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# Inverted perovskite solar cells based on potassium salt-modified NiO<sub>x</sub> hole transport layers†

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Perovskite solar cells (PSCs) have been attracting increasing attention in recent years because of their exceptional high efficiency with incredible developments. However, the poor contact between the hole transport layer (HTL) and perovskite layer still limits the further development of power conversion efficiency (PCE) for inverted solar cells, particularly for inorganic HTL based devices. Herein, potassium salt (KI, KSCN and KNO<sub>3</sub>) modified nickel oxides are selected as HTLs to improve the photovoltaic performance of an inverted device. We found that all potassium salts play positive roles in optimizing the photovoltaic parameters. Through the modification of potassium salts, higher charge recombination impedance and strong photoluminescence quenching were achieved, which means retarded carrier nonradiative recombination and quick charge transfer at the interface. Besides, KI modified NiO<sub>x</sub> could improve the perovskite film coverage and reduce trap densities. Thus, the champion device based on a KI modified NiO<sub>x</sub> film attained a PCE of 20.10% with an enhanced fill factor of 0.812. The findings demonstrate that potassium doping is an effective route to improve the performance of inverted planar PSCs.

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## Introduction

Organometal halide perovskites have attracted plenty of attention in solar cells, light-emitting diodes (LEDs), lasers, detectors and other fields due to their excellent performance.<sup>1–6</sup> As a light absorption material, the application of perovskites has developed rapidly in the field of photovoltaics. In particular, the power conversion efficiencies (PCEs) of perovskite solar cells (PSCs) have been increased from 3.8%<sup>7</sup> in 2009 to the current world record of 25.5%.<sup>8</sup> Perovskite materials represent both n-type and p-type characteristics, which provides a basis for the diversity of solar cell structure.<sup>9</sup> Generally, the structure of PSCs is divided into two categories. In conventional devices, mesoporous or compact n-type oxide semiconducting materials work as an electron transport layer (ETL) deposited on conducting substrates under the light absorber layer, and a hole transport layer (HTL) acts as the p-type contact to collect holes on the top. Alternatively, the n- and p-type transport layers are fabricated in the reverse sequence for the inverted structure.<sup>10</sup> Although the conventional PSCs demonstrate higher PCEs than the inverted PSCs, conventional

efficient solar cells usually require a mesoporous titanium oxide ETL as a support layer for electron transport, which causes complicated assembly processes and UV-induced degradation of perovskite.<sup>11–13</sup> Small molecules,<sup>14–16</sup> such as 2,2',7,7'-tetrakis[*N,N*-di(4-methoxyphenyl)amino]-9,9'-spirobifluorene (spiro-OMeTAD), have been used as a benchmark HTL for conventional PSCs, in which Li and Co salts also hamper the device stability.<sup>17–19</sup> Moreover, its cost is about 5 times more than that of gold, which would significantly limit the commercialization of PSCs. Therefore, it is critical to search for other low-cost and efficient hole transport materials to replace expensive spiro-OMeTAD. Herein, we took inverted PSCs as the main research object and explored the influence of different HTLs.

Poly(3,4-ethylenedioxythiophene)-poly(styrenesulfonate) (PEDOT:PSS) is a widespread hole transport material in inverted devices,<sup>20</sup> but its hydroscopicity and high acidity have a serious negative impact on device stability. In this regard, different inorganic hole transport materials are considered to replace PEDOT:PSS.<sup>21–23</sup> Nickel oxide is one of the most popular inorganic HTL materials due to its prominent properties such as high optical transmittance, appropriate work function and lower valence band position.<sup>24</sup> Unfortunately, the PCEs of the inverted PSCs based on a nickel oxide HTL have been greatly improved, but are still lower than those of conventional structure devices.<sup>25,26</sup> This is mainly induced by the inferior crystallinity of perovskite and the poor contact between the HTLs and perovskite layers. Interface modification is a simple

