



Cite this: *Nanoscale*, 2021, **13**, 4537

Single-step chemical vapour deposition of anti-pyramid MoS₂/WS₂ vertical heterostructures†

Xueyin Bai,^{a*} Shisheng Li,^b Susobhan Das,^a LuoJun Du,^a Yunyun Dai,^a Lide Yao,^c Ramesh Raju,^a Mingde Du,^a Harri Lipsanen^a and Zhipei Sun^{a,d}

Van der Waals heterostructures are the fundamental building blocks of electronic and optoelectronic devices. Here we report that, through a single-step chemical vapour deposition (CVD) process, high-quality vertical bilayer MoS₂/WS₂ heterostructures with a grain size up to ~60 μm can be synthesized from molten salt precursors, Na₂MoO₄ and Na₂WO₄. Instead of normal pyramid vertical heterostructures grown by CVD, this method synthesizes an anti-pyramid MoS₂/WS₂ structure, which is characterized by Raman, photoluminescence and second harmonic generation microscopy. Our facile CVD strategy for synthesizing anti-pyramid structures unveils a new synthesis route for the products of two-dimensional heterostructures and their devices for application.

Received 20th November 2020,
Accepted 29th January 2021

DOI: 10.1039/d0nr08281c

rsc.li/nanoscale

1. Introduction

Two-dimensional (2D) materials are a series of crystalline materials with a thickness at atomic scale, such as graphene, transition metal dichalcogenides (TMDs), hexagonal boron nitride (h-BN) and other materials. Due to their unique properties, these 2D materials are expected to play important roles in future electronics, optoelectronics and photonics.^{1–6} By vertically stacking different 2D materials, various 2D van der Waals heterostructures (2D heterostructures) have been demonstrated showing improved performance or even completely novel properties, and potential for various applications.^{7–10} For example, by the stacking of graphene, h-BN and MoS₂ as a heterostructure, Withers *et al.* have achieved light emitting diodes with a quantum efficiency of 8.4%,¹¹ while monolayer MoS₂ based light emitting diodes typically only have that of less than 1%.^{12,13} Hence, 2D heterostructures are very attractive to researchers, and how to obtain scalable and high-quality heterostructures and their devices is an important question urgently waiting for answers.

In the synthesis of 2D materials and their heterojunctions, chemical vapour deposition (CVD) is a common and facile method due to its best balance between scalable and controlla-

ble growth, high quality and large area production, and economic cost.^{10,14} Lots of efforts have been made to synthesize various TMDs, including single or multilayer (homo)structure and heterostructure stacking with different TMD layers. Usually, a heterostructure is fabricated by a multi-step CVD growth method, in which two or more different precursors are injected into the system in sequence.^{15–18} The only single-step process, to the best of our knowledge, is a reported growth method of a WS₂/MoS₂ heterostructure, in which additional tellurium is added as a promoter to accelerate the melting of tungsten powder.¹⁹ Nevertheless, similar to other CVD growth methods,²⁰ transition metals in this work are offered as a vapour of metal oxides, leading to difficulties in controlling the thickness and size of TMDs. Hence, finding a facile fabrication method for the heterostructure which consists of monolayer TMDs is still challenging. Recently, Li *et al.* demonstrated a novel vapour-liquid-solid (VLS) growth method of TMDs.^{21,22} In sharp contrast to the vapour of transition metal oxides in other growth methods, the non-volatile salts, *e.g.*, Na₂MoO₄ and Na₂WO₄, used in the VLS growth method are in the liquid phase during the reaction, yielding solid monolayer TMDs after being saturated with chalcogens. This VLS growth method has been proved as a promising method for synthesizing various monolayer TMDs with a large area and high quality.^{21–23} However, to date, no heterostructures have been demonstrated by this process.

In this work, by carefully choosing the precursors, we demonstrated a versatile, reproducible one-step growth method for the 2D TMD heterostructures without any additional promoters. This is the first report of fabrication of a vertical anti-pyramid heterostructure by the CVD growth method. The synthesized MoS₂/WS₂ heterostructure has a

^aDepartment of Electronics and Nanoengineering, Aalto University, Finland.

