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Effect of single electrons on the excited state dynamics of rod-shaped Au₂₅ nanoclusters†

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The excited state dynamics of small-sized metal nanoclusters are dependent on their crystal structures, while the effect of the charge state remains largely unknown. Here, we report the influence of single electrons on the excited-state dynamics of non-superatom Au clusters by comparing the transient absorption isotropy and anisotropy dynamics of two rod-shaped Au₂₅ nanoclusters protected by organic ligands. Two decay lifetimes (0.9 ps and 2.3 μs) can be identified in the excited state relaxation of Au₂₅²⁺ rods, which are assigned to the internal conversion from a higher to lower excited state and the relaxation to the ground state, respectively. With the addition of one electron, an additional 660 ps decay is observed in Au₂₅⁺, which should originate from the presence of a single electron occupied molecular orbital. Transient anisotropy measurements reveal a 500 ps rotational diffusion process in both the nanoclusters, while the initial dipole moment orientation is found to be highly dependent on the charge state. These results are of importance to understanding the effect of the charge state on the optical properties of metal nanoclusters.

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Introduction

Metal nanoclusters (NCs) that bridge nanoparticles and small metal complexes in terms of their structures and electronic properties have attracted broad research interest.^{1–8} In plasmonic gold and silver nanoparticles, photo-excitation will induce the collective excitation of conduction band electrons, which gives rise to a single surface plasmon resonance peak in their electronic absorption spectra.^{9,10} Ultra-small gold and silver NCs, on the other hand, exhibit single-exciton transition after photoexcitation, showing multiple peaks in their absorption spectra.^{7,11} With the advent of atomically precise gold NCs with well-defined crystal structures, it is possible to correlate their optical properties with atomic packing modes.^{11–14} The excited state dynamics of gold and silver NCs has received intense research interest because it helps to unravel the excited state energy flow and conversion.^{15–22} Based on a previous study, small sized gold NCs ($n < 50$) are non-scalable and their optical properties including steady state and excited state behaviours are very sensitive to their structures, compositions, shapes and surface protecting ligands.²³ For example, the optical absorption, photoluminescence and photodynamics of spherical-shaped [Au₂₅(SR)₁₈]^q NCs are drastically different from those of rod shaped Au₂₅ NCs.^{24,25} Therefore, it is possible to manipulate their electronic and optical properties via “molecular surgery” on these small-sized metal NCs.^{26,27}

According to previous studies, the charge state can also have a significant influence on the electronic properties of Au NCs. Zhu *et al.* reported that the crystal structure of $[\text{Au}_{25}(\text{SR})_{18}]^-$ is similar to that of $[\text{Au}_{25}(\text{SR})_{18}]^0$ while their optical absorption spectra and magnetism are very different.^{28,29} Qian *et al.* demonstrated that the excited state lifetimes of two $[\text{Au}_{25}(\text{SR})_{18}]^q$ ($q = 0, -1$) NCs were found to differ by a factor of 10^3 .²⁴ The effect of the valence number on excited state relaxation was also identified for $[\text{MAu}_{24}(\text{SR})_{18}]^q$ (M stands for foreign atoms other than gold) NCs.³⁰ Lee and co-workers investigated Jahn–Teller distortion on 6-electron $[\text{MAu}_{24}(\text{SR})_{18}]^0$ clusters, which gave rise to the splitting of 1P orbitals and narrowing of energy gaps.³¹ Knappenberger and co-workers illustrated that $[\text{Au}_{25}(\text{SR})_{18}]^q$ NCs with different charge states show different spin relaxation dynamics.³² The structures and stability of a series of metal NCs can be predicted and explained by using a superatom model, and these NCs have a closed-electron-shell configuration. Superatom theory is used to describe the metal cluster that has a filled spherical electric shell and major energy gap to unoccupied states, wherein the properties of clusters, such as stability, ionization energy, reactivity, *etc.*, should depend on the valence of superatom orbitals.³³ The superatom model can well explain the stability of Au NCs with a spherical metal core such as $[\text{Au}_{25}(\text{SR})_{18}]^q$, while it is not applicable to those with a non-spherical metal core.³⁴

Rod-shaped $[\text{Au}_{25}(\text{PPh}_3)_{10}(\text{SR})_5\text{Cl}_2]^{2+}$ (Au_{25} rod in short) is a typical “non-superatom” nanocluster, which contains 16 valence electrons.^{35–37} The excited state dynamics of the Au_{25} rod and its derivatives have been intensively investigated. Solvent dependent dynamics has been reported on the Au_{25} rod and it exhibits strong near-infrared photoluminescence (PL) with a maximum at 990 nm and the excited state lifetime is as long as 2.3 μs .^{25,38,39} Silver alloy $\text{Au}_{25-x}\text{Ag}_x$ NCs ($x > 13$) were reported to show very intense PL at around 700 nm (quantum yield, QY, is 40%) and the excited state lifetime is 7.3 μs .^{40,41} Despite this progress in the understanding of photophysics of the Au_{25} rod, the effect of the charge state on its excited state dynamics remains largely unknown. Recently, selenolate-capped Au_{25} rod NCs with different charge states have been reported,⁴² which allows for a comparative study of the charge state effect on excited state properties.