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and efficient way to facilitate the contact and charge transfer, so as to obtain better device performance.<sup>27</sup> Yang *et al.*<sup>28</sup> modified a nickel oxide layer by a sol-gel method with diethanolamine (DEA) to promote the interface contact and perovskite crystallinity. As a result, the PCE was significantly lifted from 11.2 to 15.9%. In addition to the small molecular modification, a buffer layer is also constructed to reduce carrier recombination at the interface.<sup>29,30</sup> For instance, introducing an aluminum oxide buffer layer, the device showed less interface carrier recombination and surface defects of the perovskite layer, thus enhancing the device performance.<sup>10</sup> In addition, treatment with metal ions in the perovskite layer is competent for passivating perovskite crystal defects and improving device performance.<sup>31–33</sup> In particular, on adding potassium ions to a perovskite layer, both the charge transport and hysteresis are optimized.<sup>34</sup> However, few reports have researched these metal salts directly at the interface,<sup>35,36</sup> where the majority of defects exist.

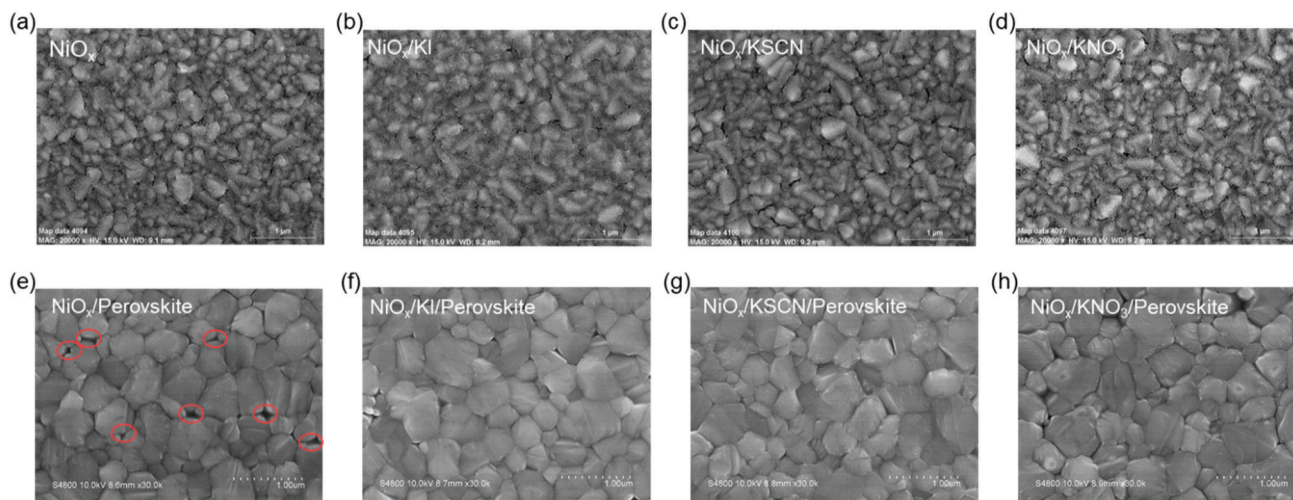
In this work, we developed inverted structure PSCs with inorganic potassium salt modified NiO<sub>x</sub> as the HTL. The devices after interface modification achieved optimized photovoltaic parameters. The inverted PSCs using KI treatment exhibited a champion PCE of 20.10% with a high fill factor (FF) of 0.812. This finding provides an effective strategy to fabricate high efficiency inverted PSCs.