E-mail: xueyin.bai@aalto.fi

^bInternational Center for Young Scientists (ICYS), National Institute for Materials Science (NIMS), Japan

^cDepartment of Applied Physics, Aalto University, Finland

^dQTF Centre of Excellence, Department of Applied Physics, Aalto University, Finland

†Electronic supplementary information (ESI) available. See DOI: 10.1039/d0nr08281c



lateral grain size of up to 60 μm and a structure of AA' stacking. Optical microscopy, micro-Raman and photoluminescence (PL) spectroscopy, atomic force microscopy (AFM), as well as non-linear optical spectroscopy are used to fully characterize the structural and optical qualities of the as-synthesized heterostructures.

2. Results and discussion

To grow the MoS_2/WS_2 heterostructure, 5 mg mL^{-1} Na_2MoO_4 and Na_2WO_4 are firstly spin-coated on two sapphire or SiO_2/Si substrates, respectively. Then, the two substrates with the precursors are placed in a dual-zone tube furnace side by side, as illustrated in Fig. 1a. Meanwhile, ~ 100 mg sulphur is loaded at the centre of Zone 1 in upstream. 70 sccm Ar is chosen as the carrier gas. The temperature ramping rate of Zone 2 is $10^\circ\text{C min}^{-1}$. When Zone 2 reaches the growth temperature (825°C for sapphire and 775°C for SiO_2/Si), the temperature of Zone 1 reaches 200°C for sulphur evaporation. Both zones are stabilized for 10 min and cooled to room temperature naturally. The MoS_2/WS_2 heterostructure is only grown on the substrate with Na_2WO_4 , while the substrate with Na_2MoO_4 has only triangular flakes of MoS_2 (Fig. S2[†]). Some typical optical images of MoS_2/WS_2 heterostructures are shown in Fig. 1b. The dark triangular centre of the flakes is considered as the MoS_2/WS_2 heterostructure region, exhibiting a typical 3-fold symmetry with straight edges. The lateral size of the heterostructure is up to $60\ \mu\text{m}$.

Raman and PL spectroscopy are usually used for determining the thickness and composition of TMDs and their heterostructures. Here, we measure the Raman and PL spectra of two different heterostructures, as shown in Fig. 2a and d, respectively. It is noticeable that the flake in Fig. 2a has two different parts, a dark triangular bilayer and a surrounding monolayer.

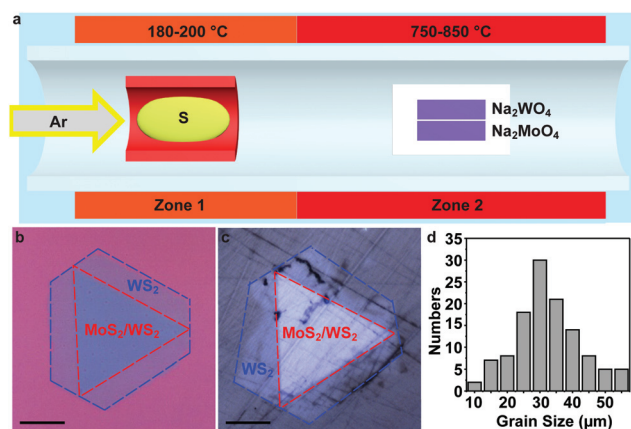


Fig. 1 Schematic illustration of the growth setup and optical images of the MoS_2/WS_2 heterostructures. (a) Schematic illustration of the growth system. (b and c) Typical optical microscopy images of the synthesized MoS_2/WS_2 heterostructures on SiO_2/Si (b) and sapphire (c) substrates, respectively (scale bar, $20\ \mu\text{m}$). (d) A histogram of grain sizes from 124 heterostructures from randomly selected areas of the sample.

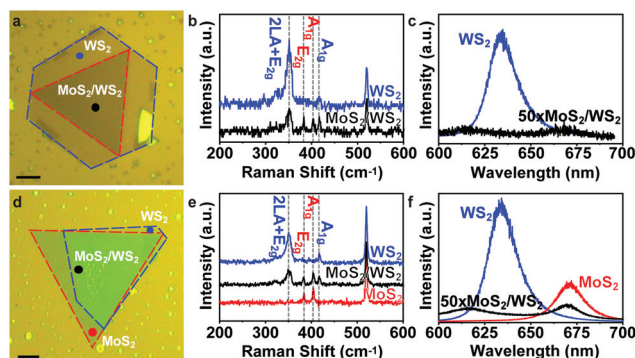


Fig. 2 Optical characterization of the MoS_2/WS_2 heterostructure. (a) Optical microscopy images of the heterostructures (scale bar, $10\ \mu\text{m}$). (b) Raman and (c) PL spectra at the marked dots with the same colours, respectively (black PL spectrum with 50 times magnification). (d) Optical microscopy images of a partial stacking heterostructure (scale bar, $25\ \mu\text{m}$). (e) Raman and (f) PL spectra at the marked dots with the same colours, respectively (black PL spectrum with 50 times magnification).

