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Sign inversion in magnetic circularly polarised luminescence of fused aromatics with 1.6 T N-up/S-up Faraday geometry†

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Although 12 diamagnetic fused aromatics with or without substituents exhibit mirror-symmetric magnetic circularly polarised luminescence (MCPL) through N-up and S-up Faraday geometries under a magnetic field intensity of 1.6 T, their signs (single and multiple) and magnitudes depend strongly on either the aromatic structures or the peripheral positions of the substituents.

Introduction

In recent years, highly controlled chirogenesis from organic, inorganic, and polymeric luminophores in the photoexcited state (ES) and the ground state (GS) has received considerable attention in photonic materials science and engineering.¹ In particular, sophisticated organic luminophores that can emit circularly polarised luminescence (CPL) with a high dissymmetry ratio (g_{CPL}), high quantum yield (Φ_{F}), and additional functionality remain challenging to produce and are in great demand from the perspective of extensibility.²

Generally, to afford the ambidextrous (+ and −) signs of CPL signals, enantiomerically pure luminophores with twisted and/or chirally distorted configurations and conformations designed rationally in the ES and GS are imperative. This can be achieved by choosing rigid chiral π -conjugated frameworks associated with both (*S*)- and (*R*)-stereogenic centres and/or both *P*- and *M*-stereogenic bonds at their peripheral positions. Such enantiomeric pairs are not always available as the starting source materials, however, and therefore, enantiomeric CPL-functioned substances often require multiple-step synthesis.

Recently, we demonstrated that simple control of ambidextrous CPL signs using semi-rigid chiral luminophores with few rotor axes is possible if appropriate achiral solvents and chiral additives with few rotor axes are chosen.³

Contrarily, for affording such ambidextrous CPL in the ES, interaction between unpolarised light and external magnetic

field,⁴ the so-called Faraday and polar-Kerr effects, is a versatile physical source, as is the use of electric field and vortex fluidic solvents. In fact, we showcased mirror-image magnetic circularly polarised luminescence (MCPL) from D_{2h} -symmetrical planar pyrene and six-coordinate C_3 -symmetrical lanthanides with achiral tris(β -diketonate) complexes.⁵ Even C_2 -symmetrical binaphthyl bearing two achiral pyrene rings showed nearly mirror-symmetrical MCPL.⁵

Although ambidextrous MCPL spectra under 1.6 T are easy to achieve by choosing both N-up and S-up Faraday geometries,⁵ questions remain unanswered as to whether (i) the resulting MCPL sign among all diamagnetic aromatics is commonly monosignate, (ii) MCPL signs are determined solely by the N-up/S-up geometry, (iii) the MCPL sign is affected by the substituents at peripheral positions, and (iv) the MCPL magnitude is determined by the number of fused rings.

To address these queries, we experimentally investigated the spectral characteristics of MCPL and unpolarised photoluminescence (PL) with $H_0 = 1.6$ T and those of CPL and PL with $H_0 = 0.0$ T, including the dissymmetry ratio (g_{MCPL}), spectral profiles, and Φ_{F} of several fused aromatic molecules dissolved in common organic solvents; some of the solvents were dispersed in a PMMA (poly(methyl methacrylate)) film by applying both the N-up and S-up geometries.

Experimental

Materials

PHE, NP, NP-1 and PMMA were purchased from FUJIFILM Wako Pure Chemical (Osaka, Japan). ANT-1 was purchased from Combi-Blocks (San Diego, USA). ANT, ANT-2, NP-2, BP and COR were purchased from Tokyo Chemical Industry (Tokyo, Japan). PHE-2 and PHE-3 were purchased from Merck-Sigma-Aldrich Japan (Tokyo, Japan). Spectroscopic grade solvents (CHCl_3 , THF, cyclohexane (CHX), and dimethyl sulfoxide (DMSO)) were purchased from Dojindo Laboratories (Kumamoto, Japan).

