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One-pot synthesis of *N*-substituted benzannulated triazoles *via* stable arene diazonium salts†

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A mild and effective one-pot synthesis of 1,2,3-benzotriazin-4(3*H*)-ones and benzothiatriazine-1,1(2*H*)-dioxide analogues has been developed. The method involves the diazotisation and subsequent cyclisation of 2-aminobenzamides and 2-aminobenzenesulfonamides *via* stable diazonium salts, prepared using a polymer-supported nitrite reagent and *p*-tosic acid. The transformation was compatible with a wide range of aryl functional groups and amide/sulfonamide-substituents and was used for the synthesis of pharmaceutically important targets. The synthetic utility of the one-pot diazotisation–cyclisation process was further demonstrated with the preparation of an α -amino acid containing 1,2,3-benzotriazin-4(3*H*)-one.

Introduction

Benzotriazin-4(3*H*)-ones and benzothiatriazine-1,1(2*H*)-dioxides are privileged heterocyclic moieties in medicinal chemistry and are key components of a wide range of pharmaceutically-relevant compounds (Fig. 1).^{1,2} For example, benzotriazin-4(3*H*)-one **1** bearing a naphthalene-substituted piperazine side-chain was found to be a high affinity agent for the 5-HT_{1A} receptor.^{2d} Benzotriazin-4(3*H*)-one **3** is a potent anaesthetic with an IC₅₀ value comparable to lidocaine,^{2c} while benzothiatriazine-1,1(2*H*)-dioxide **4** is a well-known diuretic agent.^{2a} Benzannulated triazoles have also been widely used for other applications, such as rubber components, dyes and as photoluminescence agents.¹ Furthermore, these benzannulated heterocycles are useful synthetic precursors in organic chemistry.¹ The groups of Murakami,³ Liu,⁴ and Cheng⁵ have shown that these compounds can undergo nickel-catalysed denitrogenative insertion reactions, with alkenes, alkynes and allenes, while palladium-catalysed denitrogenation and subsequent reaction with isocyanides^{6a} or alkynes^{6b} have allowed the preparation of 3-isoindolin-1-ones and *o*-alkynylated benzamides, respectively. Other nickel-catalysed denitrogenative substitution processes⁷ and visible-light-activated annulation reactions have also been reported.⁸ Specific reactions of benzothiatriazine-1,1(2*H*)-dioxides have been described for the synthesis of biaryl sultams, using thermal,⁹ visible light¹⁰ or acid-promoted¹¹ methods.

The importance of 1,2,3-benzotriazin-4(3*H*)-ones and 1,2,3-benzothiatriazine-1,1(2*H*)-dioxides have resulted in the development of a wide range of methods for their synthesis.¹ Traditionally, these compounds were prepared from 2-aminobenzamides or 2-aminobenzenesulfonamides by reaction with sodium nitrite and hydrochloric acid (Scheme 1a).^{2b,12} Due to the harsh acidic conditions that limits substrate scope, unstable diazonium salts and the use of sodium nitrite that can lead to the release of toxic nitrogen oxides, milder reagents and reaction conditions for this transformation have been reported.¹³ In addition, new approaches have been developed for the preparation of 1,2,3-benzotriazin-4(3*H*)-ones and 1,2,3-benzothiatriazine-1,1(2*H*)-dioxides.¹⁴ For example,

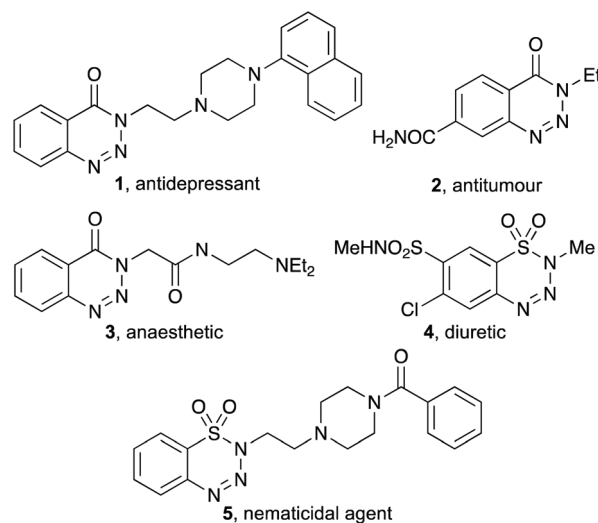


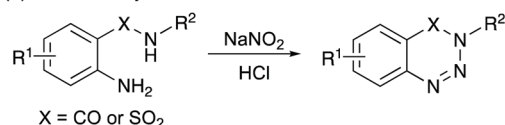
Fig. 1 Biologically active benzotriazin-4(3*H*)-ones and benzothiatriazine-1,1(2*H*)-dioxides.

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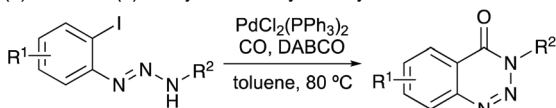
† Electronic supplementary information (ESI) available: Experimental procedures and data for starting materials, ¹H and ¹³C NMR spectra for all compounds. See DOI: 10.1039/d1ob00968k



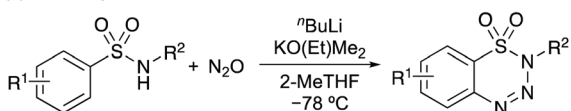
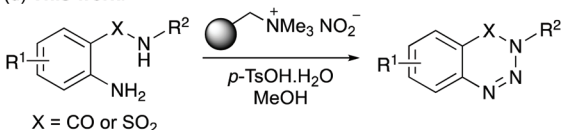
(a) Traditional synthesis via a diazonium salt intermediate.



(b) Palladium(0)-catalysed carbonylative cyclisation.



(c) Redox cyclisation with nitrous oxide.

(d) **This work:****Scheme 1** Methods for the synthesis of benzotriazin-4(3H)-ones and benzothiatriazine-1,1(2H)-dioxides.

Chandrasekhar and Sankararaman reported the preparation of *N*-substituted 1,2,3-benzotriazin-4(3H)-ones *via* the palladium(0)-catalysed carbonylation of 1,3-diaryltriazenes (Scheme 1b),¹⁵ while Cui and co-workers prepared both series of compounds by a redox cyclisation of amides and sulfonamides with nitrous oxide (Scheme 1c).¹⁶ The Song group have reported the preparation of 1,2,3-benzotriazin-4(3H)-ones by the oxidative rearrangement of 3-aminindazoles.¹⁷

Despite the wide range of approaches now developed for the synthesis of these heterocycles, there is still a need for a mild and general approach from readily available starting materials, that avoids the requirement of harsh acidic or strongly oxidising conditions and precious transition metal catalysis. In recent years, issues associated with standard diazonium salt formation (NaNO_2 , HCl) of aromatic compounds have been overcome with the use of polymer-supported nitrite reagents and milder acids.¹⁸ For example, Filimonov and co-workers demonstrated that aryl diazonium tosylate salts could be prepared using nitrite supported on a tetraalkylammonium functionalised resin, such as Amberlyst A-26, in the presence of *p*-tosic acid.^{18c,d} As well as the mild reaction conditions, the aryl diazonium tosylate salts were found to have high thermal and aging stability. Our group have exploited this safe and mild approach for aryl diazonium salt formation in one-pot processes for (radio)iodination,¹⁹ Heck–Matsuda reactions,²⁰ and the synthesis of benzotriazoles.²¹ We now report the one-pot synthesis of 1,2,3-benzotriazin-4(3H)-ones and 1,2,3-benzothiatriazine-1,1(2H)-dioxides *via* stable aryl diazonium salts, by reaction of readily available 2-aminobenzamides and 2-aminobenzenesulfonamides with a polymer-supported nitrite reagent and *p*-tosic acid

(Scheme 1d). As well as demonstrating the scope of these processes, we describe the application of this transformation for the synthesis of various biologically active compounds, including the synthesis of an α -amino acid containing 1,2,3-benzotriazin-4(3H)-one.

Results and discussion

The study began with the optimisation of a one-pot diazotisation and cyclisation of 2-aminobenzamide (**6a**) for the preparation of 1,2,3-benzotriazin-4(3H)-one (**7a**) (Table 1). Following our previous work,^{19–21} a polymer-supported nitrite reagent, readily prepared by the ion exchange of Amberlyst A-26 with an aqueous solution of sodium nitrite, was used in combination with *p*-tosic acid.^{18c,d} The reaction was initially trialled using one equivalent of both reagents and methanol as the reaction solvent (entry 1). While the transformation proceeded at room temperature, a reaction time of 48 hours was required for completion and gave **7a** in 39% yield (entry 1). Using three equivalents of both reagents led to a substantial improvement, generating **7a** in 87% yield after 2 hours (entry 2). A further increase in the number of equivalent of reagents resulted in a faster, but less efficient process (entry 3). A solvent screen was also performed (entries 4–6), and while these reactions gave **7a** in good yields, methanol was deemed the optimal solvent. It should be noted that as well as avoiding toxic reagents and harsh conditions, an advantage of this process is the simple work-up procedure and purification of products. On reaction completion, the polymer resin is removed by filtration (and recycled), and the product then purified by flash column

Table 1 Optimisation of the one-pot diazotisation and cyclisation of 2-aminobenzamide (**6a**)

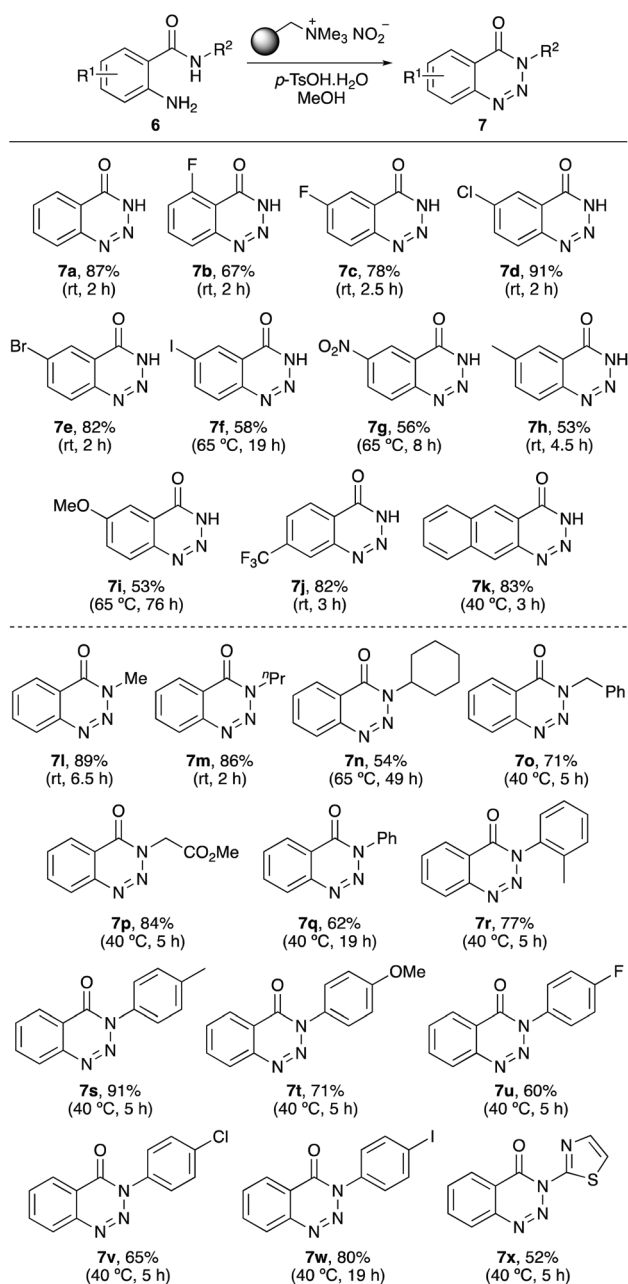
Entry	Reagents ^a (equiv.)	Solvent	Time (h)	Yield ^b (%)
1	1	MeOH	48	39
2	3	MeOH	2	87
3	6	MeOH	1	62
4	3	MeCN	2	55
5	3	THF	2	73
6	3	EtOAc	2	65
7 ^c	3	MeOH	2.5	56

^a Polymer supported nitrite and *p*-tosic acid. ^b Isolated yield. ^c Reaction was performed using sodium nitrite.



chromatography. By comparison, if the polymer-supported nitrite reagent is replaced with sodium nitrite in this process, while the reaction is completed in a similar time, the requirement of performing a work-up involving aqueous washes to remove *p*-TsOH, results in the isolation of **7a** in 56% yield (entry 7).

Following optimisation, the scope of the reaction was explored for the preparation of other *N*-unsubstituted 1,2,3-benzotriazin-4(3*H*)-ones (Scheme 2). 2-Aminobenzamides **6a–6k** bearing a range of *o*-, *m*- and *p*-substituted functional groups were found to be substrates for the reaction, providing



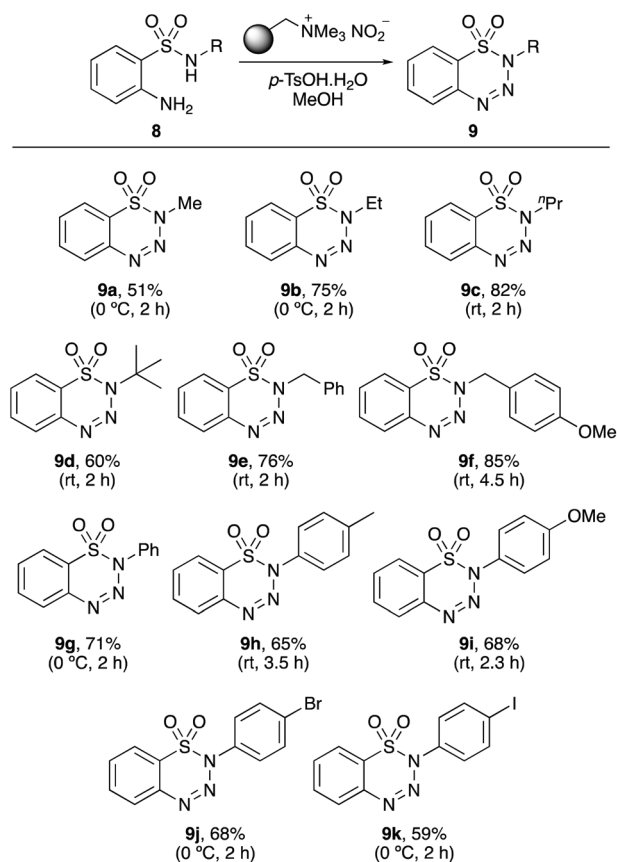
Scheme 2 Reaction scope for the synthesis of 1,2,3-benzotriazin-4(3*H*)-ones.

the corresponding 1,2,3-benzotriazin-4(3*H*)-ones **7a–7k** in 53–91% yields. For the majority of the 2-aminobenzamides, the transformation proceeded at room temperature in reaction times of 2–4.5 h. Only methoxy-substituted 2-aminobenzamide **6i** required both elevated temperatures (65 °C) and a long reaction time (76 h). This is likely due to the reduced electrophilic nature of the diazo intermediate prior to cyclisation.

