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Thymol derivatives with antibacterial and cytotoxic activity from the aerial parts of *Ageratina adenophora*†

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Three new thymol derivatives, 7-formyl-9-isobutyryloxy-8-hydroxythymol (**1**), 7,9-di-isobutyryloxy-8,10-dehydrothymol (**2**) and 2 α -methoxyl-3 β -methyl-6-methylol-2,3-dihydrobenzofuran (**3**), along with five known ones (**4–8**), were isolated from the aerial parts of the invasive plant *Ageratina adenophora*. Their structures were elucidated by extensive spectroscopic analysis and they were all isolated from the aerial part of *A. adenophora* for the first time. These compounds, except **8**, selectively showed *in vitro* antimicrobial activity against three Gram-(+) and two Gram-(–) bacterial strains. In particular, compounds **1** and **5** showed notable *in vitro* antimicrobial activity against all five bacterial strains with IC₅₀ values ranging from 3.9 to 15.6 $\mu\text{g mL}^{-1}$, as compared to reference compound kanamycin sulfate with a MIC value 1.9–3.9 $\mu\text{g mL}^{-1}$. Compounds **1** and **5** were further revealed to show *in vitro* cytotoxic activity against three tested human tumor (MCF-7, NCI-H460 and HeLa) cell lines, with IC₅₀ values ranging from 7.45 to 28.63 μM . Compounds **7** and **8** selectively showed slight but detectable *in vitro* cytotoxicity toward MCF-7 and NCI-H460 cell lines, with IC₅₀ values 44.65–83.19 μM . No cytotoxic effects were detected in the bioassay of the other four thymol derivatives. The present results provide new data to support that the aerial parts of *A. adenophora* are a rich source of bioactive chemicals valuable in medicinal applications.

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Introduction

Ageratina adenophora (Sprengel) King & Robinson (synonym: *Eupatorium adenophorum* Sprengel) is a perennial, herbaceous invasive plant, native to Mexico and Costa Rica.¹ As a well-known invasive species, this plant has successfully invaded more than thirty countries and regions in tropical and temperate zones of the world, including America, Australia, Europe, India, South Africa, Southeast Asia, and the southwest part of China.^{2,3} At its invasion places, the rapid spread of *A. adenophora* has caused serious economic losses to agriculture, forestry and livestock, and intensely damaged the local ecosystem and the original biodiversity.^{4,5}

In nature, *A. adenophora* is seldom attacked by microorganisms and insects. This suggests a rich defense related to bioactive chemicals, that possibly might also be pharmaceutically valuable, would exist in this plant. Previously, some literature reported that this plant was used in Nigeria and India as traditional or folk herb medicine for treatment of many human diseases, like fever, diabetes, inflammation, *etc.*^{6,7} To date, some terpenoids, flavonoids, phenylpropanoids, coumarins, sterols and alkaloids have been reported from this plant,^{8–10} with part of them exhibiting phytotoxic,¹¹ allelopathic,^{12,13} antifungal¹⁴ and antifeedant¹⁵ activities. In our recent study, some bioactive compounds, including phytotoxic phenolics,¹⁶ antifungal monoterpenes¹⁷ and antibacterial quinic acid derivatives,¹⁸ were also revealed from the roots or aerial parts of this plant. In continuation of the work to clarify those potentially new and bioactive chemicals in the aerial parts of *A. adenophora*, eight thymol derivatives including three new (**1–3**) and five known (**4–8**) ones are here further obtained (Fig. 1). We herein report the isolation and structural elucidation of these compounds, as well as describe their *in vitro* antimicrobial activity against five bacteria and cytotoxicity toward three human cancer cell lines.

Results and discussion

The petroleum ether- and EtOAc-soluble fractions of the ethanol extract of the aerial parts of *Ageratina adenophora* (Spreng.) were

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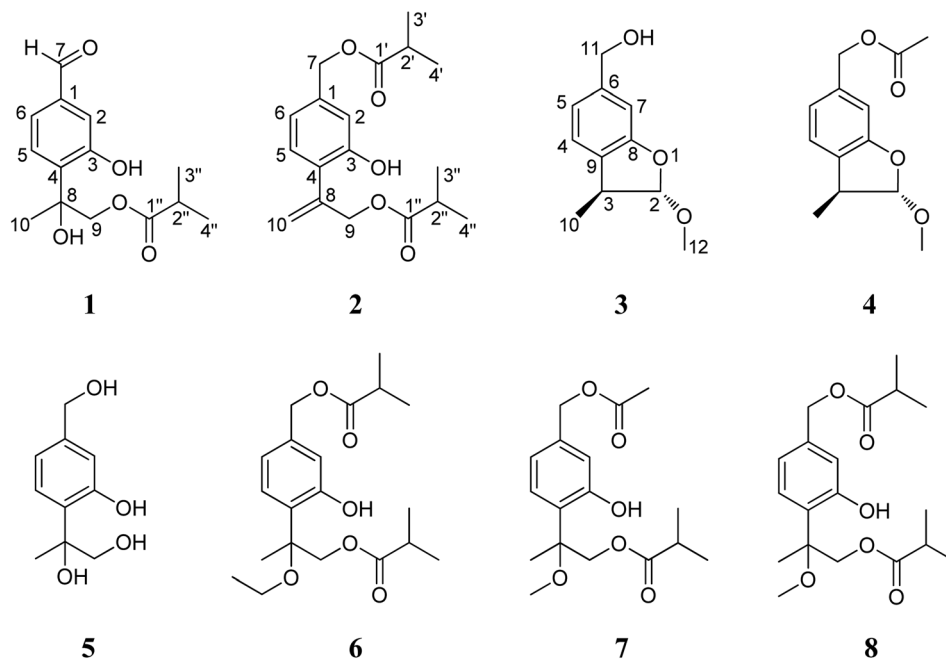