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Instrumentations

MCPL, CPL and PL measurements. MCPL, CPL and PL spectra were acquired using a JASCO (Hachioji-Tokyo, Japan) CPL-300 spectrofluoro-polarimeter equipped with and without a JASCO PM-491 permanent magnet (1.6 T) at room temperature at a forward scattering angle with 0° upon excitation by unpolarised monochromatic incident light with a 10 nm bandwidth. PMMA film samples were prepared by spin-coating **PHE**-containing PMMA CHCl_3 solution (1.0×10^{-2} M) at 3000 rpm (Opticoat MS-A100, Mikasa, Tokyo, Japan). The excitation wavelengths were 355 nm for **ANT-1** and **ANT-2** in DMSO; 340 nm for **ANT** in DMSO; 280 nm for **NP-1**, **NP-2**, and **NP** in DMSO; and 270 nm for **PHE** in DMSO, CHCl_3 , THF, and the PMMA film, as well as for **PHE-2** and **PHE-3** in DMSO. A 5 mm and 10 mm pathlength cell were used for solution-state MCPL and CPL measurements, respectively. **BP** in DMSO was excited at 350 nm and **COR** in DMSO was excited at 340 nm.

UV-Vis absorption measurement. Solution UV-Vis absorption spectra of the luminophores were measured with a JASCO V-670KU spectrophotometer with 10 mm pathlength. The value of Φ_F was obtained with a quantum yield spectrometer (Hamamatsu Photonics, Hamamatsu, Shizuoka, Japan, model C9920-02).

Results and discussion

We chose six unsubstituted aromatics: naphthalene (**NP**), anthracene (**ANT**), phenanthrene (**PHE**), benzo[ghi]perylene (**BP**), coronene (**COR**), and pyrene (**PY**).^{5a} To investigate substituent effects, we also studied six **NP**, **PHE**, and **ANT** derivatives carrying electron-withdrawing carboxylic acids at two different peripheral positions (**NP-1** and **NP-2**, **PHE-2** and **PHE-3**, and **ANT-1** and **ANT-2**), as shown in Fig. 1.

We observed that the 12 fused aromatics with or without the substituents exhibit mirror-symmetric MCPL spectra under N-up and S-up Faraday geometries. Among these aromatics, **BP**

and **COR** exhibit unexpectedly complex MCPL spectral profiles with multiple signs. The other aromatics have simple monosignate MCPL spectra, although their signs and magnitudes largely depend on either the topology of the aromatics or the position of the substituents. All UV-visible absorption spectra of the aromatic luminophores in solutions before the MCPL, CPL, and PL spectral measurements are depicted in Fig. S1–S5, ESI.†

Firstly, we compared the MCPL and PL spectra under 1.6 T with the N-up and S-up geometries associated with the corresponding CPL/PL spectra without the magnetic field (Fig. 2 and Table 1). When PL intensity is normalised so that $\text{PL} = (I_L + I_R)/2 = 1.0$ from the raw PMT DC voltage, the MCPL and CPL spectra can be normalised by the common equations g_{MCPL} at 1.6 T = $(I_L - I_R)/[(I_L + I_R)/2]$ and g_{CPL} at 0 T = $(I_L - I_R)/[(I_L + I_R)/2]$, where I_L and I_R denote the intensities of the left- and right-MCPL and CPL components, respectively, upon excitation by unpolarised light. Secondly, the g_{MCPL} values at 1.6 T from an apparent MCPL spectrum were normalised to be g_{MCPL} per 1.6 T and expressed in units of T^{-1} . This is because g_{MCPL} may depend on the strength of the magnetic field to some extent.