## Results and discussion

To understand the potential influence on morphology of introducing potassium salts (KI, KSCN and KNO<sub>3</sub>), we performed scanning electron microscopy (SEM) and measurements of NiO<sub>x</sub> thin films with or without modification. As shown in Fig. 1a–d, all NiO<sub>x</sub> films are uniform and have no obvious change after modification of all potassium salts. Then CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> perovskite films were formed by a two-step continuous spin-coating and annealing procedure.<sup>37</sup> Compared with a pristine sample,

Fig. 1e–h display that the perovskite films based on modified NiO<sub>x</sub> films have fewer holes and better surface coverage. Comparable research on HTLs has also suggested that small amounts of metal salt could smooth the NiO<sub>x</sub> films, promoting the formation of highly crystalline, uniform perovskite layers,<sup>38,39</sup> which is consistent with the observations in the atomic force microscopy (AFM) images (Fig. S1, ESI†). This is certainly devoted to optimizing the photoexcited charge carrier generation and transportation. To further investigate the crystal structures of the perovskite films prepared by a two-step continuous method, we employed X-ray diffraction (XRD) to characterize the HTL/PbI<sub>2</sub> and HTL/Perovskite films (Fig. 2a and b). The peak positions of PbI<sub>2</sub> are nearly identical deposited on various samples with and without potassium salt modification. Meanwhile, the main diffraction peaks at  $2\theta = 14.2^\circ$ ,  $28.6^\circ$  and  $32.0^\circ$  are assigned to the (110), (220) and (310) planes of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> perovskite<sup>40</sup> without obvious shift. This indicates that the potassium cations post-treatment could hardly change the crystal structure of the PbI<sub>2</sub> and perovskite. Fig. S2 (ESI†) exhibits absorbance spectra of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> films based on different treated HTLs. An increased light absorbance intensity is detected for KSCN post-treatment over the entire wavelength, which could be benefited for the formation of photo-induced charge carriers.

Inverted PSCs were fabricated to investigate the influence of potassium salts on photovoltaic performance. The device structure is fluorine-doped tin oxide (FTO)/NiO<sub>x</sub>/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> (MAPbI<sub>3</sub>)/[6,6]-phenyl-C61-butyric acid methyl ester (PCBM)/bathocuproine (BCP)/Ag (Fig. 3a). Current density–voltage (*J*–*V*) curves of PSCs with different HTLs are represented in Fig. 3b measured under standard air mass 1.5 global (AM 1.5G) illumination with a reverse scan rate of 0.15 V s<sup>-1</sup>. These detailed performance parameters are summarized in Table 1. The device treated by KI attained the highest PCE of 20.10% with a photocurrent density (*J*<sub>SC</sub>) of 22.81 mA cm<sup>-2</sup>, an open circuit voltage (*V*<sub>OC</sub>) of 1085 mV and a FF of 0.812.



**Fig. 1** SEM images of (a) NiO<sub>x</sub> HTLs and (b) KI, (c) KSCN and (d) KNO<sub>3</sub> modified NiO<sub>x</sub> HTLs. SEM images of perovskite layers grown on (e) NiO<sub>x</sub> HTL, (f) NiO<sub>x</sub>/KI HTL, (g) NiO<sub>x</sub>/KSCN HTL and (h) NiO<sub>x</sub>/KNO<sub>3</sub> HTL.

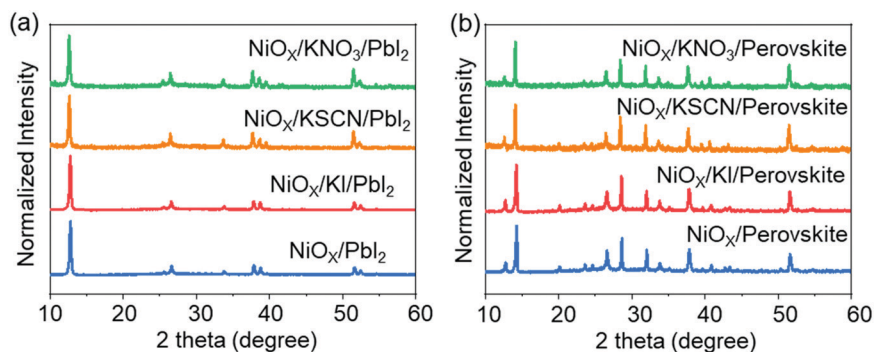


Fig. 2 XRD patterns of (a)  $\text{PbI}_2$  and (b) perovskite thin films grown on the FTO/ $\text{NiO}_x$  substrates with and without potassium salt modification.