The next stage of the project investigated the application of the one-pot method for the synthesis of *N*-substituted 1,2,3-benzotriazin-4(3*H*)-ones (Scheme 2). Unlike most of the simple 2-aminobenzamides used in the first part of this study that were commercially available, many of the *N*-alkyl and *N*-aryl substituted benzamides required synthesis. These were prepared in one-step by the reaction of isatoic anhydride with various amines.^{22,23} Reaction of *N*-alkyl benzamides with polymer-supported nitrite reagent and *p*-tosic acid, under optimised conditions gave the corresponding *N*-alkyl 1,2,3-benzotriazin-4(3*H*)-ones **7l–7p** in 54–89% yields. With these compounds, it was found that cyclisation of more sterically hindered secondary amides required more forcing conditions. For example, *N*-cyclohexyl benzamide **6n** required a reaction time of 49 hours and a reaction temperature of 65 °C, while a *t*-butyl analogue demonstrated the limitation of this approach with no reaction observed. *N*-Aryl benzamides also required slightly elevated temperatures (40 °C) but all substrates investigated were converted to the corresponding *N*-aryl 1,2,3-benzotriazin-4(3*H*)-ones (**7q–7x**). The process was efficient for the preparation of 1,2,3-benzotriazin-4(3*H*)-ones bearing both electron-rich and electron-deficient *N*-aryl substituents and was also compatible with *N*-heterocyclic moieties (e.g. **7x**). The synthesis of **7r** demonstrates that *ortho*-substituted *N*-aryl groups are also tolerated during the cyclisation step.

The research programme then examined the scope of the one-pot diazotisation and cyclisation for the preparation of *N*-substituted 1,2,3-benzothiazine-1,1(2*H*)-dioxides **9** (Scheme 3). The *N*-substituted benzenesulfonamides **8** were prepared in two steps by reaction of 2-nitrobenzenesulfonyl chloride with an amine, under basic conditions, followed by reduction of the nitro-group with tin dichloride.²³ The *N*-substituted 2-aminobenzenesulfonamides **8** were found to be more reactive to diazotisation and cyclisation than the corresponding benzamides **6**. Reactions proceeded at 0 °C or room temperature and with relatively short reaction times (2–4.5 h). For 2-amino-*N*-methylbenzenesulfonamide (**8a**) and *N*-ethyl analogue **8b**, room temperature reactions led to denitrogenation of the diazo intermediates and isolation of the deaminated *N*-alkylbenzenesulfonamides as the major products. However, performing these reactions at 0 °C, allowed clean transformations and the isolation of **8a** and **8b** in 51% and 75% yields, respectively. Reaction of the other *N*-alkylbenzenesulfonamides proceeded smoothly at room temperature and gave 1,2,3-benzothiazine-1,1(2*H*)-dioxides **9c–9f**, in 60–85% yields. A demonstration of the increased reactivity of this class of compound was the synthesis of *t*-butyl analogue **9d**. While the corresponding 2-aminobenzamide gave no product, **9d** was isolated in 60% yield. Similarly, the

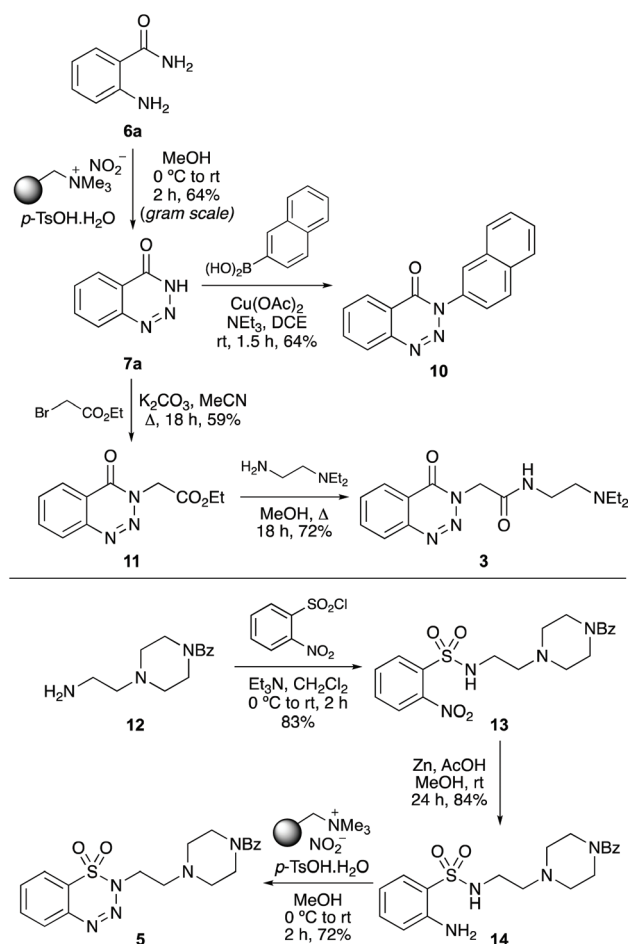




Scheme 3 Reaction scope for the synthesis of 1,2,3-benzothiazine-1,1(2H)-dioxides.

reaction of 2-amino- N -arylbenzenesulfonamides also proceeded at 0 °C or room temperature and gave the corresponding N -aryl 1,2,3-benzothiazine-1,1(2H)-dioxides **9g–9k** in 59–71% yields. Several of the compounds prepared in this part of the study are used for the production of materials. For example, **9a**, **9d** and **9g** are blowing agents for the formation of cellular rubber,²⁴ while **9a** and **9g** are also used as catalysts for free-radical polymerisation.²⁵

We next examined the application of the one-pot process for the synthesis of biologically active benzannulated triazoles. Initially, the optimised conditions for diazotisation and cyclisation (Table 1, entry 2) were used for the gram scale synthesis of 1,2,3-benzotriazin-4(3H)-one (**7a**). This gave **7a** in 64% yield (Scheme 4). Copper(II)-catalysed N -arylation of **7a** with naphthalene-2-boronic acid under basic conditions completed the two-step synthesis of chorismate mutase inhibitor **10**.²⁶ 1,2,3-Benzotriazin-4(3H)-one (**7a**) was also used for the three-step synthesis of anaesthetic compound **3**, which has similar activity to lidocaine.^{2c} N -Alkylation of **7a** with ethyl bromoacetate, followed by aminolysis with N,N -diethylethylenediamine gave **3** in good overall yield. As well as the synthesis of biologically active 1,2,3-benzotriazin-4(3H)-ones, the one-pot method was also used for a new synthesis of benzothiazine-1,1(2H)-dioxide **5**, a compound with nematicidal activity and a poten-

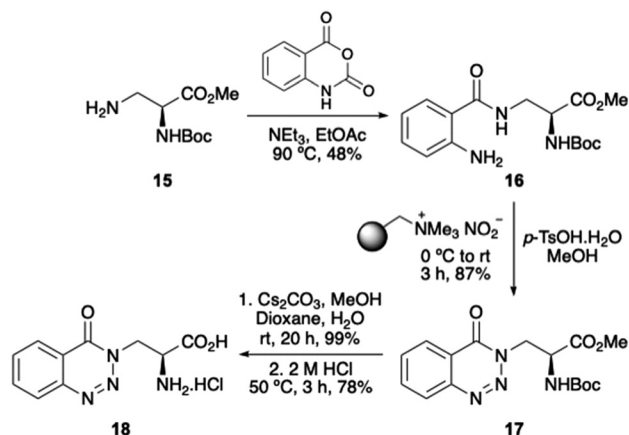


Scheme 4 Synthesis of biologically active benzannulated triazoles **3**, **5** and **10**.

tial pesticide with low-toxicity.^{2e} Initially, amine **12** was prepared in two steps.²³ N -Benzoylpiperazine was alkylated with N -(2-bromoethyl)phthalimide and this was followed by removal of the phthalimide protecting group with hydrazine. Reaction of **12** with 2-nitrobenzenesulfonyl chloride and nitro-group reduction with zinc/acetic acid gave sulfonamide **14** in high yields. Application of the one-pot diazotisation and cyclisation method with **14**, proceeded at room temperature and after a reaction time of 2 hours, gave nematicidal agent **5** in 72% yield.

We have an interest in the development of novel α -amino acids that can be used as biological probes.²⁷ Thus, as a proof-of-concept experiment, we were interested in exploring whether the one-pot diazotisation and cyclisation procedure could be used for the preparation of an α -amino acid bearing a 1,2,3-benzotriazin-4(3H)-one side-chain. Key intermediate **16** was prepared by the reaction of known L-3-aminoalanine derivative **15**²⁸ with isatoic anhydride (Scheme 5). Diazotisation and cyclisation of **16** was complete in 3 hours and gave 1,2,3-benzotriazin-4(3H)-one **17** in 87% yield. Despite the use of p -tosic acid, no Boc-group deprotection was observed under the mild diazotisation reaction conditions.



Scheme 5 Synthesis of α -amino acid 18.

Ester hydrolysis with caesium carbonate at room temperature, followed by acid-mediated removal of the Boc-protecting group gave parent amino acid 18.

Conclusions

In summary, a mild and general one-pot process has been developed for the synthesis of both 1,2,3-benzotriazin-4(3H)-ones and 1,2,3-benzothiazotriazine-1,1(2H)-dioxides *via* stable aryl diazonium tosylate salts. The use of a polymer-supported nitrite reagent and *p*-tosic acid facilitated the rapid diazotisation and cyclisation of 2-aminobenzamides and 2-aminobenzenesulfonamides, yielding the target compounds in good yields. The process was found to be compatible for a wide range of substrates,²⁹ bearing *N*-aryl or *N*-alkyl substituted amides or sulfonamides. The mild nature of this transformation allowed the preparation of various agents for rubber and polymer production, as well as the synthesis of medicinally important compounds. The use of this process for the preparation of a benzotriazin-4(3H)-one derived amino acid serves as a platform for future work and the synthesis of more complex analogues for chemical biology applications.

Experimental

All reagents and starting materials were obtained from commercial sources and used as received. All dry solvents were purified using a PureSolv 500 MD solvent purification system. Brine refers to a saturated aqueous solution of sodium chloride. Flash column chromatography was performed using Merck Geduran Si 60 (35–70 μ m) silica gel. Merck aluminium-backed plates pre-coated with silica gel 60F₂₅₄ were used for thin layer chromatography and were visualised with a UV lamp or by staining with KMnO₄ or ninhydrin. ¹H NMR spectra were recorded on a Bruker DPX spectrometer at either 400 or 500 MHz and data are reported as follows: chemical shift in

ppm relative to tetramethylsilane as the internal standard, multiplicity (integration, s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet or overlap of non-equivalent resonances). ¹³C NMR spectra were recorded on a Bruker DPX spectrometer at either 101 or 126 MHz and data are reported as follows: chemical shift in ppm relative to tetramethylsilane or the solvent as internal standard (CDCl₃: δ 77.0 ppm, CD₃OD: δ 49.0 ppm, DMSO-*d*₆: δ 39.5 ppm), multiplicity with respect to hydrogen (deduced from DEPT experiments, C, CH, CH₂ or CH₃). Infrared spectra were recorded on a Shimadzu FTIR-84005 spectrometer; wavenumbers are indicated in cm⁻¹. Mass spectra were recorded using a JEOL JMS-700 Spectrometer or a Bruker microTOFq High Resolution Mass Spectrometer. Melting points were determined on a Gallenkamp melting point apparatus and are reported uncorrected.

General procedure for preparation of polymer supported nitrite resin

To a stirred solution of sodium nitrite (5.50 g, 80.0 mmol) in water (200 mL) was added Amberlyst A26 hydroxide form resin (10.0 g, 40.0 mmol). The reaction mixture was stirred at rt for 1 h. The polymer-supported resin was filtered and washed with water until the filtrate was of neutral pH. The content of the polymer-supported nitrite was 3.5 mmol of NO₂ per g.¹⁸

General Procedure for synthesis of 1H-benzotriazinones

To a stirred solution of the 2-aminobenzamides (1 equiv.) in methanol (10 mL mmol⁻¹) at 0 °C was added polymer-supported nitrite (containing 3.0 equiv. of NO₂) and *p*-toluenesulfonic acid monohydrate (3 equiv.). The reaction mixture was stirred for 1 h at 0 °C. The reaction mixture was then warmed to room temperature and stirred until completion. The resin was filtered and washed with methanol (20 mL mmol⁻¹). The reaction mixture was concentrated *in vacuo*. Purification by flash column chromatography gave the 1H-benzotriazinones.

1,2,3-Benzotriazin-4(3H)-one (7a).³⁰ The reaction was carried out as described in the general procedure using 2-aminobenzamide (6a) (0.303 g, 2.20 mmol), polymer-supported nitrite (1.89 g, containing 6.60 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (1.26 g, 6.60 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for 1 h. Purification by flash column chromatography, eluting with 35% ethyl acetate in hexane gave 1,2,3-benzotriazin-4(3H)-one (7a) (0.294 g, 87%) as a white solid. Mp 212–215 °C (lit.³⁰ 215–217 °C); δ_{H} (400 MHz, DMSO-*d*₆) 7.91 (1H, br t, *J* 7.7 Hz, 7-H), 8.08 (1H, br t, *J* 7.7 Hz, 6-H), 8.18 (1H, br d, *J* 7.7 Hz, 8-H), 8.22 (1H, br d, *J* 7.7 Hz, 5-H), 14.94 (1H, br s, NH); δ_{C} (101 MHz, DMSO-*d*₆) 120.2 (C), 124.3 (CH), 127.8 (CH), 132.6 (CH), 135.5 (CH), 144.2 (C), 155.6 (C); *m/z* (EI) 147 (M⁺, 100%), 104 (22), 92 (69), 76 (72), 63 (83).