In this work, we chose two rod-shaped $[\text{Au}_{25}(\text{PPh}_3)_{10}(\text{SePh})_5\text{Cl}_2]^q$ ($q = +1$ and $+2$) NCs (**Au₂₅ I** and **Au₂₅ II** in short, respectively) as model systems to probe the effect of the charge state on the photodynamics of “non-superatom” NCs. Although the **Au₂₅ I** nanocluster has a smaller energy gap than the **Au₂₅ II** nanocluster, their nanosecond transient absorption spectra show similar profiles and their lifetimes are comparable. Femtosecond transient absorption isotropy in two Au NCs shows different photodynamics. In **Au₂₅ II**, the excited state dynamics can be fitted by two decay components (0.9 ps and >1 μs), which is explained by internal conversion from the $\text{LUMO}+n$ to the LUMO and the relaxation to the ground state, respectively. In **Au₂₅ I**, three decay components (0.8 ps, 660 ps, >1 μs) were identified and the 660 ps decay was assigned to an

additional decay channel that occurs from the lowest excited state. Transient absorption anisotropy dynamics in both the NCs involves a 500 ps rotational diffusion process, which indicates that the transition moment is aligned along the axis of the rod-shaped nanoclusters. Different anisotropy dynamics can be observed in the initial few picoseconds of two NCs, which suggests that the transition dipole moment of two NCs is highly dependent on the charge state. These results are of great importance to understanding the effect of the charge state on the electronic properties of “non-superatom” gold NCs.

Results and discussion

As reported in previous work, the crystal structures of two rod-shaped Au_{25} NCs are very similar, both consisting of two icosahedral Au_{13} building blocks by sharing one vertex.⁴² The major difference between two NCs is the alignment of the benzene rings on Ph–Se ligands (Fig. 1a): in **Au₂₅ I**, those benzene rings are aligned in the same plane (blue dashed line); in **Au₂₅ II**, those benzene rings are distorted (red dashed line). Despite very similar crystal structures, different optical absorption spectra were observed for the two NCs (Fig. 1b). Between 300 and 800 nm, the two rod-shaped Au_{25} NCs exhibit similar UV-vis absorption profiles, with a peak at around 450 nm slightly stronger in **Au₂₅ II**. Between 800 and 1100 nm, an additional absorption peak at around 870 nm can be observed in **Au₂₅ I**. Based on steady state absorption, the optical gaps of **Au₂₅ I** and **II** are determined to be 1.2 eV and 1.6 eV, respectively (Fig. S1, ESI†). The difference in the optical bandgap can also be supported by recent PL measurements on two NCs, which are centered at around 990 nm (**Au₂₅ II**) and 1100 nm (**Au₂₅ I**), respectively.³⁹ Based on previous theoretical calculations, the presence of additional peaks (870 nm) in **Au₂₅ I** should originate from an additional single electron occupied molecular orbital (SMO) which lies between the normal HOMO and LUMO.⁴²

To understand the excited state dynamics, we first probe the nanosecond transient absorption (TA) of two NCs dissolved in dichloromethane (CH_2Cl_2) using flash photolysis. Under similar experimental conditions after 400 nm excitation, the TA spectra of two NCs at 30 ns are almost identical between 300 nm and 800 nm (Fig. 1c), showing ground state bleaching (GSB) at around 350 nm, 450 nm and 700 nm and excited state absorption (ESA) at around 620 nm. The ns-TA kinetics probed at 620 nm and the single exponential fitting convoluted with the instrument response function (IRF) are shown in Fig. 1d. The excited state lifetimes are determined to be 3 μs and 2.3 μs , respectively. The very long-lived TA lifetime of **Au₂₅ II** is identical to that of thiolate capped $[\text{Au}_{25}(\text{PPh}_3)_{10}(\text{SR})_5\text{Cl}_2]^{2+}$ (**Au₂₅ III** in short, the TA lifetime is around 2.3 μs),²⁵ which indicates that the replacement of capping ligands on the waist of the Au_{25} rod shows little influence on the excited deactivation kinetics. Surprisingly, **Au₂₅ I**, which has a much