NP, **ANT**, and **PHE** in CHX emitted clear MCPL (Fig. 2(a–c)). Among these unsubstituted aromatics, **PHE** in CHX exhibited the highest $|g_{\text{MCPL}}|$ value, on the order of 10^{-3} T^{-1} . The other two aromatics (**NP** and **BP**) in CHX have $|g_{\text{MCPL}}|$ values on the order of 10^{-4} T^{-1} . In addition, these luminophores in DMSO also exhibited clear MCPL signals (Fig. 2(d–f)). In particular, MCPL from **PHE** in DMSO is as strong as MCPL from **PHE** in CHX. For **PHE**, solvent effects and external matrix effects were also examined. **PHE** emitted clear MCPL in THF and CHCl_3 solutions and in PMMA film (Fig. S6†). Interestingly, cyclic π -conjugated **BP** and **COR** also exhibited strong MCPL in DMSO (Fig. 2(g and h)).

The sign of MCPL from **PHE** and **ANT** is clearly negative (red lines; upper panel in Fig. 2, S7† and Table 1), and **NP** appears to exhibit positive-sign signals for an N-up Faraday geometry; this indicates that the $\text{N} \rightarrow \text{S}$ magnetic field relative to the incident light is applied. Conversely, a negative sign (blue lines; upper panel in Fig. 2 and S7† and Table 1) occurs for the S-up geometry, as expected. Both the N- and S-up Faraday geometries enable the verification of nearly mirror-image MCPL signals from achiral aromatic luminophores. Notably, the MCPL signals from **NP** in CHX and DMSO are oppositely signed, although the reason for this is unknown. This shows that the MCPL signs can be controlled not only by the direction of magnetic field but also by the solvent.

NP, **ANT**, **PHE**, **BP**, and **COR** do not exhibit clear CPL signals (green lines; upper panel in Fig. 2), reflecting the achiral planarity in the ES as well as in the GS in the absence of the 1.6 T magnet. Thus, our MCPL instrument with a magneto-optical set-up and the CPL instrument with a chiro-optical set-up are guaranteed to detect MCPL and CPL signals precisely from achiral luminophores, regardless of the presence or absence of the 1.6 T magnetic field.

Among the planar aromatics, the g_{MCPL} , MCPL extremum wavelength (λ_{MCPL}), and Φ_F (at 0.0 T) of **PHE** depend to some extent on the nature of the fluidic solvents and solidified film used, although the MCPL signs remain unchanged. For

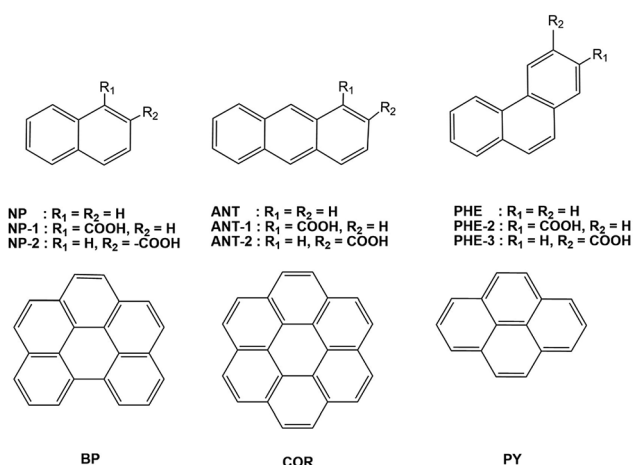


Fig. 1 Chemical structures of naphthalene (**NP**), anthracene (**ANT**), phenanthrene (**PHE**), benzo[ghi]perylene (**BP**), coronene (**COR**), and pyrene (**PY**) without or with substituents at two different peripheral positions (**NP-1**, **NP-2**, **ANT-1**, **ANT-2**, **PHE-2**, and **PHE-3**).