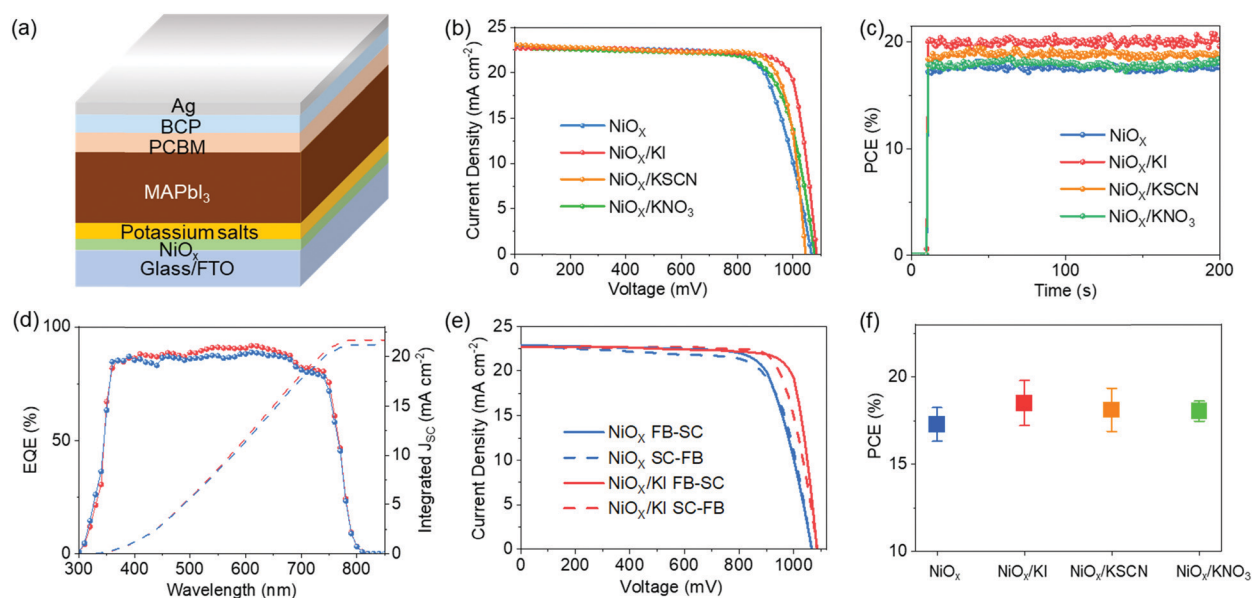


Fig. 3 (a) Device architecture of PSCs. (b)  $J$ - $V$  curves of PSCs based on  $\text{NiO}_x$ ,  $\text{NiO}_x/\text{KI}$ ,  $\text{NiO}_x/\text{KSCN}$  and  $\text{NiO}_x/\text{KNO}_3$  HTLs. (c) Steady-state power output of the device with and without modification measured at a fixed maximum power point (MPP) voltage as a function of time. (d) EQE spectra and the corresponding integrated photocurrent of the pristine (blue) and KI treated (red) device. (e) FB-SC and SC-FB  $J$ - $V$  curves of KI treated and pristine champion devices measured with a scan rate of  $0.15 \text{ V s}^{-1}$ . (f) Statistical photovoltaic parameters of PCE for devices based on different HTLs.