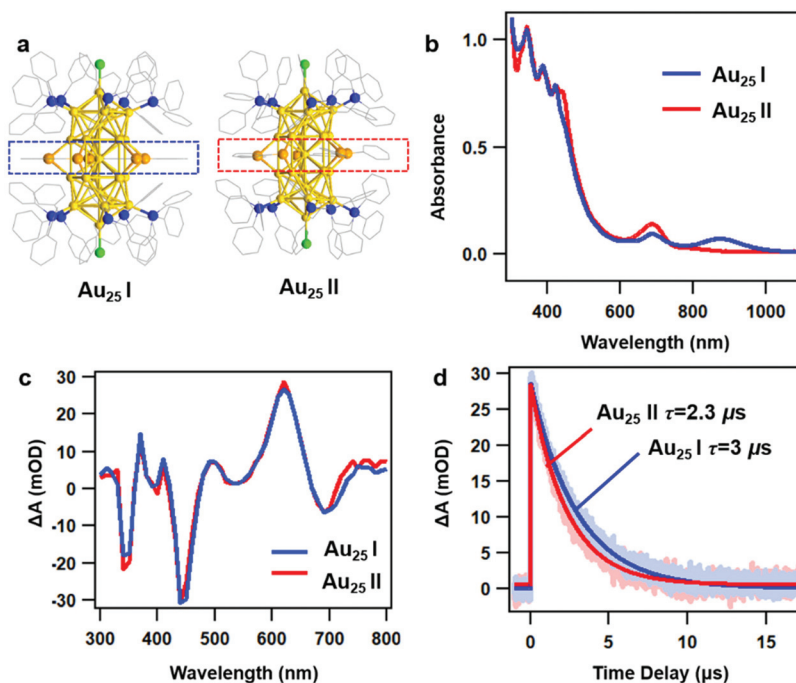


Fig. 1 (a) Crystal structures of Au₂₅ I (left) and Au₂₅ II (right); dashed lines indicate the structural difference between two NCs; (color labels: Au = yellow; P = blue; Cl = green; and Se = orange; the C atoms are in wireframe); (b) UV-vis absorption spectra of two NCs, solvent: CH₂Cl₂; (c) TA spectra of two NCs probed at 30 ns; (d) nanosecond TA kinetics of two NCs probed at 620 nm and the corresponding fits.

smaller optical gap (1.2 eV) compared to Au₂₅ II (1.6 eV), exhibits a slightly longer excited state lifetime.

In general, a smaller bandgap will shorten the excited state lifetime (faster nonradiative process) of Au NCs with a similar structure.⁴³ Previous density functional theory (DFT) calculations on Au₂₅ III indicated that the HOMO is aligned along the z-axis of the Au₂₅ rod while the LUMO is mainly contributed by the 6s orbital on the central Au atom shared by two Au₁₃ units.³⁶ As a result, it is difficult for the excited state energy to dissipate into the environment. Based on previous calculation results, the LUMO of Au₂₅ II and the SMO of Au₂₅ I are also contributed by the 6s orbital on the central Au atom. Therefore, both the Au₂₅ rods exhibit very long-lived excited state lifetimes and similar excited-state dynamics in the nanosecond to microsecond time range.

The slightly longer excited-state lifetime observed in Au₂₅ I (3 μs) than that in Au₂₅ II (2.3 μs) can be ascribed to the different intrinsic radiative decays in the two NCs. Compared to Au₂₅ II, Au₂₅ I exhibits stronger near-infrared (NIR) photoluminescence and a smaller absorption coefficient of the band-edge transition, which will lead to a slower radiative rate and longer excited state lifetime.^{24,44} Therefore, Au₂₅ I with a smaller E_g value shows a slightly longer excited-state lifetime than Au₂₅ II. In this regard, the photoluminescence mechanisms of the two Au₂₅ rod NCs are needed to elucidate this point, which calls for further investigations in the future.

Despite similar nanosecond photodynamics, different relaxation dynamics were observed in the femtosecond to pico-

second time ranges. We first look into the TA isotropy dynamics by setting the pump and probe pulses to the magic angle (54.7°) so that the rotational contribution can be ruled out and only the population dynamics is obtained. Fig. 2 shows the TA data map and spectra of the two NCs probed between 350 nm and 750 nm at different time delays with excitation at 400 nm (3.01 eV). In Au₂₅ II, the major evolution of the TA spectra occurs between 0 and 5 ps: the ESA peak at around 500 nm decays to give rise to ESA at around 620 nm, which remains unchanged between 10 ps and 7 ns (Fig. 2a). Such an observation is very similar to that of Au₂₅ III which also shows non-decaying ESA peak at around 610 nm. In Au₂₅ I, similar evolution in the TA spectra was observed between 0 and 5 ps, while an additional decay can be found between 100 ps and 1000 ps. Between 1000 ps and 7000 ps, the TA spectra remain unchanged (Fig. 2b). It is worth noting that in the initial 10 ps, a strong overlap of ESA and GSB can be found in the TA spectra for both the NCs. After comparing the TA kinetics probed at 508 nm of the two NCs, one can observe an additional decay between 0 and 3000 ps in Au₂₅ I (Fig. S2, ESI†).