5-Fluoro-1,2,3-benzotriazin-4(3H)-one (7b). The reaction was carried out as described in the general procedure using 6-fluoro-2-aminobenzamide (6b) (0.200 g, 1.30 mmol), polymer-supported nitrite (1.11 g, containing 3.89 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.740 g,



3.89 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for 1 h. Purification by flash column chromatography, eluting with 70% diethyl ether in hexane gave 5-fluoro-1,2,3-benzotriazin-4(3*H*)-one (**7b**) (0.144 g, 67%) as a white solid. Mp 200–204 °C; $\nu_{\max}/\text{cm}^{-1}$ (neat) 3161 (NH), 2975 (CH), 2160, 2012, 1700 (C=O), 1480, 1262, 815; δ_{H} (500 MHz, DMSO- d_6) 7.68 (1H, dd, J 10.5, 9.0 Hz, 6-H), 8.00 (1H, br d, J 8.0 Hz, 8-H), 8.07 (1H, ddd, J 10.5, 8.0, 5.0 Hz, 7-H), 14.97 (1H, br s, NH); δ_{C} (126 MHz, DMSO- d_6) 109.9 (C, $^2J_{\text{CF}}$ = 8.8 Hz), 118.6 (CH, d, $^2J_{\text{CF}}$ = 20.2 Hz), 124.0 (CH, $^4J_{\text{CF}}$ = 3.8 Hz), 136.5 (CH, $^3J_{\text{CF}}$ = 10.1 Hz), 145.7 (C), 152.7 (C, $^3J_{\text{CF}}$ = 1.3 Hz), 158.7 (C, d, $^1J_{\text{CF}}$ = 264.6 Hz); m/z (ESI) 166.0410 (MH^+). $\text{C}_7\text{H}_5\text{FN}_3\text{O}$ requires 166.0411.

6-Fluoro-1,2,3-benzotriazin-4(3*H*)-one (7c).¹⁷ The reaction was carried out as described in the general procedure using 2-amino-5-fluorobenzamide (**6c**) (0.150 g, 0.970 mmol), polymer-supported nitrite (0.836 g, containing 2.92 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.556 g, 2.92 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for 1.5 h. Purification by flash column chromatography, eluting with 30% ethyl acetate in hexane gave 6-fluoro-1,2,3-benzotriazin-4(3*H*)-one (**7c**) (0.125 g, 78%) as a white solid. Spectroscopic data were consistent with the literature.¹⁷ Mp 206–208 °C; δ_{H} (400 MHz, DMSO- d_6) 7.91–8.00 (2H, m, 5-H and 7-H), 8.31 (1H, ddd, J 8.8, 5.0, 0.4 Hz, 8-H), 15.04 (1H, br s, NH); δ_{C} (101 MHz, DMSO- d_6) 109.5 (CH, $^2J_{\text{CF}}$ = 24.0 Hz), 122.4 (C, $^3J_{\text{CF}}$ = 9.4 Hz), 123.9 (CH, d, $^2J_{\text{CF}}$ = 24.4 Hz), 131.6 (CH, d, $^3J_{\text{CF}}$ = 9.5 Hz), 141.5 (C, d, $^4J_{\text{CF}}$ = 2.2 Hz), 155.0 (C, d, $^4J_{\text{CF}}$ = 3.1 Hz), 163.2 (C, d, $^1J_{\text{CF}}$ = 253.4 Hz); m/z (EI) 165 (M^+ , 20%), 122 (10), 94 (21), 78 (88), 63 (100).

6-Chloro-1,2,3-benzotriazin-4(3*H*)-one (7d).³¹ The reaction was carried out as described in the general procedure using 2-amino-5-chlorobenzamide (**6d**) (0.150 g, 0.880 mmol), polymer-supported nitrite (0.759 g, containing 2.65 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.504 g, 2.65 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for 1 h. Purification by flash column chromatography, eluting with 35% ethyl acetate in hexane gave 6-chloro-1,2,3-benzotriazin-4(3*H*)-one (**7d**) (0.145 g, 91%) as a white solid. Mp 190–192 °C (lit.³¹ 195–196 °C); δ_{H} (400 MHz, DMSO- d_6) 8.12 (1H, dd, J 8.7, 2.4 Hz, 7-H), 8.18 (1H, d, J 2.4 Hz, 5-H), 8.22 (1H, d, J 8.7 Hz, 8-H), 15.11 (1H, br s, NH); δ_{C} (101 MHz, DMSO- d_6) 121.7 (C), 123.5 (CH), 130.2 (CH), 135.6 (CH), 136.9 (C), 142.8 (C), 154.6 (C); m/z (EI) 181 (M^+ , 18%), 138 (7), 110 (11), 78 (88), 63 (100).

6-Bromo-1,2,3-benzotriazin-4(3*H*)-one (7e).³² The reaction was carried out as described in the general procedure using 2-amino-5-bromobenzamide (**6e**) (0.150 g, 0.700 mmol), polymer-supported nitrite (0.601 g, containing 2.10 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.399 g, 2.10 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for 1 h. Purification by flash column chromatography, eluting with 25% ethyl acetate in hexane gave 6-bromo-1,2,3-benzotriazin-4(3*H*)-one (**7e**) (0.129 g, 82%) as a white solid. Spectroscopic data were consistent with the literature.³² Mp 207–209 °C; δ_{H} (400 MHz,

DMSO- d_6) 8.12 (1H, d, J 8.6 Hz, 8-H), 8.24 (1H, dd, J 8.6, 2.2 Hz, 7-H), 8.31 (1H, d, J 2.2 Hz, 5-H), 15.10 (1H, br s, NH); δ_{C} (101 MHz, DMSO- d_6) 121.7 (C), 125.6 (C), 126.5 (CH), 130.1 (CH), 138.3 (CH), 142.9 (C), 154.3 (C); m/z (EI) 227 (M^+ , 23%), 225 (23), 184 (22), 182 (21), 156 (17), 154 (18), 78 (78), 63 (100).

6-Iodo-1,2,3-benzotriazin-4(3*H*)-one (7f). The reaction was carried out as described in the general procedure using 2-amino-5-iodobenzamide (**6f**) (0.150 g, 0.570 mmol), polymer-supported nitrite (0.497 g, containing 1.72 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.372 g, 1.72 mmol). The reaction mixture was stirred for 1 h at 0 °C and then heated under reflux for 18 h. Purification by flash column chromatography, eluting with 40% ethyl acetate in hexane gave 6-iodo-1,2,3-benzotriazin-4(3*H*)-one (**7f**) (0.0910 g, 58%) as a white solid. Mp 127–130 °C; $\nu_{\max}/\text{cm}^{-1}$ (neat) 3071 (NH), 2945 (CH), 1672 (C=O), 1227, 1188; δ_{H} (400 MHz, CD_3OD) 7.90 (1H, d, J 8.6 Hz, 8-H), 8.37 (1H, dd, J 8.5, 2.0 Hz, 7-H), 8.64 (1H, d, J 2.0 Hz, 5-H); δ_{C} (101 MHz, CD_3OD) 99.5 (C), 122.8 (C), 130.5 (CH), 134.6 (CH), 145.2 (C), 145.6 (CH), 156.6 (C); m/z (ESI) 295.9288 (MNa^+). $\text{C}_7\text{H}_4\text{IN}_3\text{NaO}$ requires 295.9291.

6-Nitro-1,2,3-benzotriazin-4(3*H*)-one (7g).³³ The reaction was carried out as described in the general procedure using 2-amino-5-nitrobenzamide (**6g**) (0.0800 g, 0.442 mmol), polymer-supported nitrite (0.378 g, containing 1.32 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.252 g, 1.32 mmol). The reaction mixture was stirred for 1 h at 0 °C and heated under reflux for 7 h. Purification by flash column chromatography, eluting with 60% diethyl ether in hexane gave 6-nitro-1,2,3-benzotriazin-4(3*H*)-one (**7g**) (0.0480 g, 56%) as a yellow solid. Mp 189–194 °C (lit.³³ 194 °C); δ_{H} (400 MHz, DMSO- d_6) 8.42 (1H, dd, J 8.9, 0.5 Hz, 8-H), 8.76 (1H, dd, J 8.9, 2.6 Hz, 7-H), 8.82 (1H, dd, J 2.6, 0.5 Hz, 5-H), 15.42 (1H, br s, NH); δ_{C} (101 MHz, DMSO- d_6) 120.3 (CH), 121.1 (C), 129.4 (CH), 130.1 (CH), 146.4 (C), 148.5 (C), 154.8 (C); m/z (EI) 192 (M^+ , 28%), 149 (28), 84 (82), 66 (100).

6-Methyl-1,2,3-benzotriazin-4(3*H*)-one (7h).³⁴ The reaction was carried out as described in the general procedure using 2-amino-5-methylbenzamide (**6h**) (0.100 g, 0.660 mmol), polymer-supported nitrite (0.573 g, containing 2.00 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.380 g, 2.00 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for 3.5 h. Purification by flash column chromatography, eluting with 30% ethyl acetate in petroleum ether gave 6-methyl-1,2,3-benzotriazin-4(3*H*)-one (**7h**) (0.0570 g, 53%) as a white solid. Mp 219–220 °C (lit.³⁴ 217–218 °C); δ_{H} (400 MHz, DMSO- d_6) 2.54 (3H, s, 6- CH_3), 7.89 (1H, dd, J 8.2, 1.6 Hz, 7-H), 8.01 (1H, br s, 5-H), 8.07 (1H, d, J 8.2 Hz, 8-H), 14.86 (1H, br s, NH); δ_{C} (101 MHz, DMSO- d_6) 21.3 (CH_3), 120.1 (C), 123.6 (CH), 127.8 (CH), 136.7 (CH), 142.6 (C), 143.4 (C), 155.6 (C); m/z (EI) 161 (M^+ , 8%), 104 (8), 89 (6), 78 (82), 63 (100).

6-Methoxy-1,2,3-benzotriazin-4(3*H*)-one (7i). The reaction was carried out as described in the general procedure using 5-methoxy-2-aminobenzamide (**6i**) (0.100 g, 0.600 mmol), polymer-supported nitrite (0.515 g, containing 1.80 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.342 g,



1.80 mmol). The reaction mixture was stirred for 1 h at room temperature and then heated under reflux for 75 h. Purification by flash column chromatography, eluting with 40% ethyl acetate in hexane gave 6-methoxy-1,2,3-benzotriazin-4(3*H*)-one (**7i**) (0.0560 g, 53%) as a white solid. Mp 190–192 °C; $\nu_{\max}/\text{cm}^{-1}$ (neat) 3159 (NH), 2959 (CH), 1662 (C=O), 1296, 1141, 895, 833; δ_{H} (400 MHz, DMSO-*d*₆) 3.96 (3H, s, 6-OCH₃), 7.55 (1H, d, *J* 2.8 Hz, 5-H), 7.62 (1H, dd, *J* 9.0, 2.8 Hz, 7-H), 8.12 (1H, d, *J* 9.0 Hz, 8-H), 14.79 (1H, br s, NH); δ_{C} (101 MHz, DMSO-*d*₆) 56.2 (CH₃), 104.2 (CH), 122.0 (C), 124.6 (CH), 130.1 (CH), 139.4 (C), 155.6 (C), 161.9 (C); *m/z* (EI) 177.0534 (*M*⁺. C₈H₇N₃O₂ requires 177.0538), 134 (12%), 106 (38), 78 (79), 63 (100).

7-Trifluoromethyl-1,2,3-benzotriazin-4(3*H*)-one (7j). The reaction was carried out as described in the general procedure using 2-amino-4-trifluorobenzamide (**6j**) (0.150 g, 0.730 mmol), polymer-supported nitrite (0.630 g, containing 2.20 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.418 g, 2.20 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for 2 h. Purification by flash column chromatography, eluting with 30% ethyl acetate in petroleum ether gave 7-trifluoromethyl-1,2,3-benzotriazin-4(3*H*)-one (**7j**) (0.130 g, 82%) as a white solid. Mp 185–187 °C; $\nu_{\max}/\text{cm}^{-1}$ (neat) 3136 (NH), 1694 (C=O), 1651 (C=C), 1319, 1134, 864; δ_{H} (400 MHz, DMSO-*d*₆) 8.20 (1H, br d, *J* 8.3 Hz, 6-H), 8.40 (1H, br d, *J* 8.3 Hz, 5-H), 8.57 (1H, br s, 8-H), 15.25 (1H, br s, NH); δ_{C} (101 MHz, DMSO-*d*₆) 123.1 (C, q, ¹*J*_{CF} = 273.4 Hz), 123.1 (C), 125.2 (CH, q, ³*J*_{CF} = 3.2 Hz), 126.3 (CH), 128.1 (CH, q, ³*J*_{CF} = 3.2 Hz), 134.8 (C, d, ²*J*_{CF} = 33.1 Hz), 143.9 (C), 154.8 (C); *m/z* (EI) 215.0301 (*M*⁺. C₈H₄F₃N₃O requires 215.0306), 172 (6%), 160 (8), 144 (9), 78 (81), 63 (100).

3*H*-Naphtho[2,3-*d*][1,2,3]triazin-4-one (7k). The reaction was carried out as described in the general procedure using 2-aminonaphthalene-3-carboxamide (**6k**) (0.0330 g, 0.177 mmol), polymer-supported nitrite (0.152 g, containing 0.532 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.101 g, 0.532 mmol). The reaction mixture was stirred at 0 °C for 1 h and heated to 40 °C for 2 h. Purification by flash column chromatography, eluting with 80% ethyl acetate in hexane with 1% triethylamine gave 3*H*-naphtho[2,3-*d*][1,2,3]triazin-4-one (**7k**) (0.0290, 83%) as a white solid. Mp 254–257 °C; $\nu_{\max}/\text{cm}^{-1}$ (neat) 3006 (NH), 2850 (CH), 2325, 2072, 1823, 1667 (C=O), 1203, 747; δ_{H} (400 MHz, DMSO-*d*₆) 7.73–7.86 (2H, m, 7-H and 8-H), 8.27–8.38 (2H, m, 6-H and 9-H), 8.84 (1H, s, 10-H), 8.91 (1H, s, 5-H), 14.64 (1H, s, NH); δ_{C} (101 MHz, DMSO-*d*₆) 117.7 (C), 125.8 (CH), 127.9 (CH), 128.9 (CH), 129.1 (CH), 129.3 (CH), 129.4 (CH), 133.7 (C), 135.7 (C), 140.5 (C), 155.8 (C); *m/z* (ESI) 198.0662 (*M*⁺. C₁₁H₈N₃O requires 198.0662).

