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Influences of a standardized food matrix and gastrointestinal fluids on the physicochemical properties of titanium dioxide nanoparticles

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The fast-growing applications of engineered titanium dioxide nanoparticles (e-TiO₂-NPs) in the food and pharmaceutical industry in production, packaging, sensors, nutrient delivery systems, and food additives enhance the possibility of oral exposure. Physicochemical transformations may occur when e-TiO₂-NPs are incorporated into a food matrix and pass through the human gastrointestinal tract (GIT), which may redefine the toxic effects of the e-TiO₂-NPs. In this study, a standardized food model (SFM) and simulated gastrointestinal fluids have been used to study the fate of e-TiO₂-NPs following a three-step digestion model *in vitro*, and a case study was carried out to assess the toxicity of the digested e-TiO₂-NPs using an *in vitro* cellular model. In the absence and presence of the SFM, the transformations of the tristimulus color coordinates, size, agglomeration state, surface charge and solubility of the e-TiO₂-NPs in the salivary, gastric and intestinal digestion fluids were compared with those before digestion. The results demonstrate that the presence of the SFM impacted the physicochemical properties of the e-TiO₂-NPs significantly. The SFM stabilized the e-TiO₂-NP suspensions and acted as a dispersant during each digestive phase. The e-TiO₂-NPs showed differentiated transformations of their physicochemical properties after each step of the digestive process. The pH shifts and variable concentrations of enzymes and salts in gastrointestinal fluids induced the transformations of the physicochemical properties of the e-TiO₂-NPs. The transformed e-TiO₂-NPs could release titanium ion in the gastrointestinal tract. Also, the cell viability induced by e-TiO₂-NPs was found to be strongly affected by the presence of the SFM and simulated human GIT fluids. It can be concluded that the physicochemical transformations of the e-TiO₂-NPs that were found when they were incorporated into an SFM and passed through the GIT consequently strongly affected the biological effects of the e-TiO₂-NPs, which highlights that the toxicity assessment of ingested NPs should use appropriate standardized food models and take realistic physiological conditions into account.

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Introduction

Presently, nanoscale structure materials exhibiting unique physicochemical properties are increasingly being incorporated into commercial food and pharmaceutical products for the purposes of improving the properties of the products, including visual presentation, product quality, food shelf life, safety, and nutritional absorption.^{1–6} Addition of nanoparticles (NPs) to food and pharmaceutical products enhances the possibility of oral exposure. After being ingested, NPs will pass through the human gastrointestinal tract (GIT) before being absorbed and entering the circulatory system (*e.g.*, mouth, esophagus, stomach, small intestine, and large intestine).⁷ The properties of the NPs can be influenced after passing through the GIT, which may redefine the toxic effects of the NPs.⁸ Therefore, concerns over the potential

human health effects of oral exposure to NPs should be carefully assessed further⁹ to examine not only the characteristics of the material itself, but also the interaction between the material and the real environment medium.

Extensive studies have shown that the unique physicochemical properties of NPs may cause adverse health effects both *in vitro* and *in vivo*. For instance,¹⁰ ZnO NPs could cause slight toxicity to the liver and kidneys in IBD mice. Ag NPs were found to be toxic to different cells depending on their size, morphology, surface chemical, exposure dose and time.¹¹ A 14 day study was carried out by oral gavage administration of Al₂O₃ and CuO NPs to mature female Wistar albino rats with a dose series of 0, 0.5, 5, and 50 mg per kg b.w. per day; the results showed that Al₂O₃ and CuO NPs could pass through the intestinal gut wall, enter the bloodstream, accumulate in the liver, and cause some significant alterations in the activities of antioxidant enzymes, which suggests that the antioxidant system of rats suffered after NP administration. In recent studies, transformations of the properties and biological effects of NPs

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resulting from their interactions with the food matrix and simulated human gastrointestinal fluids have received focus, including NPs with specific food components (protein, sugar, lipids, phytochemicals, mineral), NPs with whole foods (skim milk, coffee mix, sports drink, bread, meat), and NPs with complex food matrices. For instance,¹² albumin reduced the hydrodynamic radii and PDI values of titanium dioxide nanoparticles (TiO₂ NPs) and increased the colloidal stability of TiO₂ NPs as a consequence, while glucose and arsenic(III) resulted in agglomerates.^{12,13} The UV-Vis absorbance of zinc oxide nanoparticles (ZnO NPs) was increased by the presence of palmitate.¹⁴ The stronger dispersion stability of ZnO NPs dispersed in skim milk was compared to those in individual components of skim milk (*i.e.*, casein, lactose, and calcium).¹⁵ The hydrodynamic radii and zeta potential values of ZnO NPs decreased in the presence of sugar mixture solutions.¹⁶ However, the dissolution of ZnO NPs was not influenced by albumin, glucose honey, or sugar mixture solutions.^{12,16} In an *in vivo* study, the oral absorption of silica nanoparticles (SiO₂ NPs) by rats was enhanced by the presence of albumin and glucose, probably due to the interactions between the food components and the NPs.¹⁷ The solubility of SiO₂ NPs dispersed in *in vitro* simulated gastric fluid and orally administered to rats was not influenced by the presence of albumin or glucose. Gold nanoparticles (Au NPs) aggregated and presented negligible changes in fasted or fed simulated intestinal fluid by DLS and UV-Vis spectrometry analysis.¹⁸ Different levels of aggregation and dissolution of silver nanoparticles (Ag NPs) digested in fed and fasted fluids and three GIT fluids were measured by UV-Vis spectroscopy, DLS, TEM, and inductively coupled plasma mass spectrometry (ICP-MS).¹⁹ However, high stability of the engineered silver sulfide nanoparticles (e-Ag₂S-NPs) was observed after each phase of *in vitro* digestion.²⁰ As such, knowledge of the processes that influence the fate of NPs with food components/matrices or the GIT digesta is important to better assess the toxicity of NPs following oral exposure.

Engineered titanium dioxide nanoparticles (e-TiO₂-NPs) with photocatalytic activity and ultraviolet absorption activity at specific wavelengths are being used increasingly in the food and pharmaceutical industry (*i.e.*, production, packaging, sensors, nutrient delivery systems and food additives).²¹ As food additives, e-TiO₂-NPs are widely present in various food products, including gum, candies, puddings, cheeses, sauces, skimmed milk, ice cream, pastries, dressings and certain powdered foods, with the goal of improving the white color, brightness, and flavor of food products.^{22,23} This may increase the possibility of exposure to e-TiO₂-NPs *via* human oral routes. Inevitably, humans may be co-exposed to e-TiO₂-NPs and food ingredients or matrices. In this study, a standardized food model (SFM) as a food matrix and simulated gastrointestinal fluids have been used to study the fate of e-TiO₂-NPs following a three-step digestion model *in vitro* with the aim to study the physicochemical transformations and toxicity of e-TiO₂-NPs when incorporated into a food matrix and passed through the GIT.²⁴ The SFM was established to conform to the dietary habits of Chinese residents based on the compositions and proportions of Chinese dietary reference intakes and Chinese dietary guidelines from the averages for men and women aged 18 years or older to better imitate the intake of e-TiO₂-NPs in real life.²⁵ The compositions of the SFM included daily recommended intakes of protein, fat, carbohydrates, dietary fiber, and minerals. To indicate the impact of the food matrix and gastrointestinal effects on the physicochemical properties of the e-TiO₂-NPs, the transformations of the tristimulus color coordinates, hydrodynamic size, agglomeration state, UV-Vis spectra, zeta potential, and solubility of the e-TiO₂-NPs present in each step suspension were measured using colorimetry, UV-Vis spectroscopy, DLS, TEM-EDX and ICP-AES. Then, a nanotoxicology case study was carried out to assess the toxicity of the digested e-TiO₂-NPs using an *in vitro* cellular model of human colon epithelial Caco-2 cells (ATCC; SCSP-5027). A detailed schematic of the methodology of the study is summarized in Fig. 1.

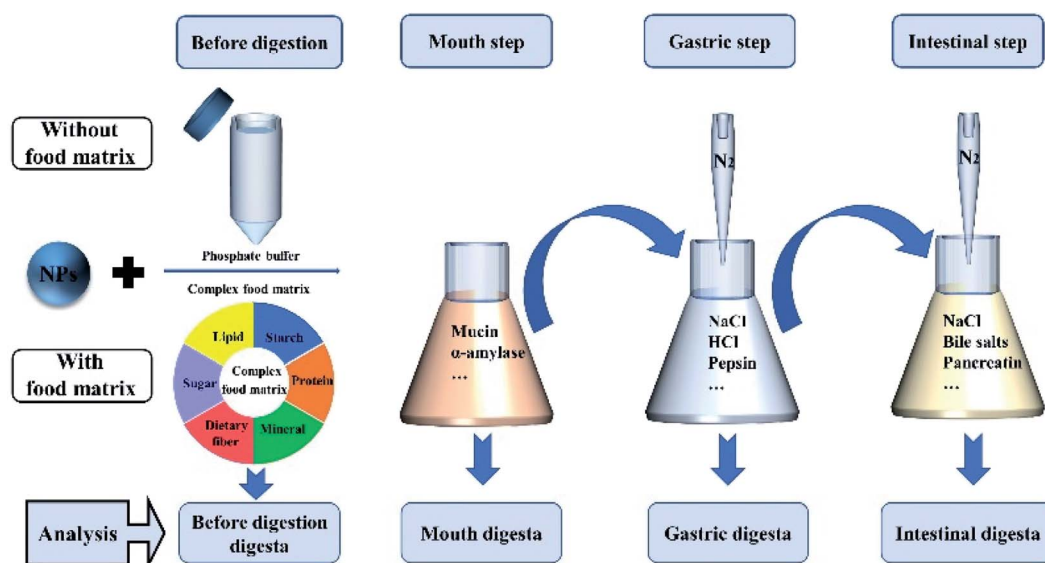


Fig. 1 Schematic overview of the *in vitro* studies of the NPs.