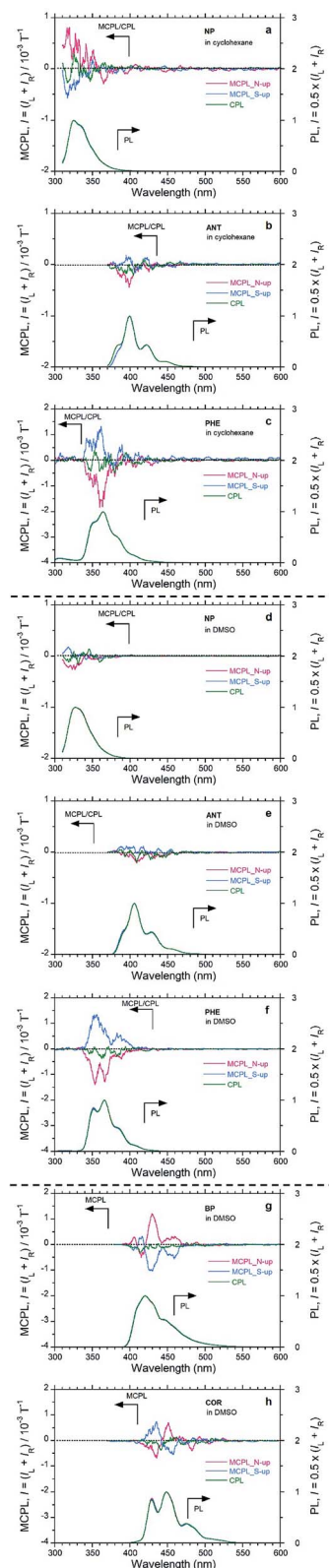


Fig. 2 MCPL spectra at $H_0 = 1.6$ T (red and blue lines denote N-up and S-up geometries, respectively), CPL spectra at $H_0 = 0.0$ T (green line), and the corresponding PL spectra in dilute solutions (1.0×10^{-3} M); pathlengths were 5 mm for MCPL and 10 mm for CPL. (a) NP in CHX excited at 280 nm, (b) ANT in CHX excited at 340 nm, and (c) PHE in CHX excited at 270 nm, (d) NP in DMSO excited at 280 nm, (e) ANT in DMSO excited at 340 nm, and (f) PHE in DMSO excited at 270 nm. (g)

example, the g_{MCPL} values of **PHE** in CHCl_3 and in PMMA are $+1.76 \times 10^{-3} \text{ T}^{-1}$ at 349 nm (S-up) and $-1.63 \times 10^{-3} \text{ T}^{-1}$ at 345 nm (N-up), respectively, while the $|g_{\text{MCPL}}|$ values in the other conditions range from $0.9 \times 10^{-3} \text{ T}^{-1}$ to $1.3 \times 10^{-3} \text{ T}^{-1}$. DMSO exhibits the highest Φ_F value, 0.108, while the other liquids and solid media exhibit lower Φ_F values of 0.03–0.05. This uniqueness of DMSO may arise from a highly polar and C_{2v} -symmetrical structure with two C–C rotor axes.

The shortest MCPL wavelengths (λ_{MCPL}) of **NP**, **ANT**, **PHE**, **BP**, **COR**, and **PY** are 316–317 nm (not prominent due to the noise), 345–361 nm, 415 nm, 435 nm, and 374 nm, respectively, corresponding to the vibronic 0–0' PL bands of the monomer species in the media. This knowledge facilitates tailoring of solely the λ_{MCPL} , which can be performed by choosing various well-known and/or well-designed aromatic luminophores ranging from the UV-visible to the near infrared (NIR) regions.

Between **ANT** and **PHE** (which are three fused-ring aromatics), **PHE** exhibits clear MCPL but **ANT** does not. We assume that it is more advantageous for a bent or curved π -structure to emerge and boost MCPL compared with a linear π -structure. **BP** and **COR** are examples of bent and curved aromatics. Furthermore, among the six aromatics, **PY** has the greatest $|g_{\text{MCPL}}|$ value, 7×10^{-3} to $8 \times 10^{-3} \text{ T}^{-1}$, which is exceptionally high compared with the other five values. **PY** is regarded as an extended **PHE** derivative with a two-fold symmetry. Furthermore, **BP** is considered as a **PHE** derivative fused with biphenyl moiety, while **COR** is considered a six-fold symmetrical structure composing two **PHE** blocks.