To obtain the decay time constants of the two NCs, we performed global fitting on the TA data and the fitting results are shown in Fig. 3 and Fig. S3.† The TA relaxation of Au₂₅ II can be well fitted by two decay components (0.9 ps and >1 μs) (Fig. 3a). One can clearly observe that the ESA intensity in the first 2 ps experiences rapid decay and the ESA intensity at around 600 nm experiences an increase at the same time (Fig. 3c). With 400 nm (3.01 eV) excitation, the electrons of

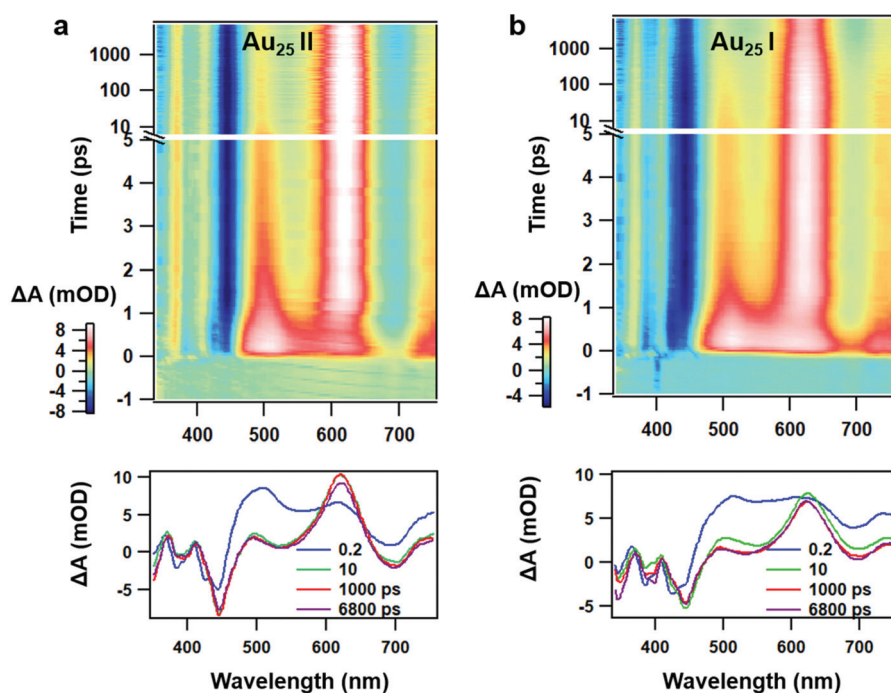


Fig. 2 (a) TA data map with 400 nm excitation and TA spectra at the selected probe time of Au_{25} II NCs; (b) TA data map with 400 nm excitation and TA spectra at the selected probe time of Au_{25} I NCs.

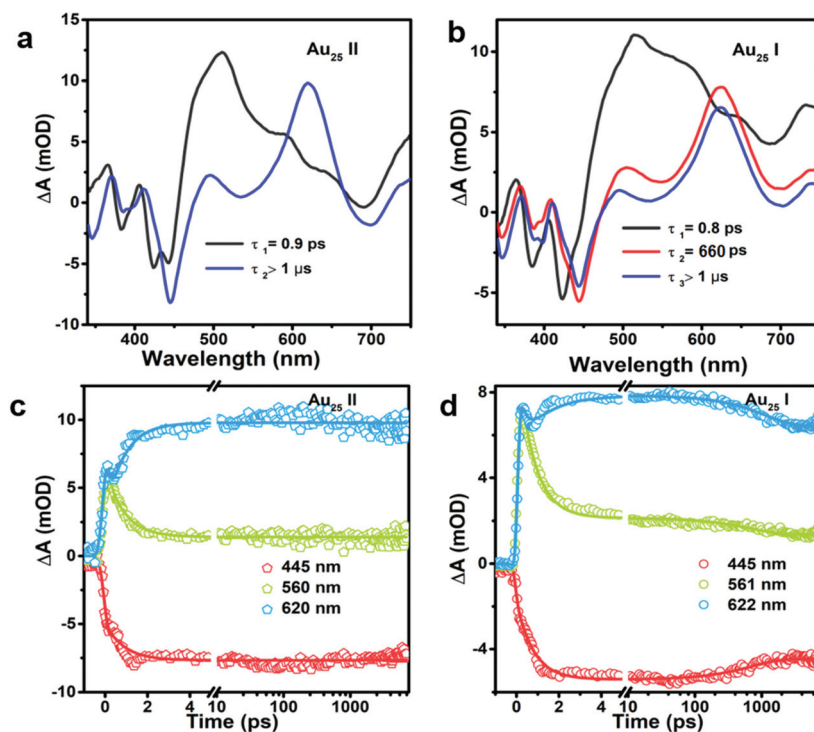


Fig. 3 Evolution associated spectra (EAS) obtained from global analysis on the TA data of Au_{25} II (a) and Au_{25} I (b) NCs; TA kinetics probed at selected wavelengths and the corresponding fits of Au_{25} II (c) and Au_{25} I (d) NCs.