***N*-Methyl-1,2,3-benzotriazin-4(3*H*)-one (7l).**¹⁶ The reaction was carried out as described in the general procedure using 2-amino-*N*-methylbenzamide (**6l**) (0.150 g, 1.00 mmol), polymer-supported nitrite (0.850 g, containing 3.00 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.571 g, 3.00 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for 5.5 h. Purification

by flash column chromatography, eluting with 25% ethyl acetate in petroleum ether gave *N*-methyl-1,2,3-benzotriazin-4(3*H*)-one (**7l**) (0.144 g, 89%) as a white solid. Mp 115–117 °C (lit.¹⁶ 118–120 °C); δ_{H} (500 MHz, DMSO-*d*₆) 3.93 (3H, s, 3-CH₃), 7.89–7.95 (1H, m, 6-H), 8.05–8.10 (1H, m, 7-H), 8.19 (1H, dd, *J* 8.1, 1.0 Hz, 8-H), 8.24 (1H, dd, *J* 8.1, 1.0 Hz, 5-H); δ_{C} (126 MHz, DMSO-*d*₆) 37.0 (CH₃), 119.2 (C), 124.3 (CH), 127.9 (CH), 132.8 (CH), 135.2 (CH), 143.9 (C), 155.1 (C); *m/z* (EI) 161 (*M*⁺. 6%), 78 (82), 63 (100).

***N*-Propyl-1,2,3-benzotriazin-4(3*H*)-one (7m).**³⁵ The reaction was carried out as described in the general procedure using 2-amino-*N*-propylbenzamide (**6m**) (0.150 g, 0.840 mmol), polymer-supported nitrite (0.721 g, containing 2.52 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.479 g, 2.52 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for 1 h. Purification by flash column chromatography, eluting with 20% ethyl acetate in hexane gave *N*-propyl-1,2,3-benzotriazin-4(3*H*)-one (**7m**) (0.136 g, 86%) as a white solid. Mp 50–55 °C (lit.³⁵ 56–57 °C); δ_{H} (500 MHz, DMSO-*d*₆) 0.92 (3H, t, *J* 7.3 Hz, 3'-H₃), 1.83 (2H, sextet, *J* 7.3 Hz, 2'-H₂), 4.33 (2H, t, *J* 7.3 Hz, 1'-H₂), 7.88–7.93 (1H, m, 6-H), 8.04–8.09 (1H, m, 7-H), 8.16 (1H, br d, *J* 8.0 Hz, 8-H), 8.22 (1H, dd, *J* 8.0, 1.0 Hz, 5-H); δ_{C} (126 MHz, DMSO-*d*₆) 10.9 (CH₃), 22.7 (CH₂), 50.7 (CH₂), 119.2 (C), 124.5 (CH), 127.9 (CH), 132.8 (CH), 135.3 (CH), 143.6 (C), 154.7 (C); *m/z* (ESI) 212 (*M*Na⁺. 100%).

***N*-Cyclohexyl-1,2,3-benzotriazin-4(3*H*)-one (7n).**^{13c} The reaction was carried out as described in the general procedure using 2-amino-*N*-cyclohexylbenzamide (**6n**) (0.150 g, 0.690 mmol), polymer-supported nitrite (0.590 g, containing 2.06 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.392 g, 2.06 mmol). The reaction mixture was stirred at room temperature for 1 h and then heated under reflux for 48 h. Purification by flash column chromatography, eluting with 25% ethyl acetate in hexane gave *N*-cyclohexyl-1,2,3-benzotriazin-4(3*H*)-one (**7n**) (0.0850 g, 54%) as a yellow solid. Mp 130–133 °C (lit.^{13c} 129–131 °C); δ_{H} (500 MHz, CD₃OD) 1.30–1.42 (1H, m, CHH), 1.49–1.62 (2H, m, CH₂), 1.76–1.84 (1H, m, CHH), 1.92–2.07 (6H, m, 3 × CH₂), 4.95–5.06 (1H, m, 1'-H), 7.89 (1H, td, *J* 8.1, 1.0 Hz, 6-H), 8.02–8.07 (1H, m, 7-H), 8.15 (1H, br d, *J* 8.1 Hz, 8-H), 8.31 (1H, dd, *J* 8.1, 1.0 Hz, 5-H); δ_{C} (126 MHz, CD₃OD) 26.5 (CH₂), 26.9 (2 × CH₂), 32.9 (2 × CH₂), 58.2 (CH), 120.6 (C), 126.0 (CH), 129.0 (CH), 133.7 (CH), 136.4 (CH), 145.2 (C), 156.6 (C); *m/z* (EI) 229 (*M*⁺. 68%), 172 (43), 158 (69), 148 (75), 130 (40), 105 (76), 78 (100), 63 (100).

***N*-(Benzyl)-1,2,3-benzotriazin-4(3*H*)-one (7o).**¹⁴ The reaction was carried out as described in the general procedure using 2-amino-*N*-(4-benzyl)benzamide (**6o**) (0.0500 g, 0.221 mmol), polymer-supported nitrite (0.190 g, containing 0.663 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.126 g, 0.663 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 4 h. Purification by flash column chromatography, eluting with 30% ethyl acetate in hexane gave *N*-(benzyl)-1,2,3-benzotriazin-4(3*H*)-one (**7o**) (0.0370 g, 71%) as a white solid. Mp 115–119 °C (lit.¹⁴ 118–120 °C); δ_{H} (400 MHz, CDCl₃) 5.63 (2H, s, PhCH₂), 7.26–7.37 (3H, m, 3'-H, 4'-H and



5'-H), 7.50–7.55 (2H, m, 2'-H and 6'-H), 7.74–7.80 (1H, m, 6-H), 7.89–7.95 (1H, m, 7-H), 8.14 (1H, br d, J 8.2 Hz, 8-H), 8.33 (1H, dd, J 8.0, 1.0 Hz, 5-H); δ_{C} (101 MHz, CDCl_3) 53.5 (CH_2), 120.2 (C), 125.3 (CH), 128.3 (CH), 128.5 (CH), 128.9 (2 \times CH), 129.0 (2 \times CH), 132.5 (CH), 134.9 (CH), 135.9 (C), 144.5 (C), 155.5 (C); m/z (ESI) 260 (MNa^+ , 100%).

***N*-(Methoxycarbonylmethyl)-1,2,3-benzotriazin-4(3H)-one (7p).**^{13c}

To a stirred solution of 2-amino-*N*-(methoxycarbonylmethyl)benzamide (**6p**) (0.127 g, 0.601 mmol) in methanol (6 mL) was added polymer-supported nitrite (0.516 g, containing 1.80 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.342 g, 1.80 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 4 h. The reaction mixture was cooled to room temperature and concentrated *in vacuo*. The reaction mixture was diluted in ethyl acetate (20 mL) and washed with water (4 \times 20 mL). The organic layer was dried (MgSO_4), filtered and concentrated *in vacuo* to give *N*-(methoxycarbonylmethyl)-1,2,3-benzotriazin-4(3H)-one (**7p**) (0.110 g, 84%) as a white solid. Mp 128–130 °C (lit.^{13c} 128–130 °C); δ_{H} (400 MHz, CDCl_3) 3.81 (3H, s, OCH_3), 5.21 (2H, s, $\text{CH}_2\text{CO}_2\text{CH}_3$), 7.80–7.86 (1H, m, 6-H), 7.95–8.02 (1H, m, 7-H), 8.19 (1H, br d, J 8.1 Hz, 8-H), 8.37 (1H, dd, J 7.9, 1.1 Hz, 5-H); δ_{C} (101 MHz, CDCl_3) 50.9 (CH_2), 52.8 (CH_3), 119.9 (C), 125.4 (CH), 128.8 (CH), 132.8 (CH), 135.3 (CH), 144.5 (C), 155.7 (C), 167.7 (C); m/z (ESI) 242 (MNa^+ , 100%).

***N*-Phenyl-1,2,3-benzotriazin-4(3H)-one (7q).**^{13c} The reaction was carried out as described in the general procedure using 2-amino-*N*-phenylbenzamide (**6q**) (0.0500 g, 0.236 mmol), polymer-supported nitrite (0.202 g, containing 0.707 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.135 g, 0.707 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 18 h. Purification by flash column chromatography, eluting with 25% ethyl acetate in hexane gave *N*-phenyl-1,2,3-benzotriazin-4(3H)-one (**7q**) (0.0330 g, 62%) as a white solid. Mp 150–153 °C (lit.^{13c} 150–152 °C); δ_{H} (400 MHz, CDCl_3) 7.49 (1H, tt, J 7.4, 1.3 Hz, 4'-H), 7.53–7.59 (2H, m, 3'-H and 5'-H), 7.63–7.68 (2H, m, 2'-H and 6'-H), 7.83–7.88 (1H, m, 6-H), 7.96–7.03 (1H, m, 7-H), 8.23 (1H, br d, J 8.1 Hz, 8-H), 8.45 (1H, dd, J 7.9, 1.1 Hz, 5-H); δ_{C} (101 MHz, CDCl_3) 120.6 (C), 125.8 (CH), 126.2 (2 \times CH), 128.7 (CH), 129.1 (CH), 129.2 (2 \times CH), 132.9 (CH), 135.2 (CH), 139.0 (C), 143.9 (C), 155.4 (C); m/z (ESI) 246 (MNa^+ , 100%).

***N*-(2'-Methylphenyl)-1,2,3-benzotriazin-4(3H)-one (7r).**^{13b}

The reaction was carried out as described in the general procedure using 2-amino-*N*-(2'-methylphenyl)benzamide (**6r**) (0.100 g, 0.442 mmol), polymer-supported nitrite (0.379 g, containing 1.33 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.252 g, 1.33 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 4 h. Purification by flash column chromatography, eluting with 100% dichloromethane gave *N*-(2'-methylphenyl)-1,2,3-benzotriazin-4(3H)-one (**7r**) (0.0810 g, 77%) as a white solid. Mp 158–160 °C (lit.^{13b} 153–155 °C); δ_{H} (500 MHz, CDCl_3) 2.21 (3H, s, 2'- CH_3), 7.33–7.49 (4H, m, 3'-H, 4'-H, 5'-H and 6'-H), 7.86 (1H, br t, J 7.5 Hz, 6-H), 8.01 (1H, br t, J 7.5, 7-H), 8.25 (1H, br d, J 7.5 Hz, 8-H), 8.45 (1H, br d, J 7.5 Hz, 5-H); δ_{C} (126 MHz, CDCl_3) 17.8

(CH_3), 120.4 (C), 125.6 (CH), 127.1 (CH), 127.8 (CH), 128.7 (CH), 129.9 (CH), 131.2 (CH), 132.8 (CH), 135.2 (CH), 135.6 (C), 138.0 (C), 144.1 (C), 155.2 (C); m/z (ESI) 260 (MNa^+ , 100%).

***N*-(4'-Methylphenyl)-1,2,3-benzotriazin-4(3H)-one (7s).**¹⁴ The reaction was carried out as described in the general procedure using 2-amino-*N*-(4'-methylphenyl)benzamide (**6s**) (0.285 g, 1.26 mmol), polymer-supported nitrite (1.09 g, containing 3.78 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.719 g, 3.78 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 4 h. The reaction mixture was cooled to room temperature and concentrated *in vacuo*. The reaction mixture was diluted in dichloromethane (20 mL) and washed with 1 M aqueous sodium hydroxide (6 \times 20 mL). The organic layer was dried (MgSO_4), filtered and concentrated *in vacuo* to give *N*-(4'-methylphenyl)-1,2,3-benzotriazin-4(3H)-one (**7s**) (0.272 g, 91%) as an orange solid. Mp 137–140 °C (lit.¹⁴ 139–141 °C); δ_{H} (400 MHz, CDCl_3) 2.45 (3H, s, 4'- CH_3), 7.33–7.39 (2H, m, 3'-H and 5'-H), 7.50–7.56 (2H, m, 2'-H and 6'-H), 7.82–7.87 (1H, m, 6-H), 7.96–8.01 (1H, m, 7-H), 8.22 (1H, br d, J 8.1 Hz, 8-H), 8.44 (1H, dd, J 7.9, 1.4 Hz, 5-H); δ_{C} (101 MHz, CDCl_3) 21.4 (CH_3), 120.6 (C), 125.8 (CH), 126.0 (2 \times CH), 128.6 (CH), 129.8 (2 \times CH), 132.8 (CH), 135.1 (CH), 136.5 (C), 139.2 (C), 143.9 (C), 155.5 (C); m/z (ESI) 260 (MNa^+ , 100%).

***N*-(4'-Methoxyphenyl)-1,2,3-benzotriazin-4(3H)-one (7t).**¹⁴ The reaction was carried out as described in the general procedure using 2-amino-*N*-(4'-methoxyphenyl)benzamide (**6t**) (0.0500 g, 0.206 mmol), polymer-supported nitrite (0.177 g, containing 0.619 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.118 g, 0.619 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 4 h. Purification by flash column chromatography, eluting with 70% diethyl ether in hexane gave *N*-(4'-methoxyphenyl)-1,2,3-benzotriazin-4(3H)-one (**7t**) (0.0370 g, 71%) as a yellow solid. Mp 147–150 °C (lit.¹⁴ 151–153 °C); δ_{H} (400 MHz, CDCl_3) 3.88 (3H, s, 4'- OCH_3), 7.02–7.09 (2H, m, 3'-H and 5'-H), 7.52–7.60 (2H, m, 2'-H and 6'-H), 7.84 (1H, td, J 8.0, 1.0 Hz, 6-H), 7.94–8.01 (1H, m, 7-H), 8.21 (1H, br d, J 8.1 Hz, 8-H), 8.43 (1H, dd, J 8.0, 1.0 Hz, 5-H); δ_{C} (101 MHz, CDCl_3) 55.7 (CH_3), 114.4 (2 \times CH), 120.5 (C), 125.7 (CH), 127.4 (2 \times CH), 128.6 (CH), 131.9 (C), 132.8 (CH), 135.1 (CH), 143.9 (C), 155.5 (C), 160.0 (C); m/z (ESI) 276 (MNa^+ , 100%).