The Raman and PL spectra of the bilayer (black dot in Fig. 2a) show 4 typical peaks (black curve in Fig. 2b). The two peaks at $385\ \text{cm}^{-1}$ and $404\ \text{cm}^{-1}$ are assigned to the E_{2g} and A_{1g} mode phonons of monolayer MoS_2 , while the other two peaks, $351\ \text{cm}^{-1}$ and $417\ \text{cm}^{-1}$, are indicated to be the $2\text{LA}(\text{M}) + \text{E}_{2g}$ and A_{1g} phonon modes of WS_2 .^{24,25} With the Raman spectrum (black curve in Fig. 2b), it is concluded that the dark triangular centre is a composite of MoS_2 and WS_2 monolayers. Meanwhile, the Raman spectrum (blue curve in Fig. 2b) measured at the blue dot of Fig. 2a only shows peaks at $351\ \text{cm}^{-1}$ and $417\ \text{cm}^{-1}$, indicating that the light outside part is the WS_2 monolayer. For the PL measurements, two PL peaks with centres at $670\ \text{nm}$ and $620\ \text{nm}$ are observed with a magnification of 50 times from the black dot, indicating that the peaks are the A and B excitons of the monolayer MoS_2 , respectively. In addition, the extremely weak peaks also indicate that the transfer of electrons and holes happens during the PL measurements,⁹ which agrees with the MoS_2/WS_2 heterostructures. At the blue dot, the high-intensity PL peak with a centre at $630\ \text{nm}$ is assigned to the A exciton of the monolayer WS_2 and indicates that the light outside part is the monolayer WS_2 . Therefore, the flake in Fig. 2a is confirmed as a heterostructure with the monolayer MoS_2 (area in red dashed lines) and the monolayer WS_2 (area in blue dashed lines).

To exclude other possibilities (e.g., MoS_2/WS_2 alloy), a special heterostructure shown in Fig. 2d is found on the substrate and is measured to clarify the structure. The flake shows a partial stacking by two layers marked with red and blue dashed lines, respectively. In the PL and Raman spectra at the left part of the monolayer (red dot), two phonon modes at $385\ \text{cm}^{-1}$ and $404\ \text{cm}^{-1}$ and a luminescence peak at $670\ \text{nm}$ are observed, as shown in Fig. 2e and f, indicating that the left part (i.e., the red dotted area of Fig. 2d) is monolayer MoS_2 . Similarly, the luminescence at $630\ \text{nm}$ and the phonon modes at $351\ \text{cm}^{-1}$ and $417\ \text{cm}^{-1}$ at the right part of the monolayer (blue dotted area in Fig. 2d) indicate that it is monolayer WS_2 .



Note that the PL and phonon modes measured at the black dot in Fig. 2d are very similar to the results measured at the black dot in Fig. 2a. Considering the left and right parts are MoS₂ and WS₂ monolayers, respectively, the middle part is believed to be a vertical stacking part of the MoS₂ and WS₂ monolayers. Hence, we firmly conclude that our product is a MoS₂/WS₂ heterostructure composed of monolayer MoS₂ and monolayer WS₂.

The vertical stacking order of our MoS₂/WS₂ heterostructure is important for understanding the formation mechanism and finding future applications. To clarify the vertical stacking order, Raman spectroscopy is employed to study a typical MoS₂/WS₂ heterostructure (Fig. 3a). Our heterostructure shows an abnormal enhancement in the Raman intensity mapping using a 532 nm laser. Fig. 3b and c show the intensity mapping of the E_{2g} mode of MoS₂ (385 cm⁻¹) and E_{2g} and 2LA (M) modes of WS₂ (351 cm⁻¹), respectively. A clear enhanced intensity of 2LA(M) + E_{2g} modes is observed at the edges of the MoS₂/WS₂ heterostructure, which is not observed when using a 488 nm laser (Fig. 3e). The enhancement is related to the lattice deformation of the top WS₂ layer, as shown by the top-view and cross-section schematic illustration in Fig. 3f and g, respectively. According to the theoretical calculations and experimental reports,^{26–28} the lattice deformation of the monolayer WS₂ leads to an obvious and coincident adjustment of the electronic band structure. As a result, both the direct tran-