Materials and methods

Materials

Sodium caseinate, corn oil, pectin, sucrose, sodium chloride, and e-TiO₂-NPs (25 nm) were purchased from Macklin Biochemical Co. (Shanghai, China). Na₂HPO₄ and NaH₂PO₄ were purchased from Sigma-Aldrich (Shanghai, China). Modified starch was obtained from a local food supplier.

Preparation of the complex food matrix (SFM)

The nutrient formula of the SFM was established in this study according to the method described by Yan Li *et al.*,²⁴ and the preparation of the complex food matrix was carried out using procedures described previously.^{26–29} Sodium caseinate (1% w/w) was dissolved in phosphate buffer solution (10 mM, pH 7.0) with constant stirring until complete dissolution at room temperature. Insoluble powder particles in sodium caseinate solution were removed by sterile filtering before further use (the protein concentration was not considerably changed). Corn oil (3.21% w/w) was then dispersed in sodium caseinate solution with continuous shaking for 10 min. The mixture was homogenized using a high-speed mixer at 16 000 rpm for 2 min (IKA Instruments, Germany) to prepare a coarse emulsion. A fine emulsion was then prepared by homogenizing the coarse emulsion using a high-pressure homogenizer (APV Instruments, Germany) at an operational pressure of 12 000 psi for 3 passes. The SFM was prepared by sequentially and slowly dispersing additional sodium caseinate and other dietary components into the fine emulsion with continual stirring to ensure complete dissolution (fat = 3.21% w/w, protein = 2.89% w/w, pectin = 1.32% w/w, starch = 13.8% w/w, sucrose = 1.44% w/w, sodium chloride = 0.37% w/w).

Suspension of e-TiO₂-NPs

Engineered TiO₂-NPs were suspended in 10 mM phosphate buffer (without food matrix) and the SFM (with food matrix) at 0.5% w/w, 1% w/w and 1.5% w/w, respectively.³⁰ The e-TiO₂-NPs were vortexed (30 s) and then sonicated for 10 s before mixing according to the method described previously.

In vitro digestion model

Engineered TiO₂-NPs suspensions without or with the presence of the food matrix were exposed to a simulated three-step GIT model, including the mouth, gastric and intestine phase, which was adopted from some previous studies with slight modifications as described below.^{26,31–34}

Before the digestion step. e-TiO₂-NPs suspensions without or with the presence of food matrix (20.0 g) were preheated in a temperature-controlled incubator shaker with continual stirring at a rate of 100 rpm at 37 °C for 15 min (SHA-CA, QIqian, Shanghai, China).

Mouth step. Simulated saliva fluid (20.0 g) containing 0.03 g mL^{−1} mucin and 150 units per mL amylase was adjusted to pH 6.8 using 1.0 M NaOH before being preheated to 37 °C and then mixed with both e-TiO₂-NPs suspensions without or with the

presence of food matrix (20.0 g) with continuous shaking. The mixture was re-adjusted to pH 6.8 and then shaken at 100 rpm at 37 °C for 10 min in the shaker to mimic mouth conditions.

Gastric step. The obtained digesta (20.0 g) resulting from the salivary step was added to simulated gastric fluid (pH 2.5, 37 °C) containing 0.002 g mL^{−1} NaCl, 0.7% (v/v) HCl, 0.0032 g mL^{−1} pepsin. The mixture was adjusted to pH 2.5 using 1.0 M NaOH or HCl and then flushed by N₂ for 30 s with continuous shaking at 37 °C for 2 h in the shaker to mimic stomach conditions.

Intestinal step. The obtained digesta (30.0 g) resulting from the gastric step was adjusted to pH 7.00 using 1.0 M NaOH and then pre-heated to 37 °C. After that, the digesta was mixed with 1.50 mL of simulated intestinal fluid containing 0.25 M CaCl₂·2H₂O, 3.75 M NaCl and 3.50 mL of bile salt solution (0.05 g mL^{−1}) with continual stirring in the shaker, and the pH was adjusted to 7.00. After that, 2.50 mL of pancreatin solution (0.036 g mL^{−1}) was added to the digestion solution and then flushed with N₂ for 30 s. A pH-stat (S220, METTLER TOLEDO, Shanghai, China) was used to maintain the pH of the resulting mixture at 7.0 by dropping 1 M NaOH at 37 °C for 2 h with continuous shaking in the shaker to mimic small intestine conditions. The samples after each digestion step were collected and used to characterize the e-TiO₂-NPs according to the described methods in previous studies with slight modifications.

Characterization of the engineered TiO₂-NPs suspensions

Colorimetry of e-TiO₂-NPs in different suspensions. The tristimulus color coordinates of the e-TiO₂-NPs suspensions were measured using a ColorFlex EZ colorimeter (HunterLab Instruments, US) to research the optical properties as described previously.²⁶ *L** (lightness), *a** (red to green), and *b** (yellow to blue) are the tristimulus color coordinates.

Hydrodynamic sizes and PDIs of e-TiO₂-NPs in different suspensions. The hydrodynamic sizes and PDIs of the e-TiO₂-NPs suspensions were determined by Dynamic Light Scattering (DLS) using a NanoBrook 173Plus (Brookhaven Instruments, USA). The refractive indices were set at 2.55. Samples were properly diluted with phosphate buffer solution prior to measurement to avoid multiple scattering effects. All measurements were performed at 25 °C.

Transmission electron microscopy of e-TiO₂-NPs in different suspensions. Engineered TiO₂-NPs before digestion and after each step of the digestive process were visualized using a JEM-2100 transmission electron microscope (TEM, JEOL, Tokyo, Japan) and were collected using a method described in previous work with minor modifications. Energy-dispersive X-ray spectroscopy (EDX, AMETEK, USA), a technique of elemental analysis, was used to detect the elemental compositions of the samples. In the absence of the food matrix, briefly, the e-TiO₂-NPs suspension before digestion and the resultant mouth, stomach, and small intestine digesta fluids were centrifuged at 10 000g for 40 min. Then, the resultant mouth, stomach, and small intestine digesta fluids were first centrifuged at 1500g for 5 min to remove the insoluble particles and then centrifuged at 10 000g for 40 min to collect the e-TiO₂-NPs pellets, followed by



TEM imaging. In the presence of the food matrix, briefly, the e-TiO₂-NPs suspension before digestion and the digesta fluids from each phase of the simulated human GIT were first centrifuged three times at 2000g for 1 min to remove the insoluble particles. Then, the e-TiO₂-NPs pellets collected from centrifugation at 10 000g for 1 h were visualized by TEM imaging.

Zeta potentials of e-TiO₂-NPs in different suspensions. The zeta potentials of the e-TiO₂-NPs suspensions were measured by laser Doppler velocimetry using a Zetasizer Nano S90 (Malvern Instruments, UK). Samples were properly diluted with phosphate buffer solution prior to measurements to avoid multiple scattering effects.

UV-Vis spectra of e-TiO₂-NPs in different suspensions. The UV-Vis absorbance of the e-TiO₂-NPs suspensions were recorded by a UV-Vis spectrophotometer (TU-1901, PERSEE, Beijing, China), and water was used as a blank.

Solubility of e-TiO₂-NPs in different suspensions. The solubility of the e-TiO₂-NPs suspensions was measured using an ICP-AES730 (Agilent Technologies, Santa Clara, CA, USA); the supernatants were collected by ultracentrifugation (10 000 × g) for 1 h as described for the ICP-AES analysis.

Cell culture

Human colon epithelial Caco-2 cells (ATCC; SCSP-5027) were grown in Dulbecco's modified Eagle's medium/high glucose medium (Hyclone, Shanghai, China) supplemented with 10% heat-inactivated fetal bovine serum (Gibco, Shanghai, China), 1% penicillin-streptomycin solution (Hyclone, Shanghai, China) and non-essential amino acids (1/100 dilution of 100× solution, Gibco, Shanghai, China). The seeding density was 1.5×10^4 per well on 96-well plates, and the cells were grown for 24 h before exposure.

Cell viability analysis

The resulting digesta from the intestinal phase were mixed with serum-free DMEM media in a ratio of 1 : 9, added to the Caco-2 cells, and then incubated for 24 h. A Pierce MTS cytotoxicity

assay kit (BestBio, Shanghai, China) was used to measure the cell viability at the wavelength of 490 nm, and then the cytotoxicity percentage was calculated.

Data analysis

All results are presented as the mean ± standard deviation (SD) of the triplicates for each prepared sample, and they were analyzed using SPSS software (IBM SPSS Statistics 25, Armonk, NY, USA). One-way analysis of variance (ANOVA) followed by Dunnett's multiple comparison tests were used for the statistical analysis of results. Different letters in the table also indicated significant difference ($P < 0.05$).

Results

Changes of optical properties

The optical properties of the e-TiO₂-NPs suspensions (1% w/w) before digestion and after each step (mouth, stomach and small intestine) of the digestive process were determined using an instrumental colorimeter, as shown in Fig. 2. The optical properties of e-TiO₂-NPs without the presence of the food matrix were measured as $L^*, a^*, b^* = 22.88 \pm 0.07, -0.52 \pm 0.04, -0.08 \pm 0.01$, which were obviously changed by the presence of the food matrix to $L^*, a^*, b^* = 23.65 \pm 0.02, -0.62 \pm 0.04, 3.13 \pm 0.03$, respectively ($P < 0.01$).