***N*-(4'-Fluorophenyl)-1,2,3-benzotriazin-4(3H)-one (7u).**¹⁴ The reaction was carried out as described in the general procedure using 2-amino-*N*-(4'-fluorophenyl)benzamide (**6u**) (0.0500 g, 0.217 mmol), polymer-supported nitrite (0.187 g, containing 0.652 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.124 g, 0.652 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 4 h. Purification by flash column chromatography, eluting with 30% ethyl acetate in hexane gave *N*-(4'-fluorophenyl)-1,2,3-benzotriazin-4(3H)-one (**7u**) (0.0320 g, 60%) as a white solid. Mp 136–139 °C (lit.¹⁴ 138–140 °C); δ_{H} (400 MHz, CDCl_3) 7.20–7.28 (2H, m, 2'-H and 6'-H), 7.61–7.68 (2H, m, 3'-H and 5'-H), 7.82–7.89 (1H, m, 6-H), 7.96–8.01 (1H, m, 7-H), 8.22 (1H, br d, J 8.0 Hz, 8-H), 8.43 (1H, dd, J 8.0, 1.1 Hz, 5-H); δ_{C} (101 MHz, CDCl_3) 116.1 (2 \times CH, d, $^2J_{\text{C-F}}$ = 23.0 Hz), 120.4 (C), 125.8 (CH), 128.0 (2 \times CH, d, $^3J_{\text{C-F}}$ =



8.8 Hz), 128.7 (CH), 133.0 (CH), 134.9 (C, d, $^4J_{C-F}$ = 3.3 Hz), 135.3 (CH), 143.8 (C), 155.4 (C), 162.7 (C, d, $^1J_{C-F}$ = 249.1 Hz); m/z (ESI) 264 (MNa⁺, 100%).

***N*-(4'-Chlorophenyl)-1,2,3-benzotriazin-4(3*H*)-one (7v).^{13b}**

The reaction was carried out as described in the general procedure using 2-amino-*N*-(4'-chlorophenyl)benzamide (6v) (0.150 g, 0.608 mmol), polymer-supported nitrite (0.522 g, containing 1.82 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.347 g, 1.82 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 4 h. The reaction mixture was cooled to room temperature and concentrated *in vacuo*. The reaction mixture was diluted in ethyl acetate (20 mL) and washed with 1 M aqueous sodium hydroxide (6 × 20 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo* to give *N*-(4'-chlorophenyl)-1,2,3-benzotriazin-4(3*H*)-one (7v) (0.102 g, 65%) as a yellow solid. Mp 174–176 °C (lit.^{13b} 173–175 °C); δ_H (400 MHz, CDCl₃) 7.50–7.55 (2H, m, 2'-H and 6'-H), 7.61–7.67 (2H, m, 3'-H and 5'-H), 7.84–7.90 (1H, m, 6-H), 7.98–8.03 (1H, m, 7-H), 8.23 (1H, br d, J 8.2 Hz, 8-H), 8.44 (1H, dd, J 7.9, 1.2 Hz, 5-H); δ_C (101 MHz, CDCl₃) 120.4 (C), 125.8 (CH), 127.4 (2 × CH), 128.8 (CH), 129.4 (2 × CH), 133.1 (CH), 135.0 (C), 135.4 (CH), 137.4 (C), 143.7 (C), 155.3 (C); m/z (ESI) 280 (MNa⁺, 100%).

***N*-(4'-Iodophenyl)-1,2,3-benzotriazin-4(3*H*)-one (7w).¹⁴** The reaction was carried out as described in the general procedure using 2-amino-*N*-(4'-iodophenyl)benzamide (6w) (0.0900 g, 0.266 mmol), polymer-supported nitrite (0.229 g, containing 0.799 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.152 g, 0.799 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 18 h. Purification by flash column chromatography, eluting with 25% ethyl acetate in hexane gave *N*-(4'-iodophenyl)-1,2,3-benzotriazin-4(3*H*)-one (7w) (0.0740 g, 80%) as a white solid. Mp 187–190 °C (lit.¹⁴ 191–193 °C); δ_H (400 MHz, CDCl₃) 7.42–7.47 (2H, m, 2'-H and 6'-H), 7.83–7.92 (3H, m, 6-H, 3'-H and 5'-H), 7.97–8.04 (1H, m, 7-H), 8.23 (1H, br d, J 8.1 Hz, 8-H), 8.44 (1H, dd, J 7.9, 1.0 Hz, 5-H); δ_C (101 MHz, CDCl₃) 94.5 (C), 120.4 (C), 125.8 (CH), 127.8 (2 × CH), 128.8 (CH), 133.1 (CH), 135.4 (CH), 138.4 (2 × CH), 138.7 (C), 143.7 (C), 155.2 (C); m/z (ESI) 372 (MNa⁺, 100%).

***N*-(Thiazol-2'-yl)-1,2,3-benzotriazin-4(3*H*)-one (7x).^{13c}** The reaction was carried out as described in the general procedure using *N*-(2-aminobenzoyl)-2-aminothiazole (6x) (0.147 g, 0.670 mmol), polymer-supported nitrite (0.578 g, containing 2.01 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.387 g, 2.01 mmol). The reaction mixture was stirred at 0 °C for 1 h and then heated to 40 °C for 4 h. The reaction mixture was cooled to room temperature and concentrated *in vacuo*. The reaction mixture was diluted in ethyl acetate (20 mL) and washed with 1 M aqueous sodium hydroxide (6 × 20 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo* to give *N*-(thiazol-2'-yl)-1,2,3-benzotriazin-4(3*H*)-one (7x) (0.0800 g, 52%) as an orange solid. Mp 170–173 °C (lit.^{13c} 174–176 °C); δ_H (400 MHz, CDCl₃) 7.44 (1H, d, J 3.5 Hz, 5'-H), 7.89 (1H, d, J 3.5 Hz, 4'-H), 7.88–7.95 (1H, m, 6-H), 8.02–8.09 (1H, m, 7-H), 8.32 (1H, br d, J 8.1 Hz, 8-H), 8.51 (1H, dd, J 7.9, 1.4 Hz, 5-H); δ_C (101 MHz, CDCl₃) 119.1 (CH), 119.4 (C), 126.0

(CH), 129.5 (CH), 133.8 (CH), 135.9 (CH), 140.2 (CH), 142.6 (C), 154.3 (C), 156.3 (C); m/z (ESI) 253 (MNa⁺, 100%).

General procedure for synthesis of 2*H*-benzothiatriazin-1,1-dioxides

To a stirred solution of 2-aminobenzenesulfonamides (1 equiv.) in methanol (10 mL mmol⁻¹) at 0 °C was added polymer-supported nitrite (containing 3.0 equiv. of NO₂) and *p*-toluenesulfonic acid monohydrate (3 equiv.). The reaction mixture was stirred at 0 °C until completion or stirred at 0 °C for 1 h and then warmed to room temperature and stirred until completion. The resin was filtered and washed with methanol (20 mL mmol⁻¹). The reaction mixture was concentrated *in vacuo*. Purification by flash column chromatography gave the 2*H*-benzothiatriazin-1,1-dioxides.

***N*-Methyl-1,2,3,4-benzothiatriazin-1,1(2*H*)-dioxide (9a).¹⁶** To a stirred solution of 2-amino-*N*-(methyl)benzenesulfonamide (8a) (0.0840 g, 0.451 mmol) in methanol (6 mL) was added polymer-supported nitrite (0.387 g, containing 1.35 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.257 g, 1.35 mmol). The reaction mixture was stirred at 0 °C for 2 h. The reaction mixture was diluted with ethyl acetate (30 mL) and washed with water (4 × 30 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 50% diethyl ether in hexane gave *N*-methyl-1,2,3,4-benzothiatriazin-1,1(2*H*)-dioxide (9a) (0.0450 g, 51%) as a yellow solid. Mp 70–72 °C (lit.¹⁶ 68–72 °C); δ_H (400 MHz, CDCl₃) 3.91 (3H, s, NCH₃), 7.79 (1H, td, J 7.7, 1.2 Hz, 7-H), 7.87–7.92 (1H, m, 6-H), 8.02–8.07 (2H, m, 5-H and 8-H); δ_C (101 MHz, CDCl₃) 34.0 (CH₃), 120.6 (CH), 125.4 (C), 129.5 (CH), 132.8 (CH), 134.3 (CH), 141.8 (C); m/z (ESI) 220 (MNa⁺, 100%).

***N*-Ethyl-1,2,3,4-benzothiatriazin-1,1(2*H*)-dioxide (9b).³⁶** To a stirred solution of 2-amino-*N*-(ethyl)benzenesulfonamide (8b) (0.0500 g, 0.250 mmol) in methanol (2.5 mL) was added polymer-supported nitrite (0.215 g, containing 0.749 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.142 g, 0.749 mmol). The reaction mixture was stirred at 0 °C for 2 h. The reaction mixture was diluted with ethyl acetate (30 mL) and washed with water (4 × 30 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 25% ethyl acetate in hexane gave *N*-ethyl-1,2,3,4-benzothiatriazin-1,1(2*H*)-dioxide (9b) (0.0410 g, 75%) as a dark orange solid. Spectroscopic data were consistent with the literature.³⁶ Mp 75–78 °C; δ_H (400 MHz, CDCl₃) 1.59 (3H, t, J 7.2 Hz, 2'-H₃), 4.34 (2H, q, J 7.2 Hz, 1'-H₂), 7.77 (1H, td, J 7.8, 1.1 Hz, 6-H), 7.88 (1H, td, J 7.8, 1.4 Hz, 7-H), 8.01 (1H, dd, J 7.8, 1.4 Hz, 5-H), 8.03 (1H, dd, J 7.8, 1.1 Hz, 8-H); δ_C (101 MHz, CDCl₃) 16.0 (CH₃), 43.7 (CH₂), 120.4 (CH), 125.7 (C), 129.3 (CH), 132.6 (CH), 134.1 (CH), 141.8 (C); m/z (ESI) 234 (MNa⁺, 100%).

***N*-Propyl-1,2,3,4-benzothiatriazin-1,1(2*H*)-dioxide (9c).** To a stirred solution of 2-amino-*N*-(propyl)benzenesulfonamide (8c) (0.0650 g, 0.303 mmol) in methanol (3 mL) was added polymer-supported nitrite (0.261 g, containing 0.910 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.173 g,



0.910 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for a further 1 h. The reaction mixture was diluted with ethyl acetate (20 mL) and washed with water (4 × 30 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo* to give *N*-propyl-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9c**) (0.0560 g, 82%) as a red solid. Mp 83–86 °C; $\nu_{\text{max}}/\text{cm}^{-1}$ (neat) 2967 (CH), 1573 (C=C), 1471, 1446, 1329, 1313, 1185, 1160, 1111, 946, 766; δ_{H} (400 MHz, CDCl₃) 1.04 (3H, t, *J* 7.4 Hz, 3'-H₃), 2.01 (2H, sext, *J* 7.4 Hz, 2'-H₂), 4.23 (2H, t, *J* 7.4 Hz, 1'-H₂), 7.77 (1H, td, *J* 7.8, 1.1 Hz, 6-H), 7.88 (1H, td, *J* 7.8, 1.4 Hz, 7-H), 8.02 (1H, dd, *J* 7.8, 1.4 Hz, 5-H), 8.04 (1H, dd, *J* 7.8, 1.1 Hz, 8-H); δ_{C} (101 MHz, CDCl₃) 11.2 (CH₃), 23.7 (CH₂), 50.0 (CH₂), 120.5 (CH), 125.7 (C), 129.3 (CH), 132.6 (CH), 134.1 (CH), 141.8 (C); *m/z* (ESI) 248.0468 (MNa⁺. C₉H₁₁N₃NaO₂S requires 248.0464).

***N*-(*tert*-Butyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9d**).¹⁶**

To a stirred solution of 2-amino-*N*-(*tert*-butyl)benzenesulfonamide (**8d**) (0.0260 g, 0.114 mmol) in methanol (1.1 mL) was added polymer-supported nitrite (0.0980 g, containing 0.342 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.0650 g, 0.342 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for a further 1 h. The reaction mixture was diluted with ethyl acetate (10 mL) and washed with water (4 × 10 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 20% ethyl acetate in hexane gave *N*-(*tert*-butyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9d**) (0.0160 g, 60%) as an orange solid. Mp 78–82 °C (lit.¹⁶ 78–82 °C); δ_{H} (400 MHz, CDCl₃) 1.82 (9H, s, NC(CH₃)₃), 7.74 (1H, td, *J* 7.8, 1.2 Hz, 7-H), 7.85 (1H, td, *J* 7.8, 1.4 Hz, 6-H), 7.96–8.03 (2H, m, 5-H and 8-H); δ_{C} (101 MHz, CDCl₃) 30.5 (3 × CH₃), 67.9 (C), 120.4 (CH), 126.3 (C), 128.6 (CH), 132.3 (CH), 134.0 (CH), 141.2 (C); *m/z* (ESI) 262 (MNa⁺. 100%).

***N*-Benzyl-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9e**).³⁶** To a stirred solution of 2-amino-*N*-benzylbenzenesulfonamide (**8e**) (0.130 g, 0.496 mmol) in methanol (5 mL) was added polymer-supported nitrite (0.426 g, containing 1.49 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.283 g, 1.49 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for a further 1 h. The reaction mixture was diluted with ethyl acetate (20 mL) and washed with water (4 × 20 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo* to give *N*-benzyl-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9e**) (0.104 g, 76%) as an orange solid. Spectroscopic data were consistent with the literature.³⁶ Mp 99–102 °C; δ_{H} (400 MHz, CDCl₃) 5.42 (2H, s, PhCH₂), 7.30–7.38 (3H, m, 3'-H, 4'-H and 5'-H), 7.46–7.51 (2H, m, 2'-H and 6'-H), 7.77 (1H, td, *J* 7.7, 1.2 Hz, 6-H), 7.87 (1H, td, *J* 7.7, 1.4 Hz, 7-H), 8.01 (1H, dd, *J* 7.7, 1.4 Hz, 5-H), 8.04 (1H, dd, *J* 7.7, 1.2 Hz, 8-H); δ_{C} (101 MHz, CDCl₃) 51.3 (CH₂), 120.6 (CH), 126.1 (C), 128.6 (CH), 128.9 (2 × CH), 129.0 (2 × CH), 129.5 (CH), 132.8 (CH), 134.2 (CH), 135.4 (C), 141.8 (C); *m/z* (ESI) 296 (MNa⁺. 100%).