sition energy gap at the K point and the indirect transition energy gap at the M point of the Brillouin zone are narrowed and this leads to a reduction of the photon energy as well as a redshift of the peak positions of the A and B excitons. Meanwhile, as the 2LA(M) mode intensity is related to the resonance coupling between the B exciton and the 532 nm exciton laser, the shift of the B exciton leads to the enhancement of the 2LA(M) intensity.²⁵ Fig. 3h and i show the normalized PL and Raman spectra and their Lorentzian fitting curves, respectively, at the marked positions in Fig. 3c. Both PL and Raman spectra clearly show a red shift confirming our discussion of the lattice deformation. Besides, from the AFM phase analysis shown in Fig. 3i and j, no different phase angle is observed between the central triangular part and the outside hexagonal part, indicating that the two parts have the same surface. Otherwise, a clear contrast between the two different TMDs should occur.²⁹ Since the outside part has already been confirmed to be the WS₂ monolayer by Raman microscopy, the top layer of the heterostructure is the monolayer of WS₂. We also have compared these results with an artificial anti-pyramid heterostructure made from a monolayer WS₂ flake overlapping on a triangular MoS₂ flake (Fig. S5†). In comparison, we notice that both enhanced Raman intensity and AFM phase coherence exist in both the artificial heterostructure and our as-grown heterostructure, which indicates that our synthesized heterostructure has the same anti-pyramid structure as that of the artificial heterostructure. Hence, it is fully confirmed that the top layer of the MoS₂/WS₂ heterostructures is monolayer WS₂ and the bottom layer is MoS₂, implying that the heterostructure is an anti-pyramid structure. In the previous studies of multilayer homo- or heterostructures, the most common stacking order is the pyramid order, in which a smaller layer lies on a bigger layer forming a pyramid structure. Meanwhile, the opposite order, anti-pyramid order, is rarely reported. However, the anti-pyramid order is more important, in which part of the top layer ignores the borders of the bottom layer. Therefore, this method has the possibility to make the flake free-standing, contributing to the study of physical properties without the influence of the substrates. Besides, due to the existence of MoS₂ underneath, the central part of WS₂ will have a different dielectric environment from the outside part, which contacts with the SiO₂/Si substrate. This difference in the environment could lead to a dielectric induced band-gap renormalization effect, which could be used to form a lateral electronic junction with atomic scales in the WS₂ layer.³⁰ Such a diode-like junction has a strong asymmetric behaviour, which can provide a new approach for engineering 2D nanoelectronic devices.³¹

Second harmonic generation microscopy (SHG) is also used to demonstrate the symmetric stacking of the heterostructure. For monolayer TMDs such as MoS₂ and WS₂, transition metal atoms and sulphur atoms form an asymmetric “hexatomic” ring without any symmetry centre and lead to a six-petal polarization pattern, as shown in Fig. 4a (left). When monolayer MoS₂ and monolayer WS₂ stack with an angle of 60° (known as AA' stacking), the Mo atoms of MoS₂ are stacked on the S