Fig. 1 Chemical structures of compounds 1–8.

isolated and purified by repeated column chromatography (CC) and HPLC to afford the three new (1–3) and five known (4–8) thymol derivatives. By comparing their NMR and MS data with those reported in literatures, the five known compounds were identified as 2 α -methoxyl-3 β -methyl-6-(acetyl-*O*-methyl)-2,3-dihydrobenzofuran (4),¹⁹ 7,8,9-trihydroxythymol (5),²⁰ 7,9-di-isobutyryloxy-8-ethoxythymol (6),²¹ 7-acetoxy-9-isobutyryloxy-8-methoxythymol (7)²¹ and 7,9-di-isobutyryloxy-8-methoxythymol (8).²¹

Compound 1 was isolated as a yellow oil and determined to have a molecular formula C₁₄H₁₈O₅ on the basis of HR-ESI-MS data (m/z 289.1056 [M + Na]⁺, calcd. for C₁₄H₁₈O₅Na, 289.1052). The IR absorptions at 3424 and 1695 cm⁻¹ revealed the presence of hydroxyl and carbonyl groups in the molecule. The existence of three methyl groups was revealed by ¹H NMR spectrum which provided methyl signals at δ_{H} 1.66 (3H, s), 1.12 (3H, s) and 1.13 (3H, s). The three aromatic proton signals at δ_{H} 7.34 (1H, d, $J = 1.2$ Hz), 7.18 (1H, d, $J = 8.4$ Hz), and 7.35 (1H, dd, $J = 8.4, 1.2$ Hz) were indicative of a typical pattern of 1,3,4-trisubstituted phenyl group.²² The ¹³C NMR (DEPT) spectra (Table 2), coupled with HSQC spectral analysis, revealed the presence of a formyl group at δ_{C} 191.9 (C-7), a carbonyl group at δ_{C} 177.7 (C-1'), an oxymethylene group at δ_{C} 70.4 (C-9), a methine group at δ_{C} 33.9 (C-2''), and an oxygenated quaternary carbon at δ_{C} 77.7 (C-8). Taken together these spectral data and the molecular formula into consideration, the existence of two hydroxyl groups could further be deduced. In the ¹H–¹H COSY spectrum, significant correlation signals for two proton systems were displayed, one at C-5 through C-6 and the other at C-3'' through C-4'' (Fig. 2), corresponding to structure fragments of –CH(5)–CH(6)– and CH₃(3'')–CH(2'')–CH₃(4''), respectively. In the HMBC spectrum, the exhibition of correlation signals of H-7 (δ_{H} 9.92) with C-2 and C-6, and of H-2, H-6 with C-7 (δ_{C} 191.9)

evidenced the connection of C-7 with C-1. The HMBC correlations of Me-10, H-9 with C-4 supported the connection of C-8 with C-4. The HMBC correlations of Me-3'', Me-4'' with C-1'' supported the connection of C-1'' with C-2''. The HMBC correlations of H-9 with C-1'' supported the ester bond linkage of C-1'' with C-9. Further consideration of the substitution pattern of the aromatic ring and the other NMR data led to the assignment of the two hydroxyl groups at C-3 and C-8, respectively. Based on these above spectroscopic analyses, we can unambiguously establish the planner structure of 1 as depicted in Fig. 1. However, it is presently difficult to determine the configuration of C-8. Thus, the structure of 1 was elucidated as 7-formyl-9-isobutyryloxy-8-hydroxythymol.

Compound 2 was assigned the molecular formula C₁₈H₂₄O₅, as deduced from ESI-MS and HR-ESI-MS data. Comparative analysis of the spectroscopic data and literature precedents^{21,23,24} supported 2 to be also a thymol derivative. Careful comparison of ¹H and ¹³C NMR data (Tables 1 and 2) of 2 with those of known compound 8 revealed the major differences of the two compounds that the resonances for Me-10, Me-11 and

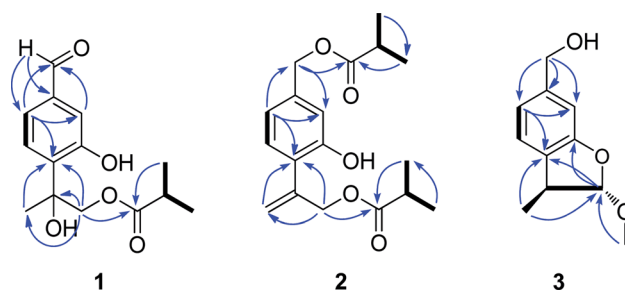


Fig. 2 Key HMBC (→) and ¹H–¹H COSY (⇌) correlations of 1, 2 and 3.