Au_{25} II will be excited to a higher excited state ($\text{LUMO}+n$), which should experience a rapid internal conversion (0.9 ps) to the LUMO and then a slow decay (2.3 μs obtained from ns-TA)

to the ground state. Such a relaxation pathway in Au_{25} II is very similar to that in Au_{25} III reported previously,²⁵ which suggests that replacing S-Ph with Se-Ph on the waist of the Au_{25} rod

does not change the excited state dynamics significantly. In **Au₂₅ I**, three decay components are required to fit the relaxation dynamics (0.8 ps, 660 ps and >1 μs) (Fig. 3b). The first sub-picosecond decay component is similar to those of **Au₂₅ II** and **III**. After comparing the evolution associated spectra of the second (660 ps) and the third component (>1 μs), we found that they have almost similar profiles and only the amplitude changes slightly. Moreover, the ESA and GSB decay together between ~3 ps and ~2000 ps (Fig. 3d), indicating that no other intermediate excited state species was observed in the TA isotropy signal during this time. Therefore, the 660 ps decay can be identified as an additional relaxation channel of the excited state. Such an additional relaxation pathway should originate from the relaxation of the SMO lying between the LUMO and HOMO. The schematic diagram of the excited-state relaxation pathways for the two NCs is shown in Fig. 4.

Since rod-shaped **Au₂₅** NCs are anisotropic in shape, we also performed transient absorption anisotropy (TAA) on the two NCs to obtain the dipole moment change during the excited state relaxation. The TAA signal $r(t)$ is calculated by:⁴⁵

$$r(t) = \frac{I_{VV} - I_{VH}}{I_{VV} + 2I_{VH}} \quad (1)$$

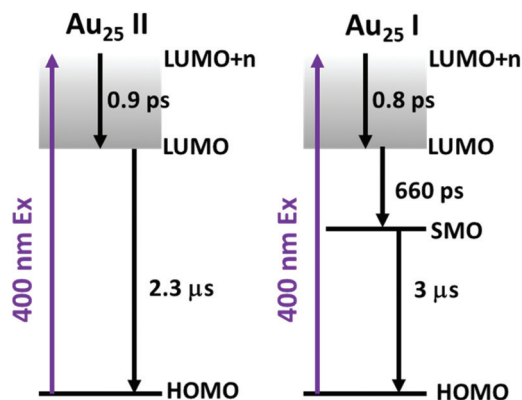


Fig. 4 The schematic diagram of the excited-state relaxation pathways for **Au₂₅ II** (left) and **Au₂₅ I** (right).

where I_{VV} and I_{VH} represent the TA signals with the probe parallel and vertical to the pump, respectively. In the TAA data map probed at different wavelengths and time delays of the two rod-shaped **Au₂₅** NCs (Fig. S4, ESI†), we observed anisotropy that persists for a long time during the entire TA measurements. The TAA spectra of the two NCs at selected time delays (1–1000 ps) are plotted in Fig. 5. In general, the initial anisotropy value at 1 ps is higher in **Au₂₅ II** than that in **Au₂₅ I** at most probe wavelengths. In **Au₂₅ II** (Fig. 5a), the anisotropy near GSB at around 700 nm at 1 ps is 0.3, which is close to the maximum allowable value (0.4). This indicates that at early time delays, the probed transition is almost parallel to the transition excited in **Au₂₅ II**. At GSB around 450 nm and ESA around 620 nm, the anisotropy is around 0.1 at 1 ps for **Au₂₅ II**. In **Au₂₅ I**, on the other hand, the anisotropy near GSB at 700 nm is around −0.1 at 1 ps. At other probe wavelengths, the anisotropy of **Au₂₅ I** is smaller than 0.1 during the whole TAA measurement, which indicates that there is a large angle between the probed transition and the transition excited. In both the rod-shaped **Au₂₅** NCs, the TAA signals decay slowly and do not reach zero at 1000 ps.

The TAA decay dynamics can reflect the change in the excited state population, excited state energy transfer and spatial rotation of the clusters in solutions. We further plotted the TAA kinetics probed at GSB around 700 nm and 450 nm and ESA around 620 nm for two NCs in Fig. 6. The TAA values at 10 ps for two NCs are summarized in Table 1. In **Au₂₅ II**, the TAA decays at 700 nm and 450 nm consist of a fast decay followed by a slow decay. If we ignore the dynamics before 1 ps, the TAA dynamics probed at 700 nm and 450 nm can be fitted by a two-exponential decay (1 ps, 500 ps). The 1 ps decay should be assigned to the internal conversion process, which will lower the TAA value probed at 450 nm and 700 nm. The second 500 ps decay should be explained as the rotational diffusion of the cluster. In a previous study, the rotational diffusion time of **Au₂₅ III** dissolved in ethanol was found to be 1.3 ns.²⁵ The rotational correlation time can be estimated from the hydrodynamic relation:⁴⁶