Compared to the results obtained before digestion, there was an apparent increase of lightness, green, and yellow of the e-TiO₂-NPs suspensions without ($L^*, a^*, b^* = 26.77 \pm 0.29, -1.30 \pm 0.04, 3.40 \pm 0.02$) and with ($L^*, a^*, b^* = 26.71 \pm 1.44, -0.87 \pm 0.09, 4.45 \pm 0.35$) the presence of the food matrix. Additionally, the lightness and green of the e-TiO₂-NPs suspensions were lower when the food matrix was present during this phase. In contrast, the yellow of the e-TiO₂-NPs suspensions was significantly increased in the presence of the food matrix ($P < 0.05$).

Without ($L^*, a^*, b^* = 23.73 \pm 0.31, -1.11 \pm 0.02, 1.77 \pm 0.02$) or with ($L^*, a^*, b^* = 26.13 \pm 0.13, -0.86 \pm 0.04, 3.13 \pm 0.01$) the presence of the food matrix, the color, lightness, green and yellow of the e-TiO₂-NPs suspension were higher in the gastric phase than before digestion. Moreover, the colors of the e-TiO₂-

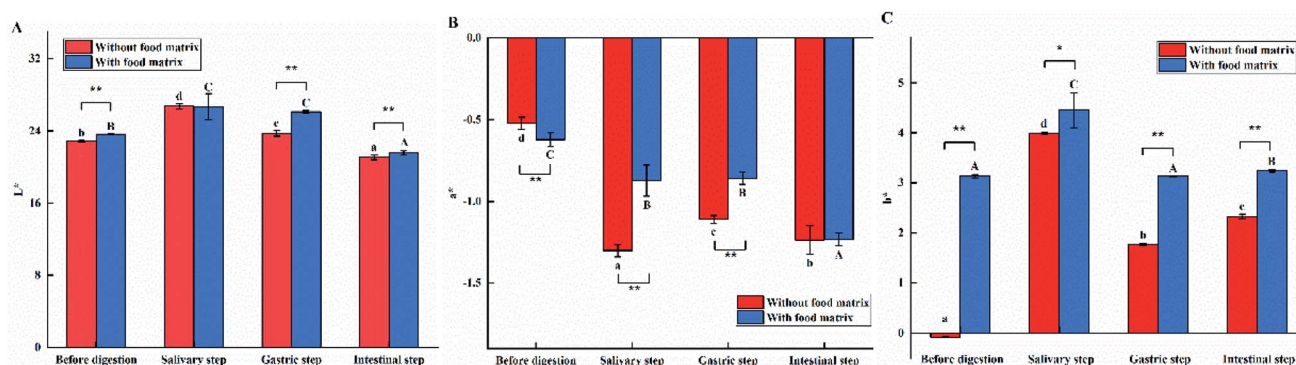


Fig. 2 The tristimulus color coordinates (L^*, a^*, b^*) of e-TiO₂-NPs suspensions (1% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion. The lower-case (a–d) and capital (A–C) letters indicate significant differences between the digestive step in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of the food matrix for each phase of digestion (* $P < 0.05$ and ** $P < 0.01$).



Table 1 The tristimulus color coordinates (L^* , a^* , b^*) of e-TiO₂-NPs suspensions (0.5%, 1%, and 1.5% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion^a

Food matrix	Digestion step	L^*			a^*			b^*		
		0.5%	1%	1.5%	0.5%	1%	1.5%	0.5%	1%	1.5%
Without	Before digestion	22.14 ± 0.04 ^c	22.88 ± 0.07 ^b	23.65 ± 0.00 ^a	-0.52 ± 0.04 ^d	-0.52 ± 0.04 ^d	-0.47 ± 0.04 ^c	0.06 ± 0.02 ^a	-0.08 ± 0.01 ^a	0.45 ± 0.02 ^a
	Salivary step	21.97 ± 0.36 ^c	26.77 ± 0.29 ^d	30.48 ± 0.33 ^b	-0.99 ± 0.06 ^b	-1.30 ± 0.04 ^a	-1.29 ± 0.03 ^a	4.06 ± 0.06 ^d	3.99 ± 0.02 ^d	4.34 ± 0.13 ^d
	Gastric step	21.43 ± 0.20 ^b	23.73 ± 0.31 ^c	36.21 ± 0.31 ^d	-0.88 ± 0.05 ^c	-1.11 ± 0.02 ^c	-1.00 ± 0.04 ^b	2.35 ± 0.03 ^b	1.77 ± 0.02 ^b	1.85 ± 0.02 ^b
	Intestinal step	20.30 ± 0.02 ^a	21.08 ± 0.26 ^a	30.96 ± 0.14 ^c	-1.06 ± 0.07 ^a	-1.24 ± 0.09 ^b	-1.29 ± 0.03 ^a	2.99 ± 0.08 ^c	2.33 ± 0.05 ^c	2.52 ± 0.01 ^c
With	Before digestion	22.96 ± 0.70 ^{*C}	23.65 ± 0.02 ^{*B}	27.81 ± 0.14 ^{**C}	-0.65 ± 0.08 ^{**C}	-0.62 ± 0.04 ^{**C}	-0.63 ± 0.02 ^{**D}	2.81 ± 0.05 ^{**A}	3.13 ± 0.03 ^{**A}	2.24 ± 0.01 ^{**A}
	Salivary step	23.50 ± 0.04 ^{**C}	26.71 ± 1.44 ^C	29.33 ± 0.05 ^{**D}	-0.85 ± 0.05 ^{**B}	-0.87 ± 0.09 ^{**B}	-0.89 ± 0.04 ^{**B}	4.46 ± 0.05 ^{**D}	4.45 ± 0.35 ^{*C}	4.53 ± 0.02 ^{**D}
	Gastric step	21.75 ± 0.65 ^B	26.13 ± 0.13 ^{**C}	24.60 ± 0.07 ^{**A}	-0.70 ± 0.02 ^{**C}	-0.86 ± 0.04 ^{**B}	-0.76 ± 0.02 ^{**C}	3.13 ± 0.06 ^{**B}	3.13 ± 0.01 ^{**A}	2.87 ± 0.03 ^{**B}
	Intestinal step	20.52 ± 0.13 ^{**A}	21.59 ± 0.22 ^{**A}	25.85 ± 0.13 ^{**B}	-0.96 ± 0.02 ^{**A}	-1.23 ± 0.04 ^A	-1.07 ± 0.04 ^{**A}	3.87 ± 0.03 ^{**C}	3.24 ± 0.03 ^{**B}	3.30 ± 0.04 ^{**C}

^a The lower-case (a-d) and capital (A-D) letters indicate significant differences between digestive step in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of food matrix for each phase of digestion (* $P < 0.05$ and ** $P < 0.01$).

NPs suspensions in lightness and yellow were more remarkable in the presence of the food matrix than in its absence during this phase ($P < 0.01$).

The e-TiO₂-NPs suspension without or with the presence of the food matrix showed a low level of lightness color ($L^* = 21.08 \pm 0.26$ and 21.59 ± 0.22) and a higher level of green or yellow color (a^* , $b^* = -1.24 \pm 0.09$, 2.33 ± 0.05 and -1.23 ± 0.04 , 3.24 ± 0.03) than that before digestion. In addition, significant increases of color in lightness and yellow were found in the presence of the food matrix during the intestinal phase ($P < 0.01$). A similar tendency was observed at different e-TiO₂-NPs concentrations (0.5%, 1%, and 1.5 w/w), as shown in Table 1.

Changes of hydrodynamic size and PDI

The hydrodynamic sizes and PDIs of the e-TiO₂-NPs suspensions (1% w/w) before digestion and after each step (mouth, stomach and small intestine) of the digestive process were determined by dynamic light scattering (DLS) analysis, as shown in Fig. 3.

The hydrodynamic diameters of the e-TiO₂-NPs dispersed in 10 mM phosphate buffer (442.38 ± 8.72 nm) and the food matrix (328.20 ± 10.96 nm) showed significant differences ($P < 0.01$). The PDI of the e-TiO₂-NPs was measured to be 0.256 ± 0.006 , which decreased to 0.128 ± 0.013 in the presence of the food matrix ($P < 0.01$).

Significant increases of the hydrodynamic size and PDI of the e-TiO₂-NPs without or with the food matrix were found after simulating salivary digestion, ranging from 442.38 ± 8.72 to 873.74 ± 11.89 nm, 0.256 ± 0.006 to 0.373 ± 0.002 nm, 328.20 ± 10.96 to 799.75 ± 64.99 nm and 0.128 ± 0.013 to 0.200 ± 0.019 nm, respectively. Additionally, the presence of the food matrix decreased the hydrodynamic size and PDI of the e-TiO₂-NPs during the salivary step.

Compared to the results obtained before digestion, the hydrodynamic diameters of the e-TiO₂-NPs increased to 1183.42 ± 30.71 nm in the absence of the food matrix and increased to 1330.62 ± 186.115 nm in the presence of the food matrix after simulating gastric digestion. In the absence of the food matrix, the PDI of the e-TiO₂-NPs (0.253 ± 0.041) was slightly decreased by the presence of gastric fluids compared to the PDI before digestion. However, in the presence of the food matrix, the e-TiO₂-NPs PDI increased significantly, varying from 0.128 ± 0.013 before digestion to 0.307 ± 0.031 . In addition, there was an increase of the hydrodynamic size and PDI of the e-TiO₂-NPs when the food matrix was present during the gastric phase.