Table 1 The ^1H NMR spectral data (δ in ppm, J in Hz) for compounds 1, 2 and 3

Position	1 ^a	2 ^a	3 ^b
2	7.34 (1H, d, 1.2)	6.92 (1H, d, 1.2)	5.21 (1H, d, 2.0)
3	—	—	3.15 (1H, brq, 7.2)
4	—	—	7.12 (1H, d, 7.6)
5	7.18 (1H, d, 8.4)	7.07 (1H, d, 7.6)	6.87 (1H, brd, 7.6)
6	7.35 (1H, dd, 8.4, 1.2)	6.84 (1H, dd, 7.6, 1.2)	—
7	9.92 (1H, s)	5.06 (2H, s)	6.79 (1H, brs)
9	4.47 (1H, d, 12.0)	4.73 (2H, s)	—
10	4.33 (1H, d, 12.0)	5.47 (1H, d, 1.2)	1.23 (3H, d, 7.2)
	1.66 (3H, s)	5.28 (1H, d, 1.2)	—
11	—	—	4.53 (2H, s)
12	—	—	3.47 (3H, s)
2'	—	2.62 (1H, sept, 7.2)	—
3'	—	1.20 (3H, d, 7.2)	—
4'	—	1.20 (3H, d, 7.2)	—
2''	2.57 (1H, sept, 7.2)	2.62 (1H, sept, 7.2)	—
3''	1.12 (3H, d, 7.2)	1.20 (3H, d, 7.2)	—
4''	1.13 (3H, d, 7.2)	1.20 (3H, d, 7.2)	—

^a Recorded at 600 MHz in CDCl_3 . ^b Recorded at 400 MHz in CD_3OD .

the sp^3 quaternary carbon C-8 in compound 8 were replaced by signals [δ_{H} 5.28 (1H, $\text{H}_{\text{a}}-10$), 5.47 (1H, $\text{H}_{\text{b}}-10$); δ_{C} 141.6 (C-8), 116.8 (C-10)] for a terminal double bond in 2. In the HMBC spectrum, significant correlation signals of H_2-10 to C-4 and C-9 were observed, which confirmed the location of the terminal double bond between C-8 and C-10. The $^1\text{H}-^1\text{H}$ COSY spectrum and HMBC correlations (Fig. 2) of Me-3' and Me-4' with C-1', of Me-3'' and Me-4'' with C-1'', of H-7 with C-1', and of H-9 with C-1'', evidenced the ester bond linkages of C-7 and C-9 with an individual isobutyryl group, respectively. The presence of

diagnostic proton signals at δ_{H} 6.92 (1H, d, $J = 1.2$ Hz), 7.07 (1H, d, $J = 7.6$ Hz), and 6.84 (1H, dd, $J = 7.6, 1.2$ Hz) supported the appearance of the core 1,3,4-trisubstituted phenyl moiety. The HMBC correlations of H-2 and H-6 with C-7, of H-10 and H-9 with C-4, and of H-5 with C-8 confirmed the connections of C7 with C-1, C-4 with C-8, and the location of a hydroxyl group at C-3. Therefore, compound 2 was determined as 7,9-di-isobutyryloxy-8,10-dehydrothymol.

The molecular formula of compound 3 was determined as $\text{C}_{11}\text{H}_{14}\text{O}_3$ by the HR-EI-MS, due to a quasi-molecular ion peak at m/z 194.0938 (calcd. for $\text{C}_{11}\text{H}_{14}\text{O}_3$, 194.0937). The IR absorption at 3448 cm^{-1} revealed the presence of hydroxyl group in the molecule. The ^1H and ^{13}C (DEPT) NMR spectra displayed closely related signals with those of 2α -methoxyl- 3β -methyl-6-(acetyl-*O*-methyl)-2,3-dihydrobenzofuran,¹⁹ a known thymol compound which was also obtained as compound 4 in the present study. Careful comparison of their NMR spectral data indicated the major difference that the spectroscopic resonances for the acetoxy group connected to C-11 in 4 were absent in 3, suggesting that the acetoxy group located at C-11 in 4 was replaced by a free hydroxyl group in 3. Accordingly, we can preliminarily establish the structure of compound 3 as shown in Fig. 1, and this deduction was further well supported by $^1\text{H}-^1\text{H}$ COSY and HMBC analysis (Fig. 2). Furthermore, the presented coupling constant of H-2 ($J_{2,3} = 2.0$ Hz) and the NOE correlation between H-2 and H_3-10 in the NOESY spectrum supported the β -orientation of Me-10 and the α -orientation of the methoxy group at C-2.^{19,25} Therefore, compound 3 was assigned as 2α -methoxyl- 3β -methyl-6-methylol-2,3-dihydrobenzofuran.

Among these thymol derivatives, 1–3 are new compounds that are here reported from nature for the first time. Compound 3 is structurally characterized with a dihydrobenzofuran skeleton and this type of monoterpene is rare in natural products.²⁵ Compound 5 was previously only reported from plant *Eupatorium fortunei* and this is the first time for it being isolated from