Contrary to expectations, we observed unique MCPL spectra with multiple signs for **BP** and **COR** in DMSO (Fig. 2(g and h)): a (+/–/+)-quadruple-sign sequence for **BP** with the N-up geometry and *vice versa* with the S-up geometry; a (–/+/–)-triple-sign sequence for **COR** with the N-up geometry, and *vice versa* with the S-up geometry. The apparent MCPL sign inversion characteristics appear to occur at the 0–0', 0–1', 0–2', and 0–3' vibronic transitions. A clear explanation for the origin of the multiple sign inversion could not be found. Our experimental results indicate that MCPLs induced by an external magnetic field are related to the photoexcited vibronic transition states arising from distorted C=C bonds in the fused ring and in-plane/out-of-plane C–H bending modes.

The MCPL sign is significantly affected by the peripheral positions of the substituents (Fig. 3 and S8,† and Table 2). In the **PHE** derivatives in DMSO with the N-up geometry, **PHE** and **PHE-2** exhibit (–)-sign MCPL, while **PHE-3** exhibits (+)-sign MCPL. In the **NP** derivatives in DMSO with the N-up geometry, **NP** exhibits (–)-sign MCPL, while **NP-1** and **NP-2** exhibit commonly (+)-sign MCPL. In the **ANT** derivatives in DMSO with the N-up geometry, **ANT**, **ANT-1**, and **ANT-2** exhibit the common (–)-sign MCPL. In addition to the dependence of MCPL sign inversion on substituents at peripheral positions, there is a trend in MCPL sign among **PY**, **PHE** and **NP**, which have successively fewer fused rings.

BP in DMSO excited at 350 nm and (h) **COR** in DMSO excited at 340 nm.



Table 1 MCPL characteristics (normalised at 1.6 T) of **NP**, **ANT**, **PHE**, **BP**, **COR**, and **PY** in common organic solvents (1.0×10^{-3} M) and PMMA film cast from a dilute CHCl_3 solution

| Entry | Medium | Φ_F | $g_{\text{MCPL}} (\lambda_{\text{MCPL}} \text{ in nm})/10^{-3} \text{ T}^{-1}$ | |
|--------------------------|------------------------------|----------|--|--|
| | | | N-up | S-up |
| NP ^{a,b} | CHX | 0.005 | +0.77 (317) | −0.56 (316) |
| | DMSO | 0.025 | −0.26 (327) | +0.29 (318) |
| ANT ^b | CHX | 0.196 | −0.23 (399) | +0.07 (399) |
| | DMSO | 0.198 | −0.22 (409) | +0.12 (405) |
| PHE ^a | CHX ^b | 0.033 | −1.08 (361) | +0.85 (361) |
| | DMSO ^a | 0.108 | −1.06 (353) | +0.97 (353) |
| | THF ^a | 0.048 | −1.00 (349) | +0.86 (349) |
| | CHCl_3 ^a | 0.032 | −1.06 (347) | +1.76 (349) |
| | PMMA ^{a,c} | 0.045 | −1.63 (345) | +1.27 (345) |
| BP | DMSO | 0.382 | −0.27 (415) ^a +0.86 (429) ^b | +0.17 (415) ^a −0.71 (428) ^b |
| | DMSO ^a | 0.115 | −0.58 (435) ^a +0.46 (451) ^b | +0.68 (435) ^a −0.38 (458) ^b |
| PY ^{a,d} | CHCl_3 | n.d. | −8.2 (374) | +8.2 (374) |
| | PMMA | n.d. | −7.0 (374) | +7.0 (374) |

^a At 0–0' band. ^b At 0–1' band. ^c Fig. S6, ESI. ^d Data taken from ref. 5a.