$$\tau = \frac{\nu\eta}{k_B T} \quad (2)$$

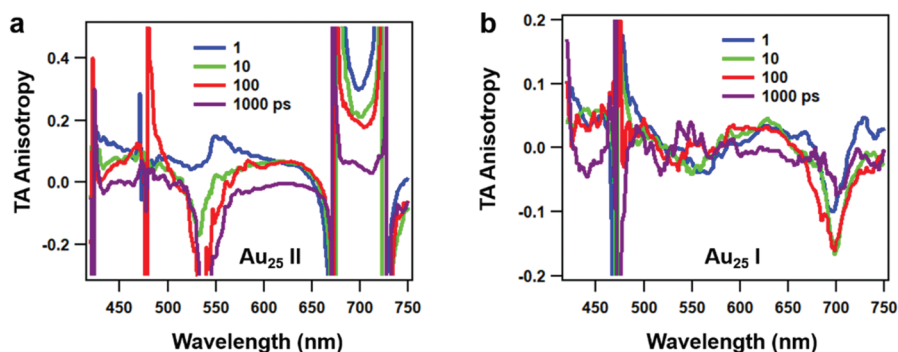


Fig. 5 Transient absorption anisotropy (TAA) spectra of (a) **Au₂₅ II** and (b) **Au₂₅ I** NCs at selected time delays.

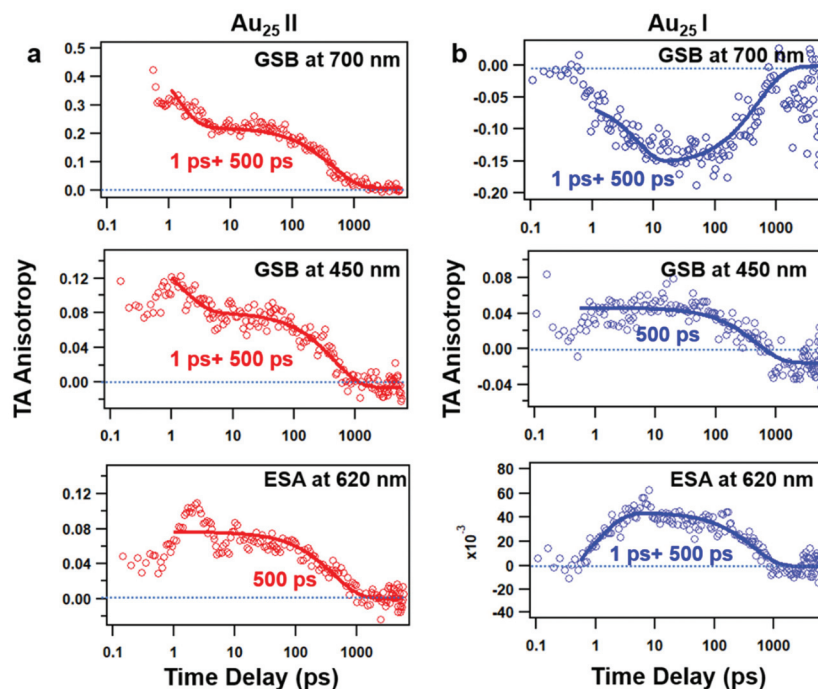


Fig. 6 TAA kinetics at different probed wavelengths and the corresponding fits of Au_{25} II (a) and Au_{25} I (b) NCs.

Table 1 Transient absorption anisotropy values of two rod-shaped Au_{25} NCs probed at different wavelengths with a pump–probe delay of 10 ps

	GSB at 700 nm	GSB at 450 nm	ESA at 620 nm
Au_{25} I	0.21	0.07	0.07
Au_{25} II	−0.15	0.04	0.04

where ν represents the particle volume and η is the viscosity of the solvent. The viscosity of dichloromethane and ethanol at room temperature is around 0.45 and 1.08,⁴⁷ respectively, which are consistent with the 500 ps and 1.3 ns rotational diffusion times of rod-shaped Au_{25} NCs in two solvents. The single exponential rotational correlation time (500 ps) indicates that the transition moment is aligned with the long axis of the clusters.

In Au_{25} I, on the other hand, the TAA kinetics probed at 700 nm and 620 nm experience a fast increase followed by a very slow decay (Fig. 6b). The TAA kinetic traces probed at 700 nm and 620 nm can be fitted by a two-exponential function (1 ps rise and 500 ps decay), and the slow decay should be assigned to rotational diffusion similar to that in Au_{25} II. In Au_{25} I, the GSB around 700 nm is strongly overlapped with ESA, which may complicate the analysis of the TAA dynamics. One can observe the anisotropy value around 700 nm which increases from 0 to −0.15 with a time constant of 1 ps (Fig. 6b), which agrees with the 0.9 ps internal conversion in TA isotropy dynamics (Fig. 3c). This suggests that the probed transition (700 nm GSB) is almost vertical to the transition excited (400 pump) at 10 ps in Au_{25} I.