Table 2 The ^{13}C (DEPT) NMR spectral data (δ in ppm) for compounds 1, 2 and 3

Position	1 ^a	2 ^a	3 ^b
1	137.4 (C)	138.2 (C)	—
2	118.9 (CH)	115.5 (CH)	115.5 (CH)
3	157.1 (C)	153.7 (C)	44.4 (CH)
4	132.5 (C)	125.1 (C)	124.7 (CH)
5	127.0 (CH)	129.6 (CH)	120.8 (CH)
6	120.5 (CH)	119.3 (CH)	143.3 (C)
7	191.9 (CH)	65.5 (CH_2)	109.3 (CH)
8	77.7 (C)	141.9 (C)	159.3 (C)
9	70.4 (CH_2)	65.5 (CH_2)	131.8 (C)
10	25.8 (CH_3)	116.8 (CH_2)	18.9 (CH_3)
11	—	—	65.2 (CH_2)
12	—	—	56.2 (CH_3)
1'	—	176.9 (C)	—
2'	—	34.0 (CH)	—
3'	—	18.8 (CH_3)	—
4'	—	18.8 (CH_3)	—
1''	177.7 (C)	177.8 (C)	—
2''	33.9 (CH)	34.0 (CH)	—
3''	18.8 (CH_3)	18.9 (CH_3)	—
4''	18.8 (CH_3)	18.8 (CH_3)	—

^a Recorded at 150 MHz in CDCl_3 . ^b Recorded at 100 MHz in CD_3OD .



Table 3 MIC values of compounds 1–8 in $\mu\text{g mL}^{-1}$ against five bacterial strains^a

Compounds	<i>Staphylococcus aureus</i>	<i>Bacillus cereus</i>	<i>Bacillus thuringiensis</i>	<i>Escherichia coli</i>	<i>Salmonella enterica</i>
1	7.8	3.9	7.8	15.6	15.6
2	>100	62.5	>100	>100	>100
3	31.3	31.3	62.5	>100	>100
4	31.3	62.5	62.5	>100	>100
5	7.8	7.8	15.6	15.6	15.6
6	62.5	62.5	62.5	>100	>100
7	>100	62.5	62.5	>100	>100
8	>100	>100	>100	>100	>100
KS	1.9	3.9	3.9	3.9	3.9

^a KS = kanamycin sulfate.

A. adenophora.²⁰ The other four known compounds 4, 6, 7 and 8 are here also isolated and identified from the aerial parts of *A. adenophora* for the first time.

Aimed to explore their potential and undiscovered pharmacological activity of these isolated thymol derivatives, compounds 1–8 were evaluated for their *in vitro* antimicrobial activity by testing their MIC values, using a bioassay method as previously we used and described.^{26,27} Table 3 lists the results obtained for these compounds on the viability of five tested bacterial strains, compared to kanamycin sulfate (KS) as a reference compound. Among them, compounds 1 and 5 were found to be strongly active against all the five assayed microorganisms with MIC values ranging from 3.9 to 15.6 $\mu\text{g mL}^{-1}$, which were comparable to the positive control KS (MICs = 1.9 to 3.9 $\mu\text{g mL}^{-1}$). Compounds 2, 3, 4, 6 and 7 only selectively showed antibacterial activity (MIC 31.3–62.5 $\mu\text{g mL}^{-1}$) against three tested Gram-(+) bacteria, *i.e.* *S. aureus*, *B. cereus* and *B. thuringiensis*. While no antibacterial activity was detected for compound 8 toward all the five tested bacterial strains. From the result, it seems to show that the existence of both the hydroxyl group at C-3 and C-8 would be important for this group of thymol compounds to display their antibacterial potentials.

Compounds 1–8 were further tested for their *in vitro* cytotoxicity against human cancer MCF-7, HeLa and NCI-H460 cell lines, using a microdilution titre technique as recently we described.²⁸ The resulting IC₅₀ values are displayed in Table 4, compared to Adriamycin as positive control. Compounds 1 and 5 were found to show strong or moderate cytotoxicity against all the three tested cancer cell lines, with IC₅₀ values ranging from 7.45 to 28.63 μM . Compounds 7 and 8 showed slight but detectable cytotoxicity toward MCF-7 and NCI-H460 cell lines, with IC₅₀ values ranging from 44.65 to 83.19 μM . While, no obvious cytotoxic activity was detected for the other compounds in this bioassay. Comparison of the chemical structures and the cytotoxic activity of these compounds indicated that the existence of both the hydroxyl group at C-3 and C-8 would be important for this group of thymol derivatives to fully display their cytotoxic potentials.

As a well-known invasive plant, *A. adenophora* has attracted much attention of scientists to investigate its invasion

mechanisms. Since that this plant is accumulating a huge biomass at its invasion areas, to explore the potential utilization of *A. Adenophora* is gradually also concentrated and emphasized by some researchers. Up to date, phytochemical studies have indicated that structurally diverse chemicals exist in this invasive species. Our present findings further support that the aerial part tissue of this plant is rich in bioactive natural products valuable to be explored for medicinal usage. It is interesting to note that compound 5 was reported as a natural product capable of strongly inhibiting the growth of *Microcystis aeruginosa*,²⁰ suggesting that these thymol derivatives, at least for compound 5, might have some allelopathic potential to contribute the invasion success of *A. adenophora*. Noteworthy, for all these thymol derivatives identified from *A. adenophora*, the carbon C-7 (or C-11 in compounds 3 and 4) is generally appeared as an oxygenated carbon. While for those thymol compounds reported from some other *Eupatorium* plants, such as thymol compounds from *E. fortunei* and *E. cannabinum*,^{22–24} usually exhibited at C-7 is a methyl group (an unoxygenated carbon). Thus, the general oxygenation extent of thymol compounds at C-7 might have some chemotaxonomic significance for plants in *Adenophora* and (or) *Eupatorium* genus. Furthermore, it is evident that compounds 3 and 4 contain a typical dihydrobenzofuran skeleton. Dihydrobenzofuran type monoterpene is rare in nature and the discovery of dihydrobenzofuran type new compound 3 suggests that some so far undiscovered rare monoterpenes would still exist in the aerial parts of *A. adenophora* worthy of further investigation.