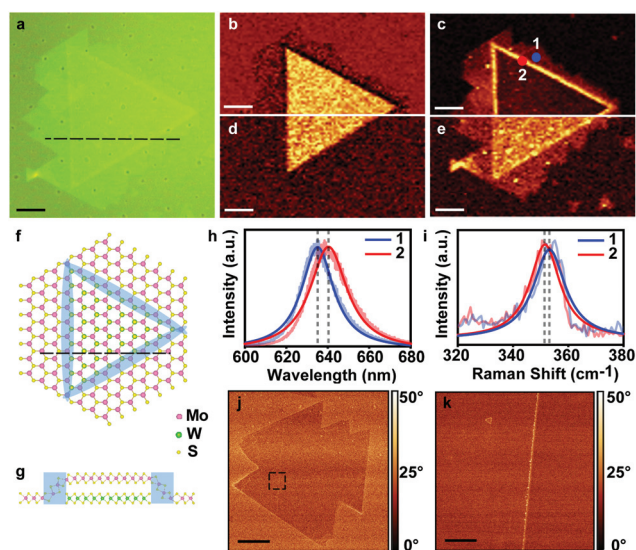


Fig. 3 The enhanced Raman intensity mapping and AFM phase images. (a) Optical image of the MoS₂/WS₂ heterostructure (scale bar: 10 μm). (b–e) Raman intensity mapping at 385 cm⁻¹ (b) and 351 cm⁻¹ (c) using a 532 nm laser as well as at 385 cm⁻¹ (d) and 351 cm⁻¹ (e) using a 488 nm laser (scale bar: 10 μm). (f) Atomic schematic illustration of the flake shown in (a). (g) Cross-section at the dashed line in (f). (h) Normalized PL spectra (light blue and red) and their Lorentzian fitting curves (blue and red) at the marked positions in (c). (i) Normalized Raman spectra at 2LA(M) + E_{2g} modes (light blue and red) and their Lorentzian fitting curves (blue and red) at the marked positions in (c). (j) AFM phase image of a typical heterostructure (scale bar: 10 μm). (k) High resolution AFM phase image of the dashed line area in (j) (scale bar: 1 μm).



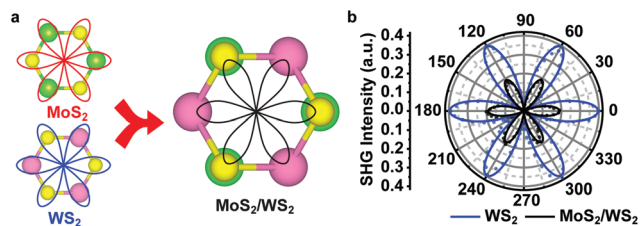


Fig. 4 Second harmonic generation of the MoS₂/WS₂ heterostructure. (a) Schematic illustration of second harmonic generation on MoS₂, WS₂ and MoS₂/WS₂. (b) Measured second harmonic generation intensity of the WS₂ part (blue) and MoS₂/WS₂ part (black).

atoms of WS₂ and the S atoms of MoS₂ are stacked on the W atoms of WS₂, leading to a slight increase of centrosymmetry and a decrease of SHG intensity, as shown in Fig. 4a (right). The measured SHG intensity of the heterostructure is shown in Fig. 4b, and it is clear that the SHG of the MoS₂/WS₂ heterostructure has the same orientation as that of WS₂ outside with a lower intensity, indicating that the MoS₂/WS₂ heterostructure has AA' stacking. Besides, the total SHG intensity in the heterostructure with an angle between the two layers, $I_h(\theta)$, can be expressed as:^{32,33}

$$I_h(\theta) \propto (\chi_1^{(2)})^2 + (\chi_2^{(2)})^2 + 2|\chi_1^{(2)}||\chi_2^{(2)}|\cos 3\theta, \quad (1)$$

where $I_h(\theta)$ stands for the SHG intensity in the stacking heterostructure; θ is the stacking angle, which is defined as the different angle between the two stacking layers and is 60° in AA' stacking; and $\chi^{(2)}$ is the second-order susceptibility. According to ref. 33 and 34, $\chi_{\text{MoS}_2}^{(2)} \approx 5.5 \text{ pm V}^{-1}$ and $\chi_{\text{WS}_2}^{(2)} \approx 17 \text{ pm V}^{-1}$ and the ratio of $I_{\text{WS}_2}/I_{\text{MoS}_2/\text{WS}_2}$ can be calculated to be ~ 2.19 , theoretically. The experimental SHG intensity is shown in Fig. 4b as a typical WS₂ SHG pattern with an $I_{\text{WS}_2}/I_{\text{MoS}_2/\text{WS}_2}$ ratio of ~ 2.11 . The agreement between theoretical and experimental results indicates that the symmetric stacking is AA' stacking.

After understanding the stacking of heterostructures, it is worth discussing how the formation of AA'-stacking anti-pyramid MoS₂/WS₂ heterostructures occurs. Control experiments, in which blank substrates are used instead of a substrate with Na₂MoO₄ or Na₂WO₄, respectively, are carried out at the same temperature of the heterostructure growth. Triangular MoS₂ flakes are formed on the blank substrate side-by-side with Na₂MoO₄, while no WS₂ flakes are observed in another blank substrate with Na₂WO₄, as shown in Fig. S6 and S7,[†] indicating the diffusion of Na₂MoO₄ and no diffusion of Na₂WO₄ at the growth temperature. Based on this and the different temperatures between MoS₂ and WS₂ growth, the heterostructure formation process is shown in the schematic illustration in Fig. 5 using the as-grown sample on a SiO₂/Si substrate as an example. At 700–800 °C, the melting Na₂MoO₄ transforms into a gas phase, diffuses to the substrate with Na₂WO₄, then dissolves in melting Na₂WO₄ and finally forms a liquid mixture. Later, with the dissolving of sulphur gas in the mixture droplets, MoS₂ starts to nucleate because it can