Typically, the TAA signal only experiences decay, which can reflect the relaxation of the excited states, energy transfer and the rotation of the molecule. However, the transient absorption spectrum is a combination of positive ESA, negative GSB and stimulated emission; a change in the amplitude of one component could change the anisotropy value of another component at the same probe wavelengths.⁴⁵ Therefore, the overlap in the TA spectrum complicated the analysis of TA anisotropy. Here, at 620 nm, there is less overlap and the ESA dominates the TA spectra of both the NCs. We then chose the ESA around 620 nm to compare the TAA dynamics of two Au_{25} NCs. In Au_{25} I, an increase (1 ps) in the TAA dynamics around 620 nm can be observed, and is absent in Au_{25} II. The increase in TAA indicates that an additional ultrafast energy redistribution process (1 ps) should occur in Au_{25} I. The ultrafast energy redistribution of Au_{25} I could arise from the (1) relaxation from the LUMO to the SMO; (2) generation of a new transition dipole induced by photoexcitation as a result of its paramagnetic response. The detailed assignment of the increase in TAA in Au_{25} I calls for further theoretical work in the future.

Conclusions

In summary, we have revealed the effect of single electrons on the excited state dynamics of “non-superatom” rod-shaped Au_{25} NCs by a comparative study of transient absorption isotropy and anisotropy spectroscopy. The relaxation dynamics of $[\text{Au}_{25}(\text{PPh}_3)_{10}(\text{SePh})_5\text{Cl}_2]^{2+}$ is almost identical to that of $[\text{Au}_{25}(\text{PPh}_3)_{10}(\text{SPh})_5\text{Cl}_2]^{2+}$, indicating little effect of the SePh on the waist of the Au_{25} rod. With the addition of one electron, an

additional decay of 660 ps can be observed in the population dynamics while the long-lived microsecond decay remains almost unchanged. TAA measurements reveal a 500 ps rotational diffusion process in both the NCs, while the initial TAA dynamic process is found to be highly dependent on the charge state. These results provide new insight into the charge state effect on the photophysics of “non-superatom” metal nanoclusters.

Conflicts of interest

There are no conflicts to declare.