Table 1 Photovoltaic parameters of PSCs based on different  $\text{NiO}_x$  films, measured at simulated AM 1.5G solar irradiation

		$J_{\text{SC}}$ ( $\text{mA cm}^{-2}$ )	$V_{\text{OC}}$ (mV)	FF	PCE (%)
$\text{NiO}_x$	Highest	22.82	1064	0.752	18.25
	Average	22.43	1058	0.732	17.29
$\text{NiO}_x/\text{KI}$	Highest	22.81	1085	0.812	20.10
	Average	22.51	1077	0.772	18.81
$\text{NiO}_x/\text{KSCN}$	Highest	22.95	1050	0.803	19.36
	Average	22.72	1046	0.762	18.12
$\text{NiO}_x/\text{KNO}_3$	Highest	22.81	1075	0.757	18.56
	Average	22.18	1073	0.756	18.00

The stabilized power output of the champion device yielded a stabilized PCE of 20.03% after 200 s at maximum power point (MPP) under a bias of 0.9 V (Fig. 3c). And Fig. S3 (ESI<sup>†</sup>) displays the photocurrent density as a function of time for the best PSCs based on different  $\text{NiO}_x$  HTLs held at a forward bias of MPP.

Meanwhile, the external quantum efficiency (EQE) spectra of the KI modified devices also match well with the corresponding  $J$ - $V$  measurements (Fig. 3d). Examples of  $J$ - $V$  curves recorded at a scanning rate of  $0.15 \text{ V s}^{-1}$  are shown in Fig. 3e, where the reverse and forward scanning directions yielded PCE values of 20.10% and 19.70% for KI treated devices with little hysteresis. In the same batch, the devices based on  $\text{NiO}_x/\text{KSCN}$  and  $\text{NiO}_x/\text{KNO}_3$  HTL also revealed superior performances. The KSCN treated and  $\text{KNO}_3$  treated devices gave a PCE of 19.36% with a  $J_{\text{SC}}$  of  $22.95 \text{ mA cm}^{-2}$ , a  $V_{\text{OC}}$  of 1050 mV and an FF of 0.803, and a PCE of 18.56% with a  $J_{\text{SC}}$  of  $22.81 \text{ mA cm}^{-2}$ , a  $V_{\text{OC}}$  of 1075 mV and an FF of 0.757, respectively. By contrast, the pristine sample exhibited a  $J_{\text{SC}}$  of  $22.82 \text{ mA cm}^{-2}$ ,  $V_{\text{OC}}$  of 1064 mV and an FF of 0.752, and yielded a PCE of 18.25%. These devices by using modified  $\text{NiO}_x$  HTLs also exhibit good stability, as shown in Fig. S4 (ESI<sup>†</sup>). To confirm the reproducibility of the devices, 48 individual cells based on KI, KSCN and  $\text{KNO}_3$