The hydrodynamic size or PDI of the e-TiO₂-NPs in the absence and presence of the food matrix was further raised to 2899.95 ± 38.49 nm or 0.455 ± 0.045 and 2856.43 ± 18.22 nm or 0.409 ± 0.023 , respectively, during the intestinal phase. Moreover, the presence of the food matrix decreased the hydrodynamic size and PDI of the e-TiO₂-NPs during the intestinal step.

A similar tendency was observed at different concentrations of e-TiO₂-NPs (0.5%, 1%, and 1.5 w/w), as shown in Fig. 4 and Table 2.

Changes in the agglomeration state

The size, shape, and agglomeration state of e-TiO₂-NPs (1% w/w) before digestion and after each step of the digestive

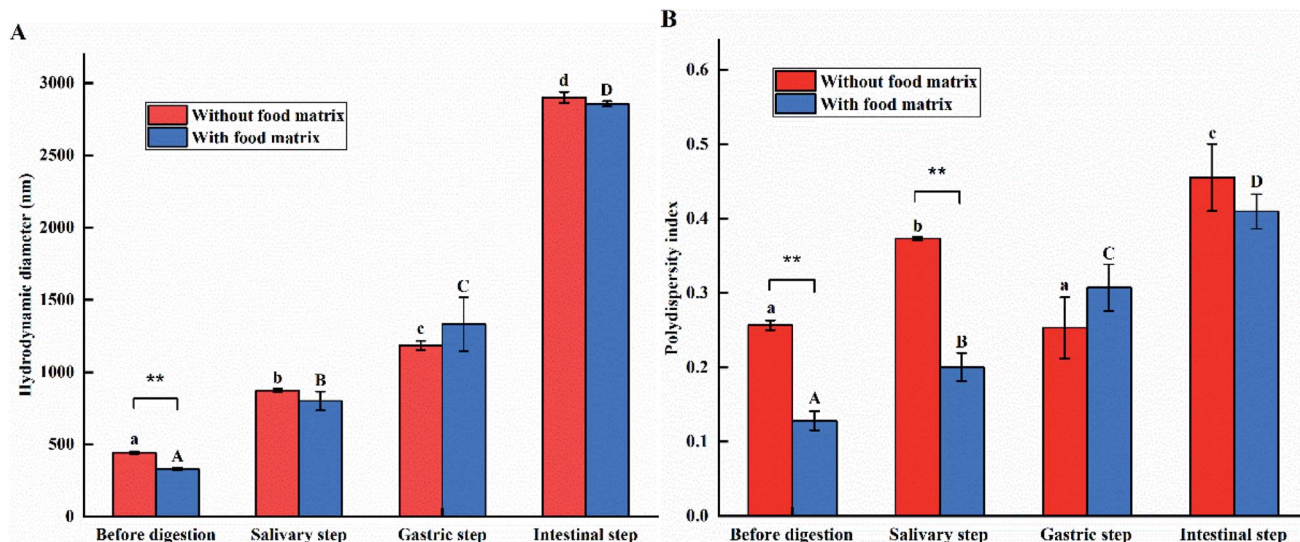


Fig. 3 The hydrodynamic diameters (A) and PDIs (B) of e-TiO₂-NPs suspensions (1% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion. The lower-case (a–d) and capital (A–D) letters indicate significant differences between the digestion step in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of the food matrix for each phase of digestion (* $P < 0.05$ and ** $P < 0.01$).

process were characterized by TEM imaging and are shown in Fig. 5. In the presence of the food matrix before digestion and during the salivary or intestinal step, greater dispersion and smaller agglomeration of the e-TiO₂-NPs were observed compared to those in the absence of the food matrix, which is contrary to the results during the gastric step. Contrary to before digestion, agglomeration of the e-TiO₂-NPs after each phase of simulated human GIT was observed, either in the absence or presence of the food matrix. Major alterations in the e-TiO₂-NPs occurred during the gastric and intestinal phase in both the absence and presence of the food matrix. In the absence of the food matrix, there was a significant increase in the particle size from before digestion to the intestinal phase. A similar tendency was observed in the

presence of the food matrix. EDX was used as a semi-qualitative method to reveal the elemental composition of the specimens. Ti element was observed in the specimens in great amount, which showed that transformed e-TiO₂-NPs could release titanium ion in the gastrointestinal tract (without food matrix: 1.22 ± 0.03 to 0.18 ± 0.05 mg kg⁻¹; with food matrix: 0.42 ± 0.01 to 0.44 ± 0.00 mg kg⁻¹) (Fig. 6).

Changes in the zeta potential

The zeta potentials of the e-TiO₂-NPs (1% w/w) suspensions are shown in Fig. 7. Before simulating the passage of the e-TiO₂-NPs suspension through the GIT, both e-TiO₂-NPs suspensions without or with the presence of the complex food matrix were negatively charged, and the zeta potentials greatly increased in

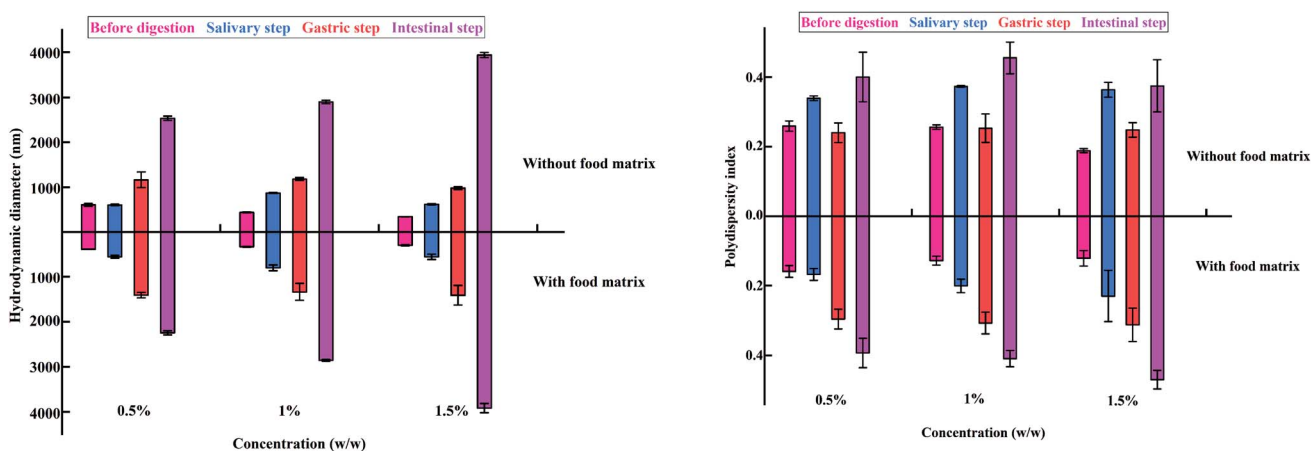


Fig. 4 The hydrodynamic diameters (left) and PDIs (right) of e-TiO₂-NPs suspensions (0.5%, 1%, and 1.5% w/w) without or with the presence of food matrix during each digestive phase *in vitro*. The lower-case (a–d) and capital (A–D) letters indicate significant differences between the digestive steps in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of the food matrix for each phase of digestion (* $P < 0.05$ and ** $P < 0.01$).



Table 2 The hydrodynamic diameters and PDIs of the e-TiO₂-NPs suspensions (0.5%, 1%, and 1.5% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion^a

Food matrix	Digestion step	Hydrodynamic size (nm)			PDI		
		0.5%	1%	1.5%	0.5%	1%	1.5%
Without	Before digestion	610.41 ± 32.11 ^a	442.38 ± 8.72 ^a	344.82 ± 0.92 ^a	0.26 ± 0.01 ^{ab}	0.26 ± 0.01 ^a	0.19 ± 0.01 ^a
	Salivary step	609.35 ± 16.89 ^a	873.74 ± 11.89 ^b	622.51 ± 13.18 ^b	0.34 ± 0.01 ^b	0.37 ± 0.00 ^b	0.36 ± 0.02 ^b
	Gastric step	1164.40 ± 172.43 ^b	1183.42 ± 30.71 ^c	983.05 ± 35.80 ^c	0.24 ± 0.03 ^a	0.25 ± 0.04 ^a	0.25 ± 0.02 ^a
	Intestinal step	2534.06 ± 47.55 ^c	2899.95 ± 38.49 ^d	3941.63 ± 56.64 ^d	0.40 ± 0.07 ^c	0.46 ± 0.05 ^c	0.37 ± 0.17 ^b
With	Before digestion	381.53 ± 5.81 ^{**A}	328.20 ± 10.96 ^{**A}	295.75 ± 13.688 ^{**A}	0.16 ± 0.02 ^{**A}	0.13 ± 0.01 ^{**A}	0.12 ± 0.02 ^{**A}
	Salivary step	554.71 ± 33.85 ^B	799.75 ± 64.99 ^B	554.75 ± 59.61 ^B	0.17 ± 0.02 ^{**A}	0.20 ± 0.02 ^{**B}	0.23 ± 0.07 ^{*B}
	Gastric step	1402.19 ± 61.53 ^C	1330.62 ± 186.12 ^C	1405.62 ± 215.57 ^C	0.30 ± 0.03 ^B	0.31 ± 0.03 ^C	0.31 ± 0.05 ^B
	Intestinal step	2245.30 ± 46.06 ^{**D}	2856.43 ± 18.22 ^D	3917.38 ± 102.69 ^D	0.39 ± 0.04 ^C	0.41 ± 0.02 ^D	0.47 ± 0.03 ^C

^a The lower-case (a–d) and capital (A–D) letters indicate significant differences between digestive step in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of food matrix for each phase of digestion (* $P < 0.05$ and ** $P < 0.01$).

the presence of the food matrix (-25.63 ± 1.36) compared to in its absence (-41.86 ± 1.54).