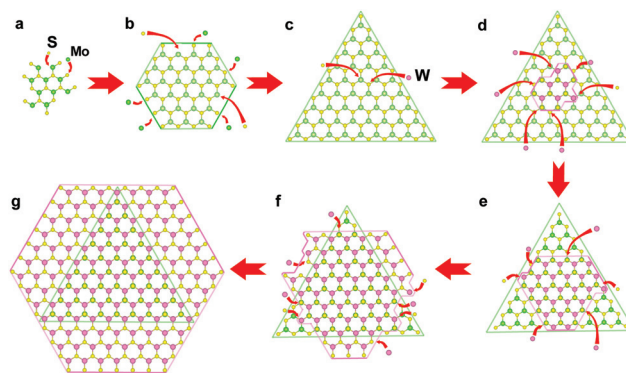


Fig. 5 Schematic illustration of the heterostructure epitaxy. (a) MoS₂ nucleation. (b) MoS₂ epitaxy. (c) MoS₂ triangular flake formation. (d) WS₂ nucleation on the MoS₂ layer. (e) WS₂ epitaxy on the MoS₂ layer. (f) WS₂ epitaxy outside the MoS₂ layer. (g) MoS₂/WS₂ heterostructure formation.

grow at a lower temperature, 700 °C and forms triangular MoS₂ flakes with S-terminal zig-zag edges. Then, the increasing temperature, 775 °C, leads to the nucleation and growth of WS₂. The selective growth of WS₂ on MoS₂ is caused by the larger energy barrier, which makes the absorbed WS₂ more immobile, facilitating nucleation and growth on MoS₂.³⁵ Based on the results of different theoretical calculations,^{36–38} it is seen that the most stable stacking of 2-layer TMDs is AA' stacking, which is the same in our heterostructures as demonstrated. In this structure, the top layer WS₂ stacks on the bottom layer MoS₂ with an orientation of difference 60°, which makes the Klein edges of WS₂ have the same epitaxial direction as that of the zig-zag edges of MoS₂. According to the previous research,³⁹ the growth kinetics of zig-zag edges and Klein edges are different and the Klein edges grow epitaxially much more faster. Because of the fast epitaxy, the Klein edges of WS₂ are able to extend without the limitation of MoS₂ boundaries and the anti-pyramid heterostructures are obtained finally. However, due to the lack of the tungsten precursor, WS₂ grows to a hexagonal structure with incomplete and uneven edges instead of the triangular structure. Nevertheless, here, the WS₂/MoS₂ heterostructure growth can be fully understood.

3. Conclusion

In conclusion, we report a facile and reproducible single-step CVD method for growing vertical bilayer MoS₂/WS₂ heterostructures with a lateral size up to $\sim 60 \mu\text{m}$. In this method, no additional chemical is required, and a relatively low growth temperature is needed, compared with other single-step MoS₂/WS₂ growth processes. In addition, the AA'-stacking with an anti-pyramid order is demonstrated by different techniques, including Raman, PL, AFM and SHG microscopy. The novel anti-pyramid order indicates that the VLS process has the potential for ignoring the borders of the substrate and forming free-standing TMDs. This method offers an opportu-



nity to synthesize heterostructures in a simple but efficient way. It could also possibly provide a potential strategy for devices based on free-standing 2D materials and a new approach for engineering 2D nanoelectronic devices.

4. Experimental

Na₂MoO₄ (99.9%, Sigma-Aldrich) and Na₂WO₄ (99.995%, Sigma-Aldrich) are used as purchased. Si substrates with 280 nm thick SiO₂ are purchased from University Wafer and cleaned with acetone and isopropanol, followed by O₂ plasma treatment before use. The growth occurs in a dual-zone tube furnace (MTI Corp.). Ar (99.995%, Agar) is chosen as the carrier gas.