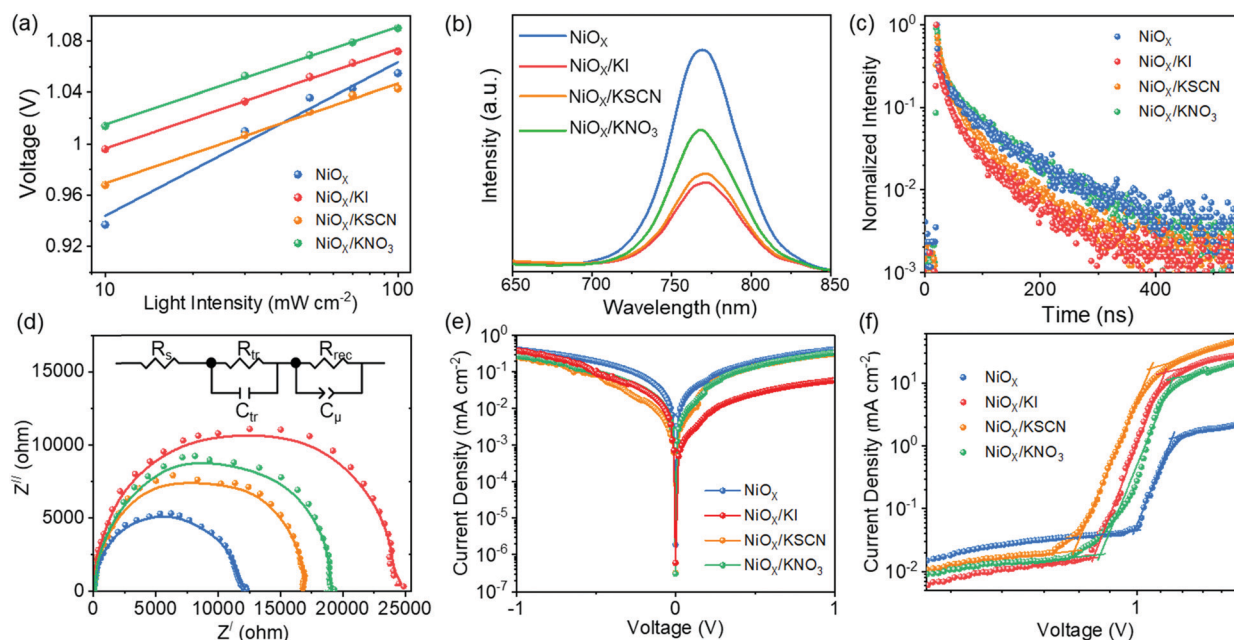
modification were measured to have a PCE of  $18.81\% \pm 6.4\%$ ,  $18.12\% \pm 5.7\%$  and  $18.00\% \pm 3.2\%$ , respectively, all higher than  $17.29 \pm 5.4\%$  of the pristine devices (Fig. 3f).

In addition, we further performed a detailed comparison between the PSCs based on  $\text{NiO}_x$  and  $\text{NiO}_x$ /potassium salt HTLs. We investigated the surface electronic properties of the pristine  $\text{NiO}_x$  film and  $\text{NiO}_x$ /KI films by ultraviolet photoelectron spectroscopy (UPS) and the results are exhibited in Fig. S5 (ESI<sup>†</sup>). The valence band positions *versus* vacuum level could be calculated to be 5.29 eV and 5.37 eV for the pristine  $\text{NiO}_x$  and KI modified  $\text{NiO}_x$ . In principle, deeper valence band maximum (VBM) of the HTLs close to the VBM of perovskite can result in better energy level alignment between the HTL and perovskite, which is beneficial for hole extraction and reducing the loss of photocarriers accumulated at the interface. The ideality factor ( $n$ ) was measured to evaluate the recombination behavior of the PSCs. The ideality factor ( $n$ ) of the devices can be inferred from the slope of  $V_{\text{OC}}$  as a function of light intensity ( $I$ ), according to the following equation:<sup>41</sup>

$$V_{\text{OC}} = \frac{E_g}{q} - \left( \frac{nkT}{q} \right) \ln \left( \frac{I_0}{I} \right)$$

where  $E_g$ ,  $k$ ,  $T$ , and  $q$  are the bandgap of the light absorber, Boltzmann constant, absolute temperature, and elementary charge, respectively. Fig. 4a shows the slopes of  $1.29 kT/q$ ,  $1.30 kT/q$  and  $1.27 kT/q$  for the KI, KSCN and  $\text{KNO}_3$  modified devices and  $2.00 kT/q$  for the pristine device, indicating that the suppression suggests reduced trap-assisted bimolecular recombination in the modified PSCs. We measured steady-state photoluminescence (PL) spectra with an excitation wavelength of