Table 4 Cytotoxic activity of compounds 1–8 (IC₅₀, μM)^a

Compounds	MCF-7	NCI-H460	HeLa
1	7.45 ± 0.22	8.32 ± 0.21	9.45 ± 0.46
2–4	>100	>100	>100
5	11.54 ± 0.86	15.67 ± 1.03	28.63 ± 1.93
6	>100	>100	>100
7	44.65 ± 4.08	52.74 ± 5.16	>100
8	83.19 ± 6.55	77.42 ± 5.06	>100
Adriamycin	0.78 ± 0.06	1.12 ± 0.05	0.54 ± 0.04

^a Values represent mean ± SD ($n = 3$) based on three individual experiments.



Materials and methods

General experimental procedures

UV spectra were recorded in MeOH on a PerkinElmer Lambda 35 UV-vis spectrophotometer. IR spectra (KBr) were recorded on a Bruker Tensor 27 spectrophotometer in cm^{-1} . ^1H (600 MHz and 400 MHz), ^{13}C (150 MHz and 100 MHz), and 2D NMR spectra were recorded in CDCl_3 and CD_3OD on a Bruker DRX-400 instrument and a Bruker AVANCE 600 instrument with TMS as an internal standard. HR-ESI-MS data were obtained on a Water Q-TOF Premier mass spectrometer and HR-EI-MS data were obtained on a Finigan MAT 95XP mass spectrometer. ESIMS were collected on an MDS SCIEX API 2000 LC/MS/MS instrument. Preparative HPLC was conducted using a P3000 HPLC pump and a UV3000 UV-VIS Detector with a Fuji-C18 column (10 μm –100A). For column chromatography (CC), silica gel (200–300 mesh, Qingdao Marine Chemical Inc., Peoples Republic of China), YMC ODS-A (50 μm , YMC Co. Ltd., Japan) were used, and Sephadex LH-20 (Pharmacia Fine Chemical Co. Ltd.) were used. Fractions were monitored by TLC, and spots were visualized by heating the silica gel plates sprayed with 10% H_2SO_4 in ethanol.

Plant materials

The aerial part material of *Ageratina adenophora* (Spreng.) were collected in a suburb of Kunming, Yunnan province, P. R. China, in July 2009, and authenticated by Prof. Fu-Wu Xing, South China Botanical Garden, Chinese Academy of Sciences. A voucher specimen (no. 20090702) was deposited at the Laboratory of Phytochemistry at the South China Botanical Garden, Chinese Academy of Sciences.

Extraction and isolation

The air-dried and powdered aerial part material of *A. adenophora* (10 kg) were extracted with 95% aqueous ethanol at room temperature for three times (3 \times 20 L, each for 24 h). The ethanol extracts were next combined and concentrated *in vacuo*, with the resulting residue suspended in water and sequentially extracted with petroleum ether (3 \times 3 L) and EtOAc (3 \times 3 L). The petroleum ether and EtOAc layers were evaporated *in vacuo* to yield petroleum-soluble fraction (93.1 g) and EtOAc-soluble fraction (80.0 g), respectively.

The petroleum ether fraction (93.1 g) was subjected to silica gel CC (1000 mm \times 120 mm i.d.), eluting with a gradient of CHCl_3 –MeOH (100 : 0 to 90 : 10, v/v) to give twelve fractions (P_1 – P_{12}) after pooled according to their TLC profiles. Fraction P_6 (20.6 g), obtained by elution with CHCl_3 –MeOH (98 : 2, v/v), was applied to silica gel CC (800 \times 75 mm i.d.) eluted with a gradient of petroleum ether–acetone (100 : 0 to 80 : 20, v/v) to obtain nine fractions (P_{6-1} – P_{6-9}). Fraction P_{6-4} (7.4 g) was subjected to silica gel CC (800 \times 50 mm i.d.) eluted with petroleum–acetone (300 : 1 to 100 : 5) in a gradient to obtain sub-fractions P_{6-4-4} – P_{6-4-9} , of which sub-fraction P_{6-4-4} (600.0 mg) was separated by preparative HPLC (flow rate 8 mL min^{-1}) using MeOH– H_2O (70 : 30, v/v) to afford compound 6 (t_{R} = 37.5 min, 2.4 mg). Fraction P_{6-7} (6.2 g) was applied to an ODS CC using

MeOH– H_2O (30 : 70 to 90 : 10) to obtain sub-fractions P_{6-7-1} – P_{6-7-12} . Subsequently sub-fraction P_{6-7-8} (19.2 mg) was purified by Sephadex LH-20 CC (1500 mm \times 25 mm i.d.) eluted with CHCl_3 –MeOH (20 : 80, v/v) to afford compounds 2 (2.1 mg) and 8 (2.5 mg), and sub-fraction P_{6-7-3} (930.0 mg) was purified by a Sephadex LH-20 CC using acetone to afford compound 7 (25.1 mg). Fraction P_{6-8} (340.0 mg) was subjected to Sephadex LH-20 CC and preparative HPLC (flow rate 8 mL min^{-1}) using MeOH– H_2O (45 : 55 to 50 : 50, v/v) as mobile phase to afford compound 1 (t_{R} = 48.0 min, 6.7 mg). Fraction P_{6-9} (600.0 mg) was subjected to silica gel CC eluted with petroleum–acetone (100 : 1 to 90 : 10, v/v) in gradient to obtain sub-fractions P_{6-8-1} and P_{6-8-2} , and sub-fraction P_{6-8-2} (45.5 mg) was further purified by an ODS CC using MeOH– H_2O (65 : 35, v/v) as eluent to afford compound 4 (3.0 mg).