The zeta potentials of the e-TiO₂-NPs without (-23.70 ± 0.71) or with (-28.35 ± 1.29) the presence of the food matrix after the salivary step significantly increased and decreased, respectively. Meanwhile, a significant difference in the zeta potentials of the e-TiO₂-NPs was found without and with the presence of the food matrix ($P < 0.01$).

Compared to the results obtained before digestion, the strongest increases in the zeta potentials of e-TiO₂-NPs without (-3.38 ± 0.31) or with (-4.22 ± 0.38) the presence of the food matrix were found after the gastric step. Meanwhile,

a significant difference was found between the zeta potentials of the e-TiO₂-NPs without and with the presence of the food matrix ($P < 0.01$).

A similar result was observed during the salivary and intestinal phases. In the absence of the food matrix, the zeta potential of the e-TiO₂-NPs was calculated as -22.45 ± 1.59 mV during the intestinal step, which markedly decreased to -27.37 ± 0.83 mV in the presence of the food matrix ($P < 0.01$).

A similar transformation of the zeta potential of the e-TiO₂-NPs at different concentrations (0.5%, 1%, and 1.5% w/w) was found, while slight differences were also observed Fig. 8 and Table 3.

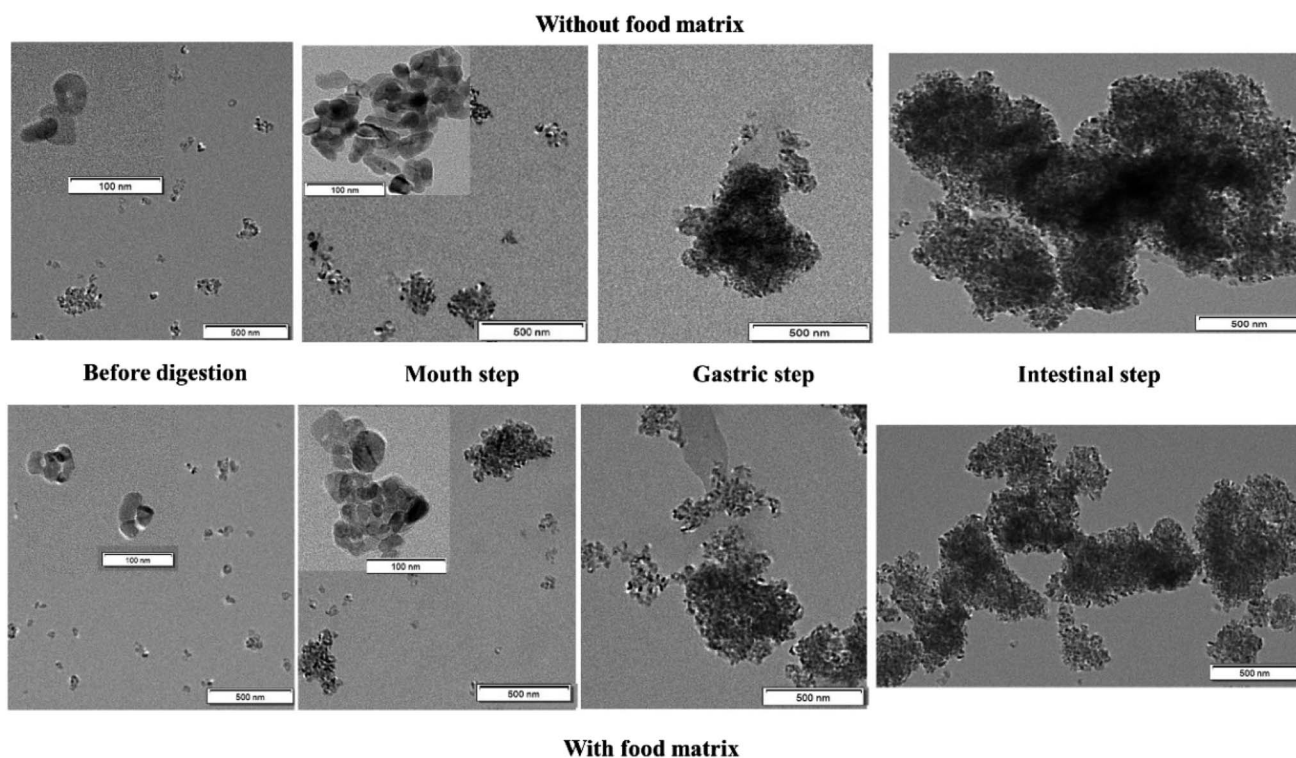


Fig. 5 The agglomeration state of e-TiO₂-NPs suspensions (1% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion. Note: the sizes, shapes, and agglomeration states of the e-TiO₂-NPs (1% w/w) before digestion and after each step of the digestive process were characterized by TEM imaging.



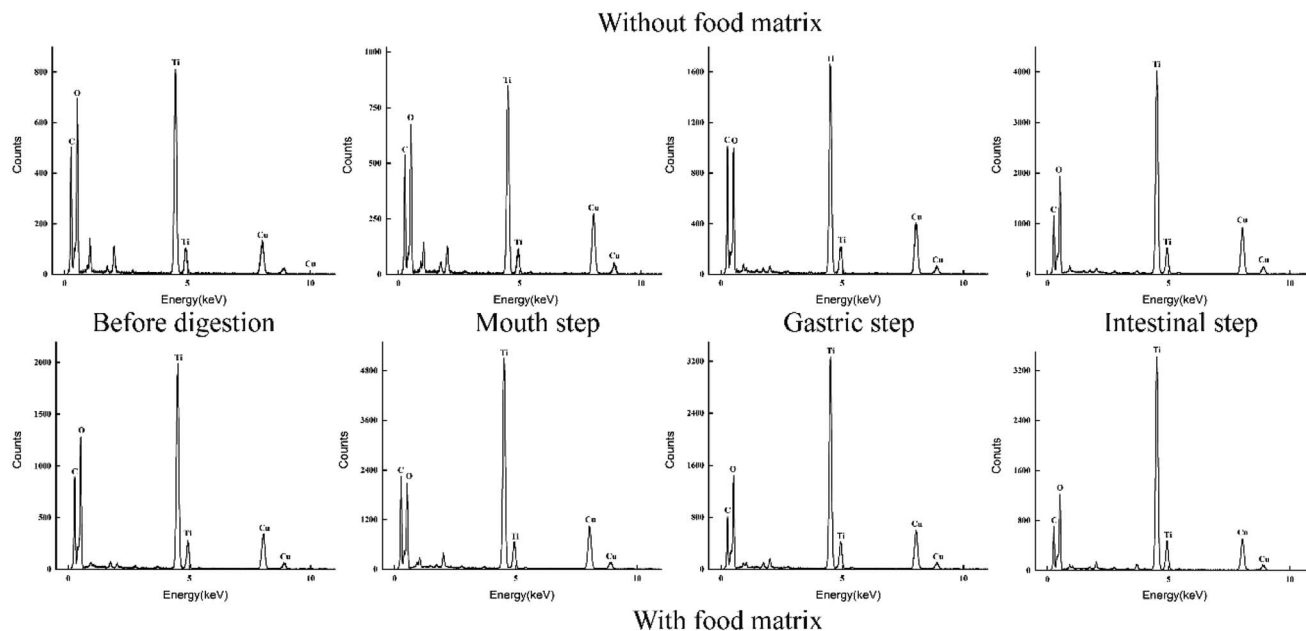


Fig. 6 The elemental compositions of the e-TiO₂-NPs suspensions (1% w/w) without or with the presence of the food matrix during each digestive phase. Note: a similar tendency was observed in the presence of the food matrix. The EDX technique was used as a semi-qualitative method to reveal the elemental compositions of the specimens. The results show that the presence of Ti element was observed in the specimens in great amounts.

Changes in the UV-Vis absorption peaks

The surface plasmon resonance (SPR) peaks of the e-TiO₂-NPs (1% w/w) with background corrections were monitored by UV-Vis

spectroscopy, as shown in Fig. 9. The SPR peak of the e-TiO₂-NPs dispersed in phosphate buffer showed a significant blueshift after the addition of e-TiO₂-NPs to the food matrix (from 339 to 332 nm). The e-TiO₂-NPs suspensions incubated in simulated salivary fluid

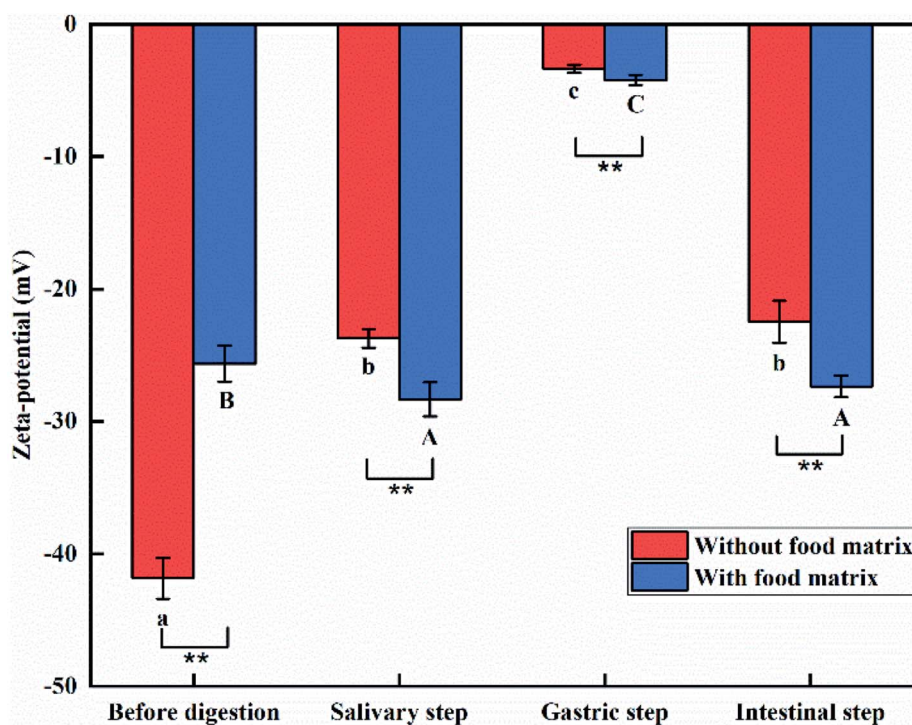


Fig. 7 The zeta-potential of the e-TiO₂-NPs suspensions (1% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion. Note: the lower-case (a–c) and capital (A–C) letters indicate significant differences between the digestive step in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of the food matrix for each phase of digestion (* $P < 0.05$ and ** $P < 0.01$).