515 nm to understand carrier transport and recombination behavior at the interface between  $\text{NiO}_x$  and the perovskite layers. As revealed in Fig. 4b, the potassium salt treated  $\text{NiO}_x$  films displayed much more efficient PL quenching than the  $\text{NiO}_x$ /KI films (59.74% for KI treated film, 55.71% for KSCN treated film and 35.99% for  $\text{KNO}_3$  treated film). A similar trend regarding PL quenching was observed in terms of the time-resolved photoluminescence (TRPL) spectra, where the electron-hole lifetime is shorter in the perovskite coated on the  $\text{NiO}_x$ /potassium salt HTL than that of the  $\text{NiO}_x$  HTL (shown in Fig. 4c). This suggests that potassium salts at the  $\text{NiO}_x$ /perovskite interface can effectively passivate defects from perovskite and thus reduce the interface nonradiative recombination. Improved hole extraction could stem from better surface morphology and enhanced hole conductivity after potassium salt treatment for the  $\text{NiO}_x$  film, which might contribute to the increased  $V_{\text{OC}}$ . We also gained insight into the interfacial charge carrier dynamics of PSCs by electrochemical impedance spectroscopy (EIS) characterization (Fig. 4d) measured in the frequency range between 1.0 MHz and 0.1 Hz under dark conditions. Higher charge recombination resistances are recorded for the PSCs with potassium salt modified  $\text{NiO}_x$  HTLs versus their pristine devices, which is consistent with the superior  $V_{\text{OC}}$  of the corresponding devices.<sup>42</sup> To assess the electronic passivation effect of KI on  $\text{NiO}_x$  films, dark current density measurements were operated. The lower dark current density of the treated device indicates the reduction of defective grain boundaries, in agreement with the PL results (Fig. 4e). We tested the space charge limited current (SCLC) characteristics of devices based on different HTLs. Fig. 4f shows the KI, KSCN and  $\text{KNO}_3$  treatments



**Fig. 4** (a) Plots of light intensity dependent  $V_{\text{OC}}$  of the pristine and potassium salt treated device. (b) Steady-state PL spectra and (c) TRPL spectra of perovskite films deposited on different  $\text{NiO}_x$  substrates. (d) Nyquist plots recorded of PSCs based on pristine (blue) and KI (red), KSCN (orange),  $\text{KNO}_3$  (green) modified  $\text{NiO}_x$  films with a frequency range of 1 MHz–100 Hz in 0.8 V. (e) Dark  $J$ - $V$  curves of devices with and without modification. (f) SCLC curves of hole only devices deposited on different HTLs.

exhibiting a trap-filled limit voltage ( $V_{TFL}$ ) of 0.82 V, 0.76 V and 0.85 V, respectively, whereas the value of the pristine device (0.99 V) is much higher. This suggests that the potassium salt modified films possess lower trap-state densities and higher carrier mobility.

## Experimental

### Substrate treatments

The patterned FTO-coated glass substrates were cleaned by ultrasonication with detergent, acetone, ethanol and deionized water in sequence for 20 min, then dried at 80 °C and finally treated with ultraviolet ozone cleaner for 15 min.

### Preparation of solution-derived NiO<sub>x</sub> films

Precursor solutions were prepared by dissolving 0.029 g nickel(II) 2-ethylhexanoate (78% w/w in 2-ethylhexanoic acid, Aldrich) in *n*-hexane (0.2436 g). The solutions were then spin-coated onto FTO at 4000 rpm for 60 s, followed by heated at 100 °C for 30 min; after cooling to room temperature, the film was calcined at 400 °C for 1 h.<sup>43</sup>

### Interfacial modification of NiO<sub>x</sub> film

KI, KSCN and KNO<sub>3</sub> were dissolved in acetonitrile, respectively, with a molar concentration of 0.0525 M. Then the NiO<sub>x</sub> film was spin-coated at 4000 rpm for 60 s, followed by a heat treatment at 60 °C for 10 min.