The EtOAc-soluble fraction (80.0 g) was subjected to silica gel CC (1000 mm \times 120 mm i.d.) using a gradient of CHCl_3 –MeOH (95 : 5 to 60 : 40, v/v) to give ten fractions (E_1 – E_{10}). Fraction E_6 (12.0 g), obtained by elution with CHCl_3 –MeOH (85 : 15, v/v), was subjected to silica gel CC eluted with CHCl_3 –MeOH (40 : 1 to 9 : 1, v/v) in gradient to obtain sub-fractions E_{6-1} – E_{6-8} . Subfraction E_{6-2} (213.1 mg) was further applied to ODS CC eluted with MeOH– H_2O (20 : 80 to 90 : 10, v/v) to afford compound 3 (7.0 mg). Fraction E_{6-5} (2.7 g) was subjected to ODS CC using MeOH– H_2O (10 : 90 to 70 : 30, v/v) to afford five sub-fractions (E_{6-5-1} – E_{6-5-5}), of which sub-fraction E_{6-5-1} (312.0 mg) was further purified by Sephadex LH-20 CC using MeOH to afford compound 5 (30.0 mg).

7-Formyl-9-isobutyryloxy-8-hydroxythymol (1). Yellowness oil; $[\alpha]_{\text{D}}^{20}$ –2.0 (*c* 0.65, CHCl_3); IR (KBr) ν_{max} 3424, 1695, 1612, 1579 cm^{-1} ; UV (MeOH) λ_{max} (log ϵ) nm: 222 (4.11), 260 (3.88), 317 (3.36); ^1H (CDCl_3 , 600 MHz) and ^{13}C NMR (CDCl_3 , 150 MHz) data: see Tables 1 and 3; ESI-MS m/z 571 $[\text{2M} + \text{K}]^+$, 555 $[\text{2M} + \text{Na}]^+$, 567 $[\text{2M} + \text{Cl}]^-$, 289 $[\text{M} + \text{Na}]^+$, 301 $[\text{M} + \text{Cl}]^-$, 265 $[\text{M} - \text{H}]^-$; HR-ESI-MS m/z 289.1056 $[\text{M} + \text{Na}]^+$ (calcd. for $\text{C}_{14}\text{H}_{18}\text{O}_5\text{Na}$, 289.1052).

7,9-Di-isobutyryloxy-8,10-dehydrothymol (2). Yellowness oil; $[\alpha]_{\text{D}}^{20}$ 0.0 (*c* 0.20, CH_3OH); IR (KBr) ν_{max} 3438, 1733, 1677 cm^{-1} ; UV (MeOH) λ_{max} (log ϵ) nm: 205 (4.34), 285 (3.42); ^1H (CDCl_3 , 600 MHz) and ^{13}C NMR (CDCl_3 , 100 MHz) data: see Tables 1 and 3; ESI-MS m/z 343 $[\text{M} + \text{Na}]^+$, 359 $[\text{M} + \text{K}]^+$; HR-ESI-MS m/z 343.1526 $[\text{M} + \text{Na}]^+$ (calcd for $\text{C}_{18}\text{H}_{24}\text{O}_5\text{Na}$, 343.1521).

2 α -Methoxyl-3 β -methyl-6-methylol-2,3-dihydrobenzofuran (3). Yellowness oil; $[\alpha]_{\text{D}}^{20}$ 2.8 (*c* 0.16, CH_3OH); IR (KBr) ν_{max} 3448, 1596, 1496, 1455 cm^{-1} ; UV (MeOH) λ_{max} (log ϵ) nm: 204 (4.09), 279 (3.32); ^1H (CD_3OD , 400 MHz) and ^{13}C NMR (CD_3OD , 100 MHz) data: see Tables 1 and 3; ESI-MS m/z 195 $[\text{M} + \text{H}]^+$, 217 $[\text{M} + \text{Na}]^+$, 289 $[\text{M} + \text{Na}]^+$, 423 $[\text{2M} + \text{Cl}]^-$; HR-EI-MS m/z 194.0938 (calcd. for $\text{C}_{11}\text{H}_{14}\text{O}_3$, 194.0937).

