FeBr₃/bpy/Na₂S₂O₈ combination as a catalytic system in 1,4-dioxane/H₂O (20 : 1) under air atmosphere, hydroxysulfenylation of *N*-allyl-*N*-sulfonamides 28 with various aromatic sulfonyl hydrazides 29 furnished the corresponding β-hydroxy sulfides 30 in modestly to high yields, ranging from 35% to 80% (Scheme 13a). In this investigation the authors found some limitations in their methodology when they attempted to use sterically hindered *N*-allyl-*N*-sulfonamides and secondary *N*-allyl-*N*-sulfonamides as the substrates. Unfortunately, in these cases, no desired products were observed. The results also revealed that the hydroxysulfenylation reaction could not proceed without sulfonyl groups in the substrates. Butane-1-sulfonylhydrazide did not work well in the reaction and therefore no other aliphatic sulfonyl hydrazides were examined in the protocol. The authors proposed mechanistic

pathway for this difunctionalization reaction is illustrated in Scheme 13b. It consists of the following key steps: (i) reaction of sulfonyl hydrazide 29 with bromine source in the presence of Na₂S₂O₈ to form the sulfenyl bromide intermediate XXI; (ii) electrophilic addition of sulfenyl bromide XXI to alkene 28 to produce thiiranium ion intermediate XXII; and (iii) ring-opening of thiiranium ion XXII with H₂O under acidic conditions to produce the desired β-hydroxy sulfides 30.

5. Catalyst-free reactions

After pioneering works by the groups of Kharasch²⁴ and Beckwith²⁵ on catalyst-free intermolecular hydroxysulfenylation of a small library of alkenes with thiols and dioxygen, the first general report of the direct synthesis of β-hydroxylated sulfides

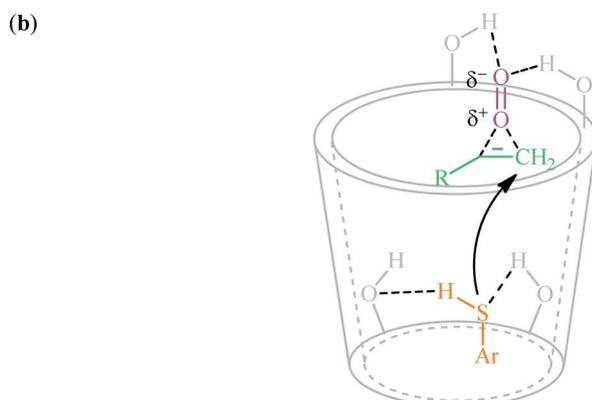
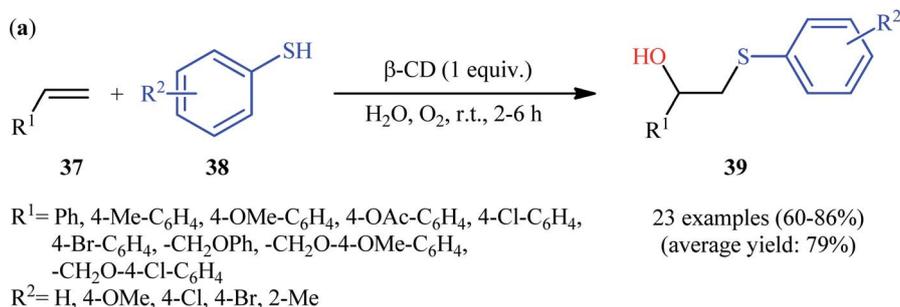


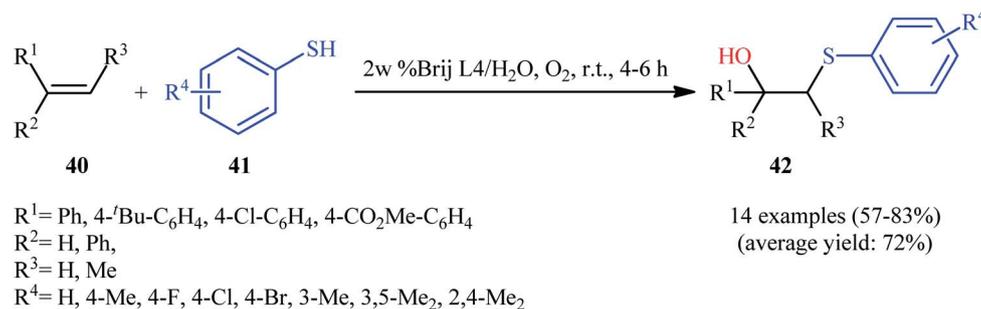
Scheme 15 Huo's synthesis of β -hydroxy sulfide derivatives **36**.

possible the further manipulation of the end products. This green protocol is also applicable for the hydroxysulfenylation of internal alkenes. It should be mentioned that β -cyclodextrin can be easily recovered from the reaction mixture and reused for the next reaction runs. As proposed by the authors, β -cyclodextrin catalyzes this hydroxysulfenylation reaction *via* reversible formation of host-guest complexes by non-covalent bonding as seen in Scheme 16b. Subsequently, Kamal and co-workers reported the synthesis of twenty β -hydroxy sulfides from various terminal and internal alkenes and aromatic thiols using a mixture of ionic liquid [bmim][BF₄] and water in the presence of aerial oxygen under neutral conditions.²⁹