### Synthesis of methylammonium iodide (CH<sub>3</sub>NH<sub>3</sub>I)

The CH<sub>3</sub>NH<sub>3</sub>I was prepared according to a reported method.<sup>44</sup>

### Device fabrication

PbI<sub>2</sub> (1.30 M) (DMSO) complex solution in DMF was prepared by mixing 0.5993 g PbI<sub>2</sub> (99.9985%, Alfa Aesar), 92 μL dimethyl sulfoxide (DMSO, 99.9%, Sigma-Aldrich) and 1 mL *N,N*-dimethylformamide (DMF, 99.9%, Alfa Aesar) with stirring at room temperature, overnight. The PbI<sub>2</sub> (DMSO) solution was spin-coated on NiO<sub>x</sub> films with or without modification at 3000 rpm for 30 s, and then MAI solution (60 mg mL<sup>-1</sup> in 2-propanol) was spin-coated on the top of the transparent PbI<sub>2</sub> (DMSO) film at 5000 rpm for 30 s after loading for 5 s. The films changed to dark brown during spin coating and the films were dried on a hot plate at 100 °C for 10 min. Then, the PCBM (dissolved in chlorobenzene of 20 mg mL<sup>-1</sup>) and BCP (0.5 mg mL<sup>-1</sup> in ethanol) were deposited by spin coating at 2000 rpm for 45 s, and 4000 rpm for 45 s. Finally, 110 nm Ag was thermally evaporated as the back electrode on the HTL.

### Characterization

The morphology and structure of the NiO<sub>x</sub> films were characterized by field emission scanning electron microscopy (FESEM, HITACHI S4800). The roughness and phase uniformity of the perovskite films were investigated by atomic force microscopy (AFM, Veeco/DI). The XRD spectra of the prepared powder were measured using powder X-ray diffraction (PXRD,

Bruker D8 Advanced Diffractometer, Cu K $\alpha$  radiation, 40 kV). The absorption spectra of the perovskite and the transmittance spectra of NiO<sub>x</sub> films were measured by using a Cary 500 UV-Vis-NIR spectrophotometer. The photoluminescence measurement was acquired at room temperature with a UV-Vis-NIR fluorescence spectrophotometer (Fluorolog-3-P) with an excitation wavelength of 515 nm. Electrochemical impedance spectroscopy (EIS) was carried out on an electrochemical workstation (Parstat 2273, Princeton) in the frequency range between 1 MHz and 100 Hz under short circuit at a forward bias of 0.2 V. The solar cells were illuminated using a solar light simulator (Solar IV-150A, Zolix) and the power of the simulated light was calibrated to 100 mW cm<sup>-2</sup> using a Newport calibrated KG5-filtered Si reference cell. *J*-*V* curves of solar cells were measured with a scan rate of 0.15 V s<sup>-1</sup> under standard AM 1.5G irradiation of 100 mW cm<sup>-2</sup> using a Keithley 2400 digital sourcemeter. Devices were masked with a metal aperture to define the active area of 0.0625 cm<sup>2</sup>. The steady state current output of the best-performing devices was measured by biasing the device at maximum power point for 200 s. The work function of the NiO<sub>x</sub> films was measured by UPS with a He source of incident energy 21.21 eV (He I line).

## Conclusions

In this research, we report a simple potassium salt modification strategy of NiO<sub>x</sub> based HTLs for inverted PSCs. This strategy can effectively increase the charge extraction capability of NiO<sub>x</sub> films and improve the perovskite films' quality, leading to improved device performance. The champion device modified *via* KI attained a PCE of 20.10% with a *J*<sub>SC</sub> of 22.81 mA cm<sup>-2</sup>, a *V*<sub>OC</sub> of 1085 mV and an FF of 0.812, compared to 18.25% of the pristine sample. Our work indicates that potassium salt modified nickel oxides are excellent candidates to obtain high performance inverted PSCs and provides a simple and efficient approach for practical applications.

## Author contributions

X. L. and Y. H. designed the study. H. G. Y., S. Y. and Y. H. directed the study. X. L. performed the main experimental work. X. L., Q. H. W. and M. C. contributed to the characterizations. Y. H., B. G. and H. W. Q. analyzed the data. X. L., Y. H. and S. Y. wrote the manuscript. All authors revised the paper.

## Conflicts of interest

There are no conflicts to declare.

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