Antimicrobial activity

Antimicrobial activity of these eight thymol compounds were evaluated, using 96-well plates, by a method based on a micro-dilution titre technique with modification in determination of the minimum inhibitory concentration (MIC) values.²⁶ Before the assay, proper amounts of the tested compounds and



kanamycin sulfate (positive control) were dissolved in methanol to prepare their 1.0 mg mL^{-1} sample solutions and the positive control solution, respectively, and resazurin (indicator) was dissolved in distilled water to prepare $100 \text{ }\mu\text{g mL}^{-1}$ indicator solution for the assay. In the test, $100 \text{ }\mu\text{L}$ indicator solution (resazurin, $100 \text{ }\mu\text{g mL}^{-1}$) was first placed into each of the sterility control wells (11th column) on the 96 well plates, and about 7.5 mL indicator solution was mixed with 5 mL test organism (10^6 cfu mL^{-1}) followed by transferring ($100 \text{ }\mu\text{L}$, each) to growth control wells (12th column) and all test wells ($1\text{--}10^{\text{th}}$ column). Then, each of $100 \text{ }\mu\text{L}$ of the sample solutions (1.0 mg mL^{-1} of tested compounds in methanol) and the positive control solution (1.0 mg mL^{-1} of kanamycin sulfate in methanol) as well as the negative control sample (pure MeOH) were allied to the wells in the 1st column of the plates. In each plate, up to six samples along with a positive control and a negative control samples were applied. Once all samples and controls were properly applied to the 1st column of wells on the plate, half of the homogenized content ($100 \text{ }\mu\text{L}$) from these wells was then parallel transferred to the 2nd column of wells, and each subsequent well was treated similarly (doubling dilution) up to the 10th column, followed by discarding the last $100 \text{ }\mu\text{L}$ aliquot. Finally, the plates were incubated at $37 \text{ }^\circ\text{C}$ for 5–6 h until the color of growth control change to pink. The lowest concentration for each tested compound at which color change occurred was recorded as its primary MIC value. The average of primary values from three individual tests were calculated and that was taken as the final MIC value for each of the test compounds.²⁷ A total of five microorganisms including three Gram-(+) bacteria (*Staphylococcus aureus*, *Bacillus cereus* and *Bacillus thuringiensis*) and two Gram-(−) bacterial species (*Escherichia coli* and *Salmonella enterica*) were used in the bioassay. The resulting MIC values of the tested compounds were listed in Table 3.

Cytotoxic assay

The cytotoxic activity of compounds 1–8 against three human tumor (MCF-7, NCI-H460 and HeLa) cell lines were assayed by using 96 well plates according to a MTT method as we used previously.²⁸ In brief, the cells were cultured in RPMI-1640 medium supplemented with 10% fetal bovine serum in a humidified atmosphere with 5% CO_2 at $37 \text{ }^\circ\text{C}$. Each well of 96-well cell culture plates was seeded with $100 \text{ }\mu\text{L}$ adherent cells (5×10^4 cell per mL) and placed in an atmosphere with 5% CO_2 at $37 \text{ }^\circ\text{C}$ for 24 h to form a monolayer on the flat bottoms. Subsequently, in each well, the supernatant was removed and $100 \text{ }\mu\text{L}$ fresh medium and $100 \text{ }\mu\text{L}$ medium containing one of the test compounds was added. Then the plate was incubated in 5% CO_2 atmosphere at $37 \text{ }^\circ\text{C}$. After 3 days, $20 \text{ }\mu\text{L}$ MTT at concentration 5 mg mL^{-1} in DMSO was added into each well and incubated for 4 h. Carefully, the supernatant in each well was removed and $150 \text{ }\mu\text{L}$ DMSO was added. Then the plate was vortex shaken for 15 min to dissolve blue formazan crystals. The OD (optical density) value of each well was tested on a Genios microplate reader (Tecan GENios, Männedorf, Switzerland) at 570 nm. All the tests were conducted by three individual experiments and adriamycin was applied as a positive control.

In a test, for each of the tumor cell lines, each of the test compounds was set at concentrations 50, 25, 12.5, 6.25, 3.125, $1.5625 \text{ }\mu\text{g mL}^{-1}$. The inhibitory rate of tumor cell growth was calculated by the formula: inhibition rate (%) = $(\text{OD}_{\text{control}} - \text{OD}_{\text{treated}})/\text{OD}_{\text{control}} \times 100\%$, and the IC_{50} values were calculated by SPSS 16.0 statistic software. The three tumor cell lines were purchased from the Kunming Institute of Zoology, CAS. The resulting IC_{50} values listed in Table 4 were based on three individual experiments and represented as means \pm standard deviation (SD).

Conclusions

Eight thymol derivatives, including three new (1–3) and five known ones (4–8), were obtained from the aerial parts of the invasive plant *Ageratina adenophora*. Their structures were elucidated by extensive spectroscopic analysis and they were all isolated from the aerial parts of *A. Adenophora* for the first time. Compounds 1 and 5 were found to show obviously *in vitro* antimicrobial activity against all the five tested bacterial strains with IC_{50} values ranging from 7.8 to $15.6 \text{ }\mu\text{g mL}^{-1}$. Other thymol compounds, except 9, only selectively showed detectable *in vitro* antimicrobial activity toward three tested Gram-(+) bacteria. Compounds 1 and 5 were further revealed to show *in vitro* cytotoxic activity against human tumor MCF-7, NCI-H460 and HeLa cell lines, with IC_{50} values ranging from 7.45 to $28.63 \text{ }\mu\text{M}$. Compounds 7 and 8 selectively showed slight *in vitro* cytotoxicity toward MCF-7 and NCI-H460 cell lines. These data supported that the aerial parts of *A. Adenophora* are rich in potential bioactive natural products valuable to be developed in medicinal field.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 S. Qiang, *J. Wuhan Bot. Res.*, 1998, **16**, 366–372.
- 2 F.-H. Wan, W.-X. Liu, J.-Y. Guo, S. Qiang, B.-P. Li, J.-J. Wang, G.-Q. Yang, H.-B. Niu, F.-R. Gui and W.-K. Huang, *Sci. China: Life Sci.*, 2010, **53**, 1291–1298.
- 3 X.-Y. Sun, Z.-H. Lu and W.-G. Sang, *J. For. Res.*, 2004, **15**, 319–322.
- 4 F.-H. Wan, J.-Y. Guo and D.-H. Wang, *Biodiversity Sci.*, 2002, **10**, 119–125.
- 5 R. Wang and Y.-Z. Wang, *Divers. Distrib.*, 2006, **12**, 397–408.
- 6 F. M. Awah, P. N. Uzoegwu, P. Ifeonu, J. O. Oyugi, J. Rutherford, X. Yao, F. Fehrmann, K. R. Fowke and M. O. Eze, *Food Chem.*, 2012, **131**, 1279–1286.