Inspired by these results, in 2017, Feng and co-workers investigated the possibility of synthesizing β -hydroxy sulfide through the auto-oxidative radical hydroxysulfenylation of the

respective alkenes in surfactant/H₂O system.³⁰ To determine the optimum conditions, they carefully investigated the activities of different surfactant (*e.g.*, SDS, tergitol, brij L4, brij L23, brij O20, brij C20, TPGS-750-M) in the hydroxysulfenylation of 1,1-diphenylethylene with 4-chlorobenzenethiol under oxygen atmosphere, as a model reaction. The optimal system was recognized using 2 wt% brij L4 at room temperature. A variety of terminal styrenes, as well as α -substituted and β -substituted styrenes **40** were reacted well with functionalized aromatic thiols **41** under the standard reaction conditions to provide the expected hydroxysulfenylated products **42** in fair to high yields (Scheme 17). 1,2-Diphenylethylene was also tested and gave product but in rather poor yield. However, neither aliphatic alkenes nor aliphatic thiols were examined in this synthetic strategy. It is important to mention that the reaction could be

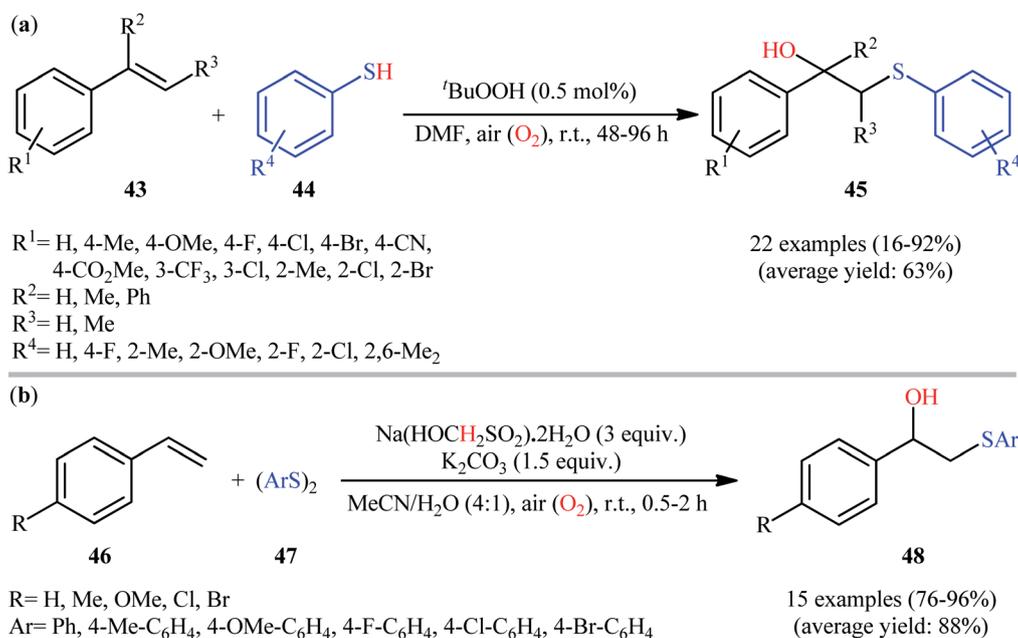
Scheme 16 (a) Synthesis of β -hydroxy sulfides **39** from terminal alkenes **37** and aromatic thiols **38** in the presence of β -CD in neat water; (b) schematic illustration of reaction mechanism for hydroxysulfenylation reactions using β -CD as a mediator.

Scheme 17 Feng's synthesis of β -hydroxy sulfides 42.

easily scaled up to the gram-scale as exemplified by the fubrication of 2-((3,5-dimethylphenyl)thio)-1,1-diphenylethanol on a 0.79 g scale (67%).