Table 2 MCPL characteristics (normalised at 1.6 T) of **NP-1**, **NP-2**, **ANT-1**, **ANT-2**, **PHE-2**, and **PHE-3** in DMSO (1.0×10^{-3} M)

| Entry | Medium | Φ_F | $g_{\text{MCPL}} (\lambda_{\text{MCPL}} \text{ in nm})/10^{-3} \text{ T}^{-1}$ | |
|--------------|-------------------|----------|--|--------------------------|
| | | | N-up | S-up |
| NP-1 | DMSO | 0.014 | (+) ^b | (−) ^b |
| NP-2 | DMSO | 0.209 | +0.12 (352) ^b | −0.20 (355) ^b |
| ANT-1 | DMSO | 0.690 | (−) ^b | (−) ^b |
| ANT-2 | DMSO | 0.609 | −0.14 (429) ^b | +0.13 (442) ^b |
| PHE-2 | DMSO | 0.312 | (−) ^b | (−) ^b |
| PHE-3 | DMSO ^a | 0.150 | +0.83 (374) | −0.88 (375) |

^a At 0–1' band. ^b Very weak signals although the MCPL sign is recognisable.

Regarding the $|g_{\text{MCPL}}|$ values, among **PHE-3**, **NP-2**, and **ANT-2**, the introduction of carboxylic acid groups slightly suppressed the $|g_{\text{MCPL}}|$ values by several times compared with the corresponding values for **PHE**, **NP**, and **ANT** (Fig. 3 and S8,† and Table 2). For the substituted derivatives, the MCPL signals were dramatically changed according to the substituent positions (Fig. 3 and Table 2). Thus, by introducing substituents at the proper peripheral positions, it was possible to suppress MCPL signals of the fused aromatics.

Conclusions

We showcased ambidextrous chirogenesis from representative planar fused aromatic luminophores (**NP**, **ANT**, **PHE**, **BP**, and **COR**) and their derivatives with substituents, with the N-up and S-up Faraday geometries in a commercial CPL/PL spectrometer. The MCPL sign depended on the nature of the fused aromatics. Furthermore, although **NP**, **ANT**, and **PHE** exhibited mono-