Raman spectra are recorded using a WITec alpha300 RA+ system with a 532 nm or 488 nm continuous wave laser. SHG spectra are recorded using a femtosecond laser (Spectra-Physics, TOPAS) at a wavelength of 800 nm (repetition rate: 2 kHz). The pulse duration of the laser pulse is ~230 fs. AFM topographic and phase images are obtained using an AFM Dimension Icon (Bruker) with a tapping mode.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The authors thank the provision of facilities and technical support by Aalto University at the OtaNano-Micronova Nanofabrication Centre and OtaNano-Nanomicroscopy Center (Aalto-NMC). The authors also acknowledge the support from the Academy of Finland (Grant No. 314810, 333982, 336144 and 336818), the Academy of Finland Flagship Programme (Grant No. 320167, PREIN), the European Union's Horizon 2020 Research and Innovation Program (Grant No. 820423, S2QUIP) and the European Research Council (Grant No. 834742).

References

- Z. Sun, A. Martinez and F. Wang, *Nat. Photonics*, 2016, **10**, 227–238.
- A. C. Ferrari, F. Bonaccorso, V. Fal'Ko, K. S. Novoselov, S. Roche, P. Bøggild, S. Borini, F. H. Koppens, V. Palermo, N. Pugno, *et al.*, *Nanoscale*, 2015, **7**, 4598–4810.
- J. Shim, H.-Y. Park, D.-H. Kang, J.-O. Kim, S.-H. Jo, Y. Park and J.-H. Park, *Adv. Electron. Mater.*, 2017, **3**, 1600364.
- X. Li, L. Tao, Z. Chen, H. Fang, X. Li, X. Wang, J.-B. Xu and H. Zhu, *Appl. Phys. Rev.*, 2017, **4**, 021306.
- W. Zhang, Q. Wang, Y. Chen, Z. Wang and A. T. Wee, *2D Mater.*, 2016, **3**, 022001.
- J. Wang, I. Verzhbitskiy and G. Eda, *Adv. Mater.*, 2018, **30**, 1802687.
- P. Xiong, R. Ma, N. Sakai, L. Nurdijayanto and T. Sasaki, *ACS Energy Lett.*, 2018, **3**, 997–1005.
- P. Xiong, X. Zhang, H. Wan, S. Wang, Y. Zhao, J. Zhang, D. Zhou, W. Gao, R. Ma, T. Sasaki, *et al.*, *Nano Lett.*, 2019, **19**, 4518–4526.
- K. Novoselov, A. Mishchenko, A. Carvalho and A. C. Neto, *Science*, 2016, **353**, aac9439.
- Z. Cai, B. Liu, X. Zou and H.-M. Cheng, *Chem. Rev.*, 2018, **118**, 6091–6133.
- F. Withers, O. Del Pozo-Zamudio, A. Mishchenko, A. Rooney, A. Gholinia, K. Watanabe, T. Taniguchi, S. Haigh, A. Geim, A. Tartakovskii, *et al.*, *Nat. Mater.*, 2015, **14**, 301–306.
- R. Sundaram, M. Engel, A. Lombardo, R. Krupke, A. Ferrari, P. Avouris and M. Steiner, *Nano Lett.*, 2013, **13**, 1416–1421.
- E. Ponomarev, I. Gutiérrez-Lezama, N. Ubrig and A. F. Morpurgo, *Nano Lett.*, 2015, **15**, 8289–8294.
- Y. Zhang, Y. Yao, M. G. Sendeku, L. Yin, X. Zhan, F. Wang, Z. Wang and J. He, *Adv. Mater.*, 2019, **31**, 1901694.
- Q. Zhang, X. Xiao, R. Zhao, D. Lv, G. Xu, Z. Lu, L. Sun, S. Lin, X. Gao, J. Zhou, *et al.*, *Angew. Chem., Int. Ed.*, 2015, **54**, 8957–8960.
- X. Zhou, N. Zhou, C. Li, H. Song, Q. Zhang, X. Hu, L. Gan, H. Li, J. Lü, J. Luo, *et al.*, *2D Mater.*, 2017, **4**, 025048.
- Y. Gong, S. Lei, G. Ye, B. Li, Y. He, K. Keyshar, X. Zhang, Q. Wang, J. Lou, Z. Liu, *et al.*, *Nano Lett.*, 2015, **15**, 6135–6141.
- T. Yang, B. Zheng, Z. Wang, T. Xu, C. Pan, J. Zou, X. Zhang, Z. Qi, H. Liu, Y. Feng, *et al.*, *Nat. Commun.*, 2017, **8**, 1906.
- Y. Gong, J. Lin, X. Wang, G. Shi, S. Lei, Z. Lin, X. Zou, G. Ye, R. Vajtai, B. I. Yakobson, *et al.*, *Nat. Mater.*, 2014, **13**, 1135–1142.
- K. Bogaert, S. Liu, J. Chesin, D. Titow, S. Gradecak and S. Garaj, *Nano Lett.*, 2016, **16**, 5129–5134.
- S. Li, Y.-C. Lin, W. Zhao, J. Wu, Z. Wang, Z. Hu, Y. Shen, D.-M. Tang, J. Wang, Q. Zhang, *et al.*, *Nat. Mater.*, 2018, **17**, 535–542.
- S. Li, Y.-C. Lin, X.-Y. Liu, Z. Hu, J. Wu, H. Nakajima, S. Liu, T. Okazaki, W. Chen, T. Minari, *et al.*, *Nanoscale*, 2019, **11**, 16122–16129.
- J. Zhou, J. Lin, X. Huang, Y. Zhou, Y. Chen, J. Xia, H. Wang, Y. Xie, H. Yu, J. Lei, *et al.*, *Nature*, 2018, **556**, 355–359.
- H. Li, Q. Zhang, C. C. R. Yap, B. K. Tay, T. H. T. Edwin, A. Olivier and D. Baillargeat, *Adv. Funct. Mater.*, 2012, **22**, 1385–1390.
- F. Wang, I. A. Kinloch, D. Wolverson, R. Tenne, A. Zak, E. O'Connell, U. Bangert and R. J. Young, *2D Mater.*, 2016, **4**, 015007.
- J. Feng, X. Qian, C.-W. Huang and J. Li, *Nat. Photonics*, 2012, **6**, 866–872.
- C. Zhu, G. Wang, B. Liu, X. Marie, X. Qiao, X. Zhang, X. Wu, H. Fan, P. Tan, T. Amand, *et al.*, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2013, **88**, 121301.