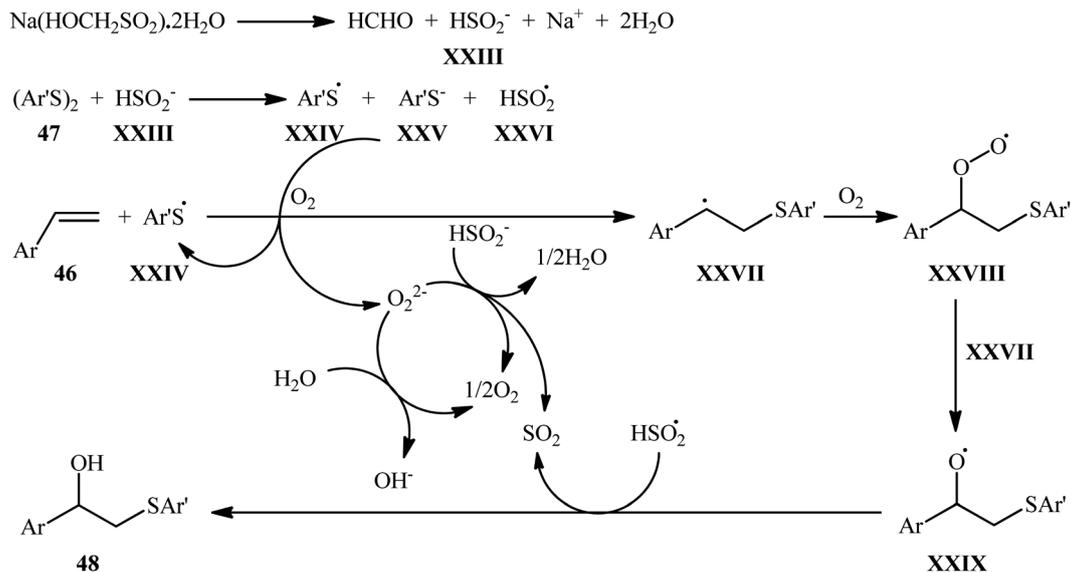
In 2015, Pan and Zou along with their colleagues developed a convenient and mild approach for regioselective synthesis of β -hydroxy sulfides 45 through the *tert*-butyl hydroperoxide (*t*-BuOOH)-initiated air oxidative radical hydroxysulfenylation of styrenes 43 with arylthiols 44.³¹ This radical difunctionalization reaction was run in DMF at room temperature under ambient air and provided substituted β -hydroxy sulfides 45 in poor to excellent yields (Scheme 18a); nonetheless the reaction failed completely for heteroaryl thiols, nitro and amino-substituted aryl arylthiols, aliphatic alkenes, and nitro-substituted styrenes. Interestingly, when styrenes bearing an easily removable leaving group at the α -position were used as substrates, β -oxosulfides were obtained as the sole products. In a similar manner, the reaction of styrenes bearing a leaving group at the β -position with arylthiols gave the corresponding β -oxythioacetals. At the same year, Yadav's research team presented the high yielding synthesis of a library of β -hydroxy sulfides 48

via rongalite-mediated hydroxysulfenylation of styrenes 46 with aromatic disulfides 47 employing 1.5 equiv. of K_2CO_3 as the base in a 4 : 1 mixture of MeCN and H_2O at room temperature (Scheme 18b).³² According to the authors proposed mechanism, this conversion is likely to follow the pathway illustrated in Scheme 19. Initially, thermal decomposition of rongalite generates formaldehyde and HSO_2^- anion **XXIII**, which reacts with the disulfide 47 to give radicals **XXIV** and **XXVI** along with the thiolate anion **XXV**. Subsequently, the thiolate anion **XXV** reacts with oxygen to form a thiyl radical **XXIV** and peroxide anion. Next, this radical regioselectively adds to the terminal C-C double bond of styrene 46 to produce benzylic radical **XXVII** that after reaction with O_2 provides a peroxy radical **XXVIII**. The newly formed reacts with **XXVII** to form radical **XXIX**. Finally, the alkoxy radical **XXIX** abstracts a hydrogen atom from **XXVI** to afford hydroxysulfenylated product 48. It should be mentioned that aliphatic alkenes and dialkyl disulfides were completely inert in the present protocol. The authors speculated that the unreactivity of aliphatic alkenes in this strategy is related to the far less stability of the alkyl free radicals formed



Scheme 18 (a) TBHP-initiated air oxidative radical hydroxysulfenylation of styrenes 43 with arylthiols 44; (b) rongalite-mediated hydroxysulfenylation of styrenes 46 with aromatic disulfides 47 under ambient air.



Scheme 19 Proposed mechanism for the generation of β -hydroxy sulfides 48.

after the attack of the thiyl radical on the aliphatic alkenes. They also explained the reason of unreactivity of dialkyl disulfides to their strong S–S bond.

Finally, it should be noted that transition metal complexes play important role in synthesis of β -hydroxy sulfides and other compounds.^{33–54}

7. Conclusion

1,2-Difunctionalization of alkenes has received continuous interests as it provides a valuable and versatile synthetic route to introduce two functional groups at the C=C bonds in a single click. Among them, intermolecular hydroxysulfenylation of alkenes is one of the most efficient and convenient pathways to the construction of biologically and synthetically important β -hydroxysulfides through the simultaneous C–O and C–S bond formation. As illustrated, various thiols, disulfides, sulfonyl hydrazides, and thiosulfates were successfully employed as the sulfenyl sources in this chemistry. Moreover, most of the developed systems were compatible with both terminal and internal alkenes. Despite the remarkable accomplishments during the past few years on this interesting page of β -hydroxysulfides synthesis, the construction of chiral β -hydroxysulfides through this chemistry, has, to date, been rarely described. Therefore, the development of chiral catalysts and ligands that allow asymmetric synthesis of titled compounds would be highly desirable. It is our hope that this Mini-Review will stimulate continued interest in the fabrication of β -hydroxysulfide derivatives directly from alkenes and make it a prolonged and prominent research arena for developing straightforward and extremely effective methodology used for pharmaceutical products syntheses.

Conflicts of interest

There are no conflicts to declare.

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