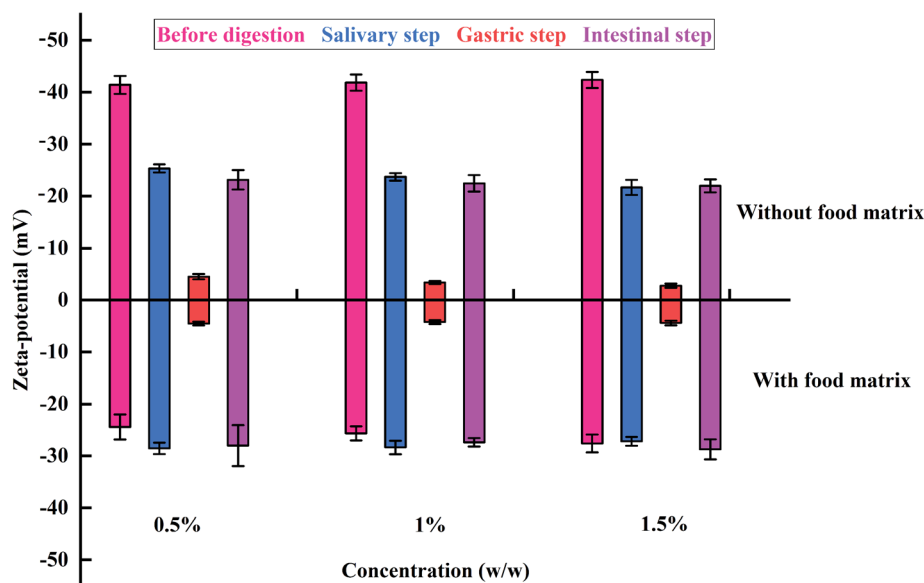


Fig. 8 The zeta potentials of e-TiO₂-NPs suspensions (0.5%, 1%, and 1.5% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion.

exhibited a redshift of the SPR peaks from 339 to 340 nm without the food matrix and from 332 to 337 nm with the food matrix. A blueshift of the SPR peaks due to the presence of the food matrix was observed during this phase as well as before digestion. There was a further successive redshift of the SPR peaks of the e-TiO₂-NPs in the salivary (from 339 to 512 nm in the absence of the food matrix and from 332 to 563 nm in the presence of the food matrix) and gastric (from 339 to 602 nm in the absence of the food matrix and from 332 to 619 nm in the presence of the food matrix) phases. Additionally, there was a redshift of the e-TiO₂-NPs SPR peaks when the food matrix was present during these phases ($P < 0.01$). A similar tendency was observed for different e-TiO₂-NPs concentrations (0.5%, 1%, and 1.5% w/w), as shown in Fig. 10 and Table 4.

Changes of solubility

The Ti ion concentrations from the dissolution of the e-TiO₂-NPs (1% w/w) were determined by ICP-AES, and the results are shown in Fig. 11.

The Ti ions released from the dissolution of e-TiO₂-NPs were measured as $1.22 \pm 0.03 \text{ mg kg}^{-1}$, which significantly decreased to $0.42 \pm 0.01 \text{ mg kg}^{-1}$ in the presence of the food matrix ($P < 0.01$).

The solubility of e-TiO₂-NPs without or with the presence of the food matrix respectively decreased to 0.48 ± 0.00 and 0.31 ± 0.06 after simulating salivary digestion, which were significantly different from each other ($P < 0.01$).

The solubility of the e-TiO₂-NPs was lower in the gastric phase than before digestion, either in the absence (0.30 ± 0.00) or presence (0.22 ± 0.01) of the food matrix. Moreover, the presence of the food matrix significantly decreased the dissolution of e-TiO₂-NPs during this phase ($P < 0.01$).

Compared to the results obtained before digestion, the largest decrease of the amount of Ti ion released from the e-TiO₂-NPs in the simulated intestinal phase was observed when the food matrix was absent (0.18 ± 0.05). An opposite result was observed when the food matrix was present (0.44 ± 0.00) during

Table 3 The zeta potentials of e-TiO₂-NPs suspensions (0.5%, 1%, and 1.5% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion^a

Food matrix	Digestion step	Zeta-potential (mV)		
		0.5%	1%	1.5%
Without	Before digestion	-41.41 ± 1.72^a	-41.86 ± 1.54^a	-42.36 ± 1.52^a
	Salivary step	-25.33 ± 0.79^b	-23.70 ± 0.71^b	-21.68 ± 1.47^b
	Gastric step	-4.52 ± 0.52^d	-3.38 ± 0.31^c	-2.78 ± 0.40^c
	Intestinal step	-23.13 ± 1.87^c	-22.45 ± 1.59^b	-21.98 ± 1.26^b
With	Before digestion	$-24.40 \pm 2.41^{**B}$	$-25.63 \pm 1.36^{**B}$	$-27.58 \pm 1.72^{**A}$
	Salivary step	$-28.53 \pm 1.09^{**A}$	$-28.35 \pm 1.29^{**A}$	$-27.17 \pm 0.86^{**A}$
	Gastric step	-4.52 ± 0.37^C	$-4.22 \pm 0.38^{**C}$	$-4.39 \pm 0.45^{**B}$
	Intestinal step	$-28.00 \pm 3.96^{**A}$	$-27.37 \pm 0.83^{**A}$	$-28.72 \pm 1.93^{**A}$

^a The lower-case (a–d) and capital (A–C) letters indicate significant differences between digestive step in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of food matrix for each phase of digestion ($*P < 0.05$ and $**P < 0.01$).



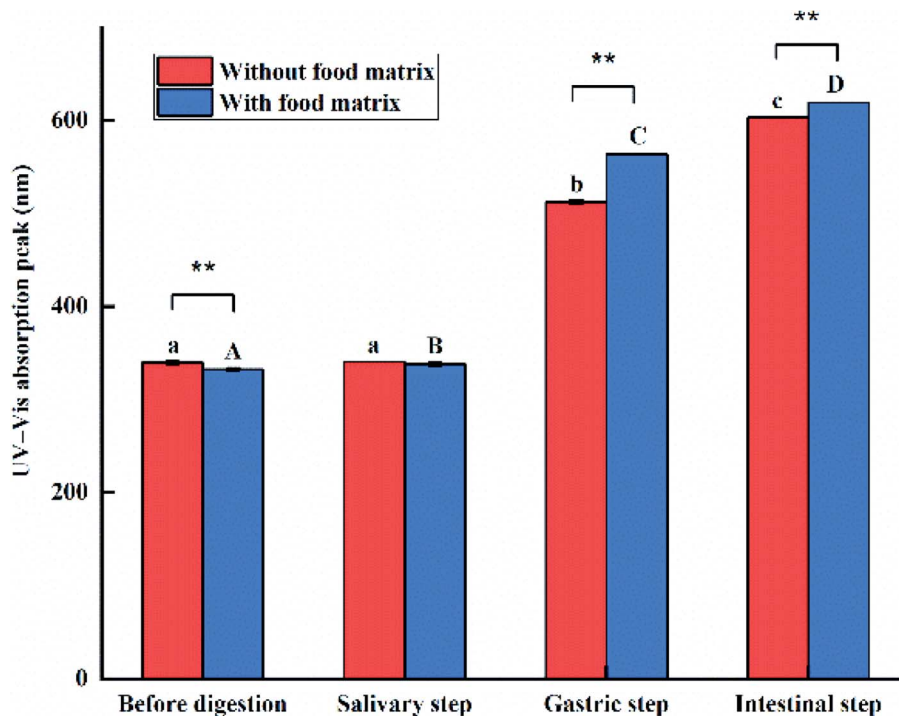


Fig. 9 The SPR peaks of e-TiO₂-NPs suspensions (1% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion. The lower-case (a–c) and capital (A–C) letters indicate significant differences between the digestive step in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of the food matrix for each phase of digestion ($*P < 0.05$ and $**P < 0.01$).

this phase. Apparently, the presence of the food matrix increased the amount of Ti ions released in the simulated intestinal phase ($P < 0.01$).