- 28 M. Zeng, J. Liu, L. Zhou, R. G. Mendes, Y. Dong, M.-Y. Zhang, Z.-H. Cui, Z. Cai, Z. Zhang, D. Zhu, *et al.*, *Nat. Mater.*, 2020, **19**, 528–533.
- 29 C.-Y. Lin, X. Zhu, S.-H. Tsai, S.-P. Tsai, S. Lei, Y. Shi, L.-J. Li, S.-J. Huang, W.-F. Wu, W.-K. Yeh, *et al.*, *ACS Nano*, 2017, **11**, 11015–11023.
- 30 F. Giannazzo, *Nat. Electron.*, 2019, **2**, 54–55.
- 31 M. I. B. Utama, H. Kleemann, W. Zhao, C. S. Ong, H. Felipe, D. Y. Qiu, H. Cai, H. Li, R. Kou, S. Zhao, *et al.*, *Nat. Electron.*, 2019, **2**, 60–65.
- 32 W.-T. Hsu, Z.-A. Zhao, L.-J. Li, C.-H. Chen, M.-H. Chiu, P.-S. Chang, Y.-C. Chou and W.-H. Chang, *ACS Nano*, 2014, **8**, 2951–2958.
- 33 A. Autere, H. Jussila, Y. Dai, Y. Wang, H. Lipsanen and Z. Sun, *Adv. Mater.*, 2018, **30**, 1705963.
- 34 A. Autere, H. Jussila, A. Marini, J. Saavedra, Y. Dai, A. Säynätjoki, L. Karvonen, H. Yang, B. Amirsolaimani, R. A. Norwood, *et al.*, *Phys. Rev. B*, 2018, **98**, 115426.
- 35 J. Zhu, W. Li, R. Huang, L. Ma, H. Sun, J.-H. Choi, L. Zhang, Y. Cui and G. Zou, *J. Am. Chem. Soc.*, 2020, **142**, 16276–16284.
- 36 A. Blumberg, U. Keshet, I. Zaltsman and O. Hod, *J. Phys. Chem. Lett.*, 2012, **3**, 1936–1940.
- 37 Q. Liu, L. Li, Y. Li, Z. Gao, Z. Chen and J. Lu, *J. Phys. Chem. C*, 2012, **116**, 21556–21562.
- 38 T. Peng, G. Huai-Hong, Y. Teng and Z. Zhi-Dong, *Chin. Phys. B*, 2014, **23**, 106801.
- 39 D. Zhu, H. Shu, F. Jiang, D. Lv, V. Asokan, O. Omar, J. Yuan, Z. Zhang and C. Jin, *npj 2D Mater. Appl.*, 2017, **1**, 8.