***N*-(4'-Methoxybenzyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9f**).³⁶** To a stirred solution of 2-amino-*N*-(4'-methoxybenzyl)

benzenesulfonamide (**8f**) (0.0700 g, 0.293 mmol) in methanol (2 mL) was added polymer-supported nitrite (0.206 g, containing 0.718 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.137 g, 0.718 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for a further 3.5 h. The reaction mixture was diluted with ethyl acetate (20 mL) and washed with water (4 × 20 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo* to give *N*-(4'-methoxybenzyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9f**) (0.0610 g, 85%) as a red solid. Spectroscopic data were consistent with the literature.³⁶ Mp 92–95 °C; δ_{H} (400 MHz, CDCl₃) 3.79 (3H, s, 4'-OCH₃), 5.36 (2H, s, NCH₂), 6.85–6.91 (2H, m, 3'-H and 5'-H), 7.40–7.46 (2H, m, 2'-H and 6'-H), 7.76 (1H, td, *J* 7.8, 1.2 Hz, 7-H), 7.86 (1H, td, *J* 7.8, 1.4 Hz, 6-H), 8.01 (1H, dd, *J* 7.8, 1.2 Hz, 5-H), 8.03 (1H, dd, *J* 7.8, 1.4 Hz, 8-H); δ_{C} (101 MHz, CDCl₃) 51.0 (CH₂), 55.4 (CH₃), 114.3 (2 × CH), 120.6 (CH), 126.1 (C), 127.5 (C), 129.5 (CH), 130.6 (2 × CH), 132.7 (CH), 134.2 (CH), 141.8 (C), 159.9 (C); *m/z* (ESI) 326 (MNa⁺. 100%).

***N*-Phenyl-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9g**).³⁷** To a stirred solution of 2-amino-*N*-(phenyl)benzenesulfonamide (**8g**) (0.0390 g, 0.157 mmol) in methanol (1.5 mL) was added polymer-supported nitrite (0.135 g, containing 0.471 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.0900 g, 0.471 mmol). The reaction mixture was stirred at 0 °C for 2 h. The reaction mixture was diluted with ethyl acetate (20 mL) and washed with water (4 × 20 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 50% diethyl ether in hexane gave *N*-phenyl-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9g**) (0.0290 g, 71%) as an orange solid. Mp 111–115 °C (lit.³⁷ 111 °C); δ_{H} (400 MHz, CDCl₃) 7.51–7.60 (3H, m, 3'-H, 4'-H and 5'-H), 7.62–7.69 (2H, m, 2'-H and 6'-H), 7.84 (1H, td, *J* 7.7, 1.1 Hz, 7-H), 7.94 (1H, td, *J* 7.7, 1.4 Hz, 6-H), 8.10–8.15 (2H, m, 5-H and 8-H); δ_{C} (101 MHz, CDCl₃) 121.1 (CH), 126.6 (C), 128.1 (2 × CH), 129.7 (2 × CH), 129.8 (CH), 130.1 (CH), 133.2 (CH), 134.4 (CH), 135.0 (C), 141.4 (C); *m/z* (ESI) 282 (MNa⁺. 100%).

***N*-(4'-Methylphenyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9h**).³⁶** To a stirred solution of 2-amino-*N*-(4'-methylphenyl)benzenesulfonamide (**8h**) (0.0650 g, 0.248 mmol) in methanol (2.5 mL) was added polymer-supported nitrite (0.213 g, containing 0.743 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.141 g, 0.743 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for a further 2.5 h. The reaction mixture was diluted with ethyl acetate (20 mL) and washed with water (4 × 20 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 20% ethyl acetate in hexane gave *N*-(4'-methylphenyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9h**) (0.0440 g, 65%) as an orange solid. Spectroscopic data were consistent with the literature.³⁶ Mp 100–102 °C; δ_{H} (400 MHz, CDCl₃) 2.45 (3H, s, 4'-CH₃), 7.35 (2H, br d, *J* 8.0 Hz, 3'-H and 5'-H), 7.50–7.55 (2H, m, 2'-H and 6'-H), 7.83 (1H, td, *J* 7.6, 1.2 Hz, 6-H), 7.93 (1H, td, *J* 7.6, 1.4 Hz, 7-H), 8.08–8.14 (2H, m, 5-H and 8-H); δ_{C}



(101 MHz, CDCl₃) 21.4 (CH₃), 121.1 (CH), 126.7 (C), 128.1 (2 × CH), 129.7 (CH), 130.3 (2 × CH), 132.3 (C), 133.1 (CH), 134.4 (CH), 140.5 (C), 141.5 (C); *m/z* (ESI) 296 (MNa⁺, 100%).

***N*-(4'-Methoxyphenyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (9i).**³⁸ To a stirred solution of 2-amino-*N*-(4'-methoxyphenyl) benzenesulfonamide (**8i**) (0.0550 g, 0.197 mmol) in methanol (2 mL) was added polymer-supported nitrite (0.170 g, containing 0.592 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.113 g, 0.592 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for a further 1.3 h. The reaction mixture was diluted with ethyl acetate (20 mL) and washed with water (4 × 20 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 50% diethyl ether in hexane gave *N*-(4'-methoxyphenyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9i**) (0.0390 g, 68%) as a yellow solid. Mp 77–82 °C (lit.³⁸ 83–84 °C); δ_H (400 MHz, CDCl₃) 3.88 (3H, s, 4'-OCH₃), 7.05 (2H, br d, *J* 8.8 Hz, 3'-H and 5'-H), 7.56 (2H, br d, *J* 8.8 Hz, 2'-H and 6'-H), 7.80–7.87 (1H, m, 6-H), 7.90–7.97 (1H, m, 7-H), 8.08–8.17 (2H, m, 5-H and 8-H); δ_C (101 MHz, CDCl₃) 55.8 (CH₃), 114.9 (2 × CH), 121.1 (CH), 126.5 (C), 127.1 (C), 129.7 (CH), 130.0 (2 × CH), 133.1 (CH), 134.4 (CH), 141.4 (C), 161.1 (C); *m/z* (ESI) 284 (MNa⁺ – N₂, 100%).

***N*-(4'-Bromophenyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (9j).** To a stirred solution of 2-amino-*N*-(4'-bromophenyl) benzenesulfonamide (**8j**) (0.100 g, 0.306 mmol) in methanol (3.1 mL) was added polymer-supported nitrite (0.263 g, containing 0.917 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.174 g, 0.917 mmol). The reaction mixture was stirred at 0 °C for 2 h. The reaction mixture was diluted with ethyl acetate (20 mL) and washed with water (4 × 20 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 40% diethyl ether in hexane gave *N*-(4'-bromophenyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9j**) (0.0700 g, 68%) as an orange solid. Mp 107–112 °C; ν_{max}/cm⁻¹ (neat) 2981 (CH), 1572 (C=C), 1471, 1448, 1337, 1168, 904, 759; δ_H (400 MHz, CDCl₃) 7.50–7.56 (2H, m, 2'-H and 6'-H), 7.65–7.71 (2H, m, 3'-H and 5'-H), 7.85 (1H, td, *J* 7.7, 1.2 Hz, 6-H), 7.95 (1H, td, *J* 7.7, 1.3 Hz, 7-H), 8.12 (1H, dd, *J* 7.7, 1.3 Hz, 5-H), 8.13 (1H, dd, *J* 7.7, 1.2 Hz, 8-H); δ_C (101 MHz, CDCl₃) 121.1 (CH), 124.4 (C), 126.6 (C), 129.5 (2 × CH), 129.9 (CH), 132.9 (2 × CH), 133.4 (CH), 134.1 (C), 134.6 (CH), 141.3 (C); *m/z* (ESI) 359.9408 (MNa⁺, C₁₂H₈⁷⁹BrN₃NaO₂S requires 359.9413).

***N*-(4'-Iodophenyl)-1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (9k).** To a stirred solution of 2-amino-*N*-(4'-iodophenyl) benzenesulfonamide (**8k**) (0.119 g, 0.318 mmol) in methanol (3.2 mL) was added polymer-supported nitrite (0.273 g, containing 0.954 mmol of NO₂) and *p*-toluenesulfonic acid monohydrate (0.182 g, 0.954 mmol). The reaction mixture was stirred at 0 °C for 2 h. The reaction mixture was diluted with ethyl acetate (30 mL) and washed with water (4 × 30 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 50% diethyl ether in hexane gave *N*-(4'-iodophenyl)-

1,2,3,4-benzothiaziazin-1,1(2*H*)-dioxide (**9k**) (0.0720 g, 59%) as an orange solid. Mp 115–118 °C; ν_{max}/cm⁻¹ (neat) 1570 (C=C), 1472, 1454, 1338, 1182, 1159, 1128, 808, 762; δ_H (400 MHz, CDCl₃) 7.37–7.41 (2H, m, 2'-H and 6'-H), 7.83–7.91 (3H, m, 6-H, 3'-H and 5'-H), 7.96 (1H, td, *J* 7.8, 1.4 Hz, 7-H), 8.12 (1H, dd, *J* 7.8, 1.4 Hz, 5-H), 8.14 (1H, dd, *J* 7.8, 1.2 Hz, 8-H); δ_C (101 MHz, CDCl₃) 96.0 (C), 121.1 (CH), 126.6 (C), 129.6 (2 × CH), 130.0 (CH), 133.4 (CH), 134.6 (CH), 134.8 (C), 138.9 (2 × CH), 141.3 (C); *m/z* (ESI) 407.9274 (MNa⁺, C₁₂H₈IN₃NaO₂S requires 407.9274).

***N*-Naphthyl-1,2,3-benzotriazin-4(3*H*)-one (10).**²⁶ To a stirred solution of 1,2,3-benzotriazin-4(3*H*)-one (**7a**) (0.100 g, 0.680 mmol), 2-naphthaleneboronic acid (0.175 g, 1.02 mmol) and copper acetate (0.123 g, 0.680 mmol) in dichloroethane (7 mL) was added triethylamine (188 μL, 1.36 mmol). The reaction was stirred at room temperature for 3 h and then heated to 45 °C for a further 2.5 h. The reaction mixture was cooled to room temperature and filtered through a pad of Celite®. The reaction mixture was concentrated *in vacuo*. Purification by flash column chromatography, eluting with 60% diethyl ether in hexane gave *N*-naphthyl-1,2,3-benzotriazin-4(3*H*)-one (**10**) (0.118 g, 64%) as a yellow solid. Mp 169–173 °C (lit.²⁶ 173–175 °C); δ_H (400 MHz, CDCl₃) 7.53–7.61 (2H, m, 4'-H and 5'-H), 7.76 (1H, dd, *J* 8.8, 2.1 Hz, 8'-H), 7.85–7.90 (1H, m, 6-H), 7.91–7.97 (2H, m, 3'-H and 6'-H), 7.98–8.04 (2H, m, 7-H and 7'-H), 8.18 (1H, d, *J* 2.1 Hz, 2'-H), 8.26 (1H, br d, *J* 8.1 Hz 8-H), 8.48 (1H, dd, *J* 8.0, 1.5 Hz, 5-H); δ_C (101 MHz, CDCl₃) 120.6 (C), 123.8 (CH), 125.1 (CH), 125.8 (CH), 126.9 (CH), 127.2 (CH), 127.9 (CH), 128.6 (CH), 128.7 (CH), 129.1 (CH), 133.0 (CH), 133.2 (C), 133.3 (C), 135.3 (CH), 136.4 (C), 143.9 (C), 155.6 (C); *m/z* (ESI) 296 (MNa⁺, 100%).

***N*-(Ethoxycarbonylmethyl)-1,2,3-benzotriazin-4(3*H*)-one (11).**^{2c} To a stirred solution of 1,2,3-benzotriazin-4(3*H*)-one (**7a**) (0.150 g, 1.02 mmol) in acetonitrile (10 mL) was added ethyl bromoacetate (0.113 mL, 1.02 mmol) and potassium carbonate (0.141 g, 1.02 mmol). The reaction mixture was heated under reflux for 18 h. The reaction mixture was cooled to room temperature, diluted in chloroform (30 mL) and washed with water (3 × 30 mL). The organic layer was dried (MgSO₄), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 25% ethyl acetate in hexane gave *N*-(ethoxycarbonylmethyl)-1,2,3-benzotriazin-4(3*H*)-one (**11**) (0.140 g, 59%) as an off-white solid. Mp 107–111 °C (lit.^{2c} 114–116 °C); δ_H (400 MHz, CDCl₃) 1.31 (3H, t, *J* 7.1 Hz, OCH₂CH₃), 4.28 (2H, q, *J* 7.1 Hz, OCH₂CH₃), 5.19 (2H, s, CH₂CO₂Et), 7.83 (1H, ddd, *J* 8.8, 7.9, 1.2 Hz, 6-H), 7.98 (1H, ddd, *J* 8.8, 8.2, 1.5 Hz, 7-H), 8.19 (1H, ddd, *J* 8.2, 1.2, 0.6 Hz, 8-H), 8.37 (1H, ddd, *J* 7.9, 1.5, 0.6 Hz, 5-H); δ_C (101 MHz, CDCl₃) 14.3 (CH₃), 51.0 (CH₂), 62.2 (CH₂), 119.9 (C), 125.4 (CH), 128.7 (CH), 132.8 (CH), 135.3 (CH), 144.5 (C), 155.7 (C), 167.2 (C); *m/z* (ESI) 256 (MNa⁺, 100%).