Cell viability analysis

A nanotoxicology case study was carried out to assess the toxicity of the digested e-TiO₂-NPs using an *in vitro* cellular

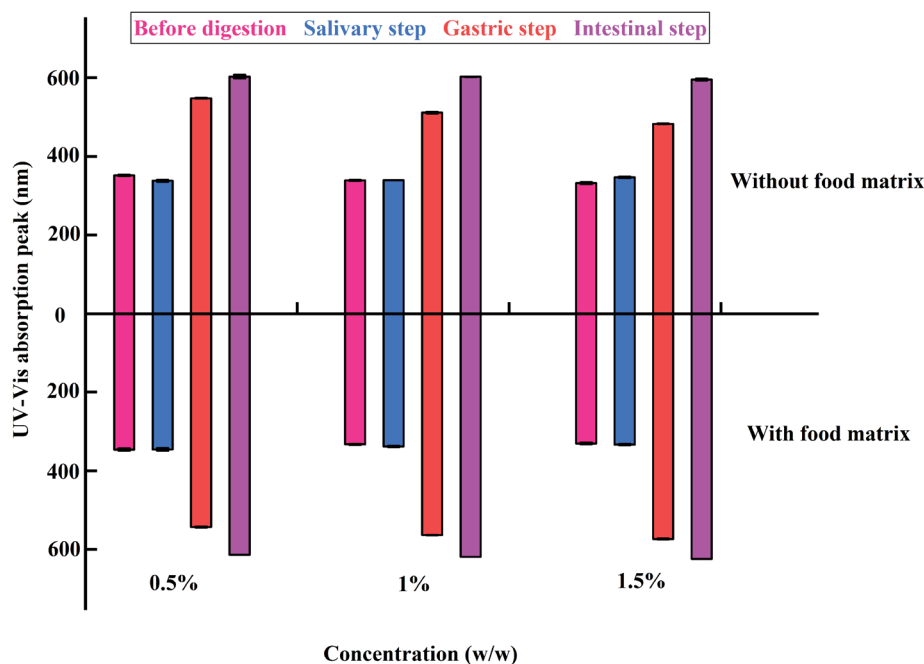


Fig. 10 The SPR peaks of e-TiO₂-NPs suspensions (0.5%, 1%, and 1.5% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion.

Table 4 The SPR peaks of e-TiO₂-NPs suspensions (0.5%, 1%, and 1.5% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion^a

Food matrix	Digestion step	UV-Vis absorption peak		
		0.5%	1%	1.5%
Without	Before digestion	352 ± 1.63 ^b	339 ± 1.49 ^a	332 ± 2.43 ^a
	Salivary step	338 ± 2.43 ^a	340 ± 0.00 ^a	347 ± 2.04 ^b
	Gastric step	548 ± 1.15 ^c	512 ± 2 ^b	483 ± 1.15 ^c
	Intestinal step	602 ± 4.62 ^d	602 ± 0.58 ^c	595 ± 2.31 ^d
With	Before digestion	345 ± 3.14 ^{**A}	332 ± 1.37 ^{**A}	330 ± 2.31 ^A
	Salivary step	345 ± 3.40 ^{*A}	337 ± 1.79 ^B	333 ± 1.53 ^{**A}
	Gastric step	543 ± 1.73 ^{**B}	563 ± 0.58 ^{**C}	573 ± 1.15 ^{**B}
	Intestinal step	614 ± 0.00 ^{*C}	619 ± 0.00 ^{**D}	624 ± 0.00 ^{**C}

^a The lower-case (a–d) and capital (A–D) letters indicate significant differences between digestive step in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of food matrix for each phase of digestion (* $P < 0.05$ and ** $P < 0.01$).

model and Pierce MTS cytotoxicity assay kit to measure cell viability. Higher cell viability of the e-TiO₂-NPs digesta in the presence of the food matrix was observed. Meanwhile, the cell viability of the e-TiO₂-NPs digesta in the presence of the food matrix was higher than that in its absence for the same dose of e-TiO₂-NPs. The cell viabilities of the e-TiO₂-NPs suspensions (0.5, 1% and 1.5% w/w) are shown in Fig. 12.

Discussion

The present study investigated the influence of a complex food matrix (SFM) and the gastrointestinal tract on physicochemical aspects of e-TiO₂-NPs, simulating the real complex, multi-

component food digestive system. A variety of analytical techniques, such as colorimetry, UV-Vis spectroscopy, DLS, TEM-EDX and ICP-AES, were used to provide valuable information, including the shapes, sizes, agglomeration states, charges and solubility of the e-TiO₂-NPs in the complex food matrix (SFM) and gastrointestinal tract milieu.

The optical characterization of the e-TiO₂-NPs revealed that the tristimulus colors before digestion and after each step of the digestive process were differently affected in the presence of the food matrix. The results revealed both apparent and different increases of the colors green and yellow in the e-TiO₂-NPs suspensions without or with the presence of the food matrix when the e-TiO₂-NPs suspensions were passed through each

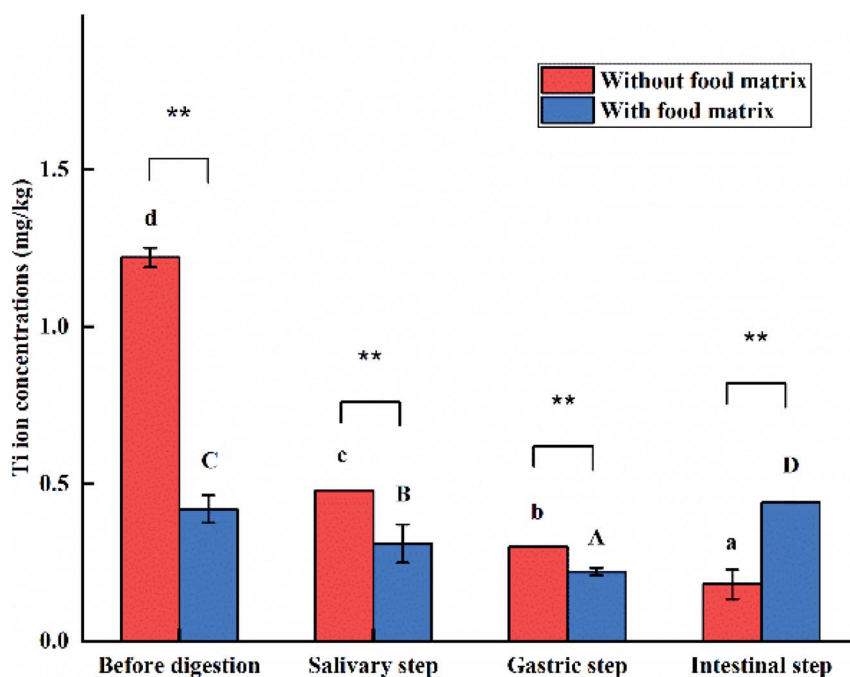


Fig. 11 The solubility of e-TiO₂-NPs suspensions (1% w/w) without or with the presence of the food matrix during each phase of *in vitro* digestion. Note: the lower-case (a–d) and capital (A–D) letters indicate significant differences between the digestive steps in the absence and presence of the food matrix ($P < 0.05$). * indicates significant differences between the absence and presence of the food matrix for each phase of digestion (* $P < 0.05$ and ** $P < 0.01$).



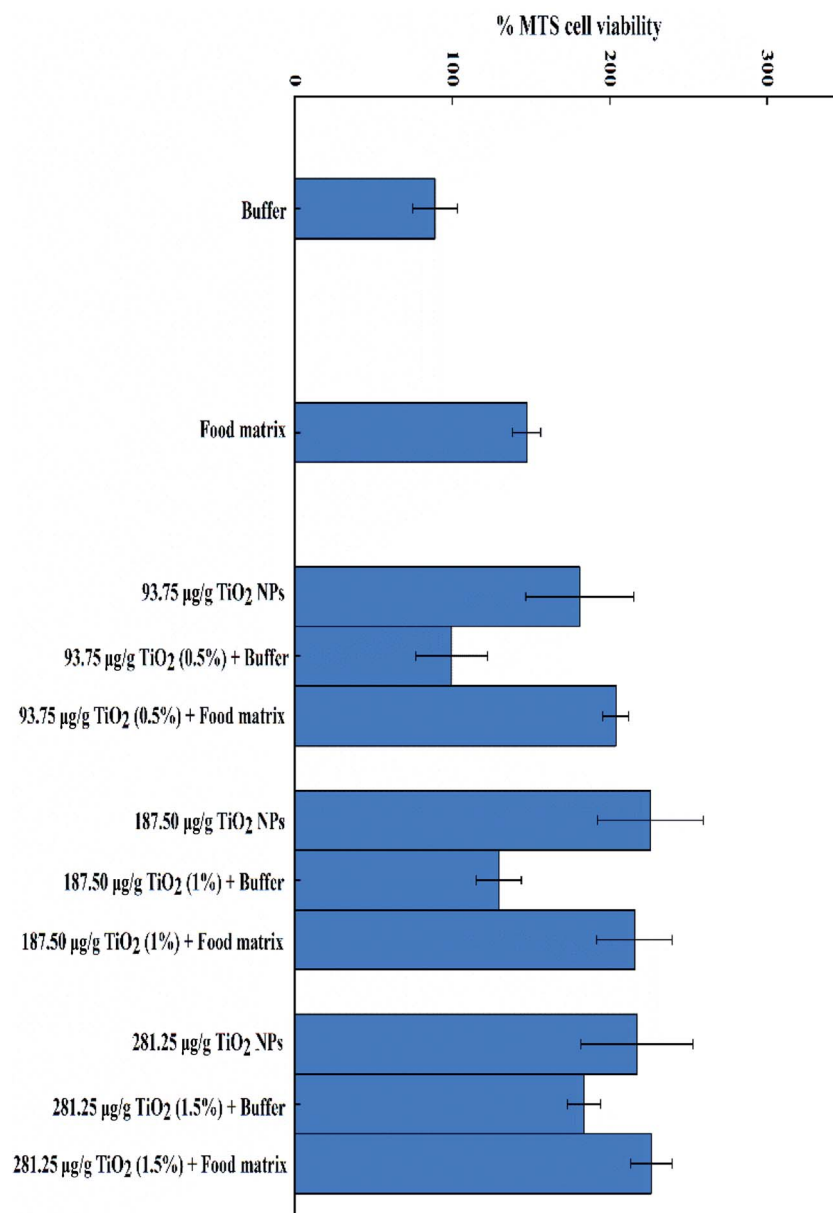


Fig. 12 The cell viabilities of e-TiO₂-NPs digesta (0.5%, 1% and 1.5% w/w) without (buffer) or with the presence of the food matrix during each phase of *in vitro* digestion.

phase of a three-stage simulated digestion process compared to those before digestion. However, the lightness (L^*) of the e-TiO₂-NPs suspensions without or with the presence of the food matrix appreciably increased during the salivary or gastric step, and the decrease was appreciable during the intestinal step (excepting 1.5 w/w). In addition, the presence of the food matrix could considerably increase the lightness and yellow of the e-TiO₂-NPs suspensions, which depend on the concentration and immediate environment of the e-TiO₂-NPs.