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References

- R. Jin, C. Zeng, M. Zhou and Y. Chen, *Chem. Rev.*, 2016, **116**, 10346–10413.
- I. Chakraborty and T. Pradeep, *Chem. Rev.*, 2017, **117**, 8208–8271.
- Y. Li, M. Zhou, Y. Song, T. Higaki, H. Wang and R. Jin, *Nature*, 2021, **594**, 380–384.
- C. Zeng, Y. Chen, K. Kirschbaum, K. J. Lambright and R. Jin, *Science*, 2016, **354**, 1580–1584.
- Z. Lei, X.-K. Wan, S.-F. Yuan, Z.-J. Guan and Q.-M. Wang, *Acc. Chem. Res.*, 2018, **51**, 2465–2474.
- J. Yan, B. K. Teo and N. Zheng, *Acc. Chem. Res.*, 2018, **51**, 3084–3093.
- H. Yang, Y. Wang, H. Huang, L. Gell, L. Lehtovaara, S. Malola, H. Häkkinen and N. Zheng, *Nat. Commun.*, 2013, **4**, 2422.
- R.-W. Huang, Y.-S. Wei, X.-Y. Dong, X.-H. Wu, C.-X. Du, S.-Q. Zang and T. C. W. Mak, *Nat. Chem.*, 2017, **9**, 689–697.
- G. V. Hartland, *Chem. Rev.*, 2011, **111**, 3858–3887.
- M. Zhou, C. Zeng, Y. Chen, S. Zhao, M. Y. Sfeir, M. Zhu and R. Jin, *Nat. Commun.*, 2016, **7**, 13240.
- M. Zhu, C. M. Aikens, F. J. Hollander, G. C. Schatz and R. Jin, *J. Am. Chem. Soc.*, 2008, **130**, 5883–5885.
- C. M. Aikens, *Acc. Chem. Res.*, 2018, **51**, 3065–3073.
- M. Zhou, T. Higaki, G. Hu, M. Y. Sfeir, Y. Chen, D.-E. Jiang and R. Jin, *Science*, 2019, **364**, 279–282.
- M. Kato, Y. Shichibu, K. Ogura, M. Iwasaki, M. Sugiuchi, K. Konishi and I. Yagi, *J. Phys. Chem. Lett.*, 2020, **11**, 7996–8001.
- M. Zhou and R. Jin, *Annu. Rev. Phys. Chem.*, 2021, **72**, 121–142.
- T. Stoll, E. Sgrò, J. W. Jarrett, J. Réhault, A. Oriana, L. Sala, F. Branchi, G. Cerullo and K. L. Knappenberger, *J. Am. Chem. Soc.*, 2016, **138**, 1788–1791.
- T. D. Green and K. L. Knappenberger, *Nanoscale*, 2012, **4**, 4111–4118.
- M. S. Devadas, V. D. Thanthirige, S. Bairu, E. Sinn and G. Ramakrishna, *J. Phys. Chem. C*, 2013, **117**, 23155–23161.
- V. D. Thanthirige, M. Kim, W. Choi, K. Kwak, D. Lee and G. Ramakrishna, *J. Phys. Chem. C*, 2016, **120**, 23180–23188.
- M. Zhou, H. Qian, M. Y. Sfeir, K. Nobusada and R. Jin, *Nanoscale*, 2016, **8**, 7163–7171.
- S. Maity, D. Bain and A. Patra, *Nanoscale*, 2019, **11**, 22685–22723.
- A. K. Das, S. Maity, T. Sengupta, D. Bista, A. C. Reber, A. Patra, S. N. Khanna and S. Mandal, *J. Phys. Chem. Lett.*, 2021, **12**, 2154–2159.
- M. Zhou, T. Higaki, Y. Li, C. Zeng, Q. Li, M. Y. Sfeir and R. Jin, *J. Am. Chem. Soc.*, 2019, **141**, 19754–19764.
- H. Qian, M. Y. Sfeir and R. Jin, *J. Phys. Chem. C*, 2010, **114**, 19935–19940.
- M. Y. Sfeir, H. Qian, K. Nobusada and R. Jin, *J. Phys. Chem. C*, 2011, **115**, 6200–6207.
- Q. Li, T.-Y. Luo, M. G. Taylor, S. Wang, X. Zhu, Y. Song, G. Mpourmpakis, N. L. Rosi and R. Jin, *Sci. Adv.*, 2017, **3**, e1603193.
- Q. Li, M. Zhou, W. Y. So, J. Huang, M. Li, D. R. Kauffman, M. Cotlet, T. Higaki, L. A. Peteanu, Z. Shao and R. Jin, *J. Am. Chem. Soc.*, 2019, **141**, 5314–5325.
- M. Zhu, W. T. Eckenhoff, T. Pintauer and R. Jin, *J. Phys. Chem. C*, 2008, **112**, 14221–14224.
- M. Zhu, C. M. Aikens, M. P. Hendrich, R. Gupta, H. Qian, G. C. Schatz and R. Jin, *J. Am. Chem. Soc.*, 2009, **131**, 2490–2492.
- M. Zhou, C. Yao, M. Y. Sfeir, T. Higaki, Z. Wu and R. Jin, *J. Phys. Chem. C*, 2018, **122**, 13435–13442.
- K. Kwak, Q. Tang, M. Kim, D.-E. Jiang and D. Lee, *J. Am. Chem. Soc.*, 2015, **137**, 10833–10840.
- L. J. Williams, P. J. Herbert, M. A. Tofanelli, C. J. Ackerson and K. L. Knappenberger, *J. Chem. Phys.*, 2019, **150**, 101102.
- M. Walter, J. Akola, O. Lopez-Acevedo, P. D. Jadzinsky, G. Calero, C. J. Ackerson, R. L. Whetten, H. Grönbeck and H. Häkkinen, *Proc. Natl. Acad. Sci. U. S. A.*, 2008, **105**, 9157.
- H. Qian, W. T. Eckenhoff, Y. Zhu, T. Pintauer and R. Jin, *J. Am. Chem. Soc.*, 2010, **132**, 8280–8281.
- K. Nobusada and T. Iwasa, *J. Phys. Chem. C*, 2007, **111**, 14279–14282.
- Y. Shichibu, Y. Negishi, T. Watanabe, N. K. Chaki, H. Kawaguchi and T. Tsukuda, *J. Phys. Chem. C*, 2007, **111**, 7845–7847.
- S. Park and D. Lee, *Langmuir*, 2012, **28**, 7049–7054.
- V. D. Thanthirige, E. Sinn, G. P. Wiederrecht and G. Ramakrishna, *J. Phys. Chem. C*, 2017, **121**, 3530–3539.
- Q. Li, C. J. Zeman IV, Z. Ma, G. C. Schatz and X. W. Gu, *Small*, 2021, **17**, 2007992.

- 40 S. Wang, X. Meng, A. Das, T. Li, Y. Song, T. Cao, X. Zhu, M. Zhu and R. Jin, *Angew. Chem., Int. Ed.*, 2014, **53**, 2376–2380.
- 41 M. Zhou, J. Zhong, S. Wang, Q. Guo, M. Zhu, Y. Pei and A. Xia, *J. Phys. Chem. C*, 2015, **119**, 18790–18797.
- 42 Y. Song, S. Jin, X. Kang, J. Xiang, H. Deng, H. Yu and M. Zhu, *Chem. Mater.*, 2016, **28**, 2609–2617.
- 43 M. Zhou and Y. Song, *J. Phys. Chem. Lett.*, 2021, **12**, 1514–1519.
- 44 Q. Li, C. J. Zeman, G. C. Schatz and X. W. Gu, *ACS Nano*, 2021, **15**, 16095–16105.
- 45 S. Wallin, J. Davidsson, J. Modin and L. Hammarström, *J. Phys. Chem. A*, 2005, **109**, 4697–4704.
- 46 L. Reynolds, J. A. Gardecki, S. J. V. Frankland, M. L. Horng and M. Maroncelli, *J. Phys. Chem.*, 1996, **100**, 10337–10354.
- 47 I. Smallwood, *Handbook of Organic Solvent Properties*, Butterworth-Heinemann, 2012.