***N*-(2'-Diethylaminoethyl)acetamide-1,2,3-benzotriazin-4(3*H*)-one (3).**^{2c} To a stirred solution of *N*-(ethoxycarbonylmethyl)-1,2,3-benzotriazin-4(3*H*)-one (**11**) (0.100 g, 0.430 mmol) in anhydrous methanol (4 mL) was added dropwise *N,N*-diethylethylenediamine (0.0610 mL, 0.430 mmol). The reaction



mixture was heated under reflux for 18 h. The reaction mixture was cooled to room temperature and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 10% methanol in dichloromethane gave *N*-(2'-diethyl-aminoethyl)acetamide-1,2,3-benzotriazin-4(3*H*)-one (3) (0.0940 g, 72%) as a white solid. Mp 147–150 °C (lit.^{2c} 155–156 °C); δ_{H} (400 MHz, CDCl_3) 0.97 (6H, t, *J* 7.1 Hz, $2 \times \text{NCH}_2\text{CH}_3$), 2.55 (4H, q, *J* 7.1 Hz, $2 \times \text{NCH}_2\text{CH}_3$), 2.61 (2H, t, *J* 5.7 Hz, Et_2NCH_2), 3.39 (2H, q, *J* 5.7 Hz, CONHCH_2), 5.13 (2H, s, CH_2CONH), 6.85 (1H, br s, NH), 7.79–7.85 (1H, m, 6-H), 7.94–8.00 (1H, m, 7-H), 8.19 (1H, dd, *J* 8.2, 0.5 Hz, 8-H), 8.36 (1H, dd, *J* 7.9, 1.5 Hz, 5-H); δ_{C} (101 MHz, CDCl_3) 11.4 ($2 \times \text{CH}_3$), 36.9 (CH_2), 46.9 ($2 \times \text{CH}_2$), 51.5 (CH_2), 52.8 (CH_2), 120.0 (C), 125.3 (CH), 128.7 (CH), 132.7 (CH), 135.2 (CH), 144.5 (C), 155.8 (C), 166.1 (C); *m/z* (ESI) 304 (MNa^+ , 100%).

***N*-[2'-(4"-Benzoylpiperazin-1-yl)ethyl]-2-nitrobenzenesulfonamide (13).** 2-Nitrobenzenesulfonyl chloride (0.0560 g, 0.253 mmol) was dissolved in dichloromethane (0.2 mL) and added dropwise to a stirred solution of 1-benzoyl-4-[2'-aminoethyl]piperazine (12) (0.0550 g, 0.236 mmol) and triethylamine (0.0340 mL, 0.248 mmol) in dichloromethane (0.4 mL) at 0 °C. The reaction mixture was warmed to room temperature and stirred for 2 h. The reaction mixture was concentrated *in vacuo*. Purification by flash column chromatography, eluting with 1% methanol in dichloromethane gave *N*-[2'-(4"-benzoylpiperazin-1-yl)ethyl]-2-nitrobenzenesulfonamide (13) (0.0880 g, 83%) as a colourless oil. $\nu_{\text{max}}/\text{cm}^{-1}$ (neat) 3335 (NH), 2941 (CH), 1627 (C=O), 1576 (C=C), 1539, 1437, 1364, 1166, 1012, 759; δ_{H} (500 MHz, CD_3OD) 2.39 (2H, br s, 2"-H₂), 2.49 (2H, br s, 6"-H₂), 2.54 (2H, br t, *J* 6.1 Hz, 2'-H₂), 3.21 (2H, br t, *J* 6.1 Hz, 1'-H₂), 3.37 (2H, br s, 3"-H₂), 3.68 (2H, br s, 5"-H₂), 7.35–7.51 (5H, m, Ph), 7.79–7.86 (2H, m, 4-H and 6-H), 7.90 (1H, br t, *J* 5.3 Hz, 5-H), 8.10–8.15 (1H, m, 3-H); δ_{C} (126 MHz, CD_3OD) 41.1 (CH_2), 43.0 (CH_2), 53.4 (CH_2), 54.0 (CH_2), 57.5 ($2 \times \text{CH}_2$), 126.1 (CH), 128.0 ($2 \times \text{CH}$), 129.7 ($2 \times \text{CH}$), 131.2 (CH), 131.7 (CH), 133.7 (CH), 134.7 (C), 135.1 (CH), 136.6 (C), 149.6 (C), 172.4 (C); *m/z* (ESI) 441.1211 (MNa^+ , $\text{C}_{19}\text{H}_{22}\text{N}_4\text{NaO}_5\text{S}$ requires 441.1203).

***N*-[2'-(4"-Benzoylpiperazin-1-yl)ethyl]-2-aminobenzenesulfonamide (14).** To a stirred solution of *N*-[2'-(4"-benzoylpiperazin-1-yl)ethyl]-2-nitrobenzenesulfonamide (13) (0.0830 g, 0.198 mmol) in methanol (2 mL) was added zinc powder (0.129 g, 1.98 mmol) and acetic acid (0.113 mL, 1.98 mmol). The reaction was stirred at room temperature for 5 h. The reaction mixture was filtered through a pad of Celite® and washed with methanol (50 mL). The reaction mixture was concentrated *in vacuo*. Purification by flash column chromatography, eluting with 5% methanol in dichloromethane gave *N*-[2'-(4"-benzoylpiperazin-1-yl)ethyl]-2-aminobenzenesulfonamide (14) (0.0650 g, 84%) as a colourless solid. Mp 65–70 °C; $\nu_{\text{max}}/\text{cm}^{-1}$ (neat) 3373 (NH), 2944 (CH), 1617 (C=O), 1575 (C=C), 1454, 1305, 1146, 1012, 711; δ_{H} (500 MHz, CD_3OD) 2.29 (2H, br s, 2"-H₂), 2.35–2.47 (4H, m, 2'-H₂ and 6"-H₂), 2.96 (2H, br t, *J* 6.4 Hz, 1'-H₂), 3.39 (2H, br s, 3"-H₂), 3.71 (2H, br s, 5"-H₂), 6.70 (1H, br t, *J* 8.1 Hz, 5-H), 6.84 (1H, br d, *J* 8.1 Hz, 3-H), 7.29 (1H, td, *J* 8.1, 1.2 Hz, 4-H), 7.35–7.51 (5H, m, Ph), 7.62 (1H, dd, *J* 8.1, 1.2 Hz, 6-H); δ_{C} (126 MHz, CD_3OD) 40.6 (CH_2), 43.1 (CH_2), 53.4

(CH_2), 53.9 (CH_2), 57.3 ($2 \times \text{CH}_2$), 117.2 (CH), 118.3 (CH), 121.7 (C), 128.0 ($2 \times \text{CH}$), 129.7 ($2 \times \text{CH}$), 130.6 (CH), 131.1 (CH), 135.0 (CH), 136.7 (C), 147.7 (C), 172.3 (C); *m/z* (ESI) 411.1463 (MNa^+ , $\text{C}_{19}\text{H}_{24}\text{N}_4\text{NaO}_3\text{S}$ requires 411.1461).

***N*-[2'-(4"-Benzoylpiperazin-1-yl)ethyl]-1,2,3,4-benzothiaziazine-1,1(2*H*)-dioxide (5).** To a stirred solution of *N*-[2'-(4"-benzoylpiperazin-1-yl)ethyl]-2-aminobenzenesulfonamide (14) (0.0760 g, 0.196 mmol) in methanol (2 mL) was added polymer-supported nitrite (0.164 g, containing 0.587 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.112 g, 0.587 mmol). The reaction mixture was stirred at 0 °C for 1 h, warmed to room temperature and stirred for a further 1 h. The reaction mixture was filtered and the resulting resin was washed with methanol (10 mL). The reaction mixture was concentrated *in vacuo*. Purification by flash column chromatography, eluting with 60% ethyl acetate in hexane (with 1% triethylamine) gave *N*-[2'-(4"-benzoylpiperazin-1-yl)ethyl]-1,2,3,4-benzothiaziazine-1,1(2*H*)-dioxide (5) (0.0560 g, 72%) as a yellow solid. Mp 60–65 °C; $\nu_{\text{max}}/\text{cm}^{-1}$ (neat) 2939 (CH), 1630 (C=O), 1575 (C=C), 1433, 1337, 1188, 1013, 761; δ_{H} (500 MHz, CD_3OD) 2.53 (2H, br s, 2"-H₂), 2.66 (2H, br s, 6"-H₂), 2.93 (2H, t, *J* 6.5 Hz, 2'-H₂), 3.41 (2H, br s, 3"-H₂), 3.74 (2H, br s, 5"-H₂), 4.41 (2H, t, *J* 6.5 Hz, 1'-H₂), 7.36–7.50 (5H, m, Ph), 7.91 (1H, td, *J* 7.7, 1.4 Hz, 7-H), 8.01 (1H, td, *J* 7.7, 1.2 Hz, 6-H), 8.06–8.11 (2H, m, 5-H and 8-H); δ_{C} (126 MHz, CD_3OD) 43.2 (CH_2), 46.2 (CH_2), 54.0 (CH_2), 54.2 (CH_2), 58.0 ($2 \times \text{CH}_2$), 121.2 (CH), 127.3 (C), 128.0 ($2 \times \text{CH}$), 129.7 ($2 \times \text{CH}$), 130.3 (CH), 131.1 (CH), 134.4 (CH), 135.6 (CH), 136.8 (C), 142.9 (C), 172.4 (C); *m/z* (ESI) 422.1253 (MNa^+ , $\text{C}_{19}\text{H}_{21}\text{N}_5\text{NaO}_3\text{S}$ requires 422.1257).

Methyl (2*S*)-2-(*tert*-butoxycarbonylamino)-3-(2'-aminobenzoyl)aminopropanoate (16). To a stirred solution of methyl (2*S*)-2-(*tert*-butoxycarbonylamino)-3-aminopropanoate (15) (0.0900 g, 0.412 mmol) in ethyl acetate (1.5 mL) was added isatoic anhydride (0.0740 g, 0.454 mmol). The reaction was heated to 90 °C and stirred for 18 h. After cooling to room temperature, the reaction mixture was concentrated *in vacuo*. Purification by flash column chromatography, eluting with 40% ethyl acetate in hexane with 1% triethylamine gave methyl (2*S*)-2-(*tert*-butoxycarbonylamino)-3-(2'-aminobenzoyl)aminopropanoate (16) (0.0664 g, 48%) as a white solid. Mp 120–124 °C; $\nu_{\text{max}}/\text{cm}^{-1}$ (neat) 3360 (NH), 2976 (CH), 2928 (CH), 1741 (C=O), 1705 (C=O), 1522, 1368, 1256, 1161; $[\alpha]_{\text{D}}^{19} +25.6$ (c 0.3, CHCl_3); δ_{H} (400 MHz, CDCl_3) 1.43 (9H, s, $3 \times \text{CH}_3$), 3.67–3.85 (5H, m, OCH_3 and 3-H₂), 4.45–4.55 (1H, m, 2-H), 5.51 (2H, br s, 2'-NH₂), 5.60 (1H, br d, *J* 4.8 Hz, 2-NH), 6.58–6.69 (2H, m, 3'-H and 5'-H), 6.78 (1H, br s, 3-NH), 7.19 (1H, br t, *J* 7.8 Hz, 4'-H), 7.33 (1H, br d, *J* 7.8 Hz, 6'-H); δ_{C} (101 MHz, CDCl_3) 28.4 ($3 \times \text{CH}_3$), 42.5 (CH_2), 52.9 (CH), 53.9 (CH_3), 80.6 (C), 115.5 (C), 116.8 (CH), 117.4 (CH), 127.5 (CH), 132.7 (CH), 148.8 (C), 156.1 (C), 169.9 (C), 171.3 (C); *m/z* (ESI) 360.1532 (MNa^+ , $\text{C}_{16}\text{H}_{23}\text{N}_3\text{NaO}_5$ requires 360.1530).

Methyl (2*S*)-2-(*tert*-butoxycarbonylamino)-3-[1',2',3'-benzotriazin-4'(3*H*)-one]propanoate (17). To a stirred solution of methyl (2*S*)-2-(*tert*-butoxycarbonylamino)-3-(2'-aminobenzoyl)aminopropanoate (16) (0.100 g, 0.297 mmol) in methanol (3 mL) at 0 °C was added polymer-supported nitrite (0.256 g,



containing 0.890 mmol of NO_2) and *p*-toluenesulfonic acid monohydrate (0.169 g, 0.890 mmol). The reaction mixture was stirred at 0 °C for 1 h and then 2 h at room temperature. The reaction mixture was filtered and the resulting resin was washed with methanol (10 mL). The reaction mixture was concentrated *in vacuo*. The reaction mixture was diluted in ethyl acetate (20 mL) and washed with water (4 × 20 mL). The organic layer was dried (MgSO_4), filtered and concentrated *in vacuo*. Purification by flash column chromatography, eluting with 60% diethyl ether in hexane gave methyl (2*S*)-2-(*tert*-butoxycarbonylamino)-3-[1',2',3'-benzotriazin-4'(3*H*)-one]propanoate (**20**) (0.0900 g, 87%) as a colourless oil. $\nu_{\text{max}}/\text{cm}^{-1}$ (neat) 3333 (NH), 2976 (CH), 2928 (CH), 1746 (C=O), 1713 (C=O), 1688 (C=O), 1508, 1368, 1302, 1163; $[\alpha]_{\text{D}}^{18}$ -18.2 (*c* 0.2, CHCl_3); δ_{H} (400 MHz, CDCl_3) 1.32 (9H, s, 3 × CH_3), 3.80 (3H, s, OCH_3), 4.72 (1H, dd, *J* 12.8, 6.8 Hz, 3-*HH*), 4.85–5.00 (2H, m, 2-H and 3-*HH*), 5.43 (1H, br d, *J* 6.4 Hz, 2-NH), 7.81 (1H, br t, *J* 8.0 Hz, 6'-H), 7.95 (1H, td, *J* 8.0, 0.9 Hz, 7'-H), 8.14 (1H, br d, *J* 8.0 Hz, 8'-H), 8.35 (1H, dd, *J* 8.0, 0.9 Hz, 5'-H); δ_{C} (101 MHz, CDCl_3) 28.3 (3 × CH_3), 50.7 (CH_2), 52.9 (CH), 53.0 (CH_3), 80.4 (C), 119.8 (C), 125.4 (CH), 128.5 (CH), 132.7 (CH), 135.1 (CH), 144.2 (C), 155.2 (C), 156.2 (C), 170.5 (C); *m/z* (ESI) 371.1323 (MNa^+ , $\text{C}_{16}\text{H}_{20}\text{N}_4\text{NaO}_5$ requires 371.1326).