The DLS data showed that the hydrodynamic sizes of the e-TiO₂-NPs were differently agglomerated and aggregated in the absence and presence of the food matrix and then were variously affected during each step of *in vitro* digestion. The hydrodynamic size of the e-TiO₂-NPs increased significantly and

gradually as the *in vitro* digestion steps increased when they were dispersed in phosphate buffer and the food matrix, respectively. Aggregation of the NPs mainly occurred in the GIT. The drop in pH, salt concentration, presence of digestive enzymes during *in vitro* digestion, hydrophobic interactions, ionic strength and bridging or depletion flocculation could have induced the agglomeration.^{7,18,35} Moreover, the hydrodynamic size of the e-TiO₂-NPs was greater before digestion and after each step of the digestive process, except for the gastric step, in the absence of the food matrix compared to in its presence; this indicates that the complex food matrix (SFM) may contribute to NPs dispersion, which was probably due to the role of multiple food components. For example, the complex protein matrices (skim milk), rather than only one component (*i.e.*, casein,

lactose, and calcium), may contribute to the dispersion of the ZnO NPs by repulsive electrostatic forces in the presence of minor minerals and individual amino acids in the complex food matrix.^{15,36} Another study showed that minor nutrients in honey contribute to agglomeration of NPs.¹⁶ Moreover, saccharides and proteins (*i.e.*, acacia honey, fructose, glucose, lactose or albumin), as NP dispersants, contribute to NP dispersion; this is associated with the type and concentration of NPs or the food matrix type and the physicochemical parameters of the surrounding medium used for assessment.^{12,16,37} Meanwhile, food components or matrices (protein, starch, fat and minerals) are thickening or gelling agents that may change the rheological properties of the gastrointestinal fluid and then change the properties of the nanoparticles in the GIT.^{38,39} For the PDI, the study results showed that there was both an apparent increase of the PDI of the e-TiO₂-NPs without or with the presence of the food matrix when the e-TiO₂-NPs suspensions were passed through a three-stage simulated digestion process. In contrast, in the absence of the food matrix, reduced PDI of the e-TiO₂-NPs after the gastric step was found compared to before digestion. In addition, the presence of the food matrix decreased the PDI of the e-TiO₂-NPs except for in the gastric step.

The size, shape, and agglomeration state of e-TiO₂-NPs were clearly demonstrated by TEM-EDX after adding e-TiO₂-NPs to phosphate buffer or the food matrix (SFM) and then respectively digesting e-TiO₂-NPs in the absence and presence of food matrix in the three GIT fluids to mimic the passage through the human mouth, stomach and small intestine. Higher dispersion before digestion and smaller agglomeration at the salivary and intestinal step were observed in the e-TiO₂-NPs dispersed in the food matrix (SFM) relative to in its absence, which suggests that the complex food matrix acts as a dispersant. Meanwhile, the agglomeration state of e-TiO₂-NPs during the three GIT fluids showed a similar increased tendency when dispersed in phosphate buffer and the food matrix. The results indicate that the size, shape, and agglomeration state of the e-TiO₂-NPs can be affected in the simulated human GIT, which is consistent with the DLS results. Previous studies have reported the aggregation of NPs in the presence of food matrices, and the simulated human GIT and food components may stabilize the NP suspensions.⁴⁰ Pepsin in gastric fluids could promote the aggregation of Ag NPs by adsorbing on the surface of the nanoparticles.⁴¹

The zeta-potential of the e-TiO₂-NPs was also found to be strongly affected by the presence of the food matrix (SFM) and the simulated human GIT fluids. The zeta-potentials of the e-TiO₂-NPs suspensions were all negative. The gastric step was more effective to reduce the negative zeta-potential of the e-TiO₂-NPs in the absence or presence of the food matrix. Moreover, the negative zeta-potential of the e-TiO₂-NPs before digestion and after each step of digestive process was found to be strongly affected by the presence of the food matrix (SFM). There was an apparent increase of the negative zeta-potential of the e-TiO₂-NPs obtained from each step of the digestive process. In contrast, the decrease of the negative zeta-potential of the e-TiO₂-NPs before digestion was obvious in the presence of the food matrix (SFM).

The UV-Vis measurements showed that the SPR peak of e-TiO₂-NPs could be affected by the presence of the food matrix and simulated GIT, which is associated with the DLS and TEM-EDX results. Meanwhile, alterations of the SPR peak of the NPs are very likely to reflect changes in the NP surface properties because the SPR peak depends on the surface plasmon resonant frequency.^{19,42} Both before digestion and after the salivary step, e-TiO₂-NPs without the presence of the food matrix showed various degrees of redshift of the SPR peak compared with those in its presence, suggesting greater dispersion of e-TiO₂-NPs in the presence of the food matrix, which is contrary to the results obtained in the gastric and intestinal phases. Additionally, continuous redshifts of the SPR peaks of the e-TiO₂-NPs with or without the presence of the food matrix after each step of *in vitro* digestion were found compared to the e-TiO₂-NPs suspension before digestion, which suggests that the e-TiO₂-NPs incessantly aggregated during these phases, especially the gastric and intestinal phases. Moreover, the formation of biomolecular corona and the increase of the ionic strength may be reasons for the redshift of the SPR peaks of the NPs.^{19,43}

The solubility of the e-TiO₂-NPs was also found to be strongly affected by the presence of the food matrix (SFM) and the simulated human GIT fluids. The amount of Ti ion decreased along the successive phases of the simulated human GIT (*i.e.*, mouth > stomach > intestine) in the absence and presence of the food matrix (SFM). Interestingly, in the presence of the food matrix (SFM), moderately high solubility was found during the intestinal step compared with the results obtained before digestion. Meanwhile, the presence of the food matrix (SFM) decreased the dissolution of the e-TiO₂-NPs differently, excepting the intestinal step. However, the results from this study showed that less than 0.1% of the e-TiO₂-NPs suspensions were dissolved in all cases, in line with previous studies.¹² Previous studies showed that the dissolution of NPs in the presence of food components or under physiological conditions can affect their bioavailability or biological fate, as in the cases of ZnO, TiO₂ and Ag NPs.^{8,12,44,45} Therefore, we need to further study the potential toxicity and underlying mechanisms using the food matrix, especially the standardized food model (SFM), and the passage of NPs through the gastrointestinal tract.

A nanotoxicology case study was carried out, and the results showed that the cell viability of the e-TiO₂-NPs was also strongly affected by the presence of the food matrix (SFM) and the simulated human GIT fluids. The presence of the food matrix and the simulated human GIT fluids may increase the cell viability of e-TiO₂-NPs, which highlights the importance of food matrix effects and GIT effects on the toxicological assessment of nanoparticles.

Conclusion

In this study, a standardized food model (SFM) and simulated gastrointestinal fluids have been used to study the fate of e-TiO₂-NPs following a three-step digestion model *in vitro*. The results evidenced that in the absence and presence of the food matrix (SFM), different levels of transformations of the tristimulus color coordinates, size, agglomeration state, surface



charge and solubility of the e-TiO₂-NPs during salivary, gastric and intestinal digestion fluids compared to the results obtained before digestion are closely related to the pH shifts and variable concentrations of enzymes and salts. In short, the presence of a food matrix (a standardized food model) significantly impacted the physicochemical properties of the e-TiO₂-NPs. Moreover, the e-TiO₂-NPs showed differentiated transformations of their physicochemical properties after each step of the digestive process. The pH shifts and variable concentrations of enzymes and salts in gastrointestinal fluids induced the transformations of the physicochemical properties of the e-TiO₂-NPs. The food matrix stabilized the e-TiO₂-NPs suspensions and acted as dispersants during each digestive phase. Transformed e-TiO₂-NPs could release titanium ion in the gastrointestinal tract. The cell viability induced by the e-TiO₂-NPs was also found to be strongly affected by the presence of the SFM and the simulated human GIT fluids. It can be concluded that physicochemical transformations of e-TiO₂-NPs occurred when they were incorporated into the SFM and passed through the GIT, which consequently strongly affected the biological effects of the e-TiO₂-NPs; this highlights that in the toxicity assessment of ingested NPs, an appropriate standardized food model should be used and realistic physiological conditions should be taken into account.

Conflicts of interest

There are no conflicts to declare.

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