- 7 V. Ahluwalia, R. Sisodia, S. Walia, O. P. Sati, J. Kumar and A. Kundu, *J. Pestic. Sci.*, 2014, **87**, 341–349.
- 8 Q.-S. Yan, J. Yang, H.-M. Li, A.-C. Cao, Q.-H. Chen, Y.-Q. Wen and L. He, *J. Beijing Norm. Univ.*, 2006, **42**, 70–73.
- 9 Y.-M. Li, Z.-Y. Li and M. Ye, *J. Yunnan Agric. Univ.*, 2008, **23**, 42–46.
- 10 L. He, J. Hou, M. L. Gan, J. G. Shi, S. Chantrapromma, H. K. Fun, I. D. Williams and H. H. Y. Sung, *J. Nat. Prod.*, 2008, **71**, 1485–1488.
- 11 G. Zheng, Y. Jia, X. Zhao, F. Zhang, S. Luo, S. Li and W. Li, *Chemoecology*, 2012, **22**, 131–138.
- 12 G.-Q. Yang, F.-H. Wan, W.-X. Liu and X.-W. Zhang, *Allelopathy J.*, 2006, **18**, 237–245.
- 13 X. Zhao, G.-W. Zheng, X.-M. Niu, W.-Q. Li, F.-S. Wang and S.-H. Li, *J. Agric. Food Chem.*, 2009, **57**, 478–482.
- 14 X. Liu, C. Ouyang, Y. Li, D. Yang, W. Fang, D. Yan, M. Guo, A. Cao and Q. Wang, *J. Plant Dis. Prot.*, 2016, **123**, 163–170.
- 15 W. Shi, S. Luo and S. Li, *Chin. J. Chem.*, 2012, **30**, 1331–1334.
- 16 Z.-Y. Zhou, W.-X. Liu, G. Pei, H. Ren, J. Wang, Q.-L. Xu, H.-H. Xie, F.-H. Wan and J.-W. Tan, *J. Agric. Food Chem.*, 2013, **61**, 111792–111799.
- 17 Q.-L. Xu, M. Zhang, Z.-Y. Zhou, W.-X. Liu, F.-H. Wan, H.-F. Wang and J.-W. Tan, *Phytochem. Lett.*, 2014, **9**, 123–126.
- 18 M. Zhang, W.-X. Liu, M.-F. Zheng, Q.-L. Xu, F.-H. Wan, J. Wang, T. Lei, Z.-Y. Zhou and J.-W. Tan, *Molecules*, 2013, **18**, 14096–14104.
- 19 B. Luo, L.-M. Dong, Q.-L. Xu, X. Zhang, Q. Zhang, W.-B. Liu and J.-W. Tan, *Phytochem. Lett.*, 2018, **24**, 67–70.
- 20 T. N. Pham, H. D. Pham, D. K. Dang, T. T. Duong, T. P. Q. Le, Q. D. Nguyen and D. N. Tien, *Nat. Prod. Res.*, 2019, **33**, 1345–1348.
- 21 L.-M. Dong, M. Zhang, Q.-L. Xu, Q. Zhang, B. Luo, Q.-W. Luo, W.-B. Liu and J.-W. Tan, *Molecules*, 2017, **22**, 592.
- 22 Y. Wang, J. Li, H. Wang, D.-Q. Jin, H. Chen, J. Xu, Y. Ohizumi and Y. Guo, *Phytochem. Lett.*, 2014, **7**, 190–193.
- 23 M. Tori, Y. Ohara, K. Nakashima and M. Sono, *J. Nat. Prod.*, 2001, **64**, 1048–1051.
- 24 J.-J. Chen, Y.-C. Tsai, T.-L. Hwang and T.-C. Wang, *J. Nat. Prod.*, 2011, **74**, 1021–1027.
- 25 H. X. Jiang, Q. Liu and K. Gao, *Nat. Prod. Res.*, 2008, **22**, 937–941.
- 26 J. Wang, H. Ren, Q.-L. Xu, Z.-Y. Zhou, P. Wu, X.-Y. Wei, Y. Cao, X.-X. Chen and J.-W. Tan, *Food Chem.*, 2015, **168**, 623–629.
- 27 M. M. Rahman and A. I. Gray, *Phytochemistry*, 2005, **66**, 1601–1606.
- 28 J.-K. Ouyang, L.-M. Dong, Q.-L. Xu, J. Wang, S.-B. Liu, T. Qian, Y.-F. Yuan and J.-W. Tan, *RSC Adv.*, 2018, **8**, 40483–40489.