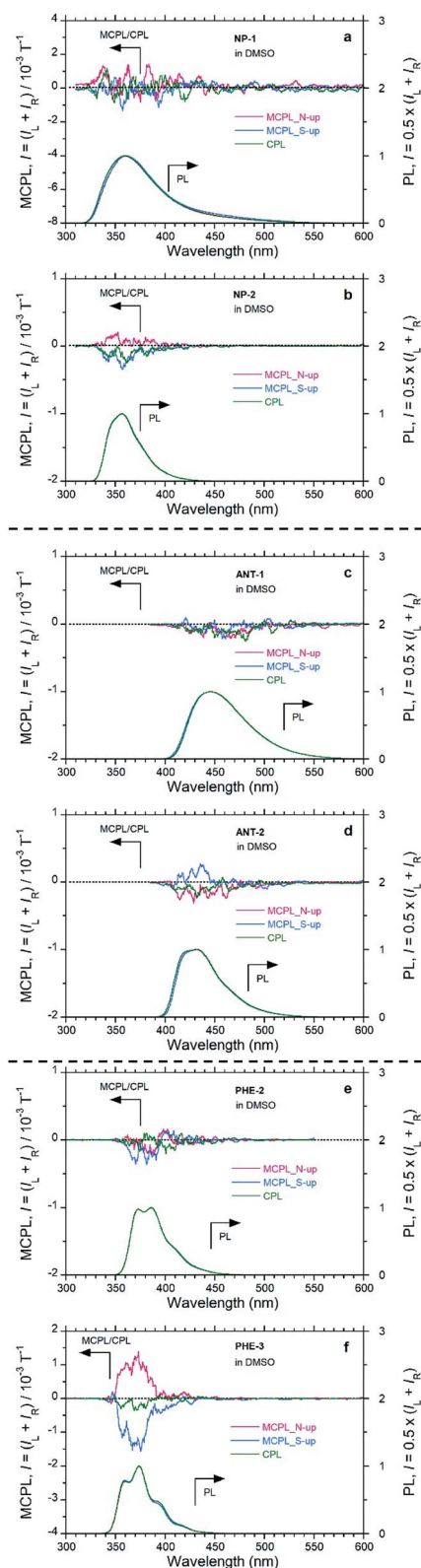


Fig. 3 MCPL with $H_0 = 1.6$ T (red and blue lines at N-up and S-up geometry, respectively), CPL (green line), and the corresponding PL spectra in dilute solutions (1.0×10^{-3} M; pathlengths: 5 mm for MCPL and 10 mm for CPL): (a) **NP-1** and (b) **NP-2** excited at 340 nm, (c) **ANT-1** and (d) **ANT-2** excited at 355 nm, and (e) **PHE-2** and (f) **PHE-3** excited at 270 nm.



signate MCPL profiles with (−) or (+) signs, **BP** and **COR** exhibited complex MCPL with multiple (+)/(−)-sign sequences. Knowledge of the relation between MCPL characteristics and the properties of fused aromatics with or without substituents should aid in the rational design of more efficient MCPL luminophores in the UV-Vis-NIR region in the future, as solution states, in the solids, and in the solid film. The external magnetic field-induced ambidextrous mirror-symmetrical MCPL from achiral planar aromatics without any stereogenic centres and/or bonds in the ES is an alternative, versatile chirogenesis approach, as is the conventional chirogenesis approach using CPL substances containing stereogenic centres and/or bonds.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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Notes and references