(2*S*)-2-(*tert*-Butoxycarbonylamino)-3-[1',2',3'-benzotriazin-4'(3*H*)-one]propanoic acid. To a stirred solution of methyl (2*S*)-2-(*tert*-butoxycarbonylamino)-3-[1',2',3'-benzotriazin-4'(3*H*)-one]propanoate (**17**) (0.0700 g, 0.201 mmol) in methanol (3.5 mL), dioxane (1.75 mL) and water (1.75 mL) was added caesium carbonate (0.0850 g, 0.261 mmol). The reaction mixture was stirred at room temperature for 18 h. The reaction mixture was concentrated *in vacuo*, diluted in water (30 mL) and acidified to pH 1 using 1 M aqueous hydrochloric acid. The reaction mixture was extracted with dichloromethane (3 × 30 mL). The organic layers were combined, dried (MgSO_4), filtered and concentrated *in vacuo* to give (2*S*)-2-(*tert*-butoxycarbonylamino)-3-[1',2',3'-benzotriazin-4'(3*H*)-one]propanoic acid (0.0660 g, 99%) as a colourless oil. $\nu_{\text{max}}/\text{cm}^{-1}$ (neat) 3351 (NH), 2973 (CH), 1689 (C=O), 1394, 1367, 1301, 1164, 779; $[\alpha]_{\text{D}}^{19}$ -52.5 (*c* 0.2, MeOH); δ_{H} (400 MHz, CD_3OD) 1.23 (9H, s, 3 × CH_3), 4.62 (1H, dd, *J* 13.2, 9.6 Hz, 3-*HH*), 4.79 (1H, dd, *J* 9.6, 4.4 Hz, 2-H), 5.02 (1H, dd, *J* 13.2, 4.4 Hz, 3-*HH*), 7.89 (1H, br t, *J* 7.9 Hz, 6'-H), 8.04 (1H, br t, *J* 7.9 Hz, 7'-H), 8.13 (1H, br d, *J* 7.9 Hz, 8'-H), 8.33 (1H, br d, *J* 7.9 Hz, 5'-H); δ_{C} (101 MHz, CD_3OD) 28.5 (3 × CH_3), 52.3 (CH_2), 54.8 (CH), 80.6 (C), 120.8 (C), 125.9 (CH), 129.1 (CH), 133.8 (CH), 136.4 (CH), 145.5 (C), 157.5 (C), 157.6 (C), 173.0 (C); *m/z* (ESI) 357.1164 (MNa^+ , $\text{C}_{15}\text{H}_{18}\text{N}_4\text{NaO}_5$ requires 357.1169).

(2*S*)-2-Amino-3-[1',2',3'-benzotriazin-4'(3*H*)-one]propanoic acid hydrochloride (18**).** A solution of (2*S*)-2-(*tert*-butoxycarbonylamino)-3-[1',2',3'-benzotriazin-4'(3*H*)-one]propanoic acid (0.040 g, 0.12 mmol) in 2 M aqueous hydrochloric acid (2 mL) was stirred at 50 °C for 3 h. The reaction mixture was cooled to room temperature and concentrated *in vacuo* to give (2*S*)-2-amino-3-[1',2',3'-benzotriazin-4'(3*H*)-one]propanoic acid hydrochloride (**18**) (0.026 g, 78%) as a white solid. Mp 175–180 °C; $\nu_{\text{max}}/\text{cm}^{-1}$ (neat) 3372 (NH), 3220 (CH), 1679 (C=O), 1651,

1122, 784; $[\alpha]_{\text{D}}^{19}$ -5.4 (*c* 0.2, MeOH); δ_{H} (400 MHz, CD_3OD) 4.63 (1H, dd, *J* 7.2, 4.4 Hz, 2-H), 4.94 (1H, dd, *J* 14.6, 7.2 Hz, 3-*HH*), 5.11 (1H, dd, *J* 14.6, 4.4 Hz, 3-*HH*), 7.95 (1H, td, *J* 8.1, 1.2 Hz, 6'-H), 8.10 (1H, td, *J* 8.1, 1.2 Hz, 7'-H), 8.21 (1H, br d, *J* 8.1 Hz, 8'-H), 8.35 (1H, dd, *J* 8.1, 1.2 Hz, 5'-H); δ_{C} (101 MHz, CD_3OD) 49.7 (CH_2), 53.3 (CH), 120.9 (C), 126.0 (CH), 129.6 (CH), 134.4 (CH), 136.8 (CH), 145.4 (C), 157.8 (C), 169.1 (C); *m/z* (ESI) 257.0643 ($[\text{MNa} - \text{HCl}]^+$, $\text{C}_{10}\text{H}_{10}\text{N}_4\text{NaO}_3$ requires 257.0645).

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- 1 Z. Khalid, H. A. Ahmad, M. A. Munawar, M. A. Khan and S. Gul, *Heterocycles*, 2017, **94**, 3.
- 2 For example, see: (a) S. M. Gadekar and J. L. Frederick, *J. Org. Chem.*, 1962, **27**, 1383; (b) A. S. Clark, B. Deans, M. F. G. Stevens, M. J. Tisdale, R. T. Wheelhouse, B. J. Denny and J. A. Hartley, *J. Med. Chem.*, 1995, **38**, 1493; (c) G. Caliendo, F. Fiorino, P. Grieco, E. Perissutti, V. Santagada, R. Meli, G. M. Raso, A. Zanesco and G. De Nucci, *Eur. J. Med. Chem.*, 1999, **34**, 1043; (d) F. Fiorino, B. Severino, F. De Angelis, E. Perissutti, F. Frecentese, P. Massarelli, G. Bruni, E. Collavoli, V. Santagada and G. Calliendo, *Arch. Pharm. Chem. Life Sci.*, 2008, **341**, 20; (e) X. Xu, Z. Li, X. Chen, W. Li, H. Jia, X. Cao, X. Shao and Z. Xu, CN108276352, 2018.
- 3 (a) T. Miura, M. Yamauchi and M. Murakami, *Org. Lett.*, 2008, **10**, 3085; (b) M. Yamauchi, M. Morimoto, T. Miura and M. Murakami, *J. Am. Chem. Soc.*, 2010, **132**, 54; (c) T. Miura, M. Morimoto, M. Yamauchi and M. Murakami, *J. Org. Chem.*, 2010, **75**, 5359.
- 4 (a) Z.-J. Fang, S.-C. Zheng, Z. Guo, J.-Y. Guo, B. Tan and X.-Y. Liu, *Angew. Chem., Int. Ed.*, 2015, **54**, 9528; (b) N. Wang, S.-C. Zheng, L.-L. Zhang, Z. Guo and X.-Y. Liu, *ACS Catal.*, 2016, **6**, 3496.
- 5 V. H. Thorat, N. S. Upadhyay, M. Murakami and C.-H. Cheng, *Adv. Synth. Catal.*, 2018, **360**, 284.
- 6 (a) T. Miura, Y. Nishida, M. Morimoto, M. Yamauchi and M. Murakami, *Org. Lett.*, 2011, **13**, 1429; (b) M. H. Balakrishnan and S. Mannathan, *Org. Lett.*, 2020, **22**, 542.
- 7 M. H. Balakrishnan, K. Sathriyan and S. Mannathan, *Org. Lett.*, 2018, **20**, 3815.
- 8 H. Wang and S. Yu, *Org. Lett.*, 2015, **17**, 4272.
- 9 M. S. Ao and E. M. Burgess, *J. Am. Chem. Soc.*, 1971, **93**, 5298.



- 10 (a) G. Ege and E. Beisiegel, *Liebigs Ann. Chem.*, 1972, **763**, 46; (b) Y.-Y. Han, H. Wang and S. Yu, *Org. Chem. Front.*, 2016, **3**, 953.
- 11 Y. Xie, G. Gong, Y. Liu, S. Deng, A. Rinderspacher, L. Branden and D. W. Landry, *Tetrahedron Lett.*, 2008, **49**, 2320.
- 12 (a) E. Van Heyningen, *J. Am. Chem. Soc.*, 1955, **77**, 6562; (b) J. P. Colomer and E. L. Moyano, *Tetrahedron Lett.*, 2011, **52**, 1561.
- 13 For example, see: (a) Y. Yan, H. Li, B. Niu, C. Zhu, T. Chen and Y. Liu, *Tetrahedron Lett.*, 2016, **57**, 4170; (b) N. G. Khaligh, M. R. Johan and J. J. Ching, *Aust. J. Chem.*, 2018, **71**, 186; (c) D. S. Barak, S. Mukhopadhyay, D. J. Dahatonde and S. Batra, *Tetrahedron Lett.*, 2019, **60**, 248; (d) Y.-M. Cai, X. Zhang, C. An, Y.-F. Yang, W. Liu, W.-X. Gao, X.-B. Huang, Y.-B. Zhou, M.-C. Liu and H.-Y. Wu, *Org. Chem. Front.*, 2019, **6**, 1481.
- 14 Y. Yan, B. Niu, K. Xu, J. Yu, H. Zhi and Y. Liu, *Adv. Synth. Catal.*, 2016, **358**, 212.
- 15 A. Chandrasekhar and S. Sankararaman, *J. Org. Chem.*, 2017, **82**, 11487.
- 16 Z. Lai, C. Wang, J. Li and S. Cui, *Org. Lett.*, 2020, **22**, 2017.
- 17 Y. Zhou, Y. Wang, Y. Lou and Q. Song, *Org. Lett.*, 2018, **20**, 6494.
- 18 (a) M. Caldarelli, I. R. Baxendale and S. V. Ley, *Green Chem.*, 2000, **2**, 43; (b) J. Merrington, M. James and M. Bradley, *Chem. Commun.*, 2002, 140; (c) V. D. Filimonov, M. Trusova, P. Postnikov, E. A. Krasnokutskaya, Y. M. Lee, H. Y. Hwang, H. Kim and K.-W. Chi, *Org. Lett.*, 2008, **10**, 3961; (d) M. E. Trusova, E. A. Krasnokutskaya, P. S. Postnikov, Y. Choi, K.-W. Chi and V. D. Filimonov, *Synthesis*, 2011, 2154.
- 19 (a) N. L. Sloan, S. K. Luthra, G. McRobbie, S. L. Pimlott and A. Sutherland, *Chem. Commun.*, 2017, **53**, 11008; (b) N. L. Sloan, S. K. Luthra, G. McRobbie, S. L. Pimlott and A. Sutherland, *RSC Adv.*, 2017, **7**, 54881.
- 20 R. J. Faggyas, M. Grace, L. Williams and A. Sutherland, *J. Org. Chem.*, 2018, **83**, 12595.
- 21 (a) R. J. Faggyas, N. L. Sloan, N. Buijs and A. Sutherland, *Eur. J. Org. Chem.*, 2019, 5344; (b) J. D. Bell, T. E. F. Morgan, N. Buijs, A. H. Harkiss, C. R. Wellaway and A. Sutherland, *J. Org. Chem.*, 2019, **84**, 10436.
- 22 (a) J. R. Feldman and E. C. Wagner, *J. Org. Chem.*, 1942, **7**, 31; (b) M. S. Manhas, S. G. Amin and V. V. Rao, *Synthesis*, 1977, 309.
- 23 See ESI† document for the synthesis of 2-aminobenzenesulfonamides and 1-benzoyl-4-[2'-aminoethyl]piperazine (**12**).
- 24 F. A. V. Sullivan, US2798055, 1957.
- 25 W. M. Thomas, N. Heights and F. A. V. Sullivan, US2834764, 1958.
- 26 K. S. Kumar, R. Adepu, S. Sandra, D. Rambabu, G. R. Krishna, C. M. Reddy, P. Misra and M. Pal, *Bioorg. Med. Chem. Lett.*, 2012, **22**, 1146.
- 27 For example, see: (a) C. M. Reid, K. N. Fanning, L. S. Fowler and A. Sutherland, *Tetrahedron*, 2015, **71**, 245; (b) L. Gilfillan, R. Artschwager, A. H. Harkiss, R. M. J. Liskamp and A. Sutherland, *Org. Biomol. Chem.*, 2015, **13**, 4514; (c) A. H. Harkiss, J. D. Bell, A. Knuhtsen, A. G. Jamieson and A. Sutherland, *J. Org. Chem.*, 2019, **84**, 2879; (d) J. D. Bell, A. H. Harkiss, D. Nobis, E. Malcolm, A. Knuhtsen, C. R. Wellaway, A. G. Jamieson, S. W. Magennis and A. Sutherland, *Chem. Commun.*, 2020, **56**, 1887.
- 28 M. R. Stojkovic, P. Piotrowski, C. Schmuck and I. Piantanida, *Org. Biomol. Chem.*, 2015, **13**, 1629.
- 29 This type of transformation cannot tolerate substrates with strongly nucleophilic substituents such as other amines or hydroxyl groups but as demonstrated by compounds **14** and **16**, protected versions of these are acceptable.
- 30 Y. Chang, J. Zhang, X. Chen, Z. Li and X. Xu, *Bioorg. Med. Chem. Lett.*, 2017, **27**, 2641.
- 31 F. G. Kathawalas, US3678166, 1972.
- 32 J. A. Zablocki, E. Elzein, X. Li, D. O. Koltun, E. Q. Parkhill, T. Kobayashi, R. Martinez, B. Corkey, H. Jiang, T. Perry, R. Kalla, G. T. Nottee, O. Saunders, M. Graupe, Y. Lu, C. Venkataramani, J. Guerrero, J. Perry, M. Osier, R. Strickley, G. Liu, W.-Q. Wang, L. Hu, X.-J. Li, N. El-Bizri, R. Hirakawa, K. Kahlig, C. Xie, C. H. Li, A. K. Dhalla, S. Rajamani, N. Mollova, D. Soohoo, E.-I. Lepist, B. Murray, G. Rhodes, L. Belardinelli and M. C. Desai, *J. Med. Chem.*, 2016, **59**, 9005.
- 33 J. G. Archer, A. J. Barker and R. K. Smalley, *J. Chem. Soc., Perkin Trans. 1*, 1973, 1169.
- 34 A. Rosowsky, R. A. Forsch and R. G. Moran, *J. Med. Chem.*, 1992, **35**, 2626.
- 35 R. J. LeBlanc and K. Vaughan, *Can. J. Chem.*, 1972, **50**, 2544.
- 36 T. Miura, M. Yamauchi, A. Kosaka and M. Murakami, *Angew. Chem., Int. Ed.*, 2010, **49**, 4955.
- 37 F. Ullmann and C. Groß, *Ber. Dtsch. Chem. Ges.*, 1910, **43**, 2694.
- 38 A. Saeed and N. H. Rama, *J. Chem. Soc. Pak.*, 1994, **16**, 31.