- (a) N. Chandrasekhar and R. Chandrasekar, *Chem. Commun.*, 2010, **46**, 2915; (b) D. Li, K. Wang, S. Huang, S. Qu, X. Liu, Q. Zhu, H. Zhang and Y. Wang, *J. Mater. Chem.*, 2011, **21**, 15298; (c) D. Dang, X. Wang, Z. Yang, D. Hao, Y. Xu, S. Zhang and L. Meng, *ACS Appl. Nano Mater.*, 2018, **1**, 2324; (d) A. Singh, A. K. Pati and A. K. Mishra, *Phys. Chem. Chem. Phys.*, 2018, **20**, 4167; (e) T. Yang, B. Liang, Z. Cheng, C. Li, G. Lu and Y. Wang, *J. Phys. Chem. C*, 2019, **123**, 18585; (f) F. Shahrokhi and Y. Zhao, *Org. Lett.*, 2019, **21**, 9306; (g) A. Singh, A. K. Pati and A. K. Mishra, *J. Phys. Chem. A*, 2019, **123**, 443; (h) M. Fujiki, L. Wang, N. Ogata, F. Asanoma, A. Okubo, S. Okazaki, H. Kamite and A. J. Jalilah, *Front. Chem.*, 2020, **8**, 685.
- (a) D. Y. Kim, *J. Korean Phys. Soc.*, 2006, **49**, 505; (b) R. Farshchi, M. Ramsteiner, J. Herfort, A. Taharaoui and H. T. Grahm, *Appl. Phys. Lett.*, 2011, **98**, 162508; (c) H. Maeda, Y. Bando, K. Shimomura, I. Yamada, M. Naito, K. Nobusawa, H. Tsumatori and T. Kawai, *J. Am. Chem. Soc.*, 2011, **133**, 9266; (d) R. Carr, N. H. Evan and D. Parker, *Chem. Soc. Rev.*, 2012, **41**, 7673; (e) Y. Yang, R. C. da Costa, M. J. Fuchter and A. J. Campbell, *Nat. Photonics*, 2013, **7**, 634; (f) S. Feuillastre, M. Pauton, L. Gao, A. Desmarchelier, A. J. Riives, D. Prim, D. Tondelier, B. Geffroy, G. Muller, G. Clavier and G. Pieters, *J. Am. Chem. Soc.*, 2016, **138**, 3990; (g) A. Homberg, E. Brun, F. Zinna, S. Pascal, M. Gorecki, L. Monnier, C. Besnard, G. Pescitelli, L. D. Bari and J. Lacour, *Chem. Sci.*, 2018, **9**, 7043.
- (a) R. A. van Delden, N. P. M. Huck, J. J. Piet, J. M. Warman, S. C. J. Merkers, H. P. J. M. Dekkers and B. L. Feringa, *J. Am. Chem. Soc.*, 2003, **125**, 15659; (b) A. Satrjio, S. C. J. Meskers and T. M. Swager, *J. Am. Chem. Soc.*, 2006, **128**, 9030; (c) T. Ikeda, T. Masuda, T. Hirao, J. Yuasa, H. Tsumatori, T. Kawai and T. Haino, *Chem. Commun.*, 2012, **48**, 6025; (d) T. Kimoto, T. Amako, N. Tajima, R. Kuroda, M. Fujiki and Y. Imai, *Asian J. Org. Chem.*, 2013, **2**, 404; (e) C. R. Wright, L. VandenElzen and T. Hopkins, *J. Phys. Chem. B*, 2018, **122**, 8730; (f) J. L. Ma, Q. Peng and C. H. Qian, *Chem.-Eur. J.*, 2019, **25**, 15441; (g) L. VandenElzen and T. A. Hopkins, *ACS Sustainable Chem. Eng.*, 2019, **7**, 16690; (h) K. Takaishi, K. Iwachido and T. Ema, *J. Am. Chem. Soc.*, 2020, **142**, 1774.
- (a) E. Krausz, G. Moran and H. Riesen, *Chem. Phys. Lett.*, 1990, **165**, 401; (b) R. Stranger, G. Moran, E. Krausz, H. Guedel and N. Furer, *Chem. Phys. Lett.*, 1990, **69**, 11; (c) M. Morita, T. Ozawa and T. Tsubokura, *J. Lumin.*, 1992, **53**, 495; (d) U. V. Valiev, J. B. Gruber, G. W. Burdick, A. K. Mukhammadiev, D. Fu and V. O. Pelenovich, *Opt. Mater.*, 2014, **36**, 1101; (e) U. V. Valiev, J. B. Gruber, G. W. Burdick, A. K. Mukhammadiev, D. Fu and V. O. Pelenovich, *J. Lumin.*, 2014, **145**, 393; (f) K. E. Knowles, H. D. Nelson, T. B. Kilburn and D. R. Gamelin, *J. Am. Chem. Soc.*, 2015, **137**, 13138; (g) T. Wu, J. Kapitan, V. Andrushchenko and P. Bour, *Anal. Chem.*, 2017, **89**, 5043; (h) H. D. Nelson, S. O. M. Hinterding, R. Fainblat, S. E. Creutz, X. Li and D. R. J. Gamelin, *J. Am. Chem. Soc.*, 2017, **139**, 6411; (i) E. L. Iechenko, *Phys. Solid State*, 2018, **60**, 1514.
- (a) D. Kaji, H. Okada, N. Hara, Y. Kondo, S. Suzuki, M. Miyasaka, M. Fujiki and Y. Imai, *Chem. Lett.*, 2020, **49**, 674; (b) H. Yoshikawa, G. Nakajima, Y. Mimura, T. Kimoto, S. Kondo, A. Suzuki, M. Fujiki and Y. Imai, *Dalton Trans.*, 2020, **49**, 9588; (c) H. Okada, N. Hara, D. Kaji, M. Shizuma, M. Fujiki and Y. Imai, *Phys. Chem. Chem. Phys.*, 2020, **22**, 13862.

